



Cite this: *Chem. Sci.*, 2025, **16**, 19910

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 8th July 2025  
 Accepted 23rd September 2025  
 DOI: 10.1039/d5sc05065k  
[rsc.li/chemical-science](http://rsc.li/chemical-science)

## Introduction

A multitude of machines in our everyday lives are designed in such a way that mechanical motion is transmitted from where it is produced to other components executing the actual function. This paradigm applies also at the nanoscale, as exemplified by Nature-made molecular machines that synthesize ATP (ATP synthase<sup>1–5</sup>) and provide intracellular transport (e.g., kinesin motor proteins).<sup>6–10</sup> In the field of artificial molecular machines,<sup>11–17</sup> it is similarly desirable to design and synthesize molecular gears<sup>18–30</sup> capable of transmitting rotary motion while also controlling the direction of rotation. Here, for the sake of facile motion, constructing gears whose receiver components undergo rotation around a single bond is a natural goal. However, as typical gear designs rely on passive thermal activation, random thermal fluctuations (Brownian motion) make it difficult for them to follow a preferred direction of rotation and, consequently, to perform actual mechanical work.<sup>31</sup>

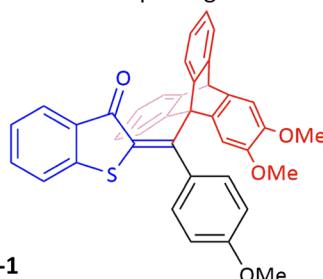
A possible solution to this problem, advocated by the Feringa group,<sup>32</sup> is to exploit the unique ability of synthetic molecular motors to convert the energy from an external source into directed rotary motion.<sup>33–41</sup> Along those lines, in 2022,<sup>42</sup> the Dube group presented a molecular photogear incorporating a photoswitch of hemithioindigo (HTI) type.<sup>43–45</sup> Specifically, this design, which is shown as **PG-1** in Fig. 1, transmits the photoinduced 180° rotation of the “rotor” moiety of the HTI

## Achieving one-step molecular photogearing in a minimal light-driven molecular motor

Enrique M. Arpa <sup>\*ab</sup> and Bo Durbejj <sup>\*a</sup>

The last decades have seen a wealth of progress in the design and synthesis of molecular motors for converting light energy into directed rotary motion around a double bond. Yet, realizing the full potential of these systems in the field of artificial molecular machines will inevitably require a breakthrough in the formidable challenge to construct molecular photogears for transmitting such motion through space and onto a remote single-bond axis, without losing control of the direction of rotation. Here, we unveil a surprisingly straightforward mechanism for achieving this goal in a single photochemical step by incorporating a propeller-shaped barrelene motif into the protonated Schiff-base skeleton of a minimal light-driven molecular motor. Corroborating the mechanism by state-of-the-art computational modeling, our study also identifies strategies for optimizing the photogearing efficiency through modulation of steric interactions. Overall, the results of this work help establish a new route for constructing molecular photogears by combining molecular-motor and propeller-shaped structures.

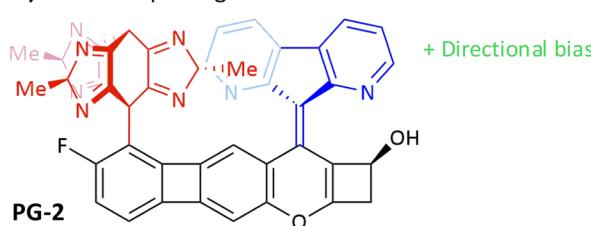
### A) Dube's 2022 photogear



**PG-1**

+ Simple structure  
 + Directional bias

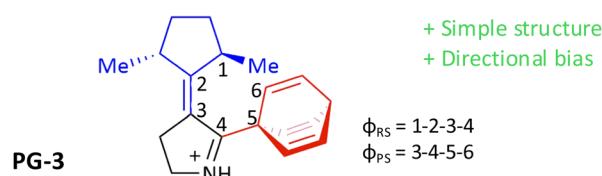
### B) Our 2024 photogear



**PG-2**

+ Directional bias

### C) This work



+ Simple structure  
 + Directional bias

<sup>a</sup>Division of Theoretical Chemistry, IFM, Linköping University, 58183 Linköping, Sweden. E-mail: [bodur@ifm.liu.se](mailto:bodur@ifm.liu.se)

<sup>b</sup>Institute of Organic Chemistry, RWTH Aachen University, 52056 Aachen, Germany. E-mail: [enrique.arpa@rwth-aachen.de](mailto:enrique.arpa@rwth-aachen.de)

Fig. 1 (A) Photogear developed by the Dube group in 2022.<sup>42</sup> (B) Photogear developed in our 2024 work.<sup>49</sup> (C) Photogear presented in this work and definitions of relevant dihedral angles. In all photogears, the rotor, propeller and stator parts are blue-, red- and black-colored, respectively.



switch around a C=C bond, into a 120° rotation of a triptycene “propeller” around a C-C bond (in Fig. 1, the rotor and propeller moieties are blue- and red-colored, respectively). Adopting a bevel-like configuration with non-parallel rotation axes, **PG-1** differs from previous molecular setups developed by the same group<sup>46,47</sup> in its ability to transmit the rotary motion through spatial interactions alone (the hallmark of true molecular gearing),<sup>19</sup> without help from a covalent chain linkage between the rotor and propeller moieties. Following excitation using blue light, the authors showed that **PG-1** achieves photogearing with a quantum yield of up to ~7%, depending on which isomer is irradiated. Moreover, in a follow-up study, the photogearing was found to occur in such a way that both the rotor rotation and the propeller rotation exhibit a rather pronounced directional bias, but no overall directionality.<sup>48</sup> This likely reflects the absence of a stable and permanent source of chiral asymmetry in **PG-1** to control the rotations.

As a potential remedy, we recently put forth the photogear design shown as **PG-2** in Fig. 1, whose triptycene-like propeller contains three stereogenic  $sp^3$  centers.<sup>49</sup> Combining this propeller with a diazafluorene rotor (fluorene being a common motif in light-driven motors of overcrowded-alkene type),<sup>50,51</sup> and linking the two *via* an extended “stator” fragment, **PG-2** adopts a spur-like configuration with parallel C=C (rotor) and C-C (propeller) rotation axes. Encouragingly, assessing the design by means of quantum chemical calculations and molecular dynamics (MD) simulations, it was found to enable a two-step photogearing cycle with alternating C=C photoisomerization and thermal helix inversion steps.<sup>49</sup> Furthermore, these steps were predicted to favor directed photogearing both dynamically (during photoisomerization) and kinetically (during thermal helix inversion). However, because of its complex structure, particularly in the stator motif, it would likely be difficult to synthesize **PG-2**.

Against this background, we herein present and computationally evaluate a new photogear design (**PG-3** in Fig. 1) that combines the main favorable attributes of **PG-1** (structural simplicity) and **PG-2** (directed photogearing) in a reaction cycle involving just one single isomer. In this cycle, photogearing is achieved when the isomer in question reforms itself in a single photochemical step, without any intermediary steps. We refer to such a process as one-step molecular photogearing. Overall, a key goal of our study is to identify the most basic structural ingredients needed to realize molecular photogearing.

## Results and discussion

Our development of **PG-3** was guided by a number of considerations. First, we changed from a spur (in **PG-2**) to a bevel design, for which it is less critical that the rotor-propeller distance is large<sup>24,28,29</sup> and, consequently, a smaller stator motif might be used. Furthermore, compared to the **PG-1** bevel design, the rotor and propeller rotation axes of **PG-3** are not directly connected. Second, in order to possibly eliminate thermal steps from the photogearing cycle, the rotor-stator core of **PG-3** was based on a protonated Schiff base akin to other such structures computationally predicted (but without

rigorous experimental verification) to produce rotary motion for molecular motors in a purely photochemical fashion.<sup>52–55</sup> Third, as for the choice of propeller, the starting point was again triptycene, but instead of replacing the benzene rings of triptycene with slightly smaller imidazole-tautomeric rings (as done for **PG-2**), in **PG-3** the benzene rings were removed altogether, resulting in a much smaller  $C_3$ -symmetric barreleno propeller. Fourth, and finally, instead of having three asymmetric Me groups in the propeller (as in **PG-2**), in **PG-3** two such groups were included in a  $C_2$ -symmetric cyclopentane rotor.

We began the computational assessment of **PG-3** by exploring whether this system has the functionality of a light-driven rotary molecular motor. This requires that the potential energy surface (PES) of the bright  $\pi\pi^*$  state populated by light absorption exhibits a directional bias for the rotor rotation around the central C=C bond. For **PG-3**, the idea is that this bias may be realized by the asymmetric rotor Me groups, which introduce steric repulsion between the rotor and the propeller. In order to test this idea, the  $\pi\pi^*$  PESs of **PG-3** and its demethylated derivative **PG-3'** (obtained by replacing the Me groups with H atoms) were mapped for both clockwise (CW) and counterclockwise (CCW) changes in the rotor-stator dihedral angle ( $\varphi_{RS}$ , see Fig. 1C) relative to the value at the ground-state ( $S_0$ ) equilibrium geometry. Here, CW/CCW is defined as the direction of rotation in which the value of  $\varphi_{RS}$  increases/decreases. Using both density functional theory (DFT) and multiconfigurational methods, the full details of these (and all other) calculations are given in Section 1 of the SI. Including the  $S_0$  ground and the  $S_1$  and  $S_2$  excited states in the analysis, the DFT results are presented in Fig. 2. Very similar results obtained with the multiconfigurational methods are shown and discussed in Section 3 of the SI.

From Fig. 2, it can be seen that the  $\pi\pi^*$  state of **PG-3'** is perfectly symmetric with respect to CW and CCW rotor rotations, which means that this system cannot function as a light-driven rotary motor. **PG-3**, on the other hand, does display the required directional bias for the rotor rotation. First, the  $S_0$  equilibrium geometry (which occurs at  $\varphi_{RS} = 0^\circ$  for **PG-3'**) is now markedly twisted in the CW direction ( $\varphi_{RS} = 11.6^\circ$ ). Second, following excitation to the  $\pi\pi^*$  state, CW rotation is a downhill process, whereas CCW rotation requires surmounting an energy barrier of 0.34 eV (*ca.* 8 kcal mol<sup>-1</sup>).

Having thus established that **PG-3** behaves like a light-driven rotary motor, it should be clarified that the  $[-90^\circ, 90^\circ]$  range of  $\varphi_{RS}$  values considered in Fig. 2 covers all possible  $\varphi_{RS}$  values, thanks to the  $C_2$  symmetry of the rotor. Combined with the  $C_3$  symmetry of the propeller, and the fact that the rotor rotation does not alter the favorable axial orientations of the rotor Me groups, this means that the photochemical rotor rotation should proceed without an accompanying thermal step. This is different from, *e.g.*, overcrowded-alkene motors, where the photochemical steps do change the orientations of the stereogenic substituents from axial to equatorial and, therefore, thermal steps are needed to revert these changes.<sup>41</sup>

Next, we decided to test whether the rotor rotation in **PG-3** induces any propeller rotation, as it would if **PG-3** also has photogearing capability. To this end, we compared the



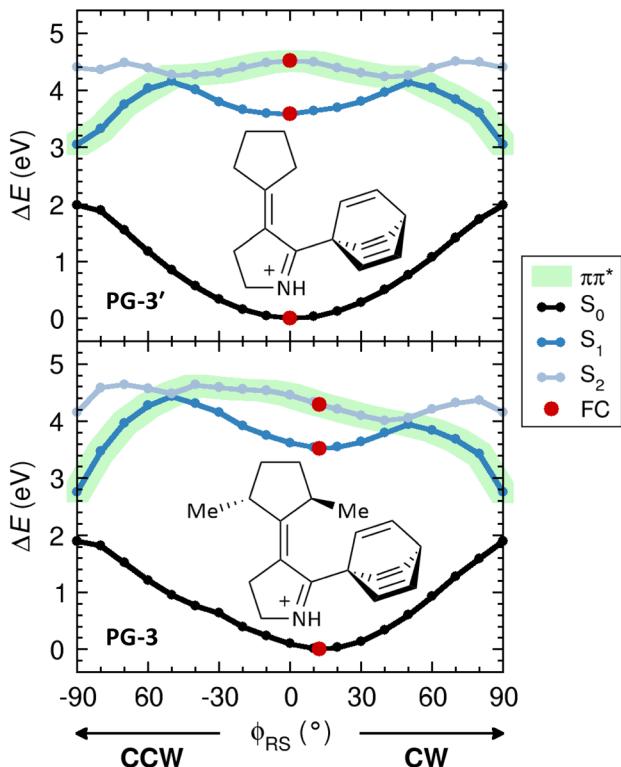


Fig. 2  $S_0$ ,  $S_1$  and  $S_2$  energy profiles for changing the  $\phi_{RS}$  dihedral angle in the **PG-3'** and **PG-3** systems, with the  $\pi\pi^*$  state highlighted in light-green color. In each system, a full red circle indicates the value of  $\phi_{RS}$  at the vertically excited Franck–Condon (FC) point (i.e., at the  $S_0$  equilibrium geometry).

optimized geometries of the  $S_0$  minimum of **PG-3** and the  $S_1/S_0$  conical intersection (CI) through which the rotor rotation of **PG-3** was found to proceed. Following typical protocols employed in the literature,<sup>56,57</sup> these calculations were done with the multi-configurational complete active space self-consistent field (CASSCF) method.<sup>57,58</sup> The resulting geometries, alongside the corresponding  $\phi_{RS}$  and propeller-stator ( $\phi_{PS}$ , see Fig. 1C) dihedral angles, are shown in Fig. 3.

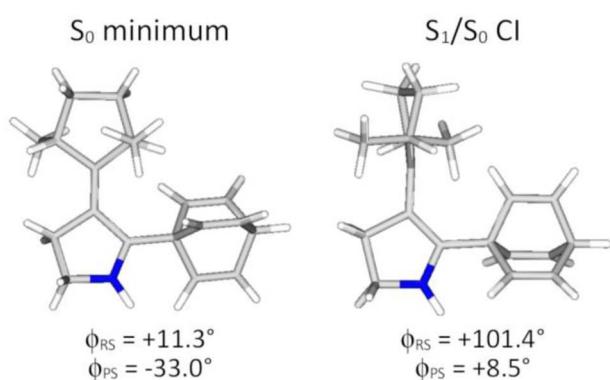


Fig. 3 Geometries of the  $S_0$  minimum and the  $S_1/S_0$  CI for rotor rotation for **PG-3**, and the corresponding values of the  $\phi_{RS}$  and  $\phi_{PS}$  dihedral angles.

Interestingly, because of steric clashing between one of the rotor Me groups and the propeller, the  $\sim 90$  degree CW rotor rotation needed to reach the  $S_1/S_0$  CI causes a  $\sim 42^\circ$  propeller rotation in the direction of increasing  $\phi_{PS}$  values, which we define as CCW. This suggests that **PG-3** could well be capable of photogearing. Furthermore, since the rotary cycle underlying the basic motor function of **PG-3** appears to require only one single photoinduced  $180^\circ$  rotor rotation to regenerate the parent  $S_0$  minimum (thanks to the rotor and propeller symmetries), it seems possible that any photogearing by **PG-3** would also proceed through a “simple” mechanism. On the downside, the high symmetry of **PG-3** would inevitably make experimental observation of the photogearing more difficult.<sup>48</sup>

In order to model the actual photodynamics of **PG-3** upon light absorption and assess the propensity of this compound to sustain photogearing in as unbiased fashion as possible, we next decided to perform non-adiabatic molecular dynamics (NAMD) simulations.<sup>59,60</sup> Continuing to use the CASSCF method and running the simulations for 500 fs, the details of the simulations are provided in Section 1 of the SI. In principle, following excitation to the  $\pi\pi^*$  state of **PG-3**, four different types of outcomes are possible and uniquely characterizable in terms of the changes  $\Delta\phi_{RS}/\Delta\phi_{PS}$  in  $\phi_{RS}/\phi_{PS}$  dihedral angles along the NAMD trajectories. Specifically, with the above definitions of CW and CCW directions of rotations for the rotor and the propeller, respectively, a positive  $\Delta\phi_{RS}$  value (i.e., an increase in  $\phi_{RS}$ ) corresponds to a CW rotor rotation, whereas a positive  $\Delta\phi_{PS}$  value (i.e., an increase in  $\phi_{PS}$ ) corresponds to a CCW propeller rotation. Thereby, the different outcomes can be described as follows:

- Photogearing: this is when a CW rotor rotation ( $\Delta\phi_{RS} = +180^\circ$ ) induces a CCW propeller rotation ( $\Delta\phi_{PS} = +120^\circ$ ), or a CCW rotor rotation ( $\Delta\phi_{RS} = -180^\circ$ ) induces a CW propeller rotation ( $\Delta\phi_{PS} = -120^\circ$ ). Here, we refer to the first scenario as forward photogearing, and the second scenario as reverse photogearing. Notably, the results from the static calculations in Fig. 2 and 3 suggest that forward photogearing is more likely than reverse photogearing.

- Slippage: this is a free-standing CW or CCW rotor rotation ( $\Delta\phi_{RS} = \pm 180^\circ$ ) that occurs without a propeller rotation ( $\Delta\phi_{PS} = 0^\circ$ ), or a free-standing CW or CCW propeller rotation ( $\Delta\phi_{PS} = \pm 120^\circ$ ) that occurs without a rotor rotation ( $\Delta\phi_{RS} = 0^\circ$ ).

- Gear clash: this is when a CW rotor rotation ( $\Delta\phi_{RS} = +180^\circ$ ) induces a CW propeller rotation ( $\Delta\phi_{PS} = -120^\circ$ ), or a CCW rotor rotation ( $\Delta\phi_{RS} = -180^\circ$ ) induces a CCW propeller rotation ( $\Delta\phi_{PS} = +120^\circ$ ). As appreciable rotor-propeller steric repulsion would have to be overcome for gear clashing to happen, this outcome seems unlikely.

- FC repopulation: this is when the rotor begins rotating in one direction, reaches the associated  $S_1/S_0$  CI, decays to the  $S_0$  state, but then continues rotating back in the opposite direction to repopulate the parent  $S_0$  minimum ( $\Delta\phi_{RS} = 0^\circ$  and  $\Delta\phi_{PS} = 0^\circ$ ).

The calculated NAMD trajectories are summarized in Fig. 4, which shows all possible  $\Delta\phi_{RS}$  and  $\Delta\phi_{PS}$  values among the 20 trajectories at each time step. An alternative way of presenting these results is through Fig. S6 of the SI, which plots the  $\Delta\phi_{RS}$  and  $\Delta\phi_{PS}$  values for the individual trajectories separately. Before



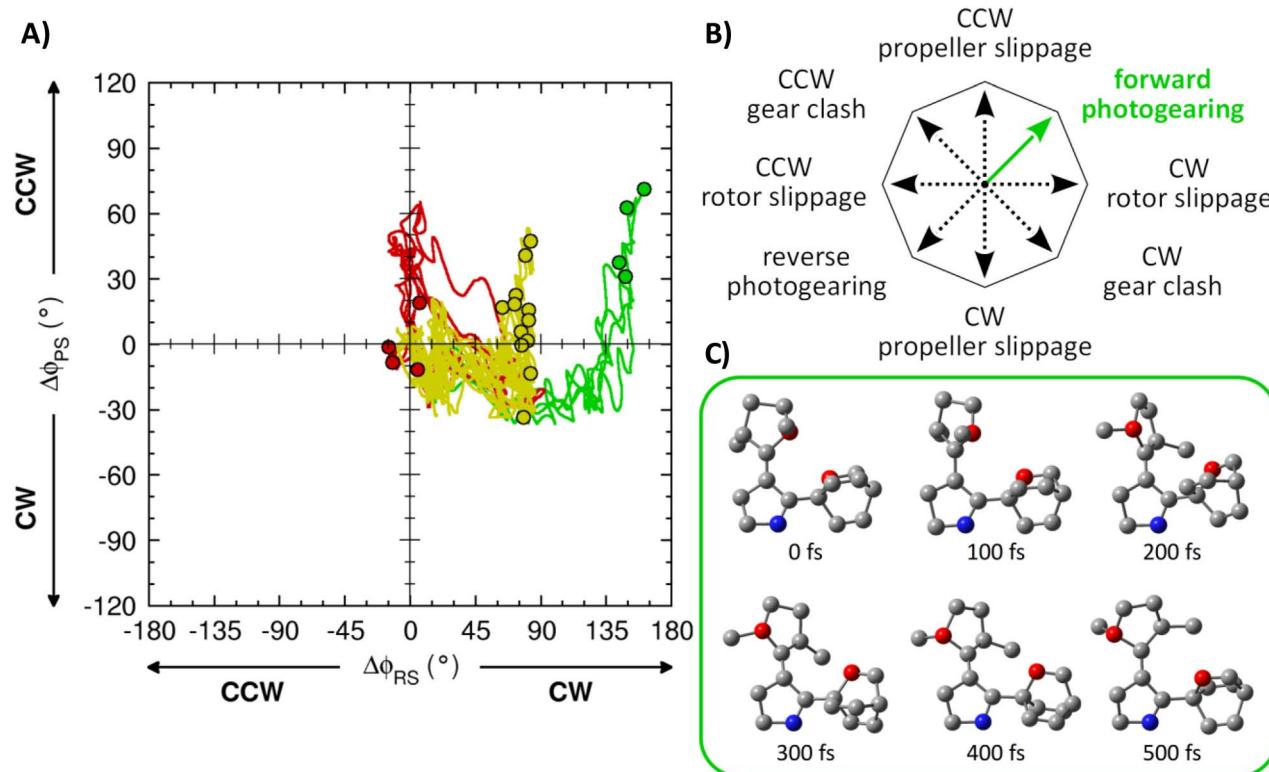


Fig. 4 (A) Key structural changes along the 20 NAMD trajectories calculated for PG-3. The color-coding used for plotting the trajectories is explained in the main text. For ease of interpretation, the final structure of each trajectory is encircled. The plot shows the changes in the  $\varphi_{RS}$  and  $\varphi_{PS}$  dihedral angles ( $\Delta\varphi_{RS}$  and  $\Delta\varphi_{PS}$ ) relative to the unique  $\varphi_{RS}$  and  $\varphi_{PS}$  values at the starting point of each trajectory. Accordingly, each trajectory begins at  $\Delta\varphi_{RS} = 0^\circ$  and  $\Delta\varphi_{PS} = 0^\circ$ . (B) Schematic illustration of all possible reaction outcomes (except FC repopulation) following photoexcitation of PG-3. (C) Structural snapshots at 100 fs intervals from a trajectory showing forward photogearing, with all hydrogen atoms hidden and two carbon atoms highlighted in red color for ease of interpretation. Notice how the positions of the highlighted carbon atoms gradually change from right to left (the rotor carbon), and from back to front (the propeller carbon), during the CW rotor and CCW propeller rotations.

commenting on the results, it should be noted that the purpose of the NAMD simulations is not to obtain a quantitative estimate of the photogearing quantum yield of PG-3 – this would require calculating many more and much longer trajectories, which is not yet affordable for CASSCF-based NAMD simulations of systems the size of PG-3 (44 atoms). Rather, the purpose is to probe, qualitatively, whether photogearing is a viable reaction channel. Besides emphasizing this point, it may also be noted that the necessity to limit the number of states included in the simulations (2,  $S_0$  and  $S_1$ ) implies that channels involving other excited states than the bright  $\pi\pi^*$  state will not be manifested.

Finding that all of the 20 trajectories in Fig. 4 can be classified into one of three different categories, shown in green, red and yellow colors, it is notable that four of them (20%, green color) induce forward photogearing and decay to the  $S_0$  state before the 500 fs mark. In fact, the number of such trajectories is identical to the number of trajectories leading to FC repopulation (20%, red color). Combined with the fact that the remaining 12 trajectories (60%, yellow color) show no indication that other processes would outcompete photogearing, these results lend credence to the idea that photogearing is indeed a feasible reaction channel (the 12 trajectories in

question reach the  $S_1/S_0$  CI region but do not decay to the  $S_0$  state within 500 fs). At the same time, despite the absence of any imposed reaction coordinate in the NAMD simulations, it is certainly possible for the simulations to be biased toward forward photogearing, in the same way as previous studies have documented bias in the predicted outcomes of other photochemical reactions by computational modeling.<sup>61</sup>

Nonetheless, as further evidence that the four green-colored trajectories in Fig. 4 represent forward photogearing, propagating them adiabatically in the  $S_0$  state for an additional 500 fs readily produced a full CCW propeller rotation with  $\Delta\varphi_{PS} = +120^\circ$  (see Fig. S7 and Section 5 of the SI, as well as the multimedia file described in Section 7 of the SI). Analogous extended simulations were also performed for the four red-colored trajectories in Fig. 4 (see Section 5 of the SI).

Based on the results in Fig. 4, two additional comments are in place. First, we note that photogearing occurs exclusively in the forward direction, whereby a CW rotor rotation induces a CCW propeller rotation. This finding is consistent with the prediction of a preferred photogearing direction by the static calculations in Fig. 2 and 3. Second, we note that the photogearing is asynchronous, with the propeller rotation lagging behind the rotor rotation quite considerably. In fact, although



the propeller rotation is set in motion in the  $S_1$  state (as can be inferred from Fig. 3), most of the rotation occurs when the photoexcited system has decayed to the  $S_0$  state through the CI for rotor rotation, and quite large  $\Delta\phi_{RS}$  values have been attained (see Fig. 4).

Importantly, maximally efficient photogearing also requires that neither rotor rotation nor propeller rotation can be triggered thermally. Accordingly, it is advantageous if the thermal barriers for free-standing rotor and propeller rotations are large. Hence, we calculated the barriers of these “slippage” processes for **PG-3** with two different density functionals, as shown in Table S2 of the SI. From these calculations, it can be seen that the free-energy barrier for rotor slippage around the associated double bond is of such magnitude (33–40 kcal mol<sup>−1</sup>) that this process does not influence the photogearing efficiency. The barrier for single-bond propeller slippage, on the other hand, is sufficiently small (only 7–8 kcal mol<sup>−1</sup>) that this moiety is likely to start rotating in the dark, which would be detrimental to photogearing. However, this appears to be a manageable issue, because complementary calculations also included in Table S2 identified two different strategies for raising this barrier within the confines of the **PG-3** template. More specifically, these strategies are represented by the **PG-3** derivatives shown in Fig. S9 of the SI.

The first strategy is to increase the steric repulsion between the propeller and the stator, which can be done by replacing the protonated Schiff base of the stator with a methylated ditto. This was found to raise the barrier to 10–11 kcal mol<sup>−1</sup> (see Table S2). The second strategy is to instead increase the steric repulsion between the propeller and the rotor, which is readily achieved by replacing the small barrelene propeller with the larger triptycene propeller so commonly employed by thermally driven molecular gears<sup>24,28,29</sup> (and also by the **PG-1** photogear in Fig. 1).<sup>42</sup> Encouragingly, this raised the barrier to a considerable 20–23 kcal mol<sup>−1</sup> (see Table S2).

Here, it should be pointed out that the indirect, positive effect on the photogearing efficiency that these two strategies offer through inhibition of propeller slippage, could be offset by a direct, negative effect from the altered steric interactions they introduce. While an analysis of this possibility is beyond the scope of the present work (and will have to await future NAMD simulations of the larger **PG-3** derivatives in Fig. S9), it deserves to be mentioned that the electronic structure of the rotor-stator core of **PG-3** was found to be quite inert to the altered steric interactions. This may be interpreted to indicate that any such negative effect is not pronounced.

Finally, it is clear that synthetic feasibility would matter for experimental realization and testing of the **PG-3** design. Pleasingly, synthetic procedures are available for both the rotor-stator core<sup>54,62</sup> and the barrelene propeller.<sup>63–65</sup> Still, barrelenes are sometimes regarded as “synthetically challenging”.<sup>65</sup> Furthermore, barrelenes can photoisomerize into cyclo-octatetraenes and semibullvalenes,<sup>66</sup> which also would have to be dealt with. Again, as was the case for propeller slippage, a viable solution to these challenges could be to replace barrelene with triptycene (see Fig. S9). Indeed, the synthesis of triptycenes is well developed.<sup>67,68</sup> Moreover, their larger size offers

further opportunities to reducing the propeller symmetry through chemical modification, which would make it easier to observe the photogearing experimentally.

## Conclusions

In summary, we have discovered a simple mechanism for realizing molecular photogearing in a compound where a  $C_3$ -symmetric propeller is added to a minimal light-driven molecular motor of protonated Schiff-base type.<sup>52–55</sup> Specifically, we have shown that combining these building blocks produces a photogear whose through-space transmission of double-bond rotary motion into directed single-bond rotary motion occurs in a single step, and whose efficiency is tunable through steric interactions. Altogether, the results of this work identify a path toward progress on a problem of profound importance in the field of artificial molecular machines.

## Author contributions

The authors contributed equally to all parts of this work.

## Conflicts of interest

There are no conflicts to declare.

## Data availability

The data supporting this article have been included as part of the SI. Supplementary information: computational details, additional results (Fig. S1–S9 and Tables S1 and S2), description of SI multimedia file, and Cartesian coordinates and energies of optimized geometries. See DOI: <https://doi.org/10.1039/d5sc05065k>.

## Acknowledgements

We acknowledge financial support from the Swedish Research Council (grants 2019-03664 and 2022-06442), the Olle Engkvist Foundation (grant 204-0183) and the Carl Trygger Foundation (grant CTS 24:3446). The computations were enabled by resources provided by the National Academic Infrastructure for Supercomputing in Sweden (NAIIS) and the Swedish National Infrastructure for Computing (SNIC) at the National Supercomputer Centre partially funded by the Swedish Research Council (grants 2022-06725 and 2018-05973).

## Notes and references

- 1 J. E. Walker, M. Saraste and N. J. Gay, *Biochim. Biophys. Acta*, 1984, **768**, 164–200.
- 2 W. Junge, H. Lill and S. Engelbrecht, *Trends Biochem. Sci.*, 1997, **22**, 420–423.
- 3 P. D. Boyer, *Annu. Rev. Biochem.*, 1997, **66**, 717–749.
- 4 M. Saraste, *Science*, 1999, **283**, 1488–1493.
- 5 M. Yoshida, E. Muneyuki and T. Hisabori, *Nat. Rev. Mol. Cell Biol.*, 2001, **2**, 669–677.



6 N. Hirokawa, *Science*, 1998, **279**, 519–526.

7 R. D. Vale and R. A. Milligan, *Science*, 2000, **288**, 88–95.

8 R. D. Vale, *Cell*, 2003, **112**, 467–480.

9 N. Hirokawa and Y. Noda, *Physiol. Rev.*, 2008, **88**, 1089–1118.

10 N. Hirokawa, Y. Noda, Y. Tanaka and S. Niwa, *Nat. Rev. Mol. Cell Biol.*, 2009, **10**, 682–696.

11 V. Balzani, A. Credi, F. M. Raymo and J. F. Stoddart, *Angew. Chem., Int. Ed.*, 2000, **39**, 3348–3391.

12 K. Kinbara and T. Aida, *Chem. Rev.*, 2005, **105**, 1377–1400.

13 W. R. Browne and B. L. Feringa, *Nat. Nanotechnol.*, 2006, **1**, 25–35.

14 S. Erbas-Cakmak, D. A. Leigh, C. T. McTernan and A. L. Nussbaumer, *Chem. Rev.*, 2015, **115**, 10081–10206.

15 F. Lancia, A. Ryabchun and N. Katsonis, *Nat. Rev. Chem.*, 2019, **3**, 536–551.

16 S. Corra, M. Curcio, M. Baroncini, S. Silvi and A. Credi, *Adv. Mater.*, 2020, **32**, 1906064.

17 D. Dattler, G. Fuks, J. Heiser, E. Moulin, A. Perrot, X. Yao and N. Giuseppone, *Chem. Rev.*, 2020, **120**, 310–433.

18 Y. Kawada and H. Iwamura, *J. Org. Chem.*, 1980, **45**, 2547–2548.

19 H. Iwamura and K. Mislow, *Acc. Chem. Res.*, 1988, **21**, 175–182.

20 A. M. Stevens and C. J. Richards, *Tetrahedron Lett.*, 1997, **38**, 7805–7808.

21 J. Clayden and J. H. Pink, *Angew. Chem., Int. Ed.*, 1998, **37**, 1937–1939.

22 W. Setaka, T. Nirengi, C. Kabuto and M. Kira, *J. Am. Chem. Soc.*, 2008, **130**, 15762–15763.

23 J. Ściebura, P. Skowronek and J. Gawronski, *Angew. Chem., Int. Ed.*, 2009, **48**, 7069–7072.

24 D. K. Frantz, A. Linden, K. K. Baldridge and J. S. Siegel, *J. Am. Chem. Soc.*, 2012, **134**, 1528–1535.

25 F. Huang, G. Wang, L. Ma, Y. Wang, X. Chen, Y. Che and H. Jiang, *J. Org. Chem.*, 2017, **82**, 12106–12111.

26 G. Erbland, S. Abid, Y. Gisbert, N. Saffon-Merceron, Y. Hashimoto, L. Andreoni, T. Guérin, C. Kammerer and G. Rapenne, *Chem.-Eur. J.*, 2019, **25**, 16328–16339.

27 I. Liepuoniute, M. J. Jellen and M. A. García-Garibay, *Chem. Sci.*, 2020, **11**, 12994–13007.

28 X. Jiang, S. Yang, M. J. Jellen, K. N. Houk and M. Garcia-Garibay, *Org. Lett.*, 2020, **22**, 4049–4052.

29 M. J. Jellen, I. Liepuoniute, M. Jin, C. G. Jones, S. Yang, X. Jiang, H. M. Nelson, K. N. Houk and M. Garcia-Garibay, *J. Am. Chem. Soc.*, 2021, **143**, 7740–7747.

30 Y. Gisbert, S. Abid, C. Kammerer and G. Rapenne, *Chem.-Eur. J.*, 2021, **27**, 12019–12031.

31 M. Baroncini and A. Credi, *Science*, 2017, **356**, 906–907.

32 P. Štacko, J. C. M. Kistemaker, T. van Leeuwen, M.-C. Chang, E. Otten and B. L. Feringa, *Science*, 2017, **356**, 964–968.

33 J.-P. Sauvage, *Acc. Chem. Res.*, 1998, **31**, 611–619.

34 N. Koumura, R. W. J. Zijlstra, R. A. van Delden, N. Harada and B. L. Feringa, *Nature*, 1999, **401**, 152–155.

35 N. Koumura, E. M. Geertsema, M. B. van Gelder, A. Meetsma and B. L. Feringa, *J. Am. Chem. Soc.*, 2002, **124**, 5037–5051.

36 M. Schliwa and G. Woehlke, *Nature*, 2003, **422**, 759–765.

37 E. R. Kay, D. A. Leigh and F. Zerbetto, *Angew. Chem., Int. Ed.*, 2007, **46**, 72–191.

38 S. Saha and J. F. Stoddart, *Chem. Soc. Rev.*, 2007, **36**, 77–92.

39 S. Kassem, T. van Leeuwen, A. S. Lubbe, M. R. Wilson, B. L. Feringa and D. A. Leigh, *Chem. Soc. Rev.*, 2017, **46**, 2592–2621.

40 V. García-López, D. Liu and J. M. Tour, *Chem. Rev.*, 2020, **120**, 79–124.

41 D. R. S. Pooler, A. S. Lubbe, S. Crespi and B. L. Feringa, *Chem. Sci.*, 2021, **12**, 14964–14986.

42 A. Gerwien, F. Gnannt, P. Mayer and H. Dube, *Nat. Chem.*, 2022, **14**, 670–676.

43 M. Guentner, M. Schildhauer, S. Thumser, P. Mayer, D. Stephenson, P. J. Mayer and H. Dube, *Nat. Commun.*, 2015, **6**, 8406.

44 R. Wilcken, M. Schildhauer, F. Rott, L. A. Huber, M. Guentner, S. Thumser, K. Hoffmann, S. Oesterling, R. de Vivie-Riedle, E. Riedle and H. Dube, *J. Am. Chem. Soc.*, 2018, **140**, 5311–5318.

45 C. Petermayer and H. Dube, *Acc. Chem. Res.*, 2018, **51**, 1153–1163.

46 E. Uhl, S. Thumser, P. Mayer and H. Dube, *Angew. Chem., Int. Ed.*, 2018, **57**, 11064–11068.

47 E. Uhl, P. Mayer and H. Dube, *Angew. Chem., Int. Ed.*, 2020, **59**, 5730–5737.

48 F. Gnannt, A. Gerwien, S. Waldmannstetter, S. Gracheva and H. Dube, *Angew. Chem., Int. Ed.*, 2024, **63**, e202405299.

49 E. M. Arpa, S. Stafström and B. Durbeej, *Chem.-Eur. J.*, 2024, **30**, e202303191.

50 M. M. Pollard, A. Meetsma and B. L. Feringa, *Org. Biomol. Chem.*, 2008, **6**, 507–512.

51 B. L. Feringa, *Angew. Chem., Int. Ed.*, 2017, **56**, 11060–11078.

52 J. Wang, B. Oruganti and B. Durbeej, *Phys. Chem. Chem. Phys.*, 2017, **19**, 6952–6956.

53 J. Wang and B. Durbeej, *ChemistryOpen*, 2018, **7**, 583–589.

54 I. Schapiro, M. Gueye, M. Paolino, S. Fusi, G. Marchand, S. Haacke, M. E. Martin, M. Huntress, V. P. Vysotskiy, V. Veryazov, J. Léonard and M. Olivucci, *Photochem. Photobiol. Sci.*, 2019, **18**, 2259–2269.

55 J. Wang and B. Durbeej, *Comput. Theor. Chem.*, 2019, **1148**, 27–32.

56 S. Gozem, F. Melaccio, A. Valentini, M. Filatov, M. Huix-Rotllant, N. Ferré, L. M. Frutos, C. Angeli, A. I. Krylov, A. A. Granovsky, R. Lindh and M. Olivucci, *J. Chem. Theory Comput.*, 2014, **10**, 3074–3084.

57 I. F. Galván, M. G. Delcey, T. B. Pedersen, F. Aquilante and R. Lindh, *J. Chem. Theory Comput.*, 2016, **12**, 3636–3653.

58 B. O. Roos, P. R. Taylor and P. E. M. Siegbahn, *Chem. Phys.*, 1980, **48**, 157–173.

59 R. Crespo-Otero and M. Barbatti, *Chem. Rev.*, 2018, **118**, 7026–7068.

60 S. Mai, P. Marquetand and L. González, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2018, **8**, e1370.

61 T. Fischer, J. Leitner, A. Gerwien, P. Mayer, A. Dreuw, H. Dube and J. Wachtveitl, *J. Am. Chem. Soc.*, 2023, **145**, 14811–14822.



62 R. R. Paccani, D. Donati, S. Fusi, L. Latterini, G. Farina, V. Zanirato and M. Olivucci, *J. Org. Chem.*, 2012, **77**, 1738–1748.

63 H. E. Zimmerman and R. M. Paufler, *J. Am. Chem. Soc.*, 1960, **82**, 1514–1515.

64 H. E. Zimmerman, G. L. Grunewald, R. M. Paufler and M. A. Sherwin, *J. Am. Chem. Soc.*, 1969, **91**, 2330–2338.

65 J. M. Bloch, E. Savelson, A. Q. Meng, M. N. Ericson, I. U. Patel, D. A. Dickie, J. J. Tepe and W. D. Harman, *J. Am. Chem. Soc.*, 2025, **147**, 30146–30153.

66 H. E. Zimmerman and G. L. Grunewald, *J. Am. Chem. Soc.*, 1966, **88**, 183–184.

67 Y. Jiang and C.-F. Chen, *Eur. J. Org. Chem.*, 2011, **2011**, 6377–6403.

68 M. Woźny, A. Mames and T. Ratajczyk, *Molecules*, 2022, **27**, 250.

