


Cite this: *RSC Adv.*, 2025, 15, 17580

Versatile near-infrared polarization-sensitive ionic liquid-gated organic electrochemical phototransistor†

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In response to the rising demand for diversified detection capabilities, multi-dimensional and multi-functional optoelectronic devices have become a significant focus in scientific research. The organic electrochemical phototransistor (OECPT) is a pioneering photoelectric conversion device whose unique operating mechanism positions it as a strong candidate for applications in areas such as biological systems, sensing, and artificial neural network modeling. In this study, polarization-responsive OECPT devices were fabricated by combining thermally oriented anisotropic thin films with ionic liquid gating. The resulting devices demonstrated a distinct polarization sensitivity in the near-infrared region, achieving a photogenerated current dichroic ratio of 1.52. Furthermore, by modulating device non-volatility vis gate voltage, we explored the potential of OECPTs in neural synapse emulation and optoelectronic memory storage. These findings provide valuable insights for advancing the design and application of polarization-sensitive OECPT devices in multifunctional optoelectronic systems.

Received 17th February 2025

Accepted 19th May 2025

DOI: 10.1039/d5ra01167a

rsc.li/rsc-advances

Introduction

In response to the rapid advancements in science and technology, the replacement cycle for electronic devices has significantly shortened, necessitating continuous improvement in the performance of core electronic components. Key areas of focus include ensuring high reliability, multifunctionality, and minimal energy consumption.^{1–3} These goals demand concerted research and development efforts aimed at pushing the boundaries of current technologies. Among emerging devices, organic phototransistors (OPTs) stand out as promising optoelectronic devices, leveraging semiconductor materials with photosensitive properties to replace conventional conductive channel layers.^{4,5} This substitution enables OPTs to generate light-responsive current signals, with gate modulation offering effective noise suppression, signal amplification, and additional storage functions based on the dielectric layer.⁶ These properties have positioned OPTs at the forefront of research, promising new applications in fields such as photodetection,^{7,8} photoelectric storage,^{9,10} and neural simulation.^{11–14} Traditional transistors, however, face limitations due to their high operating voltage, motivating research into various approaches to overcome these constraints. Innovations have included the use

of high-dielectric-constant materials,^{15,16} reduced device charge trap density,^{17,18} dynamic voltage regulation,¹⁹ and the exploration of novel transistor structures.^{20–22}

One such approach is the organic electrochemical transistor (OECT),²³ a three-terminal device introduced in the 1980s that utilizes an electrolyte layer with mobile ions as a gate-regulation medium.²⁴ Unlike conventional field-effect transistors, OECTs rely on ion-mediated gate field modulation to control charge transport in the channel. The distinctive properties of the electric double layer (EDL) formed by ions in the OECTs enable reduced gate dielectric thickness (nanometer level), resulting in notable low operating voltages. The flexibility in selecting gate electrolytes for OECTs also enables biocompatibility, environmental compatibility, and cost-effectiveness,^{25,26} expanding potential applications in environmental monitoring,²⁷ photo-detection,²⁸ and biomedical fields.^{29,30}

Combining a photosensitive channel layer with an OECT structure leads to organic electrochemical phototransistors (OECPTs), which exhibit high photo-responsivity under low-voltage operation. Researchers have achieved notable advances in OECPT design. Lhuillier *et al.* in 2014 synthesized CdSe/CdS nanosheets with a core/shell structure for use as a photoactive semiconducting layer in an electrochemical transistor, achieving a responsivity of approximately 10 mA W^{-1} at an operating voltage below than 2 V.³¹ In 2017, Xu *et al.* developed large-capacitance ionic gels to serve as gate electrolytes in a bulk heterojunction (BHJ)-based OECPT, resulting in a maximum responsivity of approximately $1.5 \times 10^3 \text{ A W}^{-1}$.³² In 2019, Lan *et al.* demonstrated an OECPT device that achieved

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† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d5ra01167a>



a maximum responsivity of 7.3 A W^{-1} under the illumination of a sunlight simulator.³³ In 2022, Sun *et al.* showcased a responsivity of 3.56 A W^{-1} for an OECPT when exposed to near-infrared light, with potential applications in erasable photoelectric storage devices.³⁴ By 2023, Yang *et al.* further improved OECPT technology to create a stretchable, ultrasensitive, low-energy phototransistor capable of real-time human pulse monitoring.³⁵ In 2024, Lin *et al.* reported that the OECPT, designed as an electrical double-layer transistor (EDLT), exhibited an extremely low energy consumption of just 0.59 nW , along with a responsivity of up to 6.12 A W^{-1} .³⁶ The relevant properties of the above devices are listed in Table 1 for comparison.

The recent development of polarized light detectors enhances light polarization detection beyond intensity and wavelength, allowing devices to distinguish between linear and circular polarizations based on anisotropic optical absorption.³⁷ In 2011, anisotropic absorption in poly(3-hexylthiophene) (P3HT) films was achieved by Yang *et al.* through the application of unidirectional thermal rubbing.³⁸ The resulting photodetector exhibited an electric current dichroic ratio (DR_i) of 3.54. Subsequently, various film fabrication methods, such as blade-coating,^{39,40} off-center spin-coating,^{41,42} unidirectional floating film transfer method,^{43–46} and strain-induced orientation,^{47–49} have been employed to impart polarization sensitivity to polymer films, leading to advances in polarization-sensitive photodetectors. However, further research is required to fully integrate electrochemical phototransistors with polarization detection.

In this work, oriented poly[[1,2,5]thiadiazolo[3,4-*c*]pyridine-4,7-diyl (4,4-dihexadecyl-4*H*-clopenta[2,1-*b*:3,4-*b'*]dithiophene-2,6-diyl)[1,2,5] thiadiazolo[3,4-*c*]pyridine-7,4-diyl(4,4-dihexadecyl-4*H*-cyclopenta [2,1-*b*:3,4-*b'*]dithiophene-2,6-diyl)] (PCDTPT) films with dichroic optical absorption were prepared using thermal rubbing orientation. These films were then integrated into an electrochemical transistor with a top-gate-bottom-contact configuration and an ionic liquid ([EMIM][TFSI]) as the gate electrolyte. This approach enables the creation of polarization-sensitive electrochemical phototransistors capable of distinguishing linearly polarized light, exhibiting distinct photocurrent response to different polarization angles. While ordering the polymer backbone through rubbing orientation is a common method for preparing oriented aligned layers, most of the existing studies have focused only on the change in mobility and oriented films' performance in two-terminal devices.^{50–52} In contrast, this work applies oriented polymer films to ionic liquid-gated

electrochemical transistors and explores their device performance. The resulting electrochemical phototransistor devices exhibit a significant anisotropic photoelectric response. Although the gating results are somewhat routine, the novel aspect lies in the device's distinct response to incident light at different polarization angles. Furthermore, the non-volatile nature of the optical response shows a clear dependence on the gate voltage. In conclusion, this work may offer valuable insights for the development of novel multifunctional optoelectronic devices that integrate anisotropic thin films with electrolyte-gated electrochemical phototransistors.

Experimental section

Materials

The gate electrolyte material, 1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide ([EMIM][TFSI], $\text{C}_8\text{H}_{11}\text{F}_6\text{N}_3\text{O}_4\text{S}_2$, 99%), was obtained from Innochem. The active layer material, PCDTPT ($M_w = 75\,000$), was sourced from 1-material. Chloroform (CH_3Cl , >99.8%) was purchased from Amethyst Chemicals. Custom-designed patterned indium tin oxide (ITO, $\sim 135 \text{ nm}$ thick, $\leq 15 \, \Omega$ square resistance) glass substrates were provided by South China Science & Technology Co. Ltd, while velvet fabric for rubbing orientation was obtained from Taobao. All materials were used as received without further treatment.

Fabrication of the oriented PCDTPT film

A 10 mg mL^{-1} chloroform solution of PCDTPT was stirred overnight at $60 \, ^\circ\text{C}$ under a nitrogen atmosphere to ensure complete dissolution. Substrate was subjected to ultrasonic cleaning for 20 min in ITO cleaner, deionized water, and anhydrous ethanol, sequentially, and then dried under nitrogen flow. Prior to spin-coating, substrates were treated in a vacuum oxygen plasma cleaner to remove surface residues. The film was then spin-coated at an acceleration of 1000 rpm s^{-1} and a rotation speed of 2000 rpm for 60 s within a nitrogen-filled glovebox. Following spin-coating, the PCDTPT film-coated substrate was annealed at $200 \, ^\circ\text{C}$ on a hot plate for 8 min to enhance crystallinity. The device was then placed in a fixed groove mold and annealed at $220 \, ^\circ\text{C}$ (slightly higher than the annealing temperature³⁸) for 1 min. Subsequently, the film was gently rubbed unidirectionally using velvet fabric for approximately 1 min. A final post-annealing treatment was performed for 3 min.

Table 1 Summary of reported device performances

Active layer material	Electrolyte material	Response wavelength	Responsivity	Reference
CdSe/CdS nanosheets	LiClO_4 : PEG	405 nm	10 mA W^{-1}	31
PDPP2T: PC ₆₁ BM	P(VDF-HFP): [EMIM][TFSI]	808 nm	$1.5 \times 10^3 \text{ A W}^{-1}$	32
PCDTBT	[EMIM][TFSI]	Sunlight	7.3 A W^{-1}	33
PCDTPT	[EMIM][TFSI]	885 nm	3.56 A W^{-1}	34
SEBS: PDPP2T: PC ₆₁ BM	P(VDF-HFP): [EMIM][TFSI]: LiTFSI	808 nm	$2 \times 10^3 \text{ A W}^{-1}$	35
DNTT	Indigo carmine: P4VP- <i>b</i> -PEO	450 nm	6.12 A W^{-1}	36

Device fabrication

Pre-deposited patterned ITO electrodes on glass substrate served as the source and drain electrodes, with a channel length (L) of 50 μm and width (W) of 1000 μm . The oriented PCDTPT film was fabricated as described. Device assembly was carried out in a nitrogen-filled glovebox. For testing, approximately 2.5 μL of [EMIM][TFSI] ionic liquid was applied atop the conducting channel as the gate electrolyte layer, ensuring full coverage of the channel with a large contact angle. A tungsten probe, inserted into the [EMIM][TFSI] droplet, was used as the gate electrode to apply gate bias voltage.

Characterization

Substrate surface cleaning was conducted using a vacuum oxygen plasma cleaning machine (TS-PL02, Shenzhen Tonson Tech Automation Equipment Co., Ltd). Film absorption spectra were measured with an Ocean Optics USB4000 spectrometer, using a DH-2000-BAL deuterium-halogen light source. Film surface morphology characterization was performed using atomic force microscopy (Cypher, VRS). Characterization of the organic polarization-sensitive electrochemical phototransistor was carried out using a semiconductor analyzer (Keithley 4200 SCS). A self-assembled coaxial incident light system was used, consisting of a near-infrared LED light source (Oeabt, LED-D1-850), a polarizer (Daheng Optics, GLC-051012), and an electrically driven rotation stage (LETEK, ICSM-RT1). Incident light intensity was calibrated with an OPHIR Photonics PD300-UV-ROHS power meter.

Results and discussion

The molecular structures of the materials are illustrated in Fig. 1a. PCDTPT was selected as the active layer material due to its exceptional light absorption and exciton dissociation properties, functioning effectively as a donor-acceptor copolymer. Its infrared response range also demonstrates significant potential for signal sensing applications. Additionally, PCDTPT functions as a P-type semiconductor with notable charge carrier mobility. Consequently, it combines efficient photoconversion capabilities with high charge transport, making it suitable as the active layer in the OECPT devices. [EMIM][TFSI], a commonly utilized ionic liquid, is capable of forming electric double layers at the semiconductor interface with a unit area capacitance of approximately 10 nF cm^{-2} .³³ This ionic liquid also exhibits high ionic conductivity, which enables rapid ion migration in response to electric field changes. Furthermore, [EMIM][TFSI] possesses excellent stability under ambient conditions, low volatility, and environmental compatibility, rendering it a suitable choice as the gate electrolyte material.

The polarized absorption spectra of PCDTPT films, before and after orientation treatment, are depicted in Fig. 1b and c, respectively. The non-oriented film exhibits optically isotropic properties, as reflected in its polarized absorption spectrum. Conversely, the oriented PCDTPT film demonstrates distinct absorption characteristics depending on the polarization state of the incident light. For incident light polarized parallel to the

rubbing direction (referred to as parallel light), higher absorbance is observed, whereas lower absorbance is recorded for incident light polarized perpendicular to this direction (referred to as perpendicular light). This phenomenon occurs because the randomly arranged polymer chain segments in the film attain a certain degree of order after uniaxial orientation. The film preferentially absorbs light with polarization parallel to the main chain, even surpassing the absorbance of unpolarized light. This intrinsic mechanism accounts for the polarization sensitivity observed in oriented organic polymer films. Following orientation treatment, the film's thickness is reduced due to the rubbing process, resulting in lower absorbance for unpolarized light compared to non-oriented films. The dichroic ratio (DR) is employed to quantify the film's optical anisotropy and is defined as follows:

$$\text{DR} = A_{\text{max}}/A_{\text{min}} \quad (1)$$

where A_{max} and A_{min} represent the film's maximum and minimum absorbance, respectively, for different polarization states of incident light. Specifically, in this work, they correspond to the absorbance of the oriented film for parallel and perpendicular light, respectively. The DR value of the oriented PCDTPT film is calculated according to eqn (1), with a comparison to the non-oriented film displayed in Fig. 1d. As illustrated, the non-oriented film maintains a DR value of 1 across the absorption spectrum, indicating its isotropic nature. In contrast, the oriented film displays a DR value of approximately 3 at 865 nm. Although the DR value remains elevated near 1000 nm, absorption in this wavelength range is weaker. Therefore, 850 nm was selected as the testing wavelength to ensure adequate light absorption by the film while maintaining a higher DR value.

Atomic force microscope (AFM) characterization was performed to further compare the morphology of the PCDTPT films before and after orientation. The results for the non-oriented film are shown in Fig. 2a. The film's surface is flat and free of significant structural defects, with a root mean square (RMS) surface roughness of only 3.402 nm. Overall, no ordered orientational arrangement is observed in the non-oriented films. In contrast, the oriented film demonstrates clear alignment of polymer chains, as coarse, fiber-like aggregates are evident in AFM images of the film's surfaces (Fig. 2b). These unidirectionally aligned polymer backbones contributed to anisotropic absorption in the films. However, it is noteworthy that the morphology of the oriented film was compromised due to the mechanically destructive nature of velvet friction orientation, which increases the RMS roughness to 23.985 nm and created an uneven film surface. This morphological change is observed to hinder both carrier transport and gate modulation within the film.

The OECPT device was designed with a top-gate-bottom-contact structure, as shown in Fig. 3a. Due to the limited dependence of the electrochemical working mechanism on electrode materials, pre-deposited patterned ITO (with a work function of 4.7 eV)³⁴ was employed as the materials for source and drain electrodes thus simplifying the fabrication process.



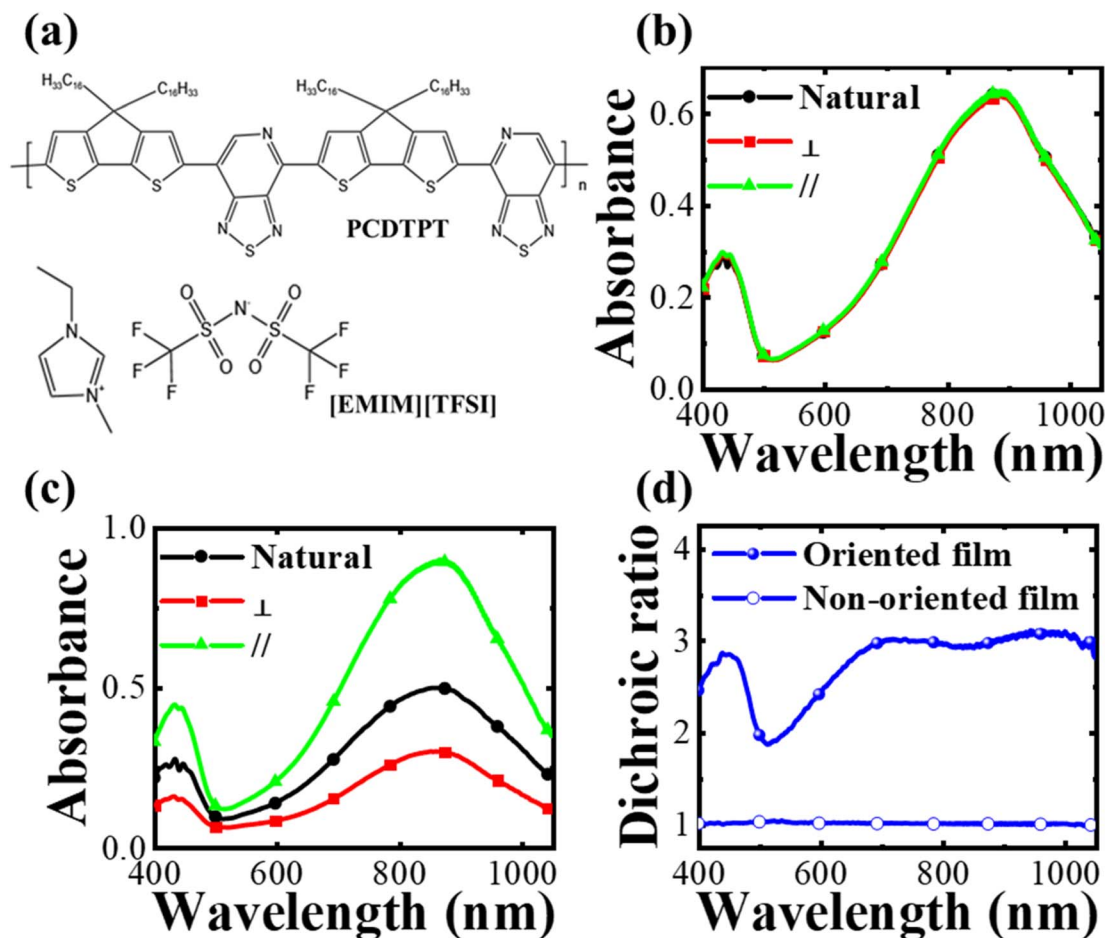


Fig. 1 (a) Molecular structures of PCDTPT and [EMIM][TFSI]. Absorption spectra of PCDTPT thin films (b) before and (c) after orientation treatment under varying polarization states of incident light, where “ \perp ” represents light polarized perpendicular to the film’s rubbing alignment direction, and “ \parallel ” represents light polarized parallel to the alignment direction. (d) Comparison of optical absorption dichroic ratio between oriented and non-oriented thin films.

The dimensions of the conducting channel are $W/L = 1000/50$ μm . The channel was fully covered by [EMIM][TFSI], ensuring effective gate modulation across the entire channel.

Subsequently, electrical characterization of the device under various illumination conditions was conducted in ambient atmosphere with a fixed source-drain voltage (V_{DS}) of -0.8 V.

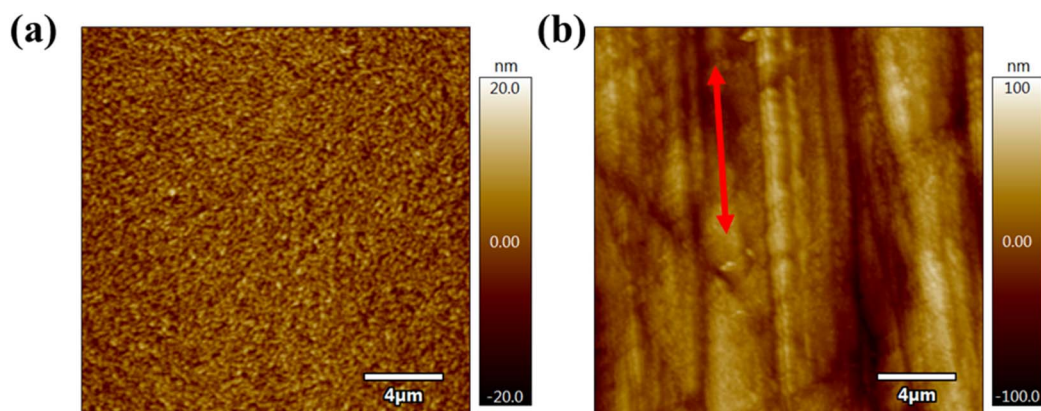


Fig. 2 Surface morphology of PCDTPT thin films observed using AFM: (a) before and (b) after orientation treatment by rubbing. The red arrow indicates the rubbing direction for orientation.

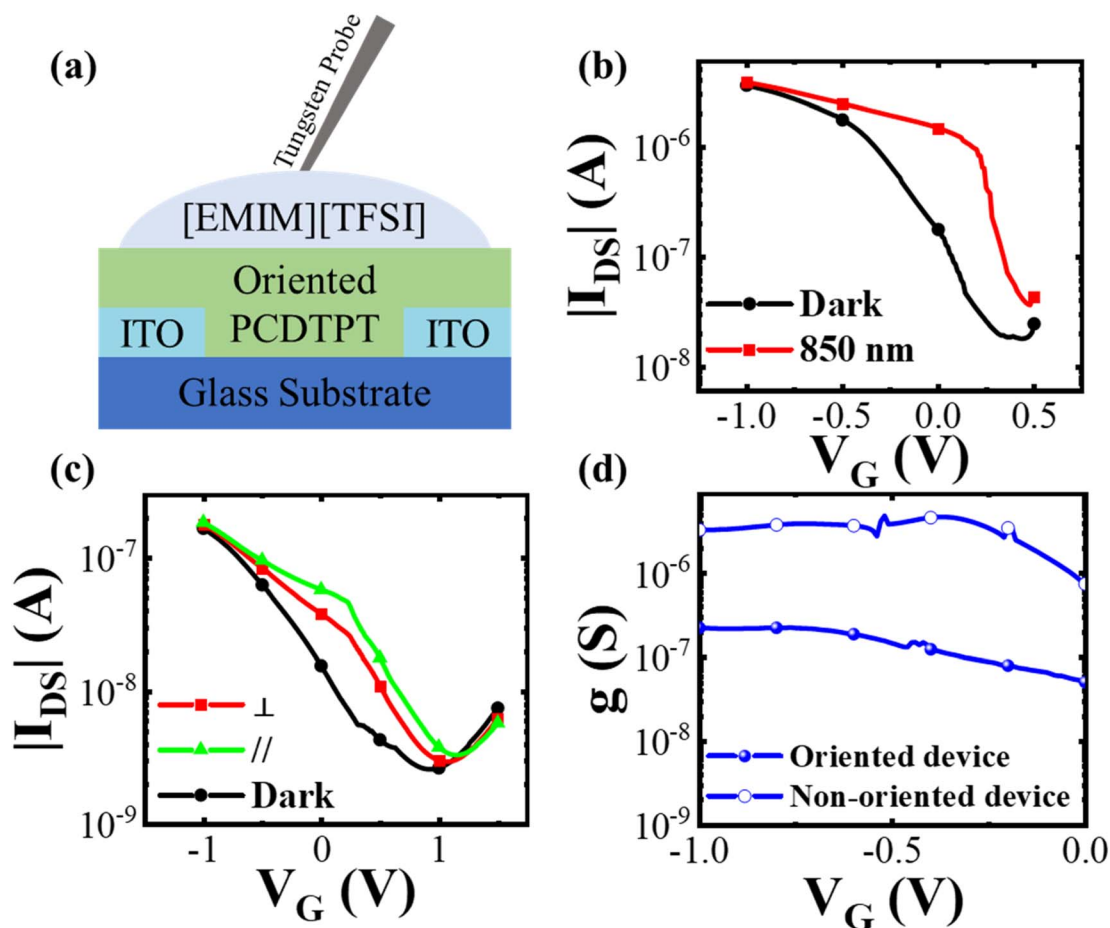


Fig. 3 (a) Schematic of the OECPT device structure. Transfer curves of (b) the non-oriented device and (c) the oriented device, with $V_{DS} = -0.8$ V under a constant 1 mW cm^{-2} incident light intensity at 850 nm. (d) Comparison of transconductances values between the two device types.

The transfer characteristics are compared in Fig. 3c. The device exhibits typical P-type transistor characteristics. In comparison to the non-oriented device (Fig. 3b), the oriented device shows a decrease in modulation performance, attributed to morphological damage in the film caused by the orientation process, which adversely impact charge transport. In OECT, maximum transconductance (g_m) is commonly used to evaluate device performance and is defined as follows:

$$g = \frac{\partial I_{DS}}{\partial V_G} \quad (2)$$

where I_{DS} is the channel current and V_G is the gate voltage. The calculated g_m values for the oriented and non-oriented PCDTPT devices are $2.25 \times 10^{-7} \text{ S}$ and $3.29 \times 10^{-6} \text{ S}$, respectively (Fig. 3d). The decrease in transconductance is attributed to the thermal rubbing orientation, which partially disrupted the film's surface morphology, introducing additional structural defects that diminish the gate modulation capability. This disruption also necessitates a larger positive gate voltage at the start of the scan to eliminate the residual defect charge from previous scans, thereby reducing the off-state current. It should be noted that the initial higher I_{DS} is attributed to ion currents arising from ion redistribution rather than electron currents.

The dichroic ratio of photocurrent, DR_{I_p} , is an optoelectronic parameter used to quantify the dichroism of photogenerated currents. This ratio can be derived from the transfer characteristics and is defined as:

$$DR_{I_p} = \frac{I_{//} - I_d}{I_{\perp} - I_d} \quad (3)$$

where I_d , $I_{//}$, and I_{\perp} represent the channel current in the dark state, under parallel light and perpendicular light illuminations, respectively. The resulting DR_{I_p} curve is shown in Fig. 4a. The maximum DR_{I_p} value, approximately 3.17, is observed in the off-state region of the device. As the gate voltage increased, the DR_{I_p} value gradually decreases in the on-state region. This decrease occurs because, under on-state conditions, the gate-modulated current becomes dominant in the channel, effectively eliminating the variation in photocurrent induced by different polarization states of incident light. Eventually, DR_{I_p} decreases to a value of 1.38. Responsivity (R) reflects the capacity of a photonic device to convert light signals into electrical signals, and it is defined as:

$$R = \frac{I_p}{P_{in} \times A} \quad (4)$$



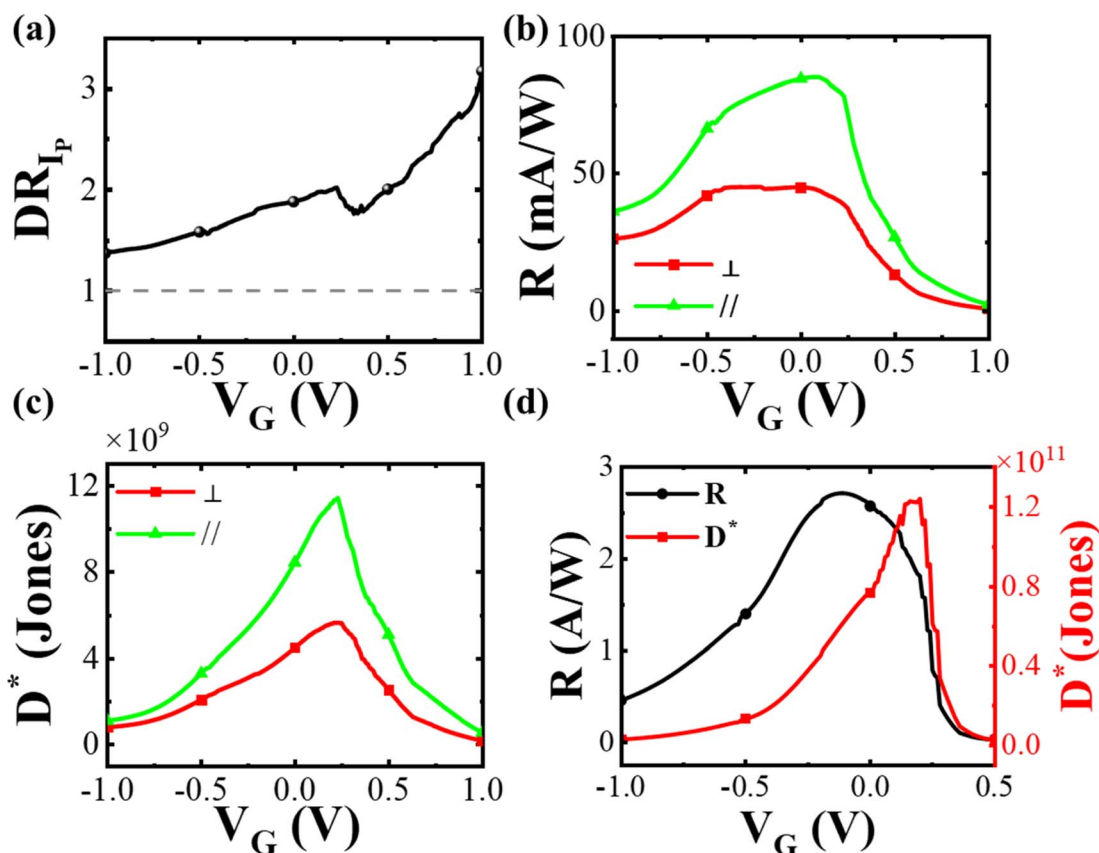


Fig. 4 (a) Photogenerated current dichroic ratio (DR_{I_p}), (b) responsivity (R), and (c) normalized detectivity (D^*) of the oriented device. (d) Corresponding optoelectronic performance parameters for the non-oriented device.

where I_p is the photocurrent (the difference between the light-state and dark-state currents), P_{in} is the incident light power density, and A represents the active area of the device. The relationship between device responsivity and gate voltage is depicted in Fig. 4b. It can be observed that the device exhibits higher responsivity under parallel light than under perpendicular light. The maximum responsivity for parallel light reaches 84.92 mA W^{-1} , while for perpendicular light, under identical voltage conditions, the responsivity is 44.12 mA W^{-1} . Normalized detectivity (D^*) serves as a photonic parameter to characterize the sensitivity of a device to weak light and can be calculated for phototransistor devices using the equation:

$$D^* = R \sqrt{\frac{A}{2eI_d}} \quad (5)$$

where R denotes the responsivity, A is the active area, e is the elementary charge (approximately $1.6 \times 10^{-19} \text{ C}$), and I_d is the dark channel current. The calculated values are shown in Fig. 4c. Under parallel light illumination, the maximum D^* value reaches $1.14 \times 10^{10} \text{ Jones}$, whereas under perpendicular light illumination, it reaches $5.66 \times 10^9 \text{ Jones}$. The data are consistent, confirming that the oriented device demonstrates the anticipated polarization-sensitive response. The corresponding parameters for the non-oriented device are shown in Fig. 4d. Table 2 summarizes these optoelectronic parameters.

To assess the performance stability of the devices, approximately 20 randomly selected devices, both oriented and non-oriented, were tested. The electrical parameters, such as threshold voltage, on-state current, off-state current, and the on/off ratio, were extracted from their transfer curves (ESI Fig. S1†). The results indicate that, following the friction orientation treatment, the threshold voltage of the devices exhibits a wider fluctuation range, and the overall on/off ratio decreases. This behavior is attributed to the disruption of the film morphology. A comparison of these results is provided in Table S1.† These findings are consistent with previous studies. For further investigation into device stability, the variation in current at steady-state voltage was also analyzed (ESI Fig. S2†).

Table 2 Summary of photonic parameters for both oriented and non-oriented devices

	Oriented device	Non-oriented device
I_{on}/I_{off}	6.43×10^1	1.99×10^2
V_{th}	0.45 V	0.21 V
g_m	$2.25 \times 10^{-7} \text{ S}$	$3.29 \times 10^{-6} \text{ S}$
R	84.92 mA W^{-1}	2.72 A W^{-1}
D^*	$1.14 \times 10^{10} \text{ Jones}$	$1.24 \times 10^{11} \text{ Jones}$
	$5.66 \times 10^9 \text{ Jones}$	

The data show that non-oriented devices exhibit more stable off-state currents, which remain nearly constant for over 500 seconds. However, the on-state current continues to increase without reaching saturation within a short period. This phenomenon is attributed to the unique electrochemical working mechanism of organic electrochemical transistors (OECTs). Over time, removable ions migrate into the channel, which continuously enhances channel conductance, thereby leading to a gradual increase in on-state current. In contrast, electrochemical processes in oriented devices are more susceptible to perturbations due to the altered film morphology, which leads to both on- and off-state currents increasing with time. While device stability has been demonstrated to some extent, there remains room for improvement, which we plan to address in future work.

Fig. 5(a) and (b) illustrate the device's operation in the dark. When a positive gate voltage is applied, the cations $[\text{EMIM}]^+$ in the ionic liquid are driven by the electric field toward the interface between the electrolyte and semiconductor layers. This induces electronic changes within the semiconductor layer, forming an electric double layer structure. The resulting

increase in localized electron concentration is accompanied by hole depletion, which reduces hole conductivity and hinders the transport of holes injected from the source. Consequently, the device exhibits a low I_{DS} under positive gate voltage.

As the gate voltage shifts to a more negative value, hole channels with higher hole concentration gradually form at the interface. This facilitates a high-conductance path for the injected holes, resulting in the device turning on and an increase in I_{DS} . Fig. 5(c) and (d) show the operating mechanism of the device under light. Upon illumination, photogenerated electrons and holes are created within the semiconductor layer. Some of these carriers contribute to channel formation, while others are transported as free charges. This leads to an enhancement in I_{DS} compared to the dark state, due to the additional carriers generated by the light. Notably, in the pre-on state, the device exhibits higher responsivity. This is because the photogenerated current is suppressed under positive gate voltage, while at negative gate voltage, the channel current is primarily dominated by the injected current due to gate amplification. As a result, the device achieves maximum responsivity around $V_{\text{DS}} = 0$ V.

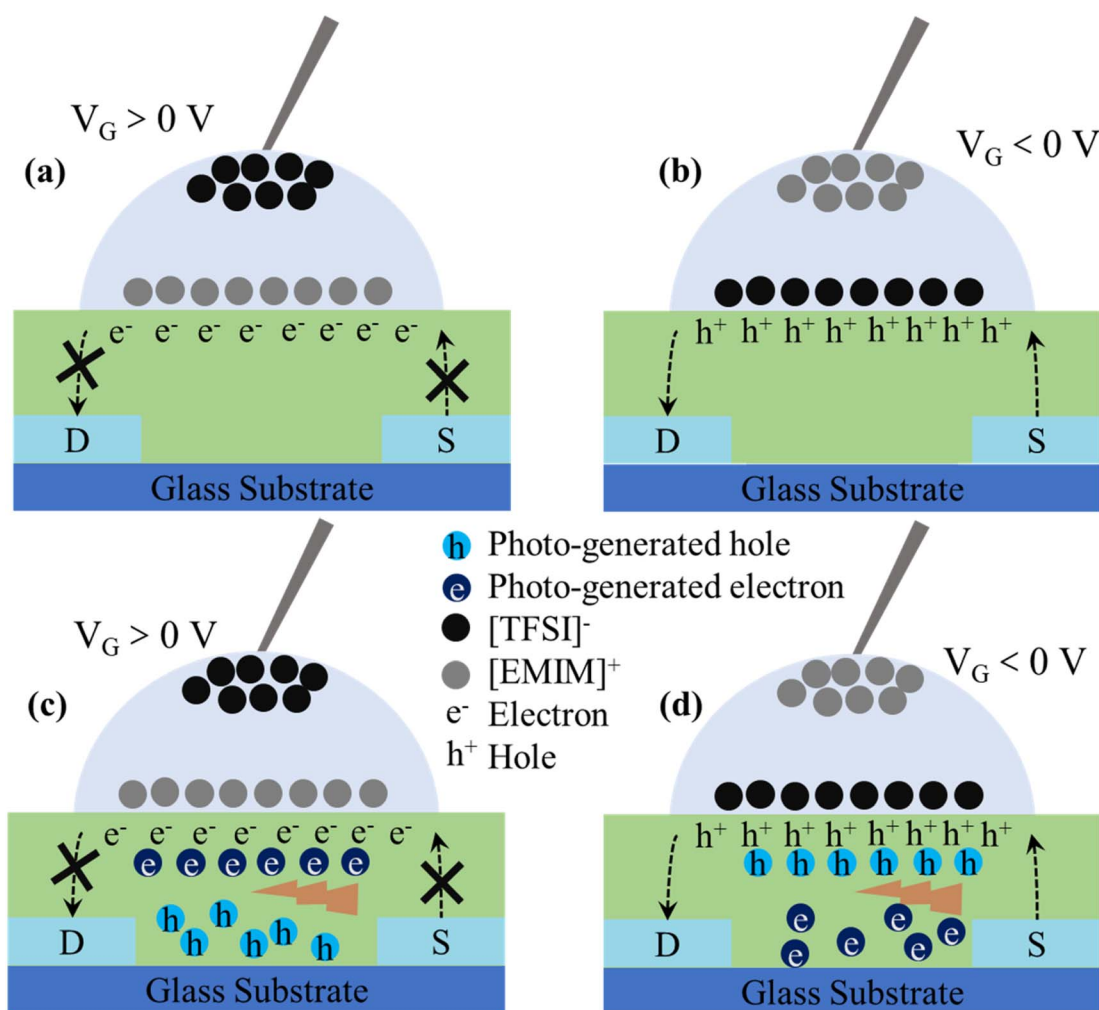


Fig. 5 Schematic of the device operation mechanism. (a) and (b) Are in the dark conditions. (c) and (d) Are in the light condition.



The dependence of the device's response on the incident light polarization angle was evaluated under constant bias conditions. In electrochemical transistors, unavoidable ion permeation typically occurs during operation, wherein ions diffuse into the active layer driven by a concentration gradient, thereby initiating a series of electrochemical redox reactions that generally increase the channel conductivity. Conversely, the on-state voltage facilitates ion migration into the active layer, enhancing conductivity and causing the channel current to increase over time, a factor that is not conducive to polarization detection by the device. Consequently, the operating region for electrochemical phototransistors is typically confined to the off-state region of the device. For polarized light response testing in this work, a constant bias voltage of $V_{DS} = -0.8$ V and $V_G = 1$ V was applied to the device. The active area was illuminated with incident light at 850 nm, with an intensity of 1 mW cm^{-2} . The resulting current signals under parallel and perpendicular polarization of light, shown in Fig. 6a, reveal a light response in the nA range, with significant differences observed between responses to parallel and perpendicular light. The photocurrent dichroic ratio DR_p under these two illumination conditions is approximately 1.52, which is slightly lower than the value derived from transfer curves at the same voltage. This outcome is attributed to the suppression of the total current flowing through the channel by the off-state gate voltage, which, although it stabilizes the light response under

the constant off-state bias, results in a smaller overall light response and a decrease in DR_p value.

The dependence of the device's response on the polarization angle of incident light is illustrated in Fig. 6b, where an electrically-driven rotating polarizer was employed to modulate the light's polarization angle. Real-time monitoring of current changes under constant bias conditions shows results consistent with Malus' law, as displayed in the polar coordinate plot in Fig. 6c. Oriented PCDTPT thin film based OECPT devices demonstrate a notable polarization response. Typically, the photoresponse of phototransistors is tested in the off-state region of the device. In the on-state voltage region, the channel current is primarily an injection current that is amplified by gate modulation. This is particularly true for transistors based on electrochemical mechanisms, where the current in the on-state region also increases over time. This increase occurs because the OECT device undergoes additional doping reactions, leading to an increase in channel conductance when it remains in the on-state for an extended period. A more stable optical response is achieved in the off-state. Applying a larger reverse gate voltage effectively releases trapped charge, allowing the device to return to its initial state as much as possible. Moreover, the photoresponse of the device exhibits different characteristics at various gate voltages. At larger reverse gate voltages, the channel current significantly decreases, which suppresses the photoresponse. However, this

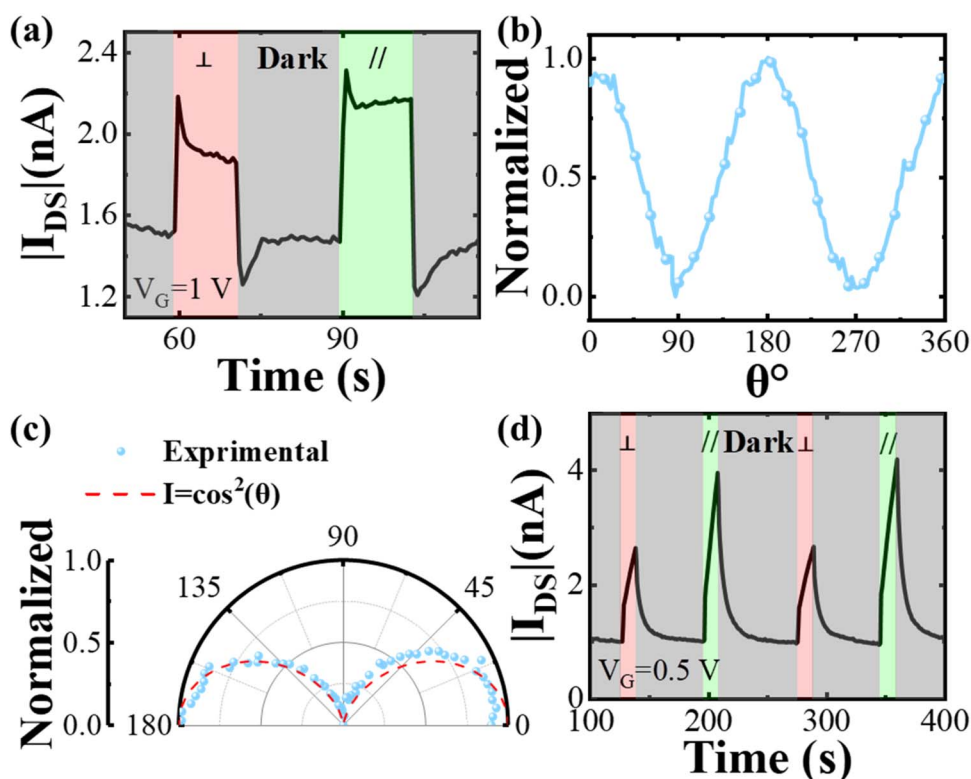


Fig. 6 (a) Device performance under constant bias ($V_{DS} = -0.8$ V, $V_G = 1$ V) with varied illumination conditions. (b) Dynamic response of device current to continuously changing incident light polarization angles under constant bias. (c) Polar coordinate plot of the response, with blue dots representing experimental data and the red dashed line indicating the theoretical curve. θ denotes the angle between the incident light polarization and the device orientation. (d) Polarized light response curve with bias condition adjusted to $V_G = 0.5$ V.

also results in an enhanced response speed, as the larger reverse gate voltage helps release residual charges. In contrast, a smaller reverse gate voltage leads to less suppression of the channel currents, resulting in larger photoresponses, but this also increases the device's response time. The polarization light response under a gate voltage of 0.5 V, illustrated in Fig. 6d, demonstrates a larger response current that does not saturate quickly, unlike the results shown in Fig. 6a. However, this increased response comes with the trade-off of a prolonged "erase" time required to restore the current to its baseline level upon the removal of illumination. In summary, the optical response of OECPT devices shows a clear dependence on gate voltage. These non-volatile properties of OECPT are valuable for simulating biological synaptic behavior and for the advancement of multidimensional optoelectronic storage devices.

Conclusions

This study confirms the effectiveness of the thermal rubbing orientation method in inducing anisotropy in organic polymer films, which, when combined with ionic liquid gate modulation, enables the creation of polarization-sensitive organic electrochemical phototransistors. The device demonstrates a notable polarization response to near-infrared light, achieving a maximum responsivity of 84.92 mA W^{-1} for parallel-polarized light and 44.12 mA W^{-1} for perpendicularly polarized light under identical voltage conditions. Under continuous bias testing, a photocurrent dichroic ratio of 1.52 was achieved, highlighting the device's sensitivity to polarization. The device's polarization response was found to be dependent on gate voltage, with device non-volatility modifiable through variations in gate voltage conditions. This work holds promise for advancements in areas such as biometric monitoring, neural synapse emulation, and optoelectronic memory storage, offering valuable insights for the development of multifunctional photodetectors.

Data availability

The data supporting the findings of this study are available within the article and its ESI.†

Conflicts of interest

The authors declare no conflict of interest.

Acknowledgements

The authors acknowledge the financial support of Fundamental Research Funds for the Central Universities (2023JBZY001); the National Natural Science Foundation of China (No.62275013, 62234004, 12274020, 62075009); Beijing Natural Science Foundation (IS23068).

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