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1. Introduction

Andrographis paniculata (A. paniculata) is an annual herb and a member of the family Acanthaceae that expands up to 1–2 meters in height mostly in shady places. Due to its exceptionally bitter flavor, it is often referred to as the "King of Bitters". A. paniculata has a rich history in traditional medical practices in Southeast Asia, India, and China, and has been used for several generations.1,2 Traditionally, the leaves and roots of this plant have been utilized for a variety of therapeutic purposes in Asia and Europe. In China, this plant extract is used to eliminate toxins and cool the body.3,4 In India, it is used to cure diabetes, dysentery, herpes, jaundice, helminth infection, and ulcers, and is topically used to treat snake bites and skin infections.^{5,6}

Synthesis of novel pyrazole acetals of andrographolide and isoandrographolide as potent anticancer agents†

Siva Kumar Rok[kam](http://orcid.org/0000-0002-6671-2022), \mathbf{D}^{a} Manohar [Bh](http://orcid.org/0000-0002-1333-9316)ujel,^a Dolly Jain,^c Lakshminath [Sri](http://orcid.org/0000-0003-3994-3615)pada,^a Srinivas Nanduri, \mathbf{D}^{b} Avinash Bajaj \mathbf{D}^{c} and Nageswara Rao Golakoti $\mathbf{D}^{\star\mathsf{a}}$

Globally, cancer is the most prevalent chronic disease-related cause of death. Although there are many anticancer drugs, some of them have adverse effects. Due to their limited side effects, natural products are preferred over synthetic drugs. Andrographolide and its derivatives are known to be potent anticancer agents. In this context, sixteen novel 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide and isoandrographolide (1a–1h, 2a–2g, 2i) from 3-aryl-1-H-pyrazole-4-carboxaldehydes (a–i) were synthesized. All the synthesized compounds were characterized using ¹H NMR, ¹³C NMR, HRMS, FT-IR, and UV-vis spectroscopy. All the compounds were evaluated against a panel of 60 different human cancer cell lines for their anticancer potential at NCI, USA. Four compounds, having promising GI₅₀s (50% growth inhibitory activity) on all 60-cell lines were selected for further in vitro studies. Out of these four compounds, compound 1g exhibited the best IC_{50} (3.08 μ M) against the colon cancer cell line, HCT-116. Cell cycle analysis, annexin V-FITC/PI, and ROS assays revealed that the apoptosis of HCT-116 cells induced by compound 1g could be mainly attributed to the elevated levels of intracellular ROS. Further, the structure–activity relationship revealed that the pyrazole moiety of andrographolide plays a key role in their anticancer properties. These compounds were further examined for in silico ADMET and Lipinski characteristics to assess their potential as lead compounds. **PAPER**
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Numerous research investigations have demonstrated that A. paniculata extracts exhibit a plethora of pharmacological activity, including anti-cancer, anti-inflammatory, anti-viral, anti-malarial, anti-oxidant, anti-diarrheal, hepatoprotective, and immunostimulatory activities.⁵⁻⁹ The observed activities of the methanolic extract are primarily attributed to the plant's major phytochemical constituent andrographolide.^{10,11} The recent research findings revealed that andrographolide and its derivatives exert their anticancer effects by cell cycle arrest via increasing the levels of tumor suppressor proteins p53, p21 and p16, inducing apoptosis.¹²–¹⁴ Andrographolide and its derivatives also suppress TNF- α , TGF- β 1, IL-1 β , IL-6 levels, angiogenic factor VEGF, NF-kB signaling pathway, and causes degradation of proto-oncogene tyrosine-protein kinase Src. Further, andrographolide acts as antiangiogenic and found to inhibit cell migration and invasion in A549, HT29, CT26, lovo, and cholangiocarcinoma cells by interfering with the production of proteins and cellular pathways that are crucial for the spread of cancer.¹⁵–¹⁸

Our previous studies on various andrographolide and isoandrographolide derivatives such as C-12-alkyl/benzyl/ sulfamoyl-phenyl/amino-14-deoxyandrographolides, 3,19benzylidene-andrographolide/isoandrographolide, 3,19-heteroaromatic acetals of andrographolide/isoandrographolide, 14 deoxy-12-hydroxy-andrographolide acetals displayed excellent

a Department of Chemistry, Sri Sathya Sai Institute of Higher Learning, Prasanthi Nilayam, Andhra Pradesh, India. E-mail: gnageswararao@sssihl.edu.in

b Department of Chemical Sciences, National Institute of Pharmaceutical Education and Research, Balanagar, Hyderabad, Telangana, 500037, India

Laboratory of Nanotechnology and Chemical Biology, Regional Centre for Biotechnology, NCR Biotech Science Cluster, 3rd Milestone, Faridabad-Gurgaon Expressway, Faridabad, Haryana, 121001, India

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anticancer activity on a panel of 60 different human cancer cell lines (at NCI).¹⁹⁻²³ Further, the mechanism of action for the most potent derivatives was determined against breast cancer cell line MDA-MB-231. Since the acetals of andrographolide and isoandrographolide acted as very good anticancer agents, in our present work, we intended to link the pyrazole moiety to andrographolide and isoandrographolide and investigate their potential as anticancer agents. Pyrazoles were chosen because of their signicant importance as pharmacophores in forming the basis for diverse pharmaceutical compounds.²⁴⁻²⁶

2. Experimental

2.1. Synthesis

All the chemicals were obtained from Sigma Aldrich, HiMedia, and Alfa Aesar, while the solvents were distilled before use.

2.1.1. Synthesis of N' -(1-arylethylidene) isonicotinohydrazides. Isoniazid (25 mmol) was added to a solution of substituted acetophenones (25 mmol) in methanol (80 mL) and stirred at room temperature. 2 mL of glacial acetic acid was then added, and the mixture was refluxed for 8 hours. The progress of the reaction was monitored by TLC analysis. Crushed ice was added once the reaction was complete, and the resultant solid was filtered and dried. The product obtained was then recrystallized with ethanol.

2.1.2. Synthesis of 3-aryl-1-H-pyrazole-4-carboxaldehydes $(A-I)$. Aliquot amounts of N' -(1-arylethylidene) isonicotinohydrazides (20 mmol) were added to the Vilsmeier-Haack reagent prepared from N,N-dimethylformide (20 mL) and POCl₃ (60 mmol) at 0 °C. The reaction mixture was stirred and heated at 65 \degree C for 4 hours after the addition. After the reaction was completed, the resultant mixture was poured into crushed ice. The solution was then neutralized with saturated $Na₂CO₃$ to obtain a solid. The product was purified by column chromatography using hexane and ethyl acetate in a 65 : 35 ratio as eluent.

2.1.3. Synthesis of 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide (1a–1h). Andrographolide (1 mmol) and 3-aryl-1H-pyrazole-4-carbaldehydes (4 mmol) were dissolved in DMSO (3 mL) at room temperature. To this, pyridinium para-toluene sulfonate (PPTS) was added in a catalytic amount (0.08 mmol) and the mixture was heated at 70 \degree C for 24 hours. After completion of the reaction, the mixture was cooled down to room temperature and neutralized with freshly prepared cold saturated sodium bicarbonate solution. The resulting solution was then extracted three times (15 mL each) with DCM and the obtained organic layer was thoroughly washed with brine solution and dried over anhydrous sodium sulfate. DCM was removed using a rota evaporator. The crude product obtained was then purified using column chromatography (hexane: acetone, 60 : 40).

2.1.4. Synthesis of 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide (2a–2g, 2i). First, isoandrographolide was synthesized by dissolving 8 g of andrographolide in 150 mL of conc. HCl and stirred the mixture for 24 hours at room temperature. After completion of the reaction, crushed ice was added and extracted three times with DCM (25 mL each). The

organic layer was thoroughly washed with freshly prepared cold saturated sodium bicarbonate and brine solutions, dried over anhydrous $Na₂SO₄$ and recrystallized from ethyl acetate to get pure isoandrographolide. Then isoandrographolide was taken as starting material along with 3-phenyl-1H-pyrazole-4 carbaldehydes and followed the same procedure as above to synthesize the compounds 2a–2g, 2i.

2.2. Instrumentation

The 1 H and 13 C-NMR spectra were recorded on BRUKER ASCEND 400 MHz and 100 MHz respectively. Tetramethylsilane (TMS) was used as an internal standard. Mass spectra were generated using AGILENT 6550 Q-TOF LC-MS (HRMS). Acetonitrile was used as a solvent to record UV-vis spectra in the range 200–600 nm using a Shimadzu 2450 spectrophotometer. Using KBr pellets, FT-IR spectra were acquired in the range 400 and 4000 cm^{-1} on AGILENT Cary 630 spectrophotometer. Agilent 1260 Infinity high-performance liquid chromatography system, equipped with a quaternary solvent delivery system, inline degasser, auto sampler, and photodiode array detector, was used. Samples were examined using water (10%) and acetonitrile (90%) as mobile phase on a Zorbax Extend-C18 (4.6 mm 250 mm, 5 µm) column at a flow rate of 1 mL min⁻¹. At 25 °C, the separation was performed with an injection volume of $15 \mu L$. A diode array detector was utilized to capture the chromatogram between 200 and 700 nm. RSC Advances

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2.3. In silico studies

All the active compounds were screened for their in silico Lipinski and ADMET properties using the pkCSM tool.²⁷

3. Biological testing

3.1. Anticancer studies

3.1.1. NCI protocol. National Cancer Institute (NCI), Bethesda, Maryland, USA, screened all the sixteen compounds for their anticancer activity against nine different cancer types, consisting of a panel of 60 human cancer cell lines. The assessment of growth inhibition percentage compared to untreated control samples was conducted using spectrophotometry. A continuous 48 h exposure approach was employed, and cell viability was determined using a sulforhodamine B protein assay. A comprehensive protocol has been provided in the ESI.†

3.1.2. Cell culture. Cancer cells (HCT-116) and Vero E6 cells were cultured in Dulbecco's modified Eagle's medium supplemented with 10% fetal bovine serum (FBS) and $1\times$ penicillinstreptomycin solution at 37 $\mathrm{^{\circ}C}$ and 5% CO_{2} . Cells were grown to 70 to 80% confluency and harvested using 1X trypsin.

3.1.3. Cell viability assay. Cancer cells (HCT-116) and Vero E6 cells $(5 \times 10^3$ cells per well) were seeded in 96-well plates. Following a 24 hour incubation period, the cells were exposed to various concentrations (ranging from 100 to 0.097 μ M) of test samples for 48 h. Subsequently, MTT (3-(4,5-dimethylthiazol-2 yl)-2,5-diphenyltetrazolium bromide) (10 mL) was introduced into each well and allowed to incubate for an additional 3 h.

After the incubation, the media was removed and the crystals were dissolved in 100 µL DMSO. Then, the absorbance of the plate was taken at 570 nm using a spectrophotometer. The values were plotted (% cell viability vs. concentration) to obtain the IC_{50} as per the previous protocol.²⁸

3.1.4. Cell cycle assay. Cancer cells $(HCT-116)$ $(3 \times 10^5$ cells per well) were seeded on a 6-well plate. After a 24 h incubation, the medium was replaced with an incomplete medium lacking FBS to synchronize the cells. After 8 h synchronization period, the incomplete medium was removed, and cells were exposed to the test samples (5 μ M and 10 μ M) for 24 h. Subsequently, the cells were harvested using 1X trypsin, washed twice with cold 1X PBS, and fixed in 70% ethanol at 4 \degree C overnight. After two additional washes with PBS, the cells were treated with a staining buffer containing ribonuclease A (100 μ g mL⁻¹) and

propidium iodide (50 $\mu\text{g}\,\text{mL}^{-1})$ in 1 \times PBS overnight. Finally, the cells were analyzed using a flow cytometer (FACSVerse, BD Biosciences, NJ, USA).²⁸

3.1.5. PI uptake assay. Cancer cells (HCT-116) $(1 \times 10^5 \text{ cells})$ per well) were seeded in a 12-well plate. Following a 24 hour incubation period, cells were treated with the test samples (10 μ M) for 48 h. Subsequently, cell harvesting was conducted using 1X trypsin, followed by a washing step with a 1X PBS solution. Staining with propidium iodide (PI) (100 µg mL⁻¹) was performed, and the stained cells were promptly acquired and analyzed using a flow cytometer (FACSVerse, BD Biosciences, NJ, USA).²⁸

3.1.6. Apoptotic assay. Cancer cells $(HCT-116)$ (3 \times 10⁵ cells per well) were seeded in a 12-well plate. After 24 h incubation, cells were treated with test samples $(10 \mu M)$, and further

Scheme 1 Synthesis of N'-(1-arylethylidene) isonicotinohydrazides (A-I) and 3-aryl-1H-pyrazole-4-carbaldehydes (a-i).

Scheme 2 Synthesis of 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide (1a-1h)

Scheme 3 Synthesis of isoandrographolide and 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide (2a–2g and 2i).

incubated for 48 h. The supernatant (for dead cells) and adherent cells were both collected by introducing 1X trypsin for 5 min, followed by centrifugation at 4000 rpm for 5 min. The collected cells were then resuspended in 100 μ L of 1X annexin V binding buffer. Annexin V–FITC staining $(1 \mu L)$ was added to the cells, which were then vortexed and left in the dark at $4 \degree C$ for 15 min. Subsequently, propidium iodide solution (100 $\mu\mathrm{g\,mL^{-1}})$ was added to each vial, incubated for 15 min, and cells were promptly analyzed using a flow cytometer (FACSVerse, BD Biosciences, NJ, USA).²⁸

3.1.7. ROS assay. Cancer cells (HCT-116) $(1 \times 10^5 \text{ cells per})$ well) were plated in a 12-well plate. Following 24 h incubation period, cells underwent a wash with 1X PBS and were exposed to 2.5 µL (10 mM) DCHF-DA (2,7-dichlorodihydrofluorescein diacetate) for 30 min in 0.2% FBS in DPBS. Subsequent to this incubation, another wash with $1 \times$ PBS was performed to eliminate excess dye solution. Cells were then treated with the test sample (10 μ M), and incubated for an additional 3 h. Cell harvesting was carried out using 1X trypsin, and cells were

observed the formation of 3-aryl-1H-pyrazole-4-carbaldehydes (a–i) instead of 1-isonicotinoyl-3-aryl-1H-pyrazole-4 carbaldehydes (Scheme 1). To investigate the effect of $POCI₃$ in this reaction, we repeated the procedure using 2 equivalents of POCl₃. However, we again observed the formation of 3-aryl- $1H$ -pyrazole-4-carbaldehydes $(a-i)$. This is the first report of the synthesis of 3-aryl-1H-pyrazole-4-carbaldehydes in a two-step reaction. Four of the nine synthesized carbaldehydes are novel $(c, e, f, and g)$ in our work. Other five of these carbaldehydes were synthesized by You et al., following a three-step reaction and patented the work.³⁰ In step-3, 3-aryl-1H-pyrazole-4carbaldehydes (a–i) were reacted with andrographolide (ADG) (Scheme 2) and isoandrographolide (ISOADG) (Scheme 3) in the presence of mild acid catalyst pyridinium para-toluene sulfonate (PPTS) to obtain the desired products (1a–1h, 2a–2g, 2i). All the products were purified using column chromatography.

4.1.1. Probable mechanism for the formation of 3-aryl-1Hpyrazole-4-carbaldehydes (a–i). Step-1: formation of iminium ion intermediate from DMF and POCl₃.

immediately analysed using a flow cytometer (FACSVerse, BD Biosciences, NJ, USA).²⁸

4. Results and discussion

4.1. Chemistry

We intend to synthesize 1-isonicotinoyl-3-aryl-1H-pyrazole-4 carbaldehydes and further prepare their acetals with andrographolide and isoandrographolide. In step-1, we synthesized nine N' -(1-arylethylidene) isonicotinohydrazides (A–I), and in step-2, they were to be converted to 1-isonicotinoyl-3-aryl-1Hpyrazole-4-carbaldehydes $(a-i)$ in the presence of DMF, POCl₃ (1.2 equivalents) at 65 °C as reported by Beniwal et al.²⁹ However, surprisingly, when we used 3 equivalents of $POCl₃$, we

Step-2: reaction of N' -(1-arylethylidene) isonicotinohydrazides with iminium ion.

Step-3: nucleophile attack on the amide carbonyl to form 3 aryl-1H-pyrazole-4-carbaldehydes.

The amide carbonyl, as seen in step-3 above, is linked to two electron-withdrawing groups (pyridine and pyrazole-4 carboxaldehyde). This makes it more prone to react with nucleophiles. Hence, the possible mechanism for the formation of 3-aryl-1H-pyrazole-4-carbaldehydes (a–i) instead of 1-isonicotinoyl-3-aryl-1H-pyrazole-4-carbaldehydes could involve nucleophilic attack at the amide carbonyl, resulting in the hydrolysis of the amide bond.

4.1.2. Characterization. The UV-vis spectra exhibited bands between 230 and 250 nm for all the acetals. In the IR spectra of

3,19-(NH-3-aryl-pyrazole) acetals of andrographolide, a peak in the range 3520–3401 cm^{-1} was observed for –OH whereas a peak at \sim 1675 cm⁻¹ corresponds to exocyclic C=C groups. These peaks were absent for 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide. –NH stretching peak in all the acetals appeared in the range 3321–3270 cm^{-1} . The peak at \sim 1755 cm⁻¹ corresponds to the carbonyl group (C=O) of the lactone moiety. Distinct absorptions at approximately 1200 to 1100 cm^{-1} can be attributed to the C–O stretching bands. Compounds with aryl chloride have bands at \sim 1150 cm⁻¹ (Ar– Cl stretch). Compounds with aryl fluorides showed a band around 1234 and 1219 cm^{-1} (Ar-F stretch), and compounds with aryl bromides have a band between 1018–1010 cm−¹ (Ar–Br stretch).

In the ¹H NMR spectra of all the acetals, the signal around δ 7.80 is attributed to the presence of H-5', while peaks within the range of δ 7.90 to 6.95 correspond to the aromatic protons

linked to the pyrazole ring $(H-7'$ to H-11'). The proton signal for the unsaturated lactone (H-14) in 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide is observable at $\sim \delta$ 7.29. Additionally, the peak at around δ 6.95 corresponds to H-12 in 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide. Across all compounds, a singlet peak at \sim δ 5.82 is for H-21.

In the 13 C NMR spectra, signals approximately within the range of δ 173-170, confirm the presence of a carbonyl group $(C=O)$. The carbons in the pyrazole ring, aromatic ring, and olefin carbons were identified in the \sim δ 163 to 110 range. Notably, in the case of 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide, the olefin carbons such as C-8 and C-17 are absent, leading to the absence of a signal around δ 110. A distinct signal around δ 90, associated with C-21 bonded to two oxygen atoms, strongly supports the formation of acetals. Carbon signals related to oxygen atoms (–O–C) were observed within the range of δ 82 to 66.^{31,32} The complete characterization and spectral data of the compounds (A–I, a–i, 1a–1h, 2a–2g, and 2i) is provided in the ESI.†

4.2. Biological evaluation

4.2.1. NCI anticancer screening. Anticancer potential of the sixteen newly synthesized 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide and isoandrographolide was tested against a panel of 60 different human cancer cell lines spanning across nine human cancer types such as leukaemia, non-small lung cancer, colon, central nervous system (CNS), melanoma, ovarian, renal, prostate, and breast cancer by National Cancer Institute (NCI, USA). Initially, all sixteen compounds (1a–1h, 2a–2g, and 2i) were tested at 10 µM (one dose) on 60 cell lines. Out of which, seven compounds $(1a-1g)$ satisfied pre-determined threshold inhibition criteria in a minimum number of cell lines (one dose study result of these seven compounds is given in the ESI (Table S1)).† Therefore, these seven compounds were further selected for their five dose screening to obtain GI_{50} (50% growth inhibitory activity), TGI (Total growth inhibition), and LC_{50} (50% lethal concentration) values on all the cell lines by NCI. Interestingly, all seven compounds (1a–1g) that are active on most of the cell lines at 10 μ M were 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide. However, 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide have exhibited less % growth inhibition indicating that these derivatives are less active on the majority of the cell lines due to which they were not selected for five dose studies. RSC Advances

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4.2.2. Five dose study. Five dose screening results revealed that four out of seven compounds: meta-substituted –Br and –Cl derivatives (1c and 1b) and para-substituted –Br and –OCH₃ derivatives (1e and 1g) are highly active and displayed excellent $GI₅₀S$ against majority of the cell lines (Table 1). Among the four compounds, compound 1c was observed to be the most potent, followed by compound 1b against almost all the cell lines.

Both meta and para substituted -Br derivatives 1c and 1e showed the highest activity with $GI₅₀S$ 1.28 μ M and 2.13 μ M respectively, against breast cancer cells MCF7, and compounds 1b and 1g demonstrated better activity on the MDA-MB-468 cell line with GI₅₀s 1.88 μ M and 2.17 μ M.

All four compounds showed promising activity on four different leukemia cancer cell lines HL-60(TB) $(1c, Gl₅₀: 1.87)$ μ M), RPMI-8226 (1b, GI₅₀: 2.25 μ M), K-562 (1e, GI₅₀: 2.24 μ M) and CCRF-CEM (1g, GI₅₀: 2.08 μ M). Compounds 1c (GI₅₀: 1.66 and 1.75 μ M) and 1b (GI₅₀: 1.79 and 1.93 μ M) have exhibited potent activity against the colon and ovarian cancer cell lines HCT-15 and NCI/ADR-RES, respectively. However, compounds 1e and 1g showed their superior activity on HCT-116 ($GI₅₀$: 1.93 and 1.86 μ M) and OVCAR-3 (GI₅₀: 2.08 and 2.26 μ M) cell lines.

The compounds 1c, 1b, and 1e have the best activity on the melanoma cancer cell line LOX IMV ($GI₅₀S$: 1.39, 1.44 and 1.37 μ M respectively), the renal cancer cell line RXF 393 (GI₅₀S: 1.81, 1.66 and 1.93 μ M), non-small lung cancer cell line NCI-H226 (GI_{50s} : 1.62, 1.73 and 1.90 μ M) and the CNS cancer cell line SF-539 (GI₅₀s: 1.74, 1.77 and 2.20 μ M). In contrast, compound 1g displayed better activity against SK-MEL-5 (GI₅₀: 1.52 μ M), UO-31 (GI₅₀: 1.70 μ M), NCI-H460 (GI₅₀: 1.91 μ M) and SNB-75

(GI₅₀: 1.75 μ M) cell lines. All four compounds were more active on the prostate cancer cell line PC-3 with $GI₅₀S$ 3.07 μ M $(1c)$, 4.08 μ M $(1b)$, 2.13 μ M $(1e)$ and 2.98 μ M $(1g)$.

Total growth inhibition (TGI) and 50% lethal concentration (LC_{50}) of these four compounds against the 60 human cancer cell lines are provided in the ESI (Tables S2 and S3).†

4.2.3. Effect of 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide on proliferation of colon cancer HCT-116 cell line. Out of the above four compounds (1b, 1c, 1g, and 1e), three compounds (1b, 1c, and 1g) that have demonstrated excellent anticancer properties in the five dose study on the 60-cell line panel at NCI were selected for further cytotoxicity studies on the colon cancer cell line HCT-116. As shown in Table 2, the compound 1g significantly inhibited cell proliferation of HCT-116 cells with an IC₅₀ of 3.08 μ M followed by 1c (3.16 μ M) and 1b (3.40 μ M). Since the compound 1g exhibited a better IC₅₀ than the other compounds, it was further selected to test against Vero E6 cells (the immortalized cell line established from kidney epithelial cells of the African green monkey). Compound 1g was found to be non-toxic to Vero E6 cells till 20 μ M concentration, and its cytotoxic concentration CC₅₀ was 47.97 μM.

4.2.4. Cell cycle analysis of 1g. As compound 1g exhibited the best IC_{50} value against the HCT-116 cell line and a low inhibitory effect on Vero E6 cells, we sought to investigate whether its inhibition was linked to its interference with the cell cycle. To identify this, HCT-116 cells were subjected to dosedependent treatments with $1g$ at concentrations of 5 μ M and 10μ M. Remarkably, our observations, as depicted in Fig. 1a-d, indicate that compound 1g induced cell accumulation in the G0/G1 phase at both 5 μ M and 10 μ M concentrations. This effect showed a concentration-dependent increase in the percentage of cells in the G0/G1 phase compared to the control (untreated cells, Fig. 1a). At a concentration of 5 μ M (Fig. 1b and d), 1g was responsible for the accumulation of HCT-116 cells in the G0/G1 phase, leading to a reduced percentage of cells in the S and G2/ M phases. At a concentration of 10 μ M (Fig. 1c and d), it was observed that a slightly higher percentage of cells were arrested in the G0/G1 as compared to the cells treated with $1g$ at 5 μ M concentration. Arresting cell cycle progression in G0/G1 phase indicates cells undergo repair mechanisms or follow the apoptotic pathway. To determine this, propidium iodide uptake, annexin V-FITC/PI, and reactive oxygen species assays (ROS) were performed.

4.2.5. Annexin V-FITC/PI assay. The extent of apoptosis triggered by compound 1g in HCT-116 cells was investigated using annexin V/propidium iodide dual staining that enables the identification of different cell states like live cells, early apoptotic cells, late apoptotic cells, and necrotic cells.³³ It was observed that compound 1g causes a noticeable increase in the percentage of early apoptotic cells (Fig. $2a-c$). These findings provide evidence that compound-induced apoptosis is responsible for inhibiting HCT-116 cell proliferation.

4.2.6. ROS assay. Maintaining a stable redox environment is crucial for various cellular functions. Cancer cells exhibit elevated levels of reactive oxygen species (ROS) compared to healthy cells. However, cancer cells manage to uphold their

Table 1 Growth inhibition (50%) of the potent compounds against the NCI human cancer cell line panel

Table 2 IC_{50} values of the potent compounds 1b, 1c, and 1g against the HCT-116 cell line

redox equilibrium due to their robust antioxidant defenses. Nevertheless, the heightened accumulation of ROS can perturb redox homeostasis and result in substantial damage to cancer cells.³⁴ Since we noticed that compound 1g was associated with

cell apoptosis, we were interested in investigating whether this effect was linked to the generation of ROS in HCT-116 cells. Interestingly, following the exposure of HCT-116 cells to 1g (10 μ M), there was a significant rise in intracellular ROS levels when compared to the control (Fig. 3). These findings suggest that the apoptosis of HCT-116 cells induced by compound 1g can be mainly attributed to the elevated levels of ROS.

4.3. Structural activity relationship based on NCI screening

As mentioned before, our group previously investigated the anticancer properties of 3,19-benzylidene/heteroaromatic

HCT-116 cells treated with 10 μ M concentration of 1g; (d) % cells in each phase after treatment with 1g.

Fig. 2 Flow cytometry plots using Annexin V-FITC/PI staining for apoptosis. HCT-116 cells were treated with (a) control (untreated cells) (b) 10 μ M concentration of 1g. (c) % of apoptotic cells after treatment with 1g.

ROS levels in the presence of 1g.

Table 3 Comparison of % growth inhibition of benzylidene/heteroaromatic/pyrazole acetals of andrographolide on a human cancer cell line panel

	Cell line	% Growth inhibition								
Cancer		ADG	Bz-ADG	Fur-ADG	Py-ADG	Thio-ADG	1c	1b	1g	
Leukaemia	CCRF-CEM	76.51	90.97	89.84	86.21	66.78	-15.57	-14.87	99.40	
	$HL-60(TB)$	$n.a^a$	51.12	1.86	13.79	n.a	-16.26	-15.08	98.41	
	K-562	34.47	56.31	55.44	47.29	23.54	-2.18	95.00	-31.21	
	RPMI-8226	16.71	71.77	82.88	58.08	29.46	-6.49	-11.69	-27.00	
	SR	20.46	63.12	70.99	83.82	23.59	97.32	84.29	-29.91	
NSCL	NCI-H522	82.86	36.48	36.90	34.87	15.89	-55.96	-59.98	-18.40	
Colon	HCT-116	39.98	61.62	48.53	38.35	15.56	-95.92	-91.58	-19.04	
	$HCT-15$	38.11	34.48	40.82	28.68	13.59	-67.53	-59.20	-35.29	
	HT29	72.99	45.20	36.91	30.12	15.95	98.48	99.69	-26.28	
	SW-620	55.71	50.33	54.62	51.49	8.40	-81.14	-57.57	n.t	
Melanoma	LOX IMVI	78.16	66.18	77.96	77.02	27.00	-100	-100	n.t	
Ovarian	OVCAR-8	15.52	43.42	42.62	26.37	21.32	-42.49	-23.31	-3.29	
	NCI/ADR-RES	16.25	42.98	23.28	12.07	6.23	-50.15	-56.43	93.33	
Breast	MCF7	42.44	53.91	51.30	36.01	27.86	-48.20	-52.32	98.00	
	BT-549	12.88	36.15	n.a	n.a	n.a	-94.24	-92.39	-32.90	

^a n.a – not active; n.t – not tested, negative values represent lethality (example: [−]15.57% growth inhibition means 15.57% lethality).

Fig. 4 The order of anticancer activity of benzylidene/heteroaromatic/pyrazole acetals of andrographolide.

Table 4 Lipinski parameters of the active compounds

	Lipinski parameters								
Compounds	MW ^a	MR	#H.D	# $H.A$	$\text{Log } P$				
1b	538	93.67	2		4.76				
1c	582	93.67	2		4.93				
1g	534	102.90	2	8	4.07				

 a MW - molecular weight, MR - molar refractivity, #H.D - no. of. hydrogen bonds, $#H.A - no.$ of. hydrogen bond acceptors. Log P partition coefficient.

acetals of andrographolide and isoandrographolide.²² These derivatives have displayed moderate to good anticancer activity. Therefore, from these series, we have selected and compared the most active compounds and the cell lines on which they showed good activity with the top three 3,19-(NH-3-arylpyrazole) acetals of andrographolide (1c, 1b and 1g) (% growth inhibition -Table 3, Fig. 4). Interestingly, on all the cell lines, the compounds 1c, 1b and 1g exhibited several-fold better % growth inhibition than the rest of the compounds. Indeed, these three compounds are lethal to most of the cancer cell lines. Here, the better anticancer activity could be attributed to the presence of pyrazole moiety in the acetals of andrographolide or isoandrographolide.

4.4. Drug-likeness and ADMET prediction

Employing the pkCSM tool, in silico drug-likeness, pharmacokinetic, and toxicity studies (Table 4 and 5) were performed for the active compounds in order to determine their potential as drug candidates by analysing their Lipinski and ADMET parameters.

From the results, we note that the *in silico* molecular properties of the active compounds are in accordance with Lipinski's rule of five (RO5) barring molecular weights. The molecular weights of all the active compounds are in the range of 534–582

Daltons, slightly exceeding the acceptable limit (<500 Dalton). However, molecular weights of natural products are frequently considered an exception to Lipinski's rules.³⁵ Molar refractivity of the compounds was found to be in the range 40–130 and Log P was observed to be <5. The total number of N–H and O–H hydrogen bond donors is <5 and the number of nitrogen and oxygen-hydrogen bond acceptors (HBA) is <10.

We also examined the adsorption, distribution, metabolism, excretion, and toxicity effects of the active compounds using pkCSM to obtain a comprehensive picture of their therapeutic efficacy.

Table 5 provides an overview of the ADMET properties of the various compounds with potential for use as drugs. The compounds showed moderate aqueous solubility, high gastrointestinal absorption, and low BBB permeability. In addition, $log K_p$ and maximum tolerated dose values were observed to be in the acceptable range, and all the compounds are noninhibitors of CYP1A2 and hERG. Importantly, none of the compounds exhibited AMES toxicity, hERG inhibition, skin sensitization, and minnow toxicity. Compliance with Lipinski's parameters and the observed ADMET properties indicate that these molecules deserve to be investigated as potential anticancer agents.

5. Conclusion

In conclusion, nine 3-aryl-1H-pyrazole-4-carbaldehydes were synthesized, and among these four are novel. These pyrazole carbaldehydes were used as starting materials to synthesize eight 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide and eight 3,19-(NH-3-aryl-pyrazole) acetals of isoandrographolide. All the synthesized compounds (1a–1h, 2a–2g, 2i) have been characterized extensively and were initially tested at the NCI (USA) for their anticancer properties on a panel of 60 human cancer cell lines. Interestingly, 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide exhibited better anticancer activity than the acetals of isoandrographolide. Seven out of the eight 3,19-(NH-3-aryl-pyrazole) acetals of andrographolide (1a–1g) were further selected for five dose study to determine $GI₅₀$ s, total growth inhibition, and LC_{50} s by NCI. The most potent among them (1b,

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1c, and 1g) were further chosen to find out their IC_{50} s against the colon cancer HCT-116 cell line. Based on the IC_{50} values, the most potent compound 1g (IC₅₀ 3.08 μ M) was selected to investigate its mechanism of action on the HCT-116 cell line. The results revealed that 1g was responsible for the cell cycle arrest at the G0/G1 phase due to the early apoptosis of HCT-116 cells. In addition, ROS assay results indicated that the apoptosis of HCT-116 cells induced by compound 1g was mainly attributed to the elevated levels of intracellular ROS in HCT-116 cells. Further, the best three active compounds (1c, 1b and, 1g) were found to be more active than 3,19-benzylidene/heteroaromatic acetals of andrographolide and isoandrographolide, which were previously synthesized by our group. This comparison indicates that the pyrazole moiety played a crucial role in improving the anticancer activity. Further, in silico Lipinski and ADMET results of the most active compounds suggest that these can be modeled as potential therapeutic leads for the treatment of cancer. Paper
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Data availability

The data supporting this article have been included as part of the ESI.†

Author contributions

Siva Kumar Rokkam carried out the synthesis, characterization, in silico studies, data analysis, and drafted the manuscript. Manohar Bhujel and Dolly Jain performed and analysed the in vitro anticancer studies: cell viability, cytotoxicity, cell cycle analysis, staining, and ROS assays. Avinash Bajaj supervised the anticancer studies. Nageswara Rao Golakoti, Lakshminath Sripada and Srinivas Nanduri conceived the study, participated in its design, analyzed the results, and contributed to drafting the manuscript. All authors read and approved the final manuscript.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 C. Aromdee, Expert Opin. Ther. Pat., 2012, 22, 169–180.
- 2 S. G. S. Kandanur, N. Tamang, N. R. Golakoti and S. Nanduri, Eur. J. Med. Chem., 2019, 176, 513–533.
- 3 M. Jiang, F. Sheng, Z. Zhang, X. Ma, T. Gao, C. Fu and P. Li, J. Ethnopharmacol., 2021, 272, 113954.
- 4 P. Sharma, N. Kumar and R. Dudhe, Pharmacologyonline, 2011, 2, 11.
- 5 S. Kumar, B. Singh and V. Bajpai, J. Ethnopharmacol., 2021, 275, 114054.
- 6 A. Okhuarobo, J. Ehizogie Falodun, O. Erharuyi, V. Imieje, A. Falodun and P. Langer, Asian Pac. J. Trop. Dis., 2014, 4, 213–222.
- 7 L. S. Chua, Phyther. Res., 2014, 28, 1589–1598.
- 8 A. Widyawaruyanti, M. Asrory, W. Ekasari, D. Setiawan, A. Radjaram, L. Tumewu and A. F. Hafid, Procedia Chem., 2014, 13, 101–104.
- 9 A. K. Jadhav and S. M. Karuppayil, Phyther. Res., 2021, 35, 5365–5373.
- 10 R. A. Kumar, K. Sridevi, N. V. Kumar, S. Nanduri and S. Rajagopal, J. Ethnopharmacol., 2004, 92, 291–295.
- 11 Y. Dai, S.-R. Chen, L. Chai, J. Zhao, Y. Wang and Y. Wang, Crit. Rev. Food Sci. Nutr., 2019, 59, S17–S29.
- 12 J. C. W. Lim, T. K. Chan, D. S. W. Ng, S. R. Sagineedu, J. Stanslas and W. S. F. Wong, Clin. Exp. Pharmacol. Physiol., 2012, 39, 300–310.
- 13 G. Liu and H. Chu, Oncol. Lett., 2018, 15, 5301–5305.
- 14 P. Udomwan, C. Pientong, P. Tongchai, A. Burassakarn, N. Sunthamala, S. Roytrakul, S. Suebsasana and T. Ekalaksananan, Int. J. Mol. Sci., 2021, 22, 6806.
- 15 Z. Malik, R. Parveen, B. Parveen, S. Zahiruddin, M. Aasif Khan, A. Khan, S. Massey, S. Ahmad and S. A. Husain, J. Ethnopharmacol., 2021, 272, 113936.
- 16 A. A. Farooqi, R. Attar, U. Y. Sabitaliyevich, N. Alaaeddine, D. P. de Sousa, B. Xu and W. C. Cho, Cancers, 2020, 12.
- 17 J. C. W. Lim, T. K. Chan, D. S. Ng, S. R. Sagineedu, J. Stanslas and W. F. Wong, Clin. Exp. Pharmacol. Physiol., 2012, 39, 300–310.
- 18 S. K. Mishra, S. Tripathi, A. Shukla, S. H. Oh and H. M. Kim, Front. Biosci., 2015, 7, 255–266.
- 19 S. G. S. Kandanur, S. Nanduri and N. R. Golakoti, Bioorg. Med. Chem. Lett., 2017, 27, 2854–2862.
- 20 S. G. S. Kandanur, S. Kundu, C. Cadena, H. S. Juan, A. Bajaj, J. D. Guzman, S. Nanduri and N. R. Golakoti, Chem. Pap., 2019, 73, 1669–1675.
- 21 S. G. S. Kandanur, N. R. Golakoti and S. Nanduri, Bioorg. Med. Chem. Lett., 2015, 25, 5781–5786.
- 22 N. Tamang, C. Andrews, S. K. Mavileti, S. Nanduri, N. R. Golakoti and B. Karanam, New J. Chem., 2022, 46, 9745–9754.
- 23 B. Karanam, N. Tamang, Y. Inabathina, S. K. Mavileti, L. Sripada, S. Nanduri and N. R. Golakoti, Cancer Res., 2023, 83, 3818.
- 24 V. Kumar, K. Kaur, G. K. Gupta and A. K. Sharma, Eur. J. Med. Chem., 2013, 69, 735–753.
- 25 M. Faisal, A. Saeed, S. Hussain, P. Dar and F. A. Larik, J. Chem. Sci., 2019, 131, 70.
- 26 A. Ansari, A. Ali, M. Asif and Shamsuzzaman, New J. Chem., 2016, 41, 16–41.
- 27 D. E. V. Pires, T. L. Blundell and D. B. Ascher, J. Med. Chem., 2015, 58, 4066–4072.
- 28 A. Kar, D. Jain, S. Kumar, K. Rajput, S. Pal, K. Rana, R. Kar, S. K. Jha, N. Medatwal, P. S. Yavvari, N. Pandey, D. Mehta, H. Sharma, D. Bhattacharya, M. K. Pradhan, R. D. Sharma, A. Srivastava, U. Agrawal, A. Mukhopadhyay, S. Sengupta, V. S. Patil, A. Bajaj and U. Dasgupta, Sci. Adv., 2023, 9, 1–22. PSC Advances

20 A. Kur, D. Jain, S. Kuran, K. Rujpan, S. Published on 10. D. Pacine G. M. Lampuan and G. S. Rich, *Introduction*

5. K. D. W. Metalwood, P. S. Voront, N. Published on 10. D. Street, 10. D. D. D. D. D. D. D
	- 29 M. Beniwal and N. Jain, J. Heterocycl. Chem., 2019, 56, 2508– 2516.
	- 30 L. You, Xu Qidong, Z. Xiaoli, Li Shiduo, Z. Hui, B. Wei Wu; X. Liu; L. Li; G. Xin; Z. Jiang and W. Xiaoke, China Pharmaceutical University, CN115960047A, 2023.
- 31 D. L. Pavia, G. M. Lampman and G. S. Kriz, Introduction to Spectroscopy-A Guide for Students of Organic Chemistry, Thomson Learning, Inc., 2001.
- 32 D. L. B. R. M. Silverstein, F. X. Webster and D. J. Kiemle, Spectrometric Identification of Organic Compounds, 8th edn, Wiley, 2014.
- 33 E. Miller, ed. S. P. Langdon, Humana Press, Totowa, NJ, 2004, pp. 191–202.
- 34 G.-Y. Liou and P. Storz, Free Radical Res., 2010, 44, 479–496.
- 35 A. Ganesan, Curr. Opin. Chem. Biol., 2008, 12, 306–317.