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Combining classical reactive scattering and the time–energy uncertainty relation

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Gaussian binning is a method for analyzing the results of classical dynamical simulations of gas-phase and gas-surface reactions that has been used since the early 2000s in order to improve predictions of state-resolved cross sections and related quantities measured in molecular beam experiments. In this method, classical trajectories are assigned Gaussian statistical weights, with much higher weights given to product energies closer to their quantized values. Here, we extend this idea to the activated complex in two ways: (i) treating it as if it were in a stationary state, just like the products, which requires as narrow a Gaussian as possible, and (ii) considering the activated complex as being in a time-dependent state, broadening the Gaussian so that the statistical properties of the activated complex are consistent with the time–energy uncertainty relation. The latter approach, which builds on a recent semiclassical study of chemical reaction thresholds [L. Bonnet, C. Crespos and M. Monnerville, *J. Chem. Phys.*, 2022, **157**, 094114], is coupled with simple calculations of tunneling through adiabatic barriers in the parabolic limit. The resulting methods are then used to calculate the reaction probability for two model processes involving, respectively, long-lived and short-lived activated complexes. The agreement with quantum probabilities appears to be very good. Based on these results, the shapes of quantum probabilities are analysed from the classical dynamics and the time–energy uncertainty relation. The question of zero-point energy violation at the transition state is examined in light of the preceding results.

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1 Introduction

Molecular beam experiments provide highly accurate data on the dynamics of chemical reactions and inelastic collisions.¹ These data are mainly the state-resolved integral and differential cross sections for gas-phase reactions, and related quantities for gas-surface processes. The most accurate way to reproduce them theoretically is to perform exact quantum scattering calculations.² More often than not, however, such calculations are unfeasible for gas-phase reactions involving more than three atoms, and for all gas-surface reactions if the surface atoms are allowed to move. Therefore, most polyatomic processes are studied theoretically by means of the quasi-classical trajectory (QCT) method, whose applicability is not limited by the number of atoms.^{3–8}

The QCT method leads to satisfactory predictions for complex-forming reactions if enough quantum states (more than 3 or 4) are available in the separated fragments. QCT calculations are also accurate for barrier reactions if the previous state number constraint is satisfied not only in the

separated fragments, but also in the activated complex (AC), defined here as the region around the transition state (TS) in which the vibrational motion orthogonal to the reaction path evolves adiabatically (see Section 2 for more details). Otherwise, the dynamics have a truly quantum nature and QCT predictions may not be realistic. This is typically the case of endothermic barrierless reactions occurring at total energies below the product zero point energy (ZPE). Such processes are quantum mechanically prohibited, but classically allowed.⁹ This also applies to barrier reactions occurring at total energies below the zero point energy of the activated complex (or TS).^{10,11} Dissociative chemisorption is another example for which the QCT method may fail. If a molecule in the internal ground state hits a surface with very low kinetic energy and some trajectories corresponding to this event are reflected back into the vacuum, it may happen that most of the reformed reagents have a vibrational energy lower than the ZPE. These unphysical paths artificially enhance the reflection at the expense of the dissociative chemisorption, thus, making the sticking probability too low compared to its quantum value.¹²

Gaussian binning (GB) is a method of analysing the trajectory outcomes that significantly improves the QCT predictions when the state number constraint is not satisfied in the separated fragments.¹³ GB is based on the idea, justified by classical *S*-matrix theory (CSMT),¹⁴ that only those trajectories

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complying with Bohr's quantization condition in the separated fragments should be taken into account in the prediction of the measurements.^{15,16} Such trajectories start from integer actions of the reagents and end with integer actions of the products (see Chapter 6.2 of ref. 17 for a clear definition of the concept of action; actions are vibrational and rotational for a gas-phase reaction, and vibrational, rotational and diffractive¹² for a molecule reflected by a surface). In order to ensure that Bohr's quantization conditions are satisfied, trajectories are launched from the reagents with integer actions (like in the standard QCT method) and should in principle be assigned Dirac delta weights, infinite if the product actions have integer values, zero otherwise.¹⁸ Clearly, this approach is intractable as such. Therefore, in practice, trajectories are assigned Gaussian weights normalized to unity such that the closer product actions are to integer values, the higher the statistical weight of the trajectories.^{12,13,15,19–21} Moreover, the full-width-at-half-maximum of the Gaussians, simply called width from now on, are taken at values much smaller than the unit spacing between two neighboring states, as they are supposed to mimic Dirac deltas. For polyatomic reactions, however, it is more efficient to make Gaussian weights depend on the vibrational energy rather than the actions, as this choice scales much better with dimensionality,^{22–24} in particular for describing threshold effects.

So far, GB has never been applied to the activated complex. Yet, its states are known to be quantized to some extent, particularly near the reaction threshold.^{11,25,26} Therefore, one may wonder whether trajectories should be assigned Gaussian weights centered at integer values of the vibrational actions of the activated complex (or at its quantized energies). The only difference with the product states is that the quantized states of the activated complex²⁵ are non-stationary states with an energy width and a lifetime, as shown by the semiclassical developments in Section IV.B.4. of ref. 11. Consequently, Gaussian widths should not be taken as small as possible, like for products whose lifetime is infinite and the quantized energies perfectly defined. Instead, they should be chosen so as to comply with the time–energy uncertainty relation, as suggested by the results in ref. 11. The aim of this study is to show how this can be done in practice within the framework of the Schatz model of H + H₂ reaction that mimics the passage from the free rotation of reactants, through bending vibration at the transition state, to the free rotation of products.¹⁰ As a preliminary step, however, we consider a recent model reaction, reminiscent of going through a canyon,¹¹ for which the lifetime of the activated complex is long enough for it to be considered as being in a stationary state, and therefore assigned a standard Gaussian weight.

2 Standard Gaussian binning of the activated complex

2.1 Canyon crossing

A detailed study of the quantum, semiclassical and classical dynamics of the present model reaction has recently been

published.¹¹ For this reason, we limit ourselves to a brief review of the details that are important for this study.

2.1.1 System and dynamics. We consider a fixed-plane of the laboratory and a rigid diatom rotating within this plane. Moreover, we force the center-of-mass of the diatom to move on a given R -axis of the plane. The total energy of the rotator in translation is E . The orientation of the diatom with respect to R is determined by the angle ϕ . The classical Hamiltonian of the system is

$$H = \frac{P^2}{2\mu} + \frac{J^2}{2I} + V(R, \phi) = E \quad (1)$$

where $P = \mu \frac{dR}{dt}$ and $J = I \frac{d\phi}{dt}$ are the momenta conjugate to R and ϕ , respectively, t is the time, μ is the mass associated with motion along the R -axis, I is the moment of inertia of the diatom and $V(R, \phi)$ is the potential energy. Two values of μ will be considered, 19 amu and 1.9 amu, corresponding to atom-diatom reactions of the heavy plus light-heavy and light plus heavy-light type, respectively. I is equal to mr_e^2 with $m = 0.5$ amu and $r_e = 0.7414$ Å. Our model potential energy surface (PES) is defined by

$$V(R, \phi) = \frac{\alpha \sin^2(\phi/2)}{\cosh(2\beta R) + \cosh(\beta \Delta)} \quad (2)$$

with $\alpha = 4.053$ eV, $\Delta = 2$ Å and $\beta = 3$ Å⁻¹. Its contour level representation is shown in Fig. 1. The right column gives the correspondence between colors and potential energy in eV. The system parameters and the analytical expression of the PES can be found in ref. 11. The PES involves a narrow channel separating the reagent and product planes on the left and right, respectively. An example of reactive trajectory starting from the reagent plane with $\phi = \phi_1$ and no rotational excitation is represented by the yellow line. We will indeed assume in the following that the initial diatom is in the rotational ground-state, with a random orientation ($-\pi \leq \phi_1 \leq \pi$). The trajectory is started at $R_1 = -3$ Å with $P_1 = \sqrt{2\mu E}$ and stopped at $R_2 = 3$ Å (see Section III.A.2 of ref. 11 for more details on the trajectory calculations). Within the channel, there is a periodic orbit (PO)

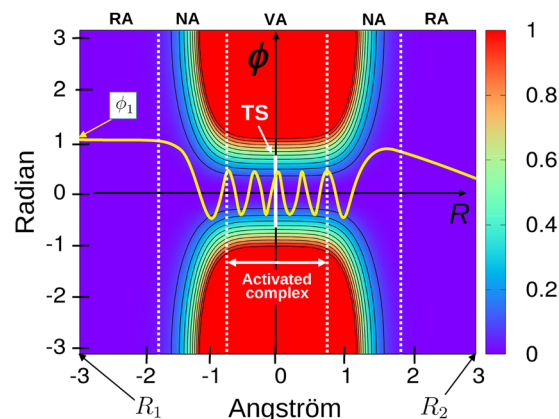


Fig. 1 Potential energy surface of the canyon-crossing model and a typical classical reactive path. See the text for details.



trapped along the ϕ -axis (see the white vertical segment centered at the origin of the (R, ϕ) frame). According to classical mechanics, the transition state (TS) is defined as the phase space surface whose projection on the (R, ϕ) configuration plane coincides with the representation of the PO in the same plane.²⁷ The TS is thus trivially defined in (R, ϕ, P, J) by $R = 0$. The reactive flux is minimum through the TS which, thus, corresponds to a phase space bottleneck. The reactive trajectory clearly mimics the passage from the free rotation of reactants, through bending vibration at the transition state, to the free rotation of products that occurs in three-dimensional bimolecular reactions. Four white vertical dashed lines define five regions labeled RA, NA and VA above the frame of Fig. 1. The dynamics are rotationally adiabatic in the RA reagent and product half planes (the PES is flat in these domains so that the rotation is free), vibrationally adiabatic in the VA central region encompassing the TS (see further below for a numerical proof of this statement, which is quite obvious from the shape of the trajectory in the VA band), and nonadiabatic in the NA regions due to the presence of strong couplings between the R and ϕ coordinates. We define the activated complex as the system orthogonal to the R -axis in the VA region, whose boundaries are $R \approx -0.75$ and 0.75 , in Å (see Fig. 1). The activated complex has a lifetime, contrary to the TS which is just crossed by classical trajectories.

Although the TS is a classical concept, it can nevertheless be treated “artificially” in a quantum framework. For our process, the quantum Hamiltonian of the TS is $\hat{H}^\ddagger = -\frac{\hbar^2}{2I} \frac{\partial^2}{\partial \phi^2} + V(0, \phi)$ ($\hbar = \frac{h}{2\pi}$ where h is Planck’s constant) and its spectrum can be obtained by diagonalization methods (see ref. 28 and references therein). Alternately, it can be semiclassically estimated as follows. Calling $E^\ddagger = \frac{J^2}{2I} + V(0, \phi)$ the energy of the TS, its vibrational action over one cycle is given by

$$N^\ddagger = \frac{\oint J d\phi}{h} - \frac{1}{2} \quad (3)$$

with

$$J = \pm \sqrt{2I[E^\ddagger - V(0, \phi)]} \quad (4)$$

(see Chapter 6.2 of ref. 17). In the harmonic limit, valid for our system in the energy range that we will be considering, $V(0, \phi)$ is equal to $\frac{1}{2}I\omega^{\ddagger 2}\phi^2$ and eqn (3) and (4) lead to

$$E^\ddagger = \hbar\omega^\ddagger \left(N^\ddagger + \frac{1}{2} \right) = E_{N^\ddagger}^\ddagger. \quad (5)$$

This illustrates clearly that N^\ddagger is the classical analog of the vibrational quantum number. The quantized energies E_n^\ddagger of the TS, where n is a positive integer or 0, are given by eqn (5) with N^\ddagger replaced by n (the latter cannot be too large, because the harmonic approximation of $V(0, \phi)$ is valid for moderate values of $|\phi|$). E_0^\ddagger is equal to 705 cm^{-1} ($2E_0^\ddagger$ typically corresponds to a bending vibration quantum) and $E_n^\ddagger = (2n + 1)E_0^\ddagger$.

The vibrational action can also be calculated along the trajectories inside the channel. It is still given by eqn (3) and (4), except that in the latter, E^\ddagger and $V(0, \phi)$ are replaced by $[E - P^2/(2\mu)]$ and $V(R, \phi)$, respectively. This action, now called N , is represented in terms of R in Fig. 2 for four randomly selected reactive trajectories at the energy $H = E = 1.5E_0^\ddagger$ (at this energy, the system is bounded along the ϕ -axis for $|R| < 1.5$, so N is given by eqn (3) and (4); for larger values of $|R|$, the rotational motion is hindered and the expression of the action along the ϕ -axis is different). As a matter of fact, N does not vary significantly between $R \approx -0.75$ and $R \approx 0.75$ compared to the unit spacing between two vibrational states. In other words, the nature of the vibrational motion does not change within the previous boundaries, *i.e.*, the dynamics are vibrationally adiabatic within the VA band (see Fig. 1). In contrast, Fig. 2 shows that N fluctuates significantly within the NA bands, due to strong couplings between the R and ϕ coordinates. We will call $N^\ddagger(\phi_1)$ the value of N at the TS for the reactive trajectory starting from ϕ_1 .

2.1.2 Reaction probabilities. The quantum mechanical (QM) and QCT reaction probabilities are represented in Fig. 3 for the heavy and light masses (details of the QM and QCT calculations can be found, respectively, in Sections III.A.1 and III.A.2 of ref. 11; see also ref. 29). Within the energy range considered, the quantum probabilities (solid lines) involve two thresholds which coincide with the quantized TS energies E_0^\ddagger and E_2^\ddagger , respectively. It is known from the research by Truhlar and co-workers²⁵ that these energies are associated with quantized states of the activated complex that ensure the passage of reactive flux as soon as they are accessible. These states are discussed further below within the framework of semiclassical reactive scattering theory. We note that there is no threshold at E_1^\ddagger , for the following reason: the initial rotational state, given by the constant value $1/\sqrt{2\pi}$, is even. Moreover, the ϕ -dependence of the potential energy is also even. Therefore, the E -resolved scattering state is necessarily symmetric with respect to the R -axis. Consequently, the coefficients of its development on the vibrational states $\Psi_n(\phi)$ of the TS are necessarily 0 for odd

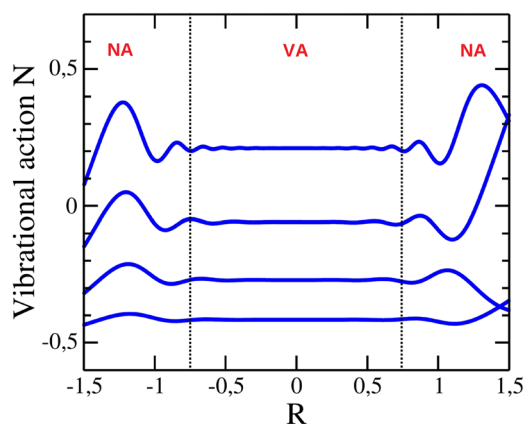


Fig. 2 Variation, for the canyon crossing model, of the vibrational action N along the reaction coordinate R , given in Å, for some reactive trajectories at $E = 1.5E_0^\ddagger$.



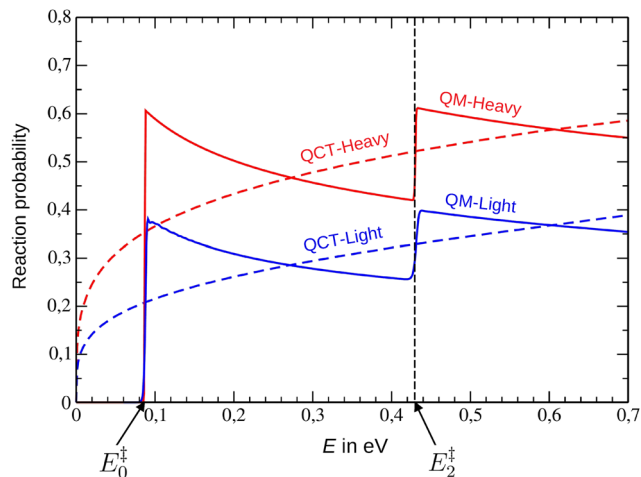


Fig. 3 QM and QCT reaction probabilities for the heavy and light masses in terms of the collision energy E ; QM: solid lines, QCT: dashed lines.

values of n (for these states are antisymmetric). The QCT probabilities (dashed lines in Fig. 3) are in poor agreement with QM ones. This is especially true around the thresholds, which classical mechanics obviously ignores. Nevertheless, we see that the disagreement decreases when the total energy increases, as it should be. In particular, QM probabilities perform damped “oscillations” around QCT probabilities in such a way that both are nearly equal at the middle of the jumps for the threshold associated with the TS excited states $n \geq 2$ (only the one corresponding to $n = 2$ can be seen in Fig. 3). We will take advantage of this observation later.

We now consider the GB procedure applied to the activated complex. We found that at $E = 1.5E_0^\ddagger$, the ratio of the average lifetime of the activated complex (time to cross the VA band) and its vibrational period is roughly equal to 12 and 4 for the heavy and light masses, respectively. Therefore, the activated complex survives for a pretty long time, making it reasonable to consider it in a stationary state. Consequently, classical trajectories may be assigned Gaussian weights centered at integer values of the bending vibrational action at the TS. The GB-QCT reaction probability (also called the GB probability in the following) is then formally given by¹⁶

$$P = \frac{1}{\pi} \int_0^{\phi_>} d\phi_1 \sum_{k=0}^{k_{\max}} G(N^\ddagger(\phi_1) - 2k), \quad (6)$$

with

$$G(x) = \frac{2e^{-(x/\varepsilon)^2}}{\sqrt{\pi\varepsilon}} \quad (7)$$

[note that $G(x) = 2\delta(x)$ in the limit where ε tends toward 0]. Reactive trajectories start from (R_1, ϕ_1) with ϕ_1 within the interval $[0, \phi_>]$ (we limit ourselves to positive angles since trajectories are initially run parallel to the R -axis ($J = 0$) and the potential is symmetric with respect to the latter). $\phi_>$ is the maximum angle leading to reaction at E (between $\phi_>$ and π , trajectories are non-reactive). k_{\max} is the maximum value of k

consistent with E , and ε should be taken at a value much lower than 1. Here, ε is taken at 0.06, its usual value,¹³ except in the vicinity of the thresholds where it needs to be significantly reduced for better accuracy—right at a threshold, however, the Gaussian, in addition to be very narrow, must also be centered at a value slightly smaller than the integer action corresponding to the threshold, so it can fully overlap a range of available vibrational actions; if the Gaussian is centered exactly at an integer action, it covers only half of the available action range, and the GB probability therefore amounts to only half of its actual value. A few hundred (thousands) trajectories are far (near) enough from thresholds. Note the presence of a factor of 2 in the numerator of the Gaussian weight (eqn (7)) in order to compensate for the fact that only even states of the activated complex need be taken into account. A similar doubling is used in the QCT approach of homonuclear diatom scattering with a third body; the parity of the rotational state of the diatom is indeed conserved throughout the collision. An illuminating semiclassical analysis of this question can be found in ref. 30. QM and GB-QCT reaction probabilities are compared in Fig. 4. The agreement between both approaches is quantitatively satisfactory, but qualitatively excellent. As a matter of fact, Gaussian binning the vibrational action of the activated complex considerably improves the shape of the reaction probability (compare the dashed curves in Fig. 3 and 4). We note in the QM curve for the light mass a slight smoothing of the second step, due to tunneling through the adiabatic barrier¹¹ of state $n = 2$. Such smoothing is obviously absent from the GB curve which assumes that the activated complex is in a perfectly stationary state.

2.1.3 Shape of the quantum probability. We now use eqn (6) and (7) to explain the shape of the quantum probability. In the limit where ε tends toward 0, these equations become

$$P = \frac{2}{\pi} \int_0^{\phi_>} d\phi_1 \sum_{k=0}^{k_{\max}} \delta(N^\ddagger(\phi_1) - 2k). \quad (8)$$

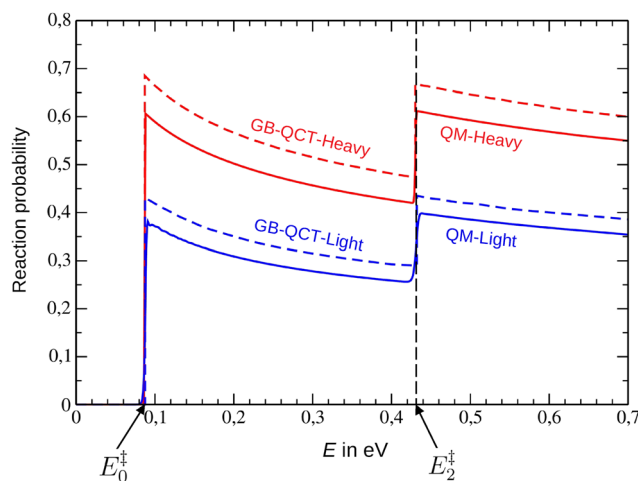


Fig. 4 QM and GB-QCT reaction probabilities for the heavy and light masses in terms of the collision energy E ; QM: solid lines, GB-QCT: dashed lines.



Let us consider the large value of μ and two energies E_a and E_b slightly larger and slightly lower than E_0^\ddagger and E_2^\ddagger , respectively. For both E_a and E_b , the TS vibrational energy is necessarily lower than E_2^\ddagger . Now, the latter corresponds to $N^\ddagger(\phi_1) = 2$ (see eqn (5) for the harmonic case). As a consequence, $N^\ddagger(\phi_1)$ is strictly lower than 2 for both E_a and E_b , and the sum in eqn (8) reduces to $\delta(N^\ddagger(\phi_1))$. $N^\ddagger(\phi_1)$ is shown in Fig. 5 for $E_a = 1.1E_0^\ddagger$ and $E_b = E_2^\ddagger - 0.1E_0^\ddagger$. For $\phi_1 = 0$, the trajectories follow the R -axis and $N^\ddagger(\phi_1)$ takes the minimum value of $-1/2$ [see eqn (3)]. Above $\phi_1 = 0$, the larger ϕ_1 , the larger the amplitude of the vibrational motion within the channel (see Fig. 1), and the larger $N^\ddagger(\phi_1)$. The maximum angle $\phi_>$ leading to reaction at E_a is labelled $\phi_>^a$ (see the right end of the blue solid curve; note for the following that if one decreases the energy from E_a to E_0^\ddagger , the right end of blue solid curve ends up touching the dashed green line). For each energy, only one value of ϕ_1 leads to $N^\ddagger(\phi_1) = 0$. These values, called ϕ_1^a and ϕ_1^b , respectively (see Fig. 5), lead to the blue and red trajectories represented in Fig. 6. Using the fact that for E_a ,

$$\delta(N^\ddagger(\phi_1)) = \frac{1}{\left| \frac{dN^\ddagger}{d\phi_1}(\phi_1^a) \right|} \delta(\phi - \phi_1^a) \quad (9)$$

(see eqn (1.158) in ref. 31), the value P_a of P at E_a , deduced from eqn (8), is

$$P_a = \frac{2}{\pi \left| \frac{dN^\ddagger}{d\phi_1}(\phi_1^a) \right|} \quad (10)$$

The value P_b of P at E_b is given by the same expression with ϕ_1^b substituted for ϕ_1^a . Fig. 6 shows that the potential at the entrance of the interaction region acts as a converging lens that focuses the reactive trajectories. We note, however, that the trajectory is more deviated for the lowest energy (blue curve)

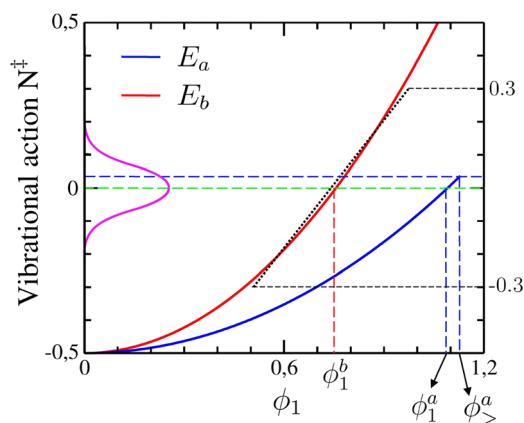


Fig. 5 ϕ_1 -dependence of the vibrational action at the transition state for $E_a = 1.1E_0^\ddagger$ (blue solid curve) and $E_b = E_2^\ddagger - 0.1E_0^\ddagger$ (red solid curve). The highest action available at E_b , given by the height of the dashed blue line, is reached at the angle $\phi_>$. A dotted black straight line is represented within the action range $[-0.3, 0.3]$, showing that the red line is nearly linear within this range. A Gaussian weight centered on the dashed green line, i.e. at 0, is represented in magenta. The action is 0 at ϕ_1^a and ϕ_1^b for E_a and E_b , respectively.

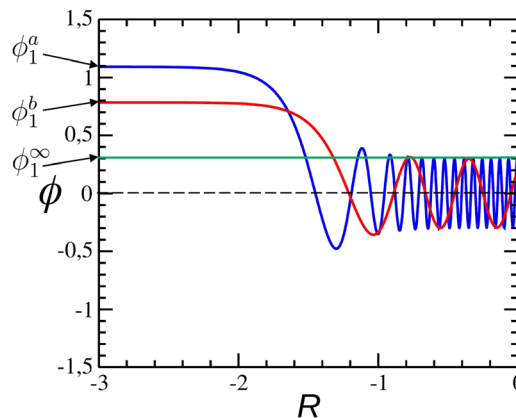


Fig. 6 Trajectories leading to $N^\ddagger(\phi_1) = 0$ at E_a (blue path), E_b (red path) and $E = +\infty$ (green path).

than for the highest (red curve). This is logical since for an infinite energy, the trajectory leading to $N^\ddagger(\phi_1) = 0$ is the straight line defined by ϕ_1^∞ (green curve), not deviated at all. Therefore, ϕ_1^a is larger than ϕ_1^b , so the slope of $N^\ddagger(\phi_1)$ is stronger for E_b than for E_a (see Fig. 5). As a consequence, the reaction probability, proportional to the inverse of the slope (see eqn (10)), decreases with E . However, it approaches a plateau at high energies for which all the trajectories converge towards the green trajectory. This justifies the shapes of each TS vibrational state contribution to the reaction probability in Fig. 4.

A comment on the value of ε in eqn (7) is in order. It can be shown that the integral of $G(f(\phi_1))$ does not depend on ε if $f(\phi_1)$ is linear (the proof is straightforward). Now, at E_b , $N^\ddagger(\phi_1)$ is relatively close to linearity within the range $[-0.3, 0.3]$ (see the solid red curve and dotted black straight line in Fig. 5). The width of $G(N^\ddagger(\phi_1))$ is equal to $2\sqrt{\ln(2)\varepsilon}$. The value of ε corresponding to the width of the previous range is ~ 0.36 . Any value of ε lower than 0.36 is thus expected to make P , as obtained from eqn (6) and (7), very close to P_b . Moreover, numerical calculations show that even for $\varepsilon \sim 0.6$, corresponding to the unit width (i.e., the classical limit³²), P remains relatively close to P_b . On the other hand, the situation is different at E_a . The maximum vibrational action $N^\ddagger(\phi_>^a)$ is slightly above 0 (see the right end of the blue curve in Fig. 5). Therefore, $G(N^\ddagger(\phi_1))$ must be very narrow for eqn (6) and (7) to agree with eqn (10). Otherwise, $G(N^\ddagger(\phi_1))$ will be non 0 for all the actions greater than $N^\ddagger(\phi_>^a)$, which are not available, so eqn (6) and (7) will significantly minimize the reaction probability. This would for instance be the case with the magenta Gaussian in Fig. 5 which partly overlaps the region of unavailable actions (above the horizontal dashed blue line). We will come back to this point in Section 3.3.

2.2 Schatz model of the H + H₂ reaction

The Hamiltonian is the same as previously (see eqn (1)), with $\mu = 1224.7$ a.u. and $I = 3600$ a.u. The analytical expression of Schatz PES is

$$V(R, \phi) = \frac{B \sin^2 \phi}{\cosh^2(R/D)} \quad (11)$$



with $B = 1.63$ eV and $D = a_0/2$.¹⁰ The system parameters are roughly appropriate to the H + H₂ reaction. Its contour level representation is shown in Fig. 7. Compared to the canyon-crossing model (Fig. 1), the interaction region is narrow. The spatial extension of the activated complex is thus necessarily shorter than previously, as well as its average lifetime. Fig. 8 shows the variation of the vibrational action N along the R -axis for four sets of a dozen trajectories having the following properties: the blue, red, green and orange trajectories are run at $E = 3E_0^\ddagger/2$, $E_2^\ddagger + E_0^\ddagger/2$, $E_4^\ddagger + E_0^\ddagger/2$ and $E_6^\ddagger + E_0^\ddagger/2$, respectively. Moreover, they cross the TS with N^\ddagger close to 0, 2, 4 and 6, respectively, thus implying that semiclassically, they strongly contribute to the reactivity (see Section IV.B.3 of ref. 11 and Section 3.1 in the present work). The rectangles delimit the VA zones, where N is nearly constant. The length of these zones, indicated above each rectangle, is obtained from the trajectories that cross the transition state with N equal to an integer.

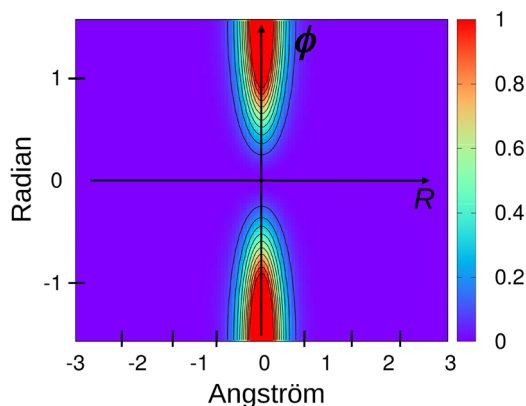


Fig. 7 Potential energy surface of the model H + H₂ reaction. ϕ is limited to the range $[-\pi/2, \pi/2]$.

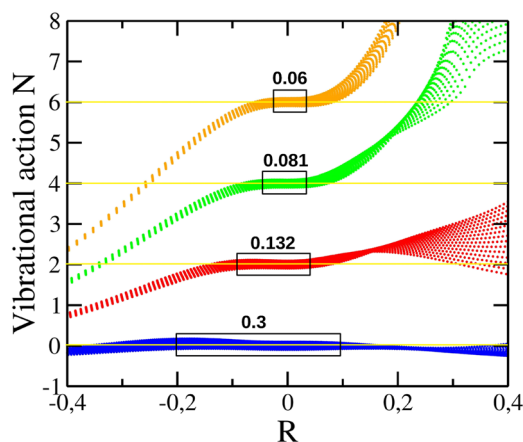


Fig. 8 Variation, for the model H + H₂ reaction, of the vibrational action N along the reaction coordinate R , given in Å, for some reactive trajectories slightly above the threshold energies E_0^\ddagger , E_2^\ddagger , E_4^\ddagger and E_6^\ddagger . Actions are represented by dots every 0.1 fs. Black rectangles locate the activated complex, or the VA region, where actions do not vary. The lengths of these regions are indicated above the rectangles.

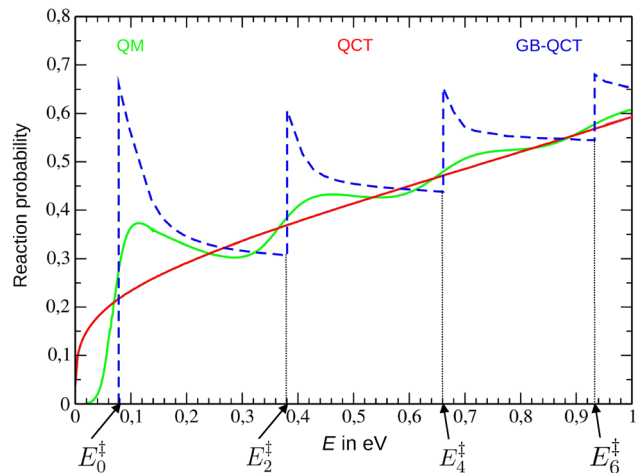


Fig. 9 Energy dependence of the reaction probability for the Schatz model of the H + H₂ reaction obtained by means of three approaches: QM calculations (solid green), the standard QCT method (solid red), and the GB-QCT method (dashed blue).

This length, hereafter denoted L , appears to decrease with N . This is because the couplings between the R and ϕ coordinates near the ϕ -axis become stronger as ϕ increases, and the larger the value of N , the greater the range of ϕ that can be accessed. Interestingly, the zones are not necessarily centered exactly at $R = 0$.

The calculations show that the ratio of the lifetime of the activated complex and its vibrational period is less to $\sim 1/3$. Consequently, unlike the canyon case, the activated complex survives for a negligible amount of time. As a result, we cannot consider it in a stationary state and use eqn (6) and (7) with $\varepsilon \sim 0$. If we do so, we obtain the dashed blue curve in Fig. 9 (GB results), which shows poor agreement with the solid green curve (QM results), in contrast with the canyon case (see Fig. 4). Therefore, we need to go beyond standard GB.

3 Uncertainty-based Gaussian binning of the activated complex

We base our new method on the semiclassical analysis of chemical reaction thresholds published in ref. 11. We first review some key results of this study and then present the method.

3.1 Quantization of the activated complex from the point of view of classical S-matrix theory

In ref. 11, we applied CSMT to the canyon-crossing model in order to calculate the probability P_{00} that reactants in the rotational ground state yield products in the same state. P_{00} is equal to $|S_{00}|^2$ where the S -matrix element S_{00} is approximated by the CSMT expression

$$S_{00} = \sum_{k=1}^{\infty} \sqrt{p_k^{\text{cl}}} e^{i\Omega_k/\hbar}. \quad (12)$$

The terms of the sum are the probability amplitudes assigned to the infinite number of real-time trajectories



starting from the reagent plane with $J = 0$ and reaching the product plane with $J = 0$ (see the blue line in Fig. 3 of ref. 11, where the final value of J is referred to as J_2 and the initial angle ϕ_1^k of the k th trajectory leading to $J_2 = 0$ is referred to as $\bar{\phi}_1^k$). In eqn (12), the modulus of each amplitude is the square root of the classical statistical weight p_k^{cl} of the corresponding trajectory,³³ while its phase Ω_k/\hbar is the action integral in the momentum space along the trajectory, expressed in units of \hbar (this phase includes the so-called Maslov index, which plays a crucial role in explaining the origin of quantum reaction thresholds^{11,18}). Quantum mechanics comes into play in eqn (12) *via* the sum of complex numbers, which is the mathematical expression of the superposition principle, and *via* \hbar , in the denominator of the phases. In particular, the sum creates interferences between probability amplitudes when taking the squared modulus of S_{00} (we will simply say that trajectories “interfere”).

We found that for $E = 1.2E_0^\ddagger$, for example, the phase Ω_k/\hbar is stationary for a given value k_0 of k , and the Ω_k 's are close to Ω_{k_0} —with respect to \hbar —for a range of values $[k_0 - K, k_0 + K]$ where K is of the order of ~ 10 . For these values, the probability amplitudes are nearly aligned in the complex plane, so they interfere constructively. Outside this range, however, Ω_k/\hbar increases rapidly with $|k - k_0|$ in such a way that the corresponding trajectories interfere destructively. Consequently, only $\sim 2K$ trajectories actually contribute to P_{00} (a few tens at $E = 1.2E_0^\ddagger$, more if E is closer to E_0^\ddagger ; details are given in ref. 11, in particular in Section IV.B.3). For each of these trajectories—hereafter referred to as dominant paths (the blue trajectories in Section IV.B of ref. 11)—we calculated the vibrational energy at the TS, $E_v^\ddagger = J^2/(2I) + V(0, \phi)$, and the lifetime T^\ddagger of the activated complex, given by

$$T^\ddagger = \frac{L}{\sqrt{2(E - E_v^\ddagger)}/\mu} \quad (13)$$

The denominator of this expression is the velocity along the R -axis within the VA band. The dependence of L on N was extrapolated from the 4 lengths indicated in Fig. 8.

We then calculated the standard deviations ΔE_v^\ddagger and ΔT^\ddagger of E_v^\ddagger and T^\ddagger for the dominant paths (see eqn (61)–(64) in Section IV.B.4 of ref. 11). A first remarkable result of these calculations is that E_v^\ddagger is distributed around E_0^\ddagger in such a way that its mean value \bar{E}_v^\ddagger is very close to E_0^\ddagger (property I). This also corresponds to N^\ddagger distributed about 0, with mean value $\bar{N}^\ddagger \sim 0$. A second remarkable result is that the product $\Delta E_v^\ddagger \Delta T^\ddagger$ is distributed around $\hbar/2$ so that it roughly satisfies the time-energy uncertainty relation (property II).³⁴ In other words, the activated complex is in a time-dependent quantum state whose average energy is E_0^\ddagger . This corroborates the findings of Truhlar and co-workers.²⁵ The previous observations are valid whatever E within the range $[E_0^\ddagger, E_2^\ddagger]$. This is also the case for E within the range $[E_2^\ddagger, E_4^\ddagger]$, except that a few tens of trajectories cross the TS with $E_v^\ddagger \sim E_0^\ddagger$, and a few other tens with $E_v^\ddagger \sim E_2^\ddagger$. A similar

scenario is valid for the following energy intervals, with more states of the activated complex involved in the reaction. The trajectories represented in Fig. 8 in the (R, N) plane are dominant paths.

Below E_0^\ddagger , the situation is similar to that within the range $[E_0^\ddagger, E_2^\ddagger]$, except that the trajectories contributing to P_{00} must be integrated in imaginary time (see Section IV.B.3 in ref. 11). This allows them to tunnel through the adiabatic ground state barrier with a negative kinetic energy $P^2/(2\mu)$ and a vibrational energy E_v^\ddagger greater than E , also distributed around E_0^\ddagger and equal on average to E_0^\ddagger . Furthermore, the product $\Delta E_v^\ddagger \Delta T^\ddagger$, where T^\ddagger is now the modulus of the imaginary time, is again close to $\hbar/2$. Similar findings are found for tunneling through adiabatic excited state barriers.

Since the semiclassical reaction probability is mainly determined by the dominant paths, one may wonder whether the QCT probability computed using only these paths could be more accurate than the standard QCT probability. However, the rigorous identification of the dominant paths (DP) is difficult for full-dimensional reactions. Consequently, in the following section we propose an approximate identification based on GB in order to estimate the DP-based QCT probability.

3.2 Method and results

For simplicity's sake, we only consider trajectories in real time, as in general, running trajectories in imaginary time poses real problems. We first assume that the total energy is slightly above the first threshold, E_0^\ddagger , so the only available state of the activated complex is the ground one. N_{traj} trajectories are run parallel to the R -axis ($J = 0$) from $R_1 = -3 \text{ \AA}$ and regularly spaced values of ϕ_1 . N_{reac} trajectories are reactive. For the k th reactive trajectory, the vibrational action and energy at the TS as well as the activated complex lifetime are denoted by N_k^\ddagger , $E_{v,k}^\ddagger$ and T_k^\ddagger , respectively [the k th trajectory here has no connection with the k th trajectory in eqn (12)].

We have seen previously that the dominant paths cross the TS with N^\ddagger distributed around 0 and $\bar{N}^\ddagger \sim 0$. We now assume that, if reactive trajectories are assigned the weight $G(N^\ddagger)$ given by eqn (7) and if ε is varied to satisfy both properties I and II, the resulting probability should be in reasonably good agreement with the DP-based QCT probability, and hopefully with the quantum one. To check the validity of this assumption, we proceed as follows.

We first set ε at a very small value, say, 0.001. $G(N^\ddagger)$ is thus an extremely narrow Gaussian centered at 0 and it is obvious that property I is satisfied, *i.e.*, $\bar{E}_v^\ddagger = E_0^\ddagger$. On the other hand, there is no dispersion in the vibrational actions and energies and, consequently, no dispersion in the lifetimes [see eqn (13)]. Therefore, both ΔE_v^\ddagger and ΔT^\ddagger tend toward 0, and property II is not satisfied. We then increase ε by a small amount, so that the Gaussian becomes slightly wider; the dispersion of both the vibrational energies and the lifetimes increases slightly, and $\Delta E_v^\ddagger \Delta T^\ddagger$ increases accordingly. We repeat this iterative calculation until $\Delta E_v^\ddagger \Delta T^\ddagger$ exceeds $\hbar/2$ —we checked that property I is still satisfied—and use



the value of ε corresponding to the last iteration to calculate the reaction probability according to

$$P_{>} = \frac{\sum_{k=1}^{N_{\text{reac}}} G_k^{\ddagger}}{N_{\text{traj}}}, \quad (14)$$

with $G_k^{\ddagger} = G(N_k^{\ddagger})$. The subscript on the left-hand-side of eqn (14) will remind us that its right-hand-side is the expression used above E_0^{\ddagger} . For an energy larger than E_2^{\ddagger} but lower than E_4^{\ddagger} , the same approach is applied twice, once for the ground state of the activated complex, as before, and once for its second excited state, with GB weights $G_k^{\ddagger} = G(N_k^{\ddagger} - 2)$. For the next interval, three states of the activated complex are considered, and so on. We will call UGB this uncertainty-based GB procedure. We wish to emphasize that during the iterative calculation, ε must not exceed the classical limit of 0.6.³²

Now, what about the reaction probability $P_{<}$ below the minimum threshold E_0^{\ddagger} ? As previously stated, we wish to avoid running trajectories in imaginary time, as this leads to tricky calculations. However, we have recently demonstrated by means of the saddle point method that, in the case of a simplified canyon-crossing model with analytically tractable dynamics, the reaction probability obtained from eqn (12) is proportional to the probability of tunneling through the adiabatic ground state barrier (see eqn (60) in ref. 11). We now approximate this probability as follows. The second order development in ϕ of the right-hand-side of eqn (11) can be written as $I\omega(R)^2\phi^2/2$ with $\omega(R) = \sqrt{2B/I}/\cosh(R/D)$. The adiabatic ground state barrier $E_0(R)$ is given by the sum of the ZPE at R and the potential energy along the reaction path. The latter being 0, we have $E_0(R) = \hbar\omega(R)/2 = E_0^{\ddagger}/\cosh(R/D)$ with $E_0^{\ddagger} = \hbar\sqrt{B/(2I)}$. If we approximate $E_0(R)$ with a parabolic barrier, we obtain a very satisfying approximation of the tunneling probability immediately below E_0^{\ddagger} from³⁵

$$P_{\text{tun}} = \frac{e^{\beta}}{1 + e^{\beta}} \quad (15)$$

with

$$\beta = \frac{2\pi D(E - E_0^{\ddagger})}{\hbar\sqrt{E_0^{\ddagger}/\mu}} \quad (16)$$

Calling $P_{>}(E_0^{\ddagger})$ the value of $P_{>}$ at E_0^{\ddagger} , and taking into account the fact that (i) $P_{<}$ is proportional to P_{tun} and (ii) $P_{>}$ and $P_{<}$ must have the same value at E_0^{\ddagger} , we find

$$P_{<} = 2P_{>}(E_0^{\ddagger})P_{\text{tun}}. \quad (17)$$

Eqn (14) with $\Delta E_v^{\ddagger}\Delta T^{\ddagger} = \hbar/2$ and eqn (17) lead to the dashed curve in Fig. 10. It appears that the slope is discontinuous at E_0^{\ddagger} , which constitutes a non-physical result. However, we found that this issue disappears when $\Delta E_v^{\ddagger}\Delta T^{\ddagger}$ is reduced to $\sim \hbar/8$. Should we be concerned about choosing $\hbar/8$ instead of $\hbar/2$? We do not believe so, for the following reason: while the reaction probabilities obtained from both CSMT and UGB are based on

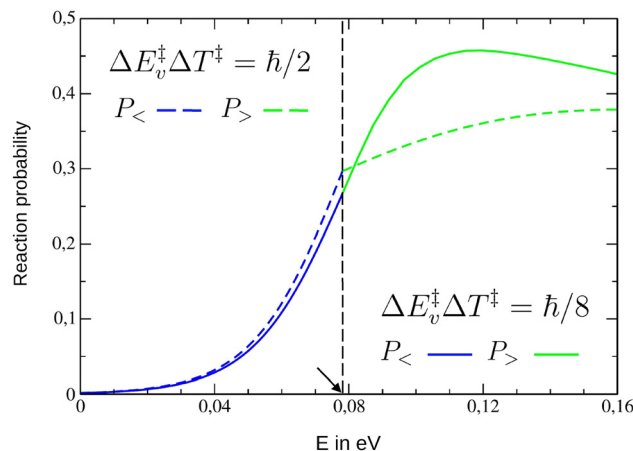


Fig. 10 Energy dependence of the reaction probability around E_0^{\ddagger} for the Schatz model of the H + H₂ reaction obtained from a treatment combining, first, the UGB-QCT method [green curves corresponding to $P_{>}$, given by eqn (14)] and, second, a simple calculation of tunneling through the adiabatic ground state barrier in the parabolic limit [blue curves corresponding to $P_{<}$, given by eqn (17)]. Dashed and solid curves correspond to $\Delta E_v^{\ddagger}\Delta T^{\ddagger}$ taken at $\hbar/2$ and $\hbar/8$, respectively. Only the second value makes the slope of the reaction probability continuous at E_0^{\ddagger} .

roughly the same set of trajectories, the dominant paths that satisfy properties I and II, the calculations leading to these probabilities differ fundamentally in nature. Indeed, the former accounts for constructive interference between trajectories as well as quantization of the final rotational motion, whereas the latter involves a straightforward summation of UGB weights and no quantization of the final rotational motion. Thus, only a qualitative agreement between the two probabilities can be expected from the standard uncertainty relation $\Delta E_v^{\ddagger}\Delta T^{\ddagger} = \hbar/2$. It was thus tempting to tune the degree of uncertainty by varying α in the relation $\Delta E_v^{\ddagger}\Delta T^{\ddagger} = \alpha\hbar/2$ so as to ensure the continuity of the slope of $P_{<}$ and $P_{>}$ at E_0^{\ddagger} and make the reaction probability more realistic in this respect. This is what we did and this led to the solid curve in Fig. 10, corresponding to $\alpha = 1/4$. We then repeated the same calculations for all the excited states of the activated complex available at E . The sum of the resulting probabilities, along with that of the ground state (solid curve in Fig. 10), is plotted in magenta in Fig. 11. As in the case of the long-lived activated complex (Fig. 4), the qualitative agreement with the quantum probability is pleasing. In particular, both the period of the oscillations and the decay of their amplitudes with energy are well reproduced (Fig. 11). Moreover, the quantitative agreement is also satisfactory.

To explain the structure of the semiclassical probability, and thereby gain insight into that of the quantum probability, we consider in Fig. 12 the TS ground state contribution to the UGB, tunneling, and GB probabilities, as well as the value of the Gaussian width (W) after the last iteration of the UGB procedure. W increases from 0.27, at E_0^{\ddagger} , to 1 (the classical limit) at 0.21 eV (vertical dashed line), beyond which it plateaus. A width of 0.27 is small enough to consider that the activated complex is quantized. Nevertheless, the UGB probability at E_0^{\ddagger} is much smaller than the GB one which, yet, results from “exact”



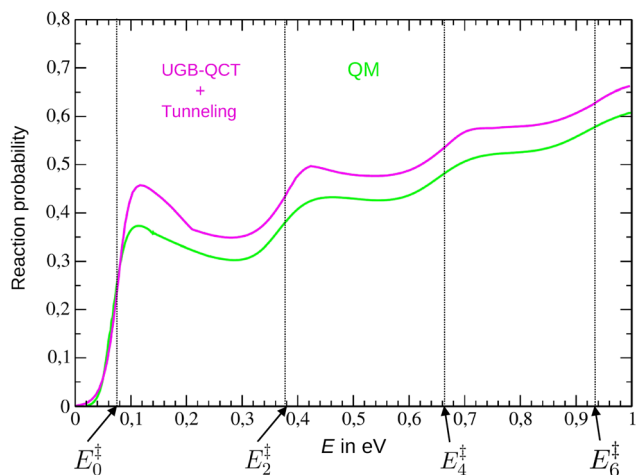


Fig. 11 Energy dependence of the reaction probability for the Schatz model of the H + H₂ reaction obtained by means of QM calculations (green curve), and the UGB-QCT + tunneling method (magenta curve).

quantization (in the sense of Bohr). Therefore, such a difference may not seem to be logical. However, in UGB, only half of the Gaussian overlaps the region of available vibrational actions (right at E_0^\ddagger), unlike in our application of GB which necessarily leads to a higher probability than UGB [see, a few lines after eqn (7), the discussion on the width and position of the Gaussian in the GB curves shown in this work]. When E increases from E_0^\ddagger to 0.21 eV, the Gaussian widens (W increases from 0.27 to 1). Concomitantly, $N^\ddagger(\phi_1)$, which is below the solid blue line in Fig. 5 at E_0^\ddagger , moves upwards (at 0.21 eV, $N^\ddagger(\phi_1)$ is an intermediate curve between the blue and red lines in Fig. 5; note that the latter concerns the canyon crossing but the scenario is similar for the model H + H₂ reaction). These two

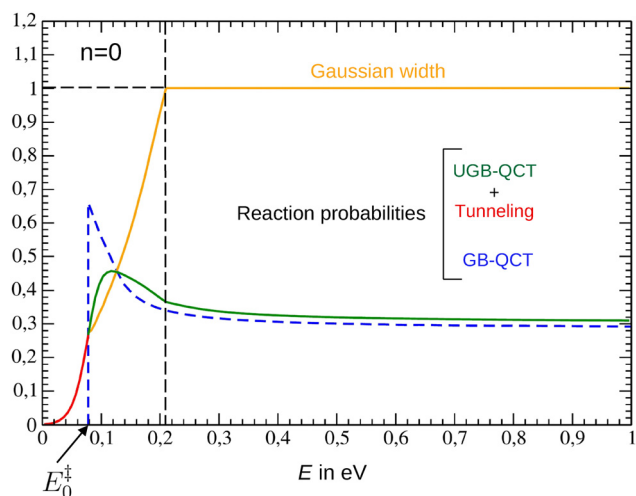


Fig. 12 Energy dependence of the contribution of the TS ground state to the reaction probability for the Schatz model of the H + H₂ reaction obtained by means of the UGB-QCT method (green curve), the tunneling model (red curve) and the GB-QCT method (dashed blue curve). The value of the Gaussian width after the last iteration of the UGB procedure is also shown (orange curve).

effects combine in a subtle way that creates the hill of the UGB probability (Fig. 12) resembling that of the quantum probability (Fig. 11). Once on the plateau ($E \geq 0.21$ eV), the classical limit is reached, since $W = 1$. Nevertheless, the GB probability remains close to the UGB probability for the reasons discussed in the penultimate paragraph of Section 2.1 (quasi-linearity of $N^\ddagger(\phi_1)$ over the range of actions overlapped by the Gaussian). We therefore conclude that quantization of the activated complex is significant only around the threshold E_0^\ddagger . In fact, all the trajectories contributing to the UGB curve in Fig. 12 cross the TS with vibrational energies close to E_0^\ddagger . As a result, the translational energy remains small only when the total energy E is sufficiently close to E_0^\ddagger . In this regime, the activated complex tends to have a relatively long average lifetime, and the standard deviation of its lifetime distribution is significant.³⁶ This, in turn, implies a small standard deviation in its vibrational energy (according to the uncertainty relation), meaning that the vibrational energy is effectively quantized. However, once E reaches the plateau, the translational energy becomes too large for the lifetime distribution to retain a significant spread. Consequently, the standard deviation of the vibrational energy is large and the activated complex is no longer quantized.

The green, blue and red curves in Fig. 13 show the value of W after the final iteration of the UGB procedure for the TS states $n = 2, 4$ and 6 (the one for $n = 0$ is repeated here for comparison). For $n = 2$ and 4 , there is still some quantization of the activated complex at the threshold, though the Gaussians are much broader than for the ground state. On the other hand, quantization no longer occurs for $n = 6$. The reason why threshold quantization is less and less effective as n increases can be found in Fig. 8: the size of the activated complex along the R -axis decreases with n , which reduces the standard deviation of the lifetimes³⁶ and consequently, increases the

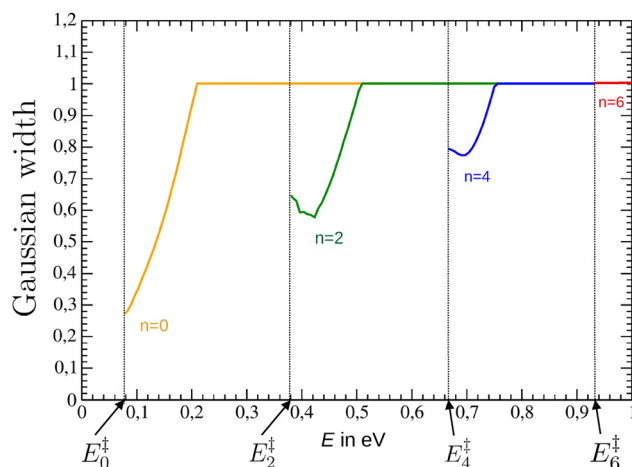


Fig. 13 Energy dependence of the Gaussian widths at the end of the iterative procedure of the UGB-QCT method for the Schatz model of the H + H₂ reaction. The orange, green, blue and red curves correspond respectively to the TS vibrational states $n = 0, 2, 4$ and 6 . The red plateau at 1 covers the blue which covers the green which covers the orange.



spreading of the vibrational energies; the activated complex is thus less and less quantized.

3.3 Semiempirical correction of the reaction probability

We return to the observation that the QCT and QM probabilities converge to the same values for large values of E , as it should be. For the canyon crossing, this necessarily makes the two probabilities nearly equal at the middle of the jumps corresponding to the thresholds of the TS excited states (see Fig. 3). We now propose a simple modification of the GB probability that enables it to satisfy the previous requirement. We consider the GB-QCT and QCT probabilities at E_2^\ddagger , E_4^\ddagger and E_6^\ddagger , divide the latter by the former, and perform a linear fitting of the obtained ratios (we found them to be nearly aligned; we thus considered the straight line through the first and third energies). The GB probabilities multiplied by this fitting, called GB', are compared in Fig. 14 with the QM probabilities. This time, the quantitative agreement is very satisfactory. Note that for the present process, SCIVR calculations turned out to be more tricky, more time-consuming and less accurate than the present calculations.

For the $H + H_2$ model reaction, the analog of the previous method consists in modifying the UGB probability in such a way that the new UGB' probability oscillates around the QCT probability in the limit of large energies. To this aim, we have found the straight line around which the UGB + tunneling curve oscillates for energies larger than $E > E_2^\ddagger$ (the dashed blue line in Fig. 15). Moreover, we have fitted the QCT probability by a straight line in the same energy range (the dashed red line in Fig. 15). The mathematical expressions of these lines are given in the caption of Fig. 15. Dividing the second by the first and multiplying by the UGB + tunneling probability leads to the UGB' + tunneling probability. The latter is compared with the quantum probability in Fig. 16. The agreement is now very satisfactory.

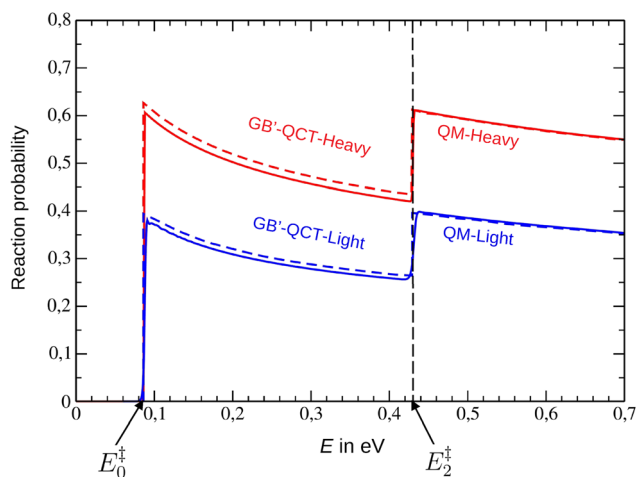


Fig. 14 QM and GB'-QCT reaction probabilities for the heavy and light masses in terms of the collision energy E ; QM: solid lines, and GB'-QCT: dashed lines.

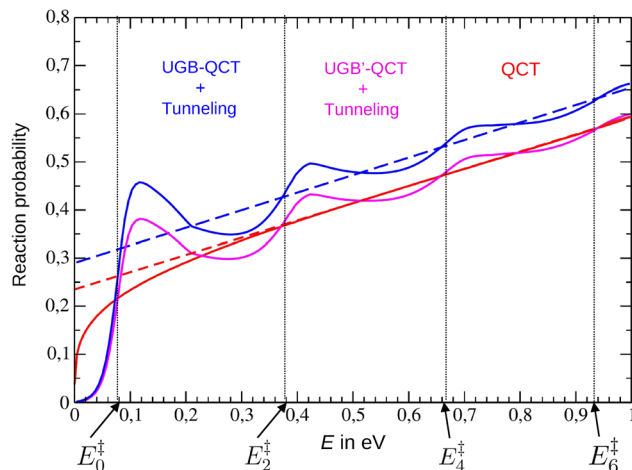


Fig. 15 Energy dependence of the reaction probability for the Schatz model of the $H + H_2$ reaction obtained by means of the UGB-QCT + tunneling method (solid blue curve) and the standard QCT method (solid red curve). The dashed blue line is a straight line chosen in such a way that the blue probability performs damped oscillations around it for $E > E_2^\ddagger$. The dashed red line is a linear fit of the red probability within the same energy range. These lines are given by $0.29 + 0.365E$ and $0.2355 + 0.3575E$, respectively. Dividing the red dashed line by the blue dashed line and multiplying by the UGB + tunneling probability leads to the UGB' + tunneling probability (magenta curve).

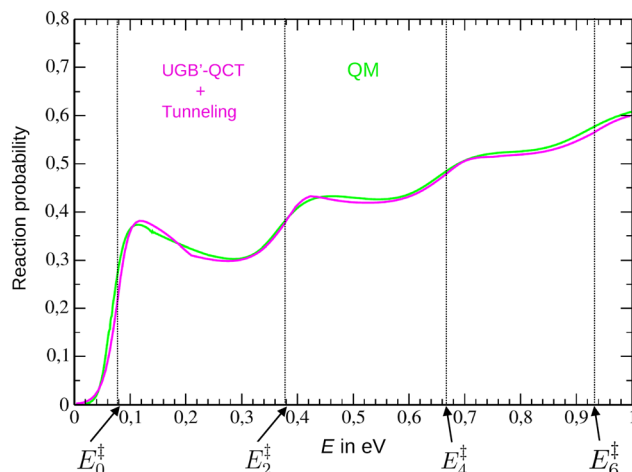


Fig. 16 Energy dependence of the reaction probability for the Schatz model of the $H + H_2$ reaction obtained by means of QM calculations (green curve), and the UGB'-QCT + tunneling method (magenta curve).

4 Conclusion

In this paper, we have introduced the idea that, in classical simulations of chemical reaction dynamics, trajectories can be assigned Gaussian weights centered at integer values of the vibrational actions of the activated complex, defined as the region around the transition state in which the vibrational motion orthogonal to the reaction path evolves adiabatically (alternatively, an energy-based Gaussian weighting scheme may be employed). The Gaussian width is then adjusted so as to



satisfy the time-energy uncertainty relation $\Delta E_v^\ddagger \Delta T^\ddagger \sim \hbar/2$ (in a loose sense), where ΔE_v^\ddagger is the standard deviation of the vibrational energy of the activated complex and ΔT^\ddagger is the standard deviation of its lifetime. In this approach, called UGB, the weight of each trajectory depends on the weight of all the others. Therefore, trajectories “feel” each other, a bit like in semiclassical approaches where trajectories are assigned phases that allow them to interfere. Consequently, UGB has a more quantum spirit than GB, which considers each trajectory independently on the others (like standard binning). Its application to the calculation of the reaction probability for a model $H + H_2$ reaction yields results in very good agreement with quantum scattering calculations.

A current limitation of UGB is that it cannot be applied below adiabatic thresholds if one uses real-time trajectories. To remedy this flaw, the approach must be coupled with a treatment of tunneling through adiabatic barriers. It would also be of interest to run imaginary-time trajectories and assess whether such calculations are not only feasible but also yield accurate UGB probabilities well below adiabatic barrier tops. Even so, a tunneling treatment would have to be applied right at the top of the adiabatic ground state barrier in order to set the value of α in the UGB uncertainty relation $\Delta E_v^\ddagger \Delta T^\ddagger = \alpha \hbar/2$. Note that at this energy, even a tunneling treatment within the parabolic limit is expected to be accurate. Of course, we intend to apply these ideas to realistic three-dimensional processes in the near future.³⁷

In the past, several so-called active methods have been proposed to prevent ZPE leakage along a trajectory.^{38–46} The present study tends to show that there is no reason why one should prevent ZPE leakage if the vibrational motion occurs below the ZPE a period of time shorter than the vibrational period (see Fig. 12 and the related discussion). Therefore, we believe that the only case where ZPE leakage is an issue is when a trajectory slowly crosses a TS below its ZPE, or reaches the products (or the reformed reagents) below their ZPE. Standard GB solves the second issue and a treatment of tunneling can help solve the first—or possibly, UGB with imaginary-time trajectories, as suggested above.

Conflicts of interest

The authors have no conflicts to disclose.

Data availability

The data that support the findings of this study are available from the authors upon reasonable request.

Appendices

The standard deviations ΔE_v^\ddagger and ΔT^\ddagger of E_v^\ddagger and T^\ddagger are given by

$$\Delta X^\ddagger = \sqrt{X^{\ddagger 2} - \bar{X}^{\ddagger 2}} \quad (\text{A.1})$$

with

$$\bar{X}^{\ddagger 2} = \frac{\sum_{k=1}^{N'} G_k^\ddagger X_k^{\ddagger 2}}{\sum_{k=1}^{N'} G_k^\ddagger} \quad (\text{A.2})$$

and

$$\bar{X}^\ddagger = \frac{\sum_{k=1}^{N'} G_k^\ddagger X_k^\ddagger}{\sum_{k=1}^{N'} G_k^\ddagger}, \quad (\text{A.3})$$

$X = E_v$, T . Note that these expressions are different from eqn (61)–(64) in Section IV.B.4 of ref. 11.

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- 29 In ref. 11, the QM curves shown in Fig. 6 do not correspond to the masses 19 (upper panel) and 1.9 (lower panel), but to one tenth of these masses. The SCIVR results had to be arbitrarily divided by 2 to be in good agreement with the (wrong) QM results and we attributed that to a failure of the SCIVR approach. This attribution was in fact erroneous since doubling the height of the green dots in Fig. 6 of ref. 11 makes them in satisfactory agreement with the (correct) QM results of the present work. Importantly, none of the analyses or conclusions of ref. 11 are affected by this mistake.
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