


 Cite this: *RSC Adv.*, 2024, 14, 14697

 Received 14th April 2024  
 Accepted 29th April 2024

DOI: 10.1039/d4ra02773f

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# Photoredox-catalyzed sulfonylation of diaryliodonium salts with DABSO and silyl enolates involving the insertion of SO<sub>2</sub>†

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A versatile photoredox-catalyzed three-component sulfonylation of diaryliodonium salts with DABSO and silyl enolates involving the insertion of SO<sub>2</sub> was developed. Moreover, by employing β-alkyl substituted silyl enolates as substrates, the sulfonylation would give α-alkyl substituted β-keto sulfones, which are difficult to accessed by previous method involving the insertion of SO<sub>2</sub>.

β-Keto sulfones are privileged frameworks in many pharmaceuticals and display remarkable biological activities and pharmacological properties such as anti-bacterial, anti-fungal, anti-hepatitis, and non-nucleoside inhibition (Fig. 1).<sup>1</sup> Additionally, β-keto sulfones are utilized as versatile intermediate synthons in diverse synthetic transformations due to the simultaneous existence of multiple functional groups including carbonyl, sulfonyl and active methylene moieties.<sup>2</sup> Therefore, there is high demand for developing efficient synthesis methods to construct β-keto sulfones.

Conventional approaches for the synthesis of β-keto sulfones include oxidation of 2-oxo-sulfides with strong oxidants<sup>3</sup> and the sulfonylation of α-halo-ketones with preinstalled sulfonyl-containing segments,<sup>4</sup> which have several drawbacks, such as the employment of poorly accessible and smelly organosulfur compounds as starting materials, limited substrate applicability, and harsh conditions. In recent years, sulfonylation involving the insertion of sulfur dioxide *via* a radical process has emerged as a powerful method to access sulfone derivatives,<sup>5</sup> in which various SO<sub>2</sub> surrogates, like DABSO, metabisulfites, rongalite reagents and SOgen, are employed instead of SO<sub>2</sub> gas.<sup>6</sup> Outstanding contributions to synthesize diverse β-keto sulfones from silyl enolates *via* sulfur dioxide insertion have been made by Wu's groups using aryldiazonium tetrafluoroborates,<sup>7</sup> aryl/alkyl halides,<sup>8</sup> and thianthrenium salts<sup>9</sup> as active radical precursors in some cases. These transformations proceed with good tolerance of functional groups, easily enabling the incorporation of various sulfonyl skeletons into ketones.

On the other hand, diaryliodonium salts can participate in various transformations as a type of versatile, easily available, non-toxic, environmentally, and air stable solid arylating reagent.<sup>10</sup> So far, diaryliodonium salts as active radical precursors in multi-component sulfonylation reactions involving the insertion of sulfur dioxide has been developed initially. Jiang and coworkers reported a straightforward protocol for the synthesis of diverse functionalized diarylannulated sulfones through SO<sub>2</sub>/I exchange of iodonium(III) salts. In this reaction, the aryl radical generated from diaryliodonium salt was captured by Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub> to form a SO<sub>2</sub> radical anion, which served as the key intermediate to realize the exchange strategy (Scheme 1a).<sup>11</sup> In 2018, Manolikakes<sup>12</sup> and Zhang<sup>13</sup> demonstrated that visible-light-induced reduction of diaryliodonium salts would generate aryl radicals, which could undergo cascade cyclization in combination with SO<sub>2</sub> for the synthesis of sulfonylated coumarins (Scheme 1b) and 3-arylsulfonylquinoline derivatives, respectively. Very recently, Piguel and coworkers established the first visible-light photoredox catalyzed C–H sulfonylation of imidazopyridines with diaryliodonium salts and DABSO for the straightforward synthesis of novel C-3

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† Electronic supplementary information (ESI) available: Experimental procedures, analytical data for products, NMR spectra of products. See DOI: <https://doi.org/10.1039/d4ra02773f>

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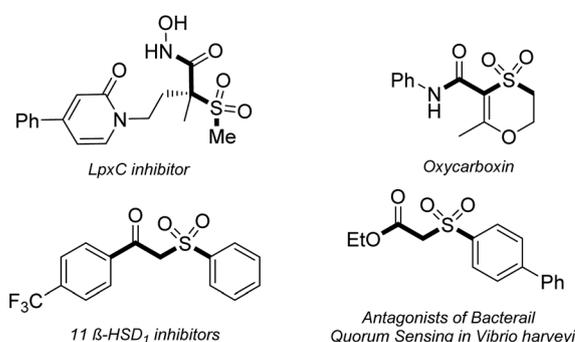
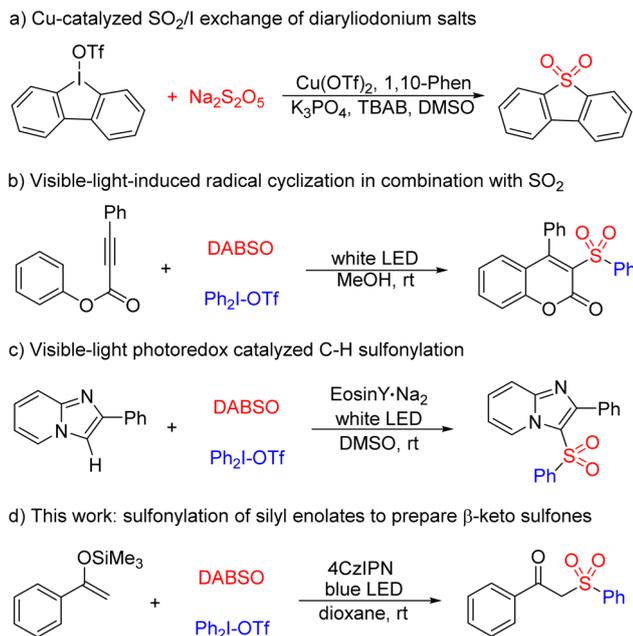


Fig. 1 Examples for β-keto sulfones and related sulfones with biological activities.





Scheme 1 Diaryliodonium salts as active radical precursors in the insertion of sulfur dioxide.

sulfonated imidazoheterocycles (Scheme 1c).<sup>14</sup> Herein, we would like to report a novel 4CzIPN-catalyzed three-component sulfenylation of diaryliodonium salts with DABSO and silyl enolates *via* a radical process involving SO<sub>2</sub> insertion. This photoredox catalysis with the assistance of visible light represents a green and sustainable approach for the synthesis of β-keto sulfones (Scheme 1d).

Trimethyl((1-phenylvinyl)oxy)silane **1a**, diphenyliodonium triflate (**2a**) and 1,4-diazabicyclo[2.2.2]octane bis(sulfur dioxide) (DABSO) were selected as starting materials to optimize the sulfenylation conditions. Initially, irradiation of all three reaction partners in 1,4-dioxane with blue LED (25 W) at room temperature for 12 hours under N<sub>2</sub> led to the desired β-keto sulfone **3a** in 47% yield (Table 1, entry 1). Other SO<sub>2</sub> surrogates such as K<sub>2</sub>S<sub>2</sub>O<sub>5</sub>, Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>, Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub> displayed lower activity (Table 1, entries 2–4). A variety of solvents were examined. 1,4-Dioxane proved to be superior to tetrahydrofuran (THF), acetonitrile, 1,2-dichloroethane (DCE), *N,N*-dimethylformamide (DMF) and toluene (Table 1, entries 1 and 5–10). When SO<sub>2</sub>-MeCN solution was used as SO<sub>2</sub> source instead of DABSO, the reaction mixture in MeCN gave **3a** in a slightly increased yield (Table 1, entries 6 and 7). Next, a range of experiments were carried out under various light sources such as blue LED (10 W, 25 W, 30 W), white LED (24 W), compact fluorescent lamp (CFL, 18 W), Kessil light (390 nm) and UV (600 W). As a result, 25 W of blue LED has the best performance (Table S1 in ESI<sup>†</sup>). Despite extending the time to 24 h, only 55% yield of **3a** can be obtained (Table S1,† entry 8). The addition of photoredox catalysts was beneficial for the transformation. For example, the yield of **3a** sharply rose to more than 80% when 2.0 mol% of either [Ir{dFCF<sub>3</sub>ppy}<sub>2</sub>(bpy)]PF<sub>6</sub> or 1,2,3,5-tetrakis(carbazol-9-yl)-4,6-dicyanobenzene (4CzIPN) was added (Table 1, entries 11 and 14). Considering cost and availability

Table 1 Optimization of three-component sulfenylation<sup>a</sup>

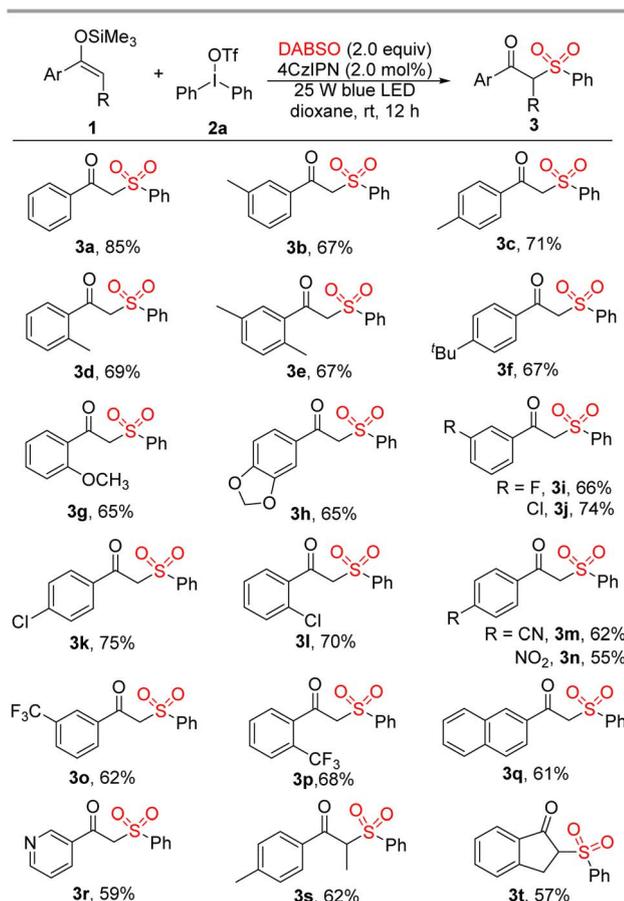
Entry	[SO <sub>2</sub> ]	PC	Solvent	Yield <sup>b</sup> (%)
1	DABSO	—	1,4-Dioxane	47
2	K <sub>2</sub> S <sub>2</sub> O <sub>5</sub>	—	1,4-Dioxane	35
3	Na <sub>2</sub> S <sub>2</sub> O <sub>5</sub>	—	1,4-Dioxane	24
4	Na <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	—	1,4-Dioxane	0
5	DABSO	—	THF	34
6	DABSO	—	MeCN	45
7	SO <sub>2</sub> solution	—	MeCN	50
8	DABSO	—	DCE	37
9	DABSO	—	PhMe	0
10	DABSO	—	DMF	0
11	DABSO	<b>Ir-1</b>	1,4-Dioxane	84
12	DABSO	<b>Ru-1</b>	1,4-Dioxane	Trace
13	DABSO	Eosin Y	1,4-Dioxane	56
14	DABSO	4CzIPN	1,4-Dioxane	85
15 <sup>c</sup>	DABSO	4CzIPN	1,4-Dioxane	60
16 <sup>d</sup>	DABSO	4CzIPN	1,4-Dioxane	0

<sup>a</sup> Reaction conditions: **1a** (2.0 equiv.), **2a** (0.1 mmol, 1.0 equiv.), [SO<sub>2</sub>] (2.0 equiv.), PC (2 mol%), and *n*-C<sub>12</sub>H<sub>26</sub> (10 μL) in solvent (1.0 mL) at room temperature for 12 h, irradiation with 25 W blue LED, under N<sub>2</sub>. <sup>b</sup> GC yields. <sup>c</sup> Under air atmosphere. <sup>d</sup> No light. **Ir-1** = [Ir{dFCF<sub>3</sub>ppy}<sub>2</sub>(bpy)]PF<sub>6</sub>, **Ru-1** = [Ru(bpy)<sub>3</sub>]Cl<sub>2</sub>·6H<sub>2</sub>O. SO<sub>2</sub> solution: SO<sub>2</sub> solution in MeCN (~7.9 M).

factors, 4CzIPN was chosen as the optimal photocatalyst. When the reaction was carried out under air atmosphere, 60% yield of **3a** was generated (Table 1, entry 15). In addition, the three-component sulfenylation cannot proceed without light (Table 1, entry 16). In addition, the introduction of various bases, such as Na<sup>t</sup>Bu, NaOH, Na<sub>2</sub>CO<sub>3</sub>, NaHCO<sub>3</sub> and NEt<sub>3</sub>, has caused the reaction yield to decrease to varying degrees (Table S2 in ESI<sup>†</sup>).

We investigated the scope of this three-component sulfenylation transformation with the optimal conditions in hand. The treatment of diverse silyl enolates **1** with diphenyliodonium triflate **2a** and DABSO afforded the corresponding β-keto sulfone derivatives in moderate to good yields (Scheme 2). Silyl enolates possessing methyl and alkoxy groups on the phenyl ring could react with **2a** and DABSO smoothly to give the desired products (**3b–3h**). Halogen atoms, especially chlorine, can be compatible under the optimal conditions (**3i–3l**), which revealed possible further transformations of the resulting halogenated products with other nucleophiles.<sup>15</sup> Representative electron-withdrawing groups, such as cyano, nitro, and trifluoromethyl, were also tolerated in this three-component sulfenylation transformation under the optimal conditions (**3m–3p**). When trimethyl((1-(naphthalen-2-yl)vinyl)oxy)silane was treated with **2a** and DABSO, the desired **3q** was isolated in 61% yield. In addition, heterocyclic substituted silyl enolate could also be transformed to **3r** in 59% yield. It is notable that α-substituted β-keto sulfones cannot be obtained from reported radical transformations involving the insertion of SO<sub>2</sub>.<sup>7–9</sup> To our



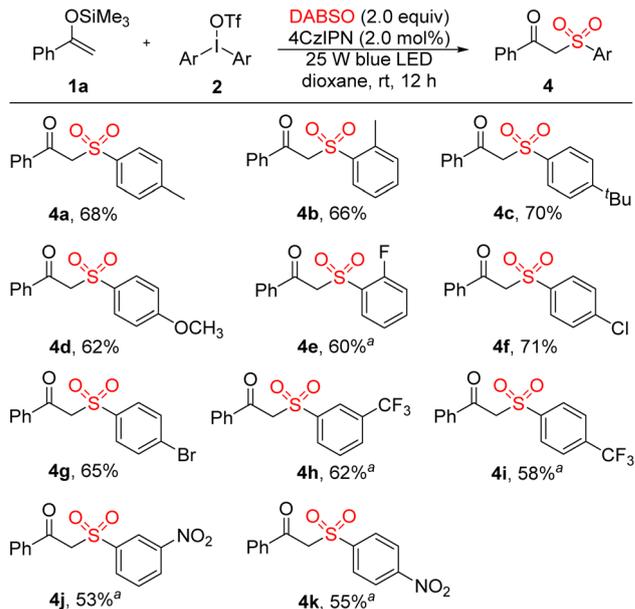
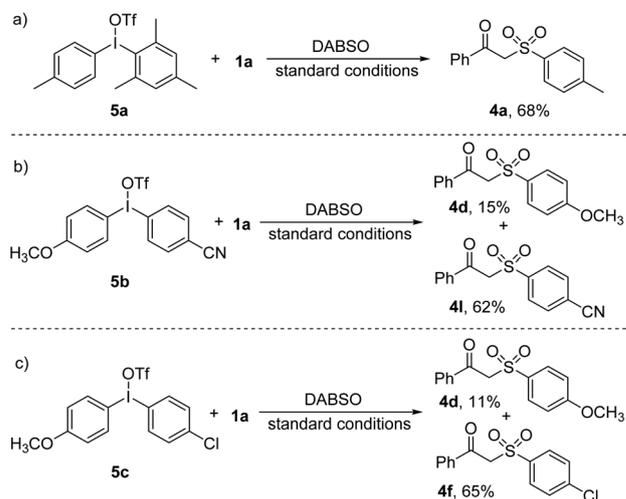


Scheme 2 The scope of silyl enolates.

delight,  $\beta$ -alkyl substituted silyl enolates could work well in this three-component sulfonylation to give  $\alpha$ -alkyl substituted  $\beta$ -keto sulfones **3s** and **3t** in acceptable yields.

Subsequently the scope of diaryliodonium salts was examined. As shown in Scheme 3, both diaryliodonium triflates and diaryliodonium tetrafluoroborates were suitable substrates for this photocatalytic reaction. Diaryliodonium salts bearing different functional groups, containing electron-donating methyl (**4a** and **4b**), *tert*-butyl (**4c**), methoxy (**4d**) and electron-withdrawing fluorine (**4e**), chlorine (**4f**), bromine (**4g**), trifluoromethyl (**4h** and **4i**) and nitro (**4j** and **4k**), were readily compatible in this 4CzIPN-catalyzed three-component sulfonylation.

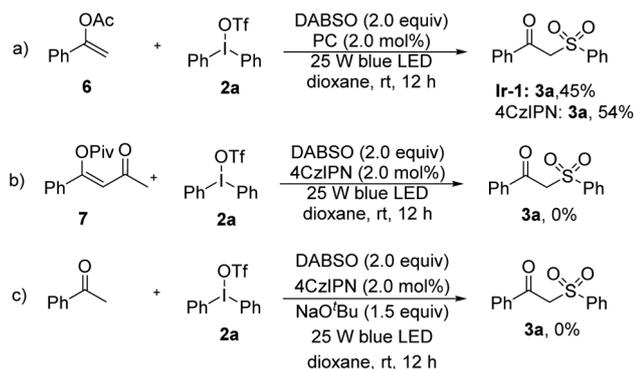
We next investigated the 4CzIPN-catalyzed three-component sulfonylation with unsymmetrical diaryliodonium salts. When unsymmetrical salt possessing a *p*-methylphenyl and a bulky mesityl group was used, only **4a** was obtained in 68% *via* a selective transfer of the *p*-methylphenyl group (Scheme 4a). The reaction of **5b** afforded methoxy substituted sulfone **4d** and cyano substituted sulfone **4l** in 15% and 62%, respectively (Scheme 4b). When unsymmetrical diaryliodonium salt **5c** reacted with **1a** and DABSO, the  $\beta$ -keto sulfone **4f** was isolated as the main product in 65% with selective transfer of the electron-deficient *p*-chlorophenyl group over the electron-rich *p*-methoxyphenyl moiety (Scheme 4c). These results indicate that

Scheme 3 The scope of diaryliodonium salts.<sup>a</sup> Diaryliodonium tetrafluoroborates was used.Scheme 4 Reactions with unsymmetrical diaryliodonium salts: (a) the sulfonylation with *p*-tolyl-mesityl iodonium salt **5a**; (b) the sulfonylation with (*p*-cyanophenyl)-(*p*-methoxyphenyl) iodonium salt **5b**; (c) the sulfonylation with (*p*-chlorophenyl)-(*p*-methoxyphenyl) iodonium salt **5c**.

the electron-deficient aryl moiety is more easily reduced by photocatalyst to generate an aryl radical.

The scope of enolates in addition to silyl enolates was checked. 1-Phenylvinyl acetate **6** can afford the desired product in moderate yield in the presence of either **Ir-1** or 4CzIPN (Scheme 5a), while 1,3-diketone derived enol ester **7** was not compatible (Scheme 5b). Under the standard conditions, the combination of acetophenone and sodium *tert*-butoxide was adopted, which can generate enolate *in situ* to replace the enol silane. As a result, the generation of **3a** was not observed (Scheme 5c).

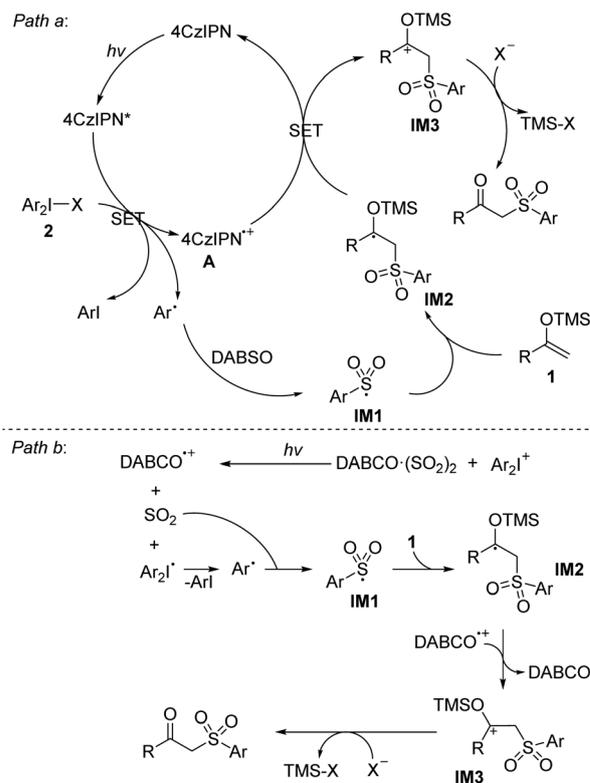




**Scheme 5** The scope of other enolates: (a) the sulfonation with 1-phenylvinyl acetate **6**; (b) the sulfonation with 1,3-diketone derived enol ester **7**; (c) the sulfonation with acetophenone in the presence of sodium *tert*-butoxide.

To gain more insight into the mechanism, the following radical inhibition experiments were performed. The standard sulfonation transformation was completely restrained in the presence of 3.0 equiv. of 2,2,6,6-tetramethylpiperidine 1-oxyl (TEMPO) (Scheme 6a). At the same time, the yield of **3a** decreased to 47% when 3.0 equiv. of butylated hydroxytoluene (BHT) was added in the reaction system (Scheme 6b). These results implied that a radical process might be involved in the mechanism.

Based on the above observations and previous work, plausible mechanism of the photoredox-catalyzed sulfonation involving the insertion of  $\text{SO}_2$  is proposed. For the pathway in the presence of photocatalyst (Scheme 7, path a):<sup>9</sup> The photoexcited 4CzIPN\* reduces diaryliodonium salt **2** through single electron transfer (SET) process to give an aryl radical and the oxidized photocatalyst **A**. Aryl radical is trapped by  $\text{SO}_2$  to generate the sulfonyl radical **IM1**. Then, the addition of sulfonyl radical **IM1** into the double bond of silyl enolate **1** affords radical intermediate **IM2**, which can be oxidized to cation intermediate **IM3** by photocatalyst **A** *via* another SET along with the regeneration of the photocatalyst 4CzIPN. Finally, cation intermediate **IM3** undergoes desilylation with nucleophilic anion species to give the desired  $\beta$ -keto sulfone **3**. For the pathway in the absence of photocatalyst (Scheme 7, path b):<sup>12</sup> the interaction of iodonium salt with DABSO ( $\text{DABCO} \cdot (\text{SO}_2)_2$ ) would produce DABCO radical cation, dioxide, and diaryl iodine radical. Fragmentation of diaryl iodine radical furnishes an aryl



**Scheme 7** Plausible mechanistic pathway.

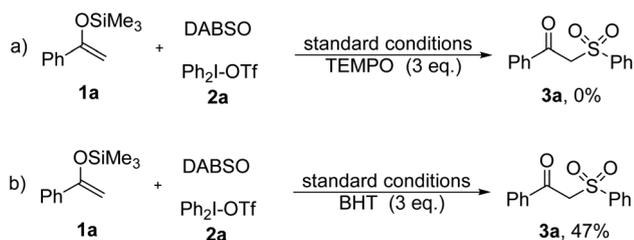
radical, which is trapped by  $\text{SO}_2$  to generate the sulfonyl radical **IM1**. Then, the addition of sulfonyl radical **IM1** into the double bond of silyl enolate **1** affords radical intermediate **IM2**, which can be oxidized to cation intermediate **IM3** by DABCO radical cation. Finally, cation intermediate **IM3** undergoes desilylation to give **3**.

## Conclusions

In summary, a versatile strategy for the synthesis of diverse  $\beta$ -keto sulfones *via* photoredox-catalyzed sulfonation of diaryliodonium salts with DABSO and silyl enolates involving the insertion of  $\text{SO}_2$  has been established. In this reaction, blue-light-induced reduction of diaryliodonium salts afford the aryl radical, which would be trapped by  $\text{SO}_2$  to form sulfonyl radical as the key intermediate. This novel photoredox catalysis with the assistance of visible light represents a green and sustainable approach for the synthesis of  $\beta$ -keto sulfones and features wide substrate scope, good reactivity, and broad functional group tolerance. In addition, by employing  $\beta$ -alkyl substituted silyl enolates, this three-component sulfonation would give  $\alpha$ -alkyl substituted  $\beta$ -keto sulfones, which cannot be accessed by previous method involving the insertion of  $\text{SO}_2$ .

## Conflicts of interest

There are no conflicts to declare.



**Scheme 6** Parallel control experiments: (a) radical inhibition experiment with TEMPO; (b) radical inhibition experiment with BHT.



## Acknowledgements

This work was supported by grants from the National Natural Science Foundation of China (No. 22101259) and the Zhejiang Provincial Natural Science Foundation of China (No. LQ22B020002).

## References

- (a) A. Markham and S. J. Keam, *Drugs*, 2018, **78**, 1271–1276; (b) E. H. Fleming, E. E. Ochoa, J. E. Nichols, M. K. O'Banion, A. R. Salkind and N. J. Roberts Jr, *J. Med. Virol.*, 2018, **90**, 26–33; (c) J. Xiang, M. Ipek, V. Suri, M. Tam, Y. Xing, N. Huang, Y. Zhang, J. Tobin, T. S. Mansoura and J. McKew, *Bioorg. Med. Chem.*, 2007, **15**, 4396–4405; (d) C. Curti, M. Laget, P. Vanelle, C. Curti, M. Laget, A. Ortiz Carle, A. Gellis and P. Vanelle, *Eur. J. Med. Chem.*, 2007, **42**, 880–884; (e) J. Xiang, M. Ipek, V. Suri, W. Masefski, N. Pan, Y. Ge, M. Tam, Y. Xing, J. F. Tobin, X. Xu and S. Tam, *Bioorg. Med. Chem. Lett.*, 2005, **15**, 2865–2869; (f) H. Peng, Y. Cheng, N. Ni, M. Li, G. Choudhary, H. T. Chou, C.-D. Lu, P. C. Tai and B. Wang, *ChemMedChem*, 2009, **4**, 1457–1468.
- (a) Y. M. Markitanov, V. M. Timoshenko and Y. G. Shermolovich, *J. Sulfur Chem.*, 2014, **35**, 188–236; (b) O. O. Shyshkina, K. S. Popov, O. O. Gordivska, T. M. Tkachuk, N. V. Kovalenko, T. A. Volovnenko and Yu. M. Volovenko, *Chem. Heterocycl. Compd.*, 2011, **47**, 923–945.
- (a) B. M. Trost and D. P. Curran, *Tetrahedron Lett.*, 1981, **22**, 1287–1290; (b) A. K. Singh, R. Chawla, T. Keshari, V. K. Yadav and L. D. S. Yadav, *Org. Biomol. Chem.*, 2014, **12**, 8550–8554; (c) X.-J. Pan, J. Gao and G.-Q. Yuan, *Tetrahedron*, 2015, **71**, 5525–5530.
- (a) Y. Y. Xie and Z. C. Chen, *Synth. Commun.*, 2001, **31**, 3145–3149; (b) Q. Lu, J. Zhang, G. Zhao, Y. Qi, H. Wang and A. Lei, *J. Am. Chem. Soc.*, 2013, **135**, 11481–11484.
- (a) G. Liu, C.-B. Fan and J. Wu, *Org. Biomol. Chem.*, 2015, **13**, 1592–1599; (b) D. Zheng and J. Wu, *Sulfur Dioxide Insertion Reactions for Organic Synthesis*. Springer, Singapore, 2017, pp. 11–77; (c) G.-S. Qiu, K.-D. Zhou, J. Wu and L. Gao, *Org. Chem. Front.*, 2018, **5**, 691–705; (d) G.-S. Qiu, L.-F. Lai, J. Wu and J. Cheng, *Chem. Commun.*, 2018, **54**, 10405–10414; (e) K. Hofman, N. W. Liu and G. Manolikakes, *Chem.–Eur. J.*, 2018, **24**, 11852–11863; (f) S.-Q. Ye, G.-S. Qiu and J. Wu, *Chem. Commun.*, 2019, **55**, 1013–1019; (g) S. Ye, X. Li, J. Wu and W. Xie, *Eur. J. Org. Chem.*, 2020, **2020**, 1274–1287; (h) D. Zeng, M. Wang, W.-P. Deng and X. Jiang, *Org. Chem. Front.*, 2020, **7**, 3956–3966; (i) S. P. Blum, K. Hofman, G. Manolikakes and S. R. Waldvogel, *Chem. Commun.*, 2021, **57**, 8236–8249; (j) E. L. S. de Souza and C. C. Oliveira, *Eur. J. Org. Chem.*, 2023, **26**, e202300073.
- (a) D. Zheng, J. Yu and J. Wu, *Angew. Chem., Int. Ed.*, 2016, **55**, 11925–11929; (b) H. Wang, S. Sun and J. Cheng, *Org. Lett.*, 2017, **19**, 5844–5847; (c) Y. Zong, Y. Lang, M. Yang, X. Li, X. Fan and J. Wu, *Org. Lett.*, 2019, **21**, 1935–1938; (d) H. Xia, Y. An, X. Zeng and J. Wu, *Org. Chem. Front.*, 2018, **5**, 366–370; (e) X. Gong, M. Wang, J. Wu and S. Ye, *Org. Lett.*, 2019, **21**, 1156–1160; (f) Y. Meng, M. Wang and X. Jiang, *CCS Chem.*, 2021, **3**, 17–24; (g) M. Zhang, L. Liu, B. Wang, Y. Yang, Y. Liu, Z. Wang and Q. Wang, *ACS Catal.*, 2023, **13**, 11580–11588; (h) H. Li, Y. Zhang and X. Zou, *ACS Catal.*, 2024, **14**, 3664–3674.
- T. Liu, D. Zheng, Y. Ding, X. Fan and J. Wu, *Chem.–Asian J.*, 2017, **12**, 465–469.
- X. Gong, Y. Ding, X. Fan and J. Wu, *Adv. Synth. Catal.*, 2017, **359**, 2999–3004.
- F.-S. He, P. Bao, Z. Tang, F. Yu, W.-P. Deng and J. Wu, *Org. Lett.*, 2022, **24**, 2955–2960.
- (a) J. Chen, H. Qu, J. Peng and C. Chen, *Chin. J. Org. Chem.*, 2015, **35**, 937–946; (b) M. Sheng, D. Frurip and D. Gorman, *J. Loss Prev. Process Ind.*, 2015, **38**, 114–118.
- M. Wang, S. Chen and X. Jiang, *Org. Lett.*, 2017, **19**, 4916–4919.
- (a) Z. Chen, N.-W. Liu, M. Bolte, H. Rena and G. Manolikakes, *Green Chem.*, 2018, **20**, 3059–3070; (b) N.-W. Liu, Z. Chen, A. Herbert, H. Ren and G. Manolikakes, *Eur. J. Org. Chem.*, 2018, **2018**, 5725–5734; (c) A. M. Nair, I. Halder, S. Khan and C. M. R. Volla, *Adv. Synth. Catal.*, 2020, **362**, 224–229.
- D. Sun, K. Yin and R. Zhang, *Chem. Commun.*, 2018, **54**, 1335–1338.
- C. Breton-Patient, D. Naud-Martin, F. Mahuteau-Betzer and S. Piguel, *Eur. J. Org. Chem.*, 2020, **2020**, 6653–6660.
- (a) C.-L. Sun and Z.-J. Shi, *Chem. Rev.*, 2014, **114**, 9219–9280; (b) P. Ruiz-Castillo, L. Stephen and S. L. Buchwald, *Chem. Rev.*, 2016, **116**, 12564–12649.

