



Cite this: *Chem. Soc. Rev.*, 2026, 55, 2959

Photoswitchable imaging contrast agents as an emerging frontier in precision bioimaging

Xiaoqiong Tan,^{†ad} Qiankun Ni,^{†b} Ting Luo,^b Ki Taek Nam,^f Hong-Bo Cheng,^{id *ae} Paul J. Dyson,^{id *e} Xing-Jie Liang^{id *b} and Juyoung Yoon^{id *cg}

Photoswitchable molecules exhibit reversible photoisomerization between distinct isomeric states under alternating light irradiation, possessing unique photophysical properties. Derived from these molecules, photoswitchable imaging contrast agents (PICAs) offer significant advantages over conventional agents: (1) background suppression through state-switching algorithms; (2) continuous imaging capability enabled by high fatigue resistance; (3) rapid signal modulation minimizing acquisition time. A comprehensive analysis of PICA advancements and persistent challenges across bioimaging applications is imperative. This review systematically evaluates several representative photoswitchable molecules as PICAs and proposes innovative molecular design and assembly strategies to address the limitations hindering the widespread implementation of PICAs in bioimaging, such as hydrophobicity, short-wavelength excitation, and suboptimal emission performance. Additionally, recent advances in PICAs for fluorescence, phosphorescence, photoacoustic, magnetic resonance and Raman imaging are summarized, highlighting their significant contributions to improving bioimaging quality and precision. Finally, an outlook is provided on the intersection of PICA design and property optimization for propelling both preclinical physiological research and clinical diagnostic and monitoring technologies.

Received 2nd January 2026

DOI: 10.1039/d6cs00003g

rsc.li/chem-soc-rev

^a State Key Laboratory of Organic-Inorganic Composites, Beijing Laboratory of Biomedical Materials, College of Materials Science and Engineering, Beijing University of Chemical Technology, Beijing 100029, P. R. China. E-mail: chenghb@mail.buct.edu.cn

^b CAS Key Laboratory for Biomedical Effects of Nanomaterials and Nanosafety, CAS Center for Excellence in Nanoscience, National Center for Nanoscience and Technology, Beijing 100190, P. R. China. E-mail: liangxj@nanoctr.cn

^c Department of Chemistry and Nanoscience, Ewha Womans University, Seoul 03760, Republic of Korea. E-mail: jyoony@ewha.ac.kr

^d Department of Biomedical Engineering, College of Future Technology, Peking University, Beijing 100871, P. R. China

^e Institute of Chemical Sciences and Engineering, Ecole Polytechnique Fédérale de Lausanne (EPFL), Lausanne CH-1015, Switzerland. E-mail: paul.dyson@epfl.ch

^f Severance Biomedical Science Institute, Brain Korea 21 PLUS Project for Medical Science, College of Medicine, Yonsei University, Seoul 03760, Republic of Korea

^g Graduate Program in Innovative Biomaterials Convergence, Ewha Womans University, Seoul 03760, Korea

[†] These authors contributed equally to this work.



Xiaoqiong Tan

Xiaoqiong Tan received MS degree in chemistry from Beijing University of Chemical Technology in 2023 under the supervision of Prof. Hong-Bo Cheng. She received joint training at National Center for Nanoscience and Technology under the supervision of Prof. Xing-Jie Liang from 2021–2023. Currently, she is pursuing a PhD in Biomedical Engineering at Peking University. Her research interests mainly focus on the photo-responsive nanomaterials for imaging-guided disease diagnosis and therapy.



Qiankun Ni

Qiankun Ni received his BS degree in Materials Chemistry from University of Science and Technology of China in 2018, followed by his PhD degree in Chemistry from Tsinghua University in 2024 under the supervision of Prof. Jinghong Li. From 2018–2024, he received joint training at National Center for Nanoscience and Technology under the supervision of Prof. Xing-Jie Liang. His research interests focus on the development of biomaterials for T cell tumour immunotherapy.



1. Introduction

As a bridge connecting molecular mechanisms and macroscopic physiological properties, bioimaging provides an indispensable tool in preclinical physiological research, medical diagnosis, and drug development. The core challenge in bioimaging technology is achieving a balance among molecular specificity, spatiotemporal resolution, and biocompatibility in complex biological systems. Imaging contrast agents (ICAs) have made considerable strides in bioimaging, driven by advances in synthetic and assembly techniques that enhance their capabilities for biological analysis and disease diagnosis. Photoswitchable molecules, which undergo reversible structural transition between *trans* and *cis* forms or between ring-opened and

ring-closed configurations resulting in significant changes in fluorescence or absorption spectra upon irradiation at two distinct wavelengths, serve as the foundation for developing photoswitchable ICAs (PICAs).¹ Owing to their distinctive photoactive and photo-controllable physical properties, PICAs outperform conventional ICAs in several key ways, offering a revolutionary solution to overcome current limitations in bioimaging.^{2–5} First, the switchable states of PICAs can effectively enhance imaging contrast and eliminate imaging background noise through specific algorithms, while most conventional ICAs emit a single and “always on” signal.^{6–8} Second, the high fatigue resistance of PICAs facilitates continuous and stable imaging,^{9,10} contrasting with some conventional dyes, such as indocyanine green (ICG), which experience rapid photobleaching even under short irradiation times.



Ting Luo

Ting Luo received her BS degree in 2017, and then MS degree in 2020 both from China Medical University. She received joint training at National Center for Nanoscience and Technology from 2021–2024 under the supervision of Prof. Xing-Jie Liang, and obtained her PhD degree from Nankai University in 2024. Her research interests mainly focus on the imaging-guided diagnosis and treatment for cancer, as well as advanced biomaterials for improving immunotherapies.



Ki Taek Nam

Dr. Nam received his veterinary training with special qualifications in veterinary pathology and mouse pathology from Seoul National University in 1991. After completing his PhD, which focused on the induction of gastric cancer in rodent models, he did his post-doctoral research at Vanderbilt University. He joined the faculty at Vanderbilt in 2010 and moved to Yonsei University College of Medicine in 2013. Currently, he serves as a professor in the Department of Biomedical Sciences and as a scientific advisor in the Department of Laboratory Animal Resource. His research interests center on mouse models for various human diseases, including cancer. He has also conducted efficacy tests on functional materials and therapeutic agents.



Hong-Bo Cheng

Hong-Bo Cheng received his PhD in organic chemistry from Nankai University in 2014 under Prof. Yu Liu. He has worked at the Chinese Academy of Sciences' Key Laboratory for Biomedical Effects of Nanomaterials & Nanosafety and was a postdoctoral fellow with Prof. Juyoung Yoon at Ewha Womans University (2017–2019). Currently, he is a professor at Beijing University of Chemical Technology and served as a visiting professor at EPFL with Prof. Paul J. Dyson (2023–2024). His research focuses on nanoprobes and photoresponsive biomaterials.



Paul J. Dyson

Paul J. Dyson is a professor at the École Polytechnique Fédérale de Lausanne (EPFL), where he has led the Laboratory of Organometallic and Medicinal Chemistry since 2002 and serves as the Dean of the Faculty of Basic Sciences since 2021. He chaired the Institute of Chemical Sciences and Engineering from 2008 to 2016. He has earned several notable awards including the Werner Prize of the Swiss Chemical Society (2004), the Luigi Sacconi Centennial Medal of the Italian Chemical Society (2011), the Bioinorganic Award of the Royal Society of Chemistry (2015), European Sustainable Chemistry Award (2018) and the Green Chemistry Award of the Royal Society of Chemistry (2020).



Third, the fast photo-response of PICAs promotes instantaneous switching between two different signals, drastically shortening imaging time.^{11–13} Additionally, rational molecular design, such as group modification, structural adjustment, and chemical conjugation, along with advanced assembly strategies, has further expanded the bioimaging applications of PICAs.^{14–18} Further advancements could aim at improving their stability, hydrophilicity, biocompatibility, targeted delivery, deep penetration capability under long-wavelength excitation, emission performance and photoisomerization efficiency.^{19–23}

To date, an increasing number of studies have demonstrated the utility of PICAs across diverse imaging modalities, including fluorescence imaging (FI), phosphorescence imaging (PI), photoacoustic imaging (PAI), magnetic resonance imaging (MRI), and Raman imaging (RI) (Fig. 1). Despite these developments, a comprehensive review summarizing the role of PICAs in various bioimaging techniques is still lacking. Therefore, a systematical evaluation of recent progress in PICA-based bioimaging is both timely and necessary.

This review systematically evaluates several representative photo-switchable molecules functioning as PICAs and elucidates their underlying photoisomerization mechanisms. It also describes diverse optimization strategies focused on addressing current challenges that hinder the broader applications of PICAs across different bioimaging modalities. Finally, an outlook on this emerging field at the intersection of PICA design and performance optimization is presented. By consolidating these advancements, this review aims to support both preclinical physiological research and clinical diagnostic and monitoring technologies for various diseases.

2. Photoswitchable molecules as photoswitchable imaging contrast agents

The photoswitchable (photochromic) phenomenon was first described by Y. Hirshberg in the 1950s, who proposed that

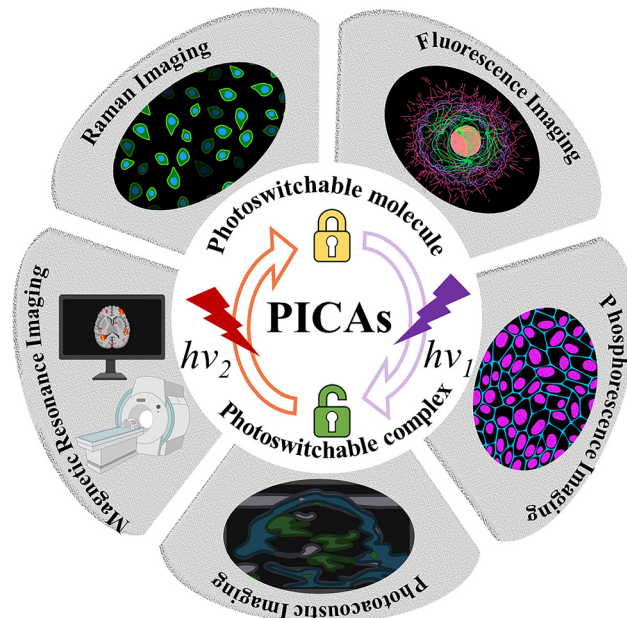


Fig. 1 Schematic illustration of photoisomerization of PICAs and their applications across multiple imaging modalities, including fluorescence, phosphorescence, photoacoustic, magnetic resonance, and Raman imaging. Created with BioRender.com.

the reversible structural transformation of dianthrone under ultraviolet (UV) irradiation was related to photo-responsive switching behavior.^{24,25} Photochromism is now typically defined as the photo-induced reversible switching between two isomers, each displaying distinct physical properties when exposed to specific wavelengths of electromagnetic radiation, such as UV, visible (Vis), or infrared light.²⁶ Consequently, these molecules are designated as photoswitchable or photochromic molecules.

The reversible transition of photoswitchable molecules is driven by various mechanisms, including ring-opening/closing reaction, *trans*-to-*cis* or *cis*-to-*trans* isomerization, intramolecular proton transfer, intramolecular electron transfer, and



Xing-Jie Liang

Xing-Jie Liang received his PhD from the National Key Laboratory of Biomacromolecules, Institute of Biophysics at CAS, Beijing, China. He finished his postdoc at the Center for Cancer Research, NCI, NIH, and worked as a Research Fellow at the Surgical Neurology Branch, NINDS. Professor Liang is the director of laboratory of controllable nanopharmaceuticals, National Center for Nanoscience and Technology of China. He is the Member of

Academia Europaea and the elected Fellow of American Institute for Medical and Biological Engineering (AIMBE). His research interests are in elucidating mechanisms to improve drugability and nanomedicinal bioavailability by nanotechnology.



Juyoung Yoon

Juyoung Yoon received his PhD (1994) from The Ohio State University. After completing post-doctoral research at UCLA and at Scripps Research Institute, he joined the faculty at Silla University in 1998. In 2002, he moved to the Ewha Womans University, where he is currently a Distinguished Professor of the Department of Chemistry and Nano Science. His research interests include investigations of fluorescent probes, activatable photosensitizers, theranostics and organic functional materials. He is listed as a highly cited researcher in chemistry of 2014–2024.



Common photoswitchable molecules

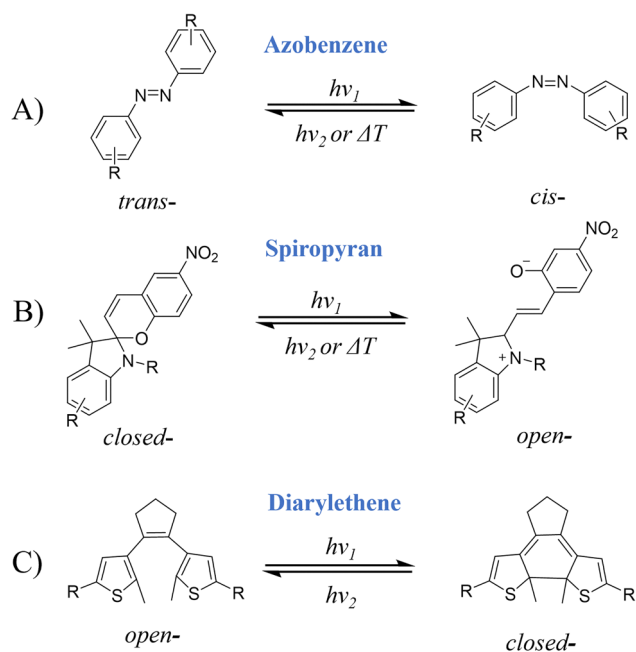


Fig. 2 Common photoswitchable molecules concluding (A) azobenzene, (B) spiropyran, (C) Diarylethene and their photoisomerization behavior.

cycloaddition.^{27,28} Photoswitchable molecules have been extensively studied for preclinical physiological research, disease diagnosis and monitoring. These photoswitchable molecules and their derivatives function as PICAs themselves or as switches that regulate the photophysical properties of additional contrast agents. This section discusses several representative photoswitchable molecules either served directly as PICAs or used to construct PICAs, including azobenzene (Azo), spiropyran (SP), diarylethene (DAE), and photoswitchable proteins, with particular elucidation on their photoswitching processes and mechanisms (Fig. 2).

2.1. Azobenzene

Azo is the most prominent photoswitchable molecule and has been studied extensively. Azo exists as the thermodynamically stable *trans* configuration at room temperature. It undergoes *trans*-to-*cis* isomerization upon UV irradiation and can reversibly return to the *trans* isomer through Vis light irradiation, slow thermal relaxation, or heating (Fig. 2A).²⁹ This *trans*-to-*cis* photoisomerization proceeds by torsion around N=N bond or torsion accompanied with in-plane inversion (Fig. 3A).^{30,31} The photoswitching behavior of Azo results in distinct light absorption profiles: the *trans* isomer absorbs light irradiation below 400 nm, while the *cis* isomer exhibits a red-shifted absorption above 440 nm, attributable to the occurrence of a $n-\pi^*$ transition (Fig. 3B).³² Considering these unique optical properties, combined with its structural tunability and inherent stability, Azo holds great potential for developing advanced PICAs.^{33,34}

Because of its intrinsic non-fluorescent nature, Azo is typically employed as a building block to develop Azo-based PICAs,

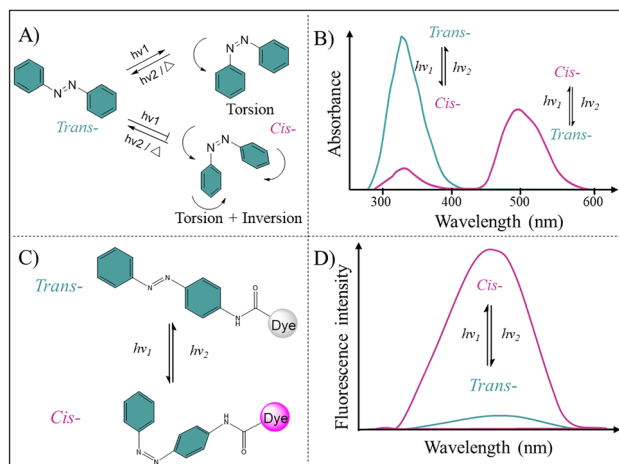


Fig. 3 (A) The *trans*-to-*cis* photoisomerization of Azo proceeds by torsion around N=N bond or torsion accompanied with in-plane inversion. (B) Absorption spectral changes of Azo between the *trans* and *cis* states. (C) Fluorescence modulation of the dye by the conjugated Azo moiety upon UV and Vis light irradiation. (D) Fluorescence intensity of the Azo-modified dye derivative decreases upon *trans*-to-*cis* isomerization and recovers following *cis*-to-*trans* conversion.

through covalent conjugation or non-covalent interaction with fluorescent dyes or other imaging contrast agents. In general, photoisomerization of Azo regulates the fluorescence intensity or wavelength of adjacent dyes through altering their electronic environment and rigidity constraints (Fig. 3C). In its *trans* state, Azo functions as an energy acceptor, quenching dye fluorescence through Förster resonance energy transfer (FRET) effect (Fig. 3D).^{35,36} When Azo transforms to the *cis* state under UV light irradiation, the conjugated system is disrupted, thus enhancing the molecular dipole moment and polarity. This change eliminates the absorption-emission spectral overlap between Azo and dye, restoring the dye's emission. Conversely, Vis light irradiation reverts Azo to its *trans* isomer, re-establishing the conjugated system to recover the non-fluorescence state. Moreover, the *trans*-to-*cis* isomerization increases molecular steric hindrance, disrupting the planar, conjugated structure of *trans*-Azo dye. This alleviates its aggregation-caused quenching (ACQ) effect and promotes aggregation-induced emission (AIE) behavior. Due to the distinct spatial conformation change, *trans*-*cis* isomerization of Azo has been widely employed to reversibly modulate macromolecular or supramolecular assembly structures for precise control over their emission properties.³⁷⁻³⁹

However, the broader application of Azo-based PICAs in bioimaging is still hindered by their intrinsic hydrophobicity, short-wavelength excitation, and fast thermal relaxation. The second section will elaborate on the optimization strategies through structural engineering and functional modification to address these challenges.

2.2. Spiropyran

SP represents another promising candidate for photoswitchable bioimaging, featuring a distinct photoisomerization mechanism



from Azo. SP maintains a thermal-stable ring-closed form in its native state, with UV absorption (<400 nm). Upon appropriate light irradiation, SP undergoes reversible photochemical cleavage of the C–O bond to open the pyran ring, transforming into the merocyanine (MC) isomer (Fig. 2B).⁴⁰ Moreover, MC exhibits *cis-trans* (*Z/E*) isomerism at its C=C bond, with the *trans* (*E*) configuration typically being thermodynamically favored arising from decreased steric strain (Fig. 4A). Notably, the SP → MC isomerization generates an extended π -conjugated system with charge separation, resulting in a large electric dipole moment.⁴¹ This unique structural rearrangement red-shifts the absorption spectral to the 500–600 nm region (Fig. 4B) and dramatically enhances fluorescence enhancement from a weakly emissive state (low intensity at <450 nm) to a strong fluorescent one (high intensity at >600 nm) (Fig. 4C). Conversely, the MC → SP isomerization occurs through spontaneous thermal relaxation in the dark, heating, or Vis light exposure, with reduced absorption and fluorescence.^{41,42} The reversible photoisomerization of SP enables precise control over the on/off state of fluorescence, or facilitates switching between multiple emission wavelengths for monochromatic or multicolor FL.^{43,44} In addition to serving as a direct photoswitchable fluorophore, SP provides three common modification sites for covalent dye conjugation, thereby allowing modulation of the fluorescence of adjacent dyes *via* FRET (Fig. 4D).

As an important photoswitchable molecule developed after SP, spirooxazine (SO) has emerged as a research hotspot in the frontier fields such as super-resolution imaging, target labeling, and *in vivo* tracking, due to its unique photophysical properties and excellent stability. Sharing similar structure with SP, SO also exhibits reversible photoisomerization where UV light irradiation cleaves the C–O bond to open the oxazine ring, and Vis light irradiation triggers reversible ring closing through an electrocyclic reaction that reforms the C–O bond (Fig. 5A).⁴⁰ Unlike SP, which contains only oxygen atom in

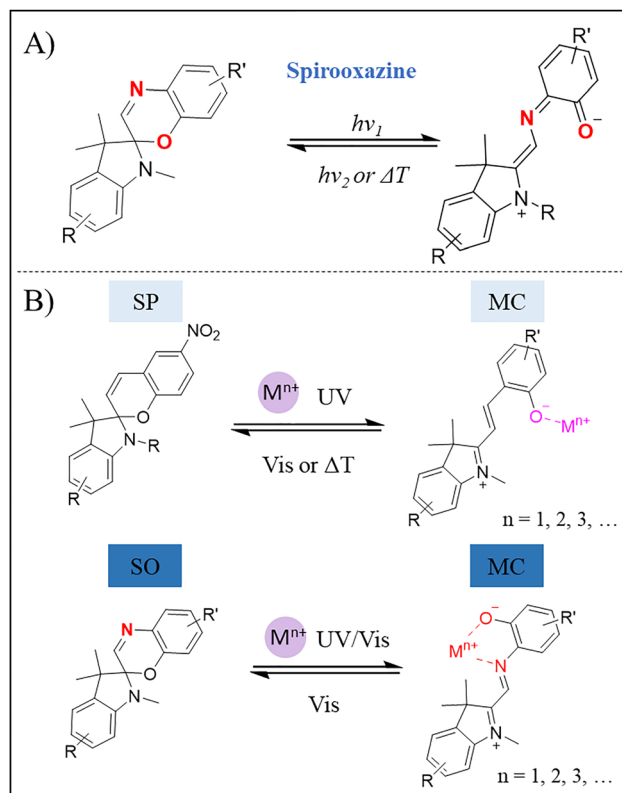


Fig. 5 (A) The ring-opening/ring-closing reactions of SO. (B) Responsiveness of SP and SO to metal ions.

the pyran ring, SO contains both oxygen and nitrogen atoms in the heterocycle, along with an extended π -conjugated system. As a result, SO has longer half-life and higher photostability compared to SP. The thermal stability of SO in its ring-opened form lays the foundation for its application in time-resolved imaging.^{45–49} Furthermore, SO exhibits excellent fatigue resistance, withstanding over 10^4 switching cycles without significant photodegradation, far surpassing SP that typically achieves fewer than 10^3 cycles.^{50,51} This superior performance primarily stems from the electronic effect and steric hindrance effect of its oxazine ring, which together effectively suppress photobleaching.⁵²

Apart from functioning as photoswitchable fluorescent probes, both SP and SO perform superior potential for biosensing. This depends on the distinct responsiveness of their molecular structures: the oxygen atom in the pyran/oxazine ring enables sensitive detection of metal ions, while the reactive C=C bond allows for the recognition of various analytes (*e.g.*, CN^- , H_2S , SO_2).^{53,54} Metal ion binding promotes the UV light-induced ring-opening, stabilizes the resultant ring-opened isomer and acts as an auxochrome with unique absorption and fluorescence signals (Fig. 5B). The coordination affinity of the oxygen and nitrogen atoms in SP or SO varies considerably among different metal ions, such as Li^+ , Na^+ , K^+ , Mg^{2+} , Cu^{2+} , Ca^{2+} , Fe^{2+} , Fe^{3+} , Zn^{2+} , and Ag^+ . The coordination of metal ions alters the geometry and π -conjugation extent of SO, leading to distinct fluorescence signals upon external stimulation.

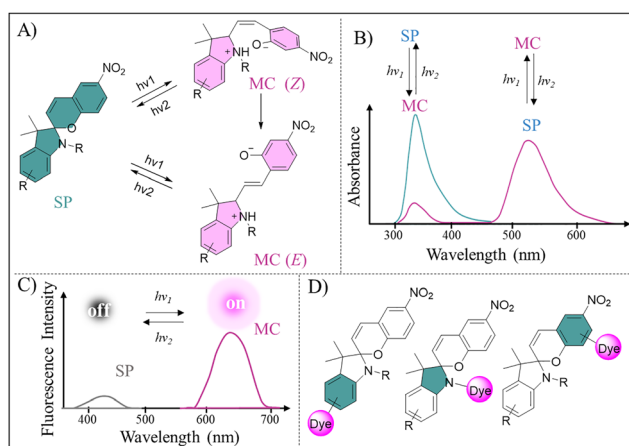


Fig. 4 (A) Photoisomerization from SP to MC (*Z/E* conformation) occurs through ring-opening with or without C=C bond torsion. (B) Absorption spectral changes of SP and MC upon different light irradiation. (C) The fluorescence changes of SP and MC upon different light irradiation. Spot size is positively correlated with fluorescence intensity; color indicates emission state (black = non-fluorescent, red = fluorescent). (D) Three common modified sites on SP for covalent tethering to dye molecules.



This property positions SO as a promising probe for detecting diverse metal ions in living organisms.^{55,56}

Despite ongoing molecular optimization of SP and SO, its applications in bioimaging remain underexplored, primarily due to several persistent challenges: (1) excitation wavelengths are largely confined to the UV-Vis region (<600 nm); (2) enhanced thermal stability may paradoxically limit the photoisomerization efficiency; (3) suboptimal biocompatibility hinders effective cellular uptake; and (4) low fluorescence quantum yield. In the second section, the corresponding solutions strategies will be discussed.

2.3. Diarylethene

DAE has a core structure consisting of two aromatic rings (typically thiophene or benzothiophene units) bridged by the ethylene bond in cyclopentene. Upon irradiation with altering lights, DAE undergoes reversible photocyclization, transforming from an unconjugated, ring-opened isomer to a conjugated, ring-closed isomer (Fig. 2C).⁵⁷ The ring-opened DAE isomer exists in two conformational states: parallel and anti-parallel (Fig. 6A). Notably, photoisomerization occurs exclusively in the anti-parallel conformation, whereas the parallel conformation remains photochemically inactive.^{57,58} This conformational dependence is a critical consideration in the molecular design and application of DAE.

UV light exposure dynamically rearranges the π -electron cloud of the vinyl double bond and the π - π conjugated system in the DAE skeleton. According to the Woodward–Hoffmann rules, the photocyclization of the ring-opened DAE proceeds by the photochemical conrotatory electrocyclicization of its central 6π -electron 1,3,5-hexatriene system to form a new C–C bond.⁵⁹

This process results in the isomerization from the ring-opened state to the ring-closed state to augment intramolecular conjugation, consequently red-shifting the absorption spectrum (λ_{\max} from ~ 300 nm to ~ 550 nm) along with visible color change (Fig. 6B). Conversely, irradiation with Vis light or near-infrared (NIR) light triggers the ring-opening reaction of DAE, cleaving the newly formed C–C bond and reverting the molecule to its initial ring-opened state.

Diverse molecular design strategies have been proposed to precisely tune the photoswitching properties of DAE. The reversible photocyclization activity is governed by the core architecture of DAE, where the two core aryl groups offer the foundation for electron conjugation. Their electronic properties and steric hindrance determine the overall stability and reaction activity of the molecule. DAE derivatives exist in either symmetric configuration (with identical aryl ring type) or asymmetric configuration (with distinct aryl ring type) (Fig. 6C). Typical aryl ring units include thiophene, benzothiophene, sulfonyl-substituted thiophene, thiazole, and oxazole. Asymmetric DAE commonly improves photoisomerization efficiency and photostability, yet their synthetic complexity is a challenge. Furthermore, the structure of the vinyl bridge plays a crucial role in modulating the photo-switching performance of DAE, such as photoisomerization efficiency, chemical stability, fatigue resistance, and thermal half-life (Fig. 6D).

In addition, DAE derivatives display high value in bioimaging because of their exceptional thermal stability (both isomers are indefinitely stable at room temperature in the darkness) and fatigue resistance (allowing hundreds to thousands of switching cycles without pronounced degradation). These properties render DAE particularly suitable for long-term FI, where they support repeated on/off toggling and diverse imaging modalities.^{20,60,61}

As a molecular switch, DAE regulates the fluorescence emission of conjugated dyes *via* a FRET process.⁶² Fluorophores are typically attached to one or both terminal thiophene units of the DAE core, which allows effective fluorescence modulation through a photocyclization-induced quenching mechanism (Fig. 6E). Moreover, DAE can be co-encapsulated with fluorophores into nanoparticles *via* supramolecular interactions, thereby achieving FRET-based reversible fluorescence switching for bioimaging.¹² However, similar to other photoswitchable molecules, DAE faces inherent challenges in bioimaging applications, such as short-wavelength activation, hydrophobicity, and weak emission. The main strategy for optimizing the properties of DAE is molecular engineering through the introduction of functional groups, such as: (1) heterocyclic aryl groups to red-shift absorption; (2) fluorophores to improve emission; and (3) hydrophilic moieties to increase solubility in biological media.^{10,14,63–66}

2.4. Other photoswitchable molecules

In addition to the development based on classical molecule scaffolds, other photoswitchable molecules, such as rhodamine, dihydropyrene (DHP), hexaarylbiimidazole (HABI), and donor–acceptor Stenhouse adduct (DASA), have been explored as PICAs for photoswitchable bioimaging.

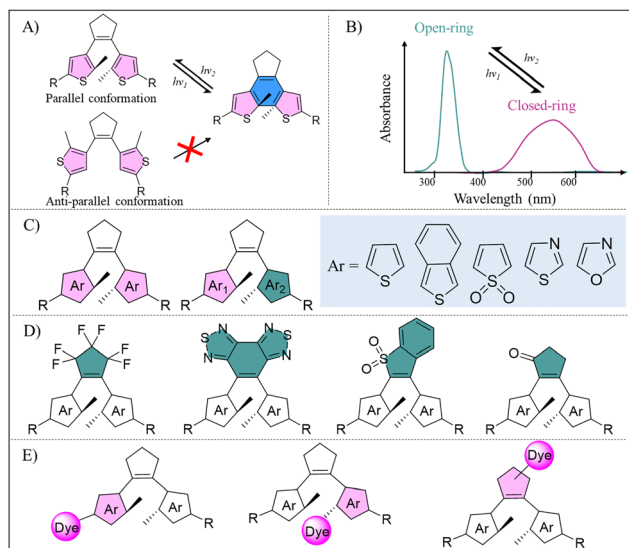


Fig. 6 (A) Parallel and anti-parallel configurations of DAE. (B) Absorption spectra of DAE in ring-opened and ring-closed states. (C) Symmetric and asymmetric DAE conformations with identical or distinct aryl ring types. (D) Diverse DAE derivatives based on novel vinyl bridge types, including perfluorocyclopentenes, dithiazolopyrazine, benzothiophene sulfone, and cyclopentenone. (E) Three common modified sites on DAE enabling covalent attachment to dye molecules.



The photoswitchable reaction of rhodamine amides was first proposed by Knauer and Gleiter in the 1970s, a discovery that has since spurred the development of numerous photoswitchable rhodamine derivatives.^{67,68} The photoswitching mechanism of rhodamine depends on the reversible cleavage and reformation of the spirocyclic bond, which interconverts between a fluorescent zwitterionic (ring-opened) state and a non-fluorescent spirolactone (ring-closed) state. Rhodamine offers advantages over conventional fluorescent probes in super-resolution imaging, including high fluorescence quantum yield and excellent photochemical stability.⁹ However, the short lifetime of its fluorescent ON-state and inherent switching stochasticity fundamentally constrain the extension of temporal imaging windows and precise spatiotemporal coordination, necessitating a trade-off between spatiotemporal super-resolution performance. A novel rhodamine derivative, Rh-Gly, was developed with remarkable fluorescence brightness and prolonged ON-state stability under UV light irradiation.⁶⁹ The presence of a carboxyl group in Rh-Gly stabilizes its zwitterionic ring-opened conformation *via* intramolecular hydrogen bonding. This yields a prolonged ON-state lifetime (68.5 ms) and enhanced fluorescence brightness (3.57×10^4 photons per s), enabling simultaneous high spatial resolution (~ 50 nm) and temporal resolution (10 s) in live-cell super-resolution imaging.

HABI is another class of photoswitchable molecule characterized by the reversible dimerization of 2,4,5-triphenylimidazolyl radical (TPIR).^{70–73} Upon 365 nm light irradiation, HABI dissociates into TPIR radicals, which can subsequently revert to HABI through thermal transformation.⁷⁴ HABI has attracted increasing attention due to its rapid photo-responsive behavior and robust fatigue resistance. Owing to this reversible dimerization involving bond cleavage and reformation, HABI has been covalently attached to BODIPY dye to modulate their fluorescence.⁷⁵

Compared to conventional UV-triggered photoswitchable molecules, negative photoswitchable molecules such as DHP and DASA undergo photoisomerization upon Vis/NIR light irradiation. This reduces the phototoxicity to cells or living organisms, rendering them more suitable for prolonged imaging. Owing to the rigid planar structure and extended π -system, DHP absorbs at longer wavelength (> 600 nm) than conventional photoswitchable molecules.^{76,77} Its negative photoswitching behavior involves a 6π -electrocyclic rearrangement that disrupts the π -conjugation, yielding a decolorized cyclopentadiene upon Vis/NIR light irradiation. However, photoisomerization under illumination exceeding 750 nm remains challenging. One viable synthetic strategy is to introduce donor and acceptor groups at the 2,7- or 4,9-substituted positions of DHP, which red-shift the absorption spectrum into the NIR region reaching 900 nm.⁷⁸ In addition, DASA represents a new class of negative photoswitchable molecules with a D- π -A architecture. This structure allows reversible photoisomerization between a colored (open) form and a colorless (closed) form upon Vis-NIR light (490–700 nm) irradiation.⁷⁹

Despite the development of numerous novel photoswitchable molecules, their applications in bioimaging remain limited and require further exploration.

2.5. Photoswitchable proteins

The advent of reversibly photoswitchable fluorescent proteins (RSFPs) has revolutionized bioimaging, particularly by super-resolution microscopy to visualize subcellular dynamics at the nanoscale in living systems.^{80–82} The first RSFP, Dronpa, was discovered by Ando *et al.* in bacterial studies.⁸³ Since then, a wide variety of RSFPs have been rapidly developed to meet the specific demands of advanced bioimaging modes. Under alternating light irradiation at two distinct wavelengths (typically UV light and Vis light), RSFPs undergo reversible positive or negative photoswitching between fluorescent and non-fluorescent states (Fig. 7A).⁸⁴ This is primarily driven by light-induced conformational change of the chromophore in RSFPs, though it may also involve protonation or other complex photophysical processes.⁸⁵ Fig. 7B illustrates the photoswitching mechanisms of three classical RSFPs (Dronpa, Padron, and Dreiklang) based on distinct structural changes of chromophores: *cis-trans* isomerization, *trans-cis* isomerization, and hydration/dehydration reaction, respectively. The chromophore of Dronpa follows by a negative photoswitching mechanism: Vis light irradiation (*e.g.*, 488 nm) transforms it from a fluorescent deprotonated *cis*-state to a non-fluorescent protonated *trans*-state, whereas UV or blue light irradiation (*e.g.*, 405 nm) reverses this process to restore the fluorescence.⁸⁶ Padron, a Dronpa variant, maintains the same chromophore structure but displays positive photoswitching process: UV or blue light irradiation (*e.g.*, 405 nm) triggers *trans*-to-*cis* isomerization and deprotonation, consequently activating bright green fluorescence at 522 nm.⁸⁷ Conversely, Vis light (*e.g.*, 488 nm) returns the chromophore to a non-fluorescent protonated *trans*-state. Dreiklang is another RSFP that holds a distinctive photoswitching mechanism. Unlike the *cis-trans* isomerization typical of many RSFPs, Dreiklang relies on a reversible hydration/dehydration reaction occurring at a specific carbon atom in the imidazolidinone ring.⁸⁸ Specifically, 405 nm light irradiation triggers a hydration reaction that converts the chromophore from a fluorescent state to a non-fluorescent state, while 365 nm light irradiation drives a dehydration reaction to restore the fluorescence. However, most RSFPs are limited by intrinsic photophysical properties, including finite switching cycles, low photon output, and short-wavelength-light activation, which collectively compromise their applications in live-cell super-resolution imaging and *in vivo* imaging.

Over the past decade, considerable efforts have been directed toward engineering improved RSFPs for super-resolution and *in vivo* imaging. The Padron-derived mutant, Kohinoor, exhibited rapid switching kinetics, enhanced photostability, and high quantum yields, enabling successful super-resolution imaging in living cells *via* reversible saturable optical fluorescence transition (RESOLFT).⁷ Moreover, Kohinoor showed high photostability with minimal fluorescence decay per switching cycle ($\sim 0.1\%$, compared to Padron's 2.5%). It also achieved 17-fold more cycles than Padron (340 cycles *versus* 20 cycles) before 50% intensity loss, demonstrating accelerated switching kinetics. Kohinoor exhibited substantially higher quantum yield than Padron in both states: the open state ($\Phi_{\text{on}} = 0.02$) and closed



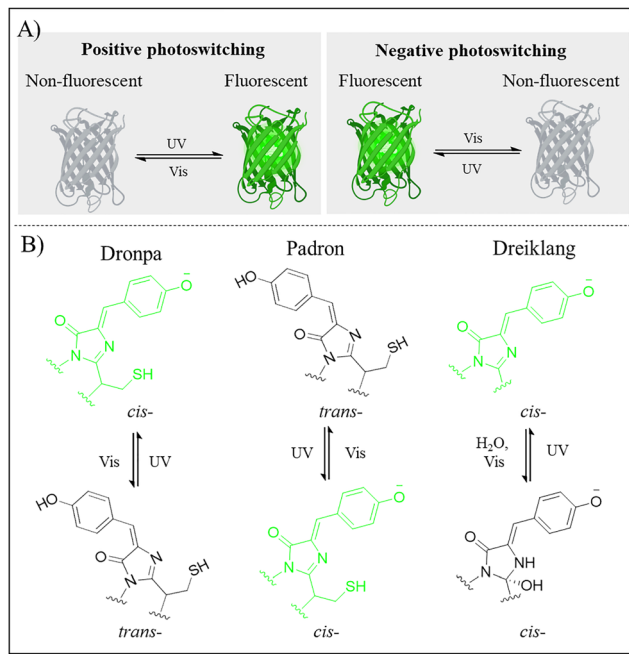


Fig. 7 (A) Light-induced positive or negative photoswitching of RSFPs between their non-fluorescent and fluorescent states. (B) Three photoswitching mechanisms involving *cis*–*trans* isomerization, *trans*–*cis* isomerization, and hydration/dehydration reactions. The color of molecular structure indicates emission state (black = non-fluorescent, green = fluorescent).

state ($\Phi_{\text{off}} = 0.15$) exceeded Padron's corresponding values ($\Phi_{\text{on}} \approx 0.003$, $\Phi_{\text{off}} \approx 0.04$) by 6.7-fold and 3.75-fold, respectively. These improvements enabled biocompatible live-cell imaging at ultralow laser intensity (0.004 J cm^{-2}).

Most RSFPs are activated by UV light, raising concerns over phototoxicity and scattering effect. Therefore, there is a demand for the development of RSFPs excited by longer wavelength. Several red RSFPs, including rsCherry, rsCherryRev, and rsTagRFP, have been used as PICAs for super-resolution FI under Vis light.⁸⁹ More recently, three novel RSFPs derived from FusionRed (a red-emitting RSFP) enable reversible switching upon the irradiation of green light (510 nm) and orange light (590 nm).⁹⁰ These rsFusionReds exhibited outstanding fatigue resistance (> 1000 photoswitching cycles) and a rapid photoswitching rate ($58.3 \pm 0.4 \text{ ms}$), facilitating nanoscale live-cell imaging. Their operation at wavelengths beyond 500 nm also enabled the RESOLFT implementation with mitigated DNA damage and photodamage. Furthermore, the advantages of NIR light have directed increasing attention toward NIR fluorescence proteins for light-controlled cellular and *in vivo* imaging.⁹¹ Advancements in the photoacoustic properties of NIR-induced photoswitchable proteins have also motivated studies exploring NIR RSFPs for high-resolution PAI.^{92,93}

Despite progress in developing biologically safe and genetically encodable NIR RSFPs, their broad bioimaging adoption remains constrained by some shortcomings, such as low brightness, poor photostability, insufficient fatigue resistance, and complex genetic circuitry.^{48,89} Future efforts should address

these challenges to enhance the overall photoswitching performance of NIR RSFPs.

In brief, these representative classical photoswitchable molecules, including Azo, SP, DAE, and RSFPs, have demonstrated superior potential for development into PICAs. Nonetheless, their widespread adoption in bioimaging confronts considerable challenges. The photoisomerization of most photoswitches is driven by short-wavelength light, which falls outside the optical window of biological tissues. To overcome this limitation, further molecular modifications are necessary to red-shift their absorption toward longer wavelength. Strategies for elongating the π -conjugated structure or introducing appropriate fluorophores are viable. Alternatively, integrating photoswitchable molecules with upconversion materials enables excitation by NIR light. Since most photoswitchable molecules are inherently hydrophobic, incorporating hydrophilic moieties can improve biocompatibility.^{28,94} Additionally, coupling photoswitchable molecules with other functional materials is a promising route to tailor specific physicochemical properties.^{95,96}

3. Optimization strategies on photoswitchable imaging contrast agents

Despite considerable improvements in photoswitchable molecule design, some challenges continue to impede the practical application of PICAs in bioimaging, particularly for *in vivo* use. These include: (1) poor stability in complex biological environments hinders prolonged imaging; (2) hydrophobicity-induced aggregation impairs photoswitching sensitivity and raises safety concerns; (3) dependence on short-wavelength excitation limits their effectiveness in whole-organism and deep-tissue imaging; (4) inadequate emission performance; and (5) low photoisomerization efficiency diminishes imaging resolution. Molecular design strategies, such as functional group modification, structural adjustment, and chemical conjugation, have been employed to improve the stability, water solubility, and optical performance of photoswitchable molecules. Alternatively, intermolecular assembly *via* hydrophilic-hydrophobic interactions, electrostatic forces, host-guest complexes, or π - π stacking provides a complementary method to address these challenges.

In this section, the challenges are categorized, and an exploration of how molecular design and assembly strategies can address them to optimize the properties of PICAs for bioimaging applications is provided (Fig. 8).

3.1. Strengthening the stability

The stability of PICAs, encompassing chemical stability, thermal stability, photostability, and fatigue resistance, is essential for their application in bioimaging. However, these properties are often compromised under harsh external conditions. First, the chemical stability of PICAs is particularly challenged by complex biological milieus. For example, enzyme-mediated



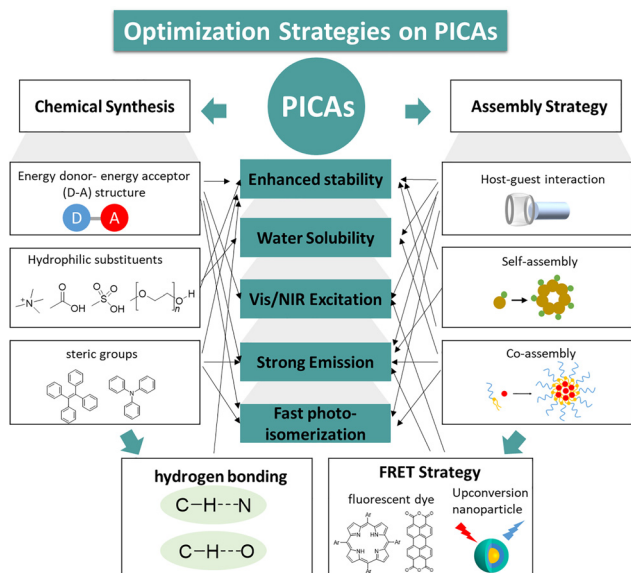


Fig. 8 Schematic illustration of optimization strategies on PICAs aimed at improving stability, water solubility, Vis/NIR excitation, emission properties, and photoisomerization efficiency.

modification or reduction can occur in Azo molecules, which deactivates their photoswitching ability.^{97,98} Rapid thermal relaxation also favors the transition of PICAs back to the ground state, while photobleaching resulting from continuous irradiation hampers long-term stable imaging. Furthermore, repeated photoswitching cycles necessitates PICAs with robust fatigue resistance.

Significant progress has been made in improving the chemical stability, photostability, and thermal half-life of PICAs to facilitate high-quality photoswitching imaging *via* structural modification.²⁰ Chemical modification with certain groups on photoswitchable molecule itself is a viable strategy.^{99,100} First, the introduction of hydrogen bonds with specific functional groups (*e.g.* hydroxyl, carboxyl, and nitrogen heterocycle) can markedly enhance the thermal/photo-stability. For example, a series of Azo derivatives with phenyl rings substituted by pyrazole rings at different positions, significantly extends the thermal half-life from hours to years (Fig. 9A).¹⁰¹ These Azo derivatives containing pyrazole rings produce intramolecular C-H...N hydrogen bonds between the pyrazole rings further stabilizes the *cis* form. Second, introducing strong electron-withdrawing groups lowers the molecular oxidation potential and inhibits the photo-oxidation reaction, leading to the improved chemical stability of photoswitching molecules. Taking DAE derivatives as an example, the use of perfluorocyclopentenes as a vinyl bridge moiety with strong electron-withdrawing and steric characteristics markedly enhances their thermal stability and photobleaching resistance. While non-fluorinated DAEs only have 20–30 photoswitching cycles, fluorinated DAEs possess high fatigue resistance with hundreds to thousands of photoswitching cycles.¹⁰² In addition, perfluorocyclopentene-contained DAEs exhibit thermal half-lives of their ring-closed isomers reaching several years.¹⁰³

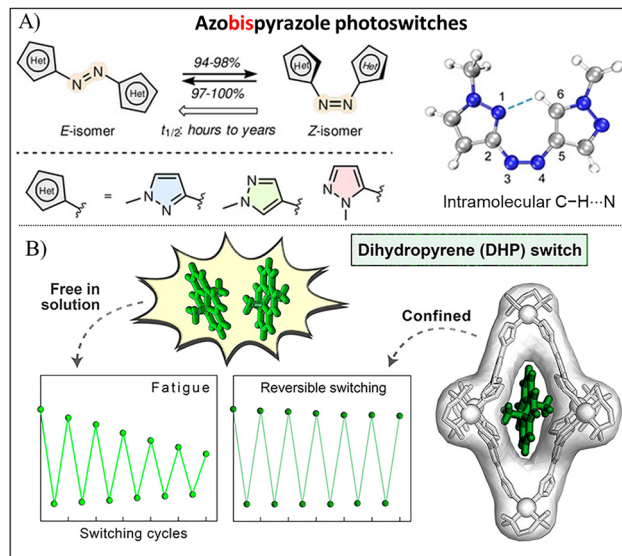


Fig. 9 (A) Photoswitchable isomerization of dipyrazole-substituted Azo derivatives with an intramolecular C-H...N hydrogen bond between the two pyrazole rings. Reproduced with permission.¹⁰¹ Copyright 2021, Wiley-VCH. (B) Enhanced fatigue resistance of DHP confined within a Pd^{II}₆L₄ coordination cage. Reproduced with permission.¹⁹ Copyright 2020, American Chemical Society.

Supramolecular assembly between photoswitchable molecules and biomaterials or synthetic materials offers a straightforward approach to improve their chemical stability and fatigue resistance while preserving their intrinsic photophysical properties. For example, peptide/protein-based assemblies can provide a confined environment that protects Azo compounds from enzymatic damage and inhibit *cis-trans* thermal relaxation, thereby enhancing fluorescence emission.^{37,104} Assembly strategies have been demonstrated to confine photoswitchable molecules within a limited space, which prevents molecular aggregation and improves fatigue resistance.²² Furthermore, molecular confinement successfully avoids the quenching of photoswitchable molecules. For example, DHP was encapsulated within a Pd^{II}₆L₄ coordination cage, significantly improving its fatigue resistance (Fig. 9B).¹⁹ When exposed to Vis light irradiation, DHP underwent isomerization to form the CPD isomer, which was stabilized in a conformation similar to its initial state due to the molecular confinement effect. After UV light irradiation or thermal relaxation in the dark, the CPD isomer converted back to DHP, suggesting reversible photoswitching within the coordination cage. Following encapsulation in the cage, DHP maintained a rapid photoswitching response after 10 switching cycles, with its attenuation rate decreasing from 28% to 8%, indicating enhanced fatigue resistance.

3.2. Improving the water solubility

The inherent hydrophobicity of many photoswitchable molecules restricts their applications in imaging within cells and organisms. Hydrophobicity-induced molecular aggregation can



impair the photoswitching properties and increase potential safety risks. A common countermeasure is to functionalize photoswitchable molecules with hydrophilic groups, such as sulfonate groups, carboxylic acid groups, quaternary amine groups, polyethylene glycol, and water-soluble cavitands.^{4,10,20,105–111} In one study, DAE was modified with a short tetra(ethylene glycol) methyl ether (mTEG) group.¹⁰ The incorporation of the neutral mTEG group significantly improved the water solubility of the PICAs, facilitating cellular uptake and biosafety. Additionally, this modification increased the on-off contrast in FI under reversible photoswitching conditions. The introduction of quaternary ammonium salts onto the DAE structure also increases its hydrophilicity.^{4,112–114} Naren *et al.* synthesized a water-soluble DAE derivative with two quaternary amine substituents for cellular FI (Fig. 10A).⁴ Separately, Hell *et al.* engineered two novel hydrophilic PICAs by functionalizing DAE derivatives with either four or eight carboxylic acid groups (Fig. 10B).²⁰ Both PICAs exhibited high aqueous solubility, robust photostability, and reversible photoswitching behavior, rendering them effective for labeling cellular structures. While hydrophilic modification can reduce the hydrophobicity of photoswitchable molecules, entirely inhibiting molecular aggregation in aqueous media remains difficult. To overcome this limitation, Kim *et al.* designed a water-soluble DAE-based dendritic nanocluster covalently linked to Cy5.5.¹⁰ The dendritic architecture effectively mitigated molecular aggregation while showing reversibly photoswitchable NIR fluorescence (Fig. 10C). Using confocal microscopy, NIR fluorescence signal was detected in living HeLa cells and zebrafish following the internalization of dendritic nanocluster. Irradiation with 365 nm UV light reverted DAE to the ring-closed state (absorption band within 450–750 nm), leading to efficient FRET quenching of the Cy5.5 NIR fluorescence at 700 nm (excitation at 630 nm). Subsequent exposure to 532 nm visible light restored the NIR fluorescent signal to its initial level.

However, even with hydrophilic functionalization, some photoswitchable molecules still show poor aqueous solubility, thereby impeding their applications in bioimaging. This limitation can be addressed by assembling photoswitches with amphiphilic molecules through non-covalent interactions.^{12,63} Qi *et al.* used a nanoprecipitation method to develop a photoswitchable nanomicelle based on amphiphilic lipid-PEG2000 (Fig. 10D).⁶³ During assembly process, DAE was loaded into the lipid layer through hydrophobic interactions to form the core, while the hydrophilic PEG in the outer layer supported colloidal stability. This photoswitchable nanomicelle allowed dual-mode FI and PAI to guide tumor resection in mice. In addition to lipid-based systems, amphiphilic macrocycles have been used to assemble hydrophobic photoswitchable molecules.^{115–117} Notably, supramolecular host-guest assembly with amphiphilic macrocyclic compounds (having hydrophilic surface and hydrophobic cavity), including cucurbit[7]uril, sulfonatocalix[4]-arene, and cyclodextrin, or with large-cavity coordination cages, can markedly improve the hydrophilicity of photoswitchable molecules.^{118–120} For instance, hydrophobic SP 6, which is poorly soluble in aqueous media, became highly water-soluble

upon forming the host-guest complex **6**⊂**4** with flexible coordination cage **4** (Fig. 10E).¹²¹

3.3. Extending the activation wavelength to the visible/near-infrared region

Currently, most photoswitchable molecules require UV light irradiation for photoisomerization. This short-wavelength-dependent activation imposes some critical challenges for bioimaging: (1) limited tissue penetration – UV light is extensively absorbed by biological tissues as a result of the high extinction coefficient of biomolecules (such as nucleic acids and proteins) within the UV spectrum region. Therefore, activation is confined to superficial cells or tissues, accompanied by rapid energy dissipation over micrometer-scale depth. (2) Phototoxicity – long-term UV light exposure induces oxidative stress *via* reactive oxygen species (ROS) production, along with direct DNA damage (*e.g.*, thymine dimer formation). These side effects compromise cell viability during live imaging, generally manifesting as reduced proliferation or apoptosis of target cells. (3) High autofluorescence background – UV excitation triggers intrinsic autofluorescence from endogenous components, such as NADH (450–500 nm) and collagen (305–450 nm). This interference degrades the signal-to-noise ratio (SNR), necessitating higher excitation intensities or complex spectral filtering to distinguish target signals. Therefore, it is crucial to develop photoswitchable molecules responsive to Vis/NIR light for *in vivo* bioimaging. Current approaches to red-shift the activation wavelength of PICAs can be categorized into four main aspects: chemical modification of electron-donating or electron-accepting groups; design of negatively photoswitchable molecules; integration with upconversion nanoparticles (UCNPs); and utilization of two-photon excitation technology.^{122–125}

Constructing a donor-acceptor (D-A) structure by chemical modification of electron-donating or electron-accepting groups on the skeleton of photoswitchable molecules offers a promising approach to red-shift the absorption wavelength.^{99,126–129} Electron-donating moieties (such as aromatic groups,¹⁰¹ heterocyclic groups,^{130,131} and ammonium ions^{21,132}) can extend the π -conjugation system of photoswitchable molecules and lower the HOMO-LUMO energy gap. Woolley *et al.* is dedicated to modifying Azo derivatives with electron-donating moieties to red-shift absorption wavelength.^{133–135} After substitution with *ortho* or *meta*-methoxy groups, *p*-amino-substituted Azo derivatives were converted to azonium ions, resulting in a pronounced red shift in absorption toward the NIR and far-red spectral regions (>700 nm) for *trans-cis* photoisomerization (Fig. 11A).²¹ Moreover, the formation of intramolecular H-bond between the *ortho*-methoxy groups and the azonium protons further stabilized the azonium ions. Upon red light irradiation, *trans-cis* photoisomerization disrupted intramolecular H-bonds, which generated neutral *cis* isomer and slowed thermal relaxation. This azonium ion-based PICA provides a promising alternative for NIR-excited photoswitchable imaging *in vivo*.

Negatively photoswitchable molecules differ fundamentally from conventional photoswitches: they undergo photoisomerization



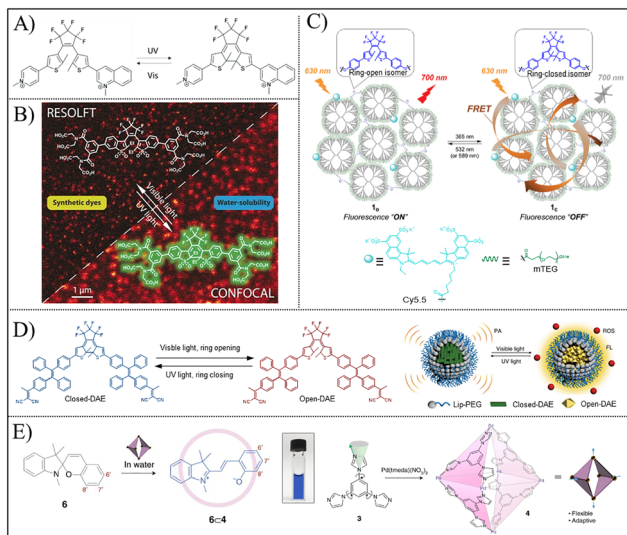


Fig. 10 (A) Structure of a DAE derivative functionalized with two quaternary amine substituents. Reproduced with permission.⁴ Copyright 2008, American Chemical Society. (B) Structure and photoisomerization of a hydrophilic DAE derivative engineered with eight carboxylic acid groups. Reproduced with permission.²⁰ Copyright 2016, Wiley-VCH. (C) Structure of a water-soluble DAE-based dendritic nanocluster. Reproduced with permission.¹⁰ Copyright 2022, Wiley-VCH. (D) Assembly of hydrophilic DAE-based nanoparticles that exhibit photoswitchable fluorescence emission and photoacoustic signal. Reproduced with permission.⁶³ Copyright 2018, Springer Nature. (E) Enhanced solubility of hydrophobic SP (**6**) via host-guest assembly in a flexible coordination cage (**4**). Reproduced with permission.¹²¹ Copyright 2018, Springer Nature.

upon Vis/NIR light irradiation and return to the ground state through thermal relaxation in the dark or upon short-wavelength light irradiation.^{45,136,137} DASA is a prominent class of such negative photochromic molecules.^{138,139} The initial DASA adopts an elongated triene structure with strong color and hydrophobicity, while Vis light (550 nm) irradiation converts it back to a colorless, hydrophilic, cyclized cyclopentenone isomer. Based on the characteristics of DASA, Senthilkumar *et al.* developed a negative photoswitchable polymer nanoparticle as a delivery vehicle by conjugating DASA to the terminal ends of a poly(*p*-phenylene vinylene) structure (Fig. 11B).¹⁴⁰ Upon Vis light irradiation, DASA converted to a ring-closed cyclopentenone isomer, inducing the disassembly of the nanoparticles and facilitating drug release. Structural modification of traditional photoswitchable molecules represents a general approach for generating negatively photoswitchable variants.^{141,142} Irie *et al.* introduced an oxidized thiophene ring at the 2-position of the thiophene ring in DAE to extend the π -conjugation length of the ring-opened DAE.¹⁴³ The resultant negatively photoswitchable DAE derivative underwent photocyclization upon irradiation with Vis light ($\lambda > 430$ nm) and then a ring-opening reaction under UV light irradiation, with a red-shifted absorption profile.

Another strategy is integrating UCNP capable of converting Vis/NIR light into shorter-wavelength emission, which spectrally overlaps with the absorption required for photoswitchable molecules.¹⁴⁴ UCNP are primarily composed of inorganic materials represented by rare-earth fluorides and oxides.

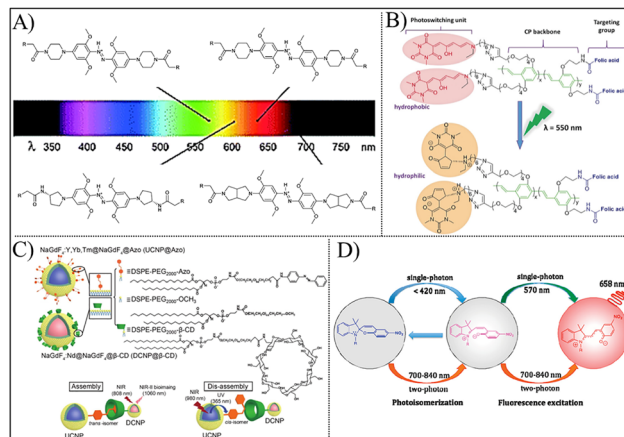


Fig. 11 (A) Structure of Azo derivatives (azonium ions) bearing *ortho* or *meta*-methoxy and *para*-amino substituents on the phenyl rings. Reproduced with permission.²¹ Copyright 2015, Royal Society of Chemistry. (B) Construction of a DASA-conjugated polymer and its photoswitching behavior upon irradiation ($\lambda = 550$ nm). Reproduced with permission.¹⁴⁰ Copyright 2018, Wiley-VCH. (C) Schematic illustration of NIR laser-regulated assembly and disassembly of a NIR-II Azo-based nanoprobe. Reproduced with permission.¹⁴⁹ Copyright 2018, Wiley-VCH. (D) Two-photon-induced SP \rightarrow MC photoswitching behavior and fluorescence emission. Reproduced with permission.¹⁵³ Copyright 2011, American Chemical Society.

Characterized by a large Stokes shift, UCNP absorb NIR light (typically at 808 nm and 980 nm) and upconvert this energy into UV or Vis emission.^{145–148} To broaden the window of PICAs for *in vivo* bioimaging, it is desirable to further red-shift the excitation wavelength into the second NIR region (NIR-II, 1000–1700 nm), because NIR-II light provides deeper tissue penetration capacity and reduced background interference. Zhang *et al.* designed a novel photoswitchable UCPN (ps-UCNP) system for NIR-II-excited photoisomerization of Azo (Fig. 11C).¹⁴⁹ Specifically, *trans*-Azo-modified UCNP were assembled with cyclodextrin-modified nanoparticles through host-guest interactions. Upon 980 nm light irradiation, the UCNP emitted UV light to drive the isomerization of the *trans*-Azo to its *cis*-isomer, leading to nanoparticle disassembly.

In addition to chemical modification or structural adjustment that red-shifts the absorption of photoswitchable molecules into the Vis/NIR region, two-photon excitation offers an efficient route to operate photoswitching with NIR light.¹⁵⁰ Two-photon excitation is a nonlinear process in which molecules simultaneously absorb two photons to reach an excited state with higher energy.¹⁵¹ Generally, two-photon excitation occurs in the NIR range, while emission falls in the Vis range.¹⁵² There is growing interest in developing PICAs that can be activated using two-photon Vis/NIR light irradiation.^{153–155} Zhu *et al.* proposed that two-photon excitation using NIR light facilitated the photoisomerization of the SP to the MC isomer, accompanied by red fluorescence emission, comparable to that achieved through single-photon excitation with UV light (Fig. 11D).¹⁵³ In another study, Azo was covalently attached to a two-photon-absorbing unit, specifically a triarylamine antenna, which enabled NIR light-driven (750 nm) isomerization from the *cis* isomer to the *trans* form.¹⁵⁶



3.4. Enhancing the emission performance

High emission intensity at long-wavelength region of PICAs is crucial for effective bioimaging. PICAs can benefit from flexible molecular design strategies based on structure–property relationships or host-guest assembly.

Introducing electron-donating groups to extend the π -conjugation system has been shown to red-shift the absorption, while enhancing emission intensity or broadening emission wavelength range. Irie *et al.* synthesized a series of DAE derivatives and found that, the ring-closed derivatives with phenyl or thiophene substituents displayed high fluorescence emission (Fig. 12A).¹⁵⁷ These substituents function as strong electron donor, while the central benzobis(thiadiazole) bridge acts as efficient electron acceptor, resulting in the formation of a D- π -A structure along with an intramolecular charge transfer (ICT) process. During ICT process, electron transfers from the donor (D) to the acceptor (A) to activate a polarized excited state and suppress non-radiative decay. The nonradiative decay rate constant (k_{nr}) of substituted DAE in the ring-closed state dramatically reduced from $6.5 \times 10^8 \text{ s}^{-1}$ for the non-substituted structure to $4.3 \times 10^7 \text{ s}^{-1}$. Eventually, the substitution strategy significantly improved the fluorescence quantum yield (Φ_F) of ring-closed DAE from ~ 0.2 to ~ 0.9 . In addition to improved fluorescence quantum yield, extending the π -conjugation system has been demonstrated to endow photoswitchable molecules with dual-color FI capability.¹⁵⁸

Most small-molecule photoswitches are hydrophobic, thereby tending to aggregate in aqueous media to cause ACQ that diminishes both photoswitching and emission capacities.

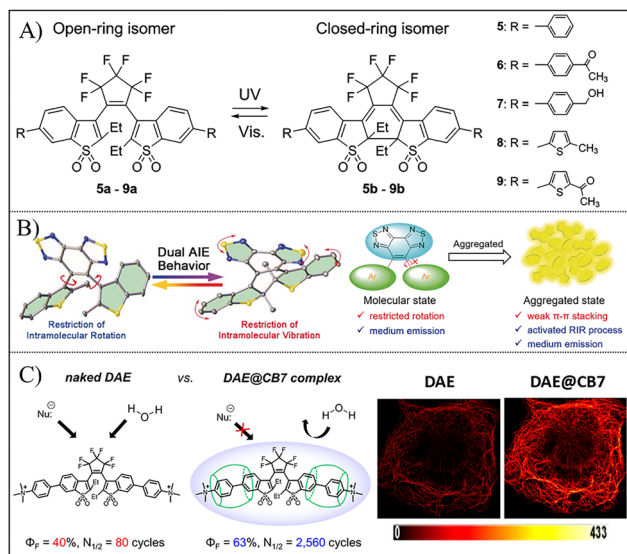


Fig. 12 (A) Structures of DAE derivatives functionalized with various aromatic groups. Reproduced with permission.¹⁵⁷ Copyright 2011, American Chemical Society. (B) Schematic illustration of a sterically hindered DAE derivative that exhibits AIE behavior by preventing π - π stacking in the aggregated state. Reproduced with permission.¹⁶³ Copyright 2020, Wiley-VCH. (C) Enhanced emission of DAE@CB7 through host-guest assembly strategy. Reproduced with permission.²² Copyright 2022, American Chemical Society.

Introducing hydrophilic groups²⁰ and regulating steric hindrance (*e.g.*, rigid planar structures and AIE motifs) are effective approaches to prevent aggregation.^{159–162} Zhu *et al.* synthesized a novel DAE derivative through functionalizing it with two bulky benzothiophene groups, which showed characteristic AIE behavior, including restriction of intramolecular vibration (RIV) and rotation (RIR), as well as intermolecular stacking (Fig. 12B).¹⁶³ The steric effect generated a rigid aggregation structure, decreasing the nonradiative decay rate to enhance the fluorescence emission of ring-opened DAE. Upon UV light irradiation, the AIE-type DAE underwent a cyclization reaction to cause fluorescent quenching owing to reduced steric hindrance and increased π - π stacking, which eventually resulted in negligible emission intensity caused by ACQ. Therefore, modulating the nonradiative decay rate and aggregation state of PICAs provides a promising tool for improving the fluorescence emission performance.

Host-guest supramolecular assembly is capable to modulate the spatial confinement, environmental polarity, energy transfer, and dynamic reversibility of photoswitchable molecules, leading to an enhanced emission performance.^{22,164,165} For example, encapsulation by CB7 reduced the interaction between DAE and water molecules, inhibited non-radiative transitions, increased the Φ_F from 0.40 to 0.63, and prolonged the fluorescence lifetime (Fig. 12C).²² The nonradiative decay rate constants of DAE and DAE@CB7 were calculated to be $k_{nr} = 3.3 \times 10^8 \text{ s}^{-1}$ and $1.6 \times 10^8 \text{ s}^{-1}$, respectively, indicating the enhanced emission property.

In photoswitching imaging, FRET not only turns the on/off state of fluorescence,⁶¹ but also manipulates fluorescence properties through two distinct interaction pathways between the photoswitchable molecule and the fluorophore: (1) enhancing the emission intensity of photoswitchable molecule *via* energy transformation from fluorophore; (2) red-shifting the emission wavelength of photoswitchable molecule by activating the fluorescence of fluorophore (Fig. 13). FRET is a non-radiative energy transfer process that needs spatial coupling between the donor and acceptor.^{166,167} As shown in Fig. 13A, FRET proceeds from fluorophore (as a donor, D) to photoswitchable molecule (photoswitch, as an acceptor, A) when the emission spectra of the fluorophore overlaps with the absorption spectra of the isomerized photoswitch at off-state.¹⁶⁸ In this way, the fluorescence of donor is quenched while the emission of photoswitch is activated, generating dual color fluorescence signals. On the contrary, when the emission spectra of the photoswitch at on-state overlaps with the absorption spectra of fluorophore, FRET occurs from the photoswitch (now as donor) to the fluorophore (now as acceptor) (Fig. 13B). The second FRET process enables fluorophore activation through energy transfer from the photoswitch, thereby both enhancing the overall fluorescence intensity¹⁶⁹ and red-shifting the emission from the Vis region to the NIR region¹⁵ for improved bioimaging.

3.5. Improving the photoisomerization efficiency

Photoisomerization is a defining characteristic of PICAs distinctive from conventional contrast agents. Photoisomerization efficiency serves as a critical criterion for assessing the potential of PICAs in photoswitchable bioimaging. Photoswitchable molecules proceed a reversible isomerization process wherein



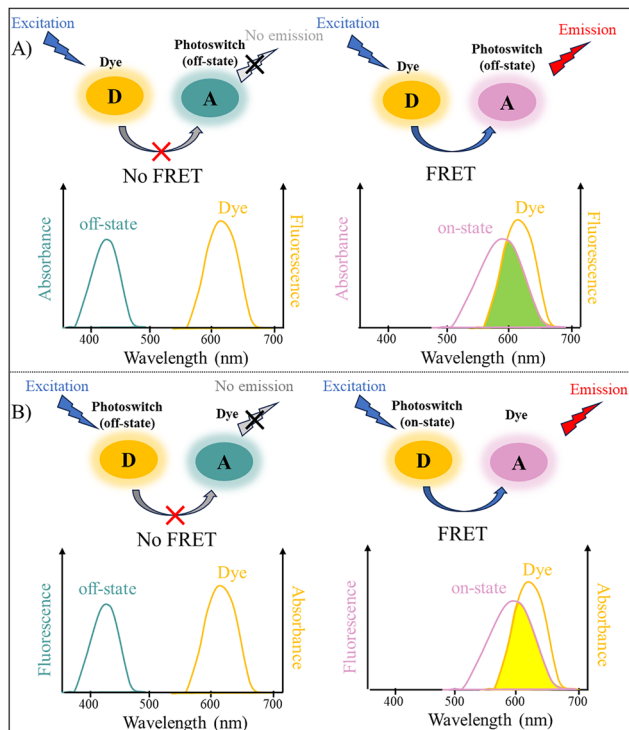


Fig. 13 Two FRET pathways within a photoswitch-fluorophore pair enabling fluorescence manipulation through: (A) energy transfer from the dye (donor) to the on-state photoswitch (acceptor), and (B) the reverse transfer from the on-state photoswitch (donor) to the dye (acceptor).

they populate an excited singlet state upon light irradiation and subsequently return to the ground state. However, during this sequence, competing pathways, such as non-radiative decay and intermolecular energy/charge transfer, inevitably dampen photoisomerization efficiency. In addition, poor thermal stability of the two ground-state isomers further attenuates the apparent quantum yield. Molecular design optimization and supramolecular assembly have been shown to overcome these limitations, thereby strengthening the practical viability of photoswitching molecules as PICAs for bioimaging.

A promising method is introducing electron-donating groups (such as amino and methoxy) or electron-accepting groups (such as nitro and cyano), to promote the D–A character of photoswitchable molecules and reduce excitation energy required for photoisomerization. Conventional Azo compounds typically display insufficient *trans* → *cis* photoisomerization efficiency, with a low quantum yield of approximately 10–30%.³¹ Modifying a pair of D–A substituents (such as methoxy/cyano and amino/nitro) on both sides of the Azo markedly decreased the excitation energy and raised the photoisomerization quantum yield to 35% (Fig. 14A).²³ In addition, extending the π -conjugated system with electron-rich groups enlarges the π -electron delocalization range, which can also reduce the excitation energy to improve the photoisomerization efficiency.¹⁷⁰ For example, replacing one phenyl ring with thiophene generated a lone-pair (S atom) ··· π (Azo) interaction between the S atom and the Azo system, strengthening ICT from the thiophene donor to

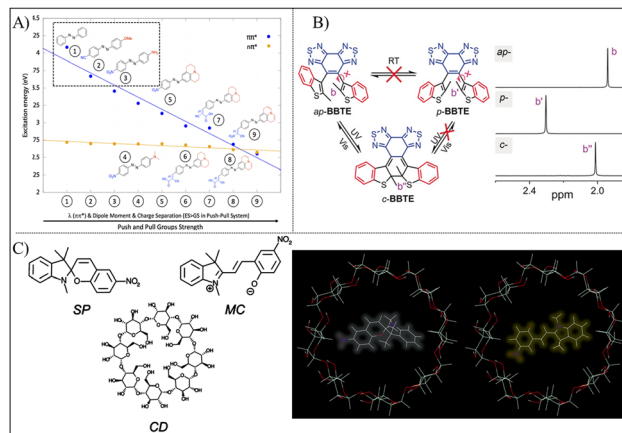


Fig. 14 (A) Excitation energy of Azo derivatives functionalized with various electron donor groups. Reproduced with permission.²³ Copyright 2020, American Chemical Society. (B) Photoswitching behavior of anti-parallel DAE functionalized with two benzothiophene substituents. Reproduced with permission.¹⁶³ Copyright 2020, Wiley-VCH. (C) Stable host-guest interaction between CD and SP/MC. Reproduced with permission.¹⁷⁶ Copyright 2015, American Chemical Society.

the phenyl acceptor. This change raised the quantum yield to 40% for *trans* → *cis* and 65% for *cis* → *trans* photoisomerization.¹⁷¹

Incorporating steric groups into the skeleton of photoswitchable molecules possess a twist conformation and increases steric hindrance, thereby enhancing photoisomerization efficiency.¹⁷² For instance, DAE generally has a photoisomerization (ring-opened → ring-closed) quantum yield below 50%, owing to the coexistence of parallel (photoinactive) and antiparallel (photoactive) conformations. After introducing large steric-hindrance groups (*e.g.*, benzothiophene) on both sides of the DAE skeleton, the obtained twisted structure constrained non-radiative pathways such as intramolecular rotation and vibration, which in turn compelled the molecule to preferentially adopt the anti-parallel conformation (Fig. 14B).¹⁶³ This symmetric benzothiophene-based DAE promoted cyclization reaction, raising the photoisomerization quantum yield to 88% for ring-closed state and 90% for ring-opened state.

Supramolecular assembly strategy *via* non-covalent interactions (including hydrophobicity, hydrogen bonding, π - π stacking, and host-guest recognition) between photoswitchable molecules and large-cavity host molecules, tailors the aggregation state of photoswitchable molecules, thus leading to enhanced photoisomerization efficiency.^{173–175} For example, hydrophobic SP could insert into the cavity of cyclodextrin (CD) by host-guest interaction, resulting in a stable assembly complex (Fig. 14C).¹⁷⁶ Due to the stabilization of MC state by CD, the half-life of MC state prolonged and the emission intensity was enhanced.

4. PICAs for bioimaging systems

Bioimaging technologies are crucial for the early diagnosis and treatment of diseases because they enable sensitive and precise detection of changes in biological and pathological



signals.^{177–179} Bioimaging also contribute fundamentally to understanding the structure of living organisms and elucidating their underlying physiological functions.¹⁸⁰ However, achieving high accuracy and resolution in biological imaging remains challenging owing to the complexity of environment. PICAs, with their dynamic signal modulation and controllable performance, can reduce background interference and improve the quality of photo-controllable bioimaging. Recent advances in fluorescence microscopy have driven the widespread use of PICAs in fluorescence sensing and imaging, particularly in super-resolution imaging.^{81,163,181} Additionally, PICAs have expanded into other imaging modalities, such as PI,¹⁶⁴ PAI,⁶³ MRI,¹⁸² and RI (Table 1).¹⁴

4.1. Fluorescence imaging

FI is the most widely used imaging modality owing to its non-invasive nature, high spatiotemporal resolution, operational simplicity, and capacity for real-time monitoring. FI has demonstrated a remarkable success in diverse domains, including the elucidation of cellular pathophysiological processes, early-stage disease diagnosis, and real-time intraoperative imaging guidance.^{183–185} Compared to conventional fluorescent probes, PICAs offer several key advantages: (1) reversible on-off switching enables enhanced imaging resolution by breaking the optical diffraction limit; (2) multiple switching cycles statistically eliminate random noise for improved SNR; (3) selective illumination of regions of interest minimizes nonspecific background signals; and (4) on-demand light activation mitigates photobleaching and phototoxicity, permitting long-term and dynamic tracking. Consequently, PICAs have been employed as effective alternatives to conventional fluorescent probes in super-resolution imaging, biosensing, dual/multi-color imaging, and multiplexed imaging.

4.1.1. Super-resolution imaging. Super-resolution FI surpasses the diffraction-limited resolution of conventional microscopy, allowing precise delineation of subcellular structures.^{186–188} Nonlinear fluorescence response to the intensity of excitation light is the primary technique for super-resolution imaging, typically originating from the on- and off-transitions of fluorescence or the saturation of fluorescence.¹⁸⁹ Accordingly, photoswitchable molecules with high on-off contrast and fatigue resistance are the optimal candidate for super-resolution imaging.^{190,191} To date, small molecule-based PICAs and RSFPs have been widely utilized in single-molecule localization microscopy (SMLM) and RESOLFT nanoscopy. In SMLM, precise control of illumination conditions enables sparse activation of photoswitchable molecules within the sample, ensuring spatially isolated emission spots for super-resolution image reconstruction. Meanwhile, the temporally sequential activation of photoswitchable molecules circumvents signal overlap between adjacent emitters, thereby decoding unresolvable information within the diffraction-limited region. In contrast, RESOLFT relies on high-intensity depletion light to saturate photoswitchable molecules, restricting fluorescence emission to a nanoscale central region. Photoswitchable materials functioning as PICAs in super-resolution fluorescence microscopy require high photostability, fluorescence quantum yields (Φ_F),

on/off ratios and rapid switching speeds.^{5,16,181–183} To date, PICAs used in super-resolution FI can be classified into three categories: RSFPs, organic small molecules, and photoswitchable molecule-based assemblies.^{5,69,192–195}

RSFPs are suitable for RESOLFT-based dynamic imaging owing to their superior live-cell compatibility and switching cyclability, as well as lower light dose requirement for saturation transition. Testa's group developed a multi-sheet RESOLFT microscope to overcome the limitations of conventional light-sheet fluorescence microscopy (LSFM) in subcellular imaging.¹⁹⁶ The technology employs wide-field illumination to activate all RSFPs, followed by sinusoidal interference-patterned depletion light pulse to confine on-state RSFPs to a tightly defined sheet. This allows emission sheets with a thickness far below the diffraction limit (< 100 nm)—10 times thinner than conventional excitation sheets. By parallelizing the imprinting of multiple sheets and rapid readout, the technique achieved fast volumetric imaging at 1–2 Hz with a spatial resolution below 250 nm. It required only 10–20 switching cycles and is compatible with a wide range of RSFPs. Using a classical RSFP, rsEGFP (N205S), multi-sheet RESOLFT succeeded in high-resolution live-cell 3D tracking of subcellular dynamics, including actin dynamics, cell division, and virus-like particle movement. Future research can focus on incorporating other modulation patterns and image rotators to further improve spatial resolution and expanding its applications to 3D culture and multicellular organisms.

The performance of RESOLFT nanoscopy relies on multiple properties of RSFPs, including switching rate, photostability, brightness, and residual fluorescence in their off state. It is challenging to achieve optimal performance due to inherent trade-offs between these properties. For instance, enhanced brightness often compromises switching rate, and a lower off-switching rate increases the signal-to-background ratio (SBR). To reach a balance, Sun *et al.* developed a novel monomeric RSFP, namely GMars-Q, for long-term RESOLFT imaging in living cells (Fig. 15A).¹⁹⁷ GMars-Q was activated by 405 nm light and switched off by 488 nm light, respectively. Significantly, the off-state residual fluorescence of GMars-Q was 6–9-fold lower than that of rsEGFP2 and N205S, two classical reversibly photochromic fluorescent proteins, enabling significantly reduced SNR and improved imaging contrast. Moreover, GMars-Q exhibited higher fatigue resistance and SBR compared with N205S. These improvements may stem from an unusual biphasic photobleaching feature. When tagged with the KDEL targeting sequence, GMars-Q achieved stable live-cell observation of endoplasmic reticulum tubulation for 2 minutes, which was an order of magnitude longer than that achieved by rsEGFP (N205S) at equivalent resolution. This study provides a methodological advancement for long-term super-resolution imaging in living cells. Nevertheless, green-emitting RSFPs generally need violet-blue light for photoswitching, which potentially causes scattering effect and phototoxicity. Red-shifted RSFPs with strong fatigue resistance and rapid switching rate are needed. Testa *et al.* engineered three green photo-activated red-emitting rsFusionRed variants (rsFusionRed1, rsFusionRed2, and rsFusionRed3) *via* site-specific amino acid mutagenesis.⁹⁰





Table 1 Summary and comparison of PICAs in various imaging modalities

Photoswitchable imaging modality	Photoswitching moiety		Photoswitching light (nm)	Half-life of isomer ($t_{1/2}$)	Photoswitching speed	Photoswitching quantum yield	Photoswitching ratio	Photoswitching cycle	Imaging object	Ref.
	Molecule type	Original name								
FI	RSFP	GMars-Q	405/488	120 min of off-state	2 ms for on-switching/24 ms for off-switching	—	—	>50	Various cell structures and organelles with fwhm of ~80 nm	197
	RSFP	rsFusionRed variants	510/590	>20–40 min	—	—	—	>1000	F-Tractin of U2OS cells with fwhm of <80 nm	90
	SP	DSA-2SP	365 or 302/561	>24 h	—	—	—	>10	Cylindrical micelles formed with fwhm of ~50 nm	195
	DAE	PMI-3DTE	302/532	—	2–5 s for ring-closing/2 s for ring-opening	0.459 of closed-state/0.007 of open-state	3026	>10 000	Polymer film with fwhm of 36 nm	5
	DAE	DAE@CB7	355/485	—	—	0.35 of closed-state/ 9.2×10^{-3} of open-state	—	2560	Microtubules of Cos7 cells with fwhm of 70–90 nm	22
	SP	SP-PCL	<420/>420	—	1 s for ring opening	—	—	>6	HeLa cells with sub-50 nm resolution	198
	SP	SP-CE	300–400/Vis	—	—	—	—	—	Li^+ detection in HeLa cells and zebrafish	106
	SP	Ly-NT-SP	UV/Vis	—	—	—	—	>3	SO_2 detection in lysosome during heat shock	210
	SP	NpG@HSA	365/550	—	—	—	—	>5	Endogenous β -Gal detection in SKOV-3 cells and WI-38 cells with fwhm of <80 nm	16
	SP	SPMA	302/525	—	—	—	—	>5	A549 cells	224
PI	DAE	DBTEO	365/>420	—	—	—	—	>4	RAW264.7 cells with dual-color fluorescence and fwhm of 64 nm	12
	RSFP	BOSA-SP	365/524	—	—	—	—	>5	HeLa cells with three-color fluorescence and fwhm of 31 nm	225
	DAE	DTE-C3 = CB[8]@Cy5	365/>600	—	—	—	—	>10	A549 cells	15
	DAE	DTE-TP/CB[8]	365/>600	—	—	—	—	>6	A549 cells	164
	RSFP	RpBppP1	730–790/630–690	~210 s	1.56 s^{-1}	—	~4.3	>10	U87 tumor in a mouse kidney or liver with 34 times contrast enhancement	92
	RSFP	DrBppP-PCM	630–690/730–790	—	0.54 s^{-1}	—	—	>5	Brain, kidney, and liver tumors in mice	93
	RSFP	BppP1	620–680/740–800	—	—	—	—	>7	4T1 tumor in mice	255
	DAE	DTE-TPECM	365/610	—	—	—	—	>10	4T1 tumor in mice	63
	Azo	wsT-AzHCy	—	<10 ps	—	—	—	—	Liver, kidney and spleen in mice	17
	Azo	UCNPs@PAA-Azo	365/450	—	—	—	—	—	Nude mouse	260
MRI	SP	Gd-SPDO3A	365/Vis	—	—	—	—	—	—	269
	SP	spitopyran-DO3A-Gd	532/—	—	200–300 ms for ring-closing	—	—	>13	—	271
Azo	Azopyridine functionalized Ni-porphyrin	500/435	—	—	—	—	>10 000	—	18	

Table 1 (continued)

Photoswitchable imaging modality	Photoswitching moiety		Photoswitching light (nm)	Half-life of isomer ($t_{1/2}$)	Photoswitching speed	Photoswitching quantum yield	Photoswitching ratio	Photoswitching cycle	Imaging object	Ref.
	Molecule type	Original name								
	Azo	Bacteriochlorin 11	720/505	—	—	—	—	> 100	—	277
	Azo	Azoimidazole functionalized Ni-II porphyrins	365/—	—	—	—	—	—	Phantom	276
RI	Azo	KMR-Az	365/—	—	—	—	—	—	—	278
	SP	CLADIO-NH-SP	UV/Vis	—	—	—	—	—	—	279
	DAE	DAE620	375/640	—	515 ns for ring-closing, 437 ns for ring-opening	—	> 25	> 8	HeLa cells	285
	DAE	DTE-alkyne derivatives	360/633	—	—	—	—	> 100	HeLa cells	14
	DAE	Asymmetric DAE polymers	365(405)/640	—	< 200 μ s for ring-closing	—	—	> 100	U2OS cells	286

Compared to switching under short light irradiation (405 or 488 nm), all variants could be efficiently switched on through 510 nm green light irradiation, exhibiting slightly slower on-switching kinetics but lower fatigue. Upon 590 nm orange light irradiation, rsFusionRed3 showed substantially faster off-switching kinetics—30-fold faster than rsTagRFP (487 ± 2 ms), and 6-fold faster than rsCherryRev1.4. Importantly, the improved off-switching kinetics of rsFusionRed3 did not compromise their fatigue resistance (> 1000 consecutive switching cycles) and thermostability (half-life, > 20–40 min). Using RESOLFT nanoscopy, rsFusionRed3 tagged with either F-Tractin or vimentin enabled live-cell visualization of actin networks or vimentin filaments with full width at half-maximum (fwhm) of ~ 80 nm under low intensity irradiation (~ 1 kW cm $^{-2}$). This demonstrates the excellent potential of rsFusionRed variants in ref-shifted non-invasive imaging *in vivo* by RESOLFT nanoscopy.

In addition to RSFPs, small-molecule photoswitches also display superior capacity for super-resolution imaging owing to their chemical tunability and photophysical diversity, which can be tailored through precise synthesis and modification based on structure–property relationships. However, as mentioned above, most small-molecule photoswitches tend to aggregate in aqueous solutions as a result of molecular stacking and hydrophobicity, causing ACQ effect to diminish their photoswitching behavior. To address this issue, Qi *et al.* coupled two SP units with an AIE moiety, 9,10-distyryl-anthrance (DSA), to develop a photoswitchable SP derivative (DSA-2SP).¹⁹⁵ To improve its water solubility, DSA-2SP was loaded into a nanomicelle assembled from amphiphilic polymer, polystyrene-*block*-poly(ethylene oxide). The nonplanar molecular structure of DSA offers large free space to prevent molecular stacking, which together with intramolecular hydrogen bonds, support reversible photoisomerization between DSA-2SP and DSA-2MC upon altering irradiation with 302 nm UV light and 561 nm Vis light. Notably, intramolecular energy transfer from the DSA donor to the MC acceptor resulted in a bright red fluorescence at 670 nm with a fwhm of ~ 49.5 nm for efficient photoswitchable super-resolution imaging (Fig. 15B). Zhu *et al.* coupled a single perylenemonoimide (PMI) molecule with three DAE units *via* covalent oxygen bridge to develop a trident photoswitchable probe (PMI-3DTE). (Fig. 15C).⁵ In this design, PMI offers high fluorescence quantum yield and photostability, while DAE functions as an efficient switch with excellent photoisomerization behavior. The three-dimensional structure of this trident photoswitchable probe efficiently prevents the ACQ effect. Upon 2–5 s of 302 nm UV light irradiation, > 95% of PMI-3DTE isomerized to its ring-closed state. This isomerization, combined with energy transfer from PMI to the ring-closed DAE, triggered rapid fluorescence quenching, with an efficiency close to 100% and a photoswitchable fluorescence quenching ratio in the photostationary state exceeding 10 000. In addition, under alternating irradiation with UV light and Vis light, the probe showed superior fatigue resistance over 10 000 switching cycles, holding great potential for super-resolution bioimaging.

Assembly strategies have been demonstrated to improve the performance of small-molecule photoswitches for super-resolution



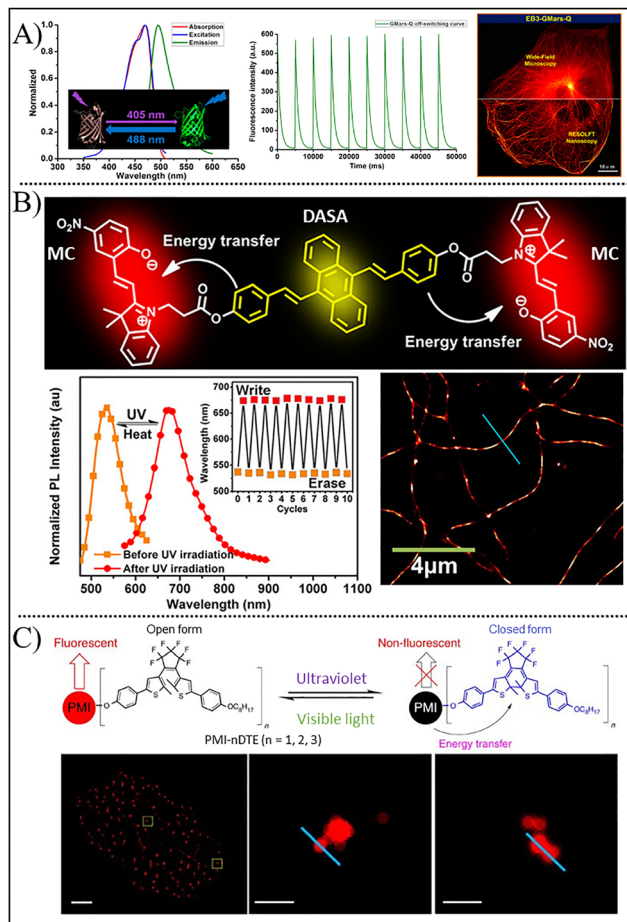


Fig. 15 (A) Reversible photoswitching behavior and super-resolution image of GMars-Q. Reproduced with permission.¹⁹⁷ Copyright 2016, American Chemical Society. (B) Intramolecular energy transfer in DSA-2MC and super-resolution imaging via photoswitching between DSA-2MC and DSA-2SP. Reproduced with permission.¹⁹⁵ Copyright 2017, American Chemical Society. (C) Photoswitching mechanism and super-resolution fluorescence images of PMI-3DTE. Reproduced with permission.⁵ Copyright 2014, Springer Nature.

imaging, such as photostability, water solubility, and photoisomerization capacity. Hell *et al.* developed a supramolecular assembly probe (DAE@CB7), in which DAE was encapsulated in cucurbit[7]uril (CB7) via the host-guest interactions, for super-resolution RESOLFT imaging.²² DAE@CB7 exhibited rapid, reversible photoisomerization upon altering irradiation of 355 nm light and 485 nm light, and increased fluorescence quantum yield of 0.63 compared to 0.40 for unimolecular DAE. Possibly ascribing to reduced aggregation in aqueous solution, DAE@CB7 demonstrated robust fatigue resistance that endured over 2560 switching cycles before half-bleaching, whereas unimolecular DAE persisted only 8 switching cycles. After conjugation with targeting antibodies, the probe succeeded in labelling cellular microtubules and photo-controllable super-resolution imaging by RESOLFT, with a 70–90 nm resolution. Dual-color fluorescence switching of PICAs offers another approach to achieve super-resolution. In addition to photoswitchable performance, the aqueous biocompatibility of assemblies is crucial for their use in bioimaging. Liu *et al.* designed a biodegradable

nanoparticle probe (SP-PCL) for super-resolution FI through the assembly of poly(ϵ -caprolactone) terminated with a SP unit via ring-opening polymerization.¹⁹⁸ Upon altering irradiation of 365 nm UV light and > 420 nm Vis light, the nanoparticle probe underwent reversible photoisomerization between the SP-PCL state and the MC state, thereby achieving fluorescence switching between green emission at 530 nm and red emission at 650 nm. Owing to the lipophilic nature of PCL, the probe was readily internalized into HeLa cells, resolving cellular structures with sub-50 nm resolution by FI.

4.1.2. Biosensing. Biosensing offers a vital strategy to visualize and analyze dynamic chemical composition, biochemical reactions, and physiological processes in cells and organisms, as well as their surroundings.^{199,200} However, conventional fluorescence biosensors lack temporal control over their sensing activity, resulting in unintended binding to encountered targets during transport process.²⁰¹ In contrast, biosensors based on photo-switchable molecules allow photo-controllable selective and rapid activation once they reach the desired regions, thereby improving detection accuracy. Furthermore, their photochromic behavior can help discriminate target signals from autofluorescence background or noise interference, while alleviating potential phototoxicity caused by continuous light exposure.

The utilization of PICAs for detecting biomarkers such as metal ions, specific gases, and proteins, has gained growing interest in exploring cellular processes and disease mechanisms.^{202–208} SP has been widely used to detect metal ions, particularly alkali metal ions, owing to the chelating ability of its colored MC isomer. In one study, SP was functionalized with aza-12-crown-4 to constructed a red-emitting probe (SP-CE) for Li⁺ detection (Fig. 16A).¹⁰⁶ Once encountering Li⁺, the crown ether of SP-CE rapidly chelated with it, resulting in the photoisomerization of SP to the colored MC state without the requirement for light irradiation. The resultant phenolate ion in MC also chelated with Li⁺ to further stabilize the complex, which significantly enhanced the red fluorescence emission to achieve accurate Li⁺ sensing (Fig. 16B). Following exposure to Vis light, MC reverted to SP and released Li⁺. This photoswitchable sensor subsequently succeeded in Li⁺ imaging in living HeLa cells and zebrafish (Fig. 16C), indicating the potential of SP for metal ion biosensing *in vivo*.

Beyond chelating with metal ions, the MC isomer contains a polarized unsaturated C=C bond that serves as a nucleophilic sites, enabling Michael addition reactions with nucleophiles such as sulfur dioxide (SO₂), hydrogen sulfide (H₂S), and sulfite ion (SO₃²⁻).^{209–212} Yin *et al.* synthesized a SP-based fluorescent probe, Ly-NT-SP, for detecting SO₂ in lysosomes (Fig. 16D).²¹⁰ Ly-NT-SP consisted of a lysosome-targeted group, a fluorophore (naphthalimide with green fluorescence), and a SP moiety. After UV light irradiation, Ly-NT-SP isomerized to the Ly-NT-MC with a weak red-shifted emission at 630 nm. The C=C-C=N⁺ fragment of MC could react with SO₂ to generate an adduct that emitted strong green fluorescence at 535 nm. Using this sensor, the overproduced SO₂ in lysosomes (induced by heat shock) was successfully visualized with a detection limit of 0.47 μM. Notably, tissue imaging showed that increased SO₂ alleviated



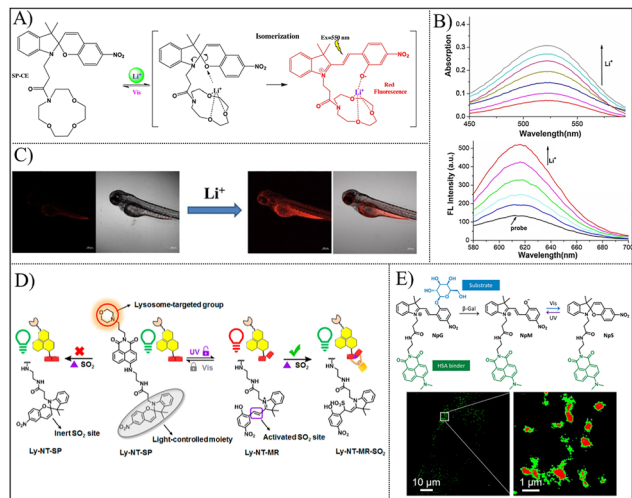


Fig. 16 (A) The isomerization of SP-CE is triggered by Li^+ binding and is reversed by Vis light irradiation. Reproduced with permission.¹⁰⁶ Copyright 2021, Elsevier. (B) Absorption spectra and fluorescence intensity of SP-CE in Li^+ solution. Reproduced with permission.¹⁰⁶ Copyright 2021, Elsevier. (C) Red fluorescence images of SP-CE in zebrafish. Reproduced with permission.¹⁰⁶ Copyright 2021, Elsevier. (D) A lysosome-targeted, SP-derived PICA (Ly-NT-SP) for SO_2 sensing. Reproduced with permission.²¹⁰ Copyright 2020, American Chemical Society. (E) A SP-based PICA (NpG@HSA) enabling super-resolution detection of β -Gal in cells. Reproduced with permission.¹⁶ Copyright 2020, American Chemical Society.

small intestine injury caused by heat shock *via* modulating oxidative stress. Additionally, Tian's group reported an amphiphilic photoswitchable molecule through modifying the naphthalimide-SP conjugate with a hydrophilic D -galactose (Fig. 16E).¹⁶ This molecule could assemble into a nanomicelle sensor for SO_3^{2-} detection in living cells. Subsequently, the group functionalized the similar naphthalimide-MC conjugate with a β -galactosidase (β -Gal)-responsive galactose unit to quench the red fluorescence emission (620 nm) of MC. The resulting molecule was bound to human serum albumin (HSA) through host-guest interactions, for increased green fluorescence emission at 520 nm as well as better biocompatibility. In the presence of β -Gal, the shielding unit was cleaved to expose the free phenol motif, thereby restoring the red fluorescence emission and photoisomerization function of the system. The reversible photoswitchable behavior of the sensor enabled on/off photoblinking for STORM imaging, which precisely visualized the subcellular localization of β -Gal in different cell lines. Collectively, these studies demonstrate the significance of small-molecule photoswitchable biosensors for detecting biomarkers related to physiological homeostasis or pathological processes, such as metal ions, small molecules, and enzymes.^{213–217}

In addition to small-molecule photoswitches, RSFPs have been used as genetically encoded indicators (GEIs) for photocontrollable sensing of specific metabolites and ions. This mechanism depends on photoisomerization-induced signal changes in the readout moiety of GEIs following target binding. Stiel's team constructed a prototype photoswitchable GEI (rsGCaMP) based on a common Ca^{2+} indicator GCaMP5G,

which enables reversible *cis/trans* photoisomerization in its Ca^{2+} -bound form upon altering irradiation with 405 nm and 488 nm light.¹¹ The key sensing function stems from interaction between the β -barrel structure of rsGCaMP and its calcium receptor moiety. rsGCaMP exhibited a fluorescent quantum yield of 9.2% and robust fatigue resistance at various pH values. It achieved super-resolution imaging of Ca^{2+} concentration in cultured cells and visualization of Ca^{2+} distribution in the endoplasmic reticulum, as well as photoacoustic Ca^{2+} imaging in HeLa cells implanted in mice. Notably, the universality of this photoswitchable molecular sensor concept was validated by successful construction of RSFP-based maltose sensor and dopamine sensor, offering a new approach for mapping analytes at the nanoscale and in live animal tissues. Subsequently, Dedecker's team engineered a series of GCaMP variants for cellular activity sensing, by retaining a Ca^{2+} -sensitive moiety (CaM) domain and a CaM-binding peptide (RS20) while introducing photochromic mutations.²¹⁸ GCaMP6s-Q was selected as the best variant with higher molecular brightness and Ca^{2+} response. Since Ca^{2+} concentration affects the overall fluorescence intensity and photochromic properties of the sensor, the team proposed a photochromism-enabled absolute quantification (PEAQ) method for Ca^{2+} detection. PEAQ correlated the fluorescence intensity changes before and after light irradiation with Ca^{2+} concentration and established a calibration correlation curve, enabling accurate determination of absolute Ca^{2+} concentration by simply measuring the fluorescence intensity. This strategy addressed the limitation of conventional photoswitchable probes that rely on complete switching cycles, thereby significantly lowering the required light dosage and shortening imaging acquisition time.

4.1.3. Dual/multiple-color imaging. Dual/multiple-color imaging enables simultaneous observation of different biomolecules or cellular events, substantially improving experimental throughput and data richness. PICAs have been used for photoswitchable dual/multiple-color bioimaging, especially in the field of super-resolution microscopy.^{12,219,220} Unlike monochromatic PICAs that cease signal emitting immediately once switching to their off state, dual/multiple-color PICAs allow continuous target tracking by cycling between distinct fluorescence colors, while maintaining spatial alignment with dynamic biological targets (*e.g.*, migrating cells). Dual/multiple-color imaging is generally preceded by combining a photoswitchable fluorescent molecule with a spectrally distinct fluorescent dye. This permits simultaneous acquisition in which the fluorescent dye emits continuously while the photoswitchable signal undergoes optical modulation (on/off cycling), consequently facilitating reversible toggling between single- and dual-channel imaging.

However, the key challenge is that spectral overlap between photoswitchable molecules and fluorescent dyes often induces FRET process, resulting in selective fluorescence quenching of the donor molecules. Therefore, fluorescent dyes featuring large Stokes shifts minimize spectral overlap with photoswitchable molecules by significantly separating their excitation and emission bands. A dual-color photoswitchable nanoparticle was



engineered by integrating a blue-emitted ($\lambda_{em} = 498 \text{ nm}$, $\Phi_F = 0.55$) DAE (DBTEO) at closed state and a large Stokes shift dye (3-(1-phenyl-1*H*-phenanthro[9,10-*d*]imidazol-2-yl)naphthalen-2-ol, HPNIC) with absorption at 337 nm and emission at 590 nm ($\Phi_F = 0.11$) (Fig. 17A).¹² This design achieved 100% switching on/off of blue emission from DBTEO upon alternating irradiation at 365 nm (on state) and $>420 \text{ nm}$ (off state). The persistent orange emission of HPNIC enabled long-term nano-carrier trafficking analysis in living cells, while the photoswitchable blue fluorescence of DBTEO provided spatiotemporally precise target localization *via* UV/vis light control (Fig. 17B).

Conversely, an alternative strategy for dual-color PICAs exploits the FRET effect between a photoswitchable molecule and spectrally coupled fluorescent dye.^{44,87,158,221–223} Jiang *et al.* synthesized a photoswitchable amphiphilic polymer nanoparticle that contains conjugated SP and BODIPY for FRET-mediated dual-color FI with high contrast in living cells (Fig. 17C).²²⁴ Upon exposure to Vis light, BODIPY emitted green fluorescence during cell imaging, while SP remained nonfluorescent. When UV light was applied, the green fluorescence disappeared, and red fluorescence emerged owing to the FRET process from BODIPY to the MC isomer.

Compared to multi-component PICA systems, multicolor imaging that employs a single photoswitchable molecule can efficiently suppress intermolecular energy transfer (*e.g.*, FRET) or electron transfer through confining all optical transitions within one molecule skeleton. In one study, SP was covalently with a fluorophore (distyryl anthracene, BOSA) *via* a flexible ester chain to develop a single-component probe for photoswitchable three-color FI (Fig. 17D).²²⁵ This probe displayed wavelength-dependent color modulation through the synergistic interaction between SP/MC photoisomerization and energy acceptor (BOSA). Upon 405 nm light irradiation, SP underwent ring-opening reaction to form MC, activating FRET effect from the BOSA to MC. This change red-shifted the emission from 520 nm (green) to 620 nm (red). Reversal to the SP state was achieved by 532 nm light irradiation, which terminated the FRET effect to recover green fluorescence. Notably, 800 nm two-photon excitation can directly trigger a yellow fluorescence (at 570 nm) of BOSA. To enhance aqueous solubility, BOSA-SP was encapsulated within amphiphilic F127 polymer-based nanoparticles to achieve efficient cellular uptake. This nanosystem programmed triple-fluorescence switching (green/yellow/red) in confocal microscopy, allowing significant suppression of cellular autofluorescence interference and high-contrast live-cell imaging with improved spatial resolution (fwhm = 31 nm).

However, realizing multiple-color fluorescence switching *via* a single photoswitchable molecule faces persistent challenges, such as inefficient switching, poor photostability, and low fluorescence quantum yield. Innovative strategies on molecular design hold promise to overcome these challenges, such as developing integrated chromophore systems combining complementary photoswitching mechanisms, devising stable covalent immobilization schemes, and refining energy transfer networks to suppress non-radiative decay. Progress in this field will provide improved spatiotemporal precision for multicolor

super-resolution imaging, thereby achieving real-time visualization of intricate biomolecular interactions and cellular dynamics with minimal background noise.

4.1.4. Multiplexed imaging. Multiplexed FI is constrained by the limited discrimination capacity among fluorophores.²²⁶ The absorption-emission cycle of fluorophores provides two spectral dimensions (absorption and emission spectra) along with an additional fluorescence lifetime parameter for label differentiation.²²⁷ However, broad excitation/emission bands tend to cause spectrum overlap, while fluorescence lifetime faces challenges due to narrow lifetime dispersion.²²⁸ Although deconvolution or subtraction algorithms offer feasible solutions, they inevitably compromise photon efficiency and increase computational complexity.²²⁹ Recent advances have leveraged the distinct photoisomerization kinetics of photoswitchable fluorophores (*e.g.*, photoactivated-state lifetimes and relaxation dynamics of photochemical responses) to discriminate spectrally overlapping probes.⁸ This principle has promoted the development of novel imaging techniques, including optical lock-in detection (OLID), synchronously amplified fluorescence image recovery (SAFIRE), and out-of-phase imaging after optical modulation (OPIOM).^{230–232}

Marriott *et al.* firstly employed OLID technology to isolate the characteristic fluorescence signal of photoswitchable probes (such as nitrospirobenzopyran and Dronpa) from background or overall fluorescence.²³⁰ By modulating the fluorescent and non-fluorescent states of photoswitchable probes and performing cross-correlation analysis with the switching waveform, OLID generates correlation images that can discriminate these probes from conventional fluorophores. This approach achieved high-contrast imaging, as evidenced in zebrafish neurons labeled with Dronpa, where the SBR was improved approximately 5-fold. In contrast, SAFIRE relies on the amplitude of fluorescence modulation induced by a secondary light source, which depopulates the dark state of the photoswitchable probe.²³³

Jullien *et al.* developed OPIOM, a technique using periodically modulated monochromatic illumination to drive oscillations in photoswitchable fluorophores between their two states.²³¹ Through phase-sensitive detection and resonance matching of illumination parameters and photoswitching kinetics, OPIOM isolates the specific fluorescence signal from overall out-of-phase fluorescence. In HEK293 cells co-expressing membrane-localized EGFP and nuclear-localized Dronpa-3, OPIOM rapidly removed EGFP interference to extract the Dronpa-3 signal. Similarly, in Dronpa-3-labeled zebrafish embryos, OPIOM also filtered out yolk autofluorescence and proved compatibility with light sheet fluorescence microscopy for 3D image reconstruction of the Dronpa-3 signal. OPIOM provides an effective approach for selective imaging of photoswitchable fluorescent probes. However, a key issue is its suboptimal photoswitching rate and extent. Jullien *et al.* developed speed OPIOM by optimizing RSFP switching using two antiphase-modulated light sources (Fig. 18A).²³⁴ This approach doubled the signal intensity of OPIOM while reducing the



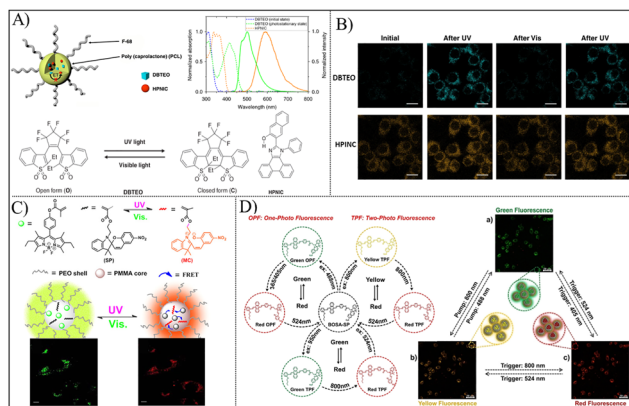


Fig. 17 (A) Construction of a DBTEO-based nanoparticle and its corresponding normalized absorption/emission spectra, as well as structure of DBTEO and HPNIC. Reproduced with permission.¹² Copyright 2019, Springer Nature. (B) Dual-color photoswitchable FI in living cells by using F127-encapsulated nanoparticles containing DBTEO-based nanoparticles, featuring emission at 498 nm (closed DBTEO) and 590 nm (HPNIC). Reproduced with permission.¹² Copyright 2019, Springer Nature. (C) Photoswitchable dual-color FI using SP-based nanoparticles in living cells. Reproduced with permission.²²⁴ Copyright 2015, American Chemical Society. (D) Schematic illustration of three-color fluorescence switching (green/red/yellow) of BOSA-SP and corresponding photoswitchable FI in HeLa cells. Reproduced with permission.²²⁵ Copyright 2022, Wiley-VCH.

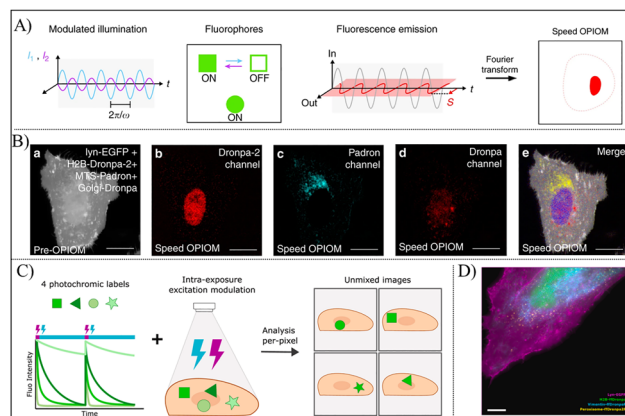


Fig. 18 (A) Schematic illustration of the speed OPIOM method. Reproduced with permission.²³⁴ Copyright 2017, Springer Nature. (B) Independent speed OPIOM images showing the distribution of four spectrally similar fluorescent proteins in HeLa cells. Localization: H2B-Dronpa-2 (nucleus), Mito-Padron (mitochondria), GTS-Dronpa (Golgi), and Lyn11-EGFP (cell membrane). Reproduced with permission.²³⁴ Copyright 2017, Springer Nature. (C) Schematic illustration of the NEEMO approach. Reproduced with permission.²³⁸ Copyright 2024, Elsevier. (D) NEEMO image showing the distribution of four differently targeted fluorescent proteins expressed in HeLa cells. Reproduced with permission.²³⁸ Copyright 2024, Elsevier.

imaging time to the millisecond level that is compatible with image acquisition rate of up to 50 Hz. Additionally, speed OPIOM effectively suppressed autofluorescence and scattered light, enhancing the contrast between photoswitchable fluorophores and spectrally overlapping non-photoactivatable fluorophores or ambient light by a factor of 10^2 – 10^3 . The technique also enabled discrimination of spectrally similar RSFPs (Dronpa, Dronpa-2, Dronpa, Kohinoor, and Padron) through modulated resonant illumination. In HeLa cells, speed OPIOM achieved real-time monitoring of Kohinoor-tagged ERK2 nuclear translocation in response to epidermal growth factor stimulation. Notably, speed OPIOM clearly distinguished Dronpa-2 (nucleus), EGFP (cell membrane), Padron (mitochondria), and Dronpa (Golgi) in U2OS cells, overcoming the spectral limitations of multiplexed imaging (Fig. 18B). This technology was subsequently implemented in macroscale fluorescence and fiber-optic epifluorescence imaging.^{235,236} Further improvements in temporal resolution and multiplexing capabilities could be achieved by developing brighter and faster-switching fluorescent probes.

The aforementioned methods demand the photoswitching response rate to match the temporal resolution of instruments, which is intrinsically constrained by the camera's exposure and readout time in widefield imaging. To address this constraint, Dedecker's team proposed a novel intra-exposure excitation modulation (NEEMO) technique to distinguish spectrally overlapping fluorophores.²³⁷ Through rapid modulation of excitation light, NEEMO induced photoisomerization in photoswitchable fluorophores while acquiring time-integrated fluorescence (Fig. 18C).²³⁸ Using merely 2–4 captured images, the technique successfully differentiated bacteria individually

labeled with distinct fluorescent proteins (EGFP, ffDronpa, ffDronpaF, and ffDronpa2F). Afterwards, the team developed an extended NEEMO approach to discriminate these four differently targeted fluorescent proteins expressed in HeLa cells, achieving fluorophore assignment accuracy of 80–95% (Fig. 18D).²³⁸ Unlike conventional methods, NEEMO bypasses direct kinetic analysis and instead leverages solely the modulation rate of the light source, allowing multiplexed imaging based on broad temporal dynamics.

Current dynamic-contrast FI generally employs a single temporal dimension, necessitating prolonged acquisition time and characteristic time span to distinguish multiple photoswitchable fluorophores. Jullien's team developed light-tunable time-gated reading-out of photocycles for multiplexed FI (LIGHTNING), a technique that substantially increases the number of distinguishable and identifiable RSFPs.²³⁹ LIGHTNING capitalizes on the primary light- and thermally-driven reactions of RSFPs, exploiting the dependence of photochemical reaction rate on light intensity to extract rich, RSFP-specific kinetic information. By applying multi-color illumination sequences at varying intensities, multiple characteristic times of RSFPs are collected, parameterizing the evolution of their fluorescence signals into a LIGHTNING kinetic fingerprint for each RSFP. LIGHTNING successfully distinguished 20 out of 22 spectrally similar green RSFPs in solution. Moreover, when tested on 16 *Escherichia coli* (*E. coli*) strains, each expressing a distinct RSFP, it accurately identified 9 strains with 93% accuracy. These results suggest that LIGHTNING can reliably discriminate fluorescent labels in complex samples, underscoring its potential for chromatic-aberration-free, multiplexed FI without the requirement for special optics or extensive data processing.



FRET offers an effective tool for understanding molecular interaction mechanisms, but its use in multiplexed imaging is limited by the difficult establishment of multiple donor-acceptor FRET pairs due to spectral overlap. Dedecker *et al.* proposed a strategy to distinguish spectrally similar FRET pairs by harnessing the photochromic parameters of the donor.²⁴⁰ This technique relies on cyclically switching the donor between fluorescent and non-fluorescent states by cyclic light irradiation, followed by calculating the photoswitching ratio from fluorescence changes of the donor. By constructing a photochromic FRET biosensor for cAMP-dependent protein kinase (PKA) activity (rsAKARev) and a spectrally overlapping but non-photochromic FRET biosensor for extracellular signal-regulated kinase (ERK) activity (EKARev), simultaneous detection of both kinases in the same cell was achieved. Additionally, combining this separation strategy with a red-shifted calcium sensor (RCaMP) enabled triple measurement of PKA, ERK, and calcium dynamics, demonstrating the efficacy of this method for FRET-based multiplexed imaging.

4.2. Phosphorescence imaging

PI is achieved by detecting the release of stored photons from phosphors after illumination ceases. The emission lifetime of phosphorescence (milliseconds to seconds) is significantly longer than that of fluorescence (nanoseconds), effectively suppressing autofluorescence interference with improved SNR.^{241–245} Currently, PI often suffers from weak signal intensity and environmental quenching under biological environments. Moreover, most phosphors emit at short wavelengths, limiting their applications in deep-tissue imaging.²⁴⁶ Although photoswitchable molecules lack intrinsic PI ability, they can spatiotemporally modulate phosphor performance by reversible on-off switching.

Recent progress has centered on developing photoswitchable phosphorescent probes with red-shifted emission *via* phosphorescence resonance energy transfer (PRET). Unlike FRET, PRET enable energy transfer from a triplet state to a singlet state, thus extending the lifetime of conventional phosphors. Leveraging this mechanism, Liu's team constructed a series of DAE-derived supramolecular assemblies for NIR PI.^{117,164,247–250} For instance, a DAE derivative (DTE-C3) modified with binary phenylpyridinium units was assembled with cucurbit[8]uril (CB[8]) to generate a photoswitchable phosphorescent probe (DTE-C3 < CB[8]) (Fig. 19A).¹⁵ In this system, the host-guest complex between CB[8] and DTE, described as a "single molecular dual-fold", exhibited strong room-temperature phosphorescence. Additionally, the FRET interaction between ring-opened DTE and co-encapsulated Cy5 extended the phosphorescence peak from 600 to 673 nm, improving penetration depth. Upon UV or Vis light irradiation, the phosphorescence signal was switched off/on owing to the isomerization between the non-luminescent ring-closed DTE and luminescent ring-opened DTE, which was further tested for photoswitchable PI in cells (Fig. 19B). On the basis, DTE-C3 was functionalized through substituting the terminal bromine atoms with 5-methyl thio-phenylene moieties, followed by assembly with CB[8] (Fig. 19C).¹⁶⁴

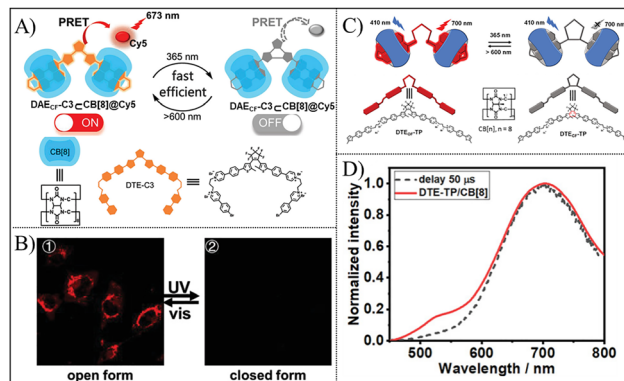


Fig. 19 (A) Structure of DAE-C3 and its supramolecular assembly exhibiting photoswitchable phosphorescence at 673 nm. Reproduced with permission.¹⁵ Copyright 2022, Wiley-VCH. (B) Photoswitchable PI in living cells under UV/Vis irradiation. Reproduced with permission.¹⁵ Copyright 2022, Wiley-VCH. (C) Photoswitchable phosphorescence of DAE-TP-based supramolecular assembly (DTE-TP/CB[8]) at 700 nm. Reproduced with permission.¹⁶⁴ Copyright 2022, Wiley-VCH. (D) Phosphorescence spectra of DTE-TP/CB[8]. Reproduced with permission.¹⁶⁴ Copyright 2022, Wiley-VCH.

The optimized probe demonstrated extended phosphorescence emission at 700 nm, achieving photo-controllable NIR PI of lysosomes in living cells. The NIR dye IR780 was incorporated to further extend the emission wavelength to 817 nm through the PRET effect (Fig. 19D). Collectively, photoswitchable phosphorescent probes offer a potential alternative for deep-tissue imaging with spatiotemporal precision.

4.3. Photoacoustic imaging

PAI combines the high sensitivity and contrast of optical imaging with the superior spatial resolution of ultrasound, circumventing the depth limitation typical of optical imaging (~ 1 mm).^{251,252} In PAI, biological samples are typically illuminated with a nanosecond-pulsed laser. The molecules within the samples absorb the light energy and convert it into thermal energy, causing a localized temperature increase that generates a photoacoustic signal.²⁵³ However, endogenous chromophores in blood, such as hemoglobin and melanin, introduce background interference, which severely undermines detection sensitivity and resolution. This highlights the necessity for developing high-contrast PAI probes.²⁵⁴ Benefiting from the photoswitching behavior, PICAs can effectively eliminate background signals by subtracting the PAI signals respectively acquired in the on and off states.^{92,93} Furthermore, the controllable switching between these two distinct isomeric states considerably improves detection sensitivity and the robustness of image reconstruction.

Genetically encoded photochromic chromoproteins, notably bacterial phytochromes (BphPs), offer advantages due to their high photostability, harmless non-invasiveness, and ability to be expressed by tissue-specific promoters. BphPs consist of a photosensory core module (PCM) and various effector domains. BphPs bind to biliverdin IX α chromophore, enabling differential PAI because of its two isomeric states, Pfr (ground state)



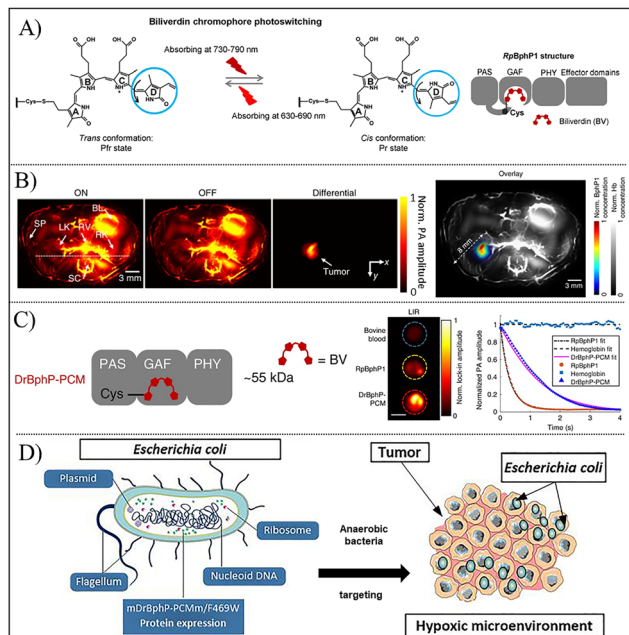


Fig. 20 RSFP-based PICAs for PAI. (A) Structure of RpBphP1 and photoisomerization behavior of the photoswitchable moiety (BV chromophore) between Pfr and Pr states. Reproduced with permission.⁹² Copyright 2016, Springer Nature. (B) Deep-tissue PAI of RpBphP1 in the kidney region of nude mouse. Reproduced with permission.⁹² Copyright 2016, Springer Nature. (C) Structure and PAI signals of the small-size RSFP, DrBphP-PCM. Reproduced with permission.⁹³ Copyright 2018, Springer Nature. (D) Schematic illustration of tumor-homing bacteria engineered with RSFPs for photoswitchable PAI on tumor. Reproduced with permission.²⁵⁸ Copyright 2022, The National Academy of Sciences.

and Pr. Reversible switching between these two states is mediated by *cis-trans* isomerization of the D-ring around the C15=C16 double bond. Wang's group developed a NIR photochromic BphP from *Rhodospseudomonas palustris* (referred to as RpBphP1).⁹² The ground-state RpBphP1 (on state) transitioned to the Pr isomer (off state) upon illumination with 780 nm light, and reversed back upon illumination with 630 nm light or through natural relaxation with a half-life of ~ 210 s (Fig. 20A). The robust and repeatable photoswitching could be maintained over many cycles without photobleaching. In scattering media, the PAI signal amplitude of RpBphP1 at a depth of 10 mm was 10 times higher than that of rsTagRF, a convention RSFP that can be toggled between 440 nm and 570 nm light illumination. This enhancement is attributed to the ~ 200 -nm red-shifted absorption and twofold higher extinction coefficient of RpBphP1. Differential PAI of RpBphP1 showed 21-fold increase in contrast-to-noise ratio (CNR) compared with on-state imaging alone. U87 human glioblastoma cells were engineered to express RpBphP1 for monitoring tumor growth in mice. After 20 switching cycles, differential PAI could clearly indicate kidney tumors at depths of up to 8 mm and brain tumors at ~ 2 mm depth, with an average CNR of 20. Furthermore, PAI provided accurate longitudinal tracking of both primary hepatic tumor progression and tumor metastases over a one-month period (Fig. 20B). This study provides groundbreaking guidance

for noninvasive multi-contrast deep-seated PAI. Subsequently, the same group developed another NIR photochromic BphP from *Deinococcus radiodurans* (referred to as DrBphP-PCM).⁹³ DrBphP-PCM exclusively preserved the 55-kDa PCM region, resulting in improved folding efficiency and a 2.3-fold higher expression level in U87 cells than RpBphP1 (Fig. 20C). Along with significantly increased NIR absorption and photoswitching contrast, DrBphP-PCM demonstrated a 10-fold enhancement in CNR compared to RpBphP1. Through single-impulse panoramic photoacoustic computed tomography, brain, kidney, and liver tumors expressing DrBphP-PCM were successfully detected *in vivo* with a resolution of 125 μm . Therefore, reducing the size of protein-based probes offers a potential pathway to resolution enhancement.

The heterogeneous distribution in endogenous biliverdin poses a challenge for BphP-mediated PAI of specific tissues or organs. Kasatkina *et al.* employed a Cre recombinase-dependent method to express BphP1 with temporal and spatial targeting.²⁵⁵ In BphP1-Cre transgenic mice, differential PAI of BphP1 efficiently eliminated blood background noise, and indicated distinct BphP1 expression patterns across organs, such as higher signals in liver and spleen compared to stomach. The intracellular BphP1 concentration within these organs was estimated at ~ 100 nM. This system was also successfully applied for non-invasive imaging of developing embryos and longitudinal tracking of viral-mediated gene expression in the kidney and liver. These results demonstrate the high sensitivity and deep penetration of PAI achieved by Cre-dependent BphP1 expression. In this study, beyond its PAI role, BphP1 also functioned as an optogenetic promoter to activate target gene expression.

Targeted delivery of exogenous BphPs or their encoding genes represents an alternative strategy. Chen *et al.* engineered a protein assembly, DrBphP-CBD, by fusing the photosensory module of DrBphP with a collagen-binding domain (CBD).^{256,257} DrBphP-CBD exhibited high affinity for type I and III collagen, and preferentially accumulated and retained in collagen-rich tumors following intravenous administration. This construct maintained superior reversible photoswitching properties, including high photoswitching contrast, photobleaching resistance, and an absorption coefficient ratio (Pfr/Pr) of 11 at 760 nm. In aqueous media, it achieved a 20-fold increase in the SNR of differential PAI compared to the on-state Pfr. Ten hours after administration, DrBphP-CBD demonstrated approximately 7-, 10.6-, and 2.5-fold higher differential PAI signals than DrBphP in MDA-MB-231, MC38, and 4T1 solid tumor models, respectively. The differential accumulation of DrBphP-CBD across tumor models is attributed to higher collagen expression in MDA-MB-231 and MC38 than 4T1. Compared to non-photo-switchable IR780 nanoparticles, DrBphP-CBD significantly enhanced imaging contrast and background discrimination for real-time dynamic monitoring and accurate quantitative analysis. This study advances photoswitchable phytochrome-based PAI imaging and provides a novel strategy for precision tumor diagnosis. Notably, although the endogenous blood protein-derived CBD may reduce immunogenicity, the tissue



toxicity of DrBphP-CBD still needs more comprehensive evaluation. In a study, tumor-homing *Escherichia coli* (*E. coli*, MG1655) was reprogrammed to express a genetically encoded BphP, mDrBphP-PCMm/F469W (Fig. 20D).²⁵⁸ mDrBphP-PCMm/F469W maintained most of the superior photoswitching properties of BphPs, while exhibiting 1.75-fold and 4.2-fold higher photoswitching contrast compared to DrBphP-PCM and RpBphP1, respectively. After being intravenously administered to 4T1 tumor-bearing mice, *E. coli* preferentially migrated toward the solid tumor and started to proliferate. The tumor-tropic accumulation of *E. coli* enabled differential PAI of deep-seated tumor tissues with significantly enhanced imaging contrast. In contrast, only negligible PAI signals were observed in normal tissues or organs, demonstrating the remarkable tumor-targeting ability of *E. coli* for PAI probe delivery. Collectively, spatiotemporally selective PAI of BphPs is expected to benefit early tumor detection and tumor progression monitoring.

PAI probes based on organic photoswitchable molecules surpass BphPs in chemical tunability and diversity, while offering simpler screening and preparation advantages. In one study, two (1-(4-(1,2,2-triphenylvinyl)phenyl)ethylidene)-malononitrile units were attached to the terminal positions of a DTE core (DTE-TPECM).⁶³ In its ring-closed form, the planar conjugated structure formed by the two thiophene rings promoted intramolecular energy transfer and interactions, redirecting energy dissipation toward thermal relaxation for efficient PAI. To boost biocompatibility *in vivo*, DTE-TPECM was assembled with an amphiphilic lipid-PEG2000 into nanoparticles (DTE-TPECM NPs). These nanoparticles showed photoacoustic intensity 2.0-fold and 1.8-fold higher than that of methylene blue and polymer nanoparticles, respectively, and markedly enhanced resistance to reactive oxygen and nitrogen species compared to ICG. Following conjugation with the targeting peptide YSAYPDSVPMMS, DTE-TPECM NPs preferentially enriched in tumors that overexpress the EphA2 protein. In 4T1-tumor-bearing mice, intravenous injection of the peptide-modified DTE-TPECM NPs successfully achieved precise PAI of deep-seated tumors to guide tumor resection. On the other hand, upon 610 nm light irradiation for 5 minutes, these nanoparticles underwent photoisomerization to the ring-opened state, thereby restoring fluorescence to indicate postoperative tumor residues.

However, most organic photoswitchable molecules require high-energy light (e.g., UV light) for isomerization, restricting their applicability in deep-tissue differential PAI. To overcome this issue, UCNP have been utilized to convert low-energy NIR light into UV energy for the reversible photoisomerization of small-molecule photoswitches. One study designed a photoswitchable hybrid nanoprobe by integrating a dithienylethene-containing β -diketone (3ThacacH, exhibiting strong photoacoustic signal under the ring-closed state) and a core-shell structured UCNP.²⁵⁹ The UCNP consisted of a non-doped NaYF₄ shell, and a NaYF₄ nanocrystal core co-doped with ytterbium (Yb) for NIR light harvesting and thulium (Tm) for UV emission. The UCNP surface was coated with an amphiphilic polymer (PSMA-M2070) to form a hydrophobic matrix layer for loading 3ThacacH. Upon 980 nm light irradiation, the

UCNP efficiently converted NIR light to 365 nm UV emission, which spectrally overlaps with the absorption of the ring-opened 3ThacacH, thus driving its photocyclization to the closed state. This mechanism enabled reversible photoacoustic signal switching: 980 nm light switched the signal on, while 640 nm light returned the system to the ring-opened state to turn it off. The nanoprobe showed a photoacoustic on/off signal ratio of ~ 10 , and maintained a CNR over 3 in hemoglobin solution even at a low concentration of 0.05 nM, indicating its superior ability to suppress background interference by differential imaging. When used to HeLa cell labeling, nanoprobe-based differential imaging by 20 switching cycles achieved a detection sensitivity of 5×10^3 cells *in vitro* and 10^4 subcutaneously implanted cells *in vivo*, a two-order-of-magnitude increase over conventional photoacoustic contrast agents such as ICG. Notably, employing low-energy 980 nm light irradiation to switch on PAI minimized phototoxicity and extended imaging depth to ~ 1.8 cm. In another study, analogous NaYF₄ UCNP were functionalized with a photoswitchable Azo-grafted polymer (PAA-Azo) *via* ligand exchange method (Fig. 21A).²⁶⁰ Upon 808 nm light irradiation, UCNP simultaneously emitted 365 nm UV light and 475 nm blue light. This dual-wavelength output not only triggered reversible *trans-cis-trans* isomerization of PAA-Azo to produce vibrational energy and heat, but also transformed it into a liquid state with a higher thermal expansion coefficient. These effects synergistically amplified photoacoustic signal by sixfold compared to uncoated UCNP (Fig. 21B). The probe allowed high-resolution PAI of mouse femora artery and rapid lymph node detection, clearly visualizing tissue structure and function. Moreover, by combining the benefits of PAI and NIR-II imaging, the probe supported real-time, precise guidance during lymph node resection.

In addition, molecular switches have been integrated with existing organic photoacoustic contrast agents to address their limitations in PAI, such as photostability and nonlinear

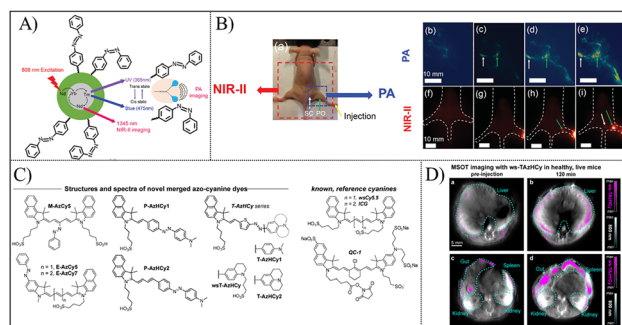


Fig. 21 Photoswitchable molecule-based PICAs for PAI. (A) Principal concept for photoacoustic signal enhancement through the construction of Azo-UCNP-based polymer nanoparticles. Reproduced with permission.²⁶⁰ Copyright 2019, Wiley-VCH. (B) Dual photoacoustic and NIR-II imaging of lymph nodes in mice using Azo-UCNP-based polymer nanoparticles. Reproduced with permission.²⁶⁰ Copyright 2019, Wiley-VCH. (C) Structures of Azo-cyanine dyes. Reproduced with permission.¹⁷ Copyright 2024, Wiley-VCH. (D) *In vivo* PAI showing the biodistribution of wsT-AzHCy in nude mice. Reproduced with permission.¹⁷ Copyright 2024, Wiley-VCH.



responses to excitation intensity caused by slow $S1 \rightarrow S0$ electronic relaxation. Azo, with its non-fluorescent emission and picosecond $S1 \rightarrow S0$ relaxation half-time, serves as an excellent energy acceptor for nonradiative heat dissipation.^{261,262} Furthermore, Gambhir's work demonstrated the complex Azo BHQ3 as a strong photoacoustic emitter.²⁶³ However, no water-soluble Azo derivatives have been available in the ideal NIR window for PAI. Therefore, Thorn-Seshold *et al.* developed a series of azohemicyanines by grafting the Azo switch onto the middle or terminal position of the NIR cyanine dyes (strong absorption to enter $S1$ state) for enhanced photo-switchable PAI (Fig. 21C).¹⁷ Benefiting from ultrafast vibrational relaxation (<10 ps, >100 -fold faster than ICG), most synthetic azohemicyanines with NIR absorption demonstrated >10 -fold higher photostability compared to ICG. Furthermore, the ultrafast vibrational relaxation enabled multiple $S1 \rightarrow S0 \rightarrow S1$ excitation cycles within a single laser pulse, significantly enhancing the photoacoustic signal generation efficacy of these probes to over 3 times that of ICG. The improved signal longevity and loudness effectively addressed the nonlinear response of cyanine photoacoustic dyes to illumination intensity, thus achieving higher photoacoustic signal output for improved resolution and lower background. Among all azohemicyanin, the hydrophilic thiophenylazo-bridged probe (designated as wsT-AzHCy, absorption peak at 729 nm) showed the optimal potential for *in vivo* PAI, with >1900 -fold enhanced photostability and 3.2-fold signal intensity. Two hours after intravenous administration of wsT-AzHCy, PAI successfully separated the probe signal from endogenous background signal, suggesting the sufficient survival time of wsT-AzHCy for practical applications (Fig. 21D). Furthermore, the photoacoustic signal was predominantly observed in the liver, spleen, and gut, but absent in the kidneys—a promising indication for future molecular imaging. This work proposes a novel chemical concept that leverages molecular switches to accelerate the $S1 \rightarrow S0$ relaxation of photoacoustic contrast agents for improved PAI.

4.4. Magnetic resonance imaging

Magnetic resonance imaging (MRI) is a noninvasive, non-ionizing radiation, and multi-dimensional imaging modality.²⁶⁴ The signal amplification of MRI contrast agents stems from paramagnetic centers that perturb proton relaxation kinetics. Specifically, hydration-shell ^1H nuclei under an external applied field experience magnetic field distortion, which enhances signal amplitude *via* modulating T_1/T_2 (longitudinal/transverse) relaxation time.²⁶⁵ In T_1 -weighted imaging, short T_1 contrast agents (*e.g.* Gad-type) produce bright signals; conversely, in T_2 -weighted imaging, long T_2 contrast agents (*e.g.* Iron-type) enhances MRI intensity. Since the initial clinical approval of gadopentetate dimeglumine (Gd-DTPA) in 1988, over 200 million doses of MRI contrast agents have been administered globally.²⁶⁶ However, the suboptimal relaxivity of current MRI contrast agents often requires high doses for adequate image differentiation.

Over the past decades, photo-activated MRI contrast agents have enabled spatiotemporally precise signal modulation and

enhanced proton relaxivity.^{267,268} Unlike unidirectional photo-contrast agents that are limited to single-use contrast release, PICAs support repeatable signal switching, thereby facilitating longitudinal MRI tracking and higher SNR. Upon light irradiation, photoisomerization modulates proton relaxivity by two primary mechanisms: (1) photoisomerization of photoswitchable moiety changes the symmetry of the primary hydration shell around paramagnetic centers (*e.g.* Gd^{3+} and Ni^{2+}), which directly governs proton relaxivity; (2) light-triggered conversion of the metal coordination geometry (when photoswitchable molecule as a ligand) induces the switching from a diamagnetic state to a paramagnetic state, thereby amplifying proton relaxivity.

Photoswitchable MRI probes primarily employ SP or Azo as photoswitching moieties, leveraging their pronounced conformation changes to dynamically modulate MRI signals. Since 2007, Louie *et al.* have systematically studied SP-based PICAs for MRI.^{269–273} The photoisomerization between the SP state and the MC state perturbs the molecular microenvironment to modulate water-proton relaxivity. In their initial study, they designed a photoswitchable MRI contrast agent (Gd-SPDO3A) through coordinating SP with Gd-DO3A, enabling reversible photoswitching of MRI signals (Fig. 22A).²⁶⁹ Under dark-adapted equilibration, coordination between Gd^{3+} and the phenolate oxygen promoted the $\text{SP} \rightarrow \text{MC}$ isomerization and stabilized

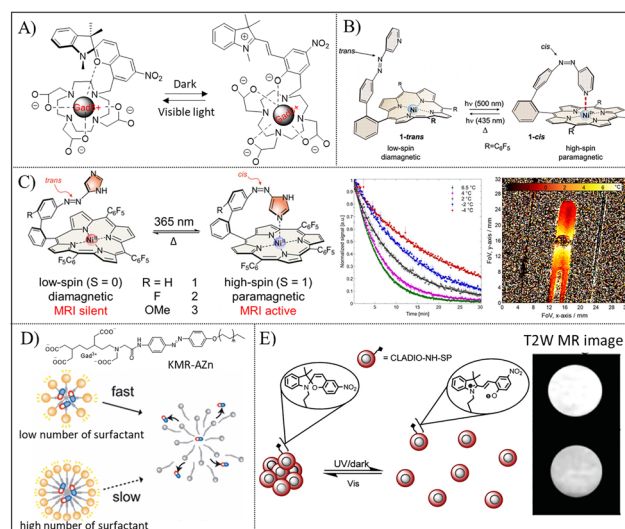


Fig. 22 (A) Photoisomerization of Gd-SPDO3A (left) and Gd-MCDO3A (right). Reproduced with permission.²⁶⁹ Copyright 2007, Royal Society of Chemistry. (B) Light-induced coordination and de-coordination between azopyridine and Ni^{2+} for magnetic signal switching between diamagnetism and paramagnetism *via trans-cis* isomerization of azopyridine. Reproduced with permission.¹⁸ Copyright 2011, American Association for the Advancement of Science. (C) Schematic illustration of the photoisomerization of azoimidazole-based PICAs and time-dependent thermal MRI mapping. Reproduced with permission.²⁷⁶ Copyright 2021, American Chemical Society. (D) Structure of amphiphilic Azo derivatives (KMR-AZn) and KMR-AZn-based micelles. Reproduced with permission.²⁷⁸ Copyright 2012, Wiley-VCH. (E) Photo-induced aggregation of CLADIO-NH-SP with reduced T_2 MRI signal. Reproduced with permission.²⁷⁹ Copyright 2010, American Chemical Society.



the metastable MC form in Gd-MCDO3A, thereby generating an absorption peak at 486 nm. Upon Vis light irradiation, the MC \rightarrow SP photoisomerization increased the relaxivity (r_1) of Gd-MCDO3A by 21.2% (from 3.72 to 2.93 $\text{mM}^{-1} \text{s}^{-1}$). This increase resulted from the shortened distance between the indoline unit of SP and Gd^{3+} through enhanced electrostatic interaction, which restricted water access to the paramagnetic center— Gd^{3+} ion.

Currently, the *in vivo* application of photoswitchable MRI agents is limited by potential phototoxicity under high-intensity illumination. Louie's group developed a SP-based MRI contrast agent that undergoes MC \rightarrow SP isomerization upon irradiation with low-intensity Vis light.²⁷¹ Specifically, just 1 min of exposure to blue LED light ($\lambda = 465 \text{ nm}$, $< 2.37 \times 10^{12} \text{ photons s}^{-1}$) converted MC-DO3A-Gd into SP-DO3A-Gd, decreasing its r_1 relaxivity from 2.93 to 2.63 $\text{mM}^{-1} \text{s}^{-1}$. Notably, this light intensity matches the emission level of luciferase-luciferin systems, indicating the feasibility of bioluminescence-driven photoisomerization. Although luciferase emission alone failed to trigger the MC \rightarrow SP isomerization, this approach retains potential for future endogenously activated MRI.

Unlike SP-based MRI agents that attenuate MRI signals by disrupting the interaction between water and paramagnetic center, Azo-based MRI agents exhibit enhanced paramagnetism. Their *trans* \rightarrow *cis* photoisomerization enables switching between diamagnetic state and paramagnetic state *via* light-driven coordination-induced spin-state switching (LD-CISS). Herges' group has been dedicated to developing Azo-based photoswitchable MRI probes and utilizing *trans-cis* isomerization to modulate the spin state of coordinated transition metals.^{3,18,274–276} For instance, they designed an Azo-based PICA (1-*trans*) in which *trans-to-cis* isomerization of the phenylazopyridine (PAP) modulates axial coordination between the pyridyl nitrogen at the Azo terminal and the square-planar Ni-porphyrin, thereby driving the spin transition from low-spin ($S = 0$) to high-spin ($S = 1$) (Fig. 22B).¹⁸ Irradiation with 500 nm light converted the diamagnetic 1-*trans* isomer to the paramagnetic 1-*cis* isomer with a yield of $\sim 65\%$. Conversely, exposure to 435 nm light reversed the isomerization process with a yield of $\sim 90\%$. This reversible photoswitching persisted for $> 10\,000$ cycles, indicating the outstanding fatigue resistance of PAP-based PICAs. Notably, the magnetic moment of the 1-*cis* isomer was $2.989 \pm 0.084 \text{ B.M.}$, consistent with paramagnetic Ni^{2+} complexes (expected range: 2.8 to 3.4 B.M.), while the resultant 1-*trans* isomer upon blue light irradiation exhibited diamagnetism behavior with a negligible magnetic moment ($\mu \approx 0 \text{ B.M.}$). To improve clinical translation potential, the group further presented an optimized design of a molecular spin-state switch that is water-soluble and switchable in the far-red and NIR regions.²⁷⁷ By engineering porphyrins into substituted dihydroporphyrin (chlorin) or tetrahydroporphyrin (bacteriochlorin), the newly synthesized PICAs demonstrated enhanced photoswitching performance: *cis* \leftrightarrow *trans* (also diamagnetic \leftrightarrow paramagnetic) switching efficiency reached 72–78%, while spin-state switching efficiency in aqueous systems raised from 19% to 41%. Moreover, both chlorin and bacteriochlorin probes exhibited a blue-shift absorption and reversible photoisomerization

upon long-wavelength illumination: 505 nm light irradiation triggered *cis-to-trans* conversion, whereas 620/720 nm light irradiation induced *trans-to-cis* switching. Such NIR-activatable switching offers an attractive translational prospect for deep-tissue MRI diagnosis.

MRI signals depend on multiple interdependent parameters (*e.g.*, contrast agent concentration, temperature sensitivity, and relaxation time). Body temperature is a critical biomarker in clinical diagnosis for tissue inflammation, metabolic disorders, and tumors. It is important to separating temperature contribution to the generated MRI signals through independent quantification of these parameters. However, current methods impose technical barriers to hinder clinical translation. Herges *et al.* designed a series of azoimidazole-based photoswitchable MRI probes to assess time-dependent thermal MRI mapping, superseding conventional static relaxation time measurement (Fig. 22C).²⁷⁶ Upon 365 nm light irradiation, *cis*-PICAs ($\sim 60\%$ yield) showed accelerated thermal relaxation, reverting completely to the *trans* configuration within seconds to minutes at room temperature. The thermo-driven *cis-trans* isomerization of the azoimidazole triggered dissociation of the imidazole ligand from Ni^{2+} , resulting in a spin-state switching in the Ni-porphyrin from paramagnetic (MRI-detectable) to diamagnetic (MRI-silent). This succeeded in the acquirement of MRI signals with standard clinical sequences by exploiting the intrinsic correlation between temperature and kinetic parameters (reaction rate and $t_{1/2}$), which is independent of probe concentration. These thermo-responsive MRI PICAs enabled high-resolution temperature mapping ($130 \times 130 \mu\text{m}$) with $\pm 0.22 \text{ }^\circ\text{C}$ accuracy *in vitro* through monitoring the thermal relaxation kinetics of *cis*-Azo.

In addition, photoswitchable molecules enable dual-level MRI signal modulation: (1) molecular-level signal control by photoswitchable metal-ligand coordination; and (2) nanoscale regulation between aggregated state and dispersed state by photoisomerization. Compared with free molecular agents, nanoscale contrast agents demonstrate significantly higher MRI signal intensity due to increased local concentration of paramagnetic centers and restricted molecular motion upon molecular aggregation. To improve photoswitchable MRI contrast, a series of nanomicelle probes (KMR-AZn) were developed through the self-assembly of an engineered amphiphilic molecule that couples a hydrophobic Azo derivative with hydrophilic Gd-DTPA, while also encapsulating a hydrophobic drug substitute, Nile Red (Fig. 22D).²⁷⁸ Upon 365 nm UV irradiation, the Azo derivative underwent *trans-to-cis* photoisomerization, evidenced by attenuation of the 350 nm absorption peak. This conversion triggered Nile Red release, indicating photocontrollable disassembly of the aggregated architecture. KMR-AZn exhibited high longitudinal relaxivity ($14.5\text{--}16.5 \text{ mM}^{-1} \text{ s}^{-1}$), surpassing clinical Gd-DTPA ($4.1 \text{ mM}^{-1} \text{ s}^{-1}$) by 3–4-fold. Following UV irradiation, r_1 value decreased from 15.3 to 12.9 $\text{mM}^{-1} \text{ s}^{-1}$ for KMR-AZ6 and from 16.5 to 14.7 $\text{mM}^{-1} \text{ s}^{-1}$ for KMR-AZ8, demonstrating reversible MRI signal photoswitching.

Similar to T_1 -weighed MRI, the MC \rightarrow SP isomerization also decreases the signal intensity of T_2 -weighed MRI. Louie *et al.*



designed a photoswitchable MRI probe based on SP-modified iron oxide nanoparticles (CLADIO-NH-SP) to modulate the T_2 relaxation time (Fig. 22E).²⁷⁹ CLADIO-NH-SP was initially incubated in the darkness for >1 h to ensure complete SP \rightarrow MC photoisomerization. Upon Vis light irradiation, the absorption at 563 nm decreased concomitant with the nanoparticle hydrodynamic diameter transiting from 200 nm in the dispersed state (CLADIO-NH-MC) to 520 nm in the aggregated state (CLADIO-NH-SP). After 2-hour dark incubation, CLADIO-NH-SP gradually depolymerized, accompanied by increased absorption at 563 nm and restoration of the CLADIO-NH-MC with dispersed size distribution, indicating reversible SP \leftrightarrow MC photoisomerization. CLADIO-NH-MC exhibited a T_2 relaxation time of 37.09 ± 3.09 ms in the darkness, while CLADIO-NH-SP upon Vis light irradiation showed a shortened T_2 relaxation time of 24.55 ± 1.86 ms. This change in magnetic response stems from enhanced interparticle magnetic coupling by nanoparticle aggregation. In T_2 -weighted imaging, Vis light exposure induced a 15% signal decline from 249 ± 5 to 211 ± 3 arbitrary units, correlating with reduced T_2 relaxation time. This photoswitchable probe holds promise as an MRI gene reporter for deep-tissue luciferase expression system.

4.5. Raman imaging

Raman scattering is an inelastic scattering phenomenon where photons interact with molecules, resulting in energy exchange that reveals molecular vibrational information.^{272,273} RI acquires high-resolution maps of the chemical composition and molecular structure of a sample by spatially tracking its Raman scattering signals. This technique supports both label-free and targeted small-molecule imaging, and exhibits superior multiplexing capability because Raman spectral linewidths are ~ 50 times narrower than those of FI.^{280,281} Stimulated Raman scattering (SRS) overcomes the low sensitivity limitation of spontaneous RI, while electronic pre-resonance SRS (epr-SRS) further improves sensitivity to 250 nM and multiplexing of up to 24 colors.^{226,282} Nevertheless, conventional Raman probes suffer from static signals and are susceptible to photodegradation during prolonged illumination, which undermines imaging stability. By contrast, photoswitchable Raman probes can dynamically alter their vibrational signatures *via* photoisomerization, generating spectral shift that improves imaging contrast and multiplexing capability. Additionally, the reversible on/off switching of Raman signals allows spatiotemporally controlled dynamic imaging.

Small-molecule Raman probes offer distinct advantages such as single-wavelength excitation for analyzing multiple vibrational modes, exceptional photostability, and reaction specificity, but suffer from relatively low sensitivity. Gao's team designed LysoAzo-pH, an Azo-based resonance Raman pH probe for lysosomal pH sensing (Fig. 23A).²⁸³ The intrinsic Raman signal of LysoAzo-pH was 10^4 times greater than that of a conventional alkyne-based Raman reporter molecule, 5-ethynyl-2'-deoxyuridine. By substituting electron-donating and -withdrawing groups on the Azo moiety, simultaneous fluorescence quenching and resonance-enhanced Raman scattering

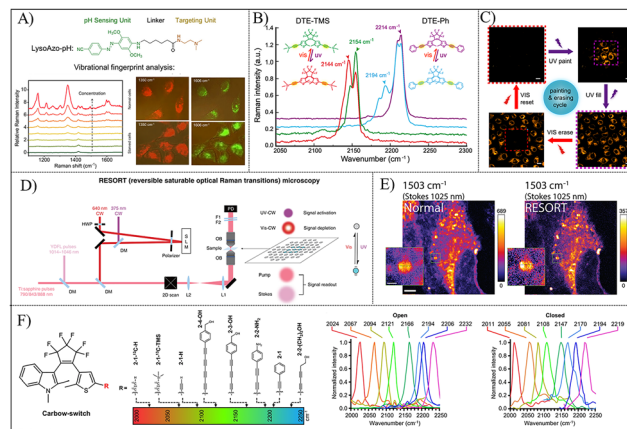


Fig. 23 (A) Raman spectra at varying concentrations and RI of Azo-based PICA (LysoAzo-pH) in lysosomes. Reproduced with permission.²⁸⁷ Copyright 2021, American Chemical Society. (B) Spontaneous and stimulated Raman signal response in the photoisomerization of DTE-alkyne derivatives (DTE-TMS and DTE-Ph). Reproduced with permission.¹⁴ Copyright 2021, Springer Nature. (C) Live-cell painting and erasing with SRS microscopy using DTE-Ph-Mito. Reproduced with permission.¹⁴ Copyright 2021, Springer Nature. (D) Principle and design of RESORT microscopy. Reproduced with permission.²⁸⁵ Copyright 2023, American Association for the Advancement of Science. (E) Normal SRS imaging and RESORT imaging of live HeLa cells at 1503 cm^{-1} (Raman on-resonance). Reproduced with permission.²⁸⁵ Copyright 2023, American Association for the Advancement of Science. (F) Chemical structure of 8 DAE derivatives with their normalized SRS spectra in both the open and closed forms. Reproduced with permission.²⁸⁶ Copyright 2024, Springer Nature.

was achieved, with an enhancement factor reaching approximately 10^3 . The protonation/deprotonation-induced vibrational fingerprint shift of the Azo group enabled precise visualization of proton distribution dynamics in autophagic cells, achieving a sensitivity of < 0.2 pH units. In addition, Ji *et al.* constructed photochromic vibrational probes for SRS imaging by modifying both ends of a DTE scaffold with strong Raman-tag acetylene groups (Fig. 23B).¹⁴ For the representative probe (DTE-Ph), UV light irradiation (365 nm) induced ring-closing photoisomerization of the conjugated DTE moiety, resulting in a 20 cm^{-1} red shift of the narrow acetylene Raman peak. Subsequent visible light irradiation (633 nm) completely converted DTE-pH to the ring-opened isomer, restoring the origin sharp Raman peak at 2214 cm^{-1} . The reversible photoisomerization of DTE-Ph enabled high-contrast SRS imaging with a switching ratio of 50%. When used for targeting mitochondria, DTE-Ph achieved light-controlled “drawing” and “erasing” of cells, as well as dual-label multiplexed imaging (Fig. 23C). Furthermore, DTE-Ph exhibited excellent fatigue resistance (> 100 switching cycles), promoting long-term tracking of mitochondrial dynamics. However, SRS beam-caused photobleaching of this probe, likely due to two-photon excited cycloreversion, remained a limitation in bioimaging. Further optimization of molecular design should focus on enhance photostability and imaging sensitivity.

Wei's team proposed a general design strategy for EPR-SRS probes that exploits molecular electronic and vibrational coupling to switch the molecular electronic state among four



different configurations, thereby enabling photoswitchable control of ground-state EPR-SRS signal.²⁸⁴ Unlike small molecules, photoswitchable proteins possess intrinsic biocompatibility and markedly higher fatigue resistance. For example, the NIR photoswitchable DrBphP-PCM could shift its electronic resonance toward or away from the pump photon energy. After switching to the on state (Pfr state) under 690 nm light, DrBphP-PCM generated a Raman signal 200-fold stronger than that of 5-ethynyl-2'-deoxyuridine. In addition, it displayed outstanding stability (>40 switching cycles) and detection sensitivity below 1 μM , demonstrating its suitability for bioimaging based on this universal EPR-SRS strategy.

Traditional vibrational microscopy techniques possess certain limitations, such as requiring cell fixation and high-power illumination. Ozeki's team combined photoswitchable SRS with RESOLFT technology to develop reversible saturable optical Raman transitions (RESORT) microscopy, an approach that enabled super-resolution vibrational imaging (Fig. 23D).²⁸⁵ RESORT exploits a low-power (microwatt level) continuous-wave laser to modulate the SRS detection of photoswitchable Raman probes (under competitive conditions, only the photoswitchable probes at the center of the donut-shaped beam focus remain activating). The team designed a photoswitchable DAE derivative (DAE620) that reversibly toggles between on/off states by alternating irradiation with 375 nm light and 640 nm light, thereby activating or depleting Raman signals. Its ring-closed isomer displayed a strong SRS peak at 1503 cm^{-1} , with an intensity 25 times higher than that of its ring-opened isomer. Significantly, DAE620 showed excellent fatigue resistance (>20 switching cycles) and ultrafast photoswitching kinetics (time constant, $\tau < 1 \mu\text{s}$), providing critical foundation for RESORT imaging. In DAE620-labeled HeLa cells, RESORT reached a spatial resolution of 151 nm, approximately double that of conventional SRS imaging, surpassing the optical diffraction limit and revealing nanoscale organelle details in living cells (Fig. 23E). Moreover, RESORT demonstrated high contrast, chemical specificity, and minimal phototoxicity ascribing to its low-power irradiation. With the expanded library of photoswitchable Raman probes, RESORT is expected to propel highly multiplexed super-resolution bioimaging.

At present, research on reversible photo-controllable multiplexed SRS imaging remains limited. Hu's group developed a super-multiplexed Carbow palette (Carbow-switch) consisting of 16 photoswitchable polyene-based Raman probes. These probes were built on an asymmetric DAE backbone and incorporated molecular design strategies such as polyene conjugation, heterocyclic incorporation, and terminal functionalization (Fig. 23F).²⁸⁶ Their absorption peaks lied within a favorable range (560–720 nm), allowing efficient SRS excitation and signal amplification. Particularly, DAE derivatives based on the 1-methylindole skeleton displayed outstanding performance, including rapid photoswitching kinetics under low-power irradiation (microwatt level), high light-to-signal conversion yield (>70%, up to 93%), and robust fatigue resistance (>95% after 100 switching cycles). Reversible photoisomerization of these probes (triggered by alternating illumination with

405 nm light and 640 nm light) resulted in a significant Raman shift between 12–24 cm^{-1} and SRS signal enhancement up to 90-fold. With negligible spectral crosstalk and high signal resolution, Carbow-switch showed great potential for high-resolution multiplexed imaging. By targeted modifications, the team achieved photo-controllable 6-color and 9-color SRS imaging of specific organelles in living cells, and successfully observed dynamic changes of mitochondria under oxidative stress, lysosomal changes during stress granule formation, and intercellular organelle transport. Carbow-switch opens up a new avenue for studying subcellular dynamics and interactions in living cells with high spatiotemporal precision and multiplexing ability.

5. Conclusions

Photoswitchable molecules undergo reversible photoisomerization upon light irradiation at two distinct wavelengths, allowing precision bioimaging through tunable photophysical properties. These molecules serve as intrinsic PICAs or modulators for conventional ICAs, offering unique advantages: (1) background suppression *via* state-switching algorithms; (2) high fatigue resistance for stable long-term imaging; and (3) instantaneous signal control shortening acquisition time.

Small-molecule photoswitches occupy a larger share in PICA market. They exhibit superior chemical tunability, enabling precise control over physicochemical parameters such as absorption/emission wavelengths, photoswitching rates, hydrophilicity, and targeting property through synthesis and modification. Concurrently, their high stability and diverse photoswitching mechanisms allow adaptation to multiple imaging modalities. Azo exhibits rapid *trans-cis* isomerization for dynamic imaging, but suffer from thermal instability and UV light-dependent photoisomerization. Strategies such as introducing steric hindrance groups and metal coordination can stabilize the *cis* isomer. The thermal relaxation of Azo plays a dual role: although it impedes fluorescence switching for super-resolution imaging, it contributes to photoacoustic signal generation.²⁶⁰ Moreover, light-controlled coordination of Azo to Ni^{2+} enables reversible switching between paramagnetic and diamagnetic MRI signals.¹⁸ The temperature-responsive relaxation of *cis*-Azo has been leveraged for 3D temperature MRI.²⁷⁶ Concurrently, the inherent molecular vibrational energy of Azo scaffold enhances RI signals.^{288,289}

SP exhibits bright fluorescence in its MC state, which underpins super-resolution and dual-color imaging.^{195,224} Its multiple stimuli-responsive (ions/pH/temperature) ability expands biosensing applications,²⁹⁰ and reversible metal coordination allows MRI signal modulation. However, the slow switching kinetics and susceptibility to photodegradation require further electronic and steric optimization.^{291,292}

DAE offers exceptional thermal stability and fatigue resistance, rendering it suitable for super-resolution and PI.^{22,107,164} The NIR absorption of its closed form supports PAI,^{63,66} while alkyne-functionalized derivatives enable photoswitchable RI.¹⁴



However, the relatively low cyclization quantum yield and limited magnetic moment variation impede its utility in MRI, necessitating further molecular design optimization.

In contrast, RSFPs are expressed in specific organelles or proteins *via* gene transfection, offering high biocompatibility and suitability for long-term dynamic imaging. They have been extensively utilized in precise single-molecule and cellular tracking, as well as protein-protein interaction research by super-resolution imaging.^{293–295} Recent studies have focused on improving photoswitching efficiency, photostability, and fluorescence quantum yield of RSFPs.^{296,297} RSFPs with NIR absorption demonstrate superior photoacoustic conversion capacity, enabling high-contrast PAI at tissue depth up to 8 cm.^{92,298} However, the instability and complex structure of RSFPs hinder their combination with exogenous contrast agents (*e.g.*, inorganic nanoparticles, organic dyes, and metal chelates). Therefore, their broad implementation in PI, MRI, and RI still faces technical obstacles.

Recent progress has yielded numerous photoswitchable molecules that surpass classical counterparts mentioned above, greatly enriching the molecular repertoire for super-resolution imaging.^{299,300} However, photoswitchable imaging is still predominantly constrained to fluorescent technique. Advancing this field forward requires the development of next-generation PICAs capable of supporting multimodal applications. In clinic, accurate disease diagnosis and assessment often necessitates the integration of multiple imaging modalities, which can be achieved by the spatiotemporally programmable signal control *via* PICAs. For example, certain PICAs combine NIR absorption and fluorescence switching, enabling simultaneous photoswitchable PAI and FI.^{63,92} Similarly, coupling fluorescent photoswitchable molecules with magnetic metals facilitates dual-mode MRI and FI, where the on/off states of fluorescence and MRI signals are controlled independently.^{271,273} Notably, PICAs supporting concurrent imaging modes can benefit patients by reducing the requirement for separate administration of single-function contrast agents. Beyond imaging, rationally designed PICAs can also modulate cellular metabolism, enabling simultaneous disease diagnosis and therapeutic intervention. For instance, some PICAs serve as activatable photosensitizers for controlled generation of ROS—a key mediator in cellular damage, aging, and tumor progression.⁴⁷ Furthermore, photoswitchable molecules can function as optogenetic actuators to reversibly regulate ion-channel gating, thereby precisely manipulating cellular behavior.²⁹⁴ Future efforts should prioritize the development of photoswitchable systems that combine multimodal imaging with diagnostics and therapeutic ability.

Critical limitations still persist in the design of photoswitchable molecules, including structural instability, hydrophobicity, short-wavelength excitation, and suboptimal photophysical properties. While functional group modification and structural adjustment offer partial solutions, they often trade off other essential properties, owing to the inherent spatial constraint of small molecule frameworks. Self-assembly strategies present a facile and scalable alternative to tackle these issues, albeit at the cost

of losing precise molecular-level switching control, which potentially compromising imaging accuracy.

Another challenge is the application of PICAs in living systems. Developing PICAs with red-shifted excitation is promising to address phototoxicity and poor tissue penetration. Equally important is controlling their *in vivo* distribution. Modification with targeting groups and employing nano-material-based delivery systems can enhance the accumulation of PICAs at interested sites, thereby improving imaging accuracy while mitigating off-target side effects to healthy tissues. However, the potential toxicity from the introduced targeting groups or carriers should be thoroughly evaluated to promote the clinical translation of PICAs.

In summary, PICAs are rapidly evolving. With ongoing optimization and deep learning-directed predictive design, next-generation PICAs hold promise to revolutionize precision bioimaging across diverse modalities.

Conflicts of interest

No conflicts to declare.

Data availability

No primary research results, software or code have been included and no new data were generated or analysed as part of this review.

Acknowledgements

This study was supported by the National Natural Science Foundation of China (NSFC) (Grant No. 22371012, 82430067, and 32030060) and the National Key Research & Development Program of China (Grant No. 2021YFA1201000 and 2023YFC2605000). The authors also acknowledge support from the Major Project of Guangzhou National Laboratory (Grant No. GZNL2024 A03010), the Key Grant of the Beijing Natural Science Foundation (Grant No. L258028), and the Science Fund for Creative Research Groups of Nature Science Foundation of Hebei Province (Grant No. B2021201038). H. B. C. received financial support from the China Scholarship Council (CSC) (Grant No. 202306880018). J. Y. thanks the Nano & Material Technology Development Program through the National Research Foundation of Korea (NRF) funded by Ministry of Science and ICT (Grant No. RS-2024-00407093) and the National Research Foundation of Korea (NRF) grant funded by the Korean government (MSIT) (Grant No. 2022R1A2C3005420).

Notes and references

- 1 L. Wang and Q. Li, *Chem. Soc. Rev.*, 2018, **47**, 1044–1097.
- 2 A. M. Rice, C. R. Martin, V. A. Galitskiy, A. A. Berseneva, G. A. Leith and N. B. Shustova, *Chem. Rev.*, 2020, **120**, 8790–8813.



- 3 M. Dommaschk, M. Peters, F. Gutzeit, C. Schutt, C. Nather, F. D. Sonnichsen, S. Tiwari, C. Riedel, S. Boretius and R. Herges, *J. Am. Chem. Soc.*, 2015, **137**, 7552–7555.
- 4 G. Naren, W. Larsson, C. Benitez-Martin, S. Li, E. Perez-Inestrosa, B. Albinsson and J. Andreasson, *Chem. Sci.*, 2021, **12**, 7073–7078.
- 5 C. Li, H. Yan, L. X. Zhao, G. F. Zhang, Z. Hu, Z. L. Huang and M. Q. Zhu, *Nat. Commun.*, 2014, **5**, 5709.
- 6 M. Hofmann, C. Eggeling, S. Jakobs and S. W. Hell, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 17565–17569.
- 7 D. K. Tiwari, Y. Arai, M. Yamanaka, T. Matsuda, M. Agetsuma, M. Nakano, K. Fujita and T. Nagai, *Nat. Methods*, 2015, **12**, 515–518.
- 8 T. J. Chozinski, L. A. Gagnon and J. C. Vaughan, *FEBS Lett.*, 2014, **588**, 3603–3612.
- 9 J. Fölling, V. Belov, R. Kunetsky, R. Medda, A. Schönle, A. Egner, C. Eggeling, M. Bossi and S. W. Hell, *Angew. Chem., Int. Ed.*, 2007, **119**, 6382–6386.
- 10 H. Y. Jung, B. Kim, M. H. Jeon and Y. Kim, *Small*, 2022, **18**, e2103523.
- 11 K. Mishra, J. P. Fuenzalida-Werner, F. Pennacchietti, R. Janowski, A. Chmyrov, Y. Huang, C. Zakian, U. Klemm, I. Testa, D. Niessing, V. Ntziachristos and A. C. Stiel, *Nat. Biotechnol.*, 2022, **40**, 598–605.
- 12 D. Kim, K. Jeong, J. E. Kwon, H. Park, S. Lee, S. Kim and S. Y. Park, *Nat. Commun.*, 2019, **10**, 3089.
- 13 J. Chen, W. Zhong, M. Xue, H. Wang, M. Yu, P. Zhang and P. Yi, *Polym. Chem.*, 2017, **8**, 6520–6526.
- 14 J. Ao, X. Fang, X. Miao, J. Ling, H. Kang, S. Park, C. Wu and M. Ji, *Nat. Commun.*, 2021, **12**, 3089.
- 15 C. Wang, X. K. Ma, P. Guo, C. Jiang, Y. H. Liu, G. Liu, X. Xu and Y. Liu, *Adv. Sci.*, 2022, **9**, e2103041.
- 16 X. Chai, H. H. Han, A. C. Sedgwick, N. Li, Y. Zang, T. D. James, J. Zhang, X. L. Hu, Y. Yu, Y. Li, Y. Wang, J. Li, X. P. He and H. Tian, *J. Am. Chem. Soc.*, 2020, **142**, 18005–18013.
- 17 M. Müller, N. Liu, V. Gujrati, A. Valavalkar, S. Hartmann, P. Anzenhofer, U. Klemm, A. Telek, B. Dietzek-Ivanšić, A. Hartschuh, V. Ntziachristos and O. Thorn-Seshold, *Angew. Chem., Int. Ed.*, 2024, **63**, e202405636.
- 18 S. Venkataramani, U. Jana, M. Dommaschk, F. D. Sönnichsen, F. Tuzcek and R. Herges, *Science*, 2011, **331**, 445–448.
- 19 M. Canton, A. B. Grommet, L. Pesce, J. Gemen, S. Li, Y. Diskin-Posner, A. Credi, G. M. Pavan, J. Andréasson and R. Klajn, *J. Am. Chem. Soc.*, 2020, **142**, 14557–14565.
- 20 B. Roubinet, M. L. Bossi, P. Alt, M. Leutenegger, H. Shojaei, S. Schnorrenberg, S. Nizamov, M. Irie, V. N. Belov and S. W. Hell, *Angew. Chem., Int. Ed.*, 2016, **55**, 15429–15433.
- 21 M. Dong, A. Babalhavaeji, M. J. Hansen, L. Kálmán and G. A. Woolley, *Chem. Commun.*, 2015, **51**, 12981–12984.
- 22 D. Kim, A. Aktalay, N. Jensen, K. Uno, M. L. Bossi, V. N. Belov and S. W. Hell, *J. Am. Chem. Soc.*, 2022, **144**, 14235–14247.
- 23 F. Aleotti, A. Nenov, L. Salvigni, M. Bonfanti, M. M. El-Tahawy, A. Giunchi, M. Gentile, C. Spallacci, A. Ventimiglia, G. Cirillo, L. Montali, S. Scurti, M. Garavelli and I. Conti, *J. Phys. Chem. A*, 2020, **124**, 9513–9523.
- 24 M. Natali and S. Giordani, *Chem. Soc. Rev.*, 2012, **41**, 4010–4029.
- 25 G. Kortüm, W. Theilacker and V. Braun, *J. Chem. Phys.*, 1955, **23**, 1723–1724.
- 26 J. Zhang, Q. Zou and H. Tian, *Adv. Mater.*, 2012, **25**, 378–399.
- 27 H. Nie, J. L. Self, A. S. Kuenstler, R. C. Hayward and J. Read de Alaniz, *Adv. Opt. Mater.*, 2019, **7**, 1900224.
- 28 I. Yildiz, E. Deniz and F. M. Raymo, *Chem. Soc. Rev.*, 2009, **38**, 1859–1867.
- 29 H. Zhou, C. Xue, P. Weis, Y. Suzuki, S. Huang, K. Koynov, G. K. Auernhammer, R. Berger, H.-J. Butt and S. Wu, *Nat. Chem.*, 2017, **9**, 145–151.
- 30 A. Cembran, F. Bernardi, M. Garavelli, L. Gagliardi and G. Orlandi, *J. Am. Chem. Soc.*, 2004, **126**, 3234–3243.
- 31 C.-W. Chang, Y.-C. Lu, T.-T. Wang and E. W.-G. Diau, *J. Am. Chem. Soc.*, 2004, **126**, 10109–10118.
- 32 G. A. Woolley, *Chem. Soc. Rev.*, 2011, **40**, 4422–4437.
- 33 H. B. Cheng, S. Zhang, J. Qi, X. J. Liang and J. Yoon, *Adv. Mater.*, 2021, **33**, e2007290.
- 34 H.-J. Yu, H. Wang, F.-F. Shen, F.-Q. Li, Y.-M. Zhang, X. Xu and Y. Liu, *Small*, 2022, **18**, 2201737.
- 35 H. Jin, F. Liu, P. Wu, Z. Sun, P. Sui, Y. Cao, Y. Zhou and S. Lin, *Small*, 2025, **21**, 2408147.
- 36 G. Park, C. Ettles, M. Charles and R. H. E. Hudson, *J. Photoch. Photobio. A*, 2023, **441**, 114653.
- 37 K. Punia, D. Britton, K. Hüll, L. Yin, Y. Wang, P. D. Renfrew, M. L. Gilchrist, R. Bonneau, D. Trauner and J. K. Montclare, *Soft Matter*, 2023, **19**, 497–501.
- 38 D. Chao, Y. Yang, X. Jia and E. B. Berda, *Dyes Pigm.*, 2019, **167**, 77–82.
- 39 H. Xiong, K. A. Alberto, J. Youn, J. Taura, J. Morstein, X. Li, Y. Wang, D. Trauner, P. A. Slesinger, S. O. Nielsen and Z. Qin, *Nano Research*, 2023, **16**, 1033–1041.
- 40 G. Berkovic, V. Krongauz and V. Weiss, *Chem. Rev.*, 2000, **100**, 1741–1754.
- 41 R. Klajn, *Chem. Soc. Rev.*, 2014, **43**, 148–184.
- 42 A. D. Pugachev, I. V. Ozhogin, M. B. Lukyanova, B. S. Lukyanov, I. A. Rostovtseva, I. V. Dorogan, N. I. Makarova, V. V. Tkachev, A. V. Metelitsa and S. M. Aldoshin, *Spectrochim. Acta, Part A*, 2020, **230**, 118041.
- 43 Y. Wu, C. Yin, W. Zhang, J. Chao and F. Huo, *Dyes Pigm.*, 2021, **193**, 109467.
- 44 Z. Tian, W. Wu, W. Wan and A. D. Q. Li, *J. Am. Chem. Soc.*, 2009, **131**, 4245–4252.
- 45 G. Copley, J. G. Gillmore, J. Crisman, G. Kodis, C. L. Gray, B. R. Cherry, B. D. Sherman, P. A. Liddell, M. M. Paquette, L. Kelbaskas, N. L. Frank, A. L. Moore, T. A. Moore and D. Gust, *J. Am. Chem. Soc.*, 2014, **136**, 11994–12003.
- 46 Y. Xiong, A. Vargas Jentzsch, J. W. M. Osterrieth, E. Sezgin, I. V. Sazanovich, K. Reglinski, S. Galiani, A. W. Parker, C. Eggeling and H. L. Anderson, *Chem. Sci.*, 2018, **9**, 3029–3040.
- 47 T. Zhang, X.-Y. Lou, X. Li, X. Tu, J. Han, B. Zhao and Y.-W. Yang, *Adv. Mater.*, 2023, **35**, 2210551.



- 48 A. T. Frawley, V. Wycisk, Y. Xiong, S. Galiani, E. Sezgin, I. Urbancic, A. Vargas Jentzsch, K. G. Leslie, C. Eggeling and H. L. Anderson, *Chem. Sci.*, 2020, **11**, 8955–8960.
- 49 A. T. Frawley, K. G. Leslie, V. Wycisk, S. Galiani, D. Shrestha, C. Eggeling and H. L. Anderson, *ChemPhysChem*, 2023, **24**, e202300125.
- 50 H. Xia, K. Xie and G. Zou, *Molecules*, 2017, **22**, 2236.
- 51 V. K. Seiler, N. Tumanov, K. Robeyns, J. Wouters, B. Champagne and T. Leysens, *Crystals*, 2017, **7**, 84.
- 52 K. Schöller, S. Küpfer, L. Baumann, P. M. Hoyer, D. de Courten, R. M. Rossi, A. Vetushka, M. Wolf, N. Bruns and L. J. Scherer, *Adv. Funct. Mater.*, 2014, **24**, 5194–5201.
- 53 Y. Wang, T. Lan, N. Ji, Q. Meng and W. He, *Dyes Pigm.*, 2025, **238**, 112654.
- 54 C. Gong, P. Zhang, Y. Liu, R. Chen, K. Huang, W. Yang, L. Gao, M. Xu, X. Yang, Y. Liu, Q. Liu, B. Yan, B. Zhao and G. Jiang, *J. Hazard. Mater.*, 2025, **493**, 138343.
- 55 X. Tan, Y. Tang, T. Yang, G. Dai, C. Ye, J. Meng and F. Li, *Anal. Chem.*, 2023, **95**, 3309–3316.
- 56 J.-T. Ye, L. Wang, H.-Q. Wang, Z.-Z. Chen, Y.-Q. Qiu and H.-M. Xie, *RSC Adv.*, 2017, **7**, 642–650.
- 57 M. Irie, *Chem. Rev.*, 2000, **100**, 1685–1716.
- 58 H. Tian and S. Yang, *Chem. Soc. Rev.*, 2004, **33**, 85–97.
- 59 C. Shi, R. Zhu, K. Guo, L. Zhang, Z. Liu, X. Liu, Q. Ai and X. Hu, *Adv. Opt. Mater.*, 2023, **11**, 2301844.
- 60 T. Fukaminato, T. Hirose, T. Doi, M. Hazama, K. Matsuda and M. Irie, *J. Am. Chem. Soc.*, 2014, **136**, 17145–17154.
- 61 Z. Zheng, H. Hu, Z. Zhang, B. Liu, M. Li, D.-H. Qu, H. Tian, W.-H. Zhu and B. L. Feringa, *Nat. Photonics*, 2022, **16**, 226–234.
- 62 T. Fukaminato, T. Doi, N. Tamaoki, K. Okuno, Y. Ishibashi, H. Miyasaka and M. Irie, *J. Am. Chem. Soc.*, 2011, **133**, 4984–4990.
- 63 J. Qi, C. Chen, X. Zhang, X. Hu, S. Ji, R. T. K. Kwok, J. W. Y. Lam, D. Ding and B. Z. Tang, *Nat. Commun.*, 2018, **9**, 1848.
- 64 M. Estrader, J. Salinas Uber, L. A. Barrios, J. Garcia, P. Lloyd-Williams, O. Roubeau, S. J. Teat and G. Aromí, *Angew. Chem., Int. Ed.*, 2017, **56**, 15622–15627.
- 65 K.-X. Qin, Y.-S. Su, M.-Q. Zhu and C. Li, *ChemBioChem*, 2024, **25**, e202400326.
- 66 X. Tan, Y. Wang, H. Li, Y. Duan, B. Wen, J. Zhao, H. Kim, J. Y. Lee, L. Zhou, H.-B. Cheng and J. Yoon, *Small*, 2025, **21**, 2409027.
- 67 K.-H. Knauer and R. Gleiter, *Angew. Chem., Int. Ed. Engl.*, 1977, **16**, 113.
- 68 K. Li, Y. Xiang, X. Wang, J. Li, R. Hu, A. Tong and B. Z. Tang, *J. Am. Chem. Soc.*, 2014, **136**, 1643–1649.
- 69 Z. Ye, H. Yu, W. Yang, Y. Zheng, N. Li, H. Bian, Z. Wang, Q. Liu, Y. Song, M. Zhang and Y. Xiao, *J. Am. Chem. Soc.*, 2019, **141**, 6527–6536.
- 70 H. Miyasaka, Y. Satoh, Y. Ishibashi, S. Ito, Y. Nagasawa, S. Taniguchi, H. Chosrowjan, N. Mataga, D. Kato, A. Kikuchi and J. Abe, *J. Am. Chem. Soc.*, 2009, **131**, 7256–7263.
- 71 Y. Kobayashi, T. Katayama, T. Yamane, K. Setoura, S. Ito, H. Miyasaka and J. Abe, *J. Am. Chem. Soc.*, 2016, **138**, 5930–5938.
- 72 S. Honda and T. Toyota, *Nat. Commun.*, 2017, **8**, 502.
- 73 Y. Liu, Y. Yang, D. Shi, M. Xiao, L. Jiang, J. Tian, G. Zhang, Z. Liu, X. Zhang and D. Zhang, *Adv. Mater.*, 2019, **31**, 1902576.
- 74 Z. Chen, Y. Sun, X. Wang, W. Zhang and Z. Zhang, *Macromol. Rapid Commun.*, 2023, **44**, 2300198.
- 75 Q. X. Hua, B. Xin, Z. J. Xiong, W. L. Gong, C. Li, Z. L. Huang and M. Q. Zhu, *Chem. Commun.*, 2017, **53**, 2669–2672.
- 76 R. H. Mitchell, *Eur. J. Org. Chem.*, 1999, 2695–2703.
- 77 R. H. Mitchell, T. R. Ward, Y. Chen, Y. Wang, S. A. Weerawarna, P. W. Dibble, M. J. Marsella, A. Almutairi and Z.-Q. Wang, *J. Am. Chem. Soc.*, 2003, **125**, 2974–2988.
- 78 K. Klaue, W. Han, P. Liesfeld, F. Berger, Y. Garmshausen and S. Hecht, *J. Am. Chem. Soc.*, 2020, **142**, 11857–11864.
- 79 S. Dubuis, A. Dellai, C. Courdurié, J. Owona, A. Kalafatis, L. Vellutini, E. Genin, V. Rodriguez and F. Castet, *J. Am. Chem. Soc.*, 2023, **145**, 10861–10871.
- 80 K. I. Willig, W. Wegner, A. Muller, V. Calvet-Fournier and H. Steffens, *Cell Rep.*, 2021, **35**, 109192.
- 81 X. Zhang, X. Chen, Z. Zeng, M. Zhang, Y. Sun, P. Xi, J. Peng and P. Xu, *ACS Nano*, 2015, **9**, 2659–2667.
- 82 G. U. Nienhaus, *Angew. Chem., Int. Ed.*, 2012, **51**, 1312–1314.
- 83 R. Ando, H. Mizuno and A. Miyawaki, *Science*, 2004, **306**, 1370–1373.
- 84 M. Andresen, M. C. Wahl, A. C. Stiel, F. Gräter, L. V. Schäfer, S. Trowitzsch, G. Weber, C. Eggeling, H. Grubmüller, S. W. Hell and S. Jakobs, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 13070–13074.
- 85 A. Acharya, A. M. Bogdanov, B. L. Grigorenko, K. B. Bravaya, A. V. Nemukhin, K. A. Lukyanov and A. I. Krylov, *Chem. Rev.*, 2017, **117**, 758–795.
- 86 S. Habuchi, R. Ando, P. Dedecker, W. Verheijen, H. Mizuno, A. Miyawaki and J. Hofkens, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 9511–9516.
- 87 M. Andresen, A. C. Stiel, J. Folling, D. Wenzel, A. Schonle, A. Egner, C. Eggeling, S. W. Hell and S. Jakobs, *Nat. Biotechnol.*, 2008, **26**, 1035–1040.
- 88 T. Brakemann, A. C. Stiel, G. Weber, M. Andresen, I. Testa, T. Grotjohann, M. Leutenegger, U. Plessmann, H. Urlaub, C. Eggeling, M. C. Wahl, S. W. Hell and S. Jakobs, *Nat. Biotechnol.*, 2011, **29**, 942–947.
- 89 D. M. Shcherbakova, O. M. Subach and V. V. Verkhusha, *Angew. Chem., Int. Ed.*, 2012, **51**, 10724–10738.
- 90 F. Pennacchietti, E. O. Serebrovskaya, A. R. Faro, I. I. Shemyakina, N. G. Bozhanova, A. A. Kotlobay, N. G. Gurskaya, A. Boden, J. Dreier, D. M. Chudakov, K. A. Lukyanov, V. V. Verkhusha, A. S. Mishin and I. Testa, *Nat. Methods*, 2018, **15**, 601–604.
- 91 K. G. Chernov, T. A. Redchuk, E. S. Omelina and V. V. Verkhusha, *Chem. Rev.*, 2017, **117**, 6423–6446.
- 92 J. Yao, A. A. Kaberniuk, L. Li, D. M. Shcherbakova, R. Zhang, L. Wang, G. Li, V. V. Verkhusha and L. V. Wang, *Nat. Methods*, 2016, **13**, 67–73.
- 93 L. Li, A. A. Shemetov, M. Baloban, P. Hu, L. Zhu, D. M. Shcherbakova, R. Zhang, J. Shi, J. Yao, L. V. Wang and V. V. Verkhusha, *Nat. Commun.*, 2018, **9**, 2734.



- 94 J. Zhang, J. Wang and H. Tian, *Mater. Horiz.*, 2014, **1**, 169–184.
- 95 J.-L. Zhao, M.-H. Li, Y.-M. Cheng, X.-W. Zhao, Y. Xu, Z.-Y. Cao, M.-H. You and M.-J. Lin, *Coord. Chem. Rev.*, 2023, **475**, 214918.
- 96 P. Hong, N.-H. Xie, K. Xiong, J. Liu, M.-Q. Zhu and C. Li, *J. Mater. Chem. A*, 2023, **11**, 5703–5713.
- 97 A. A. Beharry, L. Wong, V. Tropepe and G. A. Woolley, *Angew. Chem., Int. Ed.*, 2011, **50**, 1325–1327.
- 98 K.-T. Chung, *J. Environ. Sci. Health, Part C*, 2016, **34**, 233–261.
- 99 P. Hong, J. Liu, K.-X. Qin, R. Tian, L.-Y. Peng, Y.-S. Su, Z. Gan, X.-X. Yu, L. Ye, M.-Q. Zhu and C. Li, *Angew. Chem., Int. Ed.*, 2024, **63**, e202316706.
- 100 Z. Zhang, W. Wang, P. Jin, J. Xue, L. Sun, J. Huang, J. Zhang and H. Tian, *Nat. Commun.*, 2019, **10**, 4232.
- 101 Y. He, Z. Shangguan, Z.-Y. Zhang, M. Xie, C. Yu and T. Li, *Angew. Chem., Int. Ed.*, 2021, **60**, 16539–16546.
- 102 K. Uno, M. L. Bossi, M. Irie, V. N. Belov and S. W. Hell, *J. Am. Chem. Soc.*, 2019, **141**, 16471–16478.
- 103 T. I. Lansakara, F. Tong, C. J. Bardeen and A. V. Tivanski, *Nano Lett.*, 2020, **20**, 6744–6749.
- 104 L. Zhang, S. Jeong, J. Lee, J. Kim, J. S. Lee, J. Park, J. Hong, J.-H. Eom, H. Kim, Y. M. Rhee, H. Lee and H.-S. Lee, *Small*, 2024, **20**, 2401480.
- 105 S. He, S. Schog, Y. Chen, Y. Ji, S. Panitz, W. Richtering and R. Göstl, *Adv. Mater.*, 2023, **35**, 2305845.
- 106 J. Kang, E. Li, L. Cui, Q. Shao, C. Yin and F. Cheng, *Sens. Actuators B Chem.*, 2021, **327**, 128941.
- 107 B. Roubinet, M. Weber, H. Shojaei, M. Bates, M. L. Bossi, V. N. Belov, M. Irie and S. W. Hell, *J. Am. Chem. Soc.*, 2017, **139**, 6611–6620.
- 108 K. Uno, A. Aktalay, M. L. Bossi, M. Irie, V. N. Belov and S. W. Hell, *Proc. Natl. Acad. Sci. U. S. A.*, 2021, **118**, e2100165118.
- 109 S. H. Um, H. J. Kim, D. Kim, J. E. Kwon, J. W. Lee, D. Hwang, S. K. Kim and S. Y. Park, *Dyes Pigm.*, 2018, **158**, 36–41.
- 110 S. Qi, X. Lu, W. Mei, G. Gu, W. Li and A. Zhang, *Nanoscale*, 2023, **15**, 18053–18067.
- 111 H. Kong, X. Xie, Y. Bao, F. Zhang, L. Bian, K. Cheng, Y.-D. Zhao and J. Xia, *Angew. Chem., Int. Ed.*, 2025, **64**, e202419538.
- 112 Y. Zou, T. Yi, S. Xiao, F. Li, C. Li, X. Gao, J. Wu, M. Yu and C. Huang, *J. Am. Chem. Soc.*, 2008, **130**, 15750–15751.
- 113 S. Qiu, A. T. Frawley, K. G. Leslie and H. L. Anderson, *Chem. Sci.*, 2023, **14**, 9123–9135.
- 114 J. Chen-Wu, D. B. Guzmán-Ríos, P. Remón, J. A. González-Delgado, A. J. Martínez-Martínez, F. Nájera, J. F. Arteaga and U. Pischel, *Adv. Mater.*, 2023, **35**, 2300536.
- 115 S. Song, H. Zhang and Y. Liu, *Acc. Mater. Res.*, 2024, **5**, 1109–1120.
- 116 W.-L. Guan, J.-F. Chen, J. Liu, B. Shi, H. Yao, Y.-M. Zhang, T.-B. Wei and Q. Lin, *Coord. Chem. Rev.*, 2024, **507**, 215717.
- 117 X. Y. Dai, Y. Y. Hu, Y. Sun, M. Huo, X. Dong and Y. Liu, *Adv. Sci.*, 2022, e2200524.
- 118 X. Y. Dai, Y. Y. Hu, Y. Sun, M. Huo, X. Dong and Y. Liu, *Adv. Sci.*, 2022, e2200524.
- 119 S. Wan, Z. Ma, C. Chen, F. Li, F. Wang, X. Jia, W. Yang and M. Yin, *Adv. Funct. Mater.*, 2016, **26**, 353–364.
- 120 T. Takeshita, A. Yano and M. Hara, *ChemistrySelect*, 2017, **2**, 11288–11292.
- 121 D. Samanta, D. Galaktionova, J. Gemen, L. J. W. Shimon, Y. Diskin-Posner, L. Avram, P. Král and R. Klajn, *Nat. Commun.*, 2018, **9**, 641.
- 122 J.-X. Wang, C. Li and H. Tian, *Coord. Chem. Rev.*, 2021, **427**, 213579.
- 123 L. Li, Y.-T. Yu, N.-N. Zhang, S.-H. Li, J.-G. Zeng, Y. Hua and H. Zhang, *Coord. Chem. Rev.*, 2024, **500**, 215526.
- 124 K. Hakouk, O. Oms, A. Dolbecq, J. Marrot, A. Saad, P. Mialane, H. El Bekkachi, S. Jobic, P. Deniard and R. Dessapt, *J. Mater. Chem. C*, 2014, **2**, 1628–1641.
- 125 C.-C. Ko and V. W.-W. Yam, *Acc. Chem. Res.*, 2018, **51**, 149–159.
- 126 S. Samanta, A. A. Beharry, O. Sadovski, T. M. McCormick, A. Babalhavaeji, V. Tropepe and G. A. Woolley, *J. Am. Chem. Soc.*, 2013, **135**, 9777–9784.
- 127 Z. Li, X. Zeng, C. Gao, J. Song, F. He, T. He, H. Guo and J. Yin, *Coord. Chem. Rev.*, 2023, **497**, 215451.
- 128 S. Lin, K. G. Gutierrez-Cuevas, X. Zhang, J. Guo and Q. Li, *Adv. Funct. Mater.*, 2020, **31**, 2007957.
- 129 F. Li, M. Li, Y. Shi, X. Bian, N. Lv, S. Guo, Y. Wang, W. Zhao and W.-H. Zhu, *Chem. Sci.*, 2025, **16**, 14270–14277.
- 130 M. Dong, A. Babalhavaeji, S. Samanta, A. A. Beharry and G. A. Woolley, *Acc. Chem. Res.*, 2015, **48**, 2662–2670.
- 131 C. Li, K. Xiong, Y. Chen, C. Fan, Y. L. Wang, H. Ye and M. Q. Zhu, *ACS Appl. Mater. Interfaces*, 2020, **12**, 27651–27662.
- 132 S. Samanta, A. Babalhavaeji, M.-X. Dong and G. A. Woolley, *Angew. Chem., Int. Ed.*, 2013, **52**, 14127–14130.
- 133 M. Dong, A. Babalhavaeji, C. V. Collins, K. Jarrah, O. Sadovski, Q. Dai and G. A. Woolley, *J. Am. Chem. Soc.*, 2017, **139**, 13483–13486.
- 134 D. Wu, M. Dong, C. V. Collins, A. Babalhavaeji and G. A. Woolley, *Adv. Opt. Mater.*, 2016, **4**, 1402–1409.
- 135 A. A. Beharry and G. A. Woolley, *Chem. Soc. Rev.*, 2011, **40**, 4422–4437.
- 136 S. Aiken, R. J. L. Edgar, C. D. Gabbutt, B. M. Heron and P. A. Hobson, *Dyes Pigm.*, 2018, **149**, 92–121.
- 137 V. A. Barachevsky, *Rev. J. Chem.*, 2017, **7**, 334–371.
- 138 M. M. Lerch, W. Szymański and B. L. Feringa, *Chem. Soc. Rev.*, 2018, **47**, 1910–1937.
- 139 C. A. Reyes, H. J. Lee, C. Karanovic and E. Picazo, *Nat. Commun.*, 2024, **15**, 5533.
- 140 T. Senthilkumar, L. Zhou, Q. Gu, L. Liu, F. Lv and S. Wang, *Angew. Chem., Int. Ed.*, 2018, **57**, 13114–13119.
- 141 Z.-X. Zhang, F.-Q. Bai, L. Li and H.-X. Zhang, *New J. Chem.*, 2015, **39**, 1634–1642.
- 142 R. Li, B. Mou, M. Yamada, W. Li, T. Nakashima and T. Kawai, *Molecules*, 2024, **29**, 2007957.
- 143 T. Fukaminato, M. Tanaka, L. Kuroki and M. Irie, *Chem. Commun.*, 2008, 3924–3926.
- 144 T. Yang, Q. Liu, J. Li, S. Pu, P. Yang and F. Li, *RSC Adv.*, 2014, **4**, 15613–15619.



- 145 V. Marturano, J. Kozłowska, A. Bajek, M. Giamberini, V. Ambrogio, P. Cerruti, R. Garcia-Valls, J. M. Montornes and B. Tylkowski, *Coord. Chem. Rev.*, 2019, **398**, 213013.
- 146 J. Kim, S. Lee, Y. Jeong, K. Kim, K. Nam, H. Jin, Y. Choi, H.-J. Kim, H. Ryu, K. H. Kim, J.-I. Kim, J. Park, J. Joo and J.-H. Park, *Adv. Mater.*, 2025, **37**, 2502739.
- 147 X. Zhu, Q. Su, W. Feng and F. Li, *Chem. Soc. Rev.*, 2017, **46**, 1025–1039.
- 148 Y. Mi, H. B. Cheng, H. Chu, J. Zhao, M. Yu, Z. Gu, Y. Zhao and L. Li, *Chem. Sci.*, 2019, **10**, 10231–10239.
- 149 M. Zhao, B. Li, P. Wang, L. Lu, Z. Zhang, L. Liu, S. Wang, D. Li, R. Wang and F. Zhang, *Adv. Mater.*, 2018, **30**, e1804982.
- 150 C. Benitez-Martin, S. Li, A. Dominguez-Alfaro, F. Najera, E. Pérez-Inestrosa, U. Pischel and J. Andréasson, *J. Am. Chem. Soc.*, 2020, **142**, 14854–14858.
- 151 G. S. He, L.-S. Tan, Q. Zheng and P. N. Prasad, *Chem. Rev.*, 2008, **108**, 1245–1330.
- 152 F. Helmchen and W. Denk, *Nat. Methods*, 2005, **2**, 932–940.
- 153 M.-Q. Zhu, G.-F. Zhang, C. Li, M. P. Aldred, E. Chang, R. A. Drezek and A. D. Q. Li, *J. Am. Chem. Soc.*, 2011, **133**, 365–372.
- 154 K. Mutoh, K. Yamamoto and J. Abe, *Photoch. Photobio. Sci.*, 2022, **21**, 1445–1458.
- 155 L. Lin, Z. Zhang, Z. Lu, Y. Guo and M. Liu, *J. Phys. Chem. A*, 2016, **120**, 7859–7864.
- 156 J. Moreno, M. Gerecke, L. Grubert, S. A. Kovalenko and S. Hecht, *Angew. Chem., Int. Ed.*, 2016, **55**, 1544–1547.
- 157 K. Uno, H. Niikura, M. Morimoto, Y. Ishibashi, H. Miyasaka and M. Irie, *J. Am. Chem. Soc.*, 2011, **133**, 13558–13564.
- 158 X. Hu, B. Fang, P. Li, J. Wang, J. Li, Y. Dai and M. Yin, *Adv. Funct. Mater.*, 2025, **35**, 2423793.
- 159 J. He, H. Zhao, H. Wu, Y. Yang, Z. Wang, Z. He and G. Jiang, *Phys. Chem. Chem. Phys.*, 2021, **23**, 17939–17944.
- 160 Z. Xiong, X. Zhang, L. Liu, Q. Zhu, Z. Wang, H. Feng and Z. Qian, *Chem. Sci.*, 2021, **12**, 10710–10723.
- 161 Y. Su, X. Li, D. Zheng, J. Andréasson, H. Wang, L. Yu, J. Chen, J. Ma and Y. Fang, *Chem. Sci.*, 2025, **16**, 23121–23128.
- 162 A. T. Muhammed Munthasir, P. Rani, P. Dhanalakshmi, S. Geremia, N. Hickey and P. Thilagar, *Inorg. Chem.*, 2025, **64**, 5878–5892.
- 163 H. Yang, M. Li, C. Li, Q. Luo, M. Q. Zhu, H. Tian and W. H. Zhu, *Angew. Chem., Int. Ed.*, 2020, **59**, 8560–8570.
- 164 C. Wang, Y. H. Liu and Y. Liu, *Small*, 2022, e2201821.
- 165 R. Zhang, Z. Lei, Z. Yu, Y. Chen and Y. Liu, *Adv. Sci.*, 2025, **12**, e07090.
- 166 L. Wu, C. Huang, B. P. Emery, A. C. Sedgwick, S. D. Bull, X.-P. He, H. Tian, J. Yoon, J. L. Sessler and T. D. James, *Chem. Soc. Rev.*, 2020, **49**, 5110–5139.
- 167 N. Asadi-Zaki, H. Mardani, H. Roghani-Mamaqani and F. Wang, *Coord. Chem. Rev.*, 2024, **500**, 215518.
- 168 Y. Xiong, P. Rivera-Fuentes, E. Sezgin, A. Vargas Jentzsch, C. Eggeling and H. L. Anderson, *Org. Lett.*, 2016, **18**, 3666–3669.
- 169 Z.-M. Yue, S.-M. Zhai, B.-Y. Wei, B.-X. Zhao and Z.-M. Lin, *Dyes Pigm.*, 2025, **233**, 112544.
- 170 J. Wu, L. Kreimendahl and J. L. Greenfield, *J. Am. Chem. Soc.*, 2025, **147**, 17549–17554.
- 171 C. Slavov, C. Yang, A. H. Heindl, H. A. Wegner, A. Dreuw and J. Wachtveitl, *Angew. Chem., Int. Ed.*, 2020, **59**, 380–387.
- 172 R. Yang, Y. Jiao, B. Wang, B. Xu and W. Tian, *J. Phys. Chem. Lett.*, 2021, **12**, 1290–1294.
- 173 S. Dutta Choudhury, *Langmuir*, 2022, **38**, 14819–14826.
- 174 G. Das, T. Prakasam, N. Alkhatib, R. G. AbdulHalim, F. Chandra, S. K. Sharma, B. Garai, S. Varghese, M. A. Addicoat, F. Ravaux, R. Pasricha, R. Jagannathan, N. I. Saleh, S. Kirmizialtin, M. A. Olson and A. Trabolsi, *Nat. Commun.*, 2023, **14**, 3765.
- 175 Z. Ye, Z. Yang, L. Wang, L. Chen, Y. Cai, P. Deng, W. Feng, X. Li and L. Yuan, *Angew. Chem., Int. Ed.*, 2019, **58**, 12519–12523.
- 176 V. Dryza and E. J. Bieske, *J. Phys. Chem. C*, 2015, **119**, 14076–14084.
- 177 T. Pan, D. Lu, H. Xin and B. Li, *Light Sci. Appl.*, 2021, **10**, 124.
- 178 H. Xu, D. Kim, Y.-Y. Zhao, C. Kim, G. Song, Q. Hu, H. Kang and J. Yoon, *Adv. Mater.*, 2024, **36**, 2402806.
- 179 J. Huang, J. Li, Y. Lyu, Q. Miao and K. Pu, *Nat. Mater.*, 2019, **18**, 1133–1143.
- 180 J. Du, S. Yang, Y. Qiao, H. Lu and H. Dong, *Biosens. Bioelectron.*, 2021, **191**, 113478.
- 181 D. Hu, Z. Tian, W. Wu, W. Wan and A. D. Q. Li, *J. Am. Chem. Soc.*, 2008, **130**, 15279–15281.
- 182 L. Z. Cai, Q. S. Chen, C. J. Zhang, P. X. Li, M. S. Wang and G. C. Guo, *J. Am. Chem. Soc.*, 2015, **137**, 10882–10885.
- 183 S. Wang, W. X. Ren, J. T. Hou, M. Won, J. An, X. Chen, J. Shu and J. S. Kim, *Chem. Soc. Rev.*, 2021, **50**, 8887–8902.
- 184 H. Li and J. C. Vaughan, *Chem. Rev.*, 2018, **118**, 9412–9454.
- 185 M. Koch and V. Ntziachristos, *Annu. Rev. Med.*, 2016, **67**, 153–164.
- 186 B. Huang, M. Bates and X. Zhuang, *Annu. Rev. Biochem.*, 2009, **78**, 993–1016.
- 187 Z. Yang, A. Sharma, J. Qi, X. Peng, D. Y. Lee, R. Hu, D. Lin, J. Qu and J. S. Kim, *Chem. Soc. Rev.*, 2016, **45**, 4651–4667.
- 188 Y. Xu, R. Xu, Z. Wang, Y. Zhou, Q. Shen, W. Ji, D. Dang, L. Meng and B. Z. Tang, *Chem. Soc. Rev.*, 2021, **50**, 667–690.
- 189 J. Kwon, M. S. Elgawish and S. H. Shim, *Adv. Sci.*, 2022, **9**, e2101817.
- 190 H. Chang, M. Zhang, W. Ji, J. Chen, Y. Zhang, B. Liu, J. Lu, J. Zhang, P. Xu and T. Xu, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 4455–4460.
- 191 M. Minoshima and K. Kikuchi, *J. Biol. Inorg. Chem.*, 2017, **22**, 639–652.
- 192 B. Wang, M. Xiong, J. Susanto, X. Li, W.-Y. Leung and K. Xu, *Angew. Chem., Int. Ed.*, 2022, **61**, e202113612.
- 193 Q. Qi, W. Chi, Y. Li, Q. Qiao, J. Chen, L. Miao, Y. Zhang, J. Li, W. Ji, T. Xu, X. Liu, J. Yoon and Z. Xu, *Chem. Sci.*, 2019, **10**, 4914–4922.
- 194 H.-I. D. Lee, S. J. Lord, S. Iwanaga, K. Zhan, H. Xie, J. C. Williams, H. Wang, G. R. Bowman, E. D. Goley,



- L. Shapiro, R. J. Twieg, J. Rao and W. E. Moerner, *J. Am. Chem. Soc.*, 2010, **132**, 15099–15101.
- 195 Q. Qi, C. Li, X. Liu, S. Jiang, Z. Xu, R. Lee, M. Zhu, B. Xu and W. Tian, *J. Am. Chem. Soc.*, 2017, **139**, 16036–16039.
- 196 A. Bodén, D. Ollech, A. G. York, A. Millett-Sikking and I. Testa, *Nat. Methods*, 2024, **21**, 882–888.
- 197 S. Wang, X. Chen, L. Chang, R. Xue, H. Duan and Y. Sun, *ACS Nano*, 2016, **10**, 9136–9144.
- 198 M.-Q. Zhu, G.-F. Zhang, Z. Hu, M. P. Aldred, C. Li, W.-L. Gong, T. Chen, Z.-L. Huang and S. Liu, *Macromolecules*, 2014, **47**, 1543–1552.
- 199 Y. Kim, J. Gonzales and Y. Zheng, *Small*, 2021, **17**, e2004988.
- 200 C. Xu and K. Pu, *Nat. Rev. Bioeng.*, 2024, **2**, 425–441.
- 201 M. Li, J. Zhao, H. Chu, Y. Mi, Z. Zhou, Z. Di, M. Zhao and L. Li, *Adv. Mater.*, 2019, **31**, 1804745.
- 202 D. Yi, J. Zhao and L. Li, *Angew. Chem., Int. Ed.*, 2021, **60**, 6300–6304.
- 203 W. Yu, H. Jin, C. Tang, J. Du and Z. Zhang, *Br. J. Pharmacol.*, 2018, **175**, 1114–1125.
- 204 H. Zhao, G. Patterson and P. Schuck, *Biophys. J.*, 2016, **110**, 347a.
- 205 W. N. Harrington, M. V. Novoselova, D. N. Bratashov, B. N. Khlebtsov, D. A. Gorin, E. I. Galanzha and V. P. Zharov, *Sci. Rep.*, 2019, **9**, 12439.
- 206 Z. Qiao, H. Qi, H. Zhang, Q. Zhou, N. Wei, Y. Zhang and K. Wang, *Anal. Chem.*, 2020, **92**, 1934–1939.
- 207 N. Wang, J. Wang, S. Huang, C. Zheng, S. Cui and S. Pu, *Anal. Chim. Acta*, 2026, **1382**, 344848.
- 208 Y. Wu, Z. Gong, X. Wu, Y. Huang, L. Sun, H. Ding, Y.-L. Zeng, C. Fan, G. Liu and S. Pu, *Anal. Chim. Acta*, 2025, **1359**, 344114.
- 209 W. Zhang, F. Huo and C. Yin, *Org. Lett.*, 2019, **21**, 5277–5280.
- 210 W. Zhang, F. Huo, Y. Yue, Y. Zhang, J. Chao, F. Cheng and C. Yin, *J. Am. Chem. Soc.*, 2020, **142**, 3262–3268.
- 211 Y. Hong, P. Zhang, H. Wang, M. Yu, Y. Gao and J. Chen, *Sens. Actuators, B*, 2018, **272**, 340–347.
- 212 Y. Fu, X. Zhang, J. Liu, G. Qian, Z. P. Xu and R. Zhang, *J. Mater. Chem. B*, 2022, **10**, 3366–3374.
- 213 X. Xie, G. Mistlberger and E. Bakker, *J. Am. Chem. Soc.*, 2012, **134**, 16929–16932.
- 214 X. Chai, H. H. Han, Y. Zang, J. Li, X. P. He, J. Zhang and H. Tian, *Beilstein J. Org. Chem.*, 2019, **15**, 2380–2389.
- 215 Y. Fu, H. H. Han, J. Zhang, X. P. He, B. L. Feringa and H. Tian, *J. Am. Chem. Soc.*, 2018, **140**, 8671–8674.
- 216 J. Zhang, Y. Fu, H. H. Han, Y. Zang, J. Li, X. P. He, B. L. Feringa and H. Tian, *Nat. Commun.*, 2017, **8**, 987.
- 217 M. Yu, W. Zhao, F. Ni, Q. Zhao and C. Yang, *Adv. Opt. Mater.*, 2022, **10**, 2102437.
- 218 F. Bierbuesse, A. C. Bourges, V. Gielen, V. Mönkemöller, W. Vandenberg, Y. Shen, J. Hofkens, P. Vanden Berghe, R. E. Campbell, B. Moeyaert and P. Dedecker, *Nat. Commun.*, 2022, **13**, 1850.
- 219 L. Liu, S. Wang, C.-H. Zhang, J. Jiang, Y. Gao, P. Zhang, R. Zeng and J. Chen, *Dyes Pigm.*, 2021, **191**, 109370.
- 220 M. Bates, B. Huang, G. T. Dempsey and X. Zhuang, *Science*, 2007, **317**, 1749–1753.
- 221 M. Deng, Y. Zhang, Y. Wang and S. Jiang, *J. Mater. Chem. C*, 2020, **8**, 15697–15704.
- 222 W. Zhong, X. Zeng, J. Chen, Y. Hong, L. Xiao and P. Zhang, *Polym. Chem.*, 2017, **8**, 4849–4855.
- 223 Q. Yan, Z. Qiao, W. Zhao, J. Ren, Y. Wang, W. Yang and S. Wang, *J. Mater. Chem. C*, 2022, **10**, 7024–7030.
- 224 J. Chen, W. Zhong, Y. Tang, Z. Wu, Y. Li, P. Yi and J. Jiang, *Macromolecules*, 2015, **48**, 3500–3508.
- 225 R. Yang, X. Ren, L. Mei, G. Pan, X.-Z. Li, Z. Wu, S. Zhang, W. Ma, W. Yu, H.-H. Fang, C. Li, M.-Q. Zhu, Z. Hu, T. Sun, B. Xu and W. Tian, *Angew. Chem., Int. Ed.*, 2022, **61**, e202117158.
- 226 L. Wei, Z. Chen, L. Shi, R. Long, A. V. Anzalone, L. Zhang, F. Hu, R. Yuste, V. W. Cornish and W. Min, *Nature*, 2017, **544**, 465–470.
- 227 R. Chouket, A. Pellissier-Tanon, A. Lemarchand, A. Espagne, T. Le Saux and L. Jullien, *Chem. Sci.*, 2020, **11**, 2882–2887.
- 228 T. Niehörster, A. Löscherger, I. Gregor, B. Krämer, H.-J. Rahn, M. Patting, F. Koberling, J. Enderlein and M. Sauer, *Nat. Methods*, 2016, **13**, 257–262.
- 229 A. M. Valm, S. Cohen, W. R. Legant, J. Melunis, U. Hershberg, E. Wait, A. R. Cohen, M. W. Davidson, E. Betzig and J. Lippincott-Schwartz, *Nature*, 2017, **546**, 162–167.
- 230 G. Marriott, S. Mao, T. Sakata, J. Ran, D. K. Jackson, C. Petchprayoon, T. J. Gomez, E. Warp, O. Tulyathan, H. L. Aaron, E. Y. Isacoff and Y. Yan, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 17789–17794.
- 231 J. Querard, T.-Z. Markus, M.-A. Plamont, C. Gauron, P. Wang, A. Espagne, M. Volovitch, S. Vriz, V. Croquette, A. Gautier, T. Le Saux and L. Jullien, *Angew. Chem., Int. Ed.*, 2015, **54**, 2633–2637.
- 232 J.-C. Hsiang, A. E. Jablonski and R. M. Dickson, *Acc. Chem. Res.*, 2014, **47**, 1545–1554.
- 233 Y.-C. Chen, A. E. Jablonski, I. Issaeva, D. Bourassa, J.-C. Hsiang, C. J. Fahrni and R. M. Dickson, *J. Am. Chem. Soc.*, 2015, **137**, 12764–12767.
- 234 J. Quérard, R. Zhang, Z. Kelemen, M.-A. Plamont, X. Xie, R. Chouket, I. Roemgens, Y. Korepina, S. Albright, E. Ipendey, M. Volovitch, H. L. Sladitschek, P. Neveu, L. Gissot, A. Gautier, J.-D. Faure, V. Croquette, T. Le Saux and L. Jullien, *Nat. Commun.*, 2017, **8**, 969.
- 235 R. Zhang, R. Chouket, M.-A. Plamont, Z. Kelemen, A. Espagne, A. G. Tebo, A. Gautier, L. Gissot, J.-D. Faure, L. Jullien, V. Croquette and T. Le Saux, *Light Sci. Appl.*, 2018, **7**, 97.
- 236 R. Zhang, R. Chouket, A. Tebo, M.-A. Plamont, Z. Kelemen, L. Gissot, J.-D. Faure, A. Gautier, V. Croquette, L. Jullien and T. Saux, *Optica*, 2019, **6**, 972–980.
- 237 H. Valenta, S. Hugelier, S. Duwé, G. Lo Gerfo, M. Müller, P. Dedecker and W. Vandenberg, *Biophys. Rep.*, 2021, **1**, 100026.
- 238 H. Valenta, F. Bierbuesse, R. Vitale, C. Ruckebusch, W. Vandenberg and P. Dedecker, *Talanta*, 2024, **269**, 125397.
- 239 R. Chouket, A. Pellissier-Tanon, A. Lahlou, R. Zhang, D. Kim, M.-A. Plamont, M. Zhang, X. Zhang, P. Xu,



- N. Desprat, D. Bourgeois, A. Espagne, A. Lemarchand, T. L. Saux and L. Jullien, *Nat. Commun.*, 2022, **13**, 1482.
- 240 T. Roebroek, W. Vandenberg, F. Sipieter, S. Hugelier, C. Stove, J. Zhang and P. Dedecker, *Nat. Commun.*, 2021, **12**, 2005.
- 241 B. Chang, D. Li, Y. Ren, C. Qu, X. Shi, R. Liu, H. Liu, J. Tian, Z. Hu, T. Sun and Z. Cheng, *Nat. Biomed. Eng.*, 2022, **6**, 629–639.
- 242 W. Zhao, Z. He and B. Z. Tang, *Nat. Rev. Mater.*, 2020, **5**, 869–885.
- 243 J. Huang, L. Su, C. Xu, X. Ge, R. Zhang, J. Song and K. Pu, *Nat. Mater.*, 2023, **22**, 1421–1429.
- 244 C. Xu, J. Huang, Y. Jiang, S. He, C. Zhang and K. Pu, *Nat. Biomed. Eng.*, 2023, **7**, 298–312.
- 245 L. Yang, M. Zhao, W. Chen, J. Zhu, W. Xu, Q. Li, K. Pu and Q. Miao, *Angew. Chem., Int. Ed.*, 2024, **63**, e202313117.
- 246 F. Xiao, H. Gao, Y. Lei, W. Dai, M. Liu, X. Zheng, Z. Cai, X. Huang, H. Wu and D. Ding, *Nat. Commun.*, 2022, **13**, 186.
- 247 X.-Y. Dai, M. Huo and Y. Liu, *Nat. Rev. Chem.*, 2023, **7**, 854–874.
- 248 Y.-Y. Hu, X.-Y. Dai, X. Dong, M. Huo and Y. Liu, *Angew. Chem., Int. Ed.*, 2022, **61**, e202213097.
- 249 Y. Zhang, C. Zhang, Y. Chen, J. Yu, L. Chen, H. Zhang, X. Xu and Y. Liu, *Adv. Opt. Mater.*, 2022, **10**.
- 250 H.-J. Wang, W.-W. Xing, H.-Y. Zhang, W.-W. Xu and Y. Liu, *Adv. Opt. Mater.*, 2022, **10**, 2201178.
- 251 X. L. Dean-Ben, S. Gottschalk, B. Mc Larney, S. Shoham and D. Razansky, *Chem. Soc. Rev.*, 2017, **46**, 2158–2198.
- 252 C. Liu, X. Gong, R. Lin, F. Liu, J. Chen, Z. Wang, L. Song and J. Chu, *Theranostics*, 2016, **6**, 2414–2430.
- 253 A. B. E. Attia, G. Balasundaram, M. Moothanchery, U. S. Dinish, R. Bi, V. Ntziachristos and M. Olivo, *Photoacoustics*, 2019, **16**, 100144.
- 254 H. B. Cheng, Y. Li, B. Z. Tang and J. Yoon, *Chem. Soc. Rev.*, 2020, **49**, 21–31.
- 255 L. A. Kasatkina, C. Ma, M. E. Matlashov, T. Vu, M. Li, A. A. Kaberniuk, J. Yao and V. V. Verkhusha, *Nat. Commun.*, 2022, **13**, 2813.
- 256 S. Chen, K. Li, X. Chen, S. Lei, J. Lin and P. Huang, *Sci. Adv.*, 2024, **10**, eadn8274.
- 257 K. Mishra, M. Stankevych, P. Fuenzalida-Werner Juan, S. Grassmann, V. Gujrati, Y. Huang, U. Klemm, R. Buchholz Veit, V. Ntziachristos and C. Stiel Andre, *Sci. Adv.*, 2020, **6**, eaaz6293.
- 258 R. Gao, F. Liu, W. Liu, S. Zeng, J. Chen, R. Gao, L. Wang, C. Fang, L. Song, A. C. Sedgwick, J. L. Sessler, J. Chu, F. Yan and C. Liu, *Proc. Natl. Acad. Sci. U. S. A.*, 2022, **119**, e2121982119.
- 259 C. Liu, X. Zheng, T. Dai, H. Wang, X. Chen, B. Chen, T. Sun, F. Wang, S. Chu and J. Rao, *Angew. Chem., Int. Ed.*, 2022, **61**, e202116802.
- 260 S. He, J. Song, J. Liu, L. Liu, J. Qu and Z. Cheng, *Adv. Opt. Mater.*, 2019, **7**, 1900045.
- 261 H. M. D. Bandara and S. C. Burdette, *Chem. Soc. Rev.*, 2012, **41**, 1809–1825.
- 262 J. Garcia-Amorós, B. Maerz, M. Reig, A. Cuadrado, L. Blancafort, E. Samoylova and D. Velasco, *Chem. Eur. J.*, 2019, **25**, 7726–7732.
- 263 J. Levi, S. R. Kothapalli, T.-J. Ma, K. Hartman, B. T. Khuri-Yakub and S. S. Gambhir, *J. Am. Chem. Soc.*, 2010, **132**, 11264–11269.
- 264 M. Weiger and K. P. Pruessmann, *Prog. Nucl. Magn. Reson. Spectrosc.*, 2019, **114–115**, 237–270.
- 265 C. Fu, Y. Yu, X. Xu, Q. Wang, Y. Chang, C. Zhang, J. Zhao, H. Peng and A. K. Whittaker, *Prog. Polym. Sci.*, 2020, **108**, 101286.
- 266 D. Hao, T. Ai, F. Goerner, X. Hu, V. M. Runge and M. Tweedle, *Magn. Reson. Imaging*, 2012, **36**, 1060–1071.
- 267 J. Wahsner, E. M. Gale, A. Rodriguez-Rodriguez and P. Caravan, *Chem. Rev.*, 2019, **119**, 957–1057.
- 268 I. M. Welleman, C. L. F. van Beek, I. Belcin, A. M. Schulte, R. A. J. O. Dierckx, B. L. Feringa, H. H. Boersma and W. Szymański, *Smart Mol.*, 2024, **2**, e20230029.
- 269 C. Tu and A. Y. Louie, *Chem. Commun.*, 2007, 1331–1333, DOI: [10.1039/b616991k](https://doi.org/10.1039/b616991k).
- 270 C. Tu, E. A. Osborne and A. Y. Louie, *Tetrahedron*, 2009, **65**, 1241.
- 271 K. Kruttwig, D. R. Yankelevich, C. Brueggemann, C. Tu, N. L'Etoile, A. Knoesen and A. Y. Louie, *Molecules*, 2012, **17**, 6605–6624.
- 272 M. Gao, B. Shen, J. Zhou, R. Kapre, A. Y. Louie and J. T. Shaw, *ACS Omega*, 2020, **5**, 14759–14766.
- 273 B. Shen, M. Gao, F. C. Franco, Jr., R. Kapre, J. Zhou, X. Li, J. Garcia, J. T. Shaw and A. Y. Louie, *J. Org. Chem.*, 2020, **85**, 7333–7341.
- 274 S. Thies, H. Sell, C. Schutt, C. Bornholdt, C. Nather, F. Tuzek and R. Herges, *J. Am. Chem. Soc.*, 2011, **133**, 16243–16250.
- 275 G. Heitmann, C. Schütt, J. Gröbner, L. Huber and R. Herges, *Dalton Trans.*, 2016, **45**, 11407–11412.
- 276 V. Wellm, J. Groebner, G. Heitmann, F. D. Sonnichsen and R. Herges, *Angew. Chem., Int. Ed.*, 2021, **60**, 8220–8226.
- 277 V. Wellm, C. Nather and R. Herges, *J. Org. Chem.*, 2021, **86**, 9503–9514.
- 278 Y. Heta, K. Kumaki, H. Hifumi, D. Citterio, A. Tanimoto and K. Suzuki, *Photochem. Photobiol.*, 2012, **88**, 876–883.
- 279 E. A. Osborne, B. R. Jarrett, C. Tu and A. Y. Louie, *J. Am. Chem. Soc.*, 2010, **132**, 5934–5935.
- 280 Q. Jin, X. Fan, C. Chen, L. Huang, J. Wang and X. Tang, *Anal. Chem.*, 2019, **91**, 3784–3789.
- 281 S. Stewart, R. J. Priore, M. P. Nelson and P. J. Treado, *Annu. Rev. Anal. Chem.*, 2012, **5**, 337–360.
- 282 R. C. Prince, R. R. Frontiera and E. O. Potma, *Chem. Rev.*, 2017, **117**, 5070–5094.
- 283 Y. Tang, X. Chen, S. Zhang, Z. J. Smith and T. Gao, *Anal. Chem.*, 2021, **93**, 15659–15666.
- 284 D. Lee, C. Qian, H. Wang, L. Li, K. Miao, J. Du, D. M. Shcherbakova, V. V. Verkhusha, L. V. Wang and L. Wei, *J. Chem. Phys.*, 2021, **154**, 135102.
- 285 J. Shou, A. Komazawa, Y. Wachi, M. Kawatani, H. Fujioka, S. J. Spratt, T. Mizuguchi, K. Oguchi, H. Akaboshi,



- F. Obata, R. Tachibana, S. Yasunaga, Y. Mita, Y. Misawa, R. Kojima, Y. Urano, M. Kamiya and Y. Ozeki, *Sci. Adv.*, 2023, **9**, eade9118.
- 286 Y. Yang, X. Bai and F. Hu, *Nat. Commun.*, 2024, **15**, 2578.
- 287 Y. Tang, X. Chen, S. Zhang, Z. J. Smith and T. Gao, *Anal. Chem.*, 2021, **93**, 15659–15666.
- 288 Y. Tang, Y. Zhuang, S. Zhang, Z. J. Smith, Y. Li, X. Mu, M. Li, C. He, X. Zheng, F. Pan, T. Gao and L. Zhang, *ACS Cent. Sci.*, 2021, **7**, 768–780.
- 289 Y. Shen, G. Xue, Y. Dai, S. M. Quintero, H. Chen, D. Wang, F. Miao, F. Negri, Y. Zheng and J. Casado, *Nat. Commun.*, 2021, **12**, 6262.
- 290 A. A. Ali, R. Kharbush and Y. Kim, *Anal. Chim. Acta*, 2020, **1110**, 199–223.
- 291 M. Du and C. Li, *Adv. Mater.*, 2024, **36**, 2408484.
- 292 H.-W. Chien, C.-H. Yang, M.-T. Tsai and T.-L. Wang, *J. Photochem. Photobiol., A*, 2020, **392**, 112303.
- 293 R. S. Gormal, P. Padmanabhan, R. Kasula, A. T. Bademosi, S. Coakley, J. Giacomotto, A. Blum, M. Joensuu, T. P. Wallis, H. P. Lo, S. Budnar, J. Rae, C. Ferguson, M. Bastiani, W. G. Thomas, E. Pardon, J. Steyaert, A. S. Yap, G. J. Goodhill, M. A. Hilliard, R. G. Parton and F. A. Meunier, *Proc. Natl. Acad. Sci. U. S. A.*, 2020, **117**, 30476–30487.
- 294 F. Pennacchietti, J. Alvelid, R. A. Morales, M. Damenti, D. Ollech, O. S. Oliinyk, D. M. Shcherbakova, E. J. Villablanca, V. V. Verkhusha and I. Testa, *Nat. Commun.*, 2023, **14**, 8402.
- 295 K. H. Rainey and G. H. Patterson, *Proc. Natl. Acad. Sci. U. S. A.*, 2019, **116**, 864–873.
- 296 J. Lee, S. Lai, S. Yang, S. Zhao, F. A. Blanco, A. C. Lyons, R. Merino-Urteaga, J. F. Ahrens, N. A. Nguyen, H. Liu, Z. Liu, G. G. Lambert, N. C. Shaner, L. Chen, K. F. Tolia, J. Zhang, T. Ha and F. St-Pierre, *Nat. Commun.*, 2025, **16**, 3241.
- 297 V. Adam, K. Hadjidemetriou, N. Jensen, R. L. Shoeman, J. Woodhouse, A. Aquila, A.-S. Banneville, T. R. M. Barends, V. Bezchastnov, S. Boutet, M. Byrdin, M. Cammarata, S. Carbajo, N. Eleni Christou, N. Coquelle, E. De la Mora, M. El Khatib, T. Moreno Chicano, R. Bruce Doak, F. Fieschi, L. Foucar, O. Glushonkov, A. Gorel, M. L. Grünbein, M. Hilpert, M. Hunter, M. Kloos, J. E. Koglin, T. J. Lane, M. Liang, A. Mantovanelli, K. Nass, G. Nass Kovacs, S. Owada, C. M. Roome, G. Schirò, M. Seaberg, M. Stricker, M. Thépaut, K. Tono, K. Ueda, L. M. Uriarte, D. You, N. Zala, T. Domratcheva, S. Jakobs, M. Sliwa, I. Schlichting, J.-P. Colletier, D. Bourgeois and M. Weik, *ChemPhysChem*, 2022, **23**, e202200192.
- 298 P. Vetschera, K. Mishra, J. P. Fuenzalida-Werner, A. Chmyrov, V. Ntziachristos and A. C. Stiel, *Anal. Chem.*, 2018, **90**, 10527–10535.
- 299 Q. Qi, Y. Liu, V. Puranik, S. Patra, Z. Svindrych, X. Gong, Z. She, Y. Zhang and I. Aprahamian, *J. Am. Chem. Soc.*, 2025, **147**, 16404–16411.
- 300 G.-e Go, U. Jeong, H. Park, S. Go and D. Kim, *Angew. Chem., Int. Ed.*, 2024, **63**, e202405246.

