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## Correction: Enhanced production of dimethyl carbonate from the alternating polarity electrolysis of methanol and carbon dioxide

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Correction for 'Enhanced production of dimethyl carbonate from the alternating polarity electrolysis of methanol and carbon dioxide' by Momoko Ishii *et al.*, *Green Chem.*, 2025, **27**, 14513–14521, <https://doi.org/10.1039/D5GC02358K>.

The authors regret that the image presented as Fig. 6c was a duplicate of Fig. 6a. The corrected Fig. 6 is provided below.

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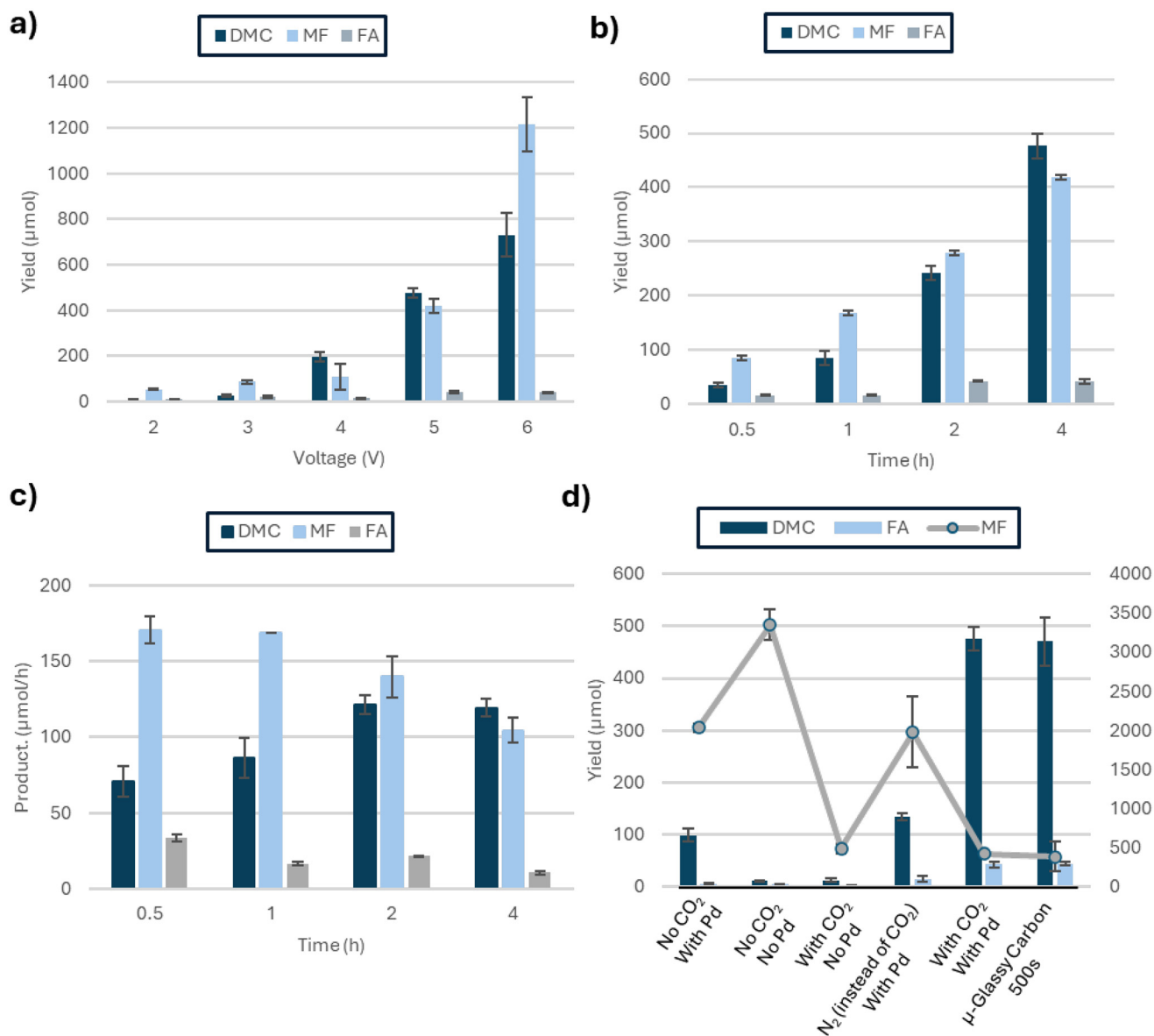
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**Fig. 6** Yield of dimethyl carbonate (DMC), formaldehyde (FA), and methyl formate (MF) after a reaction using 500 s alternating polarity (a) at various voltages after 4 h and (b) sampled at various times using 5 V; (c) productivity of DMC, FA, and MF under same conditions as in (b); (d) yield of DMC, FA and MF under different control conditions using, unless otherwise indicated, 5 V and 500 s polarity switching and a 4 h reaction time. Surface area of the  $\mu$ GC electrodes reduced by approx. 58% assuming no porosity (regular GC: 52.5 mm  $\times$  8 mm  $\times$  2 mm vs.  $\mu$ GC: 56 mm  $\times$  3 mm  $\times$  1 mm). Error bars representing  $\pm$  one standard deviation.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

