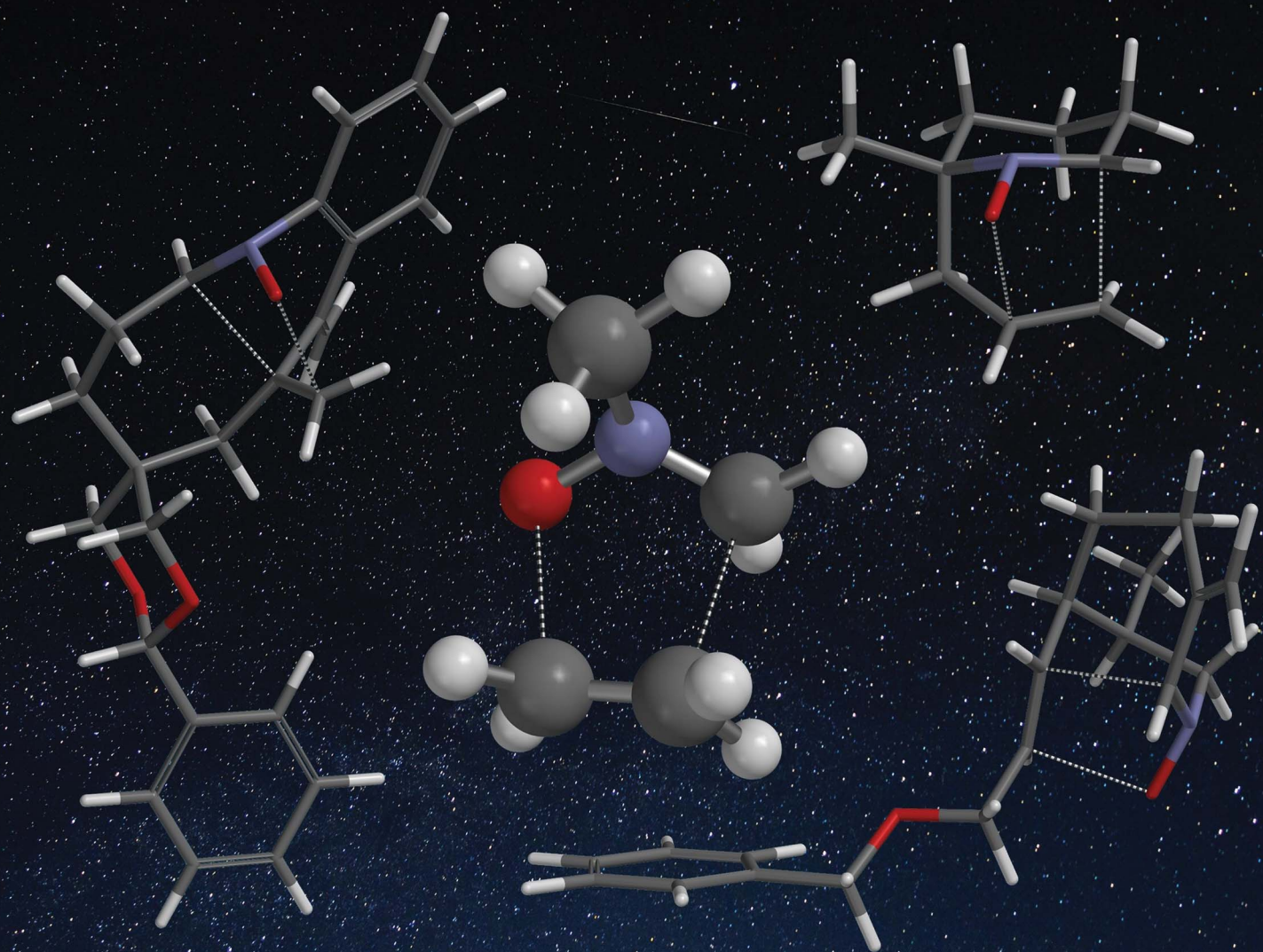


Natural Product Reports

rsc.li/npr



ISSN 0265-0568

REVIEW ARTICLE

Satoshi Yokoshima
Intramolecular cycloaddition of nitrones in total synthesis of
natural products

REVIEW

View Article Online
View Journal | View Issue



Cite this: *Nat. Prod. Rep.*, 2025, 42, 1071

Intramolecular cycloaddition of nitrones in total synthesis of natural products

Satoshi Yokoshima *

Covering 2015 to 2024

Cycloaddition of nitrones with alkenes forms isoxazolidines, which are five-membered heterocycles containing nitrogen and oxygen atoms. This transformation functionalizes alkenes by forming C–C and C–O bonds. The N–O bond in the resultant isoxazolidines is easily cleaved. Additionally, when the cycloaddition is carried out intramolecularly, the regioselectivity of the reaction is influenced by the tether connecting the nitron and alkene and can differ from the selectivity governed by frontier molecular orbital interaction. These features make the intramolecular cycloaddition of nitrones attractive in the synthesis of complex molecules. In this review, we discuss the intramolecular cycloaddition of nitrones used in the total synthesis of natural products.

Received 5th November 2024

DOI: 10.1039/d4np00062e

rsc.li/npr

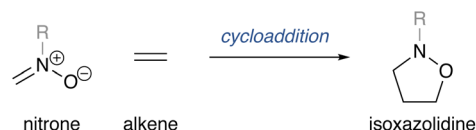
1. Introduction
2. Construction of isoxazolidine rings in natural products
 - 2.1. Synthesis of alsamaphorazine B
 - 2.2. Synthesis of virosaine A
 - 2.3. Synthesis of lycojaponicum A
3. Synthesis *via* intramolecular cycloaddition of acyclic nitrones
 - 3.1. Syntheses *via* cycloaddition of C-linked unsaturated nitrones
 - 3.1.1. Synthesis of himalensine A
 - 3.1.2. Synthesis of pleosporol A
 - 3.1.3. Synthesis of isolysergol
 - 3.1.4. Synthesis of thebainone A
 - 3.2. Syntheses *via* cycloaddition of N-linked unsaturated nitrones
 - 3.2.1. Synthesis of lasubine II
 - 3.2.2. Synthesis of palhinines A and D
 - 3.2.3. Synthesis of (19*Z*)-taberpsychine, isodihydrokoumine, and isodihydrokoumine *N*₄-oxide
4. Synthesis *via* intramolecular cycloaddition of cyclic nitrones
 - 4.1. Synthesis of 19-hydroxyibogamine
 - 4.2. Synthesis of *N*-methyl-euphococcine
 - 4.3. Synthesis of cylindricine C
 - 4.4. Synthesis of sarain A
 - 4.5. Synthesis of kopsone
 - 4.6. Synthesis of stephadiamine A
 - 4.7. Synthesis of putative melognine
5. Discussion on regioselectivity

- 5.1. Intramolecular cycloaddition of acyclic nitrones
- 5.2. Intramolecular cycloaddition of cyclic nitrones
6. Conclusions
7. Conflicts of interest
8. Acknowledgements
9. Notes and references

1. Introduction

A nitron is a 1,3-dipole consisting of carbon, nitrogen, and oxygen atoms arranged in that sequence.¹ It is usually depicted as a structure in which carbon and nitrogen atoms form a double bond, with the oxygen atom negatively charged and the nitrogen atom positively charged (Scheme 1). A nitron undergoes cycloaddition with an alkene to form an isoxazolidine.² In this transformation, the C–C double bond is functionalized to form C–O and C–C bonds. The N–O bond in the isoxazolidine is readily cleaved. These observations show that the cycloaddition of a nitron is a useful method to introduce oxygen and nitrogen functionalities with concomitant formation of a C–C bond.

In the cycloaddition step, there is an issue of regioselectivity, which can be understood by frontier molecular orbital (FMO) interaction.^{1a,h,3} Typical examples of intermolecular

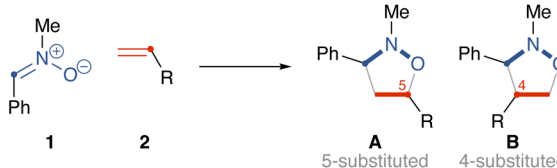


Scheme 1 Cycloaddition of nitron.

Graduate School of Pharmaceutical Sciences, Nagoya University, Nagoya 464-8601, Japan. E-mail: yokoshima.satoshi.v7@f.mail.nagoya-u.ac.jp



Table 1 Regioselectivity in cycloaddition of nitrones



Entry	R	Yield (%)	A : B	Cis : trans
1 ^a	<i>n</i> -C ₆ H ₁₃	77	100 : 0	1 : 1
2 ^b	Ph	79	100 : 0	3 : 1
3 ^c	OE _t	78	100 : 0	1 : 1
4 ^d	CO ₂ Et	97	82 : 18	1 : 7 ^f , 2 : 7 ^g
5 ^e	NO ₂	—	0 : 100	2 : 1

^a Toluene, reflux. ^b Toluene, 100 °C. ^c 80 °C. ^d Benzene, reflux. ^e 60 °C. ^f For regioisomer A. ^g For regioisomer B.

cycloaddition of a nitrone with various alkenes are summarized in Table 1. Reactions of nitrone **1** with electron rich alkenes produce 5-substituted isoxazolidines **A** (entries 1–3).⁴ These can be understood by the matching of the largest coefficients both in the LUMO_{nitrone}–HOMO_{alkene} and HOMO_{nitrone}–LUMO_{alkene} interactions (Fig. 1a). In the reaction of **1** with nitroethylene (entry 5),⁵ the HOMO_{nitrone}–LUMO_{alkene} interaction is dominant, selectively producing 4-substituted isoxazolidine **B** (Fig. 1b). In the case of ethyl acrylate (entry 4),⁶ which is a less electron-deficient alkene than nitroethylene, the less energetically preferred LUMO_{nitrone}–HOMO_{alkene} also contributes to the regioselectivity in addition to the HOMO_{nitrone}–LUMO_{alkene} interaction, leading to the formation of a mixture of 5- and 4-substituted isoxazolidines.

These regioselectivities are often changed in the case of intramolecular cycloaddition where the nitrone and alkene are connected by a tether. This tether regulates the approach between the nitrone and the alkene, determining the selectivity.⁷

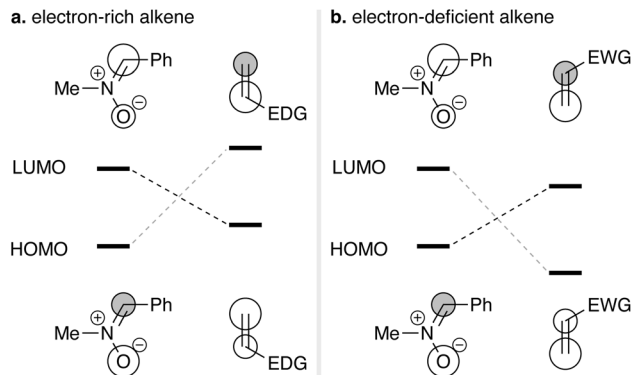


Fig. 1 HOMO–LUMO interaction of cycloaddition of nitrones. (a) Cycloaddition with an electron-rich alkene. (b) Cycloaddition with an electron-deficient alkene.

In the case of intramolecular cycloaddition, the stereo-selectivity, including facial and endo/exo-selectivities, tends to also be controlled by the tether. This feature makes the intramolecular cycloaddition of nitrones attractive for the synthesis of complex molecules, including natural products.

This review summarizes intramolecular cycloaddition reactions of nitrones in natural product syntheses reported between 2015 and 2024.^{8–11} The preparation of the nitrones and the transformations after the cycloadditions are also described.

2. Construction of isoxazolidine rings in natural products

Although some natural products contain isoxazolidine rings (Fig. 2), such natural products are rare.^{1e} The biosynthetic pathway might involve the generation of a nitrone and its cycloaddition.¹² In this section, three syntheses of natural products containing an isoxazolidine ring are demonstrated.

2.1. Synthesis of alsmaphorazine B

Alsmaphorazine B (**15**) is a hexacyclic indole alkaloid containing an isoxazolidine ring (Scheme 2). Although a biosynthetic



Satoshi Yokoshima

an associate professor and was promoted to professor in 2017. His current research interests focus on the synthesis of natural and unnatural molecules with polycyclic systems.

Satoshi Yokoshima received his B.S. in 1997, and his PhD in 2002 from The University of Tokyo under the guidance of Professor Tohru Fukuyama. After working for Mitsubishi Pharma Corporation as a medicinal chemist (2002–2004), he joined Professor Fukuyama's group at The University of Tokyo as an assistant professor. He was promoted to lecturer in 2008 and to associate professor in 2011. In 2012 he moved to Nagoya University as

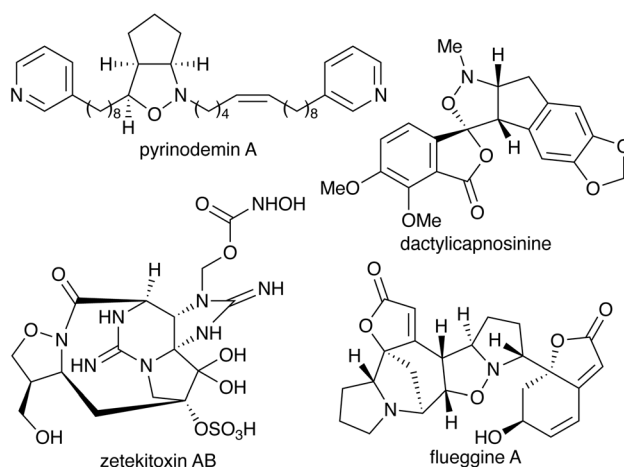
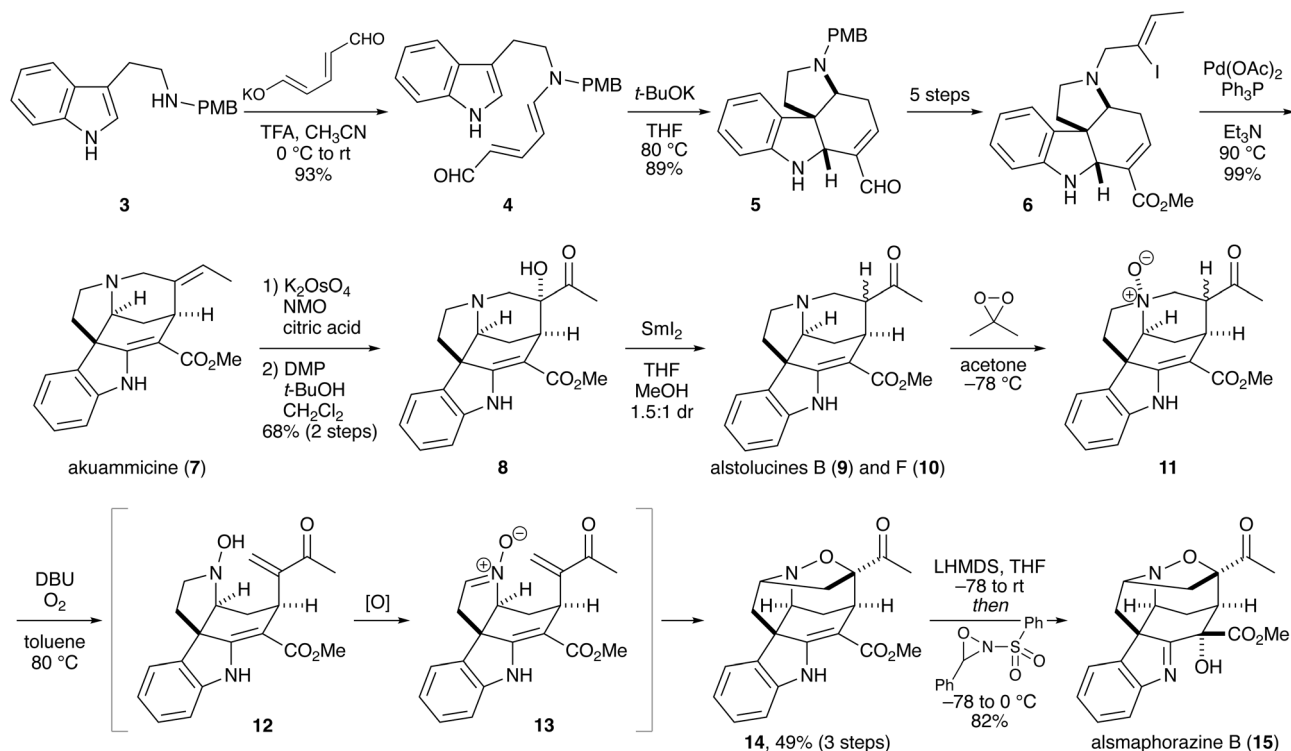


Fig. 2 Selected Natural products containing an isoxazolidine ring.





Scheme 2 Synthesis of alsmaphorazine B.

pathway of alsmaphorazine B from an akuammicine-type molecule *via* multiple stepwise oxidation and rearrangements had been proposed in a paper reporting isolation of alsmaphorazine B,¹³ Vanderwal and coworkers independently devised an alternative biosynthetic pathway, along which they achieved the synthesis of alsmaphorazine B (racemic, 15 steps).¹⁴

They first prepared akuammicine (7) from tryptamine in nine steps. One of the key reactions is base-mediated cycloaddition of Zincke aldehyde 4 to afford tetracyclic compound 5. After the conversion that included introducing an iodoalkene unit onto the nitrogen atom, the resultant unsaturated ester 6 was subjected to an intramolecular Heck reaction to give akuammicine (7).

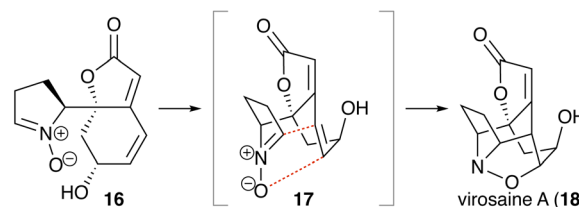
The next task was to convert the ethylidene into a ketone moiety. After extensive investigations, they established a three-step sequence involving osmium-mediated dihydroxylation, oxidation with Dess–Martin periodinane (DMP) in the presence of *tert*-butanol, and reduction of α -ketol 8 with samarium(II) iodide. The products—a 1.5 : 1 mixture of alstolucines B (9) and F (10)—were oxidized with dimethyldioxirane (DMDO) to afford *N*-oxide 11, which was heated with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) under an oxygen atmosphere. Under these conditions, base-mediated E1cb elimination or Cope elimination occurred to liberate an enone and a hydroxylamine.¹⁵ Oxidation of the hydroxylamine with gaseous oxygen furnished nitron 13, which underwent intramolecular cycloaddition with the enone moiety to give cycloadduct 14 in 49% yield. The structure of 14 was confirmed by X-ray crystallographic analysis. In this transformation, a nitron isomer was also isolated in 29% yield.

Further oxidation of the vinylogous carbamate moiety by treatment with lithium hexamethyldisilazide (LHMDS) and Davis oxaziridine afforded alsmaphorazine B (15).

2.2. Synthesis of virosaine A

Virosaine A (18) is a pentacyclic natural product containing an isoxazolidine ring (Scheme 3). In the proposed biosynthetic pathway, the isoxazolidine ring is constructed at the end of the biosynthesis *via* intramolecular cycloaddition of a five-membered cyclic nitron 16.¹⁶ Hughes and Gleason reported a synthesis of virosaine A (enantioselective, 15 steps),^{17,18} in which the formation of the isoxazolidine ring was conducted in an early stage of the synthesis *via* cycloaddition of a five-membered cyclic nitron.

An asymmetric Diels–Alder reaction of 2-bromoacrolein (19) with furan (20) catalyzed by oxazaborolidinone 21 yielded bicyclic aldehyde 22,¹⁹ which was reacted with organolithium reagent 23 to furnish a 2.7 : 1 diastereomeric mixture of alcohols

Scheme 3 A proposed biosynthetic pathway of virosaine A *via* intramolecular cycloaddition of a five-membered cyclic nitron.

24 (Scheme 4). The diastereomeric ratio and the enantiomeric excess could be improved by recrystallization. Acidic hydrolysis of the acetal moiety, followed by a reaction of the resultant cyclic acetal with *O*-TBS hydroxylamine (TBS = *tert*-butyldimethylsilyl), afforded *O*-silyl oxime **25**. Treatment of **25** with sodium hydride induced epoxide formation to give **26**. When **26** was heated in acetic acid at 120 °C under microwave irradiation, the nitrogen atom intramolecularly attacked the epoxide to form nitrone **27**, which underwent cycloaddition to give pentacyclic compound **28** in 92% yield.

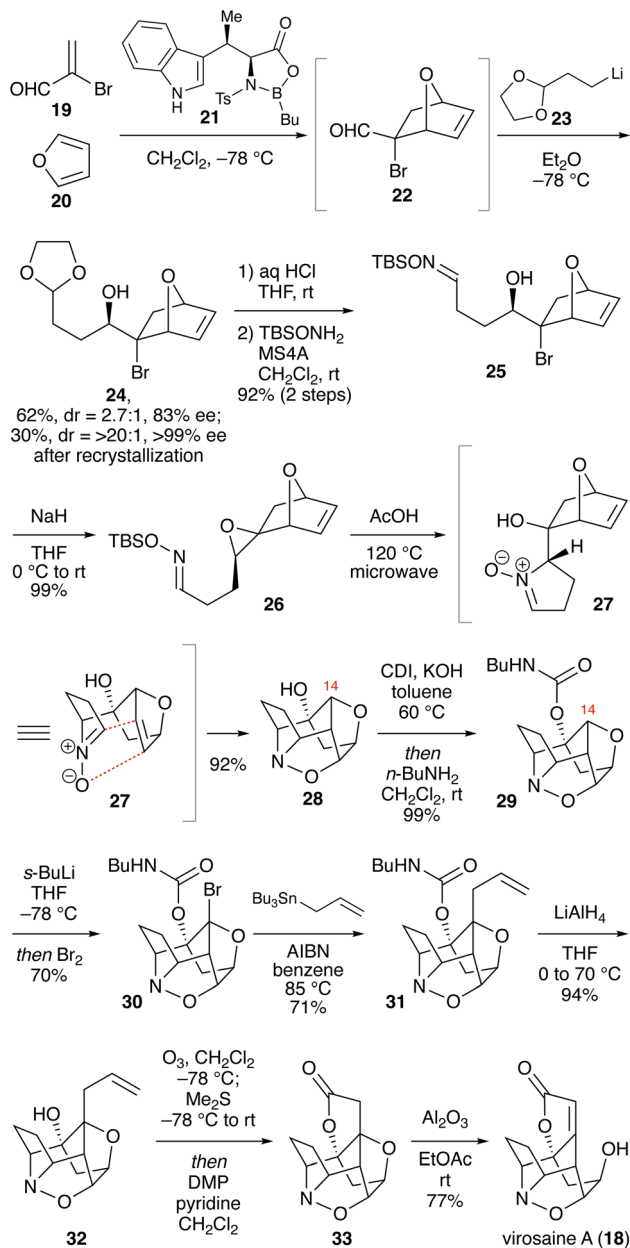
The remaining task for the synthesis of virosaine A was functionalization of C14 for the construction of the butenolide moiety. The first attempts involved the intramolecular C–H insertion of carbene or nitrene at C14, but the desired products

were not obtained. The insertion reactions of carbene or nitrene occurred at undesired positions. Finally, the authors found a site-directed deprotonation was effective for the transformation. Thus, reaction of the hydroxy group in **28** with 1,1'-carbonyldiimidazole (CDI) and then with *n*-butylamine afforded carbamate **29**. Selective lithiation at C14 by treatment with *sec*-butyllithium, followed by addition of bromine, yielded bromide **30**. Radical allylation under Keck's conditions introduced an allyl group at the bridgehead position.²⁰ Reductive cleavage of the carbamate moiety with lithium aluminum hydride liberated alcohol **32**. Ozonolysis of the terminal alkene followed by oxidation of the resultant hemiacetal with DMP furnished lactone **33**, which was exposed to activated neutral Al₂O₃ to afford virosaine A (**18**).

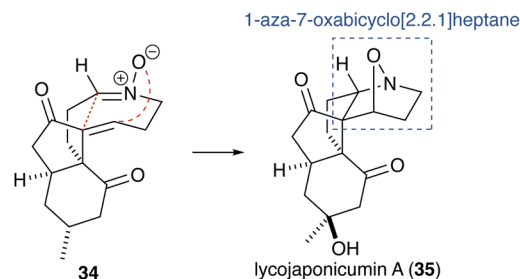
2.3. Synthesis of lycojaponicum A

Lycojaponicum A (**35**) is an alkaloid containing a bicyclo [2.2.1] system within which an isoxazolidine ring is embedded as a 1-aza-7-oxabicyclo[2.2.1]heptane (Scheme 5).²¹ A biosynthetic pathway of lycojaponicum A has been proposed, in which the bicyclo[2.2.1] system is formed *via* intramolecular cycloaddition of nine-membered cyclic nitrone **34**. Zhang, Tu and coworkers reported a synthesis of lycojaponicum A (racemic, 23 steps).²² They constructed the 1-aza-7-oxabicyclo [2.2.1] system *via* cycloaddition of an NH nitrone and subsequent intramolecular alkylation.

A reaction of amide **36** with triflic anhydride formed a kete-niminium salt, which underwent intramolecular [2 + 2] cycloaddition with the terminal alkene moiety to give cyclobutanone **37** (Scheme 6). Introduction of a dihydrofuran unit afforded tertiary alcohol **38**. Upon treatment of **38** with BF₃·OEt₂, a ring expansion reaction occurred *via* selective migration of the quaternary carbon to furnish cyclopentanone **39** with a good diastereoselectivity (11 : 1 dr at C4). After a five-step conversion, the tetrahydrofuran ring in **40** was reductively cleaved by treatment with samarium(II) iodide. The resulting enolate was transformed into silyl enol ether **41** *via* addition of TBS triflate and subsequent thermal isomerization. Phenylselenenylation of the silyl enol ether **41**, followed by one-pot oxidative elimination afforded *Z*-enone **42**, which was subjected to photoinduced isomerization with an LED lamp (395 nm), giving, after cleavage of the TBS group with TBAF, *E*-enone **43**. A three-step conversion afforded aldehyde **44**, which was reacted with hydroxylamine

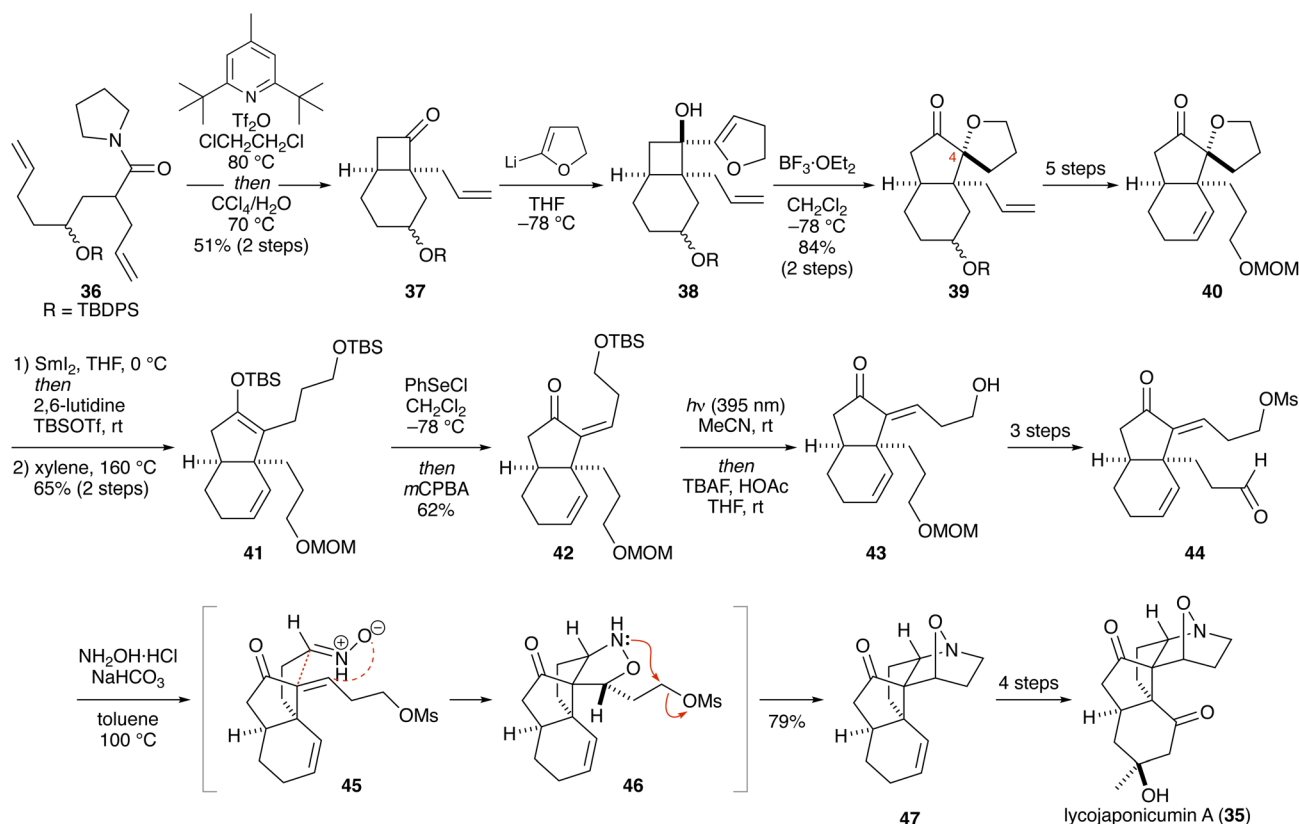


Scheme 4 Synthesis of virosaine A.



Scheme 5 A proposed biosynthetic pathway of lycojaponicum A *via* intramolecular cycloaddition of a nine-membered cyclic nitrone.



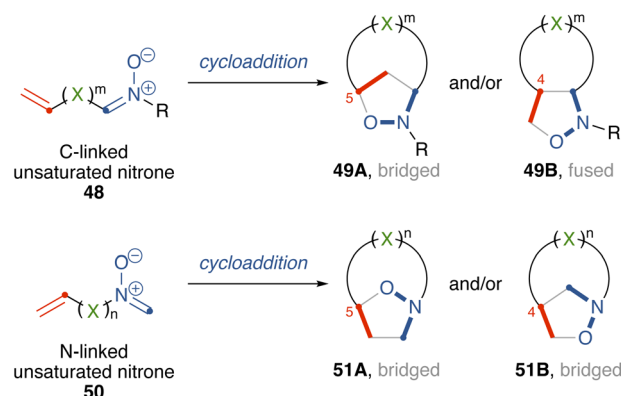


Scheme 6 Synthesis of lycojaponicum A.

hydrochloride and sodium bicarbonate in toluene at 100 °C. Under these conditions, NH nitron 45 was formed and underwent intramolecular cycloaddition with the enone moiety.²³ Subsequent intramolecular S_N2 reaction between the nitrogen atom in the resulting isoxazolidine ring and the mesylate moiety furnished pentacyclic compound 47. Sequential oxidation of the cyclohexene ring in four steps produced lycojaponicum A (35).

3. Synthesis *via* intramolecular cycloaddition of acyclic nitrones

Acyclic nitrones for the intramolecular cycloaddition can be classified into two groups depending on whether the unit containing the alkene moiety is attached to the carbon atom or the nitrogen atom of the nitron. The former (C-linked unsaturated nitron, 48) can be basically prepared from aldehydes bearing an alkene moiety by condensation with a hydroxylamine, and its cycloaddition produces bridged and/or fused isoxazolidines (49A and/or 49B, Scheme 7). The latter (N-linked unsaturated nitron, 50) can be derived from hydroxylamines bearing an alkene moiety, leading to bridged isoxazolidines (51A and/or 51B). Isoxazolidines 49A and 51A are 5-substituted isoxazolidines similar to isoxazolidine A described in Table 1, whereas isoxazolidines 49B and 51B are 4-substituted isoxazolidines resembling isoxazolidine B. This section presents syntheses *via* intramolecular cycloaddition of acyclic nitrones



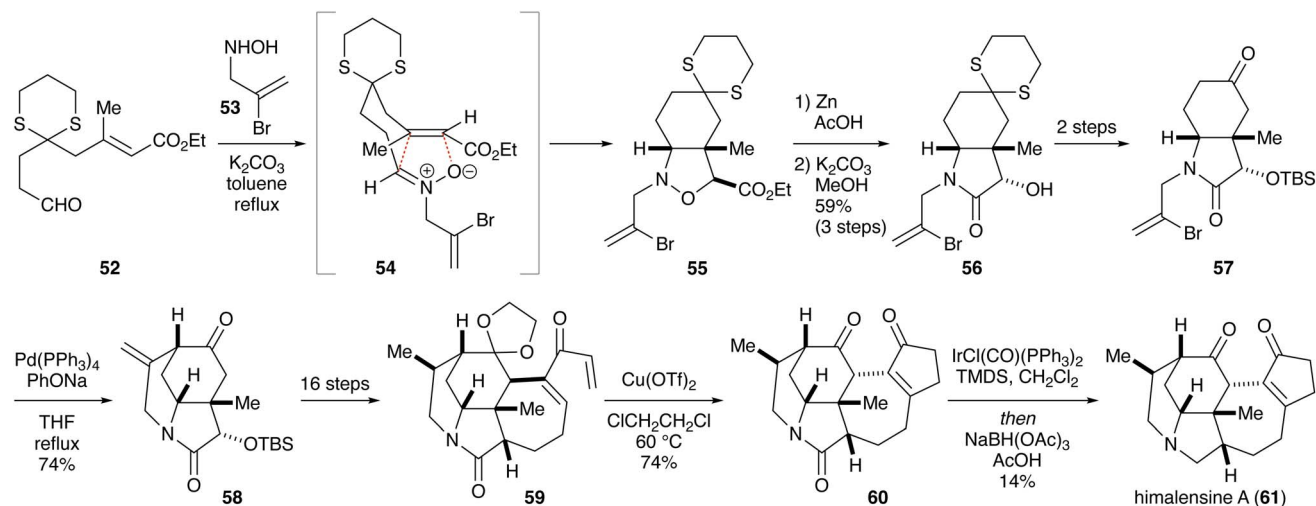
Scheme 7 C-linked and N-linked nitrones.

using C-linked unsaturated nitrones followed by N-linked unsaturated nitrones.

3.1. Syntheses *via* cycloaddition of C-linked unsaturated nitrones

3.1.1. Synthesis of himalensine A. Gao and coworkers reported a synthesis of himalensine A (racemic, 29 steps; Scheme 8).²⁴ Heating aldehyde 52 with hydroxylamine 53 yielded a nitron, which underwent intramolecular cycloaddition *via* a transition state 54 to give *cis*-fused isoxazolidine 55. Reductive cleavage of the N–O bond with zinc in acetic acid,





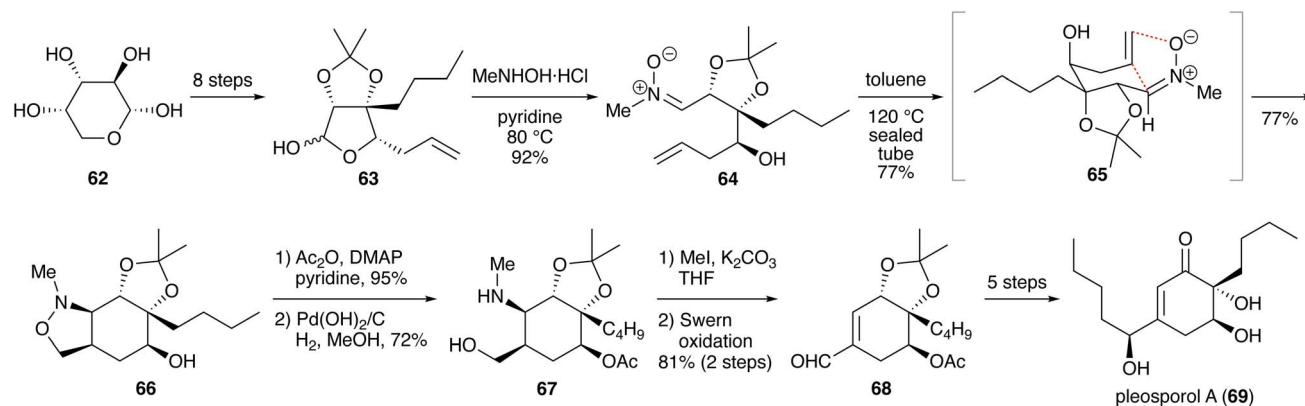
Scheme 8 Synthesis of himalensine A.

followed by basic treatment, produced lactam **56**. A two-step conversion of compound **56** including protection of the secondary alcohol with a TBS group and removal of the 1,3-dithiane group afforded ketone **57**, which underwent Pd-mediated intramolecular vinylation to produce tricyclic compound **58**. Ring-closing metathesis constructed a seven-membered ring, and Nazarov cyclization of **59** led to a cyclopentenone ring. The selective reduction of the lactam in **60** with Vaska's catalyst gave himalensine A (**61**).²⁵

3.1.2. Synthesis of pleosporol A. Xu and coworkers reported a synthesis of pleosporol A (19 steps; Scheme 9).²⁶ Starting from *L*-arabinose (**62**), compound **63** was prepared. Treatment with *N*-methylhydroxylamine hydrochloride in pyridine at 80 °C opened the lactol moiety to furnish nitron **64** in 92% yield. Cycloaddition of the nitron was induced by heating in toluene at 120 °C in a sealed tube to give *cis*-fused isoxazolidine **66** in 77% yield. The transition state **65** explains the stereoselectivity of the cycloaddition. After acetylation of the hydroxy group, the N–O bond was cleaved by hydrogenolysis in the presence of Pearlman's catalyst. Formation of the corresponding quaternary ammonium salt by treatment with iodomethane, followed by

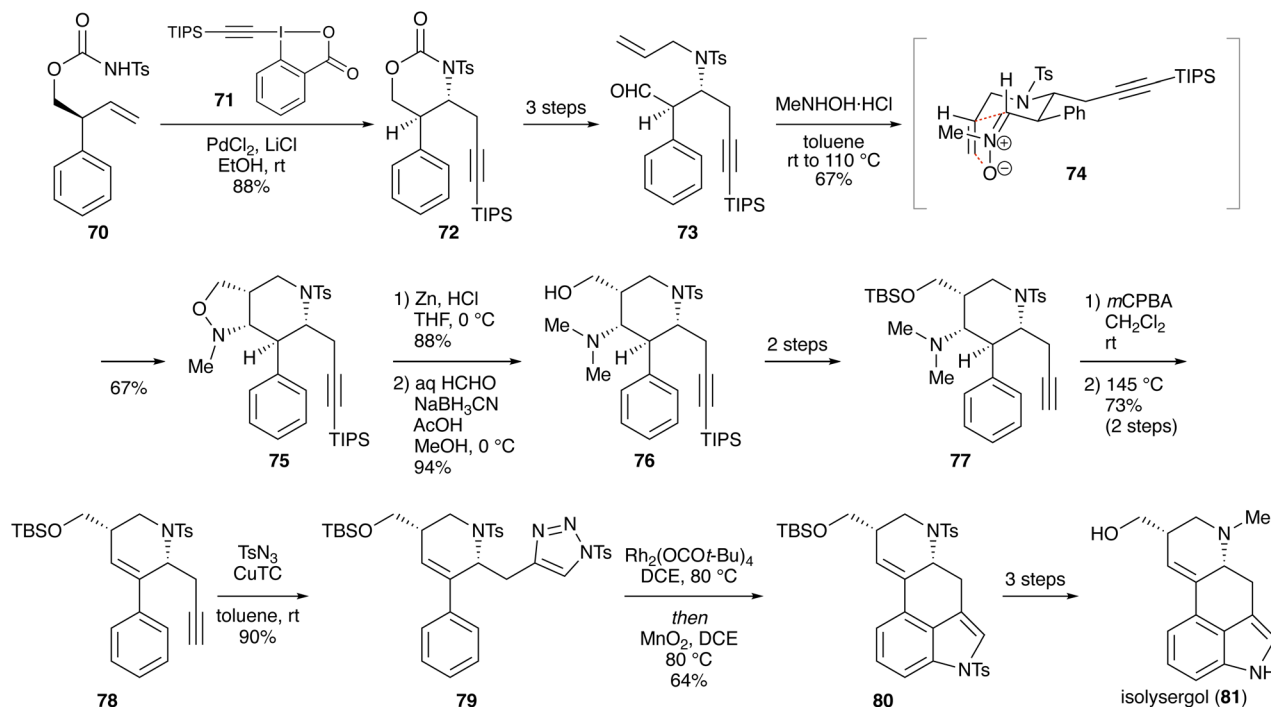
Swern oxidation, gave α,β -unsaturated aldehyde **68**, which was converted into pleosporol A (**69**) in five steps.

3.1.3. Synthesis of isolysergol. Wang and coworkers reported a synthesis of an ergot alkaloid, isolysergol (18 steps from (2*R*)-phenyloxirane; Scheme 10).²⁷ Palladium-catalyzed aminoalkynylation of carbamate **70** with 1-[(triisopropylsilyl) ethynyl]-1,2-benziodoxol-3(1*H*)-one (TIPS-EBX, **71**) proceeded stereoselectively to afford 4,5-*trans*-substituted product **72**,²⁸ which was converted into aldehyde **73** via three steps including *N*-allylation. Treatment of **73** with *N*-methylhydroxylamine led to nitron formation, and subsequent cycloaddition via transition state **74** gave *cis*-fused isoxazolidine **75** in 67% yield. Reductive cleavage of the N–O bond with zinc, followed by reductive amination with formalin, furnished compound **76** with a dimethylamino group. After a two-step conversion including removal of the TIPS group and protection of the primary alcohol with a TBS group, the dimethylamino group was oxidized with *m*-chloroperbenzoic acid (*m*CPBA) to produce the corresponding *N*-oxide, which was heated at 145 °C in neat conditions to induce *syn*-elimination (Cope elimination), giving **78** in 73% yield over two steps. Copper-catalyzed Huisgen



Scheme 9 Synthesis of pleosporol A.





Scheme 10 Synthesis of isolysergol.

cycloaddition with *p*-toluenesulfonyl azide (TsN_3) afforded *N*-tosyl-1,2,3-triazole **79**, which was treated with a rhodium catalyst in 1,2-dichloroethane at 80 °C to give, after oxidation with manganese(IV) oxide, 3,4-fused indole **80**.²⁹ Deprotection and *N*-methylation afforded isolysergol (**81**).

3.1.4. Synthesis of thebainone A. Metz and coworkers reported a synthesis of thebainone A (racemic, 22 steps; Scheme 11).³⁰ An intramolecular Heck reaction of **82** afforded spiro lactone **83**. After cleavage of the acetal moieties by treatment with a mild Lewis acid, $\text{Pd}(\text{MeCN})_2\text{Cl}_2$, the resultant keto-aldehyde **84** was treated with *N*-methylhydroxylamine without purification. The formation of a nitron and subsequent cycloaddition proceeded at low temperature with high diastereoselectivity *via* a transition state **85**. Reduction of the ketone and the lactone moieties in **86** with lithium aluminum hydride, followed by reductive cleavage of the N–O bond by treatment with RANEY® nickel under a hydrogen atmosphere, furnished aminotetraol **88**, which could be converted into compound **89** *via* an intramolecular Mitsunobu reaction between the amino group and the primary alcohol moiety. Elaboration of the oxygen functionalities on the cyclohexane ring in eight steps produced thebainone A (**90**).

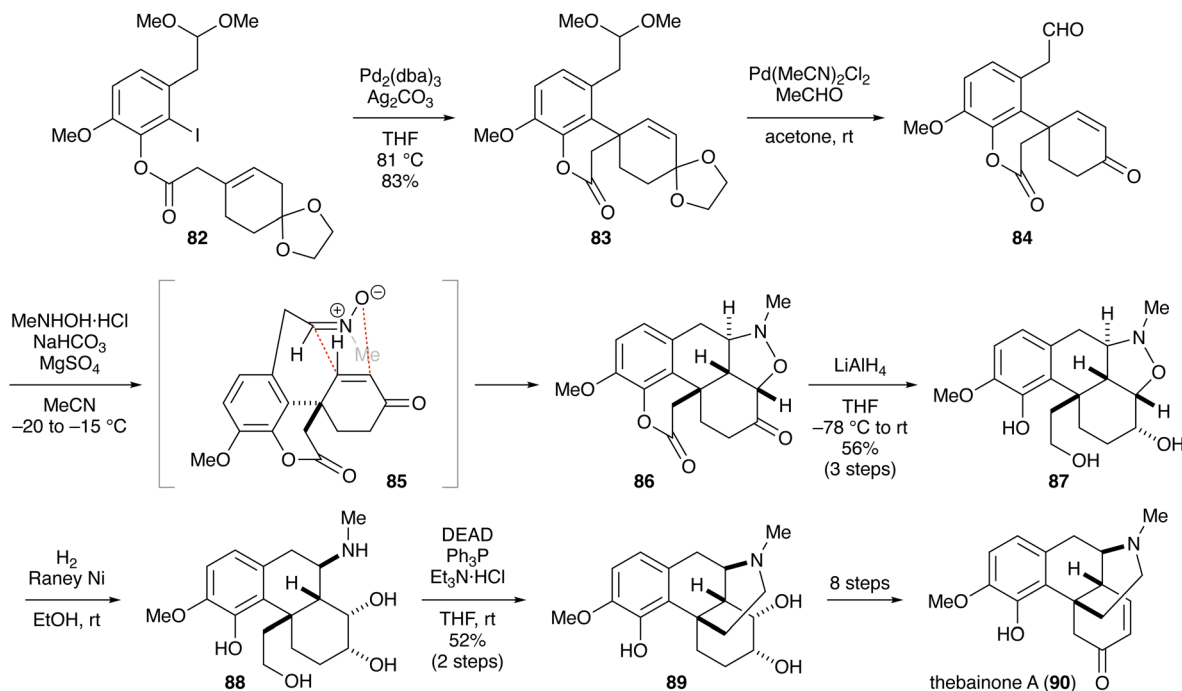
3.2. Syntheses *via* cycloaddition of N-linked unsaturated nitrones

3.2.1. Synthesis of lasubine II. Sato, Chida and coworkers reported a synthesis of lasubine II (racemic, 7 steps; Scheme 12).³¹ 3,4-Dimethoxybenzaldehyde (**91**) was converted into *N*-methoxylamine **92** *via* the formation of an oxime ether and its reduction. Three-component coupling of **92** with aldehyde **93** and allyltributylstannane in the presence of $\text{Gd}(\text{OTf})_3$

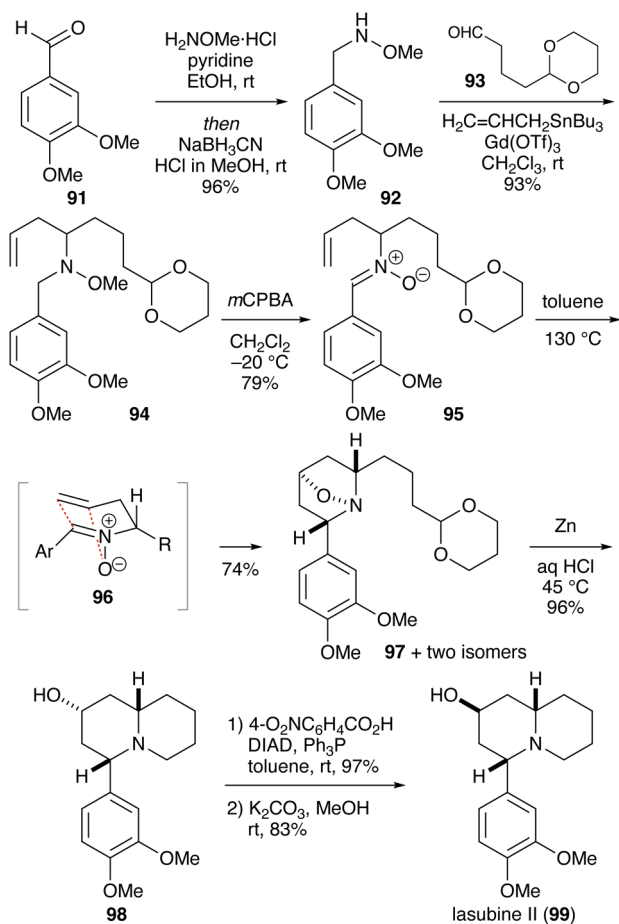
produced the desired product **94** in 93% yield. Oxidation with *m*CPBA selectively yielded nitron **95** without forming the other isomer. When nitron **95** in toluene was heated at 130 °C in a sealed tube, intramolecular cycloaddition occurred to form isoxazolidine **97** in 74% yield, along with two other diastereomers in 10% yield. Reduction with zinc in hydrochloric acid cleaved the N–O bond and the acetal, and cyclization *via* reductive amination also occurred to furnish quinolizidine **98**. Inversion of the secondary alcohol moiety *via* a Mitsunobu reaction, followed by methanolysis, produced lasubine II (**99**).

3.2.2. Synthesis of palhinines A and D. Palhinine A (**111**, Scheme 13) is an alkaloid characterized by a unique skeleton that includes a nitrogen-containing nine-membered ring. Fan and coworkers successfully constructed the nine-membered ring *via* nitron cycloaddition, thereby completing the synthesis of palhinine A (racemic, 23 steps).³² Starting from readily available enone **100**, compound **101** was prepared through 6 steps including Hosomi–Sakurai allylation and Nozaki–Hiyama–Kishi coupling.³³ Treatment of **101** with trimethylsilyl triflate and triethylamine produced siloxy-diene **102**, which was heated in *p*-xylene at 180 °C to induce an intramolecular Diels–Alder reaction that gave tricyclic compound **103** in 65% yield. After conversion in six steps including a Wittig olefination, the resultant aldehyde **104** was transformed into hydroxylamine **105** through the formation and reduction of an oxime. Condensation of the hydroxylamine with aqueous formaldehyde in a biphasic system afforded nitron **106**, which was subjected to the subsequent step without further purification because of its instability. Thus, nitron **106** was heated in *o*-dichlorobenzene at 150 °C under microwave irradiation. The cycloaddition of nitron occurred intramolecularly, and





Scheme 11 Synthesis of thebainone A.

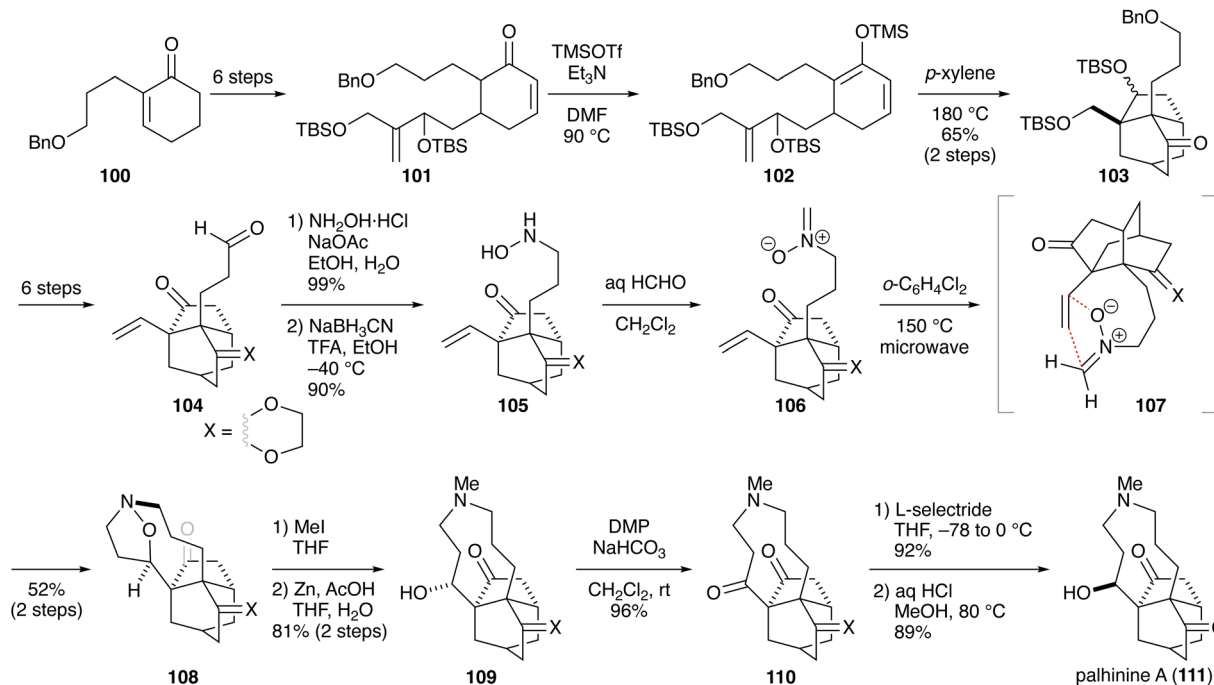


Scheme 12 Synthesis of lasubine II.

isoxazolidine **108** was obtained in 52% yield as a sole isomer. Fan and coworkers attributed the high regio- and stereo-selectivity to the minimal steric interaction and the favorable dipole-dipole attraction in the transition state (Fig. 3). *N*-Methylation, followed by reductive cleavage of the *N*-O bond with zinc, yielded aminoalcohol **109**. Inversion of the secondary alcohol in **109** was induced by an oxidation/reduction sequence. Acidic hydrolysis of the cyclic acetal afforded palhinine A (**111**). When *N*-allylation of **108** was carried out instead of *N*-methylation, palhinine D (**117**) was also synthesized according to the same sequence (Scheme 14). The allyl group was removed by treatment with ruthenium(III) chloride in a mixture of acetonitrile and water.

3.2.3. Synthesis of (19*Z*)-taberpsychine, isodihydrokoumine, and isodihydrokoumine *N*₄-oxide. Kerkovius and Kerr reported a synthesis of (19*Z*)-taberpsychine (racemic, 11 steps; Scheme 15).³⁴ Diol **118** was prepared through a sequence including hydrostannylation of an alkyne, Stille coupling, and copper-catalyzed conjugate addition. A Mitsunobu reaction of diol **118** with *N,O*-bis(Boc)-hydroxylamine (Boc = *tert*-butoxycarbonyl) occurred selectively at the allylic position to furnish protected hydroxylamine **119** in 83% yield.³⁵ Cleavage of the Boc groups by treatment with trifluoroacetic acid (TFA), followed by condensation with aldehyde **120** in the presence of triethylamine, afforded nitron **121**, which was heated in refluxing benzene to give isoxazolidine **123** (2:1 ratio about the 2-hydroxyethyl chain, 1:1 ratio about the indoline). To understand the reactivity of the hydroxylamine, the reader should note that when the hydroxylamine salt was neutralized with triethylamine in the absence of aldehyde **120**, *N*-hydroxypyrrolidine **129** was produced *via* a Cope-type hydroamination (Scheme 16).³⁶





Scheme 13 Synthesis of palhinine A.

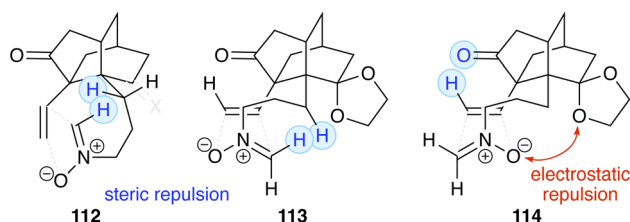
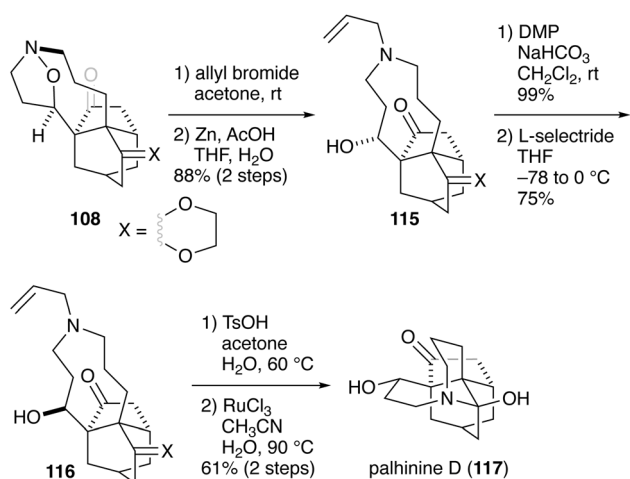


Fig. 3 Discussion on regio- and stereoselectivity.



Scheme 14 Synthesis of palhinine D.

After cleavage of the Ts group, the resultant compound was subjected to Swern oxidation. Under these conditions, simultaneous oxidation of the primary alcohol and the indoline

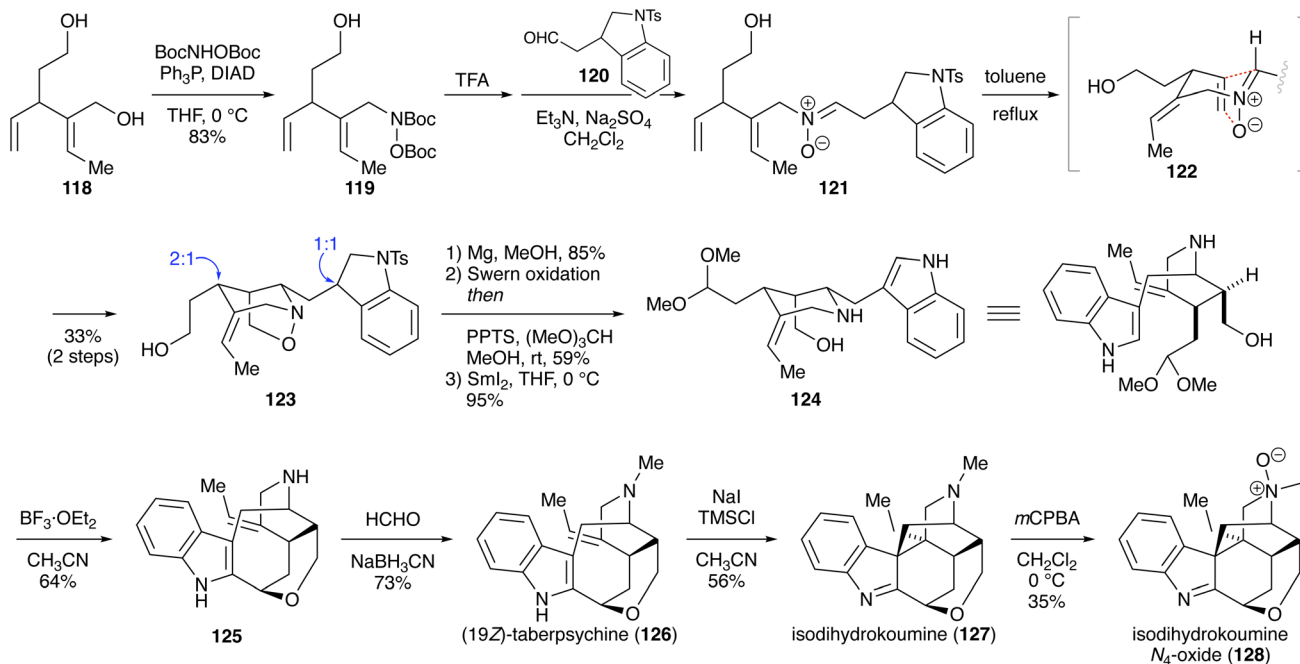
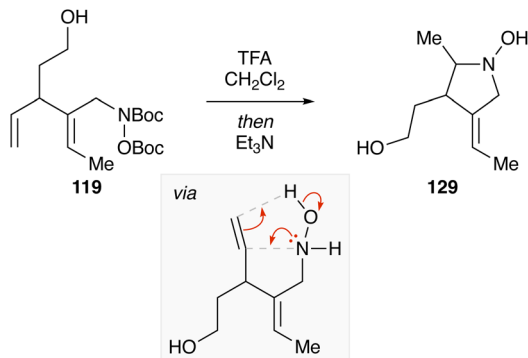
occurred; the aldehyde was then protected as its dimethyl acetal. At this stage, the diastereomers could be separated. The N–O bond was cleaved by a reaction with samarium(II) iodide. Upon treatment with $\text{BF}_3 \cdot \text{OEt}_2$ in acetonitrile, the dimethyl acetal moiety in **124** was activated and then intramolecularly attacked by the primary alcohol. Subsequent Friedel–Crafts reaction with the indole nucleus gave **125**. Reductive methylation of the secondary amine afforded (19*Z*)-taberpsychine (**126**). Treatment of (19*Z*)-taberpsychine (**126**) with trimethylsilyl iodide led to a Conia-ene-type reaction between the indole and the ethylidene moiety to furnish isodihydrokoumine (**127**). Oxidation with *m*CPBA produced isodihydrokoumine *N*₄-oxide (**128**).

4. Synthesis *via* intramolecular cycloaddition of cyclic nitrones

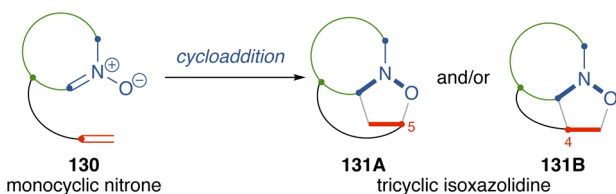
Intramolecular cycloaddition of mono-cyclic nitrones produces tricyclic isoxazolidines, beneficial for the synthesis of complex molecules (Scheme 17). Although preparation of cyclic nitrones has some difficulties, such as introducing multiple functional groups in a molecule, various methods to generate cyclic nitrones have been explored. This section presents syntheses *via* intramolecular cycloaddition of cyclic nitrones.

4.1. Synthesis of 19-hydroxyibogamine

Alkayar and Coldham reported a formal synthesis of 19-hydroxyibogamine (racemic, 16 steps; Scheme 18).³⁷ Aldehyde **132** was prepared *via* malonic ester synthesis and alkylation of a 1,3-dithiane. Oxime formation, followed by intramolecular alkylation, formed six-membered cyclic nitrone **134**, which

Scheme 15 Synthesis of (19Z)-taberpsychine, isodihydrokoumine, and isodihydrokoumine N_4 -oxide.

Scheme 16 Cope-type hydroamination of hydroxylamine.



Scheme 17 Intramolecular cycloaddition of cyclic nitrones.

underwent intramolecular cycloaddition to give isoxazolidine **135**. Hydrolysis of the 1,3-dithiane moiety under oxidative conditions afforded ketone **136**. Reductive cleavage of the N–O bond with zinc and acetic acid afforded aminoalcohol **137**, which was condensed with indoleacetic acid to give **138**. Conversion of the ketone moiety into an acetal, followed by acetylation of the hydroxy group, gave the reported intermediate

139, which could be converted into 19-hydroxyibogamine (**140**) in three steps.

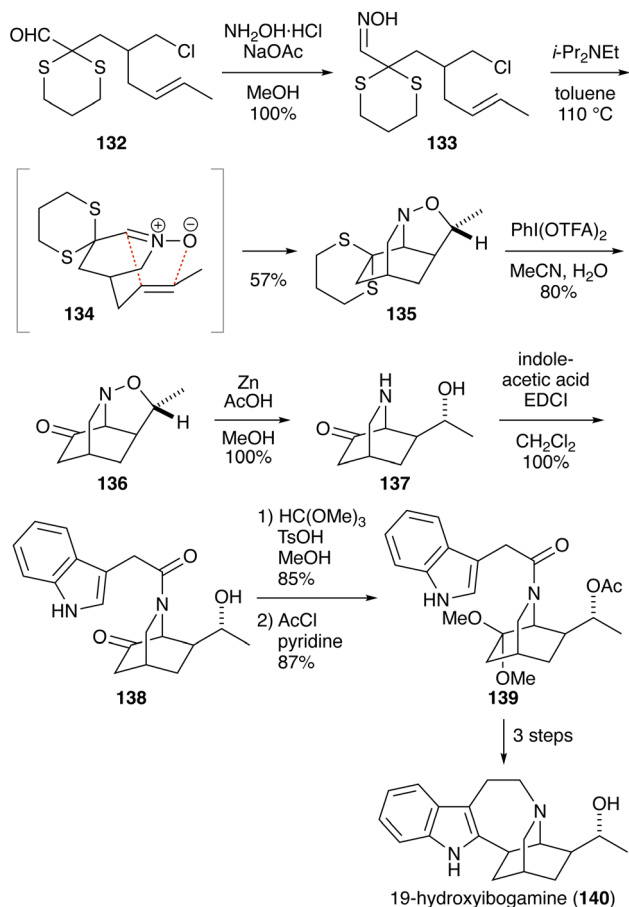
4.2. Synthesis of *N*-methyl-euphococcine

Kürti and coworkers developed a method for *C*-allylation of unprotected oximes and applied the method to the synthesis of *N*-methyl-euphococcine (racemic, 6 steps; Scheme 19).³⁸ Oxime **141** was treated with diisopropyl allylboronate in the presence of 3,5-difluorophenylboronic acid as a catalyst to give hydroxylamine **142**. After acidic hydrolysis of the acetal moiety and subsequent basic workup, the resultant six-membered cyclic nitron was heated in toluene at 110 °C to produce isoxazolidine **144** in 57% yield in two steps. *N*-methylation, followed by reduction with zinc, cleaved the N–O bond, and oxidation of the resultant secondary alcohol **146** with DMP afforded *N*-methyl-euphococcine (**147**).

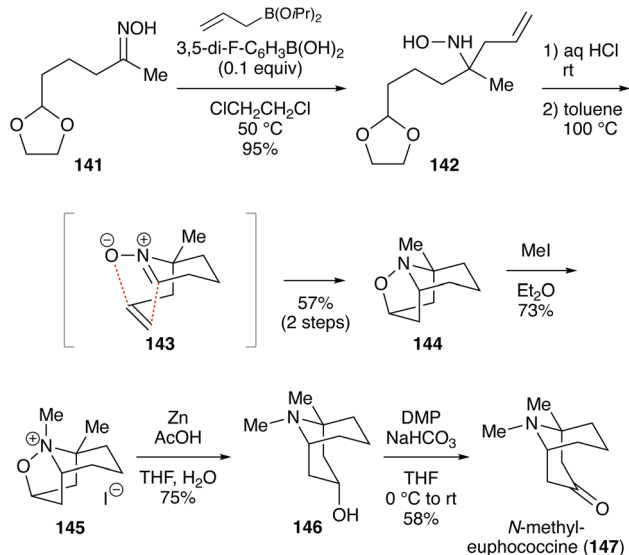
4.3. Synthesis of cylindricine C

Sato, Chida and coworkers developed a method to synthesize cyclic nitrones from *N*-hydroxylactams, and applied it to the synthesis of cylindricine C (asymmetric, 12 steps; Scheme 20).³⁹ *O*-Methylation of known lactam **148** with Meerwein's reagent yielded imidate **149**, which was converted into *N*-hydroxylactam **150** via oxidation with *m*CPBA, acidic hydrolysis of the resultant oxaziridine, and lactam formation under basic conditions. After the hydroxy group was protected with a 2-(trimethylsilyl)ethoxymethyl (SEM) group, the addition of hexenyllithium and subsequent acidic treatment afforded five-membered cyclic nitrone **152**. When heated in *tert*-butylbenzene, **152** underwent intramolecular cycloaddition to give a 1.3 : 1 mixture of regioisomers **155** and **156**. After separation

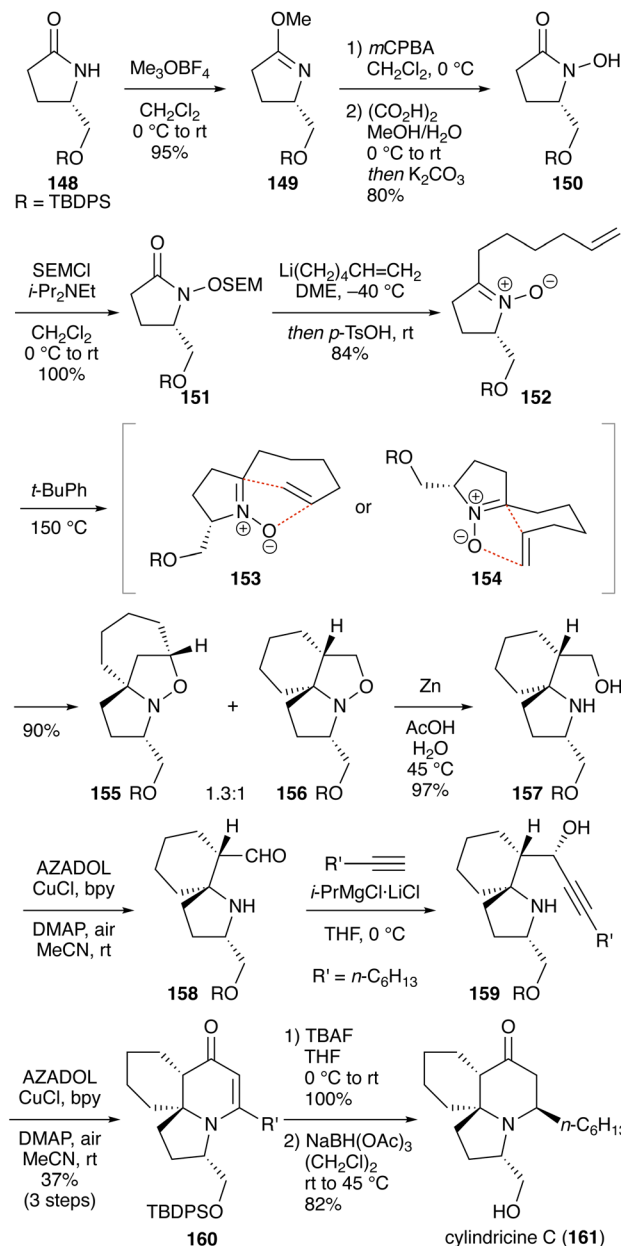




Scheme 18 Synthesis of 19-hydroxybogamine.

Scheme 19 Synthesis of *N*-methyl-euphococcine.

of the isomers, the N–O bond in **156** was reductively cleaved with zinc in aqueous acetic acid. The resultant alcohol was converted into aldehyde **158** in the presence of the unprotected secondary amine under the conditions developed by Iwabuchi



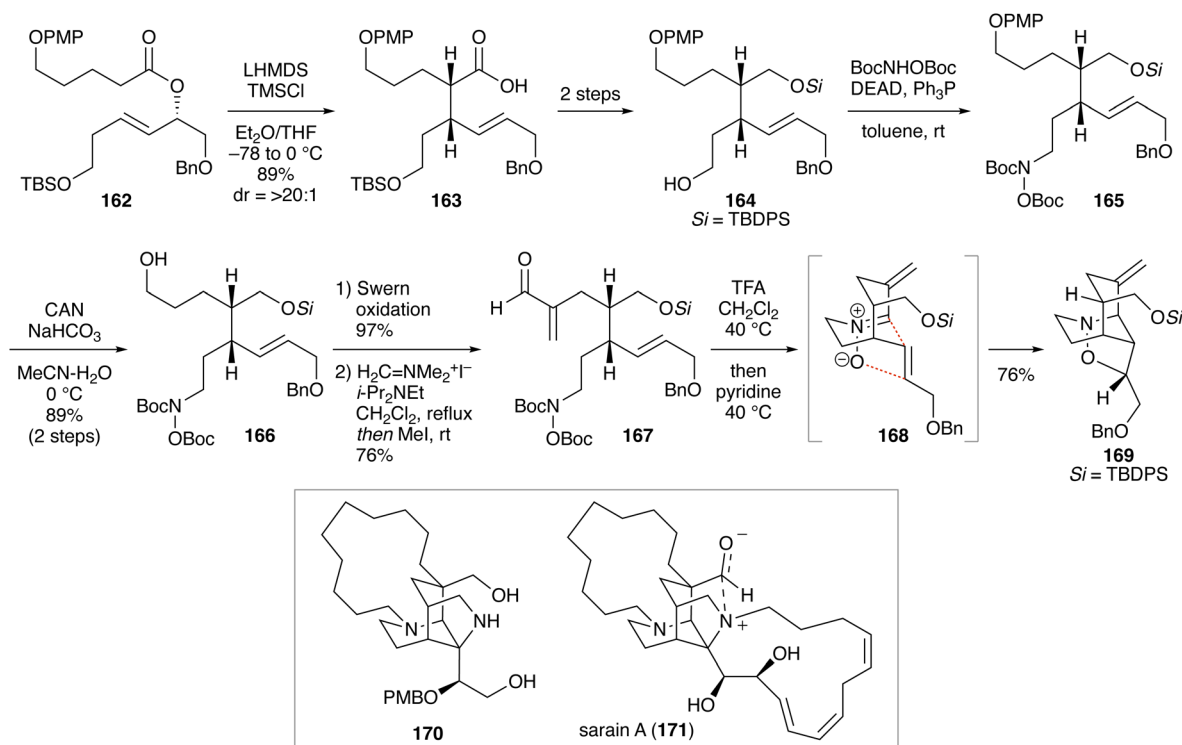
Scheme 20 Synthesis of cylindricine C.

and coworkers.⁴⁰ Addition of an acetylide to aldehyde **158** produced **159**, which was oxidized again under Iwabuchi's conditions, where intramolecular Michael addition also proceeded to furnish dihydro-4-pyridone **160**. Removal of the *tert*-butyldiphenylsilyl (TBDPS) group, followed by stereoselective reduction of the enamine moiety directed by the hydroxy group, afforded cylindricine C (**161**).

4.4. Synthesis of sarain A

Yokoshima, Fukuyama and coworkers reported a formal synthesis of sarain A (enantioselective; Scheme 21).⁴¹ Claisen–Ireland rearrangement of **162** selectively constructed the contiguous stereogenic centers to produce **163**. After conversion of **163** into alcohol **164**, a hydroxylamine unit was introduced





Scheme 21 Synthesis of sarain A.

via a Mitsunobu reaction. The *p*-methoxyphenyl (PMP)-protected alcohol was transformed into α,β -unsaturated aldehyde **167** via steps including oxidative cleavage of the PMP group, Swern oxidation, a Mannich reaction, and a methylation-E1cb elimination process. Treatment of **167** with TFA cleaved the two Boc groups to liberate the hydroxylamine moiety. Upon neutralization with pyridine, condensation of the hydroxylamine with the aldehyde moiety proceeded intramolecularly to form 8-membered cyclic nitron **168**, which underwent intramolecular cycloaddition with the alkene moiety, giving tricyclic compound **169**. This intermediate could be converted into an advanced intermediate (compound **170**) in Overman's synthesis of sarain A.⁴²

4.5. Synthesis of kopsone

Yokoshima and coworkers reported a synthesis of kopsone (enantioselective, 19 steps; Scheme 22).⁴³ Claisen–Ireland rearrangement and subsequent methylation produced compound **173**. Reduction of the ester moiety in **173**, followed by a Mitsunobu reaction with *N,O*-bis(Boc)-hydroxylamine, furnished **174**. The TBS-protected alcohol was converted into α,β -unsaturated aldehyde **176** in three steps. Upon treatment of **176** with TFA, the two Boc groups in **176** were cleaved. Neutralization with triethylamine induced the formation of eight-membered cyclic nitron **177**, which underwent intramolecular cycloaddition to give tricyclic isoxazolidine **178** in 76% yield. The diastereomeric compound **176'**, which was prepared from *Z*-isomer of **172** by an analogous sequence, could also be converted into tricyclic isoxazolidine **178'** in 57% yield. After reductive cleavage of the N–O bond in **178** with zinc and acetic acid, the resultant secondary

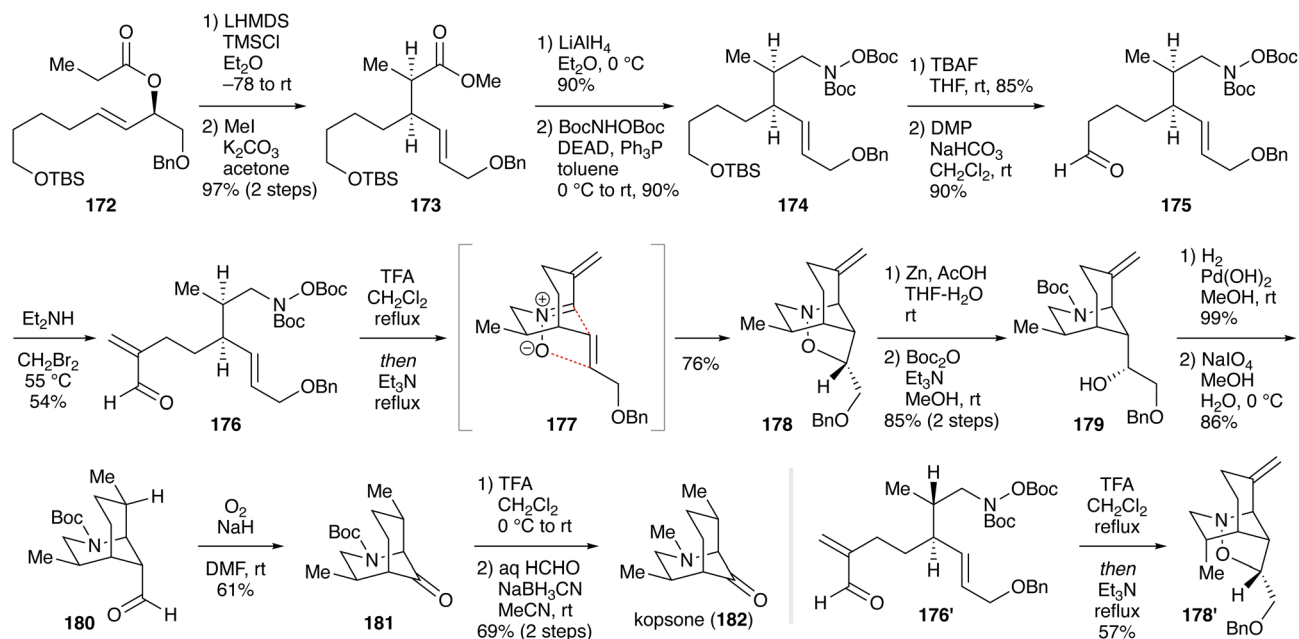
amine was protected with a Boc group. Hydrogenation of the C–C double bond and hydrogenolysis of the benzyl ether were carried out under a hydrogen atmosphere in the presence of palladium(II) hydroxide. The resultant 1,2-diol was oxidatively cleaved to yield aldehyde **180**, which was converted into ketone **181** by treatment with sodium hydride and oxygen.⁴⁴ Cleavage of the Boc group and reductive methylation afforded kopsone (**182**).

4.6. Synthesis of stephadiamine A

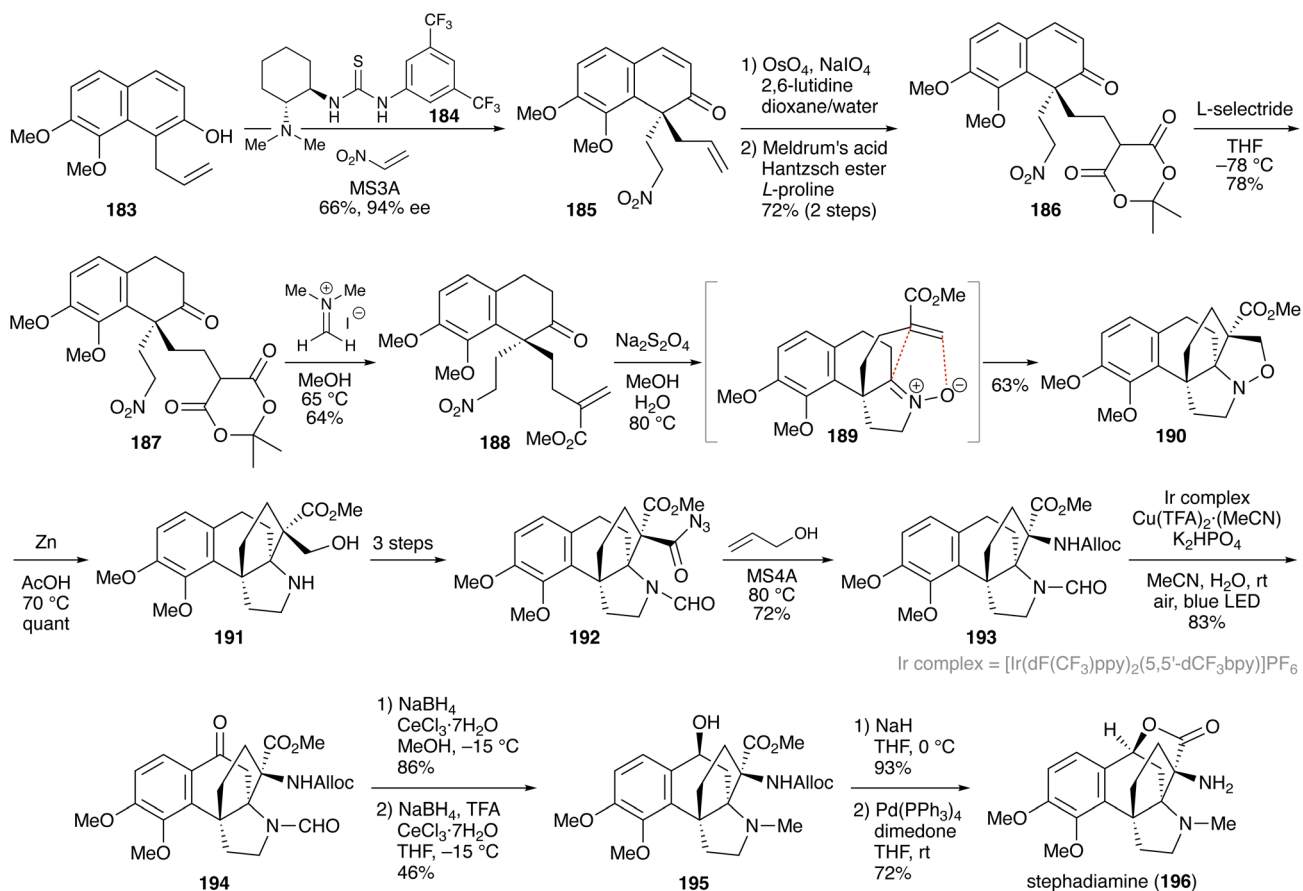
Stephadiamine (**196**) is a densely-functionalized pentacyclic natural product (Scheme 23). Zhu and coworkers used nitron cycloaddition to efficiently construct the skeleton and introduce functional groups.⁴⁵ Asymmetric Michael addition of phenol **183** to nitroethylene mediated by Takemoto catalyst **184** produced ketone **185** with a quaternary stereocenter.⁴⁶ Oxidative cleavage of the terminal alkene formed an aldehyde, which was reductively condensed with Meldrum's acid in the presence of Hantzsch ester and *L*-proline. After 1,4-reduction of the enone moiety in **186** with *L*-selectride, a reaction with Eschenmoser's salt in methanol afforded α,β -unsaturated ester **188**. Upon treatment of **188** with an excess amount of sodium dithionite in a mixed solvent system of methanol and water at 80 °C, a domino sequence involving reduction of the nitro group into a hydroxylamine, nitron formation, and intramolecular cycloaddition with the unsaturated ester moiety gave isoxazolidine **190** in 63% yield.

Cleavage of the N–O bond with zinc in acetic acid liberated aminoalcohol **191**, which was converted into acyl azide in three steps. When heated in allyl alcohol, **192** underwent Curtius rearrangement to produce compound **193**. Benzylic oxidation was carried out using a photocatalytic method reported by Yoon





Scheme 22 Synthesis of kopsone.



Scheme 23 Synthesis of stephadiamine.



and coworkers,⁴⁷ giving ketone **194**, which was stereoselectively reduced under Luche's conditions to afford an alcohol. The formyl group was also reduced with sodium borohydride in THF in the presence of cerium(III) chloride and TFA to yield **195**. Lactone formation under basic conditions, followed by palladium-catalyzed cleavage of the allyloxycarbonyl (Alloc) group, produced stephadiamine (**196**). The synthesis was completed in 18 steps.

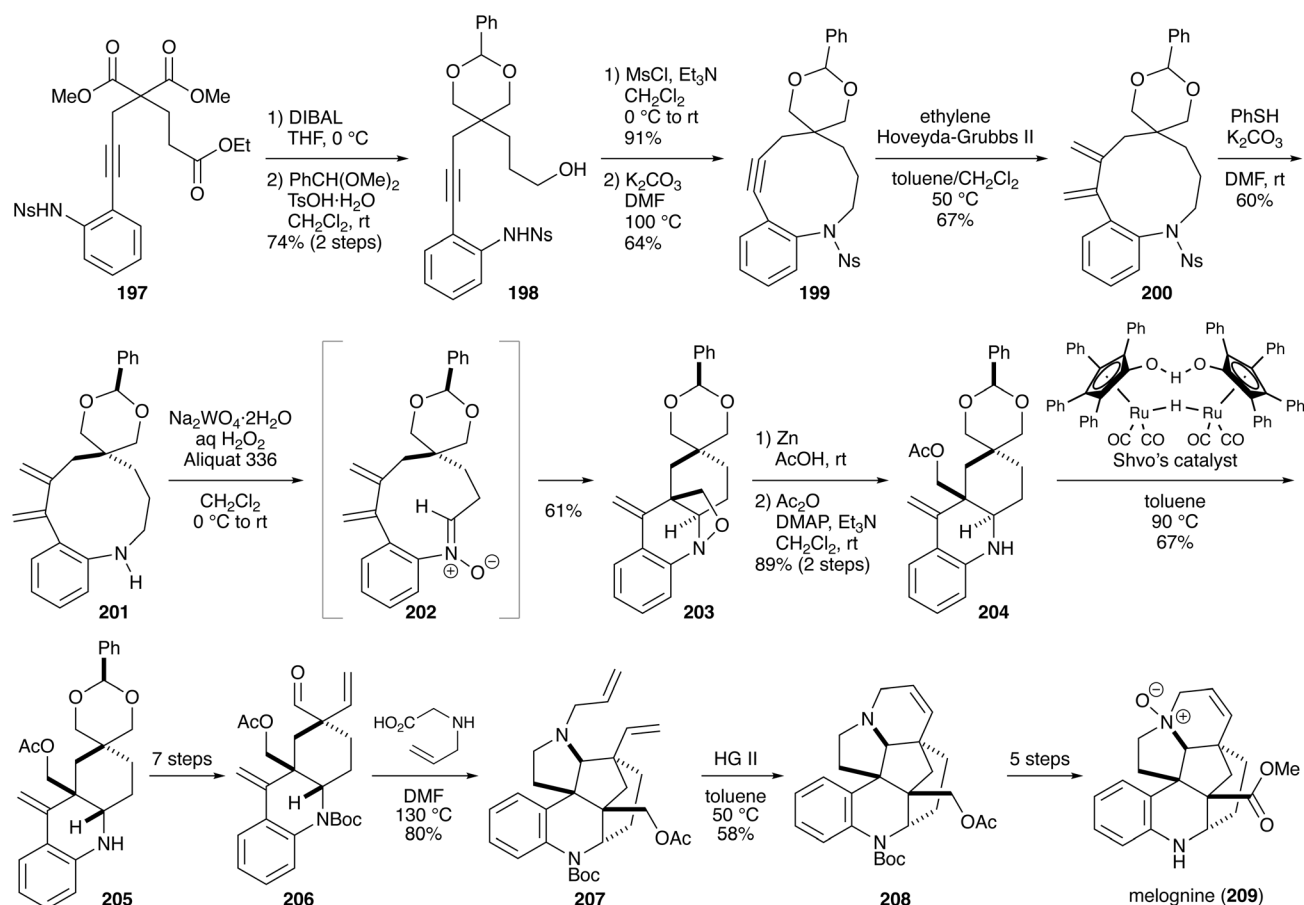
4.7. Synthesis of putative melognine

Irie and Yokoshima reported a synthesis of putative melognine (racemic, 29 steps; Scheme 24).⁴⁸ Compound **197** was prepared using malonic ester synthesis and Sonogashira coupling. Reduction of the ester moieties with diisobutylaluminum hydride (DIBAL), followed by selective protection of the 1,3-diol as its benzylidene acetal, furnished **198**. After mesylation of the primary alcohol moiety in **198**, the resultant mesylate was heated with potassium carbonate in *N,N*-dimethylformamide (DMF) at 100 °C to induce an intramolecular S_N2 reaction, giving ten-membered cyclic alkyne **199**. Enyne metathesis with ethylene in the presence of Hoveyda–Grubbs second-generation catalyst (HG II) yielded 1,3-diene **200**.⁴⁹ Cleavage of the Ns group liberated the aniline. In this step, the diastereomers derived from the benzylidene acetal could be separated. Treatment of

201 with hydrogen peroxide in the presence of sodium tungstate formed ten-membered cyclic nitrone **202**,⁵⁰ which underwent intramolecular cycloaddition to give isoxazolidine **203** as a sole isomer. Considering the stereochemistry of the product (compound **203**), the nitrone formed *in situ* must be the *Z*-isomer (Scheme 25). The diastereomeric product (compound **210**), which is derived from the *E*-isomer, was not detected. The regioisomeric product (compound **211**) was also not obtained. Interestingly, in the absence of the constraint imposed by the ten-membered ring, the opposite regioselectivity has been observed.⁵¹

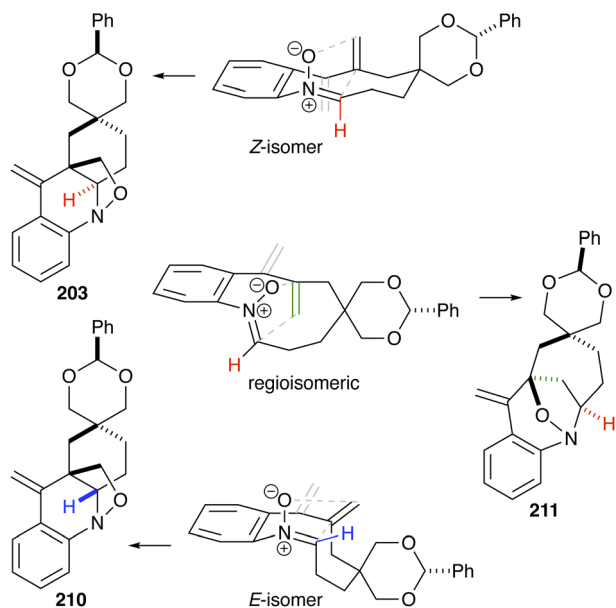
When the oxidation of **201** was carried out in methanol,⁵² isoxazolidine **203** and *N*-oxide **213** were obtained in 23% and 39% yields, respectively (Scheme 26). The latter is speculated to have formed *via* a Cope-type hydroamination of hydroxylamine **212**.⁵⁶

Cleavage of the N–O bond in **203** and protection of the resultant hydroxy group with an acetyl group furnished compound **204**, which was heated in the presence of Shvo's catalyst to induce isomerization at the α -position of the nitrogen atom.⁵³ A seven-step transformation of compound **205** gave aldehyde **206**, which was reacted with *N*-allylglycine in DMF at 130 °C. Under these conditions, the formation of an azomethine ylide and its cycloaddition proceeded to yield pentacyclic compound **207**. Ring-closing metathesis with HG II

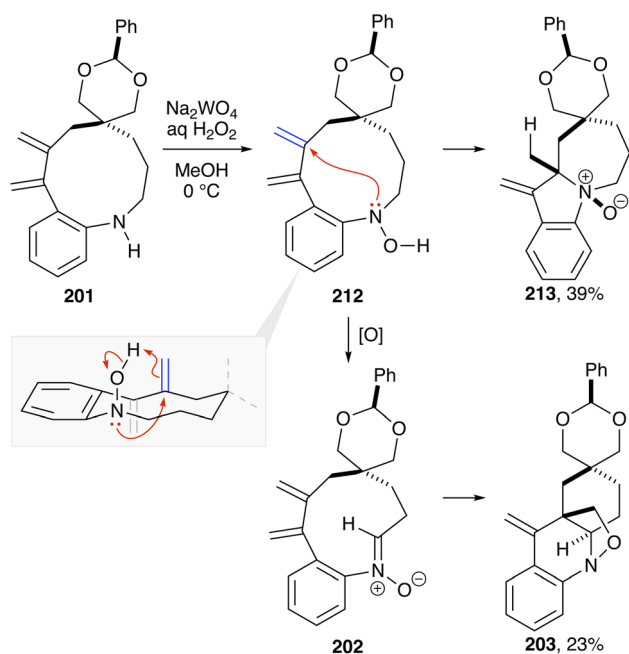


Scheme 24 Synthesis of putative melognine.





Scheme 25 Stereo- and regioselectivity in the cycloaddition.



Scheme 26 Cope-type hydroamination of hydroxylamine.

constructed a piperidine ring, and further transformation yielded putative melongine (209).

5. Discussion on regioselectivity

5.1. Intramolecular cycloaddition of acyclic nitrones

In this review, acyclic nitrones for intramolecular cycloaddition are classified into two groups: C-linked unsaturated nitrones and N-linked unsaturated nitrones. The C-linked unsaturated nitrones discussed in Section 3 (nitrones 54, 65, 74, and 85) are

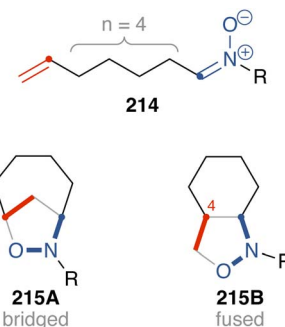


Fig. 4 C-linked unsaturated nitronone.

classified as having a four-carbon tether, with a structure like 214 (Fig. 4), and their cycloaddition produced fused isoxazolidines, represented by 215B (4-substituted isoxazolidine). There are many examples of producing fused isoxazolidines as shown in this review. However, formation of bridged isoxazolidine 215A (5-substituted isoxazolidine) has also been reported.⁵⁴ The regioselectivity is not primarily determined by the tether length. Instead, it depends on the substituents on the nitronone, alkene, and tether.

Cycloaddition of N-linked unsaturated nitrones can produce two types of regio isomers (51A vs. 51B, Scheme 7). For a two-carbon tether (216, $n = 2$; Fig. 5), 217A (5-substituted isoxazolidine) is typically the major or sole product,⁵⁵ consistent with the results observed in the synthesis of lasubine II (Scheme 12). For a three-carbon tether (218, $n = 3$), the regioselectivity varies depending on the substituents.^{51,55a,56} In the synthesis of isodihydrokoumine (Scheme 15), a 4-substituted isoxazolidine, corresponding to 219B (4-substituted isoxazolidine) was obtained. The nitronone used in the synthesis of palhinines (Scheme 13) is formally designated as 220 ($n = 5$), and the product of the cycloaddition was a 5-substituted isoxazolidine, corresponding

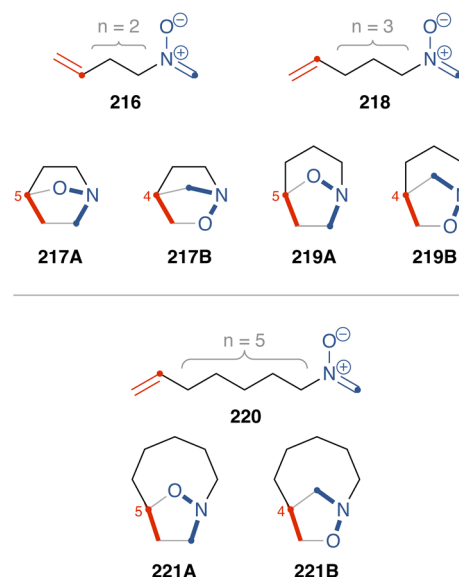


Fig. 5 N-linked unsaturated nitronone.

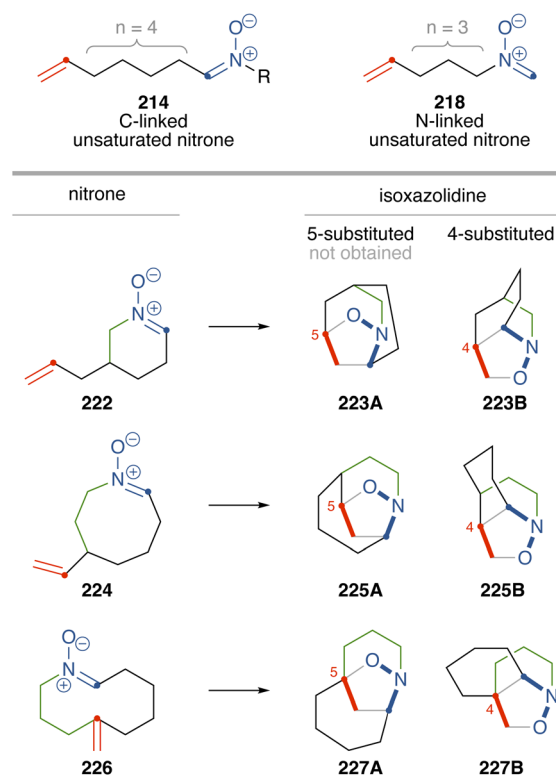
to **221A**. Fan and coworkers explain the regioselectivity using steric and electrostatic repulsion (Fig. 3).

5.2. Intramolecular cycloaddition of cyclic nitrones

In contrast to acyclic nitrones, cyclic nitrones have an additional linkage that imposes constraints on the interaction between the nitron and the alkene during intramolecular cycloaddition (Fig. 6), which can affect regioselectivity.

Nitrones **134**, **168**, **177**, and **202** are considered as both C-linked unsaturated nitrones with a four-carbon tether (**214**) and N-linked unsaturated nitrones with a three-carbon tether (**218**). Although the cycloaddition of acyclic unsaturated nitrones **214** or **218** can yield both regioisomers (**215A/B** or **219A/B**) depending on substituents, cycloaddition of cyclic nitrones **134**, **168**, **177**, and **202** produced sole regioisomers, corresponding to **223B**, **225B**, and **227B** (4-substituted isoxazolidines, Scheme 27).

Fig. 7 illustrates the transition states involved in the cycloaddition of acyclic C-linked unsaturated nitron **214**. Nitron **214** has two alkyl chains attached to the carbon and nitrogen atoms of the nitron. When the two alkyl chains are positioned on opposite sides, the configuration is referred to as “*trans*”, while when they are positioned on the same side, it is referred to as “*cis*”. In the transition state **228** from *trans*-nitron, the nitron and alkene moieties can smoothly interact to produce bridged isoxazolidine **215A**, whereas in the *cis*-nitron, interaction between the nitron and alkene moieties such as to form bridged isoxazolidine **215A** is geometrically restricted. This makes transition state **231** unfavorable. On the other hand, *cis*-nitron undergoes intramolecular cycloaddition to form fused isoxazoline **215B** through the transition state **232** or **233**.⁵⁷ Since cyclic nitrones **222** and **224** are *cis*-nitrones, cycloaddition



Scheme 27 Cycloaddition of cyclic nitrones.

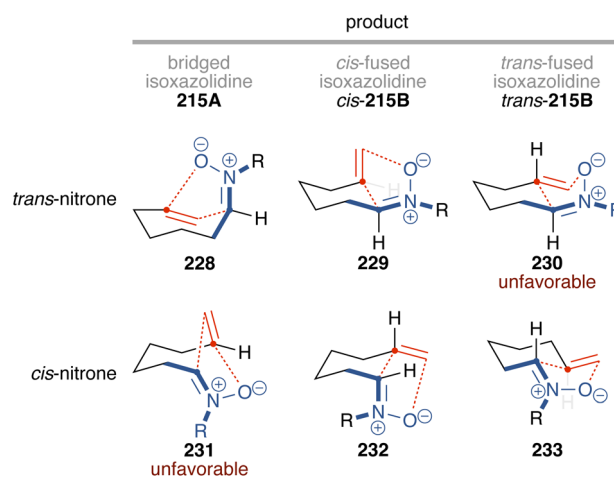


Fig. 7 Transition states in cycloaddition of C-linked unsaturated nitron **214**.

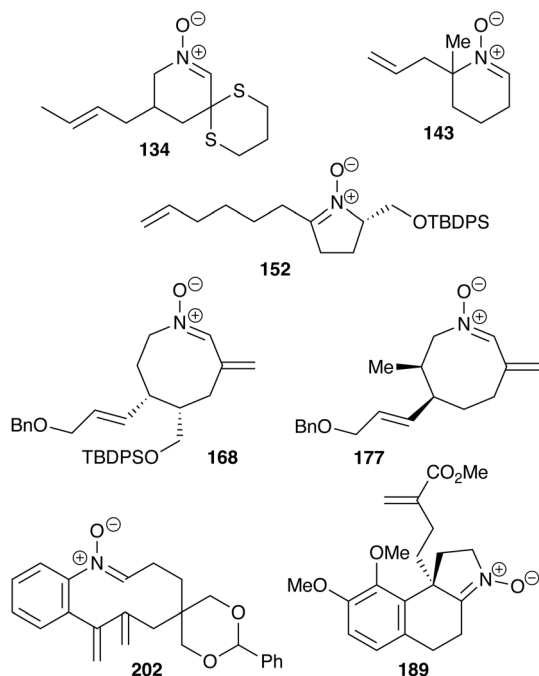


Fig. 6 Cyclic nitrones.

through a transition state such as **231** to form **223A** or **225A** is unfavorable. However, cycloaddition through a transition state such as **232** is expected to proceed. This discussion explains the selective formation of **223B** or **225B** in the cycloaddition of **222** or **224**. The transition states of these cycloaddition (**222^{TS}** and **224^{TS}**) can be depicted by adding a linkage to **232** (Fig. 8). The transition state **222^{TS}** adopts a boat-shaped six-membered conformation.

Cyclic nitron **226** is a *trans*-nitron, and therefore the above argument cannot be adopted to explain the selective formation



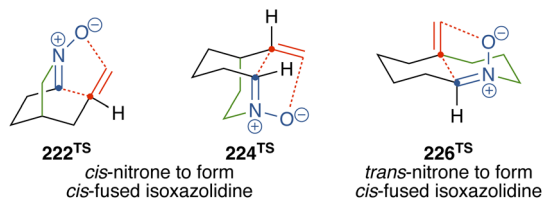


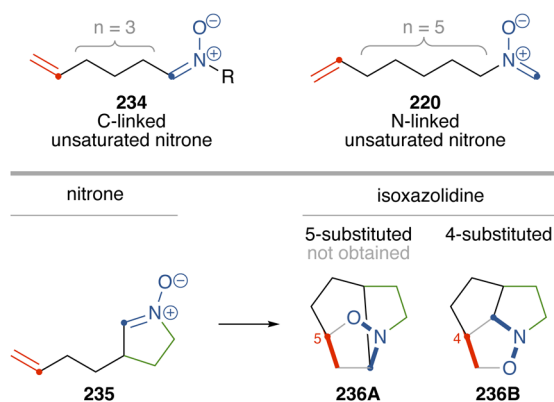
Fig. 8 Transition states in cycloaddition of cyclic nitrones.

of **227B**. In a transition state forming **227B** (**226^{TS}**, Fig. 8), newly formed rings other than the isoxazolidines are six-membered rings, while in the transition state to yield the regioisomers (**227A**), newly formed rings become seven-membered rings, causing strains in the transition state of the cycloaddition.

Nitrone **189** is considered as both C-linked unsaturated nitrones with a three-carbon tether (**234**) and N-linked unsaturated nitrones with a five-carbon tether (**220**), represented by **235** (Scheme 28). In the cycloaddition of acyclic nitrone **234**, the formation of a bridged isoxazolidine from a *cis*-nitrone is unfavorable, as is the transition state **231**, while in the transition state to form a fused isoxazolidine, the nitrone and alkene moieties can sufficiently interact. Since cyclic nitrone **235** is a *cis*-nitrone, this explains why the formation of bridged (5-substituted) isoxazolidine **236A** is unfavorable, leading to the selective formation of **236B** in the cycloaddition of **235**.⁵⁸

Nitrone **152** is a 5-membered cyclic nitrone, and can be considered as a C-linked unsaturated nitrone with a four-carbon tether (**214**, Fig. 4 and Scheme 27). Since the alkenyl chain is attached to the carbon atom of the nitrone, the geometry of the nitrone is necessarily *trans*, allowing for the formation of bridged and fused isoxazolidines (Fig. 7). Indeed, cycloaddition of nitrone **152** produced a mixture of bridged and fused isoxazolidines (**155** and **156** in Scheme 20).⁵⁹

Nitrone **143** is regarded as a N-linked unsaturated nitrone with a two-carbon tether (**216**, Fig. 5). The cycloaddition of **143** regioselectively produces isoxazolidine **144**, a 5-substituted isoxazolidine. This is consistent with the regioselectivity in cycloaddition of **216** to yield **217A** (5-substituted isoxazolidine, Fig. 5) as the major or sole product.⁶⁰



Scheme 28 Cycloaddition of cyclic nitrones.

6. Conclusions

Various ring systems of natural products have been constructed *via* the intramolecular cycloaddition of nitrones. The cycloaddition of nitrones can be carried out under simple conditions, typically heating in a solvent. This reaction enables the functionalization of alkenes through the formation of both C–O and C–C bonds, making it possible to construct even quaternary carbon centers. As discussed in Section 5, regioselectivity of intramolecular cycloaddition is often different from that of intermolecular cycloaddition of nitrones. In particular, intramolecular cycloaddition of cyclic nitrones tends to proceed in a regioselective manner as the additional bond forming the cyclic nitrone may limit the access between the nitrone and the alkene.

The preparation of nitrones is also an important issue in this review. Condensation of hydroxylamines with aldehydes or ketones is a simple method for the preparation of nitrones. In addition, the oxidation of amines (Schemes 2, 12 and 24), the *N*-alkylation of oximes (Schemes 4 and 18), and the transformation of an *N*-hydroxylactam (Scheme 20) are included in this review.⁶¹ For the preparation of hydroxylamines, various methods have been demonstrated, including reduction or *C*-allylation of an oxime (Schemes 12, 13 and 19), introduction of a hydroxylamine unit *via* a Mitsunobu reaction (Schemes 15, 21 and 22), and reduction of a nitro group (Scheme 23). After the cycloaddition, reduction with zinc, samarium(II) iodide, or RANEY® nickel is widely used to cleave the N–O bond in the isoxazolidine, yielding versatile 1,3-amino alcohols. Oxidative cleavage of the N–O bond has also been reported but is not included in this review.⁶² Further accumulation of results related to the cycloaddition of nitrones could lead to the synthesis of various molecules with complex structures.

7. Conflicts of interest

There are no conflicts to declare.

8. Acknowledgements

This work was supported by JSPS KAKENHI (Grant Numbers JP23H02602 and JP24H01767) and by the Research Support Project for Life Science and Drug Discovery [Basis for Supporting Innovative Drug Discovery and Life Science Research (BINDS)] from the Japan Agency for Medical Research and Development (AMED) under Grant Number JP24ama121044.

9. Notes and references

- For selected review on nitrones, see: (a) P. N. Confalone and E. M. Huie, in *Organic Reactions*, 2004, p. 1; (b) V. Nair and T. D. Suja, *Tetrahedron*, 2007, **63**, 12247; (c) A. Brandi, F. Cardona, S. Cicchi, F. M. Cordero and A. Goti, *Chem.–Eur. J.*, 2009, **15**, 7808; (d) J. M. B. Adam and C. Iain, *Curr. Org. Synth.*, 2010, **7**, 312; (e) M. Berthet, T. Cheviet, G. Dujardin, I. Parrot and J. Martinez, *Chem. Rev.*, 2016, **116**, 15235; (f) A. Padwa and S. Bur, in *Advances in*



- Heterocyclic Chemistry*, ed. E. F. V. Scriven and C. A. Ramsden, Academic Press, 2016, vol. 119, p. 241; (g) L. L. Anderson, *Asian J. Org. Chem.*, 2016, 5, 9; (h) A. Brandi, F. Cardona, S. Cicchi, F. M. Cordero and A. Goti, in *Organic Reactions*, 2017, p. 1; (i) S.-I. Murahashi and Y. Imada, *Chem. Rev.*, 2019, 119, 4684; (j) C. Kouklovsky, *Vietnam J. Chem.*, 2020, 58, 20; (k) S. Thakur, A. Das and T. Das, *New J. Chem.*, 2021, 45, 11420; (l) O. Tamura, *Chem. Pharm. Bull.*, 2024, 72, 731; (m) K. Tanaka, *Eur. J. Org. Chem.*, 2024, e202400202.
- 2 (a) N. A. LeBel, M. E. Post and J. J. Whang, *J. Am. Chem. Soc.*, 1964, 86, 3759; (b) W. Oppolzer and K. Keller, *Tetrahedron Lett.*, 1970, 11, 1117.
 - 3 (a) K. V. Gothelf and K. A. Jørgensen, *Chem. Commun.*, 2000, 1449; (b) T. Hashimoto and K. Maruoka, *Chem. Rev.*, 2015, 115, 5366.
 - 4 (a) F. Casuscelli, U. Chiacchio, A. Rescifina, R. Romeo, G. Romeo, S. Tommasini and N. Uccella, *Tetrahedron*, 1995, 51, 2979; (b) G. Floresta, C. Talotta, C. Gaeta, M. De Rosa, U. Chiacchio, P. Neri and A. Rescifina, *J. Org. Chem.*, 2017, 82, 4631; (c) P. DeShong, C. M. Dicken, R. R. Staib, A. J. Freyer and S. M. Weinreb, *J. Org. Chem.*, 1982, 47, 4397.
 - 5 (a) J. Sims and K. N. Houk, *J. Am. Chem. Soc.*, 1973, 95, 5798; (b) A. Padwa, L. Fisera, K. F. Koehler, A. Rodriguez and G. S. K. Wong, *J. Org. Chem.*, 1984, 49, 276.
 - 6 T. Shimizu, M. Ishizaki and N. Nitada, *Chem. Pharm. Bull.*, 2002, 50, 908.
 - 7 (a) L. R. Domingo, M. Ríos-Gutiérrez, A. I. Adjieufack, I. M. Ndassa, C. N. Nouhou and J. K. Mbadcam, *ChemistrySelect*, 2018, 3, 5412; (b) M. Bakthadoss and M. Mushaf, *Org. Biomol. Chem.*, 2020, 18, 9653; (c) A. Mondal, L. R. Domingo and N. Acharjee, *J. Phys. Org. Chem.*, 2024, 37, e4574.
 - 8 For natural product syntheses using intermolecular cycloaddition of nitron, see: (a) T. Ueda, M. Inada, N. Morita and O. Tamura, *Heterocycles*, 2015, 90, 1179; (b) T. Hirai, K. Shibata, Y. Niwano, M. Shiozaki, Y. Hashimoto, N. Morita, S. Ban and O. Tamura, *Org. Lett.*, 2017, 19, 6320; (c) M. Pieczykolan, B. Furman and M. Chmielewski, *J. Antibiot.*, 2017, 70, 781; (d) C. Wang, Z. Wang, X. Xie, X. Yao, G. Li and L. Zu, *Org. Lett.*, 2017, 19, 1828; (e) T. Pecchioli, F. Cardona, H.-U. Reissig, R. Zimmer and A. Goti, *J. Org. Chem.*, 2017, 82, 5835; (f) F. Xue, H. Lu, L. He, W. Li, D. Zhang, X.-Y. Liu and Y. Qin, *J. Org. Chem.*, 2018, 83, 754; (g) T. Malatinský, B. Otočková, L. Dikošová and R. Fischer, *ChemistrySelect*, 2019, 4, 4233; (h) B. D. A. Shennan, P. W. Smith, Y. Ogura and D. J. Dixon, *Chem. Sci.*, 2020, 11, 10354; (i) L. Dikošová, B. Otočková, T. Malatinský, J. Doháňošová, M. Kopáčová, A. Ďurinová, L. Smutná, F. Trejtnar and R. Fischer, *RSC Adv.*, 2021, 11, 31621; (j) X. Tian, T. Xuan, J. Gao, X. Zhang, T. Liu, F. Luo, R. Pang, P. Shao, Y.-F. Yang and Y. Wang, *Nat. Commun.*, 2024, 15, 6429.
 - 9 For selected synthetic studies of natural product using intramolecular cycloaddition of nitron, see: (a) S. R. Paladugu and R. E. Looper, *Tetrahedron Lett.*, 2015, 56, 6332; (b) E. Ideue, J. Shimokawa and T. Fukuyama, *Org. Lett.*, 2015, 17, 4964; (c) J. Rautschek and P. Metz, *Heterocycles*, 2017, 95, 1106; (d) J. S. Cannon, *Org. Lett.*, 2018, 20, 3883; (e) R.-J. Zhou, G.-Y. Dai, X.-H. Zhou, M.-J. Zhang, P.-Z. Wu, D. Zhang, H. Song, X.-Y. Liu and Y. Qin, *Org. Chem. Front.*, 2019, 6, 377; (f) J. Zhong, H. He and S. Gao, *Org. Chem. Front.*, 2019, 6, 3781.
 - 10 For a synthesis of natural product analogs using intramolecular cycloaddition of nitron, see: M. Yamaguchi, A. Matsuda and S. Ichikawa, *Org. Biomol. Chem.*, 2015, 13, 1187.
 - 11 For scalable syntheses using intramolecular cycloaddition of nitron, see: (a) S. P. Kolis, M. M. Hansen, E. Arslantas, L. Brändli, J. Buser, A. C. DeBaillie, A. L. Frederick, D. W. Hoard, A. Hollister, D. Huber, T. Kull, R. J. Linder, T. J. Martin, R. N. Richey, A. Stutz, M. Waibel, J. A. Ward and A. Zamfir, *Org. Process Res. Dev.*, 2015, 19, 1203; (b) P. Garcia-Losada, A. C. DeBaillie, J. Eugenio de Diego, S. J. Green, M. M. Hansen, C. Jaramillo, M. Johnson, T. Kaoudi, J. Li, P. J. Lindsay-Scott, C. Mateos, D. J. Mergott, J. A. Rincon, R. R. Rothhaar, K. D. Seibert, B. M. Watson, L. L. Winneroski, S. Gangula, D. Jing, H. Sun, L. Zhang and M. O. Frederick, *Org. Process Res. Dev.*, 2020, 24, 306.
 - 12 P. P. Painter, R. P. Pemberton, B. M. Wong, K. C. Ho and D. J. Tantillo, *J. Org. Chem.*, 2014, 79, 432.
 - 13 K. Koyama, Y. Hirasawa, A. E. Nugroho, T. Hosoya, T. C. Hoe, K.-L. Chan and H. Morita, *Org. Lett.*, 2010, 12, 4188.
 - 14 A. Y. Hong and C. D. Vanderwal, *J. Am. Chem. Soc.*, 2015, 137, 7306.
 - 15 For the Cope elimination to occur, the oxygen atom of the *N*-oxide and the α -proton of the ketone must be in a *cis*-relationship.
 - 16 B.-X. Zhao, Y. Wang, D.-M. Zhang, X.-J. Huang, L.-L. Bai, Y. Yan, J.-M. Chen, T.-B. Lu, Y.-T. Wang, Q.-W. Zhang and W.-C. Ye, *Org. Lett.*, 2012, 14, 3096.
 - 17 (a) J. M. E. Hughes and J. L. Gleason, *Angew. Chem., Int. Ed.*, 2017, 56, 10830; (b) J. M. E. Hughes and J. L. Gleason, *Tetrahedron*, 2018, 74, 759.
 - 18 For the syntheses of virosaines A and B reported before 2015, see: (a) H. Wei, C. Qiao, G. Liu, Z. Yang and C.-c. Li, *Angew. Chem., Int. Ed.*, 2013, 52, 620; (b) H. Miyatake-Ondozabal, L. M. Bannwart and K. Gademann, *Chem. Commun.*, 2013, 49, 1921.
 - 19 (a) E. J. Corey and T.-P. Loh, *Tetrahedron Lett.*, 1993, 34, 3979; (b) C. S. Schindler, C. R. J. Stephenson and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2008, 47, 8852; (c) H. F. Zipfel and E. M. Carreira, *Org. Lett.*, 2014, 16, 2854.
 - 20 G. E. Keck and J. B. Yates, *J. Am. Chem. Soc.*, 1982, 104, 5829.
 - 21 X.-J. Wang, G.-J. Zhang, P.-Y. Zhuang, Y. Zhang, S.-S. Yu, X.-Q. Bao, D. Zhang, Y.-H. Yuan, N.-H. Chen, S.-g. Ma, J. Qu and Y. Li, *Org. Lett.*, 2012, 14, 2614.
 - 22 H. Shao, K. Fang, Y.-P. Wang, X.-M. Zhang, T.-M. Ding, S.-Y. Zhang, Z.-M. Chen and Y.-Q. Tu, *Org. Lett.*, 2020, 22, 3775.
 - 23 (a) A. Hassner, R. Maurya, A. Padwa and W. H. Bullock, *J. Org. Chem.*, 1991, 56, 2775; (b) A. Hassner, R. Maurya,



- O. Friedman, H. E. Gottlieb, A. Padwa and D. Austin, *J. Org. Chem.*, 1993, **58**, 4539.
- 24 J. Zhong, K. Chen, Y. Qiu, H. He and S. Gao, *Org. Lett.*, 2019, **21**, 3741.
- 25 H. Shi, I. N. Michaelides, B. Darses, P. Jakubec, Q. N. N. Nguyen, R. S. Paton and D. J. Dixon, *J. Am. Chem. Soc.*, 2017, **139**, 17755.
- 26 H. Mao, P.-M. Wang and J. Xu, *Tetrahedron*, 2021, **81**, 131913.
- 27 J.-T. Lu, Y. Zong, X. Yue and J. Wang, *J. Org. Chem.*, 2023, **88**, 8761.
- 28 S. Nicolai, C. Piemontesi and J. Waser, *Angew. Chem., Int. Ed.*, 2011, **50**, 4680.
- 29 T. Miura, Y. Funakoshi and M. Murakami, *J. Am. Chem. Soc.*, 2014, **136**, 2272.
- 30 (a) Y. Wang, A. Hennig, T. Küttler, C. Hahn, A. Jäger and P. Metz, *Org. Lett.*, 2020, **22**, 3145; (b) T. Erhard, G. Ehrlich and P. Metz, *Angew. Chem., Int. Ed.*, 2011, **50**, 3892; (c) J. Rautschek, A. Jäger and P. Metz, *Org. Lett.*, 2018, **20**, 832.
- 31 T. Yokoyama, Y. Fukami, T. Sato and N. Chida, *Chem.-Asian J.*, 2016, **11**, 470.
- 32 F.-X. Wang, J.-Y. Du, H.-B. Wang, P.-L. Zhang, G.-B. Zhang, K.-Y. Yu, X.-Z. Zhang, X.-T. An, Y.-X. Cao and C.-A. Fan, *J. Am. Chem. Soc.*, 2017, **139**, 4282.
- 33 G.-B. Zhang, F.-X. Wang, J.-Y. Du, H. Qu, X.-Y. Ma, M.-X. Wei, C.-T. Wang, Q. Li and C.-A. Fan, *Org. Lett.*, 2012, **14**, 3696.
- 34 J. K. Kerkovius and M. A. Kerr, *J. Am. Chem. Soc.*, 2018, **140**, 8415.
- 35 M. A. Staszak and C. W. Doecke, *Tetrahedron Lett.*, 1993, **34**, 7043.
- 36 A. M. Beauchemin, *Org. Biomol. Chem.*, 2013, **11**, 7039.
- 37 Z. T. I. Alkayar and I. Coldham, *Org. Biomol. Chem.*, 2019, **17**, 66.
- 38 J. H. Siitonen, P. V. Kattamuri, M. Yousufuddin and L. Kürti, *Org. Lett.*, 2020, **22**, 2486.
- 39 S. Hiraoka, T. Matsumoto, K. Matsuzaka, T. Sato and N. Chida, *Angew. Chem., Int. Ed.*, 2019, **58**, 4381.
- 40 Y. Sasano, S. Nagasawa, M. Yamazaki, M. Shibuya, J. Park and Y. Iwabuchi, *Angew. Chem., Int. Ed.*, 2014, **53**, 3236.
- 41 T. Higo, T. Ukegawa, S. Yokoshima and T. Fukuyama, *Angew. Chem., Int. Ed.*, 2015, **54**, 7367.
- 42 M. H. Becker, P. Chua, R. Downham, C. J. Douglas, N. K. Garg, S. Hiebert, S. Jaroch, R. T. Matsuoka, J. A. Middleton, F. W. Ng and L. E. Overman, *J. Am. Chem. Soc.*, 2007, **129**, 11987.
- 43 K. Tomoya, K. Komiya, D. Nakajima, N. Umekubo and S. Yokoshima, *Org. Lett.*, 2023, **25**, 2718.
- 44 S. A. Shipilovskikh, A. E. Rubtsov and A. V. Malkov, *Org. Lett.*, 2017, **19**, 6760.
- 45 B. Yang, G. Li, Q. Wang and J. Zhu, *J. Am. Chem. Soc.*, 2023, **145**, 5001.
- 46 (a) T. Okino, Y. Hoashi and Y. Takemoto, *J. Am. Chem. Soc.*, 2003, **125**, 12672; (b) S.-G. Wang, X.-J. Liu, Q.-C. Zhao, C. Zheng, S.-B. Wang and S.-L. You, *Angew. Chem., Int. Ed.*, 2015, **54**, 14929.
- 47 B. J. Lee, K. S. DeGlopper and T. P. Yoon, *Angew. Chem., Int. Ed.*, 2020, **59**, 197.
- 48 Y. Irie and S. Yokoshima, *J. Am. Chem. Soc.*, 2024, **146**, 9526.
- 49 (a) S. T. Diver and A. J. Giessert, *Chem. Rev.*, 2004, **104**, 1317; (b) S. Kotha, M. Meshram and A. Tiwari, *Chem. Soc. Rev.*, 2009, **38**, 2065.
- 50 S. Murahashi, H. Mitsui, T. Shiota, T. Tsuda and S. Watanabe, *J. Org. Chem.*, 1990, **55**, 1736.
- 51 (a) S. L. Gómez Ayala, E. Stashenko, A. Palma, A. Bahsas and J. M. Amaro-Luis, *Synlett*, 2006, **2006**, 2275; (b) S. Gómez-Ayala, J. A. Castrillón, A. Palma, S. M. Leal, P. Escobar and A. Bahsas, *Bioorg. Med. Chem.*, 2010, **18**, 4721; (c) L. M. Acosta, A. Palma and A. Bahsas, *Tetrahedron*, 2010, **66**, 8392; (d) L. M. Acosta Quintero, A. Palma, M. Nogueras and J. Cobo, *Synthesis*, 2012, **44**, 3765; (e) S. A. Guerrero, J. E. Ramírez, C. M. Sanabria, L. M. Acosta, J. Cobo, M. Nogueras and A. Palma, *Synthesis*, 2020, **53**, 1315; (f) Y. Li, J. S. Ng, B. Wang and S. Chiba, *Org. Lett.*, 2021, **23**, 5060.
- 52 H. Mitsui, S.-i. Zenki, T. Shiota and S.-I. Murahashi, *J. Chem. Soc., Chem. Commun.*, 1984, 874.
- 53 (a) Y. Shvo, D. Czarkie, Y. Rahamim and D. F. Chodosh, *J. Am. Chem. Soc.*, 1986, **108**, 7400; (b) B. L. Conley, M. K. Pennington-Boggio, E. Boz and T. J. Williams, *Chem. Rev.*, 2010, **110**, 2294.
- 54 For selected examples of C-linked unsaturated nitrones, see: (a) J. R. Hwu and J. A. Robl, *J. Chem. Soc., Chem. Commun.*, 1986, 704; (b) K. Vanhessche, C. G. Bello and M. Vandewalle, *Synlett*, 1991, **1991**, 921; (c) S. Saito, T. Ishikawa and T. Moriwake, *J. Org. Chem.*, 1994, **59**, 4375; (d) N. A. Lebel and N. Balasubramanian, *Tetrahedron Lett.*, 1985, **26**, 4331; (e) A. Arnone, G. Broggini, D. Passarella, A. Terraneo and G. Zecchi, *J. Org. Chem.*, 1998, **63**, 9279; (f) E. M. Beccalli, G. Broggini, C. La Rosa, D. Passarella, T. Pilati, A. Terraneo and G. Zecchi, *J. Org. Chem.*, 2000, **65**, 8924; (g) Q. Cheng, W. Zhang, Y. Tagami and T. Oritani, *J. Chem. Soc., Perkin Trans. 1*, 2001, 452; (h) D. Basso, G. Broggini, D. Passarella, T. Pilati, A. Terraneo and G. Zecchi, *Tetrahedron*, 2002, **58**, 4445; (i) J. Marcus, J. Brussee and A. van der Gen, *Eur. J. Org. Chem.*, 1998, **1998**, 2513; (j) Y. Liu, A. Maden and W. V. Murray, *Tetrahedron*, 2002, **58**, 3159; (k) T. K. M. Shing, W. F. Wong, T. Ikeno and T. Yamada, *Org. Lett.*, 2007, **9**, 207; (l) M. Würdemann and J. Christoffers, *Synlett*, 2011, 2841; (m) F. Rabasa-Alcañiz, A. Asensio, M. Sánchez-Roselló, M. Escolano, C. del Pozo and S. Fustero, *J. Org. Chem.*, 2017, **82**, 2505; (n) T. Tanino, M. Yamaguchi, A. Matsuda and S. Ichikawa, *Eur. J. Org. Chem.*, 2014, **2014**, 1836.
- 55 (a) W. Oppolzer, S. Siles, R. L. Snowden, B. H. Bakker and M. Petrzilka, *Tetrahedron*, 1985, **41**, 3497; (b) A. Budzińska and W. Sas, *Tetrahedron*, 2001, **57**, 2021; (c) N. Saha and S. K. Chattopadhyay, *J. Org. Chem.*, 2012, **77**, 11056.
- 56 (a) H. A. Dondas, M. Frederickson, R. Grigg, J. Markandu and M. Thornton-Pett, *Tetrahedron*, 1997, **53**, 14339; (b) Z. Xiang, *Comput. Theor. Chem.*, 2012, **992**, 128.
- 57 The transition state **230**, where a *trans*-fused isoxazolidine is produced from a *trans*-nitron, is also not preferred.
- 58 A. J. M. Burrell, I. Coldham and N. Oram, *Org. Lett.*, 2009, **11**, 1515.



- 59 R. Saruengkhanphasit, D. Collier and I. Coldham, *J. Org. Chem.*, 2017, **82**, 6489.
- 60 (a) R. E. Looper and R. M. Williams, *Tetrahedron Lett.*, 2001, **42**, 769; (b) R. E. Looper, M. T. C. Runnegar and R. M. Williams, *Tetrahedron*, 2006, **62**, 4549.
- 61 Reduction of *N*-hydroxyamide and transformation of nitro compounds have also been reported for the synthesis of nitrones (a) S. Katahara, S. Kobayashi, K. Fujita, T. Matsumoto, T. Sato and N. Chida, *Bull. Chem. Soc. Jpn.*, 2017, **90**, 893; (b) S. Katahara, S. Kobayashi, K. Fujita, T. Matsumoto, T. Sato and N. Chida, *J. Am. Chem. Soc.*, 2016, **138**, 5246; (c) G. Bartoli, E. Marcantoni, M. Petrini and R. Dalpozzo, *J. Org. Chem.*, 1990, **55**, 4456; (d) H. Shimizu, K. Yoshinaga and S. Yokoshima, *Org. Lett.*, 2021, **23**, 2704.
- 62 For selected examples on oxidative cleavage of isoxazolidines, see: (a) J. J. Tufariello, G. B. Mullen, J. J. Tegeler, E. J. Trybulski, S. C. Wong and S. A. Ali, *J. Am. Chem. Soc.*, 1979, **101**, 2435; (b) J. J. Tufariello and J. M. Puglis, *Tetrahedron Lett.*, 1986, **27**, 1489; (c) T. Ishiwata, T. Hino, H. Koshino, Y. Hashimoto, T. Nakata and K. Nagasawa, *Org. Lett.*, 2002, **4**, 2921; (d) K. Nagasawa, T. Ishiwata, Y. Hashimoto and T. Nakata, *Tetrahedron Lett.*, 2002, **43**, 6383; (e) S.-H. Yang and V. Caprio, *Synlett*, 2007, 1219; (f) S.-H. Yang, G. R. Clark and V. Caprio, *Org. Biomol. Chem.*, 2009, **7**, 2981; (g) Y. Fukahori, Y. Takayama, T. Imaoka, O. Iwamoto and K. Nagasawa, *Chem.-Asian J.*, 2013, **8**, 244.

