

# Sustainable Food Technology

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### Sustainability Spotlight Statement

This research advances sustainable materials and food packaging technologies by valorizing rice straw, an underutilized agricultural residue, into high-performance all-cellulose nanocomposite films. Rice straw, which is often burned or discarded after harvest, is converted into a fully bio-based, biodegradable packaging material through cellulose extraction, regeneration, and nanofiber reinforcement. The study contributes to agricultural waste reduction while creating value-added materials for sustainable packaging applications. By enhancing mechanical strength, barrier performance, and biodegradability through cellulose nanofiber incorporation, this work supports the development of renewable alternatives to petroleum-based plastics. The research aligns with circular economy principles by promoting resource efficiency and closed-loop material use and demonstrates potential benefits for sustainable food packaging industries.



1 **Development and Characterization of Rice Straw-Derived All-Cellulose Nanocomposite**  
2 **Films Reinforced with Cellulose Nanofibers Obtained via High-Intensity**  
3 **Ultrasonication and High-Shear Dispersion**

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30

31 **Abstract:** Agricultural residue, rice straw, represents an underutilized source of cellulose with  
32 potential for value-added applications. In the present study, all-cellulose nanocomposite films  
33 were developed by incorporating cellulose nanofibers (CNFs) as reinforcing nanofillers within  
34 a regenerated cellulose matrix. The films were fabricated using a lithium chloride/N, N-  
35 dimethylacetamide (LiCl/DMAc) solvent system, providing a homogeneous dispersion of  
36 CNFs and facilitating strong interfacial interactions between the matrix and the nanofillers. The  
37 influence of CNF concentrations (0%, 3%, 5%, 7%, and 9%) on the morphology, barrier  
38 properties, crystallinity, mechanical performance, optical transparency, and thermal properties  
39 of the all-cellulose nanocomposite (ACNC) films was systematically evaluated.

40 The results indicated that tensile strength and modulus increased significantly with higher CNF  
41 concentrations, although the films exhibited brittleness at 9% CNF. FTIR analysis showed that  
42 the functional groups in the cellulose structure remained intact in the nanocomposites. The  
43 surface morphology of ACNCs studied through the Field Emission-Scanning Electron  
44 Microscopy (FE-SEM) showed the uniform distribution of CNF within the cellulose matrix.

45 The XRD analysis indicated that the incorporation of CNFs increased the crystallinity index of  
46 ACNC films, with CNF7 exhibiting the highest CI of 61.50%. The films were also  
47 characterized for their density, porosity, and moisture content, which were found to be  
48 influenced by CNF concentration. The water vapour transmission rate (WVTR) and oxygen  
49 transmission rate (OTR) of CNF7 were  $44.7 \pm 3.0$  g/m<sup>2</sup>/day and  $5.6 \pm 0.8$  cm<sup>3</sup>/m<sup>2</sup>/day,  
50 respectively, which were significantly lower than those of CNF0, likely due to increased  
51 tortuosity arising from CNF reinforcement. Although optical transmittance decreased with the  
52 incorporation of CNFs, the films retained sufficient transparency for food packaging  
53 applications. Thermal stability was also enhanced upon CNF addition, with the peak  
54 degradation temperature reaching 335.4 °C in CNF7. Biodegradability assessment using  
55 enzymatic degradation showed complete degradation of CNF7 within 75 days. Overall, the  
56 results highlight strong intermolecular interactions between CNFs and the cellulose matrix,  
57 leading to enhanced functional properties and demonstrating the potential of CNF-reinforced  
58 films for sustainable food packaging applications.

59  
60 **Keywords:** all-cellulose nanocomposite, cellulose nanofibers, rice straw, tensile strength,  
61 biodegradability

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## 71 1. Introduction

72 Rice is a major staple crop that feeds millions of more than half of the world's population, but  
73 its cultivation generates a significant amount of agricultural waste in the form of rice straw.  
74 Approximately 1 billion tonnes of rice straw are generated globally, with about 91% produced  
75 in Asia.<sup>1</sup> Among Asian countries, China is the largest producer of rice straw, followed by India,  
76 Indonesia, and Bangladesh.<sup>2</sup> The large-scale generation of rice straw presents a significant  
77 global management challenge, while also offering substantial opportunities for sustainability.  
78 However, a significant proportion of rice straw continues to be burned in countries such as  
79 Indonesia, India, Vietnam, China, and Pakistan.<sup>3</sup> In India, around 140 million tonnes of rice  
80 straw are generated annually, of which 84 million tonnes are burned in open fields due to the  
81 lack of efficient disposal and utilization practices.<sup>1,4</sup> Rice straw, with its high silica and lignin  
82 content, has poor nutritional value and limited use as animal fodder, causing large-scale  
83 accumulation on farms. Open field burning of straw releases harmful pollutants (CO<sub>2</sub>, CO,  
84 NO<sub>x</sub>, SO<sub>x</sub>, VOCs, etc.), worsening air quality, contributing to respiratory diseases, and  
85 depleting soil nutrients.<sup>5</sup> Developing sustainable strategies to convert this agricultural residue  
86 into cellulose-based materials could provide an eco-friendly alternative for packaging  
87 applications, reducing environmental impact while supporting circular bioeconomy goals.  
88 Cellulose is a linear biopolymer recognized for its biodegradability, non-toxicity, and high  
89 strength, which enhances its wide applicability in diverse fields. In nanoforn, cellulose exists  
90 as cellulose nanocrystals (CNCs) and cellulose nanofibers (CNFs), which differ in structure  
91 and morphology. Owing to their higher aspect ratio, CNFs impart superior tensile strength and  
92 modulus to composites compared to CNCs, highlighting their potential in reinforcing  
93 packaging materials.<sup>6</sup> CNFs form an entangled, network-like structure through hydrogen  
94 bonding and van der Waals interactions, which contribute to their reinforcing ability in  
95 composite materials.<sup>7,8</sup> Their high surface area, aspect ratio, and mechanical strength,  
96 combined with biodegradability, renewability, and stability, make CNFs highly attractive as  
97 nanofillers for sustainable packaging materials. Growing concern over environmental impact  
98 has accelerated the search for sustainable packaging alternatives.<sup>9</sup> CNF-based materials,  
99 derived from renewable biomass, provide strong mechanical properties, improved barrier  
100 performance, and biodegradability, making them promising candidates for food packaging.  
101 Their use supports product preservation and freshness and aligns with global sustainability  
102 goals by reducing carbon footprint and environmental burden.<sup>10</sup>  
103 All cellulose nanocomposites (ACNCs) are a subclass of green nanocomposites, where both  
104 the matrix and reinforcement phase are derived from cellulose.<sup>11</sup> This chemical compatibility  
105 enhances interfacial adhesion and stress transfer, resulting in superior mechanical strength and  
106 barrier performance. ACNC films, therefore, hold strong potential for application in fresh food  
107 packaging as sustainable, bio-based, and biodegradable alternatives.<sup>12</sup> ACNCs using  
108 microcrystalline cellulose and CNFs derived from wood pulp were fabricated by Guzmán-  
109 Puyol *et al.*<sup>11</sup> and observed significant improvements in mechanical strength and water vapor  
110 barrier properties, making them suitable for food packaging.  
111 Similarly, Nascimento *et al.*<sup>12</sup> employed bacterial cellulose nanofibers as a matrix, reinforced  
112 with CNCs (0–5%), to produce ACNCs. The fabricated films exhibited high crystallinity,  
113 enhanced mechanical performance, improved water resistance, and reduced vapor  
114 permeability, thereby underscoring their potential for food packaging applications. In another  
115 study, Rader *et al.*<sup>13</sup> prepared ACNCs with Cellulose nanocrystals (CNCs) and Hydroxypropyl

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116 cellulose (HPC). The film was transparent, with improved tensile strength and extensibility.  
117 The composites retained mechanical properties up to 75% relative humidity. Oxygen  
118 permeability decreased nearly tenfold at a 70% CNC concentration to levels comparable to  
119 those of PET, though water resistance remained limited, which indicated the need to optimize  
120 water barrier properties, important for food packaging. Yang *et al.*<sup>14</sup> demonstrated the  
121 development of all-cellulose nanocomposite films by mixing aqueous TEMPO-oxidized  
122 cellulose nanofibril dispersion obtained from softwood bleached kraft pulp with  
123 NaOH/urea/cellulose solutions prepared from cotton linter. The NaOH/urea/water solvent  
124 system used in their study represented an environmentally benign dissolution medium. The  
125 ACNCs have also been prepared from the components derived from various agricultural  
126 residues, including wheat straw<sup>15</sup>, sugarcane bagasse<sup>16</sup>, and corncobs<sup>17</sup>.

127 The characteristic properties of cellulose and cellulose nanofibers depend on the biomass  
128 source of the fibers and the methods of their preparation, while the mechanical and barrier  
129 properties of the ACNC films prepared from them are affected by the concentration of  
130 nanofibers used as filler and the characteristics of the cellulose matrix<sup>18, 15</sup>. Rice straw, an  
131 abundantly available agricultural residue with associated disposal challenges, can be a  
132 promising low-cost raw material for the development of ACNCs. Despite extensive studies on  
133 the utilization of rice straw-derived CNF as a reinforcement agent in various polymer matrices  
134<sup>19,20,21</sup>, the studies investigating the combined utilization of rice straw-derived cellulose and  
135 CNFs within a single material system remain scarce and inconclusive. In our earlier work,  
136 cellulose was isolated from rice straw via alkali extraction, and the CNFs were prepared using  
137 an ultrasonication-assisted extraction technique coupled with high-shear dispersion<sup>22</sup>. In the  
138 present study, the ACNC films were developed from rice-straw-derived cellulose and various  
139 concentrations of CNFs, and the films were comprehensively characterized for their  
140 mechanical, barrier, optical, thermal, and biodegradation properties, considering their potential  
141 use in food packaging within a circular economy framework.

## 142 Experimental

### 143 2.1 Materials

144 Rice straw as obtained from the Experimental Agricultural Farm, Department of Agriculture,  
145 Sharda University, Greater Noida, India. Further, it was properly washed with deionized water  
146 and dried in a hot air oven at 40 °C for 24 h. It was then cut manually into smaller sizes and  
147 stored properly in airtight polyethylene bags at ambient conditions for further investigation.  
148 Chemicals used in the study included Sodium hydroxide (NaOH), Sodium chlorite (NaClO<sub>2</sub>),  
149 Sodium chloride (NaCl), and N, N-Dimethylacetamide (DMAc), which were purchased from  
150 Thermo Fischer India, and Lithium chloride (LiCl) was purchased from Loba Chemie Pvt Ltd.,  
151 India.

### 153 2.2 Isolation of CNFs

154 Cellulose nanofibers (CNFs) were isolated following the procedure described in our previous  
155 work, Jadaun *et al.*<sup>22</sup> In brief, rice straw of short length of 4-5 cm was immersed in hot deionized  
156 water for 1 h and subjected to continuous stirring for an additional hour. Afterward, the straw  
157 was thoroughly washed with deionized water and dried overnight at 40 °C in a hot air oven.  
158 The dried straw was then finely ground using a RETSCH Knife Mill Grindomix GM300  
159 (Fischer Scientific) and sieved through a 60-mesh sieve. The ground straw was alkali-treated



160 with 12% NaOH at 121 °C for 1 h. After treatment, the alkali-treated residue was washed  
161 repeatedly with deionized water to obtain a neutral pH and then dried in an oven at 50 °C  
162 overnight. Next, bleaching was performed using 10% acetified sodium chlorite (pH 3-5,  
163 adjusted with acetic acid) at 70 °C for 1 h. This process was repeated five times to effectively  
164 remove lignin. The resulting bleached fibers were filtered thoroughly with deionized water to  
165 obtain a neutral pH and then dried overnight at 50 °C in a hot air oven to obtain purified  
166 cellulose.

167 A 0.1% cellulose fiber suspension was prepared by dispersing dried cellulose fibers in  
168 deionized water under continuous stirring. The dispersion was homogenized using IKA T18  
169 Digital ULTRA-TURRAX at 17000 rpm for 2 h, with 10 min intermittent intervals.  
170 Subsequently, 60 ml of homogenized suspension was subjected to a high-intensity probe  
171 sonicator (VCX 750, Sonics and Materials Inc., USA) at 80% amplitude and 20 kHz frequency  
172 for 2h using a 13 mm probe. The resulting CNFs were freeze-dried and stored for further  
173 analysis.

### 175 2.3 Development of All-Cellulose Nanocomposite Films

176 A predetermined amount of cellulose was activated in 100 ml of *N, N*-dimethylacetamide  
177 (DMAc) at 80 °C for 1 h under continuous stirring. After activation, 8 g of LiCl was added to  
178 the solution, and stirring was constant at 80 °C for an additional 2 h. Subsequently, the  
179 temperature was lowered to 40 °C, and stirring was continued overnight to form transparent  
180 solutions. To assess the effect of CNF incorporation, varying concentrations of CNF, viz.,  
181 3.0%, 5.0%, 7.0%, and 9.0%, denoted as CNF3, CNF5, CNF7, and CNF9, respectively, were  
182 incorporated. CNFs were incorporated into this solution by adding a predetermined amount of  
183 CNF suspension under continuous stirring at 60 °C for 90 min to achieve uniform dispersion.  
184 The resulting solutions were stored under refrigeration to ensure consistency. A control sample  
185 (CNF0) was prepared using only cellulose, without any CNF. The prepared viscous solution  
186 was then cast on the Petri dishes and air-dried overnight to form a gel. These gels were then  
187 immersed in deionized water, with the solvent replaced every 2 hours for the complete removal  
188 of the DMAc/LiCl. The purified gels were transferred onto silicon mats and affixed to the glass  
189 plates using binder clips to avoid shrinkage. The films were subsequently left to air dry at room  
190 temperature.

### 192 2.4 Characterization

#### 193 2.4.1 Chemical composition and Yield of cellulose

194 The chemical composition of the raw material, including cellulose, hemicellulose, and lignin,  
195 was determined before alkali and bleaching treatments. The analysis was performed using a  
196 slightly adapted Van Soest and Wine method<sup>23</sup>, with modifications based on the procedure  
197 described by Pradhan and Bhatia<sup>24</sup>, as detailed in our previous work<sup>22</sup>. In order to investigate  
198 the efficiency of alkaline and bleaching treatment in removing amorphous parts, the yield of  
199 cellulose was determined via the gravimetric method. The yield of extracted cellulose after  
200 these treatments was calculated using the formula given in equation (1):

$$202 \text{Yield (\%)} = \text{Final weight g} \div \text{Initial weight (g)} \times 100 \quad (1)$$

#### 204 2.4.2 Sedimentation stability and Characterization of CNF



205 A cellulose suspension (1% w/w) was prepared in distilled water and subjected to high-shear  
 206 dispersion followed by high-intensity ultrasonication, as described in our previous work<sup>21</sup>. The  
 207 sedimentation stability of the resulting CNF suspension was evaluated by visual observation  
 208 over time. The presence of nanofiber and diameter of CNFs were confirmed by FE-SEM,  
 209 whereas parameters such as crystallinity, chemical composition, zeta potential, and thermal  
 210 degradation behaviour were analysed by XRD, FTIR, zeta potential analyzer, and TGA as per  
 211 the methodology described in Jadaun *et al.*<sup>22</sup>

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### 2.4.3 Viscosity-average degree of polymerisation ( $DP_v$ ) and molecular weight ( $M_w$ )

214 Viscometric analysis of the Molecular weight ( $M_w$ ) and viscosity-average degree of  
 215 polymerisation ( $DP_v$ ) of samples was conducted according to the procedure described in  
 216 previous studies by Kadivar *et al.*<sup>25</sup> and Maraghechi *et al.*<sup>26</sup>, respectively, using a Cannon–  
 217 Fenske viscometer and LiCl/DMAc solvent. Initially, bleached cellulose fibers, CNFs, and  
 218 cellulose-CNFs solutions were prepared in 8% LiCl/DMAc solvent system under constant  
 219 stirring speed to obtain a uniform solution.

220 Using LiCl/DMAc as the solvent, the solvent efflux time of the solvent and each solution was  
 221 measured at 25 °C to determine intrinsic viscosity. From which, reduced viscosities of each  
 222 solution (0.95, 1.15, 1.5, 1.85, and 3 (mg/cm<sup>3</sup>)) were determined according to Hao *et al.*<sup>27</sup> using  
 223 the equation (2) given below:

$$\eta_{red} = \frac{t-t_0}{ct_0} \quad (2)$$

225 where  $t_0$  denotes the solvent efflux time,  $t$  is the efflux time of the solution, and  $c$  represents  
 226 the cellulose solution concentration (g/cm<sup>3</sup>).

227 The intrinsic viscosities were obtained by linear extrapolation of reduced viscosity versus  
 228 concentration to zero concentration. From the intrinsic viscosity  $[\eta]$  of the samples, the  
 229 corresponding values of the molecular weight were obtained by applying the Mark-Houwink-  
 230 Sakurada equation (3):

$$[\eta] = KM_w^\alpha \text{ (cm}^3 \text{ /g)} \quad (3)$$

232 where the two empirical constants used were  $K = 1.278 \times 10^{-4}$  and  $\alpha = 1.19$  (McCormick *et al.*<sup>28</sup>)  
 233 for the LiCl/DMAc solvent system.

234 For the determination of Viscosity-average degree of polymerisation ( $DP_v$ ), the relation  
 235 between the intrinsic viscosity ( $\eta$ ) with Viscosity-average degree of polymerisation ( $DP_v$ )  
 236 established through the Mark-Houwink equation was used, as depicted in equation (4):

$$[\eta] = K (DP_v)^\alpha \quad (4)$$

### 2.4.4 Field Emission- Scanning Electron Microscopy (FE-SEM)

239 The surface morphology of pure cellulose film (CNF0) and ACNC films was observed using  
 240 FE-SEM (JSM-7610F Plus JEOL, Japan) at an accelerating voltage of 15 kV at magnifications  
 241 of 8,000x and 10,000x. Prior to imaging, film samples were affixed to aluminium stubs using  
 242 double-sided conductive carbon tape. Using a sputter coater (DII-29030SCTR Smart Coater  
 243 JEOL, Japan), samples were sputter-coated with a thin conductive/gold layer at 5nm/min to  
 244 enhance conductivity.

### 2.4.5 Fourier Transform Infrared Spectroscopy (FT-IR)

247 The functional groups of the film samples were identified by FTIR (Agilent Cary 630 FTIR,  
 248 USA) equipped with an Attenuated Total Reflectance (ATR) unit. Spectral data were recorded



249 over the range 4000 – 400 cm<sup>-1</sup>, with each sample scanned 32 times at a spectral resolution of  
 250 2 cm<sup>-1</sup>.

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#### 252 2.4.6 XRD analysis

253 XRD patterns of all the films were recorded using an X-ray diffractometer (SmartLab 3kW/  
 254 Rigaku) equipped with Cu K $\alpha$  radiation ( $\lambda=1.54\text{ \AA}$ ). Scans were performed over a 2 $\theta$  range of  
 255 5-90° at an accelerating voltage of 40 kV with a current of 40 mA. The films were mounted in  
 256 a sample holder, and measurements were carried out in a static configuration. The XRD data  
 257 were analyzed using Origin Pro 2026 (64-bit) software. The crystallinity index (CI) was  
 258 calculated using the peak height method proposed by Segal *et al.*<sup>29</sup>

$$259 \quad CI (\%) = \left( \frac{I_{002} - I_{am}}{I_{002}} \right) \times 100 \quad (5)$$

260 where  $I_{002}$  corresponds to the peak intensity of the (002) crystalline plane and  $I_{am}$  represent  
 261 the minimum intensity located in the amorphous region between (002) and (001) peaks.  
 262

262

#### 263 2.4.7 Moisture content

264 The moisture content of the films was obtained using the gravimetric method. Film specimens  
 265 (2×2 cm<sup>2</sup>) were initially weighed ( $M_o$ ) and kept at 23 °C and 50% relative humidity (RH) until  
 266 a constant weight was achieved ( $M_f$ ). The moisture content was determined using the formula  
 267 given below:

$$268 \quad \text{Moisture content} = \frac{(M_o - M_f)}{M_o} \times 100\% \quad (6)$$

#### 269 2.4.8 Density and Porosity

270 The thickness of each film was measured at 8 randomly selected points per film using a digital  
 271 micrometer (Model 293-240-30, Mitutoyo Corporation, Japan) to the nearest 0.001 mm. The  
 272 mean value of these measurements was calculated to determine the overall thickness of each  
 273 film.

274 The density of the films ( $\rho_s$ ) was determined by dividing the dry mass of the film by its  
 275 geometric dimensions<sup>30,31</sup> as presented in equation (7) below.

$$276 \quad \rho_s = \frac{m}{A \times \delta} \quad (7)$$

277 where  $m$  represents the dry mass of the sample (g),  $A$  denotes the surface area of the films (2  
 278 ×2 cm<sup>2</sup>), and  $\delta$  corresponds to the thickness (cm) of the film.

279 The porosity ( $\varepsilon$ ) was assessed based on the previously reported gravimetric method by Liang  
 280 *et al.*<sup>32</sup> using the following equation (8) given as:

281

$$282 \quad \varepsilon (\%) = \left( 1 - \frac{\rho_s}{\rho_c} \right) \times 100 \quad (8)$$

283

284 where  $\rho_s$  indicate the density of the film (g/cm<sup>3</sup>) and  $\rho_c$  is the density of the cellulose ( $\rho_c =$   
 285 1.5 g/cm<sup>3</sup>;(Yang *et al.*<sup>33</sup>) respectively.

286

#### 287 2.4.9 Mechanical properties



288 The mechanical properties (Tensile strength, Elongation at break, and Tensile modulus) were  
 289 tested using a Universal Testing Machine (Zwick Roell, Germany, Static UTM Z010). Film  
 290 strips were cut into 55 mm × 10 mm in length and width with a tensile rate of 5 mm/min at  
 291 room temperature.

292

#### 293 **2.4.10 Water Vapour Transmission Rate (WVTR)**

294 Water vapour transmission rate (WVTR) was measured using a gravimetric method according  
 295 to the modified method of the ASTM E96/E96 M-16, as described by Souza *et al.*<sup>34</sup> Initially,  
 296 films were cut into circular size and sealed onto a glass vial with a diameter of 6.18 mm  
 297 containing anhydrous calcium chloride desiccant. These sealed vials were kept in a desiccator  
 298 filled with a saturated NaCl solution (maintained at 25 °C and 75% relative humidity) at the  
 299 bottom part. The desiccator was then sealed properly to avoid the passage of air inside the  
 300 desiccator. The weight of each vial was recorded at the start (time zero) and subsequently every  
 301 24 h for 3 days. Water vapor permeated through the films and was absorbed by the desiccant;  
 302 evaluations were determined by measuring the weight gain per unit area. The water vapor  
 303 transmission rate (WVTR) was calculated using equation (9).

$$304 \quad WVTR = \frac{G}{tA} \quad (9)$$

305

306 where  $G$  is the weight change in g,  $t$  is the time in h, and  $A$  is the effective area of the film in m<sup>2</sup>.

307

308 To determine Water Vapour Permeability (WVP), in g Pa<sup>-1</sup> m<sup>-1</sup> s<sup>-1</sup>, the following equation (10)  
 309 can be employed:

310

$$311 \quad WVP = \frac{WVTR \times L}{\Delta P} \quad (10)$$

312

313 where  $L$  is the thickness of the film in meters, and  $\Delta P$  is the partial pressure difference of water  
 314 vapor across the film in Pascals.

315

#### 316 **2.4.11 Oxygen Transmission Rate**

317 The Oxygen Transmission rate (OTR) of the control film and ACNC films was determined  
 318 with an automated Oxygen Transmission Rate Test System (PERME® OX2/230 Labthink,  
 319 China). According to ASTM D3985, the measurements were carried out using high-purity  
 320 oxygen (99.999%) at 38 °C and 90% relative humidity. Each sample had a test area of 65 cm<sup>2</sup>.

321

#### 322 **1.4.12 Color properties and Optical transmittance**

323 The color properties ( $L^*$ ,  $a^*$ ,  $b^*$ ) of the developed films were measured using a Chroma meter,  
 324 CR-400 (Konica Minolta Optics, Inc., Japan) under CIE 1931 2° standard observer with  
 325 illuminant C/D65 conditions. The  $L^*$  (lightness/brightness),  $a^*$  (redness/greenness), and  $b^*$   
 326 (yellowness/blueness) were recorded using this device. The chroma ( $C^*$ ) and hue angle ( $H^\circ$ )  
 327 values were calculated according to Equations (11 & 12), as mentioned by Sganzerla *et al.*<sup>35</sup>:

$$328 \quad C^* = \sqrt{a^{*2} + b^{*2}} \quad (11)$$

$$329 \quad H^\circ = \tan^{-1} \frac{b^*}{a^*} \quad (12)$$



330 The total color difference ( $\Delta E_{ab}$ ) of the films relative to the control (taken as control) was  
 331 calculated using Eq. (13). View Article Online  
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$$332 \quad \Delta E_{ab} = \sqrt{(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2} \quad (13)$$

333 Where,  $\Delta L^* = L^*_{\text{standard}} - L^*_{\text{sample}}$ ,  $\Delta a^* = a^*_{\text{standard}} - a^*_{\text{sample}}$ ,  $\Delta b^* = b^*_{\text{standard}} - b^*_{\text{sample}}$

334  
 335 The optical transmittance (Tr) of ACNC films was determined with a double-beam UV-Vis  
 336 spectrophotometer (UV-Vis Spectrophotometer, LMSP-UV1900S, Labman Scientific  
 337 Instruments Pvt. Ltd., Chennai) at the wavelength of 800 nm, as per the procedure explained  
 338 by Zhao *et al.*<sup>36</sup> The average thickness of the composite films was about 40  $\mu\text{m}$ . The  
 339 transmission spectra were acquired using air as a blank.

#### 341 2.4.13 Thermal gravimetric analysis (TGA)

342 The thermal stability and thermodegradation behavior were analyzed using a thermal  
 343 gravimetric analyzer (TGA 2 STARE system, Mettler Toledo, Switzerland). The sample of ~5  
 344 mg or less was placed in the crucible and heated in the temperature range of 30 to 600  $^{\circ}\text{C}$  at a  
 345 heating rate of 10 $^{\circ}\text{C}/\text{min}$  under a nitrogen atmosphere. Thermogravimetric analysis (TGA)  
 346 provided the information about the overall weight loss as a function of time/temperature, while  
 347 Derivative Thermogravimetry (DTG) represented the first derivative curve, which showed the  
 348 rate of mass change with respect to temperature and time.

#### 349 2.4.14 Biodegradability

350 The biodegradability of all the films was assessed using the technique explained by Kim *et al.*<sup>37</sup>  
 351 Films were cut into 3  $\times$  3  $\text{cm}^2$  specimens, weighed, and immersed in reagent bottles containing  
 352 a cellulose solution (3 mg/mL in PBS, pH 6.0). The reagent bottles were maintained in an  
 353 incubator at 37  $^{\circ}\text{C}$  and monitored for 75 days. Additionally, ACNC films were tested in distilled  
 354 water without the cellulase enzyme as a negative control. At 15-day intervals, films were  
 355 removed from cellulase solution, washed with DI water to remove residual enzyme, dried in a  
 356 hot air oven at 40  $^{\circ}\text{C}$ , and weighed. The dry weight was recorded, and the percentage weight  
 357 loss was calculated using equation (14) given below:

$$359 \quad \text{Weight loss (\%)} = \frac{W_0 - W_1}{W_0} \times 100 \quad (14)$$

360  
 361 where  $W_0$  represents the initial weight of a specimen, and  $W_1$  is the weight of the specimen on  
 362 a given day of sampling.

#### 364 2.5 Statistical analysis

365 Experiments were conducted in triplicate, and values are presented as mean  $\pm$  standard  
 366 deviation (SD). Data was analysed by one-way analysis of variance (ANOVA) under a  
 367 completely randomized design (CRD). The sample mean values were compared using Fisher's  
 368 least significant difference (LSD) test at a significant level of 0.05 ( $p \leq 0.05$ ). The obtained  
 369 instrumental data were interpreted and analysed using Origin Pro SR1 software.

### 371 3. Results and discussion



### 3.1 Chemical composition and yield of cellulose

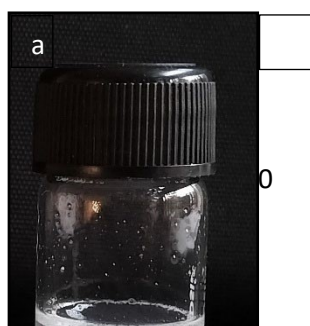
The cell wall of natural fibers is composed of reinforced cellulose microfibrils embedded in the amorphous matrix made up of hemicellulose and lignin. The complex intermolecular interactions and structural arrangement of cellulose, hemicellulose, and lignin provide the natural recalcitrance of lignocellulosic biomass. To overcome this recalcitrance and isolate cellulose fibers, chemical treatment, i.e., alkaline treatment (12% NaOH) followed by bleaching (10 % acidified NaClO<sub>2</sub>), was performed according to the protocol described in our earlier work, Jadaun *et al.*<sup>22</sup> As has already been reported in that communication, the hemicellulose and lignin in rice straw were 19.33±0.03% and 10.03±0.05%, respectively. After alkali treatment, the contents of hemicellulose and lignin were reduced to 6.26±0.01% and 2.23±0.02%, respectively, whereas cellulose percentage in the residue was increased from 35.03±0.15% to 78.70±0.1%. This enhancement reflected that alkali treatment partially disrupted the complex matrix through the cleavage of ester linkages in lignin-hemicellulose, which led to the dissolution of hemicelluloses. Consequently, hemicellulose and lignin were partially removed, and a further bleaching process eliminated insoluble lignin along with residual hemicellulose from the rice straw pulp, which enhanced cellulose content to 92.08±0.1%. It indicated that the combined treatments efficiently removed non-cellulosic components, resulting in the extraction of purified cellulose fibers, with cellulose yield approximately 42%, which was found to be comparable to the values reported in the literature for garlic stalk, corncob, and giant cane cut-up.<sup>38</sup>

### 3.2 Sedimentation stability and Characterization of CNF

Fig. 1 illustrates the sedimentation behaviour of the 1% (w/w) CNF suspension after high shear dispersion (HSD), and after combined treatment of high intensity ultrasonication (HIU) and HSD. After one hour of HSD, the suspension was observed to be partially translucent in appearance, with noticeable sedimentation at the bottom of the vial (Fig. 1a). When HSD was followed by HIU, the cellulose suspension became homogenous, stable, and exhibited slight opalescence rather than full transparency (Fig. 1b). Wang *et al.*<sup>39</sup> reported high transparency for low (0.1%) concentration levels of cellulose, while at high concentration levels, the suspension appeared to be turbid because of the increased fiber content. The effectiveness of the combined treatment could be due to the effect of HSD that partially disrupted the hydrogen bonding among the microfibrils and loosened the amorphous region, while HIU treatment resulted in nano-fibrillation by cavitation effect. The nanofibrillation is also evident in the FESEM image (Fig. 1c), where a web-like structure was observed with a nanoscale fibrillar network. It has already been communicated in our earlier communication<sup>22</sup> that the individual fibrils exhibited an average diameter of 34.2 nm and length of 1765.9 ± 42.1 nm, a mean zeta potential of -12 mV, and the maximum degradation temperature of 329 °C. The crystallinity index (CI) of CNFs was 80.5%, which was significantly improved as compared to cellulose after NaOH (CI: 50%) and NaClO<sub>2</sub> bleaching (CI: 68%) treatments. These reported results for optical appearance, nanoscale dimension, fiber morphology, and thermal stability confirmed that the prepared fibers were CNFs rather than MFC.

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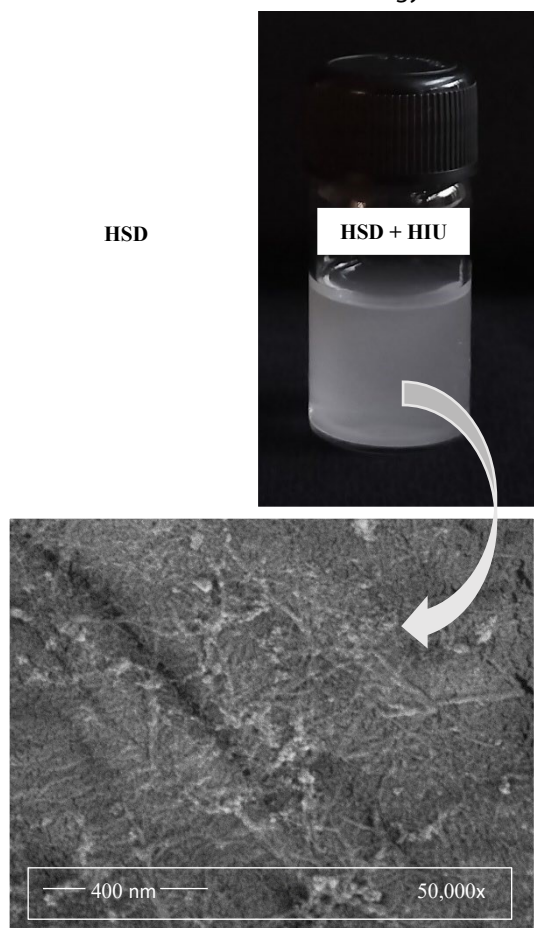
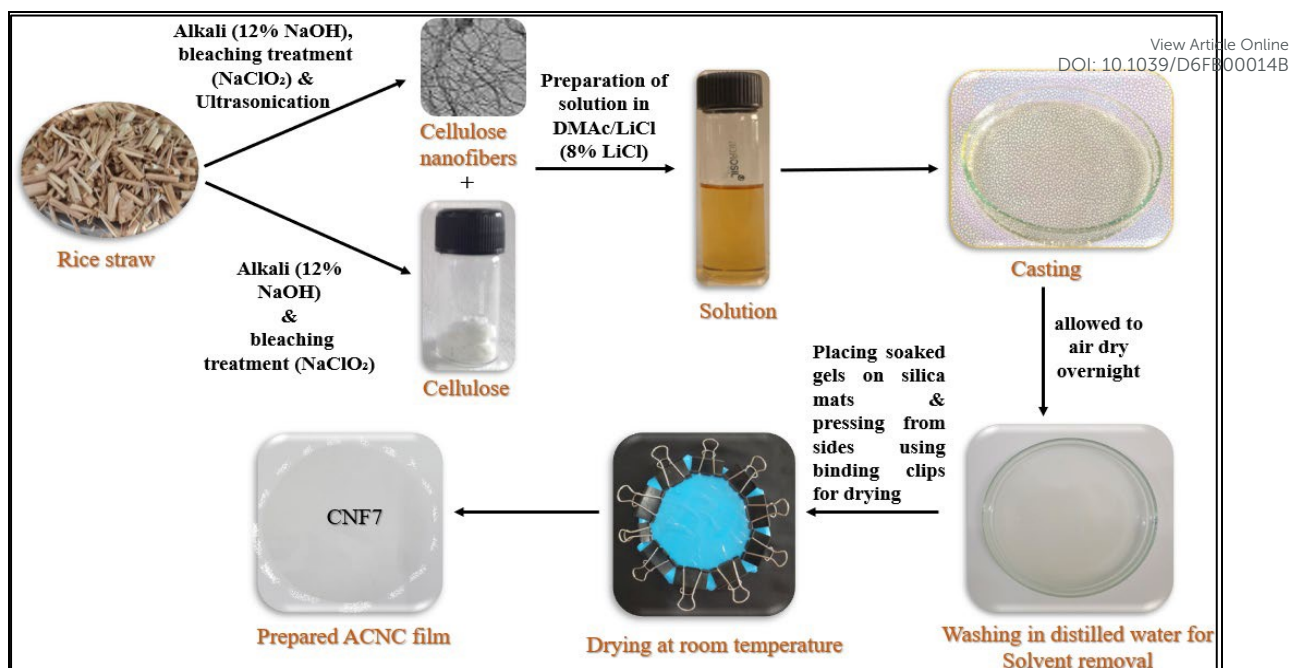


Fig. 1. (a) CNF dispersion after HSD, showing partial sedimentation, (b) CNF dispersion obtained after combined treatment of HSD and HIU, showing improved homogeneity and stability, and (c) FESEM-image of CNFs after HSD + HIU treatment, revealing a nanofibrous surface morphology of fibers observed at a magnification of 50,000x

### 3.3 Homogeneity and visual appearance of the films

After the isolation of CNF, uniform and transparent CNF films were successfully fabricated via the DMAc/LiCl dissolution-regeneration method. Fig. 2 illustrates the overall procedure of nanocomposite film preparation from rice straw. The ACNC films developed through solution





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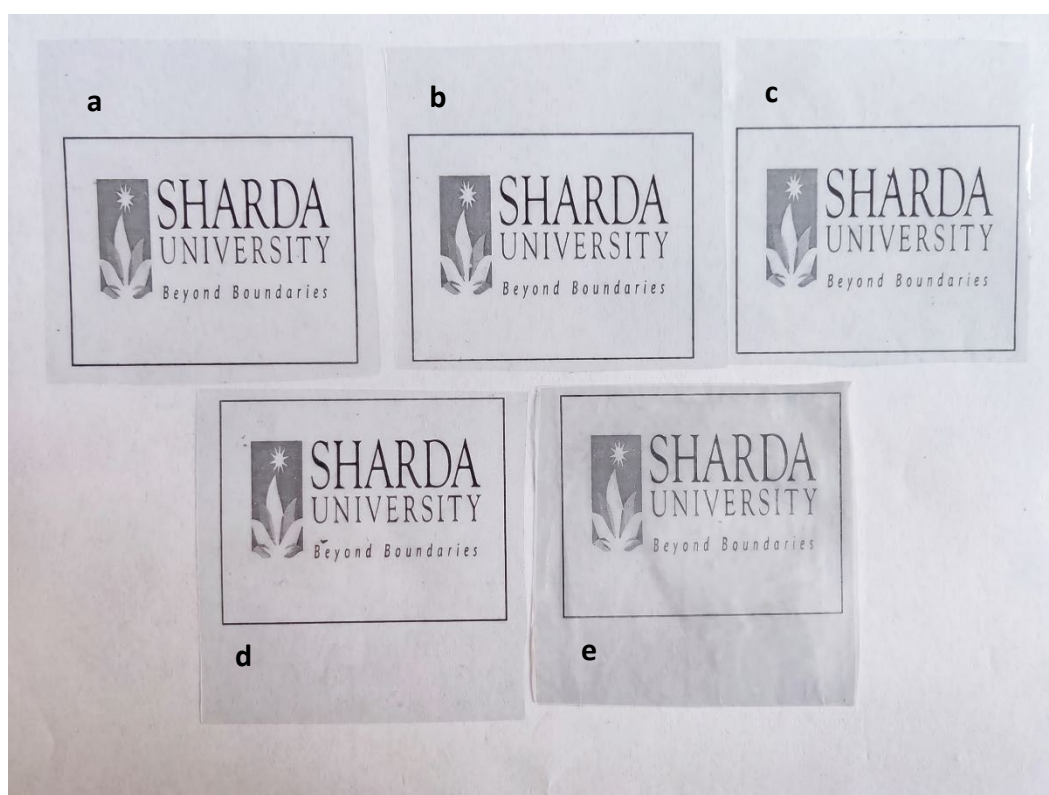
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Fig. 2. An overview of the preparation of nanocomposite films from rice straw. The extracted cellulose and CNFs from rice straw were introduced in DMAc/LiCl to achieve cellulose dissolution and CNF dispersion, followed by casting. The prepared gels were washed and dried at room temperature to obtain the ACNC films.



444

445 Fig. 3. Digital images of ACNC films (a) CNF0, (b) CNF3, (c) CNF5, (d) CNF7, and (e)

446

CNF9 showing clarity of the text and images through the film samples.



447 casting, as shown in Fig. 3, illustrate the visual appearance of the films at different CNF  
 448 concentrations. All films were uniform and smooth throughout, as can be observed from these  
 449 images, except the CNF9 film, which showed relatively decreased transparency and rigidity  
 450 due to the presence of agglomerated CNFs as observed by FE-SEM analysis (see Fig. 4). These  
 451 visual observations aligned with the UV-Vis results, which quantitatively demonstrated the  
 452 transparency differences. However, the solution appeared yellowish in color (as shown in  
 453 casting steps in Fig. 2) when first spread onto the mold, but became transparent after solvent  
 454 removal. The control film (CNF0), containing only cellulose, was flexible, whereas adding  
 455 CNF significantly improved the mechanical strength of the films. The ACNCs could be rolled  
 456 and folded without breaking and remained stable without flaking at room temperature. No  
 457 visible traces of undissolved CNF were seen on the surface (consistent with FE-SEM results),  
 458 suggesting uniform dispersion within the matrix. After drying, all films could be easily peeled  
 459 from the casting frame without damage.

### 3.4 Viscosity average degree of polymerisation and molecular weight

461 The molecular weight ( $M_w$ ) of cellulose affects the properties of cellulose-based materials (e.g.,  
 462 films), in terms of mechanical properties and crystallisation behaviour.<sup>40</sup> It forms the basis for  
 463 understanding the structure-property relationship of cellulose-based materials, as  $M_w$  reflects  
 464 the average chain length of cellulose molecules, which defines the dissolution behaviour<sup>41</sup> and  
 465 chain entanglement<sup>42</sup> in LiCl/DMAc. The viscosity-average degree of polymerization ( $DP_v$ ) is  
 466 directly related to cellulose chain length and is measured from intrinsic viscosity measurements  
 467 that reflect the hydrodynamic dimensions of cellulose chains in solution. The intrinsic viscosity  
 468 ( $\eta$ ), molecular weight ( $M_w$ ), and viscosity average degree of polymerisation ( $DP_v$ ) of bleached  
 469 cellulose fibers, CNF, and the cellulose-CNF solutions are summarized in Table 1. However, it  
 470 was not possible to determine the viscosity-average degree of polymerization of the cellulose  
 471 after NaOH pulping because the pulp after alkali treatment was not purified cellulose and could  
 472 not be dissolved in DMAc/LiCl due to the presence of lignin.

474 **Table 1. Polymer characteristics of treated fibers**

| Sample                                 | $[\eta]$ ( $\text{mL}\cdot\text{g}^{-1}$ ) | $M_w$ ( $\times 10^5$<br>$\text{g}\cdot\text{mol}^{-1}$ ) | $DP_v$ ( $\times 10^3$ ) |
|--|--|---|--------------------------|
| Bleached Cellulose fibers (rice straw) | $1023 \pm 3.0^a$                           | $6.32 \pm 0.02^a$   | $3.90 \pm 0.01^a$        |
| CNF                                    | $958 \pm 2.6^c$                            | $5.98 \pm 0.03^c$   | $3.69 \pm 0.01^c$        |
| Cellulose + CNF (3% CNF)               | $994 \pm 1.0^d$                            | $6.18 \pm 0.02^b$   | $3.81 \pm 0.03^b$        |
| Cellulose + CNF (5% CNF)               | $997 \pm 2.0^c$                            | $6.19 \pm 0.03^b$   | $3.82 \pm 0.02^b$        |
| Cellulose + CNF (7% CNF)               | $998 \pm 2.0^c$                            | $6.19 \pm 0.01^b$   | $3.82 \pm 0.03^b$        |
| Cellulose + CNF (9% CNF)               | $1002 \pm 1.7^b$                           | $6.21 \pm 0.02^b$   | $3.83 \pm 0.01^b$        |

475  $\eta$ -intrinsic viscosity,  $M_w$  -Molecular weight, and  $DP_v$ -Viscosity degree of polymerisation.  
 476 Mean values with different superscript letters in the same column are significantly different  
 477 ( $p \leq 0.05$ ).

479 The intrinsic viscosity of the bleached cellulose fibers was  $1023 \pm 3.0 \text{ mL}\cdot\text{g}^{-1}$ , which provided  
 480  $M_w$  of  $6.32 \times 10^5 \text{ g}\cdot\text{mol}^{-1}$  and  $DP_v$  of  $3.90 \times 10^3$ . The intrinsic viscosity of the CNF was  
 481 significantly decreased to  $958 \pm 2.6 \text{ mL}\cdot\text{g}^{-1}$  ( $M_w = 5.98 \times 10^5 \text{ g}\cdot\text{mol}^{-1}$ ;  $DP_v = 3.69 \times 10^3$ ) after  
 482 mechanical fibrillation. The observation implied that ultrasonication caused a decrease in the  
 483 length of cellulose fibres due to chain scission by cleavage of the glycosidic bonds. A similar



484 reduction in degree of polymerisation after ultrasonic treatment or homogenisation has been  
485 reported in previous studies by Dilamian and Noroozi<sup>43</sup>, and Du *et al.*<sup>44</sup>. However, when CNFs  
486 were mixed with cellulose in LiCl/DMAc, the intrinsic viscosity values were found to be  
487 increased relative to CNF alone ( $994 \pm 1.0$ - $1002 \pm 1.7 \text{ mL}\cdot\text{g}^{-1}$ ), which corresponded to  $M_w$   
488 values of  $6.18$ - $6.21 \times 10^5 \text{ g}\cdot\text{mol}^{-1}$  and  $DP_v$  values of  $3.81$ - $3.83 \times 10^3$ . The absolute  $M_w$  values  
489 depend on the solvent-specific Mark-Houwink constant ( $K, \alpha$ ). The  $M_w$  values with  
490 LiCl/DMAc system suggested that cellulose did not undergo significant degradation during the  
491 dissolution process, as the solvent system is known to solubilize high molecular weight  
492 cellulose without degradation.<sup>45,46</sup> The relatively high  $M_w$  and  $DP_v$  values in the cellulose-CNF  
493 solutions demonstrated that the polymer chains remained sufficiently long to contribute to the  
494 entanglement-driven mechanical reinforcement, and to allow strong chain entanglement and  
495 network formation in the regenerated films. The slight decrease in  $M_w$  in CNF could be due to  
496 defibrillation that occurred during ultrasonication. The high molecular weight of the cellulose  
497 chains supported effective chain entanglement during regeneration, while increased CNF  
498 concentration introduced additional structural interactions within the cellulose matrix that may  
499 have resulted in improved mechanical performance in the ACNC films.

500

### 501 3.5 FE-SEM Analysis

502 Fig. 4 (a-e) presents the morphology of pure cellulose film (CNF0) and CNF reinforced  
503 nanocomposite films. The CNF0 (as shown in Fig. 4(a)) exhibited a smooth, dense, and uniform  
504 surface without visible defects. A similar successful film formation has been reported by  
505 Moreira *et al.*,<sup>47</sup> who attributed it to the strong intermolecular cohesion among cellulose chains.  
506 Incorporation of CNF did not significantly alter the surface morphology of the nanocomposite  
507 films (ACNCs) up to 7% CNF loading.

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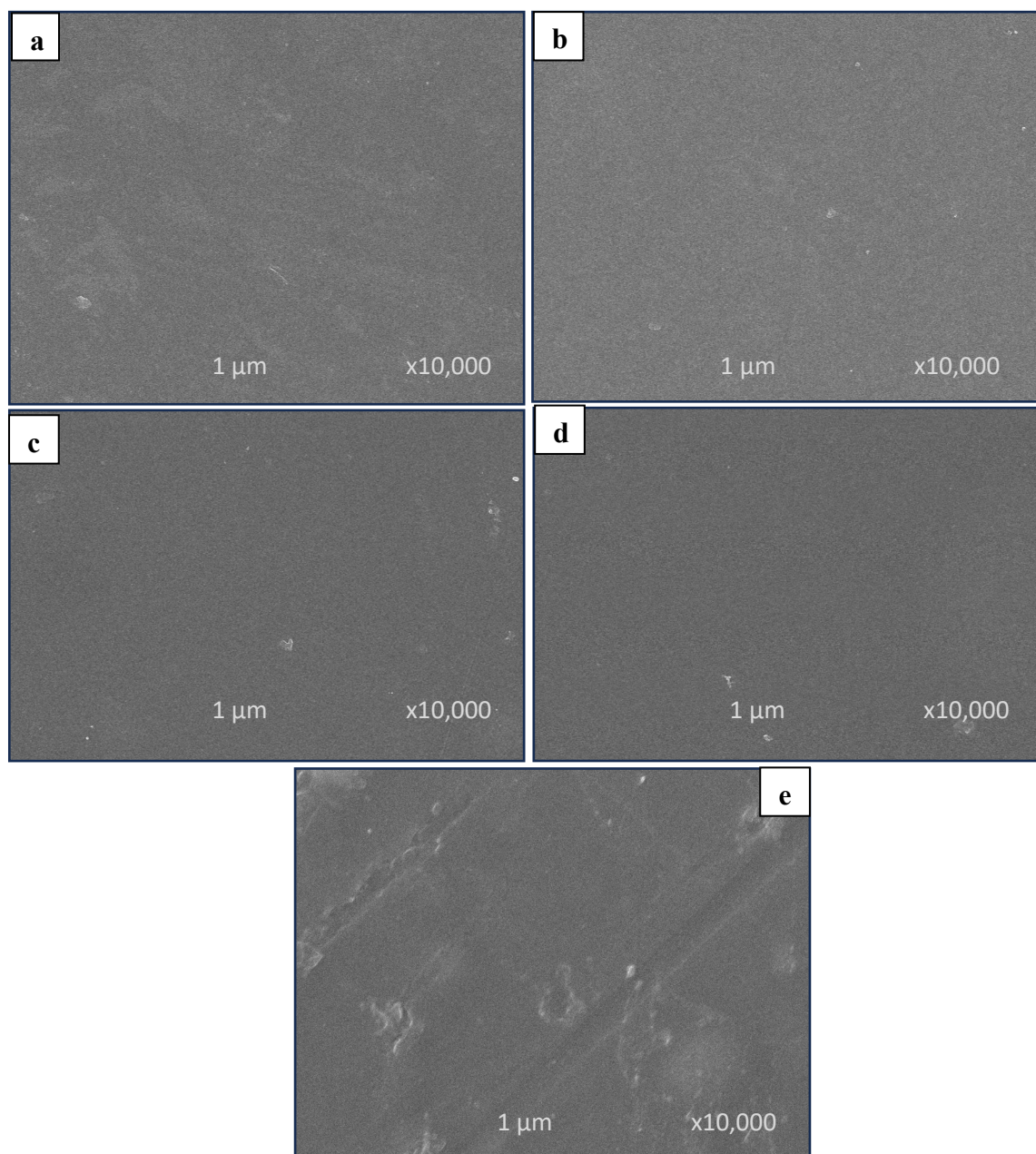
509 However, at 9% CNF (Fig. 4e), surface roughness became evident, likely caused by CNF  
510 agglomeration within the cellulose matrix. This behaviour suggested that higher CNF  
511 concentrations promoted aggregation due to strong intermolecular interactions, leading to non-  
512 homogeneous dispersion and potential deterioration of film properties.<sup>48</sup> Comparable findings  
513 were reported by Shazleen *et al.*<sup>49</sup> for CNF-reinforced PLA (Polylactic acid) nanocomposite  
514 films, where high CNF content impeded polymer chain mobility.

515 All films except CNF9 displayed compact, pore-free surfaces with no detectable surface  
516 irregularities, suggesting uniform dispersion and effective incorporation of CNFs into the  
517 matrix. The internal structure appeared dense and cohesive, without any visible traces of the  
518 original fibrous morphology of the nanofibers, suggesting that CNFs were well dispersed in  
519 the cellulose matrix during processing and subsequently underwent structural reorganization  
520 upon heating of the cellulose-CNF solution for the development of ACNCs. The absence of  
521 discrete CNF regions further confirmed that the films behaved as molecularly integrated  
522 systems rather than conventional fiber-reinforced composites, as evident in Fig. 4 (b, c, & d).  
523 Interestingly, the higher-magnification FE-SEM micrograph of the CNF7 film (Fig. 4e)  
524 revealed a finely textured surface, indicative of nanoscale features potentially associated with  
525 CNF residues. Overall, increasing the CNF concentration up to the optimum level of 7%  
526 preserved the compact morphology while enhancing internal structuring and surface  
527 uniformity.<sup>50</sup> This molecular-level reinforcement is likely responsible for the improved  
528 structural integrity, and uniformity observed in the nanocomposite films.



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548 Fig. 4. FE-SEM images x 10,000 of CNF0 and ACNC films (a) CNF0, (b) CNF3, (c) CNF5,  
549 (d) CNF7, and (e) CNF9, showing smooth and dense surfaces of film samples from CNF0 to  
550 CNF7, while agglomeration was observed in CNF9.

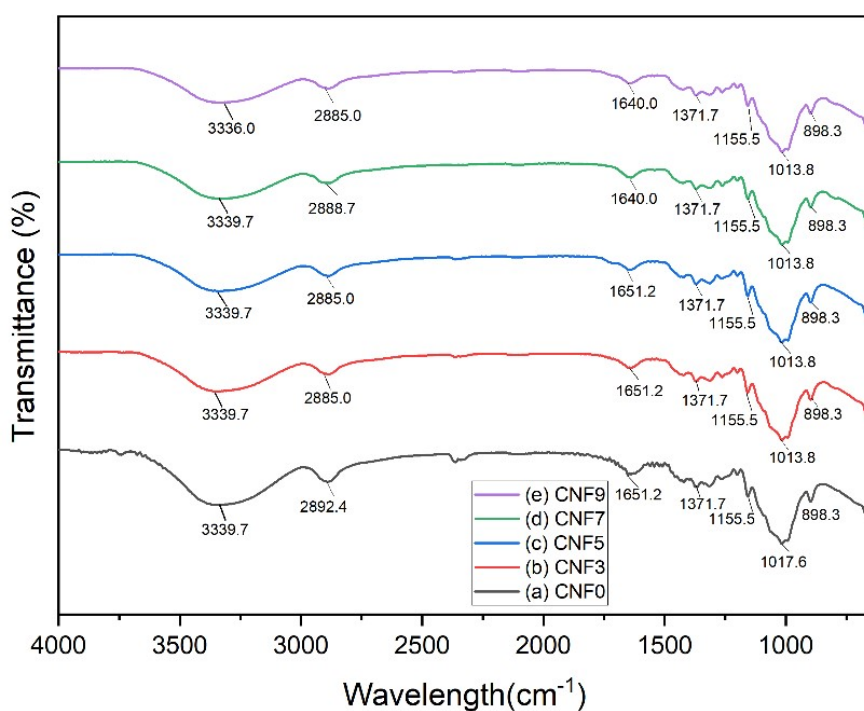
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### 552 3.6 Fourier transform infrared spectroscopy (FTIR)

553 FTIR analysis was performed to evaluate the influence of CNF concentration and the solution  
554 casting method on film structure (Fig. 5). In all samples, the broad dominant peaks observed  
555 around  $3300\text{ cm}^{-1}$  corresponded to OH group stretching vibrations, indicating the presence of  
556 intra- and intermolecular hydrogen bonding within the cellulose structure.<sup>51</sup> Additionally, in  
557 the region of  $2880 - 2900\text{ cm}^{-1}$ , C-H stretching vibrations were observed, corresponding to the  
558 primary functional groups of cellulose.



559 The intensity of the -OH band increased with CNF addition, indicating enhanced hydrogen  
 560 bonding between the cellulose matrix and CNF. The region between  $1629\text{ cm}^{-1}$  -  $1640.0\text{ cm}^{-1}$   
 561 was ascribed to O-H bending vibrations of water molecules absorbed by the cellulose structure.  
 562 The  $\text{CH}_2$  bending vibrations of the pyranose ring were observed at  $1423.8\text{ cm}^{-1}$  and  $1428.06\text{ cm}^{-1}$   
 563  $\text{cm}^{-1}$ , reflecting the crystalline structure of cellulose.<sup>52</sup> The majority of distinctive bonds in the  
 564 fingerprint region ( $800 - 1500\text{ cm}^{-1}$ ) of cellulose remain consistent in all spectra.<sup>53</sup> The C-O-C  
 565 stretching vibration detected at peak  $1155.5\text{ cm}^{-1}$  was found to be constant in all film samples.<sup>54</sup>  
 566 Additionally, the CH deformation vibration was slightly shown at the peak of  $1371.7\text{ cm}^{-1}$  in  
 567 the spectra.<sup>55</sup> The prominent peaks at  $1013.8\text{ cm}^{-1}$  and  $1017.6\text{ cm}^{-1}$  were related to C-O  
 568 vibration stretching in cellulose. The tiny peaks seen at  $895.41\text{ cm}^{-1}$  and  $898.3\text{ cm}^{-1}$  depicted  
 569 the C-O-C stretching vibration of the  $\beta$ -glycosidic link in cellulose. This indicated that  
 570 regardless of different processing treatments, the fundamental structure of cellulose remained  
 571 intact throughout, and an increase in intensity validated the existence of intermolecular  
 572 interaction and compatibility.



573

574 Fig. 5. FTIR spectra of (a) CNF0, (b) CNF3, (c) CNF5, (d) CNF7, and (e) CNF9 at different  
 575 CNF concentrations.

576

### 577 3.7 Moisture content

578 Moisture content strongly influences the mechanical and barrier performance behaviour,  
 579 particularly in hydrophilic polymers. The moisture of cellulose film without reinforcement was  
 580 estimated to be  $15.78 \pm 0.08\%$  (Table 2) and was significantly higher than ACNC films  
 581 ( $p \leq 0.05$ ). With increasing CNF concentration, the moisture content of ACNC decreased  
 582 markedly, with up to 63.6 % reduction in CNF7 relative to the control. However, the  
 583 agglomeration in CNF9, as is evident in FESEM image (Fig. 4e), likely resulted in slightly  
 584 higher moisture content in CNF9 as shown in Table 2. This decrease in moisture content with  
 585 an increase in CNF could likely be due to the restricted space available for moisture because



586 of the interaction between CNF and cellulose matrix<sup>56</sup>. Faisal et al.<sup>51</sup> also reported that the  
 587 moisture content in amylose films decreased by CNF incorporation due to the entanglement  
 588 behaviour of CNF. Deepa et al.<sup>57</sup> and Kesari et al.<sup>58</sup> found that the interaction of CNF and  
 589 polymer matrix led to a decreased number of active sites for water binding. The formation of a  
 590 dense fibrous CNF network further limited water absorption and led to reduced moisture  
 591 content.

592

593 **Table 2. Structural properties and crystallinity of the films**

| Samples | Thickness (mm)              | Moisture (%)              | Density (g/cm <sup>3</sup> ) | Porosity (%)              | CI (%) |
|---------|-----------------------------|---------------------------|------------------------------|---------------------------|--------|
| CNF0    | 0.033 ± 0.003 <sup>c</sup>  | 15.78 ± 0.08 <sup>a</sup> | 1.02 ± 0.03 <sup>d</sup>     | 31.82 ± 0.33 <sup>a</sup> | 39.90  |
| CNF3    | 0.038 ± 0.004 <sup>b</sup>  | 11.23 ± 0.03 <sup>b</sup> | 1.12 ± 0.01 <sup>c</sup>     | 27.60 ± 0.21 <sup>b</sup> | 51.02  |
| CNF5    | 0.041 ± 0.002 <sup>ab</sup> | 8.61 ± 0.08 <sup>c</sup>  | 1.20 ± 0.02 <sup>b</sup>     | 23.31 ± 0.23 <sup>c</sup> | 50.00  |
| CNF7    | 0.043 ± 0.002 <sup>ab</sup> | 5.75 ± 0.05 <sup>c</sup>  | 1.25 ± 0.01 <sup>a</sup>     | 18.40 ± 0.19 <sup>c</sup> | 61.50  |
| CNF9    | 0.046 ± 0.003 <sup>a</sup>  | 6.12 ± 0.02 <sup>d</sup>  | 1.21 ± 0.04 <sup>b</sup>     | 22.50 ± 0.17 <sup>d</sup> | 57.80  |

594 Values are represented as mean ± SD (n=3). Mean values with different superscript letters in  
 595 the same column are significantly different (p≤0.05).

596 **3.8 Thickness, Density, and Porosity**

597 Determination of film thickness is pivotal, as it influences the density and barrier properties of  
 598 the film. The measured thickness of the control and ACNC films ranged from 0.033± 0.003 to  
 599 0.046 ± 0.003 mm. There was a significant difference observed due to the incorporation of  
 600 CNF in the matrix. The significant increase in thickness from CNF0 to CNF9 can be attributed  
 601 to the higher CNF loading and enhanced intermolecular interactions caused by CNFs, which  
 602 led to higher viscosity of the film-forming solution during the casting process. The results of  
 603 thickness are similar to earlier findings reported for sugarcane bagasse-derived CNF-reinforced  
 604 starch films prepared using the solvent casting method by Ribeiro et al.<sup>59</sup> For a nanocomposite  
 605 film, density indicates the structural compactness, which depends on the arrangement of  
 606 nanofillers and the packing of polymer chains. As given in Table 2, the density significantly  
 607 increased with increasing CNF concentration, and ranged from 1.020 ± 0.03 to 1.21 ± 0.04  
 608 g/cm<sup>3</sup> among the film samples. Among all the samples, CNF0 had relatively low density, which  
 609 could likely be due to its less compact microstructure and presence of microvoids throughout  
 610 the film. Gashawtena *et al.*<sup>60</sup> reported that the presence of CNF reduced microvoids and formed  
 611 denser structures. The incorporated CNF imparted a noticeable impact in the matrix by  
 612 occupying the intermolecular spaces and strengthening H-bonding across the fibre network. In  
 613 the present investigation, these interactions within the cellulose matrix might have resulted in  
 614 the reduction of free space in the film, which led to a tightly packed structure with lower  
 615 porosity.

616 The porosity of the films declined significantly with the increase in the CNF concentration, as  
 617 presented in Table 2, with CNF7 showing a reduction of 42.17% compared to CNF0. This  
 618 decrement is probably linked to the creation of a tighter and more highly interconnected  
 619 microstructure owing to the hydrogen bonding and CNF network consolidation, which is  
 620 prevalent in films prepared with nanocellulose<sup>61</sup>. Wakabayashi *et al.*<sup>62</sup> observed that the higher  
 621 degree of fibrillation in nanocellulose films led to a significant decline in porosity, i.e.,  
 622 approximately 70% to 20%. The values of porosity obtained in the present study are relatively  
 623 lower than those reported for hot-pressed CNF films made through natural drying<sup>32</sup>, and for  
 624 the cellulose acetate/lignin-rich CNF-based nanocomposite film.<sup>63</sup>



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### 626 3.9 XRD (X-ray diffraction)

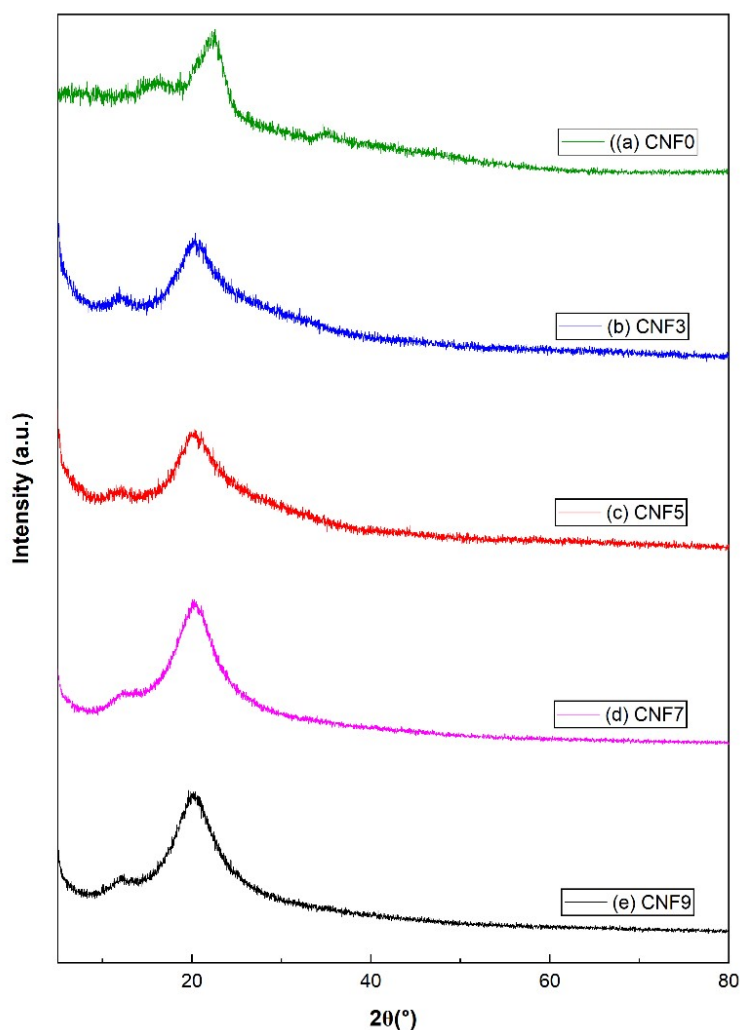
627 The XRD patterns of the fabricated ACNC films are presented in Fig. 6, and the corresponding  
628 crystallinity index (CI) values, determined using the Segal method, are summarized in Table 2.  
629 The dominant peaks at approximately  $2\theta = 14^\circ$  and  $20^\circ$  were observed in all ACNC samples,  
630 indicating a polymorphic transformation from cellulose I to cellulose II, although a minor  
631 fraction of cellulose I was also present. This indicated that dissolution/regeneration of cellulose  
632 in DMAc/LiCl led to alteration of the cellulose I crystal structure, consistent with the  
633 observations made by Zhang *et al.*<sup>64</sup>. Table 2 displays a considerable difference in crystallinity  
634 index between the film with reinforcement and without reinforcement. The crystallinity index  
635 further increased with CNF concentration, which indicated the strong interaction between  
636 CNFs and the cellulose matrix. With an increase in CNF concentration (3-7%), diffraction  
637 spectra were found to be refined at the peak (002). This indicated that a crystalline ordered  
638 structure was formed inside the matrix, revealing the role of nanofibers as nucleating for in  
639 nanocomposites.<sup>65,66</sup> This enhancement is reflected in the increased CI from 39.9% in CNF0 to  
640 a maximum of 61.5% in CNF7, demonstrating that the presence of nanofibers induced the rise  
641 of crystallinity in nanocomposite films.<sup>67</sup> However, for CNF9, a lower CI compared to CNF7  
642 indicated the beginning of agglomeration of nanofibers (revealed by FE-SEM images). This  
643 agglomeration may have led to a reduction in the internal order and resulted in lower  
644 crystallinity compared to CNF7. Similar CI values have been reported by Roy *et al.*<sup>68</sup> who  
645 demonstrated that nanocomposite films containing CNF/Zinc oxide, CNF/grapefruit seed  
646 extract, and CNF/Zinc oxide/grapefruit seed extract had CI values of 59.31%, 61.26%, and  
647 60.45%, respectively. In contrast, Lal *et al.*<sup>69</sup> reported significantly lower CI values ranging  
648 from 9.3% -22.5% for tamarind seed polysaccharide films incorporated with 0.1-1% old  
649 corrugated box-derived CNF, which was found to be lower than the CI obtained in this study.  
650 The improved crystallinity obtained in this study is likely to influence the reinforcing ability  
651 of the nanoparticles, which is expected to enhance the mechanical and barrier properties of the  
652 composite films.<sup>70</sup>

653

654 Since no intact nanofibrous morphology was observed in FE-SEM images, the reinforcing  
655 effect may not be directly inferred from the microstructure observations. However, CI showed  
656 a systematic increase in crystallinity with the addition of CNF, indicating that although CNFs  
657 lost their original fibrous morphology in the LiCl/DMAc system, they acted as effective  
658 nucleation and ordering sites during cellulose regeneration<sup>56</sup>. The presence of CNFs promoted  
659 recrystallization and the formation of ordered crystalline domains during the regeneration  
660 process, which led to structure-induced reinforcement in the regenerated films.

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Fig. 6. XRD spectra of CNF0, CNF3, CNF5, CNF7, and CNF9 films, depicting the effect of increasing CNF concentration on the crystalline structure.

### 666 3.10 Mechanical properties

667 Table 3 presents the mechanical properties of CNF0 and CNF reinforced nanocomposite films.  
668 The mechanical properties of the films are further illustrated in Fig. 7 (a, b). Incorporation of  
669 CNFs into the cellulose matrix significantly increased both tensile strength and tensile  
670 modulus, while reducing elongation at break, consistent with previous studies. For instance,  
671 Faisal *et al.*<sup>51</sup> reported a similar increase in mechanical properties due to CNF incorporation in  
672 amylose-based films. In another study by Amri *et al.*<sup>71</sup>, the incorporation of CNF improved  
673 the mechanical performance of the Jatropha Oil-Based Waterborne Polyurethane Film, with  
674 0.5% CNF presence leading to maximum enhancement in tensile strength and Young's  
675 modulus.

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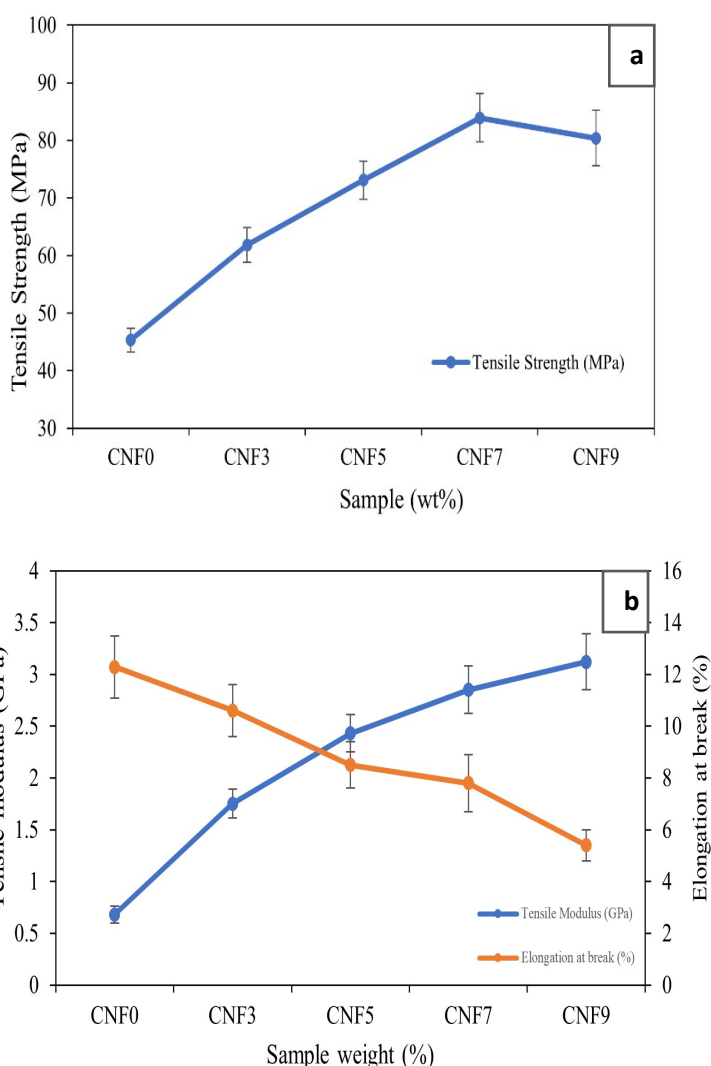


683 **Table 3. WVTR, OTR, and mechanical properties of CNF0 and the developed ACNC**  
 684 **films**

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| Samples | WVTR<br>(g/m <sup>2</sup> /day) | OTR<br>(cm <sup>3</sup> /m <sup>2</sup> /day) | Tensile<br>Strength<br>(MPa) | Elongation at<br>Break<br>(%) | Tensile<br>Modulus<br>(GPa) |
|---------|---------------------------------|---|------------------------------|-------------------------------|-----------------------------|
| CNF0    | 180.4±4.8 <sup>a</sup>          | 52.7±2.2 <sup>a</sup>                         | 45.3 ± 2.1 <sup>e</sup>      | 12.3 ± 1.2 <sup>a</sup>       | 0.68 ± 0.08 <sup>e</sup>    |
| CNF3    | 129.8 ±5.2 <sup>b</sup>         | 43.8±2.4 <sup>b</sup>                         | 61.8 ± 2.0 <sup>d</sup>      | 10.6 ± 1.0 <sup>a</sup>       | 1.75 ± 0.14 <sup>d</sup>    |
| CNF5    | 85.3 ±4.1 <sup>c</sup>          | 24.7±1.7 <sup>c</sup>                         | 73.1 ± 2.3 <sup>c</sup>      | 8.5 ± 0.9 <sup>b</sup>        | 2.43 ± 0.18 <sup>c</sup>    |
| CNF7    | 44.7±3.0 <sup>e</sup>           | 5.6± 0.8 <sup>e</sup>                         | 83.9 ± 1.5 <sup>a</sup>      | 7.8 ± 1.1 <sup>b</sup>        | 2.85 ± 0.15 <sup>b</sup>    |
| CNF9    | 52.4 ±3.2 <sup>d</sup>          | 8.8± 0.1 <sup>d</sup>                         | 80.4 ± 1.2 <sup>b</sup>      | 5.4 ± 0.6 <sup>c</sup>        | 3.12 ± 0.13 <sup>a</sup>    |

685 The values are given as the mean of three replicates ± SD (n=3). Mean values with different  
 686 superscript letters in the same column are significantly different (p≤0.05).  
 687



713 Fig. 7. (a) Tensile strength, and (b) Tensile modulus and elongation at break of CNF0 and the  
 714 developed ACNC films, showing that increasing CNF concentrations led to higher tensile  
 715 strength and tensile modulus, while elongation at break was decreased.

716

717 As per the data given in Table 3, the tensile strength increased from  $45.3 \pm 2.1$  MPa for CNF0  
 718 to  $83.9 \pm 1.5$  MPa for CNF7, which represented a statistically significant enhancement ( $p \leq$   
 719  $0.05$ ). Similar enhancements have been reported in previous studies by Kesari *et al.*<sup>58</sup>; González



720 *et al.*<sup>72</sup>, and Bian *et al.*<sup>15</sup>. It was attributed to strong interfacial interactions between the  
721 nanofibrils and the cellulose matrix, facilitated by their similar chemical composition, which  
722 effectively reinforced the film structure.<sup>73</sup>

723 As illustrated in the Fig. 9 (b), the reinforcing effect of CNFs arises from the formation of the  
724 entangled fibrous network within the cellulose matrix due to strong hydrogen bonding. As CNF  
725 concentration increases, these hydrogen-bonding interactions were enhanced, enabling an  
726 efficient stress transfer from the matrix to the nanofibers, thereby improving the tensile strength  
727 of the films.<sup>74, 75</sup> Shahi *et al.*<sup>6</sup> reported that improvement in tensile properties was due to the  
728 nanosize of CNF and homogenous dispersion, resulting in a compact and interconnected  
729 network in the matrix. This observation aligns with the present study, where CNF7 was found  
730 to be tougher than the CNF0 film. However, further increase in CNF concentration above 7 %  
731 led to a significant decrease in tensile strength, which could be the result of agglomeration (FE-  
732 SEM image Fig. 4e) and poor dispersion of CNF in the cellulose matrix.<sup>50, 68</sup>

733 The tensile modulus of CNF0 was measured to be  $0.68 \pm 0.08$  GPa (Table 3), and the  
734 incorporation of CNFs significantly increased it to  $3.12 \pm 0.13$  GPa in CNF9. The tensile  
735 modulus of the films was also affected by film density and moisture content (measured at 23  
736 °C/50 % RH). With the increase in CNF content from CNF3 to CNF7, film density and modulus  
737 significantly increased, indicating that efficient nanofiber network formation within the matrix  
738 was established, which led to better support for the transfer of stress. The lower moisture  
739 content contributed to the higher stiffness of the films by reducing the mobility of the fibrils.  
740 However, in CNF9, density slightly decreased while moisture content increased because of  
741 fibril agglomeration that led to the formation of dense local bundles, and at the same time  
742 generated microvoids that promoted water absorption. Despite the higher moisture content in  
743 CNF9, tensile modulus remained high, as the stiff agglomerated fiber clusters dominated the  
744 mechanical response.

745 However, the elongation at break decreased with the nanofiber addition, with the most  
746 pronounced reduction observed in CNF9. At higher CNF concentrations, the increased fiber  
747 density likely reached a saturation point, restricting polymer chain mobility. A similar reduction  
748 in elongation at break with increasing CNF content was also observed by Wang *et al.*<sup>76</sup>, who  
749 isolated CNF from jute fibers for reinforcement in PVA film and attributed the decrease to the  
750 formation of a rigid network and increased stiffness that restricts chain mobility, thereby  
751 lowering film flexibility.<sup>77</sup> These results demonstrated that the incorporation of CNFs into  
752 cellulose nanocomposite films effectively improved their tensile strength and modulus;  
753 however, maintaining an optimal concentration is essential to prevent aggregation and resulting  
754 loss of flexibility.

755

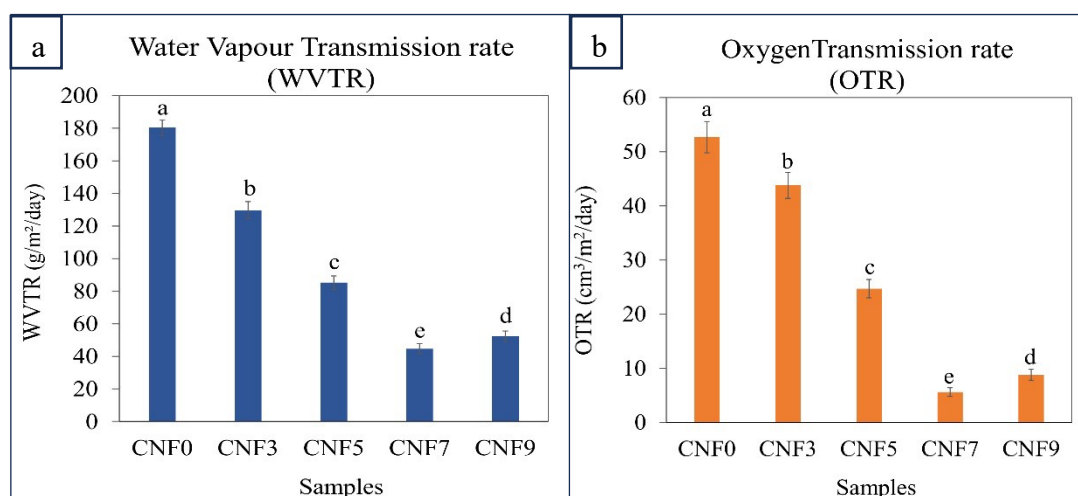
### 756 3.11 Water Vapor Transmission Rate

757 Water vapor transmission rate (WVTR) is one of the most fundamental indicators of the barrier  
758 characteristics of packaging films, with lower values indicating greater water vapor resistance,  
759 which directly influences the texture, quality, and shelf life of food products.<sup>78</sup> Excess moisture  
760 in the films promotes microbial growth and accelerated food spoilage. The challenges,  
761 particularly concerning insufficient moisture barrier, significantly limit the usage of presently  
762 accessible biodegradable polymers in food packaging.<sup>79</sup> Incorporation of CNFs into the  
763 polymer matrix reduces WVTR<sup>10, 80</sup>, likely by creating a physical barrier that hinders water  
764 vapor diffusion along the polymer–nanofiber interface. As shown in Fig. 8 (a), CNF0 exhibited  
765 a WVTR of  $180.4 \pm 4.8$  g/m<sup>2</sup>/day, which decreased significantly to  $44.7 \pm 3.0$  g/m<sup>2</sup>/day in  
766 CNF7, representing a 75.2% reduction. The WVTR values were observed to be significantly



767 different for all the films, indicating that the incorporation of CNFs progressively improved the  
 768 water vapor barrier properties of films. The results are consistent with the findings of Rincon  
 769 *et al.*<sup>81</sup> where CNFs were obtained from horticultural residues (mixture of bell pepper, tomato,  
 770 and eggplant) and vine shoots, and reinforced in CMC films, which resulted in a 20-30%  
 771 reduction in WVP compared to a neat CMC film. Guzman-Puyol *et al.*<sup>11</sup> reported  
 772 approximately a 40% reduction in WVP for ACNC films containing 30% CNF, prepared using  
 773 a solvent system comprising a mixture of Trifluoroacetic Acid and Trifluoroacetic Anhydride.  
 774 Zhang *et al.*<sup>82</sup> reported that CNFs addition to polyhydroxybutyrate (PHB) films resulted in a  
 775 47.0% reduction in WVTR, decreasing from 298.7 to 140.4 g/(m<sup>2</sup> · h). Lu *et al.*<sup>83</sup> reported that  
 776 the WVTR of commercial polyethylene films ranges from 0.1 to 40 g/m<sup>2</sup>/day, and the lowest  
 777 value observed in the present study was comparable to this range of commercial packaging  
 778 films. The enhanced moisture barrier is attributed to CNF-induced densification of the cellulose  
 779 matrix, which restricts water vapor passage. High surface, dense network formation, and  
 780 interfacial adhesion collectively and significantly enhanced the efficiency of the film by  
 781 increasing the tortuosity of the diffusion path when positioned perpendicular to the direction  
 782 of diffusion<sup>79, 84, 85</sup>, as illustrated in Fig. 9 (b). Optimal CNF loading also minimizes polymer  
 783 swelling and promotes the formation of an extensive hydrogen-bonded network, further  
 784 strengthening structural integrity and interfacial interactions.<sup>48, 86</sup> Furthermore, CNFs  
 785 incorporation resulted in a reduction of the moisture content of ACNC films (Table 2), which  
 786 likely contributed to improved water vapor barrier properties. Since cellulose is hydrophilic  
 787 and sensitive to moisture, water absorption triggers polymer swelling and plasticization, which  
 788 leads to increased polymer chain mobility and remarkably reduced water vapor barrier  
 789 properties of the film.<sup>87,88</sup>

790 The incorporation of nanofillers reduces the moisture content by reducing the free volume  
 791 available in the matrix.<sup>89,90</sup> Similarly, Faisal *et al.*<sup>51</sup> reported a decrease in moisture content and  
 792 WVTR due to the addition of CNF in amylose films. In the present investigation, moisture  
 793 content, film density, and porosity have also been found to be related to WVTR, as higher levels  
 794 of CNF reduced density, decreased porosity, and resulted in decreased WVTR. A positive  
 795 correlation between CNF-induced variation in porosity and reduction in water vapor barrier  
 796 properties has also been observed by Wei *et al.*<sup>91</sup> and Cainglet *et al.*<sup>92</sup>.



810 Fig. 8. (a) WVTR and (b) OTR of CNF0 and the developed ACNC films, showing a significant  
 811 and progressive decrease with increasing CNFs concentration up to CNF7 (7%). Different

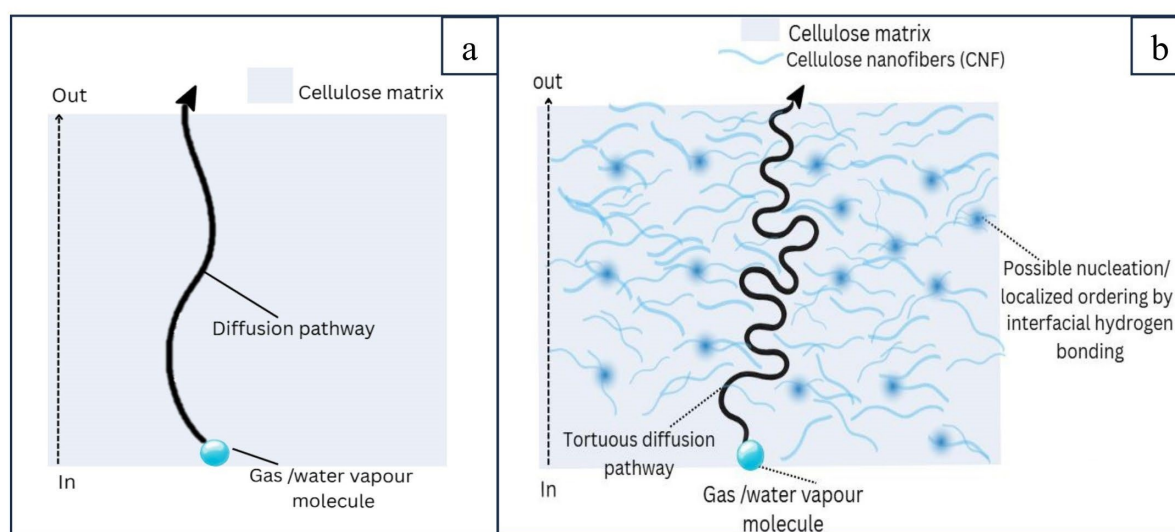


812 letters above the bars indicate statistically significant differences ( $p \leq 0.05$ ) among the  
 813 treatments. DOI: 10.1039/D6FB00014B

814

### 815 3.12 Oxygen Transmission Rate

816 Oxygen is a critical extrinsic factor in food preservation, as its presence can promote oxidative  
 817 degradation, microbial growth, rancidity, and browning, thereby compromising food  
 818 quality.<sup>93, 94</sup> Like WVTR, oxygen transmission rate (OTR) is also crucial for assessing film  
 819 barrier properties, which help extend shelf life, prevent spoilage, and preserve food  
 820 quality.<sup>10,95,96</sup> The improvement in the oxygen barrier properties of the composite film due to  
 821 the presence of CNF is significantly supported by previous studies.<sup>51,97,98,99</sup> Fig. 8 (b) displays  
 822 the OTR as a function of CNF content, where a significant decrease was observed from CNF0  
 823 to CNF7 with increasing CNF loading. As shown in Table 3, the highest value of OTR observed  
 824 in this study was  $52.7 \pm 2.2 \text{ cm}^3/\text{m}^2/\text{day}$  in CNF0, while CNF7 possessed the lowest OTR value  
 825 of  $5.6 \pm 0.8 \text{ cm}^3/\text{m}^2/\text{day}$ , showing a reduction of OTR by  $\sim 89.3 \%$ . Considering the OTR value  
 826 for food packaging, which should typically be below  $20 \text{ ml}/\text{m}^2/\text{day}$ <sup>100</sup>, the lowest OTR of  
 827 CNF7 was a good indication of its oxygen barrier property. As illustrated in Fig 9 (b), this  
 828 decreased OTR in CNF7 could be the result of tortuous path for diffusion of oxygen molecules  
 829 due to CNF-induced nucleation, and recrystallization. In contrast, the absence of CNF in CNF0  
 830 resulted in a relatively lesser obstruction in the diffusion pathway, which led to higher OTR  
 831 values (Fig 9a).  
 832



833

834

835 Fig. 9. Schematic representation of diffusion pathways in (a) CNF0 and (b) CNF7, showing  
 836 CNF-induced nucleation, recrystallization, and increased tortuosity, leading to reduced WVTR  
 837 and OTR of ACNC films.

838

839 Similar improvement in oxygen barrier property due to the incorporation of CNF has been  
 840 validated in a previous study by Faisal *et al.*<sup>51</sup>, who compared the reinforcing effect of CNC  
 841 and CNF in amylose film and found that the presence of 3% nanofibrils resulted in improved  
 842 gas barrier properties compared to the composite prepared with 3% CNC. Nguyen *et al.*<sup>77</sup>  
 843 developed composite films by adding boron nitride nanosheets to the CNF film for the meat  
 844 and cheese packaging, and observed that OTR values were reduced to  $4.7 \text{ cc}/\text{m}^2/\text{day}$  in the film  
 845 containing 5% nanosheets, which was comparable with the OTR for CNF7 obtained in the



846 present study. It was worth noting that the OTR improvement aligned well with the high tensile  
847 strength as the reinforcement effect of CNF, which promoted the creation of a strong hydrogen-  
848 bonded cohesive network due to high surface area and nanoscale size.<sup>101</sup> Interestingly, a  
849 positive correlation between OTR, porosity, density, and moisture content of the films was  
850 evident in this study. The incorporation of CNFs into the matrix led to a reduction in the porous  
851 structure of the films, with CNF7 exhibiting the lowest porosity and correspondingly reduced  
852 OTR. The nanosized fibers could have filled the voids within the polymer matrix, decreased  
853 free volume, enhanced nanofiber-polymer contact, and facilitated effective stress transfer at the  
854 interface. These observations were consistent with the measured density values, indicating that  
855 the dense structure formed in the ACNC films likely extended the diffusion path for oxygen  
856 molecules perpendicular to the film orientation.<sup>9, 102</sup> A similar trend was also observed in the  
857 moisture content values, which demonstrated a clear dependence of oxygen barrier  
858 performance on moisture content, as the presence of moisture can act as a plasticiser and disturb  
859 hydrogen bonding. This can lead to a lowering of cohesive energy with an increase in  
860 porosity<sup>103</sup>, resulting in passage through, which molecules that can permeate.<sup>104</sup> However, the  
861 presence of CNF promotes the formation of a stronger hydrogen-bonded structure, and compact  
862 fiber alignment tends to slow down the transmission<sup>105</sup> and limit polymer chain mobility. The  
863 improvement in structural properties and barrier performance correlated well with the CNF  
864 concentration, with maximum improvement observed in CNF7. As the CNF concentration  
865 increased further, the agglomeration of CNF affected the functionality of the film. This is due  
866 to the intrinsic tendency of CNF to agglomerate as a result of strong intermolecular hydrogen  
867 bonding and van der Waals interactions, which leads to uneven dispersion within the matrix,  
868 affecting film properties.<sup>48</sup> Additionally, Crystallinity also plays a significant role in restricting  
869 oxygen transport through the nanocomposite film. CNF with high crystallinity may act as a  
870 nucleating agent and interact strongly with the polymer matrix, promoting H-bonding and  
871 intercrosslinking, which can improve gas barrier properties.<sup>106</sup> A similar effect was observed in  
872 the present study, as the sample with higher crystallinity had enhanced oxygen barrier  
873 performance. Belbekhouche *et al.*<sup>107</sup> attributed lower oxygen barrier properties in cellulose  
874 nanofiber (CNF) compared to cellulose nanocrystal (CNC) films to their higher density, and  
875 enlargement of the fibrous network. Nair *et al.*<sup>108</sup> observed that the higher crystallinity and  
876 morphology of nanofibers in the film led to an intricate structure due to fibrous entanglement,  
877 forming a complex structure having low porosity and increased tortuosity, which resulted in  
878 reduced OTR.

879

### 880 3.13 Optical transmittance

881 Optical transmittance is a functional criterion for assessing the compatibility and miscibility of  
882 composite components. High transparency is particularly desirable in food packaging  
883 applications as it enhances visual appeal and consumer acceptance. Table 4 presents the optical  
884 transmittance values at a wavelength of 800 nm for all the film samples. The CNF0 exhibited  
885 a high transmittance of  $90.05 \pm 0.21$  %, demonstrating excellent transparency. The CNF3 and  
886 CNF5 showed slightly reduced transmittance values of  $88.13 \pm 0.38$  % and  $86.74 \pm 0.29$  %, respectively,  
887 while CNF9 decreased further to  $79.19 \pm 0.31$ %. The results obtained are  
888 consistent with the findings of Zhao *et al.*<sup>36</sup> where a decrease in optical transmittance was  
889 observed with increasing CNF content (5% to 20%) in the ACNC films, with the lowest



890 transmittance being 75.97% in the film containing 20% CNF. Similarly, Wang *et al.*<sup>76</sup> reported  
 891 that the optical transmittance of PVA film was slightly decreased due to CNF incorporation,  
 892 with PVA film containing 2% CNF exhibiting the lowest optical transmittance of 77.55%.

893 This reduction in transmittance most likely occurred due to the elevated light scattering caused  
 894 by the presence of CNF, as well as a slight difference in refractive index<sup>109</sup> between CNF and  
 895 the matrix and potential micro-aggregation within the film. The observed behaviour  
 896 demonstrated that, despite the slight reduction with CNF presence, the transmittance remained  
 897 relatively favourable ( $86.111 \pm 0.35\%$ , %) in CNF7, indicating that the film retained good  
 898 optical clarity, which is desirable for food packaging.

899 The color of films for food packaging is an important quality parameter. It was observed that  
 900 the optical properties of the ACNCs were influenced significantly by the increasing CNF  
 901 content (Table 4). The value of lightness (L) decreased significantly from  $95.37 \pm 0.29$  in the  
 902 control (CNF0) to  $91.28 \pm 0.33$  in CNF9, whereas the value of chroma increased significantly  
 903 from 1.14 to 2.36. The observed decrease of the lightness ( $L^*$ ) and increase of chroma ( $C^*$ ) as  
 904 a function of the CNF content could most likely be due to the increase of light scattering and  
 905 opacity caused by the nanofibril network. The significant decrease in hue angle ( $H^\circ$ ) from  
 906  $105.5^\circ$  to  $94.1^\circ$  indicated that incorporation of CNF affected saturation rather than the overall  
 907 color tone. The overall color difference ( $\Delta E_{ab}$ ) compared to the control increased from 0 to  
 908 4.28, which demonstrated the perceptible color variations with an increase in CNF content.  
 909 This finding aligns with the results reported by Fan *et al.*<sup>210</sup>, where a variation was observed in  
 910 the  $L^*$  and  $b^*$  values between hyaluronic acid composite film and hyaluronic acid-curcumin-  
 911 cellulose nanofibre composite film, with the difference being influenced by the CNF addition.  
 912 Similarly, Mohammadi *et al.*<sup>211</sup> observed an increased  $\Delta E$  value of the nanocomposites when  
 913 chitosan nanofibers were included in a carboxymethyl cellulose matrix.

914  
 915  
 916 **Table 4. Transmittance (Tr), Color parameters ( $L^*$ ,  $a^*$ ,  $b^*$ ), chroma ( $C^*$ ), hue angle ( $H^\circ$ )  
 917 and total color difference ( $\Delta E_{ab}$ ) of film samples**  
 918

| Samples | Tr (%)             | $L^*$              | $a^*$              | $b^*$             | $C^*$             | $H^\circ$          | $\Delta E_{ab}$ |
|---------|--------------------|--------------------|--------------------|-------------------|-------------------|--------------------|-----------------|
| CNF0    | $90.05 \pm 0.21^a$ | $95.37 \pm 0.29^a$ | $-0.30 \pm 0.02^c$ | $1.1 \pm 0.12^a$  | 1.14 <sup>a</sup> | 105.5 <sup>a</sup> | 0               |
| CNF3    | $88.13 \pm 0.38^b$ | $94.30 \pm 0.32^b$ | $-0.27 \pm 0.04^b$ | $1.31 \pm 0.14^a$ | 1.34 <sup>a</sup> | 101.7 <sup>a</sup> | 1.09            |
| CNF5    | $86.74 \pm 0.29^c$ | $93.77 \pm 0.20^c$ | $-0.23 \pm 0.04^a$ | $1.75 \pm 0.15^b$ | 1.77 <sup>b</sup> | 97.6 <sup>b</sup>  | 1.73            |
| CNF7    | $86.11 \pm 0.35^d$ | $93.38 \pm 0.28^c$ | $-0.21 \pm 0.03^a$ | $2.03 \pm 0.13^c$ | 2.04 <sup>c</sup> | 94.1 <sup>b</sup>  | 2.11            |
| CNF9    | $79.19 \pm 0.31^e$ | $91.28 \pm 0.33^d$ | $-0.17 \pm 0.05^a$ | $2.35 \pm 0.18^d$ | 2.36 <sup>d</sup> | 93.9 <sup>b</sup>  | 4.28            |

919 Mean values with different superscript letters in the same column are significantly different  
 920 ( $p \leq 0.05$ ).

921

### 922 3.14 Thermogravimetric Analysis

923 Fig. 10 (A, B) depicts the TGA and DTG curves of CNF0 and ACNC films, providing insight  
 924 into their thermal stability and degradation behaviour. Thermal degradation of all films  
 925 occurred in three distinct degradation stages. The initial stage, between  $33^\circ\text{C}$  and  $201^\circ\text{C}$ , was  
 926 attributed to the evaporation of physically adsorbed moisture.<sup>112</sup> The main decomposition stage,



927 observed from 206 °C to 462 °C, corresponded to pyrolysis of cellulose during which glycosidic  
 928 linkages and carboxylic groups attached to the glucose units were cleaved.<sup>68,101</sup>  
 929 The peak degradation temperature (Td) varied among the ACNC films, with CNF7 exhibiting  
 930 the highest (335.40 °C), while CNF0 had the lowest (315.68 °C). This shift is attributed to CNF  
 931 reinforcement, consistent with previous studies: Kesari *et al.*<sup>58</sup> observed an increase from  
 932 311.58 °C for neat thermoplastic starch film to 319 °C with 5% CNF, and Bian *et al.*<sup>15</sup> reported  
 933 a rise from 290 °C in wheat straw cellulose film to 327 °C with 3% CNF. Gan *et al.*<sup>113</sup> noted  
 934 that CNF reinforcement and enhanced interfacial adhesion restrict polymer chain mobility,  
 935 raising the energy barrier for thermal degradation.<sup>50</sup> TGA curves also showed that  
 936 nanocomposite films had lower weight loss rates than CNF0 films, indicating improved heat  
 937 stability. The third degradation stage occurs between 423.33 °C to 597.61 °C, which leads to  
 938 the thermal decomposition of charred residue into gaseous byproducts.<sup>114</sup> The above findings  
 939 suggested that the presence of CNF as a nanofiller reinforcement in the nanocomposite made  
 940 the film more resistant to thermal stress.<sup>76</sup>

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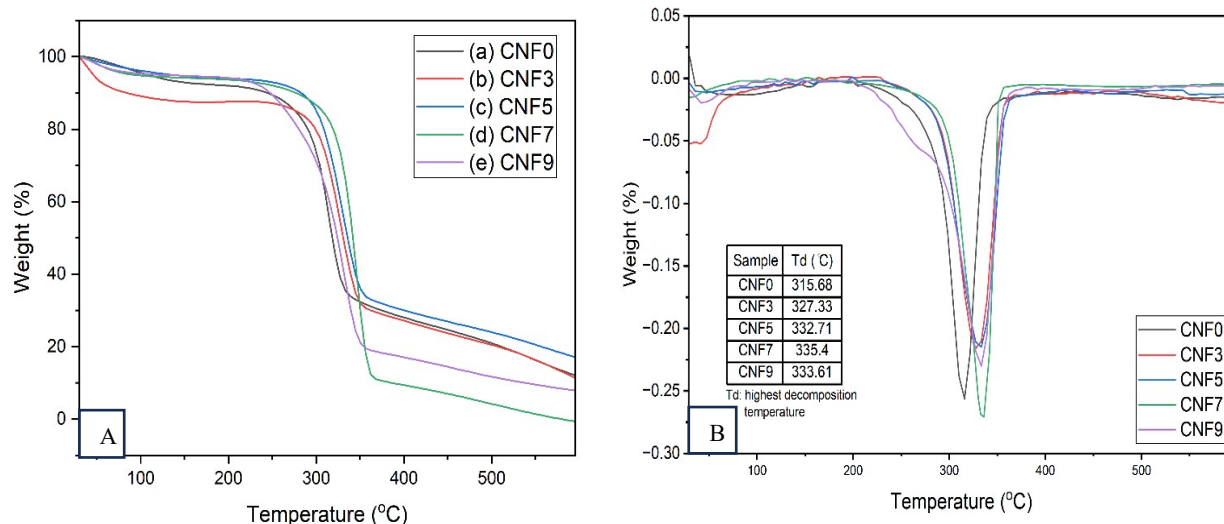
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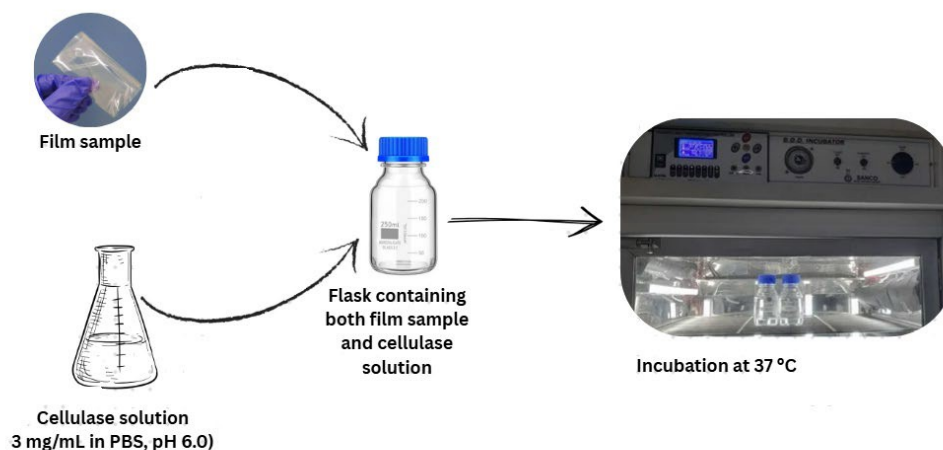
951 Fig. 10. The (A) TGA, and (B) DTG thermograms of the developed ACNC films. The insert in  
 952 the DTG thermogram shows the degradation temperatures (Td) of ACNCs, with CNF7  
 953 exhibiting the maximum value.

### 954 3.15 Biodegradability

955 The biodegradation behavior of CNF0 and CNF7 was assessed by monitoring weight loss over  
 956 time under enzymatic conditions. The biodegradability test setup, showing film samples  
 957 incubated in reagent bottles inside a controlled B.O.D. incubator, is displayed in Fig. 11.  
 958 Cellulase hydrolyzed the  $\beta$ -1,4 glycosidic bonds of cellulose, leading to the release of soluble  
 959 sugars that diffused into the medium. Consequently, the film samples became solubilized or  
 960 degraded into fragments, resulting in a gradual loss of weight over time. As shown in Fig. 12,  
 961 the incorporation of CNFs affected the degradation rate of all cellulose nanocomposite films  
 962 (CNF7) in the enzymatic solution. CNF0 exhibited a higher biodegradation rate compared to  
 963 CNF7, likely due to the dense hydrogen-bonded network formed between the cellulose matrix  
 964 and CNFs, which slowed enzymatic breakdown.

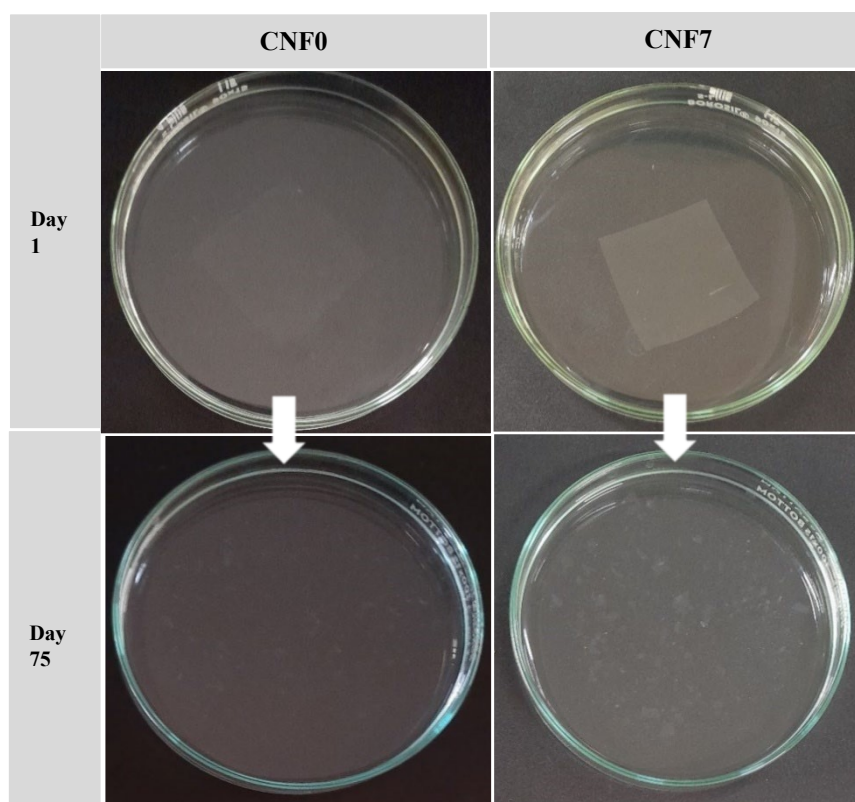
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966  
967 Fig. 11. Experimental setup for the enzymatic degradation of films. Film samples were placed  
968 in the cellulase solution and incubated at 37 °C. The enzymatic degradation of the films was  
969 monitored with time.



983 Fig. 12. Biodegradation of CNF0 and CNF7 films in cellulase enzyme solution. A higher  
984 number of fragments was observed in the CNF0 petriplate compared to the CNF7, which  
985 indicated relatively slightly faster disintegration during enzymatic degradation in CNF0

986

987

988

989 **Table 5. Biodegradable behaviour (% weight loss) of CNF0 and CNF7 films**



| Sample | Incubation Time (Days) |                        |                        |                        |                        |
|--------|------------------------|------------------------|------------------------|------------------------|------------------------|
|        | 15                     | 30                     | 45                     | 60                     | 75                     |
| CNF0   | 17.3±1.1 <sup>eA</sup> | 36.3±1.4 <sup>dA</sup> | 56.8±2.4 <sup>cA</sup> | 72.2±2.1 <sup>bA</sup> | 94.4±1.5 <sup>aA</sup> |
| CNF7   | 14.7±0.9 <sup>eB</sup> | 32.6±1.2 <sup>dB</sup> | 49.8±0.9 <sup>cB</sup> | 69.5±2.2 <sup>bB</sup> | 89.4±2.2 <sup>aB</sup> |

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990 Values are represented as mean ± SD (n=3). Means values with different small superscripts  
991 letters within the row and capital superscripts letters within the column of a parameter are  
992 significantly different (p≤0.05).

993 As shown in Table 5, CNF0 exhibited a significantly higher biodegradation rate than CNF7,  
994 and this trend persisted throughout the incubation period. After 75 days of incubation, there  
995 was 94.4% biodegradation of CNF0, while it was 89.4% for CNF7. The slower enzymatic  
996 degradation of CNF7 could likely be due to variation in film morphology, as the addition of  
997 CNF may have resulted in a strong hydrogen-bonded network between the cellulose matrix and  
998 CNFs due to the homogenous dispersion of CNF. Additionally, the higher crystallinity of CNFs,  
999 as reported in our earlier study<sup>22</sup> may have also contributed to slightly reduced biodegradability  
1000 of the ACNC films. Since enzymatic hydrolysis generally initiates at the amorphous regions,  
1001 followed by gradual degradation of the crystalline regions<sup>115</sup>, the reinforcement of CNF  
1002 reduced the proportion of amorphous cellulose available for enzymatic degradation, and thus  
1003 slowed the degradation rate. A similar observation was reported by Louis et al.<sup>116</sup>, where  
1004 nanocellulose-reinforced starch films exhibited a slower degradation rate than pure starch films  
1005 under the soil burial method.

#### 1007 4. Conclusion

1008 The present study successfully demonstrated the development of all-cellulose nanocomposite  
1009 films derived from rice straw and reinforced with cellulose nanofibers (CNFs) using a  
1010 LiCl/DMAc solvent system. The ACNC films were prepared with process parameters  
1011 optimized to prevent dimensional shrinkage. The study provided a comprehensive evaluation  
1012 of the films' structural, mechanical, optical, thermal, barrier, and biodegradability properties,  
1013 highlighting their relevance to sustainable food packaging applications. The study also offered  
1014 a systematic and comprehensive understanding of the structure-property relationship that  
1015 controlled the performance of the films. The film containing 7% CNFs showed the highest  
1016 tensile strength and tensile modulus, while maintaining adequate extensibility. The low OTR  
1017 and WVTR of CNF7 were a good indication of its barrier properties, which were comparable  
1018 to the range of other commercial packaging films. The developed ACNC film possessed lower  
1019 moisture content and sufficiently good biodegradability. Overall, the ACNC film (CNF7) could  
1020 be considered as a promising biodegradable and eco-friendly alternative to commercial non-  
1021 biodegradable polyethylene films or the other packaging materials. The use of abundantly  
1022 available underutilised rice straw for the development of the film meets the single feedstock  
1023 circular approach. The ability to achieve desirable film properties at low CNF loading of 7%  
1024 highlights the cost-effectiveness and sustainability of the developed system for food packaging  
1025 applications.

1026 Further investigations are required to evaluate the heat-sealability and self-bonding  
1027 characteristics of ACNC films to establish their commercial viability. However, for large-scale  
1028 applications, the LiCl/DMAc solvent system may be replaced with more environmentally  
1029 friendly and scalable alternatives, such as NaOH/urea/water or deep eutectic solvents, to



1030 enhance sustainability, safety, and regulatory compliance for food packaging applications. In  
 1031 addition, efficient recovery and recycling of solvents after film production will be critical for  
 1032 promoting environmental compatibility and industrial scalability. Future research should also  
 1033 investigate the performance of the developed ACNC films across a wider range of food systems  
 1034 under diverse environmental and storage conditions.

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### Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

