



Cite this: *Nanoscale Horiz.*, 2024, 9, 278

Received 26th October 2023,
Accepted 21st November 2023

DOI: 10.1039/d3nh00480e
rsc.li/nanoscale-horizons

Carrier density and delocalization signatures in doped carbon nanotubes from quantitative magnetic resonance†

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High-performance semiconductor materials and devices are needed to supply the growing energy and computing demand. Organic semiconductors (OSCs) are attractive options for opto-electronic devices, due to their low cost, extensive tunability, easy fabrication, and flexibility. Semiconducting single-walled carbon nanotubes (s-SWCNTs) have been extensively studied due to their high carrier mobility, stability and opto-electronic tunability. Although molecular charge transfer doping affords widely tunable carrier density and conductivity in s-SWCNTs (and OSCs in general), a pervasive challenge for such systems is reliable measurement of charge carrier density and mobility. In this work we demonstrate a direct quantification of charge carrier density, and by extension carrier mobility, in chemically doped s-SWCNTs by a nuclear magnetic resonance approach. The experimental results are verified by a phase-space filling doping model, and we suggest this approach should be broadly applicable for OSCs. Our results show that hole mobility in doped s-SWCNT networks increases with increasing charge carrier density, a finding that is contrary to that expected for mobility limited by ionized impurity scattering. We discuss the implications of this important finding for additional tunability and applicability of s-SWCNT and OSC devices.

New concepts

Semiconducting single-walled carbon nanotubes (s-SWCNTs) are an important class of π -conjugated organic semiconductors (OSCs) that can enable emerging opto-electronic applications. Understanding charge transport mechanisms in s-SWCNTs (and OSCs generally) is essential for material and device design. Applications like photovoltaics, sensors, light-emitting diodes, field-effect transistors, and thermoelectric devices require good electrical conductivity and carrier mobility. Common methods to measure conductivity in OSCs do not allow to independently measure charge carrier density or mobility, making it difficult to answer important fundamental and applied questions and hindering performance optimization. To address this knowledge gap, we develop a combined approach, using optical absorption and nuclear magnetic resonance spectroscopy, to directly measure charge carrier density in doped s-SWCNTs. We demonstrate that carrier density impacts charge delocalization, leading to a carrier density-dependent mobility that is contrary to that expected for mobility limited by ionized impurity scattering. Combining simulations with our experimental data produces a correlation curve that enables determination of carrier density in doped s-SWCNTs by rapid and readily available absorption spectroscopy measurements. The results provide a valuable roadmap to the OSC community for tuning, quantifying, and optimizing carrier density for a broad range of energy harvesting and optoelectronic applications.

Introduction

Versatile semiconductors are needed to supply growing energy and computing demands.^{1,2} Organic semiconductors (OSCs)

are attractive for many applications such as photovoltaic cells, field-effect transistors, thermoelectrics, sensors and light-emitting diodes because they are inexpensive, easy to fabricate, and flexible.^{3–6} Semiconducting single-walled carbon nanotubes (s-SWCNTs) – an important class of nanoscale OSCs with high carrier mobility, strong optical transitions, and chemical stability^{16–19} – can enable emerging applications in digital logic, energy harvesting, and quantum information science.^{20–24} Importantly, understanding operational mechanisms and optimizing the performance of s-SWCNTs and other OSCs in such applications relies on both tuning and quantifying charge carrier type (electron/hole), density, and conductivity.

The most common way to tune carrier density and conductivity in OSCs is by doping with p- or n-type molecular dopants.^{7–11} Unfortunately, most common methods to measure

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† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3nh00480e>



OSC conductivity do not allow to separately extract charge carrier density and mobility. For example, Hall effect measurements that are commonly used to extract carrier density and mobility in crystalline inorganic semiconductors, are not well suited for amorphous OSCs.²⁵ As such, most studies on OSC (and s-SWCNT) transport must make several assumptions to interpret carrier density-dependent mechanisms or performance trends.^{12–15}

Nuclear magnetic resonance (NMR) has been used to indirectly estimate charge carrier density in inorganic semiconductors^{26,27} such as Cu_{2-x}Se nanoparticles,²⁸ based on the so-called Knight shift induced by free holes.²⁹ The dependence of the spin-lattice relaxation rate on temperature and carrier density, based on the Korringa behavior,³⁰ can be used to quantify free carriers. Inspired by these studies, we introduce a combined method, utilizing ¹⁹F NMR and absorption measurements on s-SWCNTs doped p-type with F-containing molecular dopants, to directly quantify free charge carriers and extract carrier mobility. Our measurements demonstrate that hole mobility increases substantially with increasing carrier density in doped s-SWCNT networks, reaching *ca.* $40 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ at high carrier density. Consistently, we observe a carrier density-dependent NMR chemical shift for dopant counterions, consistent with both the carrier density-dependent mobility of s-SWCNT holes and a Coulomb screening-induced modulation of the NMR ring current effect. This method can be broadly applied to a wide variety of π -conjugated semiconductors by utilizing dopants with suitable NMR-active nuclei as reporters.

Results and discussion

Bleaching of the first excitonic optical transition,³¹ S_{11} , induced by the phase space filling effect, is commonly used as a proxy for the relative free charge carrier density in SWCNTs.^{12,32} The absorption spectra shown in Fig. 1a show this bleaching upon doping SWCNTs p-type with non-covalently adsorbed dodeca-borane clusters (DDB-F72).^{33–35} Doping also produces a new envelope of overlapping, red-shifted absorption peaks, attributed to the absorption of positively charged excitons (trions, X^+). Plasma torch (PT) SWCNTs used in this work yield several chiralities resulting in the aforementioned envelope (*i.e.* sum of all excitonic transitions, S_{11} , corresponding to each chirality). Fig. 1b schematically depicts the shift in Fermi level induced by DDB-F72 doping, that produces the exciton bleach and progressively increases conductivity.³⁶ We attribute incomplete bleaching at high doping concentrations to steric constraints imposed by the large DDB-F72 molecules, given that the redox potential of the dopant has been measured previously^{34,35} (Fig. 1 and Fig. S1, ESI[†]) and it is expected to efficiently remove an electron from the s-SWCNTs. While the exciton bleach extent offers valuable insights, it cannot independently quantify carrier density. Complementary quantitative methods are thus essential for a more comprehensive understanding and validation of molecular doping mechanisms.^{37,38}

In this work, we quantified free charge carriers using solution-phase ¹⁹F NMR (Fig. 2a) since DDB-F72 contains 72

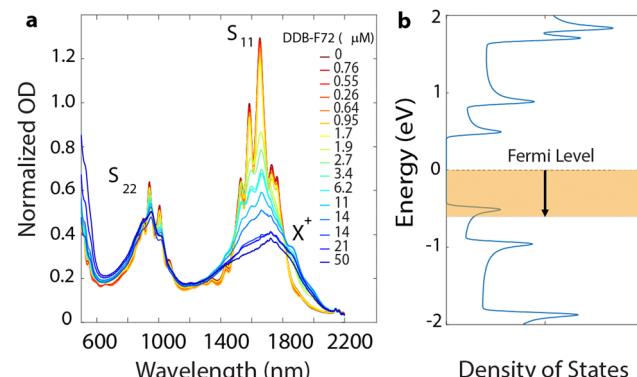
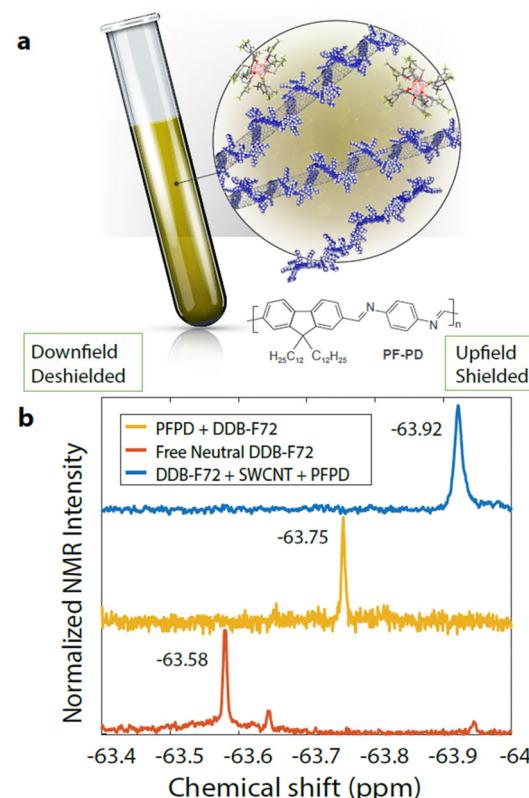


Fig. 1 (a) Absorption spectra of DDB-F72 doped s-SWCNT. Undoped sample (dark red, 0 μM DDB-F72) transitions to the highly doped sample (dark blue, 50 μM DDB-F72). Color code guide corresponds to dopant concentration for each trace. (b) Fermi energy shift because of molecular doping, modelling the phase-space filling effect. Calculated density of states (DOS) corresponds to (10,8) chirality.

fluorine atoms and the bound dopant counter anion has a unique ¹⁹F NMR chemical shift (*vide infra*). Fig. 2b shows ¹⁹F NMR spectra for control samples of neutral DDB-F72 in deuterated (d_8)-toluene,



as well as DDB-F72 mixed with the polymer³⁹ used to disperse the s-SWCNTs. The ¹⁹F chemical shift of DDB-F72 doped s-SWCNTs is shifted substantially up-field (higher ppm) from both control samples. The chemical shifts for each sample arise from the unique environments of atoms. The large up-field shift of the s-SWCNT sample, relative to the neutral molecule, is consistent with the electronegativity and π -donating ability introduced by the SWCNT, which serve to shield the magnetic field.⁴⁰

Additionally, the linewidth of the signal reports on the nature of the interaction of the dopant with other species in solution. Freely tumbling unbound molecules typically display narrow linewidths, while rotationally hindered bound molecules display broadened linewidths.⁴¹ From the chemical shifts and linewidths in Fig. 2b we can conclude that the signal for the s-SWCNT sample is dominated by bound dopant anions that have removed an electron from the occupied density of states of the s-SWCNTs. Control polymer doping experiments where the excitonic transition of the wrapping polymer was monitored and compared to the s-SWCNTs transition, Fig. S2a (ESI[†]), also support the claim that only dopant injecting charge onto the SWCNTs is quantified. The polymer excitonic transition shows no measurable bleaching, even when the S_{11} has been fully bleached (Fig. S2b, ESI[†]).

Fig. 3a displays ¹⁹F NMR spectra for the s-SWCNT doping series shown in Fig. 1. With increasing dopant concentration, the $[\text{DDB-F72}]^{-1}$ peak intensity increases, and the area under each curve provides the total number of fluorine atoms. Interestingly, increasing dopant concentration also induces a downfield chemical shift (Fig. 3b) that we assign to charge delocalization of the hole distribution in the s-SWCNTs. For higher doping concentrations, the hole distributions in s-SWCNTs for individual dopants start to merge and delocalize.^{32,34} This carrier density-dependent delocalization should reduce the coulomb attraction between hole and $[\text{DDB-F72}]^{-1}$ counterion, resulting in a larger distance between the SWCNT holes and their counterions, $[\text{DDB-F72}]^{-1}$, as the dopant

concentration increases (Fig. 3a). Additional interactions between dopants are also expected at very high dopant concentrations (above 40 μM) due to aggregation. Aggregation shows as additional NMR signal as seen for the polymer control NMR experiment (Fig. S6, ESI[†]) and as seen in the top two NMR spectra in Fig. 3a. The aggregation contribution is avoided by only integrating under the curve of the main peak.

A known NMR effect for π -conjugated systems is the ring current effect,^{42–44} where nuclei in proximity to (above or below) the plane of benzene ring(s) are shielded from the magnetic field and move up-field in chemical shift. This is an effect only measurable at adsorption distances, *i.e.*, bound dopants. The shielding is inversely proportional to the distance between the probing nucleus and benzene rings.⁴⁵ As the distance increases between these two species, the ring current-induced shielding decreases and the chemical shift moves downfield. Thus, the observation of a strong concentration-dependent ring current effect confirms the conclusion from our prior solid-state doping study that the s-SWCNT hole delocalization increases with increasing carrier density.³⁴ In contrast, the polymer doping series does not shift downfield with increased DDB-F72 concentration (Fig. 3b), suggesting weak physical interaction and ineffective doping, consistent with the expected band alignment³⁴ and absence of DDB-induced bleach of the polymer's exciton transition (Fig. S2, ESI[†]).

To calculate charge density, the area under each ¹⁹F NMR spectrum is integrated, with a known concentration standard serving as reference. The area under the curve is equal to the number of F atoms in the solution. Each DDB-F72 molecule contains 72 F atoms, thus the number of DDB-F72 molecules is the integrated area divided by 72. For the s-SWCNT samples, we do not observe other peaks that correspond to unbound DDB-F72 molecules (refer to Fig. S7, ESI[†]), *i.e.*, the calculated DDB-F72 number of molecules correspond only to bound dopants. Since each dopant contributes one hole, the number of holes (charges on the SWCNTs) is equal to the number of DDB-F72 anions calculated from NMR. The correlation between NMR-extracted hole density and exciton bleach is presented in Fig. 4, together with a fit modelled on the phase-space filling model introduced by Eckstein *et al.*³² Complete derivation and fitting is shown in the ESI.[†] The model uses the density of states from (10,8) SWCNTs as a representative s-SWCNT species found in high abundance from a detailed analysis of photoluminescence excitation (PLE) maps (Fig. S4, ESI[†]).

The best fit was obtained for an exciton size of 3 nm and hole effective mass of $0.06 m_e$, see Fig. S10 and S11 (ESI[†]). Good agreement is found between the experimental data and the model. DDB-F72 doping does not achieve 100% exciton bleaching due to steric hindrance of the bulky counterion, producing minor deviation at high doping levels. The exciton bleach magnitude was corrected by removing the contribution of the trion band, X^+ , to the total area under the absorption curve (refer to Fig. S12 and calculations shown in ESI[†]).³² Since PT SWCNTs yield several chiralities, we expect nanotubes with larger diameter/smaller band gaps to dope faster, also observed initially in Fig. 1a. As dopant concentration increases we expect

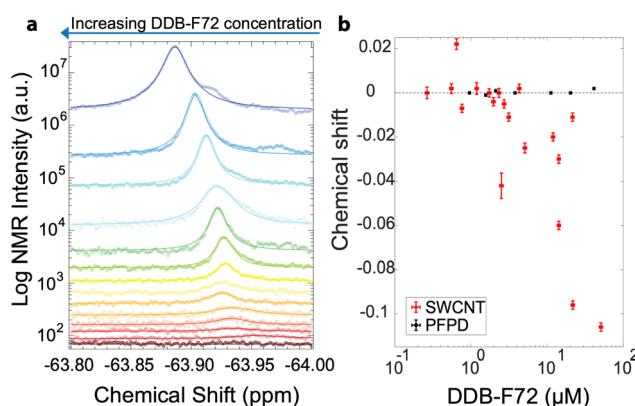


Fig. 3 (a) Spectra corresponding to ¹⁹F NMR for doping series of s-SWCNT. Spectra are arbitrarily displaced along the y axis to show the different doping steps clearly. Lower dopant concentration (red) to higher dopant concentrations (blue). (b) Change (Δ) in chemical shift, relative to most weakly doped samples, as a function of DDB-F72 concentration comparing doped s-SWCNT (red circles) and doped PF-PD polymer (black squares).



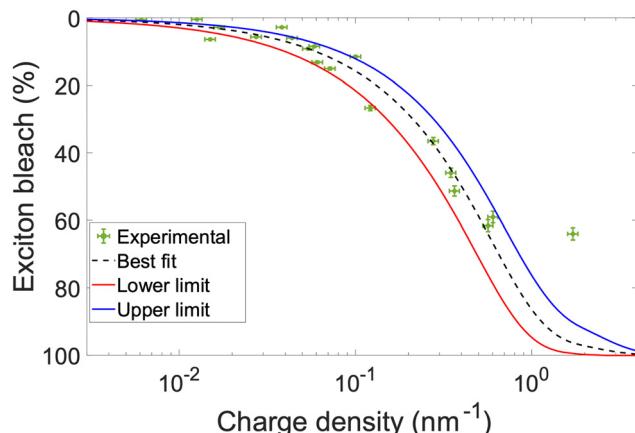


Fig. 4 Correlation between ^{19}F NMR-extracted charge carrier density versus exciton bleach obtained from absorption measurements (green data) and the homogeneous doping phase-space filling model fit: best fit (black dashed curve) obtained with 3 nm exciton size and hole effective mass of $0.06 m_e$, upper limit (blue curve) and lower limit (red curve) showing standard deviation.

all SWCNTs to dope effectively. The calculations in the ESI[†] took into account several chiralities and corresponding contribution to find the best fit to correctly calculate the length of nm for each nanotube chirality present.

From the model fit in Fig. 4, additional information can be obtained. If the exciton bleach is calculated from a simple absorbance measurement, charge carrier density can be obtained for similar diameter distributions. Since the exciton bleach induced by phase-space filling occurs independent of the local SWCNT environment, the model can be used to determine carrier density in both solution-phase and solid-state SWCNT samples. As such, the extracted carrier densities can be paired with concomitantly measured conductivity values to evaluate charge carrier mobility. We demonstrate this approach in Fig. 5 by extracting dopant concentration-

dependent charge carrier densities and mobilities from the exciton bleach and conductivity values reported in our recent doping study of plasma torch thin films.¹² Two different dopants are compared, F4TCNQ and DDB-F72.

Importantly, Fig. 5 demonstrates that hole mobility does not remain static, but instead steadily increases, as charge carrier density increases. This is an important finding, since charge carrier density has often been calculated from conductivity measurements based on static mobility values.^{17,46,47} The increase in mobility with carrier density may have several sources. Zaumseil *et al.*⁴⁸ have attributed a similar trend in s-SWCNT field-effect transistors to energetic disorder in the SWCNT network. The semi-localized transport (SLoT) model,⁴⁹ applied recently to s-SWCNT networks and a variety of semiconducting polymer films, suggests that charge carriers become more delocalized (*i.e.* mobility increases) as carrier density increases and the Coulomb potential wells of individual charge carriers overlap. Finally, recent calculations predict that increasing dopant concentration substantially increases the dielectric constant of the local environment in doped π -conjugated semiconductors, concomitantly decreasing the carrier-counterion binding energy.⁵⁰ With a means to now quantify carrier density and mobility in these doped s-SWCNT networks, we plan to explore these hypotheses in further studies.

Conclusions

We have demonstrated an NMR-based method to directly quantify charge carrier density in redox-doped s-SWCNTs, a method that can be extrapolated to other OSCs systems and dopants with NMR active nuclei. Concentration-dependent NMR shifts for the dopant counterions suggest that the SWCNT hole density becomes more delocalized, reducing the carrier-counterion Coulomb attraction, with increasing carrier density. The calculated mobility values show a charge carrier density dependence arising from enhanced delocalization, potentially in line with recent calculations predicting a doping-induced increase of local environment dielectric constant. Mobility is an important parameter for the performance of OSCs and nanoscale semiconductors in opto-electronic devices. The proper determination of this variable is crucial to understand charge transport leading to improvements in material design or device design.

Experimental section

PT SWCNT dispersion preparation

Polymer solution is first prepared with concentration of 2 mg mL^{-1} in deuterated toluene. Poly[(9,9-di-*n*-dodecyl-2,7-fluorendiyl-dimethine)-(1,4-phenylene-dinitrilomethine)] (PF-PD) was synthesized according to literature procedure.³⁹ Heated bath sonication of the polymer solution is necessary to ensure full solvation of the polymer. Plasma-torch (PT) SWCNTs (RN-020), were purchased from NanoIntegris, Quebec, Canada. The raw soot was first rinsed with toluene and centrifuged (2500 rpm) to remove soluble

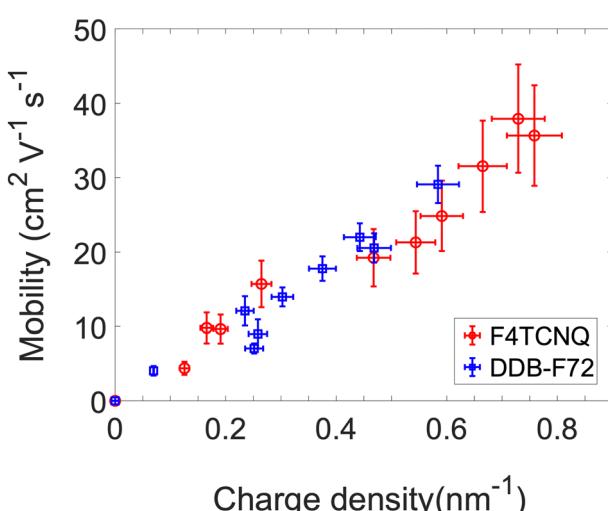


Fig. 5 Mobility calculated from conductivity measurements based on phase-space filling doping model. PT PF-PD doped with F4TCNQ (red circles) and DDB-F72 (blue squares).

fullerenes. The rinsed SWCNTs were left to dry overnight under low N₂ flow. The polymer solution is added to the SWCNTs at a concentration of 2 mg mL⁻¹ with maximum volume of *ca.* 15 mL in a 20 mL borosilicate scintillation vial. The mixture is then processed in the vial with a probe tip ultrasonic processor (Cole Parmer CPX-750, $\frac{1}{2}$ " tip) for 20 min at 40% amplitude. During tip sonication, the vial is submerged in a slurry of ethanol/dry ice mixture to increase the yield of dispersed SWCNTs. Immediately after tip sonication, the contents of the vial are transferred to a centrifuge tube and processed at 20 °C, 13 200 rpm, for 5 min (Beckman Coulter L-100 XP ultracentrifuge, SW-32 Ti rotor). The supernatant containing the dispersed SWCNTs, and excess polymer is collected *via* pipette. No polymer removal steps were performed to ensure SWCNT dispersion stability. The supernatant was concentrated *ca.* 5 times the original concentration to improve the signal-to-noise ratio (S/N) in the ¹⁹F NMR studies. PT SWCNTs contain several chiralities, the different chiralities and their distribution was determined based on PLE experiments and their fit, refer to Fig. S3–S5 and the ESI,† for a complete list and methodology.

Doping

Dodecaborane clusters substituted with fluorinated aryl groups were synthesized according to literature procedure.³⁵ The anhydrous deuterated toluene required additional molecular sieves treatment to ensure stability of the doping solution. A stock solution of 1 mg mL⁻¹ was prepared for the different doping steps. DDB-F72 molecule shown with van der Waals radii included in Fig. S1 (ESI,†) for reference.

Absorption measurements

Absorption spectra were measured for each SWCNT dispersion, both before and after achieving the desired doping level (Cary 5000 UV-Vis-NIR spectrophotometer, Agilent). Deuterated toluene was used for all measurements to prevent strong C–D stretch in the 2000–2200 nm region, and to prevent NMR line broadening by coupling to other active nuclei.

¹⁹F NMR measurements

Measurements were performed on a Bruker Ascend™ 300 spectrometer operating at 300 MHz (7 Tesla). In all experiments, the spin-lattice relaxation times were considered to measure NMR spectra under quantitative conditions. Deuterated toluene was used for all samples. The internal standard used was 1,1,1,3,3,3-hexafluoro-2-propanol.

Photoluminescence excitation (PLE) spectroscopy

PLE was measured using an in-house developed setup comprising a pulsed Xe-lamp for excitation (Edinburgh Instruments, custom adapted Xe900/XP920) equipped with a 300 mm focal length monochromator (Acton SpectraPro 2355). The excitation wavelength is varied in 5 nm steps from 600 nm to 1200 nm using either a 1200 g mm⁻¹ grating with a slit-selected average resolution of 6 nm (600–1140 nm) or a 600 g mm⁻¹ grating with a slit-selected average resolution of 12 nm (1140–1200 nm).

The PL is detected under a 90° angle using a 150 mm focal length spectrograph (Acton SpectraPro 2156) with a 150 g mm⁻¹ grating with a slit-selected average resolution of 16 nm, and with a liquid nitrogen cooled extended InGaAs photodiode array detector with a sensitivity up to 2.2 μm (Princeton Instruments OMA V:1024/LN-2.2). Samples were contained in a 60 μL quartz microcell with 3 mm path length. PLE spectra were corrected for variations in lamp excitation power, sensitivity of combined detector, gratings, and filters, and for reabsorption in the optical excitation and emission paths. To this end, samples were diluted in deuterated toluene to an absorbance of less than 0.3 (3 mm pathlength) over the full excitation and emission region to be in the linear correctable reabsorption regime.

Author contributions

J. Blackburn and A. Hermosilla-Palacios: conception of the experiments, methodology, data curation, visualization, and writing original draft. A. Hermosilla-Palacios: phase-space simulation calculations. M. Martinez: NMR experiments methodology. S. Cambré and W. Wenseleers: PLE experiments and PLE fitting. E. Doud and A. Spokoyny: DDB-F72 dopants synthesis. Y. Kim: density of states calculations. T. Hertel: supervised phase-space simulation calculations. A. Ferguson: absorption fitting and analysis.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was authored, in part, by the National Renewable Energy Laboratory, operated by the Alliance for Sustainable Energy, LLC, for the US Department of Energy (DOE) under contract no. DE-AC36-08GO28308. This study was primarily supported by the Solar Photochemistry Program, Division of Chemical Sciences, Geosciences, and Biosciences, Office of Basic Energy Sciences, US DOE. This funding supported work by M. A. H. P., M. M., A. J. F., and J. L. B. on SWCNT dispersion and doping development, absorption and NMR measurements, and experimental analysis and simulation. Y. H. K. was supported by the National Research Foundation of Korea (NRF-2019M3D1A1078302), for DFT calculations. A. M. S. thanks NIGMS (MIRA, R35GM124746) for funding the synthesis of the dopant molecules. S. C. and W. W. acknowledge funding from the Research Foundation – Flanders (FWO), through its project G036618N, for work on PLE spectroscopy and simulation. The views expressed in the article do not necessarily represent the views of the DOE or the US government.

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