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Correction: Enhanced production of dimethyl carbonate from the alternating polarity electrolysis of methanol and carbon dioxide

Momoko Ishii,^{a,b} Maria M. Paulsen,^{a,c} Remi A. Mellinghoff,^{a,b} Heather O. LeClerc,^{a,b} Ho Yin Tse,^a Hanno C. Erythropel,^{a,b} Julie B. Zimmerman,^{a,b} Paul T. Anastas^{*a,d} and Darren S. Lee^{*a,e}

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Correction for 'Enhanced production of dimethyl carbonate from the alternating polarity electrolysis of methanol and carbon dioxide' by Momoko Ishii *et al.*, *Green Chem.*, 2025, **27**, 14513–14521, <https://doi.org/10.1039/D5GC02358K>.

The authors regret that the image presented as Fig. 6c was a duplicate of Fig. 6a. The corrected Fig. 6 is provided below.



^aCenter for Green Chemistry & Green Engineering, School of the Environment, Yale University, New Haven, CT, 06511, USA. E-mail: darren.lee@ntu.ac.uk, paul.anastas@yale.edu

^bDepartment of Chemical and Environmental Engineering, Yale University, New Haven, CT, 06511, USA

^cDepartment of Energy, Aalborg University, 9220 Aalborg Øst, Denmark

^dSchool of Public Health, Yale School of Medicine, New Haven, CT, 06511, USA

^eSchool of Science & Technology, Nottingham Trent University, Clifton Lane, Nottingham, NG11 8NS, UK

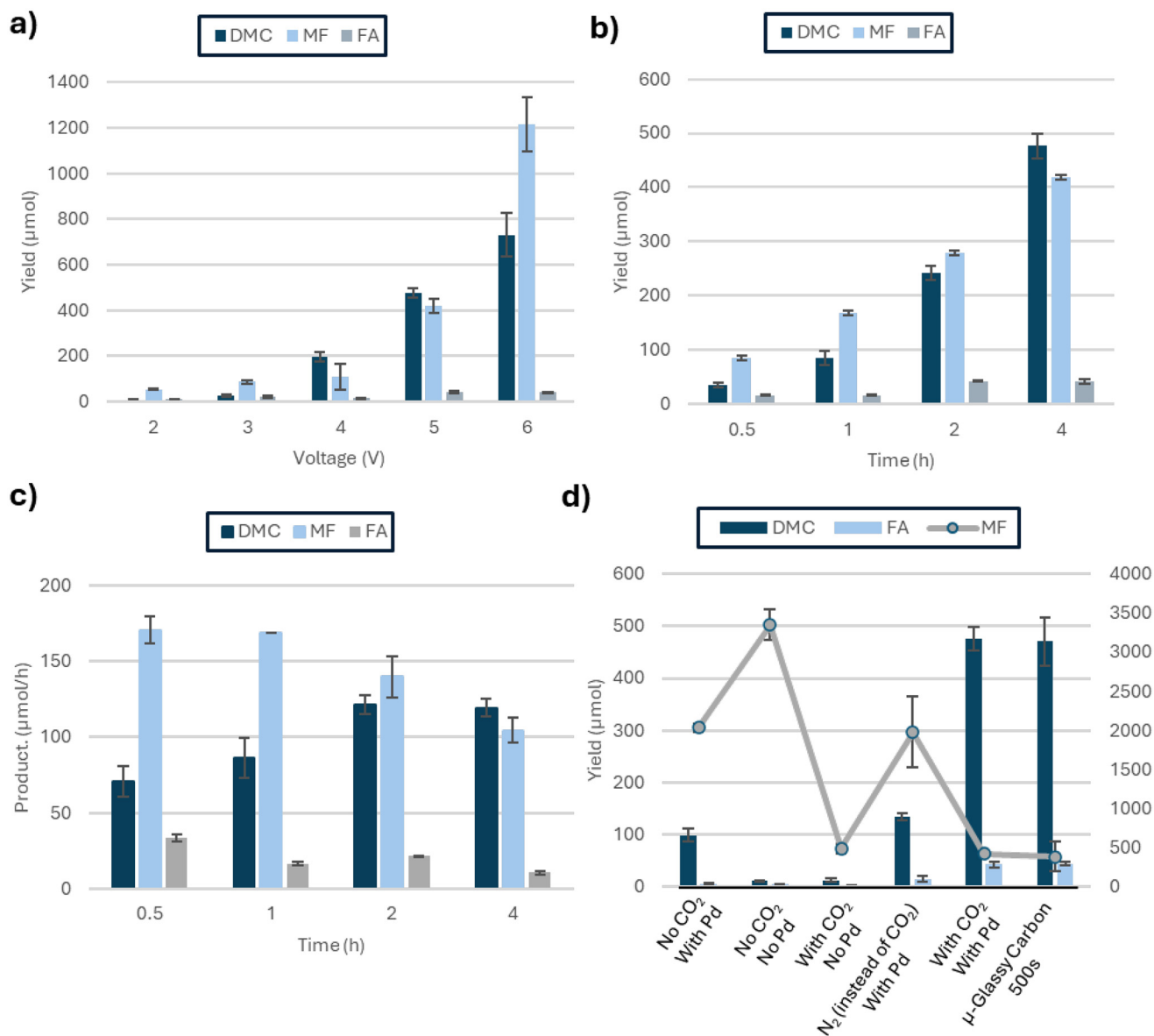


Fig. 6 Yield of dimethyl carbonate (DMC), formaldehyde (FA), and methyl formate (MF) after a reaction using 500 s alternating polarity (a) at various voltages after 4 h and (b) sampled at various times using 5 V; (c) productivity of DMC, FA, and MF under same conditions as in (b); (d) yield of DMC, FA and MF under different control conditions using, unless otherwise indicated, 5 V and 500 s polarity switching and a 4 h reaction time. Surface area of the μ GC electrodes reduced by approx. 58% assuming no porosity (regular GC: 52.5 mm \times 8 mm \times 2 mm vs. μ GC: 56 mm \times 3 mm \times 1 mm). Error bars representing \pm one standard deviation.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

