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Slippery liquid-like surfaces as a promising solution for sustainable drag reduction†

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Drag reduction is crucial for many industries, ranging from aerospace to microfluidics, to enhance the energy efficiency and reduce costs. This work is the first to study drag reduction enabled by novel slippery liquid-like surfaces fabricated from flexible polymers. We experimentally characterized the drag reduction performance of slippery liquid-like surfaces in the laminar flow regime. Our results indicate that liquid-like surfaces can reduce fluid drag regardless of surface wettability and have achieved nearly 20% drag reduction. Furthermore, the durability tests show that these surfaces can maintain slipperiness over a month when exposed to air or water and the drag reduction capability for at least one week under a fluid flow. These findings highlight the

potential of slippery liquid-like surfaces as a promising solution for sustainable drag reduction.

1. Introduction

Drag reduction plays key roles in various engineering and industrial applications, particularly in the petroleum industry,^{1,2} aerospace engineering,^{3,4} microfluidic devices,^{5,6} liquid cooling systems,^{7,8} and marine transportation.^{9,10} Reducing the flow drag not only improves the energy efficiency, but also increases the overall performance and lifespan of systems by minimizing friction between fluid and solid surfaces, for example in underwater vehicles,¹¹ aviation,² pipeline transportation,¹² wind turbine blade design,¹³ automotive aerodynamics,³ and droplet manipulation^{14–24} where fluid flow resistance can lead to significant energy losses. Effective drag reduction strategies can lead to substantial cost savings and environmental benefits. In microfluidics, drag reduction can improve the pumping efficiency and achieve higher throughput²⁵ during DNA analysis,²⁶ cell sorting²⁷ and blood diagnosis,²⁸ as well as limiting shear damage, aggregation and sedimentation in medical flows.²⁹

Surface engineering has been widely employed to reduce drag by generating fluid slip at the fluid–surface interface.^{30,31} When the fluid flows over a solid surface, the no-slip boundary condition (zero flow velocity at the fluid–solid interface) is generally applied. However, if fluid slip appears at the interfaces, as shown in Fig. 1(a), the Navier slip boundary condition is applicable. This condition states that the slip velocity (the flow velocity along the streamwise direction at the slipping interface) is proportional to the rate of strain at the interface. Mathematically, it can be expressed as:

$$u_{\text{slip}} = b \frac{\partial u}{\partial y} \Big|_{\text{w}} \quad (1)$$

where u_{slip} is the slip velocity, b is the slip length (up to the order of μm), and $\frac{\partial u}{\partial y} \Big|_{\text{w}}$ is the shear rate at the wall. In existing

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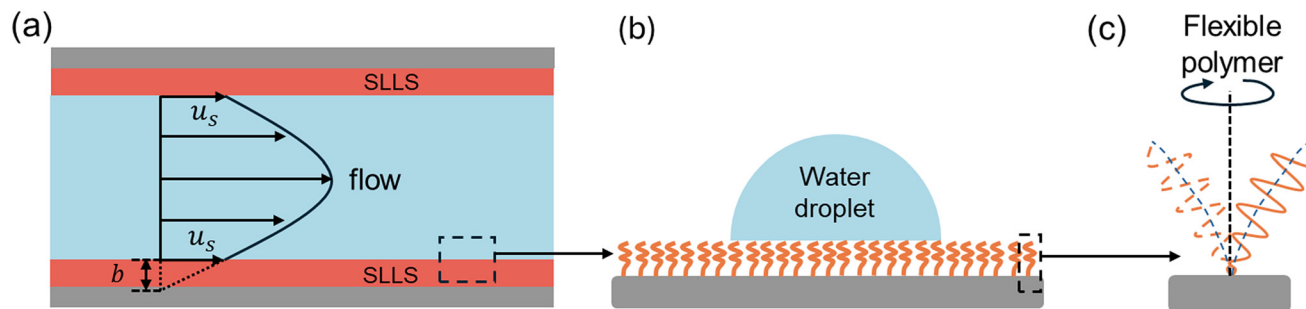


Fig. 1 (a) The schematic of a microchannel with slippery liquid-like surfaces (SLLSs) as the channel surfaces. When liquid is running through the microchannel, a fluid slip will be induced at the liquid–SLLS interface. The u_s is the fluid slip velocity at the liquid–SLLS interface, and b is the slip length which is the extended length to which the fluid velocity vanishes. (b) Zoom-in image of the SLLS made by grafting (c) flexible polymers to a substrate.

studies, various surface treatments and coatings have been developed to induce fluid slip for drag reduction. For example, superhydrophobic surfaces^{32–34} (SHSs) utilize micro/nanostructures and low surface energy materials to trap air when the liquid flows over them. The liquid is in contact with solids only at the peaks of the micro/nanostructures, while the rest of the surface is exposed to air, which has nearly zero viscosity. The configuration enables the liquid to move over the trapped air with significantly reduced friction, creating apparent fluid slip at liquid–air interfaces.^{33,35–39} Benefiting from the ultra-low viscosity of air, SHSs can achieve high drag reduction performance. Lee *et al.* summarized experimental slip measurement on SHSs and reported 5%–60% drag reduction performance.⁴⁰ However, the air–liquid interface of SHSs will break up and result in the loss of trapped air under increased external pressure.⁴¹ In addition, while the hydrostatic pressure and water flow rate increase, the dissolution ability of air in water will increase due to the mass transfer forced convection.⁴² Therefore, SHSs often fail to maintain the fluid slip due to the air depletion under a liquid flow. For example, Xiang *et al.* demonstrated that SHSs lost non-wettability after 30 minutes under a liquid flow.⁴³ To restore the slipperiness of SHSs, various methods for regenerating the trapped air in SHSs have been proposed, including *in situ* gas generation via water decomposition, heating, and gas injection through gas permeable base materials. For instance, Lloyd *et al.* demonstrated an electrochemical controlling strategy to replenish hydrogen from water decomposition.⁴⁴ Sarannndhi *et al.* regenerated the plastron of SHSs by activated heating of SHSs to establish the Leidenfrost state at a relatively low superheat temperature.⁴⁵ Breveleri *et al.* restored the plastron in 0.3 seconds by injecting a high-pressure gas from the bottom of a porous material.⁴⁶ Similarly, Zhao *et al.* injected bubbles to recover the plastron by utilizing the sparse nanostructure with low surface energy.⁴⁷ Nevertheless, these active methods for enhancing the stability of SHSs increase the complexity level of operation and make SHSs not suitable in certain applications, such as microfluidic electrokinetic energy conversion devices. Besides, slippery liquid-infused porous surfaces⁴⁸ (SLIPs) have been designed to mimic the slipperiness of natural surfaces, such as Nepenthes pitcher plants, by infusing

liquid lubricants into the surface structures. An immiscible liquid can move over the lubricant layer with a significant fluid slip.^{49–53} Studies reported that SLIPs can achieve 7%–50% drag reduction by optimizing the surface structure design and infused liquid lubricant.⁵ Despite their effectiveness, SLIPs suffer from lubricant loss due to shear force under a fluid flow and would fail within minutes to 24 hours under a flow.^{54–59} The lack of robustness has historically impeded the successful implementation of SHSs and SLIPs.

Recently, a novel type of slippery liquid-like surface has been developed by grafting flexible polymers onto substrates.^{60–66} These surfaces exhibited several unique properties. First, slippery liquid-like surfaces have flexible polymers that behave like a liquid lubricant layer as shown in Fig. 1(b) and (c). Zhao *et al.* reported the macroscopic evidence of the Couette-like flow behavior of flexible polymers under shear with an effective viscosity,⁶⁷ which enables slippery liquid-like surfaces to induce fluid slip when liquids flow over them. Second, slippery liquid-like surfaces have excellent liquid repellency. These surfaces can repel liquids with various surface tensions and contact angles. As flexible polymers behave like a liquid lubrication layer, the friction between the surface and liquids in contact is greatly reduced. Wang and McCarthy *et al.* reported that slippery liquid-like surfaces show ultralow contact angle hysteresis (CAH < 1°).⁶⁶ Zhang *et al.* also reported that slippery liquid-like surfaces show negligible contact angle hysteresis (CAH < 1.5°) to water and highly wetting liquids, e.g. FC72 ($\gamma = 10 \text{ mN m}^{-1}$), Krytox 101 ($\gamma = 17 \text{ mN m}^{-1}$), and hydrocarbon oils.⁶³ Lastly, slippery liquid-like surfaces also have shown exceptional durability. By circumventing the use of fragile micro/nano-textures and air/liquid lubricants, the non-textured all-solid features of slippery liquid-like surfaces mitigate the mechanical damage and loss of the lubricant. Zhang *et al.*⁶³ showed that slippery liquid-like surfaces remained durable even after 1000 cycles of mechanical abrasion/adhesion tests and scratch damage, while the SLIPs lost the liquid repellency in the same test. Evidently, slippery liquid-like surfaces have the potential to solve the long-standing durability issue associated with SHSs and SLIPs for sustainable drag reduction benefiting from the



abovementioned unique properties. Despite these advantages, there is little work on drag reduction over slippery liquid-like surfaces. Currently, most studies still focus on liquid-repellency,^{68,69} anti-icing,^{62,70–74} anti-fouling,^{75,76} and enhanced condensation^{77,78} enabled by slippery liquid-like surfaces. Thus, this work is the first to experimentally characterize the drag reduction performance and durability of SLLSs under a liquid flow.

In this study, we experimentally quantified the drag reduction enabled by both hydrophobic and hydrophilic slippery liquid-like surfaces at different flow rates and Reynolds numbers within the laminar flow regime. We also tested the durability of slippery liquid-like surfaces under hydrostatic pressure and hydrodynamic flow conditions. Our results demonstrated that slippery liquid-like surfaces could induce drag reduction regardless of their wettability. The tested surfaces have generated effective drag reduction in the range of 6.4%–21.1%. Moreover, slippery liquid-like surfaces have shown remarkable durability, especially the hydrophilic one has maintained its drag reduction capability for at least one week under a continuous flow.

2. Methods

2.1. Liquid-like surface fabrication

2.1.1. Hydrophobic slippery liquid-like surface. We fabricated the hydrophobic slippery liquid-like surface by grafting chlorine-terminated poly(dimethylsiloxane) (Gelest Inc.) polymer brushes onto a silicon (Si) substrate with the vapor deposition approach.⁶⁰ The Si substrate was pre-cleaned with acetone, isotropy alcohol (IPA), and deionized (DI) water and dried with an air nozzle subsequently. Then, a 20 minute oxygen-based plasma treatment was applied to the Si substrate to clean the substrate and create hydroxyl groups. The plasma treated surface was then placed upside down on the cover of a Petri dish, containing 800 μL of liquid chlorine-terminated with the corresponding molar weight (425 g mol^{-1}) which is denoted as V₄₂₅ below. The assembled Petri dish was placed in a vacuum oven at 60 °C for 1 hour. Finally, the substrates were rinsed in a toluene bath in an ultrasonic shaker for 2 min, cleaned with acetone, IPA, and DI water sequentially to remove untethered residue chemicals and dried with an air nozzle.

2.1.2. Hydrophilic slippery liquid-like surface. We fabricated the hydrophilic liquid-like surface⁶⁴ by covalently grafting polyethylene glycol (PEG) brushes to bare Si substrates. The Si substrates were cleaned with acetone and DI water and then dried in air. The cleaned surfaces were exposed to oxygen plasma for 10 minutes for hydroxylation. The hydroxylated substrates were then immersed in a mixture of solution consisting of 2 μL of 2-methoxy polyethyleneoxy (6–9) propyl trimethoxysilane (Gelest Inc.), 12 μL of hydrochloric acid (Fisher Inc.), and 45 mL of toluene (Sigma-Aldrich) in a glass Petri dish. Finally, the grafted samples were rinsed in a toluene bath for 2 minutes in an ultrasonic shaker and cleaned with DI water and ethanol (Sigma-Aldrich) sequentially and dried in air.

2.2. Surface characterization

The wetting properties of a surface consist of contact angle (CA), contact angle hysteresis (CAH), and sliding angle (SA). The CA was measured by depositing 5 μL DI water droplets onto flat testing surfaces using an automated dispenser and a goniometer (Ramehart 290) at room temperature under ambient conditions. The test surface was positioned horizontally. A 5 μL DI droplet was first dispensed very slowly to avoid splashing or irregular shape formation. Then, the droplet was placed on the testing surface and the dispenser was carefully removed. A slight vibration (*i.e.*, expansion and contraction) of the droplet was initially observed after the droplet was placed onto the testing surface. Once there was no obvious vibration of the droplet, the representative CA was quickly measured 10 times to mitigate the effects of droplet evaporation. The representative CA is the value by averaging the 10 readings of the contact angles at the left and right edges. This process was repeated at three independent positions on the same testing surface to check the variation in wetting properties on the surface. (The CA at the left and right edges are provided in ESI S2.1.†) The CAH was measured by tilting the surface slowly with 0.3 degrees per s until the droplet started sliding on the surface.⁷⁹ The angle at which the surface was tilted was the SA. The difference between the contact angle at the advancing edge (θ_{adv}) and the receding edge (θ_{rec}) is defined as CAH. All CAH and SA measurements were obtained on 3 independent positions on each surface. The surface roughness was measured by atomic force microscopy (AFM) with a tapping mode. The surface morphology was examined with scanning electric microscopy (SEM) (see ESI S1†). The coating thickness was analyzed with an ellipsometer at a 70-degree inclined angle with a He–Ne laser ($\lambda = 900 \text{ nm}$). The Cauchy model was applied to calculate the thickness of the polymer brush layer. The polymer brush layer thickness measurements are also obtained on 3 different positions on the surfaces.

2.3. Drag reduction measurement

Pressure drop measurement in a microchannel is a standard method to quantify the drag reduction for engineered surfaces.⁸⁰ We designed a 3-layer sandwich structured microchannel to perform drag reduction measurement. As shown in Fig. 2, the top polycarbonate plane and the bottom test surface are separated using a silicone rubber sheet (thickness = 0.005", McMaster-Carr). A rectangular shape (length = 76.5 mm, width = 10 mm) was cut at the center of the rubber sheet to define the microchannel shape and dimensions. In a flow experiment, the DI water was pumped through a microchannel using a syringe pump (Havard Apparatus Pump 11 Elite) using 5 controlled flow rates of 20, 22.5, 25, 27.5, and 30 mL min^{-1} . A differential pressure sensor with a high accuracy of 2 Pa (Omega Engineering PX459-10WDWU5 V) was used to measure the pressure drop across the microchannel. The drag reduction measurement was performed fifteen times at each flow rate and the average value of all measurements was used (see ESI S4†). Because the channel height slightly



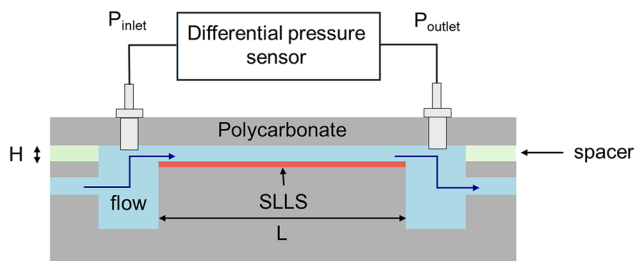


Fig. 2 Schematic of the drag reduction measurement setup (assembling details see ESI Fig. S3†). The top polycarbonate plate and testing slippery liquid-like surface (SLLS) are separated by a thin silicone rubber spacer. The microchannel has dimensions with length $L = 76.5$ mm and width $W = 10$ mm. In drag reduction measurement, water was pumped through the microchannel at controlled flow rates using a syringe pump. The differential pressure sensor was pumped through the microchannel at controlled flow rates using a syringe pump. The differential pressure sensor was connected to the ports in the inlet and outlet reservoirs to measure the pressure drop across the microchannel.

changes for each assembled microfluid channel, the channel height for each experiment was also measured independently. To accurately determine the channel height, after the flow experiment of each test surface, liquid silicone rubber (Ecoflex 00-50) was injected into the microchannel without opening the microchannel.⁵⁹ After 12 hours of curing, the solidified silicone rubber was removed from the disassembled microchannel device. The thickness of the silicone rubber was measured at three independent positions with a digital profilometer (Keyence, VHX-6000).

2.4. Durability test

2.4.1. Durability test under hydrostatic pressure. To test the durability of slippery liquid-like surfaces under hydrostatic pressure, one set of surfaces was exposed to air and another set of surfaces was immersed in DI water for 30 days. The wetting properties (*i.e.*, CA, CAH, and SA values) were characterized every day on 3 different positions respectively using a goniometer. We also examined the durability of slippery liquid-like surfaces in acidic and alkaline solutions by immersing each surface in solutions with different pH values (pH = 1, 4, 10, 13) (see ESI S7†).

2.4.2. Durability test under a hydrodynamic flow. To test the durability of slippery liquid-like surfaces under hydrodynamic flow, we connected the assembled microchannel with a circular pump and used a flow rate of 20 mL min^{-1} to continuously pump water flows over slippery liquid-like surfaces for one week or until the surface failed (depending on which comes earlier). The pressure across the microchannel was measured four times every day.

3. Results and discussion

3.1. Surface slipperiness and drag reduction performance

Table 1 and Fig. 3 show the surface characterization results. As shown in Table 1 and Fig. 3(a) and (d), the hydrophobic liquid-

Table 1 Surface characterization results of the fabricated slippery liquid-like surfaces

	V_425	PEG
CA ($^{\circ}$)	104.7 ± 0.5	42.2 ± 2.3
SA ($^{\circ}$)	5.6 ± 1.7	6.5 ± 0.4
Advancing angle ($^{\circ}$)	105.6 ± 0.4	43.8 ± 2.3
Receding angle ($^{\circ}$)	104.0 ± 0.5	42.0 ± 2.3
CAH ($^{\circ}$)	1.5 ± 0.5	1.7 ± 0.4
Thickness (nm)	6.5	22.0
Roughness (nm)	2.14	0.81

like surface has a CA of around 104.7° and the hydrophilic liquid-like surface has a CA of around 42.2° . Both hydrophobic and hydrophilic liquid-like surfaces show exceptional slipperiness from their ultra-low CAH and SA values as shown in Fig. 3(b) and (e). The standard deviation values presented in Table 1 indicate the degree of variation in the measured CA and CAH values across different positions and readings. These values demonstrate that the observed uncertainties and potential experimental errors are minimal, ensuring the reliability and reproducibility of the measurements. (All measured advancing angles and receding angles at three positions are provided in ESI S2.2.†) The ultra-low CAH of the slippery liquid-like surface is a result of its low physical and chemical inhomogeneities.⁶⁴ From the AFM measurement in Fig. 3(c) and (f), liquid-like surfaces are ultra-smooth, which induces low physical inhomogeneity. Additionally, the liquid-like surface fabrication methods we employed from the literature^{60,65} will lead to a high grafting density that also gives low chemical inhomogeneity.⁶⁴

The pressure drop across the microchannel was measured to determine the drag reduction performance of the test surface. For a solid surface with the no-slip boundary condition, in the laminar flow regime, the measured pressure drop ($\Delta P_{\text{no-slip}}$) will follow the theoretical value of the Poiseuille flow:

$$\Delta P_{\text{no-slip}} = \frac{12\mu LQ}{WH^3} \quad (2)$$

where Q represents the flow rate in the microchannel; L , W , and H denote the microchannel length, width, and height, respectively; and μ is the dynamic viscosity of DI water which is about 0.001 Pa s .⁸¹ We first validated the accuracy of our experiment setup by measuring the pressure drop when water flowed over a smooth Si surface which functioned as the reference no-slip surface. The measured results showed no drag reduction on the smooth Si surface (see ESI S5†).

When fluid slip appears at the test surfaces, the measured pressure drop ΔP_{slip} is expected to be lower than $\Delta P_{\text{no-slip}}$ at the no-slip condition. The drag reduction (DR) coefficient of the test surface can be calculated from:⁵⁹

$$\text{DR} = \frac{\Delta P_{\text{no-slip}} - \Delta P_{\text{slip}}}{\Delta P_{\text{no-slip}}} \quad (3)$$

Fig. 4 shows the experimental results of pressure drop using different flow rates for hydrophobic (V_425) and hydro-



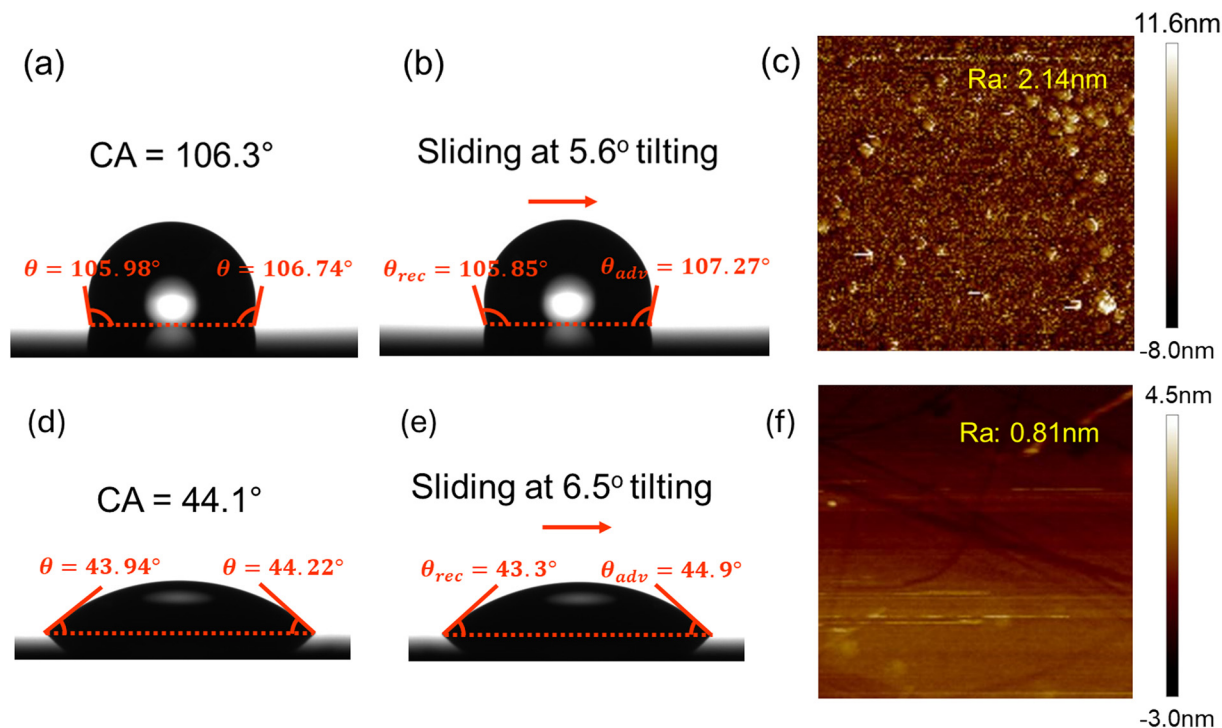


Fig. 3 (a)–(c) present the single measurement of CA, CAH and SA values and the AFM roughness measurement on a $3 \mu\text{m} \times 3 \mu\text{m}$ area for the hydrophobic slippery liquid-like surface. (d)–(f) show the single measurement of CA, CAH and SA values and the AFM roughness measurement using a $3 \mu\text{m} \times 3 \mu\text{m}$ area for the hydrophilic slippery liquid-like surface.

philic (PEG) slippery liquid-like surfaces. The pressure drop approximately increased linearly with the applied flow rate. As illustrated in Fig. 4, for both hydrophobic and hydrophilic slippery liquid-like surfaces, the experimentally measured pressure drop values are smaller than $\Delta P_{\text{no-slip}}$. This suggests that both surfaces can reduce the pressure drop across the microchannel at a fixed flow rate and induce drag reduction.

From the pressure drop measurements in Fig. 4, we can obtain the dependence of the drag reduction (DR) coefficient

on the flow Reynolds number (Re), which is shown in Fig. 5. Re for a rectangular microchannel is defined as:

$$Re = \frac{UD_h}{\nu} \quad (4)$$

where U is the average flow velocity, ν is the kinematic viscosity of the fluid (*i.e.*, water in our experiment), and $D_h = 2W \times H / (W + H)$ is the hydraulic diameter of a rectangular channel.

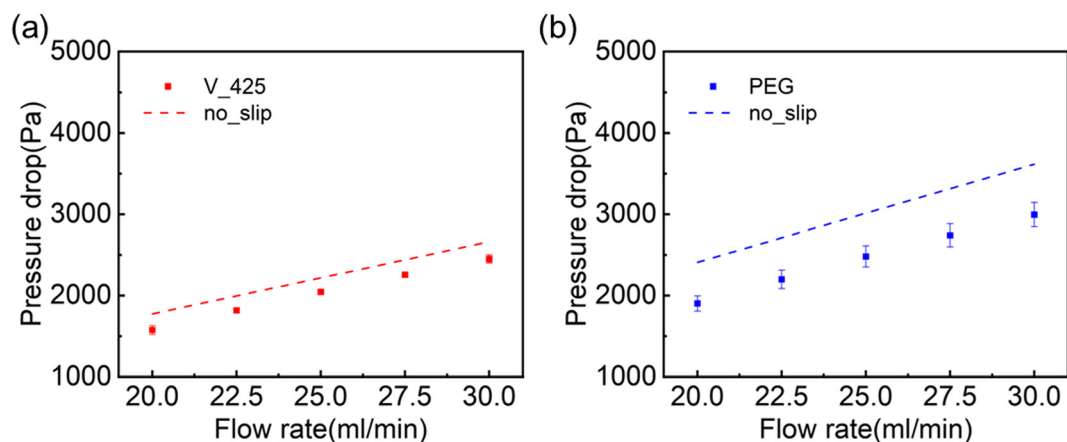


Fig. 4 Shows the pressure drop measurements for both (a) hydrophobic (V_425) and (b) hydrophilic (PEG) slippery liquid-like surfaces at different flow rates. The dots represent the experimental results, while the dashed lines represent the theoretical pressure values at the no-slip boundary condition.



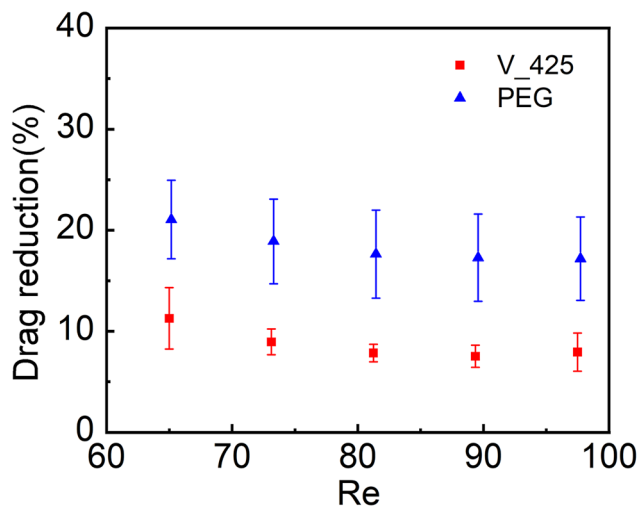


Fig. 5 Drag reduction coefficients at different Re numbers for slippery liquid-like surfaces.

As illustrated in Fig. 5, the hydrophilic liquid-like surface (PEG) has achieved a DR coefficient in the range of 17.2%–21.1%, while the hydrophobic liquid-like surface (V_425) has achieved a DR coefficient in the range of 6.4%–10.2%. The measured drag reduction performance of the slippery liquid-like surfaces could be comparable to those of SHSs (DR coefficient^{40,82,83} is 5%–60%) and SLIPSs (DR coefficient^{5,48} is 7%–50%). These results also demonstrated that slippery liquid-like surfaces could induce drag reduction regardless of their wettability. This is mainly because the key to generating drag reduction is the presence of a slipping interface. As the slippery liquid-like surfaces behave like a liquid lubricant under shear,⁶⁷ a slipping liquid-like/liquid interface will be created once water flows over these surfaces. Thus, both hydrophobic and hydrophilic slippery liquid-like surfaces can achieve drag reduction. It is also noteworthy that the DR coefficient slightly decreased as the Re number (or flow rate) increased. The fluid slip at a slipping interface is $b \sim \frac{\mu_f}{\mu_l} h_0$

(where μ_f is the flowing fluid dynamic viscosity, μ_l is the fluid lubricant dynamic viscosity, and h_0 is the lubricant layer thickness). According to Zhao *et al.*, the apparent viscosity of the slippery liquid-like surface slightly decreases with increasing shear rate.⁶⁷ Therefore, μ_l of the liquid-like lubricant layer will decrease slightly when the flow rate (*i.e.*, Re that is varied by changing the flow rate in our experiment) increases. However, at the same time, the larger shear rate at a higher applied flow rate will cause a higher degree of deformation or bending of the flexible polymers, thus reducing the effective lubrication layer thickness h_0 . Combining both the effects of μ_l and h_0 , the reason for the slightly reduced DR coefficient at a higher Re (or flow rate) could be the domination of reduction in h_0 over the decrease of μ_l when increasing the shear rate in our current applied range. Due to the competing effect between μ_l and h_0 on the fluid slip, we also suspect that there could exist an optimal shear rate at which the DR coefficient reaches the

maximum value, which will be tested in a future study. Furthermore, the higher DR coefficient achieved by the hydrophilic liquid-like surface compared to that of the hydrophobic one could be due to its higher coating thickness or h_0 which will lead to a larger fluid slip.

3.2. Exceptional durability of liquid-like surfaces

In the air environment of the hydrostatic pressure durability test, as shown in Fig. 6(a)–(c), the hydrophobic liquid-like surface remained hydrophobic and its slipperiness after 30 days. For the hydrophilic liquid-like surface, although its CA value started rising after 20 days (but still being hydrophilic), the surface still had low SA and CAH after 30 days and sustained good slipperiness. In the DI water environment, as shown in Fig. 6(d)–(f), the hydrophilic liquid-like surface kept its hydrophilicity and slipperiness after 30 days. For the hydrophobic liquid-like surface, the CA decreased but the surface still had good slipperiness based on its low CAH and SA. The hydrostatic pressure durability testing results demonstrated that the slippery liquid-like surface could sustain its slipperiness for at least 30 days in both air and liquid environments. We also assessed the acidic and alkaline resistance on slippery liquid-like surfaces (see ESI S7† for the testing results). Both hydrophobic and hydrophilic slippery liquid-like surfaces can maintain the surface wetting properties for over hundreds of hours in relatively weak acid (pH = 4) and base (pH = 10) solutions but fail within a couple of hours in strong acid (pH = 1) and base (pH = 13) solutions. In general, the PEG surface demonstrated higher resistance in an acidic environment, and the V_425 surface exhibited higher resistance in a base environment.

The adhesion of foulants, such as dust, organic oil and biomolecules, can lead to the loss of original surface properties.⁷⁶ For example, some smooth surfaces, such as metal and rare earth oxides, can transition from being hydrophilic to hydrophobic due to the adsorption of organic contaminants present in the air within 4 days.⁸⁴ To mitigate this contamination, a clean storage strategy utilizing ultraclean nanotextured storage media as a getter has been proposed.⁸⁵ The proposed storage device can maintain surface cleanliness for more than 1 week and passively clean contamination on the sample.⁸⁵ Remarkably, slippery liquid-like surfaces have been reported to reduce the adhesion of various contaminations in previous studies. For instance, Dai *et al.* obtained ultra-low CAH with Krytox101 (CAH = 0.5°), FC40 (CAH = 0.4°), and FC72 (CAH = 0.2°) droplets on slippery liquid-like surfaces, which indicated high repellency to various organic oils.⁶¹ Besides, our static durability test results in air, as shown in Fig. 6(a)–(c), further confirm that the hydrophobic slippery liquid-like surfaces remained hydrophobic and maintained their slipperiness after 30 days. For hydrophilic slippery liquid-like surfaces, although its CA value started rising after 20 days (but still being hydrophilic), the surface still had a low SA and CAH after 30 days and sustained good slipperiness. The results indicated that both slippery liquid-like surfaces can maintain wettability and slipperiness in the air environment for a long time, which demonstrates mitigated adsorption of organic contamination



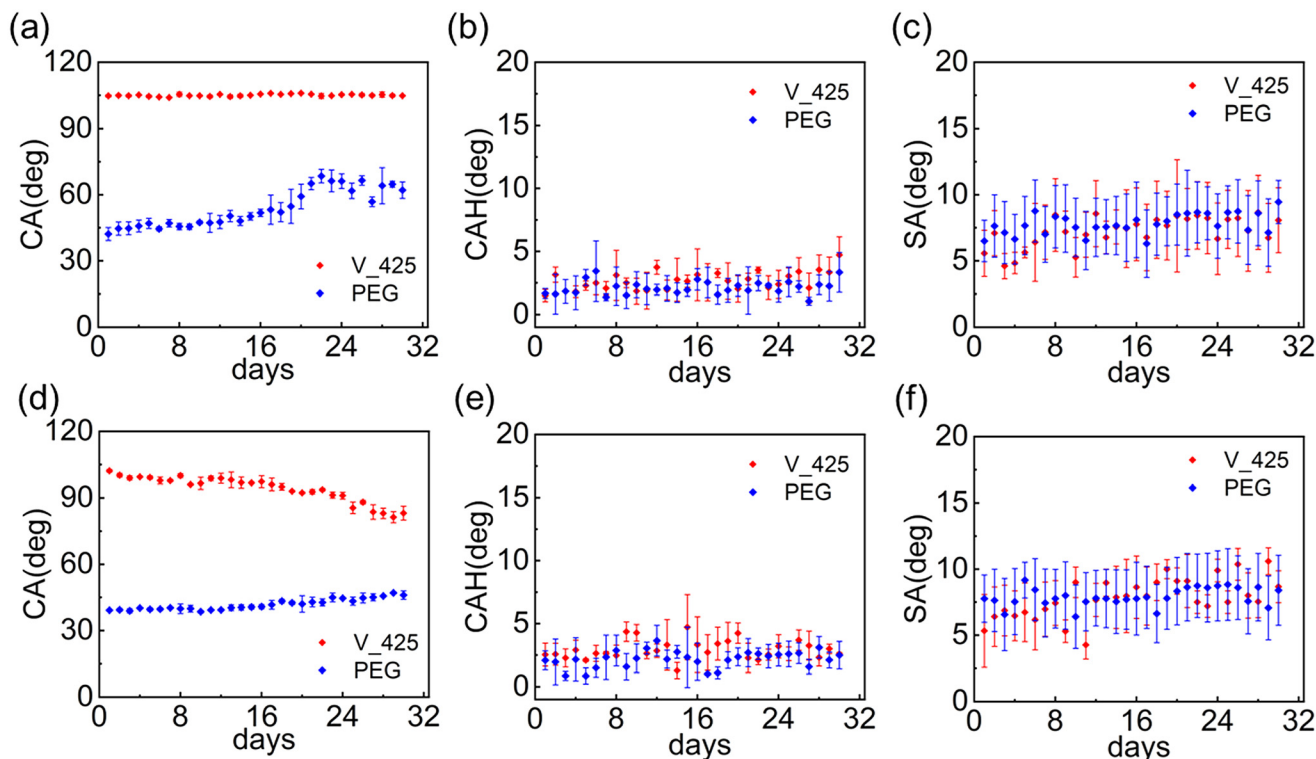


Fig. 6 (a)–(c) Change in the wetting properties *i.e.*, CA, CAH, and SA, of slippery liquid-like surfaces with time while exposed in an air environment; (d)–(f) change in wetting properties, *i.e.*, CA, CAH, and SA, of the slippery liquid-like surface while being immersed in DI water.

in air. Furthermore, Peng *et al.* embedded PDMS chains into the fluorinated ethylene-(hydroxyl-alkyl) vinyl ether (FEVE) matrix to develop solar cells with self-cleaning properties, addressing the issue of dust contamination.⁸⁶ Their group also presented exceptional biofilm suppression, such as bacterial cells, on hydrophobic slippery liquid-like surfaces.⁸⁷

Fig. 7 shows the results of the durability test under a hydrodynamic flow. When the surface loses its drag reduction capability, the pressure drop will increase significantly compared

to the initial value. From the testing results in Fig. 7, the hydrophobic liquid-like surface (V_425) could keep its drag reduction capability for 56 hours at which its DR coefficient reduced from 9% to 0%. Remarkably, the hydrophilic liquid-like surface (PEG) could still achieve 10% after 160 hours (one-week continuous flow, initial DR coefficient = 19%). In contrast, the SHSs would lose their slipperiness after 30 minutes under a liquid flow⁴³ and SLIPs lost the drag reduction capability after 14 minutes or 24 hours under laminar flow in a

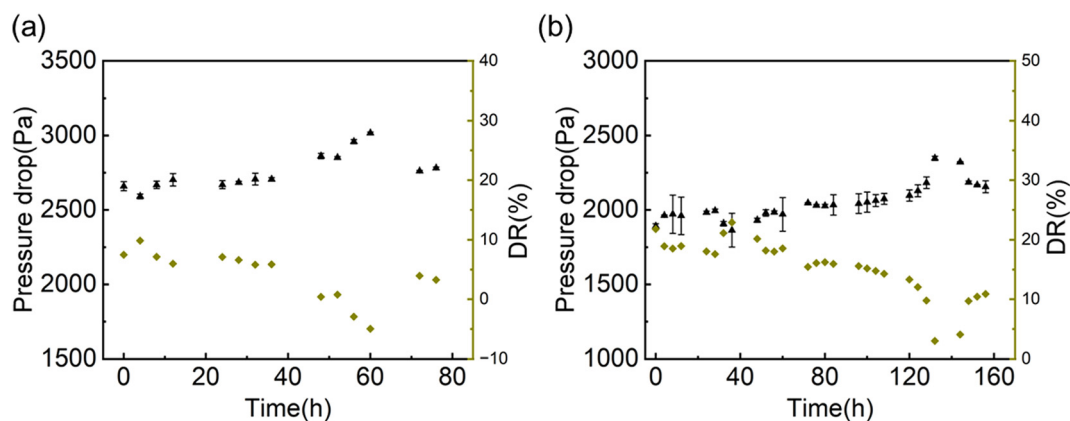


Fig. 7 (a) Hydrodynamic durability test shows the change in the pressure drop (left axis) and drag reduction coefficient (right axis) over time for the hydrophobic slippery liquid-like surface. The drag reduction reduced below 0 at 56 h. (b) Hydrodynamic durability test shows the change in pressure drop (left axis) and drag reduction (right axis) over time for the hydrophilic slippery liquid-like surface. The drag reduction remained around 10% after 1 week.



microchannel.^{54–59} Our results suggest that liquid-like surfaces, regardless of their wettability, have much better durability compared to SHSs and SLIPs. The possible cause of the degradation of liquid-like surfaces under liquid flow could be the loss of flexible polymers with time. To further increase the durability of liquid-like surfaces, the multi-point grafting method (*i.e.*, one flexible polymer will be grafted to the substrate *via* several grafting points) can be employed in the future to reduce flexible polymer loss.^{88–92}

4. Conclusion

This study represents the first demonstration of the significant potential of slippery liquid-like surfaces for drag reduction and their outstanding durability. The drag reduction coefficient could reach as high as 20%. Our experimental results also highlighted the effectiveness of slippery liquid-like surfaces in maintaining a stable, low-friction interface that significantly diminished the overall drag force experienced by the fluid. The comparison with SHSs and SLIPs revealed the distinct advantages of slippery liquid-like surfaces. While SHSs and SLIPs have shown impressive drag reduction capabilities, their long-term durability issue presents a considerable challenge for practical applications. In contrast, slippery liquid-like surfaces provide a more robust and reliable solution, making them highly suitable for a wide range of microfluidic applications, including those in biomedical devices and energy conversion technologies. For example, in microfluidic energy devices (*e.g.*, electrokinetic energy converters), drag reduction can improve liquid transport and thus enhance the energy conversion efficiency.^{93–96} In microfluidic biomedical devices, drag reduction can improve blood circulation, aiding the treatment of cardiovascular diseases.⁹⁷ The findings of this study contribute to the fundamental understanding of fluid dynamics over novel slippery liquid-like surfaces and open new avenues for the design and optimization of high-performance microfluidic devices. They will also trigger the interest of broad research communities in this direction to push it moving forward. To facilitate the utilization of slippery liquid-like surfaces in flow applications, future research will focus on systematic fundamental understanding of the mechanism of drag reduction and surface degradation of liquid-like surfaces, effects of the design factors of liquid-like surfaces on their drag reduction performance to optimize the surface design, their drag reduction performance in the turbulent flow regime, large-scale surface fabrication, and further improvement of slippery liquid-like surface durability. The insights gained from this work will be instrumental in advancing the field of drag reduction and pave the way for innovative solutions to improve the performance of broad microfluidic technologies.

Author contributions

H. L. and B. F. conceived the ideas. The experiments were mainly performed by H. L. All authors contributed to the ana-

lysis and interpretation of the results and the writing of the paper.

Data availability

Data for this article are available at Dryad at <https://datadryad.org/stash/share/MiFUYDWs51WXmrBpTqh1ck3HwdDJzM1hnkCEbgs2NWw>.

Conflicts of interest

There are no conflicts to declare.

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