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# Mass production of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles for glutathione responsive off-on T<sub>1</sub>/T<sub>2</sub> switching magnetic resonance imaging and tumor theranostics

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Individual theranostics with an integrated multifunction holds considerable promise for clinical application compared with multicomponent regimes. Mn<sub>3</sub>O<sub>4</sub> nanoparticles with an ultrasmall size (4 nm) and mass production capability were developed with dual function of integrated tumor magnetic resonance imaging (MRI) and therapy. The high valence state of Mn<sub>3</sub>O<sub>4</sub> nanocrystals enables a sensitive reaction with the glutathione (GSH) molecule and favorable decomposition ability, which further induces a unique, favorable, variable T<sub>1</sub> turn-off and T<sub>2</sub> turn-on MRI property. In addition, ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles reacted with high-level GSH in the tumor microenvironment induces responsive and enhanced variable T<sub>1</sub>- and T<sub>2</sub>-MRI imaging capability for accurate cancer diagnosis. Moreover, the synthesized ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles exhibit considerable ferroptosis effect towards tumor cells and excellent *in vivo* biocompatibility, thus indicating promising effective cancer treatment application. The developed ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles with integrated dual functions of GSH-responsive variable T<sub>1</sub> and T<sub>2</sub> MRI imaging effects and ferroptosis capability show promising potential as a candidate for tumor theranostics in clinical applications.

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## 1. Introduction

The current epidemiology of cancer shows that the incidence rate and mortality rate of cancer are increasing annually.<sup>1,2</sup> The accurate diagnosis of cancer, especially in the early stage, is of great importance for its treatment.<sup>3</sup> Among the various diagnostic techniques, MRI has been recognized as one of the most powerful medical diagnostic tools with comprehensive and detailed detection ability in the clinic.<sup>4,5</sup> MRI is proposed as an effective noninvasive method with high spatial resolution and real-time monitoring capability, which can reflect whole tumor biological features and account for accurate diagnosis and treatment.<sup>6,7</sup> In addition, numerous recent preclinical studies report that the combination of MRI detection and tumor therapy holds great promise for enhancing both cancer treatment efficiency and effectiveness.<sup>8–10</sup> Although MRI tests

possess huge advantages, their sensitivity and resolution ratio for differentiating various diseases in some cases remains to be improved using a well-designed contrast agent.<sup>11,12</sup> Therefore, developing advanced MRI contrast agents integrated with high cytotoxicity towards tumor cells has become one of the most urgent tasks for improving the efficiency of synergistic MRI diagnosis and cancer treatment.<sup>13–15</sup>

Manganese-based nanomaterials, a hot research target in recent years, have aroused considerable interest from researchers owing to their unique characteristic and broad application in biology and medicine, especially for tumor MRI theranostics.<sup>16–18</sup> According to the MRI principle, MRI contrast agents are generally classified into two types: positive and negative signal improvements, which are T<sub>1</sub>- and T<sub>2</sub>-weighted MRI, respectively.<sup>19,20</sup> Moreover, the integration of T<sub>1</sub>- and T<sub>2</sub>-mode MRI in one operation in the clinic could immensely promote diagnosis accuracy and efficiency.<sup>21</sup> Previous research has shown that Mn-based nanomaterials including MnO, Mn<sub>3</sub>O<sub>4</sub> and MnO<sub>2</sub> possess an ideal T<sub>1</sub>-mode MRI enhancement ability, thus enabling the development of a competitive alternative contrast agent.<sup>22–24</sup> Nevertheless, the development strategy for Mn-based nanomaterials for T<sub>1</sub> and T<sub>2</sub> dual-mode MRI mainly relies on combination with the T<sub>2</sub> MRI contrast agent of Fe<sub>3</sub>O<sub>4</sub>, which requires a rigorous process and extensive work. The translation of an Mn-based T<sub>1</sub> and T<sub>2</sub> dual-mode MRI

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agent to clinical application is challenging as there exist many obstacles that need to be overcome, including scaling up the production, improving stability and exporting the standard.<sup>25</sup> Furthermore, Mn-based nanomaterials have been utilized for tumor treatment for their effective antitumor effect, negligible immunogenicity, and excellent MRI contrast enhancement.<sup>26,27</sup> The tumor microenvironment is usually characterized by a high level of glutathione (GSH) and H<sub>2</sub>O<sub>2</sub>. Based on these features, the development of Mn-based nanomaterials as tumor microenvironment responsive nanoagents for both tumor T<sub>1</sub>/T<sub>2</sub> MRI and therapy can help enhance cancer treatment and promote their clinical translation.<sup>28,29</sup>

Herein, we well designed advanced theranostic ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles that possess tumor microenvironment responsive off-on T<sub>1</sub>/T<sub>2</sub> switching MRI imaging capability and an effective ferroptosis therapy effect. The synthesized ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles are about 4 nm in size with excellent uniformity and mass production ability. The high valence state Mn in Mn<sub>3</sub>O<sub>4</sub> nanocrystals offer the effective glutathione (GSH) responsive ability and variable MRI imaging property. The sensitive reaction of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles with a GSH molecule enable the functional T<sub>1</sub> turn-off to T<sub>2</sub> turn-on MRI switching effect, which furnish better signal quality and diagnose accuracy in an *in vitro* and *in vivo* environment. Moreover, ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles could effectively induce the ferroptosis effect to tumor cells with favorable *in vivo* biocompatibility. The finely developed individual ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles with an integrated responsive MRI effect and ferroptosis function make it an ideal candidate for tumor theranostics in the clinic.

## 2. Experimental methods

### 2.1 Mass production and modification of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles

Oleylamine, oleyl alcohol, dimercaptosuccinic acid (DMSA), manganese(II) acetate tetrahydrate and anhydrous sodium carbonate were purchased from Shanghai Macklin Biochemical Co. Ltd Ethanol, tetrahydrofuran and *n*-hexane were purchased from Sinopharm Chemical Reagent Co. Ltd

Ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles were synthesized according to previous research with some modifications.<sup>30</sup> Briefly, 60 g oleylamine, 100 g oleyl alcohol and 300 mL 1-octadecene were mixed with 4.9 g manganese(II) acetate tetrahydrate. The mixture was then homogenized and slowly heated to 100 °C. Thereafter, 2 mL manganese acetate tetrahydrate aqueous (0.1 M) was added to the solution. The reaction was maintained for 30 min and then cooled to room temperature. Mn<sub>3</sub>O<sub>4</sub> nanoparticles were obtained by adequately washing the solution with ethanol and *n*-hexane.

Ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles were modified with the DMSA molecule to endow better dispersity in water. Briefly, 10 mg of the obtained Mn<sub>3</sub>O<sub>4</sub> nanoparticles were added to 1 mL tetrahydrofuran, and 10 mg DMSA was added to 1 mL anhydrous sodium carbonate aqueous. The two solutions were well mixed and applied for ultrasonic treatment for 30 min. Mn<sub>3</sub>O<sub>4</sub>-DMSA was purified *via* centrifugation and the freeze-drying process.

### 2.2 Characterization

The size and morphology of the synthesized nanomaterials were detected using TEM and HRTEM (JEM-2100, JEOL, Tokyo, Japan). The structure information of the nanomaterials was tested using XRD (D8 Advance, Bruker, Ettlingen, Germany). The atomic valence and element composition information were characterized through XPS. FTIR spectroscopy (Nicolet Nexus 670, Thermo Fisher Scientific, Inc., Waltham, MA) was used to characterize the functional group of nanomaterials. Size distribution and zeta potential value were tested using a Malvern Zetasizer Nano Series. The absorbance spectrum of nanomaterials was characterized using a UV-vis spectrophotometer (UV-6100, Meipuda, Xi'an, China). The MRI property was tested using a 3.0-T clinical MRI scanner (GE Signa HDx 3.0 T MRI, USA) in a 16-channel brain coil.

### 2.3 Cells and animals

Mouse 4T1 cells were obtained from the Institute of Biochemistry and Cell Biology (Shanghai, China). Experiment mice (BALB/c, male) were acquired from Jinan Peng Yue Laboratory Animal Co., Ltd, Jinan, China. The 4T1 cells and mice were cultured as per manufacturer instructions. All mouse research has the approbation of Laboratory Animal Ethical and Welfare Committee of Shandong University, Cheeloo College of Medicine, China (accreditation number: SYXK:20190005) and was performed in line with the guidelines.

The subcutaneous tumor-bearing model was constructed by injecting 1 × 10<sup>6</sup> 4T1 cells in the right hind leg of BALB/c male mice (6–8 week-old). The liver metastatic tumor model was constructed by injecting 5 × 10<sup>4</sup> 4T1 cells in 2 mL normal saline into the tail vein of mice.

### 2.4 MRI property measurement

The T<sub>1</sub> and T<sub>2</sub> MRI imaging and relaxivity property of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles (6.25, 12.5, 25, 50, 100, and 200 μg mL<sup>-1</sup>) were tested using a 3.0-T MR scanner. Furthermore, 5 mM and 10 mM GSH were reacted with ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles (6.25, 12.5, 25, 50, 100, and 200 μg mL<sup>-1</sup>), and MRI imaging and relaxivity were recorded using the same method. The T<sub>1</sub> and T<sub>2</sub> MRI imaging abilities of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles for cancer cells (4T1) were assessed by incubating 4T1 cells with 50 μg mL<sup>-1</sup> Mn<sub>3</sub>O<sub>4</sub> nanoparticles for 2 h, 4 h, 8 h, and 12 h. The treated 4T1 cells were resuspended in 1 mL of the cell culture medium and characterized using an MRI scanner.

The *in vivo* MRI imaging effect of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles was verified in both the 4T1 subcutaneous tumor and liver metastatic tumor models. The tumor-bearing mouse was treated with 5 mg kg<sup>-1</sup> ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles through intravenous injection and then examined using an MRI scanner; the obtained images and signal intensity were recorded at the time points of 0, 2 and 4 h.

### 2.5 Cell viability assay of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles

The cell viability of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles was tested using live/dead staining. 4T1 cells were cocultured with Mn<sub>3</sub>O<sub>4</sub>



nanoparticles (6.25, 12.5, 25, and 50  $\mu\text{g mL}^{-1}$ ) for 24 h and then stained with a live/dead reagent, and the cells were observed using an inverted fluorescence microscope. The cytotoxicity of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles was assayed using CCK-8 and flow cytometry, and 4T1 cells were cocultured with  $\text{Mn}_3\text{O}_4$  nanoparticles (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) for 24 h and then detected using a CCK-8 kit as well as counting the live and dead cells using flow cytometry. The cleaved caspase-3 level in 4T1 cells after incubation with  $\text{Mn}_3\text{O}_4$  nanoparticles (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) for 24 h was measured using western blot analysis.

## 2.6 Ferroptosis therapy measurement

The 4T1 cells were cocultured with  $\text{Mn}_3\text{O}_4$  nanoparticles (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) for 24 h and stained using a ferroptosis marker. Dichlorofluorescein diacetate (DCFH-DA, 10  $\mu\text{M}$ ) was utilized to detect the intracellular ROS level *via* incubation with the 4T1 cells for 20 min. The lipid ROS sensor C11 BODIPY<sup>581/591</sup> (8  $\mu\text{M}$ ) was used to assess the lipid peroxidation level in 4T1 cells, and the treated cells were observed using fluorescence microscopy. The GPX4 level in 4T1 cells after incubation with  $\text{Mn}_3\text{O}_4$  nanoparticles was measured using western blot analysis.

## 2.7 Biosafety evaluation

The *in vivo* biocompatibility of  $\text{Mn}_3\text{O}_4$  nanoparticles was assayed before additional MRI and treatment application. The hemolytic property of  $\text{Mn}_3\text{O}_4$  nanoparticles (6.25, 12.5, 25, 50,

and 100  $\mu\text{g mL}^{-1}$ , experiment group) was tested by comparing the hemolysis rate in the group of isolated red blood cells and the group treated with PBS (positive control) and pure water (negative control). The primary organs of mice after injection with  $\text{Mn}_3\text{O}_4$  nanoparticles for 24 h were stained with hematoxylin-eosin (HE), and tissue slicing was observed using fluorescence microscopy.

## 2.8 Statistical analysis

The data were shown as the average  $\pm$  standard deviation. Comparison between two group statistics was performed using Student's *t*-test. Comparison in different group statistics was used with the analysis of variance, and the data were defined with \**p* < 0.05, \*\**p* < 0.01 and \*\*\**p* < 0.001.

# 3. Results and discussion

## 3.1 Synthesis and characterizations of ultrasmall $\text{Mn}_3\text{O}_4$ nanoparticles

Fig. 1a shows the mass production procedure scheme of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. The size and morphology of the synthesized  $\text{Mn}_3\text{O}_4$  nanoparticles were characterized using TEM (Fig. 1b). The TEM result shows that the  $\text{Mn}_3\text{O}_4$  nanoparticles are about 4 nm with an ultrasmall size and uniform shape. The enlarged HRTEM image (Fig. 1c) indicates that the obtained ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles are of good crystallinity. After modification with the DMSA molecule, the  $\text{Mn}_3\text{O}_4$

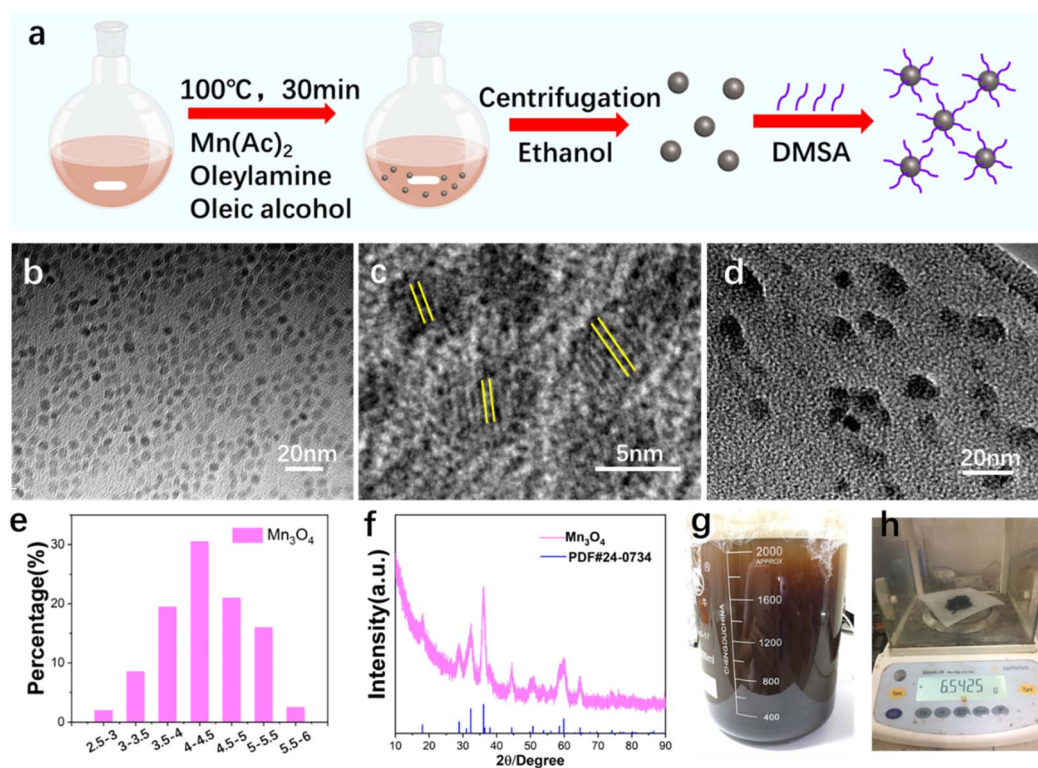


Fig. 1 Synthesis and characterization of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (a) The synthesis scheme of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (b) TEM, (c) HRTEM image of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (d) HRTEM image of  $\text{Mn}_3\text{O}_4$ -DMSA nanoparticles. (e) Size distribution statistic and (f) XRD pattern of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (g) Washing procedure and (h) mass production of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles.



nanoparticles retain their original size and morphology (Fig. 1d). The size distribution statistic of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles was analyzed (Fig. 1e), and the result indicates that the size is around 2–6 nm, and the main size is about 4.5 nm. The XRD pattern (Fig. 1f) of the synthesized nanoparticles confirms that nanoparticles are of excellent crystallinity and correspond to the pure  $\text{Mn}_3\text{O}_4$  phase (standard PDF card: 24-0734). The washing procedure (Fig. 1g) and final production of 6.54 g of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles (Fig. 1h) confirm their large-scale production and clinical translation potential.

The DMSA molecule modification endows ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles with better dispersity, which is essential for their bioapplications. XPS was utilized to analyze the Mn valence state of  $\text{Mn}_3\text{O}_4$  (Fig. 2a) and  $\text{Mn}_3\text{O}_4$ -DMSA (Fig. 2b). XPS spectrum results confirm the existence of  $\text{Mn}^{2+}$  and  $\text{Mn}^{3+}$  in  $\text{Mn}_3\text{O}_4$  and  $\text{Mn}_3\text{O}_4$ -DMSA, which agrees with the element composition. The DMSA molecule modification of  $\text{Mn}_3\text{O}_4$  was further studied using the FTIR spectrum (Fig. 2c). The absorption peaks of  $\text{Mn}_3\text{O}_4$  located at 2981.7 and 2849.6  $\text{cm}^{-1}$  correspond to N–H bond stretching vibration, which indicates the existence of the residual oleylamine molecule. For  $\text{Mn}_3\text{O}_4$ -DMSA, the peaks of N–H vibration disappeared, suggesting the successful

substitution of the DMSA molecule with the oleylamine molecule on the surface of  $\text{Mn}_3\text{O}_4$  nanoparticles. The synthesized  $\text{Mn}_3\text{O}_4$ -DMSA nanoparticles show excellent dispersity in water (Fig. 2d, inside) and intense absorbance around 300–600  $\text{cm}^{-1}$  (Fig. 2d). The hydrated particle size (%number) of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS (Fig. 2e) shows a normal distribution for 10–50 nm and is mainly located at 20 nm, which is beneficial for its utilization as a bioagent *in vivo*. The zeta potential values of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS (pH = 7, pH = 6.5 and pH = 5.5) are as high as  $-41.7 \pm 6.9$  mV,  $-23.3 \pm 4.57$  mV and  $-28 \pm 4.51$  mV (Fig. 2f), indicating the nanoparticles are of excellent dispersity and stability in aqueous PBS. The size distribution (%intensity) of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS in Fig. 2g reveals that most  $\text{Mn}_3\text{O}_4$  nanoparticles are monodispersed and there are few aggregates in the aqueous solution. The size distribution (%number) of  $\text{Mn}_3\text{O}_4$ -DMSA after long-term stability placement (2 weeks) in PBS (Fig. 2h) and PBS + BSA (Fig. 2i) conditions shows similar distribution tendency with small size differences. The hydrated particle size distribution and zeta potential values of  $\text{Mn}_3\text{O}_4$ -DMSA in different conditions demonstrate its promising bioapplication *in vivo*.

The MRI imaging property of  $\text{Mn}_3\text{O}_4$  nanoparticles *in vitro* was researched in detail. As shown in Fig. 3a, the phantom

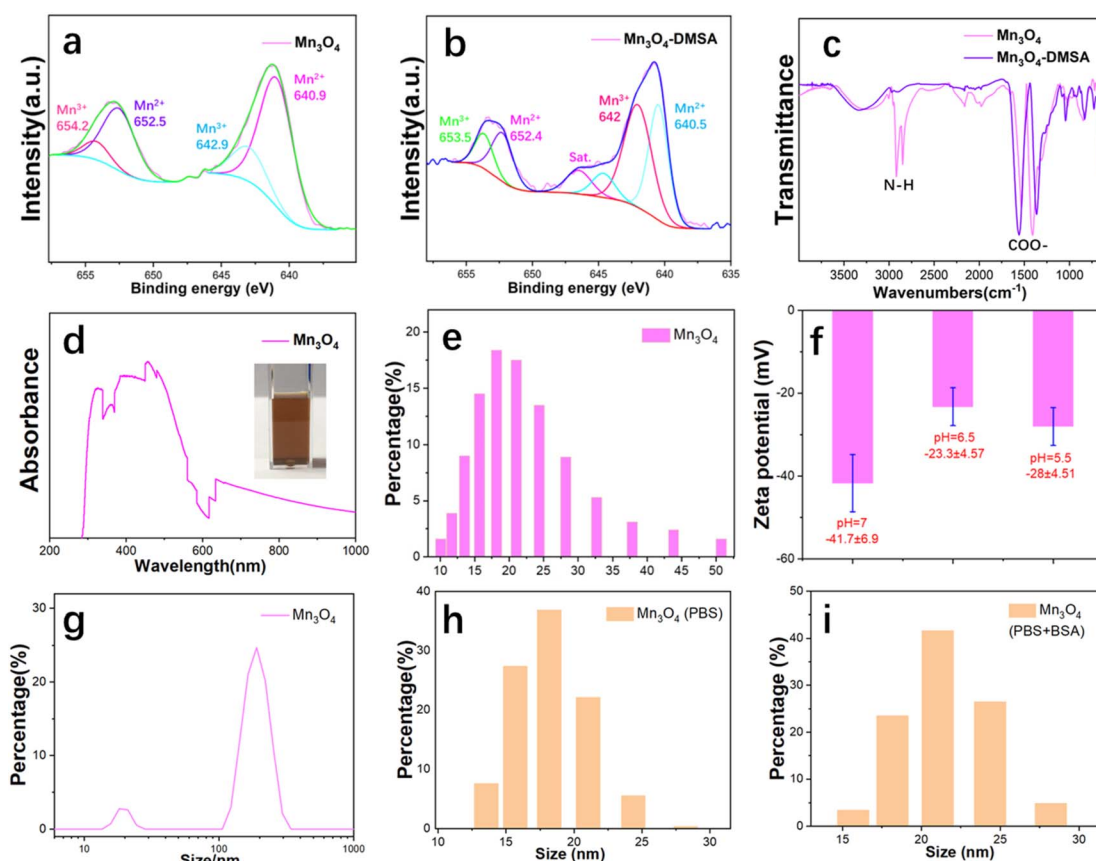
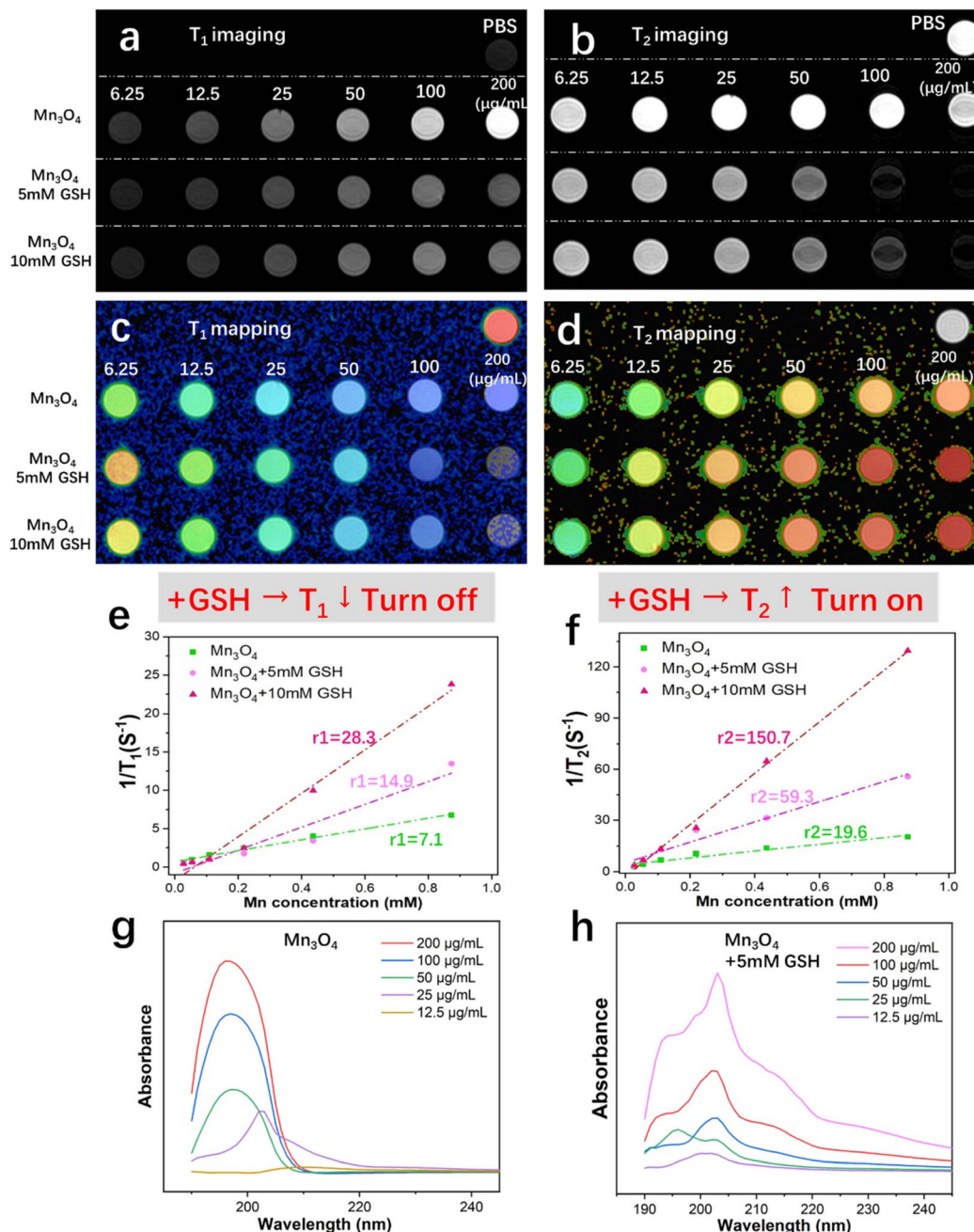


Fig. 2 Characterization of ultrasmall  $\text{Mn}_3\text{O}_4$  and  $\text{Mn}_3\text{O}_4$ -DMSA nanoparticles. (a) XPS pattern of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (b) XPS pattern of  $\text{Mn}_3\text{O}_4$ -DMSA nanoparticles. (c) FTIR pattern of  $\text{Mn}_3\text{O}_4$  and  $\text{Mn}_3\text{O}_4$ -DMSA nanoparticles. (d) UV-vis absorbance of  $\text{Mn}_3\text{O}_4$ -DMSA aqueous and the real image in water (inset). (e) Size distribution (%number) of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS. (f) Zeta potential values of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS with different pH. (g) Size distribution (%intensity) of  $\text{Mn}_3\text{O}_4$ -DMSA in PBS. Size distribution (%number) of  $\text{Mn}_3\text{O}_4$ -DMSA after long-term stability placement (2 weeks) in (h) PBS and (i) PBS + BSA condition.





**Fig. 3** GSH-responsive MRI variation measurement. (a)  $T_1$  and (b)  $T_2$  MRI imaging of  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, 25, 50, 100, and 200  $\mu\text{g mL}^{-1}$ ) as well as the MRI image after reaction with 5 mM and 10 mM GSH. (c)  $T_1$  and (d)  $T_2$  MRI mapping image of  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, 25, 50, 100, and 200  $\mu\text{g mL}^{-1}$ ) as well as the MRI mapping image after reaction with 5 mM and 10 mM GSH. (e) The  $r_1$  and (f)  $r_2$  values of  $\text{Mn}_3\text{O}_4$  after reaction with 0, 5 and 10 mM GSH. (g) UV-vis absorbance of a gratitude concentration of  $\text{Mn}_3\text{O}_4$ . (h) UV-vis absorbance of gratitude concentration  $\text{Mn}_3\text{O}_4$  after reaction with 5 mM GSH.

image of  $T_1$ -weighted imaging shows an obvious increased lightness as the concentration of  $\text{Mn}_3\text{O}_4$  increased from 6.25 to 200  $\mu\text{g mL}^{-1}$ , presenting a typical  $T_1$  MRI imaging characteristic. When  $\text{Mn}_3\text{O}_4$  was reacted with 5 mM GSH, the phantom image lightness shows a slight increased tendency with increasing concentration and darker intensity compared with the untreated group (pure  $\text{Mn}_3\text{O}_4$ ). As the treatment concentration of GSH increased to 10 mM, the phantom image lightness becomes darker and weaker compared with the treatment

of 5 mM GSH, showing a GSH-responsive  $T_1$  turn-off effect. The phantom image of  $T_2$ -weighted imaging is shown in Fig. 3b. The results show that pure  $\text{Mn}_3\text{O}_4$  has almost no  $T_2$  weight-imaging effect, indicating that  $\text{Mn}_3\text{O}_4$  is an effective and pure  $T_1$  MRI contrast agent and the state of  $T_2$  MRI imaging is off. After treatment with 5 mM and 10 mM, the darkness of  $T_2$  imaging becomes deeper as the concentration of GSH increased, indicating a sensitive GSH responsive  $T_2$  turn-on property. To summarize, ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles possess a unique



and favorable GSH responsive capability of  $T_1$  turn-off and  $T_2$  turn-on the MRI switching effect.

The  $T_1$  and  $T_2$  relaxation times of  $Mn_3O_4$  nanoparticles treated with the gradient concentration of GSH were measured, and the results were shown as  $T_1$  and  $T_2$  mapping (Fig. 3c and d). The corresponding  $r_1$  and  $r_2$  relaxivity values of  $T_1$  and  $T_2$  relaxation times were calculated from the plots of relaxation time *versus*  $Mn_3O_4$  concentration (Fig. 3e and f). The  $r_1$  value of  $T_1$  MRI imaging in the  $Mn_3O_4$  group (6.25–200  $\mu\text{g mL}^{-1}$ ) is 7.1  $\text{mM}^{-1} \text{s}^{-1}$ ; with the reaction of 5 mM and 10 mM GSH, the  $r_1$  values increase to 14.9 and 28.3  $\text{mM}^{-1} \text{s}^{-1}$ , respectively. The  $r_2$  value of  $T_2$  MRI imaging for the  $Mn_3O_4$  group (6.25 to 200  $\mu\text{g mL}^{-1}$ ) is 19.6  $\text{mM}^{-1} \text{s}^{-1}$ ; after treatment with 5 mM and 10 mM GSH,  $r_2$  values increase to 59.3 and 150.7  $\text{mM}^{-1} \text{s}^{-1}$ , respectively. The  $T_1$  and  $T_2$  MRI mapping and relaxivity values indicate the GSH-responsive  $T_1$  turn-off and  $T_2$  turn-on property. As  $Mn_3O_4$  nanoparticles show a brown color and possess obvious

light absorption, the UV-vis absorbance spectrum of  $Mn_3O_4$  nanoparticles at different concentrations (12.5–200  $\mu\text{g mL}^{-1}$ ) was obtained (Fig. 3g). Absorption intensity during 180–200 nm increases with an increasing in  $Mn_3O_4$  concentration. Fig. 3h shows the UV-vis absorbance spectrum of  $Mn_3O_4$  nanoparticles at different concentrations (12.5–200  $\mu\text{g mL}^{-1}$ ) after reaction with 5 mM GSH. These results confirm the intense reaction between  $Mn_3O_4$  nanoparticles and the GSH molecule as well as MRI variation evidence.

### 3.2 Cell viability and ferroptosis therapy of ultrasmall $Mn_3O_4$ nanoparticles

Before ultrasmall  $Mn_3O_4$  nanoparticles were used as nano-medicine *in vivo*, their cell viability was assessed. The living/dead staining images of 4T1 cells incubated with 6.25, 12.5, 25 and 50  $\mu\text{g mL}^{-1}$   $Mn_3O_4$  are shown in Fig. 4a. As the

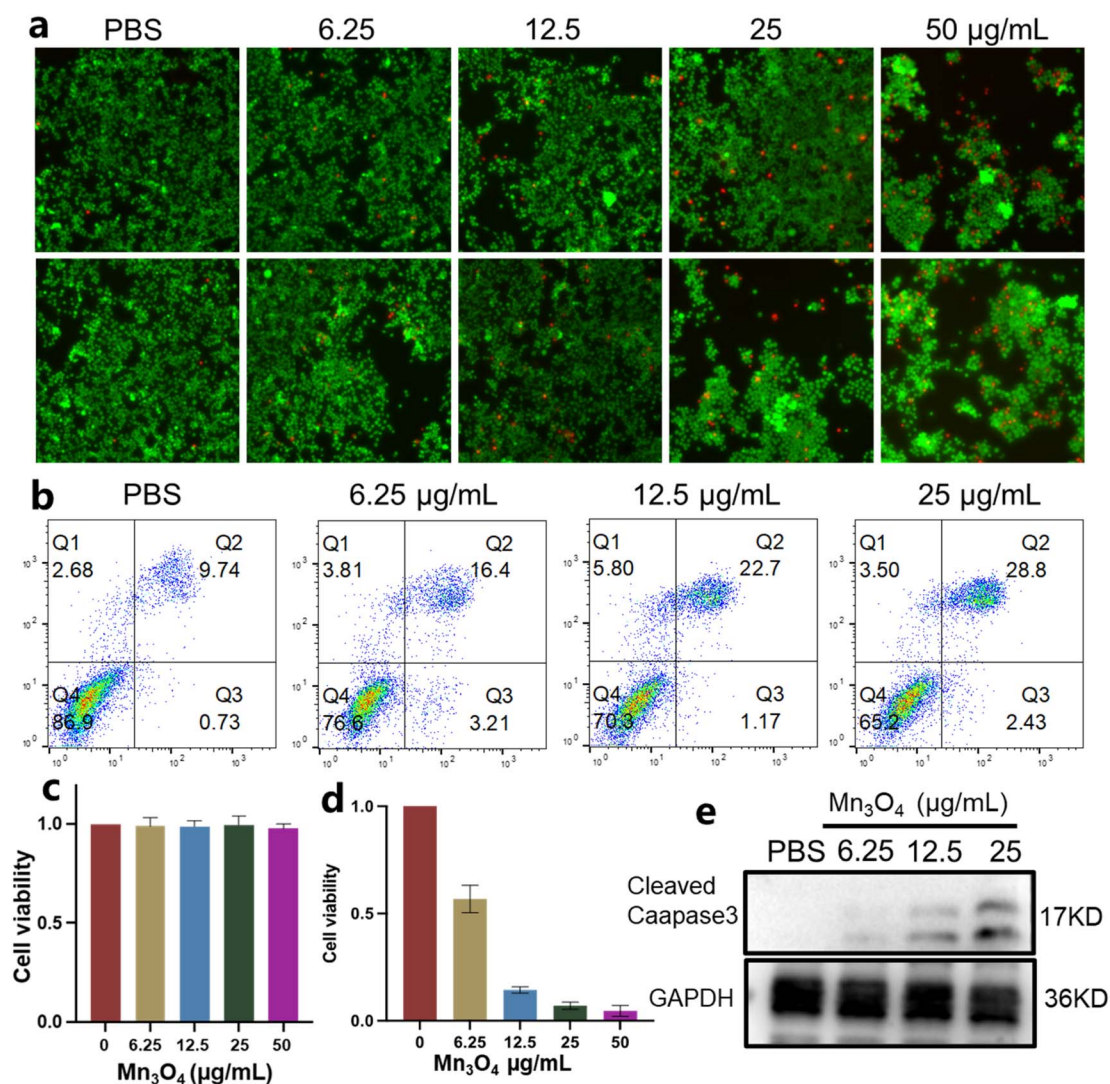


Fig. 4 Cell viability test of ultrasmall  $Mn_3O_4$  nanoparticles. (a) Living/dead staining of 4T1 cells after incubation with  $Mn_3O_4$  (6.25, 12.5, 25, and 50  $\mu\text{g mL}^{-1}$ ). (b) Flow cytometry assay of 4T1 cells after coculture with  $Mn_3O_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ). (c) CCK-8 test for endothelial cells after treating with  $Mn_3O_4$  (6.25, 12.5, 25, and 50  $\mu\text{g mL}^{-1}$ ). (d) CCK-8 test for 4T1 cells after treating with  $Mn_3O_4$  (6.25, 12.5, 25, and 50  $\mu\text{g mL}^{-1}$ ). (e) The cleaved caspase-3 express level in 4T1 cells after treating with  $Mn_3O_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ).



concentration of  $\text{Mn}_3\text{O}_4$  nanoparticles increase, the numbers of living 4T1 cells decrease and that of the dead cells obviously increase, showing an outstanding antitumor effect. The treated 4T1 cells were analyzed using flow cytometry, and the results are shown in Fig. 4b, which confirms the good antitumor effect of  $\text{Mn}_3\text{O}_4$  nanoparticles. The CCK-8 test was utilized to evaluate the biocompatibility of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles (Fig. 4c and d). With an increase in  $\text{Mn}_3\text{O}_4$  concentration, the cell viability of endothelial cells remains unchanged, but the cell viability of 4T1 cells dramatically decreases, indicating the excellent biocompatibility of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles with normal cells and distinct growth inhibition and a toxicity effect towards tumor cells. Cleaved caspase-3 is the typical marker of apoptosis, and the expression level of cleaved caspase-3 indicates the associated apoptosis state of cells. As shown in Fig. 4e, the treatment of  $\text{Mn}_3\text{O}_4$  nanoparticles could induce elevated expression of cleaved caspase-3, and the expression level was promoted by increased  $\text{Mn}_3\text{O}_4$  concentration. The cell viability assays confirm that ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles possess obvious cytotoxicity toward tumor cells and an intense antitumor effect.

To classify the ferroptosis antitumor mechanism of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles, the ROS staining of 4T1 cells

treated with  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) was performed, and the results are shown in Fig. 5a. The corresponding relative fluorescence intensity is shown in Fig. 5c. The ROS level in 4T1 cells after treatment with ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles is considerably promoted, and the promotion degree rises with the increasing of  $\text{Mn}_3\text{O}_4$  concentration. The ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles could react with GSH and  $\text{H}_2\text{O}_2$ . Therefore, the increased ROS level is induced using the GSH consumption ability and Fenton-like reaction between ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles and the high level of intracellular  $\text{H}_2\text{O}_2$ . For the ferroptosis pathway, the elevated ROS level could induce lipid peroxidation in tumor cells. Therefore, the BODIPY 581/591 C11 of the lipid peroxidation sensor was applied to detect the lipid status in 4T1 cells. The staining result is shown in Fig. 5b, and the corresponding relative fluorescence intensity is shown in Fig. 5d. These statistical data indicate that the ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles enhance lipid peroxidation in tumor cells. GPX4 in tumor cells could effectively decrease the ROS and lipid peroxidation levels, which could be considered an important marker of ferroptosis. The GPX4 expression level in 4T1 tumor cells after incubation with ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) is shown in Fig. 5e. With an increase in  $\text{Mn}_3\text{O}_4$  concentration, the GPX4 expression level

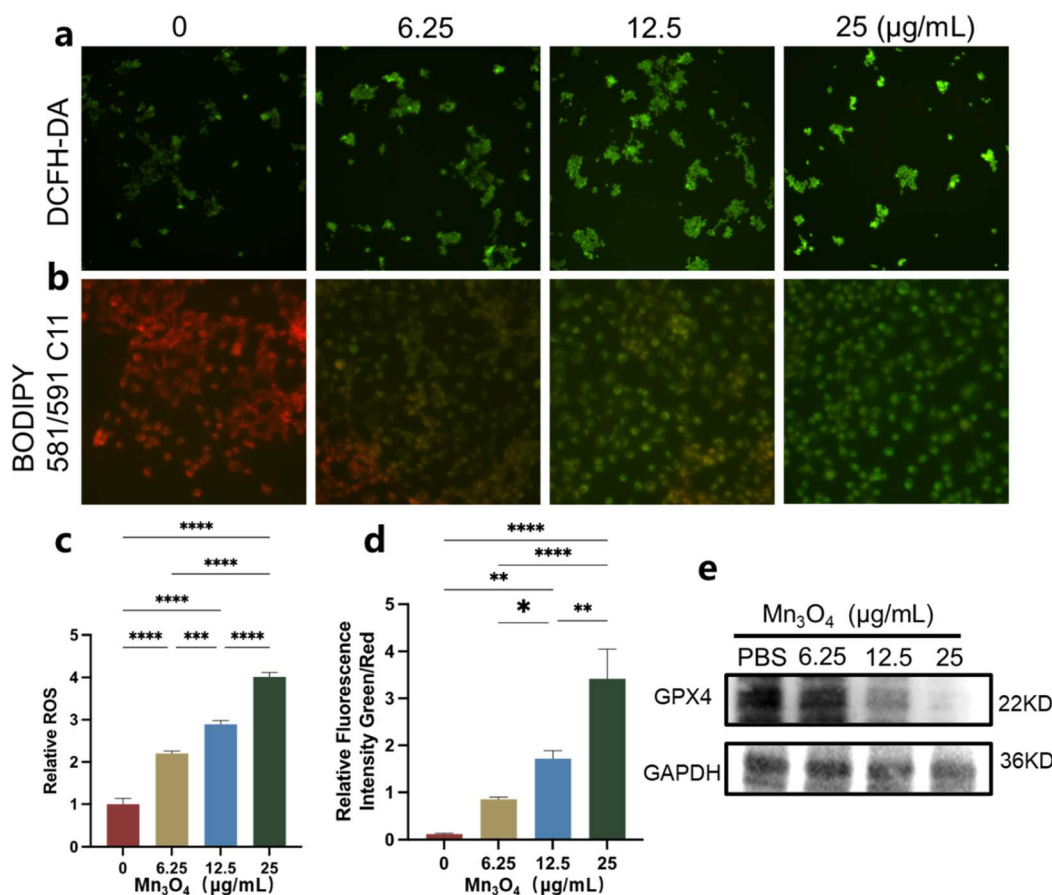


Fig. 5 Ferroptosis therapy of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles. (a) ROS staining of 4T1 cells after treating with  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) and (c) corresponding relative fluorescence intensity. (b) Lipid peroxidation staining of 4T1 cells after treating with  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) and (d) corresponding relative fluorescence intensity comparison. (e) GPX4 expression level in 4T1 cells after incubation with  $\text{Mn}_3\text{O}_4$  (6.25, 12.5, and 25  $\mu\text{g mL}^{-1}$ ) assessed using western blot analysis.



remarkably decreases, which is induced by an increased ROS level and reduced GSH concentration. These results confirm the efficient ferroptosis antitumor effect of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles.

### 3.3 $T_1$ and $T_2$ MRI signal variation for ultrasmall $\text{Mn}_3\text{O}_4$ nanoparticles *in vivo*

The MRI imaging of 4T1 tumors both *in vitro* and *in vivo* using ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles was characterized in detail. The  $T_1$  and  $T_2$  MRI imaging properties of 4T1 cells *in vitro* are shown

in Fig. 6a and b. After incubation of 4T1 cells with  $\text{Mn}_3\text{O}_4$  nanoparticles for 2, 4, 8 and 12 h, the lightness intensity of  $T_1$  MRI image, as shown in Fig. 6a, gradually elevates with increase in incubation time, which is due to the enhanced accumulation of  $\text{Mn}_3\text{O}_4$  nanoparticles in 4T1 tumor cells. The darkness of the  $T_2$  MRI image in Fig. 6b is distinctly promoted together with the incubation time scale, which could be caused by the reaction with high-level GSH in 4T1 cells. The MRI imaging of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles for 4T1 tumor cells identified  $T_1$  and  $T_2$  MRI variation at the cellular level. The ultrasmall  $\text{Mn}_3\text{O}_4$

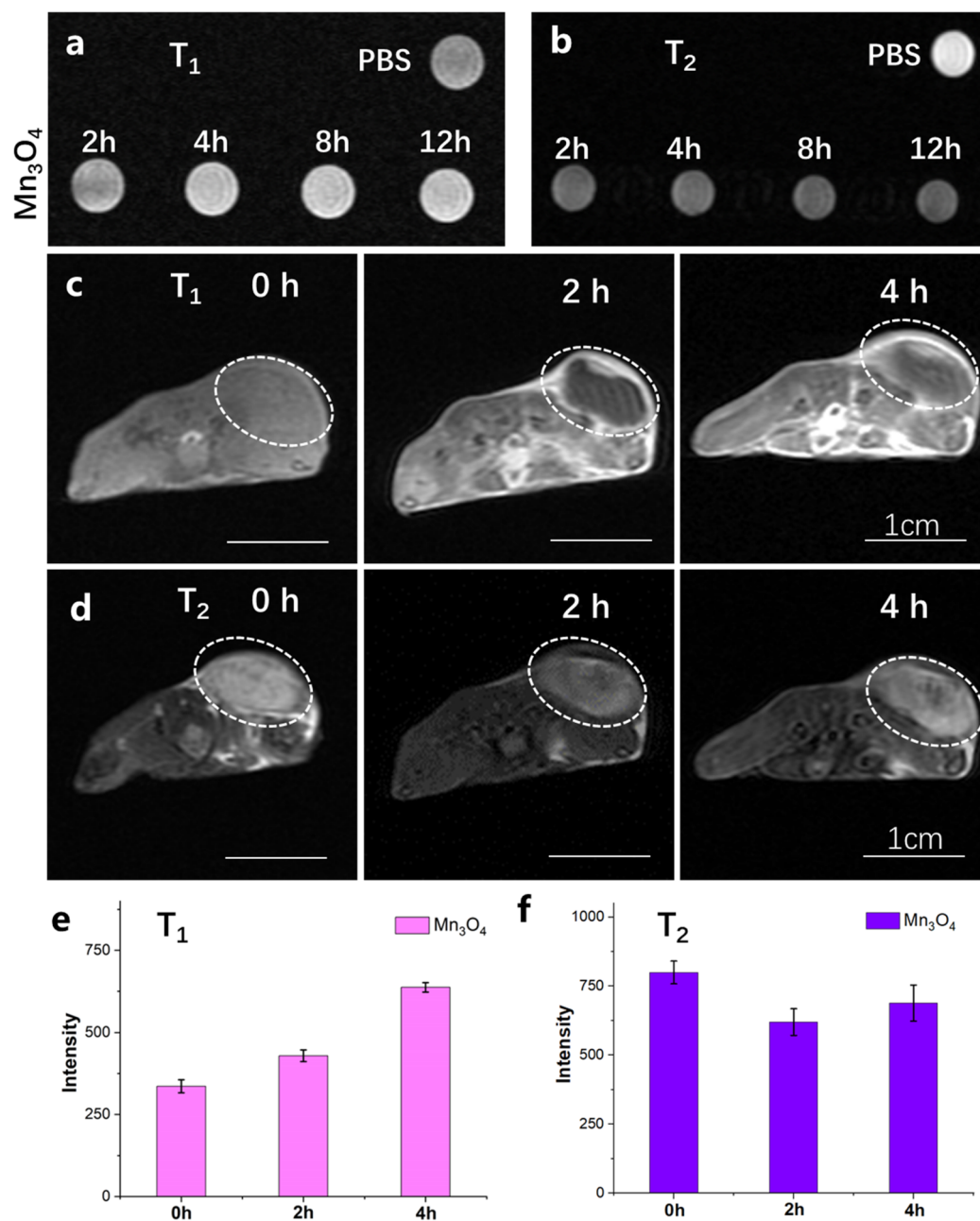


Fig. 6  $T_1$  and  $T_2$  MRI signal variation of ultrasmall  $\text{Mn}_3\text{O}_4$  nanoparticles in 4T1 cells and subcutaneous tumors. (a)  $T_1$  and (b)  $T_2$  MRI imaging and signal variation of 4T1 cells after incubation with  $\text{Mn}_3\text{O}_4$  for 2, 4, 8 and 12 h. (c)  $T_1$  and (d)  $T_2$  MRI imaging and signal variation of the subcutaneous tumor after injection with  $\text{Mn}_3\text{O}_4$  (5 mg  $\text{kg}^{-1}$ ) for 0, 2 and 4 h. (e) The corresponding  $T_1$  and (f)  $T_2$  MRI imaging signal intensity variation of the subcutaneous tumor.



nanoparticles for *in vivo* tumor MRI imaging were then detected in the 4T1 subcutaneous tumor. The  $T_1$  and  $T_2$  MRI imaging and signal variation results of tumors after intravenous injection of  $Mn_3O_4$  ( $5 \text{ mg kg}^{-1}$ ) for 0, 2 and 4 h are shown in Fig. 6c and d, respectively. The corresponding  $T_1$  and  $T_2$  MRI imaging signal intensity variations are shown in Fig. 6e and f, respectively. After the treatment of  $Mn_3O_4$  for 2 h and 4 h, the  $T_1$  signal intensity increases to 127.6% and 189.6%, respectively, compared with the control group, indicating the effectively enhanced  $T_1$  MRI imaging ability. In addition, the  $T_2$  signal intensity decreases to 77.5% and 86.1%, respectively, compared with the control group, showing a promoted  $T_2$  MRI imaging effect, which could be induced by the high level of GSH in the tumor microenvironment. The *in vitro* tumor microenvironment mimic condition (5 and 10 mM GSH in Fig. 3) is static without a continued blood flow. After the tail vein injection of  $Mn_3O_4$ ,  $Mn_3O_4$  gradually accumulated in the tumor site and induced the  $T_1$  MRI effect. Furthermore, GSH in the tumor environment induced the  $T_1$  turn-off and  $T_2$  turn-on effect. The cooperative results endow both the  $T_1$  turn-on and  $T_2$  turn-on effects. In the *in vitro* mimic condition, no  $Mn_3O_4$  accumulation process occurs, which directly induces the  $T_1$  turn-off and  $T_2$  turn-on effects. The MRI imaging results of tumors both *in vitro* and *in vivo*

collaboratively illustrate the variable  $T_1$  and  $T_2$  MRI imaging capability of ultrasmall  $Mn_3O_4$  nanoparticles.

Tumor metastasis is the main reason of recurrence and low cure rate of cancer. To excavate the MRI imaging capability of ultrasmall  $Mn_3O_4$  nanoparticles for tumors, the liver metastatic tumor was applied to perform the evaluation. MRI imaging results after the construction of the liver metastatic tumor model and the intravenous injection of  $Mn_3O_4$  ( $5 \text{ mg kg}^{-1}$ ) are shown in Fig. 7a. The lightness in the liver part shows graininess, and the intensity is significantly enhanced (inside the white circle). Compared with the PBS group, the signal intensity increases 159.7%, which conveys effective MRI detection ability for the metastatic tumor. The real image of the liver metastatic tumor (Fig. 7b) and the corresponding histological tissue sections of the liver and liver metastatic tumor (Fig. 7c) are consistent with the MRI image results, thus positioning ultrasmall  $Mn_3O_4$  nanoparticles as a promising candidate for overcoming the obstacle of accurate diagnosis for tumor metastasis.

### 3.4 Biosafety assessment of ultrasmall $Mn_3O_4$ nanoparticles

The *in vivo* biosafety of ultrasmall  $Mn_3O_4$  nanoparticles is evaluated using the hemolysis ratio and HE staining of main

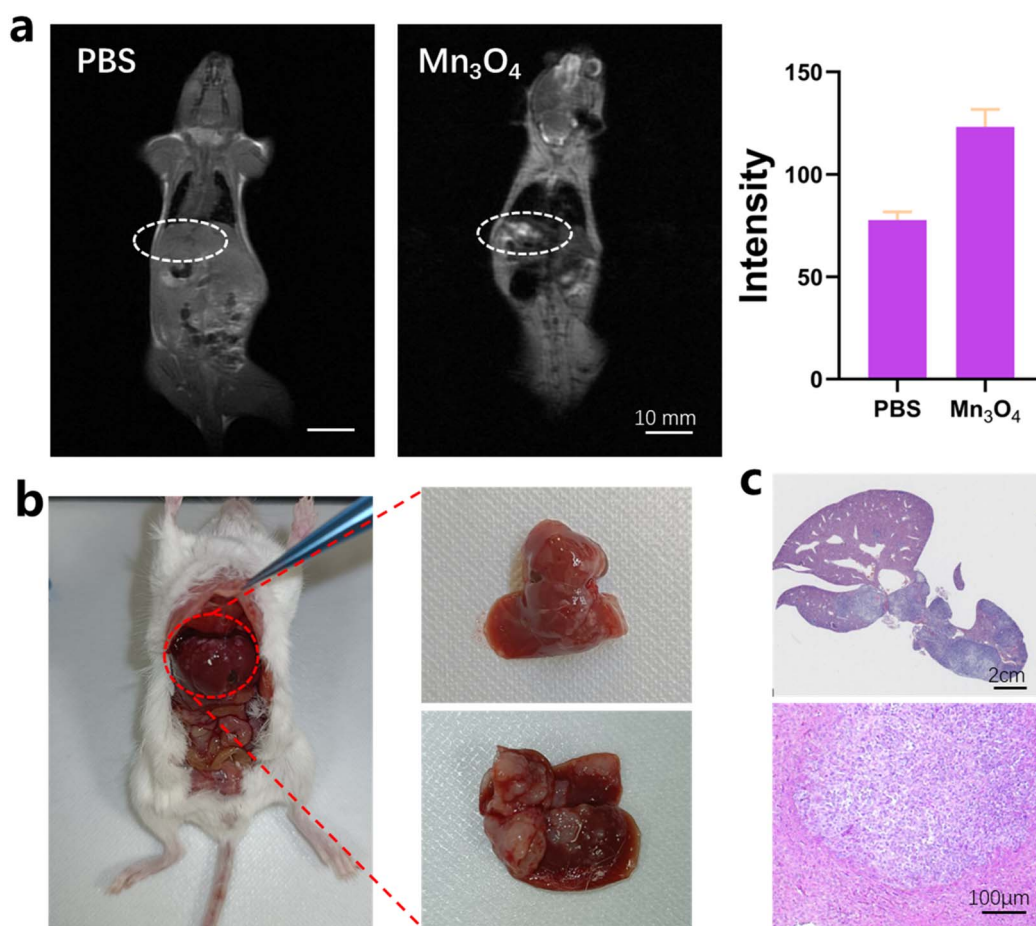


Fig. 7 MRI imaging, real image and tissue slicing staining of the liver metastatic tumor. (a) MRI imaging of the liver metastatic tumor and intensity statistic. (b) The real image of the liver metastatic tumor and (c) the corresponding histological tissue sections staining of the liver and metastatic tumor.



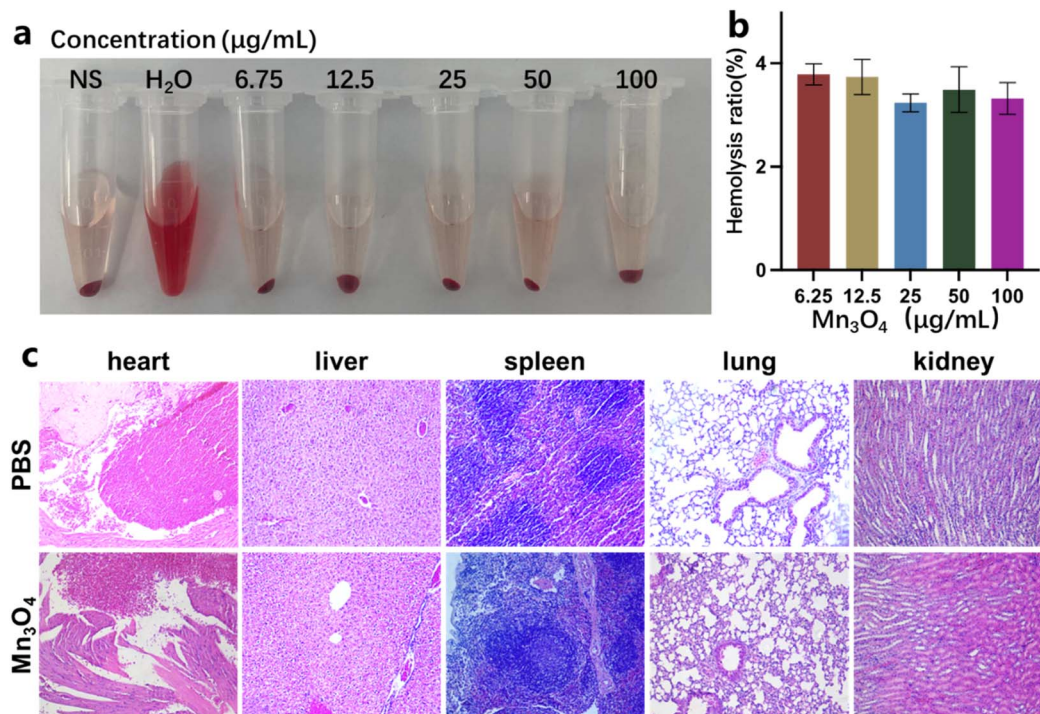


Fig. 8 Biosafety assessment of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles. (a) The real image of the hemolysis experiment and (b) hemolysis ratio. (c) HE staining of histological tissue sections of major organs after injecting Mn<sub>3</sub>O<sub>4</sub> nanoparticles.

organs. The hemolysis ratio (Fig. 8a and b) of Mn<sub>3</sub>O<sub>4</sub> nanoparticles (6.25, 12.5, 25, 50, and 100  $\mu\text{g mL}^{-1}$ ) remains smaller than 4%, revealing a good hemocompatibility for *in vivo* application. As shown in Fig. 8c, after treatment with Mn<sub>3</sub>O<sub>4</sub> nanoparticles *in vivo*, the sections from the main organ and tissues have no histological difference compared with the PBS group. Moreover, no histopathological damage, including necrosis and inflammatory response, is observed. These biosafety results demonstrate that the designed ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles possess excellent biocompatibility *in vivo* and could be used for further biomedical applications.

## 4. Conclusion

In summary, an effective and novel design of ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles with an integrated multifunction is proposed. Ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles with excellent uniformity and stability could be produced on a large scale. The designed Mn<sub>3</sub>O<sub>4</sub> nanoparticles could react and be decomposed by GSH owing to the existence of high valence Mn in the nanocrystals, which endow the responsive capability in the tumor microenvironment. Moreover, the GSH-responsive property enable Mn<sub>3</sub>O<sub>4</sub> with the  $T_1$  turn-off and  $T_2$  turn-on MRI imaging variation effects and effectively enhance the imaging intensity and quality. The unique  $T_1/T_2$  MRI imaging could be applied to subcutaneous tumors and liver metastases tumors for promoting accurate tumor diagnosis. Moreover, the ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles possess an excellent ferroptosis therapy effect toward tumor cells, which presents superb prospects for tumor treatment. The well-designed integrated multifunction

in individual ultrasmall Mn<sub>3</sub>O<sub>4</sub> nanoparticles demonstrates an innovative strategy for effective tumor treatment under MRI guidance and holds considerable potential for cancer treatment in the clinic.

## Data availability

Data will be made available upon reasonable request.

## Author contributions

Q. H. X. and J. Z. D completed most of the experiments and manuscript. G. R. Z. and D. W. Z. performed partial experiments, including the MRI test and cell staining. H. L. and D. X. Y. provided careful guidance and adequate experimental support for this work.

## Conflicts of interest

The authors declare no known competing financial interests or personal relationships.

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## References

- Q. He, H. Hu, Q. Zhang, T. Wu, Y. Zhang, K. Li, *et al.*, *Chem. Eng. J.*, 2020, **397**, 125478.
- G. Yang, L. Xu, Y. Chao, J. Xu, X. Sun, Y. Wu, *et al.*, *Nat. Commun.*, 2017, **8**(1), 902.
- W. Yang, C. Xiang, Y. Xu, S. Chen, W. Zeng, K. Liu, *et al.*, *Biomaterials*, 2020, **255**, 120186.
- H. B. Na, J. H. Lee, K. An, Y. I. Park, M. Park, I. S. Lee, *et al.*, *Angew. Chem., Int. Ed.*, 2007, **46**(28), 5397–5401.
- T.-H. Shin, P. K. Kim, S. Kang, J. Cheong, S. Kim, Y. Lim, *et al.*, *Nat. Biomed. Eng.*, 2021, **5**(3), 252–263.
- Y. Luo, J. Yang, J. Li, Z. Yu, G. Zhang, X. Shi, *et al.*, *Colloids Surf., B*, 2015, **136**, 506–513.
- B. H. Kim, N. Lee, H. Kim, K. An, Y. I. Park, Y. Choi, *et al.*, *J. Am. Chem. Soc.*, 2011, **133**(32), 12624–12631.
- S. Zhao, J. Duan, Y. Lou, R. Gao, S. Yang, P. Wang, *et al.*, *Nanoscale*, 2021, **13**(45), 19109–19122.
- Q. Xie, X. Wang, G. Zhang, D. Zhou, Y. Zhao, H. Liu, *et al.*, *Sci. Rep.*, 2024, **14**(1), 10646.
- Q. Jia, J. Ge, W. Liu, X. Zheng, S. Chen, Y. Wen, *et al.*, *Adv. Mater.*, 2018, **30**(13), 1706090.
- D. Wang, H. Wu, W. Q. Lim, S. Z. F. Phua, P. Xu, Q. Chen, *et al.*, *Adv. Mater.*, 2019, **31**(27), 1901893.
- L. Gao, Y. Zhang, L. Zhao, W. Niu, Y. Tang, F. Gao, *et al.*, *Sci. Adv.*, 2020, **6**(29), eabb1421.
- L. Feng, B. Liu, R. Xie, D. Wang, C. Qian, W. Zhou, *et al.*, *Adv. Funct. Mater.*, 2020, **31**(5), 2006216.
- S. I. Han, Sw Lee, M. G. Cho, J. M. Yoo, M. H. Oh, B. Jeong, *et al.*, *Adv. Mater.*, 2020, **32**(31), 2001566.
- Y. Chen, Q. Yin, X. Ji, S. Zhang, H. Chen, Y. Zheng, *et al.*, *Biomaterials*, 2012, **33**(29), 7126–7137.
- J. Xiao, X. M. Tian, C. Yang, P. Liu, N. Q. Luo, Y. Liang, *et al.*, *Sci. Rep.*, 2013, **3**(1), 2045.
- R. Wei, X. Gong, H. Lin, K. Zhang, A. Li, K. Liu, *et al.*, *Nano Lett.*, 2019, **19**(8), 5394–5402.
- Z. Yi, Z. Luo, N. D. Barth, X. Meng, H. Liu, W. Bu, *et al.*, *Adv. Mater.*, 2019, **31**(37), 1901851.
- X. Li, S. Lu, Z. Xiong, Y. Hu, D. Ma, W. Lou, *et al.*, *Adv. Sci.*, 2019, **6**(19), 1901800.
- S. Bai, N. Yang, X. Wang, F. Gong, Z. Dong, Y. Gong, *et al.*, *ACS Nano*, 2020, **14**(11), 15119–15130.
- X. Xu, X. Zhou, B. Xiao, H. Xu, D. Hu, Y. Qian, *et al.*, *Nano Lett.*, 2021, **21**(5), 2199–2206.
- J. Shi, W. Yu, L. Xu, N. Yin, W. Liu, K. Zhang, *et al.*, *Nano Lett.*, 2019, **20**(1), 780–789.
- H. J. Kwon, K. Shin, M. Soh, H. Chang, J. Kim, J. Lee, *et al.*, *Adv. Mater.*, 2018, **30**(42), 1704290.
- B. Ding, P. Zheng, Pa Ma and J. Lin, *Adv. Mater.*, 2020, **32**(10), 1905823.
- Y. Zhan, S. Shi, E. B. Ehlerting, S. A. Graves, S. Goel, J. W. Engle, *et al.*, *ACS Appl. Mater. Interfaces*, 2017, **9**(44), 38304–38312.
- J. Duan, S. Zhao, Y. Duan, D. Sun, G. Zhang, D. Yu, *et al.*, *Adv. Healthc. Mater.*, 2024, 2303963.
- N. Singh, M. A. Savanur, S. Srivastava, P. D'Silva and G. Mugesh, *Angew. Chem., Int. Ed.*, 2017, **129**(45), 14455–14459.
- Y. Liu, J. Yang, B. Liu, W. Cao, J. Zhang, Y. Yang, *et al.*, *Nano-Micro Lett.*, 2020, **12**(1), 127.
- X. Zeng, S. Yan, P. Chen, W. Du and B.-F. Liu, *Nano Res.*, 2020, **13**(6), 1527–1535.
- W. Wang, J. Duan, W. Ma, B. Xia, F. Liu, Y. Kong, *et al.*, *Adv. Sci.*, 2023, **10**(17), 2205859.

