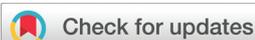


## RESEARCH ARTICLE

View Article Online  
View Journal | View IssueCite this: *Org. Chem. Front.*, 2024,  
11, 7235Received 15th August 2024,  
Accepted 7th October 2024

DOI: 10.1039/d4qo01505c

rsc.li/frontiers-organic

## C–H acylation as an enabling tool to tag phenolic drugs†

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The site-selective functionalization of value-added compounds while implementing atom-economical C–H coupling partners represents an unmet challenge of utmost importance within organic synthesis. Herein, we report a Pd-catalyzed directed C–H acylation of a collection of relevant phenol-containing compounds with ethanol and other alcohols and aldehydes. This tagging technique is distinguished by its water compatibility and predictable regioselectivity and features the use of ethanol as renewable feed-stock for the modification of intricate phenols, including estrogens and other top-selling pharmaceuticals. Mechanistic studies support the intermediacy of a challenging 6-membered dimeric palladacycle that undergoes the addition of nucleophilic acyl radical species.

## Introduction

Phenols are privileged scaffolds of widespread presence in a vast number of natural products, agrochemicals, pharmaceuticals, dyes, fragrance compounds and other value-added products (Fig. 1).<sup>1</sup> As a result, the development of efficient synthetic tools aimed at increasing the molecular complexity of phenol-containing compounds poses a challenging task of capital importance within organic chemistry.

Friedel–Crafts acylation<sup>2</sup> is one of the most useful methods to functionalize phenols through a classical electrophilic aromatic substitution mechanism and is included in all the undergraduate-level organic chemistry textbooks (Scheme 1a). Likewise, Fries rearrangement offers an alternative route to *ortho*-hydroxyaryl ketones from the corresponding *O*-acylated phenols.<sup>3</sup> Despite their extensive use even in industrial environments,<sup>4</sup> these well-known transformations suffer from severe drawbacks such as the lack of regioselectivity, use of highly reactive acyl chlorides and toxic chemical waste derived from the use of stoichiometric amounts of halide-containing Lewis acids. Accordingly, the introduction of novel yet sustainable processes for the practical acetylation of phenols featuring other acyl surrogates represents a prime goal of paramount synthetic significance. Indeed, acetyl groups have been demon-

strated to act as valuable linkers for further chemical ligation<sup>5</sup> and hence they stand out as versatile handles within organic synthesis.

C–H functionalization has changed the landscape of retrosynthetic analysis, thereby enabling the direct conversion of otherwise unreactive hydrocarbon backbones into functionalized compounds in a straightforward fashion.<sup>6</sup> In this respect, the last few years have witnessed the surge of a plethora of metal-catalyzed C(sp<sup>2</sup>)-H functionalizations of phenols,<sup>7</sup> which outcompete classical methods in terms of sustainability and site-selectivity. Most of the methods within the existing portfolio leverage the electron-rich nature of the phenol framework to further add electrophilic counterparts at the aromatic ring in a predictable manner. Conversely, the appendance of nucleophilic coupling partners requires the use of distinct tactics such as chelation assistance,<sup>8</sup> which facilitates their addition toward electrophilic organometallic intermediates formed upon a directed C–H activation event.

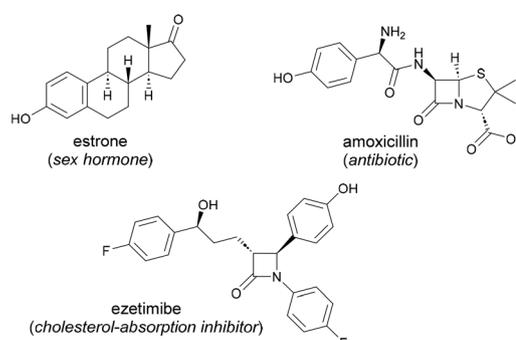
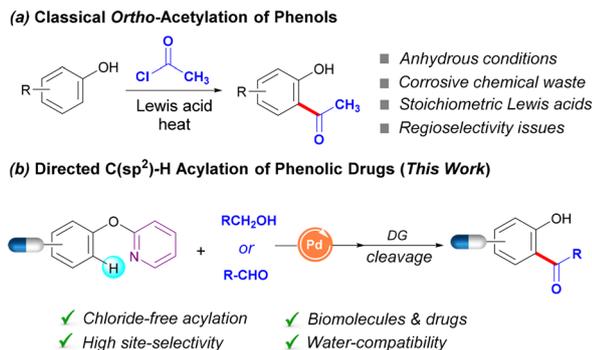


Fig. 1 Phenol-containing bioactive molecules.

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†Electronic supplementary information (ESI) available: Experimental procedures, syntheses and characterization of all new compounds, and tables with details of several optimization studies. CCDC 2386321. For ESI and crystallographic data in CIF or other electronic format see DOI: <https://doi.org/10.1039/d4qo01505c>





Scheme 1 Acylation of phenols.

In this context, while acyl radicals exhibit an amphiphilic character,<sup>9</sup> they are predominantly nucleophilic in nature and hence they can undergo different fundamental reactions than those of the corresponding acylium species involved in classical Friedel–Crafts acylation. Therefore, the directed dehydrogenative radical C(sp<sup>2</sup>)-H acylation of strategically modified phenols represents an attractive alternative for accessing biologically relevant hydroxyaryl ketones. As part of our interest in metal-catalyzed radical C–H functionalization,<sup>10</sup> we have recently unlocked the synthetic potential of simple and abundant ethanol as a C2-chemical feedstock<sup>11</sup> to perform the late-stage acetylation of a library of tyrosine-containing peptides.<sup>12</sup> Despite its importance within the realm of bioconjugation, the method was not explored for other value-added phenol-containing biomolecules or pharmaceuticals and the proposed mechanism was merely speculative based on indirect evidence.

Owing to their presence within female sex hormones, estrogens represent a key family of phenol-containing biomolecules including estrone and estradiol. Unfortunately, the chemical diversification of estrogens has been overlooked and general methods toward their late-stage modification are still elusive. In this communication, we explore the full synthetic potential of a Pd-catalyzed C(sp<sup>2</sup>)-H acylation manifold featuring EtOH and other alcohols as well as aldehydes to tag a collection of estrogens and other phenolic drugs. Notably, the required directing group (DG) can be easily installed and cleaved, which illustrates the utility of the method to prepare *ortho*-acylated phenols. Importantly, the full characterization of an acetate-bridged cyclopalladated dimer and its catalytic and stoichiometric experiments supported its intermediacy within a Pd(II)–Pd(IV) catalytic cycle.

## Results and discussion

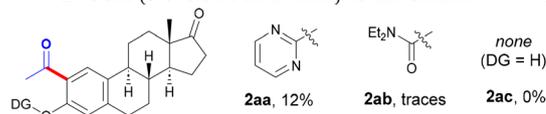
Driven by the successful use of 2-pyridyl ether as an efficient directing group (DG) in the modification of phenol compounds upon metal-catalyzed C–H functionalization,<sup>10c,12,13</sup> we first selected estrone derivative **1a** as the model substrate to test the feasibility of an acetylation reaction featuring the advantageous use of EtOH as a sustainable surrogate of corrosive

acetyl chloride. After careful evaluation of the reaction conditions,<sup>14</sup> we obtained the exclusive formation of mono-acetylated product **2a** in 58% yield in the presence of 10 mol% of Pd(OAc)<sub>2</sub> and an excess of an aqueous solution of cheap *tert*-butyl hydroperoxide (TBHP) at 120 °C (Table 1, entry 1). Control experiments underpinned the crucial role of both the catalyst and the oxidant, as the process was entirely inhibited in their absence (entries 2 and 3). The performance of the reaction at lower temperature or under air resulted in much lower yields (entries 4 and 5).

The optimal amount of EtOH was found to be 25 equiv., and the use of a lower (entries 6 and 7) or a higher amount of EtOH (up to 50 equiv.) resulted in comparatively lower yields. While other related Pd sources could be utilized, Pd(OAc)<sub>2</sub> showed a superior catalytic activity (entries 8–10). Likewise, the nature of the solvent played a crucial role and trifluorotoluene outperformed other solvents such as toluene (entry 11) or 1,2-dichloroethane (entry 12), among others.<sup>14</sup> Regarding the oxidant, while the use of dicumyl peroxide (entry 13) or di-*tert*-butyl peroxide (entry 14) resulted in no reaction, the use of a solution of *tert*-butyl hydroperoxide was key within the reaction outcome. Importantly, the inexpensive aqueous solution of TBHP afforded a higher yield than the use of the parent reagent in decane (entry 16), which represents an added bonus

Table 1 Pd-catalyzed C–H acetylation of **1a**<sup>a</sup>

| Entry | Change from standard conditions                         | <b>2a</b> <sup>b</sup> (%) |
|-------|---|----------------------------|
| 1     | None  | 58 (46) <sup>c</sup>       |
| 2     | Without Pd(OAc) <sub>2</sub>                            | 0                          |
| 3     | Without <sup>t</sup> BuOOH                              | 0                          |
| 4     | Under air   | 31                         |
| 5     | <i>T</i> = 80 °C  | 22                         |
| 6     | With EtOH (15 equiv., 120 μL)                           | 27                         |
| 7     | With EtOH (5 equiv., 40 μL)                             | Traces                     |
| 8     | Pd(OPiv) <sub>2</sub> instead of Pd(OAc) <sub>2</sub>   | 43                         |
| 9     | Pd(TFA) <sub>2</sub> instead of Pd(OAc) <sub>2</sub>    | Traces                     |
| 10    | PdCl <sub>2</sub> instead of Pd(OAc) <sub>2</sub>       | 26                         |
| 11    | PhMe instead of PhCF <sub>3</sub> as the solvent        | 45                         |
| 12    | 1,2-DCE instead of PhCF <sub>3</sub> as the solvent     | 41 <sup>d</sup>            |
| 13    | DCP instead of <sup>t</sup> BuOOH as the oxidant        | 0                          |
| 14    | DTBP instead of <sup>t</sup> BuOOH as the oxidant       | 0                          |
| 15    | <sup>t</sup> BuOOH (5.0 equiv.)                         | 36                         |
| 16    | <sup>t</sup> BuOOH (5.0–6.0 M in decane) as the oxidant | 48                         |



<sup>a</sup> Reaction conditions: **1a** (0.15 mmol), EtOH (3.75 mmol, 0.2 mL), Pd(OAc)<sub>2</sub> (10 mol%), and <sup>t</sup>BuOOH (0.90 mmol, 70 wt% in H<sub>2</sub>O) in PhCF<sub>3</sub> (1 mL) at 120 °C for 16 h under Ar. <sup>b</sup> Yield of the isolated product after column chromatography. <sup>c</sup> Reaction performed with 1.44 mmol (500 mg) of **1a**. <sup>d</sup> NMR yield obtained using the product as the internal standard.

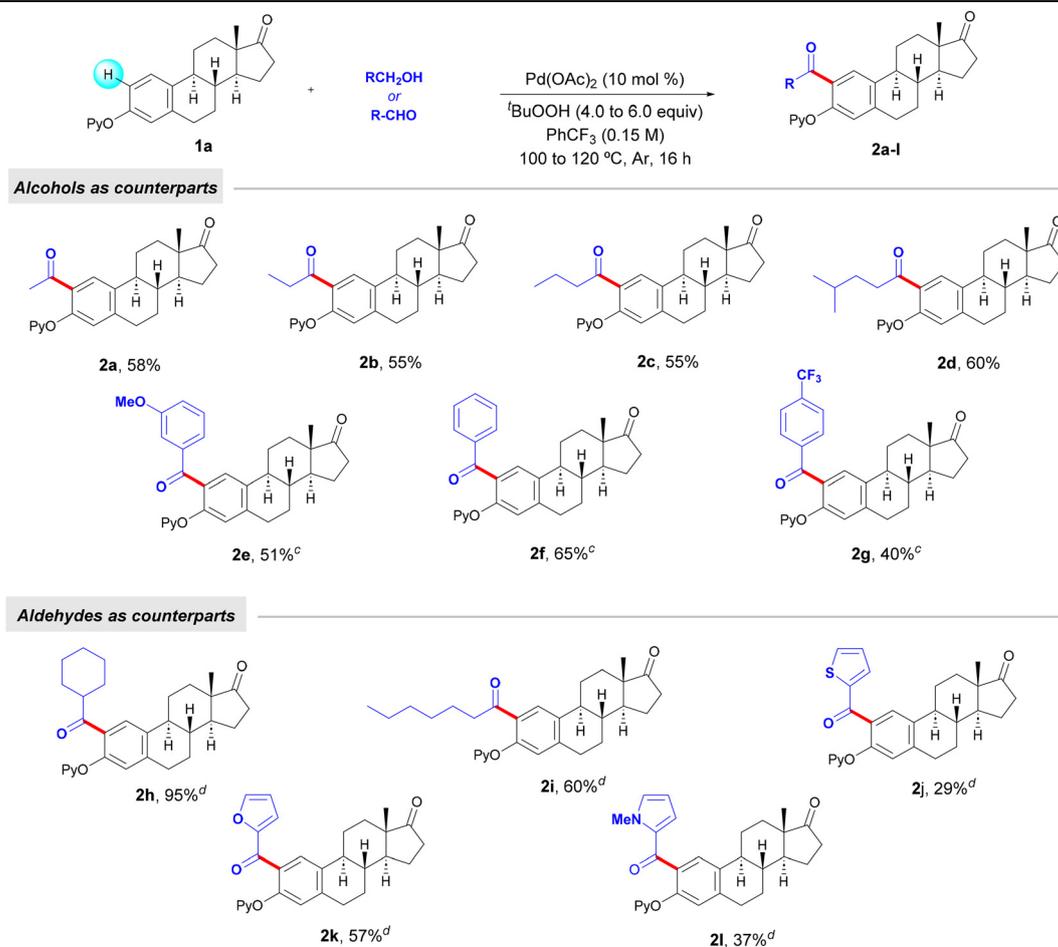


from a sustainable perspective. Accordingly, the drawback derived from using up to 6.0 equiv. of an oxidant (entry 1 vs. entry 15) is balanced by its low price and water-compatible character. In order to evaluate the role of the DG, control experiments were performed with other estrone derivatives. As expected, the absence of a DG resulted in the entire inhibition of the process; however, other related DGs such as pyrimidine (**2aa**) or a carbamate (**2ab**), previously used in the metal-catalyzed functionalization of phenol-based compounds,<sup>15</sup> exhibited an inferior reactivity to that of the pyridyl motif. Notably, the reaction could be scaled up and performed with 1.44 mmol (500 mg) of estrone derivative **1a** to selectively afford acetylated **2a** in a synthetically useful 46% yield. Despite the presence of two chemically different C–H bonds, exclusive functionalization at the less hindered C2 site was always observed, which was verified through NOE experiments after cleavage of the DG (*vide infra*).<sup>14</sup>

Prompted by the successful use of EtOH as an appealing surrogate of conventional acetyl chloride, we next evaluated the use of other alkyl and benzyl alcohols in the acylation of estrone derivative **1a** (Table 2).

Alcohols such as *n*-propanol and *n*-butanol, common solvents in a vast array of organic reactions, as well as 4-methylpentan-1-ol, could be used to perform the mono-acylation of estrone **1a**. Owing to their higher boiling point than that of EtOH, in those cases the amount of the alcohol could be reduced to 5.0 equivalents. Moreover, the reaction using easily oxidizable benzyl alcohols could be carried out with only 3.0 equiv of the corresponding alcohols. In all cases, mono-acylated estrone compounds **2a–g** were obtained in up to 65% yield. Based on the absence of *ortho*-acylation methods for estrogen derivatives, we further explored the use of certain aldehydes as counterparts in order to broaden the structural complexity within the estrone core. Interestingly, besides alkyl

**Table 2** Pd-catalyzed C–H acylation of estrone with alcohols and aldehydes<sup>a,b</sup>



<sup>a</sup> Reaction conditions: **1a** (0.15 mmol), RCH<sub>2</sub>OH (0.75 mmol), Pd(OAc)<sub>2</sub> (10 mol%), and <sup>t</sup>BuOOH (0.90 mmol, 70 wt% in H<sub>2</sub>O) in PhCF<sub>3</sub> (1 mL) at 120 °C for 16 h under Ar. <sup>b</sup> Yield of the isolated product after column chromatography, with the average of at least two independent runs with a variation in yield of no more than 5% between runs. <sup>c</sup> RCH<sub>2</sub>OH (0.45 mmol). <sup>d</sup> Reaction conditions: **1a** (0.15 mmol), aldehyde (0.75 mmol), Pd(OAc)<sub>2</sub> (10 mol%), and <sup>t</sup>BuOOH (0.60 mmol, 70 wt% in H<sub>2</sub>O) in PhCF<sub>3</sub> (1 mL) at 100 °C for 16 h under Ar.



residues (**2h** and **2i**), ketones containing relevant heterocyclic fragments such as thiophene (**2j**), furan (**2k**) and pyrrole (**2l**) could be installed site-selectively in moderate to good yields. When employing aldehydes, the reaction could take place at 100 °C with 4.0 equivalents of TBHP. Of particular importance is the use of cyclohexanecarboxaldehyde, which led to the corresponding decorated estrone derivative **2h** in excellent yield.

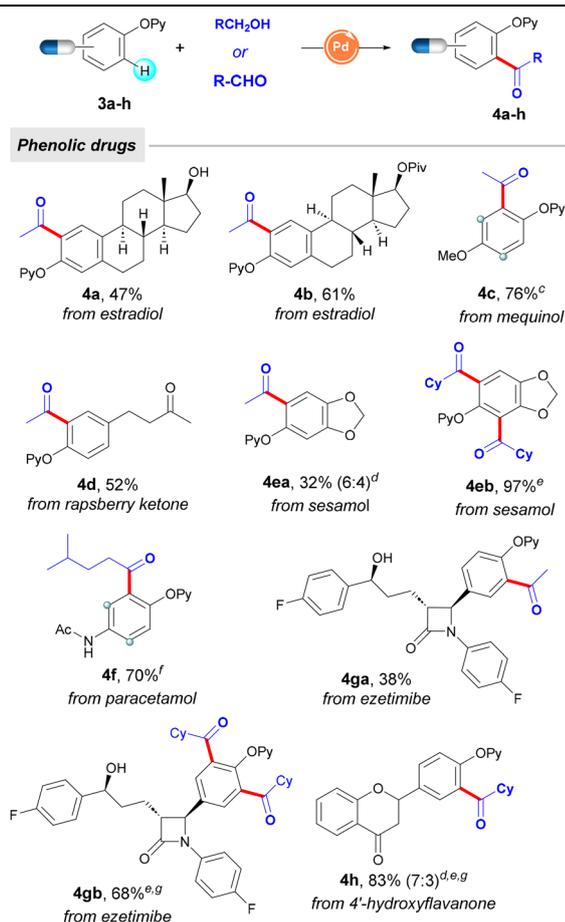
Encouraged by the utility of this acylation platform to tag estrone, we further tackled the modification of a collection of other phenol-based biomolecules and top-selling phenolic drugs, with a special emphasis on the advantageous use of EtOH as chemical feedstock. As depicted in Table 3, structurally related estradiol compounds (**4a** and **4b**) could selectively undergo C–H acylation even in the presence of a free tertiary alcohol (**4a**), which remained intact throughout the oxidative

process. Other phenolic drugs such as mequinol (**4c**, used for skin depigmentation), raspberry ketone (**4d**, dietary supplement), sesamol (**4e**, food ingredient), paracetamol (**4f**, analgesic), or 4'-hydroxyflavanone (**4h**, inhibitor of SREBP) could be smoothly tagged upon the Pd-catalyzed acylation manifold. It is important to highlight that the observed *ortho*-selectivity would not be achieved by classical Friedel–Crafts acetylation on free phenols, which would result in mixtures of regioisomers due to the presence of multiple functional groups within the corresponding arene (mostly within mequinol **4c** and paracetamol **4f** containing electron-donating groups). Remarkably, another drug containing multichiral carbon centers, such as the cholesterol absorption inhibitor ezetimibe, was also shown to be a competent coupling partner, thereby providing modified drugs in moderate to good yields (**4ga** and **4gb**). Notably, while sesamol or ezetimibe afforded low yields in the acetylation event with EtOH (**4ea** and **4ga**, respectively), they successfully underwent parent acylation with CyCHO, selectively forming the corresponding diacylated phenol compounds **4eb** and **4gb** in higher yields. Collectively, the phenol partner scope depicted in Tables 2 and 3 showcases the functional group tolerance in post-synthetic chemical modifications and illustrates the utility of this labelling technique for the selective diversification of a wide variety of intricate phenol compounds.

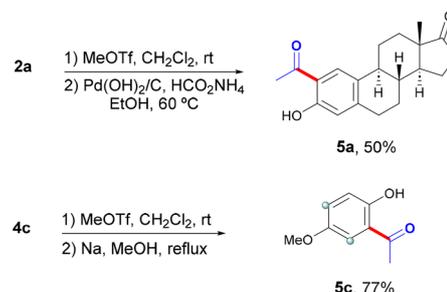
Although the cleavage of the OPy group has been often accomplished in simple arenes upon treatment with methyl triflate followed by an alcoholic solution of sodium,<sup>13</sup> its application in complex biomolecules could pose a major drawback. In this respect, the cleavage of the DG in simple mequinol derivative **4c** could be performed under those classical conditions to afford *ortho*-acetylated phenol **5c** in 77% yield (Scheme 2). However, the cleavage of the DG within more complex estrone derivative **2a** was alternatively achieved through a methylation/hydrogenation sequence<sup>13a</sup> to produce the corresponding *ortho*-acetylated estrone **5a** in 50% yield.

In order to gain some insights into the reaction mechanism, we conducted the following set of experiments. First, we treated estrone derivative **1a** with a stoichiometric amount of Pd(OAc)<sub>2</sub> in trifluorotoluene at 100 °C to obtain complex **I** as a yellow crystalline solid in 76% yield. As shown in Fig. 2, X-ray crystallographic analysis verified the dimeric nature of this palladacycle intermediate and its head-to-tail geometric struc-

Table 3 Acylation of phenolic drugs<sup>a,b</sup>



<sup>a</sup> As for Table 1, entry 1. <sup>b</sup> Yield of the isolated product after column chromatography, with the average of at least two independent runs with a variation in yield of no more than 5% between runs. <sup>c</sup> Reaction time of 24 h. <sup>d</sup> Ratio of mono- and diacylated products. <sup>e</sup> Reaction conditions: **3** (0.15 mmol), cyclohexanecarboxaldehyde (0.75 mmol), Pd(OAc)<sub>2</sub> (10 mol%), and <sup>t</sup>BuOOH (0.60 mmol, 70 wt% in H<sub>2</sub>O) in PhCF<sub>3</sub> (1 mL) at 100 °C for 1 h under Ar. <sup>f</sup> Reaction conditions: **3** (0.15 mmol), 4-methylpentanol (0.75 mmol), Pd(OAc)<sub>2</sub> (10 mol%), and <sup>t</sup>BuOOH (0.90 mmol, 70 wt% in H<sub>2</sub>O) in PhCF<sub>3</sub> (1 mL) at 120 °C for 1 h under Ar. <sup>g</sup> Reaction time = 16 h.



Scheme 2 DG cleavage.



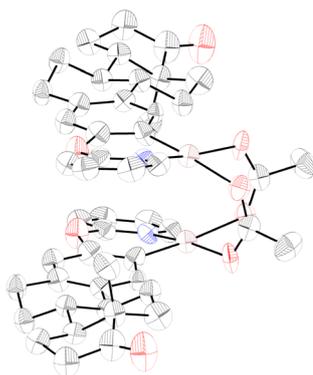
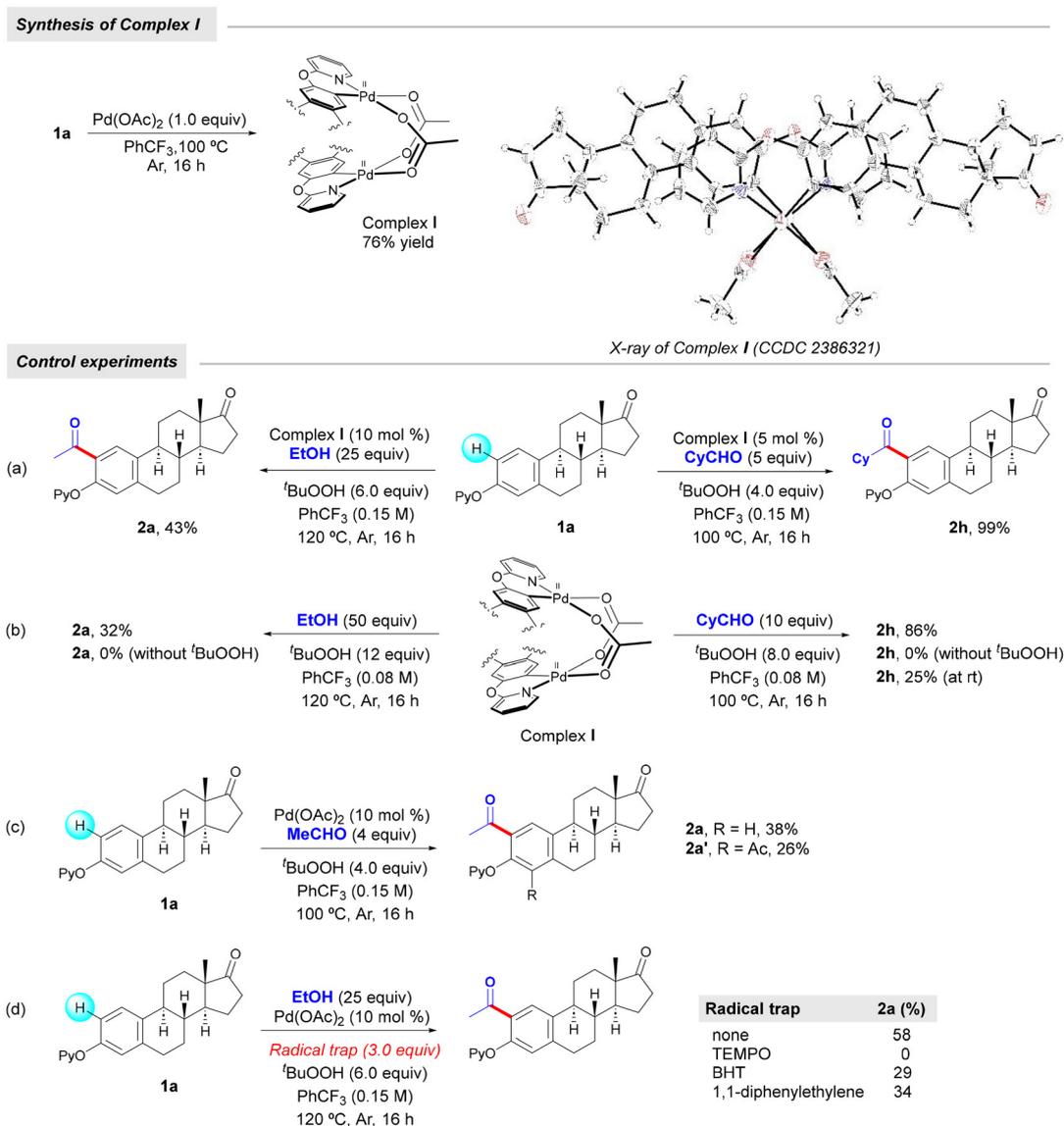


Fig. 2 ORTEP drawing of complex I.

ture<sup>16</sup> and the observed Pd–Pd distance of 2.8491(17) Å was consistent with a bond order of zero. Likewise, its characterization was complemented by <sup>1</sup>H NMR analysis, which evidenced the successful *ortho*-palladation as two singlets at 6.67 and 6.37 ppm appeared within the aromatic region.<sup>14</sup> With this acetate-bridged cyclopalladated dimer in hand, we performed some control experiments to support its intermediacy within the dehydrogenative acylation event (Scheme 3). When performing acetylation with EtOH of estrone derivative **1a** in the presence of a catalytic amount of complex I instead of Pd(OAc)<sub>2</sub>, the corresponding *ortho*-acetylated compound **2a** was obtained in 43% yield. Likewise, its use in a catalytic amount enabled *ortho*-acylation with highly reactive cyclohexanecarboxaldehyde to afford acylated compound **2h** in quantitative yield. Notably, stoichiometric experiments with complex I underpinned its key role in the radical acylation process. While the



Scheme 3 Synthesis of intermediates and control experiments.



acetylation with EtOH successfully occurred to afford **2a** in 32% yield, the acylation with CyCHO provided acylated compound **2h** in a remarkable 86% yield. It is important to note that the process only occurred in the presence of TBHP and in its absence the corresponding cyclopalladated dimer remained unreactive, which supported the crucial role of the oxidant to generate the transient acyl radical species. Furthermore, **2h** could also be obtained at room temperature, albeit in comparatively lower yields. These results are in sharp contrast to those obtained by Wu and co-workers in related mechanistic studies on simple 2-phenoxy pyridines with benzaldehydes.<sup>17</sup> In fact, owing to the low yields obtained when starting from a related cyclopalladated dimer, the intermediacy of an aroyl palladium  $\pi$  complex was proposed instead as the more plausible mechanistic scenario. Conversely, the excellent results obtained when submitting complex **I** to the acylation reaction conditions with alcohols and aliphatic aldehydes may support its key role in our acylation events. Importantly, estrone derivative **1a** underwent the acetylation reaction with acetaldehyde to form acetylated product **2a** in 38% yield together with 26% yield of diacetylated product **2a'**, which was never observed when using EtOH as the coupling partner. This experiment may support an initial oxidation of EtOH to MeCHO and tentatively indicate that the latter oxidation event or its further oxidation to the acyl radical species could be crucial within the process. Finally, in accordance with the studies by Wu,<sup>17</sup> the results obtained when performing the acetylation of **1a** with ethanol in the presence of radical traps strongly supported a radical reaction pathway. Indeed, the process was entirely inhibited when adding TEMPO as a radical scavenger and significantly disfavored in the presence of BHT and 1,1-diphenylethylene to afford **2a** in 29% and 34% yields, respectively.

Based on the mechanistic experiments mentioned above and the existing literature,<sup>9</sup> we proposed a plausible reaction pathway for the directed Pd-catalyzed acylation with alcohols and aldehydes. As depicted in Scheme 4 for the acetylation of estrone derivative **1a**, the initial coordination of Pd(OAc)<sub>2</sub> with

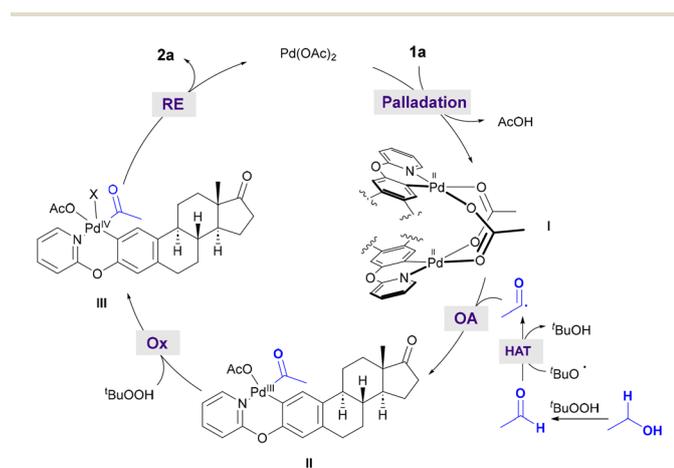
the starting phenol-compound following *ortho*-C–H bond activation would form the corresponding acetate-bridged cyclopalladated species **I**,<sup>16</sup> which has been identified as a dimer by X-ray analysis. It is well known that primary alcohols can easily undergo oxidation to the corresponding aldehydes.<sup>18,19</sup> The latter could undergo an hydrogen-atom transfer (HAT) event with electrophilic *tert*-butoxyl radical species to afford nucleophilic acyl radical species,<sup>9,20</sup> which could undergo addition to electrophilic bimetallic palladium complex **I**. Despite the existence of mechanistic studies on the formation of high valent Pd(IV) complexes *via* either mono- or bi-metallic pathways,<sup>21</sup> for simplicity reasons, we proposed that dimeric intermediate **I** may likely undergo dissociation to a monomeric Pd complex prior to its oxidation toward the formation of high valent species. Accordingly, the initially formed monomeric Pd(III) intermediate **II** could evolve into Pd(IV). Eventually, reductive elimination would furnish *ortho*-acetylated compound **2a**, thereby regenerating the active Pd(II) catalyst. However, an alternative bimetallic pathway cannot be ruled out and may also be a feasible scenario.<sup>22</sup> It is important to note that direct experimental evidence of the formation of high valent Pd species has not been achieved and its intermediacy is proposed based on literature precedents.

## Conclusions

In summary, we have unlocked the synthetic potential and versatility of EtOH as a cost-efficient promising acyl surrogate to tag a wide variety of phenol-containing bioactive compounds. In particular, we have explored the utility of a directed Pd-catalyzed C–H acylation manifold for the modification of phenols featuring a removable pyridine unit as the DG. Unlike the classical acetylation methods available for the assembly of *ortho*-hydroxyaryl ketones, this protocol enables the rapid installation of an acetyl group from a chloride-free coupling partner in a site-selective fashion. As a result, this tagging platform represents an attractive means for the diversification of intricate phenolic units, which can be applied to the use of other aliphatic alcohols and aldehydes. The strength of this method lies in the facile introduction and removal of the required pyridyl ether as the DG, water compatibility and use of abundant chemical feedstock. Mechanistic studies supported a radical pathway involving the intermediacy of an acetate-bridged palladacyclic dimer as a key species within the acylation process. We anticipate that this Pd-catalyzed C–H acylation manifold could become a useful tool for the downstream functionalization of phenol-containing value-added products employing readily available C–H coupling partners.

## Author contributions

C. Girón-Elola conducted the experiments and A. Correa supervised the project. The manuscript was written with contri-



**Scheme 4** Proposed reaction pathway for Pd-catalyzed C–H acylation with EtOH.



Contributions from both authors and both of them have given approval to the final version of the manuscript.

## Data availability

Experimental details and characterization of the complexes can be found in the ESI.†

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

We are grateful to the Ministerio de Ciencia e Innovación (PID2021-122889NB-I00 and MCI/AEI/FEDER, UE) and Basque Government (IT1741-22) for financial support. We acknowledge technical and human support provided by SGIker of UPV/EHU and European funding (ERDF and ESF). C.G.-E. acknowledges the Ministerio de Universidades for a FPU contract (FPU22/00522). Dr Miguel A. Huertos is kindly acknowledged for insightful mechanistic discussions. Dr Leire San Felices and Dr Jose Ignacio Miranda from SGIker are acknowledged for their support with the X-ray and NMR analysis, respectively.

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