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Theoretical infrared spectroscopy of protonated methane isotopologues†

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The vibrational spectroscopy of protonated methane and its mixed hydrogen/deuterium isotopologues remains a challenge to both experimental and computational spectroscopy due to the iconic floppiness of CH5+. Here, we compute the finite-temperature broadband infrared spectra of CH5+ and all its isotopologues, i.e. $CH_nD_{5-n}^+$ up to CD_5^+ , from path integral molecular dynamics in conjunction with interactions and dipoles computed consistently at CCSD(T) coupled cluster accuracy. The potential energy and dipole moment surfaces have been accurately represented in full dimensionality in terms of high-dimensional neural networks. The resulting computational efficiency allows us to establish CCSD(T) accuracy at the level of converged path integral simulations. For all six isotopologues, the computed broadband spectra compare very favorably to the available experimental broadband spectra obtained from laser induced reactions action vibrational spectroscopy. The current approach is found to consistently and significantly improve on previous calculations of these broadband vibrational spectra and defines the new cutting-edge for what has been dubbed the "enfant terrible" of molecular spectroscopy in view of its pronounced large-amplitude motion that involves all intramolecular degrees of freedom.

Protonated methane, ${\rm CH_5}^+$, the prototype of hypercoordinate non-classical carbocations with outstanding relevance to chemistry, persists as an enduring challenge to both experiment and theory, even after decades of inquiry. The underlying cause of these difficulties is the unusually flat potential energy surface (PES) of ${\rm CH_5}^+$ which allows for large-amplitude motion in terms of pseudorotational dynamics. This leads to structural fluxionality called hydrogen scrambling that involves

archetypical floppy molecule³ for a long time. It has been dubbed more recently "the enfant terrible of chemical structures", given that even "the very concept of molecular structure becomes problematic".4 Yet, after a period of confusion, it has been established that despite its fluxionality⁵ even in the quantum ground state, 7 CH₅ can structurally be described in terms of a "CH3 tripod" to which an "H2 moiety" is attached via a three-center two-electron bond⁹ that constantly exchange their constituting protons due to hydrogen scambling. Despite enormous experimental difficulties, infrared (IR) spectroscopy of many different flavors is - so far - the only experimental probe to tackle the structural dynamics of CH₅⁺. ¹⁰⁻¹⁶ This includes high-resolution spectroscopy in narrow frequency windows carried out at cryogenic conditions as well as broadband spectra recorded at much higher temperatures spanning the entire mid-IR range. Systematic isotope substitution, where all five protons are

the many equivalent minima separated by only tiny energy

barriers. 5-8 For these reasons $\mathrm{CH_5}^+$ has been considered the

Systematic isotope substitution, where all five protons are replaced stepwise by deuterons yielding a total of six $\mathrm{CH}_n\mathrm{D}_{5-n}^+$ isotopologues, does not merely imprint structural differences¹⁷ but turned out to be a most powerful tool to unfold the spectral richness of protonated methane^{14,18–20} as proposed earlier.²¹ The pronounced nuclear quantum delocalization effects on the flat PES, different for heavy deuterons *versus* light protons, have been shown to lead to preferential, noncombinatorial populations of the different isotopomers within a given isotopologue^{14,20} as anticipated.²¹ This leads to drastic changes in the overall IR lineshape of the isotopologues $\mathrm{CH}_n\mathrm{D}_{5-n}^+$ compared to the CH_5^+ ref. 14 and 18–20 in stark contrast to simple frequency shifts as they are well known from stiff, quasirigid molecules.

When it comes to computing IR spectra of protonated methane and its isotopologues while including full anharmonicity and possibly also temperature effects, an array of complementary methods has been deployed to tackle that challenge for such floppy molecules. ^{12–14,18,20,22–36} Out of these, path integral-based simulations ^{37,38} rigorously tackle large-amplitude

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motion, hydrogen scrambling dynamics, and permutations of identical particles.36,39 In conjunction with spectral decomposition,³⁹ this approach has been demonstrated to be particularly successful in accurately assigning the peaks, shoulders, and other features of all isotopologue IR spectra to the underlying molecular motion in terms of generalized normal coordinates, 14,20 as reviewed some time ago.29 Yet, these previous path integral simulations were limited along multiple computational dimensions as outlined in the following. First of all, they either relied on a quantum-reweighting technique of classical IR spectra or on path integral centroid molecular dynamics (CMD)⁴⁰ to seamlessly generate the approximate quantum time-evolution of the different isotopologues. While the first approach is quasiclassical by construction, the CMD method has been shown by us to be plagued by increasingly pronounced artifical frequency red-shifts and peak broadenings due to the curvature problem when lowering the temperature;⁴¹ see ref. 42 and 43 for lucid reviews of approximate path integral dynamics methods including recent advances. Furthermore, the electronic structure has been computed earlier^{14,20} on-the-fly along the finite-temperature path integral simulations using density functional theory (DFT). Although the previously chosen density functional has been repeatedly shown to perform very well for CH₅⁺, the pronounced computational demand of on-the-fly electronic structure evaluation in the framework of path integral simulations³⁸ hampered both path integral and statistical convergence of the IR spectra.

Today, methodological advances offer unprecedented access to converged path integral simulations down to ultralow temperatures at the level of CCSD(T) coupled cluster electronic structure theory. This applies not only to interactions and thus the PES, but also the DMS (dipole moment surface) can be obtained from machine learning (ML) representations at coupled cluster quality; 44-49 see ref. 50 for a Perspective on converged path integral simulations combined with neural network techniques. The IR spectroscopic accuracy of our consistent coupled cluster-based ML approach to PES + DMS representations has been demonstrated recently for small molecules when combined with quasiexact wavefunction-based variational and quantum dynamics techniques to compute rovibrational energy levels and IR spectra. 51,52 Unfortunately, it is still impossible to rigorously compute finite-temperature IR spectra (and not merely a limited set of rovibrational states) of very floppy molecules such as protonated methane and its isotopologues using quasiexact quantum techniques. This arises from the intricacies of the large-amplitude motion of the fluxional CH₅⁺ species, leading to hydrogen scrambling that involves all internal degrees of freedom and entangled rovibrational states.31,33-35 As shown in the following, revisiting approximate path integral dynamics with our latest methodological improvements offers a viable path to achieve progress. In contrast to practical quasiexact wavefunction-based techniques, large-amplitude motion of all coupled degrees of freedom of fluxional molecules such as protonated methane are readily included in path integral-based techniques when computing broadband IR spectra at finite temperatures. Despite their virtues, finite-temperature path integral simulations are known to be unable to provide

high-resolution IR spectra, thus offering no insight into rovibrational energy levels and couplings. Computing such stateresolved spectra, however, is the domain of quasiexact wavefunction-based techniques^{26,31,34} where nuclear spin statistics of all identical particles are imposed to filter out the low-lying Pauli-allowed rovibrational states as required for high-resolution IR spectra close to the ground state. Incidentally, this complementary situation as to path integral-based versus wavefunction-based techniques is similar in experimental spectroscopy: those spectroscopic methods that provide broadband spectra at elevated temperatures 12,14,16,53 and those that yield high-resolution state-resolved spectra at low temperatures 11,13,15 are very different in nature. Our present focus clearly is on computing as accurately as possible the broadband IR spectra of all isotopologues of protonated methane at an elevated temperature of 100 K as obtained from laser induced reactions (LIR) action vibrational spectroscopy. 12,14

In this short note, we apply converged path integral simulations^{37,38} in the framework of ring polymer molecular dynamics (RPMD)⁵⁴ to compute the broadband IR spectra of all six $CH_nD_{5-n}^+$ isotopologues at 100 K close to the experimental temperature using ML-based potential energies and dipole moments at CCSD(T) accuracy; we refer to the "Computational methods and details" section at the end of the main text for the details of the underlying explicitly correlated CCSD(T*)-F12a/ aug-cc-pVTZ and DF-CCSD(T*)-F12a/aug-cc-pVDZ reference calculations of potential energies and dipole moments, respectively, abbreviated by "CCSD(T)" throughout this text, and to the ESI† for the neural network parameterization. Based on our results, we demonstrate that this approach currently provides the most accurate broadband IR spectra of protonated methane and its isotopologues as judged by comparing to the experimental broadband spectra.

We start the discussion connecting back to our earlier work²⁰ by comparing in panel (a) of Fig. 1 the IR spectrum of CH5+ obtained from CMD to the experimental LIR (laser induced reactions) action vibrational spectrum¹² recorded at roughly 110 K. At variance with our earlier work, we can first of all now improve the path integral convergence (using 150 Trotter slices to discretize the path integral at 200 K versus only 32 beads more than a decade ago and a total of 8 ns CMD trajectories versus 162.5 ps earlier to compute the spectrum). Secondly, we used DFT earlier to compute on-the-fly the interactions that govern the path integral time evolution as well as the dipole moments required to compute the IR spectrum from its auto-correlation function. Here, we now use CCSD(T) theory to describe both, the PES and DMS. Whereas the first improvement merely provides more thoroughly converged data, the latter enables one to clearly see the red-shift of the CMD spectrum in the C-H stretching window relative to LIR, which is an artifact due to the curvature problem; 41 we note in passing that 200 K is roughly the lowest temperature in the "linear regime",41 in the case of CH5+ before CMD fully breaks down. Earlier, 20 we clearly discussed this caveat while lacking at that time any means to quantify the CMD-shift: the CMD spectrum obtained from DFT matched LIR very well but only fortuitously

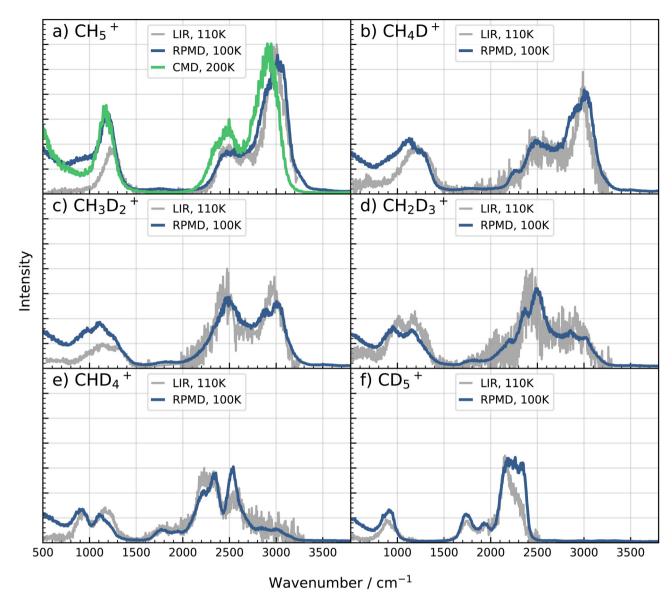


Fig. 1 Infrared spectra of CH_5^+ and its deuterated isotopologues up to CD_5^+ computed from RPMD (blue) and CMD (green) path integral simulations at CCSD(T) accuracy, see text. The corresponding experimental reference spectra (gray) were obtained from laser induced reactions (LIR) action vibrational spectroscopy as published in ref. 14 the arbitary intensity scale of the LIR spectra has been adjusted individually to allow for easy comparison to the computed IR spectra.

since the CMD-induced red-shift perfectly compensated the DFT-induced blue-shift of the C-H stretches. In addition, the splitting of the bending peak around 1200 cm⁻¹ as observed with DFT but not seen in experiment vanishes when using CCSD(T) in conjunction with CMD and, thus, is not a CMD artifact.

As opposed to CMD, RPMD does not feature any curvature problem by construction and thus no artifical red-shifts of vibrational spectra, yet artificial resonances between physical and fictitious modes⁴¹ might appear depending on the specific system and simulation parameters; see ref. 42 and 43 for reviews on deep analyses of this phenomenon and possible remedies. For CH₅⁺, we did not encounter such interferences at the relevant temperature and, thus, refrained from using

thermostats⁵⁵ or applying other modifications of bare RPMD propagation. 42,43 Indeed, the CCSD(T)-based RPMD spectrum of CH5+ turns out to be very close to the LIR ref. 12 at the experimental temperature, see Fig. 1(a). In this context, we recall that LIR56 is an intricate variant of action spectroscopy where an IR excitation, $\hbar\omega$, induces a bimolecular reaction of a mass-selected parent molecule, here CH₅⁺, with a reactant, CO₂ in this case, to generate CH₄ + HCO₂⁺ in a variable-temperature ion trap according to the reaction rate $k^*(\omega)$ of the excited ion.⁵³ The amount of HCO₂⁺ generated by the reaction is a measure of the absorption intensity as a function of the IR laser excitation wavelength ω , thus yielding the LIR action spectrum. This implies in the first place that the LIR intensities are not directly governed by the oscillator strength of rovibrational

transitions as is the case for linear IR absorption spectra. In addition, the LIR signal is proportional to the product of the linear absorption cross section, $\alpha(\omega)$, and the rate coefficient $k^*(\omega)$. While the former is directly computed in theoretical spectroscopy, the frequency-dependence of the latter is unknown but $k^{\star}(\omega)$ is estimated for the above laser induced reaction to decrease significantly below 1000 cm⁻¹ and to quickly drop toward zero according to approximate modeling. 12,53 In consequence, these LIR spectra are known to significantly lose sensitivity at frequencies below roughly 1000 cm⁻¹; we note in passing a recent proposal⁵⁷ of an empirical approach to approximately correct for this experimental phenomenon (see Section 4.4.2 and the Appendix in ref. 57). Yet, in view of all inherent uncertainties, we refrain from applying such ad hoc models of $k^{\star}(\omega)$ to the computed linear absorption cross section, $\alpha(\omega)$.

At this stage, we conclude that converged RPMD simulations using CCSD(T)-quality interactions and dipole moments satisfactorily reproduce the broadband IR spectrum of $\mathrm{CH_5}^+$ without any further adjustments.

Given that favorable situation, we now turn to the deuterated isotopologues including the fully deuterated CD₅⁺ species as well, see Fig. 1(b)-(f). The partially deuterated species have been shown earlier to provide deep insights into the assignment of the IR peak structure of protonated methane.¹⁴ This can be achieved in terms of generalized normal coordinates³⁹ by decomposing the IR spectra of all individual isotopomers underlying the fully scrambling $CH_nD_{5-n}^+$ isotopologues (which provide the observable IR spectra) into molecular motions in real space. 14,20 The current simulations provide overall a very convincing match with the LIR spectra¹⁴ which outperforms the earlier comparisons obtained from path integral-based methods14,20 as analyzed in the following (see Fig. 7 in ref. 29 for one-to-one comparisons to experiment offered by the previous calculations including the DMC/VCI IR spectra^{18,19} which (i) do not reproduce the clearly observed bimodal bending band of CH₂D₃⁺ and CHD₄⁺ and (ii) provide artificially symmetric lineshapes of the prominent stretching band of these two isotopologues that are very asymmetric according to experiment).

At first glance it may be surprising that the rather high intensity of the blue-wing part of the tripod stretching band of CD_5^+ depicted in Fig. 1 appears to visibly distort the peak structure. This most intense peak of the CD₅⁺ spectrum has been assigned previously¹⁴ to consist (from lower to higher frequencies) of the in-plane, symmetric out-of-plane and antisymmetric out-of-plane C-D stretching modes within the CD₃ tripod fragment of the perdeuterated species (labelled by t_D^i , t_D^s and $t_{\rm D}^{\rm a}$, respectively, and visualized in Fig. S1 of ref. 14). The lineshape of that peak computed from RPMD reflects (but overemphasizes) the significant skewness of the tripod stretching band that is clearly seen in the experimental LIR spectrum of CD_5^+ , whereas the corresponding peak of CH_5^+ (spanned by the $t_{\rm H}^{\rm i}$, $t_{\rm H}^{\rm s}$ and $t_{\rm H}^{\rm a}$ modes) is very symmetric according to both, experiment and simulation according to Fig. 1(a). We therefore assign the experimentally observed skewness of the tripod C-D

stretching band of ${\rm CD_5}^+$, unseen in previous calculations of that IR spectrum (see Fig. 7 in ref. 29), to the antisymmetric out-of-plane C–D stretching mode $t_{\rm D}^a$.

The present RPMD simulations that are entirely based on CCSD(T) quality electronic structure theory moreover reproduce perfectly the unimodal bending band (in the range of roughly 1000 to 1550 cm⁻¹) for CH_5^+ , CH_4D^+ , $CH_3D_2^+$ and CD_5^+ as well as the rather symmetric bimodal peak structure in case of the CH₂D₃⁺ and CHD₄⁺ isotopologues in full accord with experiment, see Fig. 1. This systematic agreement greatly improves previous calculations where, depending on the approach, the unimodal peak in LIR is found computationally to split into a greatly asymmetric dublet structure or the bimodal peak in LIR is confluent and yields a single maximum (see the IR spectra of all isotopologues in the original literature 14,18-20 or compiled in Fig. 7 of the review²⁹). Last but not least, the relative intensities of the different mode contributions to the significantly modulated lineshape functions of the $CH_nD_{5-n}^+$ species are found to closely match the experimental peak heights, which is another noteworthy improvement with respect to all previous calculations.29

In conclusion, our results demonstrate that path integral molecular dynamics simulations at CCSD(T) accuracy provide the most accurate set of theoretical broadband IR spectra for protonated methane and its partially deuterated isotopologues up to the perdeuterated species as judged by directly comparing to the respective experimental spectra at the same temperature of about 100 K. This is promising progress given the remaining lack of quasiexact methods that would be able to ultimately provide accurate IR spectra of extremely floppy molecules such as protonated methane, while this route has been successfully demonstrated recently for much less demanding species. 51,52 Moving away from practical path integral molecular dynamics simulations with all their inherent difficulties and limitations in systematically approximating exact quantum dynamics, 42,43 our finding provides the perfect basis for future work on highresolution rather than broadband vibrational spectroscopy of protonated methane. The goal in this case would be to assign in terms of state-to-state transitions the entangled rovibrational states31,33-35 that have been measured15 a decade ago with utmost accuracy - and yet await assignment! The necessary methodological advances in theoretical vibrational spectroscopy might come from improvements in multi-configurational timedependent Hartree (MCDTH), vibrational matrix product state (MPS) or tree tensor networks state (TTNS) techniques^{51,58} to eventually enable quasiexact calculations of IR spectra even of floppy molecular systems such as protonated methane.

Computational methods and details

All reported path integral simulations were carried out using the CP2k software package^{59,60} in conjunction with the RubN-Net4MD interface⁶¹ to high-dimensional neural network (NN) representations.⁶² We use our published NN-PES of CH₅⁺ trained on CCSD(T) energies computed from an augmented

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correlation-consistent basis set up to triple zeta functions^{63,64} (aug-cc-pVTZ or AVTZ) in combination with the explicitly correlated F12a method^{65,66} and an adequate scaling (*) of the triples, 66 thus providing CCSD(T*)-F12a/AVTZ accuracy for our NN-PES of CH₅⁺ as validated and benchmarked in detail in ref. 67. This potential energy surface is combined with a neural network-based dipole moment surface (NN-DMS), developed for this study (see below), to compute the reported IR spectra as usual from the Fourier transform of the dipole time-autocorrelation function generated from either RPMD⁵⁴ or CMD⁴⁰ path integral trajectories; 42,43 all reported IR spectra have been smoothed to a minimal extent by applying a Hann window that is identical to the length of the underlying RPMD or CMD trajectories. To correctly sample the quantum canonical ensemble each system was propagated using path integral Langevin equation (PILE) thermostatting⁶⁸ at T = 100 as well as 200 K with the time constant $\tau = 200$ fs and friction coefficient $\lambda = 0.5$; Trotter discretizations in terms of P = 300 and 150 beads were used at 100 and 200 K, respectively. After an equilibration period of 20 ps, path integral positions and velocities were recorded every 500 fs and used to start both, RPMD and CMD simulations at 100 and 200 K, respectively; in case of CMD we used a centroid adiabatic decoupling parameter of γ = 16 in accordance with the definition provided in ref. 41 corresponding to a decrease in the non-centroid bead masses by a factor of $\gamma^2 = 256$. To correctly integrate the resulting fast dynamics, a timestep of $\Delta t = 0.1$ fs was chosen for the RPMD simulations, whereas CMD was carried out with $\Delta t = 0.01$ fs at this level of adiabatic decoupling. In total, 400 such RPMD and CMD trajectories were generated for each of the six isotopologues and propagated for 20 ps, generating a total of 8 ns per isotopologue for both, RPMD and CMD sampling in order to compute the respective dipole time-autocorrelation functions. 42,43 For the newly parameterized NN-DMS of CH₅⁺ we refer to the ESI† where we compile all NN parameters and also report our validation tests to support the quality. The dipole moment vectors underlying this NN-DMS have been computed based on CCSD(T) data computed from an augmented correlationconsistent basis set up to double zeta functions^{63,64} (aug-ccpVDZ or AVDZ) in combination with the explicitly correlated F12a method^{65,66} and an adequate scaling (*) of the triples⁶⁶ within the density fitting (DF) approximation⁶⁹ thus providing DF-CCSD(T*)-F12a/AVDZ accuracy for our NN-DMS of CH₅⁺ which is consistent in quality with the NN-PES. In ref. 49 we show for a large data set of 10 000 structures of the Zundel cation that the AVDZ basis set results in a negligible mean absolute error of 0.003 D compared to the AVTZ basis set when computing dipole moments from DF-CCSD(T*)-F12a/AVDZ theory. The Molpro software package^{70,71} has been used to compute these dipole moments for which reference input is also provided in the ESI.†

Data availability

All CC calculations were performed using the Molpro quantum chemistry package. 70,71 The NN-DMS has been constructed

using the RubNNet4MD package,61 while the NN-PES used is available within the ESI† of ref. 67 The data used to train and validate the NN-DMS presented in this paper together with the final NN-DMS paremeterization have been included as part of the ESI.†

Conflicts of interest

There are no conflicts to declare.

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