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New phenanthrenequinones from *Cymbidium* ensifolium roots and their anti-inflammatory activity on lipopolysaccharide-activated BV2 microglial cells†

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The roots of *Cymbidium ensifolium* yielded a total of 17 compounds, comprising two new compounds (1–2), one new natural product (3), and 14 known compounds (4–17). The structures of new compounds were determined through the analysis of their spectroscopic data, including NMR, MS, UV, FT-IR, optical rotation, and CD. The anti-inflammatory activity of the isolated pure compounds was assessed using lipopolysaccharide-activated BV2 microglial cells. Compounds 1, 3, 6, 12, 14, and 16 showed the ability to reduce LPS induced NO release in BV2 microglial cells, with IC₅₀ values of 9.95 \pm 2.13, 8.77 \pm 3.78, 2.39 \pm 0.91, 6.69 \pm 2.94, 2.96 \pm 1.38, 8.42 \pm 2.99 μ M, respectively and reduced the secretion of proinflammatory mediators (TNF- α , IL-6, MCP-1) in a concentration-dependent manner. Furthermore, the mechanistic role of the compound 3 was determined, which demonstrated its ability to inhibit the nuclear factor- κ B (NF- κ B) pathway through decreasing phosphorylation of p65 subunits.

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Introduction

Neuroinflammation has gained significant interest in recent years, particularly as it has been linked to chronic and progressive neurodegenerative conditions like Alzheimer's disease, Parkinson's disease, and multiple sclerosis. Microglia, the primary innate immune cells in the central nervous system, play crucial roles in supporting the health of neurons by contributing to cellular maintenance and innate immune

responses.² Microglia activation is evident in brain injuries and is triggered by various stimuli such as lipopolysaccharides (LPS), interferon- γ , or β -amyloid exposure.³ Activated microglia, considered a hallmark of neurodegeneration, can potentially exacerbate neurodegenerative processes by releasing proinflammatory and/or cytotoxic factors like IL-1 β , TNF- α , NO, and reactive oxygen intermediates (ROS).⁴ These neurotoxic factors contribute to neuronal cell damage and the development of neurodegenerative diseases.⁵ Hence, the development of agents capable of inhibiting the production of inflammatory mediators holds promise as a therapeutic approach for treating neurodegenerative diseases.⁶

Cymbidium, a genus belonging to the orchid family, is widely distributed throughout Southeast Asia, China, Japan, and Northern Australia. Certain species of Cymbidium, including Cymbidium finlaysonianum Lindl., Cymbidium aloifolium (L.) Sw., and Cymbidium ensifolium (L.) Sw., have been traditionally utilized as herbal remedies by Thai practitioners. Compounds isolated from C. finlaysonianum have been evaluated for cytotoxic activity against human small cell lung cancer (NCI-H187) cells. The ethanolic leaf extract of C. aloifolium has been documented for its analgesic and anti-inflammatory properties. The roots of Cymbidium ensifolium, referred to as "nang kham" or "chulan" in Thai, are utilized in traditional Thai medicine to alleviate liver dysfunction and nephropathy. In our previous studies, we investigated compounds isolated from the aerial parts of C. ensifolium for cytotoxic effects against lung

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cancer H460, breast cancer MCF7, and colon cancer CaCo2 cells. As part of our ongoing research on bioactive compounds from orchids, 2-14 we investigated the chemical constituents of the roots of *C. ensifolium* and assessed their effectiveness in inhibiting neuroinflammation.

Results and discussion

Structural determination

A phytochemical investigation of the methanolic extract of the roots of *Cymbidium ensifolium* resulted in the isolation of 17 compounds, including two new compounds (1–2), one new natural product (3), and 14 known compounds (4–17). The structures of known compounds were identified as 9,10-dihydro-2,5-dimethoxy-1,7-phenanthrenediol (4), 15 *p*-hydroxybenzaldehyde (5), 16 coelonin (6), 17 hircinol (7), 18 3,7-dihydroxy-2,4,6-trimethoxyphenanthrene (8), 19 bulbophyllanthrin (9), 20 *p*-hydroxybenzoic acid (10), 21 vanillic acid (11), 22

parviphenanthrine C (12), 23 denthyrsinin (13), 24 1-(4-hydroxylbenzyl)-4-methoxy 9,10-dihydrophenanthrene 2,7-diol (14), 25 4-hydroxymethyl benzoic acid (15), 26 blestriarene C (16), 27 and *N-trans*-feruloyl tyramine (17) 28 (Fig. 1).

Compound 1 was obtained as an orange-colored amorphous solid. The molecular formula $C_{18}H_{16}O_6$ was analyzed from its $[M+H]^+$ at m/z 329.1019 (calcd for $C_{18}H_{17}O_6$ at 329.1020) in the HR-ESI-MS. Furthermore, it had another major peak at m/z 679.1789 that coincided with the calculated 2M peak, $[2M+Na]^+$ at m/z 679.1791. FT-IR revealed absorption bands at 3396 cm⁻¹ (hydroxyl), 1726 cm⁻¹ (carbonyl), 2922, 1461, and 1376 cm⁻¹ (aromatic). The compound revealed UV absorptions at 210, 240, 305, and 395 nm indicated an aromatic system.²⁹ The ¹³C NMR and DEPT spectra revealed eighteen carbon signals comprising four methoxy, four methines, and ten quaternary carbons. The presence of two carbonyl carbons can be supported by the chemical shifts δ_C 179.4 and 185.4. The ¹H NMR spectrum of 1 also exhibited two singlet proton signals at δ_H 6.29 (1H, s, H-3)

Fig. 1 Chemical structures of isolated compounds 1–17 from the roots of Cymbidium ensifolium.

and 7.29 (1H, s, H-8), two ortho-coupled doublet protons at $\delta_{\rm H}$ 8.01 (1H, d, J = 8.8 Hz, H-9) and 7.95 (1H, d, J = 8.8 Hz, H-10), four methoxy groups at $\delta_{\rm H}$ 3.94 (3H, s, MeO-2), 4.05 (3H, s, MeO-5), 3.91 (3H, s, MeO-6) and 4.06 (3H, s, MeO-7). The HMBC correlations of C-1 ($\delta_{\rm C}$ 179.4) with H-3 and H-10; C-4 ($\delta_{\rm C}$ 185.4) with H-3 indicated the quinone structure at ring A of compound **1.** The first methoxy group ($\delta_{\rm H}$ 3.94) was positioned on ring A at C-2, as deduced from its HSQC correlation with MeO-2 carbon ($\delta_{\rm C}$ 55.7), and NOESY interaction with H-3. It was supported by the HMBC correlation of C-2 ($\delta_{\rm C}$ 158.2) with H-3 and MeO-2 protons. On ring B, one singlet proton at $\delta_{\rm H}$ 7.29 was assigned as H-8 based on its HMBC correlation with C-9 ($\delta_{\rm C}$ 131.2) and NOESY interaction with H-9 (Fig. 2). The position of second methoxy ($\delta_{\rm H}$ 4.06) at C-7 was supported by its HSQC correlation with MeO-7 carbon ($\delta_{\rm C}$ 60.8), HMBC correlation with C-7 ($\delta_{\rm C}$ 156.2), and NOESY correlation with H-8. The third methoxy group ($\delta_{\rm H}$ 3.91) was assigned to C-6 based on its HSQC correlation with MeO-6 carbon ($\delta_{\rm C}$ 60.0); the HMBC correlations of C-6 ($\delta_{\rm C}$ 143.8) with MeO-6 and H-8. The HSQC correlation between the methoxy group at $\delta_{\rm H}$ 4.05 and MeO-5 carbon ($\delta_{\rm C}$ 55.6); the HMBC correlation between MeO-5 protons and C-5 ($\delta_{\rm C}$ 150.5) confirmed the fourth methoxy group at C-5. Based on the above spectra, compound 1 was characterized as 2,5,6,7-tetramethoxy-1,4-phenanthrenequinone and given the trivial name cymensifin D.

Compound 2 was obtained as a pale yellow amorphous solid. The molecular formula $C_{18}H_{18}O_6$ was analyzed from its $[M + H]^{\dagger}$ at m/z 331.1176 (calcd for $C_{18}H_{19}O_6$ 331.1181) in the HR-ESI-MS. At m/z 683.2095, it likewise displayed a significant peak that matched the calculated 2M peak, $[2M + Na]^+$ at m/z 683.2105. FT-IR revealed absorption bands at 3333 cm⁻¹ (hydroxyl), 1717 and 1679 cm⁻¹ (carbonyl), 2923, 1464, and 1316 cm⁻¹ (aromatic).²⁹ The UV absorption bands exhibited at 205, 230, 280 and 332 nm indicate an aromatic system.29 The 1H NMR spectrum revealed a pair of *ortho*-coupled protons at $\delta_{\rm H}$ 7.84 (d, J=8.8 Hz, H-9), and 7.94 (d, J=8.5 Hz, H-10); an aromatic proton at $\delta_{\rm H}$ 7.02 (s, H-8); one oxymethine at $\delta_{\rm H}$ 4.67 (dd, J = 8.4, 6.0 Hz, H-3); four methoxy groups at $\delta_{\rm H}$ 3.64 (s, MeO-3), 4.11 (s, MeO-5), 3.95 (s, MeO-6) and 4.05 (s, MeO-7); methylene protons at $\delta_{\rm H}$ 3.42 (dd, J = 17.6, 6.0 Hz, H-2a) and 3.15 (dd, J = 17.6, 8.4 Hz, H-2b). The ¹³C NMR and DEPT spectra also exhibited eighteen carbon signals. Among them, two carbonyl carbons at $\delta_{\rm C}$ 192.5 (C-1) and 201.3 (C-4); one methylene at $\delta_{\rm C}$ 46.8 (C-2) and one oxymethine carbon at $\delta_{\rm C}$ 82.2 (C-3) were positioned on ring A based on their HMBC correlations [C-1 with H₂-2, H-3, and H-10; C-2 with H-3; C-3 with H_2 -2 and MeO-3; C-4 with H_2 -2] (Fig. 2). In the COSY spectrum, a correlation between H2-2 and H-3 was observed. The position of the methoxy group (δ_H 3.64) at C-3 was

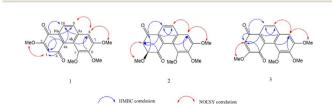


Fig. 2 HMBC and NOESY correlations of compounds 1-3.

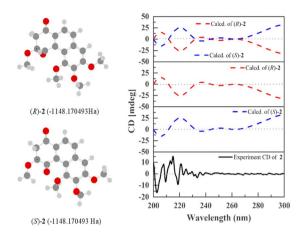


Fig. 3 Calculated and experimental ECD spectra of 2.

confirmed by its NOESY correlation with H2-2. The positions of the second ($\delta_{\rm H}$ 4.05), third ($\delta_{\rm H}$ 3.95) and fourth ($\delta_{\rm H}$ 4.11) methoxy groups were placed on ring B at C-7, C-6, and C-5 based on their HMBC and NOESY interactions as similar as compound 1. The experimental electronic circular dichroism (ECD) spectrum of compound 2 exhibited a negative cotton effect at 202.7 nm, along with positive cotton effect at 215.0 nm. These spectral features were consistent with the (S)-2 curve in the calculated ECD (Fig. 3), suggesting that the absolute configuration at C-3 of compound 2 was proposed as S. On the basis of the spectral data, compound 2 was identified as 3,5,6,7tetramethoxy-2,3-dihydro-1,4-phenanthrenequinone named cymensifin E.

Compound 3 was obtained as a yellow amorphous solid. The molecular formula $C_{18}H_{16}O_6$ was analyzed from its $[M + H]^+$ at m/z 329.1020 (calcd for $C_{18}H_{17}O_6$ at 329.1020) in the HR-ESI-MS. Moreover, it revealed a major peak at m/z 679.1783, which corresponded with the calculated 2M peak, $[2M + Na]^+$ at m/z679.1791. The UV absorptions and IR bands of 3 are similar to 1 and 2, suggesting of a phenanthrenequinone skeleton. The 1H, ¹³C NMR, and DEPT spectra of 3 also exhibited similarities with those of 1, except that the methoxy group is at C-3 in compound 3 instead of C-2 in compound 1 (Table 1). Compared with compound 1, the downfield chemical shift of C-3 in compound 3 ($\delta_{\rm C}$ 163.5) was observed due to the presence of a methoxy group at that position. The position of the methoxy group ($\delta_{\rm H}$ 4.00) at C-3 was confirmed by its HSQC correlation with MeO-3 carbon ($\delta_{\rm C}$ 56.2), the HMBC correlations of C-3 with H-2 ($\delta_{\rm H}$ 6.13) and MeO-3 and the NOESY correlation of H-2 and MeO-3. Similar to compound 1, the HMBC correlation of MeO-7 with C-7 ($\delta_{\rm C}$ 155.7), NOESY correlation with H-8 ($\delta_{\rm H}$ 7.32) and HSQC correlation with MeO-7 carbon ($\delta_{\rm C}$ 60.5) confirmed the substitution of methoxy ($\delta_{\rm H}$ 4.05) at C-7. The methoxy group ($\delta_{\rm H}$ 3.92) was located at C-6 based on the HMBC correlations of C-6 ($\delta_{\rm C}$ 143.9) with MeO-6 protons and H-8 ($\delta_{\rm H}$ 7.32), as well as its HSQC correlation with the MeO-6 carbon ($\delta_{\rm C}$ 60.1). The MeO-5 ($\delta_{\rm H}$ 4.05) was verified at C-5 by its HMBC correlation with C-5 ($\delta_{
m C}$ 149.9) and HSQC correlation with MeO-5 carbon ($\delta_{\rm C}$ 55.6). Based on the above spectral data, compound 3 was characterized as 3,5,6,7-tetramethoxy-1,4-phenanthrenequinone. Prior to

Table 1 1 H (400 MHz) and 13 C NMR (100 MHz) spectral data of 1–3

	1 (in acetone- d_6)		2 (in CDCl ₃)		3 (in acetone- d_6)	
Position	1 H (multiplicity, J in Hz)	¹³ C	1 H (multiplicity, J in Hz)	¹³ C	1 H (multiplicity, J in Hz)	¹³ C
1	_	179.4	_	192.5	_	183.7
2	_	158.2	3.42 (dd, 17.6, 6.0) 3.15 (dd, 17.6, 8.4)	46.8	6.13 (s)	106.1
3	6.29 (s)	110.1	4.67 (dd, 8.4, 6.0)	82.2	_	163.5
4	_	185.4	_	201.3	_	181.6
4a	_	134.3	_	137.3	_	132.5
4b	_	119.8	_	120.1	_	119.8
5	_	150.5	_	149.6	_	149.9
6	_	143.8	_	142.9	_	143.9
7	_	156.2	_	156.2	_	155.7
8	7.29 (s)	103.0	7.02 (s)	102.8	7.32 (s)	103.1
8a	_	135.2	_	134.8	_	134.4
9	8.01 (d, 8.8)	131.2	7.84 (d, 8.8)	130.7	8.06 (d, 8.8)	132.1
10	7.95 (d, 8.8)	120.7	7.94 (d, 8.8)	121.9	7.95 (d, 8.8)	120.8
10a	_ ` `	128.9	_ ` `	133.1	_ ` `	131.0
MeO-2	3.94 (s)	55.7	_	_	_	_
MeO-3	_	_	3.64 (s)	58.0	4.00 (s)	56.2
MeO-5	4.05 (s)	55.6	4.11 (s)	60.4	4.05 (s)	55.6
MeO-6	3.91 (s)	60.0	3.95 (s)	60.8	3.92 (s)	60.1
MeO-7	4.06 (s)	60.8	4.05 (s)	56.1	4.05 (s)	60.5

this study, the natural occurrence of **3** was not known. So, the trivial name cymensifin F was given to this compound. However, the synthesis of compound **3** was previously reported by C. L. Lee *et al.*, 2012.³⁰

Cytotoxicity profiles of the test compounds in BV2 microglial cells

A series of concentrations of isolated compounds from the roots of *Cymbidium ensifolium* underwent testing in BV2 microglial cells. Treatment with compounds over a range of concentrations $(0, 5, 10, 20, 40, 80 \mu M)$. The concentration that did not

cause toxicity to the cells was considered a safety concentration for the cells. In the present study, concentrations below 20 μ M were observed as the safety thresholds for all tested compounds and were subsequently selected for examination in activated microglial cells. Compounds 2 and 13 were not assessed due to insufficient amounts (Table 2).

The effects of the test compounds on nitric oxide release in activated microglial cells

Nitric oxide is one of the inflammatory mediators that plays an essential role in CNS diseases.³¹ Physiologically, nitric oxide acts

Table 2 Cytotoxicity profiles of isolated compounds from the roots of C. ensifolium in BV2 microglial cells^a

	Percentage viability (%)								
Compound	0 μΜ	5 μΜ	10 μΜ	20 μΜ	40 μΜ	80 μΜ			
1	100 ± 0	100.67 ± 3.33	101.51 ± 0.67	104.57 ± 7.09	101.75 ± 8.6	$73.72 \pm 18.53*$			
3	100 ± 0	99.88 ± 1.96	95.62 ± 7.44	94.93 ± 3.76	88.84 ± 6.76	$55.15 \pm 26.41**$			
4	100 ± 0	102.2 ± 3.53	99.34 ± 5.77	101.48 ± 2.3	95.34 ± 3.88	$49.15 \pm 24.65***$			
5	100 ± 0	105.43 ± 1.81	104.16 ± 7.68	100.52 ± 7.37	101.3 ± 14.9	94.45 ± 21.56			
6	100 ± 0	106.33 ± 4.65	108.08 ± 5.54	105.55 ± 7.76	$76.69 \pm 2.86**$	$44.78 \pm 13.23***$			
7	100 ± 0	105.04 ± 4.37	108.16 ± 3.16	112.38 ± 2.51	111.09 ± 1.88	103.99 ± 9.12			
8	100 ± 0	106.26 ± 0.36	100.28 ± 6.16	108.21 ± 0.53	113.19 ± 1.57	94.43 ± 21.2			
9	100 ± 0	101.76 ± 2.99	101.31 ± 4.09	104.51 ± 3.47	104.84 ± 5.59	111.8 ± 5.8			
10	100 ± 0	101.21 ± 2.85	96.63 ± 1.58	103.54 ± 7.79	95.79 ± 4.1	$85.14 \pm 9.87*$			
11	100 ± 0	102.94 ± 0.86	99.73 ± 1.55	98.77 ± 0.86	$94.8 \pm 2.82*$	$86.43 \pm 4.09***$			
12	100 ± 0	103.63 ± 3.27	101.21 ± 5.91	98.78 ± 4.41	94.28 ± 4.79	$87.5 \pm 8.17*$			
14	100 ± 0	107.82 ± 3.2	97.91 ± 8.91	91.92 ± 1.47	$65.89 \pm 4.08***$	$26.73 \pm 14.68***$			
15	100 ± 0	103.32 ± 1.97	105 ± 3.91	111.46 ± 2.26	107.65 ± 2.12	104.46 ± 3.05			
16	100 ± 0	104.25 ± 2.01	98.35 ± 3.48	96.41 ± 0.97	$44.5 \pm 11.83***$	$10.42 \pm 5.37***$			
17	100 ± 0	106.54 ± 6.44	$\textbf{97.31} \pm \textbf{7.01}$	100.77 ± 8.73	101.42 ± 5.33	$84.92 \pm 5.53*$			

^a Data are expressed as mean \pm SD (n=3). The differences in treatment with the vehicle group were statistically assessed using one-way ANOVA followed by Dunnett *post hoc* test. Statistically significant reductions in cell viability are denoted as *, **, *** representing significance levels of p < 0.05, p < 0.01, and p < 0.001, respectively.

Table 3 Profiles of the ability of isolated compounds from the roots of $Cymbidium\ ensifolium\ to\ diminish\ 50\%$ of nitric oxide release in LPS-induced BV2 microglial cells^a

Compound	$IC_{50}\pm SD\left(\mu M\right)$
1	9.95 ± 2.13
3	8.77 ± 3.78
4	>20
5	>20
6	2.39 ± 0.91
7	>20
8	>20
9	>20
10	>20
11	>20
12	6.69 ± 2.94
14	2.96 ± 1.38
15	>20
16	8.42 ± 2.99
17	>20

^a The data are expressed as mean \pm SD (n = 3).

to regulate neurotransmitter release and communication between neurons.³² However, an excessive amount of nitric oxide facilitates the progression of neurodegenerative diseases by promoting excitotoxic cell death and inducing mitochondrial dysfunction.³³ As such, alleviating nitric oxide is an available option to improve and delay the progression of CNS diseases. In the present study, a series of active compounds from the roots of *Cymbidium ensifolium* were assessed in LPS-induced microglial cells. As shown in Table 3, cymensifin D (1), cymensifin F (3), coelonin (6), parviphenanthrine C (12), (1-(4-hydroxylbenzyl)-4-methoxy 9,10-dihydrophenanthrene 2,7-diol) (14) and blestriarene C (16) demonstrated promising IC₅₀ (50%

inhibition of nitric oxide releases) values with the percentage nitric oxide inhibition is 9.95 \pm 2.13, 8.77 \pm 3.78, 2.39 \pm 0.91, 6.69 \pm 2.94, 2.96 \pm 1.38, and 8.42 \pm 2.99 μ M, respectively.

The effects of the test compounds on the expression of proinflammatory mediators (TNF- α , IL-6 and MCP-1) in activated microglial cells

Proinflammatory cytokines, including TNF-α and IL-6, are the major contributing factors in neuroinflammation and have been reported to contribute to the pathogenesis of CNS diseases, including Alzheimer's disease, Parkinson's disease, and other neuropsychiatric disorders.34,35 Microglia, CNSresident immune cells, are the cells that contribute to the significant release of proinflammatory cytokines and are pathophysiologically involved in the progression of diseases.36 In addition to cytokines, chemokines, including MCP-1, have been reported to cause neuronal loss and have also been reported to be involved in CNS-associated diseases.³⁷ Hence, exploring active compounds with the characteristic of suppressing proinflammatory mediators is essential for discovering neuroinflammation-modulating agents. These flammatory mediators are significantly elevated upon LPS stimulation.38 In the present study, the potential activity of the test compounds on modulating TNF-α, IL-6 and MCP-1 was assessed in LPS-stimulated BV2 cells. As shown in Fig. 4 and 5, LPS treatment increased the release of proinflammatory mediators, yet pretreatment with cymensifin D (1), cymensifin F (3), coelonin (6), parviphenanthrine C (12), (1-(4-hydroxylbenzyl)-4methoxy 9,10-dihydrophenanthrene 2,7-diol) (14) and blestriarene C (16) significantly reduced proinflammatory mediator secretions in activated microglia in a concentration-dependent manner. Coelonin (6), a dihydrophenanthrene compound isolated from Bletilla striata, has been reported for its anti-

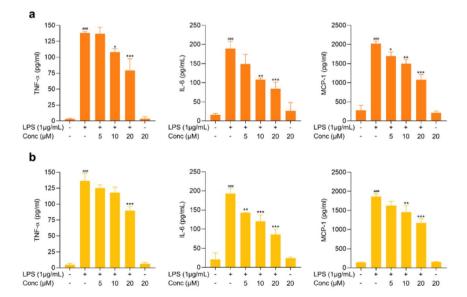
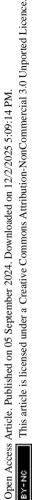


Fig. 4 The effects of new compounds, cymensifin D (a) and cymensifin F (b), on releases of TNF- α , IL-6, MCP-1 in LPS-induced BV2 microglial cells. The data are expressed as mean \pm SD (n=3). The differences between groups were statistically assessed using one-way ANOVA followed by Bonferroni post hoc test. ### denotes significant difference between vehicle and LPS group with statistical significance p<0.001. *, ***, *** denote significant differences of p<0.05, p<0.01, and p<0.001, respectively, between the LPS group and treatment groups.

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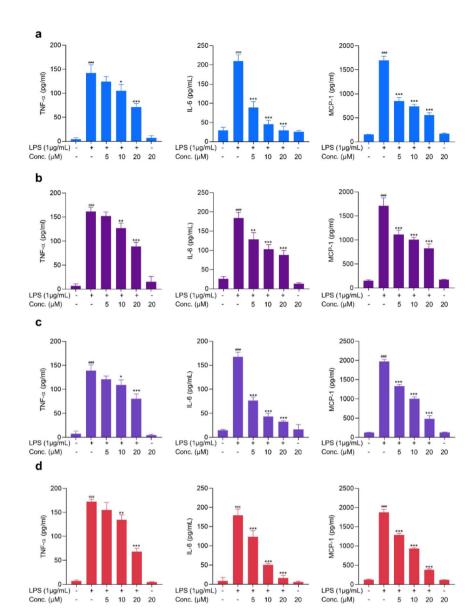


Fig. 5 The effects of coelonin (a), parviphenanthrine C (b), (1-(4-hydroxylbenzyl)-4-methoxy 9,10-dihydrophenanthrene 2,7-diol) (c) and blestriarene C (d) on releases of TNF- α , IL-6, MCP-1 in LPS-induced BV2 microglial cells. The data are expressed as mean \pm SD (n=3). The differences between groups were statistically assessed using one-way ANOVA followed by Bonferroni post hoc test. ### denotes significant difference between vehicle and LPS group with statistical significance p < 0.001. *, **, *** denote significant differences of p < 0.05, p < 0.01, and p < 0.001, respectively, between the LPS group and treatment groups.

inflammatory activity on LPS-induced RAW264.7 cells.39 Numerous studies have illustrated the remarkable antiinflammatory activity of phenanthrene, dihydrophenanthrene, derivatives.39,40 phenanthrenequinone Interestingly, compounds 1, 3, 6, 12, 14, and 16 belong to those categories. Taken together, these compounds possess the ability to modulate proinflammatory mediators in activated microglia and could potentially be further used as neuroinflammationmodulating agents.

Cymensifin F (3) modulates NF-кВ activation in LPS-treated microglia cells

To study the anti-inflammatory mechanism of the novel phenanthrenequinones, cymensifin F (3) was selected based on its IC₅₀ value (8.77 \pm 3.78 μ M). The potential anti-inflammatory mechanism of cymensifin F (3) was determined using immunoblotting. Activated microglia in CNS diseases, manifested with increased releases of proinflammatory mediators, are associated with the stimulation of the NF-κB pathway.41 NF-κB activation controls the production of TNF-α, IL-6, and MCP-1.6 Moreover, pro-inflammatory cytokines can be downregulated by inhibiting NF-κB transcriptional activity in the microglial nucleus.42 Consequently, we investigated the potential antiinflammatory effect of cymensifin F (3) on NF-κB activity. In our study, NF-κB (p65) was phosphorylated in response to LPS. Furthermore, pretreatment with cymensifin F (3) significantly suppressed LPS-induced NF-κB phosphorylation (Fig. 6). These findings suggest that the anti-neuroinflammatory properties of

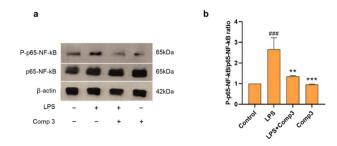


Fig. 6 Cymensifin F (**3**) inhibited LPS-induced NF- κ B activation by attenuating the phosphorylation of p65. (a) The expression of proteins was measured by western blot. (b) P-p65-NF- κ B/p65-NF- κ B ratio. Data show the mean \pm SD values of three independent experiments. ###p < 0.001 denotes significant compared to control. **p < 0.01, and ***p < 0.001 compared to LPS, using one-way ANOVA followed by Bonferroni post hoc test.

cymensifin F (3) may be attributed to its ability to inhibit NF- κB phosphorylation.

Experimental

General experimental procedures

Optical rotation was measured by a Jasco P-2000 digital polarimeter (Easton, MD, USA). UV spectra were captured using a Milton Roy Spectronic 3000 Array spectrophotometer (Rochester, Monroe, NY, USA). CD spectra were obtained from a Jasco J-815 CD spectrophotometer (Hachioji, Tokyo, Japan). FT-IR spectra were acquired using a PerkinElmer FT-IR 1760X spectrophotometer (Boston, MA, USA). Mass spectra were recorded by a Bruker MicroTOF mass spectrometer (ESI-MS) (Billerica, MA, USA). NMR spectra were obtained by a Bruker Avance Neo 400 MHz NMR spectrometer (Billerica, MA, USA). Vacuum-liquid chromatography (VLC) and column chromatography (CC) were performed by silica gel 60 (no. 1.07734.2500), size 0.063-0.200 mm, and (no. 1.09385.2500), size 0.040-0.063 mm (Merck, NJ, USA). Gel filtration chromatography was conducted by Sephadex LH-20 (Merck, NJ, USA). The initial assessment of the purity of isolated compounds was conducted by thin-layer chromatography (TLC), silica gel 60 F₂₅₄ plates (Merck, NJ, USA) under UV light. 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), lipopolysaccharide (LPS), and other chemicals were acquired from Sigma-Aldrich (St. Louis, MO, USA). NF-kB p65 (cat. no. 8252S), P-NF- κ B p65 (cat. no. 3033S), β -actin (cat. no. 4970S), and anti-rabbit antibody (cat. no. 7074S) were purchased from Cell Signaling Technology (Massachusetts, USA). Nitrocellulose membrane (0.45 µ) was obtained from HiMedia Laboratories Pvt. Ltd. (Mumbai, India). Clarity Western ECL Substrate (cat. no. 1705061) was acquired from Bio-Rad Laboratories, United States.

Plant material

The roots of *Cymbidium ensifolium* were purchased from Chatuchak Market in September 2020. The authentication was performed by Dr Boonchoo Sritularak. A voucher specimen BS-CE-092563 has been deposited at the Department of

Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Chulalongkorn University.

Extraction and isolation

Dried roots of Cymbidium ensifolium (1 kg) were macerated with methanol (MeOH) (5 \times 3 L), and the methanolic extract (92.2 g) was obtained. The methanolic extract was subsequently subjected to vacuum liquid chromatography (silica gel, EtOAchexane, gradient) to give seven fractions (A-G). Fraction B (1.5 g) was separated by Sephadex LH-20 (acetone) to yield 5 fractions (BA-BE). Fraction BC (92 mg) was separated on silica gel eluting with dichloromethane (CH₂Cl₂) to yield compounds 4 (3 mg) and 5 (5.4 mg). Fraction BE (73.1 mg) was separated on a silica gel column (acetone-CH₂Cl₂, 0.2:9.8) to yield compound 6 (3.2 mg). Fraction BD (76.4 mg) was fractionated on a silica gel column (acetone-CH2Cl2, 0.2:9.8) to give 2 fractions (BDA-BDB). BDB (30 mg) was purified on Sephadex LH-20 (acetone) to yield compound 7 (5.2 mg). Fraction C was separated on a silica gel column (acetone-CH2Cl2, 0.5:9.5), and six fractions (CA-CF) were obtained. CA fraction (261.9 mg) was separated on Sephadex LH-20 (acetone) to yield compound 8 (5.9 mg) and CAA fraction which was separated again on a silica gel column (acetone-hexane, 2.5:7.5) to produce compound 9 (1.3 mg). The CC fraction (103.7 mg) was separated on Sephadex LH-20 (acetone) and then silica gel (acetone-CH2Cl2, 0.2:9.8) to produce three fractions (CCA-CCC) fractions. CCA fraction (10 mg) underwent purification five times on preparative thin layer chromatography (PTLC) using dichloromethane as the mobile phase and compound 1 (3 mg) was produced. CCB fraction (3 mg) was purified on PTLC (acetone-hexane, 2:8, 2 times) to furnish compound 2 (0.8 mg). Fraction CCC (103.7 mg) was separated on Sephadex LH-20 (acetone) and then purified on purified on PTLC (EtOAc-CH₂Cl₂, 0.2:9.8) to produce compound 3 (2.3 mg). Fraction CD (389.3 mg) was purified on Sephadex LH20 (methanol), and then Sephadex LH-20 (acetone) to yield CDA (22.6 mg) which was further separated on a silica gel column (acetone-CH2Cl2, 1:9) to produce compound 10 (1.5 mg). Fraction CF (131.7 mg) was separated on Sephadex LH-20 (acetone) and 2 fractions (CFA and CFB) were obtained. CFA (40 mg) was separated on a silica gel column (EtOAc-CH₂Cl₂, 2: 8) and then purified again on Sephadex LH-20 (methanol) to give compound 11 (3.4 mg). CFB (29 mg) was separated on a silica gel column (MeOH-CH₂Cl₂, 0.1:9.9) and then purified again on PTLC by developing four times with dichloromethane to yield compound 12 (1.7 mg). Fraction D (1.3 g) was separated on a silica gel column (acetone-CH2Cl2) with gradient elution, and 10 fractions (DA-DJ) were obtained. Fraction DA (17 mg) was separated on a silica gel column by eluting with dichloromethane to produce compound 13 (0.8 mg). Compound 14 (1.5 mg) was obtained by purifying DI fraction (87 mg) with Sephadex LH-20 (acetone) and then Sephadex LH-20 (methanol). Compound 15 (3 mg) was produced by purifying DJ (157.3 mg) with Sephadex LH-20 (methanol). Fraction E (1.5 g) was separated on a silica gel column (acetone-CH₂Cl₂, 0.5:9.5), and 5 fractions (EA-EE) were obtained. Fraction EE (137.9 mg) was purified on Sephadex LH-20 (acetone) to produce EEA and EEB.

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Fraction EEB (15.3 mg) was separated again on Sephadex LH-20 (methanol) to furnish compound 16 (1.7 mg). Fraction F (2.8 g) was separated on a silica gel column (EtOAc-hexane) with gradient elution to give three fractions (FA-FC). Compound 17 (2.7 mg) was obtained from fraction FC (10 mg) by purifying with Sephadex LH-20 (acetone) and then Sephadex LH-20 (methanol).

Cymensifin D (1): orange-coloured amorphous solid; UV (MeOH) λ_{max} (log ε) 210 (3.96), 240 (4.10), 305 (3.85), 395 (3.04); HR-ESI-MS: $[M + H]^+$ at m/z 329.1019 (calcd for $C_{18}H_{17}O_6$ 329.1020); IR: ν_{max} at 3396, 2922, 1726, 1461, 1376 cm⁻¹.

Cymensifin E (2): pale yellow amorphous solid; $[\alpha]_D^{20}$ –21.5 (c 0.04, MeOH); UV (MeOH) λ_{max} (log ε) 205 (3.92), 230 (3.83), 280 (3.92), 332 (3.22); HR-ESI-MS; $[M + H]^+$ at m/z 331.1176 (calcd for $C_{18}H_{19}O_6$ 331.1181); ECD (MeOH): $\lambda_{max}(\Delta \varepsilon)$ 202.7 (-1.61), 215.0 (+1.52) nm; IR: ν_{max} at 3333, 2919, 2923, 1717, 1679, 1464, 1316 cm⁻¹.

Cymensifin F (3): yellow amorphous solid; UV (MeOH) λ_{max} $(\log \varepsilon)$ 220 (3.91), 245 (4.09), 300 (3.77), 390 (3.01); HR-ESI-MS: $[M + H]^+$ at m/z 329.1020 (calcd for $C_{18}H_{17}O_6$ 329.1020); IR: $\nu_{\rm max}$ at 3333, 2921, 1732, 1470, 1378 cm⁻¹.

Computational detail of ECD calculation

In this study, computational approaches were employed to optimize various configurations of compound 2. Density Functional Theory (DFT) calculations were conducted at the B3LYP/6-31g(d,p) level. Subsequently, Electron Circular Dichroism (ECD) spectra were computed using time-dependent DFT (TD-DFT) at the B3LYP/6-31+G(d,p) level, incorporating solvation effects modeled with the Continuum Model (PCM) employing methanol. All computational analyses were performed using Gaussian16 software.43 Additionally, the ECD spectra were simulated using overlapping Gaussian functions parameterized with a fitting parameter ($\sigma = 0.25$ eV) via the SpecDis 1.64 program, 44 utilizing the length gauge representation to enhance reliability.

Cell culture

BV2 microglial cells were obtained from AcceGen Biotechnology (Fairfield, NJ, USA). The cells were grown in DMEM media (PAN Biotech, Aidenbach, Germany) and supplemented with 10% fetal bovine serum (PAN Biotech, Aidenbach, Germany), 1% antibiotics (penicillin-streptomycin) (GIBCO, Carlsbad, CA, USA). The cells were kept at 37 °C, 5% CO_2 , and 95% humidity. The cells were passaged at 70-80% confluence.

Cell treatment

For cytotoxicity, the cells were seeded at a concentration of 2×10^4 cells per well in 96-well plates for 24 hours. The cells were then treated with various concentrations of the test compounds (0, 5, 10, 20, 40 and 80 μM) for another 24 hours. The selected safety concentrations were further used to evaluate the efficacy of the test compounds on proinflammatory mediator release in LPSstimulated cells. Briefly, cells were pretreated with the safety concentrations of the test compounds for 1 hour followed by the addition of 1 µg per mL LPS for 24 hours. The media was then collected for further assessment of the proinflammatory mediators

(NO, TNF-a, IL-6, and MCP-1). In the western blot, the cells were pretreated with the highest safety dose of the test compound for 1 hour and exposed to LPS for another 1 hour. The cells were collected and extracted for further use in the western blot.

Cell viability assay

The cytotoxicity of the test compound was determined by assessing the viability of the cells using the MTT assay. After treatment, the cells were added with MTT solution at a concentration of 500 µg mL⁻¹ for 3 hours. The crystal formazan was then dissolved with DMSO, followed by the measurement of the absorbance in the microplate reader. The absorbance was determined at a wavelength of 570 nm.

Nitrite assay

The levels of nitric oxide were measured using a nitrite assay. Griess reagent (Sigma-Aldrich®, St. Louis, MO, USA) was used to measure nitrite levels in the media as an indicator of nitric oxide releases. Briefly, after treatment, 100 µL of media was mixed with 100 μL Griess reagent and incubated for 10 min. The absorbance was then measured on a microplate reader at a wavelength of 520 nm.

ELISA assay

Released inflammatory mediators, including TNF-α, IL-6, and MCP-1, after treatment with the test compounds in activated microglia were determined using an ELISA assay (Biolegend, San Diego, USA). The assays were performed according to the manufacturer's instructions.

Western blot analysis

The cells were seeded at a density of 2×10^6 cells in 6 cm cell culture dishes. Following the treatments, the cell culture medium was removed, and the cells were washed with PBS. The cells were then added with 60 µL of lysis buffer (supplemented with 1% protease/phosphatase inhibitor) to the cells to cause lysing. The cell lysates were collected using scrapers, centrifuged (4 °C, 12 000 rpm, 10 min), and the supernatants were collected. A Bradford assay was used to calculate each lysate's protein content. An equal amount of proteins (40 µg) were loaded and electrophoresed on 10% sodium dodecyl sulfatepolyacrylamide gels (SDS-PAGE). A semi-dry transfer machine was used to transfer proteins onto nitrocellulose membranes, which were then treated with 5% skim milk in TBST for 1 hour at room temperature with gentle agitation in order to prevent non-specific binding sites. Following an overnight incubation at 4 °C with primary antibodies; P-p65-NF-κB (1:1000), p65-NF-κB (1:1000) and β -actin (1:2000), the membranes were treated for 2 hours at room temperature with horseradish peroxidaselabeled anti-rabbit secondary antibodies (1:1000). After washing with TBST, the chemiluminescence (ECL) solution was allowed to incubate on the membranes. Subsequently, a medical X-ray cassette, Kodak Green 400 Screen (Rochester, NY, USA), was used to observe the protein bands. ImageJ software was used to quantify the intensity of the protein bands.

Data and statistical analysis

The data were analyzed and visualized using GraphPad Prism (San Diego, CA, USA). The data shown in the graphs are expressed as means \pm SD. The statistical differences between groups were analyzed using a one-way ANOVA followed by Dunnett, Bonferroni *post hoc* test. A *p*-value below 0.05 was considered statistically significant.

Conclusions

In the present study, a total of 17 compounds were extracted from the roots of Cymbidium ensifolium, including two new phenanthrenequinones (1-2), one new natural product phenanthrenequinone (3), and 14 known compounds (4-17). The isolated pure compounds were assessed for their antineuroinflammatory properties using lipopolysaccharideactivated BV2 microglial cells. Compounds 1, 3, 6, 12, 14, and 16 demonstrated the capability to decrease lipopolysaccharideinduced nitric oxide (NO) release in BV2 microglial cells. Additionally, they reduced the secretion of proinflammatory mediators such as TNF-α, IL-6, and MCP-1 in activated microglia in a concentration-dependent manner. Cymensifin F (3) further investigated for its possible neuroinflammatory mechanism. In LPS-stimulated BV2 microglia, the anti-inflammatory effect of cymensifin F (3) is mediated through the downregulation of phosphorylated NF-κB. All of these findings suggest that C. ensifolium roots contain bioactive compounds that may alleviate neuroinflammation and aid in the treatment of neurodegenerative diseases.

Data availability

The authors confirm that the data supporting the findings of this study are available within the article and its ESI.†

Author contributions

B. S. conceived, designed, and supervised the research project, as well as revised the manuscript; P. T. supervised the antineuroinflammatory activity and revised the manuscript; H. H. supervised and assisted on anti-neuroinflammatory activity as well as prepared the manuscript; M. T. T. performed the experiments and prepared the manuscript; P. P. and S. J. analyzed the ECD experimental data; P. R. and C. B. provided comments and suggestions on the preparation of the manuscript and performing the experiments.

Conflicts of interest

The authors declare no conflict of interest.

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