## **Green Chemistry**



## CORRECTION

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## Correction: Ethanol-assisted mechanochemical asymmetric cross-dehydrogenative coupling reaction with recoverable chiral amine/NaCl for accessing chiral $\alpha$ -alkyl $\alpha$ -glycine derivatives

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Correction for 'Ethanol-assisted mechanochemical asymmetric cross-dehydrogenative coupling reaction with recoverable chiral amine/NaCl for accessing chiral  $\alpha$ -alkyl  $\alpha$ -glycine derivatives' by Jingbo Yu *et al.*, *Green Chem.*, 2024, https://doi.org/10.1039/d4gc00947a.

The authors regret that inaccuracies were present in the mechanism and associated discussion in the original article. In Scheme 4F, 1a and 2a were labelled incorrectly. The corrected Scheme 4 is shown below.

The following sentence within the main article should refer to 2a instead of 1a, and should read as follows:

"The enamine intermediate **III** is formed *via* the condensation of **2a** with chiral amine **C8**, thus facilitating a stereoselective nucleophilic addition to intermediate **II** to give chiral intermediate **IV**."

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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C8 (10 mol%)

EtOH (η = 1.3), AcOH (6.5 mol%)

N<sub>2</sub>, NaCl (2.0 g)

MM, 20 Hz, (30 + 50) min

OHN PMP

3aa, n.d.

PMP

N

BHT (2.0 eq.)

standard conditions

3aa, n.d. +

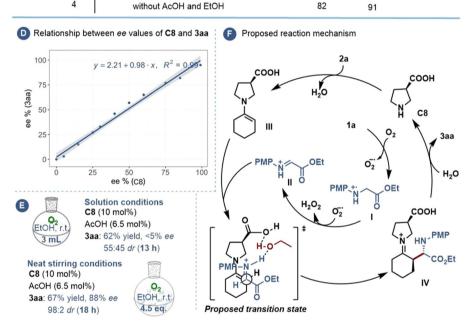
1a-BHT, 42%

Standard conditions
without O<sub>2</sub>

OEt

OEt

1-1 3aa variation from "standard conditionsa (without O2)" ee (%)a entry vield (%) 85 82 2 without EtOH 78 70 3 without AcOH 86 90 4



Scheme 4 Control experiments and plausible mechanism. (A) Mechanochemical reactions of 1a and 2a under a nitrogen atmosphere. (B) Radical trapping experiment. (C) Mechanochemical reaction of 1a and imine 1-1'. (D) Relationship between the ee values of C8 and 3aa. (E) Comparative reactions. (F) Proposed reaction mechanism. <sup>a</sup> Yields were those of the isolated products, ee values were determined by HPLC.