

Cite this: *Chem. Sci.*, 2025, 16, 7320

All publication charges for this article have been paid for by the Royal Society of Chemistry

Machine learning-driven global optimization reveals nanometre-scale mixed phases of borophene on Ag(100)[†]

Yunlei Wang, Haifeng Lv * and Xiaojun Wu *

Metal-supported borophene exhibits significant polymorphism and an inherently complex potential energy landscape, posing challenges to exploring its structural diversity. In this study, we integrate a neural network-driven machine learning potential with stochastic surface walking global optimization and an active learning framework to comprehensively map the potential energy surface (PES) of large-size borophene on an Ag(100) substrate. Our exhaustive search identifies 59 857 local minima across 556 distinct supercells, revealing a PES segmented into multiple energy basins and three major funnels. Among the low-energy configurations, 1391 low-energy structures extend to the nanometre scale, showcasing a diverse array of mixed-phase borophene architectures, including monolayer ribbons (β_{12} and χ_3) and bilayer fragments (BL- α_5 , BL- α_7 , BL- α_1 , BL- α_6 , and BL- $\alpha_1\alpha_6$). Notably, the global minimum structures feature monolayers composed of alternating χ_3 and β_{12} ribbons and bilayers formed from BL- α_5 , BL- $\alpha_1\alpha_6$, and BL- α_1 fragments. All mixed-phase borophenes exhibit metallic properties, and their simulated scanning tunneling microscopy (STM) images are provided to facilitate future experimental validation. These findings highlight the extraordinary structural complexity and rich polymorphism of borophene on extended metal surfaces, offering valuable insight into their formation, stability, and potential for tunable electronic properties.

Received 20th December 2024
Accepted 13th March 2025

DOI: 10.1039/d4sc08616c

rsc.li/chemical-science

Introduction

Borophene has attracted extensive attention due to its structural diversity^{1,2} and intriguing physical properties,^{3–5} suggesting potential applications in nanoelectronics. Owing to its electron-deficiency, borophene often adopts frameworks composed of mixed triangles and hexagons^{6,7} resulting in a high degree of polymorphism. Unlike many two-dimensional (2D) materials, borophene does not have a bulk counterpart and generally requires a metallic substrate for stabilization.^{8,9} In experiment, various polymorphs of borophene have been synthesized on different substrates, with the atomic arrangements highly influenced by experimental conditions, highlighting the potential to achieve specific polymorphs with tailored functionalities.^{8,10} However, the identification of synthesized borophene often relies on comparing experimental scanning tunneling microscopy (STM) images with simulated ones. This method is limited in capturing the detailed structural features

of large superlattices, such as mixed phases,^{11,12} ribbon dislocation,¹³ and twin boundaries.¹⁴

Accurately predicting the structures of these large superlattices is essential to capture the complex atomic patterns in borophene.^{15–17} So far, theoretically reported phases are limited to be single-phase at the Angstrom scale. This is because of the exponential increase of local minima with the number of atoms and the rapid growth of computational cost with the number of atoms.^{18,19} Machine learning methods are powerful tools to overcome these limitations, enabling data-driven exploration of broader configurational spaces at a fraction of the computational cost.^{18,20} Machine learning-driven simulations have offered valuable insights into realistic structural motifs, properties,^{21–24} and complex structure–property relationships in a variety of systems.^{25–27} Specifically, the realization of mixed-ribbon monolayer phases on Ag(100) motivates us to fully explore the configurational space. While machine learning-based global structure prediction has successfully discovered new phases of bulk boron^{28–30} and boron clusters,³¹ applying these strategies to borophene on metal substrates remains challenging due to the intricate interplay between boron–boron and boron–metal bonding.

In this work, we develop an on-the-fly neural network (NN) potential combined with stochastic surface walking (SSW) global optimization and an active learning framework to comprehensively explore the PES of borophene on an Ag(100)

State Key Laboratory of Precision and Intelligent Chemistry, School of Chemistry and Materials Sciences, Key Laboratory of Materials Sciences for Energy Conversion, Collaborative Innovation Center of Chemistry for Energy Materials (iChEM), CAS Center for Excellence in Nanoscience, University of Science and Technology of China, Hefei, Anhui, 230026, China. E-mail: hflv@ustc.edu.cn; xjwu@ustc.edu.cn

[†] Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d4sc08616c>



substrate. Our global dataset comprises 177 023 data points, revealing a complex energy landscape with multiple basins and three distinct energy funnels. By examining 59 857 local minima across 556 diverse supercells, we identify 1391 low-energy configurations extending to nanometre-scale lattice constants. Detailed structural analysis shows that the first two energy funnels predominantly host monolayer phases, while the third funnel is characterized by bilayers. Notably, both the first and third funnels display a range of mixed phases, whereas the second funnel remains predominantly single-phase borophene. The first funnel reproduces experimentally observed monolayer structures, validating our computational approach. In particular, the global minimum mixed phases feature monolayers composed of alternating χ_3 and β_{12} ribbons and bilayers formed from $BL-\alpha_5$, $BL-\alpha_1\alpha_6$, and $BL-\alpha_1$ fragments. All these mixed-phase borophenes exhibit metallic properties, and their simulated STM images are provided to support future experimental investigations. This study establishes a general framework for leveraging ML-driven global optimization to uncover the structural characteristics of experimental mixed phases, aiming to achieve the targeted polymorph of borophene.

Methods and models

First-principles calculations

All first-principles calculations were performed using the Vienna *Ab initio* Simulation Package (VASP)^{32,33} with the projector augmented wave (PAW)^{34,35} method to describe ionic cores and valence electrons. The exchange-correlation functional is treated within the generalized gradient approximation (GGA), in the Perdew–Burke–Ernzerhof (PBE) functional.^{36,37} The D3 correction³⁸ was employed to describe the van der Waals interaction between substrates and borophenes. The kinetic energy cutoff for the plane wave basis is 400 eV and the first

Brillouin zone is sampled using the Γ -centered Monkhorst–Pack scheme with a k -space density of $2\pi \times 0.04 \text{ \AA}^{-1}$.

Structure search driven by the neutral network potential

The stochastic surface walking (SSW) global optimization^{39,40} implemented in the Large-scale Atomistic Simulation with neural network Potential (LASP) software⁴¹ is used to sample and search the global potential energy surface (PES). The high dimensional neural network (HDNN) scheme in LASP software is employed to train the global dataset. For more detailed information about SSW methods and HDNN architecture, refer to the ESI.†

Structural models

The Ag(100) substrate model contains two layers with the bottom layer fixed during the SSW optimization. To evaluate the stability of borophene on Ag(100), formation energy (E_{form}) is used, which is defined as $E_{\text{form}} = (E_{\text{tot}} - E_{\text{sub}})/N_{\text{B}}$. E_{tot} and E_{sub} denote the energy of the total system and Ag(100) substrate, respectively, and N_{B} is the number of boron atoms. For additional details, refer to the ESI.†

Results and discussion

Construction of the neutral network potential and global dataset

Fig. 1 illustrates the workflow for constructing the NN potential used to identify global minima of borophene on an Ag(100) surface. This procedure involves three main steps. First, we generate metal substrate supercells of varying sizes and shapes by redefining a matrix based on the primitive cell, with the supercell size denoted by M . To ensure balanced cell dimensions and avoid extremely small angles, only those supercells

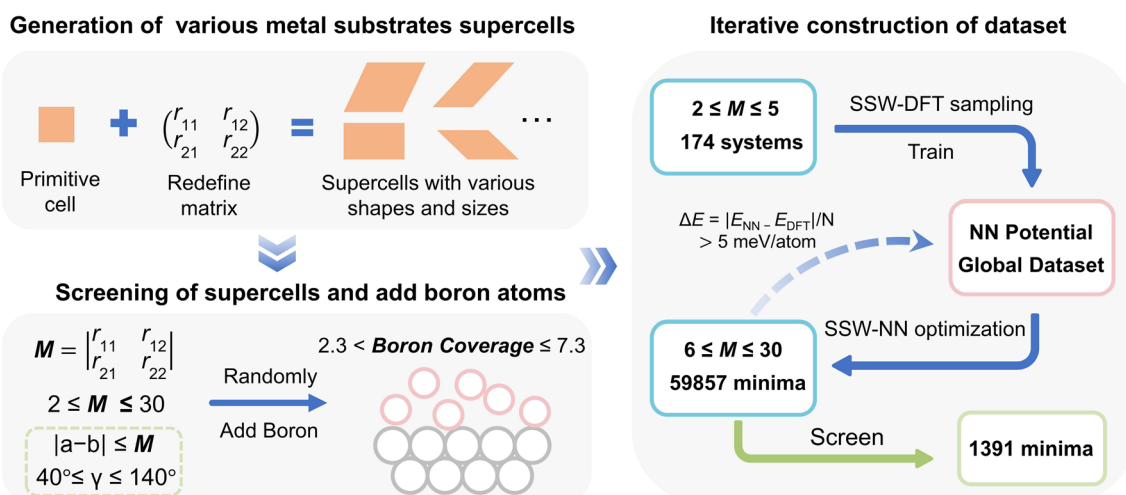


Fig. 1 Workflow for building the on-the-fly neural network (NN) potential of the borophene/Ag(100) system. M denotes the determinant of the redefined matrix and represents the multiple of the supercell relative to the primitive cell. a , b and γ are lattice parameters. The boron coverage on the Ag(100) surface is defined as the number of boron atoms/ M . During iterative refinement of the global dataset, the energy convergence criterion is $\Delta E \leq 5 \text{ meV per atom}$, in which $\Delta E = |E_{\text{NN}} - E_{\text{DFT}}|/N$. E_{NN} and E_{DFT} denote the energies calculated by the NN potential and DFT method, respectively, N is the number of total atoms. The screening of minima is conducted based on the formation energy (E_{form}).



meeting the criteria $|a - b| \leq M$ and $40^\circ \leq \gamma \leq 140^\circ$ are selected. In total, 556 supercells with M values ranging from 2 to 30 are chosen (Table S1†). Then, boron atoms are deposited onto these supercells at coverage (BC) ranging from 2.3 to 7.3, as detailed in Fig. S1.†

In the second step, we iteratively build a global dataset that captures the diverse bonding environments encountered in borophene on Ag(100). We initiate this by performing stochastic surface walking (SSW) pathway sampling on 174 smaller systems ($2 \leq M \leq 5$) containing up to 47 atoms, each starting from random configurations. This comprehensive sampling generates approximately 7 million raw structure data points, from which around 120 000 are selected to form the initial global dataset (Fig. S2†).

The initial global dataset is used to train the NN potential. Once trained, the NN potential is applied to predict 59 857 larger systems ($6 \leq M \leq 30$, BC = 2.3 to 7.3). If inaccuracies are detected in predicting minima, additional structures are extracted from SSW trajectories and subjected to *ab initio* calculations. These newly calculated structures are then added to the global dataset, and the NN potential is retrained. This iterative, active learning-like approach^{27,42} ensures the robustness and reliability of the NN potential for exploring the global PES. Performance metrics of the resulting NN potential are provided in Fig. S3, S4 and ESI.†

Overview of the global dataset

The final global dataset includes 177 023 structures, with the trained NN potential achieving root-mean-square errors (RMSE) of 11.281 meV per atom for energies and $0.456 \text{ eV \AA}^{-1}$ for forces, demonstrating high accuracy (Fig. S3†). Fig. 2a presents the formation energy (E_{form}) versus Steinhardt-type order parameter (OP_2), spanning from -6.5 to -1 eV per atom and 0 to 6, respectively. The broad distribution of E_{form} and OP_2 values (Fig. S5†) highlights the dataset's extensive diversity and the NN potential's flexibility. Additional details regarding M , BC, and OP_2 are provided in Fig. S6.†

Focusing on the low-energy region ($E_{\text{form}} < -6.0 \text{ eV per atom}$), which comprises approximately 130 000 structures, we utilize BC and OP_2 as descriptors and apply dimensionality reduction techniques^{43,44} to visualize the multidimensional PES (Fig. 2b). This analysis uncovers three distinct energy funnels at BC ranges of 2.7 to 2.9, 3.4 to 3.5, and 6.3 to 6.8, alongside numerous energy basins across different BC values, indicating the presence of multiple stable structural motifs in borophene on Ag(100). An energy funnel represents regions where the lowest-energy local minimum, including the global minimum, is most likely to occur, facilitating the steepest descent pathways on the PES. In contrast, the energy basin indicates clusters of local minima.^{43,45} The first two funnels exhibit similar energy levels, while the third energy funnel is slightly deeper, suggesting that borophene configurations with varying boron coverage in the first two funnels can coexist with comparable stability and the configurations in the third funnel are inherently more stable. OP_2 values for funnels and basins primarily lie between 1 and 2 (the most stable and ordered structures), with $OP_2 > 4$ corresponding to high-energy ridges (less stable and more disordered configurations). These findings provide a conceptual map of the global PES, aiding in the identification of stable minima from highly disordered starting configurations and enhancing the predictive power for discovering new and experimentally relevant phases.

Identified local minima and low-energy configurations

Using the established PES, we identified 59 857 local minima from random initial structures, as shown in Fig. 3a. The distribution of these minima in terms of OP_2 and BC is color-coded by E_{form} . The corresponding minima exhibit mean absolute errors of $3.278 \text{ meV per atom}$ for energies and $0.299 \text{ eV \AA}^{-1}$ for forces, indicating the reliability of our NN potential.

To isolate the most stable low-energy configurations, we screened these minima by selecting, at each BC value, the structure with the lowest E_{form} across all supercell shapes and sizes. This screening yielded 1391 low-energy minima, whose OP_2 -BC distribution and E_{form} data are shown in Fig. 3b. Three

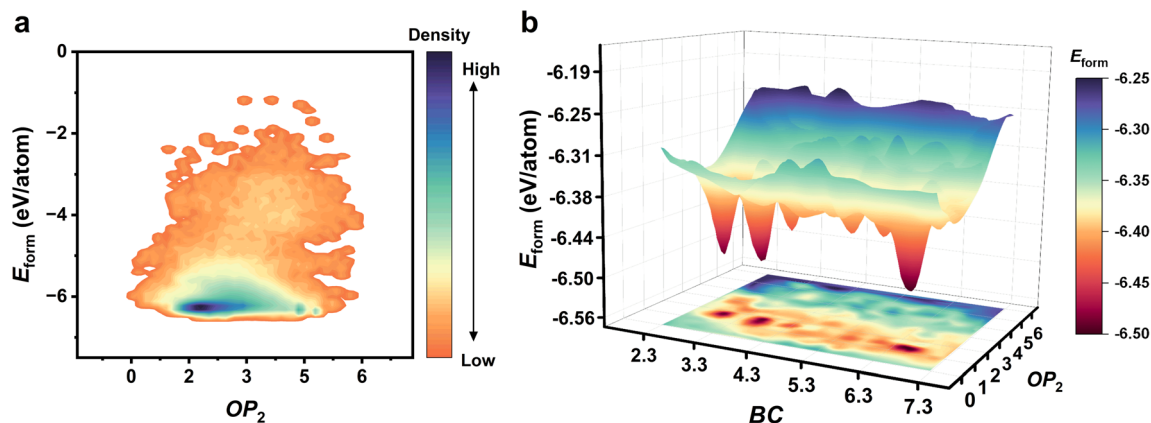


Fig. 2 (a) Illustration of the global dataset from first principles for borophene on Ag(100) systems shown as a contour map of formation energy (E_{form}) vs. OP_2 . The density is indicated by color. The OP_2 parameter reflects the averaged geometrical environment of bonded boron atoms. (b) Schematic diagram of the potential energy surface (PES) for borophene on Ag(100) systems scaled by BC and OP_2 dimensionality.



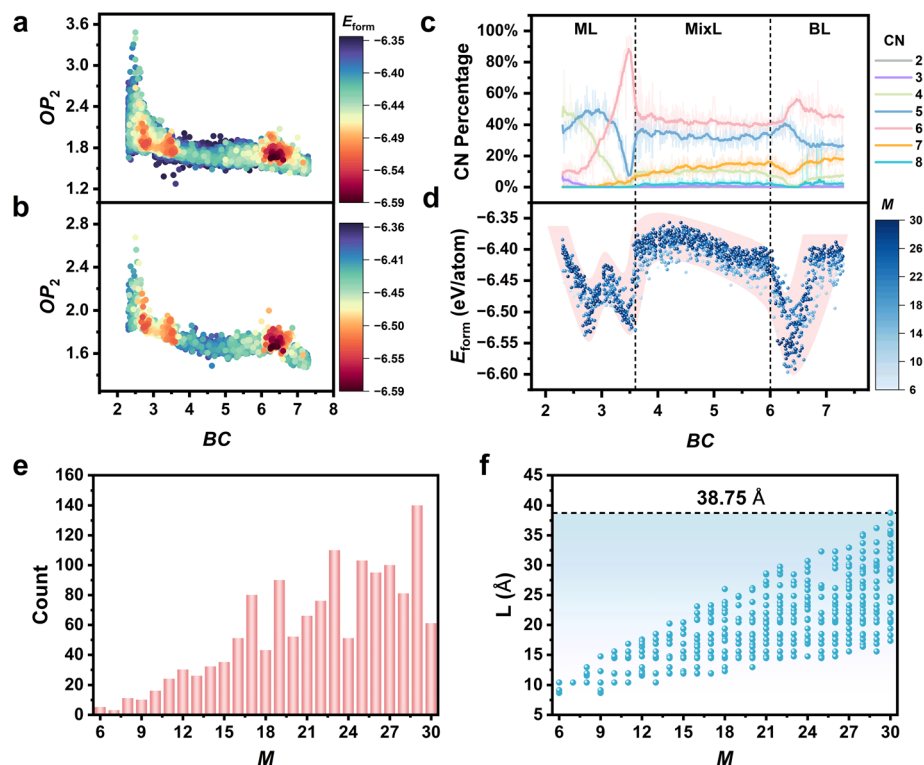


Fig. 3 The OP_2 – BC distribution of (a) minima and (b) lowest-energy minima borophenes on $Ag(100)$ colored by the formation energy (E_{form}). (c) The proportion of different coordination numbers (CN) ranging from 3 to 8 and (d) E_{form} of low-energy minima structures as a function of boron coverage (BC). In (c), the transparent colored lines represent raw data, and the solid lines are the result of applying an adjacent-averaging filter. The ML, MixL, and BL denote the monolayer, mixed-layer, and bilayer, respectively. The black dashed line located at BC of 3.6 separates the ML from the MixL, and that at BC of 6 separates the MixL from the BL. Each data point under specific BC value represents the most stable configuration. The luminance of light blue spheres denotes the magnitude of the M value, and the higher luminance corresponds to the smaller M value. The (e) statistical count and (f) the maximal lattice constant (L) of the low-energy minima under different M values ranging from 6 to 30.

prominent low-energy regions emerge at OP_2 values between 1.6 and 2.0, consistent with the three energy funnels identified in the global PES. Additionally, the overall OP_2 – BC distribution of these minima also aligns with the previously mapped low-energy landscape, confirming that our global dataset effectively captures the key atomic environments of borophene on $Ag(100)$.

To elucidate the bonding environments within these low-energy minima, we calculated the boron coordination number (CN) using a 2.3 Å boron–boron distance cutoff. Fig. 3c shows the relative percentages of each CN value as a function of BC. The presence of CN values ranging from 2 to 8 underscores the complex multi-center bonding, akin to bulk boron.²⁹ While CN = 2 and CN = 8 are relatively rare, CN = 3 briefly appears at low BC values (2.33 to 2.5) before disappearing. In contrast, CN values of 4, 5, 6, and 7 become increasingly prominent across the entire BC range, indicating evolving coordination patterns in these low-energy minima.

As BC increases from 3.6 to 6, the CN of 7 gradually rises, indicating the formation of borophene structures beyond a monolayer. Distinct shifts in CN distribution are observed around BC = 3.6 and BC = 6, as evident from both the average CN (ave_CN) and average bond length (ave_bond) (Fig. S7a and b†). Consequently, we define BC = 3.6 as the boundary between

monolayer (ML) and mixed-layer (MixL) phases, and BC = 6 as the boundary between MixL and bilayer (BL) phases. In this classification, the ML refers to a single-atomic-layer sheet, MixL denotes an ML with attached boron nanostructures, and the BL signifies two covalently bonded MLs, following our previous work.⁴⁶ Although some structures may not conform strictly to these categories, this classification provides a useful framework for understanding the structural evolution with increasing BC.

Fig. 3d correlates these structural transitions with stability by plotting E_{form} versus BC. Three distinct energy valleys emerge, corresponding to the three PES funnels. Energy valley 1 (BC ≈ 2.7 to 2.85) and energy valley 2 (BC ≈ 3.45 to 3.6) correspond to ML configurations, while energy valley 3 (BC ≈ 6.3 to 6.45) pertains to BL configurations. The first two valleys exhibit comparable stability, whereas the third valley is approximately 57 meV per atom deeper, reflecting the enhanced structural complexity and stronger bonding networks in bilayers. Within energy valley 1, CN values of 4, 5, and 6 appear in a 25 : 50 : 25 ratio, while energy valley 2's minima exclusively feature CN = 6. Energy valley 3 displays a mix of CN = 5, 6, and 7 (40%, 52%, and 8%, respectively), resulting in a higher average CN (5.5 to 5.75) and increased stability.

The influence of supercell size M is also evident. As shown in Fig. 3e, the vast majority (88.7%) of low-energy minima are



found in supercells with M values between 15 and 30. Indeed, most structures with $E_{\text{form}} < -6.50$ eV per atom have $M > 15$ (Fig. S7c†). Fig. 3f further demonstrates that the maximal lattice constants (L) of these low-energy minima range from 8.667 Å up to 38.75 Å, underscoring the critical importance of large supercells in accurately capturing the global minima of borophene on Ag(100). These results highlight the necessity of the extended-scale supercells to reflect the inherent structural complexity of borophene phases.

Structural analysis of energy valleys

To gain deeper insight into the structural motifs within each of three energy valleys, we selected ten lowest-energy configurations from each valley. The corresponding E_{form} -BC plots are shown in Fig. 4a to c, and detailed information about these structures is listed in Tables S3 to S5.† For convenience, we label the structures in energy valleys 1, 2, and 3 as ML1 to ML10, ML11 to ML20, and BL1 to BL10, respectively, based on their

E_{form} values. Within these sets, the top ten structures in valleys 1, 2, and 3 lie within energy windows of 19.4, 11.3, and 19.7 meV per atom, respectively, and their M value ranges from 19 to 30, 16 to 29, and 16 to 27.

To unveil the atomic patterns in different energy valleys, several representative structures are displayed in Fig. 4d to k, with additional structures shown in Fig. S8 to S10.† The names, BC, E_{form} , interfacial charge transfer (C), M values, and structural descriptions of these selected structures are summarized in Table 1 (see more details in Tables S3 to S5†). Structures in the energy valleys 1 and 3 exhibit mixed phases. In energy valley 1, monolayers (MLs) feature combinations of different borophene ribbons, such as χ_3 , β_{12} , and so on (Fig. 4d–g and S8†). In energy valley 3, bilayers (BLs) are characterized by mixtures of various borophene phases, including BL- α_5 , BL- α_7 , BL- α_1 , BL- α_6 , and BL- $\alpha_1\alpha_6$ (Fig. 4i–k and S10†). Note that BL- α_5 denotes the bilayer covalently stacked by two α_5 monolayers. Interestingly, energy valley 2 shows a high degree of structural degeneracy, with ML10 to ML20 all being buckled triangular lattices (δ_6)

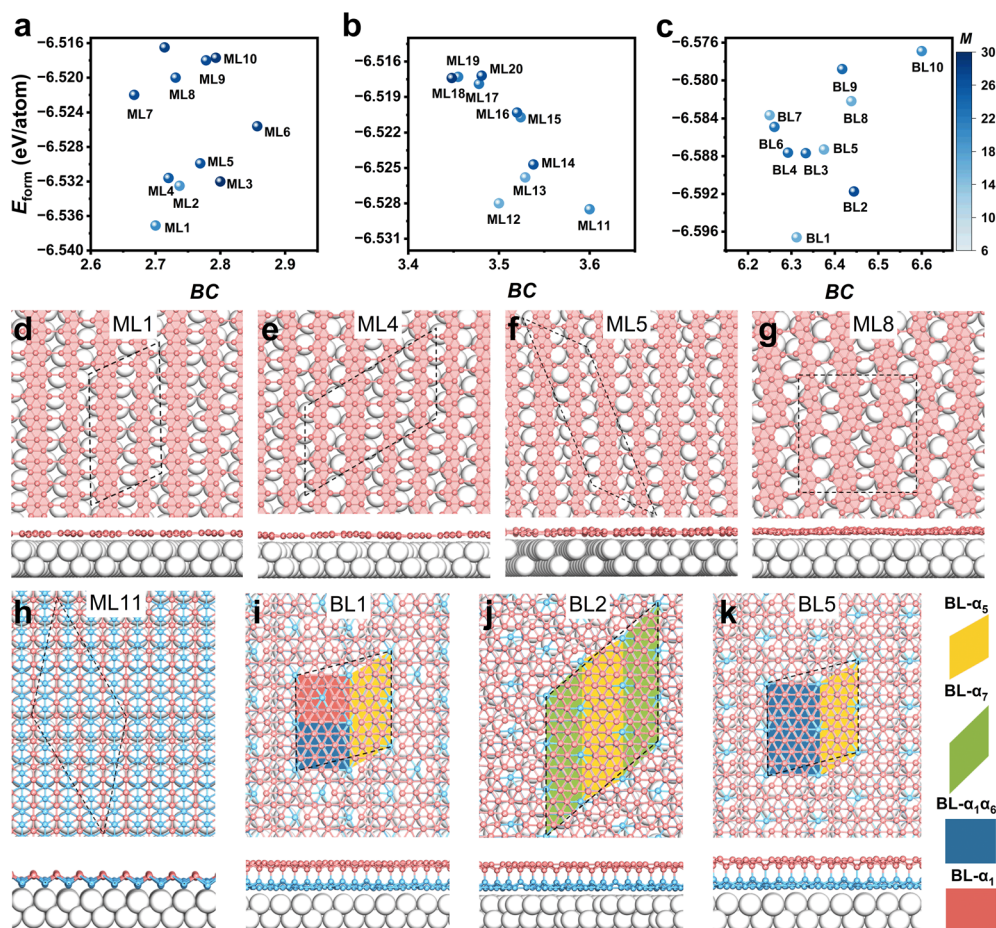


Fig. 4 The formation energy (E_{form}) varying with boron coverage (BC) for the top ten most stable configurations from energy valleys (a) 1, (b) 2, and (c) 3. The structures from energy valleys 1, 2, and 3 are numbered by ML1–ML10, ML11–ML20, and BL1–BL10, respectively. The luminance of light blue spheres denotes the magnitude of the M value, and the higher luminance corresponds to the smaller M value. (d)–(k) The atomic patterns of several representative structures (ML1, ML4, ML5, ML8, ML11, BL1, BL2, and BL5) of (a)–(c), in which the black dashed lines denote the unit cell. For ML1, ML4, ML5, and ML8, only top view is provided. For ML11, BL1, BL2, and BL5, both top and side views are provided. The light gray spheres denote Ag atoms, and the pink colors denote boron atoms in which the lower boron atoms of ML11 and the bottom sheet of BL structures are colored by light cyan. Different bilayer borophene fragments are depicted by different shapes with different colors.



Table 1 The name, boron coverage (BC), formation energy (E_{form}), and structural components of the unit cell for representative structures from three energy valleys. The structures denote the type and corresponding number of different borophene phases included in the unit cell of mixed-phase borophenes. C is the averaged number of electrons transferred from Ag to B

Name	BC	E_{form} (eV per atom)	Structures	C (e per atom)
ML1	2.700	-6.5371	$1\chi_3 1\beta_{12}$	0.0265
ML4	2.720	-6.5316	$3\chi_3 1\beta_{12}$	0.0274
ML5	2.769	-6.5299	$1\chi_3 1\beta_{12}$	0.0290
ML8	2.731	-6.5200	$1\chi_3 2\beta_{12}$ -dislocation	0.0268
ML11	3.600	-6.5285	δ_6	0.0108
BL1	6.312	-6.5966	$2\text{BL-}\alpha_5 1\text{BL-}\alpha_1 1\text{BL-}\alpha_1 \alpha_6$	0.0103
BL2	6.444	-6.5917	$3\text{BL-}\alpha_5 3\text{BL-}\alpha_7$	0.0095
BL5	6.375	-6.5870	$2\text{BL-}\alpha_5 2\text{BL-}\alpha_1 \alpha_6$	0.0110

(Fig. 4h and S9†). The simulated STM images of all these structures are provided in Fig. S11 and S12,† Fig. 5a–h. The nomenclature of all borophene structures follows the rules proposed in previous work.¹ The structures of MLs in this work, including new structures (α_6 and α_7) and borophene ribbons (χ_3 , β_{12} , and R1), are shown in Fig. S13.† Fig. S14† provides detailed atomic patterns of BL- α_5 , BL- α_7 , BL- α_1 , BL- α_6 and BL- $\alpha_1 \alpha_6$. Here, BL- α_i or BL- $\alpha_i \alpha_j$ denotes the bilayer covalently stacked by two α_i monolayers or α_i and α_j monolayers, respectively.

In energy valley 1, MLs feature combinations of different borophene ribbons, such as χ_3 and β_{12} with varying ratios

(Fig. 4d–g, S8 and Table S3†). ML1 comprises one χ_3 and one β_{12} ribbon in the unit cell (Fig. 4d), and its simulated STM image is shown in Fig. 5a. It is more stable than ML2 (Fig. S8a†) and ML3 (Fig. S8b†) by 4.6 and 5.1 meV per atom, respectively (Table 1). ML2 and ML3 comprise two χ_3 and one β_{12} ribbon and one χ_3 and two β_{12} ribbons, respectively, which closely match experimental phases C and A based on simulated STM images (Fig. S11a and b†).¹² Additionally, the simulated STM images of the pure β_{12} sheet (ML6, Fig. S8c†) are consistent with phase B (Fig. S11c†). Minor discrepancies, such as the position of a less bright stripe in the simulated STM images, may arise from undulation fluctuations due to substrate interactions. ML4 (Fig. 4e), composed of three χ_3 and one β_{12} ribbon, and ML5 (Fig. 4f), composed of one χ_3 and one β_{12} , possess comparable stability to ML2 and ML3. Note that ML5 shares the same atomic pattern as ML1, but ML1 has a lower lattice mismatch, enhancing its stability (Table S3†). The pure β_{12} (ML6, Fig. S8c†) and χ_3 (ML7, Fig. S8d†) phases, which have been experimentally synthesized, are less stable than the mixed ribbons (ML1 to ML5) according to their E_{form} values. Therefore, the newly identified ML1 and ML4 are thermodynamically feasible under experimental conditions. Additionally, structures featuring ribbon dislocation (ML8, Fig. 4g) and hybridized χ_3 - β_{12} ribbons (ML10, Fig. S8f†) have been predicted. These structural defects are inevitable in as-grown borophene,¹¹ and can only be captured in large supercells.

In energy valley 2, all structures from ML11 to ML20 adopted a buckled δ_6 lattice (Fig. 4h and S9†), despite being observed in

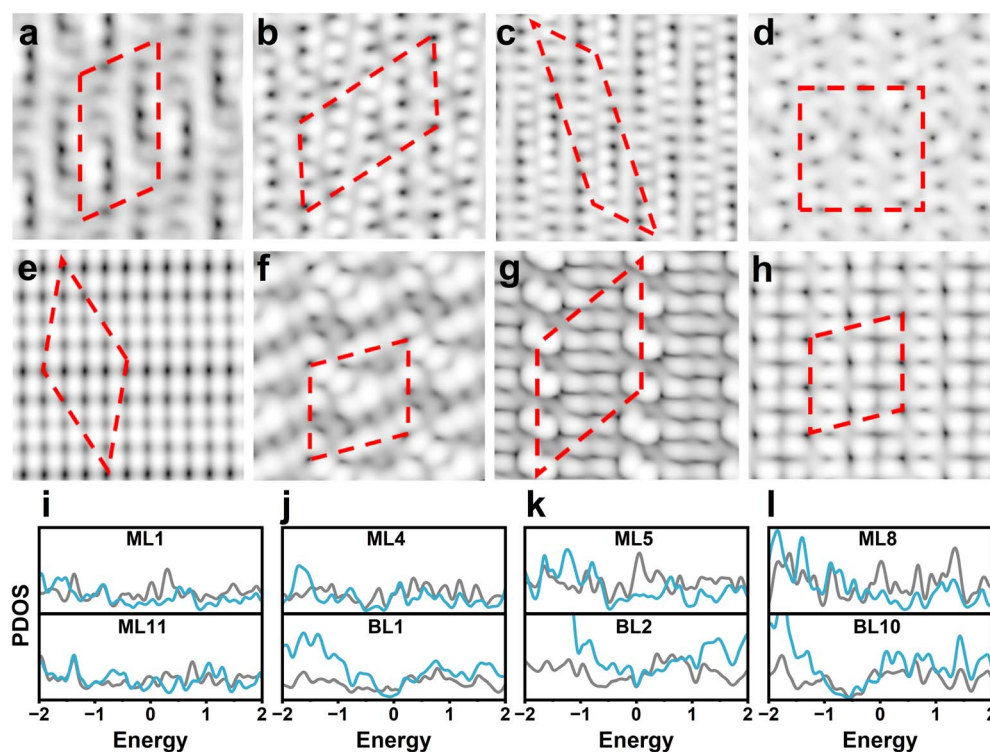


Fig. 5 Simulated images of (a) ML1, (b) ML4, (c) ML5, (d) ML8, (e) ML11, (f) BL1, (g) BL2, and (h) BL5 on Ag(100). The red dashed lines denote the primitive cell. The (i)–(l) corresponding projected density of states, in which the gray and cyan lines denote Ag and B atoms, respectively. The Fermi energy level is set as 0 eV.



different supercells (Table S4†). The simulated STM image of ML11 is provided in Fig. 5e. In energy valley 3, BLs are characterized by mixtures of various borophene phases, including BL- α_5 , BL- α_7 , BL- α_1 , BL- α_6 , and BL- $\alpha_1\alpha_6$ (Fig. 4i–k and S10†). BL1 (Fig. 4i), BL3, and BL4 are composed of varying proportions of BL- α_5 , BL- $\alpha_1\alpha_6$, and BL- α_1 . BL2 (Fig. 4j), composed of BL- α_5 and BL- α_7 , is more stable than the single-phase BL- α_7 (BL6) and BL- α_5 (BL10) by 3.8 and 19.7 meV per atom, respectively. This indicates a preference for mixed-phase bilayer borophenes on Ag(100). Additionally, BL5 (Fig. 4k) consists of BL- α_5 and BL- $\alpha_1\alpha_6$, while BL7 comprises BL- α_5 and BL- α_1 . Similarly, BL8 and BL9 are mixtures of BL- α_5 , BL- $\alpha_1\alpha_6$, and BL- α_6 in different proportions. The simulated STM images of BL1, BL2, and BL5 are provided in Fig. 5f–h, and those of the other BLs are shown in Fig. S12.† Notably, the pattern of BL10 matches well with the experimental CO-AFM images of bilayer borophene on Ag(111) in the literature.⁴⁷ All aforementioned BLs are stacked in an AB pattern, while AA-stacking variants exhibit lower stability (Fig. S15†) and their corresponding atomic arrangements are displayed in Fig. S16.†

Electronic properties of representative structures

We conducted projected density of states (PDOS) calculations for representative structures (ML1, ML4, ML5, ML8, ML11, BL1, BL2, and BL10) on the Ag(100) substrates, as shown in Fig. 5f–i. Detailed PDOS information, including those for free-standing borophene, is presented in Fig. S17 and S18.† These calculations reveal that all analyzed structures exhibit metallicity when supported on Ag(100). Notably, there is a significant increase in the density of states (DOS) near the Fermi level for ML1, ML4, ML5, and ML8, while ML11, BL1, BL2, and BL10 show a decrease in DOS in this region. These trends can be directly compared with the future experimental dI/dV spectra to validate our computational predictions.

Furthermore, all examined structures, except for BL- α_5 , retain their metallicity in free-standing forms and demonstrate good conductivity. This suggests that the mixed-phase borophenes not only maintain their electronic properties when supported on metal substrates but also exhibit promising conductivity characteristics in isolation, enhancing their potential applications in nanoelectronics.

Conclusions

In summary, we established a machine learning-driven framework to resolve the complex potential energy landscape of borophene on Ag(100), revealing unprecedented structural diversity at the nanometer scale. By integrating active learning with SSW global optimization, we systematically identified 1391 low-energy configurations across three distinct energy funnels, including mixed-phase monolayers (χ_3/β_{12} ribbons) and bilayers (BL- α_5 /BL- $\alpha_1\alpha_6$ /BL- α_1 assemblies). These findings not only reconcile experimental observations of borophene polymorphism but also unveil previously unexplored metastable phases stabilized by substrate interactions. The methodology developed here, combining scalable neural network potentials

with adaptive sampling, transcends borophene systems, offering a generalized strategy for exploring polymorphism in substrate-supported 2D materials (*e.g.*, silicene, phosphorene). Furthermore, the metallic nature of these mixed phases and their tunable electronic properties suggest promising avenues for nanoelectronic applications.

Data availability

Data for this article, including the NN potential file and the top ten most stable structures in the three energy valleys are available at GitHub at <https://github.com/yunleiw/Mixed-Phases-of-Borophene-on-Ag-100-git>.

Author contributions

Y. W., H. L., X. W. designed the study. Y. W. performed the theoretical simulations. Y. W., H. L., X. W. analyzed the data and wrote the manuscript.

Conflicts of interest

The authors declare that they have no conflict of interest.

Acknowledgements

This work is supported by the National Natural Science Foundation for Distinguished Young Scholars (grant no. 22225301), the National Natural Science Foundation of China (grant no. 22303092 and 22321001), the Strategic Priority Research Program of the Chinese Academy of Sciences (grant no. XDB0450101), the Fundamental Research Funds for the Central Universities (20720220009 and WK2490000001), and Super Computer Center of USTCSCC and SCCAS.

References

- X. Wu, J. Dai, Y. Zhao, Z. Zhuo, J. Yang and X. C. Zeng, Two-Dimensional Boron Monolayer Sheets, *ACS Nano*, 2012, **6**, 7443–7453.
- E. S. Penev, S. Bhowmick, A. Sadzadeh and B. I. Yakobson, Polymorphism of two-dimensional boron, *Nano Lett.*, 2012, **12**, 2441–2445.
- B. Mortazavi, O. Rahaman, A. Dianat and T. Rabczuk, Mechanical responses of borophene sheets: a first-principles study, *Phys. Chem. Chem. Phys.*, 2016, **18**, 27405–27413.
- E. S. Penev, A. Kutana and B. I. Yakobson, Can Two-Dimensional Boron Superconduct?, *Nano Lett.*, 2016, **16**, 2522–2526.
- X.-F. Zhou, A. R. Oganov, Z. Wang, I. A. Popov, A. I. Boldyrev and H.-T. Wang, Two-dimensional magnetic boron, *Phys. Rev. B*, 2016, **93**, 085406.
- H. Tang and S. Ismail-Beigi, Novel Precursors for Boron Nanotubes: The Competition of Two-Center and Three-Center Bonding in Boron Sheets, *Phys. Rev. Lett.*, 2007, **99**, 115501.



- 7 T. R. Galeev, Q. Chen, J.-C. Guo, H. Bai, C.-Q. Miao, H.-G. Lu, A. P. Sergeeva, S.-D. Li and A. I. Boldyrev, Deciphering the mystery of hexagon holes in an all-boron graphene α -sheet, *Phys. Chem. Chem. Phys.*, 2011, **13**, 11575–11578.
- 8 Z. Zhang, Y. Yang, G. Gao and B. I. Yakobson, Two-Dimensional Boron Monolayers Mediated by Metal Substrates, *Angew. Chem., Int. Ed.*, 2015, **54**, 13022–13026.
- 9 N. Karmodak and E. D. Jemmis, The Role of Holes in Borophenes: An Ab Initio Study of Their Structure and Stability with and without Metal Templates, *Angew. Chem., Int. Ed.*, 2017, **56**, 10093–10097.
- 10 B. Feng, J. Zhang, Q. Zhong, W. Li, S. Li, H. Li, P. Cheng, S. Meng, L. Chen and K. Wu, Experimental realization of two-dimensional boron sheets, *Nat. Chem.*, 2016, **8**, 563–568.
- 11 X. Liu, Z. Zhang, L. Wang, B. I. Yakobson and M. C. Hersam, Intermixing and periodic self-assembly of borophene line defects, *Nat. Mater.*, 2018, **17**, 783–788.
- 12 Y. Wang, L. J. Kong, C. Y. Chen, P. Cheng, B. J. Feng, K. H. Wu and L. Chen, Realization of Regular-Mixed Quasi-1D Borophene Chains with Long-Range Order, *Adv. Mater.*, 2020, **32**, 2005128.
- 13 R. T. Wu, S. Eltinge, I. K. Drozdov, A. Gozar, P. Zahl, J. T. Sadowski, S. Ismail-Beigi and I. Bozovic, Micrometre-scale single-crystalline borophene on a square-lattice Cu(100) surface, *Nat. Chem.*, 2022, **14**, 377–383.
- 14 L. R. Liu, Z. H. Zhang, X. L. Liu, X. Y. Xuan, B. I. Yakobson, M. C. Hersam and W. L. Guo, Borophene Concentric Superlattices via Self-Assembly of Twin Boundaries, *Nano Lett.*, 2020, **20**, 1315–1321.
- 15 R. Wu, I. K. Drozdov, S. Eltinge, P. Zahl, S. Ismail-Beigi, I. Bozovic and A. Gozar, Large-area single-crystal sheets of borophene on Cu(111) surfaces, *Nat. Nanotechnol.*, 2019, **14**, 44–49.
- 16 P. Sutter and E. Sutter, Large-Scale Layer-by-Layer Synthesis of Borophene on Ru(0001), *Chem. Mater.*, 2021, **33**, 8838–8843.
- 17 C. Y. Chen, H. F. Lv, P. Zhang, Z. W. Zhuo, Y. Wang, C. Ma, W. B. Li, X. G. Wang, B. J. Feng, P. Cheng, X. J. Wu, K. H. Wu and L. Chen, Synthesis of bilayer borophene, *Nat. Chem.*, 2022, **14**, 25–31.
- 18 J. Wang, H. Gao, Y. Han, C. Ding, S. Pan, Y. Wang, Q. Jia, H.-T. Wang, D. Xing and J. Sun, MAGUS: machine learning and graph theory assisted universal structure searcher, *Natl. Sci. Rev.*, 2023, **10**, nwad128.
- 19 A. R. Oganov and C. W. Glass, Crystal structure prediction using ab initio evolutionary techniques: principles and applications, *J. Chem. Phys.*, 2006, **124**, 244704.
- 20 S. Kang, W. Jeong, C. Hong, S. Hwang, Y. Yoon and S. Han, Accelerated identification of equilibrium structures of multicomponent inorganic crystals using machine learning potentials, *npj Comput. Mater.*, 2022, **8**, 108.
- 21 X. Shi, D. Cheng, R. Zhao, G. Zhang, S. Wu, S. Zhen, Z.-J. Zhao and J. Gong, Accessing complex reconstructed material structures with hybrid global optimization accelerated via on-the-fly machine learning, *Chem. Sci.*, 2023, **14**, 8777–8784.
- 22 Y. Shen, S. I. Morozov, K. Luo, Q. An and W. A. Goddard Iii, Deciphering the Atomistic Mechanism of Si(111)- 7×7 Surface Reconstruction Using a Machine-Learning Force Field, *J. Am. Chem. Soc.*, 2023, **145**, 20511–20520.
- 23 C. Schran, F. L. Thiemann, P. Rowe, E. A. Müller, O. Marsalek and A. Michaelides, Machine learning potentials for complex aqueous systems made simple, *Proc. Natl. Acad. Sci. U. S. A.*, 2021, **118**, e2110077118.
- 24 P. Mignon, A. R. Allouche, N. R. Innis and C. Bousige, Neural Network Approach for a Rapid Prediction of Metal-Supported Borophene Properties, *J. Am. Chem. Soc.*, 2023, **145**, 27857–27866.
- 25 L. H. Luo, S. D. Huang, C. Shang and Z. P. Liu, Resolving Activation Entropy of CO Oxidation under the Solid-Gas and Solid-Liquid Conditions from Machine Learning Simulation, *ACS Catal.*, 2022, **12**, 6265–6275.
- 26 D. Chen, L. Chen, Q.-C. Zhao, Z.-X. Yang, C. Shang and Z.-P. Liu, Square-pyramidal subsurface oxygen [Ag₄OAg] drives selective ethene epoxidation on silver, *Nat. Catal.*, 2024, **7**, 536–545.
- 27 Y. Han, J. Xu, W. Xie, Z. Wang and P. Hu, Comprehensive Study of Oxygen Vacancies on the Catalytic Performance of ZnO for CO/H₂ Activation Using Machine Learning-Accelerated First-Principles Simulations, *ACS Catal.*, 2023, **13**, 5104–5113.
- 28 V. L. Deringer, C. J. Pickard and G. Csányi, Data-Driven Learning of Total and Local Energies in Elemental Boron, *Phys. Rev. Lett.*, 2018, **120**, 156001.
- 29 S. D. Huang, C. Shang, P. L. Kang and Z. P. Liu, Atomic structure of boron resolved using machine learning and global sampling, *Chem. Sci.*, 2018, **9**, 8644–8655.
- 30 E. V. Podryabinkin, E. V. Tikhonov, A. V. Shapeev and A. R. Oganov, Accelerating crystal structure prediction by machine-learning interatomic potentials with active learning, *Phys. Rev. B*, 2019, **99**, 064114.
- 31 Q. C. Tong, L. T. Xue, J. Lv, Y. C. Wang and Y. M. Ma, Accelerating CALYPSO structure prediction by data-driven learning of a potential energy surface, *Faraday Discuss.*, 2018, **211**, 31–43.
- 32 G. Kresse and J. Hafner, Ab-Initio Molecular-Dynamics Simulation of the Liquid-Metal Amorphous-Semiconductor Transition in Germanium, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **49**, 14251–14269.
- 33 G. Kresse and J. Furthmüller, Efficiency of ab-initio total energy calculations for metals and semiconductors using a plane-wave basis set, *Comput. Mater. Sci.*, 1996, **6**, 15–50.
- 34 P. E. Blochl, Projector Augmented-Wave Method, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953–17979.
- 35 P. E. Blochl, P. Margl and K. Schwarz, Ab initio molecular dynamics with the projector augmented wave method, *ACS Symp. Ser.*, 1996, **629**, 54–69.
- 36 J. P. Perdew, K. Burke and M. Ernzerhof, Generalized gradient approximation made simple, *Phys. Rev. Lett.*, 1996, **77**, 3865–3868.
- 37 F. Li, P. Jin, D.-e. Jiang, L. Wang, S. B. Zhang, J. Zhao and Z. Chen, B₈₀ and B₁₀₁₋₁₀₃ clusters: remarkable stability of



- the core-shell structures established by validated density functionals, *J. Chem. Phys.*, 2012, **136**, 074302.
- 38 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, A consistent and accurate ab initio parametrization of density functional dispersion correction (DFT-D) for the 94 elements H-Pu, *J. Chem. Phys.*, 2010, **132**, 154104.
- 39 C. Shang and Z. P. Liu, Stochastic Surface Walking Method for Structure Prediction and Pathway Searching, *J. Chem. Theory Comput.*, 2013, **9**, 1838–1845.
- 40 C. Shang, X. J. Zhang and Z. P. Liu, Stochastic surface walking method for crystal structure and phase transition pathway prediction, *Phys. Chem. Chem. Phys.*, 2014, **16**, 17845–17856.
- 41 S.-D. Huang, C. Shang, P.-L. Kang, X.-J. Zhang and Z.-P. Liu, LASP: fast global potential energy surface exploration, *Wiley Interdiscip. Rev.: Comput. Mol. Sci.*, 2019, **9**, e1415.
- 42 H. W. Jung, L. Sauerland, S. Stocker, K. Reuter and J. T. Margraf, Machine-learning driven global optimization of surface adsorbate geometries, *npj Comput. Mater.*, 2023, **9**, 114.
- 43 A. R. Oganov and M. Valle, How to quantify energy landscapes of solids, *J. Chem. Phys.*, 2009, **130**, 104504.
- 44 A. R. Oganov, A. O. Lyakhov and M. Valle, How Evolutionary Crystal Structure Prediction Works-and Why, *Acc. Chem. Res.*, 2011, **44**, 227–237.
- 45 A. R. Oganov, C. J. Pickard, Q. Zhu and R. J. Needs, Structure prediction drives materials discovery, *Nat. Rev. Mater.*, 2019, **4**, 331–348.
- 46 Y. Wang, H. Lv, X. Wu and J. Yang, First-Principles Particle-Swarm Structure Calculations on Metal Substrate Modulated Monolayer to Bilayer Borophene Crystalline Structures toward Nanoelectronics, *ACS Appl. Nano Mater.*, 2023, **6**, 15871–15878.
- 47 X. L. Liu, Q. C. Li, Q. Y. Ruan, M. S. Rahn, B. I. Yakobson and M. C. Hersam, Borophene synthesis beyond the single-atomic-layer limit, *Nat. Mater.*, 2022, **21**, 35–40.

