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# Chemical fixation of CO<sub>2</sub>/CS<sub>2</sub> to access iodoallenyl oxazolidinones and allenyl thiazolidine-thiones†

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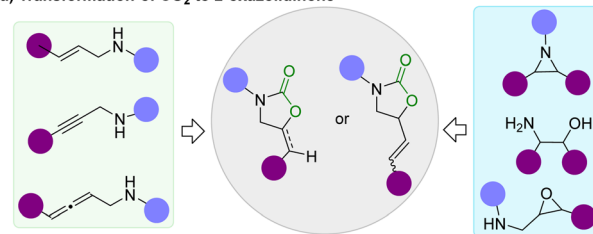
Constructing heterocyclic compounds by chemical fixation of CO<sub>2</sub>/CS<sub>2</sub> as a C1 building block is a promising approach. An efficient and environmentally friendly synthetic approach has been developed using CO<sub>2</sub>/CS<sub>2</sub> to prepare complicated allenyl heterocycles with high yields and diastereoselectivities in a metal-free manner under mild conditions. NIS promoted CO<sub>2</sub> fixation and the cyclization reaction by exclusive 1,4-*syn*-addition of 1,3-enynes rather than 1,2-addition or 3,4-addition, while CS<sub>2</sub> participated in unique 1,4-*syn*-hydrothiolation of 1,3-enynes to afford allenyl heterocycles with different reaction patterns.

With the increasing of the ever-growing carbon dioxide (CO<sub>2</sub>) level in the atmosphere, a variety of approaches have been developed for CO<sub>2</sub> capture, storage and its utilization in the past few decades.<sup>1</sup> With CO<sub>2</sub> being an abundant, nontoxic and easily available C1 building block, the fixation and post-transformations into valuable products are very hot topics related to the effective reduction of CO<sub>2</sub> emissions nowadays.<sup>2</sup> However, chemists have achieved very limited yields for the conversion of CO<sub>2</sub> to carboxylic acids, (poly)carbonates, carbamates, ureas and so on.<sup>1a,3</sup> Therefore, efficiently recycling CO<sub>2</sub> into value-added products remains a formidable challenge especially under atmospheric pressure due to the thermodynamic stability and kinetic inertness of CO<sub>2</sub>.

2-Oxazolidinones are some of the important heterocyclic compounds, and have been widely applied as chemical intermediates, chiral auxiliaries, drugs and other bioactive molecules.<sup>4</sup> To date, several methods were developed for constructing oxazolidinones using CO<sub>2</sub> as a sustainable feedstock such as carbonylations of amino alcohols with phosgene/CO<sub>2</sub>, carbamations of unsaturated amines/epoxy amines with CO<sub>2</sub>, coupling reactions between aziridine and CO<sub>2</sub>, *etc.*<sup>5–7</sup> However,

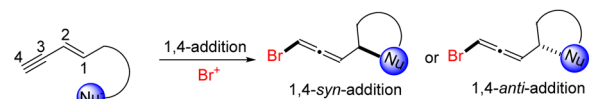
harsh reaction conditions including strong acids/bases, or high pressures, have usually been utilized to activate inert CO<sub>2</sub>. Recently, several catalytic approaches for the carboxylative cyclization of unsaturated amines with CO<sub>2</sub> have been disclosed, including using transition metal catalysts based on Ru, Pd, Ag, Au, and Cu<sup>8</sup> and organocatalysts obtained using superbases,<sup>9</sup> ionic liquids<sup>10</sup> and N-heterocyclic carbene<sup>11</sup> (Scheme 1a). Additionally, halogen promoted intramolecular nucleophilic 1,4-*syn/anti*-addition of 1,3-enynes represents one

## (a) Transformation of CO<sub>2</sub> to 2-oxazolidinone

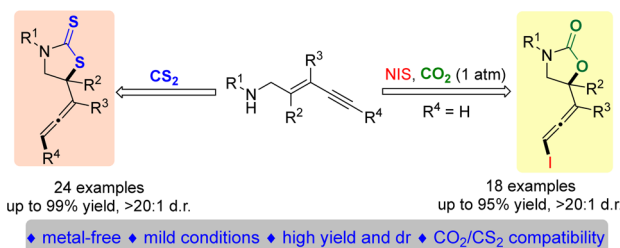


Transition metal catalyst: Ru, Pd, Ag, Au, Cu  
Organocatalyst: superbase (e.g. TBD, DBU, NHC, ILs)

## (b) Halogen promoted nucleophilic 1,4-addition of 1,3-enynes for allenyl bromides



## (c) This work: CO<sub>2</sub>/CS<sub>2</sub> fixation and nucleophilic 1,4-*syn*-addition of 1,3-enynes



**Scheme 1** Synthesis of oxazolidinones from CO<sub>2</sub>. (a) Transformation of CO<sub>2</sub> to 2-oxazolidinone. (b) Halogen promoted nucleophilic addition. (c) CO<sub>2</sub>/CS<sub>2</sub> fixation and nucleophilic addition.

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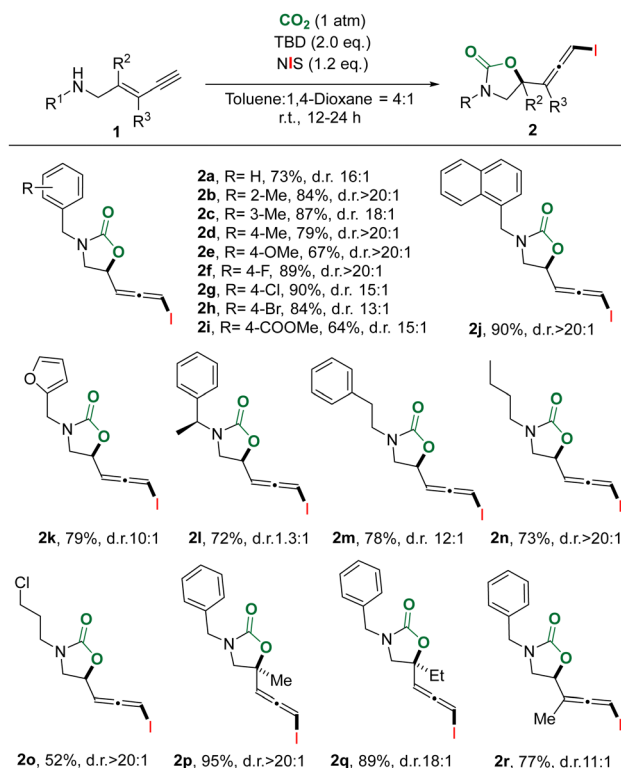
of the powerful approaches for the stereoselective introduction of allenyl bromides (Scheme 1b).<sup>12</sup> Allenes are attractive molecules, and serve as versatile synthons for preparation of a variety of bioactive compounds.<sup>13</sup> Therefore, we envisioned that 1,3-enynes could efficiently activate CO<sub>2</sub> to undergo nucleophilic 1,4-addition rather than 1,2-addition or 3,4-addition promoted by halogen. Such a designed reaction can not only afford important haloallenyl-containing 2-oxazolidones from simple starting materials, but also expand the diversity of CO<sub>2</sub> transformations (Scheme 1c).

Carbon disulfide (CS<sub>2</sub>) as an isoelectronic analogue of carbon dioxide is extensively used to prepare versatile organosulfur compounds due to its low cost, stability, and easy availability. A thiazolidine-2-thione skeleton is a useful intermediate in the fields of medicine, agriculture, and fine chemicals.<sup>14</sup> Although the synthetic methods for the preparation of thiazolidine-2-thiones using CS<sub>2</sub> have been reported,<sup>15–17</sup> the reaction between CS<sub>2</sub> and 1,3-enynes to afford allenenes is still unprecedented. Compared to CO<sub>2</sub>, CS<sub>2</sub> is considered to be more reactive due to the weaker C=S double bond.<sup>18</sup> Thus, we rationalized that CS<sub>2</sub> could rapidly react with 1,3-enynes even without additional electrophilic reagents. Herein, we proposed a concept for chemical fixation of CO<sub>2</sub>/CS<sub>2</sub> for converting these C1 building blocks to unique iodoallenyl 2-oxazolidones or allenyl thiazolidine-2-thiones by exclusive 1,4-*syn*-addition of 1,3-enynes.

To test the concept, we began our investigation by using **1a** as the model substrate, and treated with DBU and NIS under 1 atm CO<sub>2</sub> in CHCl<sub>3</sub> at room temperature. The iodoallenyl oxazolidinone product **2a** was obtained in 28% yield with 1:1 dr (Table S1, ESI,† entry 1). Then, various bases were screened (Table S1, ESI,† entries 2–6). TBD (1,5,7-triazabicyclo[4.4.0]dec-5-ene) could provide better combined yield and diastereoselectivity, generating product **2a** in 48% isolated yield and 3:1 dr. Next, different solvents and binary mixed solvents were explored and best results were obtained using a toluene/1,4-dioxane (4:1) system (Table S1, entry 25, see the ESI† for optimization details).<sup>19</sup>

With the optimal reaction conditions in hand, we investigated the substitution effect on the amines and alkenes for the substrate scope (Table 1). Various changes at nitrogen and alkene substituents could be well tolerated. For the diverse substituents of nitrogen, no obvious electronic effects were observed in most cases. The *ortho*-, *meta*- and *para*-methylbenzyl substrates provided desired products (**2b–2d**) with a high yield (79–87%) and dr (18:1 to > 20:1). These results indicate that different substituted patterns may affect the reactivity. The more electron-donating *para*-methoxy group gave the desired product (**2e**) in 67% with excellent dr (> 20:1). Therefore, more *para*-substituted substrates were examined for this CO<sub>2</sub> chemical fixation. *para*-Halo-substrates also showed high yields (84–90%) even though the diastereoselectivities from fluoro- to chloro- and bromo- slightly decreased (**2f–2h**). The desirable product (**2i**) was afforded in moderate yield and dr for the more electron-withdrawing ester substrate. To our delight, sterically demanding 1-naphthylmethyl carbamate (**2j**) was acquired in excellent yield (90%) and diastereoselectivity (dr > 20:1). In contrast, a lower dr was observed for

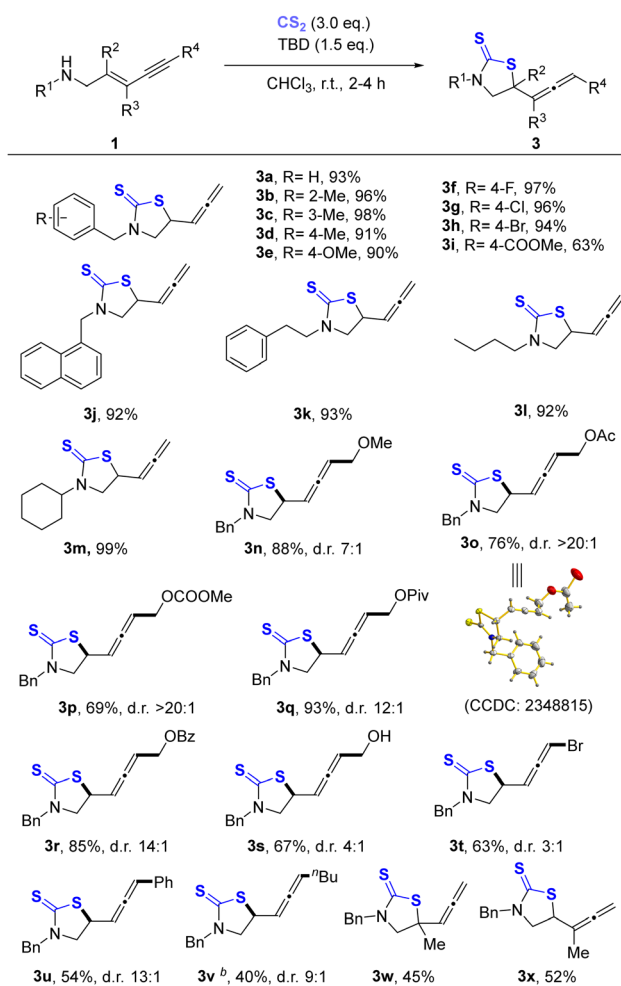
Table 1 Substrate scope for iodoallenyl oxazolidinones<sup>a</sup>



<sup>a</sup> Standard reaction conditions: **1** (0.1 mmol), TBD (0.2 mmol) and toluene:1,4-dioxane (2 mL, 4:1) were stirred for 10 min under a CO<sub>2</sub> atmosphere (1 atm) at rt before NIS (0.12 mmol) was added, then the mixture was stirred for 12–24 h with a CO<sub>2</sub> balloon (1 atm).

the heterocyclic product (**2k**) obtained from the furan-2-methyl-substrate. Besides the benzyl group, other alkyl substituted substrates, especially for enantiomerically pure methyl benzyl amine, could also provide corresponding products (**2l–2o**) in moderate to good yields and dr values. Upon substitution of the alkene near amine by a methyl or an ethyl group, quaternary carbon-containing allenyl oxazolidinones were obtained in very high yields and dr (**2p**, **2q**). In the case of substitution of the olefin away from amine by a methyl group, the product (**2r**) with a trisubstituted iodoallene was afforded in 77% yield and 11:1 dr. Unfortunately, neither aryl-substituted substrates nor strongly electron-withdrawing substituents on the amine generated desirable products. These results suggest that the nucleophilicity of nitrogen is very crucial to this transformation. Mixtures of 1,2-addition and 1,4-addition products were obtained using the internal alkynyl substrates (see the ESI† for details).

Encouraged by the successful iodoallenylation of (*E*)-*N*-benzylpent-2-en-4-yn-1-amine (**1a**) and CO<sub>2</sub>, the model reaction of CS<sub>2</sub> with **1a** was also carefully examined. In contrast, hydrothiolation of 1,3-enyne can proceed smoothly without any activation by electrophilic halogen. It may be due to more suitable nucleophilicity of dithiocarbamate than that of carbamate for this transformation because *S* was generally considered as a softer nucleophile to preferentially react with a comparable electrophile of 1,3-enyne. Desired allenyl thiazolidine-2-thione

Table 2 Substrate scope for allenyl thiazolidine-2-thiones<sup>a</sup>

<sup>a</sup> Standard reaction conditions: to an oven-dried vial equipped with a stir bar were added **1** (0.1 mmol), TBD (0.15 mmol) and CHCl<sub>3</sub> (1 mL), then CS<sub>2</sub> (0.3 mmol) was added to the mixture and stirred for 2–4 h at rt.  
<sup>b</sup> 0.3 mmol substrate and DBU (3.0 eq.) reaction for 12 h.

**3a** was achieved in 96% yield by the combination of 1.5 equiv. TBD and 3.0 equiv. CS<sub>2</sub> in CHCl<sub>3</sub> at room temperature for 4 h. For more optimal details, please see the ESI.<sup>†</sup>

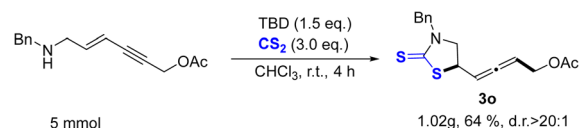
The substrate scope for TBD-promoted hydrothiolation of various 1,3-enynes with CS<sub>2</sub> is explored in Table 2. It shows that most of the changes for benzyl amine substrates were allowed, and the target products (**3a–3h**) were obtained in excellent yields (>90%). However, for the strong electron-withdrawing ester group substituted substrate, the product **3i** was obtained in 63% yield. Besides the benzyl group, other substituents on the nitrogen including the naphthyl methyl, phenylethyl, *n*-butyl or cyclohexyl group provided corresponding products (**3j–3m**) in excellent yields (>90%) as well. In addition, the activated internal alkyne substrates can be well compatible for such hydrothiolation. Disubstituted allenes (**3n–3s**) containing ether, ester and free hydroxyl groups were obtained in moderate to good yields and dr values. Unfortunately, the desired bromoallene **3t** was isolated in 63% yield and 3:1 dr. The

results (**3u**, **3v**) for aryl and alkyl substituted internal alkyne were unsatisfactory, despite extending the reaction time to 12 h, which may be attributed to the decreased electrophilicity and bulky steric hindrance of 1,3-enynes. With the substitution of alkene by the methyl group, the target products (**3w**, **3x**) were obtained in 45% and 52% yields, respectively. This may also be influenced by steric hindrance and electrophilicity of the substrate. Remarkably, some products undergoing 1,2-addition were also observed (see the ESI<sup>†</sup> for details).

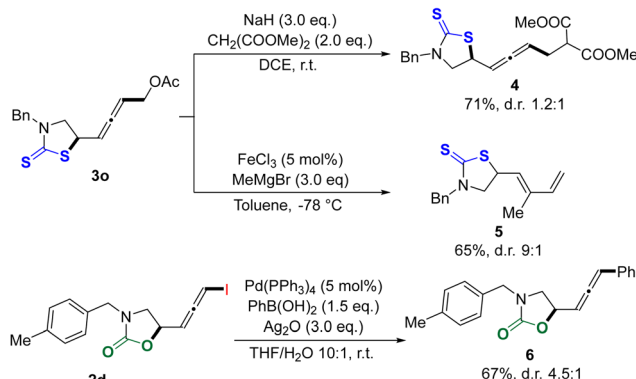
Subsequently, the gram scale synthesis and applicability for such chemical fixation of CO<sub>2</sub>/CS<sub>2</sub> are shown in Scheme 2. As shown in Scheme 2a, allene **3o** was smoothly synthesized by gram scale (1.02 g) in 64% yield with >20:1 dr. To highlight the synthetic utilities of present transformation, several post-modifications of the representative products are shown in Scheme 2b. Allene **3o** containing –OAc as a leaving group could be applied in the Pd-catalyzed Tsuji–Trost reaction with dimethyl malonate as a nucleophile to obtain a new allene **4** with 71% yield and 1.2:1 dr.<sup>20</sup> FeCl<sub>3</sub> could promote the nucleophilic addition to allene **3o** with MeMgBr as a nucleophile to afford thiazolidine-2-thione **5** containing conjugated diene substitution.<sup>21</sup> Moreover, the Suzuki cross-coupling of iodoallene **2d** with phenylboronic acid was also performed to give desired product **6** in 67% yield.<sup>22</sup>

The reaction mechanism is proposed in Scheme 3 on the basis of the current results and literature.<sup>9,12</sup> Initially, the nucleophilic addition of **1a** to the TBD–CO<sub>2</sub> adduct was carried out to obtain the carbamate intermediate **I**, followed by NIS activation of the alkyne moiety of 1,3-enyne to generate iodonium interm **II**. Then, the 1,4-*syn*-addition of 1,3-enyne with carbamate occurred to give the product **2a**. Similarly, **1a** and CS<sub>2</sub> underwent the nucleophilic addition in the presence of TBD to give intermediate **III**. The hydrothiolation reaction would be immediately performed to form product **3a** by 1,4-*syn*-addition of 1,3-enyne with dithiocarbamate.

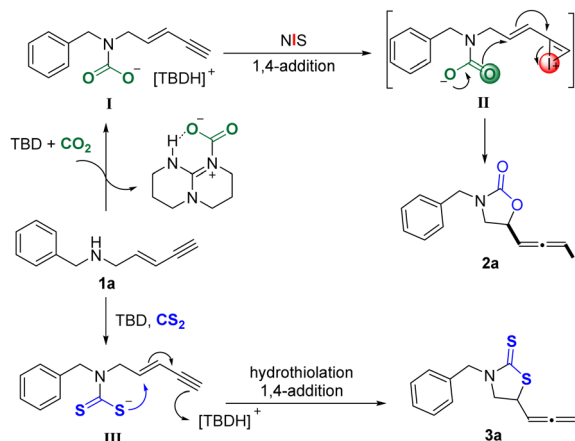
#### (a) Gram-scale preparation of **3o**



#### (b) Post-modification of products



Scheme 2 Synthetic applications and post-modifications.



Scheme 3 Proposed mechanism for chemical fixation of  $\text{CO}_2/\text{CS}_2$ .

In summary, a tandem  $\text{CO}_2/\text{CS}_2$  fixation and cyclization reaction was developed to prepare iodoallenyl oxazolidinones and allenyl thiazolidine-thiones and their derivatives with high yields and diastereoselectivities. The remarkable advantages of this strategy include a metal-free manner, mild reaction conditions, high atom-economy and wide substrate scope. Moreover, the two types of allenyl heterocyclic compounds are valuable building blocks for the synthesis of multifunctional molecules or natural products in the future.

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## Data availability

The data supporting this article have been included as part of the ESI.†

## Conflicts of interest

There are no conflicts to declare.

## References

- (a) T. Sakakura, J. Choi and H. Yasuda, *Chem. Rev.*, 2007, **107**(6), 2365–2387; (b) M. Cokoja, C. Bruckmeier, B. Rieger, W. A. Herrmann and F. E. Kuhn, *Angew. Chem. Int. Ed.*, 2011, **50**(37), 8510–8537; (c) E. S. Sanz-Pérez, C. R. Murdock, S. A. Didas and C. W. Jones, *Chem. Rev.*, 2016, **116**(19), 11840–11876; (d) X. Lu, Springer International Publishing, Switzerland, 2016.
- (a) Q. Liu, L. Wu, R. Jackstell and M. Beller, *Nat. Commun.*, 2015, **6**, 5933–5948; (b) S. Wang and C. Xi, *Chem. Soc. Rev.*, 2019, **48**(1), 382–404; (c) R. Dalpozzo, N. Della Ca, B. Gabriele and R. Mancuso, *Catalysts*, 2019, **9**(6), 511.
- (a) A. Otto, T. Grube, S. Schiebahn and D. Stolten, *Energy Environ. Sci.*, 2015, **8**(11), 3283–3297; (b) M. Aresta, A. Dibenedetto and A. Angelini, *Chem. Rev.*, 2014, **114**(3), 1709–1742.
- (a) M. Heravi and V. Zadsirjan, *Tetrahedron: Asymmetry*, 2013, **24**(19), 1149–1188; (b) C. Foti, A. Piperno, A. Scala and O. Giuffrè, *Molecules*, 2021, **26**(14), 4280–4286; (c) F. Sun, E. V. Van der Eycken and H. Feng, *Adv. Synth. Catal.*, 2021, **363**(23), 5168–5195.
- (a) Z. Zhang, J. Ye, D. Wu, Y. Zhou and D. Yu, *Chem. – Asian J.*, 2018, **13**(17), 2292–2306; (b) C. J. Dinsmore and S. P. Mercer, *Org. Lett.*, 2004, **6**(17), 2885–2888; (c) M. Yoshida, T. Mizuguchi and K. Shishido, *Chemistry*, 2012, **18**(49), 15578–15581; (d) F. Zhou, S. Xie, X. Gao, R. Zhang, C. Wang, G. Yin and J. Zhou, *Green Chem.*, 2017, **19**(16), 908.
- (a) E. García-Egido, I. Fernández and L. Muñoz, *Synth. Commun.*, 2006, **36**(20), 3029–3042; (b) S. Minakata, I. Sasaki and T. Ide, *Angew. Chem., Int. Ed.*, 2010, **49**(7), 1309–1311; (c) J. Ye, L. Song, W. Zhang, T. Ju, Z. Yin, S. S. Yan, Z. Zhang, J. Li and D. Yu, *Angew. Chem., Int. Ed.*, 2016, **55**(34), 10022–10026.
- (a) L. He, X. Dou, Z. Yang and J. Wang, *Synlett*, 2010, (14), 2159–2163; (b) X. Lin, Z. Yang, L. He and Z. Yuan, *Green Chem.*, 2015, **17**(2), 795–798; (c) X. Tian, X. Jiang, S. Hou, Z. Jiao, J. Han and B. Zhao, *Angew. Chem., Int. Ed.*, 2022, **61**(18), e202200123.
- (a) K. Sekine and T. Yamada, *Chem. Soc. Rev.*, 2016, **45**(16), 4524–4532; (b) T. Zhang, H. Zang, F. Gai, Z. Feng, M. Li and C. Duan, *New J. Chem.*, 2020, **44**(35), 15131–15139; (c) B. Lin, Y. Ruan, Q. Hou, Z. Yuan, Y. Liang and J. Zhang, *Org. Biomol. Chem.*, 2024, **22**(27), 5585–5590.
- (a) S. Zhang and L. He, *Aust. J. Chem.*, 2014, **67**(7), 980–988; (b) S. Hajra and A. Biswas, *Org. Lett.*, 2020, **22**(13), 4990–4994; (c) S. Hajra and A. Biswas, *Sci. Rep.*, 2020, **10**, 15825; (d) R. Yousefi, T. J. Struble, J. L. Payne, M. Vishe, N. D. Schley and J. N. Johnston, *J. Am. Chem. Soc.*, 2019, **141**(1), 618–625.
- Z. Yang, L. He, S. Peng and A. Liu, *Green Chem.*, 2010, **12**(10), 1850–1854.
- A. Liu, Y. Dang, H. Zhou, J. Zhang and X. Lu, *ChemCatChem*, 2018, **10**(12), 2686–2692.
- (a) W. Zhang, H. Xu, H. Xu and W. Tang, *J. Am. Chem. Soc.*, 2009, **131**(11), 3832–3833; (b) W. Zhang, N. Liu, C. M. Schienebeck, K. Decloux, S. Zheng, J. B. Werness and W. Tang, *Chem. – Eur. J.*, 2012, **18**(23), 7296–7305.
- (a) R. Blicke, M. Taillefer and F. Monnier, *Chem. Rev.*, 2020, **120**(24), 13545–13598; (b) J. Singh, A. Sharma and A. Sharma, *Org. Chem. Front.*, 2021, **8**(20), 5651–5667; (c) J. M. Alonso and P. Almendros, *Adv. Synth. Catal.*, 2023, **365**(9), 1332–1384.
- (a) A. W. DeMartino, D. F. Zigler, J. M. Fukuto and P. C. Ford, *Chem. Soc. Rev.*, 2017, **46**(1), 21–39; (b) V. Opletalova, J. Dolezel, K. Kralova, M. Pesko, J. Kunes and J. Jampilek, *Molecules*, 2011, **16**(6), 5207–5227.
- (a) H. Medini, N. H. Mekni and K. Boujlel, *J. Sulfur. Chem.*, 2015, **36**(6), 653–659; (b) A. A. Nechaev, A. A. Peshkov, K. Van Hecke, V. A. Peshkov and E. V. Van der Eycken, *Eur. J. Org. Chem.*, 2017, (6), 1063–1069; (c) A. Biswas and S. Hajra, *Adv. Synth. Catal.*, 2022, **364**(17), 3035–3042.
- (a) A. Ziyaei-Halimehjani, K. Marjani and A. Ashouri, *Tetrahedron Lett.*, 2012, **53**(27), 3490–3492; (b) K. Safa and M. Alyari, *Synthesis*, 2015, (02), 256–262.
- (a) M. Anitha and K. C. K. Swamy, *Org. Biomol. Chem.*, 2018, **16**(3), 402–413; (b) Y. Xie, C. Lu, B. Zhao, Q. Wang and Y. Yao, *J. Org. Chem.*, 2019, **84**(4), 1951–1958.
- (a) M. T. C. Ang, L. Phan, A. K. Alshamrani, J. R. Harjani, R. Wang, G. Schatte, N. J. Mosey and P. G. Jessop, *Eur. J. Org. Chem.*, 2015, (33), 7334–7343; (b) K. K. Pandey, *Coord. Chem. Rev.*, 1995, **140**, 37–114.
- (a) D. C. Whitehead, R. Yousefi, A. Jaganathan and B. Borhan, *J. Am. Chem. Soc.*, 2010, **132**(10), 3298–3300; (b) L. Zhou, C. K. Tan, X. Jiang, F. Chen and Y. Y. Yeung, *J. Am. Chem. Soc.*, 2010, **132**(44), 15474–15476; (c) H. Wang, H. Qian, J. Zhang and S. Ma, *J. Am. Chem. Soc.*, 2022, **144**(28), 12619–12626.
- Y. Tang, Q. Yu and S. Ma, *Org. Chem. Front.*, 2017, **4**(9), 1762–1767.
- (a) W. J. Kong, S. N. Kessler, H. Wu and J. E. Backvall, *Org. Lett.*, 2023, **25**(1), 120–124; (b) J. Qin, Z. Xiong, C. Cheng and M. Hu Li, *J. Org. Lett.*, 2023, **25**(51), 9176–9180.
- E. M. Woerly, A. H. Cherney, E. K. Davis and M. D. Burke, *J. Am. Chem. Soc.*, 2010, **132**(20), 6941–6943.