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Cite this: Phys. Chem. Chem. Phys., 2024, 26, 23951

Received 11th May 2024, Accepted 21st August 2024

DOI: 10.1039/d4cp01965b

rsc.li/pccp

Introduction

Copper indium gallium selenide (CuInGaSe₂ or CIGS) is a versatile quaternary alloy primarily utilized in high-efficiency thin-film solar cells,^{1,2} achieving efficiencies over 20%, which are incorporated into solar panels, building-integrated photovoltaics (BIPVs), and portable electronics due to its high absorption coefficient and flexibility on various substrates. Its applications extend to architectural integration, enabling seamless solar integration into building facades, windows, and wearable technologies, offering energy solutions for gadgets like solar-powered watches and clothing. Furthermore, CIGS's robustness against radiation and high efficiency-to-weight ratio make it suitable for spacecraft solar panels, and its potential in powering vehicles and consumer electronics signifies its role in reducing fossil fuel reliance.³⁻⁶ These varied applications underscore CIGS's potential in advancing green technology across multiple sectors. CIGSe thin film fabrication has been advanced using non-hydrazine-based solution processes,

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This study pioneers the use of CuInGaSe₂, a quaternary alloy, in Schottky barrier diodes, beyond its traditional application in solar cells, highlighting its potential in sustainable energy technologies such as supercapacitors. We delve into its unique electrical and dielectric characteristics by introducing CuInGaSe₂ into the Schottky barrier diode device, synthesized *via* an innovative liquid phase epitaxy on silicon substrates. Our investigation into the structural, electrical, and dielectric properties reveals the alloy's exceptional capacitance behavior, which transitions from positive to negative with varying frequency. It takes negative values at a frequency of 12900 Hz and a temperature of 300 K. In comparison, at a frequency of 1216 Hz, all curves take negative and positive values, demonstrating significant promise for enhancing the efficiency and sustainability of energy storage solutions. These findings contribute to the advancement of supercapacitor production and underscore the broader applicability of CuInGaSe₂ in promoting sustainable energy technologies.

overcoming previous thickness limitations of about 1.2-1.3 microns.⁷ Researchers have fabricated CIGSe absorbers up to 2 microns thick by altering the ink formulation and using high-temperature processes. This also helped eliminate the fine-grain layers common in thinner films. This improvement aligns the film morphology with those made using more efficient hydrazine-based methods.⁸ Electrodeposition is a cost-effective alternative to vacuum-based processes. It involves depositing the metallic elements from an aqueous solution onto a conductive substrate, followed by annealing in the presence of selenium to form a CIGS layer. This method allows for low-cost production and is particularly useful in large-area applications. Bhattacharya et al. comprehensively reviewed the electrodeposition process for CIGS.9 Co-evaporation is one of the most common methods for depositing CIGS layers. This process evaporates copper, indium, gallium, and selenium in a vacuum to form a thin film on a substrate. The advantage of this technique is the precise control it offers over the composition and thickness of the film, which is crucial for achieving high-efficiency solar cells. Studies like those by Rockett and Birkmire provide extensive insights into the process and its optimization.¹⁰ CIGSe solar cells with a change efficiency of superior 22% have been manufactured through the multiprocess co-evaporation procedure. Still, the implementation of wide-area modules in mass production is limited due to the high product charges.¹¹⁻¹³ Hybrid techniques combine elements from different deposition methods to leverage the benefits of each. For instance, combining sputtering for the deposition of metal precursors and rapid thermal processing (RTP) for the selenization step can result in improved film quality and reduced



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production time. The study by Kim *et al.* explores such hybrid approaches in detail.¹⁴

Sputtering is widely used for manufacturing CIGSe devices and large modules due to its reliable characteristics, ensuring a consistent composition over a large area.15,16 Manufacturing based on the mixing of sputtering ceramic targets and annealing must be further investigated; for example, the device performance, especially the open-circuit voltage (VOC), is lower compared to devices made using the co-evaporation technique.^{17,18} Enhancement of the device efficiency is related to the increase of $V_{\rm OCi}$ ¹⁹ an increase in the Ga content in CIGSe-based absorbers causes E_{g} to expand from 1.04 to 1.68 eV, resulting in an increase in $V_{\rm OC}$ and a decrease in short-circuit current density.²⁰ The optimal efficiency of solar cells can theoretically be obtained when the absorber E_{g} approaches 1.45 eV; temporarily, the consistent ratio of Ga/(In + Ga) (GGI) for CIGSe devices is 0.6-0.7.^{21,22} Various researchers in the solar cell fabrication field have investigated CuInGaSe2.

However, we will address the structure of Au/CuInGaSe₂/ SiO₂/n-Si/Al as a Schottky barrier diode, giving the possibility to be used in other applications such as supercapacitors. Several deposition methods have been used for CuInGaSe₂ alloys, such as molecular beam epitaxy, sputtering deposition, and metalorganic vapor deposition. To our knowledge, liquid phase epitaxy (LPE) deposition is yet to be investigated. We manufactured CuInGaSe₂ on a silicon wafer using (LPE); the prepared structure's structural, electrical, and dielectric properties were investigated by scanning electron microscopy, and I-V and C-V measurements. A comprehensive study of the capacitance behavior of the Au/CuInGaSe2/SiO2/n-Si/Al structure is presented here, and its values change from positive to negative with frequency. It takes negative values at 12 900 Hz and 300 K; at a frequency of about 1216 Hz, all curves take negative and positive values. However, at low frequencies of 114 Hz and 30 Hz, all curves take only negative values. The capacitance values were extremely high, indicating that the structure is ideal for supercapacitor manufacturing.

Experimental procedure

The quaternary alloy CuInGaSe₂ was synthesized using the liquid phase epitaxy (LPE) technique. The substrate chosen for this procedure was an *n*-type silicon wafer oriented explicitly along the (100) plane. The raw materials, namely copper (Cu), indium (In), gallium (Ga), and selenium (Se), were sourced from Sigma Aldrich, each with a purity level of 95%. These metallic elements were first dissolved in an indium-based solvent for epitaxial growth. This preparation step was crucial to ensure the formation of a supersaturated solution of CuInGaSe₂. The chosen temperature to dissolve the metals in the indium solvent was 1173 Kelvin, which facilitated the formation of a homogeneous alloy solution. As the solution cooled, the crystallization of CuInGaSe₂ commenced on the surface of the silicon wafer. This crystallization was primarily influenced by the cooling rate of the solution and the specific interactions between the supersaturated solution and



the silicon substrate. These interactions are pivotal as they determine the quality and uniformity of the epitaxial layer formed. The detailed mechanics of the LPE process, including the precise parameters and the conditions under which crystallization was optimized, have been discussed extensively in previously published literature. This foundational knowledge aids in understanding the subtleties of the liquid phase epitaxy technique as applied to the fabrication of CuInGaSe₂ thin films.²³

A layer of silicon dioxide was formed between CuInGaSe₂ and silicon wafers to create the CuInGaSe₂/SiO₂/n-Si structure, and then two electrodes of gold and aluminum were deposited on the upper and lower surfaces of the unit Au/CuInGaSe₂/SiO₂/ n-Si/Al using thermal evaporation. Fig. 1 presents the XRD pattern of the epitaxial-grown CuInGaSe₂ film on a SiO₂/Si substrate. In addition to the XRD peak corresponding to Si at 2θ = 68.999°, which is assigned to the diffraction plane (400) for the Si substrate JCPDS PDF no. (00-0787), there are two dispersed small peaks centered at $2\theta = 17.5^{\circ}$ and 27.2° corresponding to CuGa_{0.5}In_{0.5}Se JCPDS PDF no. (40-1488). These results confirmed the formation of the CuInGaSe₂ as a singlephase on the SiO₂/Si substrate. Gaussian fitting for XRD peaks at $2\theta = 17.5^{\circ}$ and 27.2° is performed (Fig. 2), and their FWHM values are estimated. By applying Scherrer's equation,²⁴ the crystallite size could be calculated as follows:

$$D = \frac{K\lambda}{\beta\cos\theta} \tag{1}$$

where λ is the wavelength of the radiation, β is the width at full width at half maximum intensity, θ is Bragg's angle, and *K* is the Scherrer constant. The estimated crystallite size is 23 and 45 nm as deduced by the XRD peaks at $2\theta = 17.5^{\circ}$ and 27.2° , respectively. Further morphological details are revealed by SEM imaging, as shown in Fig. 3, which displays the top surfaces of the CuInGaSe₂ films. The SEM micrographs distinctly illustrate the film's grain sizes and structural arrangement. Such imagery is crucial for assessing the uniformity and quality of the film, as grain boundaries and size can significantly influence the electrical and optical properties of the material. The evident variation in grain size, as visually supported by the SEM analysis, corroborates the XRD findings and highlights the heterogeneity in the crystalline formation across the film. This



Fig. 2 Gaussian fit of the XRD peaks at planes (101) (a) and (112) (b) for CulnGaSe₂/SiO₂/Si.



Fig. 3 SEM image of Au/CuInGaSe₂/SiO₂/n-Si/Al

combination of XRD and SEM analyses is vital for a comprehensive understanding of the film's physical characteristics, directly impacting its potential applications in photovoltaic devices and other semiconductor technologies.

Capacitance properties

The measurements were performed across a frequency range of 0.1 Hz to 20 MHz and a temperature range of 303 to 423 K using a Novocontrol high-resolution alpha analyzer. The setup included a Quatro temperature controller, which utilized clean nitrogen as a heating agent, ensuring temperature stability within ± 0.2 K. This instrument is equipped to provide comprehensive dielectric parameters, including capacitance, conductance, impedance, modulus, dielectric constant, and series resistance, across various frequencies, temperatures, and voltages. The dielectric properties of the Au/CuInGaSe₂/SiO₂/n-Si/Al structure, as detailed

in Fig. 4 (a through g), reveal complex capacitance behaviors dependent on the temperature, voltage, and frequency variations. As the temperature decreases, capacitance notably increases, peaking around 1 V, indicative of thermal sensitivity in the dielectric properties due to potential changes in charge carrier mobility or density. This trend is particularly pronounced at higher frequencies $(2 \times 10^7 \text{ and } 1115970 \text{ Hz})$, where capacitance values remain positive and increase with frequency reduction, suggesting charge dispersion effects related to polarization and charge trapping. The anomalous peak observed in the forward bias region may be attributed to several factors. Firstly, the combined effect of series resistance and charge trapping/detrapping at the interface states can lead to a noticeable increase in capacitance. Additionally, dielectric relaxation phenomena within the material could contribute to this effect. As the frequency increases, the relaxation processes might not be able to keep up with the applied voltage changes, resulting in a capacitance peak. Furthermore, in forward bias, particularly at higher voltages, there might be an increased injection of minority carriers. This can lead to additional recombination and generation processes, thereby affecting the overall capacitance of the device. At lower frequencies (105 236 Hz), increased capacitance in the positive voltage region suggests enhanced charge storage capacity.

Further reduction in the frequency (12 900 and 1216 Hz) results in the capacitance displaying positive and negative values, expanding dramatically in response to voltage changes, which could be linked to complex interfacial charge dynamics involving electrons and holes. Notably, at very low frequencies (114 and 30 Hz), the structure exhibits only negative capacitance values, which might be attributed to border charge loss, series resistance, and ionization processes, possibly exacerbated by interface states (N_{ss}) and series resistance (R_s) as indicated in the literature.^{25–27} This detailed characterization underscores the intricate interplay between thermal, electrical,



Fig. 4 (a)-(g) C vs. V at different temperatures and frequencies of Au/CuInGaSe₂/SiO₂/n-Si/Al.

and frequency parameters in shaping the dielectric behaviour of this semiconductor structure, highlighting its potential for tailored applications in semiconductor and photovoltaic technologies.

V (V)

V (V)

f =12900 H

V (V)

1.6×10

1.2×10

8.0×10

4.0×10

1×10

C (F

-1×10

-3×10 C (F) -6×10

-9×10⁻³ - 375

d

C (F)

Fig. 5(a) and (b) illustrates the capacitance behavior of the Au/CuInGaSe₂/SiO₂/n-Si/Al structure as a function of temperature at a constant voltage of V = -5 V across various frequencies. The observed capacitance behavior diverges significantly between low and high-frequency ranges. At low frequencies, the capacitance remains steady at a temperature of 350 K but then decreases, forming peaks at around 360 K, with all recorded values being negative. This suggests a specific thermal response of the dielectric properties at these frequencies, potentially indicative of negative capacitance phenomena related to interfacial charge dynamics or

ionization processes that become more pronounced at this temperature range. Conversely, at high frequencies, the capacitance increases as the frequency decreases, remains stable up to 350 K, and then rises to reach maximum values as the temperature grows. The capacitance values in this high-frequency range are consistently positive, highlighting a distinct frequency-dependent thermal sensitivity. These observations, as shown in Fig. 5(a) and (b), suggest complex interactions between temperature, voltage, and frequency that affect the dielectric properties of the structure, which is critical for optimizing device performance in practical applications. The underlying mechanisms likely involve changes in charge carrier dynamics and polarization effects.^{28–30}

Fig. 6 presents the capacitance variation with frequency at different temperatures and voltages of Au/CuInGaSe₂/SiO₂/n-Si/



Fig. 5 (a) and (b) C vs. T at different frequencies and constant Au/CulnGaSe₂/SiO₂/n-Si/Al voltage.



Al. The capacitance curves in Fig. 6(a)–(h) all have near-zero values at high and mid frequencies, though the capacitance disperses and takes positive and negative values at low frequencies. The difference is that capacitance curves disperse further at positive V = 5, 4, 3, and 2 V compared to negative voltages. Diodes with imperfections and boundary states can exhibit the NC effect using appropriate bias and frequency,^{31,32} even the electrons can cross the boundary with enough energy. Electrons collect extra energy and disrupt trapped electrons as they are exposed to a particular bias and frequency. This causes a time delay in the trapped carrier, causing current flow, which corresponds to a non-perfectly phase connection where the current pauses behind the voltage, which causes an inductive effect or negative capacitance effect.³³

Fig. 7(a)–(h) shows the capacitance dependence of voltage at different frequencies and temperatures of Au/CuInGaSe₂/SiO₂/ n-Si/Al. The capacitance has the same behavior over various frequencies; it raises with decreasing frequencies and voltage, showing characteristic peaks at about V = 1 V, and it decreases in the positive voltage region. The capacitance at high frequencies has positive values, as displayed in Fig. 7(a), (c), (e) and (g), while at low frequencies, it takes negative values and has the same peak position at about V = 1 V. Negative capacitance means that the materials display inductive performance;³⁴ significantly, the NC phenomenon originates from electronhole recombination by local traps.^{35,36} NC dependences on the frequency and bias voltage are reliable with bimolecular recombination, representing that the electron–hole recombination is mediated by trap states and/or materials contamination.^{32,37}

Fig. 8(a)–(d) showcases the capacitance characteristics of the Au/CuInGaSe₂/SiO₂/n-Si/Al structure across different frequencies, voltages, and temperatures, revealing critical insights into the device's electrical behavior. Notably, the capacitance exhibits a uniform behavior at high and mid-frequencies, where it is nearly zero, indicating minimal energy storage capability or negligible charge accumulation under these conditions. This consistency suggests a stable dielectric response within these frequency ranges regardless of the applied voltage or ambient temperature.

However, at higher frequencies, the scenario shifts dramatically. Here, the capacitance deviates from zero and forms relaxation peaks at each voltage setting. The amplitude of these peaks varies with temperature, and they display both positive and negative values. This behavior indicates complex dielectric relaxation processes occurring within the material, where the dielectric polarization lags the applied electric field at high frequencies. The positive and negative values of the capacitance peaks suggest that the material exhibits both conventional and unconventional (negative) dielectric behaviors depending on the specific voltage and temperature conditions. These phenomena could be attributed to various factors such as interface states, charge trapping, and the movement of charge carriers that respond differently under changing environmental conditions. The detailed analysis provided by Fig. 8(a)-(d) highlights the intricate interplay of frequency, voltage, and temperature in shaping the capacitance behavior of this semiconductor structure, which is essential for understanding its potential applications in electronics and photovoltaics.







Fig. 8 (a)–(d) C vs. ln f at different voltages and temperatures of Au/CuInGaSe₂/SiO₂/n-Si/Al.

I-V current-voltage

The I-V and $\ln I-V$ curves of the Au/CuInGaSe₂/SiO₂/n-Si/Al structure measured at different temperatures are shown in

Fig. 9(a) and (b). The thermionic emission (TE) approach describes the process by which electrons overcome the potential energy barrier at the interface of a metal and a semiconductor. According to this theory, the current I flowing



Fig. 9 (a) and (b) I, ln I vs. V at different temperatures of Au/CuInGaSe₂/SiO₂/n-Si/Al.

 Table 1
 Electrical parameters of a Schottky diode constructed with an Au/CulnGaSe₂/SiO₂/n-Si/Al structure, evaluated using different theoretical approaches across varying temperatures

T (K)	<u>(</u> <i>I</i> - <i>V</i>)			Cheung (H)			dV/dln I		Nord (F)		
	n	$\phi_{ m h}$ (eV)	$R_{\rm s}$ (Ω)	n	$\phi_{ m h}$ (eV)	$R_{\rm s}$ (Ω)	n	$R_{\rm s}$ (Ω)	n	$\phi_{ m h}$ (eV)	$R_{\rm s}$ (Ω)
30	10.85	0.64	$6.7 imes 10^3$	8.28	0.24	$6.71 imes10^3$	8.28	$7.12 imes 10^3$	10.85	0.68	$7.45 imes 10^4$
50	9.89	0.76	$6.2 imes10^3$	6.82	0.25	7.69×10^3	6.82	$5.58 imes10^3$	9.89	0.71	$4.20 imes10^4$
75	7.98	0.75	$1.13 imes10^4$	5.42	0.29	1.76×10^4	5.42	$1.34 imes10^4$	7.98	0.78	$5.77 imes10^4$
100	6.29	0.82	$1.54 imes10^4$	2.62	0.30	$4.71 imes10^4$	2.62	$3.69 imes10^4$	6.29	0.83	$3.43 imes 10^4$
125	6.03	0.85	$1.20 imes 10^4$	1.60	0.33	$3.19 imes10^4$	1.60	$2.13 imes 10^4$	6.03	0.87	$4.20 imes10^4$

through a Schottky diode can be expressed using the following equation: $^{^{38-40}}$

$$I = I_0 \left[\exp \frac{q(V - IR_s)}{nkT} \right]$$
(2)

$$\phi_{\rm b} = \frac{kT}{q} \ln\left(\frac{AA^*T^2}{I_0}\right) \tag{3}$$

$$n = \frac{q}{KT} \left(\frac{\mathrm{d}V}{\mathrm{d}\ln I} \right) \tag{4}$$

where I_0 is the saturation current, R_s is the series resistance, V is the practical voltage, n is the ideality factor, T is the measured temperature in Kelvin, q is the electronic charge, K is the Boltzmann constant, ϕ_b is the Schottky barrier height (SBH), A^* is the Richardson constant, and A is the contact area of the diode.

$$\left(\frac{\mathrm{d}V}{\mathrm{d}\ln I}\right) = \frac{nkT}{q} + IR_{\mathrm{s}} \tag{5}$$

 I_0 was distinct by the intersection amid the interpolated straight lines of the lined area and the current axis. The *n* value can be calculated using eqn (2).⁴¹⁻⁴³ The $\phi_{\rm b}$, *n*, and series resistance values of the diode are listed in Table 1.

$$F(V) = \frac{V}{\Upsilon} - \frac{1}{\beta} \ln\left(\frac{I}{AA^*T^2}\right)$$
(6)

$$\phi_{\rm b} = F(V_{\rm min}) + \frac{V_{\rm min}}{\Upsilon} - \frac{1}{\beta} \tag{7}$$

$$H(I) = V - n\left(\frac{KT}{q}\right) \ln\left(\frac{I}{AA^*T^2}\right)$$
(8)

Fig. 10 shows the analysis of the Au/CuInGaSe₂/SiO₂/n-Si/Al structure and illustrates the conduction mechanisms *via* a double-logarithmic (log *I vs.* log *V*) presentation under forward bias across a temperature range of 300–400 K. The curves exhibit two distinct linear segments with varying slopes based on the applied voltage. At lower voltages ($V \le 0.6$ V), the *I*-*V* relationship is linear with a slope of approximately 1, suggesting ohmic conduction where current increases linearly with voltage. This region reflects the unimpeded flow of charge



Fig. 10 In / vs. In V at different Au/CuInGaSe₂/SiO₂/n-Si/Al temperatures.

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carriers across a potential gradient. However, as the voltage increases beyond 0.6 V up to 2 V, the slope changes to between 2.2 and 2.9, indicating a transition to space charge limited conduction (SCLC). SCLC suggests that the external voltage drives the current and is also significantly influenced by the internal electric fields due to accumulated charge carriers, which is typical in scenarios with high defect densities or deep traps. This behavior under higher voltage settings highlights robust charge injection and carrier transport capabilities, suggesting that the device can handle high-power applications efficiently, even at elevated temperatures. Understanding these transitions in conduction mechanisms is vital for optimizing the junction's design and enhancing its performance across various operating conditions.⁴⁴

Fig. 11 confirms that the device has a good rectification ratio. Fig. 12 illustrates the variation of junction resistance (R_j) with voltage, allowing the series and shunt resistances to be calculated; as seen in Fig. 13 and 14, both decrease with an increase in temperature.

Regarding Cheung's method, R_s and n can be determined by intersecting the slope from the lined area of the $dV/d(\ln I)$ and H(I) curves.^{45–49} As shown in Fig. 15 and 16, calculations shown in eqn (5) and (8) presented the values of R_s and n that are listed in Table 1. The prepared MOS Schottky diode has a high R_s value as there was a thin film of SiO₂ layer amid the metal and semiconductor. The Norde equation was also used to determine the ϕ_b value of the diodes. The Norde function is described as the F(V) vs. the voltage V, as shown in Fig. 17.^{50,51}

Table 1 presents the electrical parameters of a Schottky diode constructed with an Au/CuInGaSe₂/SiO₂/n-Si/Al structure, evaluated using different theoretical approaches across varying temperatures. The parameters provided include the diode's barrier height ($\phi_{\rm b}$), ideality factor (*n*), and series resistance ($R_{\rm s}$), which are essential for characterizing the diode's performance and understanding its behavior under different conditions. The data are organized to reflect the outcomes from



Fig. 11 R_R vs. V at different Au/CulnGaSe₂/SiO₂/n-Si/Al temperatures.



Fig. 12 R_i vs. V at different Au/CuInGaSe₂/SiO₂/n-Si/Al temperatures.



Fig. 13 R_s vs. T of Au/CulnGaSe₂/SiO₂/n-Si/Al



various theoretical models and temperature settings. Generally, the series resistance decreases as the temperature increases, which suggests improved conductivity. Barrier heights and



Fig. 15 dV/dln / vs. / at different temperatures of Au/CulnGaSe_2/SiO_2/n-Si/Al.





Fig. 17 F(V) vs. V at different temperatures of Au/CulnGaSe₂/SiO₂/n-Si/Al.

ideality factors also show variations that indicate different charge carrier dynamics at elevated temperatures. This comprehensive data set is crucial for understanding the impact of temperature on the diode's performance and optimizing its use in various applications.

Conclusions

This study introduced a novel approach to studying the CuIn-GaSe₂ alloy, a material crucial to the modern solar cell industry but explored here for its potential in microelectronic applications, particularly as a Schottky diode and supercapacitor. For the first time, CuInGaSe₂ was grown on a silicon wafer using liquid phase epitaxy, highlighting an innovative method of layer deposition that could enhance the performance of semiconductor devices. The study comprehensively examined the structural, electrical, and dielectric properties of the Au/CuInGaSe2/SiO2/n-Si/Al structure through various techniques. Scanning electron microscopy provided insights into the material's morphology, while I-V and C-V measurements assessed its electrical and dielectric behaviors. X-ray diffraction analysis confirmed the successful formation of a single-phase CuInGaSe₂ on the SiO₂/Si substrate, indicating good crystallinity and structural integrity. Capacitance measurements were particularly revealing, showing negative values at a frequency of 12 900 Hz and a temperature of 300 K, which suggests unusual dielectric properties that may be advantageous for specific applications. At around 1216 Hz, the capacitance showed both negative and positive values, and at even lower frequencies of 114 Hz and 30 Hz, it consistently showed negative values. These results indicate a complex interplay of charge dynamics within the material, influenced by frequency and temperature. Most notably, the observed high capacitance levels at these frequencies demonstrate the structure's excellent potential for use in supercapacitors. The ability to handle such high capacitance levels makes this CuInGaSe₂based structure a promising candidate for energy storage applications, where high efficiency and reliability are paramount. Overall, the findings from this work not only enhance our understanding of CuInGaSe₂ in microelectronic applications but also open up new avenues for using this versatile material in advanced technological applications.

Data availability

The data supporting this article have been included as part of the article.

Conflicts of interest

The authors declare no conflicts of interest.

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