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# Rheology and coaxial extrusion of acellular and cell-laden hollow conduits of pristine kappa carrageenan and gold–kappa carrageenan nanocomposite hydrogels

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This study explores the use of kappa carrageenan ( $\kappa$ CG) hydrogels to create cell-laden hollow conduits via multi-material extrusion with a coaxial nozzle. We used an in-house setup with two syringe pumps to extrude both ionically crosslinked  $\kappa$ CG and gold– $\kappa$ CG nanocomposite ( $\kappa$ CG–AuNP) hydrogels for the conduit shell and a sacrificial core into a crosslinking bath. We tested various  $\kappa$ CG compositions and flow rates, finding that the inner filament diameters ranged from 625 to 700  $\mu$ m and the outer diameters from 1000 to 1030  $\mu$ m. Perfusion assays showed that 2% (w/v)  $\kappa$ CG–AuNP hydrogels extruded at higher flow rates maintained integrity, allowing for uniform perfusion in longer filaments. Rheological analysis indicated that gold nanoparticles (AuNPs) reduced both the storage moduli and the viscosity of the hydrogels, and the crosslinking bath improved storage moduli post-crosslinking. The 2% (w/v)  $\kappa$ CG–AuNP hydrogels demonstrated delayed breakage during high-flow rate extrusion, facilitating handling during perfusion. Importantly, the addition of A549 cells did not impact the rheological properties or pinch-off dynamics, and cell viability exceeding 60% was noted within the conduit walls. Thus, the 2% (w/v)  $\kappa$ CG–AuNP hydrogel shows promise for rapid fabrication of hollow conduits for *in vitro* modelling of tubular biological structures.

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## 1. Introduction

The overarching goal of tissue engineering is to fabricate scaffolds that recapitulate the complex microarchitecture of native tissues to be functional. This includes designing vascular artificial tissues that can mass transport fluids like blood, oxygen, and nutrients, similarly to physiological tissues.<sup>1</sup> Coaxial extrusion is a one-step strategy for creating hollow or multi-material tubular constructs by extruding two distinct biopolymers simultaneously through a coaxial nozzle.<sup>2</sup> Extrusion can either be pneumatic or piston-facilitated. Piston extrusions are generally achieved using syringe pumps.<sup>3</sup> One of the commonly employed methods for the fabrication of artificial vascular conduits is the multi-material extrusion of a cross-linkable

hydrogel along with a sacrificial core that can be dissolved or removed by physical methods.<sup>4</sup> There is a rapidly growing interest in using Pluronic F-127, a synthetic biopolymer, as a sacrificial core in coaxial extrusion. Its excellent printability, shear-thinning behaviour, and sol-like characteristics at 4 °C make Pluronic F-127 ideal for creating easily tunable hollow channels with varying inner lumen diameters. This can be achieved by adjusting the compositions of the shell biopolymer ink and their flow rates.<sup>5</sup>

The existing literature on coaxial extrusion focuses on alginate as the preferred biomaterial for fabricating hollow tubes, owing to its excellent mechanical properties and rapid ionic crosslinking using  $\text{CaCl}_2$ .<sup>6–8</sup> Kappa carrageenan ( $\kappa$ CG), like alginate, is an anionic polysaccharide composed of alternating galactose and 3,6-anhydrogalactose units. It has recently been repurposed from a food additive to the biomedical field by modifying it to form hydrogels, fibres, scaffolds, *etc.*, due to its biocompatibility, biodegradability, and thermoresponsive gelling capacity.<sup>9</sup>  $\kappa$ CG has been previously explored for both 3D printing and bioprinting by our group and by Marques and colleagues, respectively, who demonstrated high cell viability and reorganisation in 3D-printed scaffolds.<sup>10,11</sup> Previous work from our group has also demonstrated the fabrication of  $\kappa$ CG

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hydrogel beads and emulsion gels.<sup>12,13</sup> However, there is a paucity of information on fabricating  $\kappa$ CG-based hollow perfusable conduits. Therefore, based on the existing literature on  $\kappa$ CG and alginate, we hypothesize that since  $\kappa$ CG is also a negatively charged biopolymer similar to alginate, capable of crosslinking with positively charged ions, it should also be able to form self-supporting hollow tubes when crosslinked with potassium chloride (KCl). Although  $\kappa$ CG can also be ionically crosslinked with  $\text{Ca}^{2+}$ , different ions induce different gelation behaviours in  $\kappa$ CG.  $\text{K}^+$  ions are known to form stronger gels of  $\kappa$ CG compared to  $\text{Na}^+$  or  $\text{Ca}^{2+}$  ions.<sup>14</sup> In contrast,  $\text{Ca}^{2+}$  ions yield  $\kappa$ CG gels of lower stiffness, reducing the elastic modulus with increasing  $\text{Ca}^{2+}$  concentration.<sup>15</sup> Moreover, Lim and colleagues have reported no toxicity on NIH 3T3 cells in methacrylated  $\kappa$ CG hydrogels crosslinked with  $\text{K}^+$  ions during extrusion from a coaxial nozzle for the fabrication of cell-laden scaffolds.<sup>16</sup> This precedent established KCl as a safe ionic crosslinker for the fabrication of cell-laden hollow fibers.

Nevertheless, it is important to note that the mechanism of ionic crosslinking and gelation differs between alginate and  $\kappa$ CG. The  $\kappa$ CG hydrogel undergoes crosslinking in the presence of counterions *via* helix aggregation driven by ionic interactions. Monovalent cations like  $\text{K}^+$  neutralize the negatively charged sulphate groups on their polymeric chains, inducing helical secondary structure formation, resulting in gelation.<sup>17,18</sup> In contrast, alginate undergoes ionic interactions with  $\text{Ca}^{2+}$ , forming an egg-box-like structure that ultimately leads to dimer formation. These dimers further undergo multimerization, leading to gelation.<sup>19</sup> It is important to emphasize that the main objective of this study is not to prove that the  $\kappa$ CG hydrogel is superior to alginate. Instead, the goal is to investigate whether this relatively underexplored biopolymer can be optimised as a suitable material for coaxial extrusion, capable of supporting living cells during and after the extrusion process.

In addition to the above,  $\kappa$ CG, as a sulfated polysaccharide, can function as a polyelectrolyte, self-assembling during nanoparticle formation and providing stability to the nanoparticles within their soft matrices.<sup>20</sup> Previous reports have shown that the pendant hydroxyl groups of  $\kappa$ CG can be a capping and stabilizing agent for metallic nanoparticles.<sup>21,22</sup> Metallic gold nanoparticles (AuNPs) in polymer matrices confer additional biological and mechanical properties. Depending on their shape and size, AuNPs can promote cell adherence in scaffolds, induce cell differentiation and maturation, improve cell-cell coupling by imparting electrical conductivity to the scaffold system, and modulate scaffold's mechanical properties.<sup>23</sup> The presence of AuNPs in hydrogels can increase the stiffness of soft natural biopolymer-based matrices such as GelMA while simultaneously improving the shear thinning behaviour, making these nanocomposite hydrogels ideal for optimal cell attachment, adhesion, and bioprinting.<sup>24</sup> There are also many reports on *in situ* synthesized gold-chitosan nanocomposite hydrogels as candidates for biomedical applications, including tissue engineering and drug delivery.<sup>25,26</sup>

$\kappa$ CG has also been used as a reducing and stabilising agent for AuNPs, for instance, in the degradation of azo dyes.<sup>21</sup> There have been limited studies on gold- $\kappa$ CG nanocomposites exploring their biomedical applications, including antitumor activity and wound-healing properties.<sup>27–29</sup> Notably, a previous report utilised low concentrations of  $\kappa$ CG, ranging from 0.5 mg mL<sup>-1</sup> to 0.08 mg mL<sup>-1</sup>, to synthesize AuNPs but was unable to form hydrogels. The study found that AuNPs in  $\kappa$ CG did not induce significant cytotoxicity in the A549 lung cancer cell line. The AuNPs could also quench reactive oxygen species (ROS), exhibiting antioxidative behaviour.<sup>30</sup> Thus, the formation of *in situ* AuNPs and their impact on cell viability in a true hydrogel matrix comprising a higher concentration of  $\kappa$ CG, suitable for extrusion and formation of self-supporting hollow tubes, have not been addressed. Rheological analysis of a  $\kappa$ CG and locust bean gum composite hydrogel with *in situ* synthesized AuNPs revealed a decrease in viscosity with an increase in the concentration of  $\text{HAuCl}_4$  added to  $\kappa$ CG, thereby imparting superior shear thinning properties.<sup>31</sup> However, efforts remain limited in exploring gold- $\kappa$ CG nanocomposite hydrogels for bioink printability, rheological characterization, cell-laden 3D-printed constructs, and vascular constructs. We envision that AuNPs in  $\kappa$ CG hydrogels will modulate their rheological behaviour, enhance cell viability in a 3D microenvironment, and potentially improve the biological properties of the hollow conduits.

During coaxial extrusion, the material mainly experiences shear flow within the nozzle. Once the material exits the nozzle, it transitions to extensional flow as it is deposited onto the build surface. Therefore, understanding the extensional properties of complex fluids is essential not only for processing operations, such as jetting, spinning, and coating, where extensional modes of deformation are dominant, but also for analysing bio-inks used in extrusion and printing. Materials often exhibit significantly different behaviours in extensional flow compared to shear flow.<sup>32</sup> Devices like the Capillary Breakup Extensional Rheometer (CaBER) provide valuable insights by observing the thinning and breaking of a fluid filament.<sup>33</sup> In polymer solutions, extensional flows can cause flexible macromolecules to uncoil, resulting in either extensional thinning or thickening. This phenomenon significantly alters the fluid's flow response.<sup>34</sup> Slight variations in factors such as molecular weight, structure, temperature, and concentration can affect how a sample reacts under extension, making extensional rheological characterization a valuable tool for quality control. While traditional methods like CaBER are commonly used, alternative techniques, such as the dripping-on-substrate method, can also provide extensional characterization. This method involves slowly dispensing the sample from a nozzle onto a substrate surface to form a liquid bridge, which creates a filament that eventually breaks due to capillary forces.<sup>35</sup> Using a high-speed camera to record this process yields valuable data on the fluid's extensional properties, providing a practical and informative approach to rheological analysis.

A previous work from our lab demonstrated successful printing of complex shapes using 1% (w/v)  $\kappa$ CG crosslinked



with 25 mM KCl, achieving excellent biocompatibility (98%) of A549 lung cancer epithelial cells seeded on the hydrogel surface after 24 hours.<sup>36</sup> Previous work conducted by Sasikumar C. and group utilised lung epithelial cells, such as L-132 and A549, to prepare bioinks of mucin and hyaluronic acid for 3D bioprinting with the aim of fabricating printed lung tissues. Notably, κCG-based hydrogels with *in situ* synthesized AuNPs having an average particle size of  $14.3 \pm 2.1$  nm were reported to exhibit no cytotoxicity on A549 cells.<sup>30</sup> A549 cells have also been widely used for 3D bioprinting to fabricate cell-laden porous grid scaffolds as lung cancer models.<sup>37,38</sup>

However, to date, no report has employed A549 cells to coaxially bioprint hollow tubular structures as potential lung cancer models. Therefore, in this study, we demonstrate the fabrication of hollow tubular conduits from an acellular gold-κCG (κCG-AuNP) nanocomposite, pristine κCG, and A549 lung carcinoma cells encapsulated in both κCG-AuNP and κCG hydrogels. This was achieved using coaxial extrusion followed by ionic crosslinking in a post-extrusion bath, utilising a sacrificial core of Pluronic F-127. We monitored changes in channel diameters as a function of the flow rates of the sheath κCG and κCG-AuNP hydrogel inks and the uniform formation of hollow conduits, using microscopic imaging and perfusion assays. Rheological investigations and pinch-off dynamics results indicated that both the weight percentage (wt%) of κCG and the presence of AuNPs significantly influenced the storage modulus, viscosity, and filament breakage time of the inks. Additionally, we studied the effects of encapsulating lung carcinoma cells at a concentration of  $2 \times 10^6$  cells per mL in κCG and κCG-AuNP hydrogels on their rheological properties and pinch-off dynamics. Our findings suggest that the presence of AuNPs did not introduce additional toxicity into the encapsulated lung carcinoma cells, enabling the successful formation of biocompatible, cell-laden κCG-AuNP hydrogel hollow conduits.

## 2. Materials and methods

### 2.1. Sample preparation

To prepare a pristine κCG crosslinked hydrogel, 150 mg, 200 mg, and 225 mg of kappa carrageenan powder (Sigma-Aldrich) were mixed with 10 mL of 10 mM potassium chloride (KCl). This yielded hydrogels with different concentrations of κCG:  $15 \text{ mg mL}^{-1}$  (1.5% w/v),  $20 \text{ mg mL}^{-1}$  (2% w/v), and  $22.5 \text{ mg mL}^{-1}$  (2.25% w/v), respectively. The mixture was stirred in a water bath maintained at 70 °C at 800 rpm for 2 hours, until the κCG powder was completely dissolved. The pristine κCG inks will be referred to as 1.5κCG, 2κCG, and 2.25κCG throughout the manuscript.

To *in situ* synthesize AuNPs in κCG hydrogels, 150 mg, 200 mg, and 225 mg of κCG powder were dissolved in 10 mL of a 10 mM potassium chloride (KCl) solution. This preparation yielded hydrogels with different concentrations of κCG:  $15 \text{ mg mL}^{-1}$  (1.5% w/v),  $20 \text{ mg mL}^{-1}$  (2% w/v), and  $22.5 \text{ mg mL}^{-1}$  (2.25% w/v), respectively. The aqueous mixture was heated to 70 °C in a water bath while continuously stirring at 800 rpm.

Once the water bath reached 70 °C, 1 mL of a 1.52 mM gold(III) chloride trihydrate ( $\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$ ) solution (Sigma-Aldrich) was added to the mixture, which was then stirred for 24 hours. From this point onward, the *in situ* synthesized gold-κCG nanocomposite inks will be referred to as 1.5κCG-AuNP, 2κCG-AuNP, and 2.25κCG-AuNP throughout the manuscript.

### 2.2. Characterization of *in situ* synthesized AuNPs in κCG hydrogels

The synthesis of AuNPs within hydrogel matrices containing different weight percentages of κCG was confirmed through UV-visible spectrophotometry after a 24-hour reaction period. The size of the AuNPs was evaluated using atomic force microscopy (AFM). Samples were prepared as described previously and diluted to  $1 \text{ mg mL}^{-1}$  before drop-casting onto mica sheets. The prepared slides were allowed to dry in a desiccator for 24 hours before AFM imaging (Bruker NanoWizard Sense) in tapping mode. Nanoscope analysis software was utilised to process and analyse the images obtained. The hydrodynamic size of the AuNPs was also determined by dynamic light scattering (DLS) using a Brookhaven NanoBrook instrument. Similar to AFM, the samples were first diluted to  $1 \text{ mg mL}^{-1}$  in filtered deionized water before measurement.

The functional groups in κCG-AuNP and κCG hydrogels were further characterized by Fourier transform infrared (FTIR) spectroscopy using a Bruker Invenio-S FTIR spectrometer. The ATR probe was pressed directly onto the lyophilized samples placed on the crystal stage and scanned over the spectral range of  $400\text{--}4000 \text{ cm}^{-1}$ .

### 2.3. Coaxial extrusion

A pre-built coaxial nozzle from Rame-Hart Instrument Co. (USA) was used, consisting of an inner needle with an inner diameter (I.D.) of 584 μm (20G) and an outer needle with an inner diameter of 1190 μm (16G). This nozzle was connected to two 5 mL syringes: one containing a 30% (w/v) solution of Pluronic F-127 and the other containing either pristine κCG or κCG-AuNPs, as illustrated in Fig. 1. The samples were loaded into the syringe barrel while pristine κCG and κCG-AuNPs were

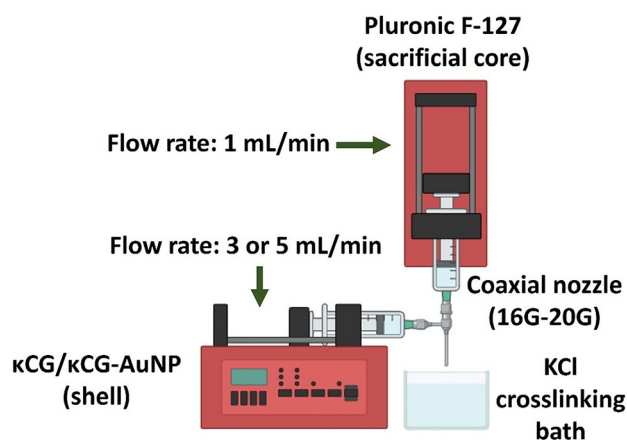


Fig. 1 Coaxial extrusion set-up. Figure created using Biorender.com.



in a sol state at elevated temperatures. This was followed by an incubation time of 10–12 minutes, during which the hydrogel temperature was allowed to decrease to 37 °C, with continuous monitoring by a temperature probe. The extrusion was initiated only after the hydrogel attained a temperature of 37 °C.

Pluronic F-127 was extruded through the inner needle using a syringe pump to create the core of the filaments at a fixed flow rate of 1 mL min<sup>-1</sup>. Simultaneously, pristine κCG or κCG–AuNP hydrogels were extruded through the outer needle using a syringe pump at different flow rates of 0.5–5 mL min<sup>-1</sup>, immediately after 37 °C was attained. The extruded filaments were deposited directly into a Petri dish containing 100 mM KCl solution, which served as a post-extrusion crosslinking bath (see the Supplementary Movie). The nozzle was positioned 4–5 mm above the crosslinking bath. The filaments immersed in the crosslinking bath were incubated at 4 °C for half an hour to remove the sacrificial core of Pluronic F-127. After incubation, the 100 mM KCl solution was discarded, and the cross-sections of the hollow filaments were imaged using a stereomicroscope. The circularity index of the lumens of the hollow channels was determined using the following equation:

$$c = \frac{4\pi A}{L^2} \quad (1)$$

where  $c$  is the circularity index,  $A$  is the cross-sectional area, and  $L$  is the perimeter of the hollow lumen. A perfect circle has a circularity index of 1; an elongated shape has a circularity index of 0.<sup>39</sup>

#### 2.4. Perfusion assay

The consistency of hollow channel fabrication after coaxial extrusion was evaluated using a perfusion assay. A green food dye was injected through the hollow lumen of the filaments using a 5 mL syringe equipped with either a 23G nozzle (inner diameter = 330 μm) or a 27G nozzle (inner diameter = 200 μm).

#### 2.5. Rheology

The rheological properties of the hydrogels forming the shell of the hollow filaments were evaluated 24 hours after preparation using a stress-controlled modular compact rheometer (Anton Paar MCR302). All rheological measurements of the pristine κCG and κCG–AuNP hydrogels were performed at an extrusion temperature of 37 °C. The samples were equilibrated at 37 °C for 10 minutes before conducting any rheological tests.

The rheological characterization of the bath-crosslinked pristine κCG and κCG–AuNP hydrogels was conducted at 25 °C on the same day as sample preparation. This involved casting and punching out hydrogel discs with a thickness of 2 mm and a diameter of 25 mm, which were then incubated in a 100 mM KCl bath at 4 °C for half an hour.

To determine the linear viscoelastic region (LVR) of both κCG and κCG–AuNP hydrogels, a large amplitude oscillatory strain sweep was performed, ranging from 0.01% to 1200% strain at a constant angular frequency of 6.28 rad s<sup>-1</sup>. The dynamic viscoelastic moduli were measured *via* a small-amplitude oscillatory frequency sweep spanning 0 to

100 rad s<sup>-1</sup> at a fixed strain within the LVR region. Additionally, the phase angle was determined from the frequency sweeps using the following equation:

$$\tan \delta = \frac{G''}{G'} \quad (2)$$

where  $G''$  is the loss modulus and  $G'$  is the storage modulus.

The flow curves of the hydrogels were obtained by increasing the shear rate logarithmically from 0.01 to 100 s<sup>-1</sup> to analyse their flow behaviour. The power law model was used to characterize the shear-thinning properties of the hydrogels. The values of  $n$  and  $K$  were calculated from the shear rate–viscosity curves by fitting them to the following power law equation:

$$\eta = K\dot{\gamma}^{n-1} \quad (3)$$

where  $\eta$  is the viscosity and  $\dot{\gamma}$  is the shear rate.

The experiments were conducted using a parallel plate geometry with a diameter of 50 mm and a gap of 0.5 mm. The bath-crosslinked hydrogel discs were tested using a parallel plate with a diameter of 25 mm and a gap of 2 mm.

#### 2.6. Mechanical characterization

The mechanical properties of the hydrogels forming the shell material of the hollow filaments were quantified using a universal testing machine (UTM) under unconfined compression. Pristine κ-carrageenan (κCG) and κCG–AuNP hydrogels were tested in the as-prepared state and after ionic crosslinking by immersion in a 100 mM KCl bath at 4 °C for 30 min. Cylindrical specimens ( $n = 3$  per condition; height  $\approx 16$  mm, diameter  $\approx 15$  mm) were placed centrally on a 70 mm compression plate and compressed using a parallel 70 mm upper plate at a crosshead speed of 1 mm min<sup>-1</sup> until failure (see the Supplementary Movie). Engineering stress–strain curves were calculated from the recorded force–displacement data, and the compressive modulus was determined from the slope of the initial linear region of the stress–strain response.

#### 2.7. Filament pinch-off dynamics experiments

A dripping-onto-substrate rheometry setup was constructed in the lab to measure the pinch-off dynamics and extensional properties of κCG and κCG–AuNP hydrogels at 37 °C. A glass slide was mounted horizontally, positioned 10 mm below a vertical syringe. The syringe was connected to a syringe pump, which extruded samples at flow rates of 3 mL min<sup>-1</sup> and 5 mL min<sup>-1</sup>, consistent with the rates used during coaxial extrusion, until the sample touched the substrate and formed a hydrogel filament. The pumping was then switched off.

An LED light source and a high-speed camera operating at 5000 frames per second recorded the thinning of the filament diameter ( $D(t)$ ) over time. These videos were analyzed using a custom MATLAB code to track the evolution of the filament diameter ( $D(t)$ ) with each frame. Up to 10 trials were conducted to ensure repeatability, and the average decay profiles of the filament diameter ( $D(t)$ ) were used for further analysis.

Surface tension ( $\sigma$ ) measurements of the κCG and κCG–AuNP hydrogels were crucial for analysing extensional



parameters. These measurements were performed using the method proposed by Adrian Daerr, as other methods, such as tensiometer measurements, posed challenges and proved unreliable for hydrogels.<sup>40</sup>

The rheological characterization of complex fluids often requires determining a constitutive parameter known as the characteristic relaxation time ( $\lambda$ ). This parameter quantifies the amount of elastic energy present in the sample. Eqn (4) describes the exponential decay of  $D(t)$ , providing valuable insights where  $\lambda$  is the characteristic extensional relaxation time,  $t$  is the observation time, and  $D_0$  is the internal diameter of the Luer lock of the syringe barrel (1880  $\mu\text{m}$ ). When stress is applied in an extensional mode, the viscosity coefficient is referred to as extensional viscosity ( $\eta_e$ ). The extensional strain rate is derived from the filament diameter and its rate of change over time, as shown in eqn (6) and (7).<sup>41–43</sup>

$$\frac{D(t)}{D_0} = \left[ e^{-t/3\lambda} \right] \quad (4)$$

$$\frac{d(D(t))}{dt} = \frac{D(t)_n - D(t)_{n-1}}{(t)_n - (t)_{n-1}} \quad (5)$$

$$\dot{\epsilon} = \frac{-2}{D(t)} \times \frac{d(D(t))}{dt} \quad (6)$$

$$\eta_e = \frac{-\sigma}{\frac{d(D(t))}{dt}} \quad (7)$$

## 2.8. Cell culture and cell-laden coaxial extrusion

Lung carcinoma cells (A549) were cultured at 37 °C and 5% CO<sub>2</sub> in Dulbecco's Modified Eagle Medium (DMEM) (Gibco), supplemented with 10% fetal bovine serum (FBS) and 1% penicillin–streptomycin. For cell-laden coaxial extrusion, the same setup was used under sterile conditions. The cells were trypsinized with 0.25% Trypsin-EDTA (Gibco) and suspended at a concentration of  $2 \times 10^6$  cells per mL in pristine  $\kappa\text{CG}$  and  $\kappa\text{CG}$ -AuNP hydrogels. The flow rate of the cell-laden  $\kappa\text{CG}$  and  $\kappa\text{CG}$ -AuNP hydrogels was maintained at 5 mL min<sup>-1</sup>, and they were deposited in a confocal dish containing a crosslinking bath of 100 mM KCl.

After post-extrusion crosslinking in 100 mM KCl at 4 °C, the filaments were washed with 1 $\times$  phosphate-buffered saline (PBS) and incubated in complete DMEM media at 37 °C for 24 hours. The extrusion process was carried out inside a laminar flow cabinet, and both the shell and sacrificial core hydrogel samples were prepared using autoclaved deionized water (DI) under sterile conditions. Additionally, the coaxial nozzle was autoclaved for half an hour before the cell-laden coaxial extrusion experiments.

## 2.9. Cell viability assay

After incubating the cell-laden hollow filaments in cell culture media at 37 °C for 24 hours, a live–dead assay was conducted to assess the viability of the seeded A549 cells. The filaments were

first washed with a 1 $\times$  PBS buffer following the gentle removal of the media. Calcein AM (2.25  $\mu\text{M}$ ) and propidium iodide (3.75  $\mu\text{M}$ ) were diluted in serum-free media to prepare the staining solution. Next, 1 mL of this staining solution was gently added to the confocal dish containing the hollow filaments and incubated at 37 °C for 15 minutes. Three representative images were captured using a Leica TCS SP8 confocal microscope at 10 $\times$  magnification. The cell viability was calculated using:

$$\text{Viability} = \frac{\text{Live cells}}{\text{Live} + \text{Dead cells}} \times 100\% \quad (8)$$

## 3. Results and discussion

### 3.1. Characterization of AuNPs

The successful synthesis of gold AuNPs within the polymer matrices of  $\kappa\text{CG}$  hydrogel was investigated using spectrophotometric analysis, as shown in Fig. 2(a). The maximum absorption values observed – approximately 532 nm, 534 nm, and 531 nm for 1.5 $\kappa\text{CG}$ -AuNP, 2 $\kappa\text{CG}$ -AuNP, and 2.25 $\kappa\text{CG}$ -AuNP, respectively – confirm the reduction of Au<sup>3+</sup> to Au<sup>0</sup> by the hydroxyl groups of  $\kappa\text{CG}$  chains. Additionally, a visible color change from a transparent  $\kappa\text{CG}$  solution to pink was noted during the formation of AuNPs, attributed to the surface plasmon resonance (SPR) phenomenon.

The characteristic peaks obtained from the height profiles of Fig. 2(b)–(d) show that the AuNPs synthesized within the  $\kappa\text{CG}$  matrix appear to be mainly spherical, along with the presence of aggregates that result in some irregular shapes. Also, the UV visible absorption spectra of all  $\kappa\text{CG}$ -AuNP compositions are all approximately close to 540 nm, which is the standard absorption wavelength of spherical AuNPs (Fig. 2(a)).<sup>44</sup> According to Bellotti *et al.*, when using AFM to measure the diameter of spherical AuNPs, the most accurate method of determining the diameter is by measuring the height ( $z$ -axis) obtained from the cross-section profiles, rather than the  $x$ - $y$  lateral dimensions. This is primarily because the AFM tip has a finite shape and size, which may lead to a broadening effect known as tip dilation. This phenomenon causes the nanoparticles to appear wider than their true diameter in the  $x$ - $y$  plane. In contrast, the height or  $z$ -axis remains unaffected by the shape and size of the tip, making the height measurements of the cross-section profiles a more reliable approach to measure AuNP diameters.<sup>45</sup> Hence, following the above approach, we considered the true diameter of spherical AuNPs to be the top height obtained from the cross-section profiles, thereby avoiding any error caused by tip dilation. The average height of the AuNPs obtained from the images ranged from 10 to 35 nm. SAXS analysis of  $\kappa\text{CG}$ -AuNP hydrogels further supported the AFM results, indicating that the AuNPs were well-dispersed within the gel network, with an average diameter of 15–16 nm across the different  $\kappa\text{CG}$  compositions (Fig. S3(c)).

The absorption peak is significantly influenced by the size and shape of the formed nanoparticles.<sup>46</sup> The presence of AuNP aggregates, or an increase in their size, has been shown to cause a shift or broadening of the maximum absorption peak.<sup>47</sup>



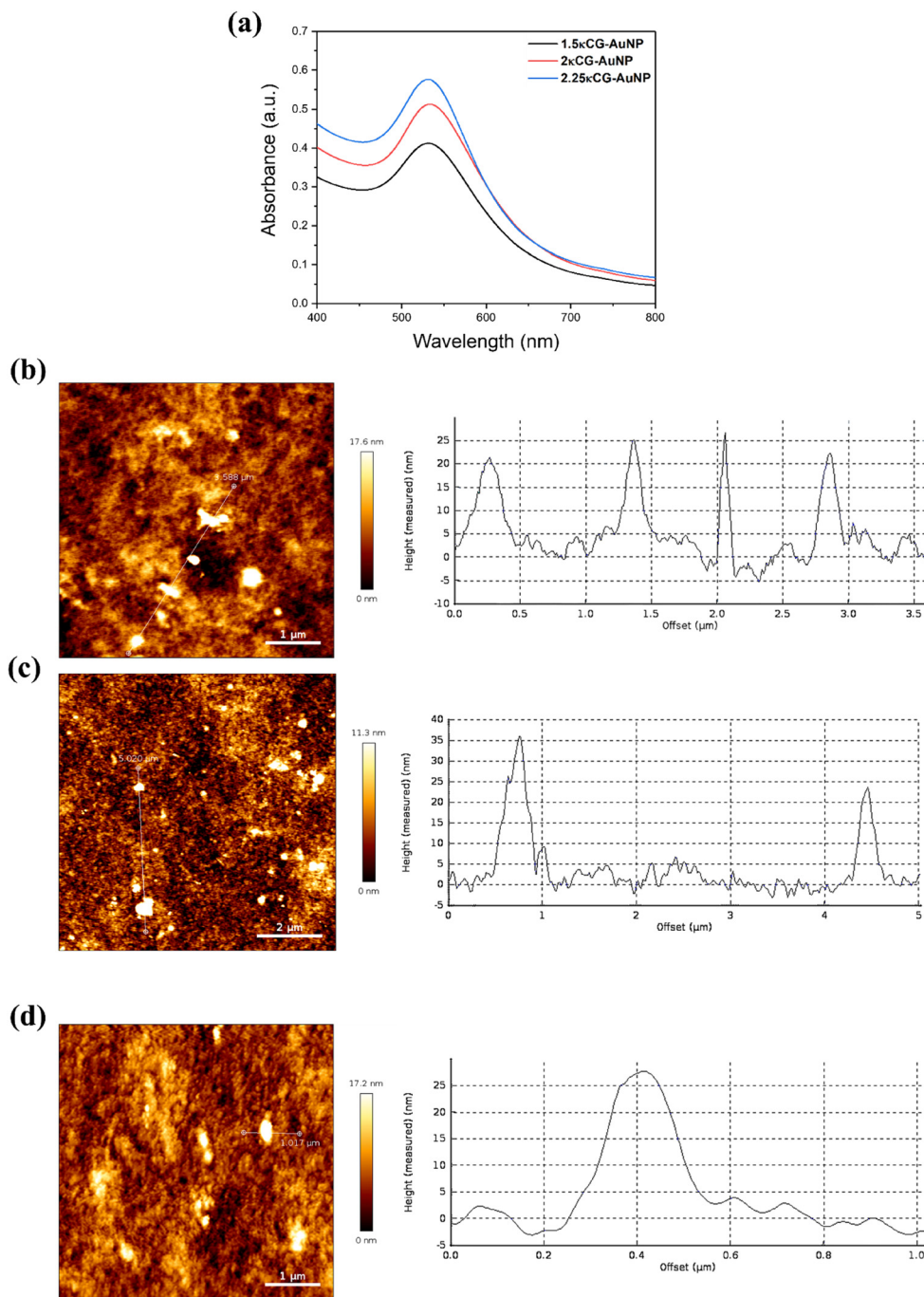


Fig. 2 (a) UV-visible spectra of AuNPs synthesized in various compositions of  $\kappa$ CG for 24 hours in a water bath at 70 °C; (b), (c) and (d) AFM image and height profiles of 1.5 $\kappa$ CG-AuNP, 2 $\kappa$ CG-AuNP and 2.25 $\kappa$ CG-AuNP hydrogels, respectively.

As illustrated in Fig. 2(a), all three compositions of  $\kappa$ CG-AuNP hydrogels display similar absorption peaks and maxima. Therefore, it is likely that the shape and size of the AuNPs synthesized in these three different  $\kappa$ CG-AuNP hydrogel compositions are also similar, as confirmed by the cross-section height profiles shown in Fig. 2(b)–(d).

Moreover, the number-weighted size distribution profiles obtained by DLS also confirm that the hydrodynamic diameter of the *in situ* synthesized AuNPs is similar across all  $\kappa$ CG compositions (Fig. S1). The average hydrodynamic diameter

of all  $\kappa$ CG-AuNP compositions is approximately 758 nm. It is known that the light scattering intensity is proportional to the sixth power of particle radius, due to which DLS is more sensitive to bigger particles.<sup>48</sup> Hence, we have reported the number-weighted size distribution of AuNPs in  $\kappa$ CG. The size derived from DLS was found to be larger than that observed by AFM, since the AuNPs are associated with their respective  $\kappa$ CG polymer compositions in the swollen state, resulting in complex shapes and increased roughness. Furthermore, various interactions like H-bonding, van der Waals and  $\pi$ - $\pi$



interactions may lead to larger coordination spheres of polymer-capped AuNPs.<sup>21</sup> The intensities of the maximum absorption peaks for the three compositions of the  $\kappa$ CG–AuNP hydrogels in Fig. 2(a) suggest that the amount of AuNPs synthesized within the hydrogels increases with a higher weight percentage of  $\kappa$ CG.

Wan *et al.* previously synthesized *in situ* AuNPs in a solution containing 18 mg mL<sup>-1</sup> (1.8% w/v)  $\kappa$ CG. The optimal reaction conditions included a temperature of 80 °C, a reaction time of 30 minutes, and a concentration of 0.4 mM HAuCl<sub>4</sub>. Transmission electron microscopy (TEM) analyses revealed that the mean size of the nanoparticles ranged from 13.5 nm to 18.6 nm.<sup>21</sup> Álvarez-Viñas and colleagues demonstrated the creation of larger, pseudo-spherical nanoparticles (ranging from 14 nm to 36 nm) in a 0.25 mg mL<sup>-1</sup> (0.025% w/v) carrageenan solution, as the concentration of chloroauric acid increased.<sup>49</sup> In addition, Débora A. de Almeida *et al.* synthesized  $\kappa$ CG-capped AuNPs with a hydrodynamic size of 563 ± 7 nm.<sup>50</sup> It is well-documented that the shape and size of AuNPs formed *in situ* depend on the reaction conditions; however, the size range of the AuNPs obtained in our samples aligns with the existing literature. Since the size range of AuNPs synthesized remained in the same range irrespective of the concentration of  $\kappa$ CG explored, any difference in rheological and mechanical properties due to variation in size of AuNPs will not be apparent, allowing us to isolate the concentration of  $\kappa$ CG as the primary driver of the mechanical stability of the hollow tubes.

The *in situ* synthesis of AuNPs was further confirmed by the subtle shifts in the characteristic O–H stretching bands compared to the pristine  $\kappa$ CG hydrogels in the FTIR spectra of all compositions of lyophilized  $\kappa$ CG hydrogels (Fig. S2). All the compositions of  $\kappa$ CG–AuNP hydrogels displayed a shift of the –O–H stretching band by 2–12 cm<sup>-1</sup> compared to the O–H bands of the pristine  $\kappa$ CG hydrogels. These shifts likely arise from the involvement of hydroxyl groups in coordinating Au<sup>3+</sup> ions, thereby contributing to the nucleation and stabilization of AuNPs within the  $\kappa$ CG matrix.<sup>51</sup> Concomitantly, the perturbation of the O–H stretching region suggests a partial disruption of intermolecular hydrogen bonding between  $\kappa$ CG chains, likely resulting from competitive interactions of  $\kappa$ CG hydroxyl groups with the AuNP surface. Secondary bands show characteristic features of  $\kappa$ CG at 2904–2910 cm<sup>-1</sup>, 1636–1640 cm<sup>-1</sup>, 1372–1374 cm<sup>-1</sup>, 1224–1227 cm<sup>-1</sup>, 1035–1036 cm<sup>-1</sup>, 918–922 cm<sup>-1</sup>, 842–844 cm<sup>-1</sup> and 698–700 cm<sup>-1</sup> corresponding to C–H stretching, bound water vibration, CH<sub>2</sub> in plane bending, S–O stretching, glycosidic linkage, C–O–C stretching of 3,6-anhydro-D-galactose, and O–S–O symmetric vibration of sulfate esters and sulfate on C-4 galactose, respectively.<sup>51,52</sup> The absence of significant changes in these bands indicates that the primary polysaccharide backbone of  $\kappa$ CG remains structurally intact following AuNP incorporation.

### 3.2. Coaxial extrusion

To identify the optimal flow rates for achieving uniform extrusion from a coaxial nozzle, extrudability tests were performed

on various compositions of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels, using flow rates ranging from 0.5 to 5 mL min<sup>-1</sup>. As shown in Fig. 3(a), a flow rate of 0.5 mL min<sup>-1</sup> was insufficient to extrude any of the compositions from the coaxial nozzle. At 1 mL min<sup>-1</sup>, extrusion was not possible for the highest weight percentages of both pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels, and only droplet formation was observed for the 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP compositions. Uniform extrusion at 1 mL min<sup>-1</sup> was achieved only for the lowest compositions of 1.5 $\kappa$ CG and 1.5 $\kappa$ CG–AuNP; however, frequent filament breakage occurred during the initial stages. At flow rates of 2, 2.5, and 3 mL min<sup>-1</sup>, filament breakage was a common issue across all compositions except for 1.5 $\kappa$ CG, which exhibited uniform extrusion with minimal breakage at the beginning. At the highest flow rate of 5 mL min<sup>-1</sup>, all compositions successfully achieved uniform extrusion with the least frequency of breakage, resulting in the production of continuous, longer filaments.

Based on the above analysis, 3 mL min<sup>-1</sup> and 5 mL min<sup>-1</sup> were selected for further studies, since continuous extrusion was observed for all compositions only at these two flow rates. Flow rates exceeding 5 mL min<sup>-1</sup> were not tested, keeping in mind the potential negative impacts of high shear rates on encapsulated cells.

The morphology of the  $\kappa$ CG–AuNP and pristine  $\kappa$ CG filaments extruded at two selected flow rates after incubation in a 100 mM KCl bath is shown in Fig. 3(b and c). The cylindrical nature of the hollow tubes is clearly visible. Longer filaments (> 50 cm) were attained using the higher flow rate of 5 mL min<sup>-1</sup> for both  $\kappa$ CG–AuNP and pristine  $\kappa$ CG hydrogels. The dimensions of coaxial-extruded hollow channels were evaluated by imaging the cross-sections using a stereomicroscope (Fig. 4(a and b)). The cross-sectional images of the filaments made from 2 $\kappa$ CG, 2.25 $\kappa$ CG, 2 $\kappa$ CG–AuNP, and 2.25 $\kappa$ CG–AuNP hydrogels showed hollow conduits, confirming the complete removal of the sacrificial core made of Pluronic F-127 at flow rates of 3 mL min<sup>-1</sup> and 5 mL min<sup>-1</sup>. Additionally, we conducted coaxial extrusion of 1.5 $\kappa$ CG–AuNP and 1.5 $\kappa$ CG hydrogels, as indicated in Fig. S4. The filaments produced from both the 1.5 $\kappa$ CG–AuNP and 1.5 $\kappa$ CG hydrogels exhibited collapsed lumens or ruptured cross-sections (Fig. S4).

The dimensions of the inner channel diameter, outer diameter, and wall thickness are generally known to depend on the flow rate of the sheath biopolymer. Previous research has indicated that increasing the speed of the sheath results in an increase in the outer diameter of hollow fibers.<sup>3,53</sup> Fig. 4(c and d) illustrate the impact of the flow rate of  $\kappa$ CG through the outer nozzle on the hollow lumen diameter, outer diameter, and wall thickness. Increasing the flow rate from 3 mL min<sup>-1</sup> to 5 mL min<sup>-1</sup> significantly enlarged the wall thickness of 2 $\kappa$ CG–AuNPs. However, other than the above, no significant differences in inner diameter, outer diameter, or wall thickness were observed in the 2.25 $\kappa$ CG–AuNP or pristine 2 $\kappa$ CG and 2.25 $\kappa$ CG hydrogel filaments. To assess the reproducibility of dimensions of hollow conduits using the setup, we performed 5 independent extrusion experiments and subsequently quantified the dimensions of 15 cross-sections by imaging under a



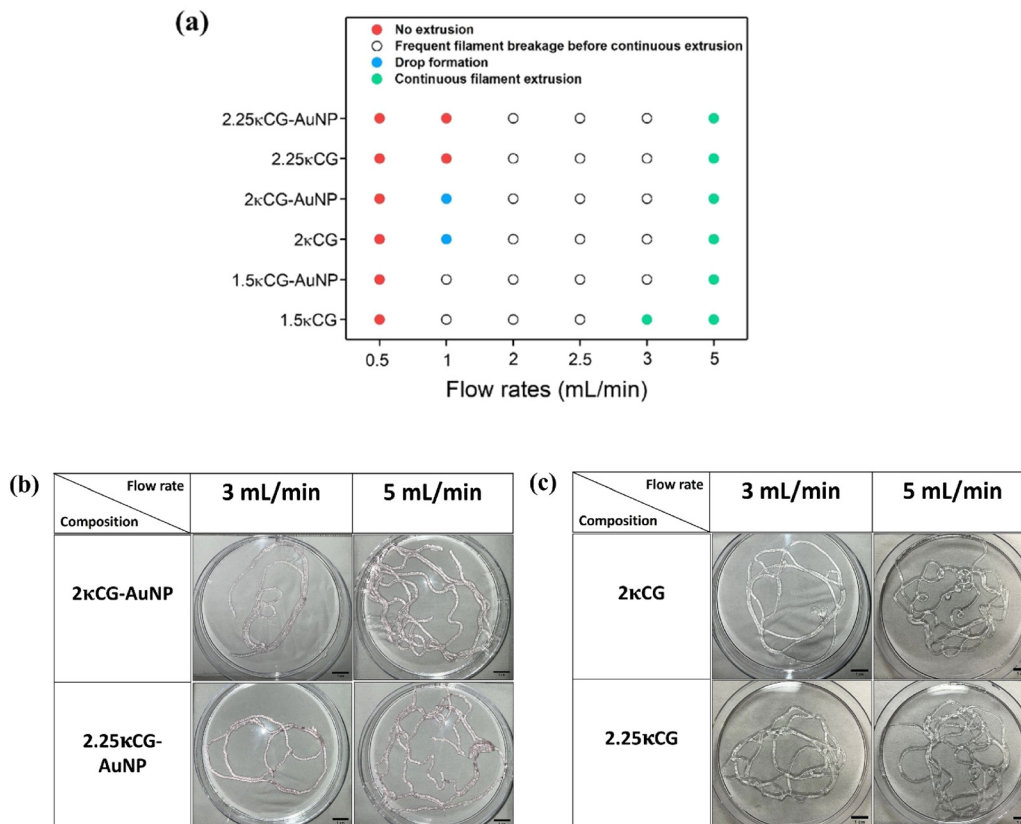


Fig. 3 (a) Extrudability map of varying compositions of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels extruded out of a pre-built coaxial nozzle (16G–20G); (b) and (c) images of extruded filaments after incubation in a KCl bath (scale bar: 1 cm).

stereomicroscope. The mean values and standard deviations for I.D., O.D., wall thickness and circularity of the hollow channels are presented in Table 1, while the coefficients of variation (CV) are reported in Table 2. According to the literature, a  $CV \leq 30\%$  is considered acceptable across experiments.<sup>54</sup> As shown in Table 2, all measured parameters exhibited CVs below this threshold, indicating that the extrusion process yielded constructs with satisfactory reproducibility and dimensional consistency.

The circularity index ranging between 0.82 and 0.92 in Table 1 indicates that the cross-sections of the extruded filaments were closer to a perfect circle (with a circularity index of 1) rather than an elongated shape (with a circularity index of 0). Therefore, all compositions of  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels, extruded at different flow rates, were able to form circular self-supporting hollow lumens after the removal of Pluronic F-127 from the core by incubation in a 100 mM KCl bath at 4 °C for 30 minutes.

### 3.3. Perfusability

A dye injection test was conducted to evaluate the formation of a uniform hollow channel in the extruded tubular structures after the removal of the sacrificial core, to determine their perfusability (Fig. 5 and the accompanying Movie in the SI). Perfusion was only successful for shorter filaments, measuring between 2 and 3 cm when extruded at a flow rate of 3 mL min<sup>-1</sup>

for all compositions of  $\kappa$ CG and  $\kappa$ CG–AuNPs. Additionally, there was a higher incidence of breakage during the manual handling of these shorter filaments.

In contrast, a continuous flow of dye was observed in longer filaments, greater than 5 cm, which were extruded at a flow rate of 5 mL min<sup>-1</sup> across all compositions of  $\kappa$ CG and  $\kappa$ CG–AuNPs. Among these, the longest filaments made from 2κCG–AuNPs were the most successful and could be handled without breakage during the perfusion assay, as illustrated in Fig. 5.

To gain a better understanding of how the concentration of  $\kappa$ CG, the influence of AuNPs, the effects of the ionic cross-linking bath, and the extrusion flow rate contribute to the successful formation of filaments with a self-supporting hollow lumen, we characterized the hydrogels using both rotational and extensional rheology, as detailed in the following sections.

### 3.4. Rheological studies

A complete rheological characterization of the pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels was performed to capture their viscoelastic and shear flow behaviours under conditions relevant to extrusion and post-extrusion crosslinking (Fig. 6). Small amplitude oscillatory frequency sweep (SAOS) experiments were conducted on pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels at an extrusion temperature of 37 °C. Large amplitude oscillatory strain sweeps (LAOS) for all ink compositions were used to determine the linear viscoelastic region (LVR), as illustrated in



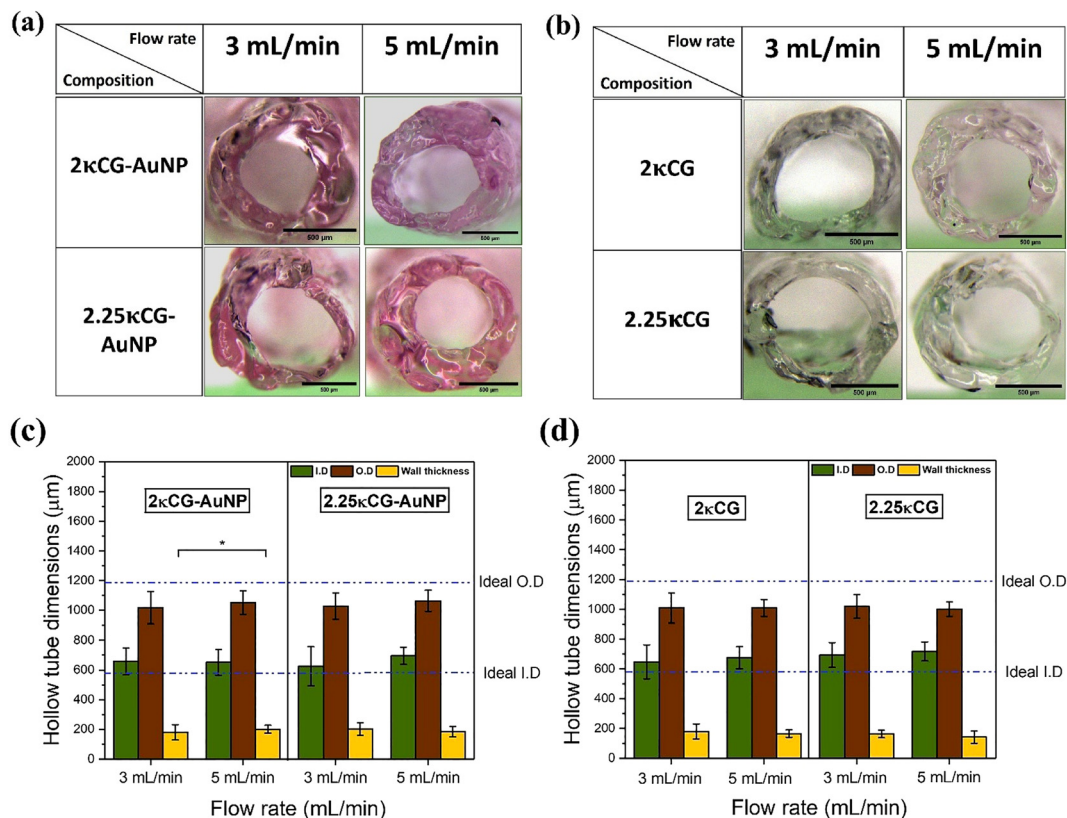


Fig. 4 Stereomicroscopic images of cross-sections of coaxially extruded hollow channels of varying compositions of (a)  $\kappa$ CG–AuNPs; (b) pristine  $\kappa$ CG (scale bar: 500  $\mu$ m); (c) and (d) the effect of the flow rate of the sheath biopolymer on the I.D., O.D. and wall thickness of hollow channels of  $\kappa$ CG–AuNP and pristine  $\kappa$ CG hydrogels. One-way analysis of variance with Tukey's test was used for comparison, and  $p < 0.05$  was considered to be statistically significant, indicated in the figure as '\*' ( $n = 10$  or  $n = 15$ ).

Table 1 Average dimensions of ID, OD, wall thickness and circularity of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hollow conduits

Flow rate Composition	3 mL min <sup>-1</sup>				5 mL min <sup>-1</sup>			
	I.D. ( $\mu$ m)	O.D. ( $\mu$ m)	Wall thickness ( $\mu$ m)	Circularity	I.D. ( $\mu$ m)	O.D. ( $\mu$ m)	Wall thickness ( $\mu$ m)	Circularity
2 $\kappa$ CG–AuNPs	658.2 $\pm$ 89.5	1018 $\pm$ 108.5	180.3 $\pm$ 51.1	0.88 $\pm$ 0.04	651 $\pm$ 86.1	1052.5 $\pm$ 78.9	200.7 $\pm$ 26.4	0.89 $\pm$ 0.06
2 $\kappa$ CG	648.4 $\pm$ 114.3	1008.4 $\pm$ 101.2	180 $\pm$ 50.3	0.82 $\pm$ 0.09	676.5 $\pm$ 74.3	1008.3 $\pm$ 57.2	166.0 $\pm$ 25.0	0.87 $\pm$ 0.08
2.25 $\kappa$ CG–AuNPs	625 $\pm$ 131.7	1028.3 $\pm$ 88.9	201.7 $\pm$ 42.2	0.85 $\pm$ 0.07	695.1 $\pm$ 58.5	1064.4 $\pm$ 71.5	184.7 $\pm$ 34.8	0.90 $\pm$ 0.03
2.25 $\kappa$ CG	693.3 $\pm$ 83.7	1018 $\pm$ 79.3	162.4 $\pm$ 25.6	0.89 $\pm$ 0.06	718.1 $\pm$ 63.4	1000.3 $\pm$ 48.1	141.1 $\pm$ 41.3	0.92 $\pm$ 0.04

Fig. S5(a and b). The shear-thinning properties of the hydrogels at an extrusion temperature of 37 °C were demonstrated through flow curves. To assess the impact of the 100 mM KCl crosslinking bath used for post-extrusion crosslinking of hydrogel filaments, small amplitude oscillatory frequency sweeps were performed on both pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogel discs after incubation in the 100 mM KCl crosslinking bath.

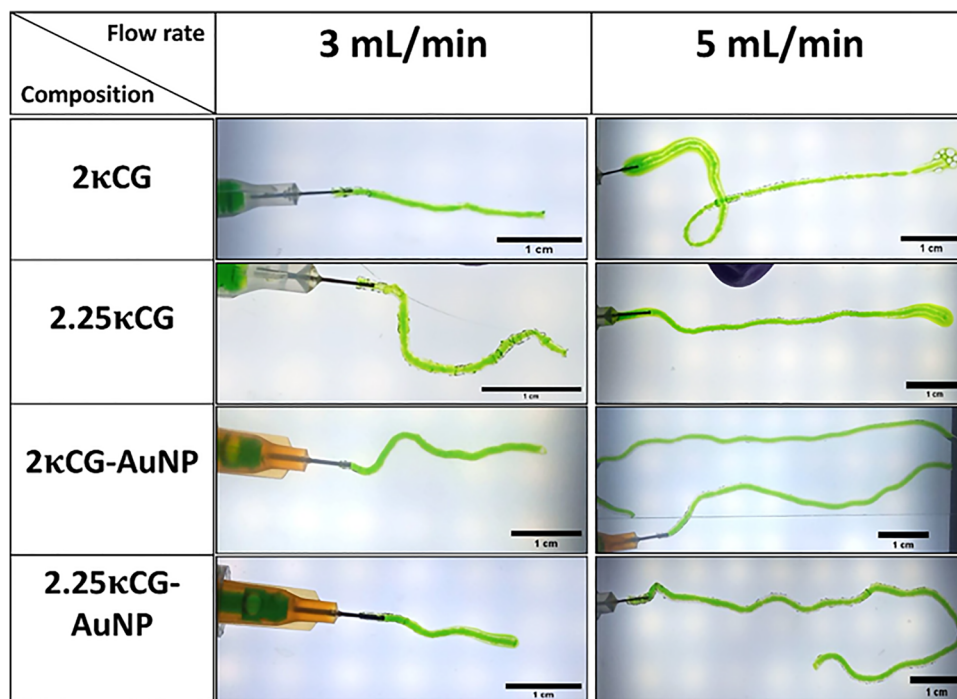
**3.4.1. Frequency sweeps of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels.** All compositions demonstrated that the storage modulus ( $G'$ ) was greater than the loss modulus ( $G''$ ), indicating an elastic gel-like behaviour throughout the measurement range of angular frequency, as shown in Fig. 6, panel (a) (i) and (ii). It was observed that as the weight percentage of  $\kappa$ CG increased, the  $G'$  values for both pristine

$\kappa$ CG and  $\kappa$ CG–AuNP hydrogels also increased. 2.25 $\kappa$ CG and 2.25 $\kappa$ CG–AuNP hydrogels exhibited the highest  $G'$  and  $G''$  values, respectively. Between the  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels, the  $\kappa$ CG compositions showed higher  $G'$  values. Specifically, the  $G'$  values at 1 rad s<sup>-1</sup> for the 1.5 $\kappa$ CG–AuNP, 2 $\kappa$ CG–AuNP, and 2.25 $\kappa$ CG–AuNP hydrogels were 1.53, 1.96, and 1.64 times lower than those of the 1.5 $\kappa$ CG, 2 $\kappa$ CG, and 2.25 $\kappa$ CG hydrogels, respectively, as depicted in Fig. 6, panel (a) (iii). Our findings align with those of prior reports written by Satish *et al.*, who synthesized gold nanospheres of size 11  $\pm$  2 nm in carboxymethyl cellulose to prepare electroconductive bioinks suitable for bioprinting.<sup>55</sup> Upon assessment of the flow curves of the hydrogels, it was observed that the viscosity of pristine carboxymethyl cellulose was higher compared to that



Table 2 CV% of I.D., O.D., wall thickness and circularity of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hollow conduits

Flow rate	$3 \text{ mL min}^{-1}$				$5 \text{ mL min}^{-1}$			
Composition	I.D. CV%	O.D. CV%	Wall thickness CV%	Circularity CV%	I.D. CV%	O.D. CV%	Wall thickness CV%	Circularity CV%
2 $\kappa$ CG–AuNPs	13.6	10.7	28.3	4.5	13.2	7.5	13.2	6.7
2 $\kappa$ CG	17.6	10	28	11	11.0	5.7	15.1	9.2
2.25 $\kappa$ CG–AuNPs	21.1	8.6	20.9	8.2	8.4	6.7	18.8	3.3
2.25 $\kappa$ CG	12.1	7.8	15.8	6.7	8.8	4.8	29.3	4.3

Fig. 5 Perfusion of food dye through the hollow lumens of tubular constructs of varying compositions of  $\kappa$ CG and  $\kappa$ CG–AuNPs coaxially extruded at different flow rates (scale bar: 1 cm).

of the *in situ* synthesized carboxymethyl cellulose. Similarly, the  $G'$  depicted in the LAOS and SAOS of the AuNP-containing hydrogel was significantly lower compared to that of the pristine hydrogel without altering the flow or viscoelastic properties of the hydrogels. The authors confirmed the reduction of Au ions by the hydroxyl (–OH) groups present in carboxymethyl cellulose, immobilising the gold nanospheres within the hydrogel matrices by XRD analysis. However, this phenomenon decreases interconnectivity between polymeric chains, leading to softer gels with lower  $G'$  and viscosity than the pristine hydrogels. Similarly, we believe that the hydroxyl groups in  $\kappa$ CG reduced  $\text{HAuCl}_4$  to  $\text{Au}^0$ , thereby stabilising the *in situ* synthesized AuNPs by acting as a capping agent. Due to the increased interaction of  $\kappa$ CG polymeric chains with AuNPs, there is a reduction in hydrogen bond interaction between polymeric chains, as shown in FTIR results, resulting in the formation of softer gels with a reduction in  $G'$  compared to the pristine compositions. Additionally, the loss tangent ( $\tan \delta$ ) of all samples ranges from 0.06 to 0.31, which suggests that they

exhibit elastic solid-like behaviour. The loss tangent is defined as the ratio of  $G''$  to  $G'$ , as indicated in eqn (2) above. When  $\tan \delta$  exceeds 1, it indicates liquid-like properties, whereas values below 1 correspond to a more solid-like behaviour.<sup>56</sup>

**3.4.2. Flow curves of pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels.** As illustrated in Fig. 6, panel (b) (i) and (ii), the viscosity of both  $\kappa$ CG–AuNP and  $\kappa$ CG hydrogels exhibits a rapid decrease when the shear rates increase from  $0.2 \text{ s}^{-1}$  to  $20 \text{ s}^{-1}$ . Additionally, the low shear viscosity measured at  $0.2 \text{ s}^{-1}$  was found to be directly proportional to the weight percentage of  $\kappa$ CG in both pristine  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels. Kim *et al.* previously demonstrated, using flow curves, that the viscosity of alginate–carrageenan composite hydrogels increases with higher concentrations of carrageenan, ranging from 0.5% (w/v) to 1.5% (w/v), when crosslinked with a fixed concentration of  $\text{CaSO}_4$ .<sup>57</sup> The shear-thinning properties were further quantified by fitting the power law mathematical model according to eqn (3). Fig. S6 shows the model fits of the flow curve data for the 1.5 $\kappa$ CG–AuNP and 1.5 $\kappa$ CG hydrogels to



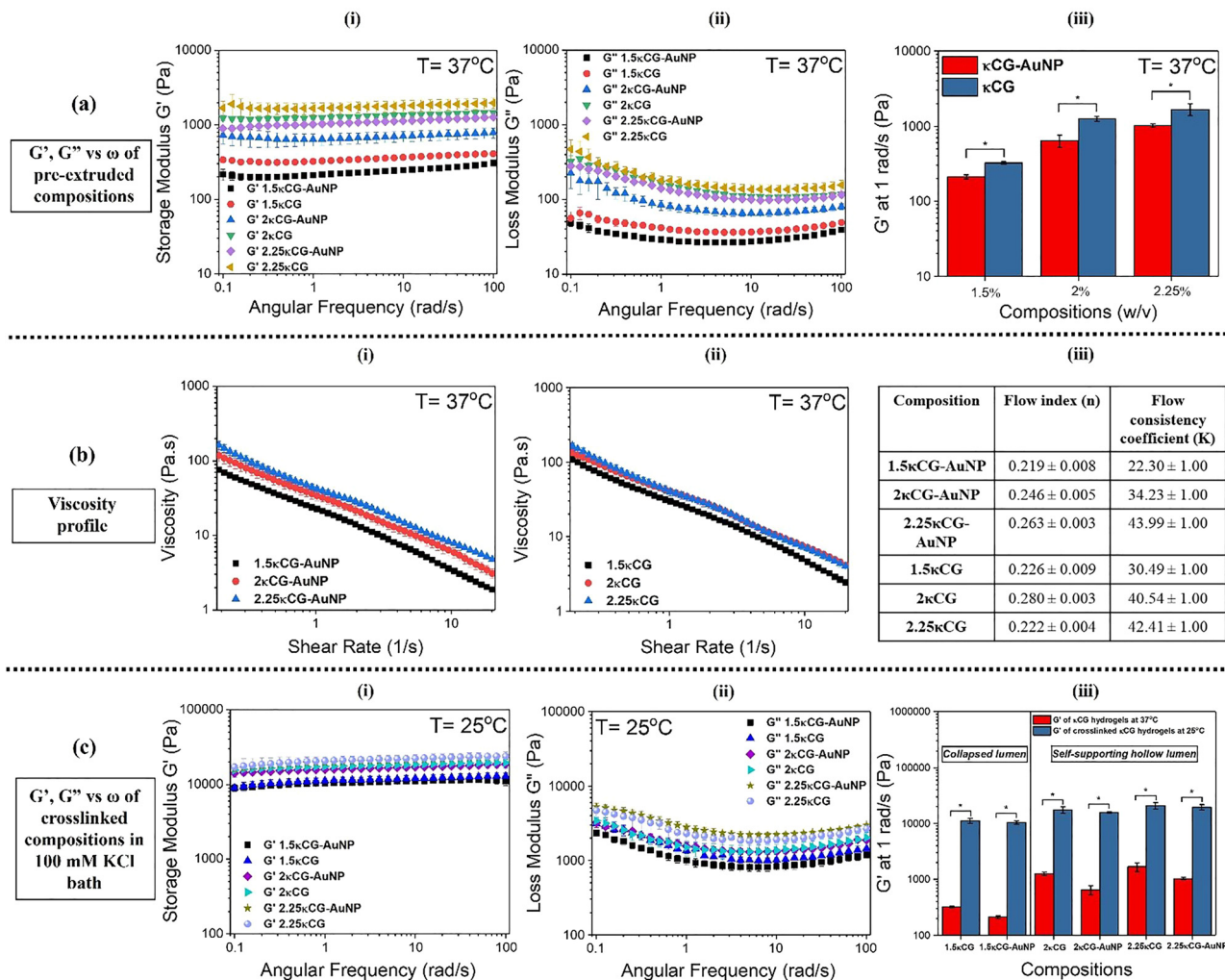


Fig. 6 Panel (a) (i) frequency sweeps depicting the storage modulus ( $G'$ ) of varying compositions of pristine  $\kappa$ CG and  $\kappa$ CG-AuNPs, (ii) frequency sweeps depicting the loss modulus ( $G''$ ) of varying compositions of pristine  $\kappa$ CG and  $\kappa$ CG-AuNPs, (iii) comparison of  $G'$  at 1 rad  $s^{-1}$  of varying compositions of pristine  $\kappa$ CG and  $\kappa$ CG-AuNPs; panel (b) flow curves showing the viscosity variation with the shear rate of varying compositions of (i) pristine  $\kappa$ CG hydrogels; (ii)  $\kappa$ CG-AuNP hydrogels, (iii)  $n$  and  $K$  table for all compositions of  $\kappa$ CG-AuNP and pristine  $\kappa$ CG hydrogels; panel (c) frequency sweeps of varying compositions of (i)  $\kappa$ CG and (ii)  $\kappa$ CG-AuNPs crosslinked in a 100 mM KCl bath, (iii) comparison of  $G'$  at 1 rad  $s^{-1}$  of varying compositions of pristine  $\kappa$ CG and  $\kappa$ CG-AuNPs at 37 °C before and after incubation in a 100 mM KCl crosslinking bath at 25 °C. One-way analysis of variance with Tukey's test was used for comparison, and  $p < 0.05$  was considered statistically significant, indicated in the figure by '\*' ( $n = 3$ ).

estimate the parameters  $n$  and  $K$ . A shear-thinning fluid is characterized by having  $n < 1$ . Thus, from Fig. 6 panel b (iii), we can infer that all compositions of  $\kappa$ CG hydrogels exhibit shear-thinning profiles with  $n$  values close to 0.2. The  $K$  values for both  $\kappa$ CG-AuNP and  $\kappa$ CG hydrogels increase with a higher weight percentage of  $\kappa$ CG. Additionally, all compositions of  $\kappa$ CG-AuNP hydrogels show lower  $K$  values compared to  $\kappa$ CG hydrogels, indicating a reduction in viscosity due to the incorporation of AuNPs within the  $\kappa$ CG hydrogel.

**3.4.3. Frequency sweep of hydrogel inks crosslinked with a 100 mM KCl crosslinking bath.** The large-amplitude oscillatory strain sweeps, which were conducted to determine the linear viscoelastic region (LVR), are shown in Fig. S5(c) and (d). After incubating 1.5 $\kappa$ CG-AuNP hydrogels in the 100 mM KCl bath for half an hour, we observed a 49-fold increase in  $G'$  compared to

$G'$  measured at the extrusion temperature of 37 °C (Fig. 6, panel c (iii)). Similarly, for the 2 $\kappa$ CG-AuNP and 2.25 $\kappa$ CG-AuNP hydrogels, we noted a 25-fold and 19-fold increase in  $G'$ , respectively.

For the 1.5 $\kappa$ CG hydrogels, incubation in the 100 mM KCl solution resulted in a 34-fold increase in  $G'$  compared to the storage modulus at 37 °C. Similarly, for the 2 $\kappa$ CG and 2.25 $\kappa$ CG hydrogels, we observed a 14-fold and 12-fold increase in  $G'$ , respectively, after incubation in the 100 mM KCl solution bath.

The increase in  $G'$  of  $\kappa$ CG hydrogels after incubation in a 100 mM KCl crosslinking bath for 30 minutes is crucial for the formation of self-supporting hollow filaments upon removal of the sacrificial core. Although there were significant increases in  $G'$  – 34-fold for the 1.5 $\kappa$ CG hydrogels and 49-fold for the 1.5 $\kappa$ CG-AuNP hydrogels – both types of filaments exhibited



collapsed lumens and ruptured cross-sections (Fig. S4). Kim *et al.* reported that a 2.5% (w/v) alginate solution, crosslinked with 0.2% (w/v) CaCl<sub>2</sub> and yielding a  $G'$  of 500 Pa, is the optimal composition for producing filaments that are closest to a straight line without core leakage during coaxial printing.<sup>58</sup> Therefore, despite the increase in  $G'$  of both 1.5 $\kappa$ CG–AuNP and 1.5 $\kappa$ CG hydrogels to over 1000 Pa after extrusion in the crosslinking bath, their relatively low  $G'$  values of 211 Pa and 324 Pa, respectively, at the extrusion temperature of 37 °C may contribute to the collapse and rupture of the filaments after they are deposited in the crosslinking bath.

The linear viscoelastic regimes of all hydrogels at an extrusion temperature of 37 °C, along with the  $\kappa$ CG hydrogels, after incubation in a crosslinking bath, were characterized through large amplitude oscillatory strain sweep tests (Fig. S5). No changes in moduli were observed at low-amplitude strains; however, as the strain increased, a decline in  $G'$  indicated deformation of the hydrogel. The point at which  $G'$  and  $G''$  intersect is referred to as the critical strain, beyond which  $G''$  dominates  $G'$ , indicating liquid-like behaviour (Fig. S7(a)). It is evident from Fig. S7(b) that the bath-crosslinked hydrogels display a decrease in the critical strain compared to the hydrogels, regardless of the weight percentage of  $\kappa$ CG or the presence of AuNPs. As demonstrated in Fig. S7(b), the critical strain of the 1.5 $\kappa$ CG–AuNP hydrogel is 83.5%, compared to just 0.65% for the bath-crosslinked 1.5 $\kappa$ CG–AuNP hydrogel. Similarly,

2 $\kappa$ CG–AuNP and 2.25 $\kappa$ CG–AuNP hydrogels showed reductions in critical strain from 49.7% and 14.4% to 0.56% and 0.47%, respectively, in the post-extrusion bath-crosslinked hydrogels. The compositions of 1.5 $\kappa$ CG, 2 $\kappa$ CG, and 2.25 $\kappa$ CG hydrogels also show a decrease in critical strain from 44.2% to 0.55%, from 30.9% to 0.63%, and from 17.9% to 0.58%, respectively.

### 3.5. Mechanical characterization

The compressive moduli, load at failure and strain at failure of  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels before and after crosslinking in the 100 mM KCl bath are presented in Fig. 7. Fig. S8 shows the model fit of the linear region of the stress–strain plot for 2 $\kappa$ CG–AuNP hydrogels.

A statistically significant increase in the compressive modulus of the 2 $\kappa$ CG hydrogel was observed after incubation in the 100 mM KCl bath. Although no statistical differences were obtained for the rest of the compositions, the average compressive modulus values of the bath-crosslinked hydrogels were consistently higher than those of the same compositions not incubated in the 100 mM KCl bath. Fig. 7(b) denotes the maximum load the hydrogels can sustain until failure, which also complies with a significant increase upon crosslinking in a 100 mM KCl bath for all the compositions. Furthermore, Fig. 7(c) shows an increasing trend in the average maximum strain at failure after bath-crosslinking; however, statistical

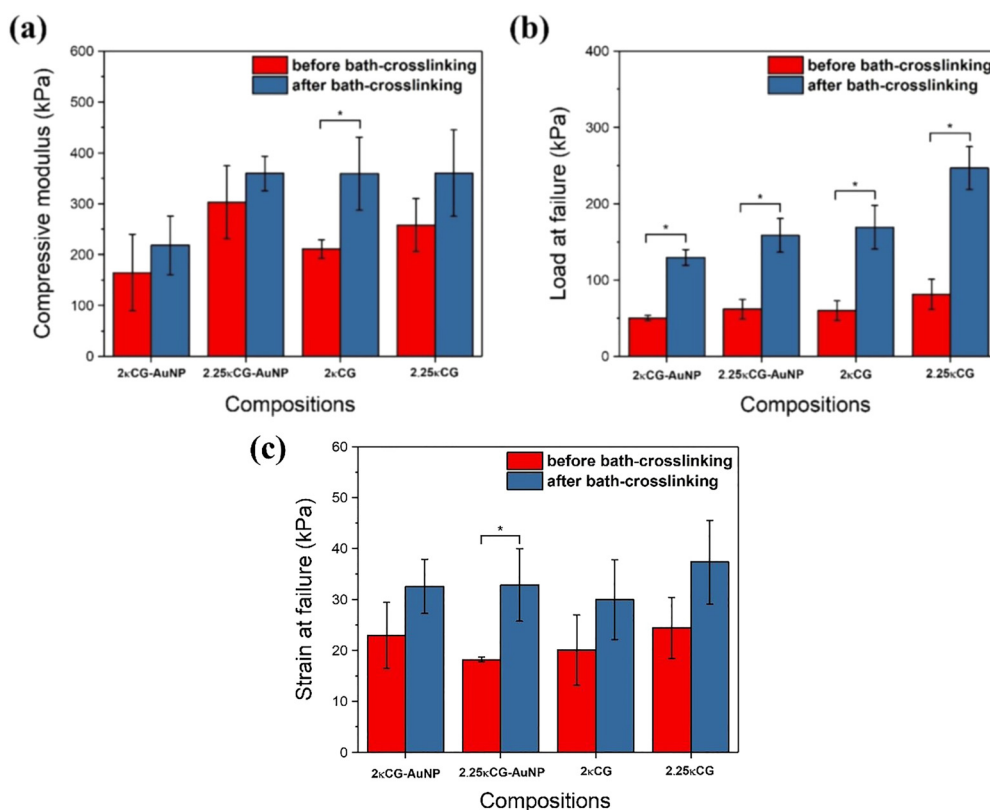


Fig. 7 Mechanical performance of varying compositions of  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels before and after crosslinking in a 100 mM KCl bath, (a) compressive moduli, (b) maximum load capacity before failure, and (c) maximum strain before failure. One-way analysis of variance with Tukey's test was used for comparison, and  $p < 0.05$  was considered statistically significant, indicated in the figure by '\*' ( $n = 3$ ).



significance was observed only for 2.25 $\kappa$ CG–AuNPs before and after crosslinking in 100 mM KCl.

Notably, all the hydrogels depicted a compressive modulus ranging between 165 and 303 kPa before incubation in the crosslinking bath and 218–360 kPa after incubation in the crosslinking bath, which are sufficiently higher compared to the compressive moduli of alginate-based hydrogels, previously used for bioprinting of perfusable constructs (34.5–50.7 kPa) and vascular tubes (6.58–8.38 kPa).<sup>59,60</sup> The above results indicate the critical role of the 100 mM KCl crosslinking bath in improving the mechanical properties of the hydrogels, which ensure the stability of the perfusable hollow channels after extrusion.

### 3.6. Pinch-off dynamics of hydrogel filaments

The filament-thinning behaviour over time was observed in all  $\kappa$ CG and  $\kappa$ CG–AuNP hydrogels. To illustrate this, Fig. 8 presents a sequence of images showing the filament thinning of  $\kappa$ CG–AuNP hydrogels at a flow rate of 5 mL min<sup>-1</sup>. A notable difference in filament pinch-off is evident when comparing the 2 $\kappa$ CG–AuNP hydrogels with the other two types. The 2.25 $\kappa$ CG–AuNP and 1.5 $\kappa$ CG–AuNP hydrogels exhibited less elongation before pinch-off. In contrast, the 2 $\kappa$ CG–AuNP sample exhibited a longer filament-thinning process, during which the filament elongated into a thinner thread just before pinch-off.

The image sequences have been quantified, and the plots are displayed in Fig. 9(a) and (b). Except for the 1.5 $\kappa$ CG and

1.5 $\kappa$ CG–AuNP hydrogels, where no filaments are formed, we observe that the presence of AuNPs increases the breakage or pinch-off time of filaments at both the 3 mL min<sup>-1</sup> and 5 mL min<sup>-1</sup> extrusion rates. The 2 $\kappa$ CG–AuNP hydrogels exhibit the longest filament breakage time at the 5 mL min<sup>-1</sup> extrusion rate. This finding aligns with our observations during the perfusion tests presented in Fig. 5. Previous studies on other biopolymers, such as xanthan gum, scleroglucan hydrogels, and nanocomposite solutions, have also examined pinch-off dynamics to investigate filament breakage time as a function of the biopolymer solutions and hydrogels' composition.<sup>61,62</sup>

In Table 3, we summarise the extensional relaxation time ( $\lambda$ ) calculated from the exponential region of Fig. 9(a) and (b). This calculation was performed by fitting eqn (4), as shown in Fig. S9. The 2 $\kappa$ CG–AuNP hydrogel filaments extruded at a flow rate of 5 mL min<sup>-1</sup> exhibit the highest extensional relaxation time. This indicates that the material takes longer to relax its stress under extensional deformation, demonstrating significant resistance to stretching and behaving more like a solid.<sup>32</sup> This solid-like behaviour can be due to the role of AuNPs promoting molecular alignment under uniaxial tension. While this mechanism has not been widely explored, our findings suggest that nanoparticle–polymer interactions can critically influence filament thinning dynamics. Particularly in 2 $\kappa$ CG–AuNPs, there may be an optimal balance of polymer–nanoparticle interactions, enabling better chain alignment and elastic recoil under uniaxial tension, giving slow filament breakup

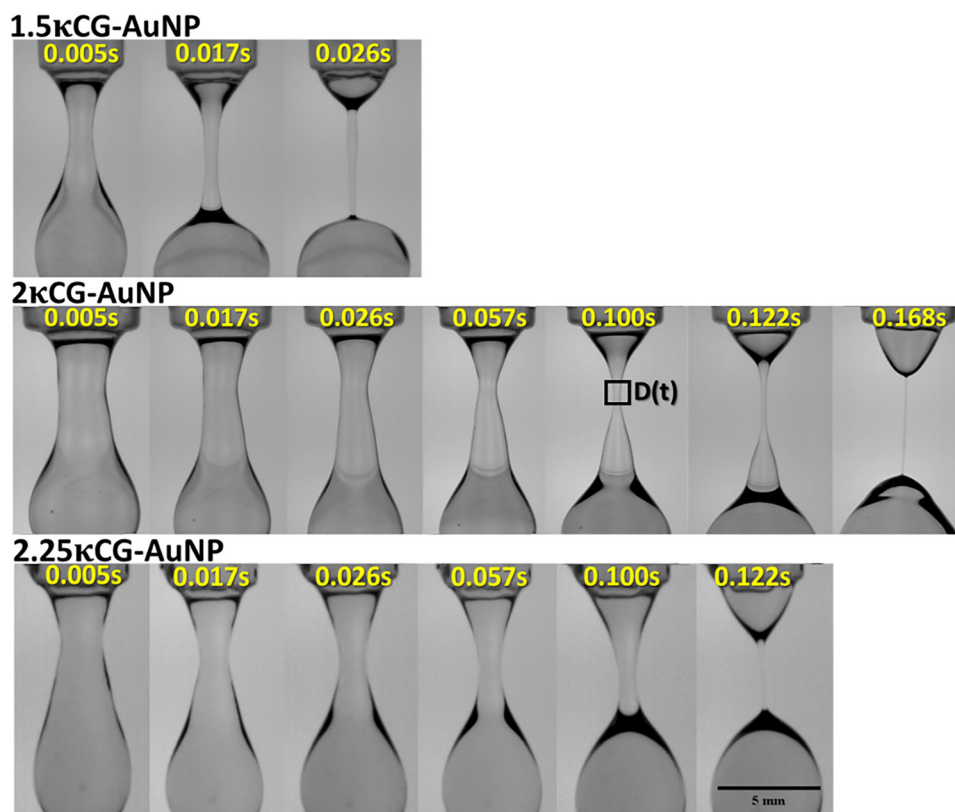


Fig. 8  $\kappa$ CG–AuNP hydrogel filament diameter thinning by dripping onto a substrate at a flow rate of 5 mL min<sup>-1</sup> (scale bar: 5 mm).



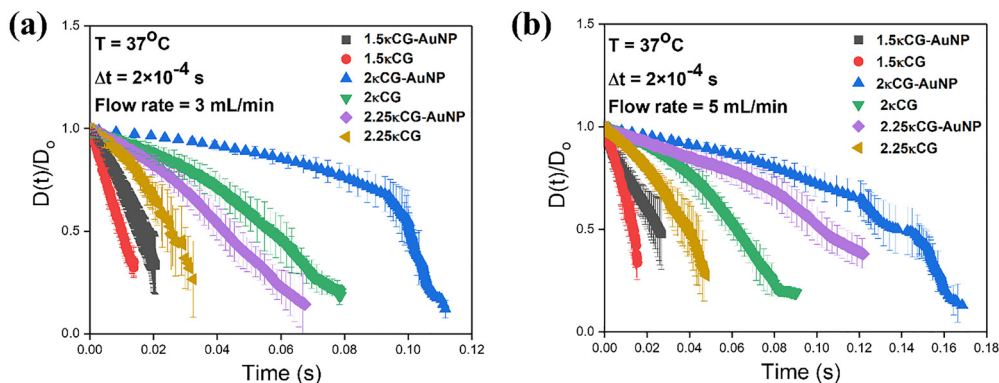


Fig. 9 Diameter evolution over time plotted for all compositions of  $\kappa$ CG and  $\kappa$ CG-AuNPs at a (a) 3 mL min<sup>-1</sup> flow rate and (b) 5 mL min<sup>-1</sup> flow rate.

Table 3 Extensional relaxation time ( $\lambda$ ) of  $\kappa$ CG and  $\kappa$ CG-AuNP hydrogels

Composition	Surface tension (mN m <sup>-1</sup> )	Relaxation time $\lambda$ (s)	
		3 mL min <sup>-1</sup>	5 mL min <sup>-1</sup>
1.5 $\kappa$ CG-AuNPs	69.11 ± 2.2	0.0130 ± 0.001	0.0121 ± 0.001
1.5 $\kappa$ CG	62.89 ± 2.4	0.007 ± 0.001	0.008 ± 0.001
2 $\kappa$ CG-AuNPs	67.20 ± 3.2	0.113 ± 0.003	0.116 ± 0.002
2 $\kappa$ CG	64.49 ± 1.7	0.046 ± 0.001	0.055 ± 0.003
2.25 $\kappa$ CG-AuNPs	68.20 ± 2.7	0.032 ± 0.002	0.078 ± 0.002
2.25 $\kappa$ CG	66.69 ± 1.9	0.021 ± 0.001	0.026 ± 0.001

under extension, leading to improved relaxation times, even though it is a softer gel (lower  $G'$ ). Although 2.25 $\kappa$ CG-AuNP exhibits higher  $G'$ , it shows shorter pinch-off times and reduced filament elongation compared to 2 $\kappa$ CG-AuNPs. This reflects that extrusion-driven filament stability is governed by a balance between stiffness and extensional viscoelasticity rather than by  $G'$  alone. A longer extensional relaxation time  $\lambda$  indicates that elastic stresses persist during necking and can markedly delay capillary breakup, yielding longer and thinner threads prior to pinch off. 2 $\kappa$ CG-AuNPs may combine sufficient shear stiffness with the highest  $\lambda$ , suggesting an optimally stretchable, elastically dominated network under uniaxial extension. In contrast, 2.25 $\kappa$ CG-AuNP likely possesses a more

constrained network in which excess polymer limits chain alignment and elastic recoil, lowering effective extensional elasticity even though  $G'$  is higher. Thus, for coaxial extrusion of hollow filaments, the softer 2 $\kappa$ CG-AuNP composition provides superior filament stability because its extensional response is better matched to the demands of the pinch off process than that of the stiffer 2.25 $\kappa$ CG-AuNP.

Although limited literature directly addresses this behaviour in nanocomposite hydrogels, studies on entangled polymer solutions demonstrate that extensional relaxation time ratios decrease with increasing polymer concentration despite rise in  $G'$ .<sup>63</sup> Also, other unrelated systems, such as protein fibrils and polysaccharide solutions, have shown that extensional rheology reveals microstructural effects not evident in shear rheology.<sup>64,65</sup>

Fig. 10 illustrates the relationship between the extensional rate and extensional viscosity, as calculated from eqn (6) and (7). A similar trend, where extensional viscosity decreases with increasing extensional rate, has been documented using the filament stretching method.<sup>66</sup> This phenomenon, characterized by reduced resistance to deformation during stretching, is known as tension-thinning behaviour or extension thinning.<sup>41,43,67</sup> The overlapping curves observed at initial extensional rates across all compositions, regardless of  $\kappa$ CG

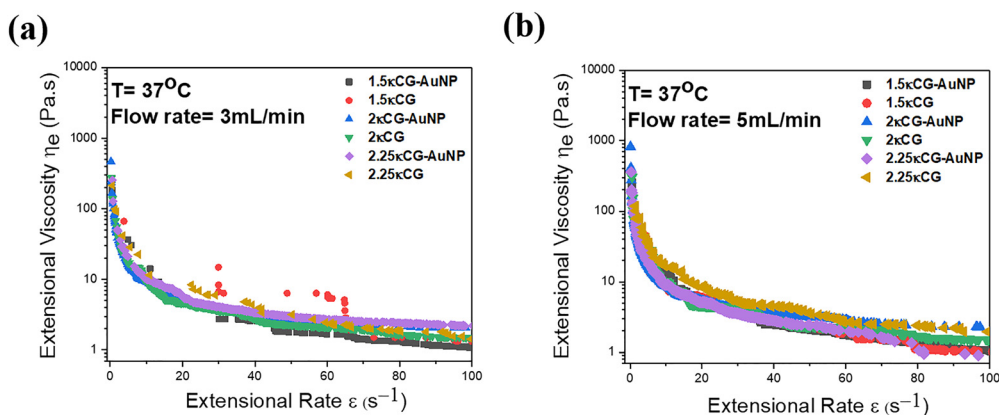


Fig. 10 Extensional viscosity versus extensional rate of  $\kappa$ CG and  $\kappa$ CG-AuNP hydrogels at a (a) 3 mL min<sup>-1</sup> flow rate and (b) 5 mL min<sup>-1</sup> flow rate.



concentration or the presence of AuNPs, suggest that the uncoiling and stretching of  $\kappa$ CG occur similarly in each case.

Perfusion assays, rotational rheological studies, and pinch-off dynamics indicated that among all hydrogel formulations, the 2 $\kappa$ CG–AuNP hydrogel extruded at a rate of 5 mL min<sup>-1</sup> successfully formed self-supporting hollow filaments. These filaments exhibited resistance to breakage during manual handling, which can be attributed to a sufficiently high initial  $G'$  that increased further after crosslinking in a 100 mM KCl bath, highlighting their predominantly elastic properties. Therefore, we chose to encapsulate A549 lung carcinoma cells within 2 $\kappa$ CG–AuNP hydrogels to create cell-laden hollow filaments using coaxial extrusion at the selected flow rate of 5 mL min<sup>-1</sup>.

### 3.7. Rheological characterization of cell-laden hydrogels

Before assessing the cell viability of encapsulated A549 lung carcinoma cells in 2 $\kappa$ CG–AuNP and 2 $\kappa$ CG filaments following coaxial extrusion, we first examined how cell encapsulation affected the rotational and pinch-off dynamics of the 2 $\kappa$ CG–AuNP and 2 $\kappa$ CG hydrogels. The results from large amplitude oscillatory strain sweeps and small amplitude oscillatory frequency sweeps for both cell-laden and acellular compositions indicated that the addition of  $2 \times 10^6$  cells per mL of A549 lung carcinoma cells led to a slight decrease in  $G'$  for both cell-laden 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP hydrogels when compared with their acellular controls (Fig. 11). The flow curves denoted an increase in viscosity at lower shear rates for the cell laden 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP hydrogels compared to the acellular gels.

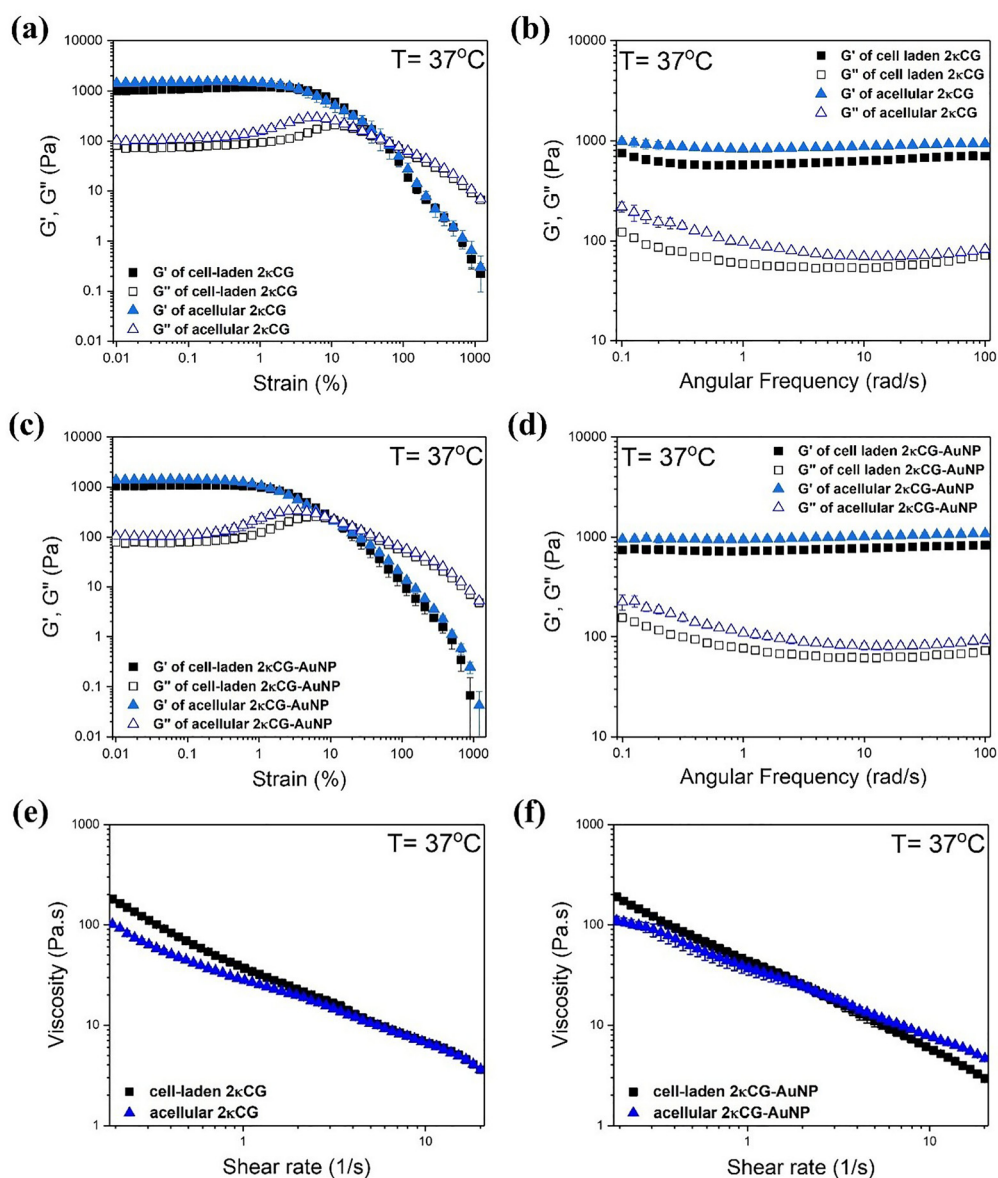


Fig. 11 (a and c) Comparison of large amplitude oscillatory strain sweeps of A549 encapsulated and acellular compositions of pristine 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP hydrogels; (b and d) comparison of frequency sweeps of A549 encapsulated and acellular compositions of pristine 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP hydrogels; (e and f) comparison of flow curves of A549 encapsulated and acellular compositions of pristine 2 $\kappa$ CG and 2 $\kappa$ CG–AuNP hydrogels.



The filament diameter decay obtained for A549 encapsulated and acellular 2κCG hydrogels and 2κCG–AuNP hydrogels from the pinch-off dynamics experiments (Fig. S10) showed a similar trend, except towards the end of the breakage process. The calculated extensional relaxation times (as shown in Table S3) for cell-encapsulated 2κCG and 2κCG–AuNP hydrogels, as well as for the acellular 2κCG and 2κCG–AuNP hydrogels, were not significantly different.

Diamantides *et al.* investigated the effect of chondrocyte inclusion on the rheological properties of collagen-based bioinks at very high cell densities ranging between  $5 \times 10^6$  and  $100 \times 10^6$  cells per mL, corresponding to a cell volume fraction of 0–0.18.<sup>68</sup> In their study, an increasing cell density led to a decrease in collagen  $G'$ . Although the cell volume fraction of the A549 lung carcinoma cells at a cell density of  $2 \times 10^6$  cells per mL in 2κCG and 2κCG–AuNP hydrogels, in our study, is low (approximately 0.0061) compared to those reported by Diamantides *et al.*, we observed a small decrease in  $G'$  and an increase in viscosity at low shear compared to acellular gels. At this low cell volume fraction, the encapsulated cells may be acting as soft inclusions, disrupting the polymeric network and thereby slightly lowering the  $G'$  while increasing hydrodynamic resistance under steady shear. Similarly, Zhang *et al.* reported that the incorporation of cells into alginate-based inks led to an increase in shear viscosity alongside no significant changes in  $G'$ .<sup>69</sup>

### 3.8. Cell viability

As shown in Fig. 12, A549 cells were successfully encapsulated in 2κCG and 2κCG–AuNP compositions at a cell density of  $2 \times 10^6$  cells per mL. The images in Fig. 12(a and b) clearly illustrate the cell-encapsulated walls and the hollow channels of the extruded filaments. Fig. S11 presents the 3D z-stacks depicting the uniform distribution of A549 cells throughout the conduit walls of 2κCG and 2κCG–AuNP hydrogels. The confocal z-stacks of 2κCG and 2κCG–AuNP hollow conduits were divided into four equal regions based on the total number of slices, and live and dead cells were quantified in each region (Tables S4 and S5). Each z-stack region indicates a sequential segment of the conduit wall along the imaging depth. The absence of any monotonic gradient in Table S4 in total cell number or viability along the z-axis indicates homogeneous encapsulation and no significant cell settling during the extrusion process. In Table S5, although a decrease in total cell number is observed with increasing imaging depth, gravitational settling should have resulted in higher cell numbers at the bottom, which was not observed. Moreover, the uniform cell viability across all regions (75–80%) indicates no depth-dependent cytotoxicity in the 2κCG–AuNP hollow conduit.

Upon encapsulation in both the 2κCG and 2κCG–AuNP hydrogels, the cells assumed a rounded morphology with a uniform distribution throughout the filament walls. This could be because of the lack of adhesion of A549 to the biopolymer chains due to the absence of adhesion sites in κCG polysaccharide.<sup>70</sup>

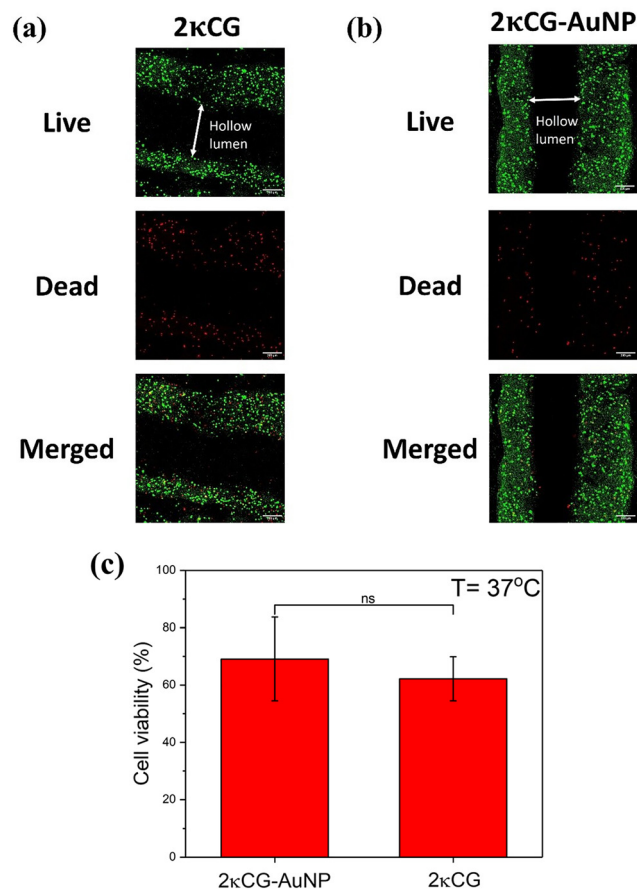


Fig. 12 (a and b) Maximum intensity projection of z-stacks of encapsulated A549 lung carcinoma cells in the walls of filaments of 2κCG and 2κCG–AuNP hydrogels, respectively, along with the presence of hollow channels; (c) cell viability after 24 hours of extrusion at a flow rate of  $5 \text{ mL min}^{-1}$ .

To quantitatively assess cell survival following the coaxial extrusion process and exposure to a 100 mM KCl crosslinking bath for 24 hours, a live/dead staining procedure was performed. The cell viability observed in the hollow conduits of 2κCG hydrogels was  $62.2 \pm 7\%$ , while the viability in 2κCG–AuNP hydrogel hollow conduits was  $69.2 \pm 14.6\%$ .

The  $G'$  of both 2κCG and 2κCG–AuNP ( $\sim 1 \text{ kPa}$ ) aligned with the  $G'$  of healthy porcine lung tissue, thereby ensuring the fabrication of physiologically relevant *in vitro* models.<sup>71</sup> Based on the above literature, to further study the biocompatibility of these cells in a complete 3D microenvironment, we decided to encapsulate A549 cells completely in the reported hydrogel compositions to evaluate their viability in a more complex shape that requires multimaterial extrusion using a simple setup. Since the current study focusses on material and process optimisation and initial biocompatibility, the choice of bulk encapsulation of cells also aligns with our goals to assess cell survival and material–cell interaction after extrusion.

Fig. 12(c) indicates that the presence of AuNPs within the κCG polymer did not confer additional toxicity to the encapsulated A549 cells. To isolate the effect of the crosslinking agent, we conducted 2D control experiments by exposing A549 cells to



Table 4 Comparison of hydrogels reported in the current study with previously investigated biopolymers

Sheath biopolymer	Core	Crosslinking bath	Coaxial nozzle size	Type of extrusion	Flow rate range/pressure on sheath biopolymers	Outer diameter (μm)	Inner diameter (μm)	Wall thickness (μm)	G' (Pa)/viscosity (Pa s)	Cell viability after 24 hours	Ref.
Alginate	CaCl <sub>2</sub>	—	14G–18G	Syringe pump	0.5–2 mL min <sup>-1</sup>	950–1400	600–1000	100–200	—	—	82
	CaCl <sub>2</sub>	CaCl <sub>2</sub>	18G–30G	Syringe pump	200–500 μL min <sup>-1</sup>	1500–2000	250	~625	—	95% (cell seeded in hollow lumen)	83
	CaCl <sub>2</sub>	CaCl <sub>2</sub>	16G–21G	Syringe pump	0.5–1.5 mL min <sup>-1</sup>	1000–1300	600–1000	~150–200	—	92.9 ± 2.4% (encapsulated cells)	6
SilkMA and GelMA	PVA	—	25.8 mm L × 4.0 mm outer diameter	Centra Printhead of RX1 bioprinter	15–250 mbar	—	~400	~45–80	—	~75% (encapsulated cells)	84
	Pluronic F-127	Pluronic F-127	5G–9G	T&R Biofab Bioprinter	120–160 kPa	3121 ± 80–2948 ± 54	—	339 ± 41–976 ± 25	40/10	HUVECs seeded in lumen	85
	Cell suspension	CaCl <sub>2</sub>	16G–21G	Syringe pump	15–30 mL h <sup>-1</sup>	870.87 ± 177.96–886.71 ± 9.83	242.89 ± 527.49 ± 13.36	313–179.61	—	96.36 ± 1.54% (cells seeded in lumen)	86
Pristine κCG and κCG–AuNPs	CaCl <sub>2</sub>	—	16G–22G	Syringe pump	0.5–1.5 mL min <sup>-1</sup>	913 ± 14–1327 ± 17	741 ± 5–1077 ± 40	86–125	—	>80% (cell seeded in lumen)	87
	Gelatin	—	17G/25G, 17G/26G, 17G/27G	Bioprinter	0.2 mL min <sup>-1</sup>	—	200–1000	—	—	>80% (encapsulated cells)	88
Pristine κCG and κCG–AuNPs	Pluronic F127	KCl	16G–20G	Syringe pump and manual	3–5 mL min <sup>-1</sup>	1000–2000	600–800	~200	1250/301, 639/262	62.2 ± 7%, 69.2 ± 14.6% (encapsulated cells)	This study

10 mM KCl for 24 hours and to 100 mM KCl for 30 minutes at 4 °C, as used for the synthesis of κCG hydrogels and the fabrication of hollow conduits. No significant cell death was observed compared with 2D control cultures, indicating that the residual KCl had no ionic or osmotic effect on cell survival, as measured 24 hours after extrusion of the hollow conduits (Fig. S12). Although AuNPs are known to have low acute toxicity both *in vitro* and *in vivo*, their size, shape, capping agent, and cell line type play critical roles in determining their toxic effects. Small spherical AuNPs (1.2 nm–1.4 nm), regardless of surface chemistry, are known to induce apoptosis and necrosis in cells.<sup>72</sup> Conversely, AuNPs of different shapes, like rods or the larger-sized flower and prism-shaped AuNPs, are generally considered safer.<sup>73</sup> Based on the current literature, AuNPs ranging from 10 nm to 40 nm with specific surface modifications have been shown to maintain good cell viability. For example, AuNPs approximately 14 nm in size, green-synthesized at low concentrations of κCG, demonstrated no significant cytotoxicity towards lung cancer epithelial cells, A549.<sup>30</sup> Similarly, carboxymethyl cellulose-capped AuNPs of size 11 nm showed moderate cytotoxicity of <80% in L929 mouse fibroblast cells only at very high concentrations of gel seeding in well plates.<sup>74</sup> Furthermore, Zhang *et al.* synthesized AuNPs in chitin gels with sizes varying from 23 to 81 nm and an average size of 40 nm. The chitin nanogels were biocompatible with HepG2 cells, resulting in cell viability exceeding 100%.<sup>75</sup> On a similar note, we also prioritised the formation of AuNPs within the safe size range (10–35 nm) to ensure no additional cytotoxicity in the cell-laden hollow tubes, as *in vitro* models. In addition, the concentration of AuNPs synthesized within 2κCG–AuNPs was quantified by using ICP-OES following complete acid digestion of lyophilised scaffolds. The composition exhibited a gold concentration of 64.6 ppm (mg kg<sup>-1</sup> of the hydrogel), confirming that the measured concentration falls within the reported biocompatible range for cell-laden scaffolds. Samadian *et al.* have previously demonstrated that AuNP loadings of 40–80 ppm did not induce any significant cytotoxicity in MG-63 cells cultured on poly(L-lactic acid (PLLA))-based hydrogel scaffolds.<sup>76</sup>

Irrespective of the lack of cytotoxicity by AuNPs, the overall cell viability of 62% and 69% for 2κCG and 2κCG–AuNP hydrogel hollow filaments, respectively, falls below the ~75–95% range reported in the literature (Table 4). However, it is critical to note that encapsulating cells in a hydrogel results in significant cell death prior to extrusion.<sup>77–80</sup> Additionally, the optimized flow rate of 5 mL min<sup>-1</sup> to obtain continuous extrusion of longer filaments is higher compared to the flow rates reported in the literature. For example, Yu *et al.* demonstrated that, as dispensing pressure increased, cells encapsulated in alginate and bioprinted with a coaxial nozzle showed a reduction in viability from 68% to 40%.<sup>81</sup> Hence, the extrusion process also imposes detrimental effects on the cells due to the high shear rates. Future applications using sophisticated coaxial extrusion bioprinters could mitigate shear-related damage during extrusion. While A549 cells assumed a spherical morphology within the walls of the hydrogel hollow filaments due



to the lack of adhesion sites, studies suggest that encapsulated cells, like fibroblasts, have the inherent potential to self-organise into 3D spheroids within the  $\kappa$ CG hydrogel over time.<sup>16</sup> Thus, extending the incubation period of encapsulated cells within  $\kappa$ CG hollow hydrogel tubes may improve cell viability *via* cell migration and spheroid formation. In spite of the above limitations, the cell viability results conclusively show that  $\kappa$ CG and its gold nanocomposite derivative have potential for the bioprinting of complex geometries in more sophisticated setups, laying the groundwork for the development of *in vitro* core-shell coculture models with potential applications in drug screening and disease modelling.

## 4. Comparison with the current literature

Table 4 summarises research from the past decade on the fabrication of hollow hydrogel conduits using syringe pumps and coaxial bioprinters. The dimensions of the hollow conduits, namely, inner diameter (I.D.), outer diameter (O.D.) and wall thickness of  $\kappa$ CG hollow hydrogel tubes, were targeted based on a literature review of the commonly fabricated dimensions of hollow channels.

As shown in Table 4, alginate-based hollow hydrogel channels have been extensively studied for creating hollow structures with wall thicknesses ranging from over 600  $\mu\text{m}$  to as thin as 45  $\mu\text{m}$ . The rapid crosslinking of alginate by  $\text{Ca}^{2+}$  allows for immediate gelation upon contact, facilitating the formation of tubular structures with strong mechanical properties. Similarly, UV-crosslinkable GelMA, another biopolymer commonly examined for coaxial printing, requires UV exposure to initiate crosslinking and enhance the mechanical properties of hollow conduits. The existing literature on alginate and GelMA primarily focuses on seeding various cell types within the hollow lumen. This seeding process can occur either directly during coaxial printing through the core nozzle or after extrusion. Such methods allow for the assessment of cell viability, spreading, and proliferation on the inner walls of the lumen. This approach helps prevent cell death caused by shear stress during extrusion and reduces the toxicity resulting from crosslinking agents used in the biopolymer. However, a few studies, including one conducted by Yin *et al.*, have attempted to encapsulate cells within the sheath biopolymer itself. This method resulted in a lower cell viability of approximately 75% after three days of culture.<sup>84</sup> While many studies emphasize the lower viscosity of commonly used biopolymers as a desirable characteristic for easier fabrication of hollow conduits, the rheological properties of these materials have not been thoroughly investigated, as shown in Table 3. Moreover, the current literature on hollow conduits primarily focuses on either pure biopolymers or biopolymer blends. Although nanomaterials have been extensively studied for 3D bioprinting, the use of nanocomposite hydrogels in coaxial bioprinting and extrusion remains limited.<sup>89,90</sup>

We have successfully fabricated hollow lumens of  $\kappa$ CG with dimensions comparable to those reported in the literature. Although some studies have achieved the fabrication of smaller I.D. values ( $\sim 200 \mu\text{m}$ ) by core-sheath extrusion, they are difficult to characterize for quantitative analysis and biological assays. For example, Homan *et al.* successfully fabricated tubules with diameters as low as 150  $\mu\text{m}$ , but quantitative assays were conducted only on tubules with diameters ranging from 400  $\mu\text{m}$  to 550  $\mu\text{m}$ .<sup>91</sup> The dimensions we targeted enabled us to perform quantitative assays such as cell viability measurements, along with perfusion of a food dye through the lumen to determine uniform formation of hollow channels. These perfusable channels not only confirm the formation of uniform hollow lumens but also facilitate proper exchange of nutrients and metabolic wastes from cells.<sup>92</sup> Moreover, the hollow tube dimensions with outer diameters ranging between 1000 and 1030  $\mu\text{m}$  and inner diameters of 625–700  $\mu\text{m}$ , fabricated in our study, closely replicate the tubular geometry of small bronchioles in the lungs.<sup>93</sup> Therefore, the obtained size range reflects the microscale architecture of the terminal airways in the lung, making the constructs well-suited for use in lung tissue engineering, disease modelling, or drug screening.

## 5. Conclusions and future outlook

In this study, we investigated key parameters, including the flow rate and composition, for pristine  $\kappa$ CG and  $\kappa$ CG-AuNPs to assess their impact on the fabrication of hollow tubular conduits. Rheological analysis showed that a low  $G'$  ( $< 500 \text{ Pa}$ ) led to the collapse of hollow lumens in both pristine 1.5 $\kappa$ CG and 1.5 $\kappa$ CG-AuNPs, whereas intact hollow lumens were observed at higher compositions. Perfusability assay and pinch-off dynamics further indicated that a higher flow rate of 5  $\text{mL min}^{-1}$  resulted in longer filaments with better mechanical integrity for manual handling, especially for 2 $\kappa$ CG-AuNPs. A significant number of cells survived the coaxial extrusion process in both pristine  $\kappa$ CG and  $\kappa$ CG-AuNP cell-laden hollow conduits. These findings highlight the importance of the investigated parameters in determining the optimal composition and extrusion conditions necessary for the formation of hollow channels with potential for scaling up to model *in vitro* biological tubular structures using more advanced coaxial setups. In addition to the above, future investigations could explore varying the concentration of chloroauric acid to study its effect on the size of *in situ* synthesized AuNPs, and consequently, on the extrusion behaviour and fidelity of hollow conduits. Further studies on the 3D encapsulation of A549 can include extended culture durations to evaluate long-term cellular behaviour and functional assays to assess tissue-specific functionality of the constructs.

## Author contributions

Sanchari Swarupa: experiments, data collection, analysis, writing; Mata Subhashita: pinch-off dynamics experiments, data



collection, analysis, writing; Nikita Chauhan: SAXS experiments, analysis; Jitendra Bahadur: SAXS analysis, writing; Sharad Gupta: cell culture experiment supervision; Prachi Thareja: overall supervision, reviewing, funding acquisition, and project administration.

## Conflicts of interest

The authors declare no competing financial interests or personal relationships that could have influenced the work reported in this article.

## Data availability

The data supporting this article have been included as part of the supplementary information (SI). Supplementary information is available: Additional experimental data including hydrodynamic size distribution of AuNPs, FTIR, SAXS pattern and fitting parameters, cross-section of hollow conduits, rheological analyses, compressive modulus calculation, relaxation time measurements, diameter decay profiles, confocal z-stack images of hollow conduits, depth-wise cell distribution analyses and 2D control experiments. See DOI: <https://doi.org/10.1039/d6ma00245e>.

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