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Achieving clear ferroelectric polarization reversal in room-temperature multiferroic ϵ -Fe₂O₃ systems through grain boundary engineering[†]

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ϵ -Fe₂O₃-related oxides are promising room-temperature multiferroic materials owing to their significant magnetization and switchable ferroelectric polarization. However, their large leakage current hinders the quality of ferroelectric polarization reversal, limiting research despite superior magnetic properties compared to the well-studied BiFeO₃ system. This study addresses these limitations through two investigations. First, we investigate the difficulty of polarization reversal by analyzing grain boundary structures in ϵ -Fe₂O₃-related epitaxial films, which inherently form due to their non-perovskite orthorhombic structure. Second, we enhance polarization reversal *via* grain boundary engineering. Our findings reveal that these films contain numerous small grains (250–770 nm²) with Fe^{2+/3+} states at grain boundaries, where approximately 40% act as ferroelectric domain walls. The high grain boundary density causes significant leakage current and hinders polarization reversal. By implementing a codoping method, we successfully reduce the grain boundary density, achieving clear ferroelectric hysteresis. This breakthrough highlights the potential of ϵ -Fe₂O₃-related oxides as room-temperature multiferroic materials with substantial magnetization and offers new prospects for research on materials distinct from the BiFeO₃ system.

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1. Introduction

Multiferroic materials are a fascinating class of materials that exhibit both ferroelectric and ferromagnetic properties within a single phase. This dual nature has the potential to advance fundamental physics and enable innovative applications, such as rapid data writing, energy-efficient devices, and secure data

storage, and hence, multiferroic materials have attracted considerable interest.^{1,2} However, the coexistence of ferroelectric and magnetic behavior at room temperature remains rare. Two categories of room-temperature multiferroic materials have been identified: oxide materials related to BiFeO₃^{3–9} and ϵ -Fe₂O₃.^{10–25} Materials in these two categories differ substantially in their crystal structures as well as their intrinsic properties.

The BiFeO₃-related oxides with a perovskite structure are distinguished by their ability to achieve clear ferroelectric polarization (P) reversal under an applied electric field (E). These materials exhibit excellent piezoelectric properties, with a remanent polarization (P_r) of approximately 60 μ C cm⁻² and a piezoelectric coefficient (d_{33}) of 70 pm V⁻¹.^{3,4} The coupling between the magnetic moment direction and polarization direction in BiFeO₃ allows for the modification of the magnetic moment orientation *via* E -induced polarization reversal at room temperature.^{5–7} However, as BiFeO₃ exhibits antiferromagnetic ordering, the remanent magnetization (M_r) is nearly zero.⁸ Although the spin structure can be tuned by substrate-induced strain or chemical substitution (e.g., Co doping), the M_r value is still limited.⁹

In contrast, ϵ -Fe₂O₃-related oxides exhibit substantial magnetization and coercive force at room temperature due to their ferrimagnetic nature.^{10–14} For instance, Rh doping leads to a

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remarkable coercive force of 24 kOe, the highest value among oxide magnets, and Ga doping increases M_r to $0.6\mu_B$ f.u. $^{-1}$ at 300 K.¹⁰ These oxides have been investigated for applications related to their magnetic properties, including photo-induced magnetization reversal memory¹¹ and millimeter-wave absorption.^{12,13} Moreover, E -induced polarization reversal has been demonstrated in thin-film ϵ -Fe₂O₃-related oxides at room temperature.¹⁵⁻²⁵ However, the reported P - E hysteresis loops indicate significant leakage currents, which hinder the quality of ferroelectric polarization reversal. ϵ -Fe₂O₃-related oxides have not been extensively investigated in the field of multiferroics because they exhibit relatively poor polarization reversal properties in comparison to BiFeO₃. Nonetheless, given the excellent magnetic properties of ϵ -Fe₂O₃-related oxides, establishing clear design guidelines for achieving reliable electric-field-induced ferroelectric polarization reversal can help significantly improve their multiferroic application prospects.

The reason for the difficulty in achieving ferroelectric polarization reversal in ϵ -Fe₂O₃-related oxides remains unclear. The challenge is attributed primarily to their unique crystal structure, which deviates from the common perovskite structure of BiFeO₃. The crystal structure of ϵ -Fe₂O₃ comprises octahedral (O_h) Fe_A, Fe_B, and Fe_C sites, as well as tetrahedral (T_d) Fe_D sites (Fig. 1(a)). The presence of grain boundaries (GBs) in the films further complicates the issue: thin films reported in the literature consist of three types of in-plane grains with the polar axis (c -axis) oriented along the out-of-plane direction (Fig. 1(b)).

This study has two key objectives. First, we aim to elucidate the grain boundary structure and identify the factors hindering polarization reversal. Second, we explore grain boundary engineering as a strategy to achieve clear ferroelectric polarization reversal. We investigate the grain structures *via* plan-view and cross-sectional transmission electron microscopy (TEM) and high-angle annular dark-field scanning TEM (HAADF-STEM) measurements on three types of ϵ -Fe₂O₃-related oxide films: Ga_{0.8}Fe_{1.2}O₃, Sc_{0.5}Fe_{1.5}O₃, and Ga_{0.4}Sc_{0.2}Fe_{1.4}O₃. Hereafter,

these films are referred to as Ga-doped, Sc-doped, and (Ga,Sc)-codoped ϵ -Fe₂O₃ films, respectively.

Our findings revealed that the films contained significantly small grains, with the average sizes ranging from 250 to 770 nm². Fe^{2+/3+} states were present at the GBs, and approximately 40% of the GBs corresponded to ferroelectric domain walls. This high density of GBs was identified as the primary source of the large leakage current and the reason for the difficulty in achieving polarization reversal. Furthermore, Ga and Sc codoping reduced the density of GBs, enabling clear ferroelectric polarization reversal in the films.

2. Results and discussion

2.1. Domain wall structure characteristics

Fig. 1(c) illustrates the 2θ - θ X-ray diffraction (XRD) patterns of the Ga-doped ϵ -Fe₂O₃ film. The 002, 004, and 006 XRD peaks were clearly observed, indicating that a c -axis-oriented film with an ϵ -Fe₂O₃ structure was obtained. As illustrated in the ϕ -scan results (Fig. 1(d)), the film exhibited sixfold in-plane symmetry, whereas the STO(111) plane exhibited threefold symmetry. Considering the orthorhombic structure of ϵ -Fe₂O₃, the film consisted of three types of in-plane grains, G_1 , G_2 , and G_3 , which were rotated by 120° relative to each other as illustrated in Fig. 1(b).

Fig. 2(a) illustrates a plan-view TEM image of the Ga-doped ϵ -Fe₂O₃ film. Stripe-like patterns were observed. Compared with the Fe_B and Fe_C sites (blue and light blue circles in Fig. 1(a)), the Fe_A and Fe_D sites (orange and green points) are expected to be more clearly observed in the plan-view TEM image because they overlap along the out-of-plane direction. Thus, the stripe-like patterns were derived from the zigzag lines of the Fe_A and Fe_D atoms, which correspond to the orange and green lines in Fig. 1. The direction of the grains in the plan-view TEM image was determined based on the direction of the stripe-like

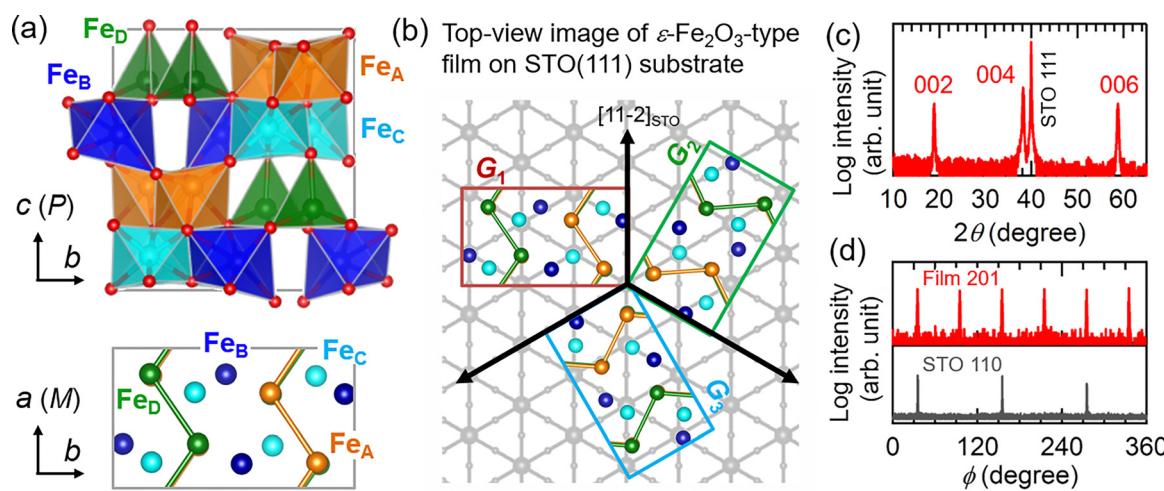


Fig. 1 (a) Crystal structures of ϵ -Fe₂O₃ and (b) top-view schematic of the ϵ -Fe₂O₃-related oxide film grown on the STO(111) substrate. (c) 2θ - θ X-ray diffraction pattern and (d) ϕ -scan of the Ga-doped ϵ -Fe₂O₃ film.



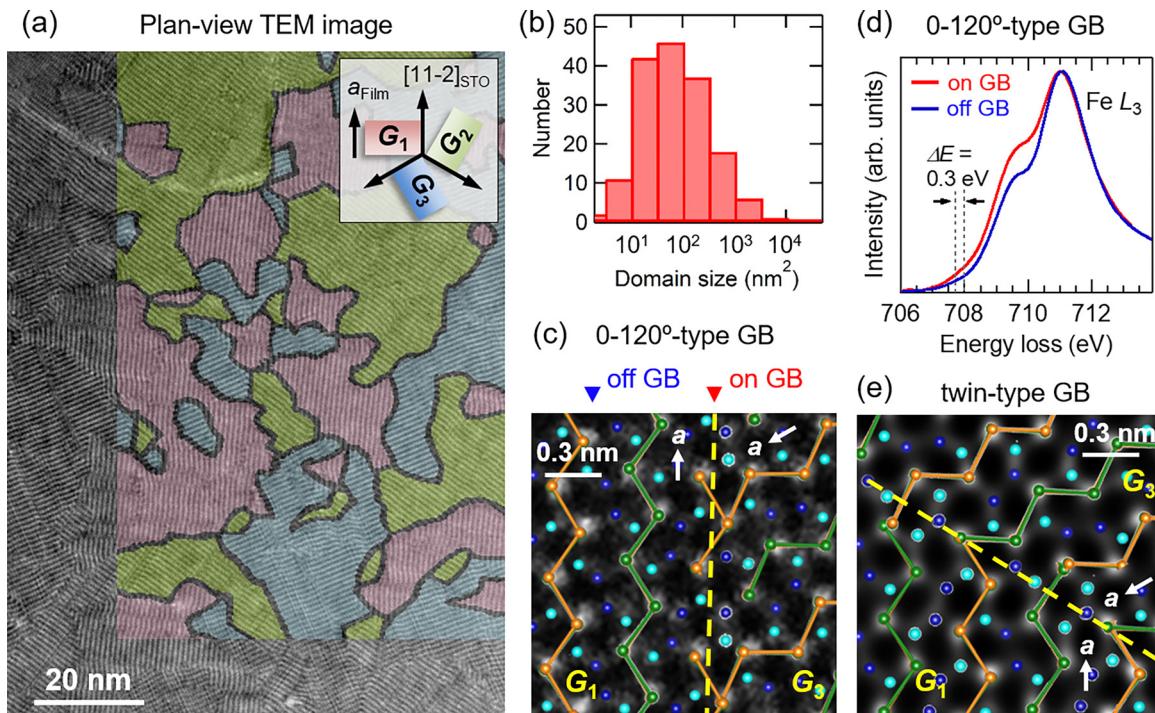


Fig. 2 (a) Plan-view TEM image. (b) A histogram of the grain size. (c) Plan-view STEM image and (d) on- and off-GB Fe L₃-edge EELS spectra near the 0–120°-type grain boundaries (GBs). (e) Plan-view STEM image near twin-type GBs for the Ga-doped ϵ -Fe₂O₃ film.

patterns. The G_1 , G_2 , and G_3 grain types are colored red, green, and blue, respectively, in Fig. 2(a). Fig. 2(b) illustrates a histogram of the grain size. The average size was as small as 250 nm². These small grains formed a nanomosaic-like pattern in the film.

The GB structure was investigated *via* HAADF-STEM measurements. Typical GB structures were 0–120°-type and twin-type, as illustrated in Fig. 2(c) and (e), respectively. The 0–120°-type GB was formed by the connection between a (100) grain and a (110) grain. This connection is preferred because the a -axis length (5.06 Å) matches well with the value of $b/\sqrt{3}$ (5.06 Å). In the twin-type GB, the mirror plane was formed in the direction parallel to the (110) plane, which crossed all cation sites. Although ϵ -Fe₂O₃ exhibits spontaneous polarization along the c -axis, and thus no intrinsic in-plane polarization is expected, the formation of twin-type grain boundaries could induce strain that may locally generate in-plane polarization components. Further investigation is required to clarify this possibility.

Fig. 2(d) and Fig. S1 (ESI†) illustrate the Fe L-edge electron energy loss spectroscopy (EELS) plots for the 0–120°-type GB (on-GB) and within the grain (off-GB). Both spectra exhibited prominent and left-shoulder peaks at 711 and 709.5 eV, respectively. The intensity of the shoulder peak in the on-GB spectrum was higher than that in the off-GB spectrum, indicating the reduced valence state of the Fe ions in the GB. The difference in the starting energy of the peak (ΔE) between the on- and off-GB regions was 0.3 eV. Thus, the valence state of the Fe ions in the GB is expected to be $\sim 2.8+$.²⁶ This implies that oxygen vacancies were generated in the GB.

The GB structure was also investigated *via* cross-sectional HAADF-STEM measurements. In the HAADF-STEM measurements, the electron beam was irradiated in the [11–2]_{STO} direction (Fig. 3(a)). Fig. 3(b) shows the cross-sectional images of ϵ -Fe₂O₃ in the G_1 and G_3 domains. The atomic arrangement, marked in yellow boxes, can be used to determine the polarization direction. For example, when ϵ -Fe₂O₃ exhibits an upward polarization (up- P) in the G_1 domain, the positions of the four atoms in the yellow box change to down-up-up-down. Meanwhile, in the up- P G_3 domain, the three atoms in the yellow box have an up-down-up arrangement.

Fig. 3(c) illustrates a cross-sectional HAADF-STEM image of the region near the GB. Based on the atomic arrangement, the left and right domains were identified as up- P G_1 and up- P G_3 , respectively. We also observed GBs between the domains with opposite P directions (Fig. 3(d)). We observed 12 GBs in the Ga-doped ϵ -Fe₂O₃ film and found that five of them were formed between grains with opposite P directions. This suggests that approximately 40% of the GBs are ferroelectric 180°-domain walls (FE DWs). ϵ -Fe₂O₃ has four layers along the c -axis (Fig. 4(a)): L1 and L3 layers consisting of Fe_A and Fe_D sites and L2 and L4 layers consisting of Fe_B and Fe_C sites. The L1 layer of one grain was connected to either the L1 or L3 layer of another grain in all the observed GBs.

2.2. The origin of the difficulty in obtaining clear ferroelectric reversal

We identified the 0–120°-type GB structure from the plan-view and cross-sectional STEM measurements. Fig. 4(b) illustrates the 0–120°-type GBs between the up- P G_1 and up- P G_3 grains.



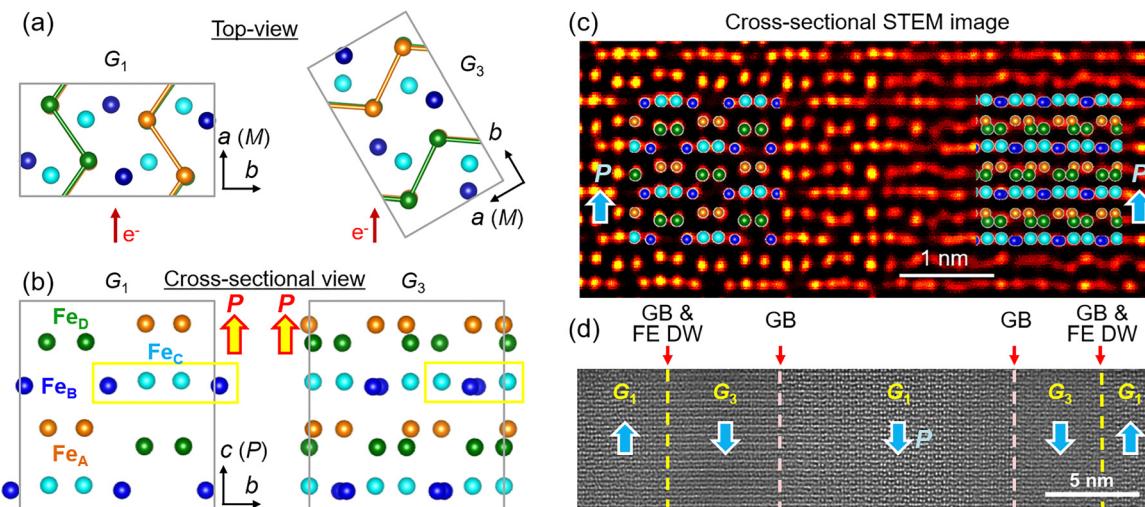


Fig. 3 (a) Top-view and (b) cross-sectional images of ϵ -Fe₂O₃. (c) and (d) Cross-sectional HAADF-STEM images of the Ga-doped ϵ -Fe₂O₃ film.

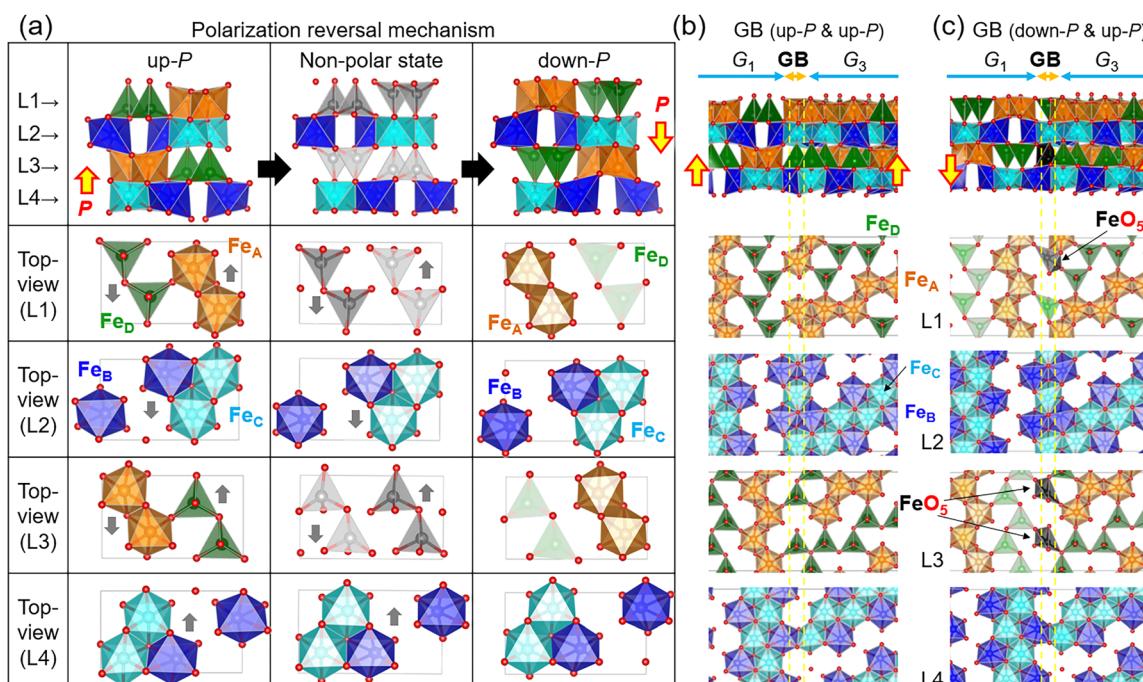


Fig. 4 (a) P-reversal mechanism in ϵ -Fe₂O₃. (b) 0–120°-type GB structure between the up-P G₁ and up-P G₃ domains. (c) Expected GB structure after P reversal in the G₁ domain.

New cationic arrangements, such as Fe_D-site tetrahedron trimers, were observed in the GB. ϵ -Fe₂O₃ is known to have a unique P-reversal mechanism.^{16,17} During P reversal, the octahedral (O_h) Fe_A and tetrahedral (T_d) Fe_D sites in the L1 and L3 layers were switched, while Fe_B and Fe_C sites in the L2 and L4 layers were shifted along the *a*-axis (Fig. 4(a)). Fig. 4(c) illustrates the expected GB structure after P reversal in the G₁ domain. Pyramidal FeO₅ appeared in the GB in the L1 and L3 layers after P reversal. In addition, the connection between the Fe_B and Fe_C sites was slightly distorted in the L2 and L4 layers. Thus, the activation energy for P reversal increased near the GB; in other words, P reversal was restricted by the GBs in the film. X. Guan *et al.* conducted *in situ*

STEM observations while applying an electric field to ϵ -Fe₂O₃ films and reported that ferroelectric switching tends to occur more readily near ferroelectric domain walls within a single grain.²⁷ This finding supports the idea that reducing the density of grain boundaries—which inhibit ferroelectric switching—and increasing grain size are crucial for improving ferroelectric properties.

2.3. Achieving clear ferroelectric reversal *via* domain engineering

Obtaining a good ferroelectric hysteresis loop in ϵ -Fe₂O₃-related oxides is challenging. We found that this difficulty is due to the high density of GBs with Fe^{2+/3+} states, which inhibits P

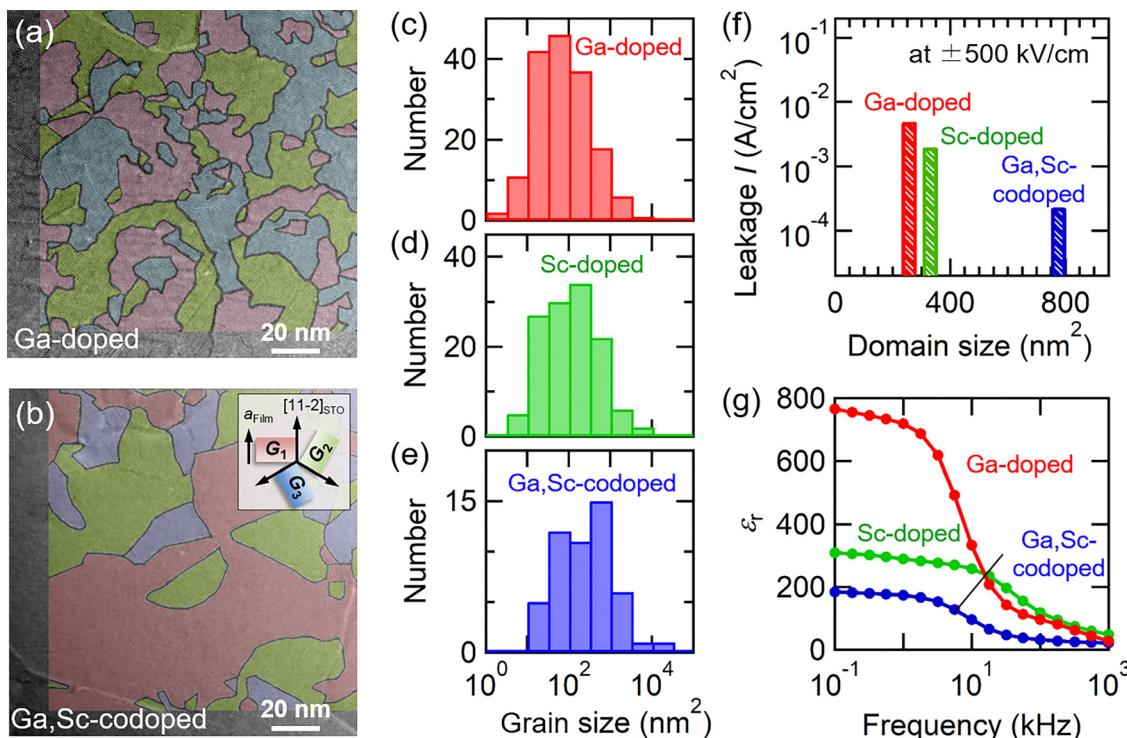


Fig. 5 Plan-view TEM images and histograms of the grain size for the (a) and (c) Ga-doped, (d) Sc-doped, and (b) and (e) Ga,Sc-codoped ϵ - Fe_2O_3 films. (f) Leakage current and (g) permittivity of the films at 300 K.

reversal, as shown in Fig. 4. Therefore, we attempted to increase the grain size through chemical substitution. Fig. 5(a)–(e) and Fig. S2 (ESI†) illustrate the plan-view TEM images and histograms of the grain size for the Ga-doped, Sc-doped, and Ga,Sc-codoped ϵ - Fe_2O_3 films. The average grain size of the Ga,Sc-codoped film (770 nm^2 , Fig. 5(e)) was three times larger than that of the Ga-doped film (250 nm^2 , Fig. 5(c)). This may be attributed to two reasons. First, new cationic arrangements, such as Fe_D -site tetrahedron trimers, are formed at the GBs. However, such cationic arrangements may be unstable depending on the elemental species, which, in turn, may increase the energy required for GB formation. Second, the difference between the a -axis length and $b/\sqrt{3}$ value leads to strain near the GBs and results in an increase in the energy required for GB formation. Further investigation is necessary to elucidate the reasons in detail.

Fig. 5(f) illustrates the leakage current (I) for the films at $\pm 500 \text{ kV}/\text{cm}^{-1}$. The I value decreases with increasing grain size, reflecting the reduction in the number of the GBs with $\text{Fe}^{2+/3+}$ state. Fig. 5(g) illustrates the frequency dependence of the dielectric constants (ϵ') of the films. The films exhibit relaxation frequencies of 10–50 kHz due to dielectric relaxation through oxygen vacancies, which are coupled with the Fe^{2+} ions.²¹ ϵ' tends to decrease, particularly below the relaxation frequency, with an increase in the grain size because of the reduction in the number of oxygen vacancies that are located at the GBs.

Fig. 6(a) illustrates the P versus electric field (P - E) hysteresis loops for the Sc-doped and Ga,Sc-codoped ϵ - Fe_2O_3 films at 300 K. The Ga-doped film did not exhibit ferroelectric behavior

in the P - E curve because of the high density of GBs, as discussed in Section 2.2. In contrast, the Sc- and Ga,Sc-codoped films exhibited clear ferroelectric hysteresis loops with minimal leakage currents in the P - E curves. The remanent polarizations (P_r) of the Sc-doped and Ga,Sc-codoped films were 3.5 and $4.5 \mu\text{C}/\text{cm}^2$, respectively. The obtained P_r values were smaller than the calculated values for GaFeO₃-type Fe_2O_3 ($21 \mu\text{C}/\text{cm}^2$) and GaFeO₃ ($25 \mu\text{C}/\text{cm}^2$).^{16,17,28} The decrease in P_r with decreasing grain size is also observed in ferroelectric $\text{Pb}(\text{Zr}_x\text{Ti}_{1-x})\text{O}_3$.²⁹ This is attributed primarily to the pinning of ferroelectric domains at the GBs due to the high activation energy of polarization reversal at the GBs. Thus, a strong relationship between the ferroelectric properties and GBs exists in ϵ - Fe_2O_3 -related oxide films, and increasing the grain size is key to improving their ferroelectric properties. The Sc-doped and Ga,Sc-codoped films also exhibited ferrimagnetic properties at 300 K (Fig. 6(b)), confirming the coexistence of spontaneous polarization and magnetization at room temperature.

3. Conclusion

The GB structures of room-temperature multiferroic ϵ - Fe_2O_3 -related oxide films were investigated via plan-view and cross-sectional TEM and STEM measurements. The grain sizes of the films were as small as 250 – 770 nm^2 . The STEM-EELS measurements revealed that the valence state of Fe was reduced at the GBs. Cross-sectional STEM measurements indicated that approximately 40% of the GBs were FE DWs. The large leakage current and the difficulty in achieving good P reversal behavior



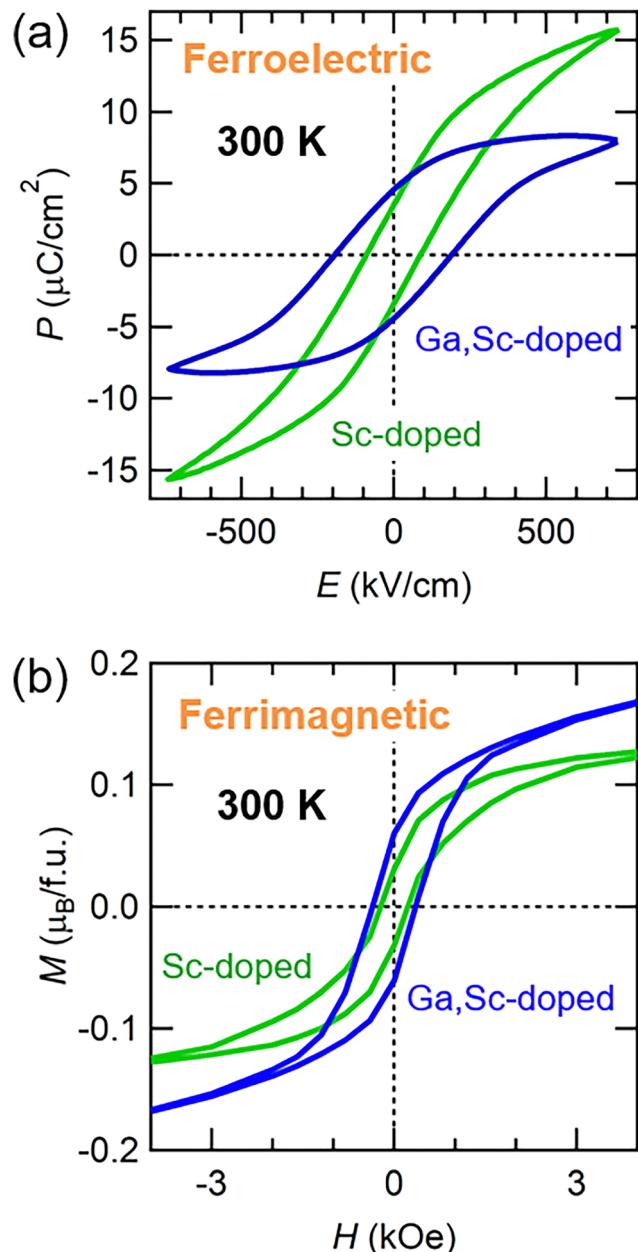


Fig. 6 (a) Polarization (P) versus electric field (E) and (b) magnetization (M) versus magnetic field (H) curves for the Sc-doped and Ga,Sc-codoped ϵ - Fe_2O_3 films at 300 K.

in ϵ - Fe_2O_3 -related oxide films were attributed to the high density of GBs. Therefore, we reduced the density of GBs *via* codoping and achieved clear ferroelectric hysteresis with minimal leakage current. The results obtained in this study are expected to catalyze further research on room-temperature multiferroic materials with large magnetization distinct from the BiFeO_3 system.

4. Experimental

The Ga-doped, Sc-doped, and Ga,Sc-codoped ϵ - Fe_2O_3 films were fabricated on $\text{SrTiO}_3(111)$ (STO) and Nb 0.5 wt%-doped

STO(111) substrates *via* pulsed laser deposition. The chemical compositions of the films were $\text{Ga}_{0.8}\text{Fe}_{1.2}\text{O}_3$, $\text{Sc}_{0.5}\text{Fe}_{1.5}\text{O}_3$, and $\text{Ga}_{0.4}\text{Sc}_{0.2}\text{FeO}_3$, respectively. The substrate temperature and oxygen partial pressure during the deposition were maintained at 700 °C and 200 mTorr, respectively. The fourth harmonic of an Nd:YAG pulsed laser (wavelength: 266 nm) was used with an energy of 12 mJ per pulse and a repetition rate of 5 Hz. The laser spot size was approximately 1 mm², and the distance between the target and the substrate was 50 mm. The typical thickness of the films was 30–40 nm.

The crystal structures of the films were determined *via* high-resolution XRD with $\text{Cu-K}\alpha_1$ radiation (Rigaku Smartlab). The microstructure was investigated *via* TEM and HAADF-STEM measurements using a JEM-ARM200F cold FEG instrument. In our EELS measurements, the acquisition points were manually selected by observing the STEM image in real time, as our system does not include an automated protocol for positioning EELS measurements. We performed the EELS measurements several times, and during the process, we observed instances of apparent chemical shifts. However, we only used data sets that showed minimal chemical shift and in which the EELS acquisition point clearly coincided with the grain boundary location in the STEM image. The leakage current was measured *via* the conventional two-probe method (Keithley 2450 SourceMeter) at room temperature. We did not use the step relaxation method for the leakage current measurements. Instead, the current–voltage (I – V) characteristics were measured by applying a voltage in steps of 0.1 seconds per voltage point. The dielectric properties were measured using a precision LCR meter (Agilent, 4284A) at room temperature. The ferroelectric properties were investigated using a ferroelectric tester (Toyo Corporation FCE-1E) at room temperature. The ferroelectric hysteresis loops were performed at a frequency of 10 kHz. Pt electrodes and Nb:STO substrates were respectively used as the top and bottom electrodes while measuring the leakage current, dielectric properties, and ferroelectric properties. The diameter of the Pt electrodes used for the ferroelectric and leakage current measurements was 100 μm , and that of the Pt electrodes used for the dielectric measurements was 200 μm . The magnetization of the films was measured using a superconducting quantum interference device magnetometer (Quantum Design, MPMS XL).

Data availability

All data supporting the findings of this study are included in the main manuscript.

Conflicts of interest

There are no conflicts to declare.

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