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Diazomethyl- λ^3 -iodane meets aryne: dipolar cycloaddition and C-to-N iodane shift leading to indazolyl- λ^3 -iodanes†

Shinya Otsuki,^a Kazuya Kanemoto,^{ID} *^a Daniel Carter Martos,^{ID} ^b Eunsang Kwon,^{ID} ^{cd} Joanna Wencel-Delord,^{ID} ^{be} and Naohiko Yoshikai,^{ID} *^a

Diazomethyl- λ^3 -iodanes have recently emerged as carbyne equivalents in organic synthesis, enabling the construction of multi-substituted carbon centers through strategic sequential activation of the diazo and iodane functional groups. Distinct from such reaction modes, we report here on the reactivity of diazomethyl- λ^3 -iodanes as iodane-bound 1,3-dipoles toward arynes. Equipped with bis(trifluoromethyl) benzyl alcohol-based benziodoxole (BX) moiety, diazomethyl- λ^3 -iodanes undergo annulation with arynes generated from *ortho*-silylaryl triflates and cyclic diarylhalonium salts, resulting in indazolyl- λ^3 -iodanes through [3 + 2] cycloaddition and carbon-to-nitrogen iodane migration. DFT calculations reveal that diazomethyl-BX prefers [3 + 2] cycloaddition with aryne over aryne insertion into the carbon–iodine(III) bond (carboiodanation) and that the subsequent iodane migration proceeds through two consecutive 1,5-iodane shifts. The utility of these indazolyl-BXs as indazole-transfer agents has been demonstrated by α -functionalization of *N,N*-dimethylaniline derivatives.

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Introduction

Organo- λ^3 -iodanes, especially those equipped with cyclic and pseudocyclic frameworks, have attracted significant attention as valuable reagents for functional group transfer in organic synthesis.¹ Within this category, diazomethyl- λ^3 -iodanes have emerged as notable carbyne surrogates, following the pioneering work of Suero and coworkers in 2018 (Scheme 1A).^{2–4} The diazo and iodane groups in these reagents can be strategically activated through two major, distinct mechanisms. In one approach, single-electron reduction of the C–I(III) bond generates a diazomethyl radical, which can then undergo alkylation of (hetero)aromatic compounds² and other radical transformations.⁵ The remaining diazo group in the product can be harnessed *via* rhodium-catalyzed carbene transfer reactions, such as X–H insertion and cyclopropanation, allowing for the divergent construction of multisubstituted tetrahedral carbons.

Alternatively, initial activation of the diazo group with rhodium leads to the formation of an α -iodanyl Rh-carbene.⁶ The excellent leaving group ability of the iodanyl moiety allows this intermediate to act as an effective Rh-carbynoid, which exhibits dual carbene-carbocation reactivity toward alkenes and alkynes.

Drawing inspirations from these literature precedents and other reported reaction modes, we demonstrate here that diazomethyl- λ^3 -iodanes can act as iodane-bound 1,3-dipoles in reactions with arynes (Scheme 1B). Supported by a benziodoxole (BX) moiety derived from bis(trifluoromethyl)benzyl alcohol, these compounds participate in coupling with arynes, generated from appropriate precursors such as *ortho*-silylaryl triflate⁷ and cyclic diaryl λ^3 -chloranes/bromanes,^{8,9} resulting in the formation of indazolyl- λ^3 -iodanes as novel *N*-based λ^3 -iodane derivatives.¹⁰ The reaction is proposed to proceed through a concerted [3 + 2] cycloaddition between the diazomethyl group and the aryne,¹¹ followed by a carbon-to-nitrogen shift of the iodane moiety. This proposal is substantiated by density functional theory (DFT) calculations, which suggest that the migration of the iodane moiety occurs through two consecutive 1,5-iodane shifts. The overall process represents a unique preparative approach to organo- λ^3 -iodanes that is distinct from simple ligand substitution on an iodine(III) electrophile with a nucleophile.¹² The indazolyl- λ^3 -iodanes proved to serve as reagents for the α -functionalization of *N,N*-dimethylaniline derivatives.

The discovery of the present reaction was guided by our recent study on the carbo- and heteroiodanation reactions of arynes, where organo- λ^3 -iodanes behave as organometallic-like

^aGraduate School of Pharmaceutical Sciences, Tohoku University, Sendai 980-8578, Japan. E-mail: kazuya.kanemoto.a1@tohoku.ac.jp; naohiko.yoshikai.c5@tohoku.ac.jp

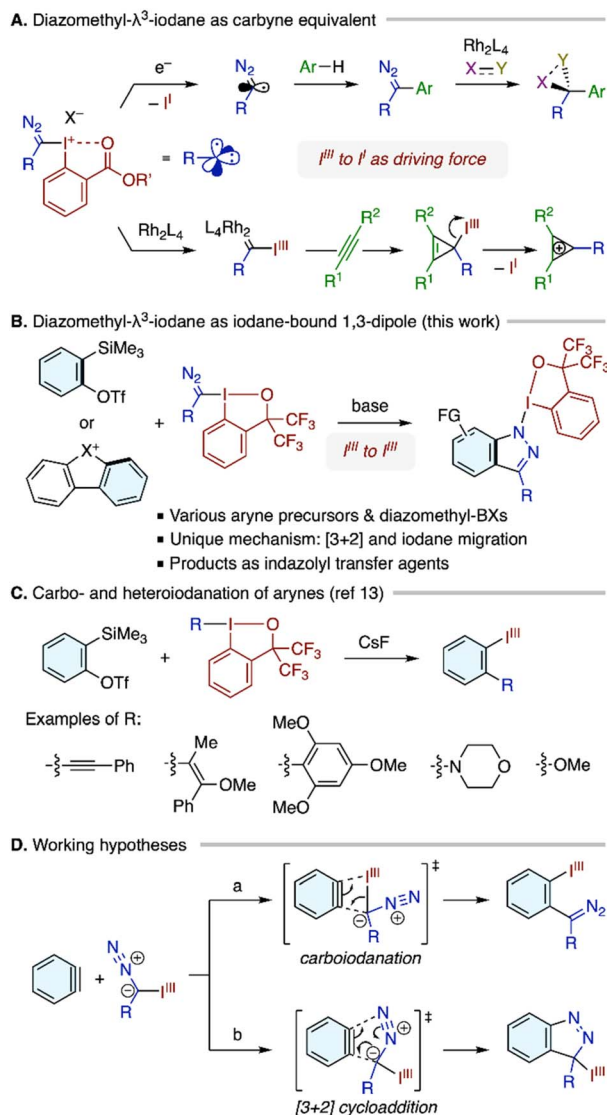
^bLaboratoire d'Innovation Moléculaire et Applications (LIMA, UMR CNRS 7042), Université de Strasbourg/Université de Haute Alsace, ECPM, 67087 Strasbourg, France

^cResearch and Analytical Center for Giant Molecules, Graduate School of Science, Tohoku University, Sendai 980-8578, Japan

^dEndowed Research Laboratory of Dimensional Integrated Nanomaterials, Graduate School of Science, Tohoku University, Sendai 980-8578, Japan

^eInstitute of Organic Chemistry, JMU Würzburg, Am Hubland, Würzburg, Germany

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Scheme 1 Background and summary of the present study.

nucleophiles toward the highly electrophilic C–C triple bond of aryne to afford *ortho*-functionalized aryl- λ^3 -iodanes (Scheme 1C).¹³ These reactions proved feasible with organobenziodoxoles having innately nucleophilic ligands such as alkynyl, electron-rich vinyl/aryl, amino, and alkoxy groups. Given this background and the intrinsic nucleophilic character of the diazomethyl carbon, we hypothesized that diazomethyl-BX may also undergo the addition across aryne to afford *ortho*-(diazomethyl)aryl-BX (Scheme 1D, path a). At the same time, an alternative hypothesis was drawn from the known reactivity of diazo compounds toward [3 + 2] cycloaddition with arynes (Scheme 1D, path b).¹¹

Results and discussion

The diazomethyl- λ^3 -iodanes requisite for the present study could be synthesized from acetoxybenziodoxole and the corresponding diazo compounds (see the ESI† for details).

Table 1 Annulation of 3-methoxybenzyne and diazomethyl-BX **2a** into indazolylium-BX **3aa**^a

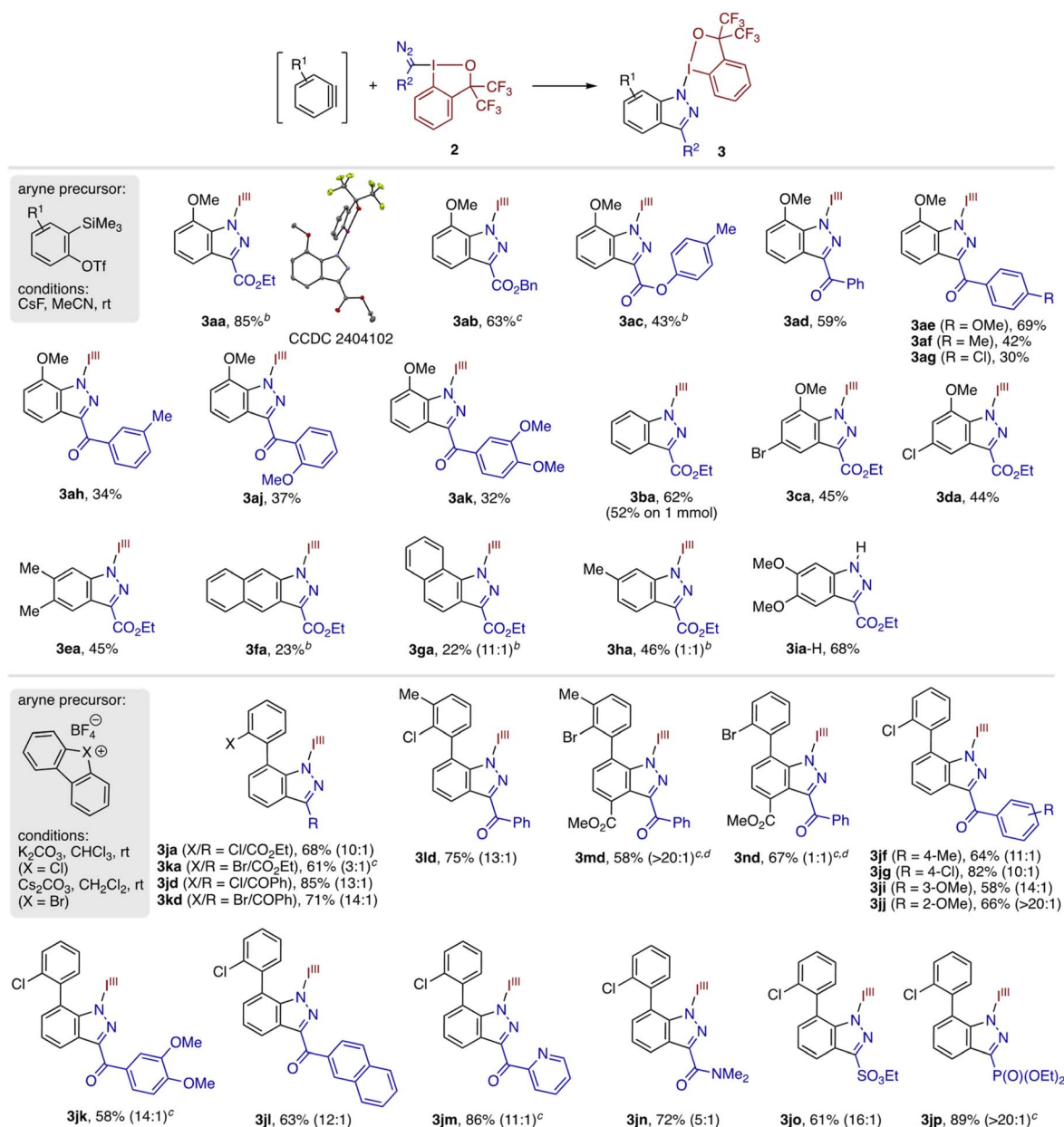
Entry	Activator	Solvent	Yield ^b [%]
1	CsF	MeCN	85 ^c
2	CsF	DME	70
3	CsF	THF	65
4	CsF	Toluene	0
5	TBAF	THF	0
6	KF/18-crown-6	THF	80
7 ^d	CsF	MeCN	91
8 ^e	CsF	MeCN	0

^a Unless otherwise noted, the reaction was performed using 0.2 mmol of **1a**, 0.1 mmol of **2a**, and 0.4 mmol of activator in a solvent at a concentration of 0.2 M. ^b Determined by ¹H NMR using 1,1,2,2-tetrachloroethane as an internal standard. ^c Isolated yield. ^d The reaction was performed using 0.1 mmol of **1a**, 0.3 mmol of **2a**, and 0.2 mmol of CsF. ^e Reagent **2a'** was used instead of **2a**.

Subsequent exploration of their reactivity toward arynes revealed the relevance of our second hypothesis (Scheme 1D, path b). Thus, the reaction between 3-methoxybenzyne, generated by activation of *ortho*-trimethylsilylaryl triflate **1a** with CsF, with diazoester derivative **2a** proceeded smoothly in acetonitrile at room temperature, affording indazolylium-BX **3aa** as a single regioisomer in 85% isolated yield (Table 1, entry 1). X-ray crystallographic analysis unambiguously established the molecular structure of **3aa**, the indazole ring of which was likely constructed through regioselective [3 + 2] cycloaddition (*vide infra*).¹⁴ The reaction became somewhat sluggish in ethereal solvents such as DME and THF (entries 2 and 3) and did not take place at all in toluene (entry 4). Among other common fluoride activators, TBAF caused the formation of a complex mixture (entry 5), whereas KF/18-crown-6 proved as effective as CsF (entry 6). A high yield of **3aa** could also be achieved by using **1a** as the limiting reagent in combination with an excess amount (3 equiv.) of **2a** (entry 7). In later studies, these “iodane-excess” conditions proved to work better for the majority of Kobayashi-type aryne precursors (*vide infra*). Notably, the diazomethylbenziodoxolone **2a'**, derived from 2-iodobenzoic acid, failed to undergo an annulation reaction with 3-methoxybenzyne but produced a complex mixture, demonstrating the importance of the cyclic skeleton of the reagent (entry 8).

Subsequently, we explored the scope of the indazolylium-BX synthesis from *ortho*-silylaryl triflates and diazomethyl-BXs (Table 2). Diazomethyl-BXs bearing benzyl and aryl ester moieties underwent the annulation with 3-methoxybenzyne to afford the corresponding products **3ab** and **3ac** in moderate to good yields. Likewise, diazomethyl-BXs bearing a variety of aryl ketone moieties participated in the reaction with 3-



Table 2 Scope of indazolyl-BX synthesis via annulation of arynes with diazomethyl-BXs^a

^a Conditions for the reaction using *ortho*-trimethylsilylaryl triflate: aryne precursor (0.1 mmol), diazomethyl-BX (0.3 mmol), CsF (0.2 mmol), MeCN (0.1 or 0.2 M), rt, 18 h. Conditions for the reaction using cyclic diarylchloronium/bromonium salt: aryne precursor (0.05 mmol), diazomethyl-BX (0.1 or 0.15 mmol), K₂CO₃ (for X = Cl, 0.15 mmol) or Cs₂CO₃ (for X = Br, 0.15 mmol), CHCl₃ (for X = Cl, 0.1 M) or CH₂Cl₂ (for X = Br, 0.1 M), rt, 18 h. Unless otherwise noted, isolated yields are shown. The ratio in the parentheses refers to the regioisomer ratio determined by ¹H NMR analysis of the crude product. ^b Aryne precursor (0.2 mmol), diazomethyl-BX (0.1 mmol), and CsF (0.4 mmol) were used. ^c Yield determined by ¹H NMR using 1,1,2,2-tetrachloroethane as an internal standard. ^d Aryne precursor (0.05 mmol), diazomethyl-BX (0.1 mmol), Cs₂CO₃ (0.1 mmol), and THF (0.05 M) were used.

methoxybenzyne, furnishing the desired products **3ad–3ak** in moderate yields. A series of unsubstituted and substituted arynes generated from the respective silylaryl triflate were capable of undergoing annulation with **2a**, affording the corresponding indazolyl-BXs **3ba–3ha** in moderate yields. The reaction between parent benzyne and **2a** could be performed on

a 1 mmol scale, albeit in somewhat diminished yield (see **3ba**; 52% (1 mmol) vs. 62% (0.1 mmol)). Similar to 3-methoxybenzyne, 3-methoxy-5-halobenzyne reacted with exclusive regioselectivity, affording the corresponding regioisomers **3ca** and **3da**. The reaction of 1-naphthalene displayed high regioselectivity (11:1), where the diazomethyl carbon was

preferentially introduced to the 2-position of the naphthalene ring (see **3ga**). Not unexpectedly, 4-methylbenzynes exhibited no regiochemical preference, producing an equimolar mixture of **3ha** and its regioisomer. Unlike these cases, the reaction of 4,5-dimethoxybenzynes and **2a** resulted in the free indazole derivative **3ia-H**, presumably through the *in situ* loss of the iodanyl group upon [3 + 2] cycloaddition. Like this case, the modest yield and failure with Kobayashi aryne precursors are partly or fully ascribed to the decomposition of the desired indazolyl-BX under the reaction conditions (Fig. S2†). Given the generally higher yields obtained with cyclic diarylhalonium precursors (*vide infra*), we suspect that fluoride ion is responsible for this undesirable decomposition, while the exact mechanism of decomposition remains unclear. Additionally, some of the products also proved to undergo partial or full decomposition during silica gel chromatography purification.

In light of the innately sensitive carbonyl groups in the diazomethyl-BX reagents, the present reaction appeared incompatible with many common methods of aryne generation that employ either a strong base or a nucleophilic organometallic reagent as an activator.^{7,15} In this respect, cyclic diaryl- λ^3 -chloranes and bromanes, capable of generating *ortho*-haloaryl-substituted arynes with the aid of mild base,^{8,9} appeared attractive as alternatives to *ortho*-trimethylsilylaryl triflates. To our delight, the present diazomethyl-BXs were found to engage successfully with these aryne precursors (Table 2). Thus, the reaction between cyclic diarylchloronium tetrafluoroborate and the diazoester reagent **2a** took place smoothly in the presence of K_2CO_3 at room temperature, affording the hetero-biaryl-type indazolyl-BX **3ja** in 68% yield and high regioselectivity (10 : 1), with C–C bond formation preferentially taking place at the distal position. The analogous diarylbromonium salt also took part in the reaction with **2a** to give **3ka**, albeit with diminished regioselectivity (3 : 1). Meanwhile, both the halonium salts reacted very smoothly with the diazoketone reagent **2d**, furnishing the desired products **3jd** and **3kd** in good yields with high regioselectivity. An additional methyl substituent on the chloronium salt was well tolerated, affording the product **3ld** with equally good yield and regioselectivity. More structurally complex aryne containing 2-bromo-3-methylphenyl and methoxycarbonyl groups at the 3- and 6-positions exhibited excellent regioselectivity (see **3md**). In contrast, analogous aryne with 2-bromophenyl group afforded an equimolar mixture of **3nd** and its regioisomer, highlighting significant influence of subtle structural change on the regioselectivity. The chloronium reagent displayed tolerance to a wide variety of diazomethyl-BXs, including those containing not only various ketones (**3jf**–**3jm**) but also amide (**3jn**), sulfonate ester (**3jo**), and phosphate ester (**3jp**), to furnish the desired biaryl-type products in good to high yields while displaying regioselectivity higher than 10 : 1 in most cases. In none of these reactions we observed oxidative decomposition of the product that could potentially be induced by the $Cl(III)/Br(III)$ salt. Notably, the 2-pyridyl ketone, amide, sulfonate, and phosphate-containing reagents failed to produce annulation products with the Kobayashi-type aryne precursor. We also note in passing that the reaction between chloronium salt **1j** and **2a** could even be performed under mechanochemical

conditions (30 Hz) to give the product **3ja** in a moderate yield (60% by 1H NMR), whereas a mechanochemical reaction between Kobayashi precursor **1a** and **2a** was unproductive. These observations underscore the outstanding performance of the chloronium salt in the present reaction.

Upon establishing the scope of the cycloaddition, we briefly evaluated the arynophilicity of the present diazomethyliodane.¹⁶ Thus, competitive reactions of **2a** using furan or benzyl azide as counterparts toward parent benzyne resulted in notable preference for **3aa** over the cycloadduct of the counterpart (see the ESI†). This demonstrates that **2a** is significantly more reactive than our previously reported organoiodane arynophiles (Scheme 1C), all of which were less arynophilic than furan.¹³

To gain insight into the mechanism of the present reaction, we explored our initial hypotheses (Scheme 1D) using DFT calculations (Fig. 1). The [3 + 2] cycloaddition between parent benzyne and model diazomethyl-BX (**SM**) proceeds *via* a concerted transition state **TS1** with an activation energy of 9.2 kcal mol^{−1}. This process is highly exergonic, leading to 3*H*-indazol-3-yl-BX (**INT1**) with a free energy change of −78.9 kcal mol^{−1}. Notably, the alternative pathway, namely, the insertion of benzyne into the carbon–iodine(III) bond, is also feasible through a four-centered transition state **TS1'**, which directly leads to *ortho*-diazomethylphenyl-BX (**INT1'**) with significant exergonicity. However, its activation energy (12.3 kcal mol^{−1}) is distinctly higher than that of the [3 + 2] cycloaddition, consistent with the lack of observation of the expected carboiodanation product. Featuring a typically unstable C(sp³)–iodine(III) bond, **INT1** undergoes a facile 1,5-iodane shift (**TS2**, $\Delta G^\ddagger = 8.5$ kcal mol^{−1}) from the C³-position to the N²-position of the indazole ring to afford 2*H*-indazol-2-yl-BX (**INT2**), which is more stable than **INT1** by over 15 kcal mol^{−1}. Finally, **INT2** undergoes a second 1,5-iodane shift (**TS3**, $\Delta G^\ddagger = 22.6$ kcal mol^{−1}) from the N² position to the N¹ position, leading to the experimentally relevant 1*H*-indazol-1-yl-BX (**PD**), with a finite but significant additional stabilization (*ca.* 5 kcal mol^{−1}) from **INT2**.

The competitive transition states of [3 + 2] cycloaddition (**TS1**) and carboiodanation (**TS1'**) merit further discussion. Reflecting the nucleophilic nature of the diazomethyl carbon, both **TS1** and **TS1'** exhibit similar distortions in the benzyne segment, with a wider bond angle (131–135°) at the carbon accepting the nucleophilic diazomethyl carbon and a narrower bond angle (117–122°) at the other carbon. Notably, the degree of aryne distortion is slightly more pronounced in **TS1'** than in **TS1**, which is also reflected in a more polarized C–C triple bond in **TS1'**, as indicated by the natural population analysis (NPA) charges (C1, +0.04; C2, −0.17 in **TS1'** and C1, −0.03; C2, −0.08 in **TS1**). This pronounced aryne distortion and polarization in **TS1'** may be attributed to the more electrostatic nature of the aryne–aryne interaction in **TS1'**, due to the highly positive iodine(III) center. The mode of aryne distortion in **TS1** aligns with the perfect regioselectivity observed with 3-methoxybenzynes (see **3aa**), whose distal carbon is distorted toward linearity to serve as a more electrophilic site.¹⁷ Our DFT calculations support this conjecture (Fig. S4†). The [3 + 2] cycloaddition between 3-methoxybenzynes and **SM**, following the experimentally observed regioselectivity, involves monotonously downhill



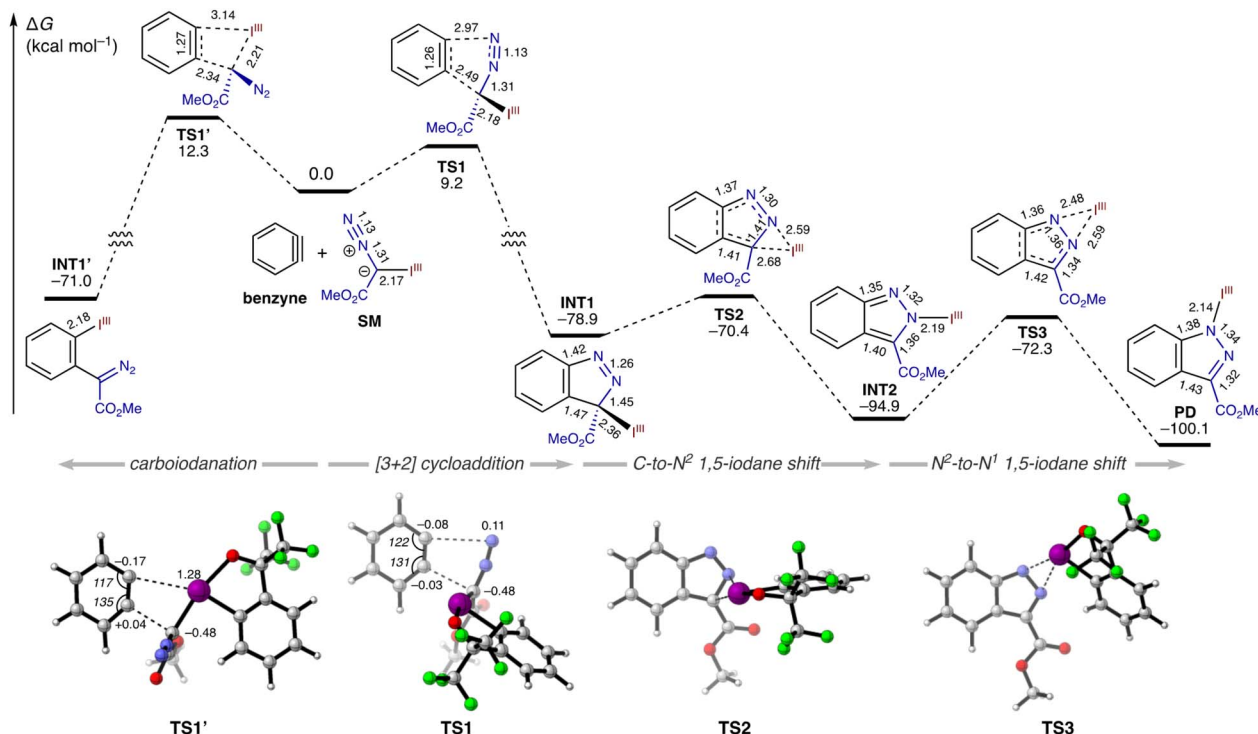


Fig. 1 Computed energy profile for the reaction between diazomethyl-BX and benzyne (level: M06-2X/6-311++G(2df,2p)-SDD(for I)/SMD(THF)//M06-2X/6-31G(d)-SDD(for I)). Bond lengths are indicated in Å.

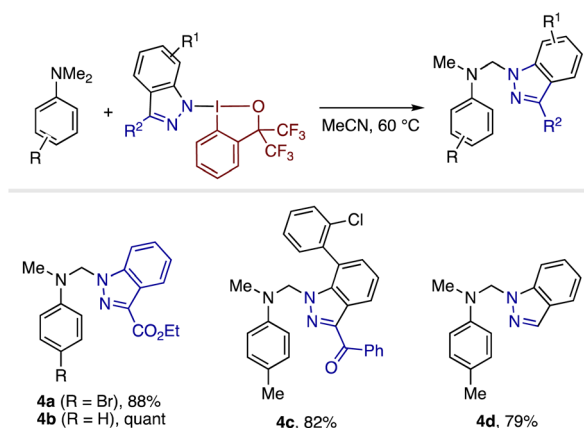
potential energy surface without a discernible transition state. In contrast, a transition state leading to the alternative regioisomer is located approximately 3 kcal mol⁻¹ higher in energy. Meanwhile, 3-(2-halophenyl)benzynes, generated from cyclic diarylchloronium and bromonium salts, have been shown to display slightly higher linearity at the proximal carbon, indicating its intrinsic inclination toward accepting nucleophilic attack.^{8c} Nevertheless, the observed regioselectivity for these arynes can be rationalized by steric factors, favoring the attack of the bulky α -iodanyl diazomethyl carbon to the distal position of 3-arylarlyne. The comparison of two possible transition states for the [3 + 2] cycloaddition between 3-(2-

chlorophenyl)benzyne and **SM** was in accordance with this regioselectivity (Fig. S5†).

Inspired by the seminal studies of Zhdkankin and Kiyokawa/Minakata on the use of nitrogen-bound benziodoxolones as reagents for radical C(sp³)-H functionalization,¹⁸ we identified the ability of the present indazolyl-BXs as indazole-transfer agents (Scheme 2). Thus, the reaction between 4-bromo-*N,N*-dimethylaniline and **3ba** in MeCN at 60 °C resulted in the installation of the indazolyl group at the α -position, affording the product **4a** in 88% yield.¹⁸ Analogous reactions took place smoothly using different dimethylanilines and indazolyl-BXs, furnishing the corresponding products **4b** and **4c** in high yields. Besides the indazolyl-BXs accessible by the present [3 + 2] reaction, parent indazolyl-BX (**3x**), which can be synthesized independently from indazole and benziodoxole chloride under basic conditions (see the ESI†), also proved to participate in the same reaction, furnishing the product **4d** in good yield. As this reaction was significantly shut down by the addition of TEMPO, we presume that the reaction proceeds through a mechanism involving radical processes initiated by single-electron transfer from *N,N*-dimethylaniline to indazolyl-BX (see Scheme S1† for a proposed reaction pathway).

Conclusions

In summary, we have demonstrated the reactivity of diazomethyl- λ^3 -iodanes supported by benziodoxole moiety (diazomethyl-BXs) as iodane-bound 1,3-dipole toward arynes, leading to the formation of indazolyl- λ^3 -iodanes through [3 + 2]



Scheme 2 Use of indazolyl-BXs as group transfer agents for the functionalization of *N,N*-dimethylaniline derivatives.



cycloaddition and carbon-to-nitrogen iodane migration. The present reaction proved compatible with Kobayashi-type *ortho*-trimethylsilylaryl triflate and cyclic diaryl- λ^3 -chloranes/bromanes as aryne precursors. The latter displayed particularly excellent tolerance to a broad range of diazomethyl-BXs, which represents a unique merger of two distinct and complementary hypervalent halogen reagents in a single transformation. The DFT calculations demonstrated the preference for the [3 + 2] cycloaddition over the alternative carboiodanation pathway as well as the stepwise nature of the subsequent iodane shift event. The resulting indazolyl-BXs proved to serve as a new class of N-based iodane group transfer agents,^{10,19} as demonstrated by the C(sp³)-H functionalization of *N,N*-dimethylanilines. The present findings are also notable considering that, despite the extensive use of azole heterocycles as stabilizing elements in cyclic and pseudocyclic iodane compounds,²⁰ λ^3 -iodanes with unsupported azolyl-iodine(III) bonds remain relatively scarce and their reactivity less explored.²¹ Further investigations into the merger of aryne chemistry and nucleophilic hypervalent iodine chemistry are ongoing in our laboratories.

Data availability

The data supporting this article have been included as part of the ESI.† Crystallographic data for **3aa** has been deposited at the CCDC under 2404102 and can be obtained from the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service.

Author contributions

K. K. and N. Y. conceived the project. S. O. performed experiments with the assistance of D. C. M. E. K. performed X-ray crystallographic analysis. N. Y. performed DFT calculations. N. Y. wrote the original draft. K. K., J. W. -D., and N. Y. reviewed and edited the manuscript. N. Y. administered the project.

Conflicts of interest

There are no conflicts to declare.

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- 14 Deposition number 2404102 (for **3aa**) contains the supplementary crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service.
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