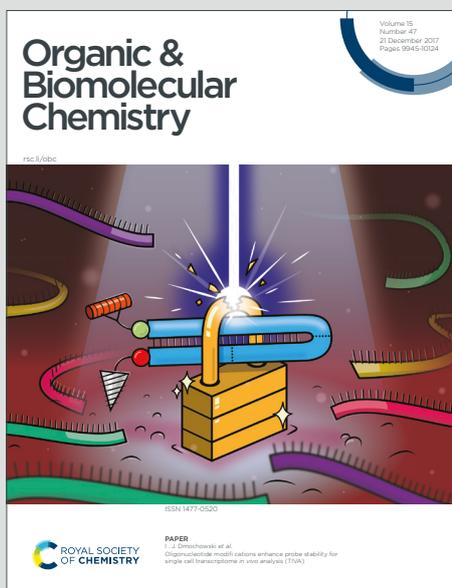


# Organic & Biomolecular Chemistry

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: R. E. Lynch, A. Lowry, G. P. McGlacken and P. Byrne, *Org. Biomol. Chem.*, 2025, DOI: 10.1039/D6OB00221H.



This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.

## COMMUNICATION

Synthesis of Tri-substituted, Aliphatic and  $^{13}\text{C}$ -Labelled  $\alpha,\beta$ -Unsaturated Carboxylic Acids *via* Wittig  $\text{CO}_2$  Utilisation ReactionsRachel E. Lynch,<sup>a</sup> Amy Lowry,<sup>b</sup> Gerard P. McGlacken,<sup>b,c</sup> Peter A. Byrne<sup>\*,a,c</sup>Received 00th January 20xx,  
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

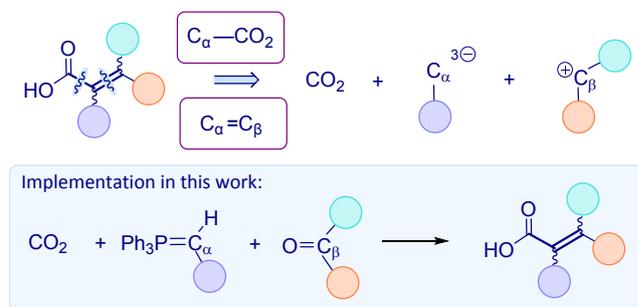
Phosphonium carboxylate ylides are formed through  $\text{CO}_2$  activation by phosphonium ylides. These can undergo Wittig reactions with carbonyl compounds, thereby enabling formation of both the  $\text{C}-\text{CO}_2$  bond and the two bonds of the  $\text{C}=\text{C}$  double bond of an  $\alpha,\beta$ -unsaturated carboxylic acid in a one-pot transformation. In this work, we have developed methodology to enable the synthesis of very useful trisubstituted alkene-containing products. Secondly, the protocol allows for the use of base-sensitive aliphatic aldehydes, avoiding the occurrence of aldol self-condensation. Thirdly, we show that this methodology enables production of a labelled  $\alpha,\beta$ -unsaturated carboxylic acid using  $^{13}\text{C}$ -labelled  $\text{CO}_2$ .

## Introduction

Utilisation of waste products as chemical feedstocks for the production of valuable chemical products has become a point of significant focus in recent years.<sup>1–3</sup>  $\text{CO}_2$  is arguably the most important waste product of the modern industrialised world, and hence development of methods for utilisation of  $\text{CO}_2$  is of particular significance. Indeed, due to its ready availability, renewability, low cost, and non-toxicity,  $\text{CO}_2$  is a highly attractive one-carbon building block for the construction of valuable target compounds.<sup>4–7</sup> Although much research has been conducted into chemical transformations of  $\text{CO}_2$ ,<sup>8–18</sup> only a small number of chemicals are made industrially using  $\text{CO}_2$ .<sup>6,19,20</sup> Development of approaches for utilisation of  $\text{CO}_2$  in the synthesis of high value entities of industrial importance, such as synthetic building blocks and Active Pharmaceutical Ingredients (APIs), is thus of paramount importance.

Recently, we reported on a novel  $\text{CO}_2$  utilisation methodology in which *three* new carbon-carbon bonds could be formed in one process, enabling formation of both the  $\text{C}_\alpha-\text{CO}_2$  single bond and the  $\text{C}_\alpha=\text{C}_\beta$  double bond of  $\alpha,\beta$ -unsaturated carboxylic acids using aromatic or vinylic aldehydes,  $\text{CO}_2$  and phosphonium ylides as the starting materials.<sup>21</sup> This approach originates from a novel retrosynthetic strategy involving an unprecedented three bond

disconnection (see Scheme 1). The first  $\text{C}-\text{C}$  bond is formed by nucleophilic addition of a phosphonium ylide (**2**) to  $\text{CO}_2$ . In the presence of base, the resulting ylide- $\text{CO}_2$  adduct (**3**) is deprotonated to give phosphonium carboxylate ylide **4**, which can undergo Wittig reactions with aldehydes to form the two bonds in the  $\text{C}_\alpha=\text{C}_\beta$  double bond. This methodology enables a very direct means of access to  $\alpha,\beta$ -unsaturated carboxylic acids to be realised, utilising  $\text{CO}_2$ . The  $\alpha,\beta$ -unsaturated carboxylic acid motif is prevalent in the structures of pharmaceutical compounds and their synthetic precursors<sup>22,23</sup> and also appears in the structures of important commodity chemicals such as acrylic acid (manufactured on a multimillion ton level).<sup>24,25</sup> The availability of the new methodology represented in Scheme 2 for direct access to compounds containing the  $\alpha,\beta$ -unsaturated carboxylic acid motif through combination of relatively simple starting materials is thus highly significant.



**Scheme 1.** Novel retrosynthetic strategy demonstrating a three-bond disconnection for  $\alpha,\beta$ -unsaturated carboxylic acids.

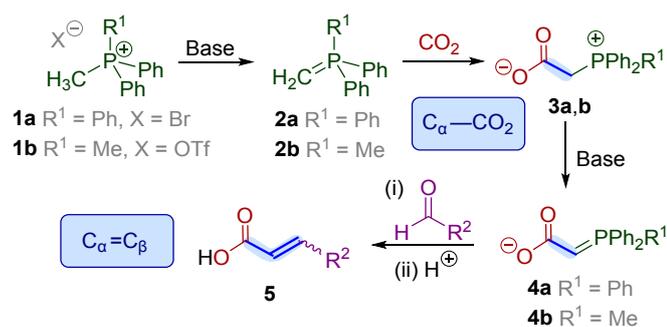
In our previously reported method for the construction of disubstituted  $\alpha,\beta$ -unsaturated carboxylic acids from phosphonium salts **1a** and **1b** (*via* derived carboxylate ylides **4a** and **4b** (formed using  $\text{CO}_2$ ); see Scheme 2), a wide substrate scope was demonstrated by synthesising a library of 37 disubstituted  $\alpha,\beta$ -unsaturated carboxylic acids from aromatic and vinylic aldehydes.<sup>21</sup> We wished to further develop this methodology to include the synthesis of  $\alpha,\beta$ -unsaturated carboxylic acids containing trisubstituted  $\text{C}=\text{C}$  entities and to investigate the applicability of the methodology in reactions

<sup>a</sup> Centre for Synthesis & Chemical Biology, School of Chemistry, University College Dublin, Belfield, Dublin 4, Ireland

<sup>b</sup> School of Chemistry & Analytical and Biological Chemistry Research Facility, University College Cork, Cork, Ireland.

<sup>c</sup> SSPC, the Research Ireland Centre for Pharmaceuticals, Ireland.





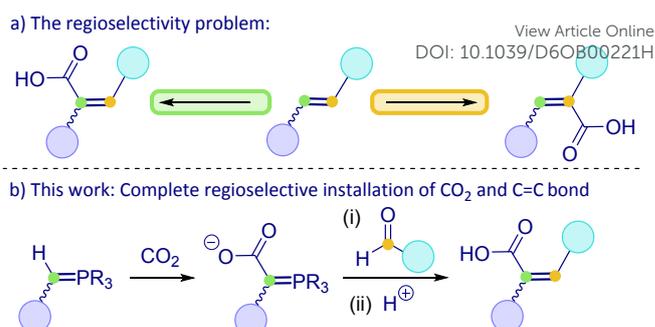
**Scheme 2.** Formation of three new carbon-carbon bonds via a novel CO<sub>2</sub> utilisation methodology mediated by phosphonium ylides.

with aliphatic aldehydes, each of which posed its own unique new challenges that had not required consideration in the development of our previously reported methodology (see further details below).

## Results and Discussion

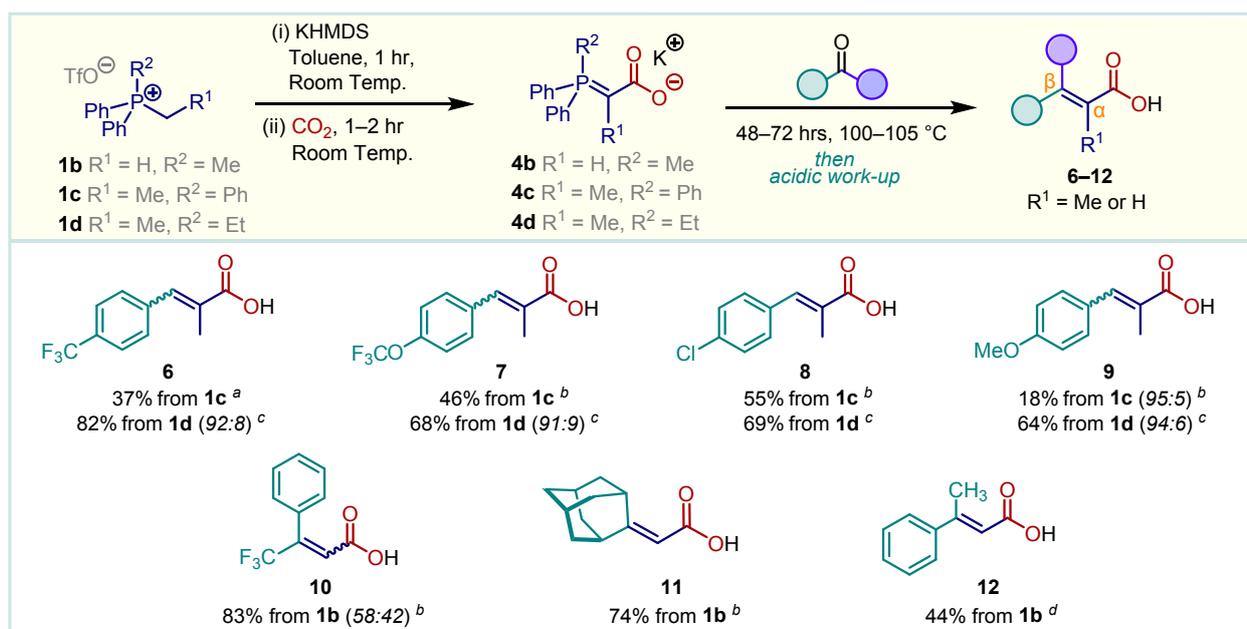
**(i) Synthesis of Trisubstituted Alkenes:** There exist many examples of regioselective carboxylation of terminal alkenes and alkynes for the synthesis of disubstituted  $\alpha,\beta$ -unsaturated carboxylic acids.<sup>18,26–33</sup> If one considers the installation of a carboxylic acid group as a substituent on an internal alkene, however, there are obvious regioselectivity concerns, with there being the possibility of carboxylation at either of the alkene carbons, as outlined in Scheme 3a. We envisaged that expanding our methodology to include trisubstituted  $\alpha,\beta$ -unsaturated carboxylic acids could allow for completely regioselective installation of not only the carboxyl group (originating from CO<sub>2</sub>) but also the C <sub>$\alpha$</sub> =C <sub>$\beta$</sub>  double bond (which is already installed in the unsaturated starting material in alkene and alkyne carboxylation methods).<sup>18,26–33</sup>

Access to  $\alpha$ -substituted  $\alpha,\beta$ -unsaturated carboxylic acids can in principle be facilitated through use of  $\alpha$ -substituted phosphonium carboxylate ylides such as **4c** and **4d** (derived from phosphonium



**Scheme 3.** Regiochemical considerations in the synthesis of trisubstituted  $\alpha,\beta$ -unsaturated carboxylic acids.

salts **1c** and **1d**; see Figure 1), while reactions of carboxylate ylides such as **4b** (derived from phosphonium salt **1b**) with ketones should enable access to  $\beta$ -substituted  $\alpha,\beta$ -unsaturated carboxylic acids. In practice, when we attempted our previously reported<sup>21</sup> Wittig CO<sub>2</sub> utilisation methodology using phosphonium carboxylate ylide **4c** (formed from phosphonium salt **1c** via ylide **2c** and ylide-CO<sub>2</sub> adduct **3c**) with several representative aromatic aldehydes, we found that compounds **6–9** were indeed accessible. However, only moderate yields were obtained for reactions of electron-deficient aldehydes (37% for **6**, 46% for **7**, and 55% for **8**), and a poor yield of 18% was obtained for the reaction of relatively electron-rich *p*-methoxybenzaldehyde to produce **9**. Thus, unfortunately, our original methodology was rendered unviable for these useful targets and we set out to design a more active ylide. Since exchanging a *P*-phenyl substituent on the phosphonium ylide for a *P*-alkyl substituent is known to enhance the nucleophilicity of the ylidic carbon,<sup>21,34,35</sup> we imagined that carboxylate ylide **4d** (generated *in situ* from **1d**; see Figure 1) might allow higher yields of trisubstituted  $\alpha,\beta$ -unsaturated carboxylic acids such as **6–9** to be achieved. Utilisation of a carboxylate ylide of enhanced nucleophilicity proved to be successful, enabling very substantial improvements in the

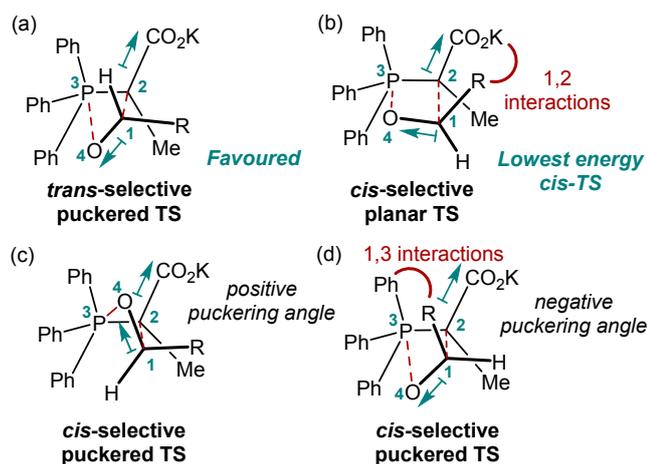


**Figure 1.**  $\alpha,\beta$ -Unsaturated carboxylic acids containing trisubstituted alkenes synthesised using phosphonium salts **1c** and **1d**. Isolated yields after chromatography are shown. Alkene *E:Z* ratios are shown in parentheses, where appropriate. Reactions were generally conducted on 1 mmol scale. <sup>a</sup> Wittig step: 24 hours, 100 °C. <sup>b</sup> Wittig step: 48 hours, 100 °C. <sup>c</sup> Wittig step: 72 hours, 100 °C. <sup>d</sup> Wittig step: 5 days 105 °C.



yields of compounds **6–9** (Figure 1), with yields of 82% for **6**, 68% for **7**, 69% for **8**, and 64% for **9**. The reactions investigated of carboxylate ylides **4c** and **4d** (derived from phosphonium salts **1c** and **1d**) all exhibited very high or even exclusive selectivity for *E*-configured C=C bonds in each of the  $\alpha,\beta$ -unsaturated carboxylic acid products (see *E/Z* ratios in Figure 1). This methodology thus affords completely regioselective and highly stereoselective access to trisubstituted alkene-containing  $\alpha,\beta$ -unsaturated carboxylic acids, thereby exploiting the major advantages conferred by the Wittig reaction while also incorporating CO<sub>2</sub> into the products.

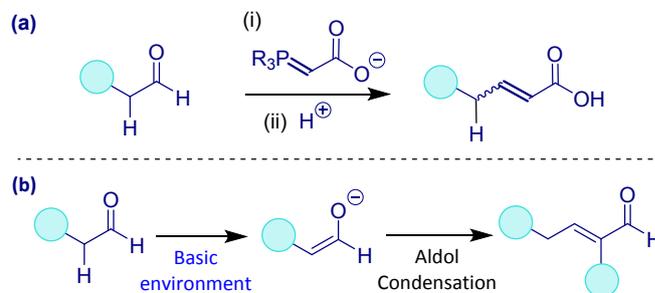
The very high *E*-selectivity in these reactions is consistent with the rationale proposed by Aggarwal, Harvey and co-workers for selectivity in Wittig reactions of  $\alpha$ -substituted ester-stabilised ylides.<sup>36–41</sup> We propose that in the transition state leading to the oxaphosphetane intermediate, puckering of the forming four-membered ring occurs because it results in a favourable antiparallel arrangement of the dipoles along the aldehyde C—O bond and ylide C—CO<sub>2</sub> bond (see Figure 2a). In this conformation, minimisation of steric interactions (between the substituents on C-1 and C-2 and between those on C-1 and P) leads to formation of the *trans*-oxaphosphetane and hence *E*-alkene being strongly favoured, particularly for reactions of Ph<sub>3</sub>P-derived ylide **4c**. Potential *cis*-selective transition states are destabilised by the occurrence of 1,2 or 1,3 steric interactions (Fig. 2b and 2d) or unfavourable dipole-dipole interactions (Figure 2c). As a caveat to this rationale, which is predicated on the stereochemistry of the alkene C=C bond being set during the Wittig reaction, we note that we did observe previously that isomerisation of a *Z*-cinnamic acid does occur under the conditions of our Wittig CO<sub>2</sub> reactions.<sup>21</sup> Consequently, we must acknowledge that isomerisation of the *Z*-isomers of our products may have occurred in the reactions discussed above, thereby potentially contributing to the observed high *E*-selectivity.



**Figure 2.** Possible transition states (TSs) for the Wittig reaction of  $\alpha$ -methyl substituted phosphonium carboxylate ylide **4c** with an aldehyde, RCHO.<sup>36</sup>

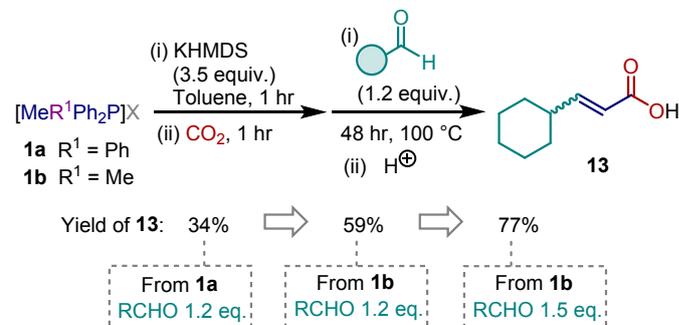
We also wished to investigate whether the methodology would be effective for formation of  $\beta$ -substituted  $\alpha,\beta$ -unsaturated carboxylic acids (with trisubstituted alkene moieties) by using ketones as the starting materials. Reactions of phosphonium carboxylate ylide **4b** (derived from **1b**, [Me<sub>2</sub>Ph<sub>2</sub>P]OTf), with representative

ketones were indeed found to give trisubstituted  $\alpha,\beta$ -unsaturated carboxylic acids **10** and **11** in high yields (see Figure 1). The Wittig reaction of phosphonium carboxylate ylide **4b** with acetophenone afforded product **12** in only a moderate yield of 44%, even after pushing the reaction time to 5 days. Interestingly, this reaction gave the *E*-isomer of product **12** exclusively.



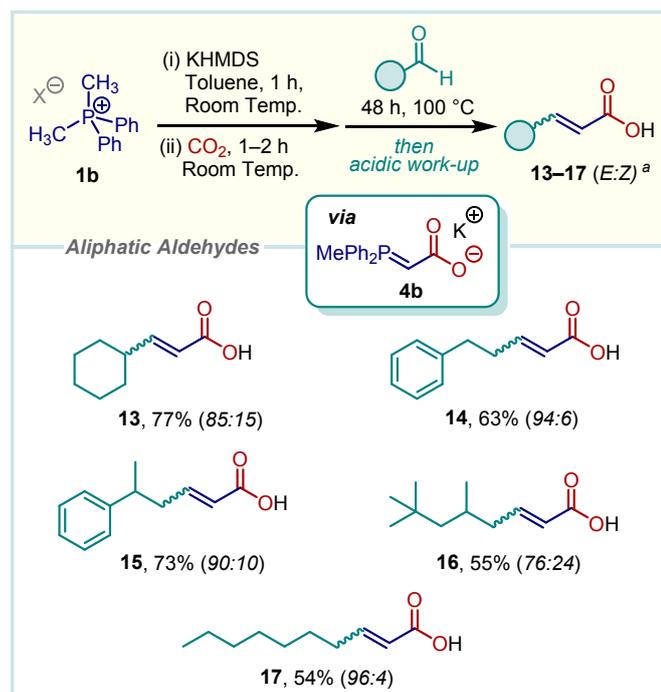
**Scheme 4.** (a) Wittig CO<sub>2</sub> utilisation reaction of an aliphatic aldehyde. (b) Aldol self-condensation of an aliphatic aldehyde under basic conditions.

**(ii) Extension of Methodology to Aliphatic Aldehydes:** In addition, we wished to test our Wittig CO<sub>2</sub> utilisation methodology further by investigating its applicability in reactions with aliphatic aldehydes (Scheme 4a). As indicated above, our previously reported methodology involved only aromatic or vinylic aldehydes,<sup>21</sup> which do not bear acidic protons at the  $\alpha$ -position. The  $\alpha$ -proton(s) of aliphatic aldehydes are susceptible to deprotonation by strong bases, and hence there is a high likelihood that aldol self-condensation reactions could occur under the basic conditions typically employed in our Wittig CO<sub>2</sub> utilisation reactions (Scheme 4b).<sup>42</sup> To investigate whether conditions could be found under which Wittig CO<sub>2</sub> utilisation reactions could be accomplished in preference to aldol self-condensation reactions, a series of optimisation experiments were undertaken using cyclohexanecarboxaldehyde as the test substrate (Scheme 5).<sup>43</sup> When the reaction was performed starting from phosphonium salt **1a** (via ylide **2a** and hence carboxylate ylide **4a**), a low yield of 34% of compound **13** was obtained (Scheme 5). Changing the starting phosphonium salt from **1a** to **1b** saw an increase in the yield of **13** (34% to 59%; Scheme 5), demonstrating a similar trend to the increase in yields obtained in other reactions discussed above when a *P*-phenyl substituent was exchanged for a *P*-methyl (compare yields for products derived from **1c** and **1d** in Figure 1). Increasing the amount of the aldehyde added from 1.2 equivalents to 1.5 equivalents, we were able to obtain a good yield



**Scheme 5.** Development of conditions for Wittig CO<sub>2</sub> utilisation reactions of aliphatic aldehyde and phosphonium salts **1a** or **1b**.



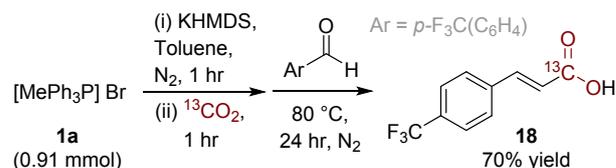


**Figure 3.** Disubstituted aliphatic  $\alpha,\beta$ -unsaturated carboxylic acids synthesised using  $[\text{Me}_2\text{Ph}_2\text{P}] \text{OTf}$ . Isolated yields are shown. The  $E:Z$  ratios of alkenes are shown in parentheses. Reactions were conducted on ca. 1 mmol scale.

of the  $\alpha,\beta$ -unsaturated carboxylic acid product (**13**, 77%; Scheme 5). The Wittig step of this reaction was carried out at 100 °C over 48 hours, starting from phosphonium salt **1b** (and proceeding via ylide **2b** and phosphonium carboxylate ylide **4b**). Several other products derived from aliphatic aldehydes, **13–17**, were generated using these reaction conditions (starting from phosphonium salt **1b**) in moderate to good yields of (54–73%; see Figure 3). Similar to the Wittig  $\text{CO}_2$  utilisation reactions discussed above and those reported in our previous publication,<sup>21</sup> the reactions of phosphonium carboxylate ylide **4b** with aliphatic aldehydes also result in preferential formation of  $E$ - $\alpha,\beta$ -unsaturated carboxylic acids (see  $E/Z$  ratios in Figure 3). The high  $E$ -stereoselectivity exhibited in these reactions is consistent with the selectivity exhibited by reactions of  $\text{MePh}_2\text{P}$ -derived ester-stabilised ylides,<sup>37,40,41</sup> which, while highly  $E$ -selective, are expected to result in formation of  $Z$ -alkene in small but significant amounts, in contrast to reactions of  $\text{Ph}_3\text{P}$ -derived ester-stabilised ylides, which are almost exclusively  $E$ -selective.

**(iii) Application for Isotopic Labelling:** The availability of novel  $\text{CO}_2$  utilisation methods provides new ways to employ  $\text{CO}_2$  as a chemical feedstock and one-carbon building block. One application in which this can be exploited to great effect is in the incorporation of isotopically labelled carbon atoms ( $^{13}\text{C}$  or  $^{14}\text{C}$ , or indeed labelled oxygen atoms,  $^{18}\text{O}$ ) into compounds through use of isotopically labelled  $\text{CO}_2$ .<sup>44–46</sup> In order to demonstrate the efficacy of the Wittig  $\text{CO}_2$  utilisation methodology for this purpose,  $^{13}\text{C}$ -labelled  $\text{CO}_2$  was employed in the synthesis of a representative  $\alpha,\beta$ -unsaturated carboxylic acid, 4-(trifluoromethyl)cinnamic acid (**18**).  $^{13}\text{CO}_2$  was used to react with ylide **2a** (generated from phosphonium salt **1a**) to produce carboxylate ylide **4a- $^{13}\text{C}$** , and this underwent reaction with 4-(trifluoromethyl)benzaldehyde at 80 °C to generate  $^{13}\text{C}$ -labelled product **18** in an isolated yield of 70% (Scheme 6). This demonstrates

the capacity of this method to enable facile incorporation of isotopically labelled atoms into products using labelled  $\text{CO}_2$  (with  $\text{CO}_2$  pressure at atmospheric pressure levels) and exemplifies the value of  $\text{CO}_2$  as a chemical feedstock.



**Scheme 6.** Synthesis of  $^{13}\text{C}$ -labelled 4-(trifluoromethyl)cinnamic acid using  $^{13}\text{CO}_2$  in a Wittig  $\text{CO}_2$  utilisation reaction.

## Conclusion

We have succeeded in extending the applicability of our Wittig  $\text{CO}_2$  utilisation methodology, developing three novel applications that had not previously been possible. The methodology was successfully applied to reactions of aliphatic aldehydes, avoiding significant competitive occurrence of aldol self-condensation, thereby allowing for the synthesis of  $\alpha,\beta$ -unsaturated carboxylic acids bearing aliphatic substituents at the  $\beta$ -position for the first time.  $\alpha$ -Substituted  $\alpha,\beta$ -unsaturated carboxylic acids (each containing a trisubstituted  $\text{C}=\text{C}$  bond) were synthesised in good yields and with high  $E$ -selectivity, enabled by the increased reactivity of a more nucleophilic ylide. The methodology was also extended to ketones to access  $\beta$ -substituted  $\alpha,\beta$ -unsaturated carboxylic acids, which are typically challenging substrates in Wittig reactions of stabilised ylides. The site of installation of both the carboxyl group and the alkene in these reactions is unambiguous; this approach thus enables utilisation of  $\text{CO}_2$  while exploiting the unique advantages of the Wittig reaction to form three new carbon-carbon bonds in a one pot telescoped process. Finally, the methodology was shown to enable the generation of labelled  $\alpha,\beta$ -unsaturated carboxylic acids through the use of labelled  $\text{CO}_2$ , providing a straightforward means of incorporating a labelled isotope into relatively complex products.

## Author Contributions

Conceptualisation, P.A.B.; Methodology, P.A.B. and G.P.M.; Investigation, A.L., R.E.L., and P.A.B.; Formal Analysis, A.L., R.E.L., and P.A.B.; Writing – Original Draft, P.A.B. and R.E.L.; Writing – Review & Editing, P.A.B., G.P.M., A.L. and R.E.L.; Funding Acquisition, P.A.B., G.P.M., A.L. and R.E.L.; Supervision, P.A.B. and G.P.M.

## Conflicts of interest

There are no conflicts to declare.

## Data availability

Synthetic details, experimental methods, and characterisation data (including copies of NMR spectra) are included in the Supplementary Information. Note: The authors have cited additional references within the Supplementary Information.<sup>47–73</sup>

## Acknowledgements

We acknowledge financial support from Research Ireland (grant numbers GOIPG/2018/169, GOIPG/2021/802, 20 FFP-P-8865), and the SSPC (the Research Ireland Centre for Pharmaceuticals; grant



number 12/RC/2275\_P2). We thank Dr Denis Lynch, Dr Lorraine Bateman, Dr Yannick Ortin and Dr Patricia Fleming for NMR spectroscopy support, and Dr Jimmy Muldoon for mass spectrometry support.

## Notes and references

- Q. Liu, L. Wu, R. Jackstell and M. Beller, *Nat. Commun.*, 2015, **6**, 5933.
- T. Sakakura, J.-C. Choi and H. Yasuda, *Chem. Rev.*, 2007, **107**, 2365–2387.
- C. Maeda, Y. Miyazaki and T. Ema, *Catal. Sci. Technol.*, 2014, **4**, 1482–1497.
- J. Artz, T. E. Müller, K. Thenert, J. Kleinekorte, R. Meys, A. Sternberg, A. Bardow and W. Leitner, *Chem. Rev.*, 2018, **118**, 434–504.
- Y. Yang and J.-W. Lee, *Chem. Sci.*, 2019, **10**, 3905–3926.
- J. Davies, J. R. Lyonnet, D. P. Zimin and R. Martin, *Chem*, 2021, **7**, 2927–2942.
- B. Grignard, S. Gennen, C. Jérôme, A. W. Kleij and C. Detrembleur, *Chem. Soc. Rev.*, 2019, **48**, 4466–4514.
- Z. Zhang, J.-H. Ye, T. Ju, L.-L. Liao, H. Huang, Y.-Y. Gui, W.-J. Zhou and D.-G. Yu, *ACS Catal.*, 2020, **10**, 10871–10885.
- Y. Zhang, T. Zhang and S. Das, *Green Chem.*, 2020, **22**, 1800–1820.
- C. N. Matthews, J. S. Driscoll and G. H. Birum, *J. Chem. Soc., Chem. Commun.*, 1966, 736–737.
- (a) H. Sahoo, L. Zhang, J. Cheng, M. Nishiura and Z. Hou, *J. Am. Chem. Soc.* 2022, **144**, 23585–23594; (b) M. Schmalzbauer, T. D. Svejstrup, F. Fricke, P. Brandt, M. J. Johansson, G. Bergonzini and B. König, *Chem* 2020, **6**, 2658–2672; (c) J. Hou, A. Ee, W. Feng, J.-H. Xu, Y. Zhao and J. Wu, *J. Am. Chem. Soc.* 2018, **140**, 5257–5263; (d) M.-M. Wang, S.-M. Lu, K. Paridala and C. Li, *Chem. Commun.* 2021, **57**, 1230–1233; (e) S. Bhatt, A. Malik, A. Soni, B. M. Abraham, A. Sen and S. L. Jain, *J. CO<sub>2</sub> Util.* 2023, **67**, 102334; (f) P. K. Giri, P. Rani, M. Kaur, P. Beniwal, T. Janardhanan, D. Kumar and C. M. Nagaraja, *ACS Appl. Mater. Interfaces* 2025, **17**, 48301–48311.
- H. Bestmann, T. Denzel and H. Salbaum, *Tetrahedron Lett.*, 1974, **15**, 1275–1276.
- (a) H. Wang, Y. Gao, C. Zhou and G. Li, *J. Am. Chem. Soc.* 2020, **142**, 8122–8129; (b) S. Zhang, L. Li, D. Li, Y.-Y. Zhou and Y. Tang, *J. Am. Chem. Soc.* 2024, **146**, 2888–2894; (c) H. Hou, M. Luo, S. Zhai, T. Yuan, M. Zheng and S. Wang, *Green Chem.* 2024, **26**, 1317–1321; (d) C. Zhou, X. Wang, L. Yang, L. Fu and G. Li, *Green Chem.* 2022, **24**, 6100–6107.
- H. Zhou, G.-X. Wang, W.-Z. Zhang and X.-B. Lu, *ACS Catal.*, 2015, **5**, 6773–6779.
- L.-L. Liao, G.-M. Cao, Y.-X. Jiang, X.-H. Jin, X.-L. Hu, J. J. Chruma, G.-Q. Sun, Y.-Y. Gui and D.-G. Yu, *J. Am. Chem. Soc.*, 2021, **143**, 2812–2821.
- J. Hou, J.-H. Xie and Q.-L. Zhou, *Angew. Chem. Int. Ed.*, 2015, **54**, 6302–6305.
- (a) S.-S. Yan, T.-Y. Gao, Y. Liu, Y.-F. Chem, J.-Z. Zuo, Q.-F. Zhang, L. Song, W. Zhang, J.-H. Ye and D.-G. Yu, *Chem. Soc. Rev.*, 2025, **54**, 11583; (b) R. Cauwenbergh, V. Goyal, R. Maiti, K. Natte and S. Das, *Chem. Soc. Rev.*, 2022, **51**, 9371–9423.
- P. K. Sahoo, Y. Zhang and S. Das, *ACS Catal.*, 2021, **11**, 3414–3442.
- E. A. Quadrelli, G. Centi, J.-L. Duplan and S. Perathoner, *ChemSusChem*, 2011, **4**, 1194–1215. DOI: 10.1039/D6OB00221H
- J. Patricio, A. Angelis-Dimakis, A. Castillo-Castillo, Y. Kalmykova and L. Rosado, *J. CO<sub>2</sub> Util.*, 2017, **22**, 330–345.
- A. Lowry, R. E. Lynch, G. P. McGlacken and P. A. Byrne, *Org. Chem. Front.*, 2025, **12**, 6273–6282.
- See <http://njardarson.lab.arizona.edu/content/top-pharmaceuticals-poster>, and N. A. McGrath, M. Brichacek, J. T. Njardarson, *J. Chem. Educ.* 2010, **78**, 1348–1349.
- See, for example: A. C. Flick, H. X. Ding, C. A. Leverett, R. E. Jr. Kyne, K. K.-C. Liu, S. J. Fink and C. J. O'Donnell, *J. Med. Chem.*, 2017, **60**, 6480–6515.
- M. Limbach, in *Advances in Organometallic Chemistry*, ed. P. J. Pérez, Academic Press, 2015, vol. 63, pp. 175–202.
- M. N. Hopkins, K. Shimmei, K. B. Uttley and W. H. Bernskoetter, *Organometallics*, 2018, **37**, 3573–3580.
- G. Burkhart and H. Hoberg, *Angew. Chem. Int. Ed.*, 1982, **21**, 76–76.
- H. Hoberg, D. Schaefer and G. Burkhart, *J. Organomet. Chem.*, 1982, **228**, C21–C24.
- S. Li, W. Yuan and S. Ma, *Angew. Chem. Int. Ed.*, 2011, **50**, 2578–2582.
- X. Wang, M. Nakajima and R. Martin, *J. Am. Chem. Soc.*, 2015, **137**, 8924–8927.
- S. Wang, P. Shao, C. Chen and C. Xi, *Org. Lett.*, 2015, **17**, 5112–5115.
- Z.-X. Yang, L. Lai, J. Chen, H. Yan, K.-Y. Ye and F.-E. Chen, *Chin. Chem. Lett.*, 2023, **34**, 107956.
- M.-M. Wang, S.-M. Lu, K. Paridala and C. Li, *Chem. Commun.*, 2021, **57**, 1230–1233.
- M. Schmalzbauer, T. D. Svejstrup, F. Fricke, P. Brandt, M. J. Johansson, G. Bergonzini and B. König, *Chem*, 2020, **6**, 2658–2672.
- H. Sabet-Sarvestani, M. Izadyar and H. Eshghi, *J. CO<sub>2</sub> Util.*, 2017, **21**, 459–466.
- H. Sabet-Sarvestani, M. Izadyar, H. Eshghi and N. Noroozi-Shad, *Energy*, 2018, **145**, 329–337.
- R. Robiette, J. Richardson, V. K. Aggarwal and J. N. Harvey, *J. Am. Chem. Soc.*, 2006, **128**, 2394–2409.
- E. Vedejs and C. Marth, *J. Am. Chem. Soc.*, 1988, **110**, 3948–3958.
- B. E. Maryanoff and A. B. Reitz, *Chem. Rev.*, 1989, **89**, 863–927.
- A. W. Johnson, Ylides and Imines of Phosphorus, Wiley, New York, 1993, pp. 221–305.
- P. A. Byrne and D. G. Gilheany, *J. Am. Chem. Soc.*, 2012, **134**, 9225–9239.
- P. A. Byrne and D. G. Gilheany, *Chem. Soc. Rev.*, 2013, **42**, 6670–6696.
- (a) A. T. Nielsen and W. J. Houlihan, *Org. React.*, 2004, **16**, 1–438; (b) H. Tsuji, F. Yagi, H. Hattori and H. Kita, *J. Catal.* **1994**, **148**, 759–770.
- See supporting information Section 6. Optimisation Reactions for the Wittig CO<sub>2</sub> Utilisation Reaction.
- A. Labiche, A. Malandain, M. Molins, F. Taran and D. Audisio, *Angew. Chem.*, 2023, **135**, e202303535.
- V. Babin, F. Taran and D. Audisio, *JACS Au*, 2022, **2**, 1234–1251.
- A. Tortajada, Y. Duan, B. Sahoo, F. Cong, G. Toupalas, A. Sallustrau, O. Loreau, D. Audisio and R. Martin, *ACS Catal.*, 2019, **9**, 5897–5901.



## COMMUNICATION

ChemComm

View Article Online  
DOI: 10.1039/D6OB00221H

47. D. B. G Williams and M. Lawton, *J. Org. Chem.* 2010, **75**, 8351–8354.
48. D. F. Shriver, *The Manipulation of Air-Sensitive Compounds, 2nd Edition*. John Wiley & Sons: New York: 1986.
49. R. E. Ireland and R. S. Meissner, *J. Org. Chem.* 1991, **56**, 4566–4568.
50. D. S. Pedersen and C. Rosenbohm, *Synthesis* 2001, **16**, 2431–2434.
51. F. Dornhaus, M. Bolte, H.-W. Lerner and M. Wagner, *Eur. J. Inorg. Chem.* 2006, 1777–1785.
52. I. Burkhardt and J. S. Dickschat, *Chem. Commun.* 2018, **54**, 3540–3542.
53. M. Zhu, W. Yu, Q. Zhong, B. Cui, C. Cao and Y. Shi, *Tetrahedron* 2023, **135**, 133321.
54. D. I. Bugaenko, A. A. Volkov, M. V. Livantsov, M. A. Yurovskaya and A. V. Karchava, *Chem. Eur. J.* 2019, **25**, 12502–12506.
55. D. Pedersen, Dry Column Vacuum Chromatography (DCVC) Tutorial. Youtube video, 2017. URL: <https://www.youtube.com/watch?v=IBNhu4kJ4Mc>.
56. T. Brégent, J.-P. Bouillon and T. Poisson, *Org. Lett.* 2020, **22**, 7688–7693.
57. R. Ruzi and W. Shu, *Org. Lett.*, 2024, **26**, 7926–7931.
58. P. Gao, L. Liu, Z. Shi and Y. Yuan, *Org. Biomol. Chem.*, 2016, **14**, 7109–7113.
59. X. Ma, X. Yan, J. Yu, J. Guo, J. Bian, R. Yan, Q. Xu and L.-B. Han, *Green Chem.*, 2025, **27**, 102–108.
60. M. Spengler, R. Y. Dong, C. A. Michal, M. Pfletscher and M. Giese, *J. Mater. Chem. C* 2017, **5**, 2235–2239.
61. M. Abe, K. Nishikawa, H. Fukuda, K. Nakanishi, Y. Tazawa, T. Taniguchi, S.-Y. Park, S. Hiradate, Y. Fujii, K. Okuda and M. Shindo, *Phytochemistry* 2012, **84**, 56–67.
62. M. Gucma, W. M. Gołębiewski and M. Krawczyk, *RSC Adv.* 2015, **5**, 13112–13124.
63. G. Zweifel and R. A. Lynd, *Synthesis* 1976, 625–626.
64. G. S. Ananthnag, J. T. Mague and M. S. Balakrishna, *Dalton Trans.* 2015, **44**, 3785–3793.
65. Y. Zhao, T. Feng, G. Li, F. Liu, X. Dai, Z. Dong and X. Qiu, *RSC Adv.* 2016, **6**, 42482–42494.
66. S. R. Kandukuri, J. A. Schiffner and M. Oestreich, *Angew. Chem. Int. Ed.*, 2012, **51**, 1265–1269.
67. B. Zhao and B. Xu, *Org. Biomol. Chem.*, 2021, **19**, 568–573.
68. Z. He, M. Hu, T. Luo, L. Li and J. Hu, *Angew. Chem. Int. Ed.*, 2012, **51**, 11545–11547.
69. C. J. Hastings, N. P. Adams, J. Bushi and S. J. Kolb, *Green Chem.*, 2020, **22**, 6187–6193.
70. E. M. Brun, S. Gil, R. Mestres and M. Parra, *Tetrahedron* 1998, **54**, 15305–15320.
71. D. Szamosvári, M. Prothiwa, C. L. Dieterich and T. Böttcher, *Chem. Commun.* 2020, **56**, 6328–6331.
72. G. Cahiez, D. Bernard and J. F. Normant, *Synthesis* 1976, 245–248.
73. R. Garrison Kinney, J. Zgheib, P.-L. Lageux-Tremblay, C. Zhou, H. Yang, J. Li, D. R. Gauthier Jr. and B. A. Arndtsen *Nature Chemistry* 2024, **16**, 556–563.



## Data availability

The data supporting this article (synthetic details, experimental methods, and characterisation data (including copies of NMR spectra)) have been included as part of the Supplementary Information. Note: The authors have cited additional references within the Supplementary Information.<sup>47–73</sup>

