



Cite this: *Chem. Commun.*, 2025, 61, 4407

Received 14th January 2025,
Accepted 7th February 2025

DOI: 10.1039/d5cc00237k

rsc.li/chemcomm

We report the first gradient aerosol chemical vapor deposition approach to fabricate vertically graded $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ thin films.

Compositionally graded films are a promising class of advanced materials with applications in a range of fields¹ such as thermoelectric energy conversion,² electrocatalysis,³ photoelectrocatalysis,⁴ photocatalysis,⁵ photovoltaics,⁶ photonics,⁷ optoelectronics,⁸ sensors,⁹ corrosion, wear,¹⁰ thermal shields,¹¹ and biomedicine.¹² Compositionally graded thin films are distinguished from homogeneous thin films by spatial variation in elemental or crystallographic composition. Spatial grading of a thin films composition may allow optimization of the physical properties of thin films such as electrical and optical properties and thermal transport.⁴ It may also result in improved efficiencies in, for example, water splitting,³ and enhanced device stability against degradation in challenging environmental conditions.¹⁰ Compositional variation can either be lateral (in the x/y -dimension)^{2,5} or vertical (in the z -dimension),^{4,6} with the latter far more useful for device fabrication.

In compound semiconductors, particularly isostructural and isoelectronic inorganic semiconductors such as $\text{Bi}_2\text{S}_3/\text{Sb}_2\text{S}_3$, MoS_2/WS_2 , or PbS/PbSe , the metal and/or non-metal components can be substituted within the crystal structure with little thermodynamic penalty, allowing control over composition, with negligible impact on phase stability and with crystallographic lattice parameters obeying Vegard's law as a function of mole fraction.⁸ However, the compositional changes in many cases does have a significant impact on the electronic and/or optical properties of these materials, allowing, for example, bandgap engineering of materials as a function of composition.^{6,8,13} Compositional control thus enables these semiconductors to be highly tuneable, enabling tailored functional and physical properties, and this has

Gradient aerosol chemical vapor deposition (GA-CVD) for vertically graded thin films of $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ †

Abraham Barde, Huda Salam Al-Jurani, Brendan Ward-O'Brien, Mark A. Buckingham, Robert Cernik and David J. Lewis *
✉

been demonstrated by myriad theoretical, computational, and experimental studies.^{2,8,13,14}

Aerosol assisted chemical vapor deposition (AA-CVD) using single source precursors is a well-developed, effective technique toward thin film deposition.^{14–16} It has become popular primarily because the requirement for volatile precursors that is crucial in other CVD techniques is relaxed by the nature of the precursor delivery in AA-CVD, and thus a wide range of precursors, often inorganic complexes, have been used to produce materials. Indeed, AA-CVD has been previously exploited to deposit films with laterally graded composition for $\text{Ga}_x\text{In}_{2-x}\text{O}_3$,¹⁷ and $\text{TiO}_2/\text{SnO}_2$ ⁵ using *in situ* precursor mixing and for producing multi-layer films *via* multiple deposition steps.¹⁸ However, to the best of our knowledge, deposition of compositionally graded films in the vertical direction has not yet been achieved using this technique. In this paper, we demonstrate a new aerosol CVD methodology – gradient aerosol chemical vapour deposition (-CVD) – which can deposit vertically graded metal sulfide thin films. We use $\text{Mo}-\text{W}-\text{S}$ alloys as an exemplar system to showcase the approach.

The experimental setup for depositing vertically graded thin films is shown in the ESI† (Fig. S3). A programmable precursor feed introduces $[\text{Mo}(\text{DTC})_4]$ and $[\text{W}(\text{S})\text{S}_2(\text{DTC})_2]$ to a digital ultrasonic humidifier. These precursors were chosen as they individually decompose to produce MoS_2 and WS_2 , which are isostructural transition metal dichalcogenides.^{19–21} They are also able to produce the full range of $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ alloys when the precursors are co-decomposed in a relatively narrow temperature range commensurate with each other, which has been demonstrated in both solventless thermolysis and in AACVD experiments.^{13,14,21} The generated aerosol from the dual feed of precursors is carried to the reactor vessel (pre-heated to 450 °C) with an argon carrier gas. This setup allowed the incorporation of both Mo and W into the deposited thin film, with the levels of each element delivered by the precursor feed changing over the course of the experiment. Table S1 (ESI†) gives details of the syringe pump parameters over the course of a typical deposition. By increasing Mo concentration and decreasing W

Department of Materials, The University of Manchester, Oxford Road, M13 9PL, UK. E-mail: david.lewis-4@manchester.ac.uk

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d5cc00237k>



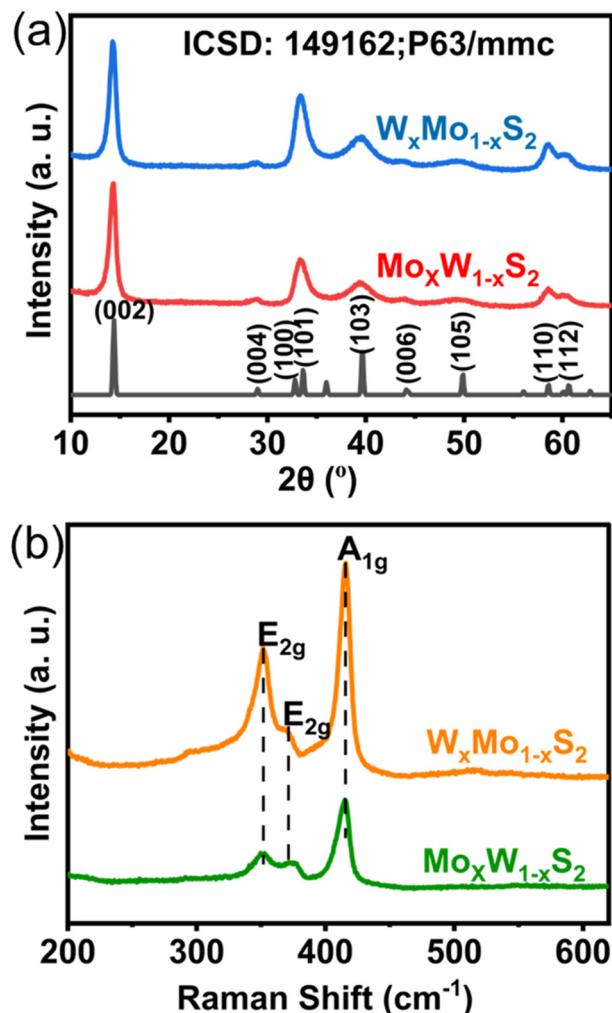


Fig. 1 (a) Grazing incidence powder X-ray diffraction (GI-XRD) patterns of $Mo_xW_{1-x}S_2$ and $W_xMo_{1-x}S_2$ thin films compared to a standard 2H-MoS₂ powder pattern. (b) Raman Spectra of $Mo_xW_{1-x}S_2$ and $W_xMo_{1-x}S_2$ thin films.

concentration over time in the precursor feed we were able to produce vertically graded $Mo_xW_{1-x}S_2$ thin films. By increasing W concentration and decreasing Mo concentration over time in

the precursor feed, we were able to produce vertically graded $W_xMo_{1-x}S_2$ thin films. In both cases, x is defined as 0 at the substrate interface with the metal chalcogenide.

Grazing incidence powder X-ray diffraction (GI-XRD), with an incidence angle of 3° was used to confirm that a single crystalline phase was achieved for both types of graded depositions as shown in Fig. 1(a). Both diffraction patterns were indexed to isostructural hexagonal 2H-molybdenite/tungstenite phases. Reflections centred at *ca.* 14.3° , 33.3° , 39.0° , 58.6° , and 60.6° were observed, corresponding to the (002), (101), (103), (110), and (112) Bragg planes, respectively.²² We compared this data to that of binary MoS₂ and WS₂ thin films and a ternary Mo_{0.5}W_{0.5}S₂ film which were deposited by standard AA-CVD and characterised by GI-XRD (Fig. S4(a), ESI[†]). All peaks present in the diffraction pattern can be assigned to the desired material suggesting there were minimal crystalline impurities. Rietveld refinements were also undertaken on all XRD patterns to extract crystalline unit cell parameters. Two isostructural intermetallic phases based on the 2H-molybdenite phase were allowed to refine independently to encompass the broad range of unit cell parameters observed in the sample, until a fit of acceptable quality ($R_{wp} < 10$) was reached (Table S2 and Fig. S5, ESI[†]). The fit predicts that both phases were significantly expanded along the unique axis in comparison to the bulk binary phases (which are very close to one another due to the similar ionic radius of Mo and W), which suggests that there may be significant residual stress within the sample. The final fit also revealed substantial preferred growth in the (002) plane for the one of the included phases and in the (101) plane for the other phase. We observed very similar behaviour when refining the patterns of the binary MoS₂ and WS₂ (Fig. S4 and S5, ESI[†]), suggesting the variation is not caused by differences in composition of the film. Preferred orientation in the (002) plane has previously been observed in MoWS thin films deposited by AA-CVD.^{13,19,20} Raman spectroscopy (Fig. 1b) was also used to further confirm molybdenite/tungstenite formation, with peaks associated with the E_{2g} (assigned to WS₂, 352 cm^{-1}), E_{2g} (assigned to MoS₂, 374 cm^{-1}), and A_{1g} (MoS₂/WS₂, 415 cm^{-1}) phonons all present, consistent with previously reported Mo–W–S₂ alloys.^{13,14}

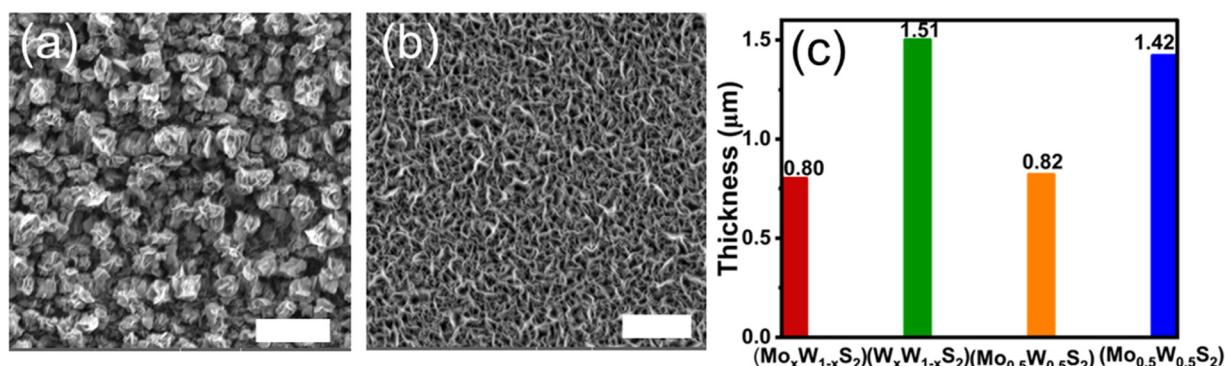


Fig. 2 Representative secondary electron (SE) micrographs of the deposited thin films. (a) $W_xMo_{1-x}S_2$, (b) $Mo_xW_{1-x}S_2$, and (c) plots of the thickness of the thin films as measured by cross-sectional SEM. $W_xMo_{1-x}S_2$, and $Mo_xW_{1-x}S_2$, were prepared by GA-CVD and the two $Mo_{0.5}W_{0.5}S_2$ by AA-CVD. The quantity of precursors used for the thin film samples with greater thickness (*i.e.*, the two taller bars) is twice the amount used in preparing the other two thin film with a thickness of $\sim 0.8\text{ }\mu\text{m}$. Scale bars represent $5\text{ }\mu\text{m}$.



Scanning electron microscopy (SEM) coupled to energy dispersive X-ray (EDX) spectroscopy was used to measure elemental homogeneity laterally across the films. SEM-EDX images are shown in Fig. S9 (ESI[†]) and show an even incorporation of Mo, W, and S across the thin films.

The morphology of the $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ film surfaces was assessed using secondary electron SEM imaging. The images in Fig. 2(a and b) reveal that densely packed, floret ($\text{W}_x\text{Mo}_{1-x}\text{S}_2$) and vertically aligned lamellar ($\text{Mo}_x\text{W}_{1-x}\text{S}_2$) crystallites were formed, which were evenly distributed on the entire substrate. These morphologies are consistent with a previous report of synthesised $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ ($0 \leq x \leq 1$) thin films fabricated *via* AA-CVD.¹³ Cross-sectional SEM (Fig. S8(a and b), ESI[†]) imaging showed that a continuous film was deposited, with a thickness of *ca.* 0.8 μm and 1.5 μm for $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ films, respectively. The discrepancy in the film thickness arises from the fact that $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ was synthesized using twice the combined molar quantity of precursors compared to that used for $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ (see Table S1, ESI[†]). This was intentionally undertaken to demonstrate that compositional gradients can be achieved at varying film thickness, showing the tailorabile nature of this method.

Glow discharge optical emission spectroscopy (GDOES) is an ablative elemental analysis technique that was used to investigate the elemental composition of the films as a function of their thickness. The lateral x/y resolution is 500 μm . Fig. 3(a and b) shows that as one metal signal linearly increases in intensity in the z -dimension the other is diminished linearly: this matches with the profile of the delivery of the precursors within the aerosol feed in the production of both $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ films. This depth profile analysis confirms the successful fabrication of crystalline thin films with a vertically graded metal sublattice. As a control experiment, the GDOES elemental depth profiling of Mo and W was also performed on the $\text{Mo}_{0.5}\text{W}_{0.5}\text{S}_2$ thin film deposited with the standard AA-CVD experiment. We found a near constant composition across the whole depth of the film in the z -dimension, with hardly any elemental variation in Mo or W, unlike the compositionally graded films produced by GA-CVD (Fig. S10, ESI[†]). This, alongside the Raman and GI-XRD data suggest that GA-CVD allows fabrication of vertically graded semiconductor thin films. Given the wide array of extant literature on AA-CVD which includes examples of deposition of inorganic, organic and hybrid materials, it is likely that myriad other functional thin films will now be possible through vertical grading.^{13,15,16,20}

In conclusion, a rapid and facile approach – gradient aerosol chemical vapour deposition (GA-CVD) – toward fabricating vertically graded thin films with full compositional control over metal stoichiometry in the z -dimension has been developed. We have demonstrated the utility using the $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and $\text{W}_x\text{Mo}_{1-x}\text{S}_2$ systems (where $x = 0$ at the substrate-metal chalcogenide interface) as exemplars. Linear control over the metal stoichiometry in the z -dimension has been achieved in these crystalline thin films in both cases. Given the now large library of metal–organic precursors available from the literature on AA-CVD, it is more than likely that GA-CVD can be easily expanded to further materials beyond the

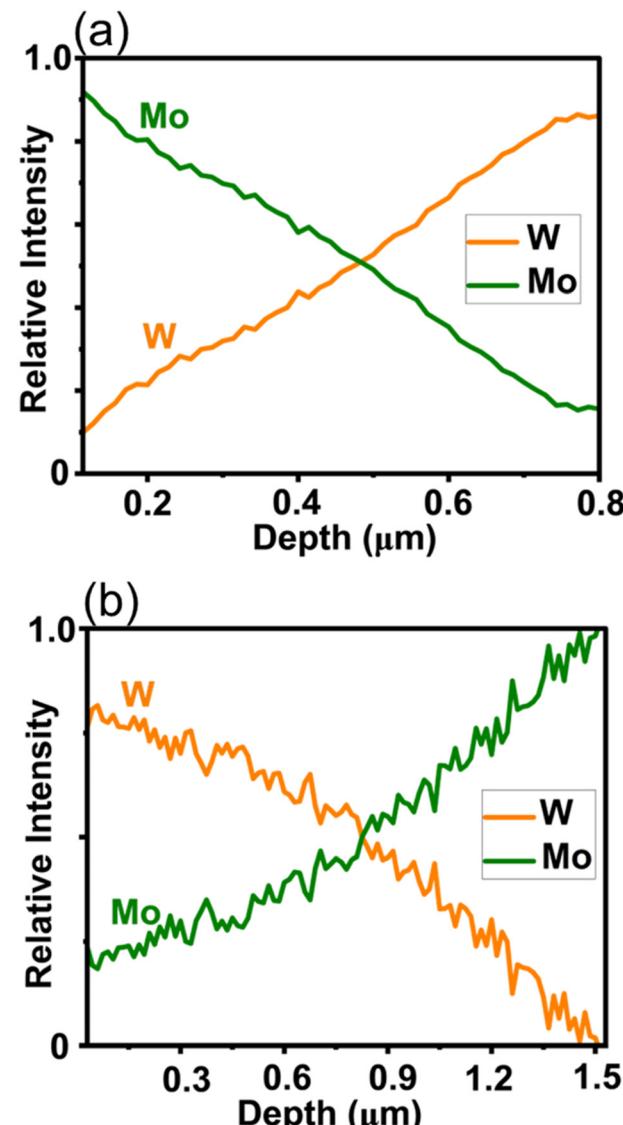


Fig. 3 Glow discharge optical emission spectroscopy (GDOES) plots revealing spatial changes in composition of Mo and W as a function of depth in deposited thin films. (a) $\text{Mo}_x\text{W}_{1-x}\text{S}_2$ and (b) $\text{W}_x\text{Mo}_{1-x}\text{S}_2$.

$\text{Mo}_x\text{W}_y\text{S}_2$ system. Broadly speaking, this new technique could also overcome the myriad challenges associated with lattice matching strain, adhesion, contamination and stability at the interfaces of successive depositions that have been observed perennially in the CVD literature, the elimination of which are important in the manufacture of semiconductor devices.²³

Conceptualisation: D. J. L.; methodology: A. B., D. J. L.; experimental design: A. B.; data collection and analysis: A. B., H. S. A.-J., B. W. O'B., M. A. B., R. C., D. J. L.; funding acquisition: A. B.; supervision: D. J. L.; validation: A. B., B. W. O'B., M. A. B., D. J. L.; visualisation: A. B., B. W. O'B.; writing original draft: A. B.; D. J. L.; review and editing: all authors.

A. B. would like to thank the PTDF Overseas Scholarship Scheme for a PhD scholarship award (PTDF/ED/OSS/PHD/AB/1782/2020PHD090). B. W-O'B. would like to thank the EPSRC for a DTP studentship.



Data availability

Data can be made available upon reasonable request.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- 1 B. Saleh, J. Jiang, R. Fathi, T. Al-hababi, Q. Xu, L. Wang, D. Song and A. Ma, *Composites, Part B*, 2020, **201**, 108376.
- 2 C. Ou, L. Zhang, Q. Jing, V. Narayan and S. Kar-Narayan, *Adv. Electron. Mater.*, 2020, **6**, 1900720.
- 3 S. M. Tan and M. Pumera, *ACS Appl. Mater. Interfaces*, 2017, **9**, 41955–41964.
- 4 Y. Xiao, Z. Fan, M. Nakabayashi, Q. Li, L. Zhou, Q. Wang, C. Li, N. Shibata, K. Domen and Y. Li, *Nat. Commun.*, 2022, **13**, 7769.
- 5 N. Chadwick, S. Sathasivam, A. Kafizas, S. M. Bawaked, A. Y. Obaid, S. Al-Thabaiti, S. N. Basahel, I. P. Parkin and C. J. Carmalt, *J. Mater. Chem. A*, 2014, **2**, 5108–5116.
- 6 S.-C. Yang, M. Ochoa, R. Hertwig, A. Arribia, A. N. Tiwari and R. Carron, *Prog. Photovolt.: Res. Appl.*, 2021, **29**, 630–637.
- 7 B. K. Singh, V. Bambole, V. Rastogi and P. C. Pandey, *Opt. Laser Technol.*, 2020, **129**, 106293.
- 8 K. Bogaert, S. Liu, T. Liu, N. Guo, C. Zhang, S. Gradečák and S. Garaj, *Sci. Rep.*, 2018, **8**, 12889.
- 9 E. Müller, Č. Drašar, J. Schilz and W. A. Kaysser, *Mater. Sci. Eng.*, 2003, **362**, 17–39.
- 10 A. Hadipour, M. Rahsepar and H. Hayatdavoudi, *Surf. Eng.*, 2019, **35**, 883–890.
- 11 M. Sathish, N. Radhika and B. Saleh, *Composites, Part B*, 2021, **225**, 109278.
- 12 H. Shi, P. Zhou, J. Li, C. Liu and L. Wang, *Front. Bioeng. Biotechnol.*, 2020, **8**, 616845.
- 13 A. A. Tedstone, E. A. Lewis, N. Savjani, X. L. Zhong, S. J. Haigh, P. O'Brien and D. J. Lewis, *Chem. Mater.*, 2017, **29**, 3858–3862.
- 14 N. Zeng, Y.-C. Wang, J. Neilson, S. M. Fairclough, Y. Zou, A. G. Thomas, R. J. Cernik, S. J. Haigh and D. J. Lewis, *Chem. Mater.*, 2020, **32**, 7895–7907.
- 15 W. Xiao, M. A. Buckingham, Y. Li, K. Hazeldine, B. Han, S. H. Cartmell, A. S. Eggeman, A. S. Walton and D. J. Lewis, *Chem. Commun.*, 2023, **59**, 12427–12430.
- 16 A. Barde and D. J. Lewis, *J. Phys. Chem. C*, 2023, **127**, 13969–13977.
- 17 C. E. Knapp, A. Kafizas, I. P. Parkin and C. J. Carmalt, *J. Mater. Chem.*, 2011, **21**, 12644–12649.
- 18 J.-G. Song, G. H. Ryu, S. J. Lee, S. Sim, C. W. Lee, T. Choi, H. Jung, Y. Kim, Z. Lee, J.-M. Myoung, C. Dussarrat, C. Lansalot-Matras, J. Park, H. Choi and H. Kim, *Nat. Commun.*, 2015, **6**, 7817.
- 19 D. J. Lewis, A. A. Tedstone, X. L. Zhong, E. A. Lewis, A. Rooney, N. Savjani, J. R. Brent, S. J. Haigh, M. G. Burke, C. A. Muryn, J. M. Rafferty, C. Warrens, K. West, S. Gaemers and P. O'Brien, *Chem. Mater.*, 2015, **27**, 1367–1374.
- 20 G. Murtaza, S. P. Venkateswaran, A. G. Thomas, P. O'Brien and D. J. Lewis, *J. Mater. Chem. C*, 2018, **6**, 9537–9544.
- 21 N. E. Richey, C. Haines, J. L. Tami and L. McElwee-White, *Chem. Commun.*, 2017, **53**, 7728–7731.
- 22 K. G. Mondal, B. S. Kar, S. Rakshit, S. Saha, P. C. Jana and M. N. Goswami, *J. Alloys Compd.*, 2023, **969**, 172356.
- 23 G.-H. Park, K. Nielsch and A. Thomas, *Adv. Mater. Interfaces*, 2018, **6**, 1800688.

