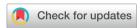
Environmental Science Water Research & Technology



CORRECTION

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Cite this: Environ. Sci.: Water Res. Technol., 2025, 11, 781

Correction: Kinetics and mechanism of hydrolysis of PF₆⁻ accelerated by H⁺ or Al³⁺ in aqueous solution

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DOI: 10.1039/d5ew90007g

rsc.li/es-water

Correction for 'Kinetics and mechanism of hydrolysis of PF_6^- accelerated by H⁺ or Al^{3+} in aqueous solution' by Takuto Miyashita et al., Environ. Sci.: Water Res. Technol., 2025, 11, 281-292, https://doi.org/ 10.1039/D4EW00758A

The authors would like to address errors in the horizontal axis of Fig. 9(a), the horizontal axis of Fig. 9(b), the title of Fig. S3(a), the title of Fig. S3(b) and the figure caption of Fig. S3. The required corrections are described below.

In Fig. 9(a), the horizontal axis should be "Reaction time (h)" instead of "Reaction time (min)".

In Fig. 9(b), the horizontal axis should be "Reaction time (h)" instead of "Reaction time (min)".

In the title of Fig. S3(a), "10 mM LiPF₆ + 100 Al-mM $Al_2(SO_4)_3$ " should be "10 mM LiPF₆ + 10 Al-mM $Al_2(SO_4)_3$ ".

In the title of Fig. S3(b), "10 mM LiPF₆ + 100 mM Al(NO₃)₃" should be "10 mM LiPF₆ + 10 Al-mM Al(NO₃)₃".

In the figure caption of Fig. S3, the first sentence should be "Concentration over time of PF₆", PO₂F₂", PO₃F²", PO₄³", and F measured by ion chromatography when 10 mM LiPF₆ solutions with (a) 10 Al-mM Al₂(SO₄)₃ and (b) 10 Al-mM Al(NO₃)₃ were kept at 90 °C." instead of "Concentration over time of PF₆", PO₂F₂", PO₃F²", PO₄", and F measured by ion chromatography when 10 mM LiPF₆ solutions with (a) 100 Al-mM Al₂(SO₄)₃ and (b) 100 mM Al(NO₃)₃ were kept at 90 °C."

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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