PAPER

Check for updates

Terrylene on monolayer WS$_2$: coverage-dependent molecular re-orientation and interfacial electronic energy levels†

Qiang Wang,[a] Sifan You,[b] Björn Kobin,[c] Patrick Amsalem,[ac] Fengshuo Zu, [ID][ad] Rongbin Wang, [ID][a] Andreas Opitz, [ID][a] Stefan Hecht, [ID][c] Lifeng Chi [ID][b] and Norbert Koch [ID] *[ad]

The electronic, optical, and functional properties of van der Waals heterostructures comprising organic and two-dimensional inorganic semiconductors depend on the structure of the molecular assembly at and near the interface. Despite the rising interest in such heterostructures, very little is known about the, potentially complex, interplay between the structure and resulting (opto)electronic properties. Herein, we demonstrate with photoemission spectroscopy and scanning tunneling microscopy experiments a coverage dependence of the molecular assembly of terrylene deposited onto monolayer WS$_2$ (with sapphire serving as the substrate) and show how this impacts interfacial electronic properties. Up to monolayer coverage, terrylene molecules adapt a flat-lying orientation, which changes to an inclined orientation for higher coverages. This re-orientation is accompanied with a reduction in terrylene ionization energy by over 400 meV and an accordingly larger energy level offset of frontier energy levels of the two semiconductors and shift of the highest occupied molecular orbital energy level away from the WS$_2$ valence band. This can, for instance, reduce the charge-separation efficiency of the heterostructure with molecular multilayer coverage compared to that with only monolayer coverage. Furthermore, the modification of monolayer WS$_2$ excitonic features through molecular film deposition was evaluated using optical spectroscopy, yielding effective dielectric constants for a series of Rydberg excitons and exciton binding energies for bare and terrylene-covered monolayer WS$_2$ supported by sapphire. Altogether, these findings allow a comprehensive and detailed understanding of the (opto)electronic properties of this prototypical van der Waals heterostructure.

Introduction

van der Waals (vdW) heterostructures comprising π-conjugated organic molecules and two-dimensional (2D) transition metal dichalcogenides (TMDCs) have recently gained increasing attention, notably in view of their applications in advanced electronics and optoelectronics.[1,2] Layered materials, such as 2D-TMDCs, feature dangling-bond-free surfaces, which support

the formation of clean interfaces with various other functional materials *via* weak vdW forces. The formation of vdW heterostructures from organic and 2D inorganic semiconductors may preserve the intrinsic properties of individual components but may also facilitate the emergence of distinct novel interfacial properties and functionalities,[3] resulting from ground-state or excited-state charge transfer. In general, interface electronic properties are strongly influenced by the structural properties of adsorbed molecular layers. Upon adsorption, many molecular materials tend to form well-ordered 2D structures, whose details depend on the fine balance between intermolecular and molecule–substrate interactions.[4] For instance, on metals or graphene as a substrate, it has been documented that organic molecules initially form wetting layers, which are characterized by either a flat-lying or edge-on configuration.[5] With increasing coverage, molecules reorient into structures that often resemble their bulk form. It should be noted that the ionization energy (IE) and electron affinity (EA) of ordered molecular layers can vary by several 100 meV depending on their molecular orientation with respect to a surface or interface owing to the specific

[a] *Institut für Physik & Center for the Science of Materials Berlin (CSMB), Humboldt-Universität zu Berlin, Berlin 12489, Germany. E-mail: norbert.koch@physik.hu-berlin.de*

[b] *Institute of Functional Nano & Soft Materials (FUNSOM), Jiangsu Key Laboratory for Carbon Based Functional Materials & Devices, Soochow University, Suzhou 215123, P. R. China*

[c] *Institut für Chemie & Center for the Science of Materials Berlin (CSMB), Humboldt-Universität zu Berlin, Berlin 12489, Germany*

[d] *Helmholtz-Zentrum Berlin für Materialien und Energie GmbH, Berlin 12489, Germany*

electrostatic potential landscape created by the molecular arrangement.[6–8] Molecular orientation thus plays a crucial role in defining the optoelectronic functionality of heterostructures. Therefore, in order to develop an improved overall-picture of heterostructures formed between organic and 2D semiconductors, a detailed understanding of possible molecular orientation changes as a function of coverage, and how this impacts other properties, is highly desirable. To date, studies on organic molecule/TMDC-monolayer vdW heterostructures have predominantly focused on employing multilayer organic films,[4,9–11] and only a few reports have attended to the structural and electronic properties of molecular monolayers and few-layers at the interface.[12–14]

2D materials introduce another level of complexity in semiconductor heterostructures compared to their three-dimensional counterparts, specifically pertaining to the energy levels. Even when an organic/TMDC interface formation does not involve covalent interactions or ground-state charge transfer, the energy level alignment at the interface may not only result from the difference in the frontier energy levels of the pristine individual components, because the TMDC's energy levels may be subject to renormalization due to the presence of additional dielectric screening by the organic overlayer.[15] This would implicate deviations compared to the energy level alignment assumed from the Schottky–Mott limit, which is the usual approach employed in the literature. This effect of molecular adsorption on the electronic and optical/excitonic properties of the TMDCs has so far received little attention.[16]

π-conjugated molecules, in general, offer opportunities for engineering the properties of interfaces, such as facilitating energy or charge transfer, or minimizing energy barriers for charge injection. They are commonly employed to form ordered mono- or few-layers by self-assembly, as their extended π-conjugated system facilitates the formation of well-organized and stable molecular layers.[2,17] For these molecules, the atomically flat and dangling-bond-free surfaces of TMDCs can serve as an ideal template for the formation of ordered self-assembled layers.[18]

Here, we investigate self-assembled mono- and few-layers of the π-conjugated molecule terrylene ($C_{30}H_{16}$) on monolayer (ML) WS$_2$ supported by different substrates, namely sapphire, highly oriented pyrolytic graphite (HOPG), and Au(111). The choice of these substrates was motivated, on the one hand, by the desire to use an insulating substrate that does not couple strongly to the semiconductor, and, on the other hand, to allow contrasting the results with those of conductive substrates and to facilitate performing scanning tunneling microscopy (STM) experiments. Angle-resolved ultraviolet and X-ray photoelectron spectroscopy (ARUPS and XPS) measurements were carried out on the heterostructures to unravel the evolution of the electronic properties as a function of the molecular coverage. Complementarily, the orientation and packing of the molecules were unveiled by STM. We found that terrylene in the monolayer (1L) on ML-WS$_2$ was flat-lying, while in the multilayer it exhibited tilted orientations in a dimerized herringbone-like structure resembling that of the single crystal. This change in orientation was accompanied by a change in the IE and overall interfacial energy level alignment by >400 meV, in both situations forming a staggered type-II

semiconductor heterojunction that should enable interfacial charge transfer upon optical excitation. Additionally, the impact of molecular adsorption on the excitonic features of the ML-WS$_2$ was investigated by optical absorption spectroscopy. While no change in the A and B exciton energies was noticed upon molecular adsorption, the higher Rydberg states of the A exciton allowed determining an increased effective dielectric constant for the terrylene-covered ML-WS$_2$ compared to the bare ML-WS$_2$ on sapphire, thereby decreasing the ML-WS$_2$ quasiparticle band gap by *ca.* 60 meV. The detailed and comprehensive knowledge obtained in this study constitutes a benchmark for the terrylene/ML-WS$_2$ vdW heterostructure and provides useful insights for future studies of this system as well as of similar heterostructures.

Results and discussion

Electronic properties

To explore the electronic properties of the terrylene/ML-WS$_2$ interface, sapphire as an isolating substrate was chosen to avoid direct coupling between ML-WS$_2$ and the conducting substrate and possible charge transfer between the substrate and the molecular overlayer.[3,19,20] The ML-WS$_2$/sapphire sample was annealed *in situ* after being introduced into ultrahigh vacuum to obtain a clean surface for the ARUPS measurements.[21] As shown in Fig. S1 of the ESI,† ARUPS revealed the electronic band dispersions of ML-WS$_2$/sapphire, with the valence band maximum located at the K point of the Brillouin zone (BZ), substantiating the ML nature of the employed WS$_2$. The observation of band dispersion from an azimuthally disordered polycrystalline ML-WS$_2$ (as used here) was enabled by the dominant photoelectron intensity coming from the two high-symmetry directions (along Γ–K and Γ–M), due to the dihedral group symmetry of the 2H-phase WS$_2$.[3,22] The annealed ML-WS$_2$/sapphire, as shown in Fig. 1a, exhibited a work function (Φ) of 4.20 eV, as determined from the secondary electron cutoff (SECO) region, and the valence band maximum (VBM) was found to be at 1.71 eV binding energy (BE) below the Fermi level (E_F) at the K point (Fig. 1b). The latter observation indicated the pronounced n-type character of the ML-WS$_2$/sapphire. The "native" n-doping resulted in free charge carriers partially filling the conduction band.[3,21,22] In the present case, a clear density of states at E_F (at the K point) was observed, as shown in Fig. 1d. We remark that the role of sulfur vacancies to explain the n-type behavior of WS$_2$ or MoS$_2$ is still debated, because theoretically sulfur vacancies should not behave as a donor state,[23] as confirmed by experiments using HOPG or graphene as a supporting substrate.[14,24] However, when using sapphire as a substrate, experimental work using ARUPS and scanning tunneling spectroscopy has reported a partial filling of the MoS$_2$ conduction band together with the formation of sulfur vacancies.[21,25]

Next, terrylene was incrementally deposited onto the ML-WS$_2$/sapphire and ARUPS measurements were performed at the Γ and K points after each step. For very low coverages of 1–2 Å, as shown in Fig. 1b, an additional state, corresponding to the highest occupied molecular orbital (HOMO) level, emerged at
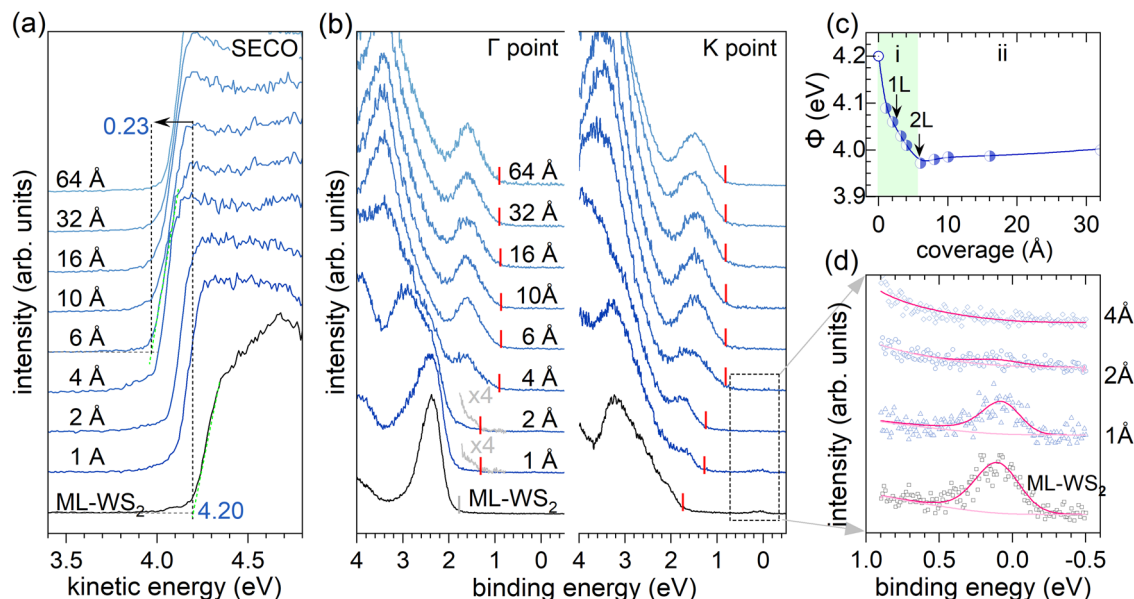
Fig. 1 Nominal coverage-dependent ARUPS spectra of terrylene/ML-WS$_2$ on sapphire showing (a) the secondary electron cutoff (SECO) and (b) valence electron region at the Γ and K points of the ML-WS$_2$ Brillouin zone. The Fermi level was at zero binding energy. At the K point, the terrylene HOMO level was clearly visible already from 1 Å coverage. (c) Plot of the work function (Φ) as a function of terrylene coverage. The two regimes (i) and (ii) discussed in the text are indicated. (d) Zoomed-in valence region near the Fermi level.

ca. 1.5 eV at the K point (*ca.* 40° emission angle) with its emission onset at 1.28 eV; this state was noticeable also at the Γ point (normal emission), but with a much lower intensity. Increasing the coverage beyond 2 Å led to a more substantial contribution of the molecules' features to the valence region signal at both emission angles. Notably, the HOMO level of terrylene became more prominent and its onset shifted to a significantly lower BE, settling at 0.80 eV BE. The first molecular layer (1L) completion was indicated by the attenuation of the partially filled WS$_2$ conduction band signal at E_F, which was completely suppressed at a nominal 4 Å coverage.

For terrylene in the low coverage regime ($<$4 Å), the VB spectra displayed substantial differences in spectral shape and intensity at the Γ and K points, indicating a pronounced angular dependence of the photoelectron emission. Notably, the valence spectra of 2 Å terrylene/ML-WS$_2$ at the Γ point resembled the momentum-integrated spectrum of bare ML-WS$_2$. This likely originated from Umklapp processes, in which photoelectrons emitted from the ML-WS$_2$ undergo elastic scattering by the molecular superlattice on top.[9,26,27] This is supported by Fig. 2b, where a momentum-integrated ML-WS$_2$ spectrum and terrylene HOMO spectrum (obtained from the
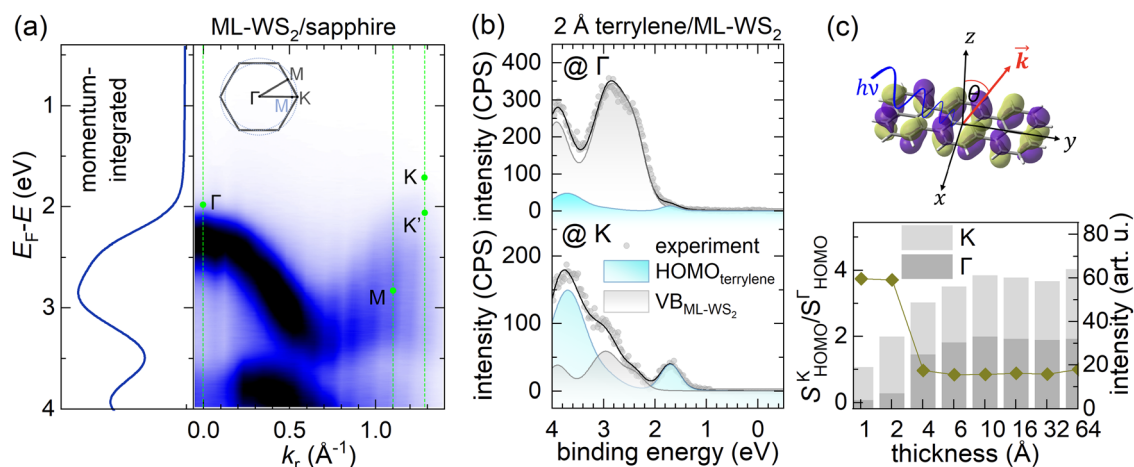

Fig. 2 (a) ARUPS map of ML-WS$_2$ and EDC curve of momentum-integrated spectra over the Γ−M−K direction. (b) Fitting of the VB spectra (Shirley background subtracted) of the 2 Å terrylene/ML-WS$_2$ sample at the Γ and K points. The terrylene HOMO component was taken from the fitting of the 2.5 Å terrylene/Au(111) spectrum. (c) Relative intensity of the HOMO peak (1.1–2.1 eV) area. The ratio of the fitting area of emissions at Γ and K points is defined as $S^{\Gamma}_{HOMO}/S^{K}_{HOMO}$.

fitting of terrylene on Au) were used to satisfactorily reconstruct the 2 Å terrylene/ML-WS$_2$ spectra at Γ and K. At the K point, the spectral weight of the terrylene HOMO was about three times larger compared to that at the Γ point, as shown in Fig. 2c. Such a behavior has previously been observed for several π-conjugated molecules with a flat-lying orientation, and explained by the fact that the photoelectrons are emitted at angles associated with the Fourier transform of the real-space distribution of the probed molecular orbital.[28,29] The maximum in emission from delocalized π-orbitals is then generally close to 40° for flat-lying molecules.[30,31] This implies, therefore, the formation of a first (well-ordered) wetting layer of terrylene with a probably flat-lying orientation. The lack of significant angle-dependence in the HOMO intensity for higher coverage (> 4 Å), as evidenced by the convergence of the HOMO intensity ratio $S_{HOMO}^{\Gamma}/S_{HOMO}^{K}$ to 1 as shown in Fig. 2c, indicated a change in the molecular orientation beyond the terrylene wetting layer (1L). We also note that the density of states from the presumably flat-lying molecules of the 1L with its peak maximum at *ca.* 1.8 eV remained discernable as a high BE shoulder of the main HOMO feature in the 4 Å coverage spectrum, while it was totally absent in the 6 Å spectrum (Fig. S2, ESI†). This suggests that the wetting itself may undergo a reorientation upon increasing coverage, instead of just being buried below differently oriented subsequent layers; this presumption is consistent with the findings from the STM measurements described further below.

The SECO spectra of terrylene with increasing coverage are shown in Fig. 1a. The work function Φ change is depicted in Fig. 1c and can be divided into two regimes: (i) for coverages up to 4 Å, where a sharp Φ decrease from 4.20 eV to 3.97 eV (*i.e.*, by 0.23 eV) occurs; (ii) with further increasing the coverage, Φ first slightly increases then settles at a constant value of about 4.02 eV. The initial Φ decrease in regime (i) apparently occurs upon completion of the terrylene 1L and is attributed to the Pauli pushback effect[32,33] of the ML-WS$_2$ with electron density spilling out of the layer, causing charge redistribution at the interface, which lowers the sample Φ. Beyond the 1L-regime of terrylene, surface de-wetting or structural defect formation[34] in the organic films might lead to the observed increase in Φ in regime (ii), as shown in Fig. 1c. However, this slight Φ increase was more likely due to a decrease in the pushback effect as the terrylene molecules change from a flat-lying to inclined orientation when multilayer formation becomes possible (*vide infra* for STM support of this presumption). The HOMO onset of the 1L terrylene was determined to be at 1.28 eV BE, giving an ionization energy (IE) of 5.30 eV for flat-lying molecules. Upon increasing the coverage beyond 6 Å, the HOMO onset shifted to 0.8 eV BE, and with the small Φ change, an IE of 4.77 eV resulted for the multilayer terrylene coverage. As explained previously, such a trend in IE variation upon the molecular orientation change was expected.[6-8]

Next, we examined the core levels by XPS in order to gain further insights into the molecule–substrate interaction, the terrylene growth mode, and the anticipated changes in molecular orientation. Fig. 3a shows the relative XPS intensities of W 4f$_{7/2}$ (ML-WS$_2$) and C 1s (terrylene) against the nominal film
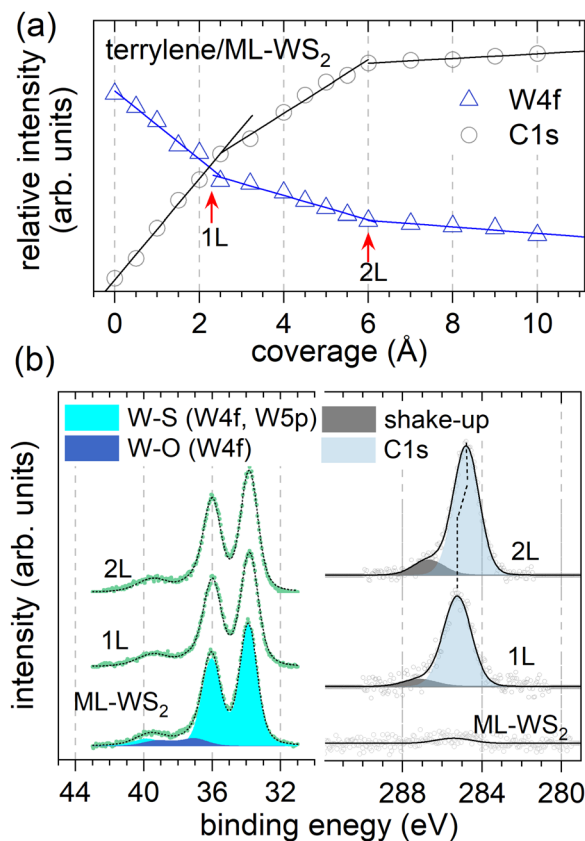

Fig. 3 (a) Variation of the relative XPS intensities of W 4f and C 1s of terrylene/ML-WS$_2$ as a function of coverage based on XPS data (Fig. S4, ESI†). For W 4f (C 1s), three linear decreasing (increasing) intensity regions could be fitted, with corresponding slope changes (red arrows). (b) W 4f and C 1s XPS spectra (Shirley background subtracted) of bare ML-WS$_2$ and 1L and 2L of terrylene/ML-WS$_2$.

coverage, revealing three distinct linear regions. The W 4f (C 1s) intensity decreased (increased) linearly up to 2.5 Å, from which point on the slope changed. Another change in slope was observed at 6 Å coverage. These characteristics suggest that the nominal thicknesses corresponding to 1L and a bilayer (2L) were 2.5 Å and 6 Å, respectively. Furthermore, the observed behavior is reminiscent of the characteristics of the Stranski–Krastanov growth mode, where the second layer starts growing only after the closure of the first layer, and weak islanding likely starts during the growth of the second layer.[35,36] As shown in Fig. 3b, the XPS W 4f spectrum of bare ML-WS$_2$ displayed the W 4f$_{7/2}$ (33.89 eV) and W 4f$_{5/2}$ (36.01 eV) doublet of W^{4+} due to the W–S bonds, together with faint contributions at 37.25 eV and 39.37 eV (3.8% of the spectral weight) corresponding to W^{6+} belonging to tungsten oxides. For the C 1s, two components were fitted, the main peak being attributed to the sp^2 carbons of the molecule and the higher BE shoulder at 1.95 eV away from the main peak to a shake-up process, likely corresponding to the HOMO–LUMO transition. Upon increasing the terrylene coverage, the C 1s peak for 2L (284.78 eV) shifted toward a lower BE by 0.47 eV with respect to that of 1L (285.25 eV) (Fig. S3, ESI†). The observed BE shift supports the molecular reorientation

suggested above, because the direction and magnitude of the C 1s peak are the same as the HOMO-onset change, underpinning the electrostatic origin of the shifts. The W 4f peaks exhibited a comparably small shift to lower BE by 90 meV upon terrylene deposition, in line with the improved screening of the final state by the adsorption of the molecular overlayer, in qualitative agreement with the change in dielectric screening found by optical spectroscopy in a related molecule/2D semiconductor system.[3]

Structural properties

To substantiate the molecular orientation changes inferred from the photoemission results, STM measurements were performed on ML-WS$_2$/HOPG (note: STM with sapphire as a support for the heterostructure was not feasible, probably due to the moderate conductivity of ML-WS$_2$ and the high current density in the STM tip vicinity), which showed a similar terrylene IE change as observed in photoemission when using the sapphire substrate as the support (Fig. S5, ESI†).

Prior to terrylene deposition, a well-defined and atomically clean ML-WS$_2$ surface was observed, as shown in Fig. 4a. The unit cell parameters were measured to be $a = (0.315 \pm 0.006)$ nm,

and $\theta = 59.5 \pm 2°$, which were consistent with the 2H phase of WS$_2$.[37] As illustrated in Fig. 4b and c, the terrylene molecules were arranged into distinct layered configurations, with the first one (1L) displaying an oblique lattice, and the second (2L) and third layers (3L) featuring hexagonal arrangements. The unit cell parameters of the 1L [$a = (1.10 \pm 0.04)$ nm, $b = (1.84 \pm 0.06)$ nm and $\theta = (60.0 \pm 2.5)°$] support the flat-lying orientation of closely packed terrylene molecules, thereby maximizing the substrate area coverage with the molecular area, a feature also observed on Au(111) (Fig. S6, ESI†). As shown in Fig. 4e, an inclined and seemingly dimerized molecular orientation, resembling the terrylene single crystal configuration (Fig. S6d, ESI†),[38] was suggested by STM for higher coverage (2L and 3L), with unit cell parameters of $a = (1.93 \pm 0.04)$ nm, $b = (1.84 \pm 0.05)$ nm, and $\theta = (59.0 \pm 3.0)°$. Such a transition from a flat-lying to inclined orientation was further supported by the relative height and corrugation of each layer, as depicted in Fig. 4g. Notably, the apparent heights with respect to ML-WS$_2$ were *ca.* (3.0 ± 0.1) Å for 1L, (9.4 ± 0.1) Å for 2L, and (19.0 ± 0.2) Å for 3L, accompanied by an increase in the in-plane corrugation [0.08 Å for 1L, and 0.23 Å for 3L]. We note that the apparent height of a terrylene layer lying
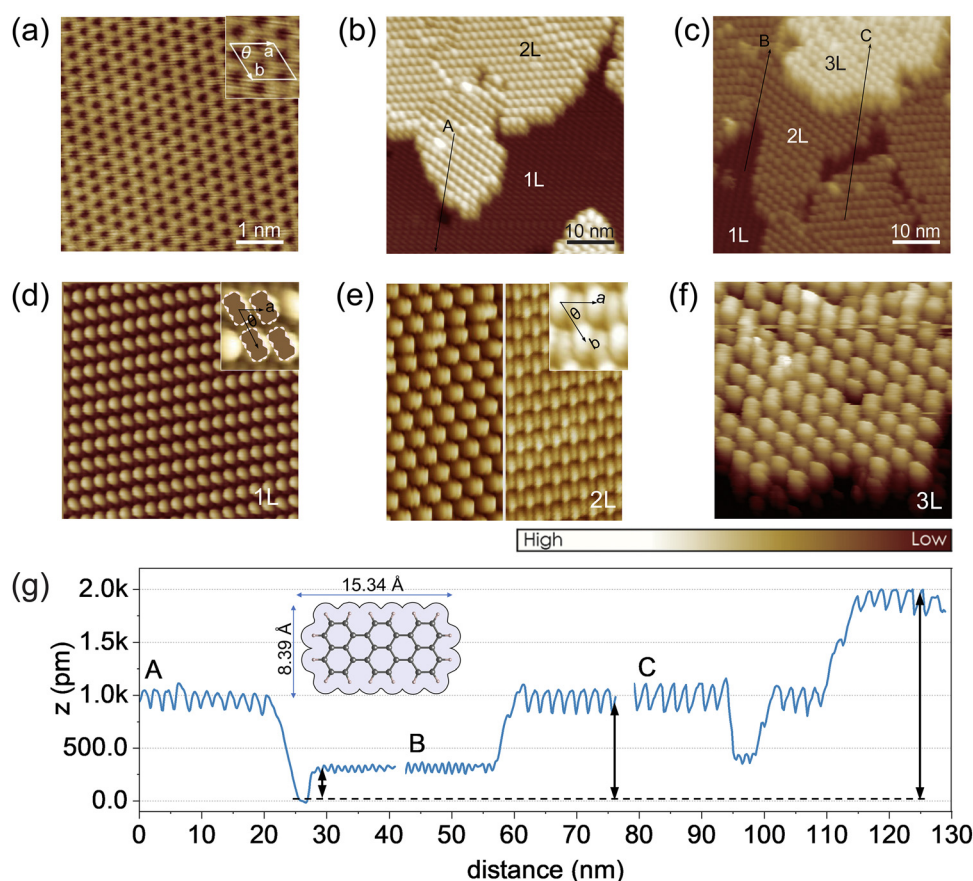

Fig. 4 STM images of the orientation change in terrylene multilayer on ML-WS$_2$/HOPG. (a) Clean surface of ML-WS$_2$ before terrylene deposition. (b) Bare ML-WS$_2$ surface (black area) next to molecular layers. (c) Multilayer terrylene on ML-WS$_2$/HOPG, with a flat-lying orientation in 1L (first layer), and inclined dimerized structures in 2L (second layer) and 3L (third layer). High resolution zoomed-in images of 1L (d), 2L (e), and 3L (f) with the corresponding unit cell marked. Two different configurations are visualized as two distinct symmetries in the STM images. Each figure is 20 nm × 20 nm. (g) Line profile in (b) and (c), marked as A, B, and C. Lower relative height and small corrugation in 1L transform to a large step height and corrugation in 3L. Set point: (a) 500 mV, 20 pA, (b), (c), and (d)–(f) 3 V, 10 pA.

flat on top of a first flat-lying monolayer was about 3.0 Å, as determined on Au(111) (see Fig. S6, ESI†). Notably, on ML-WS$_2$/HOPG, the height of the 3L corresponded to twice the height of the 2L, indicating that 3L likely corresponded to two stacked 2Ls, while the height of the 2L was about three times that of the 1L. This therefore suggests that the flat-lying molecules of the 1L undergo reorientation upon multilayer formation, transitioning to form part of the 2L with inclined molecular orientation. The STM analysis thus unambiguously revealed the change in the molecular orientation and packing from 1L to 2L. Specifically, the results confirmed a flat-lying orientation for 1L, in contrast to the inclined configuration observed for 2L and beyond.

To better visualize the molecular arrangement on ML-WS$_2$, a schematic representation of the molecular configurations, as derived from the STM data, is depicted in Fig. 5. Upon superimposing the unit cells across the interface between 1L and 2L (Fig. 5a), it was evident that 2L follows the orientation of the lattice of 1L, albeit with a different periodicity. Specifically, the volume ratio between unit cells of 1L : 2L was 4 : 7, where the dimers in 2L adopt a herringbone packing motif.[39] The lattice vectors of the terrylene 1Ls were oriented along the [120] and [1$\bar{1}$0] directions of the ML-WS$_2$ lattice and formed what resembles a $(2\sqrt{3} \times 7\sqrt{3})R30°$ superlattice (Fig. 5b), otherwise described by the superstructure matrix $G = \begin{pmatrix} 2 & 4 \\ -7 & 7 \end{pmatrix}$. For the 2L terrylene, the assembly continued along the [120] direction and changed into a dimerized structure, now resembling the single crystal structure (Fig. S6d, ESI†), which rearranged into a $(7\sqrt{3} \times 7\sqrt{3})R30°$ overlayer with $G = 7\begin{pmatrix} 1 & 2 \\ -1 & 1 \end{pmatrix}$. Based on the structural model, the calculated molecular densities of 2L to 1L was 2.3, which corresponds to the thickness (molecular weight) ratio of 2.4 obtained from the PES data (Note 2, ESI†).

Optical properties

UV-vis spectroscopy was performed to evaluate the impact of the molecular adlayer atop the ML-WS$_2$ TMDC on its excitonic properties. Detailed analysis of the spectra shown in Fig. 6 revealed that the characteristic A exciton peak maxima of ML-WS$_2$ without and with terrylene differed by only ca. 2 meV.[40] The spectral features of Rydberg states (n = 1s–5s) below the ionization threshold were enhanced by taking the secondary derivatives of the UV-vis spectra, as shown in Fig. 6. The Rydberg series can be approximated as s-type hydrogenic excitons,[41,42] following a characteristic sequence of $1/(n - 1/2)^2$. The effective dielectric constant experienced by the excitons, k_{eff}^n, with the overlayer for the nth exciton, can be calculated as:

$$k_{\text{eff}}^n = [\mu e^4/8h^2\varepsilon_0{}^2 E_{b,\text{exp}}^n(n - 1/2)^2]^{1/2} \qquad (1)$$

A reduced exciton mass μ = 0.16 m_0 for ML-WS$_2$ was adopted.[42,43] The exciton BE ($E_{b,\text{exp}}^n$) was experimentally determined by the difference between E_g and E_{ex}. In Fig. 6c, the quasiparticle E_g values were obtained by fitting the Rydberg states (n = 3–5) with the hydrogenic model, resulting in values of (2.26 ± 0.01) eV and (2.20 ± 0.01) eV for the bare and terrylene-covered ML-WS$_2$, respectively. Correspondingly, the ML-WS$_2$ A-exciton BE ($E_{b,\text{exp}}^{n=1}$) was found to decrease from 234 meV to 173 meV after terrylene deposition. As shown in Fig. 6d, the larger exciton size for the higher n resulted in enhanced dielectric screening from the surrounding material, causing the k_{eff} to decrease and eventually stabilize. Notably, the anisotropic screening of the tightly bound 1s exciton was mainly due to WS$_2$, while the higher-order (3s–5s) e–h pairs with larger radii were primarily screened by the surrounding materials as Wannier-type excitons, where k_{eff} is equal to $(\varepsilon_r^{\text{top}} + \varepsilon_r^{\text{bottom}})/2$. Using $\varepsilon_r^{\text{sapphire}}$ = 3.18, the k_{eff} values for the A-exciton (n = 1) and higher n were approximately 6.10 and 2.09, respectively. With a terrylene film (64 nm, approx. 17-MLs), $k_{\text{eff}}^{n>3}$ for higher n was fitted to 3.02.
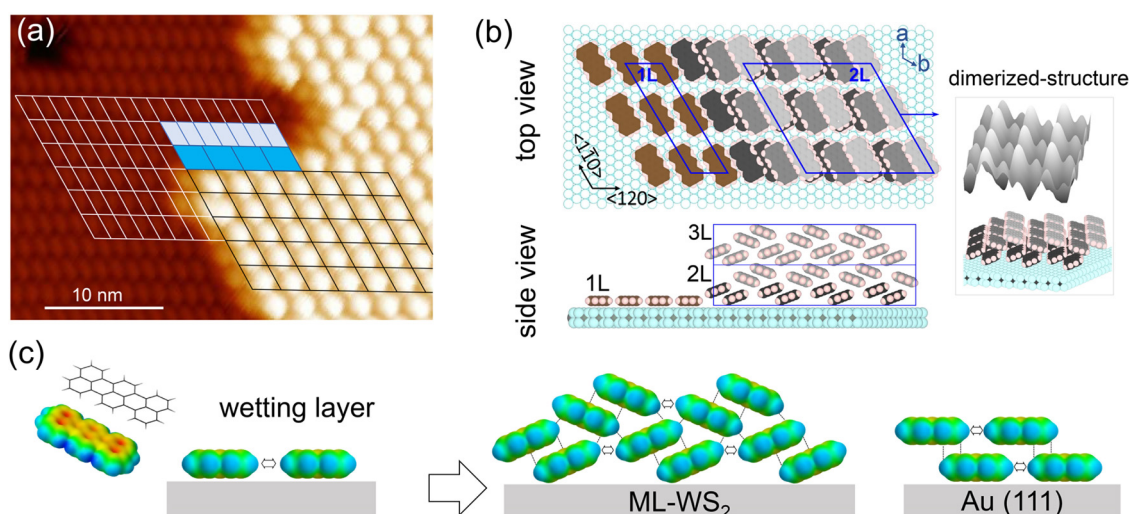

Fig. 5 (a) High-resolution STM images for 1L and 2L. The arrangement of self-assembled terrylene in 1L and 2L are marked as white and black lines, respectively. Set point: 3 V, 20 pA. (b) Top view and side view along the [100] direction of a reconstruction model based on STM results, with unit cells of $(2\sqrt{3} \times 7\sqrt{3})$ adlayer of 1L and $(7\sqrt{3} \times 7\sqrt{3})$ overlayer of 2L on ML-WS$_2$. The unit cell dimensions of ML-WS$_2$ are $a = b = 0.316$, $\theta = 120°$. (c) Formation of 2L after the completion of a wetting layer on ML-WS$_2$ and Au(111). Blue stands for positive charge and yellow stands for negative charge.
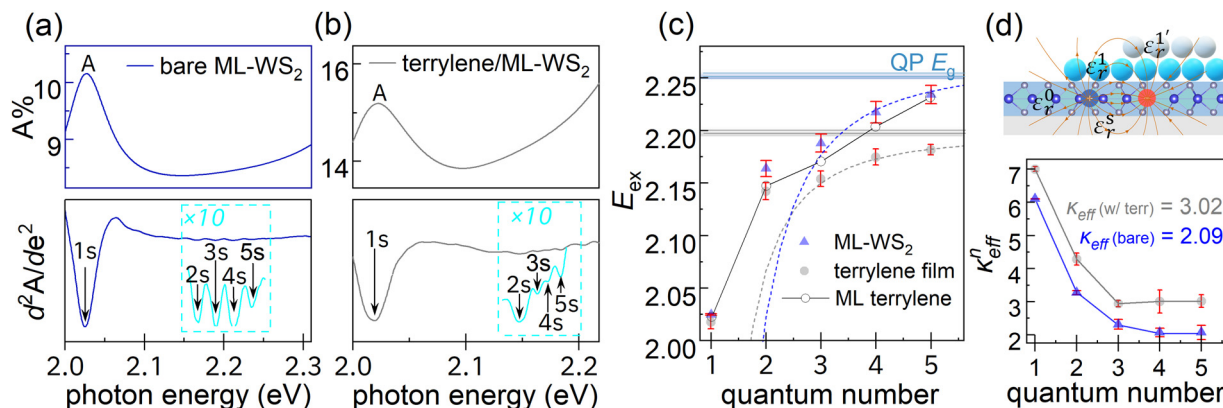
Fig. 6 UV-vis absorption spectra of (a) ML-WS$_2$/sapphire and (b) multilayer (64 Å) terrylene/ML-WS$_2$/sapphire, with the second-order numerical derivative drawn below. The Rydberg states are indicated as 1s to 5s. (c) Experimentally obtained transition energies for the exciton states as a function of the quantum number n. The error bars resulted from the fitting procedure. The absorption spectrum of sub-ML (2 Å) terrylene is shown in hollow circles. The corresponding 2D hydrogenic models before or after terrylene deposition are depicted in dashed lines. (d) Calculated effective dielectric constant as a function of n using the hydrogenic model.

Final heterostructure energy level diagram

We first remark that the ML-WS$_2$/sapphire single-particle energy gap (E_g) could not be determined by ARUPS, because of the partial CB occupation after annealing in vacuum, i.e., the E_g of ML-WS$_2$ without this strong native n-doping would be above 2 eV. The UV-vis analysis discussed above allowed estimating E_g at ca. 2.26 eV, by summing the optical gap and the exciton binding energy. This was likely because of air exposure of the UV-vis samples, which can significantly reduce the CB filling as oxygen inactivates a large part of the n-doping behavior in UHV. In Fig. 7, we summarize the energy level alignment diagram based on our ARUPS data including the occupied CB states, and use the E_g values obtained from the optical analysis to represent the unoccupied states. The optical
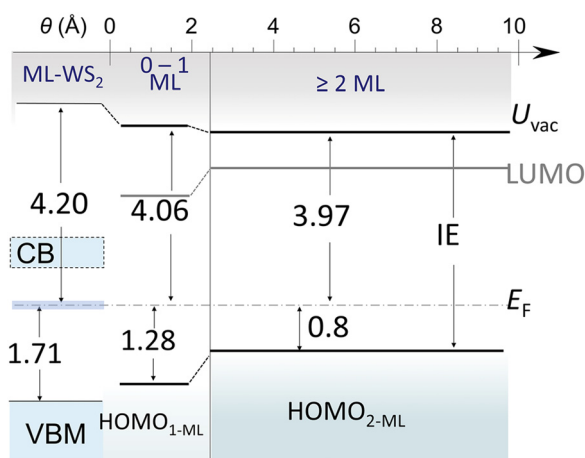
bandgap of terrylene films was determined to be 1.95 eV on sapphire, as obtained from a Tauc plot (Fig. S7b, ESI†). Accounting for a representative value of the exciton binding energy of conjugated molecules of 0.55 eV,[40,44,45] we obtained a terrylene electronic E_g of 2.5 eV. The resulting energy level alignment shows that a type-II heterojunction was formed both for the mono- and multilayer terrylene films, for which the HOMO lay at ca. 1.28 eV and 0.80 eV below E_F, respectively, due to the transition from a flat-lying to an inclined molecular orientation when entering the multilayer regime, and including a reorientation of the first wetting layer, as depicted in Fig. 5b. In addition, upon increasing the terrylene thickness, dielectric screening intensified and reduced the E_g of ML-WS$_2$ by about 0.1 eV, a value which we expect to be a reasonable guidance to estimate bandgap renormalization due to dielectric screening by organic semiconductor overlayers. This value was also in line with the 90 meV shift of the W 4f core levels observed upon the deposition of terrylene.

Conclusion

In conclusion, we examined the molecular layer thickness-dependent structural and optoelectronic properties of the terrylene/ML-WS$_2$ interface supported on sapphire, a prototypical insulator, and found a type-II energy level alignment at this interface. Thickness-dependent ARUPS and STM measurements allowed for correlating the observed change in molecular ionization energy with the molecular orientation: the reorientation from flat-lying adsorbed terrylene in the monolayer to the edge-on orientation in the multilayer film was accompanied by a reduction in terrylene IE by 0.48 eV. Furthermore, our results also suggest that the monolayer molecules change their orientation upon multilayer formation in order to form the bulk-like dimerized structure, which was possibly due to the weak molecule–substrate interaction. Additionally, we quantified the change in the TMDC quasiparticle bandgap and exciton

Fig. 7 Schematic energy level alignment in terrylene/ML-WS$_2$ heterostructures. The energy difference between energy levels and E_F is shown by arrows and numbers (unit, eV). The partially filled CB in ML-WS$_2$ is indicated by blue shading. Note that the "0–1 ML" region was no longer present as soon as the 2 ML coverage region has formed because the 1L flat-lying wetting layer of terrylene was consumed to form the 2L inclined layer (see Fig. 5 and associated text).

binding energy resulting from the increased dielectric screening promoted by the molecular adlayer. The higher-order Rydberg states were found to be strongly affected due to the large e–h radius, while the 1s exciton remained largely unchanged due to its small spatial extend. These results demonstrate the importance of coupling the structural and optoelectronic characterization of these hybrid interfaces in order to extract the relevant energy level offset occurring upon the formation of device-relevant organic/TMDC hybrid heterostructures.

Experimental section

Preparation of the ML-WS$_2$ samples

The ML-WS$_2$ films were synthesized by chemical vapor deposition using WO_3, Na_2WO_4 and sulfur as precursors on a sapphire substrate. The as-grown ML-WS$_2$ was transferred to a clean sapphire substrate and freshly cleaved HOPG substrate. The detailed synthesis and transfer for ML-WS$_2$ is described in the ESI† (note). The after-transfer ML-WS$_2$/sapphire samples were fixed to a sample holder and electrically connected to ground with screws fixing the film edges. The ML-WS$_2$/sapphire or WS$_2$/HOPG samples were loaded into an ultrahigh vacuum chamber (base pressure of 1×10^{-9} mbar) and annealed for 10 h at 350 °C to remove transfer-induced contamination.

Terrylene synthesis

The synthesis of terrylene was carried out following a route published previously.[46] Final purification was carried out by multiple subsequent steps of crystallization from 1,2,4-trichlorobenzene and sublimation to achieve a purity of 99.5%.

Photoemission measurements

The samples were annealed at high temperature (500–600 °C) for 1 h in repeated cycles to create enough sulfur vacancies, prior to transferring to the analyzing chamber equipped with a hemispherical electron analyzer (SPECS Phoibos 100). The photoemission spectra were collected using a monochromatic He Iα source (HIS-13 lamp mounted on a VUV5046 UV monochromator) and Al Kα (1486.6 eV) as the excitation source for UPS and XPS analyses, respectively, ensuring no sample charging was observed. The overall energy resolution was determined by a polycrystalline gold sample, to be 0.07 eV and 0.8 eV for UPS and XPS, respectively. For the ARPES measurements, the spectra were collected by rotating the manipulator at an angular step of 2°.

STM measurements

All the samples were prepared with a commercial low-temperature scanning tunneling microscopy system (Unisoku LT-STM 1500s) with a base pressure better than 1×10^{-10} torr. The WS$_2$/HOPG sample was annealed to 823 K for 2 h before the deposition of terrylene molecules. Terrylene molecules were evaporated at 543 K for 5 min on ML-WS$_2$/HOPG held at room temperature using a commercial Knudsen cell (Kentax, Germany). The STM tips were cleaned by e-beam heating under UHV. All the images were acquired at 77 K in the constant current mode. The STM images were processed using the WSxM software.[47]

Optical measurements

Optical absorption spectra were obtained for the large area (>92% coverage) ML-WS$_2$ films on sapphire using a Lambda 950 UV/Vis/NIR spectrophotometer (Perkin Elmer Inc.) under ambient conditions.

Data availability

Data for this article are available at the Open-Access-Publikationsserver of Humboldt-Universität at https://doi.org/10.18452/28966.

Conflicts of interest

The authors declare no conflict of interest.

Acknowledgements

References

1 K. S. Novoselov, D. Jiang, F. Schedin, T. J. Booth, V. V. Khotkevich, S. V. Morozov and A. K. Geim, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 10451–10453.

2 D. Jariwala, T. J. Marks and M. C. Hersam, *Nat. Mater.*, 2017, **16**, 170–181.

3 J. Ma, P. Amsalem, T. Schultz, D. Shin, X. Xu and N. Koch, *Adv. Electron. Mater.*, 2021, **7**, 2100425.

4 M. Dreher, D. Günder, S. Zörb and G. Witte, *Chem. Mater.*, 2020, **32**, 9034–9043.

5 B. Nickel, M. Fiebig, S. Schiefer, M. Göllner, M. Huth, C. Erlen and P. Lugli, *Phys. Status Solidi A*, 2008, **205**, 526–533.

6 S. Duhm, G. Heimel, I. Salzmann, H. Glowatzki, R. L. Johnson, A. Vollmer, J. P. Rabe and N. Koch, *Nat. Mater.*, 2008, **7**, 326–332.

7 M. L. Tietze, W. Tress, S. Pfützner, C. Schünemann, L. Burtone, M. Riede, K. Leo, K. Vandewal, S. Olthof, P. Schulz and A. Kahn, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **88**, 085119.

8 M. Schwarze, K. S. Schellhammer, K. Ortstein, J. Benduhn, C. Gaul, A. Hinderhofer, L. Perdigon Toro, R. Scholz, J. Kublitski, S. Roland, M. Lau, C. Poelking, D. Andrienko,

G. Cuniberti, F. Schreiber, D. Neher, K. Vandewal, F. Ortmann and K. Leo, *Nat. Commun.*, 2019, **10**, 2466.

9 S. Park, N. Mutz, S. A. Kovalenko, T. Schultz, D. Shin, A. Aljarb, L. J. Li, V. Tung, P. Amsalem, E. J. W. List-Kratochvil, J. Stähler, X. Xu, S. Blumstengel and N. Koch, *Adv. Sci.*, 2021, **8**, 2100215.

10 H. G. Shin, D. Kang, Y. Jeong, K. Kim, Y. Cho, J. Park, S. Hong, Y. Yi and S. Im, *ACS Nano*, 2020, **14**, 15646–15653.

11 S. Park, T. Schultz, D. Shin, N. Mutz, A. Aljarb, H. S. Kang, C. H. Lee, L. J. Li, X. Xu, V. Tung, E. J. W. List-Kratochvil, S. Blumstengel, P. Amsalem and N. Koch, *ACS Nano*, 2021, **15**, 14794–14803.

12 Z. Song, T. Schultz, Z. Ding, B. Lei, C. Han, P. Amsalem, T. Lin, D. Chi, S. L. Wong, Y. J. Zheng, M. Y. Li, L. J. Li, W. Chen, N. Koch, Y. L. Huang and A. T. S. Wee, *ACS Nano*, 2017, **11**, 9128–9135.

13 S. Park, H. Wang, T. Schultz, D. Shin, R. Ovsyannikov, M. Zacharias, D. Maksimov, M. Meissner, Y. Hasegawa, T. Yamaguchi, S. Kera, A. Aljarb, M. Hakami, L. J. Li, V. Tung, P. Amsalem, M. Rossi and N. Koch, *Adv. Mater.*, 2021, **33**, 2008677.

14 G. Reecht, N. Krane, C. Lotze and K. J. Franke, *ACS Nano*, 2019, **13**, 7031–7035.

15 W.-T. Hsu, J. Quan, C.-Y. Wang, L.-S. Lu, M. Campbell, W.-H. Chang, L.-J. Li, X. Li and C.-K. Shih, *2D Mater.*, 2019, **6**, 025028.

16 P. A. Markeev, E. Najafidehaghani, G. F. Samu, K. Sarosi, S. B. Kalkan, Z. Gan, A. George, V. Reisner, K. Mogyorosi, V. Chikan, B. Nickel, A. Turchanin and M. P. de Jong, *ACS Nano*, 2022, **16**, 16668–16676.

17 W. H. Lee, J. Park, S. H. Sim, S. Lim, K. S. Kim, B. H. Hong and K. Cho, *J. Am. Chem. Soc.*, 2011, **133**, 4447–4454.

18 C.-H. Lee, T. Schiros, E. J. G. Santos, B. Kim, K. G. Yager, S. J. Kang, S. Lee, J. Yu, K. Watanabe, T. Taniguchi, J. Hone, E. Kaxiras, C. Nuckolls and P. Kim, *Adv. Mater.*, 2014, **26**, 2812–2817.

19 S. Park, N. Mutz, T. Schultz, S. Blumstengel, A. Han, A. Aljarb, L.-J. Li, E. J. W. List-Kratochvil, P. Amsalem and N. Koch, *2D Mater.*, 2018, **5**, 025003.

20 Y. Liu, J. Guo, E. Zhu, L. Liao, S. J. Lee, M. Ding, I. Shakir, V. Gambin, Y. Huang and X. Duan, *Nature*, 2018, **557**, 696–700.

21 F. Caruso, P. Amsalem, J. Ma, A. Aljarb, T. Schultz, M. Zacharias, V. Tung, N. Koch and C. Draxl, *Phys. Rev. B*, 2021, **103**, 205152.

22 S. Park, T. Schultz, A. Han, A. Aljarb, X. Xu, P. Beyer, A. Opitz, R. Ovsyannikov, L. J. Li, M. Meissner, T. Yamaguchi, S. Kera, P. Amsalem and N. Koch, *Commun. phys.*, 2019, **2**, 68.

23 J.-Y. Noh, H. Kim and Y.-S. Kim, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2014, **89**, 205417.

24 B. Schuler, D. Y. Qiu, S. Refaely-Abramson, C. Kastl, C. T. Chen, S. Barja, R. J. Koch, D. F. Ogletree, S. Aloni, A. M. Schwartzberg, J. B. Neaton, S. G. Louie and A. Weber-Bargioni, *Phys. Rev. Lett.*, 2019, **123**, 076801.

25 P. Vancso, G. Z. Magda, J. Peto, J. Y. Noh, Y. S. Kim, C. Hwang, L. P. Biro and L. Tapaszto, *Sci. Rep.*, 2016, **6**, 29726.

26 L. Giovanelli, P. Amsalem, T. Angot, L. Petaccia, S. Gorovikov, L. Porte, A. Goldoni and J. M. Themlin, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **82**, 125431.

27 M. Krivenkov, D. Marchenko, M. Sajedi, A. Fedorov, O. J. Clark, J. Sanchez-Barriga, E. D. L. Rienks, O. Rader and A. Varykhalov, *Nanoscale*, 2022, **14**, 9124–9133.

28 P. Puschnig, S. Berkebile, A. J. Fleming, G. Koller, K. Emtsev, T. Seyller, J. D. Riley, C. Ambrosch-Draxl, F. P. Netzer and M. G. Ramsey, *Science*, 2009, **326**, 702–706.

29 X. Yang, M. Jugovac, G. Zamborlini, V. Feyer, G. Koller, P. Puschnig, S. Soubatch, M. G. Ramsey and F. S. Tautz, *Nat. Commun.*, 2022, **13**, 5148.

30 O. Endo, F. Matsui, S. Kera, W.-J. Chun, M. Nakamura, K. Amemiya and H. Ozaki, *J. Phys. Chem. C*, 2022, **126**, 15971–15979.

31 Y. Nakayama, S. Kera and N. Ueno, *J. Mater. Chem. C*, 2020, **8**, 9090–9132.

32 P. S. Bagus, V. Staemmler and C. Woll, *Phys. Rev. Lett.*, 2002, **89**, 096104.

33 H. Ishii, K. Sugiyama, E. Ito and K. Seki, *Adv. Mater.*, 1999, **11**, 605–625.

34 T. Hosokai, H. Machida, A. Gerlach, S. Kera, F. Schreiber and N. Ueno, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 195310.

35 C. Argile and G. E. Rhead, *Surf. Sci. Rep.*, 1989, **10**, 277–356.

36 P. Schroeder, C. France, B. Parkinson and R. Schlaf, *J. Appl. Phys.*, 2002, **91**, 9095–9107.

37 H. M. Hill, A. F. Rigosi, K. T. Rim, G. W. Flynn and T. F. Heinz, *Nano Lett.*, 2016, **16**, 4831–4837.

38 C. L. Hall, I. Andrusenko, J. Potticary, S. Gao, X. Liu, W. Schmidt, N. Marom, E. Mugnaioli, M. Gemmi and S. R. Hall, *Chem. Phys. Chem.*, 2021, **22**, 1631–1637.

39 S. M. Ryno, C. Risko and J. L. Bredas, *J. Am. Chem. Soc.*, 2014, **136**, 6421–6427.

40 P. Navarro, F. C. Bocquet, I. Deperasińska, G. Pirug, F. S. Tautz and M. Orrit, *J. Phys. Chem. C*, 2014, **119**, 277–283.

41 T. Olsen, S. Latini, F. Rasmussen and K. S. Thygesen, *Phys. Rev. Lett.*, 2016, **116**, 056401.

42 A. Chernikov, T. C. Berkelbach, H. M. Hill, A. Rigosi, Y. Li, O. B. Aslan, D. R. Reichman, M. S. Hybertsen and T. F. Heinz, *Phys. Rev. Lett.*, 2014, **113**, 076802.

43 T. C. Berkelbach, M. S. Hybertsen and D. R. Reichman, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **88**, 045318.

44 Y. Tanaka, N. Fukui and H. Shinokubo, *Nat. Commun.*, 2020, **11**, 3873.

45 P. K. Nayak, *Synth. Met.*, 2013, **174**, 42–45.

46 Y. Avlasevich, C. Kohl and K. Müllen, *J. Mater. Chem.*, 2006, **16**, 1053–1057.

47 I. Horcas, R. Fernández, J. M. Gómez-Rodríguez, J. Colchero, J. Gómez-Herrero and A. M. Baro, *Rev. Sci. Instrum.*, 2007, **78**, 013705.