





Cite this: DOI: 10.1039/d5fb00708a

# Valorization of sunflower (*Helianthus annuus* L.) and canola (*Brassica napus* L.) meals through protein coprecipitation: physicochemical, functional, and structural characterization

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Canola–sunflower protein coprecipitates (CSPoCos) were prepared using different flour ratios, and their physicochemical, functional, and structural properties were analyzed. The crude protein content of CSPoCos ranged from 88.19% to 91.21%, which was higher than that of canola protein precipitate (CPoPc, 82.25%) and sunflower protein precipitate (SPoPc, 86.19%). In addition, protein coprecipitation (PoC) increased the apparent and compact densities and Carr and Hausner indices while decreasing the particle size and modifying the morphology, which were measured through dynamic light scattering and scanning electron microscopy, respectively. Glutelins (52.71–60.30%) were the major protein fraction in CSPoCos. In addition, PoC increased the water holding capacity, oil holding capacity, least gelation concentration, emulsifying activity index, foam capacity and foam stability. The improvement in these functional properties could be due to the formation or combination of various covalent and non-covalent interactions between the individual protein sources. Moreover, PoC modified the secondary and tertiary protein structures of CSPoCos, which was manifested by a significant increase ( $p < 0.05$ ) in surface hydrophobicity, in comparison to CPoPc and SPoPc. On the other hand, the values for *in vitro* antioxidant capacity, *in vitro* digestibility, sulfhydryl groups and disulfide bridges of the CSPoCos were higher than those of CPoPc. Electrophoresis revealed that CSPoCos had subunits with molecular weights ranging from 15 to 49 kDa. Overall, CSPoCos had better functional properties than single canola and sunflower proteins, making the application of PoC an attractive alternative for obtaining protein powders with improved qualities for use as food ingredients.

Received 20th October 2025  
Accepted 20th March 2026

DOI: 10.1039/d5fb00708a

rsc.li/susfoodtech

## Sustainability spotlight

Oil extraction from oilseeds such as sunflower and canola is one of the processes that generates the most by-products. Among these by-products, de-oiled sunflower and canola meals stand out due to their high protein content. This research proposes the use of de-oiled sunflower and canola meals to produce protein coprecipitates with improved physicochemical, functional and structural properties compared to the individual sources. Numerous studies have reported the valorization of proteins from oilseed industry waste as a strategy for their use as food ingredients, the development of new products, or the improvement of existing products, consistent with this research. The results obtained in this study are aligned with Sustainable Development Goals 2 (Zero Hunger), 9 (Industry, Innovation, and Infrastructure), and 12 (Responsible Consumption and Production).

## 1 Introduction

The continuously growing world population demands a sufficient supply of protein through alternative sources.<sup>1</sup> In this

case, the recovery of proteins from agro-industrial by-products, including oilseed meals, represents an attractive alternative.<sup>2</sup> Canola (*Brassica napus* L.) and sunflower (*Helianthus annuus* L.) are the second and third most important oilseed crops in the world, respectively, after soybeans.<sup>3,4</sup> According to FAOSTAT,<sup>5</sup> 70 and 56 million metric tons of canola and sunflower were produced worldwide, respectively, with Canada and Ukraine being the largest producers of these crops. Oilseeds are an interesting source of protein since the meals obtained after oil extraction are by-products that can be valorized. Typically, residual oilseed meals from oil extraction may contain 35% to 40% protein on a dry basis. These oilseeds are recognized as an alternative protein source with potential use in human

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nutrition, although to date they are mainly used for animal consumption.<sup>6,7</sup>

One of the alternatives for the recovery of proteins for human consumption from oilseed meals is protein precipitates (PoPc) such as concentrates or isolates.<sup>8</sup> However, each oilseed meal has limited nutritional value due to the presence of anti-nutrients or a low functional quality, which restricts its use as a food ingredient.<sup>9,10</sup> Therefore, there has been interest in formulating food products by combining proteins to solve these drawbacks, which can be achieved through PoB (protein blending) and PoC (protein coprecipitation) using two or more sources (plant-plant, animal-animal or plant-animal proteins).<sup>11–13</sup> The differences between these protein products are based on the preparation method. PoB can be obtained through the direct mixing of protein isolates or concentrates, while PoC can be prepared by precipitation or by heating.<sup>14</sup>

The formation of PoB and PoCos involves different interactions that give rise to different structures. These structures present in PoB and PoCos can have a synergistic effect on their techno-functional, nutritional and biological properties,<sup>9,11,15,16</sup> depending mainly on the various intrinsic properties (type of proteins, protein concentration, amino acid composition, structure, conformation, molecular weight, and surface properties) of the protein sources and processing conditions such as pH, temperature, ionic force, meal: solvent ratio, *etc.*<sup>12,17–20</sup>

For example, Wu *et al.*<sup>13</sup> studied the effect of mixing soybean and zein proteins. The results of this study revealed that the soybean-zein (5 : 1; w/w at pH 7) PoB improved the solubility, foaming properties and water holding capacity, in contrast to the zein proteins. These researchers pointed out that the increase in these techno-functional properties was a consequence of the fusion of the two proteins and the modification of their secondary structure. In addition, Zhou *et al.*<sup>21</sup> observed that PoB obtained from pea and grass carp protein (1 : 1; w/w at pH 7) isolates enhanced the solubility and foaming properties compared to grass carp protein isolate. These results are due to the fact that the protein molecules of pea and grass carp have opposite charges, and thus their intermolecular electrostatic interactions are strengthened to generate soluble protein aggregates.

Conversely, in the case of PoC, Zhang *et al.*<sup>22</sup> demonstrated that the association of ginseng and *Schisandra* (6 : 1; w/w) proteins in the form of PoCo by alkaline extraction (pH 9) and isoelectric precipitation (pH 3.5) increased the water and oil holding capacities compared to individual proteins. This improvement resulted from the increase in hydrogen bonding interactions and hydrophilic groups exposed under alkaline conditions. In another study, Kristensen *et al.*<sup>16</sup> reported that pea and whey (20 : 80; v/v) PoCos, obtained by alkaline extraction (pH 11) and isoelectric precipitation (pH 4.6), produced new electrostatic interactions and disulfide bonds between these proteins, increasing their solubility and surface hydrophobicity.

To date, this work constitutes the first report on obtaining PoCos from canola and sunflower as an alternative protein source. Therefore, the objective was to determine the physico-chemical, functional, and structural attributes of CSPoCos to evaluate their potential application as a novel food ingredient.

## 2 Materials and methods

### 2.1 Material and chemicals

Canola meal (CM) was provided by Forrajes y Alimentos San Cayetano, S. de R.L. de C.V. (Tepic, Nayarit, México), while sunflower meal (SM) was obtained from New Country Organics, Inc. (Lubbock, Texas, USA). The contents of crude protein ( $N \times 6.25$ ), moisture, ash, fat, and total carbohydrate contents for CM, according to the AOAC<sup>19</sup> methods, were  $39.72\% \pm 0.32\%$ ,  $9.76\% \pm 0.14\%$ ,  $7.05\% \pm 0.04\%$ ,  $2.96\% \pm 0.25\%$ , and  $31.74\% \pm 0.81\%$ , respectively. The proximal composition of SM is presented in Section 2.2, since it was necessary to remove a significant amount of oil. All chemicals employed in this research were reagent grade and purchased from Sigma-Aldrich, Ferromont and J.T. Baker (Ciudad de México, México).

### 2.2 Preparation of defatted sunflower flour (DSF)

Because SM contained a considerable amount of lipids ( $21.23\% \pm 0.25\%$ ), it was necessary to apply a step to reduce the percentage of this component, according to the following procedure. Firstly, lots of 100 g of SM were placed in 1 L of ethyl ether with stirring at 500 rpm for 1 h, during two successive cycles and replacing the spent solvent with fresh solvent. The material obtained from the second defatting cycle was placed in a plastic container for complete desolventization, and then in an extractor hood for 12 h to obtain DSF, which was stored in airtight bags at room temperature for subsequent characterization. The contents of crude protein ( $N \times 6.25$ ), moisture, ash, fat, and total carbohydrates of DSF were  $33.54\% \pm 0.89\%$ ,  $8.81\% \pm 0.04\%$ ,  $6.76\% \pm 0.80\%$ ,  $0.49\% \pm 0.18\%$ , and  $50.40\% \pm 0.61\%$ , respectively.

### 2.3 Preparation of CPoPc, SPoPc and CSPoCos

CPoPc and SPoPc were obtained using 100% CM and DSF, respectively, as the source of proteins. CSPoCos were obtained employing batches of 200 g pastes, with the following proportions: 70%CM : 30%DSF, 50%CM : 50%DSF, and 30%CM : 70%DSF (w/w) by alkaline extraction and isoelectric precipitation, following the method reported by Tian *et al.*<sup>12</sup> Briefly, CM, DSF and mixtures thereof were suspended in distilled water (1 : 20, w/v), adjusted to pH 11 with 1 M NaOH by magnetic stirring at 20 °C for 45 min, and the insoluble residue was separated by centrifugation at  $8000 \times g$  and 14 °C for 10 min. After, the pH of the protein suspensions was adjusted to 4 (for CM) and 4.5 (for DSF and mixtures) with 1 M HCl by magnetic stirring at 20 °C for 30 min. Subsequently, the slurries were centrifuged at  $8000 \times g$  at 14 °C for 10 min and PoPc and PoCos were re-suspended in distilled water in a ratio of 1 : 10 (w/v) at pH 7 with 1 M NaOH. Finally, the protein dispersions were lyophilized in a freeze dryer model FreeZone 10 L (Labconco, USA) to obtain CPoPc, SPoPc and CSPoCos (CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub>, where the first subindex in each material represents the CM proportion and the second subindex is the DSF proportion used in this preparation).



## 2.4 Extraction yield and protein purity

The extraction yield of the samples was calculated using two parameters (weight yield and protein yield). The weight yield was determined as the ratio between the total weight of the recovered CPoPc, SPoPc and/or CSPoCos and the initial weight of CM and/or DSF used in the extraction process (eqn (1)). The protein yield was estimated by comparing the total protein content in the recovered CPoPc, SPoPc and/or CSPoCos and the initial protein content in CM and/or DSF (eqn (2)). The protein purity was calculated according to eqn (3), as follows:

$$\text{Process yield(\%)} = \frac{\text{weight of recovered PoPc or PoCos}}{\text{weight of CM or DSF}} \times 100 \quad (1)$$

Protein yield(%)

$$= \frac{\text{weight of protein in recovered PoPc or PoCos}}{\text{weight of initial protein in CM or DSF}} \quad (2)$$

$$\text{Protein purity(\%)} = \frac{\text{weight of protein in PoPc or PoCos}}{\text{weight total of PoPc or PoCos}} \times 100 \quad (3)$$

## 2.5 Composition and physicochemical characteristics

**2.5.1 Proximate analysis.** CM, DSF, PoPc and PoCos samples were subjected to proximate composition following the AOAC methods.<sup>23</sup> The total carbohydrates were estimated by subtraction of the moisture, fat, protein, and ash content. Results were registered as % on a wet basis.

**2.5.2 Bulk ( $b_p$ ) and compact ( $c_p$ ) density.**  $b_p$  was determined by adding the protein sample to the 10 mL line of a graduated test tube and noting the weight. Then, the test tube was hit 40 times, and the volume was registered to obtain the  $c_p$ .<sup>24</sup>  $b_p$  and  $c_p$  were expressed as  $\text{g cm}^{-3}$ .

**2.5.3 Carr's index (CaIn) and Hausner's ratio (HaRa).** CaIn and HaRa are the two major parameters to quantify powder flowability. CaIn and HaRa were calculated from  $b_p$  and  $c_p$  of the protein samples, as follows:

$$\text{CaIn (\%)} = \frac{c_p - b_p}{c_p} \times 100 \quad (4)$$

$$\text{HaRa} = \frac{c_p}{b_p} \quad (5)$$

**2.5.4 Water activity ( $a_w$ ).** The  $a_w$  of the samples was analyzed with an AquaLab Serie 4 TEV equip (Decagon Devices Inc., USA).

**2.5.5 Turbidity ( $T_{rb}$ ).**  $T_{rb}$  was measured based on the absorbance of 1% (w/v) protein dispersions at pH 7, using a UV-vis spectrophotometer model FI-01620 (Thermo Fisher Scientific, Vantaa, Finland) at 600 nm.

**2.5.6 Color analysis.** A model CR-300 color meter (Konica Minolta Holdings, Inc., Japan) was utilized to evaluate the colors of the samples. The measurements were represented by

means of the CIELAB color scope, determining  $L^*$  (lightness),  $a^*$  (+redness, -greenness) and  $b^*$  (+yellowness, -blueness). The  $L_0^*$ ,  $a_0^*$ , and  $b_0^*$  numbers of the white pattern plate employed as the reference were 94.44, -0.20 and 3.87, respectively. The total variation in color ( $\Delta E$ ) was estimated as:

$$\Delta E = \left[ (L_0^* - L^*)^2 + (a_0^* - a^*)^2 + (b_0^* - b^*)^2 \right]^{1/2} \quad (6)$$

In addition, the hue angle ( $H_{ab}$ ) was calculated as the arc tangent of the ratio of yellowness to redness, as shown in eqn (7).

$$H_{ab} = \tan^{-1} \left( \frac{b^*}{a^*} \right) \quad (7)$$

Chroma ( $C^*$ ) is considered as the purity of color or colorfulness of a sample. It was determined as the Euclidean distance of the color vector ( $L$ ,  $a$ ,  $b$ ) for a sample from the lightness, axis as shown in eqn (8)

$$C^* = \sqrt{a^2 + b^2} \quad (8)$$

**2.5.7 Protein fractionation.** Albumin (Al), globulin (Gl), prolamin (Pr), and glutelin (Gx) fractions were solubilized from the protein samples successively, as was reported by Serrano-Sandoval *et al.*,<sup>25</sup> with some modifications. Initially, distilled water at a 1 : 10 (w/v) ratio with 1 h of stirring at 4 °C was used to extract Al, which was recovered in the supernatant obtained by centrifugation for 20 min at 6000×g and stored under refrigeration at 6 °C for subsequent analysis. To the resulting precipitate, 30 mL of a 50 mM tris-HCl + 0.4 M NaCl solution (pH 8) was added for Gl extraction, and the mixture was centrifuged under the same conditions. Next, 30 mL of 70% ethanol was added to the new precipitate to obtain Pr, and subsequently 30 mL of 0.1 M NaOH was added to the final precipitate to recover Gx. The separation and storage of the supernatant in each extraction stage were performed under the same conditions as for the albumin fraction. The total protein content of each fraction was measured using the Bradford assay<sup>26</sup> with albumin from bovine serum as the reference.

## 2.6 Functional characteristics

**2.6.1 Protein solubility (PoSo).** PoSo was determined as described by Balderas-León *et al.*,<sup>27</sup> with modifications. Samples containing 40 mg of protein were mixed with 20 mL 0.01 M sodium phosphate buffer (pH 7) for 60 min. Then, the dispersion was centrifuged at 8000×g for 10 min at 20 °C, and the supernatant was used for protein quantification by the Bradford<sup>22</sup> microplate assay method.

**2.6.2 Water holding capacity (WaHC) and oil holding capacity (OiHC).** To evaluate WaHC and OiHC, samples containing 0.5 g of protein were dispersed in 10 mL of distilled water or canola oil within a pre-weighed 50 mL centrifuge tube. The dispersions were vigorously vortexed in an MX-S (Avante, Guadalajara, Mexico) and then allowed to stand at room temperature for 30 min before centrifugation (5000×g, 20 min,



25 °C). After centrifugation, the supernatants were discarded and the tubes containing the pellets were weighed.<sup>28</sup> WaHC and OiHC were reported as grams of water or oil absorbed per gram of protein.

**2.6.3 Least gelation concentration (LaGeCo).** LaGeCo was determined using the procedure reported by López-Mártir *et al.*,<sup>28</sup> with some changes. Protein dispersions of 4% to 20% (w/v) were prepared in volumes of 5 mL at pH 7, which were kept in test tubes and subsequently heated at 90 °C for 1 h. After heating, the tubes containing the samples were quickly cooled and refrigerated at 4 °C for 2 h. LaGeCo was identified as the concentration when the sample from the inverted test tube did not fall down or slip.

**2.6.4 Emulsifying characterization.** The emulsifying activity index (EACIn) and emulsifying stability index (ESTIn) were measured according to the method of Briceño-Islas *et al.*<sup>29</sup> For emulsion formation, 5 mL of canola oil and 16 mL of sample solution (0.1% w/v) in 0.01 M sodium phosphate buffer (pH 7) were homogenized with a T-25 Ultra-Turrax homogenizer (IKA Instruments, Germany) at 12 000 rpm for 1 min. Then, 50 µL of emulsion from the bottom of the tube was taken after 0 min (for EACIn) and 10 min (for ESTIn) and added to 5 mL of 0.1% SDoS (sodium dodecyl sulfate) solution. After shaking in a vortex mixer for 5 s, the absorbance of the emulsions was recorded at 500 nm. EACIn and ESTIn were calculated using the following equations:

$$\text{EACIn (m}^2 \text{ g}^{-1}\text{)} = \frac{2T \times A_0 \times N \times 10^{-4}}{\emptyset \times L \times C} \quad (9)$$

$$\text{ESTIn(\%)} = \frac{A_{10}}{A_0} \times 100 \quad (10)$$

where  $T$  is a constant value (2.30),  $A_0$  and  $A_{10}$  are the absorbance of the emulsion at 0 and 10 min, respectively,  $N$  is the dilution factor (100),  $\emptyset$  is the oil volume fraction (0.20),  $L$  is the path length of the cuvette (m), and  $C$  is the protein concentration in the aqueous phase ( $\text{g m}^{-3}$ ).

**2.6.5 Foaming characterization.** The foam capacity (FmC) and foam stability (FmS) of the samples were measured following the procedure of Cháirez-Jiménez *et al.*,<sup>7</sup> with small variations. Protein suspensions were prepared (5%) with 0.01 M sodium phosphate buffer (pH 7). Then, a sample of 15 mL was mixed for 1 min at 13 000 rpm and poured into a graduated tube. The foam volume at 1 min was recorded as  $V_{1\text{min}}$ , the foam volume at 20 min as  $V_{20\text{min}}$ , and FmC and FmS were calculated as follows:

$$\text{FmC(\%)} = \frac{V_{1\text{min}}}{15} \times 100 \quad (11)$$

$$\text{FmS(\%)} = \frac{V_{20\text{min}}}{V_{1\text{min}}} \times 100 \quad (12)$$

## 2.7 Biochemical properties

**2.7.1 Electrophoretic profile by sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE).** SDS-PAGE of the protein samples was performed according to the method described by Flores-Jiménez *et al.*,<sup>30</sup> with slight changes,

utilizing a Mini-Protean Tetra cell (Bio-Rad Laboratories, Inc., USA). Samples were processed on 4–20% precast polyacrylamide gel (Cat. #456-1095). Samples were prepared under reducing conditions with  $\beta$ -mercaptoethanol (+ $\beta$ -M) and non-reducing conditions without  $\beta$ -mercaptoethanol (– $\beta$ -M). A total of 20 µL of each sample with 15 µg of protein was charged in each channel of gel, including the molecular weight marker (MW) from 10 to 250 kDa (Precision Plus Protein All Blue Pre-stained Protein Standards). Electrophoresis was performed at 140 V to concentrate for 10 min, and after that at 110 V to the divided gel for approximately 2 h. The gel was stained with Coomassie brilliant blue (G-250) for 24 h. Ultimately, the gel was destained until bands were easily seen and then captured using the ImageJ software (National Health Institute, USA) for molecular mass definition.

**2.7.2 In vitro protein enzymatic digestibility (IVPoDi).** The IVPoDi method of Flores-Jiménez *et al.*<sup>24</sup> was used with slight changes. Briefly, a dispersion was prepared by dissolving 625 mg of protein sample in 100 mL of distiller water, which was adjusted with NaOH (0.1 M) to pH 8. The multienzyme preparation was elaborated with trypsin, protease and bovine pancreatin at concentrations of 1.6 mg mL<sup>-1</sup>, 0.05 mg mL<sup>-1</sup> and 1 mg mL<sup>-1</sup>, respectively, adjusting the medium to pH 8 and keeping it in an ice bath until its application. Afterwards, 1.5 mL of the multienzyme preparation was incorporated into the sample and agitation maintained for 10 min at 37 °C. The decrease in pH was determined after 10 min of reaction. The IVPoDi was estimated as follows:

$$\text{IVPoDi (\%)} = 210.464 - 18.10(Z) \quad (13)$$

where  $Z$  is the pH of the sample dispersion after 10 min of reaction with the multienzyme preparation.

**2.7.3 In vitro antioxidant capacity (IViAC).** The IViAC of the protein samples was evaluated using the 1,1-diphenyl-2-picrylhydrazyl (DPPH) assay according to the method described by Mesfin *et al.*,<sup>31</sup> with modifications. An aliquot of 250 µL of 0.06 mM DPPH was added and mixed vigorously with 30 µL of protein suspension (0.5 mg mL<sup>-1</sup>), and then the mixture was incubated in the dark for 30 min at room temperature. The absorbance was monitored at 515 nm by spectrophotometry after 30 min, using a blank (250 µL of DPPH + 30 µL distiller water). Finally, IViAC was calculated according to the following equation:

$$\text{IViAC(\%)} = 1 - \left( \frac{A_{\text{sa}}}{A_{\text{bl}}} \right) \times 100 \quad (14)$$

where  $A_{\text{bl}}$  and  $A_{\text{sa}}$  are the absorbance of the blank and the sample, respectively.

## 2.8 Structural properties

**2.8.1 Microstructure.** A scanning electron microscope (SEM; SNE-3200M, SEC Co., Ltd, South Korea) was used to observe the microstructure and shape of the samples. A small amount of samples was taken, coated with gold, and placed in the SEM for suction under a voltage of 20 kV and a magnification of 150 $\times$ . Once the samples were clearly focused under magnification, their characteristics were photographed.



**2.8.2 Molecular flexibility (MoFl).** MoFl of the samples was measured at a concentration of 1 mg mL<sup>-1</sup> (0.01 M sodium phosphate buffer, pH 7), employing the technique described by Rodríguez-Rivera *et al.*<sup>32</sup>

**2.8.3 Particle size (Pz) and ζ-potential.** The Pz and ζ-potential were measured using a Zetasizer Nano-ZS90 (Malvern Instruments, Malvern, UK). The samples were diluted to 0.2 mg mL<sup>-1</sup> with 0.01 M sodium phosphate buffer (pH 7). To measure Pz, the samples were placed in a glass cuvette (ZEN0118, Malvern Instruments) at 20 °C. The ζ-potential was determined by measuring the velocity and direction of dispersion as the particles moved along the electric field using the Smoluchowski mathematical method. Measurements were performed at 20 °C, and the samples were placed in disposable folded capillary cells (DTS1060, Malvern Instruments).

**2.8.4 Surface hydrophobicity (S<sub>u</sub>H<sub>o</sub>).** Protein S<sub>u</sub>H<sub>o</sub> was estimated using the fluorescence technique,<sup>33</sup> utilizing 8-anilino-1-naphthalenesulfonic acid (ANS) as the molecular probe on a 200 Pro spectrophotometer (Tecan Infinite, Austria). The samples were diluted to different protein concentrations (0.05–0.01 mg mL<sup>-1</sup>), and the test tubes were brought to a volume of 2 mL according to the solubility of the samples. Then, 25 μL of ANS was added, to which fluorescence was determined at specific excitation (364 nm) and emission (475 nm) wavelengths. S<sub>u</sub>H was determined as the slope of the graph of fluorescence vs. protein concentration (mg mL<sup>-1</sup>).

**2.8.5 Intrinsic fluorescence spectra (IFS).** IFS of the samples (0.2 mg mL<sup>-1</sup> protein with 0.01 M sodium phosphate buffer at pH 7) were obtained with a 200 Pro fluorescence spectrophotometer (Tecan Infinite 200 Pro, Grödig, Austria) at an excitation wavelength of 290 nm, in the spectral scanning range of 320–450 nm, with a slit of 5 nm.

**2.8.6 Free (SH<sub>Fr</sub>) and total (SH<sub>To</sub>) sulfhydryl groups and disulfide bridges (S–S<sub>Br</sub>).** The determination of the content of SH<sub>Fr</sub>, SH<sub>To</sub> and S–S<sub>Br</sub> of the protein samples was carried out according to the methods described by Wu *et al.*<sup>34</sup> and Ren *et al.*<sup>35</sup> with some adaptations. To determine the SH<sub>Fr</sub> content, 100 μL of sample (1 mg mL<sup>-1</sup> protein with 0.01 M sodium phosphate buffer at pH 7) was dissolved in 500 μL of tris-glycine buffer (0.086 M tris, 0.09 M glycine, 0.04 M EDTA-Na<sub>2</sub>, pH 8) and 10 μL of Ellman's reagent (4 mg of DTNB per mL of tris-glycine buffer), followed by incubation at 25 °C for 1 h. Then the samples were centrifuged at 8000 rpm for 10 min using an Eppendorf MiniSpin centrifuge (Hamburg, Germany) and the absorbance was measured at 412 nm with sodium phosphate buffer as the blank. For the analysis of SH<sub>To</sub> content, the same process was followed, with the addition of 8 M urea to the tris-glycine buffer. The concentration of SH<sub>Fr</sub>, SH<sub>To</sub> and S–S<sub>Br</sub> was calculated using the following equations:

$$\text{SH}_{\text{Fr}} \text{ or } \text{SH}_{\text{To}} (\mu\text{mol per g protein}) = \frac{73.53 \times A_{412} \times D}{C} \quad (15)$$

$$S - S_{\text{Br}} (\mu\text{mol per g protein}) = \frac{\text{SH}_{\text{To}} - \text{SH}_{\text{Fr}}}{2} \quad (16)$$

where  $A_{412}$  is the absorbance of the sample at 412 nm, for both SH<sub>Fr</sub> and SH<sub>To</sub>,  $C$  is the sample protein concentration (mg

mL<sup>-1</sup>), 73.53 is a constant, and  $D$  is the dilution factor ( $D = 1$ ).**2.8.7 Attenuated total reflectance fourier transform infrared spectra (ATR-FTIR).** Infrared spectra of the protein samples were obtained using an ATR-FTIR spectrometer (Agilent Cary 630, Agilent Technologies, USA). 100 mg of protein sample was placed in a ZnSe cell, using the conditions of wavelength range of 650 to 4000 cm<sup>-1</sup> with a resolution of 4 cm<sup>-1</sup> and 21 sweeps at 25 °C.

## 2.9 Statistical analysis

The physicochemical, functional, biochemical and structural characteristics of PoPc and CSPoCos were reported as the mean ± standard deviation of three batches. Data analysis involved one-way ANOVA and comparison of means was performed using Fisher's Protected LSD (statistical significance considered at  $p < 0.05$ ) using Statistica software version 7.1 (TIBCO Software, Inc., USA).

## 3 Results and discussion

### 3.1 Extraction yield and protein purity

The yield and purity of protein materials are important factors associated with their production methods, which determine their economic feasibility.<sup>35</sup> The results for the product yield, protein recovery, and purity of PoPc and CSPoCos are presented in Table 1. The product yield ranged from 5.20%–17.10%, following the order of CPoPc < CSPoCo<sub>70/30</sub> < CSPoCo<sub>50/50</sub> < CSPoCo<sub>30/70</sub> < SPoPc. These results are comparable to those of Wintersohle *et al.*,<sup>36</sup> who found a product yield ranging from 8.3% to 18.89% for mung bean PoPc obtained by alkaline extraction and isoelectric precipitation. Meanwhile, the protein recovery reached values of 11.76% to 53.07% in the order of CPoPc < CSPoCo<sub>70/30</sub> < CSPoCo<sub>50/50</sub> < CSPoCo<sub>30/70</sub> < SPoPc. The maximum protein recovery obtained in this study was similar to that reported for the PoPc of sunflower (55.67%),<sup>6</sup> *Hermetia illucens* (55%),<sup>37</sup> and *Japanese quince* (54%).<sup>38</sup> In the case of protein purity, PoC significantly increased ( $p < 0.05$ ) this parameter from 82.18% and 86.22% for CPoPc and SPoPc, respectively, to values ranging from 88.25% to 92.20% for CSPoCos (Table 1). The purity of the protein obtained in the present study for both PoPc (82.18–86.22%) and PoCos (88.25–92.20%) ranged between those reported for Japanese quince PoPc (83.88%).<sup>39</sup>

### 3.2 Composition and physicochemical characteristics

**3.2.1 Proximate composition.** The protein content of CSPoCos (88.19–91.21%) was higher than that of CPoPc (82.25%) and SPoPc (86.19%). The variation in the protein content of CSPoCos could be due to the precipitation pH used (4.5), which was the same for sunflower proteins, but different from that of canola proteins (4.0), thus also affect their techno-functional properties. In general, the contents of moisture, ash, lipids and total carbohydrates were reduced by effect of PoC, in contrast with PoPc, except CSPoCo<sub>70/30</sub>, due to the fact that its moisture value was higher than that of SPoPc, but in the case of lipid content, all CSPoCos showed no significant difference ( $p > 0.05$ ) with respect to SPoPc (Table 2).



**Table 1** Process yield, protein recovery and protein purity of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Samples	Process yield (%)	Protein recovery (%)	Protein purity (%)
CPoPc	5.20 ± 0.26 <sup>c</sup>	11.76 ± 0.60 <sup>e</sup>	82.18 ± 0.02 <sup>e</sup>
SPoPc	17.10 ± 0.36 <sup>a</sup>	53.07 ± 1.16 <sup>a</sup>	86.22 ± 0.06 <sup>d</sup>
CSPoCo <sub>70/30</sub>	11.40 ± 0.53 <sup>d</sup>	29.92 ± 1.37 <sup>d</sup>	88.25 ± 0.13 <sup>c</sup>
CSPoCo <sub>50/50</sub>	13.27 ± 0.25 <sup>c</sup>	37.18 ± 0.70 <sup>c</sup>	89.15 ± 0.03 <sup>b</sup>
CSPoCo <sub>30/70</sub>	16.23 ± 0.40 <sup>b</sup>	50.48 ± 1.27 <sup>b</sup>	92.20 ± 0.02 <sup>a</sup>

<sup>a</sup> Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same column indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.

The protein content of CSPoCos was higher than pea-grass carp (87.91%)<sup>19</sup> and soybean-brewers' spent grain (83.67%)<sup>40</sup> PoCos but lower than those of soybean-wheat (91.5–94%)<sup>12</sup> and sheep whey-casein (97.5%)<sup>41</sup> PoCos.

The proximate composition of PoPc from canola<sup>42</sup> and sunflower<sup>43</sup> was similar to that of CPoPc and SPoPc obtained in this study.

**3.2.2  $b_p$  and  $c_p$ .** The density of food powders is classified into two categories:  $b_p$  and  $c_p$ . These densities serve as references for evaluating the behavior of powders under external forces.<sup>44</sup> The  $b_p$  of CSPoCos (0.366–0.372 g cm<sup>-3</sup>) was not significantly ( $p > 0.05$ ) different from that of SPoPc (0.358 g cm<sup>-3</sup>), but it was higher than that of CPoPc (0.326 g cm<sup>-3</sup>) (Table 3).

Regarding  $c_p$ , significantly ( $p < 0.05$ ) higher values were observed for CSPoCos (0.482–0.499 g cm<sup>-3</sup>) compared to CPoPc (0.430 g cm<sup>-3</sup>) and SPoPc (0.436 g cm<sup>-3</sup>) (Table 3). According to various studies,  $b_p$  and  $c_p$  are influenced by factors such as particle size, size distribution, shape, surface area, roughness, and particle arrangement of powder products.<sup>45,46</sup> The  $b_p$  and  $c_p$  values obtained in this study for CPoPc and SPoPc of 0.326 g cm<sup>-3</sup> and 0.430 g cm<sup>-3</sup> and 0.358 g cm<sup>-3</sup> and 0.436 g cm<sup>-3</sup> were higher than those of 0.156 g cm<sup>-3</sup> and 0.200 g cm<sup>-3</sup> obtained for soursop seed PoPc, respectively.<sup>28</sup>

**3.2.3 CaIn and HaRa.** CaIn is an indirect measure of the  $b_p$ , size and shape, surface area, moisture content, and cohesion of powder materials.<sup>27</sup> According to Shah *et al.*,<sup>47</sup> powders can be classified with respect to their flow properties as follows: very good (CaIn < 15%), good (CaIn = 15–20%), fair (CaIn = 20–

35%), bad (CaIn = 35–45%) or very bad flowing (CaIn > 45%). Conforming to this scale, CPoPc (24.24%), CSPoCo<sub>70/30</sub> (25.69%), CSPoCo<sub>50/50</sub> (23.73%), and CSPoCo<sub>30/70</sub> (24.05%) had fair fluency, while SPoPc had good fluidity (17.74%) (Table 3).

Conversely, HaRa measures the cohesion of particles, and therefore its value determines high or low fluidity in powders.<sup>48</sup> Powders with an HaRa of 1.2 are classified as free-flowing, while those with values of 1.2–1.4 or >1.4 are considered to be of intermediate or high cohesion, respectively.<sup>30</sup> CPoPc, SPoPc, CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub>, and CSPoCo<sub>30/70</sub> had HaRa values of 1.32, 1.22, 1.35, 1.31 and 1.32, respectively, and therefore were classified as intermediate cohesion powders (Table 3). Specifically, the optimal CaIn and HaRa values depend on the potential applications of the powdered products. Therefore, in some cases, higher or lower CaIn and HaRa values may be preferable.<sup>48</sup>

López-Mártir *et al.*<sup>28</sup> reported for soursop seed PoPc fair flowability (CaIn of 22%) and intermedia cohesiveness (HaRa of 1.28), although Rodríguez-Rivera *et al.*<sup>32</sup> found that groundnut paste PoPc presented good flowability (CaIn of 18.33%) and intermediate cohesiveness (HaRa of 1.22).

**3.2.4  $a_w$ .**  $a_w$  is used to determine the critical storage conditions of food. This parameter indicates if powders are susceptible to deterioration changes such as collapse, stickiness and caking.<sup>49</sup> The  $a_w$  of CPoPc, SPoPc, and CSPoCos varied significantly ( $p < 0.05$ ), as shown in Table 3. CSPoCo<sub>30/70</sub> (0.11) had the smallest  $a_w$  value, followed by CSPoCo<sub>50/50</sub> (0.14), SPoPc (0.20) and CSPoCo<sub>70/30</sub> (0.25), while CPoPc presented the highest value (0.41). Flores-Jiménez *et al.*<sup>30</sup> pointed out that values

**Table 2** Proximate chemical composition of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Components (%)	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
Protein	82.25 ± 0.56 <sup>d</sup>	86.19 ± 0.32 <sup>c</sup>	88.19 ± 0.45 <sup>b</sup>	89.17 ± 0.54 <sup>b</sup>	91.21 ± 0.92 <sup>a</sup>
Moisture	4.42 ± 0.13 <sup>a</sup>	3.20 ± 0.04 <sup>c</sup>	3.63 ± 0.03 <sup>b</sup>	2.80 ± 0.09 <sup>d</sup>	3.21 ± 0.07 <sup>c</sup>
Ash	3.83 ± 0.11 <sup>a</sup>	3.67 ± 0.29 <sup>a</sup>	2.74 ± 0.08 <sup>b</sup>	2.78 ± 0.02 <sup>b</sup>	2.40 ± 0.13 <sup>c</sup>
Lipids	1.23 ± 0.30 <sup>a</sup>	0.95 ± 0.58 <sup>ab</sup>	0.91 ± 0.57 <sup>b</sup>	0.86 ± 0.02 <sup>b</sup>	0.87 ± 0.10 <sup>b</sup>
Carbohydrates	8.27 ± 0.55 <sup>a</sup>	5.99 ± 0.37 <sup>b</sup>	4.53 ± 0.73 <sup>c</sup>	4.39 ± 0.52 <sup>c</sup>	2.31 ± 0.43 <sup>d</sup>

<sup>a</sup> Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same row indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.



**Table 3** Physicochemical properties of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Properties	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
<b>Density</b>					
$b_p$ (g cm <sup>-3</sup> )	0.326 ± 0.01 <sup>b</sup>	0.358 ± 0.01 <sup>a</sup>	0.370 ± 0.01 <sup>a</sup>	0.372 ± 0.01 <sup>a</sup>	0.366 ± 0.00 <sup>a</sup>
$c_p$ (g cm <sup>-3</sup> )	0.430 ± 0.01 <sup>d</sup>	0.436 ± 0.02 <sup>d</sup>	0.499 ± 0.01 <sup>ab</sup>	0.488 ± 0.01 <sup>bc</sup>	0.482 ± 0.00 <sup>c</sup>
<b>Flow characteristics</b>					
CaIn (%)	24.24 ± 2.19 <sup>a</sup>	17.74 ± 1.72 <sup>b</sup>	25.69 ± 1.21 <sup>a</sup>	23.73 ± 1.76 <sup>a</sup>	24.05 ± 1.02 <sup>a</sup>
Flow	Fair	Good	Fair	Fair	Fair
HaRa	1.32 ± 0.04 <sup>a</sup>	1.22 ± 0.03 <sup>b</sup>	1.35 ± 0.02 <sup>a</sup>	1.31 ± 0.03 <sup>a</sup>	1.32 ± 0.02 <sup>a</sup>
Cohesion	Intermediate	Intermediate	Intermediate	Intermediate	Intermediate
$a_w$	0.41 ± 0.01 <sup>a</sup>	0.20 ± 0.03 <sup>c</sup>	0.25 ± 0.01 <sup>b</sup>	0.14 ± 0.01 <sup>d</sup>	0.11 ± 0.01 <sup>c</sup>
$T_{rb}$	0.11 ± 0.00 <sup>e</sup>	0.90 ± 0.01 <sup>a</sup>	0.34 ± 0.01 <sup>d</sup>	0.41 ± 0.01 <sup>c</sup>	0.48 ± 0.01 <sup>b</sup>

<sup>a</sup>  $b_p$  = bulk density,  $c_p$  = compact density, CaIn = Carr's index, HaRa = Hausner's ratio,  $a_w$  = water activity, and  $T_{rb}$  = turbidity. Values are the mean of three determinations of independent batches ± standard deviations. Different letters within the same row indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.

less than 0.66 for  $a_w$  are beneficial to the stability of the powders due to the reduced availability of water for microbial proliferation and biochemical reactions, which increases the useful life of food products.

The  $a_w$  values of 0.41, 0.21 and 0.31 have been reported for PoPc of canola,<sup>42</sup> guamuchil,<sup>24</sup> and gourd,<sup>50</sup> respectively.

**3.2.5  $T_{rb}$ .**  $T_{rb}$  is an important property and criterion for defining the formation of protein aggregates in macroscopic dimensions.<sup>51</sup> Table 3 shows the  $T_{rb}$  values of PoPc and PoCos. CSPoCos (0.34–0.48) were more turbid compared to CPoPc (0.11), although less turbid than SPoPc (0.90). These results could be due to the variation in the particle size (Fig. 6B) of the canola and sunflower proteins, resulting from the different interactions between the two proteins in PoC, which consequently impact their functional properties (Fig. 2A). Malik *et al.*<sup>52</sup> and Flores-Jiménez *et al.*<sup>24</sup> previously reported  $T_{rb}$  values of 0.97 and 1.05 in sunflower and Guamuchil PoPc, respectively, which were higher than those obtained for all the protein materials in this study.

**3.2.6 Color.** Color is one of the most important characteristics of a food product, which is generally considered a quality factor describing its level of freshness, ripeness, consumer acceptance, and safety.<sup>53</sup> Table 4 shows the values of the color parameters  $L^*$ ,  $a^*$ ,  $b^*$ ,  $C^*$ ,  $H$ , and  $\Delta E$  for PoPc and PoCos. The highest  $L^*$  values were observed for CPoPc (51.01) and CSPoCo<sub>70/30</sub> (54.07), which contain mainly canola proteins, compared to SPoPc (40.45), CSPoCo<sub>50/50</sub> (47.34), CSPoCo<sub>30/70</sub> (44.41) that have a greater amount of sunflower proteins (Fig. 1).

All PoPc and PoCos showed positive values for  $a^*$  (except SPoPc and CSPoCo<sub>30/70</sub>) and  $b^*$  chromaticity (Table 4). High  $b^*$  values indicate a yellow tone with a slight reddish tone due to small  $a^*$  values (Fig. 1). SPoPc and CSPoCo<sub>30/70</sub> exhibited a slightly greenish hue, as expressed by their negative  $a^*$  values, because these protein samples contained a considerable amount of sunflower proteins compared to the other samples (Fig. 1).

In a study with pea-grass carp, PoCos<sup>19</sup> were found to have  $L^*$ ,  $a^*$  and  $b^*$  values of 74.21, -0.44, and 12.92, respectively. In their study, Flores-Jiménez *et al.*<sup>42</sup> reported  $L^*$ ,  $a^*$ , and  $b^*$  values

of 51.27, 3.38, and 15.17 for canola PoPc, respectively, while Alexandrino *et al.*<sup>6</sup> obtained values of 51.66 ( $L^*$ ), 7.02 ( $a^*$ ), and 2.21 ( $b^*$ ) for sunflower PoPc.

$H_{ab}$  indicates whether an object is red (0°), yellow (90°), green (180°) and (270°).<sup>50</sup> SPoPc and CSPoCo<sub>30/70</sub> showed  $H_{ab}$  values of 100.88° and 93.14°, respectively, confirming their greenish–yellowish tones due to a higher proportion of sunflower protein. In contrast, the lower  $H_{ab}$  values for CPoPc (83.19°), CSPoCo<sub>50/50</sub> (84.13°), and CSPoCo<sub>70/30</sub> (80.65°) indicated their reddish–yellowish tones, which can be associated with a higher content of canola proteins. Furthermore, CPoPc and CSPoCo<sub>30/70</sub> presented higher  $C^*$  values compared to SPoPc, CSPoCo<sub>50/50</sub> and CSPoCo<sub>70/30</sub>, indicating their greater color intensity/vividity.

Naik *et al.*<sup>54</sup> and López-Mártir *et al.*<sup>55</sup> previously reported  $C^*$  values of 18.25 and 13.87 in melon and noni PoPc, respectively, which were higher than those obtained for SPoPc, CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub> in this study.

**3.2.7 Protein fractionation.** Storage proteins are classified according to the Osborne method into Al, Gl, Gx and Pr based on their PoSo in different solvents.<sup>50</sup> Table 5 shows the protein fractionation of PoPc and PoCos from canola and sunflower. The data showed that Gx and Al were the major fractions, while Gl and Pr were the minor fractions in both PoPc and PoCos. In comparison with CPoPc, CSPoCos had a higher Al content due to the greater inclusion of sunflower protein, while the Gl, Gx and Pr contents diminished. Interestingly, the higher Al content and lower Gx content in CSPoCos produced an improvement in PoSo (Fig. 2A).

As in this study, Gx and Al were the main protein fractions in PoPc obtained from canola (glutelins 57.18% and albumins 23.09%),<sup>42</sup> orange seeds (glutelins 70.13% and albumins 10.39%),<sup>56</sup> and noni seeds (glutelins 64.62% and albumins 24.12%).<sup>55</sup>

### 3.3 Functional characteristics

**3.3.1 PoSo.** PoSo is a function of hydrophilic–hydrophobic balance and electrostatic repulsion and strongly influences



**Table 4** Color of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Color	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
<i>L</i> <sup>*</sup>	51.01 ± 0.37 <sup>b</sup>	40.45 ± 0.54 <sup>c</sup>	54.07 ± 0.38 <sup>a</sup>	47.34 ± 0.55 <sup>c</sup>	44.41 ± 0.83 <sup>d</sup>
<i>a</i> <sup>*</sup>	2.11 ± 0.03 <sup>b</sup>	−0.92 ± 0.07 <sup>e</sup>	2.84 ± 0.21 <sup>a</sup>	1.16 ± 0.10 <sup>c</sup>	−0.45 ± 0.02 <sup>d</sup>
<i>b</i> <sup>*</sup>	18.56 ± 0.16 <sup>a</sup>	4.72 ± 0.16 <sup>c</sup>	17.21 ± 0.76 <sup>b</sup>	11.27 ± 0.18 <sup>c</sup>	8.32 ± 0.22 <sup>d</sup>
$\Delta E$	49.25 ± 0.21 <sup>d</sup>	57.68 ± 0.62 <sup>a</sup>	46.10 ± 0.56 <sup>c</sup>	51.10 ± 0.01 <sup>c</sup>	53.57 ± 0.27 <sup>b</sup>
<i>C</i> <sup>*</sup>	18.19 ± 0.32 <sup>a</sup>	4.81 ± 0.17 <sup>e</sup>	17.44 ± 0.79 <sup>b</sup>	11.33 ± 0.19 <sup>c</sup>	8.33 ± 0.22 <sup>d</sup>
<i>H</i> <sub>ab</sub>	83.19 ± 0.44 <sup>d</sup>	100.88 ± 0.30 <sup>a</sup>	80.65 ± 0.37 <sup>c</sup>	84.13 ± 0.44 <sup>c</sup>	93.14 ± 0.16 <sup>b</sup>

<sup>a</sup> *L*<sup>\*</sup> = lightness, *a*<sup>\*</sup> = +redness, −greenness, *b*<sup>\*</sup> = +yellowness, −blueness,  $\Delta E$  = total variation of color, *C*<sup>\*</sup> = chroma, *H*<sub>ab</sub> = hue angle. Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same row indicate significant differences (*p* < 0.05) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.

other techno-functional properties such as foaming, emulsifying and gelling characteristics.<sup>57</sup> The PoSo values for CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub> were 1.25, 1.58 and 1.50 mg mL<sup>−1</sup>, respectively, which are higher than those obtained with CPoPc (0.41 mg mL<sup>−1</sup>), but lower than those of SPoPc (1.69 mg mL<sup>−1</sup>); specifically, the PoSo of CSPoCos increased in the range of 204.9% (CSPoCo<sub>70/30</sub>) to 285.4% (CSPoCo<sub>50/50</sub>) compared to CPoPc (Fig. 2A).

The increase in the PSo of CSPoCos was due to the higher inclusion of sunflower proteins that contain a higher proportion of the Al protein fraction (Table 5). Moreover, the fusion of canola and sunflower polymers may have strengthened the intermolecular electrostatic interactions to generate soluble protein aggregates, as can be observed in Fig. 6B. In a study by Zhou *et al.*<sup>21</sup> with PoCo from pea and grass carp using alkaline extraction-isoelectric precipitation, an increase in PoSo (at pH 7) of 14.3% and 22.9% was observed, in contrast to the pea and grass carp proteins, respectively. In addition, Tian *et al.*<sup>12</sup> reported that PoCo from soybean and wheat proteins had a higher PoSo value (pH 7) than soybean and wheat protein isolates. This modification in PoCo from these protein sources was due to the secondary and tertiary structural changes, in contrast to soybean and wheat protein isolates.

**3.3.2 WaHC and OiHC.** WaHC is associated with the formation of gels and texture in food formulations.<sup>58</sup> Oil binding as measured by OiHC is an important functional property for improving palatability and mouthfeel, as well as

flavor retention of certain foods such as emulsions, dairy products, sausages, and bread.<sup>24</sup> Fig. 2B shows the WaHC and OiHC of PoPc and CSPoCos. The WaHC values of CSPoCos (2.76–2.95 g water per g protein) were higher than those of CPoPc (2.04 g water per g protein) and SPoPc (2.35 g water per g protein), except for CSPoCo<sub>30/70</sub> (2.54 g water per g protein). Consistent with the results obtained in this study, Zhou *et al.*<sup>19</sup> reported an improvement in WaHC of 9.8–13.04% for pea-grass carp PoCo compared to individual proteins of pea and grass carp, respectively, which could be due to the interaction between both protein sources promoting a higher affinity with water. In another study, with PoCo from pea and lentil (1 : 3; w/w), WaHC increased by 6.6% and 9.8%, in contrast to pea protein isolate and lentil protein isolate, respectively.<sup>59</sup>

Conversely, the OiHC of CSPoCos (3.88–4.22 g oil per g protein) was significantly higher (*p* < 0.05) compared to that of CPoPc (2.89 g oil per g protein) and SPoPc (3.17 g oil per g protein) (Fig. 2B). These results may be due to PoC inducing greater exposure of the hydrophobic groups (Fig. 6D), which facilitated better interaction with the oil and led to an increase in OiHC.

**3.3.3 LaGeCo.** Gelation of globular proteins occurs in two stages. In the first stage, partial denaturation or conformational modification of these polymers occurs. Then, in the second stage, gradual association or aggregation occurs in a three-dimensional structure that traps water, fat and other components.<sup>60</sup> Fig. 2C shows the results of LaGeCo for PoPc and PCos from canola and sunflower. CSPoCos exhibited a significant



**Fig. 1** Effect of the coprecipitation on color appearance. CPoPc is the canola protein precipitate. SPoPc is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower.



**Table 5** Protein fractional composition of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Protein fraction (%)	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
Albumin (Al)	9.33 ± 0.51 <sup>d</sup>	46.22 ± 1.11 <sup>a</sup>	31.28 ± 1.01 <sup>c</sup>	30.57 ± 1.64 <sup>c</sup>	38.74 ± 1.13 <sup>b</sup>
Globulin (Gl)	9.31 ± 0.63 <sup>a</sup>	9.64 ± 0.52 <sup>a</sup>	7.15 ± 0.15 <sup>c</sup>	8.45 ± 0.76 <sup>ab</sup>	7.63 ± 0.53 <sup>bc</sup>
Glutelin (Gx)	78.15 ± 0.69 <sup>a</sup>	43.77 ± 1.09 <sup>d</sup>	60.30 ± 0.43 <sup>b</sup>	59.96 ± 2.04 <sup>b</sup>	52.71 ± 1.13 <sup>c</sup>
Prolamin (Pr)	3.21 ± 0.53 <sup>a</sup>	0.98 ± 0.36 <sup>bc</sup>	1.26 ± 0.07 <sup>b</sup>	1.02 ± 0.09 <sup>bc</sup>	0.92 ± 0.06 <sup>c</sup>

<sup>a</sup> Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same row indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.

reduction ( $p < 0.05$ ) in LaGeCo with respect to CPoPc and SPoPc. The highest reduction in LaGeCo of 83% was observed with CSPoCo<sub>70/30</sub>, followed by 66.7% and 50% for CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub>, respectively. The improved gelling characteristics of CSPoCos could be due to interactions such as disulfide bonds (Table 6) and non-covalent bonds (Fig. 6D) between polypeptide chains, thus forming larger aggregates. According to Flores-Jiménez *et al.*,<sup>61</sup> proteins with a low  $\alpha$ -helix content and a high percentage of  $\beta$ -sheet structure form stronger gels (Table 7). Similarly, other researchers have reported that PoC improves the gelation properties of proteins by reducing LaGeCo. For example, Niu *et al.*<sup>62</sup> demonstrated that the PoPc of soybean and myofibrillar proteins improved their gelation properties by up to 20% compared to the individual sources. Furthermore, Wu *et al.*<sup>63</sup> observed an 83.3% reduction in LaGeCo during gel formation of PoPc from soybean and cod compared to the original protein sources.

**3.3.4 Emulsion characteristics.** One of the most important types of emulsions in the food industry are those prepared from an oil–water mixture (immiscible liquids), using proteins as interfacial stabilizers to prevent coalescence.<sup>64</sup> PoC significantly increased ( $p < 0.05$ ) the EACIn of all CSPoCos, reaching up to 44.27–63.28% for CSPoCo<sub>70/30</sub>, in comparison with CPoPc and SPoPc, respectively, followed by 37.4–55.5% for CSPoCo<sub>50/50</sub> (Fig. 2D). On the contrary, PoC decreased ESTIn, inducing a reduction of up to 20.6–28.6% for CSPoCo<sub>50/50</sub>, in contrast to CPoPc and SPoPc, respectively. According to Zhou *et al.*,<sup>21</sup> PoCo from pea and grass carp proteins improved EACIn and ESTIn compared to the pea PoPc. These authors pointed out that the improved emulsifying properties are due to the fact that the pea and grass carp protein molecules have opposite charges. This situation can cause intermolecular electrostatic interactions, and therefore the formation of soluble protein aggregates (Fig. 6B), which trigger the reconstruction of hydrophobic (Fig. 6D) and disulfide bonds (Table 6). Conversely, Awal *et al.*<sup>65</sup> studied the formation of PoCo from Bambara groundnut protein isolate and fish collagen. The results of this study showed a decrease in EACIn (47.6%) and an increase in ESTIn (63.9%) for PoCo compared to the Bambara groundnut protein isolate. The authors concluded that the increase in ESTIn is due to the fish proteins, which promote electrostatic interactions, generating more stable emulsions.

**3.3.5 Foam characterization.** Foams are dispersions of gas bubbles within a continuous phase (usually water), which can

be generated by bubbling, stirring or pouring, with relevant applications in food.<sup>60</sup> As shown in Fig. 2E, the FmC of CSPoCos was significantly improved ( $p < 0.05$ ), with the largest increase of 23.42% and 16.74% for CSPoCo<sub>50/50</sub> compared to CPoPc and SPoPc, respectively, except for CSPoCo<sub>70/30</sub>, in which a reduction in this property was observed.

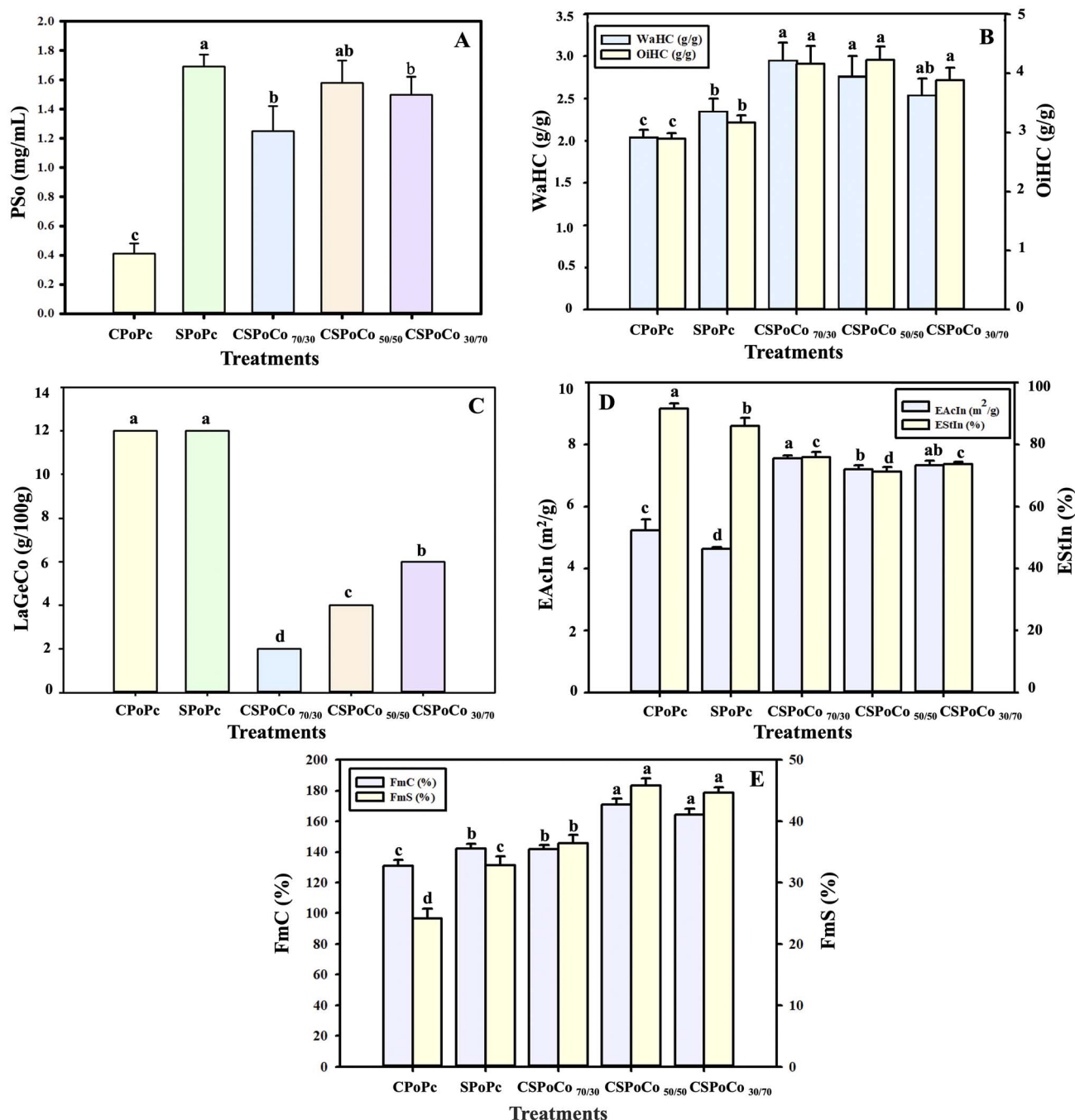
In addition, PoC significantly ( $p < 0.05$ ) increased the FmS of CSPoCos with respect to CPoPc and SPoPc. The highest increase in FmS of 47.26% and 28.44% was observed for CSPoCo<sub>50/50</sub> compared to CPoPc and SPoPc, respectively (Fig. 2E). In general, it was observed that the higher the proportion of sunflower proteins in PoCos, the better the foaming properties achieved. Furthermore, the increase in the FoP of CSPoCos was possibly due to the interaction between the canola and sunflower protein molecules, resulting in smaller particles and an increase in  $S_0H_0$  (Fig. 6B and D). In a study by Kristensen *et al.*<sup>9</sup> with a PoCo of pea and whey, an increase in foaming properties was reported, in contrast to the pea PoPc, possibly due to the interaction of the polymers used, which generated smaller particles.

### 3.4 Biochemical properties

**3.4.1 SDS-PAGE.** SDS-PAGE analysis is used to quantify, compare and characterize proteins based on their molecular weight.<sup>66</sup> The impact of PoC on the electrophoretic patterns of canola and sunflower proteins under non-reducing and reducing conditions is shown in Fig. 3.

Under non-reducing conditions (Fig. 3A), the molecular weight distribution of CPoPc consisted of four main fractions with the following molecular weights: ~49 kDa, 30 kDa, 25 kDa, and 15 kDa, which correspond to the molecular weights of cruciferin (~50–20 kDa) and napin (~7–11 kDa) of the protein fractions reported by Cháirez-Jiménez *et al.*<sup>7</sup> However, in the presence of the reducing agent  $\beta$ -mercaptoethanol (Fig. 3B), the band at ~49 kDa for CPoPc was dissociated into its two polypeptides (~20 kDa and ~30 kDa), as reported by Flores-Jiménez *et al.*<sup>42</sup> In the case of SPoPc, the main bands detected were ~49, 38, 35, 29, 25 and 8 kDa under non-reducing conditions (Fig. 3A). The subunits with a molecular weight between 20 and 50 kDa correspond to helianthinin (11S globulins), while those between 10–19 kDa correspond to 2S albumin. Under reducing conditions (Fig. 3B), 11S globulin dissociated into its two acidic subunits with a molecular weight of ~38–49 kDa and 25–29 kDa, as well as in its basic subunits with a molecular weight of 20–23 kDa.<sup>67</sup>





**Fig. 2** Effect of coprecipitation on (A) protein solubility (PoSo), (B) water holding capacity (WaHC)/oil holding capacity (OiHC), (C) least gelation concentration (LaGeCo), (D) emulsifying activity index (EAcln)/emulsifying stability index (ESltn) and (E) foam capacity (FmC)/foam stability (FmS). CPOpc is the canola protein precipitate. SPOpc is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower. Distinct letters on the bars indicate significant ( $p < 0.05$ ) difference (average  $\pm$  SD,  $n = 3$  independent batches).

Alternatively, under non-reducing conditions, CSPoCos presented seven bands with molecular weights of  $\sim$ 49, 38, 35, 29, 25, 18, and 15 kDa, which are related to the 11S globulins and 2S albumins fractions of canola and sunflower proteins (Fig. 3A). Under reducing conditions (Fig. 3B), the molecular weight distribution of CSPoCos consisted of nine main fractions, with molecular weights of  $\sim$ 49, 38, 35, 29, 25, 23, 20, 18, and 12 kDa.

Interestingly, all the protein bands from canola and sunflower were present in all CSPoCos under both reducing and non-reducing conditions. However, under non-reducing conditions, an additional 18 kDa band appeared in CSPoCos (Fig. 3A), with this band in CSPoCos<sub>70/30</sub> being more intense compared to the other coprecipitates, which is likely due to its higher canola content. According to Guidi *et al.*,<sup>59</sup> protein interactions can be



**Table 6** Free ( $\text{SH}_{\text{Fr}}$ ) and total ( $\text{SH}_{\text{To}}$ ) sulfhydryl groups and disulfide bridges ( $\text{S-S}_{\text{Br}}$ ) of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCos)<sup>a</sup>

Sulfhydryl groups ( $\mu\text{mol g}^{-1}$ protein)	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
Free ( $\text{SH}_{\text{Fr}}$ )	12.47 ± 0.48 <sup>d</sup>	17.87 ± 0.07 <sup>a</sup>	16.05 ± 0.04 <sup>b</sup>	15.02 ± 0.26 <sup>c</sup>	14.98 ± 0.19 <sup>c</sup>
Total ( $\text{SH}_{\text{To}}$ )	16.20 ± 0.68 <sup>b</sup>	20.37 ± 0.3 <sup>a</sup>	16.67 ± 0.42 <sup>b</sup>	16.18 ± 0.13 <sup>b</sup>	16.79 ± 0.49 <sup>b</sup>
Bridges ( $\text{SS}_{\text{Br}}$ )	1.87 ± 0.60 <sup>b</sup>	1.25 ± 0.28 <sup>a</sup>	0.31 ± 0.40 <sup>c</sup>	0.58 ± 0.17 <sup>c</sup>	0.91 ± 0.41 <sup>bc</sup>

<sup>a</sup> Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same row indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower pastes.

**Table 7** Secondary structure of the canola protein precipitate (CPoPc), sunflower protein precipitate (SPoPc) and canola–sunflower protein coprecipitates (CSPoCo)<sup>a</sup>

Secondary structure (%)	CPoPc	SPoPc	CSPoCo <sub>70/30</sub>	CSPoCo <sub>50/50</sub>	CSPoCo <sub>30/70</sub>
$\beta$ -sheet	49.55 ± 0.68 <sup>c</sup>	60.96 ± 1.10 <sup>a</sup>	52.13 ± 0.45 <sup>b</sup>	51.42 ± 0.28 <sup>b</sup>	52.58 ± 0.69 <sup>b</sup>
$\alpha$ -helix	8.40 ± 0.87 <sup>b</sup>	14.64 ± 0.53 <sup>a</sup>	8.79 ± 0.42 <sup>b</sup>	8.67 ± 0.77 <sup>b</sup>	8.64 ± 0.58 <sup>b</sup>
$\beta$ -turn	29.77 ± 0.54 <sup>a</sup>	10.72 ± 1.08 <sup>d</sup>	23.81 ± 0.97 <sup>c</sup>	28.39 ± 0.35 <sup>b</sup>	24.11 ± 0.44 <sup>c</sup>
Random coil	11.57 ± 0.39 <sup>b</sup>	13.67 ± 0.54 <sup>a</sup>	14.44 ± 0.32 <sup>a</sup>	12.07 ± 0.15 <sup>b</sup>	14.63 ± 0.45 <sup>a</sup>

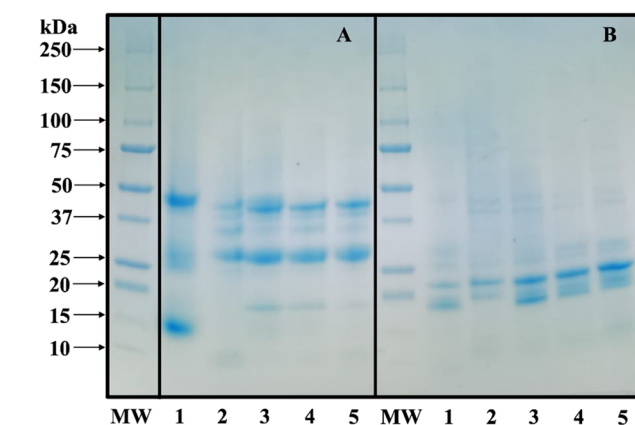
<sup>a</sup> Values are the mean of three determinations of independent batches ± standard deviation. Different letters within the same row indicate significant differences ( $p < 0.05$ ) among the groups. CSPoCo<sub>70/30</sub> = 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> = 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> = 30% canola/70% sunflower. The subscripts for each coprecipitate correspond to the proportions of canola and sunflower meals.

observed through the appearance or disappearance of bands in the electrophoretic profile. This suggests a strong interaction between the polypeptide chains of both protein sources, which is directly related to their functional properties (Fig. 2). Similar results were observed in a PoCo of Bambara groundnut protein isolate and fish collagen, which showed the presence of the characteristic bands of both polymers.<sup>65</sup>

**3.4.2 IVPoDi.** IVPoDi is a parameter used to estimate the nutritional quality of proteins, which largely depends on the nature of these biopolymers to allow or restrict the action of digestive enzymes.<sup>56</sup> The IVPoDi values of CSPoCo<sub>70/30</sub> (86.64%), CSPoCo<sub>50/50</sub> (86.22%), and CSPoCo<sub>30/70</sub> (85.25%) were significantly ( $p > 0.05$ ) comparable with that of CPoPc (84.47%), but lower than that of SPoPc (91.65%) (Fig. 4A). This could be because the IVPoDi of CPoPc is lower than that of SPoPc, and increasing the proportion of CPoPc in CSPoCos significantly decreased this parameter. This suggests that CPoPc contains a higher content of antinutritional compounds such as saponins, tannins, lectins, phytates, glucosinolates, phytic acid, and thiocyanate.<sup>68</sup> Therefore, PoC using canola and sunflower as protein sources did not result in any increase in the IVPoDi value.

Other studies have revealed IVPoDi values of 83.9% and 90.7% for canola<sup>42</sup> and sunflower<sup>6</sup> PoPc, respectively, which are similar to the values obtained for CSPoCos and PoPc in this study.

**3.4.3 IViAC.** The IViAC of PoPc and PoCos from canola and sunflower was determined by DPPH, which is a radical that has the capacity to retain an electron or oxygen from an antioxidant substance.<sup>69</sup> The DPPH radical inhibition percentages of CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub>, and CSPoCo<sub>30/70</sub> were 72.22%,



**Fig. 3** Effect of coprecipitation on the sodium dodecyl sulfate–polyacrylamide gel electrophoresis: (A) non-reducing and (B) reducing conditions. Lane MW, molecular weight marker; Lane 1, CPoPc; Lane 2, SPoPc; Lane 3, CSPoCo<sub>70/30</sub>; Lane 4, CSPoCo<sub>50/50</sub> and Lane 5, CSPoCo<sub>30/70</sub>. CPoPc is the canola protein precipitate. SPoPc is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower.

70.12% and 72.34%, respectively, which were higher than the 67.79% observed for CPoPc but lower than the 80.46% observed for SPoPc (Fig. 4B). The results obtained for the IViAC of PoPc and PoCos can be assumed as good, possibly due to the existence of phenols, flavonoids and aromatic amino acids with the capacity to give protons to radicals that lack electrons.<sup>30</sup>



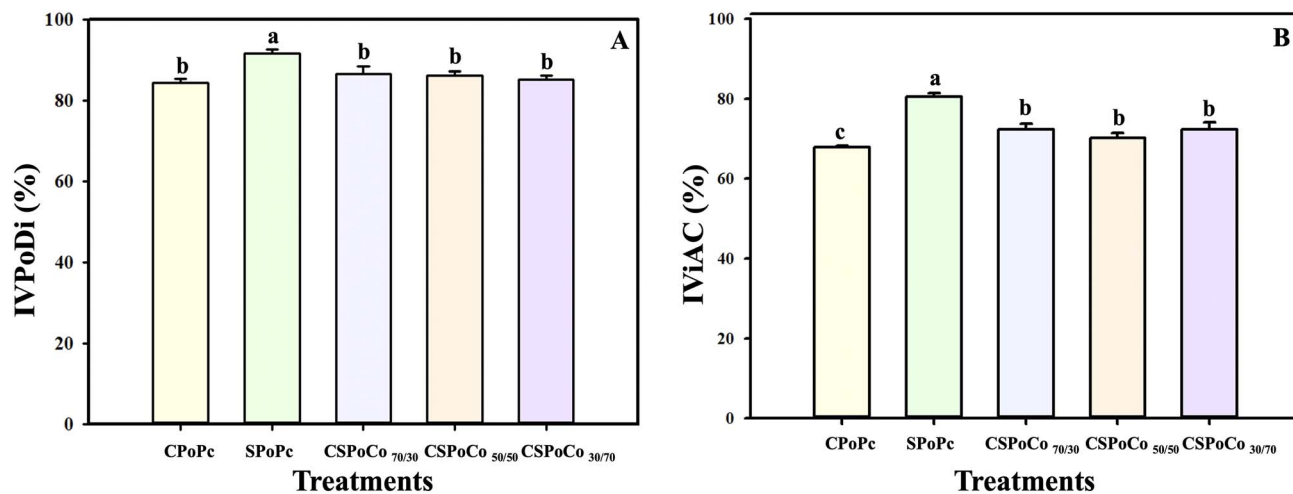


Fig. 4 Effect of the coprecipitation on (A) *in vitro* protein enzymatic digestibility (IVPoDi) and (B) *in vitro* antioxidant capacity (IViAC). CPoPc is the canola protein precipitate. SPoPc is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower. Distinct letters on the bars indicate significant ( $p < 0.05$ ) difference (average  $\pm$  SD,  $n = 3$  independent batches).

Alu'datt *et al.*<sup>18</sup> found an increase of up to 50.4% in IViAC for soybean and flaxseed PoCo compared to soybean PoPc. The authors of this study concluded that the increase in IViAC is due to the fact that PoC increases the proportion of phenolic compounds and certain amino acids with antioxidant activity.

### 3.5 Structural characterization

**3.5.1 SEM.** Morphology is one of the major analyses to characterize powder products, which influences the functional properties of protein materials.<sup>70</sup> Fig. 5 shows the granular structure of PoPc and PoCos from canola and sunflower obtained by SEM (at 150 $\times$ ). The surface of CPoPc was irregular with protuberances. Conversely, SPoPc presented larger aggregates with a flat and smooth structure compared to CPoPc and CSPoCos. In the case of CSPoCos, more heterogeneous structures with porous and smooth surfaces were presented compared to CPoPc and SPoPc. Furthermore, CSPoCo<sub>50/50</sub> had larger aggregates compared with the other coprecipitates, which could be due to the greater exposure of hydrophobic groups on the surface of the molecules (Fig. 6D), favoring interactions between them to form larger aggregates during lyophilization. Previous studies have reported that protein powders with smoother surfaces tend to have higher solubility,<sup>71–73</sup> as observed in this study (Fig. 2A).

According to a previous study, the presence and proportion of protein fractions based on solubility can influence the microstructure of proteins, as seen by SEM.<sup>30</sup> Thus, Al gives a lamellar appearance, Gx a rough surface with several tiny pores, and Gl a structure composed of small globules, which is consistent with the amount of protein fractions of Gx, Al and Gl in CPoPc, SPoPc and CSPoCos. For example, CPoPc had the highest Gx content (Table 5), presenting a more porous and rougher surface, while SPoPc had a higher proportion of Al, showing lamellar structures. In addition, CSPoCos was observed to have a greater balance between Gx and Al fractions, and thus a combination of porous/rough and soft/smooth structures was detected (Table 5 and Fig. 5). Therefore, the surface appearance of CSPoCos observed through SEM appears to be associated with the protein composition of canola and sunflower, which in turn could also influence its physico-chemical and functional properties.<sup>74</sup>

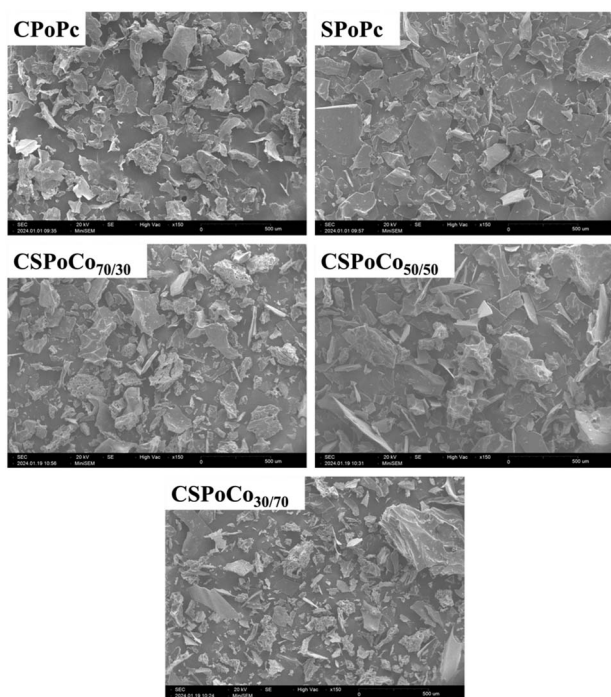


Fig. 5 Effect of the coprecipitation on the microstructure observed at 150 $\times$  magnification by scanning electron microscopy. CPoPc is the canola protein precipitate. SPoPc is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower.



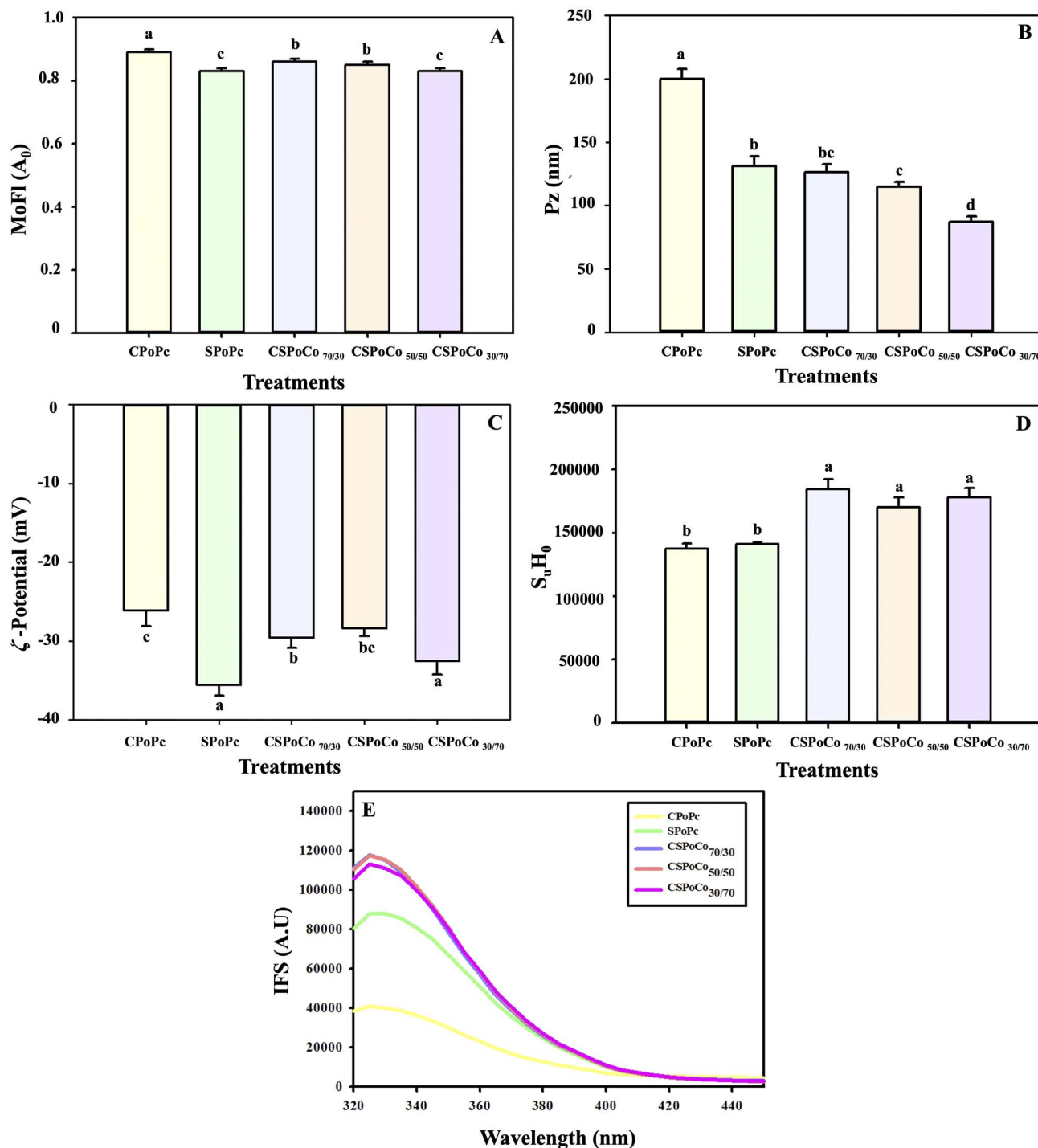


Fig. 6 Effect of coprecipitation on (A) molecular flexibility (MoFl), (B) particle size (Pz), (C)  $\zeta$ -potential, (D) surface hydrophobicity ( $S_uH_0$ ) and (E) intrinsic fluorescence spectra (IFS). CPOpC is the canola protein precipitate. SPOpC is the sunflower protein precipitate. CSPoCo is the canola–sunflower protein coprecipitate. CSPoCo<sub>70/30</sub> is 70% canola/30% sunflower, CSPoCo<sub>50/50</sub> is 50% canola/50% sunflower and CSPoCo<sub>30/70</sub> is 30% canola/70% sunflower. Distinct letters on the bars indicate significant ( $p < 0.05$ ) difference (average  $\pm$  SD,  $n = 3$  independent batches).

**3.5.2 MoFl.** MoFl refers to the adaptability of a protein to external environmental changes, which induces a reorientation of its amino acid residues, causing a change in its conformation.<sup>75,76</sup> As shown in Fig. 6A, the MoFl of CSPoCos was higher than that of SPOpC (except CSPoCo<sub>30/70</sub>), but lower compared to CPOpC ( $p < 0.05$ ). According to various studies, an increase in the

MoFl of proteins, as observed in this study for CSPoCos, causes an improvement in their functional properties (Fig. 2) due to the structural reorganization of their polypeptide chains.<sup>75,77</sup>

**3.5.3 Particle size (Pz) and  $\zeta$ -potential.** Protein Pz is a very important characteristic that influences the functional properties of food systems.<sup>32</sup> Except for CSPoCo<sub>70/30</sub>, all CSPoCos in solution



exhibited a diminished Pz compared to CPOpC and SPOpC (Fig. 6B). A higher variation in Pz was observed between CPOpC (199.3 nm) and CSPoCo<sub>30/70</sub> (86.9 nm). The decrease in the Pz of CSPoCos by PoC may be due to phenomena involving the formation of smaller aggregates or more compact structures *via* the interaction of canola and sunflower proteins.<sup>78</sup> The reduction in Pz by PoC contributed to the increase in PoSo (Fig. 2A) due to the larger surface area and the conservation of electrostatic charges, which favor protein-water interactions. Pizones Ruiz-Henestrosa *et al.*<sup>79</sup> and Chihi *et al.*<sup>80</sup> reported a reduction in Pz for PoCos of soybean (7S globulins)-whey ( $\beta$ -lactoglobulins) and pea (globulins)-whey ( $\beta$ -lactoglobulin) proteins, respectively, in agreement with the results of this study.

The  $\zeta$ -potential is related to the net surface charge and the ionic strength of proteins, affecting their adsorption and unfolding of their polypeptide chains, and therefore their functionality.<sup>81</sup> The  $\zeta$ -potential values of CPOpC, SPOpC, and CSPoCos were negative (Fig. 6C). PoC increased the negative value of  $\zeta$ -potential in the range of 8.09% (CSPoCo<sub>70/30</sub>)-19.94% (CSPoCo<sub>30/70</sub>) compared with CPOpC. On the contrary, PoC diminished the negative values for CSPoCo<sub>70/30</sub> (20.27%) and CSPoCo<sub>50/50</sub> (25.22%) with respect to SPOpC.

Changes in  $\zeta$ -potential may be associated with modifications in the amount of ionizable amino and carboxyl groups exposed on the surface caused by PoC.<sup>80</sup> Previous research has shown that protein dispersions with high absolute  $\zeta$ -potentials are characterized by strong intermolecular electrostatic repulsion, which is sufficient to resist agglomeration, thereby improving the dispersibility and stability of the protein system.<sup>10</sup> Therefore,  $\zeta$ -potential is related to the improvement in the functional properties of CSPoCos because the interactions between the two polypeptides inhibited structural folding and maintained their charged surface (Fig. 2). PoCos of rice and casein<sup>78</sup> and rice and cod<sup>10</sup> proteins also increased the negative value of  $\zeta$ -potential compared to the individual sources.

**3.5.4 S<sub>u</sub>H<sub>o</sub>.** S<sub>u</sub>H<sub>o</sub> indicates the amount of hydrophobic groups distributed along the surface of proteins, which has a great impact on the functional characteristics of these polymers.<sup>33</sup> As can be seen in Fig. 6D, CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub> exhibited an increase in S<sub>u</sub>H<sub>o</sub> by 25.5–23.6%, 19.3–17.2%, and 22.8–20.8% in comparison with CPOpC and SPOpC, respectively. The increase in S<sub>u</sub>H<sub>o</sub> in CSPoCos could be due to the structural changes brought about by the binding of the canola and sunflower proteins, which caused the main chains of these polymers to unfold to expose a polar groups on their surface. Kristensen *et al.*<sup>9</sup> and Kristensen *et al.*<sup>20</sup> demonstrated that the association of pea and whey proteins in the form of PoCo increased S<sub>u</sub>H<sub>o</sub>, as observed in this study.

**3.5.5 IFS.** Changes in the tertiary structure of proteins can be measured using fluorescence emission spectra when tyrosine, phenylalanine, and tryptophan residues of proteins are excited at a wavelength of 280 to 290 nm.<sup>34</sup> The impact of PoC on the fluorescence spectra of PoPc and PoCos from canola and sunflower is shown in Fig. 6E. The maximum emission wavelength occurred at 325 nm in all treatments. Overall, a higher IFS was observed in CSPoCos compared to CPOpC and SPOpC. In a study with soybean and whey PoCo,<sup>12</sup> it was determined that

changes in the tertiary structure by PoC, measured as IFS, are induced by the formation of stronger protein–protein interactions, in agreement with the results of this study. Furthermore, the alteration of the tertiary structure by PoC, as determined by IFS, in CSPoCos is consistent with the results regarding the increase in S<sub>u</sub>H<sub>o</sub> (Fig. 6D), which is also an indicator of the variation of the protein conformation.

**3.5.6 SH<sub>Fr</sub>, SH<sub>To</sub> and S–S<sub>Br</sub>.** Sulfhydryl groups, including SH<sub>Fr</sub>, SH<sub>To</sub> and S–S<sub>Br</sub>, are some of the most reactive functional groups in proteins, which significantly influence the functional properties of food proteins. SH<sub>Fr</sub> are the sulfhydryl groups that are located on the surface of the proteins and are easily accessible to react with the environment. SH<sub>To</sub> includes SH<sub>Fr</sub> and SH buried within the protein structure.<sup>32</sup> The results of SH<sub>Fr</sub>, SH<sub>To</sub> and S–S<sub>Br</sub> content of CPOpC, SPOpC and CSPoCos are presented in Table 6.

Overall, PoC increased the SH<sub>Fr</sub> content (20.13–28.71%) of CSPoCos, in contrast to CPOpC, whereas the S–S<sub>Br</sub> content decreased in CSPoCos in the ranges of 105.5–503.2% and 37.4–303.2% in contrast to CPOpC and SPOpC, respectively, which could imply unfolding of the polypeptide chain and thus a modification in the initial proportion of sulfhydryl groups in the new polymeric structures. Conversely, the SH<sub>To</sub> content did not change significantly ( $p > 0.05$ ) in CSPoCos with respect to CPOpC, but its value was significantly reduced ( $p < 0.05$ ) compared to SPOpC (Table 6).

Interestingly, in CSPoCos, it was observed that almost all the sulfhydryl groups corresponded to SH<sub>Fr</sub>, which also demonstrated the structural rearrangement of its proteins due to the effect of PoC, similar to the effect caused by other physical treatments such as ultrasound<sup>81</sup> and high pressure.<sup>34</sup>

**3.5.7 ATR-FTIR.** The FTIR spectrum of a protein exhibits absorption peaks primarily associated with its amide groups. The amide I peak is of great importance, as it allows the type of secondary structure ( $\beta$ -sheet,  $\alpha$ -helix,  $\beta$ -turn, and random coil) of a protein to be determined.<sup>30</sup>

Table 7 shows the secondary structure content of PoPc and PoCos from canola and sunflower. Compared with CPOpC and SPOpC, the secondary structures of CSPoCos were significantly different ( $p < 0.05$ ). In CPOpC, SPOpC, and CSPoCos,  $\beta$ -sheets were the predominant secondary structures, followed by  $\beta$ -turns, random coils and  $\alpha$ -helices. PoC increased the relative  $\beta$ -sheet content from an initial value of 49.55% for CPOpC to 52.13%, 51.42% and 52.58% for CSPoCo<sub>70/30</sub>, CSPoCo<sub>50/50</sub> and CSPoCo<sub>30/70</sub>, respectively; however, these values were lower than that of 60.96% obtained for SPOpC. In contrast, the  $\alpha$ -helix content in CSPoCos was not significantly ( $p > 0.05$ ) different from that of CPOpC. In the case of  $\beta$ -turns, PoC significantly reduced ( $p < 0.05$ ) its content in the range of 4.86% (CSPoCo<sub>50/50</sub>) to 25.0% (CSPoCo<sub>70/30</sub>) compared to CPOpC. The reason for these changes could be attributed to the proportion of different protein sources in PoCos. This composition induces a new equilibrium in the secondary structures by modifying molecular interactions, such as hydrogen bonds, disulfide bonds, and hydrophobic and van der Waals forces. Consequently, these structural transformations are linked to the enhancement of PoSo, WaHC, OiHC, LaGeCo, EAcln, FmC, and FmS (Fig. 2). Tan *et al.*,<sup>11</sup> Tian *et al.*,<sup>12</sup> and Wang *et al.*<sup>78</sup> found that the PoC of proteins from soybean-tilapia, soybean-wheat, and soybean-rice,



respectively, modified the secondary structure in contrast to individual protein sources, as found in this study.

## 4 Conclusions

PoC of canola and sunflower proteins by alkaline extraction-isoelectric precipitation altered their physicochemical, functional and structural characteristics compared to CPoPc and SPoPc. The protein content,  $\rho_b$ ,  $\rho_c$ , CaIn, HaRa, WaHC, OiHC, LaGeCo, EAcIn, FmC, FmS,  $S_uH_o$ , and IFS values of CSPoCos increased with respect to original protein source, while their Pz and EStIn values decreased. In general, the moisture,  $L^*$ ,  $a_w$ ,  $T_{rb}$ , MoFl,  $\zeta$ -potential, IVPoDi, IViAc, Al, Gx, Pr,  $SH_{Fr}$ ,  $SH_{To}$ ,  $S-S_{Br}$ ,  $\alpha$ -helix,  $\beta$ -sheet, and  $\beta$ -turn values varied between those corresponding to CPoPc and SPoPc. SDS-PAGE results revealed the characteristic bands of the two coprecipitated protein sources, in addition to the appearance of a new band, indicating an interaction between the two polymers. The improvement in the functional characteristics of CSPoCos was due to the unfolding and interactions of the canola and sunflower polypeptide chains, giving rise to new protein structures. Specifically, the values of PoSo, WaHC, FmC and FmS were higher in CSPoCo<sub>50/50</sub> compared to the other CSPoCos. Thus, PoC proved to be an effective method for modifying canola and sunflower proteins, which could considerably improve the functional quality of plant proteins, making them a good alternative to animal proteins as food ingredients. Further research into the nutritional and sensory characteristics of CSPoCos could expand their use in the production of functional foods.

## Author contributions

Nitzia Thalía Flores-Jiménez: investigation; methodology; writing – original draft preparation; editing. José Armando Ulloa: investigation; writing – original draft; review; editing; conceptualization, supervision; data curation; funding acquisition. Judith Esmeralda Urias-Silvas: software; resources; supervision; validation.

## Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this manuscript.

## Data availability

Data will be made available on request.

## Acknowledgements

The authors thank the Secretariat of Science, Humanities, Technology and Innovation (SECIHTI) of Mexico for the post-doctoral scholarship awarded to Nitzia Thalía Flores Jiménez (CVU: 601948), and the Patronage to Administer the Special Tax Destined to the Autonomous University of Nayarit (UAN) for the financial support.

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