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The distribution and enrichment characteristics of rare earth elements between coals and coal ashes from four coal-fired power plants

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Coal and coal ash (slag and fly ash) have emerged as important potential sources for rare earth elements (REY). The distribution and enrichment of REY during coal combustion from coal-fired power plants are of great significance for the recovery of REY. In this study, the concentration of REY in coal and coal ash from four coal-fired power plants in China was determined using microwave digestion-inductively coupled plasma mass spectrometry. The distribution and enrichment characteristics of REY in coal and coal ash were studied. The coal ash contains $310\text{--}683\text{ }\mu\text{g g}^{-1}$ of REY, and the proportion of critical REY was greater than 30%, which is higher than the extraction concentration ($300\text{ }\mu\text{g g}^{-1}$) recommended by the United States Department of Energy, indicating significant REY recovery potential. The concentration of REY in coal was positively correlated with ash yield, which could be explained by the fact that REY are mainly distributed in minerals. Coal and coal ash show similar REY distribution patterns, indicating that REY were not fractionated during the coal combustion. In addition, the enrichment coefficient of REY in fly ash was greater than that in slag, which revealed that REY were mainly enriched in fly ash. Therefore, the REY in fly ash has more recycling potential.

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1. Introduction

Rare earth elements (REY) are a group of metals consisting of 15 lanthanides along with yttrium.¹ REY have been included in the list of critical metals by the European Union, and the United States due to their important strategic significance.² From 2002 to 2012, China has been the primary supplier of over 90% of the world's rare earth supply.³ However, due to excessive exploitation and increased prospecting efforts in other countries, the reserves of rare earth resources in China have declined rapidly. According to the latest global rare earth reserves released by the US Geological Service in 2022, China's rare earth resources have dropped from 70% to 33.8%.⁴ Due to the increasing demand for traditional rare earth resources, it is expected that China would need to import rare earth resources from abroad to meet domestic production needs by 2040.⁵ Therefore, it is of great theoretical and practical significance to find new rare earth

metal resources for the stable supply of national mineral resources.

Research has indicated that coal and coal ash contain enriched levels of REY and are considered potential resources for their extraction.^{6,7} Based on the geochemical properties of REY in coal, REY can be divided into light (LREY-La, Ce, Pr, Nd, and Sm), medium (MREY-Eu, Gd, Tb, Dy, and Y), and heavy (HREY-Ho, Er, Tm, Yb, and Lu).⁶ In the meantime, REY can be further classified into critical (critical-Nd, Eu, Tb, Dy, Y and, Er), uncritical (uncritical-La, Pr, Sm, and Gd) and excessive (excessive-Ce, Ho, Tm, Yb, and Lu) according to the supply and demand relationship.⁸ Coal-based materials with the proportion of critical REY greater than 30% and the prospect coefficient (C_{out1} , ratio of critical to excessive) greater than 0.7 are deemed to have potential for REY recovery.⁸ In addition, the U.S. Department of Energy recommends that coal-based materials with a concentration higher than $300\text{ }\mu\text{g g}^{-1}$ can be used as raw materials for REY recovery.⁹ Coal mines that have exhibited abnormal enrichment of REY at domestic and abroad have the potential for REY recovery, such as the Russian Far East coalfield ($300\text{--}1000\text{ }\mu\text{g g}^{-1}$);¹⁰ coal seams ($500\text{--}4000\text{ }\mu\text{g g}^{-1}$) in eastern Kentucky, the United States;^{11,12} and the Daqingshan ($721\text{ }\mu\text{g g}^{-1}$), Heidaigou ($1031\text{ }\mu\text{g g}^{-1}$), Haerwusu ($1347\text{ }\mu\text{g g}^{-1}$), Zhanghe (K8, $1401\text{ }\mu\text{g g}^{-1}$ and K9, $804\text{ }\mu\text{g g}^{-1}$), Songzao (K12, $1968\text{ }\mu\text{g g}^{-1}$), Yishan, Huayingshan, Guiding, Kaili ($874\text{ }\mu\text{g g}^{-1}$), Huaqiu (K9, $239\text{ }\mu\text{g g}^{-1}$ and K16, $630\text{ }\mu\text{g g}^{-1}$) in China.¹³⁻²⁰ Moreover, REY in coal is not easy to volatilize during the coal combustion,^{21,22} so REY are further enriched in coal ash, and

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even reaches the level of industrial extraction.^{10,11} Therefore, coal and coal ash are considered to be new types of mineral resources rich in REY,^{6,7,23} and have attracted the attention of various countries. Since 2014, the United States has made significant investments in various departments to quickly initiate the project of recovering REY from coal and coal ash.^{24–26} Meantime, coal-producing countries such as Europe,^{27–29} Russia,^{30,31} Canada,³² India,^{33–35} South Africa,³⁶ and China^{37–41} have also done extensive studies on the distribution and occurrence of REY in coal. Our team has also studied the occurrence of REY in coal and coal ash, revealing that REY in coal mainly exist in minerals, while most of the very fine minerals containing REY are wrapped in the glass phase of coal ash.⁴² In addition, domestic and foreign scholars have done many researches on the distribution and enrichment of REY during the coal combustion process in power plant.^{39–49} However, the research results are inconsistent, including three main enrichment modes: the first is mainly enriched in slag;⁴³ the second is mainly enriched in fly ash;^{40,44} the third is enriched in both slag and fly ash.^{39,46}

Based on the above research, the concentration of REY in coal and coal ash is determined by microwave digestion-inductively coupled plasma mass spectrometry for four coal-fired power plants in China. This study aims to deepen the understanding of the distribution and enrichment characteristics of REY in coal and coal ash, as well as to analyze their recovery potential. This analysis provides theoretical guidance for the recovery and utilization of REY from coal-based materials.

2. Materials and methods

2.1 Materials

The coal and coal ash (slag and fly ash) samples used in this study were obtained from four coal-fired power plants in China, including P1-600 MW, P2-600 MW, P3-1000 MW, and P4-1000 MW pulverized coal power plants. All power plants use Shenhua coal. The coal was collected from the feed coal of the power plant, while the slag was collected from the boiler slag outlet, and the fly ash was collected from the first row hoppers of the electrostatic precipitator.

Nitric acid (HNO_3 , 65–68%, GR grade) and hydrofluoric acid (HF, 40%, GR grade) were purchased from Beijing Chemical Inc. Boric acid (H_3BO_3 , ≥99.8%, GR grade) was purchased from Sinopharm Chemical Reagent Co., Ltd. The experiment utilized deionized water with a resistance of 18.2 M Ω cm.

2.2 Elements analysis

The analysis of Rare Earth Elements (REY) was performed using inductively coupled plasma-mass spectrometry (ICP-MS, NEXION 2000C ICP-MS, PerkinElmer, USA) in accordance with HJ766-201. Blank and certified reference materials, *i.e.*, NIST 1632d (coal), and GSR-3 (rock), were also included in each experimental batch for quality control and assurance purposes. The recovery of REY in the reference materials ranged between 90% and 110%.

2.3 Experimental

To analyze the REY in coal and coal ash, the coal and coal ash were first dissolved by microwave digestion, followed by the concentration of the REY in the solution using ICP-MS. The specific steps of microwave digestion were follows: in the first step, 0.1 g of sample (coal or coal ash) was weighed with a precision balance, and then 5 mL of nitric acid and 2 mL of hydrofluoric acid were measured with a measuring cylinder. Finally, the sample, nitric acid and hydrofluoric acid were added to the PTFE digestion tank in turn and allowed to stand for two hours. During the standing process, the tank mouth was sealed with a fresh-keeping film to prevent debris from falling into the contaminated sample. In the second step, the digestion tank was placed in the microwave digestion instrument and the heating program was set as follows: firstly, the temperature was raised to 150 °C at a rate of 50 °C min⁻¹ and maintained at this temperature for 5 minutes. Then the temperature was raised to 180 °C at a rate of 15 °C min⁻¹ and maintained for 10 minutes. Finally, the temperature was raised to 210 °C at a rate of 6 °C min⁻¹ and maintained for 15 minutes. In the third step, to prevent the formation of fluoride precipitation in the digestion process and the corrosion of analytical instruments by excessive hydrofluoric acid in the determination process, 20 mL of saturated boric acid solution was added to the digestion tank. Then, according to the second step of the heating program, the digestion was continued to decompose the fluoride precipitate and complex the excess hydrofluoric acid.⁵⁰ After microwave digestion, the solution in the digestion tank was poured into a centrifuge bottle, and the digestion tank was rinsed with 3% dilute nitric acid for three times. The rinse solution was also poured into a centrifuge bottle and diluted to 50 mL for determination. Each sample was digested three times simultaneously.

3. Results and discussion

3.1 Physicochemical characteristics of coal

The results of industrial analysis and elemental analysis of the coal are shown in Table 1. According to MT/T 849-2000, the coal of P4 power plant is identified as mid-high volatile coal (>28.00–37.00%), whereas the coal of the other three power plants is identified as high volatile coal (>37.00–50.00%). Based on GB/15224.1-2018, the coal in three power plants (P1, P2, and P3)

Table 1 Proximate and ultimate analysis results of coal samples (wt%)^a

Sample	Proximate analysis				Ultimate analysis				
	M _{ad}	V _{daf}	A _d	FC _d	C _{daf}	H _{daf}	N _{daf}	S _d	O _{daf}
P1-coal	6.36	37.79	15.65	52.48	72.34	5.62	1.24	0.40	26.90
P2-coal	5.83	37.48	18.06	51.23	89.54	5.59	1.53	0.48	11.91
P3-coal	4.98	37.76	15.28	52.74	84.65	5.15	0.97	0.71	14.16
P4-coal	4.05	35.30	20.60	51.37	84.97	16.46	2.58	1.12	11.72

^a M moisture, V volatile matter, A ash yield, FC fixed carbon, C carbon, H hydrogen, N nitrogen, S sulfur, O oxygen, ad air-dry basis, d dry basis, daf dry and ash-free basis.



is low ash coal (>10.00–20.00%), but the coal of P4 power plant belongs to medium ash coal (>20.00–30.00%). According to MT/T 561-2008, the coal of the four power plants is low fixed carbon coal ($\leq 55.00\%$). According to GB/15224.2-2010, the coal used in P1 and P2 power plants is ultra-low sulfur coal ($\leq 0.50\%$), the coal of P3 power plant is low sulfur coal (0.51–1.00%), and the coal of P4 power plant is middle sulfur coal (1.01–2.00%).

3.2 Geochemical characteristics of REY in coal and coal ash

The concentration and characteristic parameters of REY in coal and coal ash (slag and fly ash) are shown in Table 2. The total REY concentrations in P2 and P4 power plant coals were higher than the world average value ($68.5 \mu\text{g g}^{-1}$).⁵¹ In contrast, the concentrations of REY in coal from P1 and P3 power plants were below the world average. Additionally, the concentration of REY in coal ash of each power plant was higher than that in the corresponding feed coal. The concentration of REY in slag was lower than the world average value ($403.5 \mu\text{g g}^{-1}$),⁵¹ while the concentration of REY in fly ash exceeded the world average value. Moreover, the concentration of critical REY (critical) and the prospect coefficient⁸ (C_{out1} , as shown in eqn (1)) in coal and coal ash, except for the coal and slag of P2 power plant and the coal of P4 power plant, the concentration of critical REY and the prospect coefficient in the coal and coal ash of P1 and P3 power plants, the fly ash of P2 power plant, and the coal ash of P4 power plant were all above 30% and 0.7.

$$C_{\text{out1}} = \frac{\text{Critical}}{\text{Excessive}} \quad (1)$$

where C_{out1} is the prospect coefficient, Critical represents the total concentration of critical REY (Nd, Eu, Tb, Dy, Y, and Er) and Excessive represents the total concentration of excessive REY (Ce, Ho, Tm, Yb, and Lu). Based on the market supply and demand relationship, Seredin and Dai⁶ proposed that coal ash with REY oxides concentration greater than $1000 \mu\text{g g}^{-1}$ and prospect coefficient (C_{out1}) greater than 0.7 have the potential for REY recovery. In addition, the U.S. Department of Energy (DOE) has recommended that coal ash with a total REY concentration of more than $300 \mu\text{g g}^{-1}$ has REY recovery potential.⁹ Therefore, according to the evaluation method proposed by Dai *et al.*,¹⁷ the coal ash of four power plants cannot be used as the raw material for the recovery of REY, but according to the recommended extraction value of DOE,⁹ the coal ash of four coal-fired power plants has the potential for REY recovery.

3.3 Distribution and enrichment characteristics of REY in coal and coal ash

3.3.1 Modes of occurrence of REY in coal. According to the ash yield and the concentration of REY in coal from four different power plants, the correlation between ash and REY was analyzed, as shown in Fig. 1 and 2. The ash yield in the coal of P3 and P1 power plants was similar, and the concentration of

Table 2 Concentrations and characteristics of REY in coal and coal ash ($\mu\text{g g}^{-1}$)

Element	P1			P2			P3			P4		
	Coal	Slag	Fly ash	Coal	Slag	Fly ash	Coal	Slag	Fly ash	Coal	Slag	Fly ash
La	9.3	76.1	122.9	24.4	79.2	123.7	9.7	58.8	85.3	32.6	79.0	136.0
Ce	20.4	148.6	238.6	46.1	156.6	235.5	21.3	116.3	165.6	60.6	151.7	263.5
Pr	2.3	17.1	27.1	4.9	17.4	26.1	2.4	13.7	18.6	6.4	17.3	28.8
Nd	7.9	60.6	98.9	15.3	58.3	90.8	8.0	46.5	66.3	20.7	58.8	99.9
Sm	1.4	10.9	17.2	2.9	10.6	15.9	1.5	8.8	11.9	3.8	10.5	17.3
Eu	0.3	2.0	3.2	0.5	2.1	3.0	0.3	1.7	2.4	0.7	2.0	3.2
Gd	1.4	10.4	16.0	2.7	10.0	14.2	1.4	8.3	11.2	3.6	10.0	16.0
Tb	0.2	1.6	2.5	0.4	1.5	2.3	0.2	1.3	1.7	0.6	1.6	2.5
Dy	1.3	9.3	14.4	2.3	8.6	13.2	1.2	7.0	9.7	3.2	8.6	14.2
Y	7.7	46.8	86.8	11.9	43.7	77.5	6.4	37.7	62.6	16.8	43.7	80.5
Ho	0.3	1.8	2.9	0.4	1.7	2.6	0.2	1.4	2.0	0.6	1.7	2.8
Er	0.8	5.2	8.5	1.3	4.7	7.7	0.7	4.1	5.8	1.8	4.9	8.2
Tm	0.1	0.7	1.2	0.2	0.6	1.1	0.1	0.6	0.8	0.2	0.7	1.1
Yb	0.7	4.4	7.4	1.1	4.0	6.9	0.6	3.5	5.2	1.5	4.1	7.3
Lu	0.1	0.7	1.1	0.2	0.6	1.0	0.1	0.5	0.8	0.2	0.6	1.1
LREY	41.3	313.3	504.6	93.7	322.2	492.0	43.0	244.1	347.8	124.1	317.3	545.6
MREY	11.0	70.1	122.8	17.9	65.9	110.1	9.5	56.1	87.5	24.8	66.0	116.4
HREY	2.0	12.8	21.1	3.2	11.7	19.3	1.7	10.0	14.6	4.4	11.9	20.7
REY	54.3	396.2	648.5	114.7	399.7	621.4	54.2	310.2	449.9	153.3	395.2	682.6
LREY (%)	76.2	79.1	77.8	81.6	80.6	79.2	79.3	78.7	77.3	80.9	80.3	79.9
MREY (%)	20.2	17.7	18.9	15.6	16.5	17.7	17.5	18.1	19.5	16.2	16.7	17.0
HREE (%)	3.6	3.2	3.2	2.8	2.9	3.1	3.2	3.2	3.2	2.9	3.0	3.0
Critical	18.3	125.5	214.2	31.7	119.0	194.5	16.8	98.3	148.5	43.7	119.5	208.6
Uncritical	14.4	114.5	183.1	35.0	117.2	179.9	15.1	89.5	127.0	46.4	116.9	198.1
Excessive	21.6	156.2	251.2	48.1	163.6	247.1	22.3	122.3	174.4	63.3	158.8	276.0
Critical (%)	33.6	31.7	33.0	27.6	29.8	31.3	31.0	31.7	33.0	28.5	30.2	30.6
C_{out1}	0.8	0.8	0.9	0.7	0.7	0.8	0.8	0.8	0.9	0.7	0.8	0.8



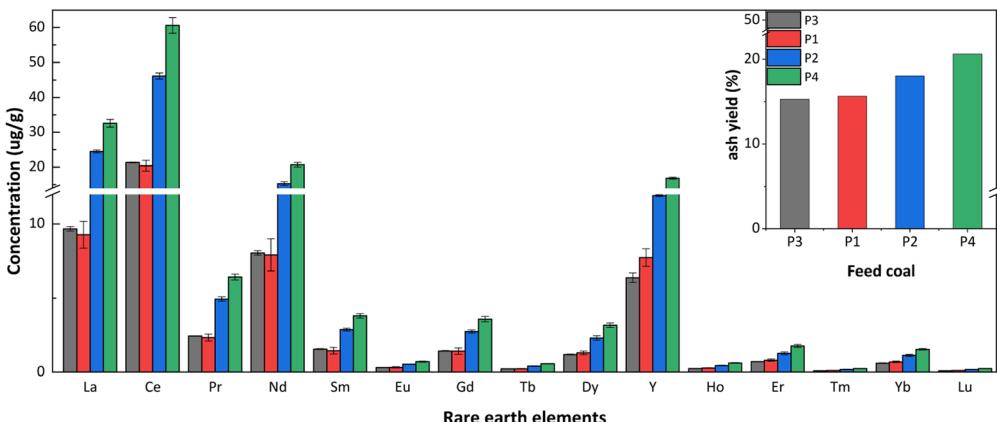


Fig. 1 Effect of ash yield in coal on the distribution of REY.

REY in corresponding coal was also approximately equal. However, from P2 to P4 power plants, as the ash yield in the coal increases significantly, and the concentration of each REY also increases substantially (Fig. 1). In addition, with the increase of the volatile matter content in the coal, the change in the concentration of each REY was not regular (Fig. 2). This indicates a positive correlation between the concentration of REY in coal and ash yield. This was consistent with the existing research results, indicating that REY were mainly distributed in REY minerals (monazite, phosphates, sulfates, carbonates, and fluorocarbonates, *etc.*), accessory minerals (zircon, apatite, *etc.*), and clay minerals in coal.^{6,52,53}

3.3.2 Distribution characteristic of REY in coal and coal ash. The concentration of each REY in coal and coal ash from four coal-fired power plants was normalized to the Upper Continental Crust (UCC),⁵⁴ respectively. The normalized REY pattern is shown in Fig. 3. The coal and coal ash showed a relatively flat REY distribution pattern, and the distribution patterns of REY in the coal, slag, and fly ash of each power plant were consistent, indicating that REY have no obvious fractionation under different coal types and combustion conditions.^{37,39} In addition, due to the volatilization and combustion of organic

matter in coal, the concentrations of REY in all ashes were higher than that in the corresponding coals, indicating that each REY was enriched in different degrees during the combustion process of different coals. Moreover, the minerals containing REY in coal were formed very fine particles during coal combustion, which may be mainly distributed in fly ash.⁵⁵⁻⁵⁹ Meantime, REY combined with organic matter may partially volatilize and condense into fly ash during combustion.⁵⁷ Studies have shown that REY exist in amorphous and graphite carbons surrounding Al-Si glass and magnetite in fly ash.^{46,60} Therefore, fly ash may be more enriched in REY than slag.

3.3.3 Enrichment characteristic of REY in coal and coal ash. The enrichment coefficient¹⁸ was used to illustrate the enrichment behavior of REY in coal and coal ash, that is, the concentration ratio of elements in one sample to corresponding elements in another sample was analyzed, as shown in Table 3. In the enrichment coefficient (CC1) of coal relative to world coal, the total REY enrichment coefficients in P1 and P3 power plants were both equal or less than 1. The enrichment coefficients of total REY in P2 and P4 power plants increased in turn and were all greater than 1. Compared with P1 and P3 power

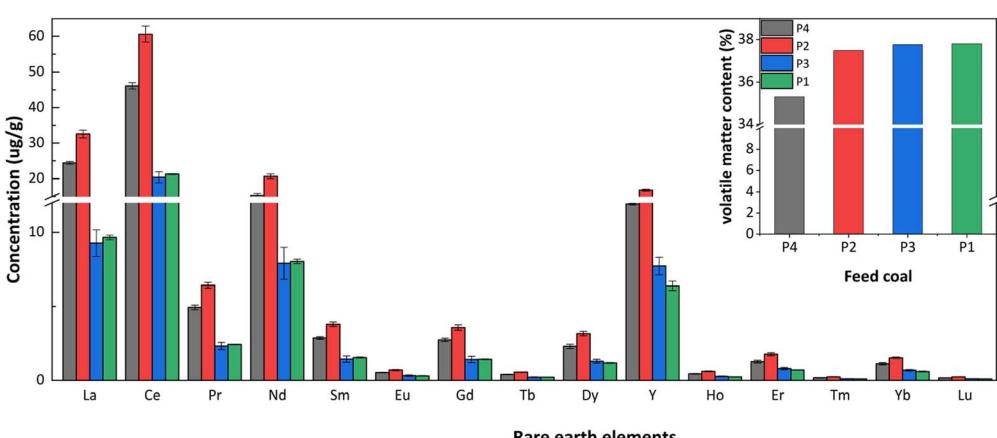


Fig. 2 Effect of volatile matter content in coal on the distribution of REY.

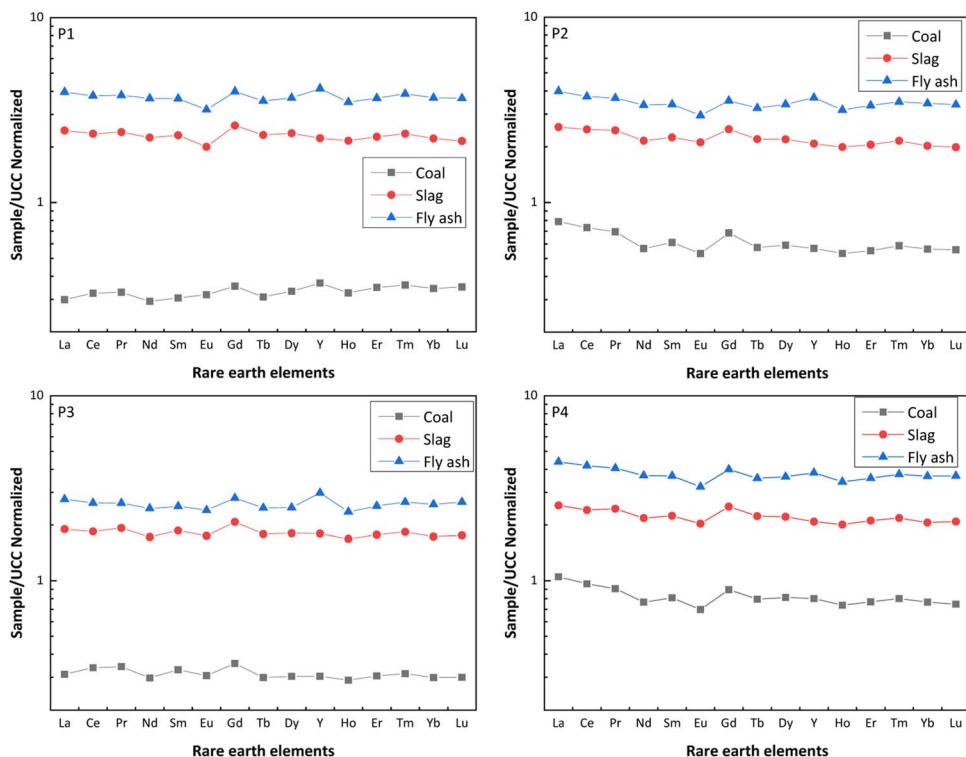


Fig. 3 UCC-normalized REY patterns of coal and coal ash sample.

plants, the REY in the coal of P4 power plant are obviously enriched. In the enrichment coefficients of coal ash (slag and fly ash) relative to world ash, the enrichment coefficient of slag relative to world ash (CC2) in each power plant was less than 1, while the enrichment coefficient of fly ash relative to world ash (CC3) was greater than 1, indicating that REY are more likely to

be enriched into fly ash during coal combustion. In addition, in the enrichment coefficients (CC4 and CC5) of coal ash (slag and fly ash) relative to the feed coal, the enrichment coefficient of coal ash relative to the coal in each power plant was all greater than 1, and the enrichment coefficients (CC6) of fly ash relative to slag were all greater than 1. This further indicates that REY

Table 3 Enrichment coefficient of REY in coal and coal ash^a

Element	P1					P2					P3					P4									
	CC1	CC2	CC3	CC4	CC5	CC6	CC1	CC2	CC3	CC4	CC5	CC6	CC1	CC2	CC3	CC4	CC5	CC6	CC1	CC2	CC3	CC4	CC5	CC6	
La	0.84	1.10	1.78	8.21	13.26	1.61	2.22	1.15	1.79	3.24	5.06	1.56	0.88	0.85	1.24	6.09	8.84	1.45	2.96	1.14	1.97	2.43	4.18	1.72	
Ce	0.89	1.14	1.84	7.29	11.70	1.61	2.01	1.20	1.81	3.39	5.10	1.50	0.93	0.89	1.27	5.46	7.78	1.42	2.64	1.17	2.03	2.50	4.35	1.74	
Pr	0.66	0.85	1.35	7.34	11.63	1.59	1.41	0.87	1.30	3.53	5.28	1.50	0.70	0.68	0.93	5.62	7.66	1.36	1.84	0.87	1.44	2.70	4.48	1.66	
Nd	0.66	0.90	1.48	7.67	12.52	1.63	1.27	0.87	1.36	3.82	5.94	1.56	0.67	0.69	0.99	5.79	8.25	1.43	1.72	0.88	1.49	2.84	4.83	1.70	
Sm	0.72	0.84	1.32	7.57	11.96	1.58	1.43	0.81	1.23	3.69	5.55	1.51	0.77	0.67	0.91	5.66	7.65	1.35	1.90	0.81	1.33	2.77	4.55	1.64	
Eu	0.68	0.80	1.27	6.31	10.03	1.59	1.13	0.85	1.18	3.97	5.55	1.40	0.65	0.70	0.96	5.70	7.85	1.38	1.49	0.81	1.29	2.91	4.61	1.58	
Gd	0.52	0.65	1.00	7.38	11.29	1.53	1.02	0.62	0.89	3.63	5.18	1.43	0.53	0.52	0.70	5.82	7.84	1.35	1.32	0.63	1.00	2.81	4.47	1.59	
Tb	0.68	0.77	1.18	7.51	11.48	1.53	1.25	0.73	1.08	3.84	5.65	1.47	0.65	0.60	0.82	5.98	8.28	1.38	1.74	0.74	1.19	2.80	4.50	1.60	
Dy	0.62	0.66	1.03	7.15	11.12	1.56	1.10	0.61	0.94	3.72	5.74	1.54	0.56	0.50	0.69	5.95	8.17	1.37	1.51	0.62	1.01	2.72	4.48	1.65	
Y	0.92	0.92	1.70	6.06	11.24	1.85	1.42	0.86	1.52	3.67	6.51	1.77	0.76	0.74	1.23	5.91	9.81	1.66	2.00	0.86	1.58	2.60	4.78	1.84	
Ho	0.50	0.45	0.73	6.65	10.76	1.62	0.82	0.41	0.66	3.74	5.94	1.59	0.44	0.35	0.49	5.82	8.13	1.40	1.13	0.42	0.71	2.72	4.64	1.71	
Er	0.86	0.95	1.54	6.52	10.56	1.62	1.36	0.86	1.40	3.72	6.07	1.63	0.75	0.74	1.06	5.80	8.29	1.43	1.90	0.88	1.50	2.75	4.66	1.70	
Tm	0.35	0.35	0.58	6.58	10.82	1.65	0.57	0.32	0.53	3.68	5.97	1.62	0.30	0.28	0.40	5.84	8.46	1.45	0.78	0.33	0.56	2.72	4.69	1.73	
Yb	0.69	0.72	1.19	6.47	10.76	1.66	1.12	0.65	1.11	3.60	6.12	1.70	0.60	0.56	0.83	5.79	8.62	1.49	1.53	0.66	1.18	2.69	4.79	1.78	
Lu	0.54	0.56	0.95	6.15	10.50	1.71	0.86	0.51	0.87	3.56	6.06	1.70	0.46	0.45	0.69	5.87	8.87	1.51	1.16	0.54	0.95	2.79	4.93	1.77	
REY	0.79	0.98	1.61	7.30	11.95	1.64	1.68	0.99	1.54	3.48	5.42	1.55	0.79	0.77	1.11	5.72	8.30	1.45	2.24	0.98	1.69	2.58	4.45	1.73	

^a CC1 represents the enrichment coefficient of coal to world coal, CC2 represents the enrichment coefficient of slag to world ash, CC3 represents the enrichment coefficient of fly ash to world ash, CC4 represents the enrichment coefficient of slag to coal, CC5 represents the enrichment coefficient of fly ash to coal, CC6 represents the enrichment coefficient of fly ash to slag, the data of REY in world coal and ash came from Ketrus and Yudovich⁵¹.



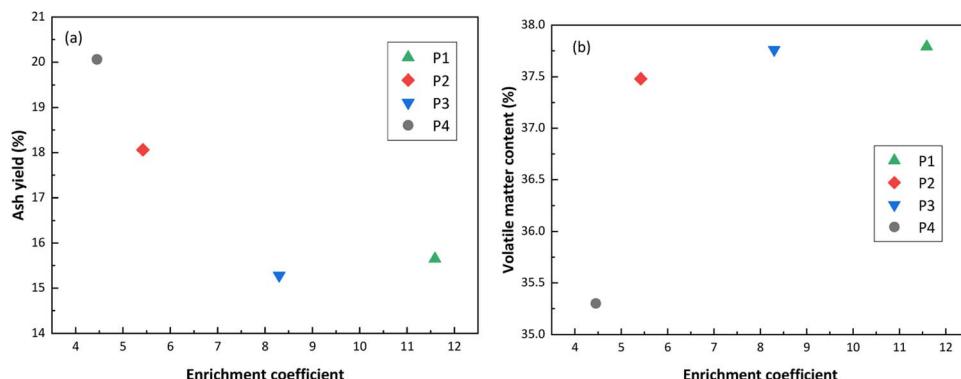


Fig. 4 Effect of the major components in coal on the enrichment coefficient of REY. (a) Represents the effect of the ash yield in coal on the enrichment coefficient (CC5) of REY. (b) Represents the effect of the volatile matter content in coal on the enrichment coefficient (CC5) of REY.

were mainly present in fine minerals and glass particles, and the amorphous and graphite carbon surrounding Al-Si glass and magnetite in fly ash may be beneficial to the enrichment of REY. Moreover, the yield of fly ash was far greater than that of slag. Therefore, REY in fly ash have more potential for recovery. In addition, with the increase of enrichment coefficient (CC5), the ash yield decreased, while the volatile matter content increased (Fig. 4). This indicates that the lower the ash yield and the higher the volatile matter content in coal, the greater the enrichment coefficient of REY in fly ash.

4. Conclusions

In this paper, the distribution and enrichment characteristics of REY in coal and coal ash from four different coal-fired power plants in China were studied. The concentrations of REY in the fly ash of the four coal-fired power plants were higher than the extraction values recommended by the US Department of Energy. Meantime, the proportion of critical REY and the prospect coefficient (C_{out1}) were both greater than 30% and 0.7, respectively, suggesting potential for these resources to be utilized as secondary resources of REY. In addition, with the increase of ash yield in coal, the concentration of REY increases. However, with the increase of volatile matter content in coal, the concentration of REY did not increase. This indicates that REY mainly exist in the minerals of coal. The coal and coal ash of each coal-fired power plant have similar REY distribution patterns, which revealed that the REY in the combustion process of the coal in each power plant do not have obvious fractionation. When the ash yield in coal was lower and the volatile matter content was higher, the enrichment coefficient of REY in coal ash was larger. The content of REY in fly ash was greater than that in slag, and it has the potential of REY recovery. These results provide a theoretical basis for the recovery of REY from coal and coal ash.

Author contributions

Guoqiang Wu: conceptualization, formal analysis, resources, funding acquisition, data analysis and writing – original draft, supervision. Zhanlong Ma: methodology, investigation,

criticism, writing – review & editing. Guosheng Li: writing – review & editing. Chunmiao Bo: criticism.

Conflicts of interest

There are no conflicts to declare.

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