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## CORRECTION

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## Correction: Membrane degassing with the combination of sweep gas and vacuum pressure for ammonia removal

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Correction for 'Membrane degassing with the combination of sweep gas and vacuum pressure for ammonia removal' by Hongsik Yoon et al., Environ. Sci.: Water Res. Technol., 2023, 9, 467-473, https://doi.org/ 10.1039/D2EW00822J

Some values presented in the text for the mass transfer coefficient were in error, missing a minus sign before the exponent. All values in Table 1 were correct. The corrected values are shown below.

In the Water impact statement on page 467, the value of  $2.03 \times 10^3$  m h<sup>-1</sup> should read  $2.03 \times 10^{-3}$  m h<sup>-1</sup>, and the value of  $3.95 \times 10^{3} \text{ m h}^{-1} \text{ should read } 3.95 \times 10^{-3} \text{ m h}^{-1}.$ 

In the Results and discussion section on page 470, the value of  $2.03 \times 10^3$  m h<sup>-1</sup> should read  $2.03 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.95 \times 10^3$  m h<sup>-1</sup> should read  $3.95 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $5.75 \times 10^3$  m h<sup>-1</sup> should read  $5.75 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.50 \times 10^{-3}$  m h<sup>-1</sup> should read  $5.75 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.50 \times 10^{-3}$  m h<sup>-1</sup> should read  $3.95 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.50 \times 10^{-3}$  m h<sup>-1</sup> should read  $3.95 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.50 \times 10^{-3}$  m h<sup>-1</sup> should read  $3.95 \times 10^{-3}$  m h<sup>-1</sup>; the value of  $3.50 \times 10^{-3}$  m h<sup>-1</sup>; the value  $10^{3} \text{ m h}^{-1}$  should read  $3.50 \times 10^{-3} \text{ m h}^{-1}$ ; and the value of  $2.46 \times 10^{3} \text{ m h}^{-1}$  should read  $2.46 \times 10^{-3} \text{ m h}^{-1}$ .

In the Conclusion section on page 472, the value of  $2.03 \times 10^3$  m h<sup>-1</sup> should read  $2.03 \times 10^{-3}$  m h<sup>-1</sup>, and the value of  $1.18 \times 10^{-3}$  $10^2 \text{ m h}^{-1} \text{ should read } 3.95 \times 10^{-3} \text{ m h}^{-1}.$ 

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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