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# Curcumin encapsulation *via* protein-stabilized emulsions: comparative formulation and characterization using whey, soy, and pea proteins

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Curcumin, a potent polyphenol, has significant limitations in food and pharmaceutical applications due to its poor water solubility, chemical instability, and low oral bioavailability. To address these challenges, this study aimed to develop and systematically characterize stable emulsion-based curcumin delivery systems using a comparative approach. The present study offers novel findings through a direct, side-by-side evaluation of a traditional milk protein (whey protein isolate, WPI) against two plant-based alternatives (soy protein isolate, SPI, and pea protein isolate, PPI) under uniform formulation conditions. The emulsions were formulated with curcumin (0.4%) in butter oil as the lipophilic core, with WPI, SPI, or PPI, and a combination of maltodextrin and gum arabic as wall materials at core-to-wall ratios of 1:1, 1:2, and 1:3. The WPI-based emulsions demonstrated superior physicochemical properties, achieving the lowest particle size ( $322 \pm 5.49 \mu\text{m}$ ) and highest zeta potential ( $-38.50 \text{ mV}$ ) at the 1:2 ratio. PPI-based emulsions exhibited the highest antioxidant activity, with 81.38% DPPH inhibition, highlighting their potential as a promising plant-based alternative. Both dairy and plant proteins effectively encapsulated curcumin, with an encapsulation efficiency of 88.16–96.26%. In contrast, the SPI-based emulsions exhibited relatively lower performance across most parameters. FT-IR spectroscopy and fluorescence microscopy confirmed successful curcumin encapsulation and emulsion integrity. These findings highlight WPI as a highly effective encapsulating agent and position PPI as a viable, sustainable, and functional plant-based alternative. This study provides valuable insights for developing stable, cost-effective, and functional curcumin delivery systems to meet the growing market demand for clean-label and plant-based foods and nutraceuticals.

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## Sustainability spotlight

This study supports sustainability by developing effective curcumin encapsulation systems using plant- and dairy-based proteins, reducing dependence on synthetic stabilizers. It shows that soy and pea proteins can match whey protein in stabilizing curcumin, promoting the use of renewable, lower-impact ingredients. The approach enables the creation of functional foods that retain nutrient potency and align with clean-label standards. By leveraging widely available agricultural resources through scalable, cost-efficient methods, the research encourages circular value chains and minimizes waste. These findings contribute to greener food technologies that balance nutritional value, consumer appeal, and environmental responsibility, fostering more resilient and sustainable food systems.

## 1 Introduction

Curcumin, a hydrophobic polyphenol derived from the rhizomes of *Curcuma longa*, is widely recognized for its multi-functional bioactivities, including powerful antioxidant, anti-inflammatory, antimicrobial, antiviral, and anticancer

activities.<sup>1,2</sup> However, its practical application in functional foods and pharmaceuticals is constrained by unfavourable physicochemical characteristics. Curcumin exhibits extremely low water solubility at acidic and neutral pH, along with poor chemical stability under alkaline conditions and upon exposure to environmental factors, such as light and heat.<sup>3</sup> The poor solubility and cellular permeability classify it as a biopharmaceutical classification system class IV molecule, which makes it difficult to deliver therapeutically relevant concentrations and results in low oral bioavailability.<sup>4</sup> To overcome these limitations, encapsulation has emerged as a promising solution to protect curcumin and enhance its stability and

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bioavailability.<sup>5,6</sup> Encapsulation is essential for protecting sensitive bioactive compounds from degradation, thereby enhancing their stability and bioavailability.<sup>7</sup> Emulsion-based microencapsulation, in particular, is a technique that has gained significant attention in the food industry because it offers a versatile and scalable method to entrap active ingredients within a protective colloidal structure, making them ideal precursors for developing advanced delivery systems.<sup>8–10</sup> It is effective in protecting lipophilic compounds, such as curcumin, from environmental degradation and can improve their sensory properties.<sup>11,12</sup> Choudhary *et al.*<sup>13</sup> suggested different encapsulation methods such as inclusion complexes, spray drying, and lipid nanoparticles to enhance the solubility, stability, and bioavailability of  $\alpha$ -lipoic acid. Moosavi-Nasab *et al.*<sup>14</sup> extended the shelf life of button mushrooms using aloe vera and gelatin edible coatings containing Shirazi thyme essential oil nano-emulsion and reported that the mushrooms maintained their quality and showed an increased shelf life during 16 days of storage at 5 °C and 90% RH.

Proteins are widely used as emulsifying agents because of their amphiphilic nature, which allows them to form protective films at the oil–water interface. Milk proteins, such as whey protein isolate (WPI), have been extensively studied for their high encapsulation efficiency and emulsion stability.<sup>15</sup> However, there is growing interest in sustainable, plant-based alternatives. Plant proteins, such as soy protein isolate (SPI) and pea protein isolate (PPI), are recognized as cost-effective and sustainable, and are functionally capable of stabilizing emulsions through their emulsifying and gelation properties.<sup>16,17</sup> Moreover, combining proteins with carbohydrates, such as maltodextrin and gum arabica, has been shown to enhance encapsulation performance and emulsion stability through synergistic interactions.<sup>18,19</sup>

Despite these advances and innovations, there is a significant research gap, with a noticeable lack of systematic comparative studies evaluating the performance of plant-based proteins, such as SPI and PPI, against conventional milk proteins, such as WPI, under uniform formulation conditions. Most of the existing studies have used a single protein source, such as whey,<sup>20,21</sup> soy,<sup>22,23</sup> and pea protein,<sup>24,25</sup> failing to provide a comprehensive, side-by-side analysis of how different protein types affect the physicochemical, structural, and functional characteristics of curcumin-loaded emulsions. Moreover, most existing studies do not incorporate a comprehensive set of analytical techniques to thoroughly assess the encapsulation performance.

To bridge these critical gaps, this study focused on the comparative formulation and characterization of curcumin-loaded emulsions stabilized by WPI, SPI, and PPI. By providing integrative and comparative experiments, this study advances the understanding of how both dairy- and plant-based proteins function in emulsion-based delivery systems. Ultimately, this study contributes to the development of cost-effective, stable, and functional encapsulation strategies for lipophilic bioactives such as curcumin, supporting the growing demand for plant protein applications in clean-label and functional food innovations.

## 2 Materials and methods

### 2.1 Chemicals and ingredients

Pure curcumin powder ( $\geq 95\%$  total curcuminoids; residual solvent: 3998 ppm) was obtained from M/s Plant Lipids (Kerala, India). Butteroil was prepared from buffalo milk procured from Dairy Farm, Banaras Hindu University, Varanasi, India, using the creamery butter method with 45% fat cream as per the procedure given by Aneja *et al.*<sup>26</sup> Maltodextrin (20% reducing sugar) and gum arabic were sourced from M/s Hi-Media Laboratories Pvt. Ltd, Mumbai, India. SPI and PPI (containing 90% protein, 1.0% fat, and 0.5% ash) were purchased from Chem-Kart, Mumbai. WPI (containing 90% protein, 1.0% fat, 1.6% carbohydrates, and 0.5% ash) was obtained from M/s Globex Enterprises, New Delhi. All analytical-grade chemicals and reagents used in this study were procured from certified vendors and freshly prepared following standard laboratory protocols.

### 2.2 Preparation of curcumin emulsion with high shear homogenization

**2.2.1 Preparation of core material.** Butter oil was selected as the lipid phase (core material) due to its high compatibility with curcumin, a lipophilic bioactive that exhibits limited solubility in aqueous systems. The triglyceride-rich matrix of butter oil enhances curcumin solubilization and stability, as confirmed in our previous study, where  $59.26 \pm 2.03\%$  solubility was achieved at 0.4% (w/v) fortification level.<sup>27</sup> In addition, butter oil is a natural dairy fat with GRAS status, confirming its suitability for improved dispersion, stability, and compatibility with dairy-based delivery systems. The mixture was heated to 60 °C and stirred continuously for 10 min using a magnetic stirrer (Narang Scientific, New Delhi) to ensure complete solubilization of curcumin.

**2.2.2 Preparation of wall material.** The wall material was formulated using protein isolates (WPI, SPI, or PPI) in combination with carbohydrates (maltodextrin with a dextrose equivalent value of  $\leq 20$  and gum arabica) at a fixed ratio of 90 : 10. The protein-to-carbohydrate ratio was standardized at 1 : 2 (w/w) based on preliminary trials conducted with soy and pea proteins, as well as our previous findings employing whey protein concentrate as a wall material for curcumin emulsions,<sup>9</sup> to achieve optimal emulsion viscosity. The dry ingredients were accurately weighed and dispersed in milk containing 6% fat and 9% solids-not-fat, followed by mixing with a hand blender (Bajaj HB21, Mumbai, India) to ensure uniform hydration and dispersion of solids prior to high-speed homogenization. The prepared mixture accounted for approximately 80% of the final emulsion volume.

**2.2.3 Formulation of curcumin emulsions.** The curcumin-loaded lipid (core) and aqueous wall phases were mixed at varying core-to-wall ratios (1 : 1, 1 : 2, and 1 : 3). Microsized emulsions were prepared using a T 25 digital Ultra-Turrax homogenizer (IKA, Germany) operated at 10 000 rpm for 10 min. The process was carried out at 60 °C to maintain the lipid phase in a molten state, thereby facilitating uniform



droplet formation and stable emulsion preparation.<sup>9,12</sup> A schematic representation of the formulation process is presented in Fig. 1. The detailed ingredient levels used in each formulation are listed in Table 1.

### 2.3 Emulsion stability

The stability of the curcumin emulsions was evaluated using a gravity-based method. Ten mL of each emulsion was transferred into a 25 mL graduated cylinder and stored at 25 °C for

24 h without disturbance. The samples were observed for visible phase separation, creaming, or sedimentation. Emulsions without observable separation were considered stable.

### 2.4 Physicochemical analyses

**2.4.1 Total solid content.** The total solids were measured using the oven-drying method, according to the AOAC protocol.<sup>28</sup> Approximately 5 g of the emulsion sample was dried

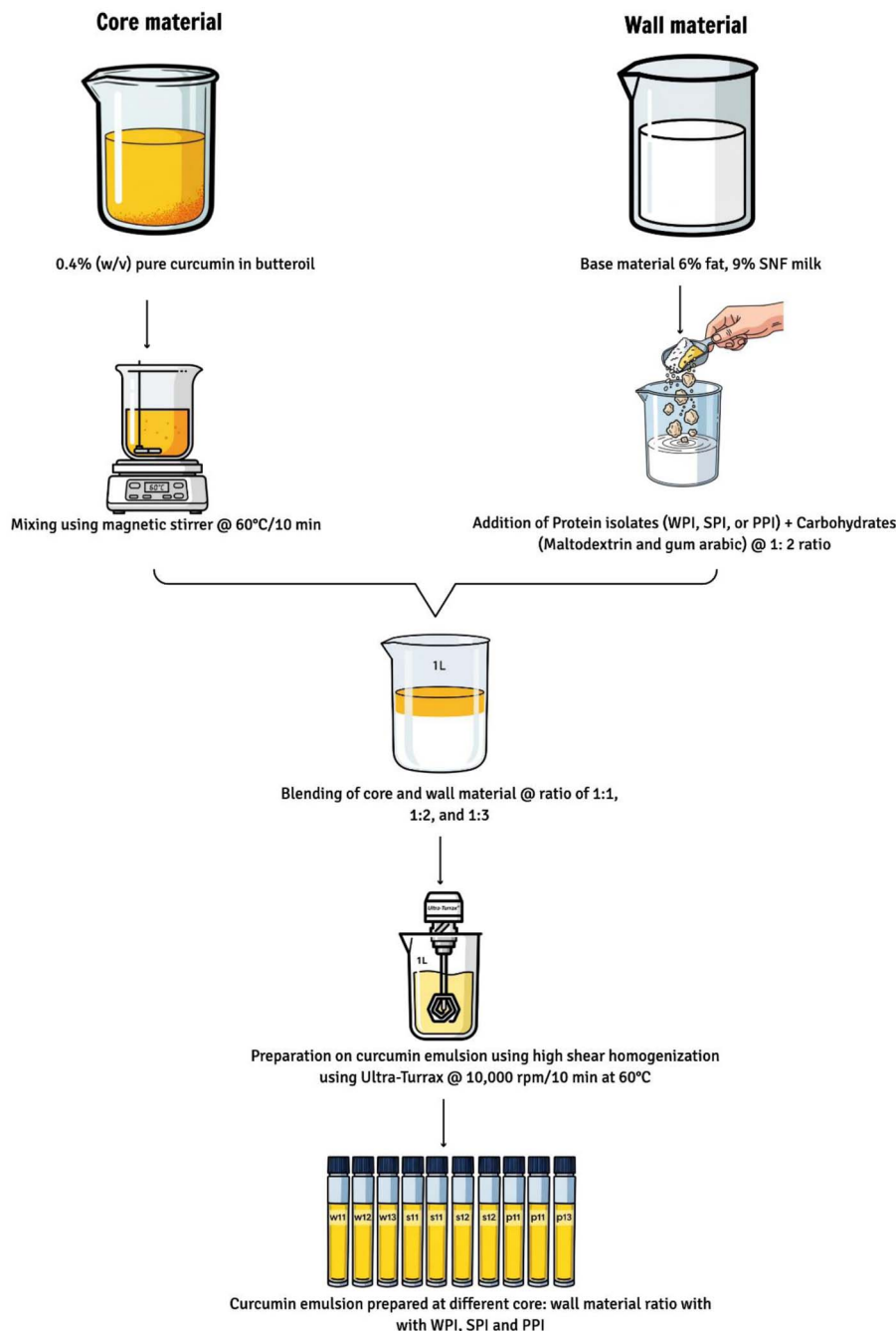
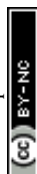


Fig. 1 Process flow diagram for the formulation of curcumin emulsions (1 litre volume) using whey, soy, and pea protein isolates as wall materials.



**Table 1** Levels of different ingredients used in the formulation of 1 L curcumin emulsion (35 w/v% total solids) using different protein isolates (whey, soy, and pea protein) in the wall material

Ratios of core : wall material	Ratios of protein : carbohydrates	Base material (milk with 6% fat and 9% SNF) (in mL)	Core material (0.4% curcumin supplemented with butteroil) (in g)	Wall material (protein and carbohydrates in combination, 1 : 2 ratio) (in g)			
				Protein isolates (whey, soy and pea)	Maltodextrin	Gum arabica	Total
1 : 1	1 : 2	800.00	100.00	33.30	60.00	6.70	100.00
1 : 2	1 : 2	800.00	66.70	44.00	80.00	8.90	133.30
1 : 3	1 : 2	800.00	50.00	50.00	90.00	10.00	150.00

at  $105 \pm 1$  °C until a constant weight was reached, and the total solids (%) were calculated by mass difference.

**2.4.2 Water activity ( $a_w$ ).** Water activity was measured using a 3TE Aqualab water activity meter (Washington, USA). The emulsion samples were poured into clean measurement cups and placed in the instrument chamber. After temperature equilibration at 25 °C, the  $a_w$  values were recorded directly from the digital readout. Each measurement was performed in triplicate.

**2.4.3 Color analysis.** The color values of the emulsions ( $L^*$  for lightness,  $a^*$  for redness/greenness, and  $b^*$  for yellowness/blueness) were determined using a ColorFlex EZ spectrophotometer (HunterLab, Virginia, USA). The emulsions were placed in a sample cup with a smooth surface to ensure the absence of air bubbles. The instrument was calibrated using a standard white tile before the analysis. Each sample was measured in triplicate.

**2.4.4 Particle size and zeta potential.** The particle size distribution and zeta potential were assessed using dynamic light scattering (DLS) with a Zetasizer Nano ZS90 (Malvern Instruments Ltd, UK). For the particle size analysis, 1.5–2 mL of the emulsion was diluted in 400 mL of distilled water. For the zeta potential, 50 mL of emulsion was mixed with 50 mL of Millipore water. Measurements were performed at 25 °C, and the particle size and zeta potential were measured in micrometres (µm) and millivolts (mV), respectively.

**2.4.5 Apparent viscosity.** The viscosity was measured at 20 °C using a dynamic rheometer (MCR 52, Anton Paar, Germany) equipped with a stainless-steel cone-plate geometry (CP75-1°, 50 mm diameter). The emulsion samples were equilibrated to the measurement temperature, and the viscosity was recorded at a shear rate of  $50 \text{ s}^{-1}$ . Each measurement was performed in triplicate.

## 2.5 2,2-Diphenyl-1-picrylhydrazyl (DPPH) inhibition activity (%)

The antioxidant activity of the curcumin emulsions was evaluated using the DPPH radical scavenging assay.<sup>29</sup> The samples were diluted in methanol to a final concentration of  $50 \text{ mg mL}^{-1}$ . A 2.5 mL aliquot of the sample was mixed with 5 mL of 0.2 mM DPPH solution and incubated in the dark at room temperature for 30 min. The absorbance was measured at 517 nm using a Jasco UV-vis double beam spectrophotometer V-

730 (Japan). All measurements were performed in triplicate. The percentage inhibition was calculated using eqn (1).

$$\% \text{ inhibition of DPPH} =$$

$$\frac{\text{absorbance of blank} - \text{absorbance of sample}}{\text{absorbance of blank}} \times 100 \quad (1)$$

## 2.6 Encapsulation efficiency

The encapsulation efficiency of the curcumin emulsion was determined as per the method described by Zhan *et al.*<sup>30</sup> with slight modifications. The freshly prepared curcumin emulsion was mixed with 80% (v/v) aqueous ethanol and vortexed for curcumin extraction. The dispersions were centrifuged (3–30K, Sigma, Germany) at 3600 rpm for 20 min to eliminate precipitates. The supernatant was diluted to an appropriate concentration with 80% ethanol, and the absorbance was recorded at 426 nm using a Jasco UV-vis double beam spectrophotometer V-730 (Japan). The encapsulation efficiency was calculated using eqn (2).

$$\text{Encapsulation efficiency (\%)} =$$

$$\frac{\text{encapsulated content of curcumin}}{\text{added content of curcumin}} \times 100 \quad (2)$$

## 2.7 FT-IR analysis

FT-IR analysis was performed using the potassium bromate (KBr) pellet method for all samples. Freeze-dried emulsion samples (2–3 mg) were mixed with micronized KBr at a ratio of 1 : 10, pressed into pellets using a hydraulic press (10 tons pressure), and stored in a desiccator for 2 h. Spectra were recorded on a Spectrum 100-FTIR spectrometer (PerkinElmer, USA) across  $4000\text{--}400 \text{ cm}^{-1}$  with 25 scans at a resolution of  $4 \text{ cm}^{-1}$ . The analysis and plotting were conducted using OriginPro 2023 software.

## 2.8 XRD pattern

The crystalline characteristics of the freeze-dried emulsions were analyzed using a Miniflex 600 X-ray diffractometer (Rigaku, Japan) with  $\text{CuK}\alpha$  radiation ( $\lambda = 1.54 \text{ \AA}$ ) operating at 40 kV and 40 mA. The samples were scanned over a  $2\theta$  range of  $5^\circ$  to  $80^\circ$  at a rate of  $30^\circ \text{ min}^{-1}$ . Data processing and graph plotting were performed using OriginPro.





Table 2 Physicochemical attributes of curcumin emulsion formulations at different core : wall material ratios with different protein isolates (WPI, SPI, and PPI) in the wall material<sup>a</sup>

Parameters	Protein isolates									
	Soya					Pea				
	Whey		Soya			Pea				
Core : wall	1 : 1 (W11)	1 : 2 (W12)	1 : 3 (W13)	1 : 1 (S11)	1 : 2 (S12)	1 : 3 (S13)	1 : 1 (P11)	1 : 2 (P12)	1 : 3 (P13)	
Total solids (%)	35.22 ± 0.11 <sup>aA</sup>	35.02 ± 0.34 <sup>aA</sup>	34.99 ± 0.24 <sup>aA</sup>	35.54 ± 0.25 <sup>aA</sup>	35.72 ± 0.38 <sup>aA</sup>	35.52 ± 0.17 <sup>aA</sup>	35.22 ± 0.54 <sup>aA</sup>	35.57 ± 0.15 <sup>a</sup>	34.98 ± 0.68 <sup>aA</sup>	
Water activity (a <sub>w</sub> )	0.991 ± 0.001 <sup>aB</sup>	0.987 ± 0.008 <sup>aB</sup>	0.994 ± 0.002 <sup>aB</sup>	0.994 ± 0.003 <sup>aB</sup>	0.888 ± 0.007 <sup>aB</sup>	0.885 ± 0.007 <sup>aA</sup>	0.992 ± 0.003 <sup>b</sup>	0.890 ± 0.007 <sup>aA</sup>	0.891 ± 0.007 <sup>aA</sup>	
<b>Colour value</b>										
L*	79.61 ± 0.01 <sup>cD</sup>	78.80 ± 0.03 <sup>bC</sup>	77.36 ± 0.06 <sup>aB</sup>	72.92 ± 1.63 <sup>aA</sup>	70.70 ± 1.11 <sup>aA</sup>	70.99 ± 1.34 <sup>aA</sup>	67.57 ± 3.93 <sup>aA</sup>	71.81 ± 1.40 <sup>aA</sup>	70.15 ± 1.63 <sup>aA</sup>	
a*	-4.05 ± 0.13 <sup>cC</sup>	-5.43 ± 0.05 <sup>aA</sup>	-4.90 ± 0.09 <sup>bB</sup>	1.09 ± 1.98 <sup>aB</sup>	1.45 ± 1.39 <sup>aB</sup>	0.89 ± 1.66 <sup>aB</sup>	4.52 ± 2.86 <sup>aB</sup>	1.44 ± 2.66 <sup>aB</sup>	2.46 ± 1.83 <sup>bB</sup>	
b*	69.55 ± 0.10 <sup>cC</sup>	65.32 ± 0.37 <sup>bB</sup>	62.77 ± 0.06 <sup>aA</sup>	74.47 ± 4.45 <sup>aC</sup>	73.52 ± 3.00 <sup>aB</sup>	70.62 ± 4.16 <sup>aC</sup>	69.39 ± 0.36 <sup>aC</sup>	72.94 ± 2.93 <sup>aC</sup>	70.58 ± 3.34 <sup>aC</sup>	
<b>Particle size</b>										
Z-average (µm)	350 ± 9.28 <sup>bB</sup>	322 ± 5.49 <sup>aA</sup>	420 ± 35.94 <sup>cB</sup>	584.6 ± 27.79 <sup>cB</sup>	391.53 ± 26.55 <sup>aB</sup>	476.2 ± 40.57 <sup>bB</sup>	512.13 ± 43.36 <sup>aB</sup>	433.2 ± 37.26 <sup>aB</sup>	410.13 ± 27.91 <sup>aB</sup>	
PdI	0.577 ± 0.068 <sup>bB</sup>	0.434 ± 0.033 <sup>aA</sup>	0.531 ± 0.045 <sup>bB</sup>	0.777 ± 0.10 <sup>aB</sup>	0.715 ± 0.06 <sup>aB</sup>	0.819 ± 0.09 <sup>aB</sup>	0.863 ± 0.144 <sup>aB</sup>	0.763 ± 0.052 <sup>aB</sup>	0.663 ± 0.083 <sup>aB</sup>	
<b>Zeta potential</b>										
Zeta potential (mV)	-36.8 ± 0.34 <sup>bB</sup>	-38.5 ± 0.43 <sup>aA</sup>	-36.23 ± 0.80 <sup>bB</sup>	-31.96 ± 0.55 <sup>aC</sup>	-34.66 ± 0.05 <sup>aC</sup>	-33.5 ± 0.8 <sup>aC</sup>	-25.63 ± 0.83 <sup>cF</sup>	-31.56 ± 0.05 <sup>aD</sup>	-30.06 ± 0.90 <sup>bE</sup>	
Electrical conductivity (mS cm <sup>-1</sup> )	0.138 ± 0.001 <sup>bC</sup>	0.112 ± 0.001 <sup>aA</sup>	0.136 ± 0.001 <sup>bB</sup>	0.120 ± 0.001 <sup>bB</sup>	0.116 ± 0.001 <sup>aB</sup>	0.143 ± 0.001 <sup>cC</sup>	0.139 ± 0.001 <sup>aC</sup>	0.175 ± 0.003 <sup>bD</sup>	0.141 ± 0.001 <sup>aC</sup>	
Viscosity (Pa s)	0.038 ± 0.010 <sup>cA</sup>	0.018 ± 0.004 <sup>bA</sup>	0.012 ± 0.003 <sup>aA</sup>	0.033 ± 0.007 <sup>aA</sup>	0.037 ± 0.007 <sup>aA</sup>	0.026 ± 0.004 <sup>aA</sup>	0.094 ± 0.035 <sup>aA</sup>	0.039 ± 0.010 <sup>aA</sup>	0.022 ± 0.004 <sup>aA</sup>	
DPPH inhibition (%)	72.91 ± 3.25 <sup>bE</sup>	61.37 ± 4.74 <sup>aD</sup>	58.33 ± 5.48 <sup>aD</sup>	47.72 ± 0.93 <sup>cC</sup>	39.17 ± 2.26 <sup>bB</sup>	28.55 ± 1.14 <sup>aA</sup>	81.38 ± 0.51 <sup>cG</sup>	77.94 ± 1.82 <sup>bF</sup>	73.76 ± 2.31 <sup>aE</sup>	
Encapsulation efficiency (%)	93.32 ± 0.47 <sup>aC</sup>	94.04 ± 0.63 <sup>aC</sup>	92.66 ± 0.22 <sup>aC</sup>	92.76 ± 0.66 <sup>bC</sup>	88.16 ± 0.54 <sup>aA</sup>	91.96 ± 0.35 <sup>bB</sup>	96.26 ± 0.45 <sup>bD</sup>	95.78 ± 0.38 <sup>bD</sup>	94.79 ± 0.31 <sup>aC</sup>	

<sup>a</sup> Note: values are presented as mean ± SD. Superscript letters (a, b, and c) indicate statistically significant differences ( $p < 0.05$ ) among different core : wall material solid ratios within the same protein wall material solid type. Superscript capital letters (A, B, and C) indicate statistically significant differences ( $p < 0.05$ ) among different core : wall material solid ratios across different protein isolates (WPI, SPI, and PPI) used in the wall material.

## 2.9 Microscopy analysis

Fluorescence microscopy was used to visualize curcumin distribution within the emulsions, as described by Yazdi and Corredig.<sup>31</sup> Approximately 10  $\mu\text{L}$  of emulsion was placed on a clean glass slide, covered with a cover slip, and observed under a fluorescence microscope (Olympus IX-83, Tokyo, Japan) at 60 $\times$  magnification.

## 2.10 Statistical analysis

All data were obtained in triplicate and are reported as mean  $\pm$  standard deviation. Statistical differences between the groups were analyzed using one-way analysis of variance (ANOVA) at  $p < 0.05$  with the statistical software package SPSS 16.0 software (SPSS INC, Chicago, IL, USA).

# 3 Results and discussion

## 3.1 Physicochemical properties of curcumin emulsions

Table 2 shows the total solids (%), water activity, and viscosity (Pa s) of curcumin emulsions prepared with varying core-to-wall ratios and different protein isolates in the wall material.

**3.1.1 Total solid content.** The total solid content of all curcumin emulsions ranged from 34.98% to 35.72%, with no significant differences observed among the different protein isolates (WPI, SPI, and PPI) or core-to-wall ratios ( $p > 0.05$ ). This

consistency in dry matter content indicates that the formulation process and compositional adjustments did not significantly alter total solids. These findings align with those of Meena *et al.*,<sup>9</sup> who reported similar results for curcumin emulsions formulated with comparable levels of wall materials. The high solid content ( $\sim 35\%$ ) is advantageous for potential downstream applications, such as spray drying or incorporation into solid food matrices, where concentrated emulsions are preferred. This characteristic enhances the versatility of emulsions in various food processing techniques.

**3.1.2 Water activity ( $a_w$ ).** The water activity ( $a_w$ ) values of the curcumin emulsions ranged from 0.885 to 0.994, with S13 exhibiting the lowest (0.885) and W12 the highest (0.994) values. Soy and pea protein-based emulsions showed significantly lower  $a_w$  at 1:2 and 1:3 core:wall ratios compared to 1:1, whereas whey protein emulsions demonstrated minimal changes across ratios. This decline in water activity with increasing wall material indicates an improved water-binding capacity of the protein-carbohydrate matrix, particularly in soy and pea protein systems. The variation in water activity is related to the level and nature of the wall material, especially the protein content.<sup>32</sup> Lower  $a_w$  values ( $< 0.9$ ) are favorable for microbial stability, whereas emulsions with  $a_w \geq 0.99$  (e.g., W12) may require refrigeration and aseptic handling to prevent microbial spoilage. However, the intrinsic antimicrobial

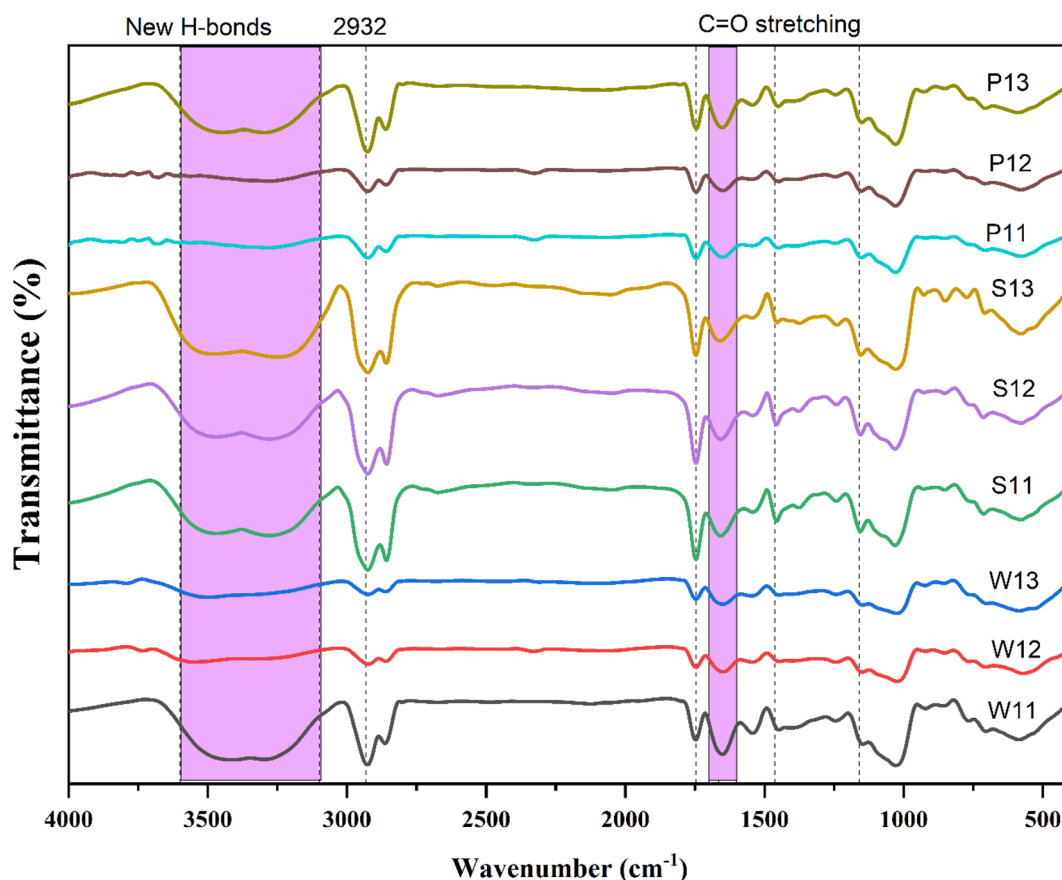


Fig. 2 FTIR spectra of curcumin emulsions prepared using different concentrations of whey, soy, and pea protein isolates as wall materials.



properties of curcumin may contribute to microbial stability and emulsion preservation to a certain extent.<sup>33</sup> In this study, water activity in curcumin emulsions was influenced by wall material type and proportion, with higher protein–carbohydrate content improving water-binding, microbial stability, and overall emulsion preservation.

**3.1.3 Viscosity.** The emulsion viscosity ranged between 0.012 and 0.094 Pa s, with the highest value observed in P11 (0.094 Pa s) and the lowest in W13 (0.012 Pa s). Although the viscosities of the SPI and PPI emulsions were not significantly different ( $p > 0.05$ ), the WPI-based emulsions showed significant variation across the ratios, with W11 exhibiting higher viscosity than W12 and W13. These results suggest that WPI may form more compact and entangled interfacial layers, thereby increasing the bulk viscosity at lower wall-to-material ratios. In contrast, soy and pea proteins may produce looser structures because of their less flexible globular configurations. These trends align with previous studies, indicating that viscosity is influenced by protein solubility, molecular flexibility, and emulsion droplet interactions.<sup>34,35</sup> The viscosity of the solution depends on several variables, including the concentration of the emulsifying and stabilizing agents, homogenization pressure, and temperature. These factors also regulate the encapsulation efficiency.<sup>34,35</sup> Notably, the viscosity values of the emulsions formulated in this study were lower than those reported by Meena *et al.*<sup>9</sup> and Tonon *et al.*<sup>36</sup> for 30% total solid content.

## 3.2 Color characteristics

Color is a crucial visual and sensory quality indicator of foods. The analysis of curcumin emulsions with different protein isolates (WPI, SPI, and PPI) as wall materials revealed significant variations in color parameters ( $L^*$ ,  $a^*$ , and  $b^*$ ) across different core-to-wall material ratios, as shown in Table 2. The  $L^*$  values (lightness) ranged from  $67.57 \pm 3.93$  (P11) to  $79.61 \pm 0.01$  (W11), with whey-based emulsions appearing significantly lighter. This enhanced lightness may be attributed to the superior color-masking ability of whey proteins, which stems from their excellent film-forming and emulsifying properties.<sup>37,38</sup> Soy and pea protein emulsions showed non-significant changes in  $L^*$  values across different core-to-wall ratios, whereas whey protein emulsions exhibited significant variations. The  $L^*$  value of the emulsion gel in the present study was lower than that of WPI/PPI-based emulsion gels<sup>39</sup> and curcumin-loaded oil body emulsions.<sup>40</sup>

The  $a^*$  values (red-green axis) ranged from  $-5.43 \pm 0.05$  to  $+4.52$ , reflecting the influence of the protein type on the emulsion hue. WPI samples displayed negative  $a^*$  values, indicating a slight greenish tint, whereas PPI emulsions showed positive values, indicating reddish tones. This difference may be due to variations in the encapsulation efficiency or free curcumin migration.

The  $b^*$  values (yellowness) were highest for S11 ( $74.47 \pm 4.45$ ) and lowest for W13 ( $62.77 \pm 0.06$ ). The strong yellowness of the soy-based emulsions indicated the presence of unencapsulated curcumin and less effective color masking. Conversely, lower

$b^*$  values in whey emulsions highlighted WPI's superior encapsulation and pigment masking abilities. These findings align with those of Meena *et al.*<sup>9</sup> and Zou *et al.*<sup>41</sup> for curcumin emulsions, who reported that WPI reduced the visual intensity of encapsulated pigments. Overall, these color parameters play a significant role in the processing, packaging, and acceptability of food products.<sup>42</sup> The results of this study provide valuable insights into the color-masking capabilities of different protein isolates in curcumin emulsions, with whey protein demonstrating the most promising results.

## 3.3 Particle size and zeta potential

**3.3.1 Particle size and polydispersity index.** The particle size of the formulation plays an important role in the stability and efficiency of the bioactive compounds present in the emulsion.<sup>43</sup> Table 2 shows the particle size ( $\mu\text{m}$ ) and zeta potential (mV) of different combinations of curcumin emulsions with different core-to-wall material ratios and different types of protein isolates in the wall material. The particle size of curcumin emulsion with different protein isolates as wall materials was observed in the range of  $322 \pm 5.49$  to  $584.60 \pm 27.7 \mu\text{m}$ . The particle size of the emulsion changed significantly ( $p < 0.05$ ) among the core : wall material ratios, except for PPI (P11, P12, and P13). The smallest particle size ( $322 \pm 5.49 \mu\text{m}$ ) was obtained for the W12 emulsion. The use of different protein isolates (WPI, SPI, and PPI) as wall materials had the least effect on the particle size of the emulsion, and no significant ( $p > 0.05$ ) changes were observed among the emulsions, except for W12. The particle size of the emulsion was previously reported to be in the range of  $356.67$ – $687.80 \mu\text{m}$  for PPI–pectin complexes for curcumin delivery<sup>44</sup> and curcumin emulsions with whey protein concentrate.<sup>9</sup> The results of the current study are within the range of those of these studies. Significant changes in particle size were observed among the different core : wall materials, which were attributed to the formation of larger/smaller oil particles with an increase/decrease in the core material in the emulsion.<sup>9,45</sup> The smaller particle size and low polydispersity index indicated that the emulsion with WPI was stable. The droplet size of curcumin emulsions was higher than that reported by Li *et al.*<sup>46</sup> for curcumin nanoparticles and Liang *et al.*<sup>47</sup> for curcumin emulsion prepared with pectin–whey protein complexes.

**3.3.2 Zeta potential.** Zeta potential is an emulsion property that helps understand the electrostatic forces governing droplet behavior, which ultimately provides stability to the emulsion.<sup>48</sup> The zeta potential of emulsions with WPI and PPI changed significantly ( $p < 0.05$ ) with the varied core : wall material ratio, whereas for SPI, it did not change significantly. The highest zeta potential value was observed for the W12 emulsion ( $-38.50$  mV), and a higher zeta potential value indicates stronger repulsion between the particles and better emulsion stability. This may be due to the uniform particle size of the W12 emulsion.<sup>49</sup> The zeta potential of the curcumin emulsion varied among the various levels of the protein wall material. A stable emulsion was obtained for WPI based on the higher zeta potential, which could be due to the better emulsifying





**Table 3** Different functional group peaks along with the mode of vibration in the curcumin emulsion formulation at different core : wall material ratios with different protein isolates (WPI, SPI, and PPI) in the wall material

Range (in cm <sup>-1</sup> )	W11	W12	W13	S11	S12	S13	P11	P12	P13	Functional group
3800–3700	–	3732	3792	–	3740	–	–	–	–	S–S, stretch
3550–3200	3424, 3295	3537	3494	3505, 3285	–	3507, 3252	3471	3285	3454, 3285	O–H stretching
3000–2800	2921, 2858	2921, 2858	2918, 2855	2925, 2856	2860	2925, 2861	2925, 2854	2928, 2855	2925, 2858	N–H and C–H stretching
3300–2500	–	–	–	2680	–	2679	–	–	–	O–H stretching
2000–1650	1651	1654	1651	–	1667	1654	1657	–	–	C–H bending
1750–1735	1740	1743	1743	1743	1743	1740	1743	1743	1743	C=O stretching
1662–1626	–	–	–	1654	–	–	1657	1644	1644	C=C stretching
1550–1500	1535	1540	1542	–	–	–	1543	1542	1535	N–O stretching
1465–1450	1452	1450	1450	1452	–	–	1452	1450	1442	C–H bending
1420–1330	–	–	–	–	1368	1373	–	–	–	O–H bending
1250–1020	1244	1020, 1148, 1240	1240, 1144, 1020	1151	1151	1240, 1150, 1022	1243, 1149, 1022	1246, 1154, 1025	1237, 1144, 1025	C–N stretching
995–985	–	–	–	–	992	–	–	–	–	C=C bending
980–960	–	–	–	961	–	–	–	–	–	C=C bending
850–550	850	850	840	–	850	847	850	845	840	C–Cl stretching
800–760	–	–	–	–	771	775	769	–	–	C–H bending
730–665	–	704	–	718	713	711	710	712	704	C=C bending
600–500	579	565	579	575	556	575	575	572	575	C–I stretching

properties of whey protein.<sup>50</sup> Another reason for the higher whey protein zeta potential is its inherent negative charge.<sup>51</sup> The zeta potential values obtained in this study were higher than those of curcumin emulsions prepared using a pectin–whey protein complex,<sup>47</sup> where the zeta potential was reported as –31.99 mV.

3.4 Antioxidant activity (DPPH scavenging)

Curcumin can interact with several reactive oxygen species and has been proven to be an effective antioxidant that minimizes the effects of oxidative stress.<sup>52</sup> Table 2 shows the DPPH inhibition activity (%) for different combinations of curcumin emulsions at different core-to-wall material ratios and different types of protein isolates in the wall material. The highest antioxidant activity was observed for P11 (81.38%), whereas the lowest activity was observed for S13 (28.55%). The antioxidant activity of the emulsion was not significantly different at different core:wall material ratios for WP, but significantly varied at different core:wall material ratios for soy and pea protein. The % DPPH of the emulsion with SPI was less than that of the emulsion reported by Fu *et al.*<sup>53</sup> The DPPH inhibition results of whey and pea proteins were supported by Mohamadian *et al.*<sup>54</sup> and Meena *et al.*<sup>9</sup>

3.5 Encapsulation efficiency (EE%)

EE is the parameter by which the performance of the encapsulation process is assessed. The high EE value ensures that bioactive compounds have excellent bioaccessibility.<sup>55</sup> It is influenced by several factors, such as particle size, wall material properties, and processing parameters (*i.e.*, pressure and intensity of treatment).<sup>56</sup> The EE results for different combinations of curcumin emulsions with different core-to-wall material ratios and different types of protein isolates in the wall material are shown in Table 2. The EE of the emulsions exhibited an excellent curcumin encapsulation effect, with an EE value above 88.16%. The core-to-wall material influenced the EE of the emulsions, except for WPI. In addition, the type of protein source also affects EE, but the results are not consistent among the types and their levels in the wall material. The higher EE and high content of soluble curcumin result from the high surface hydrophobicity and curcumin binding capacity of whey protein,<sup>56,57</sup> the high content of hydrophobic amino acids and polar and charged residues in soy protein,<sup>58</sup> and the hydrophobic interaction of curcumin and tryptophan residues of pea protein.<sup>44</sup> The results of the present study are in line with the findings of curcumin emulsions/complexes prepared by Wang *et al.*,<sup>59</sup> Solghi *et al.*,<sup>55</sup> and Vijayan *et al.*,<sup>60</sup> where values of EE (93–98%) are consistent with the present findings.

3.6 FT-IR spectroscopy

The structural properties, nature of bonding and interaction of curcumin with encapsulation materials in curcumin emulsions were evaluated by FT-IR spectroscopy in Fig. 2, and major IR spectral data/peaks are presented in Table 3. The IR spectrum of curcumin emulsion shows a broad range of absorption bands, due to the presence of various functional groups. The IR peaks between 3200 and 2500 cm<sup>-1</sup> may be attributed to the

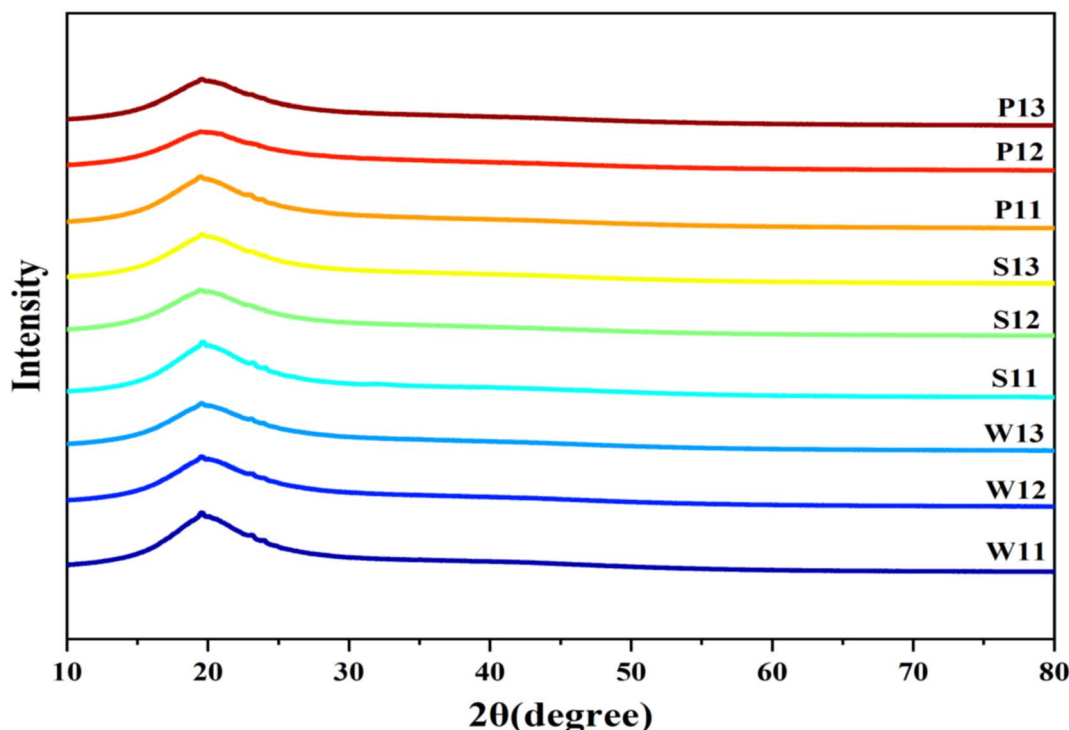


Fig. 3 XRD patterns of curcumin emulsions prepared using different concentrations of whey, soy, and pea protein isolates as wall materials.

stretching vibrations of N–H and the C–H bonds.<sup>61</sup> The sharp peak near  $2932\text{ cm}^{-1}$  probably formed due to complexing of the phenolic hydroxyl group of curcumin with amino acid residues of the protein by hydrogen bonding.<sup>62</sup> The broad bands observed at  $3600\text{--}3100\text{ cm}^{-1}$  are due to interaction between the components; the peaks slightly distorted, probably due to the formation of new hydrogen bonds.<sup>64</sup> The distinct IR absorption band at  $1700\text{--}1600\text{ cm}^{-1}$  is due to the C=O group stretching vibration of the amide I band of proteins.<sup>63</sup> The slight shift ( $1744\text{ cm}^{-1}$ ) of the distinct peaks may be due to changes in protein content during curcumin emulsion preparation. The amide II band at  $1580\text{--}1545\text{ cm}^{-1}$  is due to N–H and C–N stretching that represents the vibrational bands in the protein backbone.<sup>64</sup> Typical peaks of pure curcumin have been reported at  $1627\text{--}1462\text{ cm}^{-1}$  and  $1278\text{--}1160\text{ cm}^{-1}$ .<sup>65</sup> The prominent peak of curcumin disappeared or shifted from this region, which may be due to the formation of complexes during curcumin encapsulation, limiting the stretching vibrations of curcumin.<sup>54,65</sup> The IR spectra of emulsions with different functional peaks at  $2925\text{--}2854$ ,  $1543\text{--}1535$ ,  $1250\text{--}1020$ , and  $840\text{--}850\text{ cm}^{-1}$  due to the presence of N–H, N–O, C–N, and C–Cl stretching, respectively, for different functional groups of curcumin and protein isolates. Similar IR spectra and shifting of the curcumin peak upon encapsulation have been reported for curcumin–protein complexes, such as whey protein isolate–curcumin<sup>15,61</sup> and curcumin complex with whey, soy and pea protein.<sup>60,64,65</sup>

### 3.7 XRD analysis

The crystallinity of protein isolates *i.e.* whey, soy and pea, protein–carbohydrates complexes and free or encapsulated

curcumin was evaluated by XRD, and the patterns are shown in Fig. 3A. A broad peak at  $20^\circ$  was observed in the XRD patterns of the different emulsions, revealing the presence of  $\beta$ -sheets in the polypeptide chain structure.<sup>65</sup> The addition of a broad peak of curcumin emulsions confirmed that curcumin was properly encapsulated in all combinations of proteins, and the complexes formed during encapsulation were amorphous in nature. Guo *et al.*<sup>65</sup> and Fu *et al.*<sup>53</sup> also reported similar findings for curcumin encapsulation using pea protein and soya protein, respectively. The present findings on whey protein are in agreement with those of Mohammadian *et al.*,<sup>54</sup> who reported an amorphous nature for curcumin-loaded whey protein microgels.

### 3.8 Fluorescence microscopy

Fig. 3 shows fluorescence images (at  $60\times$  magnification) of curcumin emulsions with different protein isolates as wall materials at various core-to-wall ratios. Microscopic analysis of the emulsion revealed the type of emulsion system and droplet size using fluorescence images. The encapsulation efficiency of curcumin can be predicted by fluorescence imaging, as curcumin acts as a natural fluorescent material (emission peak at  $\sim 550\text{ nm}$  wavelength).<sup>66</sup> The fluorescence images (Fig. 4) show smaller and more uniform whey protein droplets. The images revealed the presence of free curcumin in both soy and pea proteins, but free curcumin was not observed, whereas the droplet size was uniform for soy and pea protein-based emulsions. In the optical image of the core:wall ratio (1:3), the structure is quite compact, which is probably due to protein enrichment. Recently, similar studies were conducted by Singh



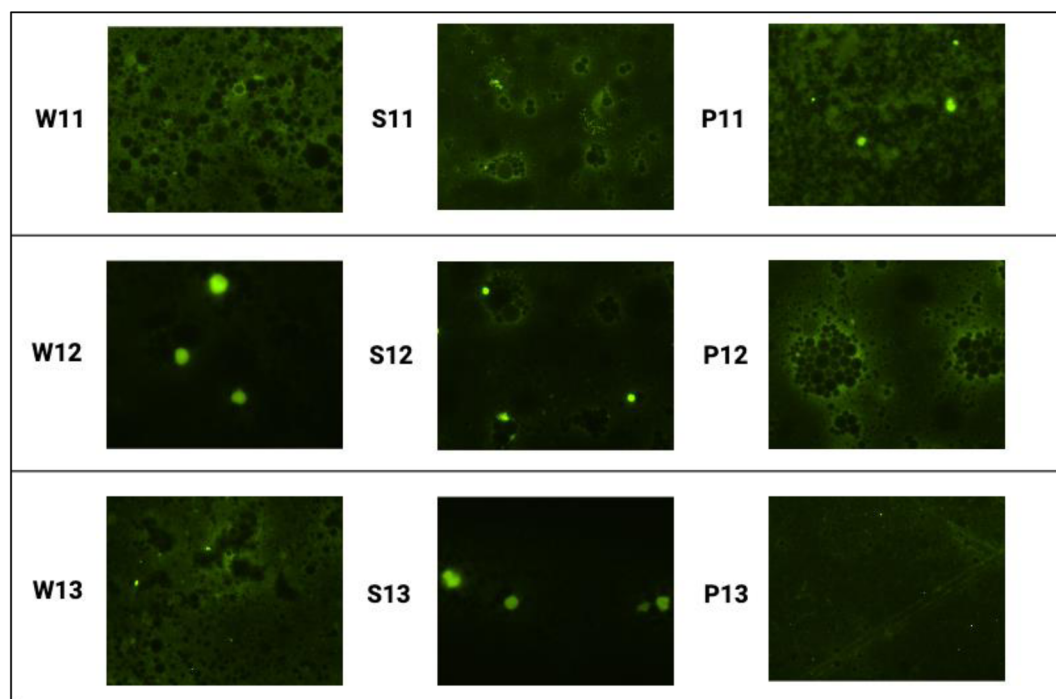


Fig. 4 Fluorescence images of curcumin emulsion with different protein isolates as wall materials: WPI (W11: 1 : 1 (core : wall ratio), W12: 1 : 2 and W13: 1 : 3); SPI (S11: 1 : 1, S12: 1 : 2 and S13: 1 : 3) and PPI (P11: 1 : 1, P12: 1 : 2 and P13: 1 : 3) at 60 $\times$  magnification.

Table 4 Results of emulsion stability for curcumin emulsions prepared using different levels of whey, soy and pea protein isolates as wall materials<sup>a</sup>

Core : wall	1 : 1 (W11)	1 : 2 (W12)	1 : 3 (W13)	1 : 1 (S11)	1 : 2 (S12)	1 : 3 (S13)	1 : 1 (P11)	1 : 2 (P12)	1 : 3 (P13)
Emulsion stability	✗	✓	✓	✗	✓	✓	✗	✓	✓

<sup>a</sup> ✗ – non-stable emulsion and ✓ – stable emulsion.

*et al.*<sup>67</sup> and Wang *et al.*<sup>68</sup> for curcumin-loaded Pickering emulsions, in which the researchers predicted the emulsion stability and interactions between the constituents of the emulsions.

### 3.9 Emulsion stability

Emulsion stability is an important practical property of emulsions that is influenced by parameters such as surfactant concentration, particle size distribution, and interactions between the emulsion constituents.<sup>69,70</sup> The stability results for the emulsions are presented in Table 4. The emulsion with a core : wall material ratio of 1 : 1 for different protein sources was unstable during the stability test, which could be due to the high amount of core material and low amount of wall material. However, the other emulsion combinations were stable.

## 4 Conclusion

This study successfully developed protein-stabilized emulsions to encapsulate curcumin, demonstrating superior physico-chemical properties with whey protein isolate (WPI) and

highlighting the high antioxidant activity of pea protein isolate (PPI). These findings provide a robust foundation for creating functional, clean-label delivery systems that address the inherent instability and poor water solubility of curcumin. Future research should build on these results by transitioning from laboratory-scale analyses to biological and industrial validations. This includes testing the formulations in dynamic *in vitro* gastrointestinal models to assess bioaccessibility and permeability and optimizing process parameters for industrial scale-up. Further investigation into synergistic protein blends of WPI and PPI is warranted to create next-generation emulsions that balance superior physical stability with enhanced nutritional and functional properties. Such work will be crucial for translating these findings into viable and long-term functional foods and nutraceuticals.

## Conflicts of interest

The authors declare no conflicts of interest relevant to this article.



## Abbreviations

WPI	Whey protein isolate
SPI	Soy protein isolate
PPI	Pea protein isolate
FT-IR	Fourier-transform infrared spectroscopy
XRD	X-ray diffraction
DLS	Dynamic light scattering
DPPH	2,2-Diphenyl-1-picrylhydrazyl
KBr	Potassium bromate
EE	Encapsulation efficiency

## Data availability

The data will be provided upon request.

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