

REVIEW

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## Recent advancements in Ni/photoredox dual catalysis for $Csp^3$ – $Csp^3$ cross-coupling reactions

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### 1. Introduction

The construction of carbon–carbon bonds is one of the core research topics in the field of organic chemistry.<sup>1</sup> A transition-metal-catalyzed cross-coupling reaction is one of the most effective methods to forge carbon–carbon bonds,<sup>1d</sup> and typical transition metal catalyzed cross-coupling reactions include the Kumada, Negishi, and Suzuki cross-coupling reactions. These reactions could produce  $Csp^2$ – $Csp^2$  cross-coupling products very efficiently in the presence of a catalytic amount of transition metal under mild conditions. However, the formation of  $Csp^3$ – $Csp^3$  bonds is usually difficult to achieve under these transition metal catalytic conditions due to the following factors:<sup>2</sup> (1) severe reaction conditions are required for the oxidative addition of more electron-rich  $Csp^3$ –X bonds with the transition metal and reductive elimination of the alkyl group from the metal center, (2) the  $Csp^3$ –M intermediate easily undergoes  $\beta$ -H elimination, affording alkene byproducts, and (3) alkyl halides are susceptible to base-assisted H–X elimination and halide exchange side reactions under cross-coupling conditions. In 2017, Fu and his co-workers summarized transition-metal-catalyzed alkyl–alkyl cross-coupling reactions,<sup>3</sup> demonstrating the initial progress of unactivated alkyl electrophilic reagents for  $Csp^3$ – $Csp^3$  coupling, which also provided a key driving force for the development of Ni/photoredox dual catalysis.

Fortunately, odd-electron processes are not uncommon for nickel complexes. Thus, nickel species have abundant

and adjustable valence states ( $Ni^0/Ni^I/Ni^{II}/Ni^{III}/Ni^{IV}$ ).<sup>4a</sup> The reactivity of the Ni species is enriched by the availability of the  $Ni^I$  and  $Ni^{III}$  states as well as the possible involvement of radical processes.<sup>4b,c</sup> Moreover, under visible light irradiation, nickel is capable of undergoing single electron transfer (SET) smoothly, and the corresponding valence state will be raised or lowered by a single valence, making it more likely for free radicals to be generated mildly when exposed to light.<sup>4d,e,f</sup>

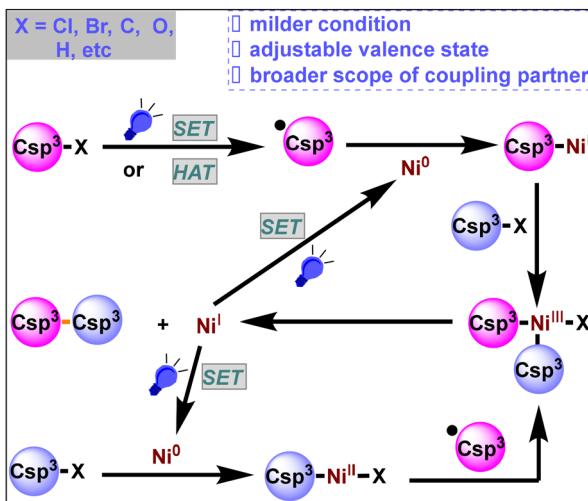
Traditional nickel-catalyzed coupling reactions involve three key steps:<sup>5a</sup> oxidative addition, transmetalation, and reductive elimination. Transmetalation is usually the decisive step in this type of reaction, with the exchange rate of different forms of carbon hybridization being  $Csp^2 > Csp^2 > Csp^3$ , making the direct construction of  $Csp^3$ – $Csp^3$  bonds particularly difficult.<sup>5b</sup> Owing to their radical-generating properties and abilities to modulate the oxidation state of metal catalysts,<sup>5c</sup> photocatalysts greatly broaden the range of electrophilic and nucleophilic reagents for cross-coupling reactions and are compatible with more reactive groups. Photocatalytic single-electron transmetalation is more suitable for  $Csp^3$ – $Csp^3$  cross-coupling reactions than the conventional nickel-catalyzed transmetalation process using metal catalysis alone.

In recent years, the introduction of photocatalysts has expanded the range of coupling partners (such as  $Csp^3$ –Cl bonds,  $Csp^3$ –Br bonds,  $Csp^3$ –C bonds,  $Csp^3$ –O bonds,  $Csp^3$ –H bonds, etc.) and has greatly changed organic synthetic chemistry (Scheme 1). In this review, we summarize recent advancements in Ni/photoredox dual catalysis for  $Csp^3$ – $Csp^3$  cross-coupling reactions over the last five years (2019–2023). We attempted to classify this context into three broad categories: neutral, reductive and oxidative cross-coupling according to the electrical properties of the coupling partners. Please note that some relevant information referred to can be found in the ESI section of the cited articles.

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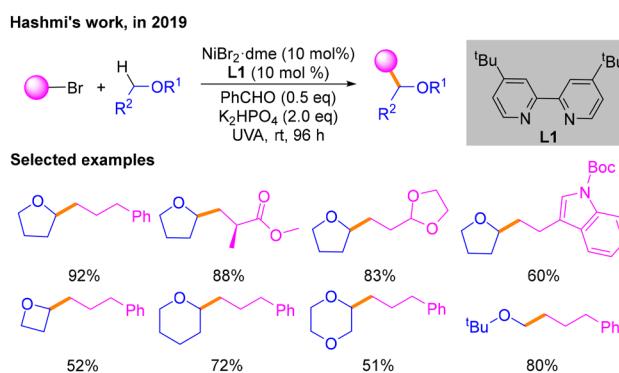


Scheme 1 Ni/photoredox dual catalyzed cross-coupling reactions.

## 2. Redox neutral cross-coupling

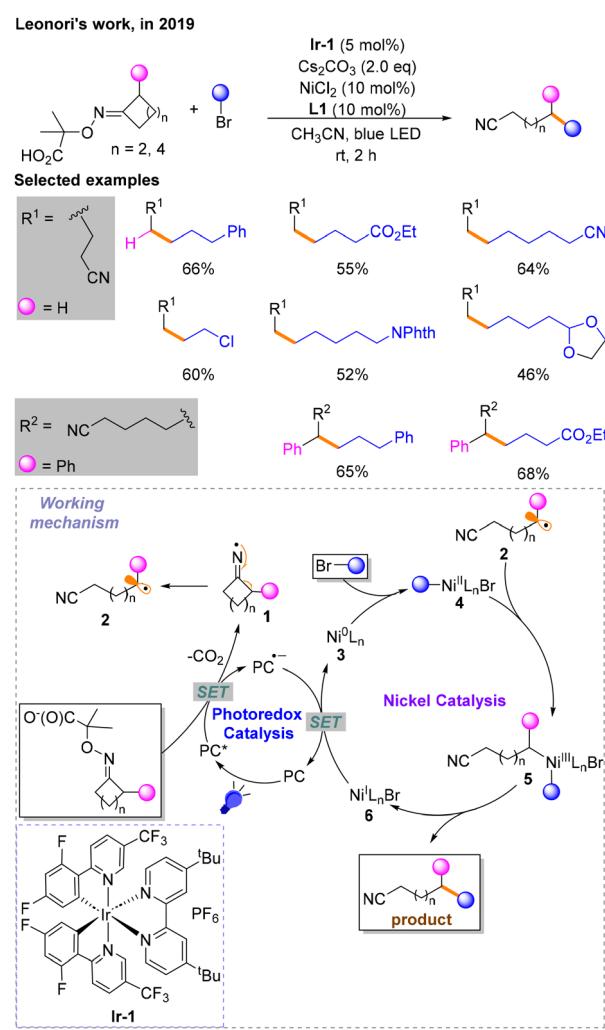
Redox neutral cross-coupling is a reaction strategy of great importance in organic synthetic chemistry.<sup>6</sup> It is characterized by the design and selection of suitable reaction conditions so that the functional groups involved in the reaction do not undergo significant oxidation state changes during the course of the reaction, thereby improving the atom economy and selectivity of the reaction. In recent years, a number of redox neutral cross-coupling reactions under photoredox/nickel dual catalysis have been developed.

In 2019, according to the fact that triplet benzaldehyde excited using ultraviolet light can interact with hydrogen-donating solvents to produce  $\alpha$ -hydroxybenzyl and solvents such as tetrahydrofuran (THF) radicals as hydrogen atom transfer (HAT) catalysts possessing a strong ability to capture hydrogen, Hashmi and co-workers explored a platform for direct alkylation of  $\text{Csp}^3$ -H bonds with alkyl halides *via* a HAT process (Scheme 2).<sup>7</sup> The optimal reaction conditions are irradiation with UVA light,  $\text{NiBr}_2\text{-dme}$  and **L1** as the nickel

Scheme 2  $\text{Csp}^3$ - $\text{Csp}^3$  cross-coupling reaction of alkyl bromides with ethers (Hashmi's work).

catalyst, benzaldehyde as both the photosensitizer and the HAT catalyst, and  $\text{K}_2\text{HPO}_4$  as the base. The substrate scope of bromide and ether coupling partners was investigated. However, the detailed mechanism requires more concrete study.

A few months later, Leonori and co-workers reported a divergent  $\text{Csp}^3$ - $\text{Csp}^3$  coupling reaction *via* iminium radical ring opening in conjunction with nickel catalysis and photocatalysis, enabling remote alkylation of nitriles (Scheme 3).<sup>8</sup> The conditions for the highest yield are identified: **Ir-1** as the photocatalyst (PC),  $\text{Cs}_2\text{CO}_3$  as the base,  $\text{NiCl}_2$  with **L1** as the nickel catalytic system and  $\text{CH}_3\text{CN}$  as the solvent under blue LED irradiation. Single-electron oxidation and decarboxylation occur to provide radical **1** in the process of photocatalyst quenching in the excited state ( $\text{PC}^*$ ) to  $\text{PC}^-$ , followed by single-electron  $\beta$ -scission to give alkyl radical **2**. The nickel catalytic cycle begins with the oxidative addition of  $\text{Ni}^0$  species **3** with brominated hydrocarbon to produce  $\text{Ni}^{II}$  species **4**, followed by the addition of **2** generated in the photocatalytic



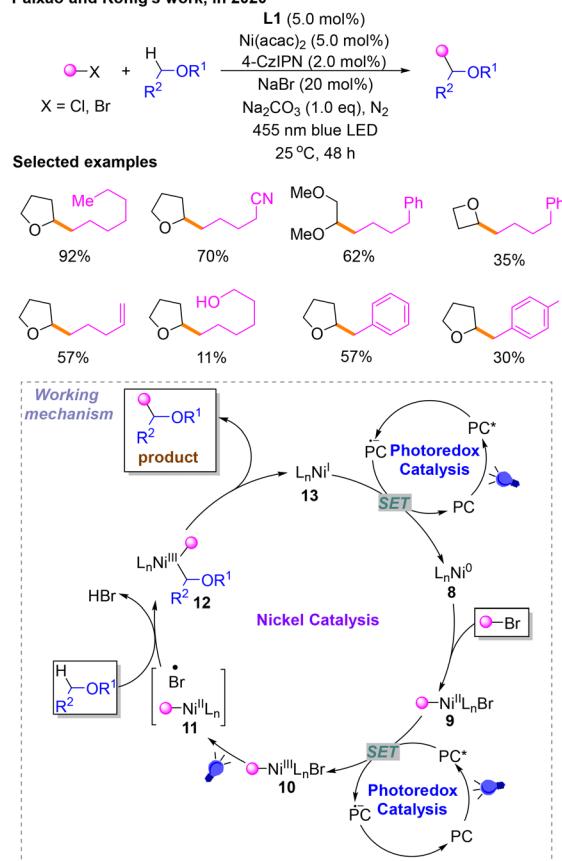
Scheme 3 Coupling reaction of iminium radical ring-opening in conjunction with Ni-catalysis.

stage to give  $\text{Ni}^{\text{III}}$  species 5, and finally reductive elimination to obtain a  $\text{Csp}^3\text{-Csp}^3$  coupling product. The  $\text{Ni}^{\text{I}}\text{Br}$  6 generated from the above process is reduced to 3, and  $\text{PC}^{\cdot-}$  is oxidized to PC simultaneously *via* a SET process.

Two months later, a method using 4-alkyl-1,4-dihydropyridines (DHPs) with oxa- and azabenzonorbornadienes to construct  $\text{Csp}^3\text{-Csp}^3$  architectures was developed by Molander's group (Scheme 4).<sup>9</sup> Notably, the regioselective and diastereoselective *cis*-1,2-dihydro-1-naphthyl alcohol backbone's structure can be achieved through this cross-coupling process under a nickel/photoredox dual catalytic system. The best conditions for the reaction are 4-CzIPN as the PC,  $\text{NiCl}_2\text{-dme}$  with L2 or L3 as the nickel catalytic system, and acetone as the solvent under blue LED irradiation. A series of mechanistic experiments have proven the impossible activation of the substrate by by-products and the product-determining-step involving no benzyl residues. According to density functional theory (DFT) calculations, in this program, a  $\pi\text{-}\sigma$  isomerization process was believed to occur in  $\text{Ni}^{\text{III}}$  species 7. The Curtin-Hammett principle supported the product produced with 1,2-regioselectivity.

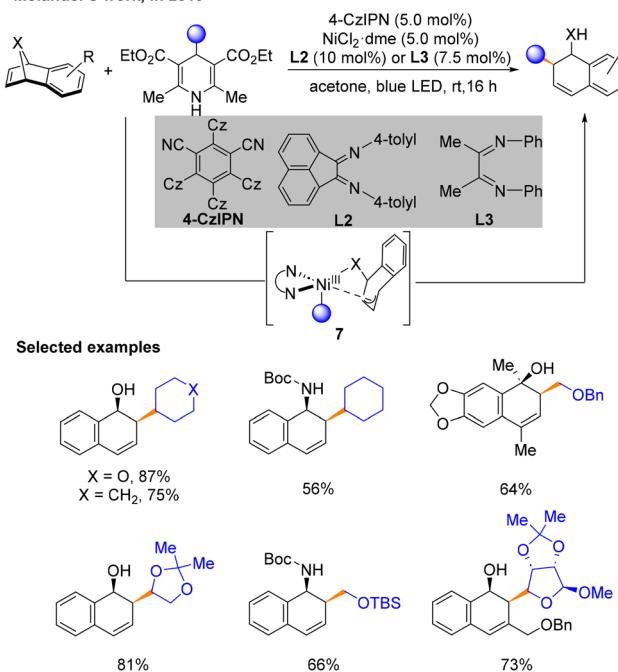
In 2020, taking into account the previously mentioned study of other researchers, a protocol was introduced for a  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reaction of alkyl bromides with ethers using a Ni/photoredox dual catalyst *via* a halide-mediated HAT process by Paixão and König (Scheme 5).<sup>10</sup> In this platform, they used 4-CzIPN as the photocatalyst to optimize the reaction conditions, and it is determined that the reaction should be carried out upon irradiation with a blue

Paixão and König's work, in 2020



**Scheme 5**  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reaction of alkyl bromides with ethers (Paixão and König's work).

Molander's work, in 2019

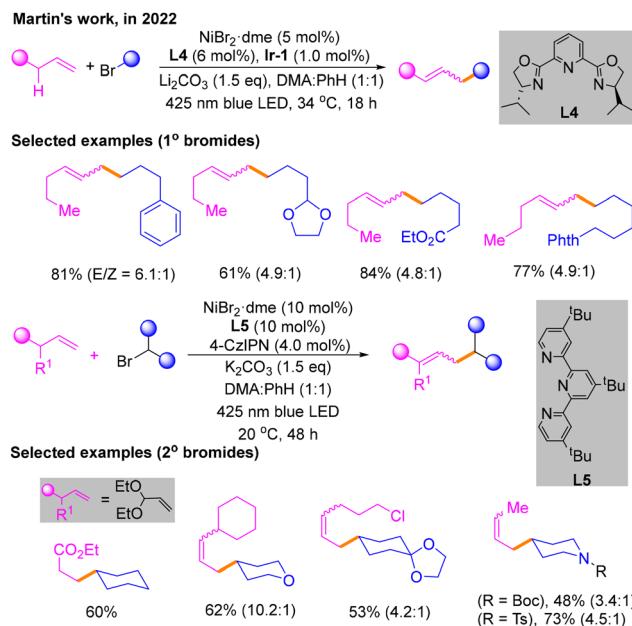


**Scheme 4** Cross-coupling of oxa- and azabenzonorbornadienes with DHP (Molander's work).

LED in the presence of  $\text{Na}_2\text{CO}_3$  using  $\text{Ni}(\text{acac})_2$  and L1 as the catalyst and 4-CzIPN as the photosensitizer in THF. The catalytic cycle begins with the oxidative addition of the  $\text{Ni}^0$  species 8 with a bromoalkane substrate to obtain the  $\text{Ni}^{\text{II}}$  species 9. Then, 9 is oxidized to afford  $\text{Ni}^{\text{III}}$  10 by  $\text{PC}^{\cdot-}$  to deliver  $\text{PC}^{\cdot-}$ . Subsequently, 10 absorbs visible light, generating the  $\text{N}^{\text{II}}$  species 11 and a halogen radical as a HAT agent. The generation of alkyl radicals is caused by the HAT of substrate ether and captured by 11 to provide 12. This HAT process is induced by the bromine radicals produced by the homolysis of the  $\text{Ni-Br}$  bond from 10 under the irradiation of a blue LED. The product comes from the reductive elimination of 12. Finally,  $\text{Ni}^{\text{I}}$  species 13 derived from 12 is reduced to  $\text{Ni}^0$  8 by  $\text{PC}^{\cdot-}$  to deliver the PC, which makes the catalytic cycle complete.

After that, combining a light-induced system with nickel catalysis, redox-neutral and site-selectivity  $\text{Csp}^3\text{-H}$  alkylation of an unactivated  $\alpha$ -olefin with an alkyl bromide was reported by Martin's group in 2022 (Scheme 6).<sup>11</sup> The reaction is able to obtain the maximum yield of primary alkyl bromides under blue LED irradiation, with  $\text{NiBr}_2\text{-dme}$  and L4 as the nickel catalytic system, Ir-1 as the photocatalyst of light,  $\text{K}_2\text{CO}_3$  as the base, and dimethylacetamide (DMA):PhH (1:1) as the solvent. Surprisingly, the optimal conditions of secondary



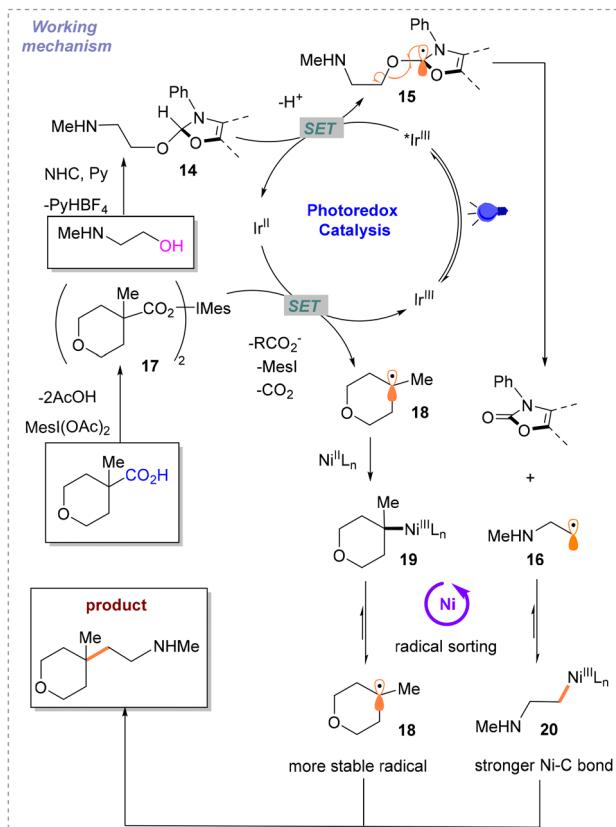
Scheme 6  $\text{Csp}^3\text{-H}$  alkylation of an unactivated  $\alpha$ -olefin.

bromide hydrocarbons are not exactly the same in ligands and photocatalysts. For secondary bromides, **L5** is the best ligand with 4-CzIPN as the photocatalyst.

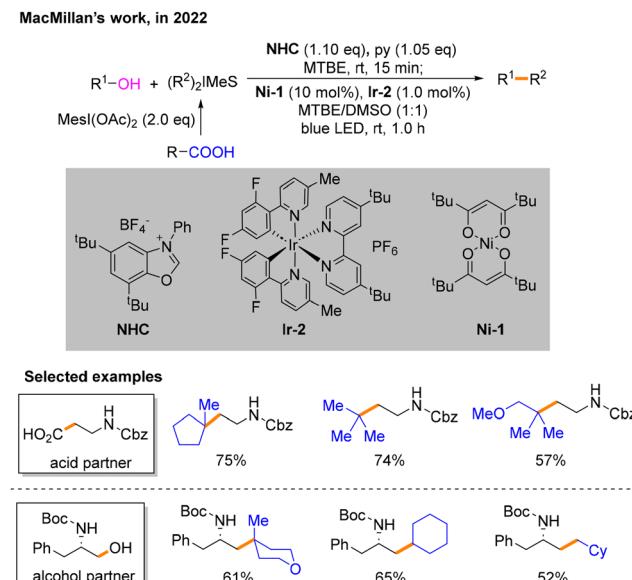
Soon after, MacMillan and co-workers developed a strategy for photo- and nickel-catalyzed  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reactions involving N-heterocyclic carbene (NHC)-mediated deoxygenation of alcohols and hypervalent iodine-mediated decarbonylation of carboxylic acids (Scheme 7).<sup>12</sup> Methyl tertiary butyl ether (MTBE):dimethyl sulfoxide (DMSO) (1:1) is used as the solvent. Different from previously reported schemes, the two  $\text{Csp}^3$  radicals involved in the reaction are both generated

during the photocatalytic SET process. In this event, the alcohol substrate reacts with NHC to afford the product **14**, which can undergo SET with  $^*\text{Ir}^{\text{III}}$ . The resulting intermediate **15** undergoes the process of  $\beta$ -breakage to generate radical **16**. Carboxylic acids are activated by iodine hypervalent to give **17**.  $\text{Ir}^{\text{II}}$  then undergoes SET with **17** to generate another free radical **18** and  $\text{Ir}^{\text{III}}$ . To maintain the catalytic cycle, blue light irradiation excites  $\text{Ir}^{\text{III}}$  to  $^*\text{Ir}^{\text{III}}$ . The difficulty of this strategy is distinguishing between the two different free radicals **16** and **18** generated instantaneously. To solve this problem, based on the difference in the relative instability of the highly substituted metal-alkyl species and different strengths of nickel–carbon bonds, a radical sorting method was conceived. That is, **20** is more stable than **19**, but their radical stabilities are the opposite with radical **18** being more stable than radical **16** (Scheme 8). After verifying the practicality of this idea, the author explored its generality with primary, secondary, and tertiary alkyl acids and alcohols under the optimal conditions.

Two months later, a platform for cross-coupling mediated by Ni and a photocatalyst is developed by Martin's group to build  $\text{Csp}^3\text{-Csp}^3$  bonds utilizing dihydroquinazolinones derived from ketones (Scheme 9).<sup>13</sup> In this metal and photocatalyst dual catalytic system, the Ni and photo-induced catalytic cycle is similar to that discussed previously. Haloalkane substrates produce  $\text{Ni}^{\text{II}}$  species **21** in the Ni catalytic cycle.



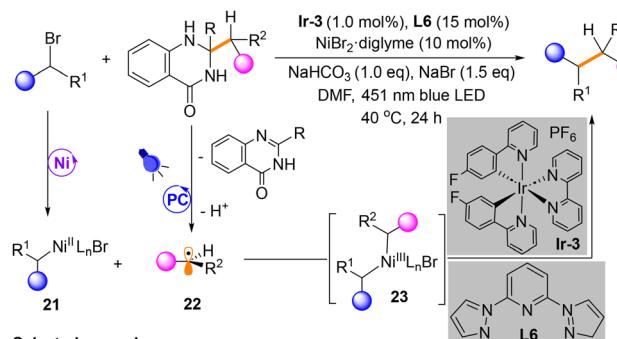
Scheme 8 Proposed reaction design of cross-coupling reactions of carboxylic acids with alcohols.



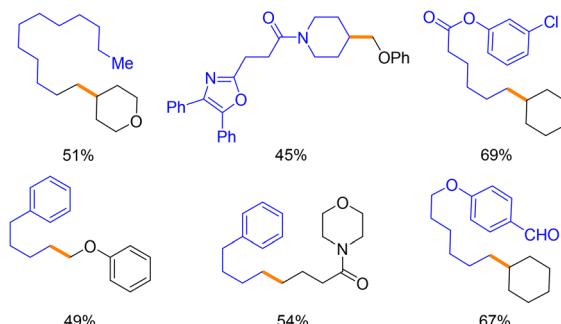
Scheme 7 Cross-coupling reaction of alcohols and carboxylic acids.



Martin's work, in 2022



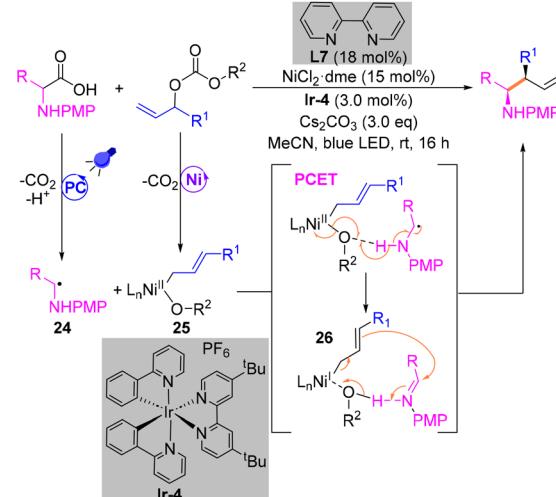
## Selected examples

Scheme 9  $Csp^3$ - $Csp^3$  cross-coupling reaction of dihydroquinazolinones with bromoalkanes.

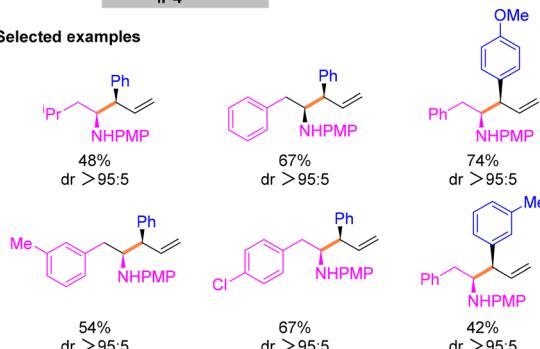
Alkyl radicals 22 are produced by the substrate dihydroquinazolinone in the photocatalytic phase. Then, 21 and 22 could undergo radical recombination to generate  $Ni^{III}$  species 23. The product is also obtained by the reductive elimination of 23. However, a distinguishing feature of this reaction is that the dihydroquinazolinone serves as a radical source causing  $\alpha$ -C-C cleavage through aromatization, which is induced by single-electron oxidation of photocatalysis. The selected examples have also been shown in Scheme 9.

Soon after, Breit and co-workers exploited a way for the construction of homoallylamines utilizing  $\alpha$ -amino acids *via* photoredox Ni-catalytic cross-coupling (Scheme 10).<sup>14</sup> This reaction enables the functionalization of allyl groups and diastereoselectively obtains branched allylation products involving  $\alpha$ -amino radical intermediates. The first investigation was conducted using model substrates and determined the most optimal conditions:  $NiCl_2\cdot dme$  and **L7** as the metal catalytic system,  $Cs_2CO_3$  as the base, and  $[Ir(ppy)_2(dtbbpy)]PF_6$  (**Ir-4**) as the PC. Subsequently, the arylallyl electrophile substrate was expanded using *N*-phenyl or *N*-*p*-methoxyphenyl (PMP) phenylalanine as the  $\alpha$ -amino radical precursor. Mechanistic experiments illustrated that the free radicals involved in the catalytic cycle are generated by SET in the photocatalytic cycle and then decarboxylation. The diastereoselectivity of products was explained by DFT calculations. A critical step is the proton coupled electron transfer (PCET) process of the radical 24 and  $Ni^{II}$  species 25 to afford 26. Then, 26 in combination with the allyl group undergoes a nucleophilic attack on the imine to give the desired product.

Bernhard Breit's work, in 2022



## Selected examples

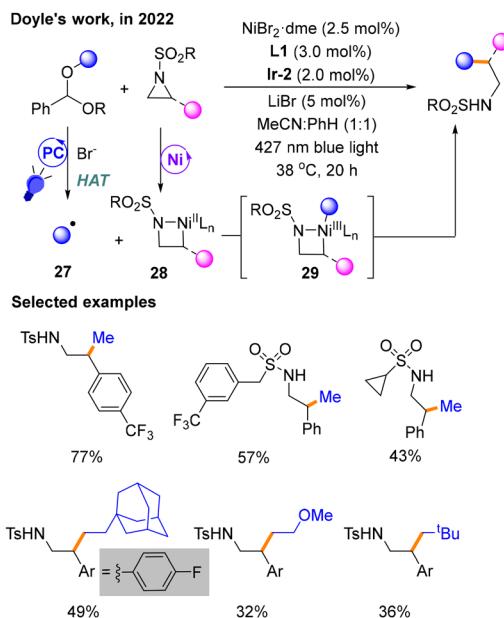


Scheme 10 Synthesis of homoallylamines by a cross-coupling reaction.

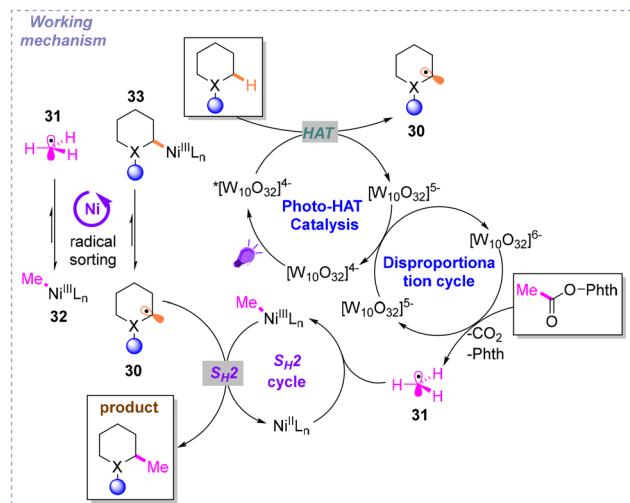
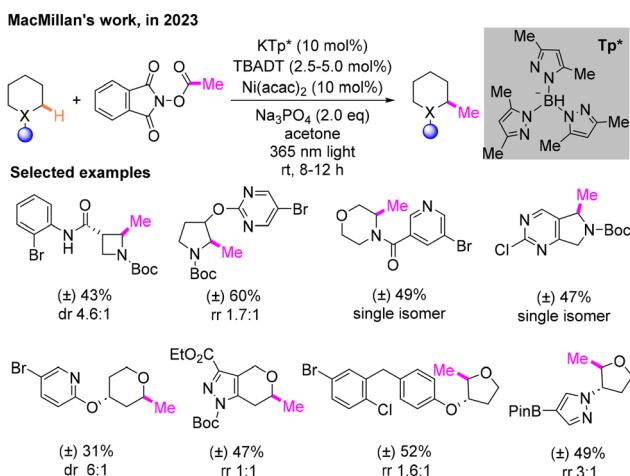
A few months later, Doyle's group utilizing benzaldehyde dialkyl acetals derived from alcohols developed a method for  $Csp^3$ - $Csp^3$  cross-coupling reactions with aziridines (Scheme 11).<sup>15</sup> The product is obtained with  $NiBr_2\cdot dme$  and **L1** as the nickel catalytic system using **Ir-2** as the PC under blue light exposure in  $MeCN:PhH$  (1:1) mixed solvent. The function of the additive  $LiBr$  is to stabilize the anionic sulfonamide and facilitate the release of the product. Benzaldehyde dialkyl ketals are employed by HAT to produce radical species 27 under the stimulation of a photocatalyst and bromide ions. Mechanistic studies show that the activation of dialkyl acetals *via* HAT by  $Br^-$  happens during the photocatalytic phase to give 27, and the oxidative addition of aziridines occurs in the Ni catalytic cycle to afford 28, explaining the differential activation of the two coupling partners. The product is released by the reductive elimination reaction from  $Ni^{III}$  species 29.

Then, inspired by the  $Csp^3$ -methyl group catalyzed by radical *S*-adenosylmethionine (SAM) methyltransferase in biological systems and some previous work, MacMillan and co-workers reported a  $Csp^3$ - $Csp^3$  coupling reaction through radical sorting and bimolecular homolytic substitution ( $S_H2$ ), in which direct methylation occurs on saturated heterocyclic  $Csp^3$ -H bonds (Scheme 12).<sup>16</sup> The most ideal conditions are irradiation under 365 nm light,  $Ni(acac)_2\cdot KTp^*$  as the metal





**Scheme 11**  $Csp^3$ – $Csp^3$  cross-coupling reaction of benzaldehyde dialkyl acetals with aziridines.

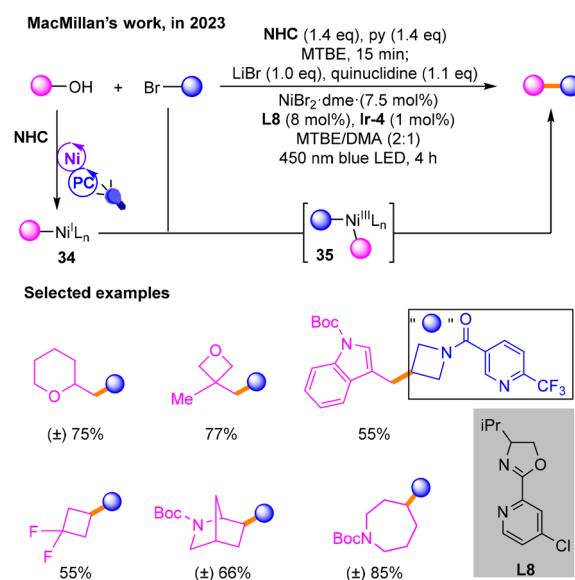


**Scheme 12** Methylation of saturated heterocyclic  $Csp^3$ -H bonds.

catalyst,  $Na_3PO_4$  as the base and tetrabutylammonium decatungstate (TBADT) as the HAT reagent and photocatalyst in acetone. The universality of this strategy is also assessed. These two alkyl radicals are respectively produced by HAT and decarboxylation in the photocatalytic cycle. The excited triplet state of decatungstate  $^3[W_{10}O_{32}]^{4-}$  under light undergoes a HAT process at the hydridic  $\alpha$ -amino C–H bond of the first substrate, resulting in reduced  $^3[W_{10}O_{32}]^{5-}$  as well as alkyl radical 30. Another radical 31 is produced during the disproportionation cycle from the second substrate. Key to the success of this mechanism is the previously mentioned radical sorting effect mediated by nickel. Radicals 30 and 32 are more stable than 31 and 33, respectively. In addition, products are obtained from the  $S_2H_2$  process.

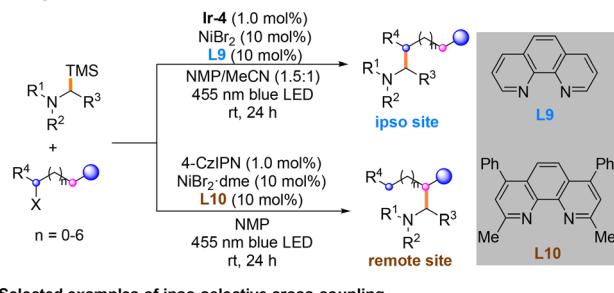
In 2023, Macmillan<sup>17</sup> used NHC to activate alcohols and performed coupling reactions with brominated hydrocarbons to build  $Csp^3$ – $Csp^3$  bonds (Scheme 13). In the photocatalytic cycle, the alcohols are activated by NHC-mediated C–O homolysis to produce alkyl radicals, which is similar to their previous reports. Then, the resulting  $Ni^I$  species 34 during nickel and photo-induced catalysis undergoes oxidative addition with halogenated hydrocarbons to produce  $Ni^{III}$  species 35, and finally the target product is obtained through reductive elimination of 35.

Later, a light/nickel-catalyzed  $Csp^3$ – $Csp^3$  cross-coupling reaction of  $\alpha$ -silylamines with unactivated alkyl halides, which could occur on the *ipso* and remote sites of alkyl halides selectively through ligand regulation, was for the first time described by Huang and Yuan in 2023 (Scheme 14).<sup>18</sup> They found that photosensitizer Ir-4 is the most suitable for the *ipso*-site reaction, and 4-CzIPN is best for the remote-site reaction. After careful screening of the reaction conditions,  $\alpha$ -N-

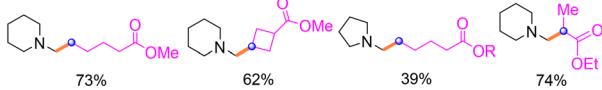


**Scheme 13** The cross-coupling reaction of activated alcohols with NHC and brominated hydrocarbons.

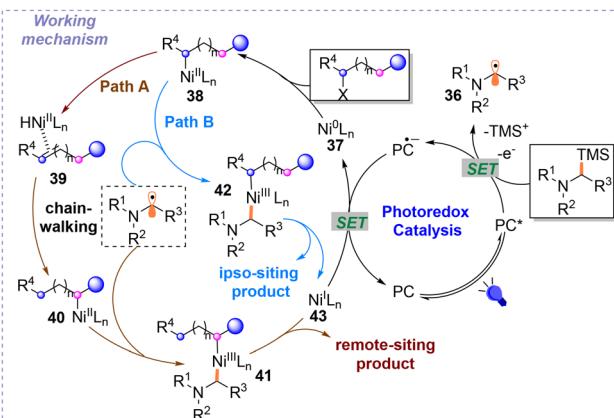
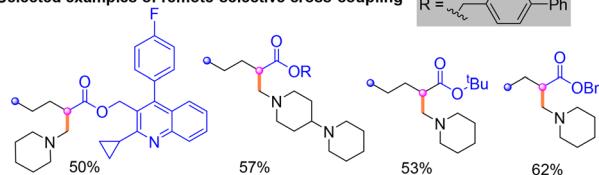
Huang and Yuan's work, in 2023



## Selected examples of ipso-selective cross-coupling



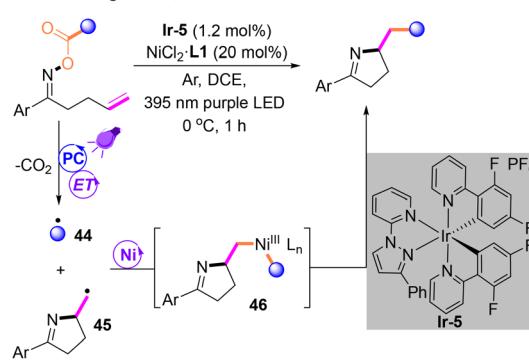
## Selected examples of remote-selective cross-coupling

Scheme 14 Csp<sup>3</sup>–Csp<sup>3</sup> cross-coupling reaction of  $\alpha$ -silylamines with alkyl bromides.

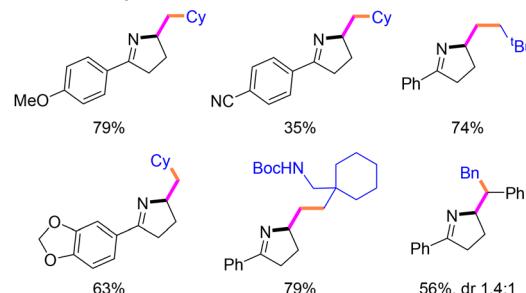
Alkyl radicals participating in the nickel catalytic cycle are generated during the photocatalytic cycle. When **L9** is used as the ligand, the olefin substrate will isomerize to form the *ipso*-site product (Path A). Moreover, the remote-site product can be output when the ligand is **L10** (Path B). Remote- and *ipso*-site products are obtained from different Ni<sup>III</sup> species: **41** and **42**. Mechanistic experiments have shown that  $\beta$ -hydrogen is necessary for the generation of remote-site products during the chain-walking process. DFT calculations proved that the steric hindrance of ligands is a factor affecting site selectivity.

After that, Liu and Fang reported a new Csp<sup>3</sup>–Csp<sup>3</sup> cross-coupling system through decarboxylation<sup>19</sup> mediated by a PC and the subsequent coupling of two radicals by a nickel catalyst (Scheme 15). This reaction was performed under the irradiation of a purple LED with dichloroethane (DCE) as the solvent, unlike most reactions that use a blue LED as the light source. Both radicals **44** and **45** are generated *via* an energy

Liu and Fang's work, in 2023



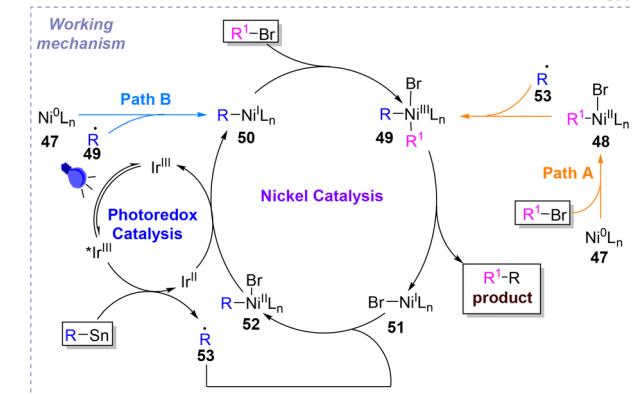
## Selected examples

Scheme 15 Csp<sup>3</sup>–Csp<sup>3</sup> decarboxylated cross-coupling reaction via E<sub>n</sub>T.

transfer (E<sub>n</sub>T) process during the photocatalytic stage, followed by a nickel-mediated cross-coupling reaction to afford Ni<sup>III</sup> species **46**. The reductive elimination of **46** occurs to produce the target product.

Subsequently, a photonickel-cocatalyzed Stille coupling reaction was developed by Zhu and Xue (Scheme 16).<sup>20</sup> After determining the reaction conditions for Csp<sup>2</sup>–Csp<sup>3</sup> coupling,

Zhu and Xue's work, in 2023



Scheme 16 Photo- and nickel-catalyzed Stille coupling reaction.

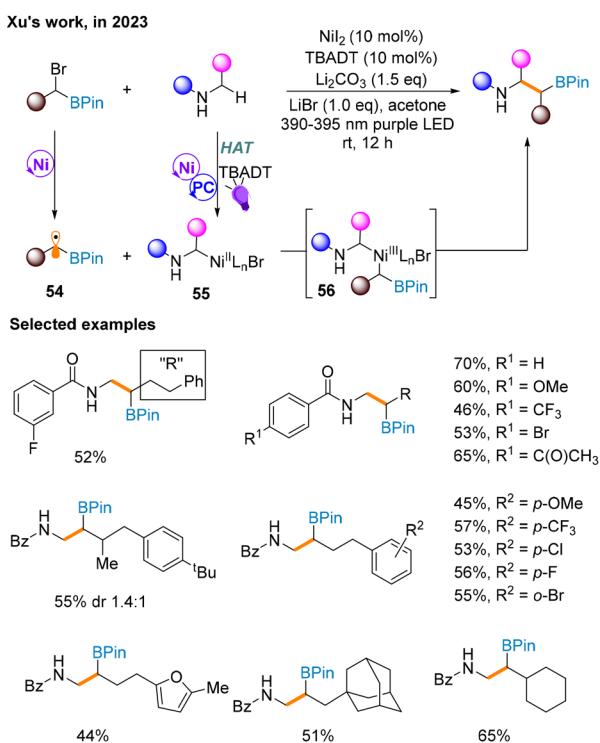
the author evaluated the scope of this protocol, in which only a  $\text{Csp}^3$ - $\text{Csp}^3$  coupling product was reported. Several mechanistic studies and DFT calculations were carried out to show how this reaction proceeds. There are two pathways A and B for this reaction. The difference between them is whether conventional oxidative addition occurs first to give **48** followed by single-electron oxidative addition to give **49** (Path A), or whether the opposite occurs to give **50**, followed by **49** (Path B). The products and  $\text{Ni}^1$  species **51** of both pathways are obtained by reductive elimination of  $\text{Ni}^{\text{III}}$  species **52**. Radical **53** could be captured by **51** to get **52**, which would be reduced to **50**.

A month later, a  $\text{Csp}^3$ - $\text{Csp}^3$  cross-coupling reaction of  $\alpha$ -bromoboronates and  $\beta$ -amino boronates through C-H bond activation is reported by Xu and co-workers to forge  $\beta$ -amino boronates (Scheme 17),<sup>21</sup> whose synthesis is currently rare. It is worth noting that the additional use of ligands will decrease the yield, and thus this strategy does not involve supernumerary ligands. In addition, unlike most of the above methods, this strategy uses the irradiation of purple light to maintain the catalytic cycle. Mechanistically,  $\alpha$ -bromoboronate undergoes a one-electron redox reaction in nickel catalysis to produce  $\alpha$ -borate radical **54**. The  $\alpha$ -*N*-alkyl radical produced by the photocatalytic cycle via the HAT process is captured by the Ni catalytic cycle to afford  $\text{Ni}^{\text{II}}$  species **55** and then combined with **54**. Finally, the product is obtained through reductive elimination of  $\text{Ni}^{\text{III}}$  species **56**.

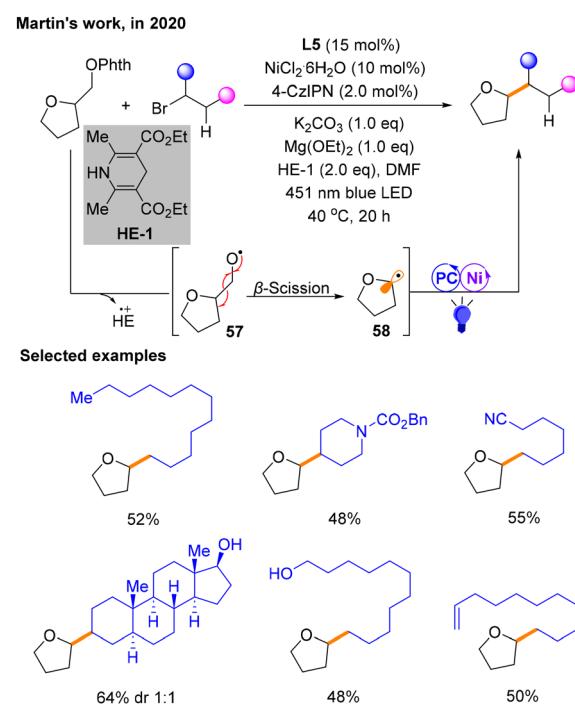
### 3. Reductive cross-coupling

Reductive cross-coupling reactions are also an important class of reactions in organic chemistry.<sup>22</sup> Reactants usually undergo electron transfer and recombination of chemical bonds in the presence of a reducing agent. In this reaction, a reducing agent is used to facilitate the formation of new carbon–carbon bonds between different organic fragments. In recent years, there have been a number of reductive coupling reactions under photoredox/nickel dual catalysis.

In 2020, Martin and co-workers demonstrated the successful development of a dual photoredox/Ni-catalytic process that gives access to alkylation of  $\text{Csp}^3$  via  $\beta$ -scission of a special class of aliphatic alcohol derivatives, *N*-phthalimide ethers (Scheme 18).<sup>23</sup> Specifically, utilizing 4-CzIPN as the PC,  $\text{K}_2\text{CO}_3$  as the base, Hantzsch ester (HE-1) as the electron donor (ED) and  $\text{NiCl}_2\text{L5}$  as the metal catalyst under the irradiation of blue light, a series of  $\text{Csp}^3$ - $\text{Csp}^3$  coupling structures were produced. The range of  $\text{Csp}^3$  alkylation was studied, explaining the influence of ligand denticity. Some examples are selected and shown in Scheme 18. In this protocol, HE-1 is not only able to reduce *N*-phthalimide ethers (Ophth) to provide radicals **57**, but it can also maintain the photocatalytic cycle. The photocatalytic cycle does not generate **57** involved in coupling, but **57** is generated by HE. Subsequently, the  $\beta$ -breakage of **57** occurs through the single electron activity of the ether with  $\text{HE}^+$ , generating alkyl radical species **58** that participate in the nickel and photocatalytic cycle to afford the product. In addition, the authors suggest that there may be an electron-



**Scheme 17** Synthesis of  $\beta$ -amino boronates via a  $\text{Csp}^3$ - $\text{Csp}^3$  cross-coupling reaction.

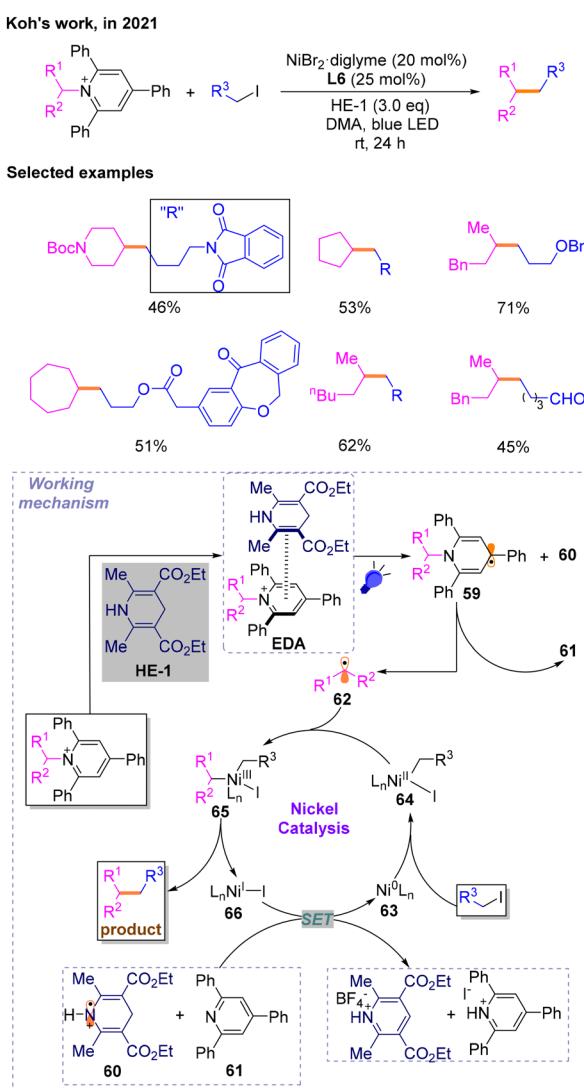


**Scheme 18** Cross-coupling reaction of  $\text{sp}^3$  alkylation via  $\beta$ -scission of *N*-phthalimide ethers.



donor-acceptor (EDA) complex between HE and the substrate prior to  $\beta$ -scission.

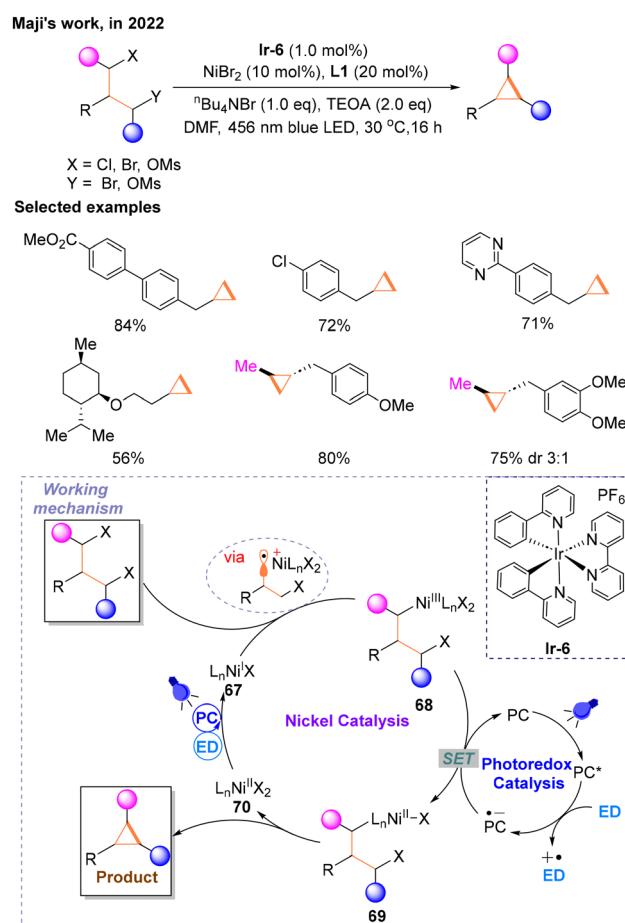
A year later, affected by the redox active characteristics of *N*-alkylpyridinium salts, a technology for constructing a  $Csp^3$ - $Csp^3$  skeleton *via* a photo-induced and nickel-catalyzed cross-electrophile coupling reaction with *N*-alkylpyridinium salts as radical precursors was disclosed by Koh in 2021 (Scheme 19).<sup>24</sup> After optimization studies, the author explored the substrate range of deamination alkylation reactions with bromides and iodides, respectively, and found that the iodides could perform  $Csp^3$ - $Csp^3$  cross-coupling reactions effectively. Then, several mechanistic experiments were conducted, and it is believed that the proposed mechanism may involve EDA complexes formed between *N*-alkylpyridinium salts and HE-1. Then, photoexcitation and SET occur to generate dihydropyridine radicals 59 and  $HE^{+}$  60, followed by deamination cleavage triggered by aromatization to produce triphenylpyridine



**Scheme 19** The coupling reaction involving *N*-alkylpyridinium salts as radical precursors.

**61** and alkane radical **62** involved in the catalytic cycle. The species **60** and **61** will reduce  $Ni^I$  **66** to  $Ni^0$  **63**. **63** would then undergo an oxidative addition reaction with halogenated hydrocarbons to produce  $Ni^{II}$  **64**, which could capture the previously generated alkyl radicals **62** affording  $Ni^{III}$  species **65**. To complete the catalytic cycle, reductive elimination of **65** occurs to produce the desired products and **66**.

In 2022, Maji and co-workers proved that metallaphotoredox catalysis can be a prominent strategy to make the cyclopropane ring *via* an intramolecular  $Csp^3$ - $Csp^3$  cross-electrophile coupling reaction (Scheme 20) of 1,3-alkyl electrophiles.<sup>25</sup> The reaction conditions were optimized using primary alkyl 1,3-dimesylate with an ester functionality as a model substrate, and the optimal reaction conditions were finally determined: **Ir-6** as the PC,  $NiBr_2$  with **L1** as the transition metal catalyst,  $^nBu_4NBr$  as the additive, triethanolamine (TEOA) as the ED, *N,N*-dimethylformamide (DMF) as the solvent and the reaction was performed at room temperature under the irradiation of a blue LED. In addition, secondary alkyl electrophiles could also react smoothly when one of the halogen atoms in the substrate is replaced with OMs. As the author speculated,  $PC^*$  was quenched to  $PC^-$  by the ED. The nickel catalytic cycle starts



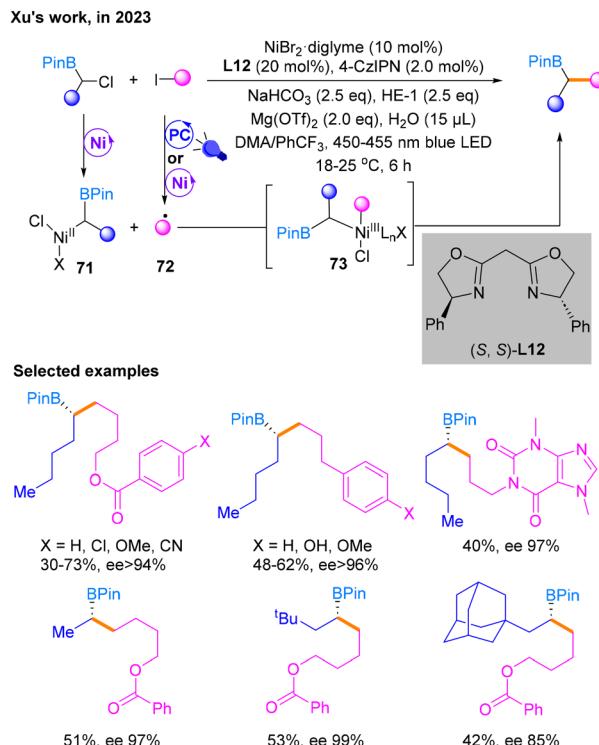
**Scheme 20** Synthesis of cyclopropanes *via* an intramolecular  $Csp^3$ - $Csp^3$  cross-coupling reaction.



with  $\text{Ni}^{\text{I}}$  **67** generated from a photocatalytic SET involving the ED, undergoing oxidative addition with halogenated hydrocarbons *via* a radical process to form  $\text{Ni}^{\text{III}}$  **68**, which is subsequently reduced to  $\text{Ni}^{\text{II}}$  **69** by  $\text{PC}^{\text{-}}$  delivering the PC.  $\text{PC}^{\text{-}}$  is regenerated under the blue light irradiation and ED to complete the photocatalytic cycle. Subsequently, **69** undergoes oxidative addition and reductive elimination to produce the target product. The new  $\text{Ni}^{\text{II}}$  species **70** produced in this step undergoes the same photocatalytic process and is reduced to  $\text{Ni}^{\text{I}}$  **67** to enter the next catalytic cycle.

Later, Oestreich's research group achieved a reductive cross-coupling reaction between  $\alpha$ -silylated alkyl bromides and an allylic sulfone using a HE as the reducing agent in 2022 (Scheme 21).<sup>26</sup> It has been experimentally verified that the key to the successful conduction of the reaction is the ability of the methylsilane at the  $\alpha$ -position of the alkyl bromide to stabilize the carbon radicals. After conditional screening, it is determined that the most qualified conditions are  $\text{NiBr}_2\text{-diglyme}$  as the nickel precatalyst and terpyridine **L11** as the ligand as well as HE-2 and  $\text{Et}_3\text{N}$  under irradiation with a blue LED in DMA as the solvent. The reaction scope of this coupling reaction was subsequently evaluated under these conditions. Notably, this process does not require a photocatalyst but only blue light irradiation to maintain the catalytic cycle. Thus, further study is needed to gain mechanistic insights.

In 2023, Xu and co-workers described  $\text{Csp}^3\text{-Csp}^3$  cross-coupling with dual Ni/photocatalysis utilizing unactivated alkyl halides and racemic  $\alpha$ -chloroboronates as substrates (Scheme 22).<sup>27</sup> This strategy enables the highly selective synthesis of chiral secondary alkylborates, important intermediates in organic synthesis, making it possible to rapidly construct enantiomerically enriched complex molecules.

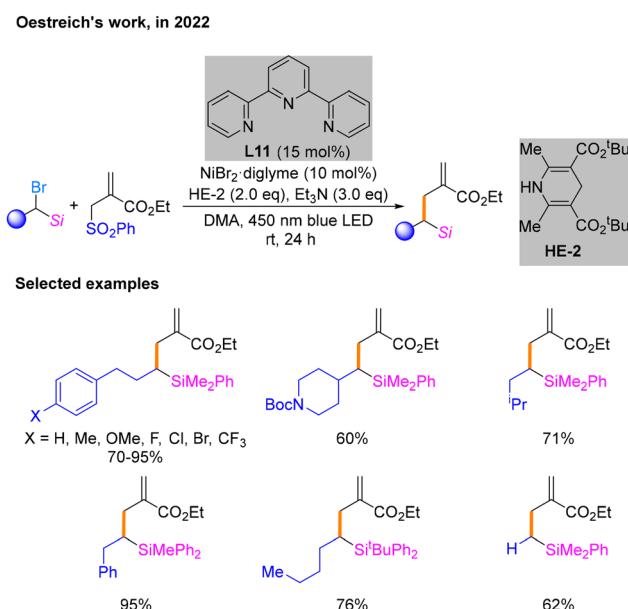


**Scheme 22** Cross-coupling of  $\alpha$ -chloroboronates with alkyl iodides

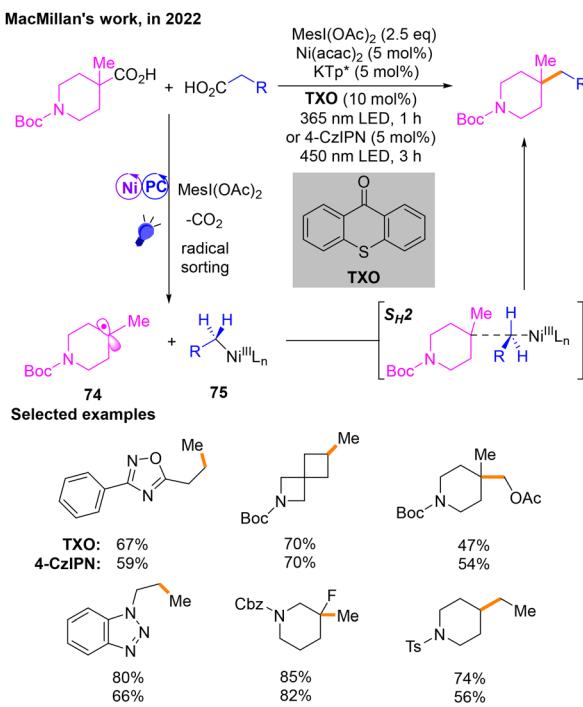
This reaction could achieve the highest yield in the  $\text{NiBr}_2$ -diglyme catalytic system using 4-CzIPN as the PC and DMA : PhCF<sub>3</sub> (1 : 1) as the solvent in the presence of some additives. Then, the substrate scope of  $\alpha$ -chloroborates and alkyl iodides for reductive cross-coupling was expanded. Based on radical-clock reactions and other mechanistic verification experiments, the author proposed that a Csp<sup>3</sup> radical was formed in the photocatalytic stage from alkyl iodides under the irradiation of blue light. In addition,  $\alpha$ -chloroboronates are activated in the nickel catalytic cycle to afford Ni<sup>II</sup> species 71. Subsequently, free radical addition and reductive elimination with 72 occur successively to generate the final product through the Ni<sup>III</sup> species 73. HE-1 also acts as the ED to maintain the photocatalytic cycle.

#### 4. Oxidative cross-coupling

Oxidative cross-coupling offers a new approach for constructing C–C bonds.<sup>28a</sup> In this type of reaction, transition-metal catalysts and oxidants are usually utilized to induce cross-coupling of two different nucleophilic organic molecules,<sup>28b,c</sup> resulting in the formation of new carbon–carbon or carbon–hetero bonds. In recent years, the construction of  $\text{Csp}^3$ – $\text{Csp}^3$  bonds by oxidative cross-coupling has been greatly developed, especially in the field of electrocatalysis.<sup>28d</sup> However, with regard to Ni/photoredox dual catalysis, there are limited cases of  $\text{Csp}^3$ – $\text{Csp}^3$  bond construction in oxidative cross-coupling.



**Scheme 21** Cross-coupling reaction of  $\alpha$ -silylated alkyl bromides and an allylic sulfone.



Scheme 23 Decarboxylative cross-coupling of aliphatic acids.

In 2022, a decarboxylative  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reaction<sup>29</sup> of aliphatic acids was developed by MacMillan's group (Scheme 23). The two free radicals of this reaction are produced after the corresponding aliphatic carboxylic acid substrates are activated by  $\text{MesI(OAc)}_2$ . Then radical 74 and  $\text{Ni}^{\text{III}}$  species 75 are produced during the radical sorting process, and they finally produce the target product through  $\text{S}_{\text{H}2}$ . When thioxanthone (TXO) is used as the photosensitizer, the reaction can occur under 365 nm violet light irradiation, and it is worth noting that when the photosensitizer is changed to 4-CzIPN, the product can also be obtained under the irradiation of a 450 nm blue LED. In addition,  $\text{MesI(OAc)}_2$  acts as the oxidant in this reaction.

## 5. Conclusion and outlook

This minireview mainly covers the research on Ni/photoredox dual catalyzed  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reactions from 2019 to 2023. Under clean and mild light irradiation conditions and cheap nickel metal catalysis, various compounds containing C-C structures can be easily obtained, avoiding the difficulty of using traditional metal-catalyzed coupling reactions. To generate  $\text{Csp}^3$  radicals, a number of chemical bonds ( $\text{Csp}^3\text{-Cl}$  bonds,  $\text{Csp}^3\text{-Br}$  bonds,  $\text{Csp}^3\text{-C}$  bonds,  $\text{Csp}^3\text{-O}$  bonds, *etc.*) have flexibly participated in redox-neutral coupling reactions and reductive cross-coupling reactions in the past five years. In some specific transformations, coupling reactions have been developed that do not require additional photocatalysts or ligands. However, most coupling reactions are driven by blue

LED light, and very few reactions are driven by purple light. In addition, the stereo- and regio-selectivities of these reactions still need further study.

In the next stage, we anticipate that the development trend of the  $\text{Csp}^3\text{-Csp}^3$  coupling reaction in Ni/photoredox dual catalysis may have the following six aspects: (a) recently, the asymmetric version of this field has experienced significant development.<sup>30</sup> It is believed that more efforts will be devoted to exploring asymmetric  $\text{Csp}^3\text{-Csp}^3$  coupling reactions by introducing chiral ligands into the catalytic system; (b) the mechanisms of some specific related reactions will be further clarified. Some new mechanisms will also emerge and become increasingly clear; (c) the formation method of C-centered  $\text{sp}^3$  hybridized radicals that can participate in  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reactions will be further expanded; (d) more classic conventional coupling reactions may be extended and also participate in the photo-nickel synergistic catalytic system to enable the formation of  $\text{Csp}^3\text{-Csp}^3$  structures; (e) site-selective  $\text{Csp}^3$  alkylation *via*  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reactions of dual photoredox/Ni-catalysis is another important direction and requires future development. We believe that the challenges in this area will eventually be overcome and more efficient, green, and gentle methods will be developed. (f) More types of light sources (such as red and green light) may be used in  $\text{Csp}^3\text{-Csp}^3$  cross-coupling reactions in the near future.

## Author contributions

Shi, M. directed the perspective and revised the manuscript. Huang, Q. Y. carried out the literature collection, organization, wrote the manuscript, drew the schemes and reviewed them.

## Data availability

No primary research results, software or code have been included and no new data were generated or analysed as part of this review.

## Conflicts of interest

There are no conflicts to declare.

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