


Cite this: *RSC Adv.*, 2025, 15, 23115

Zigzag antiferromagnetic property of two-dimensional NiPX_3 ($\text{X} = \text{S/Se}$) monolayers in their pristine structure and Janus phase†

Juntao Yang,^{id}*^{ab} XuLi Zhang,^a Zecheng Qu,^a Qing Min,^a Jian Xia,^{id}^a Hao Shang,^a Shijun Luo^a and Chengrui Wu^{*a}

Transition metal phosphorous trichalcogenides TMPX_3 ($\text{TM} = \text{transition metal}$, $\text{X} = \text{S/Se}$) with a variety of spin configurations serve as excellent platforms for studying the magnetic properties of two-dimensional systems. As an antiferromagnetic semiconductor, the coexistence of ferromagnetic and antiferromagnetic couplings between the nearest neighboring metal ions of NiPX_3 remains a debatable topic. In this work, the electronic structures and magnetic properties of NiPX_3 monolayers in their pristine structure and Janus phase were systemically investigated using first-principles calculations. It was found that the NiPX_3 system possessed an indirect band gap in the zigzag antiferromagnetic ground state with a sizable Néel temperature, as estimated by Monte Carlo simulations. Electronic structures and crystal orbital Hamilton population analyses revealed that the zigzag antiferromagnetic ordering was primarily driven by superexchange interactions through p–d hybridization. Meanwhile, the coexistence of ferromagnetic and antiferromagnetic couplings was facilitated through a combination of antibonding and bonding states below the Fermi level. This work provides a new approach to explore the diverse and intriguing magnetic properties of two-dimensional materials.

Received 24th April 2025

Accepted 18th June 2025

DOI: 10.1039/d5ra02861b

rsc.li/rsc-advances

1. Introduction

Inspired by the pioneer discovery of an intrinsic magnetic order observed in FePS_3 monolayer,^{1,2} transition metal phosphorus trichalcogenides TMPX_3 ($\text{TM} = \text{transition metals such as Cr, Mn, Fe, Co, and Ni}$; $\text{X} = \text{S/Se}$) have served as excellent platforms to explore two dimensional (2D) magnetism in recent years.³ Most TMPX_3 crystals exhibit an antiferromagnetic (AFM) behavior, with a Néel temperature (T_N) ranging from 82 K to 155 K (ref. 4 and 5); however, some of these crystals, such as MgPS_3 and ZnPS_3 (ref. 6), exhibit a paramagnetic property owing to the presence of unfilled or fully filled d electrons in the metal ions. Bulk TMPX_3 materials always exhibit a stacking sequence of “ABAB” in $C2/m$ symmetry or “ABC” in R_3 space group and are crystallized by van der Waals (vdW) forces with vdW distances ranging from 3.22 to 3.24 Å (ref. 7); therefore, TMPX_3 monolayer can be mechanically exfoliated from the bulk phase owing to the weak cleavage energy, which is smaller than 0.36 J m^{-2} for

graphite.⁸ The TMPX_3 monolayer is characterized by a hexagonal honeycomb-like lattice framework formed by TM atoms. Each TM atom is coordinated with six S/Se atoms, which constitute the upper and lower chalcogen layers. These layers are interconnected *via* P–P pairs. Specifically, the S/Se atoms are sandwiched between P–P pairs both above and below the honeycomb lattice plane. The P–P dimer vertically bisects the center of the hexagonal unit cell, forming a bipyramid-structure of $[\text{PX}_3^-]-[\text{PX}_3]$.

While 2D TMPX_3 monolayers typically demonstrate an AFM behavior, the diverse electronic configurations of TM atoms result in diverse magnetic properties and electronic structures. MnPX_3 monolayers exhibit a Néel AFM (nAFM, Fig. 1(b)) configuration as direct band gap semiconductors, while FePX_3 monolayers usually exhibit a zigzag AFM (zAFM, Fig. 1(b)) ordering as indirect band gap semiconductors; moreover, CuPX_3 system displays a paramagnetic metal character.^{9,10} In particular, multiple Dirac cones are predicated in paramagnetic NiPX_3 , PdPX_3 and PtPX_3 (ref. 11) besides ferromagnetic (FM) CrPSe_3 monolayers.¹² Some strong electron correlations have been experimentally detected, such as the Mott–Hubbard insulation in the NiPX_3 (ref. 13) system and observation of a negative charge transfer in the NiPS_3 monolayer.¹⁴ Magnetic anisotropy energy (MAE) of TMPX_3 monolayers also strongly affects their spin dimensions in Ising, XY and Heisenberg models, which have been observed in FePS_3 , MnPS_3 and CoPS_3 monolayers,^{1,15,16} respectively. Currently, 2D TMPX_3 systems

^aShiyan Key Laboratory of Quantum Information and Precision Optics, School of Optoelectronic Engineering, and Collaborative Innovation Center for Optoelectronic Technology, Hubei University of Automotive Technology (HUAT), 167 Checheng West Road, Shiyan City, Hubei, People's Republic of China. E-mail: jtyang@huat.edu.cn; crwu@huat.edu.cn

^bHubei Key Laboratory of Optical Information and Pattern Recognition, Wuhan Institute of Technology (WIT), Wuhan, 430205, China

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d5ra02861b>



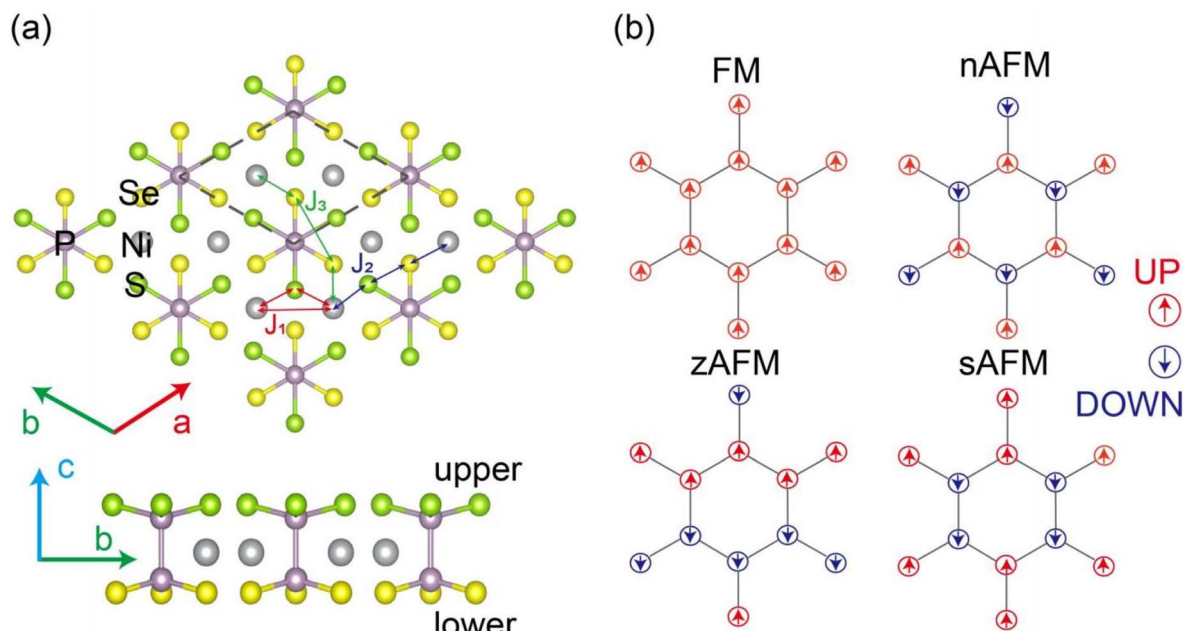


Fig. 1 (a) Crystal structure of Janus $\text{NiPS}_{3/2}\text{Se}_{3/2}$ from top and side views; (a)–(c) indicate the lattice vectors along [100], [010] and [001] directions, respectively. The dashed rhombus marks a primitive unit cell. (b) Four magnetic configurations in 2D hexagonal lattice; spin-up (UP) and spin-down (DOWN) magnetic moments are indicated by red and blue arrows, respectively.

have attracted considerable attention owing to their fascinating phenomena arising from strong coupling interactions between magnetism and other physical properties. Superconductivity can be driven into some 2D TMPX_3 , such as FePS_3 , MnPS_3 and NiPS_3 , via a pressure-driven spin-crossover.^{17–19} A giant optical linear dichroism has been observed in FePS_3 and NiPS_3 monolayers using magneto-optical spectroscopy,²⁰ while a second harmonic generation can be produced via polarization-resolved electric quadruple contribution in MnPS_3 monolayer.²¹ Furthermore, strong electron–phonon coupling in FePS_3 (ref. 22) and magnon–phonon coupling²³ and photon–matter coupling²⁴ in NiPS_3 have been directly determined using magneto-Raman spectroscopy, ultrafast spectroscopy and spectroscopic analysis in conjunction with a microscopic theory, respectively.

Despite extensive research into their unique magnetic properties, diverse electronic structures, and intriguing electromagnetic couplings of 2D TMPX_3 systems, the origin of the zAFM order in some TMPX_3 monolayers remains a subject of ongoing debate. Very recently, the magnetic ground states of NiPS_3 and FePS_3 monolayers have been demonstrated to exhibit a distinct zigzag antiferromagnetic (zAFM) ordering. Specifically, the zAFM state in NiPS_3 is primarily determined by the nearest-neighbor (NN) biquadratic exchange interaction, while in FePS_3 , it results from the competition between ferromagnetic (FM) NN and antiferromagnetic (AFM) third-nearest-neighbor (3NN) exchange interactions.²⁵ A small distortion between TM–TM distance has been predicted to result in an zAFM alignment in FePS_3 .^{9,26} To date, the reason for the co-existence of FM and AFM spin orderings between the NN TM atoms in zAFM systems is unclear. Hence a detailed investigation on the electronic

structures is imperative to reveal the mechanism of zAFM spin-ordering, where the adjacent TM chains along the zigzag direction exhibit an opposite orientation in 2D TMPX_3 .

Since chalcogen-substituted $\text{NiPS}_{3-x}\text{Se}_x$ ($0 \leq x \leq 1.3$) crystal has been successfully synthesized recently,²⁷ in this work, the magnetic properties of 2D NiPX_3 in its pristine structure in conjunction with its Janus phase of $\text{NiPS}_{3/2}\text{Se}_{3/2}$ were systematically investigated using first-principles calculations and Monte Carlo (MC) simulations, including the magnetic exchange parameters (J), MAE and Néel temperature (T_N). In addition, the electronic structures combined with crystal orbital Hamilton populations (COHP) and crystal-field splitting were discussed in detail to reveal the magnetic mechanism of zAFM spin ordering.

2 Computational methods

Based on density functional theory (DFT),²⁸ our spin-polarized first-principles calculations were carried out using the Vienna *Ab initio* simulation package (VASP).²⁹ The ion–electron interactions were characterized using the Perdew–Burke–Ernzerhof (PBE) exchange–correlation potentials within generalized gradient approximation (GGA).³⁰ The GGA+ U model was employed to consider the strong electron correlations in 3d orbitals,³¹ where the effective Hubbard parameter $U_{\text{eff}} = U - J$ encompasses both Coulomb repulsion U and Hund's rule-driven exchange interaction J . Electronic functions were described using the projector augmented wave method.³² Kinetic energy cut-off was set at 500 eV to expand the wave functions into a plane-wave basis. A 20 Å thickness vacuum space was set to avoid interlayer interactions caused by periodic



Table 1 Lattice constant a (Å), monolayer thickness h (Å), bond-length l (Å) and energy band gap E_g (eV) of NiPS₃, NiPSe₃ and NiPS_{3/2}Se_{3/2} monolayers calculated with $U = 4.0$ eV, respectively

	a	h	$l_{\text{Ni-S}}$	$l_{\text{Ni-Se}}$	$l_{\text{P-P}}$	$l_{\text{P-S}}$	$l_{\text{P-Se}}$	E_g
NiPS ₃	5.83	3.12	2.47	—	2.18	2.04	—	1.72
NiPSe ₃	6.17	3.25	—	2.58	2.20	—	2.22	1.45
NiPS _{3/2} Se _{3/2}	6.00	3.18	2.48	2.58	2.19	2.05	2.21	1.42

boundary conditions. The convergence criteria were set to be 10^{-3} eV Å⁻¹ for the force and 10^{-6} eV for the energy, respectively. The Brillouin zone (BZ) was sampled using the Monkhorst-Pack method with a Γ -centered $24 \times 24 \times 1$ k -mesh.³³ The COHP distributions were also employed to understand the chemical bonding using the lobster program based on DFT.³⁴ Phonon dispersion was calculated using the finite displacement method with phonopy program,³⁵ and Nosé-Hoover thermostat method was employed for *ab initio* molecular dynamics (AIMD) simulations.^{36,37} T_N values were simulated using MC methods based on the Metropolis algorithm with boundary conditions.^{38,39} Some data related to the pre- and post-processing processes were obtained using the Vesta program⁴⁰ and VASP-KIT package,⁴¹ respectively.

3. Results and discussion

3.1. Crystal structure and stability of 2D NiPX₃ and NiPS_{3/2}Se_{3/2}

The crystal structure of Janus NiPS_{3/2}Se_{3/2} monolayer with a 60° torsion angle between the upper S and lower Se layers (Fig. 1) has a primitive unit cell containing two formula units. Based on the distinct chalcogen atom arrangements in upper and lower sublayers, the intrinsic NiPX₃ monolayer belongs to the D_{3d} space group, while the Janus NiPS_{3/2}Se_{3/2} exhibits a $P\bar{3}1m$

symmetry. Some lattice parameters of 2D NiPS₃ and NiPSe₃ monolayers in the zAFM state are presented in Table 1. The calculated lattice constant value of a was consistent with the corresponding experimental results.^{42,43} The lengths of P-P and P-X bonds indicated a covalent character in a $[\text{PS}_3]^{2-}-[\text{PSe}_3]^{2-}$ anion pair. Owing to the larger atomic radius of Se than that of S, the lattice constant, monolayer thickness and bond length of NiPSe₃ were larger than those of NiPS₃. The structure of Janus NiPS_{3/2}Se_{3/2} was also fully optimized, as shown in Table 1. The bond lengths of Ni-X, P-X and P-P were comparable to those of pristine NiPX₃, indicating good structural stability of the Janus monolayer.

As shown in Fig. 1, four different magnetic configurations, namely, FM, nAFM, zAFM and stripy AFM (sAFM), were considered to determine the magnetic ground state of the NiPX₃ and NiPS_{3/2}Se_{3/2} monolayers. The structures were fully relaxed, and total energy was calculated for each magnetic configuration. The calculated energies followed the sequence of $E_{\text{zAFM}} < E_{\text{nAFM}} < E_{\text{FM}} < E_{\text{sAFM}}$ for different U_{eff} values, as listed in Table 2. Therefore, the pristine NiPX₃ and Janus NiPS_{3/2}Se_{3/2} monolayers were confirmed to be in the zAFM ground state. For the zAFM ordering, each adjacent Ni chain showed an opposite spin orientation along the zigzag direction, while a parallel spin arrangement perpendicular to this direction was displayed in the spin charge density map, as illustrated in Fig. S1.† Here, $U = 4.0$ eV was chosen for further discussions because at this energy, the calculated T_N and band gap of NiPX₃ monolayer were quantitatively close to the corresponding experimental values, which will be discussed in Section 3.2.

The structural stability of pristine NiPX₃ and Janus NiPS_{3/2}Se_{3/2} monolayers were examined considering the elastic constants, phonon spectra and AIMD for assessing their mechanical, dynamical and thermal performances, respectively. For a hexagonal lattice, the elastic constants C_{ij} ($i, j = 1, 2,$

Table 2 Total energy E (eV) at different magnetic configurations and magnetic exchange parameters J_i ($i = 1, 2$ and 3 meV) of 2D NiPX₃ and NiPS_{3/2}Se_{3/2} with a series of U_{eff} (eV) values calculated in $2 \times 1 \times 1$ supercells, respectively. Néel temperature T_N (K) was simulated in a $32 \times 32 \times 1$ superlattice

Compound	U_{eff}	E_{FM}	E_{nAFM}	E_{zAFM}	E_{sAFM}	J_1	J_2	J_3	T_N
NiPS ₃	1.0	-98.12	-98.48	-98.51	-98.06	7.25	0.89	-49.13	384
	2.0	-96.07	-96.34	-94.36	-96.02	5.55	0.72	-32.02	299
	3.0	-94.17	-94.37	-94.39	-94.13	4.22	0.55	-21.84	234
	4.0	-92.41	-92.57	-92.58	-92.38	3.14	0.42	-15.17	186
	5.0	-90.81	-90.92	-90.94	-90.78	2.29	0.32	-10.62	145
	6.0	-89.35	-89.43	-89.44	-89.32	1.60	0.24	-7.40	114
NiPSe ₃	1.0	-89.31	-89.68	-89.73	-89.25	12.22	0.12	-61.78	351
	2.0	-87.13	-87.42	-87.47	-87.09	7.59	-0.46	-40.42	270
	3.0	-85.12	-85.34	-85.38	-85.08	5.74	-0.16	-27.06	210
	4.0	-83.26	-83.43	-83.46	-83.23	4.33	-0.03	-18.68	160
	5.0	-81.56	-81.69	-81.71	-81.53	3.21	0.04	-12.94	125
	6.0	-80.02	-80.11	-80.13	-79.99	2.30	0.07	-8.88	95
NiPS _{3/2} Se _{3/2}	1.0	-93.49	-93.86	-93.92	-93.45	9.28	-0.59	-55.83	371
	2.0	-91.38	-91.66	-91.71	-91.34	6.81	-0.44	-36.87	291
	3.0	-89.42	-89.63	-89.67	-89.39	5.25	-0.17	-24.94	230
	4.0	-87.61	-87.77	-87.81	-87.58	4.00	-0.03	-17.30	180
	5.0	-85.96	-86.08	-86.10	-85.93	2.95	0.04	-11.99	137
	6.0	-84.45	-84.54	-84.56	-84.43	2.14	0.07	-8.35	110



Table 3 Elastic constants C_{ij} (N m^{-1}), Young's modulus Y (N m^{-1}), shear modulus G (GPa) and Poisson's ratio ν of 2D NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$

	C_{11}	C_{12}	C_{66}	Y	G	ν
NiPS_3	93.17	22.97	35.10	87.51	35.10	0.25
NiPSe_3	77.36	19.12	29.12	72.64	29.12	0.25
$\text{NiPS}_{3/2}\text{Se}_{3/2}$	88.78	21.51	33.64	83.58	33.64	0.24

and 6) were calculated as $C = [\partial^2 E / \partial \varepsilon^2] / S$, based on the energy-strain method,⁴⁴ by applying a series of small in-plane strains ε to the monolayers,⁴⁵ where E is the total energy variation with the lattice area S . As listed in Table 3, the elastic constants of C_{11} and C_{12} , with $C_{66} = (C_{11} - C_{12})/2$, satisfy the formulas $C_{11} > 0$ and $C_{11}C_{12} - C_{21}^2 > 0$ of Born-Huang criterion,⁴⁶ demonstrating that pristine NiPX_3 and Janus $\text{NiPS}_{3/2}\text{Se}_{3/2}$ exhibit good mechanical stability. Young's modulus Y , shear modulus G and Poisson's ratio ν were also calculated, as presented in Table 3; the values for $\text{NiPS}_{3/2}\text{Se}_{3/2}$ monolayers located between the corresponding values of NiPS_3 and NiPSe_3 , respectively. All the Y values were much smaller than that of graphene as $Y = 340 \pm 40 \text{ N m}^{-1}$. These results confirmed the mechanical flexibility of NiPX_3 monolayers. The phonon spectra are shown in Fig. 2 along with the density of states for each element. All phonon frequencies were positive in the entire energy range for the three monolayers, revealing excellent dynamical stability.^{47,48} Clearly, Ni, S/Se and P atoms dominated the low-frequency, mid-frequency and high-frequency branches of the phonon spectra, respectively. During AIMD simulations at 300 K, the total energy curve showed small fluctuations in a narrow range for 10 ps. In addition, the 2D lattice remained in a planar structure after each relaxation for the three monolayers, as shown in Fig. 2.

As a consequence, 2D NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ monolayers exhibited good thermal stability at room temperature.

3.2. Magnetic properties of 2D NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$

The MAE plays a crucial role in stabilizing the long-range magnetic order by preventing spontaneous spin flipping in 2D magnetic materials. Taking the spin-orbit coupling (SOC) effect into account, the MAE was calculated from the energy difference as $E_{\text{MAE}} = E_{(100)} - E_{(001)}$, with spin orientations in and out of the monolayer plane. The partial components of MAE from different atoms, *i.e.* E_{MAE} (Ni) was also calculated. The MAE results of NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ monolayers are listed in Table 4. As can be seen, the E_{MAE} was lower for NiPS_3 , while it was higher for NiPSe_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ in *ab*-plane (100) than in *c*-direction [001]. The calculated MAE results reveal distinct magnetic easy axes for the three monolayers: NiPS_3 exhibits an in-plane (*ab*-plane) easy axis along the *a*-axis, while NiPSe_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ show out-of-plane (*c*-axis) easy axes perpendicular to the *ab*-plane.²⁷ Furthermore, the absolute E_{MAE} value was larger in the *b* [010] direction than in the *a* [100] direction for the three monolayers owing to their zAFM spin configuration. From the partial component of the E_{MAE} , the magnetic anisotropy of NiPS_3 was mainly contributed by Ni atoms, whereas, Se atoms dominated the magnetic anisotropy of NiPSe_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ monolayers.

In order to describe the magnetic interactions between Ni^{2+} ions for these zAFM monolayers, the magnetic exchange parameters were evaluated using the Heisenberg Hamiltonian as follows:^{9,49}

$$H = -\frac{1}{2} \left(\sum_{\langle ij \rangle} J_1 S_i \cdot S_j + \sum_{\langle\langle ij \rangle\rangle} J_2 S_i \cdot S_j + \sum_{\langle\langle\langle ij \rangle\rangle\rangle} J_3 S_i \cdot S_j \right) \quad (1)$$

where J_1, J_2 and J_3 represent the magnetic exchange parameters in NN, second-nearest-neighboring (2NN) and 3NN distances, respectively. S_i is the magnetic moment at the atomic site i of Ni^{2+} ions. The calculated magnetic moment was about $1.40\mu_{\text{B}}$, which was in consistent with previous results for pristine NiPX_3 monolayers.^{50,51} The J_i parameters were calculated by mapping

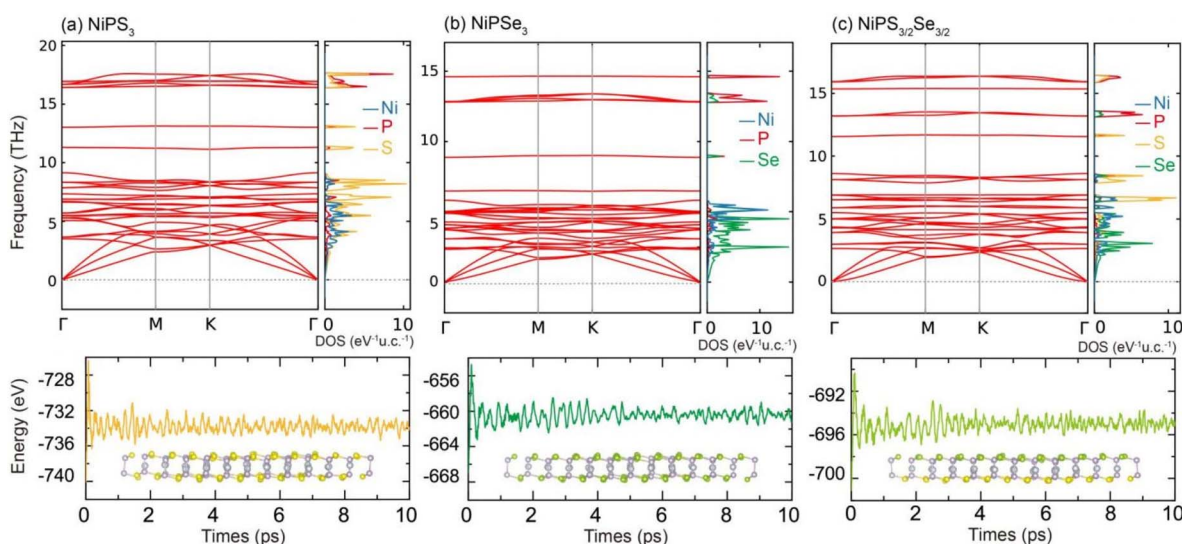


Fig. 2 Phonon spectra along with density of states and total energy oscillations with side views of the relaxed structures after AIMD simulations for (a) NiPS_3 , (b) NiPSe_3 and (c) $\text{NiPS}_{3/2}\text{Se}_{3/2}$.



Table 4 Total E_{MAE} partial component (*i.e.* E_{MAE} (Ni)) of MAE with respect to the value in (001) direction for 2D NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ in the unit of meV per u.c. (u.c. indicates a unit cell). Easy axis denotes the MAE-determined spin orientation

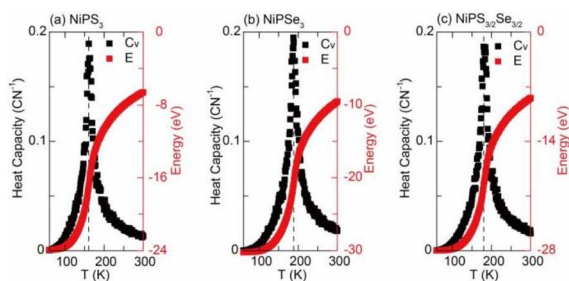
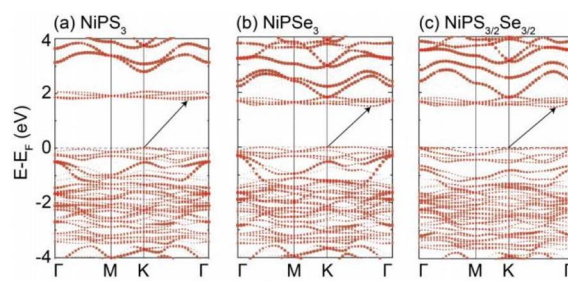
Compound	Orientation	E_{MAE}	E_{MAE} (Ni)	E_{MAE} (P)	E_{MAE} (S)	E_{MAE} (Se)	Easy axis
NiPS_3	(100)	−87	−82.1	0.7	12.6		<i>ab</i> -Plane
	(010)	−95	−52.5	0.5	1.5		
NiPSe_3	(100)	344	−6.1	0.05	—	55.3	(001)
	(010)	473	−17.2	10.4		81.1	
$\text{NiPS}_{3/2}\text{Se}_{3/2}$	(100)	185	−6.6	0.2	1.0	64.9	(001)
	(010)	369	−31.2	5.6	−0.5	140.6	

the total energies in the four distinct magnetic configurations to the Heisenberg Hamiltonian. The corresponding results for a range of U_{eff} values are presented in Table 2 for each of the three monolayers, indicating that variations in U_{eff} values significantly influence the J_i values. A larger U_{eff} value indicated stronger localization of the d-orbitals, which in turn reduced the magnetic exchange interactions between the Ni ions. In addition, the chalcogen atoms exhibited a notable impact on the magnetic exchange interactions; in particular, NiPS_3 showed smaller J_i values than those of NiPSe_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$. In the case of $U = 4.0$ eV, the calculated results were $J_1 = 3.14$ meV, $J_2 = 0.42$ meV and $J_3 = -15.17$ meV for NiPS_3 and $J_1 = 4.33$ meV, $J_2 = -0.03$ meV and $J_3 = -18.68$ meV for NiPSe_3 , which were in good agreement with previously reported theoretical values.^{25,27} For $\text{NiPS}_{3/2}\text{Se}_{3/2}$, the calculated exchange parameters were $J_1 = 4.00$ meV, $J_2 = -0.03$ meV and $J_3 = -17.30$ meV, which were close to those of NiPSe_3 . The positive J_1 value indicated FM interactions between NN Ni^{2+} ions. The 2NN J_2 was very small, indicating a negligible effect. In contrast, 3NN J_3 was negative and exhibited the largest absolute value, which was crucial for determining the ground state of these zAFM monolayers.

The specific heat capacity is defined as $C_v = \frac{\partial E}{\partial T} = \frac{\langle E^2 \rangle - \langle E \rangle^2}{K_B T^2}$ where E represents in ternal energy and k_B is the Boltzmann constant; T_N could be calculated from the second-order phase transition variable C_v curve. Using MC simulations, the critical transition temperature was evaluated in a 32×32 superlattice, with spin directions flipped randomly about 2×10^9 steps per loop. The simulated C_v and E curves as a function of temperature are plotted in Fig. 3. Under the condition of $U_{\text{eff}} = 4.0$ eV, calculated T_N values were in good

agreement with the experimental results for each pristine NiPX_3 monolayer. In particular, $T_N = 160$ K was very close to the experimental value of 155 K (ref. 52) for NiPS_3 , while $T_N = 189$ K was slightly lower than 212 K, when measured using temperature-dependent molar susceptibility⁵³ for NiPSe_3 . Notably, the T_N of NiPS_3 was lower than that of NiPSe_3 , which is attributed to the stronger magnetic exchange interactions of NiPSe_3 than that of NiPS_3 . Similarly, the T_N of Janus $\text{NiPS}_{3/2}\text{Se}_{3/2}$ is predicted to be 180 K, which represents an enhancement compared to 160 K for NiPS_3 but shows a decrease relative to 189 K for NiPSe_3 . This feature was consistent with the behavior predicated for Janus $\text{MnPS}_{3/2}\text{Se}_{3/2}$ monolayer, where the magnetic exchange interactions were dominated by transition metal elements.⁵⁴ The T_N values for a series of U_{eff} values were also calculated and are presented in Table 2 for each monolayer. As mentioned above, the T_N was much larger than the corresponding experimental value when $U > 4.0$ eV but much smaller when $U < 4.0$ eV. These results further confirmed the rationality of selecting $U = 4.0$ eV for our calculations. Along with the near-room critical temperature, NiPSe_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ monolayers, which exhibit an out-of-plane magnetization, are suitable for magnetic storage and low-power switching devices, *e.g.* spin-transfer torque for electrical switching, whereas the in-plane magnetized NiPS_3 monolayer is suited for ultrafast spin information processing, *e.g.* spin-wave logic circuits.^{55,56}

3.3. Electronic structures of 2D NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$. Band structures were calculated in a $2 \times 1 \times 1$ supercell and then projected into primitive BZ along the high-symmetry line of Γ -M-K- Γ ,⁵⁷ as illustrated in Fig. 4(a)–(c) for NiPX_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$, respectively. The three zAFM monolayers behaved as typical semiconductors, with an indirect band gap, valence

**Fig. 3** Specific heat capacity C_v and total energy E with respect to temperature T for 2D (a) NiPS_3 , (b) NiPSe_3 and (c) $\text{NiPS}_{3/2}\text{Se}_{3/2}$ calculated using Monte Carlo simulations.**Fig. 4** Band structures of 2D (a) NiPS_3 , (b) NiPSe_3 and (c) $\text{NiPS}_{3/2}\text{Se}_{3/2}$. Black arrow points from valence band maxima to conduction band minima.

band (VB) maxima located at the K point and the conduction band (CB) minima in the K- Γ path. Notably, the calculated band gap E_g was 1.71 eV for NiPS_3 and 1.44 eV for NiPSe_3 at $U = 4.0$ eV, and these values were consistent with the previously obtained DFT results.⁵⁰ Furthermore, the calculated value of 1.71 eV in the 2D structure was larger than the experimentally obtained value of 1.66 eV in bulk phase of NiPS_3 .⁵⁸ The larger bandgap of NiPS_3 than that of NiPSe_3 can be attributed to the stronger electronegativity of S than Se. Correspondingly, the bandgap of 1.42 eV of $\text{NiPS}_{3/2}\text{Se}_{3/2}$ was between the bandgap values of NiPS_3 and NiPSe_3 and was much closer to that of NiPSe_3 . Beyond the indirect band gap, the band structures of the three monolayers showed similar dispersion patterns; for instance, the four uppermost VBs and four undermost CBs near the Fermi level reached their maximum degeneracy at M point. In addition, the four undermost CBs were almost isolated from the other higher CBs.

In order to investigate the electronic structures in detail, the total density of states (DOS) and orbital projected DOS (PDOS) for NiPS_3 and $\text{NiPS}_{3/2}\text{Se}_{3/2}$ were plotted in Fig. 5(a)–(c). As the NiPS_3 system existed in an AFM ground state, the spin-up channel for each DOS or PDOS. Close to the Fermi level, the uppermost four VBs were mainly contributed by the p_x , p_y , and p_z states of chalcogen atoms, and Ni d orbitals also had some contributions for each zAFM monolayer. The VB maximum was mostly composed of chalcogen p_x and p_y orbitals; hence, NiPS_3 monolayer exhibited a smaller band gap than the NiPSe_3 monolayer as the electronegativity of the S atom was larger than that of the Se atom. It should be noted that just below the Fermi level, the Se p_x and p_y orbitals showed more concentrated

contributions than those of the S atom, resulting in a flat band along the K- Γ path for NiPSe_3 , especially for $\text{NiPS}_{3/2}\text{Se}_{3/2}$.

The electron numbers N were calculated by integrating the PDOS in the spin-up channel within the energy range from -0.50 eV to 0.0 eV by the formula of $N = \kappa \int \text{PDOS} dE$, where $\kappa = 1/2$ for pristine structures and $\kappa = 1$ for Janus phase, based on their respective molecular formulas. As listed in Table S1,[†] the electron numbers of the chalcogen elements in their p orbitals were calculated to be $N_S = 1.82$ e for NiPSe_3 and $N_{Se} = 1.96$ e for NiPSe_3 , while $N_S = 0.94$ e and $N_{Se} = 2.33$ e for $\text{NiPS}_{3/2}\text{Se}_{3/2}$. These results clearly indicated that the Se element exhibited more influence on the electronic structures and magnetic properties than the S element for $\text{NiPS}_{3/2}\text{Se}_{3/2}$. As a consequence, the bandgap, MAE value and magnetic exchange parameters of Janus $\text{NiPS}_{3/2}\text{Se}_{3/2}$ were much closer to that of NiPSe_3 than that of NiPS_3 . Here, Ni d electrons contributed more to the four undermost CBs than to the four uppermost VBs near the Fermi level. These contributions were primarily composed of the d_{xz} and d_{yz} states, with some contributions from the d_{xy} and $d_{x^2-y^2}$ orbitals. This feature deviated from that of some other AFM TMPX₃, such as CrPSe_3 and MnPS_3 , where d electrons contributed more to the uppermost VBs than to the undermost CBs.^{10,12} Throughout the energy range, P p_z orbitals strongly hybridized with those of S/Se atoms, forming the $[\text{P}_2\text{X}_6]^{4-}$ octahedrons that provided excellent mechanical stability for each monolayer. Notably, the Ni d orbitals exhibited pronounced hybridization with the chalcogen p_x and p_y orbitals, facilitating an indirect p-d exchange interaction for Ni ions.

As illustrated in Fig. S1 and S2,[†] both the spin charge density maps and charge distributions of the Ni hexagonal slide could not provide enough information for the bonding behavior between Ni ions. Here, the electronic structures were further

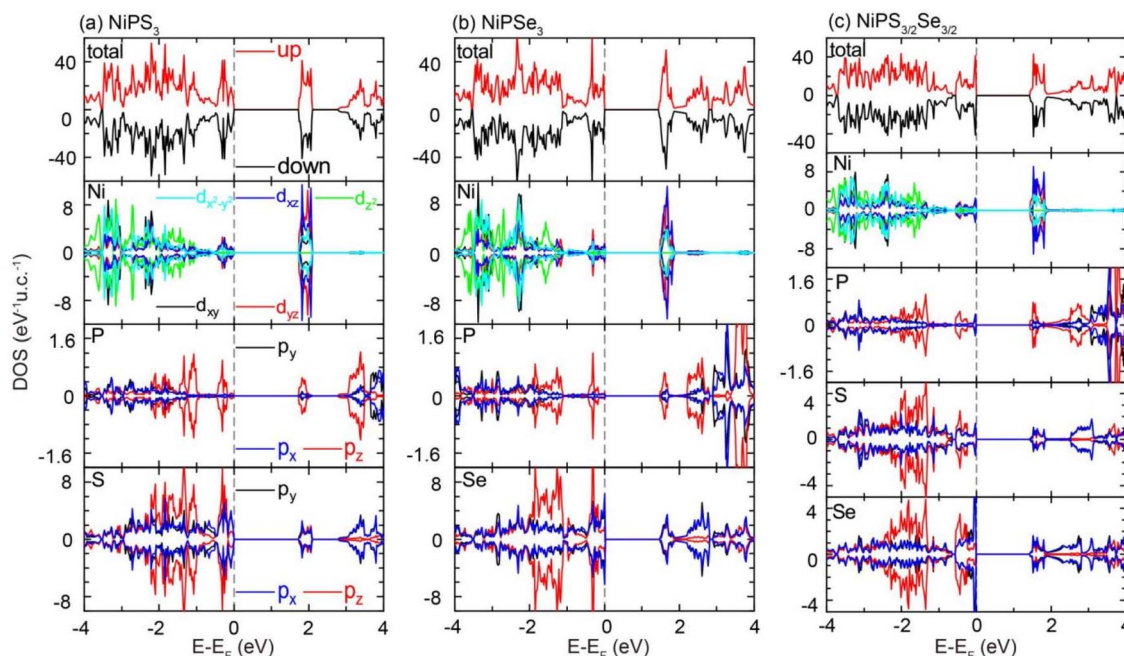


Fig. 5 Total density of states (DOS) and orbital-projected DOS of 2D (a) NiPS_3 , (b) NiPSe_3 and (c) $\text{NiPS}_{3/2}\text{Se}_{3/2}$.



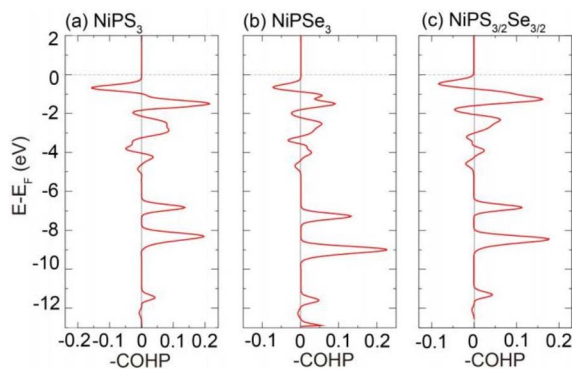


Fig. 6 –COHP distributions between NN Ni ions in 2D (a) NiPS₃, (b) NiPSe₃ and (c) NiPS_{3/2}Se_{3/2}.

evaluated by extracting the chemical bonding data from COHP diagrams for NiPX₃ and NiPS_{3/2}Se_{3/2} monolayers. The negative diagrams of NN Ni–Ni couplings are plotted in Fig. 6(a)–(c), respectively. In these –COHP diagrams, the bonding and antibonding states were denoted by positive and negative values, respectively. It was evident that there were four negative peak-curves for –COHP below the Fermi level, where the plus/minus signs of –COHP peaks alternated from –6.0 eV to Fermi energy, which was set to 0.0 eV. In particular, the first main peak, situated just under Fermi level, exhibited negative values, which played a vital role in the zAFM configuration.

The coexistence of FM and AFM couplings between NN Ni ions was discussed by considering both the crystal field effect and electronic structure to clarify the mechanism of zAFM ordering. As shown in Fig. 7(a), each Ni atom was surrounded by six S/Se atoms, forming an octahedral structure with D_{3d} symmetry.^{51,59} Owing to crystal field effect, Ni 3d orbitals could be divided into two parts: one group was named as t_{2g} of the d_{xy} ,

$d_{x^2-y^2}$ and d_{z^2} orbitals and the other one was denoted as e_g of d_{xz} and d_{yz} orbitals. According to the orbital-projected DOS in Fig. 5, the energy of the t_{2g} state was lower than that of the e_g state in VBs; thus, the first negative –COHP peak under 0.0 eV was composed of e_g states. Since Ni 3d electrons were in a configuration of $3d^8 4s^2$, the t_{2g} orbitals were fully occupied at the lower energy level, yielding no net magnetic moment. In contrast, the half-filled d_{xz}/d_{yz} orbitals had two electrons in the higher energy level with the same-direction spin alignment, resulting in a net magnetic moment with a calculated value about $1.40\mu_B$ for each Ni ion. As illustrated in Fig. 7(b), the AFM exchange interactions could be attributed to the direct hopping between the half-filled e_g orbitals for NN Ni ions, forming some bonding states. Comparatively, in Fig. 7(c), when NN Ni ions formed an FM arrangement, the FM exchange interactions could be realized by some antibonding states according to Hund's rule. As a consequence, the alternation of antibonding and bonding states led to the coexistence of AFM and FM couplings between NN Ni ions in NiPX₃ and NiPS_{3/2}Se_{3/2} monolayers.

As discussed above, the first negative –COHP peak could be considered as a result of the FM coupling between e_g orbitals in the antibonding states, resulting in positive J_1 values. In other words, the FM indirect interactions between the NN Ni ions dominated the competition against the AFM direction exchange interactions, as illustrated in Fig. 1. However, the AFM ground states of NiPX₃ and NiPS_{3/2}Se_{3/2} were predominantly determined by the 3NN superexchange interactions mediated *via* p–d hybridization. Evidently, the absolute value J_3 significantly exceeded J_1 , whereas the value of J_2 was negligible. This huge difference between J_2 and J_3 originated from their distinct superexchange pathways. As shown in Fig. 1, the 2NN interactions were mediated by two S/Se atoms residing in different sublayers (one in the upper and the other in the lower), while the 3NN interactions were mediated by two S/Se atoms within the same sublayer. Here, the magnetic mechanism of Janus NiPS_{3/2}Se_{3/2} was consistent with that of the pristine NiPX₃ system; thus, the magnetic properties, including MAE, exchange parameters and T_N values of NiPS_{3/2}Se_{3/2} were similar to those of NiPX₃ in the zAFM state. This stable magnetic feature was consistent with our previous findings on the MnPX₃ system in the nAFM state.¹⁰

4. Conclusion

In summary, the electronic structures and magnetic properties of 2D pristine NiPX₃ and Janus phase NiPS_{3/2}Se_{3/2} were comprehensively investigated using the first-principles calculations. Despite the on-site Coulomb repulsion with a series of U_{eff} values, both pristine and Janus NiPX₃ systems remained in a zAFM ground state and were characterized as indirect band gap semiconductors. The bandgaps of NiPS₃ and NiPSe₃ were 1.71 eV and 1.44 eV, respectively, while the value of NiPS_{3/2}Se_{3/2} was 1.42 eV. Three 2D monolayers showed excellent structural stability as determined using the elastic coefficients satisfying the Born criterion, phonon spectra with no negative frequency and a planar structure after AIMD relaxation. Calculated MAE

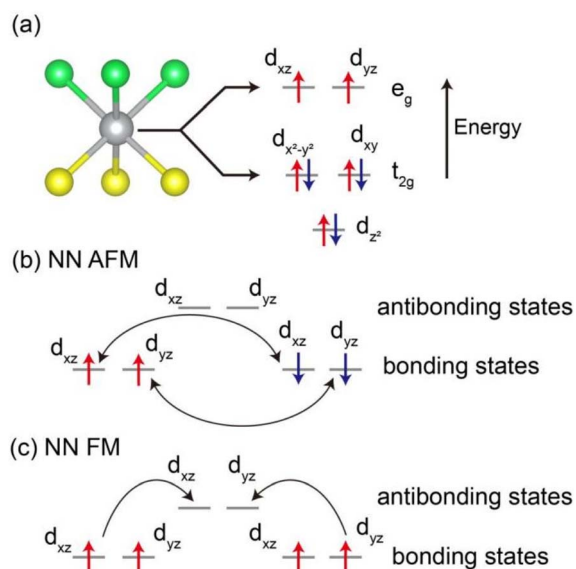


Fig. 7 Schematic of (a) octahedral crystal field and 3d electron fillings and (b) AFM, and (c) FM exchange interactions between NN Ni ions.



values showed that the magnetic easy-magnetization axis was in the z-axis direction for NiPSe₃ and NiPS_{3/2}Se_{3/2}, while it was in the xy-plane for NiPS₃. The magnetic exchange parameters for NiPS₃ was $J_1 = 3.14$ meV, $J_2 = 0.42$ meV and $J_3 = -15.17$ meV, for NiPSe₃ was $J_1 = 4.33$ meV, $J_2 = -0.03$ meV and $J_3 = -18.68$ meV and for NiPS_{3/2}Se_{3/2} was $J_1 = 4.00$ meV, $J_2 = -0.03$ meV, and $J_3 = -17.30$ meV. Néel temperatures predicted using Monte Carlo simulations were 160 K for NiPS₃, 189 K for NiPSe₃, and 180 K for NiPS_{3/2}Se_{3/2}. The magnetic properties of Janus NiPS_{3/2}Se_{3/2} were more similar to those of pristine NiPSe₃ because the Se p orbitals contributed more to the uppermost valence bands than the S p states for NiPS_{3/2}Se_{3/2}. The AFM ground state was determined by evaluating the large superexchange interactions mediated by p–d hybridization for 2D NiPX₃ in both the pristine structure and Janus phase. Furthermore, the coexistence of antibonding and bonding states under the Fermi level led to FM couplings and AFM couplings between NN Ni ions, which could be clearly clarified by the crystal field effect and crystal orbital Hamilton populations.

Data availability

The data supporting this article have been included as part of the ESI.†

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the programs of Innovation Team in Colleges for Science and Technology of Education Department of Hubei Province (No. T2021012), Doctoral Scientific Research Foundation of HUAT (BK202483, BK202208 and BK202302), Hubei key Laboratory of Energy Storage and Power Battery (No. QCCLSZK2021A06), Open Foundation Project of Hubei Key Laboratory of Optical Information and Pattern Recognition (No. 202403), WIT and College Students' Innovation and Entrepreneurship Training Program (DC2023074 and DC2024087).

References

- X. Wang, K. Du, Y. Y. F. Liu, P. Hu, J. Zhang, Q. Zhang, M. H. S. Owen, X. Lu, C. K. Gan, P. Sengupta, *et al.*, *2D Mater.*, 2016, **3**, 1.
- J.-U. Lee, S. Lee, J. H. Ryoo, S. Kang, T. Y. Kim, P. Kim, C.-H. Park, J.-G. Park and H. Cheong, *Nano Lett.*, 2016, **16**, 7433.
- P. Li, X. Li, J. Feng, J. Ni, Z.-X. Guo and H. Xiang, *Phys. Rev. B*, 2024, **109**, 214418.
- R. Brec, *Solid State Ionics*, 1986, **22**, 3.
- A. Wiedenmann, J. Rossat-Mignod, A. Louisy, R. Brec and J. Rouxel, *Solid State Commun.*, 1981, **40**, 1067.
- W. Klingner, R. Ott and H. Hahn, *Z. Anorg. Allg. Chem.*, 1973, **396**, 271.
- F. Wang, T. A. Shifa, P. Yu, P. He, Y. Liu, F. Wang, Z. Wang, X. Zhan, X. Lou, F. Xia, *et al.*, *Adv. Funct. Mater.*, 2018, **28**, 1802151.
- K.-z. Du, X.-z. Wang, Y. Liu, P. Hu, M. I. B. Utama, C. K. Gan, Q. Xiong and C. Kloc, *ACS Nano*, 2016, **10**, 1738.
- B. L. Chittari, Y. Park, D. Lee, M. Han, A. H. MacDonald, E. Hwang and J. Jung, *Phys. Rev. B*, 2016, **94**, 184428.
- J.-T. Yang, C.-J. Xu, H.-J. Wang, Q. Min, S.-J. Luo, Y.-C. Xiong, W. Ren and C. Jing, *J. Phys.: Condens. Matter.*, 2023, **35**, 395501.
- Y. Sugita, T. Miyake and Y. Motome, *Phys. Rev. B*, 2018, **97**, 035125.
- J. Yang, Y. Zhou, Y. Dedkov and E. Voloshina, *Adv. Theory Simul.*, 2020, 2000228.
- Y. Jin, M. Yan, T. Kremer, E. Voloshina and Y. Dedkov, *Sci. Rep.*, 2022, **12**, 735.
- S. Y. Kim, T. Y. Kim, L. J. Sandilands, S. Sinn, M.-C. Lee, J. Son, S. Lee, K.-Y. Choi, W. Kim, B.-G. Park, *et al.*, *Phys. Rev. Lett.*, 2018, **120**, 136402.
- S. Y. Lim, K. Kim, S. Lee, J.-G. Park and H. Cheong, *Curr. Appl. Phys.*, 2021, **21**(1).
- A. Wildes, B. Fåk, U. Hansen, M. Enderle, J. Stewart, L. Testa, H. Rønnow, C. Kim and J.-G. Park, *Phys. Rev. B*, 2023, **107**, 054438.
- Y. Wang, Z. Zhou, T. Wen, Y. Zhou, N. Li, F. Han, Y. Xiao, P. Chow, J. Sun, M. Pravica, *et al.*, *J. Am. Chem. Soc.*, 2016, **138**, 15751.
- Y. Wang, J. Ying, Z. Zhou, J. Sun, T. Wen, Y. Zhou, N. Li, Q. Zhang, F. Han, Y. Xiao, *et al.*, *Nat. Commun.*, 2018, **9**, 1914.
- H. Sun, L. Qiu, Y. Han, E. Yi, J. Li, M. Huo, C. Huang, H. Liu, M. Li, W. Wang, *et al.*, *Mater. Today Phys.*, 2023, **36**, 101188.
- Q. Zhang, K. Hwangbo, C. Wang, Q. Jiang, J.-H. Chu, H. Wen, D. Xiao and X. Xu, *Nano Lett.*, 2021, **21**, 6938.
- Z. Ni, A. Haglund, H. Wang, B. Xu, C. Bernhard, D. Mandrus, X. Qian, E. Mele, C. Kane and L. Wu, *Nat. Nanotechnol.*, 2021, **16**, 782.
- A. Ghosh, M. Palit, S. Maity, V. Dwij, S. Rana and S. Datta, *Phys. Rev. B*, 2021, **103**, 064431.
- E. Ergöçen, B. Ilyas, D. Mao, H. C. Po, M. B. Yilmaz, J. Kim, J.-G. Park, T. Senthil and N. Gedik, *Nat. Commun.*, 2022, **13**, 98.
- F. Dirnberger, R. Bushati, B. Datta, A. Kumar, A. H. MacDonald, E. Baldini and V. M. Menon, *Nat. Nanotechnol.*, 2022, **17**, 1060.
- P. Li, X. Li, J. Feng, J. Ni, Z.-X. Guo and H. Xiang, *Phys. Rev. B*, 2024, **109**, 214418.
- M. Amirabbasi and P. Kratzer, *Phys. Rev. B*, 2023, **107**, 024401.
- R. Basnet, K. M. Kotur, M. Rybak, C. Stephenson, S. Bishop, C. Autieri, M. Birowska and J. Hu, *Phys. Rev. Res.*, 2022, **4**, 023256.
- J. Hafner, *J. Comput. Chem.*, 2008, **29**, 2044.
- G. Kresse and J. Hafner, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1993, **47**, 558.
- L. Matthes, O. Pulci and F. Bechstedt, *J. Phys.: Condens. Matter*, 2013, **25**, 395305.



- 31 B. Himmetoglu, A. Floris, S. De Gironcoli and M. Cococcioni, *Int. J. Quant. Chem.*, 2014, **114**, 14.
- 32 S. Maintz, V. L. Deringer, A. L. Tchougréeff, and R. Dronskowski, *Lobster: A Tool to Extract Chemical Bonding from Plane-Wave based DFT*, 2016.
- 33 H. J. Monkhorst and J. D. Pack, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1976, **13**, 5188.
- 34 R. Dronskowski and P. E. Bloechl, *J. Phys. Chem.*, 1993, **97**, 8617.
- 35 A. Togo and I. Tanaka, *Scr. Mater.*, 2015, **108**, 1.
- 36 G. J. Martyna, M. L. Klein and M. Tuckerman, *J. Chem. Phys.*, 1992, **97**, 2635.
- 37 C. Braga and K. P. Travis, *J. Chem. Phys.*, 2005, **123**, 134101.
- 38 R. Blankenbecler, D. Scalapino and R. Sugar, *Phys. Rev. D*, 1981, **24**, 2278.
- 39 H. Haario, E. Saksman and J. Tamminen, *Bernoulli*, 2001, 223–242.
- 40 K. Momma and F. Izumi, *J. Appl. Crystallogr.*, 2011, **44**, 1272.
- 41 V. Wang, N. Xu, J.-C. Liu, G. Tang and W.-T. Geng, *Comput. Phys. Commun.*, 2021, **267**, 108033.
- 42 A. R. Wildes, V. Simonet, E. Ressouche, G. J. McIntyre, M. Avdeev, E. Suard, S. A. J. Kimber, D. Lançon, G. Pepe, B. Moubaraki, *et al.*, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2015, **92**, 224408.
- 43 G. Le Flem, R. Brec, G. Ouvard, A. Louisy and P. Segransan, *J. Phys. Chem. Solids*, 1982, **43**, 455.
- 44 F. Knuth, C. Carbogno, V. Atalla, V. Blum and M. Scheffler, *Comput. Phys. Commun.*, 2015, **190**, 33.
- 45 E. Cadelano, P. L. Palla, S. Giordano and L. Colombo, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **82**, 235414.
- 46 R. C. Andrew, R. E. Mapasha, A. M. Ukpong and N. Chetty, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 125428.
- 47 A. Togo, *J. Phys. Soc. Jpn.*, 2023, **92**, 012001.
- 48 P. Giannozzi, S. De Gironcoli, P. Pavone and S. Baroni, *Phys. Rev. B*, 1991, **43**, 7231.
- 49 H. J. Xiang, E. J. Kan, S.-H. Wei, M.-H. Whangbo and X. G. Gong, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 224429.
- 50 S. Yan, Y. Du, X. Zhang, X. Wan and D. Wang, *J. Phys.: Condens. Matter*, 2023, **36**, 065502.
- 51 Y. Gu, Q. Zhang, C. Le, Y. Li, T. Xiang and J. Hu, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2019, **100**, 165405.
- 52 Y. Lee, C. Kim, S. Son, J. Cui, G. Park, K.-X. Zhang, S. Oh, H. Cheong, A. Kleibert and J.-G. Park, *Nano Lett.*, 2024, **24**, 6043.
- 53 R. Basnet, K. M. Kotur, M. Rybak, C. Stephenson, S. Bishop, C. Autieri, M. Birowska and J. Hu, *Phys. Rev. Res.*, 2022, **4**, 023256.
- 54 J.-T. Yang, C.-J. Xu, H.-J. Wang, Q. Min, S.-J. Luo, Y.-C. Xiong, W. Ren and C. Jing, *J. Phys.: Condens. Matter*, 2023, **35**, 395501.
- 55 Q. H. Wang, A. Bedoya-Pinto, M. Blei, A. H. Dismukes, *et al.*, *ACS Nano*, 2022, **16**, 6960.
- 56 S. Toyoda, J. Kruppe, K. Yamakawa, J. Analytis and J. Orenstein, *Phys. Rev. B*, 2024, **109**, 064408.
- 57 V. Popescu and A. Zunger, *Phys. Rev. Lett.*, 2010, **104**, 236403.
- 58 P. Foot, J. Suradi and P. Lee, *Mater. Res. Bull.*, 1980, **15**, 189.
- 59 C. Autieri, G. Cuono, C. Noce, M. Rybak, K. M. Kotur, C. E. Agrapidis, K. Wohlfeld and M. Birowska, *J. Phys. Chem. C*, 2022, **126**, 6791.

