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## Backbone engineering in the hydrophobic core of villin headpiece

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Changing the backbone connectivity of proteins can impart useful new traits while maintaining essential structural and functional features. In design of artificial proteomimetic agents, backbone modification is usually isolated to sites that are solvent-exposed in the folded state, as similar changes at buried residues can alter the fold. Recent work has shown that core backbone modification without structural perturbation is possible; however, the modifications in that study were consistently destabilizing and made in a prototype of exceptionally high conformational stability. Here, we report efforts to broaden the scope and improve the efficacy of core backbone engineering by applying it to the C-terminal subdomain of villin headpiece. A series of variants are prepared in which different artificial residue types are incorporated at core positions throughout the sequence, including a crucial aromatic triad. Impacts on folding energetics are quantified by biophysical methods, and high-resolution structures of several variants determined by NMR. We go on to construct a variant with ~40% of its core modified that adopts a fold identical to the prototype while showing enhanced thermodynamic stability.

## Introduction

Engineering the chemical composition of peptides and proteins can yield biomimetic macromolecules with intricate folded structures and potent biological functions.<sup>1–4</sup> Expanding beyond ribosomally encoded amino acids as the monomer pool for construction of protein mimetics has been used to hone structure–activity relationships as well as facilitate biomedical applications.<sup>5,6</sup> One chemical change commonly applied in such work is the incorporation of amino acids with modified backbones.<sup>7–9</sup> Examples include alteration of stereochemistry (*i.e.*, D- $\alpha$ -residues), addition of new substituents (*e.g.*, N-Me- $\alpha$ -, N-amino- $\alpha$ -, C <sub>$\alpha$</sub> -Me- $\alpha$ -residues), and backbone elongation (*e.g.*,  $\beta$ -residues). Compared to side-chain substitution, altering backbone connectivity can lead to large effects on folded structure, folding energetics, and biological properties from relatively small chemical changes.<sup>10</sup>

While modification of backbone composition has proven utility in construction of artificial peptide and protein mimetics, moving from an isolated secondary structure (*e.g.*, a helical peptide) to a higher order tertiary fold (*e.g.*, a small protein) as the target for mimicry poses challenges.<sup>1</sup> In tertiary structures, local conformational preferences must be considered alongside a complex network of weak, long-range interactions critical to the fold. Related to this challenge, backbone modification in the

hydrophobic core of sequences with tertiary folding patterns has been shown to lead to a variety of unexpected effects.<sup>11–13</sup>

Seeking to broaden the scope of strategies for protein mimetic design to encompass reliable modification within the hydrophobic core, we recently reported efforts to engineer the backbone of the G-related albumin-binding module from bacterial protein PAB (1PRB).<sup>14</sup> In that work, we showed judicious backbone alteration in the core could be structurally tolerated but was consistently destabilizing. While promising, these results left open questions. Would the design strategies prove viable in a protein domain with a more delicate fold than 1PRB, which has a thermal unfolding temperature ~90 °C? Further, would expanding the pool of artificial monomers used in the construction of core-modified protein mimetics enable the creation of variants with enhanced conformational stability relative to the prototype? Here, we report efforts to address these questions through the application of backbone modification at core positions of the C-terminal subdomain of villin headpiece (**HP35**, Fig. 1).<sup>15–17</sup>

**HP35** is one of the smallest domains that folds autonomously and enjoys a rich history as a model for fundamental studies of protein folding.<sup>18–29</sup> In the realm of protein mimetics, backbone modification in **HP35** has been made at solvent-exposed sites, and the structural and energetic effects of  $\beta$ -residue and C <sub>$\alpha$</sub> -Me- $\alpha$ -residue incorporation described by Gellman<sup>30</sup> and our lab,<sup>31</sup> respectively. Backbone modification to the **HP35** core has not been reported; however, Berlicki described a computationally designed *de novo* protein mimetic

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that contained several carbocyclic  $\beta$ -residues in its core and was intended to mimic **HP35**.<sup>13</sup> Interestingly, this sequence was observed to form a domain-swapped dimer fold that deviated significantly from the design.<sup>13</sup> In the present study, we set out to systematically examine the impacts of altered backbone composition in the hydrophobic core of **HP35** on folded structure and stability of the protein. Our results show artificial residues can be accommodated at several core positions in this system—individually or in tandem—without altering folded structure and that some modifications enhance folded stability relative to the canonical backbone.

## Results and discussion

### Design of core-modified **HP35** variants

The hydrophobic core of **HP35** (Fig. 1(A) and (B)) is primarily composed of three phenylalanine side chains (F6, F10, F17), which are located within the first and second helices and form an aromatic cluster that is critical to stabilizing the tertiary fold of the domain.<sup>32</sup> Solvent-accessible surface area (SASA) analysis shows these three residues are each >90% buried. In addition to the phenylalanine triad, M12 in the loop between helix 1 and 2, L20 in the loop between helix 2 and 3, and L28 in helix 3 are partially buried (74–81%). Finally, polar residues Q25 and K29 in helix 3 occupy hydrophobic core-flanking positions and are also partially buried (67–85%). Based on the above analysis, we selected five sites from the core of **HP35** as targets for backbone modification (Fig. 1(A)): the key aromatic cluster F6, F10 and F17, buried aliphatic L28, and interfacial charged K29.

Three different artificial residue types were employed in design of core-modified **HP35** variants:  $\beta^3$ -residues, chiral  $C_\alpha$ -Me- $\alpha$ -residues, and the cyclic  $\beta$ -residue *trans*-2-aminocyclopentane

carboxylic acid (ACPC) (Fig. 1(C)). All three of these classes are known to be well accommodated in solvent-exposed helical contexts in a variety of heterogeneous-backbone tertiary structure mimetics, including **HP35**.<sup>30,31</sup> Relative to a canonical  $\alpha$ -residue, a methylene group is inserted between the carbonyl and  $C_\alpha$  in a  $\beta^3$  monomer, which adds a rotatable bond and increases backbone flexibility. In a  $C_\alpha$ -Me- $\alpha$  residue, the  $H_\alpha$  is replaced by a methyl group, which restricts backbone conformational freedom. In the cyclic  $\beta$ -residue ACPC, the cyclopentane ring constrains the central torsion, which increases rigidity relative to the  $\beta^3$  residue at the expense of a lost side chain. Collectively, the application of these three monomer types allows for a comparative analysis of the importance of rigidity and side chain functionality when modifying the backbone at buried core positions.

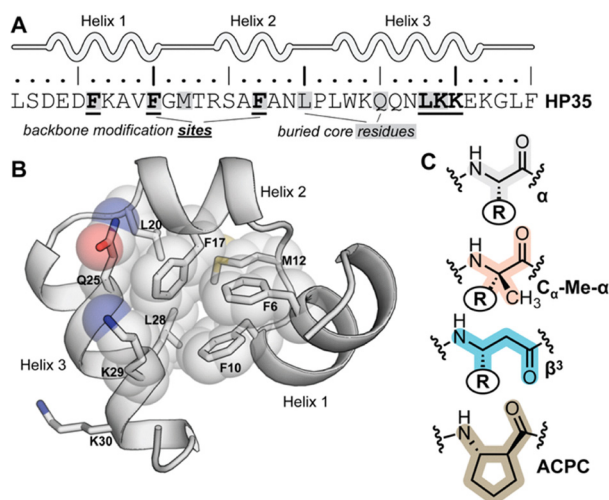
For **HP35** variants modified at core position L28 and core-flanking position K29, the canonical  $\alpha$ -residue was replaced by a  $\beta^3$ -residue or ACPC. As a control to understand context-dependent effects of these modifications at a comparable non-core site, we incorporated the same monomer types at neighboring solvent-exposed residue K30. Given the crucial role of side-chain packing of the phenylalanine triad in **HP35**, backbone substitutions at those positions were limited to  $\beta^3$  and  $C_\alpha$ -Me- $\alpha$  monomers that maintain the side chain.

The above design considerations yielded a set of 12 single-site backbone substitution variants of **HP35** for analysis. As data for the  $\beta^3$ K30 variant have been reported,<sup>30</sup> the remaining 11 peptides were prepared by Fmoc solid-phase methods. All syntheses were completed without major issues and yielded good crude purities (Fig. S1). Peptides were purified by preparative reverse phase high-performance liquid chromatography (HPLC). The identity of isolated products was confirmed by electrospray ionization mass spectrometry (ESI-MS) and purity assessed by analytical HPLC (Fig. S2–S13) prior to biophysical and structural characterization. To avoid complications from oxidation during storage and handling, M12 was replaced by norleucine in the **HP35** prototype and all variants.

### $\beta$ -Residue substitution at a core, core-flanking, and solvent-exposed site

We first examined folding behavior of the **HP35** variants modified in helix 3, comparing effects of  $\beta^3$  or ACPC substitution at three neighboring sites: core L28, core-flanking K29, and solvent-exposed K30. Results for the variants were benchmarked against our recently reported biophysical data set for prototype **HP35**.<sup>31</sup> Circular dichroism (CD) scans at 20 °C on samples consisting of 50  $\mu$ M peptide in 50 mM phosphate pH 7 show minima at  $\sim$ 208 and  $\sim$ 222 nm consistent with the predominantly helical fold of the prototype; however, the magnitude of the CD peaks in  $\beta^3$ L28 and  $\beta^3$ K29 is slightly attenuated (Fig. 2(A)).

Thermal unfolding experiments monitoring CD signal at 222 nm as a function of temperature exhibit cooperative unfolding transitions in all cases (Fig. 2(B)), with temperature midpoint ( $T_m$ ) values ranging from 47–69 °C (Table 1). Backbone modification at core position L28 results in a consistent reduction in the thermal stability of the fold ( $\Delta T_m$  –19 °C for



**Fig. 1** (A) Sequence of **HP35**. Residues shaded in gray make up the hydrophobic core, and underlined residues indicate sites selected for backbone modification. (B) Crystal structure of **HP35** (PDB 3TRY); core side chains are shown as spheres. (C) Chemical structures of an  $\alpha$ -residue and three artificial monomer types used in core-modified **HP35** analogues. The R group in each  $C_\alpha$ -Me- $\alpha$  and  $\beta^3$  residue matches the side chain of the  $\alpha$ -residue in **HP35** it replaces.



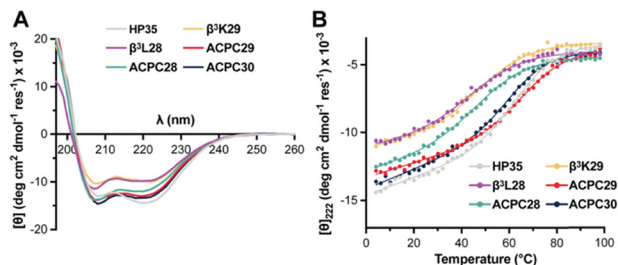


Fig. 2 CD scans at 20 °C (A) and thermal melts (B) for **HP35** and indicated variants. Conditions: 50 μM peptide in 50 mM phosphate buffer, pH 7.

**β³L28** and  $-14$  °C for **ACPC28**). At core-flanking position K29, variant **β³K29** is comparably destabilized ( $\Delta T_m -15$  °C), while **ACPC29** shows a small increase in thermal stability relative to prototype ( $\Delta T_m +3$  °C). For solvent-exposed position 30, ACPC incorporation is slightly destabilizing ( $\Delta T_m$  of  $-2$  °C), and prior results indicate  $\beta^3$  substitution is moderately destabilizing ( $\Delta T_m$  of  $-12$  °C).<sup>30</sup>

To determine the effects of backbone modification at core residues on the folded structure of **HP35**, we subjected the helix-3 variants to analysis by NMR spectroscopy. 1-Dimensional  $^1\text{H}$  and 2-dimensional  $^1\text{H}/^1\text{H}$  COSY, TOCSY, and NOESY spectra were recorded for **β³L28**, **β³K29**, **ACPC28**, **ACPC29**, and **ACPC30** at 1 mM peptide in 9 : 1  $\text{H}_2\text{O}/\text{D}_2\text{O}$  at pH 5 (Fig. S15–S20). Assignment of proton resonances followed by simulated annealing with restraints derived from the NMR measurements yielded a 10-model structure ensemble for each peptide (Fig. 3 and Tables S1–S5).

All the helix-3 variants adopt tertiary folds close to that of **HP35** (0.5–0.8 Å backbone rmsd to the X-ray structure). Further support for this finding comes from observed backbone  $\text{H}_\alpha$  chemical shifts throughout the domain, which are also very similar to the prototype (Fig. S27). Inspection of hydrophobic core packing shows that modification at L28 does not perturb

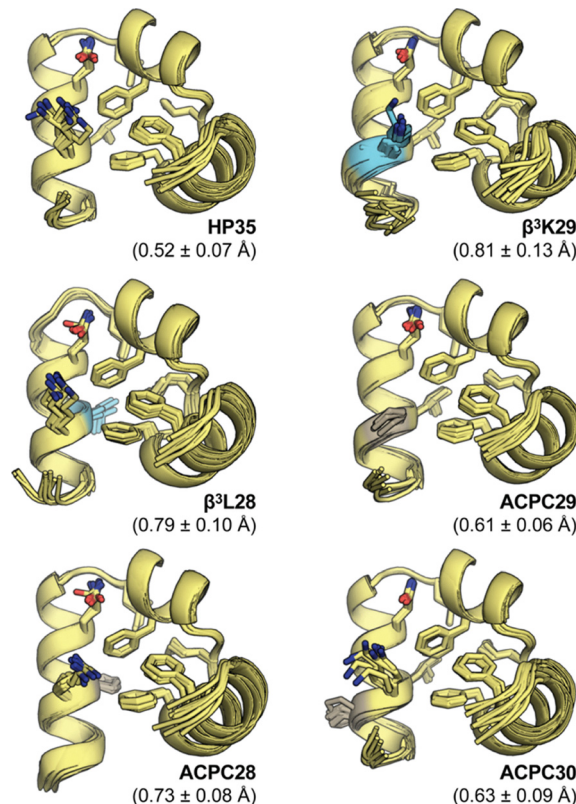


Fig. 3 NMR structure ensembles for **HP35** (PDB 9MF7) and variants. Backbone rmsd (average and standard deviation for residue range 1–33) for overlay of each ensemble to the X-ray structure of **HP35** (PDB 3TRY) is shown in parentheses.

the key aromatic triad; however, the artificial residues pack less efficiently (94%, 89%, and 70% burial for the  $\alpha$ , ACPC, and  $\beta^3$  residue at position 28, respectively) (Fig. S28). This reduction in packing efficiency may give rise to the thermal stability loss

Table 1 Biophysical properties of **HP35** and variants<sup>a</sup>

|  | $T_m$ (°C)              | $\Delta T_m$ (°C)    | $\Delta G_{\text{fold}}^\circ$ (kcal mol <sup>-1</sup> ) | $\Delta\Delta G_{\text{fold}}^\circ$ (kcal mol <sup>-1</sup> ) |
|--|-------------------------|----------------------|--|--|
| <b>HP35</b>                                      | 65.6 ± 0.7              | —                    | -2.14 ± 0.12   | 0  |
| <b>β³L28</b>                                     | 47.0 ± 2.6              | -19 ± 3              | —  | —  |
| <b>ACPC28</b>                                    | 51.2 ± 1.3              | -14 ± 1              | —  | —  |
| <b>β³K29</b>                                     | 50.6 ± 1.7              | -15 ± 2              | —  | —  |
| <b>ACPC29</b>                                    | 68.7 ± 0.7              | +3 ± 1               | -2.38 ± 0.15   | -0.2 ± 0.2   |
| <b>β³K30</b>                                     | 57.2 ± 1.7 <sup>b</sup> | -12 ± 2 <sup>b</sup> | —  | —  |
| <b>ACPC30</b>                                    | 63.3 ± 0.8              | -2 ± 1               | —  | —  |
| <b>β³F6</b>                                      | 36.3 ± 3.5              | -29 ± 4              | —  | —  |
| <b>β³F10</b>                                     | 47.3 ± 2.5              | -18 ± 3              | —  | —  |
| <b>β³F17</b>                                     | —                       | —                    | —  | —  |
| <b>α<sup>Mc</sup>F6</b>                          | 71.5 ± 0.7              | +6 ± 1               | -3.63 ± 0.26   | -1.5 ± 0.3   |
| <b>α<sup>Mc</sup>F10</b>                         | 45.3 ± 5.0              | -20 ± 5              | —  | —  |
| <b>α<sup>Mc</sup>F17</b>                         | 57.4 ± 1.3              | -8 ± 1               | -0.90 ± 0.21   | +1.2 ± 0.2   |
| <b>α<sup>Mc</sup>F6/α<sup>Mc</sup>F17/ACPC29</b> | —                       | —                    | -3.00 ± 0.24   | -0.9 ± 0.3   |

<sup>a</sup> Temperature midpoint ( $T_m$ ) of the thermal unfolding transition and change in  $T_m$  relative to **HP35** ( $\Delta T_m$ ) determined by variable temperature CD (conditions: 50 μM peptide in 50 mM phosphate buffer, pH 7). Folding free energy ( $\Delta G_{\text{fold}}^\circ$ ) and change in folding free energy relative to **HP35** ( $\Delta\Delta G_{\text{fold}}^\circ$ ) determined *via* chemical denaturation with guanidinium chloride monitored by tryptophan fluorescence (conditions: 10 μM peptide in 50 mM phosphate buffer, pH 7, 25 °C). Uncertainties for  $T_m$  and  $\Delta G_{\text{fold}}^\circ$  are standard errors from the fits and uncertainties for  $\Delta T_m$  and  $\Delta\Delta G_{\text{fold}}^\circ$  determined by error propagation. <sup>b</sup> Data sourced from published report;<sup>30</sup> the sequence (M12/H27) and buffer composition (acetate pH 5) differed slightly from the present work, so the  $\Delta T_m$  value is reported relative to the prototype  $T_m$  from that study.



observed by CD. At core-flanking position 29, both the  $\beta^3$ K side chain as well as the ACPC ring engage core residues F10 and F17 in a manner similar to K29 in **HP35**.

From a structural perspective, all the modifications examined in helix 3 of **HP35** are well tolerated. Even when backbone composition is altered at buried aliphatic L28, the variant folded structure is indistinguishable from the prototype. With respect to stability, effects of backbone modification vary with monomer type and sequence context. Results for the  $\beta^3$  variants point to a similar impact on thermal stability from substitution at the core, core-flanking, or solvent-exposed site. This suggests the thermal stability loss from  $\beta^3$  substitution arises primarily from properties inherent to the monomer, such as enhanced flexibility. In contrast, effects of ACPC substitution are context dependent—destabilizing at core site L28 but stabilizing at neighboring core-flanking K29. This finding indicates long-range interactions involving the ACPC side chain may influence fold stability in **HP35**. Prior studies have shown that methionine or norleucine substitution at K29 stabilizes **HP35** due to elimination of unfavorable electrostatic interactions.<sup>33</sup> Incorporation of ACPC at this position may provide a similar benefit. Correlations between the side chain of K29 and the aromatic rings of F10 and F17 are observed in the NOESY spectrum of **HP35**, and similar interactions are seen for the ACPC residue in variant **ACPC29** (Fig. S29). These results support the hypothesis that long-range interactions involving residue 29 are important to the folded stability of **HP35**.

### $\beta$ -Residue substitution in the **HP35** core aromatic triad

Given  $\beta$ -residue incorporation was viable at aliphatic core position L28 in **HP35**, we hypothesized it might also be tolerated in the crucial aromatic core cluster of F6, F10, and F17. To test this hypothesis, each of these residues was individually replaced by a  $\beta^3$  analogue. CD scans and thermal melts suggest that  $\alpha \rightarrow \beta^3$  substitution in the aromatic triad is consistently disruptive, though the degree depends on the site modified (Fig. 4(A), (B) and Table 1).  $\beta^3$ F6 and  $\beta^3$ F10 show evidence for a predominantly helical tertiary fold but are both significantly destabilized relative to the prototype ( $\Delta T_m = -30$  °C and  $-20$  °C, respectively). Backbone alteration in  $\beta^3$ F17 leads to CD spectral properties characteristic of a random coil and complete loss of a cooperative thermal folding transition, suggesting this variant does not adopt an ordered fold.

Comparison of the NMR spectra of  $\beta^3$ F6 and  $\beta^3$ F10 to that of **HP35** shows reduced peak dispersion and increased peak broadening (Fig. S21 and S22). This effect suggests exchange among multiple ordered conformational states on an intermediate time scale.<sup>34,35</sup> Peak overlap and broadening in  $\beta^3$ F10 precluded resonance assignment; however, we were able to generate the NMR structure for  $\beta^3$ F6 (Fig. 4(C) and Table S6).  $\beta^3$ F6 adopts a predominantly helical fold; however, tertiary packing is altered relative to prototype **HP35**. A shift in the  $\beta^3$ F side chain away from the core is accompanied by an increase in local conformational heterogeneity in helix 1 and a change in the position of helix 1 relative to the remainder of the protein (Fig. 4(D)). Additional experimental support for a

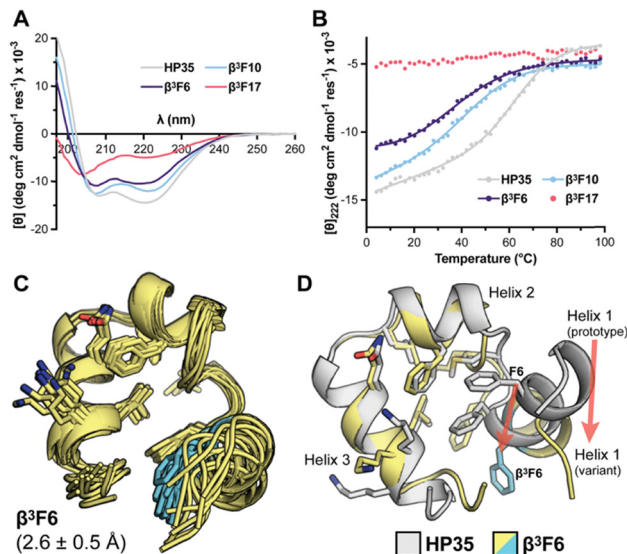


Fig. 4 (A), (B) CD scans at 20 °C (A) and thermal melts (B) for **HP35** and variants modified with  $\beta^3$  residues in the core aromatic triad. Conditions: 50  $\mu$ M peptide in 50 mM phosphate buffer, pH 7. (C) NMR structure ensemble for  $\beta^3$ F6; backbone rmsd (average and standard deviation; residue range 1–33) for overlay of the ensemble to the X-ray structure of **HP35** (PDB 3TRY) is shown in parentheses. (D) Overlay of the lowest energy model from the NMR ensemble of  $\beta^3$ F6 with the X-ray structure of **HP35** (PDB 3TRY). Alignment is based on helices 2 and 3, and arrows highlight repositioning of helix 1, which contains the F  $\rightarrow$   $\beta^3$ F modification.

greater degree of disorder in this variant comes from the  $H_\alpha$  chemical shifts, which show a lower magnitude deviation from random coil values throughout the domain than in any other folded variant (Fig. S27). The structural rearrangement of helices is accompanied by a change in composition of the hydrophobic core. Residues  $\beta^3$ F6, F10, and K29 become more exposed in the variant, while residues T13, K24, and K32 become more buried and form new long-range contacts.

Overall, these findings show that  $\beta^3$  residue incorporation in the core aromatic triad of **HP35** is significantly disruptive to both folded structure and folded stability. The differences among the three  $\beta^3$ F variants reinforce the idea that the position of modification is crucial. Among the three isomers ( $\beta^3$ F6,  $\beta^3$ F10,  $\beta^3$ F17), which differ only in the placement of a single methylene group in the backbone, one adopts an ordered but non-native fold, one a partially ordered molten globule, and the other a disordered random coil.

### $C_\alpha$ -Me- $\alpha$ -residue substitution in the **HP35** core aromatic triad

Given  $\beta^3$  substitution at core aromatic positions in **HP35** was not tolerated, we wondered whether an alternate monomer might be better accommodated at these sites. Thus, we examined a corresponding set of variants in which each of the key phenylalanine residues was individually replaced by a chiral  $C_\alpha$ -Me- $\alpha$  analogue bearing the same side chain. Given chiral  $C_\alpha$ -methyl residues are strong helix inducers,<sup>31,36–42</sup> we were interested to see how side chain retaining  $\alpha \rightarrow C_\alpha$ -Me substitution would compare to  $\alpha \rightarrow \beta^3$  replacement when targeted to critical buried core positions in **HP35**.



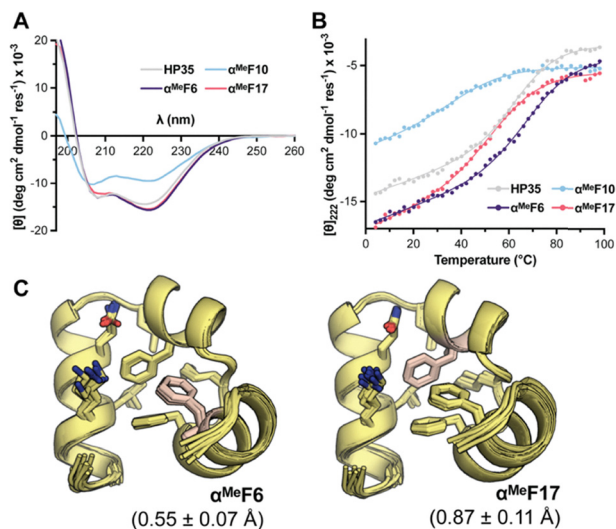


Fig. 5 (A), (B) CD scans at 20 °C (A) and thermal melts (B) for **HP35** and variants modified with  $C_{\alpha}$ -Me- $\alpha$  residues in the core aromatic triad. Conditions: 50  $\mu\text{M}$  peptide in 50 mM phosphate buffer, pH 7. (C) NMR structure ensembles for  $\alpha^{\text{Me}}\text{F6}$  and  $\alpha^{\text{Me}}\text{F17}$ ; backbone rmsd (average and standard deviation; residue range 1–33) for overlay of each ensemble to the X-ray structure of **HP35** (PDB 3TRY) is shown in parentheses.

CD spectra of variants  $\alpha^{\text{Me}}\text{F6}$  and  $\alpha^{\text{Me}}\text{F17}$  are similar to **HP35** (Fig. 5(A)), while that of  $\alpha^{\text{Me}}\text{F10}$  suggests significantly reduced helicity. Variable-temperature CD measurements exhibit cooperative folding transitions; however, the folded baseline for  $\alpha^{\text{Me}}\text{F10}$  is poorly defined (Fig. 5(B)).  $T_m$  values range from 45–72 °C (Table 1), and relative thermal stabilities follow the trend  $\alpha^{\text{Me}}\text{F6} > \text{HP35} > \alpha^{\text{Me}}\text{F17} > \alpha^{\text{Me}}\text{F10}$ . Folded structures of  $\alpha^{\text{Me}}\text{F6}$  and  $\alpha^{\text{Me}}\text{F17}$  determined by NMR show tertiary folds close to **HP35** (Fig. 5(C) and Tables S7, S8). Packing of the aromatic triad in these variants is unchanged compared to the prototype, including at the artificial  $C_{\alpha}$ -Me- $\alpha$  residues. The  $\alpha$ -methyl group at each of these sites is buried in the core, in close contact with other side chains. In  $\alpha^{\text{Me}}\text{F6}$ , the organization of the remainder of the core is indistinguishable from **HP35**. In the case of  $\alpha^{\text{Me}}\text{F17}$ , the side chains of norleucine 12 and L20 shift slightly to accommodate the new methyl group. While subtle, this change may give rise to the observed decrease in thermal stability. NMR spectra of  $\alpha^{\text{Me}}\text{F10}$  show significant peak overlap compared to the other  $\alpha^{\text{Me}}\text{F}$  variants (Fig. S23–S25), preventing resonance assignment. Taken with a low intensity CD signature and minimal thermal unfolding cooperativity, the small chemical shift dispersion suggests this variant does not adopt an ordered tertiary fold. The origin of the destabilization of  $\alpha^{\text{Me}}\text{F10}$  relative to the other two  $\alpha^{\text{Me}}\text{F}$  variants is unclear.

### Generation of an **HP35** variant with a backbone modification throughout the core

Given that several individual core sites in **HP35** proved amenable to backbone modification, we next sought to establish whether multiple core residues could be replaced in tandem. Thus, we synthesized and characterized a triple substitution

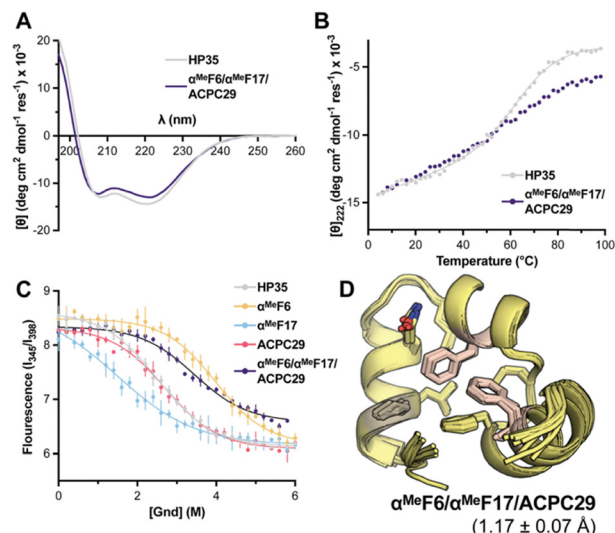


Fig. 6 (A), (B) CD scans at 20 °C (A) and thermal melts (B) for **HP35** and triple-modified variant  $\alpha^{\text{Me}}\text{F6}/\alpha^{\text{Me}}\text{F17}/\text{ACPC29}$ . Conditions: 50  $\mu\text{M}$  peptide in 50 mM phosphate buffer, pH 7. (C) Chemical denaturation of **HP35**, the triple-modified variant, and the single-site variants  $\alpha^{\text{Me}}\text{F6}$ ,  $\alpha^{\text{Me}}\text{F17}$ , and **ACPC29** by guanidinium chloride (Gnd) monitored by tryptophan fluorescence. Conditions: 10  $\mu\text{M}$  peptide in 50 mM phosphate buffer, pH 7. Data points are average and standard deviation from triplicate measurements. (D) NMR structure ensemble for  $\alpha^{\text{Me}}\text{F6}/\alpha^{\text{Me}}\text{F17}/\text{ACPC29}$ ; backbone rmsd (average and standard deviation; residue range 1–33) for overlay to the X-ray structure of **HP35** (PDB 3TRY) is shown in parentheses.

variant  $\alpha^{\text{Me}}\text{F6}/\alpha^{\text{Me}}\text{F17}/\text{ACPC29}$ . With one modification in each of the three helices, this analogue has a core that is 38% artificial in backbone composition. As detailed above, each of the corresponding individual substitutions combined in this sequence are innocuous with respect to the folded structure and two are thermally stabilizing.

CD scans suggest that the triple variant adopts a fold very similar to **HP35** (Fig. 6(A)). Thermal melts reveal an exceptionally high thermal stability, with the almost complete absence of a defined unfolding transition (Fig. 6(B)). To obtain quantitative insights into thermodynamic effects of the altered backbone composition in this analogue, we compared the free energy of folding ( $\Delta G^\circ$ ) for it, the prototype and each corresponding single-site variant (*i.e.*,  $\alpha^{\text{Me}}\text{F6}$ ,  $\alpha^{\text{Me}}\text{F17}$ , and **ACPC29**) through chemical denaturation with guanidinium chloride monitored by tryptophan fluorescence (Fig. 6(C)). All five sequences showed a cooperative unfolding transition. Folding free energies ( $\Delta G^\circ_{\text{fold}}$ ) obtained from the fits follow a similar trend as  $T_m$  (Table 1). Relative to **HP35**, single-site variant  $\alpha^{\text{Me}}\text{F6}$  is significantly stabilized ( $\Delta\Delta G^\circ -1.5$  kcal mol<sup>-1</sup>), **ACPC29** is comparable in stability ( $\Delta\Delta G^\circ -0.2$  kcal mol<sup>-1</sup>), and  $\alpha^{\text{Me}}\text{F17}$  is slightly destabilized ( $\Delta\Delta G^\circ +1.2$  kcal mol<sup>-1</sup>). The triple mutant is also more thermodynamically stable than the prototype ( $\Delta\Delta G^\circ -0.9$  kcal mol<sup>-1</sup>), and more stable than all but one of the single-site variants. NMR analysis for the  $\alpha^{\text{Me}}\text{F6}/\alpha^{\text{Me}}\text{F17}/\text{ACPC29}$  triple variant (Fig. S26 and Table S9) shows that, despite substantial fraction of the core being backbone modified, it adopts a very similar fold to **HP35** (Fig. 6(D)).



## Conclusions

In summary, we have reported here a systematic examination of the impacts of backbone modification in the hydrophobic core of the C-terminal subdomain of the villin headpiece on the folded structure and conformational stability of the protein. We find that backbone alteration through substitution of  $\alpha$ -residues with  $\beta^3$ , ACPC, or  $C_\alpha$ -Me- $\alpha$  analogues can be made at core or core-flanking positions without perturbing the fold of the protein. The sequence context for a given backbone modification is an important factor in determining its exact effect; however, even residues with side chains that engage in crucial tertiary packing interactions in the hydrophobic core can be replaced. Many of the single-site variants are destabilized relative to the prototype, but some show improved conformational stability. Combining three well tolerated modifications into a single construct yields a variant with 38% of its core residues modified, a fold virtually identical to the prototype, and significantly enhanced thermal stability and folding free energy.

Compared to prior work on the application of backbone modification at core positions in villin and other domains, the fact that many of the variants adopt folds very similar to prototype is noteworthy. Further, the enhanced stability seen for several of the variants is significant. Backbone modification in tertiary structure contexts is often destabilizing—even when made at solvent-exposed sites. Examples of tertiary structure mimetics with multiple backbone modifications and enhanced conformational stability relative to a prototype protein are rare.<sup>43–46</sup> Achieving this outcome when the modification sites are in the core of the domain is noteworthy, considering the unique composition of the hydrophobic core, the small size, and the moderate unfolding free energy of **HP35**. Collectively, the present results demonstrate that the entirety of a sequence—both buried core residues and solvent-exposed sites—should be considered fair game in the design of heterogeneous-backbone protein mimetics in the context of  $\alpha$ -helical secondary structure.

A significant challenge in development of protein mimetics based on artificial backbones is the incorporation of non-proteogenic amino acids. Although methods for ribosomal synthesis with non-canonical backbones continue to advance,<sup>47</sup> chemical synthesis remains the method of choice for larger sequences with multiple modifications. Reliance on total chemical synthesis poses limitations to the size of entities that can be produced and, thus, the complexity of folds that can be mimicked. Methods such as native chemical ligation and semi-synthesis can expand the chain lengths accessible,<sup>48</sup> however, spontaneous refolding of large synthetic constructs absent the typical cellular milieu can prove difficult.<sup>49</sup> The above technical challenges notwithstanding, studies on the folding behavior of artificial protein-like backbones—such as the present work—show what is possible in these entities and provides insights into their design. Improved understanding of these fundamental issues along with continued advances in methods for construction of protein-like artificial macromolecules will help to realize the full potential of proteomimetics as a concept.<sup>1</sup>

## Conflicts of interest

There are no conflicts to declare.

## Abbreviations

|        |   |
|--------|---|
| 1PRB   | G-Related albumin-binding module from bacterial protein PAB |
| ACPC   | <i>trans</i> -2-Aminocyclopentane carboxylic acid           |
| BMRB   | Biological magnetic resonance bank                          |
| CD     | Circular dichroism  |
| COSY   | Correlated spectroscopy                                     |
| ESI-MS | Electrospray ionization mass spectrometry                   |
| Gnd    | Guanidinium chloride  |
| HP35   | Villin headpiece  |
| HPLC   | High-performance liquid chromatography                      |
| NMR    | Nuclear magnetic resonance                                  |
| NOESY  | Nuclear Overhauser effect spectroscopy                      |
| PDB    | Protein data bank   |
| rmsd   | Root mean square deviation                                  |
| SASA   | Solvent-accessible surface area                             |
| TOCSY  | Total correlation spectroscopy                              |

## Data availability

Coordinates and experimental data for newly reported NMR structures are deposited in the PDB (9YLZ, 9YM0, 9YM1, 9YM2, 9YM3, 9YM4, 9YM5, 9YM6, 9YM7) and BMRB (31270, 31271, 31272, 31273, 31274, 31275, 31276, 31277, 31278). Other data are available from the corresponding author upon request.

Additional data supporting the findings of this study can be found in the SI for the article. Supplementary information: Fig. S1–S29, Tables S1–S9, materials and methods. See DOI: <https://doi.org/10.1039/d5cb00269a>.

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## Notes and references

- W. S. Horne and T. N. Grossmann, *Nat. Chem.*, 2020, **12**, 331–337.
- C. S. Swenson, G. Mandava, D. M. Thomas and R. E. Moellering, *Chem. Rev.*, 2024, **124**, 13020–13093.
- S. H. Hong, T. Nguyen, J. F. Ongkingco, A. Nazzaro and P. S. Arora, *Chem. Rev.*, 2025, **125**, 6819–6869.
- L. Lombardi, V. D. Genio, F. Albericio and D. R. Williams, *Chem. Rev.*, 2025, **125**, 7099–7166.
- E. Lenci and A. Trabocchi, *Chem. Soc. Rev.*, 2020, **49**, 3262–3277.
- J. L. Hickey, D. Sindhikara, S. L. Zultanski and D. M. Schultz, *ACS Med. Chem. Lett.*, 2023, **14**, 557–565.



- 7 Z. E. Reinert and W. S. Horne, *Org. Biomol. Chem.*, 2014, **12**, 8796–8802.
- 8 K. L. George and W. S. Horne, *Acc. Chem. Res.*, 2018, **51**, 1220–1228.
- 9 P. Sang and J. Cai, *Chem. Soc. Rev.*, 2023, **52**, 4843–4877.
- 10 M. M. Müller, *Biochemistry*, 2018, **57**, 177–185.
- 11 W. S. Horne, J. L. Price and S. H. Gellman, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 9151–9156.
- 12 M. W. Giuliano, W. S. Horne and S. H. Gellman, *J. Am. Chem. Soc.*, 2009, **131**, 9860–9861.
- 13 M. Bejger, P. Fortuna, M. Drewniak-Świtalska, J. Plewka, W. Rypniewski and Ł. Berlicki, *Chem. Commun.*, 2021, **57**, 6015–6018.
- 14 Y. Lin and W. S. Horne, *Chem. – Eur. J.*, 2024, **30**, e202401890.
- 15 J. C. McKnight, D. S. Doering, P. T. Matsudaira and P. S. Kim, *J. Mol. Biol.*, 1996, **260**, 126–134.
- 16 C. J. McKnight, P. T. Matsudaira and P. S. Kim, *Nat. Struct. Biol.*, 1997, **4**, 180–184.
- 17 D. E. Mortenson, K. A. Satyshur, I. A. Guzei, K. T. Forest and S. H. Gellman, *J. Am. Chem. Soc.*, 2012, **134**, 2473–2476.
- 18 Y. Duan and P. A. Kollman, *Science*, 1998, **282**, 740–744.
- 19 C. D. Snow, H. Nguyen, V. S. Pande and M. Gruebele, *Nature*, 2002, **420**, 102–106.
- 20 J. Kubelka, W. A. Eaton and J. Hofrichter, *J. Mol. Biol.*, 2003, **329**, 625–630.
- 21 M. Wang, Y. Tang, S. Sato, L. Vugmeyster, C. J. McKnight and D. P. Raleigh, *J. Am. Chem. Soc.*, 2003, **125**, 6032–6033.
- 22 A. Fernández, M.-y Shen, A. Colubri, T. R. Sosnick, R. S. Berry and K. F. Freed, *Biochemistry*, 2003, **42**, 664–671.
- 23 S. H. Brewer, D. M. Vu, Y. Tang, Y. Li, S. Franzen, D. P. Raleigh and R. B. Dyer, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 16662–16667.
- 24 H. Lei, C. Wu, H. Liu and Y. Duan, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 4925–4930.
- 25 D. L. Ensign, P. M. Kasson and V. S. Pande, *J. Mol. Biol.*, 2007, **374**, 806–816.
- 26 D. E. Shaw, P. Maragakis, K. Lindorff-Larsen, S. Piana, R. O. Dror, M. P. Eastwood, J. A. Bank, J. M. Jumper, J. K. Salmon, Y. Shan and W. Wriggers, *Science*, 2010, **330**, 341–346.
- 27 K.-N. Hu, W.-M. Yau and R. Tycko, *J. Am. Chem. Soc.*, 2010, **132**, 24–25.
- 28 J. K. Chung, M. C. Thielges and M. D. Fayer, *J. Am. Chem. Soc.*, 2012, **134**, 12118–12124.
- 29 G. Žoldák, J. Stigler, B. Pelz, H. Li and M. Rief, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 18156–18161.
- 30 D. F. Kreitler, D. E. Mortenson, K. T. Forest and S. H. Gellman, *J. Am. Chem. Soc.*, 2016, **138**, 6498–6505.
- 31 T. W. Harmon, Y. Lin, R. T. Sutton, S. W. J. Osborne and W. S. Horne, *ChemBioChem*, 2025, **26**, e202401022.
- 32 B. S. Frank, D. Vardar, D. A. Buckley and C. J. McKnight, *Protein Sci.*, 2002, **11**, 680–687.
- 33 J. Kubelka, T. K. Chiu, D. R. Davies, W. A. Eaton and J. Hofrichter, *J. Mol. Biol.*, 2006, **359**, 546–553.
- 34 H. J. Dyson and P. E. Wright, *Methods Enzymol.*, 2001, **339**, 258–270.
- 35 A. K. Mittermaier and L. E. Kay, *Trends Biochem. Sci.*, 2009, **34**, 601–611.
- 36 C. Zhang, W. Miller, K. J. Valenzano and D. J. Kyle, *J. Med. Chem.*, 2002, **45**, 5280–5286.
- 37 N. A. Tavenor, Z. E. Reinert, G. A. Lengyel, B. D. Griffith and W. S. Horne, *Chem. Commun.*, 2016, **52**, 3789–3792.
- 38 R. Islam, D. O. Sviridov, S. K. Drake, J. Tunyi, G. Abdoulaeva, L. A. Freeman, R. W. Pastor and A. T. Remaley, *Biochem. Biophys. Res. Commun.*, 2020, **526**, 349–354.
- 39 V. Bauer, B. Schmidtgal, G. Gógl, J. Dolenc, J. Osz, Y. Nominé, C. Kostmann, A. Cousido-Siah, A. Mitschler, N. Rochel, G. Travé, B. Kieffer and V. Torbeev, *Chem. Sci.*, 2021, **12**, 1080–1089.
- 40 R. A. D. Bathgate, P. Praveen, A. Sethi, W. I. Furuya, R. R. Dhingra, M. Kocan, Q. Ou, A. L. Valkovic, I. Gil-Miravet, M. Navarro-Sánchez, F. E. Olucha-Bordonau, A. L. Gundlach, K. J. Rosengren, P. R. Gooley, M. Dutschmann and M. A. Hossain, *J. Am. Chem. Soc.*, 2023, **145**, 20242–20247.
- 41 A. Chandramohan, H. Josien, T. Y. Yuen, R. Duggal, D. Spiegelberg, L. Yan, Y.-C. A. Juang, L. Ge, P. G. Aronica, H. Y. K. Kaan, Y. H. Lim, A. Peier, B. Sherborne, J. Hochman, S. Lin, K. Biswas, M. Nestor, C. S. Verma, D. P. Lane, T. K. Sawyer, R. Garbaccio, B. Henry, S. Kannan, C. J. Brown, C. W. Johannes and A. W. Partridge, *Nat. Commun.*, 2024, **15**, 489.
- 42 T. W. Harmon, J. Song, A. J. Gulewicz, Y. P. Di and W. S. Horne, *ChemBioChem*, 2025, **26**, e202400951.
- 43 K. L. George and W. S. Horne, *J. Am. Chem. Soc.*, 2017, **139**, 7931–7938.
- 44 S. D. Melton, E. A. E. Brackhahn, S. J. Orlin, P. Jin and D. M. Chenoweth, *Chem. Sci.*, 2020, **11**, 10638–10646.
- 45 J. R. Santhouse, J. M. G. Leung, L. T. Chong and W. S. Horne, *Chem. Sci.*, 2022, **13**, 11798–11806.
- 46 M. M. Wright, B. H. Rajewski, T. A. Gerrein, Z. Xu, L. J. Smith, W. S. Horne and J. R. Del Valle, *Commun. Chem.*, 2025, **8**, 76.
- 47 M. Sigal, S. Matsumoto, A. Beattie, T. Katoh and H. Suga, *Chem. Rev.*, 2024, **124**, 6444–6500.
- 48 V. Agouridas, O. El Mahdi, V. Diemer, M. Cargoët, J.-C. M. Monbaliu and O. Melnyk, *Chem. Rev.*, 2019, **119**, 7328–7443.
- 49 A. J. Callahan, A. Rondon, R. M. Reja, L. L. Salazar, S. Gandhesiri, J. Rodriguez, A. Loas and B. L. Pentelute, *J. Am. Chem. Soc.*, 2024, **146**, 28696–28706.

