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### Introduction

One emerging strategy for synthesizing complex chiral molecules is to use multiple catalysts to promote asymmetric tandem reactions, which features sequential activation of the starting materials or in situ-generated intermediates, with high levels of step- and atom-economy.<sup>1</sup> Remarkable advances have been made with asymmetric cascade reactions using three combinations of two distinct catalysts (i.e., metal/metal, metal/ organo, and organo/organo). These combinations enable enantio- and diastereoselective asymmetric ternary catalysis; however, the methods are quite limited.<sup>2</sup> As illustrated in Scheme 1a, where ternary catalysis is shown to be a cascade process in which three catalytic cycles are involved, intermediate I is formed in the first catalytic transformation and engages in the second catalyzed reaction to afford intermediate II. This is then subjected to a third catalyzed transformation to

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# Enantioconvergent and diastereoselective synthesis of atropisomeric hydrazides bearing a cyclic quaternary stereocenter through ternary catalysis†

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An efficient and highly enantioconvergent and diastereoselective ternary catalysis in a one-pot process is reported, which represents an integrated strategy for the synthesis of atropisomeric hydrazides with defined vicinal central and axial chirality from readily available racemic  $\alpha$ -amino-ynones, azodicarboxylates, and Morita–Baylis–Hillman (MBH) carbonates. This method utilizes in situ-generated racemic pyrrolin-4-ones via hydroamination of racemic  $\alpha$ -amino-ynones by AuCl catalysis as a novel and versatile C1 synthon, which engage commercially available azodicarboxylates to generate amination products in high yields and uniformly excellent enantioselectivities under the catalysis of a chiral phosphoric acid. Following amination, N-alkylation catalyzed by diastereoselective organocatalyst afforded axially chiral hydrazides with excellent diastereoselectivities (>98 : 2 dr). The synthetic utility of the amination products and axially chiral hydrazides was also demonstrated by their facile conversion to diverse molecules in high yields with excellent stereopurity. Density functional theory calculations were performed to understand the origin of diastereoselectivity. EDGE ARTICLE<br>
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deliver molecular complexity from simple starting materials. Racemic pyrrolin-4-one, which is easily accessed from readily available racemic  $\alpha$ -amino-ynones under gold catalysis,<sup>3</sup> may serve as a highly effective nucleophilic synthon for the ternary catalytic reaction. However, the functionalization of pyrrolin-4 ones using this strategy remains elusive (see Scheme 1b).

The hydrazides are a class of highly valuable compounds. Some hydrazides, such as aza-peptide analog CGP 53820, are bioactive molecules, exhibiting potent inhibition of HIV-1 and HIV-2 protease. Others are used in environmentally benign

a) A diagram of ternary catalytic reactions Cat-1  $\left(\text{Cat-2}\right)$  $\left(\text{Cat-}3\right)$  $Int.$ Int. II b) Design plan: racemic pyrrolin-4-c med in situ: novel, versatile C1 synthon for the ternary catalysis .NHBoc  $[Au]$  $\mathbf{G}$  $\epsilon$  $Cat-3$ Gouault et al.  $Cat-2$  $(+)$  $(\pm)$ 

Scheme 1 Ternary catalytic reactions for the efficient synthesis of complex chiral molecules.

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insecticides (e.g., halofenozide and derivatives; Scheme  $2(a)$ ).<sup>4</sup> However, several methods for the enantioselective synthesis of N–N axially chiral heterocyclic architectures have been reported. Nonetheless, the relevant studies focused only on their atroposelective creation.<sup>5</sup> Recently, the elegant catalytic enantioand diastereoselective synthesis of axially chiral hydrazides was achieved via two sequential asymmetric organocatalytic cycles in one pot (see Scheme 2b).<sup>6</sup> However, synthesizing enantiopure N–N axially chiral hydrazides bearing a cyclic quaternary stereocenter requires multiple steps. Rinaldi et al. reported two such methods. The first employs an enantiopure ester substrate to control the installation of N–N axial chirality, obtaining the desired product in 17% yield in five steps with a low dr of  $2.4:1$ (see Scheme 2c, eqn  $(1)$ ).<sup>7</sup> The second method employs an enantiopure diastereomeric ester to afford an N–N axially chiral product in 36% yield over three steps and a low dr of 1.2 : 1



Scheme 2 Bioactive molecules containing the hydrazides and synthesis of the optically active atropisomeric hydrazides bearing a cyclic quaternary stereocenter.

(Scheme 2c, eqn  $(2)$ ).<sup>8</sup> Notably, only azo dicarboxylates with a tert-butyl group (excessive steric hindrance) are used as the nitrogen source in both reactions. In summary, an efficient catalytic approach for the synthesis of optically pure atropisomeric hydrazides with a wide substrate scope is yet to be developed.

To improve the field of axially chiral entities<sup>9</sup> and asymmetric relay catalytic reactions,<sup>10</sup> we examined the possibility of constructing axially chiral hydrazides through ternary catalysis. Hence, this study reports an unprecedented example of ternary catalysis that integrates gold, Brønsted chiral phosphoric acid (CPA), and an organocatalyst (DMAP), and we demonstrate its effectiveness in the highly enantioconvergent and diastereoselective construction of atropisomeric hydrazides via a onepot sequence involving an intramolecular hydroamination, an intermolecular asymmetric amination, and N-alkylation (see Scheme 2d).

### Experimental section

We initially designed the enantioconvergent synthesis of 3 by employing the dual catalysis of achiral AuCl and CPA as a starting point for studying intramolecular hydroamination, followed by asymmetric amination to facilitate the subsequent development of annulation reactions. We surveyed the reaction between racemic a-amino-ynone (1a) and dibenzyl azodicarboxylate  $(2a)^{11}$  with AuCl<sup>12</sup> and CPAs<sup>13</sup> via dual catalysis strategies.<sup>14</sup> As listed in Table 1, various CPAs can enable the relay reaction to proceed smoothly and afford the desired product 3a in 5–86% yield at a high level (83–88%) of ee values





 $a$  The reaction was carried out with 1a (0.06 mmol, 1.2 equiv.), azo dicarboxylate 2a (0.05 mmol, 1.0 equiv.), AuCl (10 mol%), CPA (5 mol%) in solvent (1 mL) under a  $N_2$  atmosphere at 30 °C for 18 h. <sup>b</sup> Isolated yield of  $3a$  with respect to  $2a$ .  $\epsilon$  Determined by HPLC.  $\epsilon$  d With additive 4 Å MS (50 mg).  $\epsilon$  With additive 5 Å MS (50 mg).

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(see Table 1, entries 1–5). Hence, the common commercially available CPA3 was used as the catalyst, and the desired enantioconvergent amination 3a was obtained in 86% yield and 88% ee (see Table 1, entry 3). Moreover, different solvents and molecular sieves were tested in the presence of CPA3 (see Table 1, entries 6–10). With the optimal choice of cyclohexane and 5 Å MS, an excellent ee of 93% was observed for 3a.

### Results and discussion

With reaction conditions optimized, we next examined the generality of the catalytic reaction (see Scheme 3), starting with a series of commercially available azodicarboxylates (i.e., DEAD, DIAD, and DBAD) reacting with racemic  $\alpha$ -amino-ynone 1a to afford products (3a–3c). Good yields (92–96%) and excellent enantioselectives (97%) of products 3b and 3c were obtained for ethyl and isopropyl azodicarboxylates. The absolute configuration of 3b was unambiguously determined by single-crystal Xray diffraction (XRD). The configurations of other products were assigned as analogies; however, when azo dicarboxylates with tert-butyl groups (excessive steric hindrance) were used as the nitrogen source, the target product was not observed.

Variations in racemic a-amino-ynones <sup>1</sup> were explored next. Notably, the reaction can tolerate a wide variety of electrondonating or -withdrawing groups at different positions of the phenyl group. Furthermore, the yields and enantioselectivities of the corresponding products 3d–3i remained high. Alternatively, when using the less bulky racemic  $\alpha$ -amino-ynones 1 bearing simple methyl or benzyl substituents, the reaction provided 3j–3l in high yields and 80–91% ee. The alkyl group was replaced with an aryl group that afforded product 3m with high yield and moderate enantioselectivity (40% ee). The  $R^2$ group was replaced with a tert-butyl group, which failed to provide the desired product 3n, presumably due to the size of the tert-butyl group. Fourteen additional examples are illustrated in Scheme S1 (see the ESI† for details). Consequently, this novel cascade reaction represents an atom- and stepeconomical approach.

We then turned our attention to the N-alkylation reaction of the intramolecular hydroamination/intermolecular amination reaction to create atropisomeric hydrazides (see Scheme 4).<sup>15</sup> To our delight, the racemic 3k underwent an N-alkylation reaction in the presence of the Morita–Baylis–Hillman (MBH) carbonate 4a, achiral catalytic DMAP, and toluene to deliver an axially chiral hydrazide bearing a cyclic quaternary stereocenter 5 with 90% yield but only  $1:1$  dr (Scheme 4, eqn  $(1)$ ). Nonetheless, the dr value can be improved to 2 : 1 when the solvent is switched to cyclohexane, which we used to prepare substrate 3 via a relay catalytic reaction. Enantioenriched 3k was employed in the reaction, and the desired product 5 was obtained with the same enantiopurity. We speculate that the steric hindrance of the isopropyl groups may have affected diastereoselectivity. Notably, either racemic or enantiopure 3c was used as a substrate. Fortunately, a single diastereomer 6a was obtained at 92% yield, and high enantiopurity was retained without erosion (see Scheme 4, eqn (2)). Both CPA3 enantiomers were tested, and those corresponding to the product were obtained as single diastereomers. This result demonstrates that CPA is not effective for diastereoselectivity in N-alkylation with MBH carbonate (see Scheme 4, eqn (3)). Chemical Science<br>
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The configurational stability of this new type of N-N-linked axially chiral compound was investigated both experimentally and computationally. Heating of product (6a) in toluene for 72 h at 150 °C led to no erosion of diastereopurity, which was corroborated by the high rotational barrier of 37.0 kcal mol<sup>-1</sup> calculated for the N–N axis at 150 °C. Notably, the cyclohexane solvent used for the N-alkylation reaction was the same as that used for the AuCl/CPA relay catalytic reaction. Our next goal was to create axially chiral hydrazides bearing cyclic quaternary stereocenters via ternary catalysis. Representative optimization studies are provided in the ESI (see Table S1†).



Scheme 3 Scope of gold and CPA.



Scheme 4 Investigation of axially chiral hydrazide and the effect of CPA.

Having optimized the reaction conditions, we turned to examine the generality of the ternary catalyst, by varying racemic  $\alpha$ -amino-ynones first. A wide range of electrondonating or -withdrawing group substituents on the phenyl moiety at the 3- and 4-positions were tolerated to afford axially hydrazides 6a–6m in good yield and uniformly excellent diastereo- and enantioselectivity (see Scheme 5). Next, different MBH carbonate ester groups were evaluated, and methyl and allyl esters gave the desired products (6n and 6p) with moderate yields and excellent enantioselectivity. In contrast, bulkier isopropyl ester groups gave axially hydrazide 6o in moderate yield, but with slightly lower enantioselectivity (90% ee). Other branched alkyl groups (i.e., cyclopentyl and cyclohexyl) were introduced, and the corresponding products (6q and 6r) were obtained in a high yield with excellent enantioselectivity. A linear alkyl group  $(^{n}Pr)$  was also found to be compatible, yielding 6s in moderate yield and enantioselectivity but with a low dr  $(2:1)$ . The  $\alpha$ -amino-ynones 1 bearing the 4-nitrobenzyl substituent were also tested, and the reaction afforded 6t with 76% yield and 94% ee. The relative configuration of racemic 6t was determined by single-crystal XRD, and the absolute configuration of the quaternary carbon stereogenic center of 6 was determined to be R according to the X-ray structure analysis of  $3b$ . To further confirm that the  $R$ , $R$ -configuration product was the key experimental outcome, we compared the experimental Edge Article<br>
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Scheme 5 Scope of the atroposelective synthesis of hydrazides. Scheme 6 Control experiments.

electronic circular dichroism (ECD) spectra of enantiopure 6a with the calculated ECD spectra of the two stereoselective products (see the ESI†). The alkyl group was replaced by a phenyl group and afforded product 6u with high yield and moderate enantioselectivity (40% ee).

We hypothesized that two possible pathways exist for the AuCl/CPA relay catalytic reaction. Hence, to gain insights into the relay catalytic reaction pathway, control experiments were performed stepwise (see Scheme 6). First, racemic pyrrolin-4-one 7 was obtained as the product (90% yield) from racemic  $\alpha$ -aminoynone 1a under the catalysis of AuCl (10 mol%) without diethyl azodicarboxylate 2a or CPA3. Subsequently, treatment of 7 with diethyl azodicarboxylate 2a and chiral CPA3 (5 mol%) afforded an amination product 3b in 90% yield and 97% ee (see Scheme 6, eqn (1)). Additionally, the stepwise procedure delivered the desired products in lower two-step yields than our relay catalytic approach and efficiently avoided the loss of racemic pyrrolin-4 one intermediates during the additional purification. Alternatively, when the reaction was carried out with 1a and diethyl azodicarboxylate 2b in the presence of catalytic CPA3 (5 mol%), the desired product 8 was not observed (see Scheme 6, eqn (2)). These results demonstrated that the relay catalytic reaction followed the first pathway (see Scheme 6, eqn  $(1)$ ).

To gain a better understanding of the reaction pathways, we monitored the kinetics of the reaction between 1a and 2b using <sup>1</sup>H NMR spectroscopy (see Scheme 7). The kinetic profiles of this reaction clearly indicated that an essentially full conversion of 1a to the intermediate product 7 was observed within 1 h at 30  $\degree$ C, which was then converted to the final product, 3b, in a remarkably high yield. Consequently, the two catalytic reactions were identified as relays.

Based on the above results, a plausible triple-relay catalytic reaction mechanism was proposed: (1) activation of racemic aamino-ynones (1) by AuCl (via I) yields intramolecular cyclization product II, and protonolysis of that then produces racemic pyrrolin-4-one 7. (2) Subsequently, 3-hydroxypyrroles were formed through enolization in the presence of CPA, and the CPA catalyst activated both reactants by forming dual H-bonds with 3-hydroxypyrroles (with their 3-hydroxy group) and azodicarboxylates, which underwent an additional reaction to provide the enantioenriched amination product 3, possessing a cyclic quaternary stereocenter, which is an enantiodetermining step. (3) The first  $S_N^{2}$  reaction was triggered by







Scheme 9 Product transformation.

the addition of the Lewis base catalyst DMAP to the MBH adduct, generating a good leaving group (OBoc), which grabbed the hydrogen from compound 3, delivering HOBoc (TBHC) and **Int-V**. TBHC decomposes into  ${}^t$ BuOH and CO<sub>2</sub>. Later on, the second  $S_N^2$  reaction occurred through the addition of Int-V to the achiral pyridinium species, creating the N–N axis, and delivered the final products 6 with excellent diastereoseletivities and the organocatalyst (DMAP) to participate in the next catalytic cycle (Scheme 8).

To further demonstrate the utility of this methodology, we performed a one-mmol-scale reaction and synthetic transformations. As shown in Scheme 9(a), a representative onemmol-scale reaction easily occurred to form product 3b in 95% yield and with excellent enantioselectivity, which demonstrated that this methodology can be utilized for synthesizing chiral pyrrolin-4-ones on a large scale. Relay catalytic products are tolerant of many functional groups and are valuable for versatile structural elaborations. Several illustrative

transformations are illustrated in Scheme 9(b). As demonstrated, the Boc protecting group on 3b was easily removed under acidic conditions to generate free pyrrolin-4-one 9 in a moderate yield. Moreover, halogenation of 3b with NBS or iodine at 30 °C delivered fully-substituted pyrrolin-4-one 10 or 11 in 91 or 88% yield, respectively (see Scheme 9(b)). Brominated product 10 was coupled with phenylboronic acid to afford the desired product 12 with an 80% yield. Using this procedure, a wide range of boronic acids can be employed to afford various fully substituted pyrrolin-4-ones. Iodinated product 11 can be coupled with trimethylsilylacetylene to generate a crosscoupling product. The TMS group was efficiently removed to generate chiral terminal alkyne 13 in 75% yield over a two-step sequence in enantiopure form. Notably, the terminal alkynyl group in  $13$  is a versatile functional group,<sup>16</sup> and the bromination of axially chiral hydrazide bearing a cyclic quaternary



Scheme 8 Proposed triple relay catalytic reaction mechanism.

stereocenter 6a with NBS at 30 °C in DCM delivered the desired product 14 in 88% yield. Additionally, the hydrolysis of 6a at room temperature delivered free acid 15 in 84% yield (see Scheme 9(b)). Notably, in all of these transformations, excellent enantiopurity remained nearly without erosion.

Finally, we computationally investigated the N-alkylation of 3c in the presence of carbonate 4a and DMAP to form 6a to understand the origin of its high diastereoselectivity. Preliminary calculations demonstrated that the barrier of rotation around the N–N axis of 3c was 22.4 kcal mol<sup>-1</sup> (see the ESI†), indicating that the axis was conformationally unstable and that the reaction diastereoselectivity was governed by the N-alkylation step. Subsequently, the entire reaction profile leading to the formation of RR-6a was calculated, and the results are illustrated in Scheme 10. The DMAP activation of 4a was calculated as a facile process with a barrier of 19.4 kcal mol<sup>−1</sup>. The generated **INT2** contained a t-butyl carbonate anion, which serves as a Brønsted base to deprotonate the N–H moiety of 3c and turn it into a more potent nucleophile. The nucleophilic N–C bond formation, followed by the regeneration of the DMAP catalyst, yielded 6a. The highestenergy step was determined to be nucleophilic N–C bond formation with a  $\Delta G_{295}$  value of 32.3 kcal mol<sup>-1</sup>.<br>We then colculated the discrepselective Edge Article<br>
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We then calculated the diastereoselectivity-determining transition states for N-alkylation of 3c and 3k, and the



Scheme 11 Diastereoselectivity of N-alkylation of 3c (top) and 3k (bottom). Enclosed are rotated structures showing the steric hindrance in 3c-RS-TS4. Δ $G_{298}$  values are reported in kcal mol<sup>-1</sup> and distances in Å.

optimized structures are shown in Scheme 11. Although the calculated energy differences of 4.3 and 2.2 kcal mol<sup>-1</sup> for 3c and 3k, respectively, would predict dr values much larger than those



Scheme 10 Calculated reaction profile of N-alkylation of 3c.  $\Delta G_{298}$  values are reported in kcal mol<sup>-1</sup> and distances in Å.

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observed experimentally, there was evidence that this might be the consequence of the presence of numerous low-lying frequencies in the transition states. Hence, their treatment with quasi-harmonic approximation will lead to improved results. Furthermore, predictions based on enthalpy differences, which are less sensitive to low-lying frequencies, yielded improved results. Nevertheless, in all cases, our calculated results predicted better diastereoselectivity for 3c than 3k, which is in excellent agreement with the experimental results and can be attributed to the greater steric hindrance exerted by the i-Pr group of 3c. A very close H–H contact of 1.96 Å was observed between the i-Pr group of 3c and the C–H protons of the incoming electrophile. In comparison, the shortest H–H contact between the Me group of 3k and the electrophile was 2.56 Å.

### Conclusions

We developed a catalytic, enantioconvergent, and diastereoselective synthesis of atropisomeric hydrazides from readily available starting materials. The cascade reaction proceeded via ternary catalysis involving gold(I) chloride, CPA, and DMAP, with all three catalytic cycles interwoven. A variety of racemic aamino-ynones, azodicarboxylates, and MBH carbonates were successfully assembled in one pot in good yields and with excellent regioselectivities, which delivered the products with uniformly excellent control of central and axial chirality. DFT calculations revealed the origin of diastereoselectivity. The success of this work not only provides a useful strategy for the construction of contiguous cyclic quaternary central and N–N axial chirality in one pot but also represents the first example of a catalytic synthesis of optically pure N–N atropisomers via ternary catalysis. The development of new catalytic processes for accessing other challenging structures with different chiralities is currently under investigation.

### Data availability

The data supporting this article have been uploaded as part of the ESI.†

### Author contributions

S. L. conceived the project. X. W., S.-J. W., X. X. and H. A. performed the experiments and prepared the ESI.† H. Y. performed the DFT studies. S. L., Z. T., H. Y. and M. W. W. wrote the manuscript. All authors discussed the results.

## Conflicts of interest

There are no conflicts to declare.

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