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A unified approach to *meta*-selective methylation, mono-, di- and trifluoromethylation of arenes†

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Matched molecular series (MMS) are series of molecules that differ only by a single modification at a specific site. The synthesis of MMS is a desirable strategy in drug discovery campaigns. Small aliphatic motifs, notably methyl, mono-, di- and trifluoromethyl substituents (C_1 units), are known to have profound effects on the physicochemical properties and/or potency of drug candidates. In this context, we herein report a unique strategy for achieving direct *meta*-selective methylation, mono-, di-, and trifluoromethylation from the same parent compound. This approach takes advantage of a highly *meta*-selective ruthenium(II)-catalyzed alkylation, followed by a subsequent photocatalyzed protodecarboxylation or silver-mediated fluorodecarboxylation to reveal the (fluoro)methyl moiety. This method enables the late-stage access to MMS in small molecules bearing a variety of orienting groups as well as bio-relevant molecules containing complex functionalities, bypassing the need for *de novo* synthesis to access individual compounds in a series. Moreover, key physicochemical properties of drug candidates were successfully modulated, highlighting opportunities to accelerate medicinal chemistry programs in a sustainable fashion.

Introduction

In medicinal chemistry, incorporating small C_1 aliphatic substituents, including methyl, monofluoromethyl, difluoromethyl and trifluoromethyl groups, has proven to be a valuable strategy for enhancing the pharmacokinetic or pharmacodynamic properties of parent compounds (Fig. 1A). These derivatives are referred to as matched molecular series (MMS) as an extension of the concept of matched molecular pairs (MMP), where a series of compounds share a common scaffold but structurally differ from one another at a single site.^{1–4} The addition of a methyl group to a target molecule has the potential to enhance the potency or efficacy of a compound, widely recognized as the “magic methyl effect”.^{5,6} On a different note, fluoromethyl groups, while similar in size to methyl substituents, exert different and unique effects on drug candidate properties.^{7–10} The presence of fluorine atoms is reported to improve stability by replacing metabolically labile hydrogen atom(s),^{11–13} to modulate conformation,^{14–16} or to influence membrane permeability *via* lipophilicity modulation.¹⁷ In

addition, mono- and difluoromethyl groups may serve as bioisosteres of hydroxyl and thiol functional groups due to their ability to behave as hydrogen bond donors.^{18–20} As a result, methyl, mono-, di- and trifluoromethyl analogues of drug candidates can bring immense value in the drug development process, as important biological properties can be positively influenced while leaving core structures intact.

Benzenoid rings are by far the most privileged ring system present in small molecule drugs.^{21–24} Considering substitution patterns of arenes, *para*-substituted phenyl rings significantly outnumber analogues bearing *ortho*- or *meta*-substitution patterns, with the latter being the least represented scaffold (Fig. 1B).²⁵ The scarcity of the *meta*-substitution pattern underscores a considerable unmet need for efficient synthetic methodologies. To access a di-substituted benzenoid without the need for reconstructing the ring, chemists typically either perform functional group interconversions on one of the existing substituents of a di-functionalized ring or introduce a new group to a mono-functionalized ring. The former strategy, typically employed in C–C cross-coupling reactions,^{26–28} is heavily dependent on the availability of the starting material. However, *meta*-substituted aryl halides, the most widely used precursors for such transformations, have the poorest commercial availability.²⁹ Alternatively, electrophilic aromatic substitution (S_EAr) is one of the most conventional reactions to introduce a new substituent on a phenyl ring. However, the success of S_EAr relies on the electronic and steric nature of the existing substituents on the phenyl ring. Electron withdrawing substituents can direct S_EAr to *meta*-functionalization, but often suffer from sluggish reaction rates.^{30,31} More recently,

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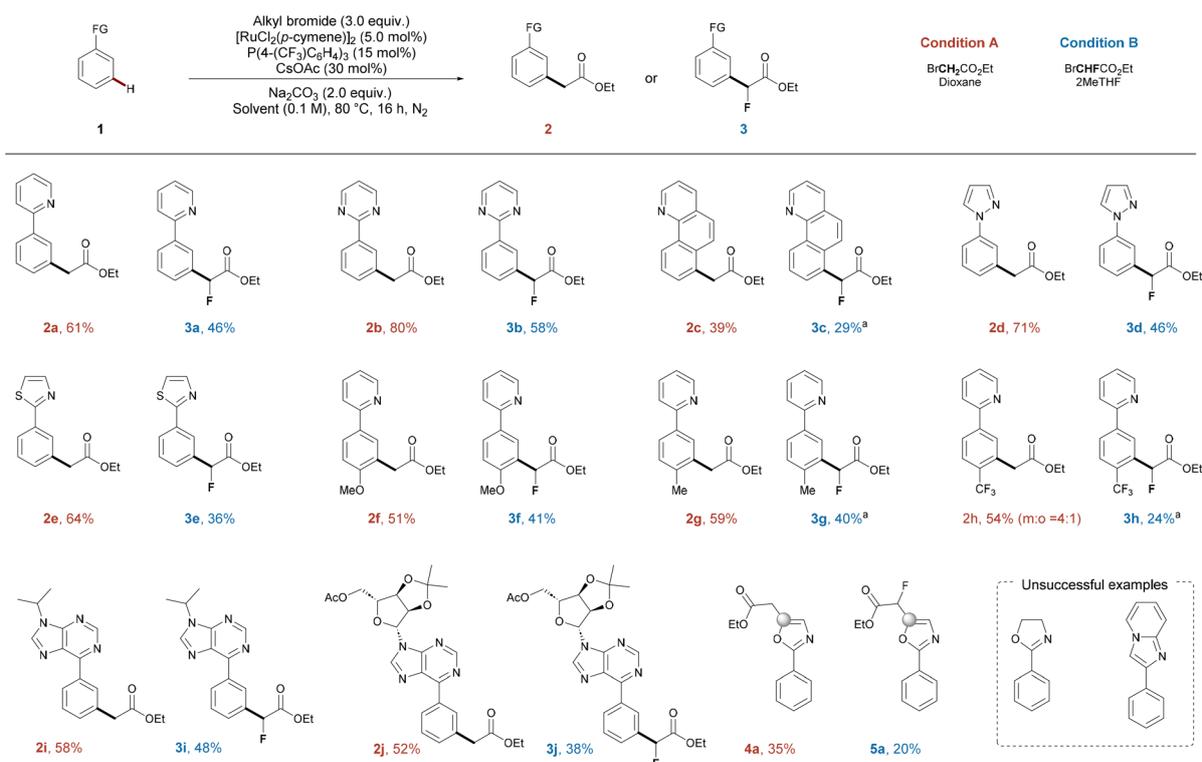


produced the desired *meta* product in 9% NMR yield. However, further optimization did not improve the yield and switching to ICH_2TMS altered the selectivity to, in this case, the *ortho*-position. After careful consideration, substrates bearing an ester stabilizing group were chosen to be the ideal candidates. First, the ester group offers excellent site-selectivity to the desired *meta* position. Second, all the non-, mono- and di-fluorinated substrates are commercially available, which could greatly reduce the cost of an industrial application. Third, an acid functional group can be revealed easily from the ester protection group through saponification, which could then be utilized in a subsequent decarboxylative protonation or fluorination step.

Next, we explored the scope and limitations of the ruthenium-mediated alkylation (Scheme 1). We were pleased to observe that the protocol can be applied to a range of different medicinally relevant heteroarenes including pyridine (**1a**), pyrimidine (**1b**), quinoline (**1c**), pyrazole (**1d**) and thiazole (**1e**). To our delight, bioactive moieties such as purine nucleobase derivative (**1i**, **1j**), including those bearing sensitive protecting groups, were well-tolerated under the optimized conditions. In addition, *para*-substituents bearing different electronic properties, including methoxy (**1f**), methyl (**1g**) and trifluoromethyl (**1h**) groups, proved feasible. Surprisingly, when switching from 2-phenylthiazole to 2-phenyloxazole, the mechanism was completely overturned from *meta*-C–H functionalisation to conventional Friedel Crafts-type reaction.^{73,74} Disappointingly,

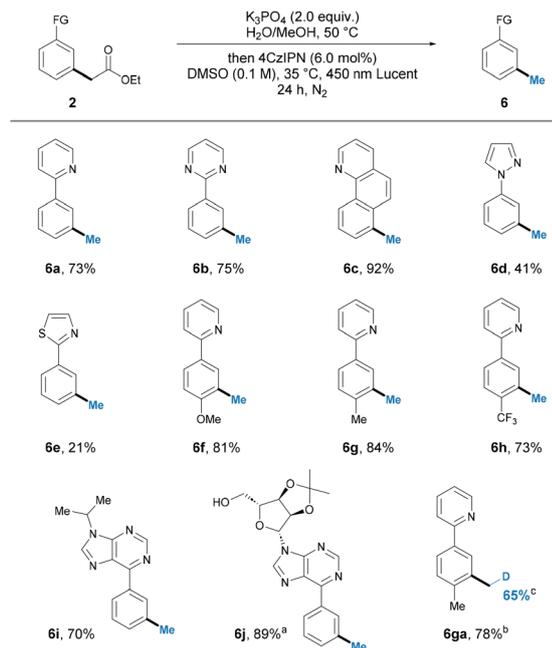
dihydrooxazole and imidazopyridine did not furnish any desired products, with only unreacted substrate observed.

We then turned our attention to investigate the one pot saponification-protodecarboxylation, which would lead to our final *meta*-methylation goal. Conventionally, decarboxylation is performed under harsh reaction conditions, typically requiring heating to over 200 °C, which would pose a challenge to molecules bearing sensitive functional groups. Inspired by a photochemical $^{12}\text{CO}_2$ to $^{13/14}\text{CO}_2$ isotope exchange strategy reported by the Audisio group, we tested our ester substrates under 450 nm blue LED irradiation.⁷⁵ The generation of alpha fluoro-radicals through photo-decarboxylation was also reported.^{76,77} A high-throughput experimentation (HTE) optimization campaign was conducted with three different esters bearing zero, one or two fluorine atoms as substrates (ESI, Scheme S2†). The substrates were subjected to three different bases for the saponification step. Then, solvents were removed under vacuum followed by addition of photocatalyst and a new set of solvents. Photocatalyzed decarboxylation was shown to suffer from increasing difficulty going from esters intermediates bearing zero, one, to two fluorine atoms. The presence of more electron-withdrawing fluorine atoms could potentially make the electropotential of the carboxylate ion more positive, thus more difficult to oxidize, hindering the radical decarboxylation. Therefore, we decided to adopt this strategy only to access the privileged methyl substituent. Of note, it is surprising that all three solvents gave satisfying conversion. DMSO was eventually selected for carrying out the scope exploration since it is



Scheme 1 Ruthenium-catalyzed *meta*-selective alkylation of arenes. ^aReactions were carried out using $[\text{RuCl}_2(p\text{-cymene})_2]$ (10 mol%) and $\text{P}(4\text{-(CF}_3\text{)C}_6\text{H}_4)_3$ (30 mol%).

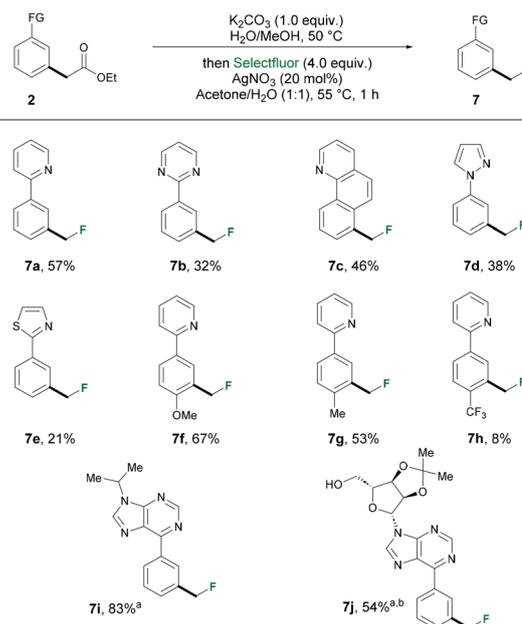




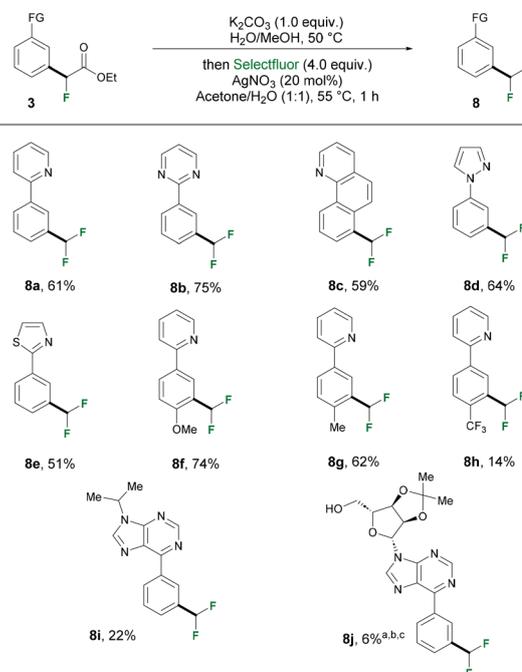
Scheme 2 One pot saponification-protodecarboxylation sequence. ^aTraceless removal of acetyl group during saponification. ^b10 equiv. D₂O were added along with 4CzIPN. ^cPercentage deuterium incorporation.

classified as a greener solvent compared to DMF or DMA. With optimized conditions in hand, we explored the versatility of photodecarboxylation (Scheme 2). All the ester intermediates obtained from the ruthenium(II/III)-catalyzed alkylation step successfully furnished the desired methylation product in satisfactory yields. Deacetylation occurred when acetate intermediate **2j** was subjected to saponification, but to our delight, the presence of free hydroxyl group was well tolerated in the subsequent protodecarboxylation. When 10 equivalents of D₂O were added in the photocatalyzed reaction, a final CH₂D group was successfully formed to yield **6ga**, offering opportunities for the late-stage selective deuterium labelling of drug molecules. The success of deuteration opens the door for potential tritiation of compounds of interest.⁷⁸ In order to obtain mono- and difluoromethylated compounds, a silver-catalyzed fluorodecarboxylation methodology was applied. This strategy also offers a unique opportunity to access trifluoromethylated products, which were inaccessible *via* protodecarboxylation, and was exemplified in the LSF of bioactive molecules. The non-fluorinated and monofluorinated intermediates that were obtained from either conditions **A** or conditions **B** (Scheme 1) were subjected to the same set of conditions for fluorodecarboxylation (Scheme 3). The ester intermediates (**2&3**) were first hydrolyzed by K₂CO₃ to furnish the acetates, then a solvent switch in the same reaction vial was performed, followed by the subsequent silver-mediated fluorodecarboxylation. The intermediates from Scheme 1 successfully underwent the saponification-fluorodecarboxylation sequence to furnish the corresponding mono- or difluoromethylated product with the exception of purine derivatives (**2i**, **2j**, **3j**). These set of

A. Decarboxylation leading to monofluorination



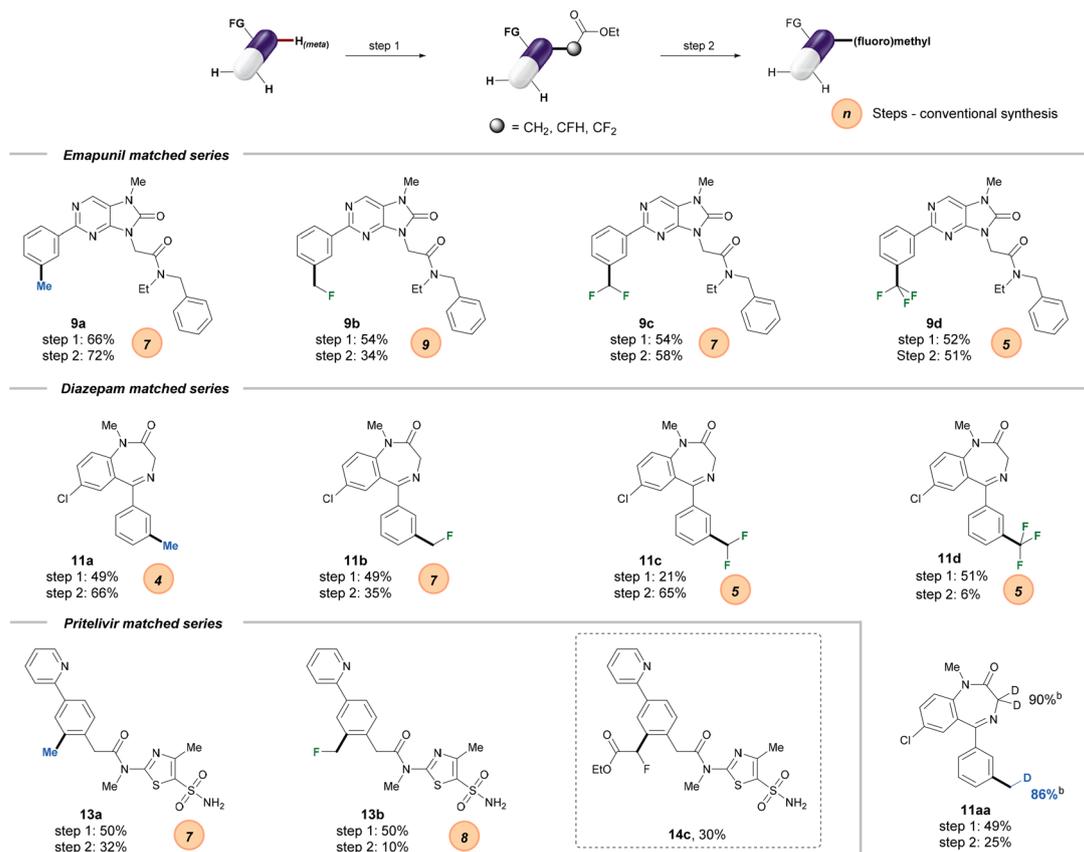
B. Decarboxylation leading to difluorination



Scheme 3 One pot saponification-fluorodecarboxylation sequence. ^aReaction was carried out with the corresponding ethyl acetate and K₂CO₃ (2.0 equiv.) in MeOH/MeOH for saponification, followed by irradiation under 450 nm blue light with (Ir[d(CF₃)ppy]₂(dtbpy))PF₆ (6.0 mol%) in DMF (0.1 M) for 24 h at 35 °C. ^bTraceless removal of acetyl group during saponification. ^cConversion determined by LC-MS.

compounds proved to be challenging in the fluorodecarboxylation step. Instead, we opted for protodecarboxylation strategy described in Scheme 2, yet still managed to furnish moderate to good yield of the desired products (**7i**, **7j**, **8j**), which highlighted



A. Application of ruthenium-mediated C–H alkylation - proto/fluorodecarboxylation sequence for LSF of biologically active substrates^a

B. Modulation of pharmaceutically relevant properties

Parent compound (R = H)	Compound	Solubility (μM, pH 7.4)	LogD (octanol, pH 7.4)	Prot. Bind. (% free)
 Emapunil series	R = H, 9	3.1	3.2	0.9
	R = Me, 9a	0.1	3.9	0.8
	R = CH ₂ F, 9b	0.3	3.5	1.3
	R = CF ₂ H, 9c	0.3	3.9	0.8
	R = CF ₃ , 9d	0.1	4.8	0.2
 Diazepam series	R = H, 11	175	2.7	2.0
	R = Me ^(ortho) , 11a^d	85	2.9	8.5
	R = Me ^(meta) , 11a	478	3.3	2.1
	R = CH ₂ F, 11b	787	2.7	6.5
	R = CF ₂ H, 11c	421	3.1	3.3
R = CF ₃ , 11d	292	3.5	2.1	
 Pritelivir series	R = H, 13	4.5	2.2	- ^c
	R = Me, 13a	13	2.4	- ^c
	R = CH ₂ F, 13b	168	2.3	- ^c

Fig. 2 Late-stage functionalization of bioactive molecules and physicochemical properties modulation. ^aDetailed reaction conditions are provided in ESI.† ^bPercentage deuterium incorporation. ^cUnstable in plasma. ^dReported in literature.⁴²

the versatility of the two-step protocol. Of note, all the respective mono- and difluoromethylated products were obtained in moderate to good yields. Interestingly, fluorodecarboxylation of

the monofluorinated intermediate (**8a–8i**) generally resulted in higher yields compared to the non-fluorinated counterpart (**7a–7i**). Li's group reported a detailed study of such transformation,



indicating that the reactivity of carboxylic acids drops in the order tertiary > secondary > primary,⁷⁹ which serves in part to explain the discrepancy in yields we observed from the two sets of intermediates.

Our newly developed (fluoro)methylation strategy was then applied to more complex bioactive molecules bearing multiple sites for potential functionalization. We were pleased to find that the ruthenium(II/III)-catalyzed alkylation successfully incorporated a masked (fluoro)methyl radical into the desired site of functionalization with traces amount (less than 5% by LC-MS) of *ortho*-product observed when ethyl bromoacetate was used as the radical precursor (Fig. 2A). In all cases, bis-functionalization was not observed. The targeted *meta*-product and undesired *ortho*-product were easily separated owing to their differences in lipophilicity brought about by the transient ester functional group, highlighting an important advantage of the indirect (fluoro)methylation. With emapunil and diazepam as starting material, all four homologues bearing one extra C₁ unit at the *meta* position could be synthesized. In addition, 86% deuterium incorporation was achieved when 10 equivalents of D₂O was added during the protodecarboxylation of diazepam-ethyl acetate intermediate to furnish **11aa** in 12% yield over 2 steps. Ruthenium-catalyzed alkylation was shown to be compatible with pritelivir to yield the unfluorinated and monofluorinated ester intermediate. Yet, fluorodecarboxylation did not yield the desired product with the latter. Nonetheless, an overall *meta*-methylation and monofluoromethylation was still achieved producing two analogues of the parent compound. Of note, trimethyltin hydroxide was used to selectively hydrolyze the transient ester functional group since saponification would also hydrolyze the tertiary amide present in the core structure. In any medicinal chemistry campaign, time is one of the most important factors. As a result, LSF is evolving to be one of the most appealing strategies for accessing synthetically challenging target compounds. We compared the preparation of derivatives of pharmaceuticals (**9a–9d**, **11a–11d**, **13a–13b**) with traditional *de novo* synthesis to produce identical compounds. In this context, expedient access to these unique sets of MMS brings significant value to the study of physicochemical properties and potency of drug candidates, thus speeding up the DMTA cycle in drug discovery.

Access to MMS of marketed pharmaceuticals enabled the generation of key DMPK data including solubility, log *D* and protein binding (Fig. 2B). A general increase in log *D* was observed when a methyl group was installed. Further replacing the hydrogen atoms by fluorine atoms reduced lipophilicity, resulting in a drop in log *D* (CH₂F and CF₂H compared to CH₃), agreeing with previous studies.^{80–82} The trifluoromethylated analogue presented a noticeable increase in lipophilicity compared to the methylated compound, which was consistent with previous report that trifluoromethyl group increased the hydrophilicity when there is no solubilizing group in close proximity.⁸³ Monofluoromethylated and difluoromethylated compounds are in general less lipophilic than the corresponding methylated and trifluoromethylated analogues due to the present of larger dipole moments.⁸⁴ As the parent compound became more lipophilic (Emapunil > Diazepam > Pritelivir), the

boost in log *D* from –CH₃ to –CF₃ also became more prominent.⁸⁵ It is also interesting to note that installation of CH₂F group may have a dramatic impact on the solubility of the target compound. Finally, the site of functionalization can modulate the properties of the parent compound differently. The difference in the medicinal chemistry relevant data of methylation occurring at *ortho* vs. *meta* position of diazepam could be explained by the perturbation of the conformation of the phenyl ring.⁴² *Ortho*-methylation has a higher probability of twisting the phenyl ring out of plane from the 7-membered ring due to higher steric congestion.

Conclusion

In summary, we have devised a strategy allowing direct access to a *meta*-selective C–H methylation and mono-, di-, and trifluoromethylation protocol of arene to quickly access a range of C₁-substituted compounds from the same parent molecule. The site-selectivity was tuned by exploiting the unique reactivity of ruthenium catalysis, and the molecular diversity was achieved by either photocatalyzed protodecarboxylation or silver-mediated fluorodecarboxylation. Moreover, this approach was successfully applied to marketed pharmaceuticals to produce MMS of the parent drug. Specifically, trifluoromethylation was achieved in the last step of synthesizing a potential API, thereby bypassing the generation of intermediate that could be classified as PFAS. An array of pharmaceutically relevant analogues was generated quickly, thus could speed up substantially the drug development campaign. In the meantime, the sustainable strategy reduces the CO₂ footprint by significantly reducing the number of step counts compared to *de novo* synthesis. Lastly, the remote incorporation of C₁ unit provided opportunities to modulate physicochemical properties of bioactive molecules, presenting the potential of this methodology to accelerate optimization towards suitable drug candidates.

Data availability

The data supporting this article have been included as part of the ESI.†

Author contributions

Conceptualization: E. Y. L., M. J. J. and L. A.; methodology: E. Y. L.; writing – original draft: E. Y. L.; writing – review & editing: E. Y. L., M. J. J. and L. A.; funding acquisition: M. J. J. and L. A.; supervision: M. J. J. and L. A.

Conflicts of interest

There are no conflicts to declare.

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