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Insight into heavy metal chemical fractions in ash collected from municipal and industrial waste incinerators in northern Vietnam†

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This investigation involved the collection of fly ash and bottom ash specimens from seven waste incinerators situated in the northern provinces of Vietnam, aimed at assessing the composition and distribution patterns of five chemical fractions of heavy metals (Pb, Cr, As, Cd, Cu, and Zn) present in incinerator waste ash. The outcomes reveal that fly ash exhibited a relatively elevated concentration of industrial waste metals (25–66%) such as As, Cd, and Pb primarily in exchangeable (F1) and carbonate fractions (F2), which are mobile forms susceptible to environmental dissolution and consequential bioaccumulation posing health risks to humans. The predominant states of the metals Cr, Cu, and Zn were identified as residual, Fe–Mn oxide, and carbonate, respectively, with their relative proportions showing minimal variation. Conversely, heavy metals were predominantly present in residual residue and Fe–Mn bound form (F3) in bottom ash derived from both residential and commercial waste incineration operations. The non-carcinogenic hazard indices (HI) associated with the examined metals, ranked for both adults and children, were as follows: Pb > Cr > As > Cd > Cu > Zn. Notably, the HI values for Pb, Cr, and As exceeded the permissible threshold (HI > 1) for children. However, the risk of As, Cd, and Pb-related cancer *via* exposure pathways remained within acceptable limits for both age groups. Conversely, the probability of carcinogenic effects attributable to Cr surpassed the permissible threshold (>10^{−4}), indicating significant health concerns associated with heavy metals in waste incinerators for humans, particularly children.

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1. Introduction

In recent years in Vietnam, the total volume of solid waste, encompassing urban, medical, and industrial waste, has experienced a rapid tenfold increase from 3.8 kg to 41 kg during the period 1990–2015. This surge is attributed to the processes of urbanization, industrialization, a robust economy, and population growth.^{1,2} Annually, Vietnam generates over 15 million tons of waste, a figure expected to escalate rapidly in the coming decade. Despite urban areas constituting only 24% of the country's population, they contribute over 6 million tons, accounting for 50% of the nation's urban waste.³ Out of the 3.6 million tons of plastic waste produced yearly, only 10 to 15%

undergo recycling, with the remainder treated through land-filling or incineration.⁴ Many developing countries face limitations in waste management activities such as collection, sorting, and processing, primarily due to insufficient funds, low technology, loose regulations, and weak community awareness, leading to restricted recycling efforts.⁵

Currently, solid waste is predominantly incinerated, a method capable of reducing waste volume by 90% and weight by approximately 70%.^{6–8} Tangri (2003) conducted a recent study on the conversion of waste into electrical energy. The research revealed that solid waste incinerators emit the highest level of greenhouse gases per unit of electricity generated (1707 g CO₂e kWh^{−1})⁹ compared to all other energy sources (ranging from 2.4 to 991 g CO₂e kWh^{−1}). Additionally, when compared to alternative energy sources like natural gas, solid waste incinerators produce higher levels of air pollutants. Factors such as poor technology, aging infrastructure, and prolonged operational hours contribute to excessive emissions, including 637.7 million tons of CO₂e, 61.9 million tons of NO_x, and 161 200 tons of SO₂. Municipal solid waste incinerators are regarded as contentious technologies in the US due to their air pollution impacts, often leading to public outrage.⁹ Incinerators serve as significant sources of harmful air emissions,

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encompassing dioxins, lead, mercury, nitrogen and sulfur oxides, particulates, and other organic pollutants, alongside greenhouse gases.⁹ Moreover, incinerators are identified as the primary contributors to air pollution in certain urban areas.¹⁰

Despite the advantages of incinerating urban and industrial waste in reducing toxicity and recovering resources, research indicated that metals underwent complex chemical transformations during high-temperature processes such as oxidation, reduction, chlorination, and other chemical processes within incineration furnaces. Metals can be released through gas pathways or self-trapped in fly ash and bottom ash. Fly ash collected in dust filter bags has a high content of heavy metals and small particle size, making it prone to environmental dispersion.¹¹ The release of heavy metals from fly ash and bottom ash into the environment can have adverse health effects on both humans and the environment.^{5,12} Recent observations showed that waste incinerators were a source of pollution in urban water and air.^{13,14} The presence of heavy metals such as As, Cd, Cr, Pb, Cu, Zn... has been reported in incinerator ash with relatively high concentrations.^{15–18} Previous reports indicated that many heavy metals concentrate in bottom ash during incineration. Metals such as Cu, and Cr, as well as volatile metals like Zn, and Pb, still predominantly remain in bottom ash, ranging from 46 to 94%.^{19,20} The risk of cancer for both children and adults from metals in ash and the surrounding soil near incinerators exceeds permissible limits, as reported by Shi *et al.* and Aendo *et al.*^{21,22} However, the total metal content alone cannot accurately assess the ecological and environmental risks. The greatest toxicity of heavy metals occurs when they exist in a labile form (or easily exchangeable form) in the environment, as labile metals can be easily released into the surrounding environment, leading to accumulation in organisms and subsequent entry into the human body through the food chain.²³

On the other hand, to minimize the large amount of ash from incinerators in solid waste management, fly ash is suggested for use as fertilizer in agriculture, and bottom ash is reused for roads and construction materials like cement and bricks.^{24,25} Therefore, understanding the characteristics of metals or their chemical bonding forms, especially the labile form, is essential for filtering metals in fly ash and bottom ash. The dynamic nature of metals depends on many factors, including bonding forms, pH, or the amount of organic matter. The Tessier sequential extraction method is widely used to analyze metal bonding forms in solid samples²⁶ This method is widely applied in many studies evaluating the ability of metals in incinerator ash,^{19,27} soil samples, road dust,^{28–30} biological reclamation of coal ash, mining waste, and biological sludge.^{31–34}

Understanding the distribution and fractions of heavy metals in ash is crucial for assessing the environmental impact of municipal and industrial incinerators. This knowledge contributes to evaluating potential risks and developing effective mitigation strategies. However, there is limited information related to the chemical fractions of heavy metals in ash from incinerators in Vietnam. Hence, it is urgent to address a gap in existing scientific literature, providing new insights and data

that contribute to the overall understanding of heavy metal behavior in incinerator ash. This can stimulate further research and inform future studies in similar contexts.

In this study, four household waste incinerators and three industrial waste incinerators were collected from four large provinces in northern Vietnam. The Tessier sequential extraction procedure was applied to provide an overview of the bonding forms, distribution, and mobility of heavy metals in fly ash and bottom ash from household and industrial waste incinerators. Additionally, the non-carcinogenic and carcinogenic risks to humans were assessed. This research aims to elucidate the characteristics of heavy metals and provides insights into the levels of contaminants that may enter soil, water, and air, impacting human health.

2. Materials and methods

2.1. Materials

A total of 42 samples, comprising fly ash and bottom ash, were collected from 7 waste incineration facilities (3 industrial and 4 municipal incinerations) located in the provinces of Northern Vietnam. Detailed information on the incinerators is presented in Fig. 1 and Table S1.† Sampling occurred in August and September 2020. Fly ash samples, representing fine particulate matter from the combustion process, were obtained from the incinerator's emission control system. Bottom ash consists of coarse particles concentrated at the incinerator's bottom, ranging in size from fine sand to pebbles (0.125 mm to 2 mm). Samples were collected as point samples using manual methods with collection tools such as brushes, shovels, and specialized trays following Vietnamese standards.³⁵ After each sampling process, the collection tools were washed and cleaned with deionized water (18.2 MΩ), acetone, and *n*-hexane to prevent cross-contamination with previous samples. The collected samples were then transferred to polyethylene (PE) bags with zip locks to avoid environmental contamination.

2.2. Methods

2.2.1. Ash properties analysis. For the analysis of total metal content in waste ash samples, sample decomposition was performed following the method of the United States Environmental Protection Agency (EPA) – method.³⁶ 500 mg sample of dust was decomposed with 9 mL of 65% HNO₃ and 3 mL of 48% HF (density = 1.15 g cm⁻³) in a Teflon container (placed in a microwave) following the program: (i) heating up to 180 °C in 5.5 minutes; (ii) holding at 180 °C for 15 minutes; (iii) cooling down to 50 °C. Blank samples were treated similarly to real samples. Metal concentrations were determined using inductively coupled plasma optical emission spectrometry (ICP-OES) with the Horiba Ultima 2 instrument, method 6010D.³⁷ The method's detection limits ranged from 0.0268 mg kg⁻¹ to 0.0350 mg kg⁻¹, with a recovery rate between 89.9% and 106%.

Scanning electron microscopy (FE-SEM, JSM-6700F, JEOL, Akishima Tokyo, Japan) in conjunction with energy dispersive X-ray spectroscopy (EDX) was used to analyze the morphology and elemental composition of specific areas inside the samples.





Fig. 1 Map of ash sampling area from waste incinerators in northern Vietnam.

2.2.2. Sequential extraction procedure (SEP). To analyze the chemical forms of metals (Cu, As, Cd, Cu, Cr, and Zn) in fly ash and bottom ash from incinerators, Tessier's sequential extraction procedure (SEP) was applied to separate metal forms.²⁶ Fig. S1.† The procedure includes 4 stages, separating into 5 main forms: exchangeable form (F1), carbonate-bound form (F2), Fe–Mn oxide-bound form (F3), sulfide and organic form (F4), and residual form (F5). The concentrations and recovery rates of the studied metals in the MESS_4 standard material using the Tessier extraction procedure, from F1 to F5, ranged from 89.8% to 110%, as determined by ICP-OES equipment.

2.2.3. Assessment of bioavailability. In this study, the Bioavailability Factor (BF) and Contamination Factor (CF) were employed to assess the bioavailability of the studied metals in dust samples. The calculated indices are based on the following formulas (1) and (2):^{38,39}

$$BF = C_{\text{Bio}}/C_{\text{total}} \quad (1)$$

$$CF = (F1 + F2 + F3 + F4)/F5 \quad (2)$$

here, C_{Bio} represents the concentration of heavy metals (Cu, As, Cd, Pb, Zn, and Cr) in bioavailable fractions (F1, F2, F3, F4). Form F5 denotes the residual fraction (mg kg^{-1}). The impact levels of the CF index are classified into the following categories: low (<1); moderate (1–3); high (3–6); very high (>6).

2.2.4. Health risk assessment. Assessing the public health implications associated with heavy metals in fly ash and bottom ash involves the application of exposure estimation methods developed by the United States Environmental Protection Agency (US EPA) for both children and adults. As stipulated by

USEPA guidelines, human exposure primarily occurs through ingestion (ID_{ing}), inhalation (ID_{inh}), and skin contact with ash particulate (ID_{dermal}).^{40,41} The estimated average daily intake (ID) is derived through three exposure pathways (eqn (3)–(5)).^{29,42,43}

$$ID_{\text{ing}} = C \times \frac{\text{IngR} \times \text{EF} \times \text{ED}}{\text{BW} \times \text{AT}} \times 10^{-6} \sqrt{a^2 + b^2} \quad (3)$$

$$ID_{\text{inh}} = C \times \frac{\text{InhR} \times \text{EF} \times \text{ED}}{\text{PEF} \times \text{BW} \times \text{AT}} \quad (4)$$

$$ID_{\text{dermal}} = C \times \frac{\text{SA} \times \text{AF} \times \text{ABS} \times \text{EF} \times \text{ED}}{\text{BW} \times \text{AT}} \times 10^{-6} \quad (5)$$

where: C represents the concentration of heavy metals in ash (mg kg^{-1}); IngR is the ingestion rate (mg per day) for children (200) and adults (100); InhR is the inhalation rate (m^3 per day) for children (7.36) and adults (12.8); ED is the exposure duration (years) for children (6) and adults (7.36); EF is the exposure frequency (days per year) set at 350; BW is the body weight (kg) for children (15) and adults (70); AT is the averaging time (days) calculated as ED multiplied by 365; PEF is the particle emission factor ($\text{m}^3 \text{kg}^{-1}$) at 1.36×10^9 ($\text{m}^3 \text{kg}^{-1}$); SA is the exposed skin area (cm^2) for children (2800) and adults (5700); ABS is the dermal absorption factor (10^3) for both children and adults; AF is the skin adherence factor (mg cm^{-1}) at 0.2 for children and 0.07 for adults. Non-carcinogenic risk (HI) and carcinogenic risks (CR) are determined through eqn (6)–(9):

$$HQ_{\text{ing/inh/derm}} = \frac{ID_{\text{ing/inh/derm}}}{\text{RfD}} \quad (6)$$



$$HI = \sum HQ_i \quad (7)$$

$$CR_i = ID_i \times CSf \quad (8)$$

$$CR = \sum CR_i \quad (9)$$

where: HQ represents the hazard quotient; RfDi is the reference dose of metal *i* (mg kg⁻¹ per day); if HQ or HI > 1, a non-carcinogenic health risk may be present. Conversely, if HQ or HI < 1, no observed health effects are associated with exposure to non-carcinogenic metals. CSf is the carcinogenicity slope factor (mg per kg per day). The range of CR values from 10⁻⁶ to 10⁻⁴ suggests manageable or acceptable carcinogenicity. Carcinogenic risk is considered significant if the value exceeds 10⁻⁴. CR values lower than 10⁻⁶ indicate no significant health risk. Reference values for RfD and CSf were drawn from previous studies.^{21,22,44}

3. Results and discussion

3.1. Concentration and characteristics of metals distribution in municipal solid waste incinerators and industrial waste incinerators

The average concentration and characteristic distribution of six studied metals were assessed in four municipal solid waste incinerators and three industrial waste incinerators (Table S2,† Fig. 2). The average metal concentrations in fly ash were arranged in the following order: Zn (2809 mg kg⁻¹) > Pb (2169 mg kg⁻¹) > Cu (1508 mg kg⁻¹) > Cr (665.8 mg kg⁻¹) > As (30.81 mg kg⁻¹) > Cd (20.95 mg kg⁻¹). These results align with trends found in previous studies.⁴⁵

The concentrations of Cu, Pb, and Cr were similar to the results reported in fly ash from municipal solid waste incinerators in Japan in 2019 (Cu: 1200 mg kg⁻¹, Pb: 3100 mg kg⁻¹, Cr: 490 mg kg⁻¹),¹⁵ and in China (Cu: 1710 mg kg⁻¹, Pb: 1537 mg kg⁻¹ in 2020 and Cu: 1055 mg kg⁻¹, Pb: 3374 mg kg⁻¹ in 2019).^{16,46} However, the concentrations of Zn and Cd were relatively low compared to studies in China (Zn: 12 187 mg kg⁻¹, Cd: 119 mg kg⁻¹ in 2019 and Zn: 9612 mg kg⁻¹, Cd: 196.8 mg kg⁻¹ in 2020),^{16,46} and in Japan.¹⁵ In bottom ash, the concentrations of As

(14.5 mg kg⁻¹), Cd (6.75 mg kg⁻¹), Pb (1611 mg kg⁻¹), and Zn (1938 mg kg⁻¹) tended to be lower than in fly ash, while Cu (1933 mg kg⁻¹) and Cr (775 mg kg⁻¹) were higher than in fly ash. The metal concentrations in the bottom ash of incinerators in Vietnam were similar to results reported in municipal solid waste incinerators in Japan (Cu: 890–3300 mg kg⁻¹, Pb: 140–2700 mg kg⁻¹, Cd: 2.8–12 mg kg⁻¹, Zn: 1500–2900 mg kg⁻¹),⁴⁷ and France (386–2825 mg kg⁻¹), but lower than the results in the bottom ash of municipal solid waste incinerators in Hangzhou, Ningbo (Zn: 2825 mg kg⁻¹, Pb: 1900 mg kg⁻¹), and Jinhua (Cd: 7.5 mg kg⁻¹, As: 26.4 mg kg⁻¹), and Wenzhou (As: 119 mg kg⁻¹, Cr: 1224 mg kg⁻¹, Zn: 3347 mg kg⁻¹) of China.^{12,19} To the best of our knowledge, there is presently scant information concerning heavy metal presence in industrial waste incinerators, rendering comparisons with the data presented in this study difficult.

The average distribution ratios of the six heavy metals (As, Cd, Cu, Cr, Pb, Zn) in fly ash from municipal solid waste incinerators were: (65–79%), (43–92%), (16–45%), (33–71%), (48–62%), (54–62%) (Fig. 3). In corresponding industrial waste incinerators, the ratios were: (42–71%), (46–64%), (42–70%), (27–52%), (51–66%), (59–63%).

In corresponding industrial waste incinerators, the ratios were: (42–71%), (46–64%), (42–70%), (27–52%), (51–66%), (59–63%). It is noticeable that metals such as As and Cd tended to occupy a higher proportion of fly ash from municipal solid waste incinerators, while metals like Cu dominated in samples collected from industrial waste incinerators. The remaining metals, Cr, Pb, and Zn, had similar percentage ratios in both types of incinerators (Anova, *p* > 0.05).

In bottom ash, the average distribution ratios of the studied metals in municipal solid waste incinerators were: As (21–35%), Cd (8–57%), Cu (55–84%), Cr (29–67%), Pb (38–52%), Zn (37–46%). For industrial waste incinerators, the corresponding ratios were: (29–58%), (36–54%), (30–58%), (48–73%), (34–49%), (37–41%). In contrast to fly ash, the average distribution ratios of metals showed no significant differences between municipal solid waste incinerators and industrial waste incinerators (*p* > 0.05). Calculated results also indicated that metals such as As, Cd, Pb, and Zn were prone to vaporize during combustion, subsequently enriching on the surface of fly ash

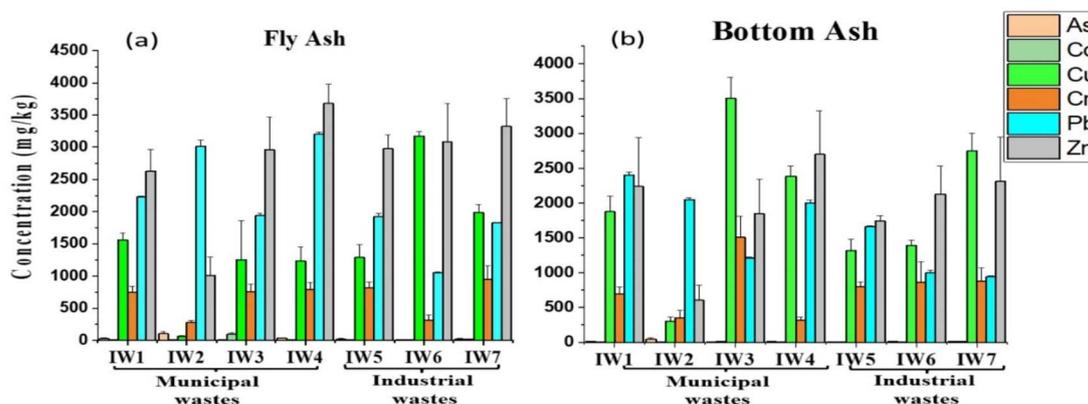


Fig. 2 Total concentration of heavy metals (a) in fly ash and (b) bottom ash collected from municipal and industrial waste incinerators (IW1: Bac Ninh 1, IW2: Hai Phong 1, IW3: Ha Noi, IW4: Hai Phong 2, IW5: Bac Ninh 2, IW6: Bac Giang; IW7: Hai Duong).



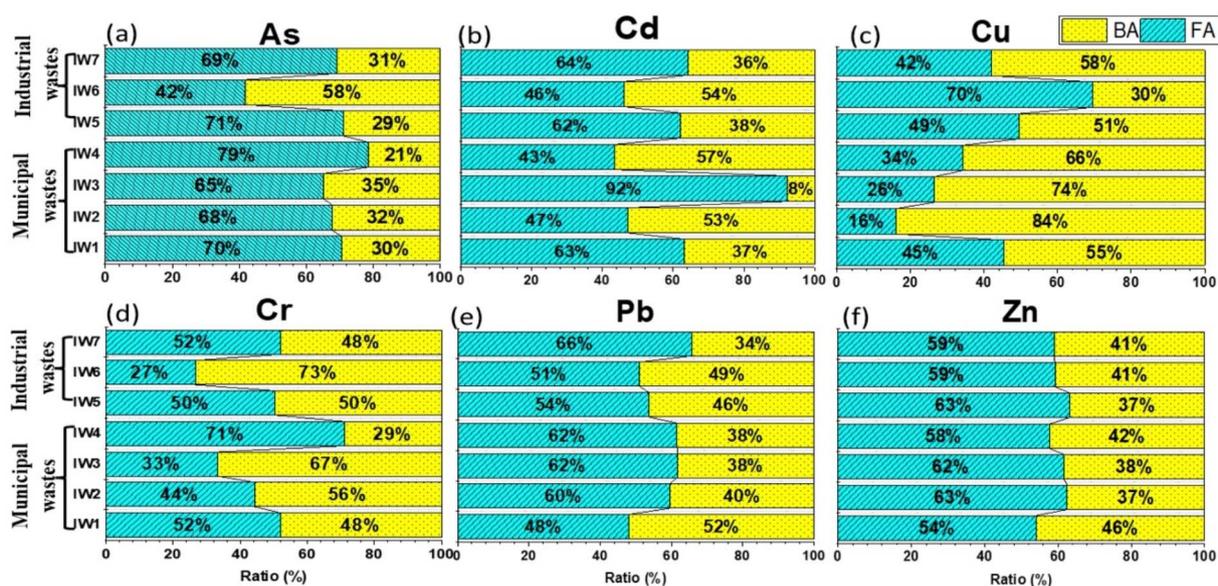


Fig. 3 Distribution ratio of total metals As – (a), Cd – (b), Cu – (c), Cr – (d), Pb – (e), and Zn – (f) in fly ash and bottom ash of waste incinerators (IW1: Bac Ninh 1, IW2: Hai Phong 1, IW3: Ha Noi, IW4: Hai Phong 2, IW5: Bac Ninh 2, IW6: Bac Giang, IW7: Hai Duong).

particles, leading to a higher concentration in fly ash. Cu, on the other hand, tended to distribute more in the bottom ash of municipal solid waste incinerators (Anova, $p < 0.05$). Chromium demonstrated an even distribution between fly ash and bottom ash, consistent with the findings of Fu. *et al.* 2018.⁴⁸ Previous studies also revealed that elements (As, Pb, Cd, and Zn) constituted approximately 40–50% of fly ash.⁴⁸ However, in some incinerators, the distribution did not align with chemical characteristics. The difference in the distribution of the same element between municipal solid waste incinerators and industrial waste incinerators may be closely related to factors

such as emissions, types of incinerated waste, and incinerator temperatures. An efficient flue gas treatment system can also reduce the metal content in fly ash.⁴⁹ High moisture content in municipal solid waste significantly affects the distribution of fly ash and bottom ash.⁴⁹

3.2. Surface characteristics of fly ash and bottom ash collected from incineration plants

The SEM analysis results indicate that the surface of the bottom ash is irregular (Fig. 4). Bottom ash from industrial incinerators

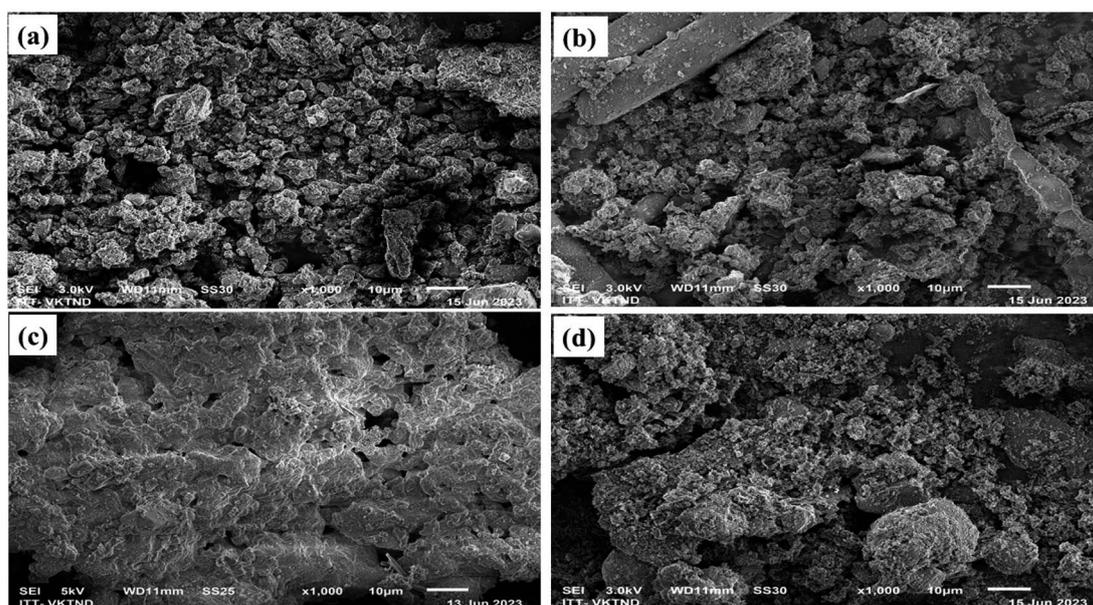


Fig. 4 SEM images of the bottom ash (a, b) and the fly ash (c, d) collected from municipal and industrial waste incinerators.



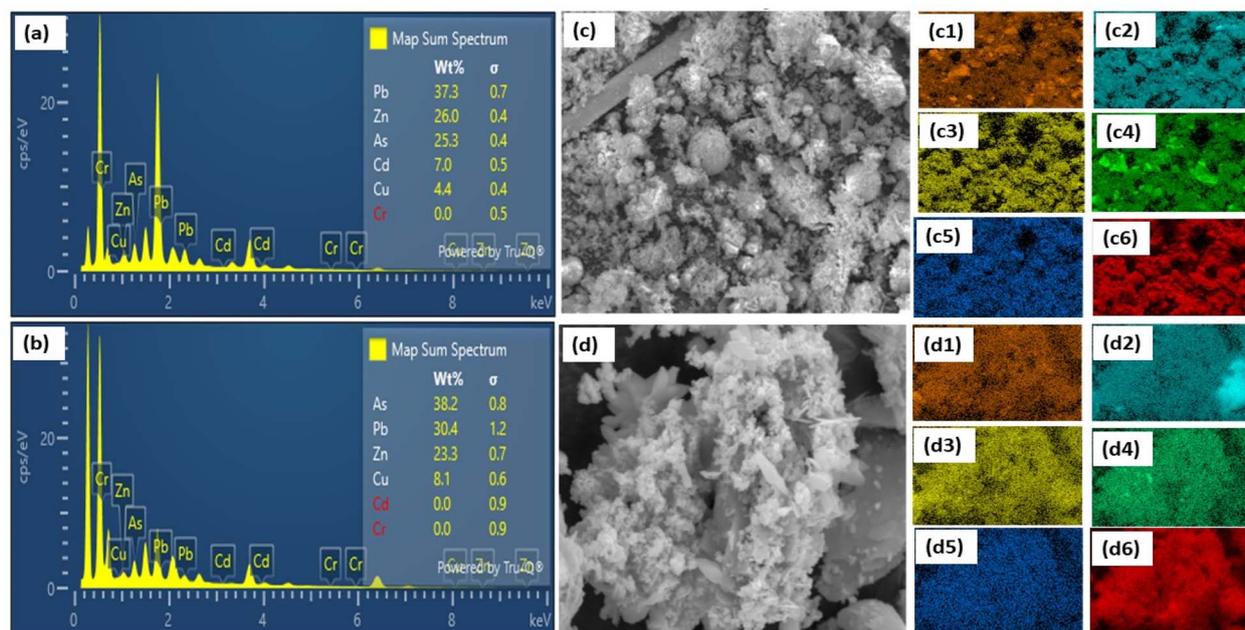


Fig. 5 The EDX map spectrum of fly ash (a) and bottom ash (b); SEM micrographs illustrate the analyzed areas of fly ash (c) and bottom ash (d). EDX mapping for Pb (c1, d1), Zn (c2, d2), Cu (c3, d3), As (c4, d4), Cd (c5, d5), and Cr (c6, d6) for the fly ash and bottom ash collected from the incinerator.

exhibits a surface consisting of particles with more uneven sizes compared to those from municipal waste incinerators. This disparity may stem from the industrial incinerators burning a more diverse array of waste types, resulting in a less uniform surface of ash particles due to variations in composition. In contrast, the particle sizes of fly ash appeared finer than those of bottom ash. Additionally, the fly ash from municipal waste incinerators was more uniform and cohesive compared to that from industrial counterparts. Thus, there was a slight disparity in the surface morphology of fly ash and bottom ash from different incinerators, which may be closely linked to differences in chemical composition as well as the concentration of heavy metals in the ash samples.

EDX analysis provides a relatively comprehensive overview of the elemental composition on the surface of the ash samples. The EDX analysis results of fly ash and bottom ash are also depicted in Fig. S2 (see ESI†) and Fig. 5. The EDX analysis indicates that the main components of bottom ash were Fe, C, O, Ca, Cl, Si, S, K, Al, Mg, and Na, while those of fly ash included Fe, C, O, S, Cd, Pb, Zn, Mn, Cl, S, K, Si (Fig. S2†). The presence of these elements was reported in previous research studies.^{50,51} The EDX mapping of the six heavy metals, including Pb, Zn, As, Cu, Cd, and Cr, in the studied fly ash and bottom ash samples is shown in Fig. 5.

3.3. Speciation characteristics and correlation of metals in fly ash and bottom ash

3.3.1. Chemical fractions of heavy metals in fly ash. The average concentrations of five chemical binding forms of metals were also assessed in two types of municipal and industrial waste, as shown in Table 1 and Fig. 6.

Results indicate that metals such as Cu and Cr were predominantly present in the residual form F5, with ratios of 39–55% and 59–78%, respectively. In fly ash, Cr primarily exists as Cr^{3+} , and both Cu and Cr can form bonds with organic compounds and sulfides, which during combustion transform into stable oxide compounds, therefore, their concentrations were relatively high in the residual form.^{48,52} In all fly ash samples, As accounted for a relatively high proportion in the carbonate form (F2) with ratios of (25–43%), followed by the residual form (F5) (25–29%), Fe–Mn oxide form F3 at 11–27%, exchangeable form F1, and organic form F4 with similar ratios (4–16%). Cd was mainly present in mobile forms F2 and F1, with the highest percentage ratios reaching 56% and 62%, respectively, while the remaining forms had comparable ratios. This finding suggests that Cd was highly mobile in the environment, and is consistent with findings reported by Chen in 2020 (with 95.90% of Cd in exchangeable and reducing forms).¹¹ Multivariate regression analysis (PCA) of metal binding forms in municipal solid waste and industrial waste incinerators was assessed using Kaiser criteria,⁵³ as shown in Fig. 7. For municipal solid waste incinerators, consisting mainly of food waste and nylon, all five binding forms accounted for the majority in PC1 (86.39%) and PC2 at 12.03%. Fractions F2 and F3 primarily originated from municipal solid waste incinerators for Zn and Cu, while F1, F4, and F5 were more pronounced for Pb. As and Cd, on the other hand, contributed to PC2, mainly in chemical forms F1 and F2, with Cr presented in chemical fraction F5. Conversely, for industrial waste incinerators, predominantly consisting of nylon, packaging, fabric, paper, and pressed sludge, fractions F1 and F2 dominated the structure of PC1 (71.82%) for the metal groups Pb, Cd, and As. Fractions F3, F4,

Table 1 Concentration of heavy metals fractions in fly ash (mg kg^{-1})

Metals	Incinerators	F1	F2	F3	F4	F5	Total metal	CF	BF
As	IW1	0.502	0.83	16.54	2.963	6.17	26.99	3.382	0.773
	IW2	22.32	38.70	14.08	0.812	31.64	107.5	2.401	0.710
	IW3	1.773	1.582	1.173	0.742	0.990	6.241	5.322	0.844
	IW4	1.844	3.181	14.95	2.270	11.39	33.63	1.955	0.662
	IW5	0.685	6.063	0.781	4.770	4.980	17.25	2.473	0.711
	IW6	0.451	0.142	2.434	0.743	4.355	8.122	0.874	0.462
	IW7	1.082	12.21	1.392	1.492	1.662	17.83	9.734	0.910
	Mean	4.093	8.962	7.333	1.970	8.741	31.09	2.560	0.724
Cd	IW1	0.841	0.921	4.102	1.102	2.341	9.305	2.981	0.753
	IW2	1.402	1.522	1.521	1.543	1.372	7.344	4.382	0.811
	IW3	65.69	13.59	3.753	1.034	6.384	90.43	13.19	0.913
	IW4	1.334	1.663	0.460	1.612	0.082	5.142	62.31	0.981
	IW5	0.760	1.554	1.482	0.511	0.633	4.923	6.88	0.871
	IW6	0.042	1.961	1.141	1.646	1.336	6.110	3.59	0.780
	IW7	0.332	11.48	0.293	1.454	2.334	15.89	5.82	0.852
	Mean	10.06	4.672	1.822	1.273	2.063	19.88	8.63	0.902
Cu	IW1	113.8	53.25	605.3	200.8	516.9	1490	1.88	0.652
	IW2	0.852	0.520	9.892	0.77	39.81	51.83	0.304	0.230
	IW3	322.6	310.2	335.5	103.7	303.5	1375	3.535	0.784
	IW4	79.13	80.44	283.3	49.23	764.9	1257	0.645	0.393
	IW5	42.20	170.9	25.35	102.8	301.6	642.8	1.131	0.532
	IW6	16.80	921.6	167.8	54.23	1929	3089	0.603	0.384
	IW7	235.6	137.8	127.4	136.0	357.4	994.3	1.784	0.640
	Mean	115.8	239.2	222.1	92.50	601.9	1272	1.113	0.532
Cr	IW1	83.80	53.15	149.9	32.97	374.9	694.8	0.852	0.465
	IW2	3.102	10.93	1.510	9.74	277.0	302.3	0.094	0.083
	IW3	47.91	71.34	176.9	41.22	377.4	714.7	0.890	0.474
	IW4	39.63	90.44	209.0	39.11	466.1	844.3	0.812	0.452
	IW5	57.35	95.90	21.02	40.20	522.1	736.6	0.413	0.294
	IW6	0.263	16.54	19.61	7.160	247.6	291.1	0.183	0.155
	IW7	1.341	94.25	52.56	8.583	727.1	883.8	0.221	0.183
	Mean	33.34	61.79	90.07	25.57	427.5	638.2	0.495	0.335
Pb	IW1	430.3	353.1	451.1	317.3	628.9	2181	2.475	0.712
	IW2	358.6	164.6	250.3	169.7	2187	3130	0.436	0.305
	IW3	461.2	497.3	71.04	1121	1172	3322	1.840	0.651
	IW4	377.5	30.01	133.4	223.2	70.19	834.3	10.89	0.923
	IW5	202.2	542.5	209.9	248.5	210.1	1413	5.731	0.850
	IW6	414.8	4563	305.1	318.3	529.9	6131	10.57	0.914
	IW7	55.60	23.82	37.93	23.62	59.11	200.1	2.383	0.702
	Mean	328.6	882.1	208.4	345.9	693.9	2459	2.544	0.721
Zn	IW1	244.7	4.513	710.5	763.0	735.7	2458	2.342	0.706
	IW2	32.22	72.85	22.12	5.100	792.2	924.5	0.170	0.140
	IW3	670.3	836.3	814.8	347.9	815.2	3485	3.274	0.773
	IW4	118.4	937.6	1003	118.9	1192.3	3370	1.832	0.655
	IW5	414.6	745.9	65.95	796.4	737.1	2760	2.743	0.734
	IW6	2.08	703.3	914.8	789.8	749.6	3159	3.226	0.760
	IW7	457.4	904.4	952.2	547.9	613.6	3475	4.660	0.822
	Mean	277.1	600.7	640.4	481.3	805.1	2804	2.481	0.713

and F5, related to Cu and Zn, contribute to PC2 (17.25%). Thus, it can be observed that in both municipal and industrial waste incinerator fly ash, potential emission sources include mobile forms F1, F2, F3 for metals such as As, Cd, Pb, and forms F4, F5 for metals such as Cr, Zn, and Cu.

3.3.2. Binding form characteristics of metals in bottom ash. The concentrations and distribution ratios of metal binding forms in bottom ash are also presented in Table 2 and Fig. 6. Regarding Cr, the distribution ratio was quite similar to that in fly ash, with the highest percentage in the residual form (68–72%) and the lowest in form F4 (2–3%). Previous leaching

studies showed that Cr exhibited weak mobility, mainly existing in poorly soluble residual forms. This could be attributed to Cr predominantly interacting with Zn and Co (ZnCr_2O_4 and CoCr_2O_4) in waste incinerators.^{28,54}

Cd ratios for chemical forms F1 and F4 were equivalent, ranging from 18–28% and 23–27%, respectively, followed by form F3 (22–23%), and the lowest was F5, with F2 having lower ratios at 13–14% and 13–18%.

The distribution in bottom ash for metals As, Cu, Pb, and Zn was in the residual fraction, with corresponding percentage ratios of (35–44%), (27–28%), (21–30%), and 28%. There was



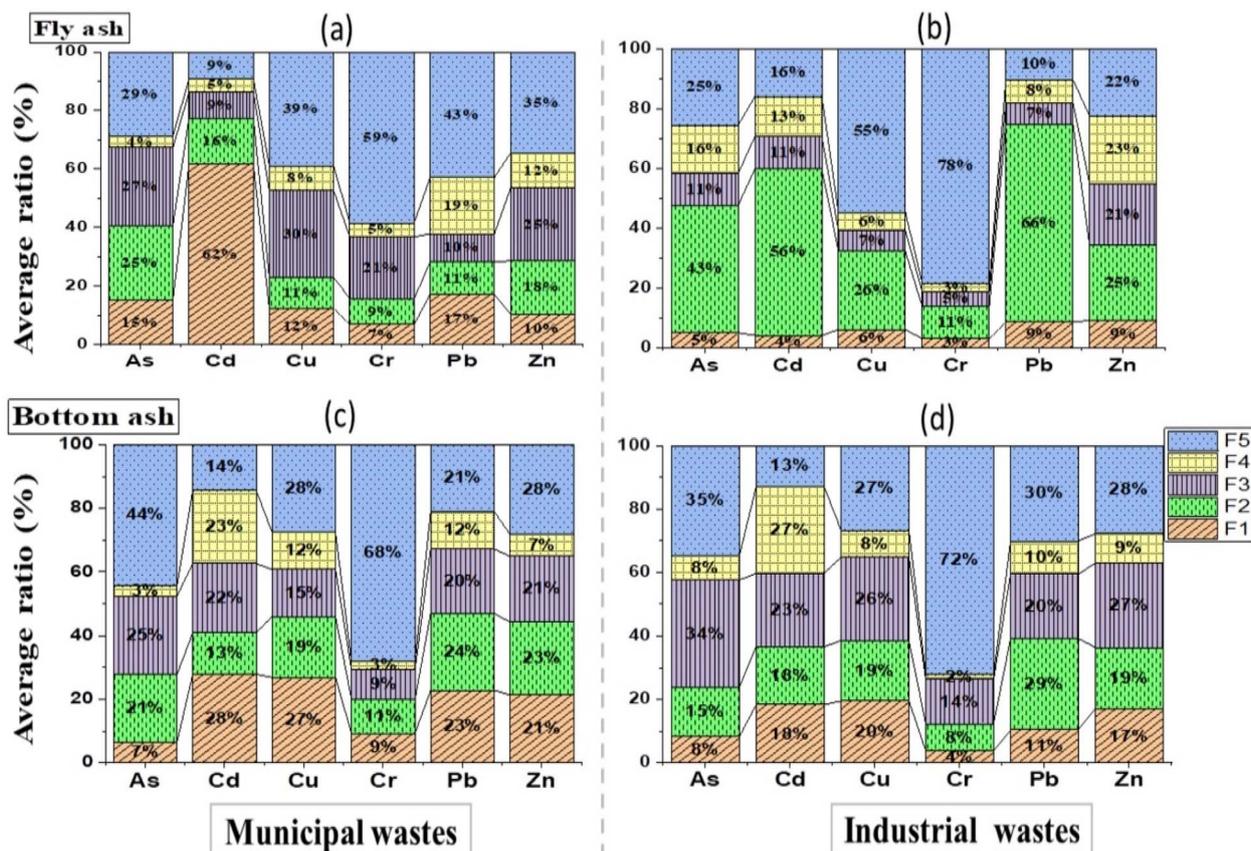


Fig. 6 Distribution ratio of heavy metals fractions in fly ash (a, b) and bottom ash (c, d) collected from municipal and industrial waste incinerators.

a noticeable difference in the percentage distribution of metals in the Fe–Mn oxide fraction in bottom ash compared to fly ash, with ratios being relatively high and uniform, namely (25–34%), (22–23%), (15–26%), (9–14%), 20%, and (21–27%). This suggests a strong association of metals with Fe–Mn oxides in the combustion waste, especially in industrial waste incinerators. The presence of stable thermal-resistant Fe and Mn oxide compounds in bottom ash contributes to the formation of bonds with metals. Total mobile fractions (F1 + F2) in bottom ash were 40–47%; 36–44%; 39–46%; 23–28%; 36–41% for Pb; Zn; Cu, and As, Cd. For Cd and Pb, chlorine compounds were formed during incineration at temperatures between 600–700 °C, easily vaporizing and self-trapping at lower temperatures. This explains the lower percentage of Cd and Pb in the F1 form in fly ash compared to bottom ash.⁴⁹ In the presence of oxygen in the flue gas, Cu in bottom ash can form carbonate and sulfate compounds, while Zn can form volatile ZnCl₂ compounds, which then transform into zinc oxide (ZnO), zinc carbonate, or zinc sulfate, condensing on ash particles.⁵⁵ Hence, for Cu and Zn, the percentage ratios among binding forms did not differ significantly (Anova, $p > 0.05$). In waste incinerators, As may bind with minerals (Al or Mg), forming compounds such as AsAlO₄ below 850 K and As₂Mg₃O₈ between 850 K and 1050 K. At temperatures higher than 1050 K, AsO(g), a gaseous oxide is the predominant species. As had intermediate solubility, resulting in lower mobility compared to Cd and Zn.⁵³

The binding forms of metals in bottom ash were also evaluated using multivariate regression analysis (PCA) for both municipal and industrial waste. In municipal waste incinerators, binding forms F1 to F4 belong to the main structure of PC1 (88.7%), with F5 in PC2 at 8.38%. Analysis results show that for municipal waste incinerators, Cu is mainly present in forms F2 and F5, while Zn and Cr were in fraction F5. Pb gradually decreased from F2 to F4. As and Cd belonged to the same group in form F1. For industrial waste incinerators, metals Pb, Cu, and Zn were concentrated in forms F1 to F4 in the main structure of PC1 (86.5%), while Cr in form F5 belonged to PC2 (10.3%).

3.3.3. Correlation analysis between heavy metal chemical fractions and pH, TC, OC, BF. Spearman correlation analysis of the five heavy metal binding forms is presented in Table 3. In fly ash samples, for As and Cd, factors such as pH, TC, and OC did not influence the distribution of forms F2, F3, F4, and F5. However, pH exhibited a significant correlation with the mobile form F1 ($p = 0.61, 0.79$ for As, Cd), indicating that the exchangeable form was notably affected by pH values. The carbonate, Fe–Mn oxide, and residual fractions of As were also significantly influenced by total metal content. Spearman correlation analysis revealed a high correlation between the carbonate form of Cu and TC ($p = 0.82$), while the oxide Fe–Mn form and residual form were predominantly controlled by the total metal content. Similarly, the forms of Cr were also governed by the total metal content. However, the distribution of



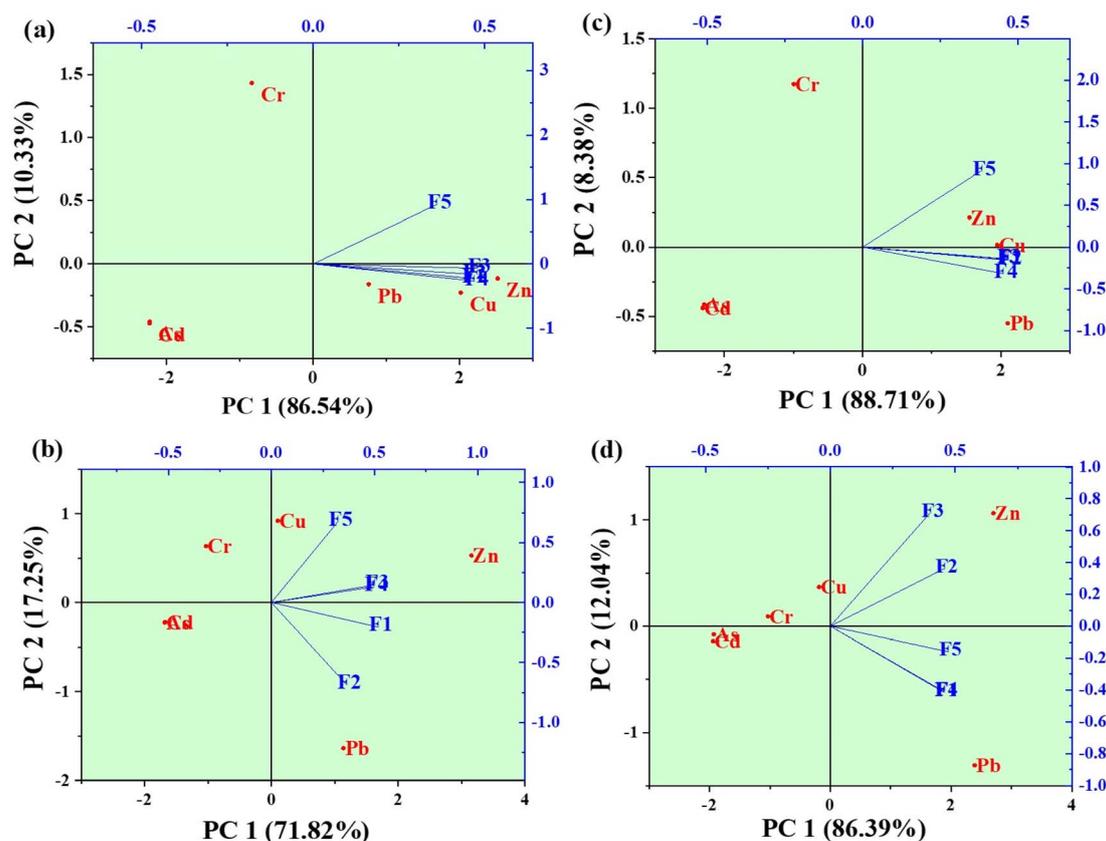


Fig. 7 Multivariate PCA analysis of heavy metals fractions in bottom ash from industrial waste (a), municipal incinerators (b) and fly-ash from industrial waste (c), municipal (d) incinerators.

the organic form showed a correlation with pH values ($p = 0.68$). Therefore, pH changes were related to the formation of organic compounds of Cr.

The total Pb content correlates positively with all five binding forms in fly ash, while TC values were closely related to the carbonate form. The organic form of Zn had a significant correlation with TC ($p = 0.64$) and OC ($p = 0.71$), indicating that an increase in total carbon content led to an increase in the organic form. These results align with the findings reported by Srivastava 2020.⁵⁶ Correlation analysis of biologically available forms in fly ash revealed insignificant correlations for mobile forms F1, F2, and F3 with volatile metals As, Cd, and Pb. However, for forms F1, F3, and F4 of Cu, Cr, and Zn, the values of BF and CF showed relatively high correlations (p values were 0.93, 0.74, and 0.61, respectively). This suggests that an increase in the concentration of these soluble forms enhances the biological availability of Cu, Cr, and Zn.

In bottom ash samples, Spearman correlation analysis results showed that pH was mostly inversely correlated with binding forms. This finding indicates that pH changes did not significantly affect the binding forms of As, Cd, Cu, Pb, and Zn in bottom ash. For Cr, however, pH had a fairly high correlation with the organic form ($p = 0.71$) and a mild correlation with the carbonate form ($p = 0.50$) and Fe–Mn oxide form ($p = 0.54$). Factors TC and OC had negligible effects on the distribution of metal forms in bottom ash, except for the correlation of TC with

the carbonate form of Cr ($p = 0.61$) and OC with the carbonate form of Pb ($p = 0.75$). According to our observations, the total metal content had a close correlation with forms F3, and F5 of As, Cd, and Cr, and forms F1, F3 of Pb and Zn. For Cu, the distribution of mobile forms F1, F2, F3, and F4 is dominated by the total metal content. From the analysis results, there was no significant positive correlation between biological availability and binding forms for As and Cd. However, for Pb, there was a high correlation between biological availability (BF, CF) and forms F1, F2, and F4. Meanwhile, Cu, Cr, and Zn had a strong correlation of biological availability with forms F1, F2, and F3.

3.4. Health risk assessment for humans from heavy metals in waste incineration ash

3.4.1. Non-carcinogenic risk. In this study, the daily intake of dust was assessed for both children and adults through three main exposure pathways: ingestion, inhalation, and dermal absorption upon contact. The non-carcinogenic and carcinogenic risks were evaluated using the Hazard Index (HI) and Cancer Risk (CR) indices. The daily intake for both children and adults in fly ash and bottom ash through the three exposure pathways is ordered as follows: ingestion > dermal contact > direct inhalation, as shown in Fig. 8 and Table S3.†

The estimated average intake for children through oral ingestion, direct inhalation, and dermal contact is 9.3 times, 6.5 times, and 2.8 times higher, respectively, than the estimated



Table 2 Concentration of heavy metals fractions in bottom ash (mg kg⁻¹)

Metals	Incinerators	F1	F2	F3	F4	F5	Total metals	CF	BF
As	IW1	0.734	0.965	4.380	0.625	4.935	11.64	1.360	0.576
	IW2	0.731	14.16	12.43	0.465	25.01	52.79	1.110	0.526
	IW4	0.569	0.838	0.513	0.613	0.681	3.210	3.720	0.788
	IW6	2.944	0.076	1.275	0.750	2.844	7.890	1.770	0.640
	IW3	0.630	2.625	0.735	0.520	3.635	8.150	1.240	0.554
	IW5	0.925	0.488	7.413	0.719	1.894	11.44	5.040	0.834
	IW7	0.863	1.350	1.656	0.988	4.500	9.360	1.080	0.519
	Mean	1.056	2.928	4.057	0.668	6.213	14.92	1.400	0.584
Cd	IW1	2.845	0.095	1.035	1.375	0.852	6.200	6.280	0.863
	IW2	1.240	1.520	1.525	1.520	1.355	7.160	4.280	0.811
	IW4	1.281	1.488	1.744	1.725	1.438	7.680	4.340	0.813
	IW6	2.350	0.544	1.675	1.763	0.325	6.660	19.48	0.951
	IW3	0.460	0.235	1.295	1.525	0.070	3.590	50.21	0.980
	IW5	1.263	1.369	1.144	1.725	0.663	6.160	8.300	0.892
	IW7	1.719	1.806	1.888	1.869	1.675	8.960	4.350	0.813
	Mean	1.594	1.008	1.472	1.643	0.911	6.630	6.280	0.863
Cu	IW1	619.3	423.7	171.4	59.45	539.8	1813	2.360	0.702
	IW2	0.820	5.020	4.270	0.895	263.2	274.2	0.040	0.040
	IW4	736.2	717.8	611.3	754.8	607.3	3427	4.640	0.823
	IW6	734.8	340.5	360.0	90.70	731.8	2258	2.090	0.676
	IW3	397.8	141.2	113.0	56.80	607.8	1317	1.170	0.538
	IW5	70.94	216.8	524.8	86.88	401.8	1301	2.240	0.691
	IW7	582.9	663.8	766.6	312.5	424.3	2750	5.480	0.846
	Mean	448.9	358.4	364.5	194.6	510.8	1877	2.670	0.728
Cr	IW1	1.415	0.710	49.68	10.38	675.8	737.9	0.090	0.084
	IW2	0.940	3.940	0.080	6.405	304.8	316.1	0.040	0.036
	IW4	204.1	296.0	215.9	61.60	826.1	1604	0.940	0.485
	IW6	73.06	22.48	10.39	5.588	235.7	347.2	0.470	0.321
	IW3	0.575	45.73	13.07	9.575	664.6	733.5	0.100	0.094
	IW5	96.00	105.3	318.1	24.40	360.3	903.9	1.510	0.601
	IW7	2.063	67.31	27.92	9.188	827.0	933.5	0.130	0.114
	Mean	54.00	77.30	90.70	18.20	556.3	796.6	0.430	0.302
Pb	IW1	312.4	252.6	674.4	203.1	785.9	2228	1.840	0.647
	IW2	478.6	562.2	341.7	180.2	620.4	2183	2.520	0.716
	IW4	261.4	408.0	248.3	183.0	158.8	1260	6.930	0.874
	IW6	725.3	664.1	310.6	350.8	71.88	2123	28.530	0.966
	IW3	201.4	704.6	256.4	220.4	426.7	1809	3.240	0.764
	IW5	104.8	63.45	418.8	97.88	398.1	1083	1.720	0.632
	IW7	89.88	314.2	86.15	75.65	307.9	873.7	1.840	0.648
	Mean	310.5	424.2	333.7	187.3	395.7	1651	3.170	0.760
Zn	IW1	518.4	528.0	443.8	323.1	510.1	2323	3.550	0.780
	IW2	11.83	76.50	26.06	4.808	504.1	623.3	0.240	0.191
	IW4	420.3	405.6	440.6	116.8	375.0	1758	3.690	0.787
	IW6	605.0	650.8	587.6	56.20	661.5	2561	2.870	0.742
	IW3	176.6	521.2	329.9	17.30	514.8	1560	2.030	0.670
	IW5	174.7	41.80	602.5	461.3	657.7	1938	1.950	0.661
	IW7	669.0	603.3	659.6	76.60	494.3	2503	4.060	0.802
	Mean	368.0	403.9	441.4	150.9	531.1	1895	2.570	0.720

values for adults. Thus, the oral ingestion pathway contributes significantly, aligning with the observed objective differences between children and adults in body weight, work habits, and slight variations in respiratory volume. In fly ash, the non-carcinogenic risk (HI) of metals for both children and adults decreases in the following order (Fig. 8): Pb > Cr > As > Cd > Cu > Zn. Notably, the HI values for Pb, Cr, and As in children are 8.05, 3.16, and 1.32, respectively, significantly exceeding the permissible values (HI > 1). This indicates a high health risk for children in areas near waste incineration facilities and ash disposal sites. The HI values for adults are all below 1

(1.31×10^{-2} to 8.72×10^{-1}). The findings align with various investigations concerning the health hazards associated with metals present in fly ash from municipal solid waste and soils contaminated by municipal solid waste, as reported by Shi *et al.* (2020). They examined the risks posed by oral ingestion, dermal contact, and inhalation, indicating Hazard Indices (HI) for children: As (1.0612), Pb (2.0676), and Cd (0.975).²¹ Similarly, Gujre *et al.* (2021) reported corresponding HI values for children: Cr (2.017) and Zn (0.060).²⁸

The average HI values in bottom ash are lower than those in fly ash, ranging from 5.99 to 0.084 for children and 0.648 to



Table 3 Spearman correlation coefficients in ash samples collected from municipal waste incinerators and industrial waste incinerators

Metals fractions		Fly ash						Bottom ash					
		Total metal	CF	BF	pH	TC	OC	Total metal	CF	BF	pH	TC	OC
As	F1	0.54	0.071	0.071	0.61	-0.54	-0.61	0.14	0.14	0.14	-0.43	-0.43	-0.071
	F2	0.54	0.29	0.29	0.036	-0.25	0.036	0.5	-0.8	-0.79	0.071	0.21	0.36
	F3	0.68	-0.32	-0.32	-0.14	-0.75	-0.43	0.93	-0.2	-0.18	0.18	-0.21	-0.39
	F4	0.38	0.13	0.13	-0.29	-0.36	0.34	-0.25	0.07	0.071	-0.5	-0.43	0.036
	F5	0.89	-0.54	-0.54	0.11	-0.61	-0.5	0.79	-0.8	-0.79	-0.36	-0.43	0.071
Cd	F1	0.32	0.43	0.43	0.79	-0.46	-0.64	0.32	-0	-0.04	-0.57	-0.89	-0.43
	F2	0.43	0.5	0.5	0.21	0.43	0.071	0.79	-0.7	-0.68	0.14	0.11	0.25
	F3	0.29	-0.36	-0.36	0.25	-0.071	-0.14	0.82	-0.4	-0.36	-0.29	-0.036	0.54
	F4	-0.18	-0.18	-0.18	-0.071	-0.14	-0.46	0.49	0.13	0.13	-0.31	0.018	0.45
	F5	0.93	-0.32	-0.32	-0.11	0.036	0.25	0.89	-0.9	-0.86	0.071	-0.29	-0.18
Cu	F1	0.29	0.93	0.93	0	-0.14	0.29	0.89	0.57	0.57	-0.36	-0.39	-0.11
	F2	0.5	0.21	0.21	0.071	0.82	0.25	0.93	0.93	0.93	-0.18	-0.36	-0.21
	F3	0.79	0.68	0.68	0.071	-0.29	-0.11	0.75	0.89	0.89	-0.036	-0.071	-0.036
	F4	0.39	0.86	0.86	-0.54	-0.036	0.71	0.89	0.82	0.82	-0.11	-0.14	-0.036
	F5	0.82	0	0	-0.25	0.18	-0.071	0.54	0.04	0.036	-0.5	-0.071	0.36
Cr	F1	0.29	0.74	0.74	0.18	-0.32	0.071	0.64	0.86	0.86	0.29	-0.036	-0.43
	F2	0.88	0.31	0.31	-0.11	0.071	0.5	0.75	0.82	0.82	0.5	0.61	0.11
	F3	0.52	0.9	0.9	0.32	-0.5	-0.21	0.82	0.68	0.68	0.54	0.21	-0.57
	F4	0.4	0.76	0.76	0.68	-0.14	-0.21	0.75	0.46	0.46	0.71	0.39	-0.54
	F5	0.9	0.19	0.19	-0.14	-0.14	0.46	0.82	0.07	0.071	0.036	0	0
Pb	F1	0.68	0	0	0.43	-0.036	-0.43	0.82	0.54	0.54	-0.29	-0.46	-0.18
	F2	0.75	0.21	0.21	0.14	0.75	0.071	0.25	0.82	0.82	-0.36	0.21	0.75
	F3	0.5	0.18	0.18	-0.21	0	-0.11	0.64	-0.5	-0.5	0.21	-0.25	-0.71
	F4	0.75	0.14	0.14	0.36	0.43	-0.14	0.61	0.64	0.64	-0.39	-0.14	0.14
	F5	0.75	-0.61	-0.61	0.43	0.036	-0.43	0.54	-0.6	-0.61	0.21	-0.036	-0.32
Zn	F1	0.61	0.61	0.61	-0.071	0.036	0.43	0.86	0.86	0.86	-0.79	-0.61	0.21
	F2	0.79	0.36	0.36	0.29	-0.11	-0.11	0.75	0.61	0.61	-0.96	-0.64	0.39
	F3	0.71	0.39	0.39	0.071	-0.18	-0.071	0.82	0.5	0.5	-0.29	-0.29	-0.071
	F4	-0.036	0.43	0.43	-0.5	0.64	0.71	0.39	0.32	0.32	0.32	-0.11	-0.71
	F5	0.18	-0.43	-0.43	0.93	-0.21	-0.93	0.32	-0.5	-0.5	-0.21	0	0.036

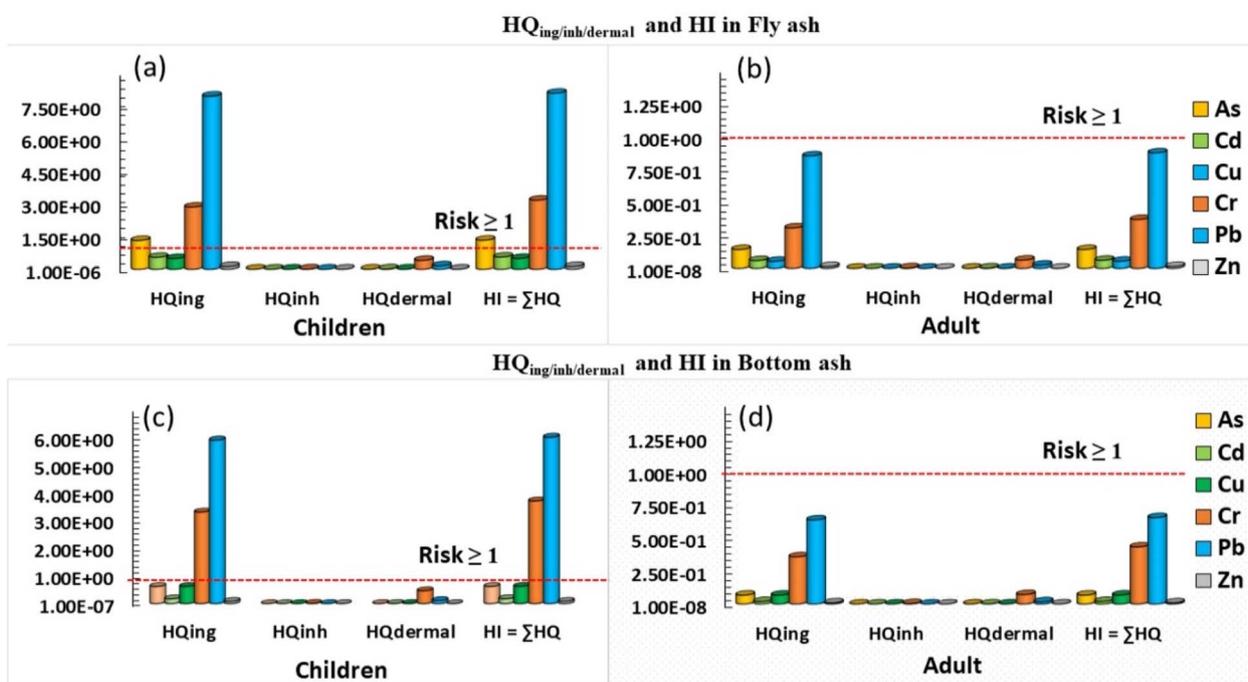


Fig. 8 Hazard quotient (HQ) and hazard index (HI) of heavy metals for children (a, c) and adults (b, d) through three main exposure pathways: ingesting (HQing), inhalation (HQinh), and dermal (HQdermal) in fly ash and bottom ash samples.



0.009 for adults. The highest non-carcinogenic risk for both children and adults was still associated with Pb, while Zn poses the lowest risk (ANOVA, $p = 0.05$, $p = 0.756$). Specifically, the HI values for children for Pb and Cr are 5.99 and 3.70, respectively, exceeding the permissible threshold ($HI > 1$). The HI values in this study are higher than the metal risk assessment in the surrounding soil of incinerators conducted by Aendo. *et al.* 2022 (HI of Pb: 2.0).²²

It is evident that both Pb and Cr negatively impact the health of children in proximity to waste incineration facilities. Thus, children living near waste incineration facilities may bear the brunt of the adverse effects of metals such as Pb, Cr, and Cd. These metals are linked to health impacts and genetic alterations (genotoxicity and epigenetic modifications). Prolonged exposure to Pb can lead to chronic and acute toxicity, resulting in immune system imbalances, hypertension, anemia, intellectual impairment, digestive issues, delayed bone development, delayed milk tooth growth, vitamin D deficiency, calcium substitution, infertility, and reduced hearing capacity.^{22,57} This highlights the need for careful consideration and investigation to propose solutions for reducing the levels of Cr and Pb in waste incineration ash.

3.4.2. Carcinogenic risk. To assess the health risks for workers directly involved in incineration facilities and residents living near waste incineration areas, this study evaluated the carcinogenic risk (CR) for four metals: As, Cd, Pb, and Cr. The

Table 4 The carcinogenic risk (CR) of heavy metals in fly ash and bottom ash from waste incinerators^a

Fly ash					
Child	As	Cd	Cr	Pb	
CR _{ing}	5.91×10^{-4}	2.01×10^{-3}	4.26×10^{-3}	2.35×10^{-4}	
CR _{inh}	1.71×10^{-7}	5.64×10^{-8}	1.20×10^{-7}	3.27×10^{-8}	
CR _{dermal}	1.65×10^{-6}	5.63×10^{-6}	1.19×10^{-5}	—	
\sum CR	5.93×10^{-4}	2.02×10^{-3}	4.27×10^{-3}	2.35×10^{-4}	
Adult					
CR _{ing}	6.33×10^{-5}	2.16×10^{-4}	4.56×10^{-4}	2.52×10^{-5}	
CR _{inh}	6.15×10^{-8}	2.03×10^{-8}	4.29×10^{-8}	1.18×10^{-8}	
CR _{dermal}	2.52×10^{-7}	8.64×10^{-7}	1.82×10^{-6}	—	
\sum CR	6.36×10^{-5}	2.16×10^{-4}	4.58×10^{-4}	2.53×10^{-5}	
Bottom ash					
Child	As	Cd	Cr	Pb	
CR _{ing}	2.78×10^{-4}	1.29×10^{-4}	4.96×10^{-3}	1.75×10^{-4}	
CR _{inh}	8.04×10^{-8}	3.75×10^{-8}	1.39×10^{-7}	2.43×10^{-8}	
CR _{dermal}	7.77×10^{-7}	3.63×10^{-7}	1.39×10^{-5}	—	
\sum CR	2.78×10^{-4}	1.30×10^{-4}	4.97×10^{-3}	1.75×10^{-4}	
Adult					
CR _{ing}	2.97×10^{-5}	6.94×10^{-5}	1.06×10^{-3}	1.88×10^{-5}	
CR _{inh}	2.90×10^{-8}	6.53×10^{-9}	9.99×10^{-8}	9.28×10^{-5}	
CR _{dermal}	1.19×10^{-7}	2.77×10^{-7}	4.24×10^{-6}	—	
\sum CR	2.98×10^{-5}	6.97×10^{-5}	1.06×10^{-3}	1.12×10^{-4}	

^a CR_{ing}, CR_{inh}, CR_{dermal}: the cancer risk from ingesting, inhalation, and dermal respectively.

remaining metals posed no carcinogenic risk. The calculated CR values for metals in fly ash and bottom ash are presented in Table 4.

The carcinogenic risk from metals in the waste ash was also assessed for both children and adults, with the risks ranked as follows for both age groups: $CR_{ing} > CR_{dermal} > CR_{inh}$. The cancer risk from oral ingestion accounted for over 99% of the risk compared to the risks from inhalation (CR_{inh}) and dermal contact (CR_{dermal}). The highest cancer risk was found for Cr (ANOVA, $p = 0.05$, $p = 0.602$), being 2.1–18 times higher in fly ash and 1.79–3.83 times higher in bottom ash compared to the other metals. The total cancer risk (CR) for all three exposure pathways in children is 1.56–9.3 times higher than in adults, ranging from 2.35×10^{-4} to 4.27×10^{-3} in fly ash and 1.30×10^{-4} to 4.97×10^{-3} in bottom ash. These results are similar to those published in the assessment of metal carcinogenic risks in waste ash and the surrounding soil of incinerators by Shi *et al.* 2020 (As: 4.09 – 1.76×10^{-5} ; Cd: 2.66 – 6.20×10^{-4} ; Cr: 1.52 – 3.54×10^{-5}),²¹ Aendo *et al.* 2020 (Cd: 2.8×10^{-4} , Cr: 2.0×10^{-4}).²² Thus, in both fly ash and bottom ash, the cancer risks from As, Cd, and Pb for both children and adults fell within acceptable limits (10^{-6} to 10^{-4}). However, for Cr, the cancer risk exceeded permissible thresholds ($>10^{-4}$). Long-term exposure to hexavalent chromium compounds (Cr VI) causes gastric ulcers, intestinal ulcers, hepatitis, and nephritis. Particularly for workers directly involved in incineration, inhaling Cr(vi) fumes and dust can lead to lung, nasal, and sinus cancers. Exposure to Cr(vi) can also impact pregnancy and lead to miscarriages. Specifically, children born to mothers exposed to Cr(vi) may experience respiratory issues and congenital defects.^{57–59} Inorganic forms of arsenic have a high potential to cause cancer in the skin, lungs, bladder, urinary tract, kidneys, and liver. Exposure to Pb and Cd, whether through occupation or the environment, increases the risk of various cancers, especially lung cancer, prostate cancer, and pancreatic cancer. Lead and cadmium poisoning leads to underdeveloped gray matter in the brain, reduced intelligence quotient (IQ) in children, and significantly impacts both physical and mental development. At very high levels, lead toxicity can result in death.^{60,61} Considering the health risks to humans, strict control policies and waste management plans are essential, especially for developing countries where incineration is a primary waste disposal method. In accordance with a 2002 report from Korea, notable concentrations of Pb and Cr metals are observed in incinerator ash constituents, including paper (Pb: 106 mg kg^{-1} , Cr: 28 mg kg^{-1}), rubber, and leather (Pb: 95 mg kg^{-1} , Cd: 31 mg kg^{-1}), wood (Pb: 33 mg kg^{-1}), plastic (Cr: 68 mg kg^{-1} , Pb: 13 mg kg^{-1}), textile (Pb: 47 mg kg^{-1}), and miscellaneous sources (Cr: 30 mg kg^{-1} , Pb: 219 mg kg^{-1}). This observation underscores the substantial emission of Pb and Cr from paper and plastic materials. Consequently, the implementation of rigorous and effective waste management protocols is imperative. In Vietnam, the annual generation of plastic waste surpasses 1.8 million tons. Per capita plastic waste has surged more than tenfold in the preceding three decades to reach 41.3 kg per capita in 2018, rendering it the third highest in Southeast Asia. Despite the government's aspiration to achieve zero single-use



plastic waste nationally by 2025, these endeavors have garnered limited attention. Immediate investment is essential in Vietnam to fortify this sector, entailing measures such as augmenting environmental levies on plastic products and bags, overhauling household waste collection and treatment procedures, and instituting a novel extended producer responsibility scheme. It is imperative to comprehensively grasp public perceptions and responses to plastic pollution to formulate behavioral interventions and awareness campaigns in collaboration with local, national, and international non-governmental organizations and governmental entities.

4. Conclusions

This study represents one of the earliest investigations into the levels of contamination by five chemical fractions of heavy metals present in waste ash originating from both industrial and residential garbage incinerators in Vietnam. Fly ash exhibited a notable concentration of volatile metals (As, Cd, Pb, and Zn) distributed homogeneously within the material, whereas bottom ash from domestic incinerators displays higher quantities of Cr and Cu. In domestic waste incinerators, Cu predominantly existed in carbonate (F2) and residue fractions (F5), while Zn and Cr were primarily present in F5; Pb exhibits a decreasing proportion from F2 to F4. Among the metals studied, As and Cd occupied a larger fraction of the mobile F1 fractions. In industrial furnaces, Pb, Cu, and Zn were predominantly concentrated in F2 and F4, while Cr was mostly found in F5. Moreover, the study demonstrates a significant association between Pb's bioavailability and fractions F1, F2, and F4. Fractions F1, F2, and F3 exhibited a relatively strong correlation with bioavailability for Cu, Cr, and Zn. For both children and adults, lifetime cancer risk (CR) and non-cancer risk (HI) were assessed. The following is the sequence of the daily intake of fly ash and bottom ash through the three exposure pathways: intake > skin contact > breathing directly. Through oral ingestion, direct inhalation, and skin contact, children are predicted to consume an average of 9.3 times, 6.5 times, and 2.8 times more than adults. For the children, the HI values for Pb, Cr, As in fly ash and Pb, Cr in bottom ash were much higher than the allowable limits (HI > 1). Every adult's HI value is less than 1. Children's and adults' carcinogenic risk from metals in the waste ash was also evaluated, and risks were ranked as follows for the two age groups: Dermal > CRing > CRinh. Adult and child As, Cd, and Pb cancer risks were within acceptable limits (10^{-6} to 10^{-4}) in both fly ash and bottom ash. But for Cr, the risk of cancer was higher than acceptable bounds ($>10^{-4}$). These findings underscore the substantial health risks posed by incinerator ash, particularly for nearby children, necessitating the implementation of effective waste management strategies for incinerator residues.

Author contributions

Conceptualization: [Thi Thu Thuy Nguyen]; methodology: [Thi Thu Thuy Nguyen]; formal analysis and investigation: [Thi Thu Ha Pham, Minh Binh Tu, Thi Hue Nguyen]; writing – original

draft preparation: [Truong Xuan Vuong]; writing – review and editing [Truong Xuan Vuong, Thi Thu Thuy Nguyen], [Anh Quoc Hoang]; funding acquisition [Thi Thu Thuy Nguyen]; resources [Thi Thu Thuy Nguyen], [Truong Xuan Vuong]; supervision: [Truong Xuan Vuong], [Minh Binh Tu]; sample preparation and analysis on SEM and EDS mapping [Thi Thu Phuong Nguyen].

Conflicts of interest

The authors have no competing interests to declare that are relevant to the content of this article.

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