


 Cite this: *Chem. Commun.*, 2024, 60, 3283

 Received 5th January 2024,  
Accepted 24th February 2024

DOI: 10.1039/d4cc00057a

rsc.li/chemcomm

## Asymmetric synthesis of enantioenriched $\alpha$ -allyl esters through Pd(BINAPHANE)-catalysed allylation of disubstituted ketenes†

 Ahmad A. Ibrahim,<sup>b</sup> Stephen C. J. O'Reilly,<sup>a</sup> Margot Bottarel<sup>a</sup> and Nessian J. Kerrigan \*<sup>a</sup>

**Pd<sub>2</sub>dba<sub>3</sub>·CHCl<sub>3</sub> (2.5 mol%)-BINAPHANE (5 mol%) was used to promote the first catalytic enantioselective allylation of disubstituted ketenes to give  $\alpha$ -allyl esters. The ester products were formed in good to excellent yields (61–93% yield for 13 examples, 16 examples in all), with moderate to good enantioselectivity (68–80% ee for 7 examples).**

One of the most important methods for the asymmetric synthesis of carbonyl compounds possessing an  $\alpha$ -stereogenic centre is the transition metal-catalysed asymmetric allylic alkylation of enolates or related nucleophiles (AAA reaction).<sup>1</sup> Stabilised 'soft' enolates have been found to be most successful as nucleophiles and the reaction has generally performed best with pronucleophiles such as cyclic  $\beta$ -ketoesters, rather than simple acyclic substrates (esters, acids, ketones, aldehydes, ketenes).<sup>1</sup> A new catalytic allylation method for such substrates would provide a powerful tool for C–C bond formation. Recently, Carreira's group and Dong's group independently demonstrated the stereodivergent synthesis of  $\alpha,\beta$ -substituted chiral aldehydes through dual catalysis involving chiral Ir(i) complex/chiral amine or chiral Rh(i) complex/chiral amine-catalysed reaction of racemic aldehydes with allylic alcohols and internal alkynes respectively.<sup>2,3</sup> Zhang's group and List's group also independently showed that *in situ*-generated enamines could be used as an alternative to enolates for enantioselective allylations of ketones and aldehydes.<sup>4</sup> Snaddon and co-workers reported the  $\alpha$ -allylation of aryl acetic acid esters to form  $\alpha$ -allylated esters bearing a tertiary chiral centre using cooperative Pd(0)/isothiourea catalysis, while Hartwig reported a chiral Ir(i)/isothiourea-catalysed stereodivergent variant of the latter reaction to form  $\alpha,\beta$ -substituted chiral esters.<sup>5,6</sup> In 2017, Ding and

co-workers communicated the asymmetric allylic alkylation of ketoesters to give products bearing vicinal tertiary and quaternary chiral centres.<sup>7</sup> However the latter method was restricted to cyclic  $\beta$ -ketoesters as substrates.

Previously, in 1986, Watanabe's group had reported the non-enantioselective  $\alpha$ -allylation of disubstituted ketenes to afford dienes or allylated esters, often obtained as mixtures of products or in low yields.<sup>8</sup> Recently our group, as part of a program of studies on the development of new reactions of ketenes, reported the Pd(0)-catalysed stereospecific reaction of enantioenriched vinyl cyclopropanes with ketenes to provide access to enantioenriched tetrahydrofurans.<sup>9</sup> However, there has been no report of a direct catalytic enantioselective allylation of ketene-derived enolates to give  $\alpha$ -allylated esters bearing an  $\alpha$ -quaternary stereogenic centre.<sup>10</sup> In this communication we describe our initial results toward that goal.

We began our studies by reinvestigating the work of Watanabe's group using Pd(PPh<sub>3</sub>)<sub>4</sub> as catalyst to promote the reaction of ethylphenylketene with allyl carbonate or allyl acetate in THF. Interestingly, in our hands only  $\alpha$ -allylated carbonyl products (*e.g.* **3a** and **3b**) were formed, with no diene **4**, in contrast to Watanabe's observations in THF (Scheme 1).<sup>8</sup>

We proposed that an aryl oxide (leaving group/counterion) generated during Pd( $\eta$ )- $\pi$ -allyl **5** formation would act as a better nucleophile than acetate or carbonate and add to the ketene to form an ester enolate **6** in stereoselective fashion (Scheme 2). We anticipated that ketene dimerisation would be minimised under the reaction conditions where the putative ester enolate would be stabilised by the Pd( $\eta$ )- $\pi$ -allyl species (perhaps through inner sphere coordination of enolate to Pd).<sup>9</sup> The ester enolate **6** would then undergo C-allylation by the associated Pd( $\eta$ )- $\pi$ -allyl species to provide the desired allyl ester product **3** (as the linear regioisomer) along with simultaneous Pd(0) regeneration (Scheme 2). The use of chiral phosphine ligands on Pd would be expected to control enantioselectivity in formation of the new  $\alpha$ -quaternary stereogenic centre in **3**.

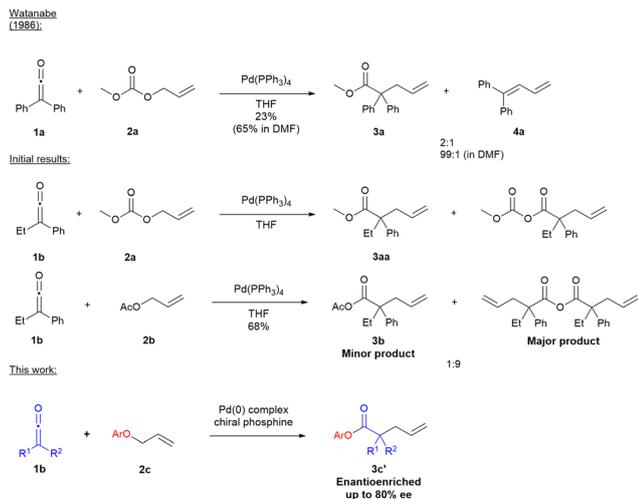
Inspired by Lectka's work on  $\alpha$ -halogenation of ketenes where an aryl oxide was used in a rebound mechanism to

<sup>a</sup> School of Chemical Sciences and Life Sciences Institute, Dublin City University, Glasnevin, Dublin 9, Ireland. E-mail: nessian.kerrigan@dcu.ie

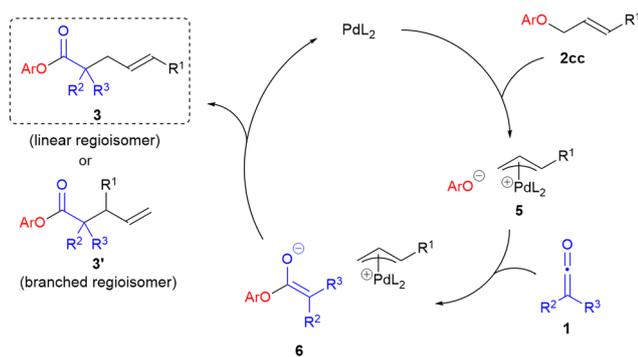
<sup>b</sup> Department of Chemistry, Oakland University, 2200 N. Squirrel Rd, Rochester, MI 48309, USA

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d4cc00057a>





Scheme 1 Precedent and initial results.

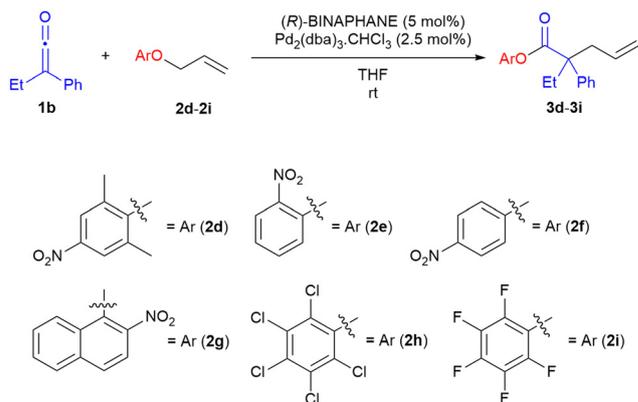


Scheme 2 Proposed reaction mechanism.

regenerate nucleophilic catalyst for the catalytic cycle, we examined a number of allyl aryl ethers bearing electron withdrawing substituents on the aryl moiety.<sup>11</sup> We determined that allyl pentafluorophenoxide gave best results in terms of yield of **3** and ee when Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub>-BINAPHANE was used as catalyst (Table 1, entry 6). To our delight, ketene dimerisation was not found to proceed to any great extent. The lack of ketene dimer implied that phosphonium enolate formation, formed through addition of dissociated BINAPHANE ligand to ketene, was not a significant reaction pathway (we have previously demonstrated that BINAPHANE is an excellent catalyst for ketene dimerisation and related cycloadditions).<sup>12</sup>

With the desired allyl ester product being favoured, we examined, in parallel, other chiral phosphines in combination with Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> in order to attempt to enhance enantioselectivity in the reaction. However, surprisingly, BINAPHANE (and BINAPINE) was found to be virtually unique in providing enantioselectivity of >30% ee, as 26 other chiral ligands in our study were found to provide ee generally <10% (Table 2, see ESI† for full details). Axial chirality associated with the phosphine skeleton appeared to be essential for enantioselectivity, with the only ligands providing ee of ≥20% being phosphines (Table 2 entries 8–10). Decreasing temperature of the Pd-BINAPHANE

Table 1 Optimization of aryl ether structure for Pd(0)-catalysed ketene allylation reaction



| Entry | Ar        | Yield <sup>a</sup> [%] | ee <sup>b</sup> [%] | <b>3</b>  |
|-------|-----------|------------------------|---------------------|-----------|
| 1     | <b>2d</b> | 50                     | 0                   | <b>3d</b> |
| 2     | <b>2e</b> | 80                     | 27                  | <b>3e</b> |
| 3     | <b>2f</b> | 50                     | 1                   | <b>3f</b> |
| 4     | <b>2g</b> | 51                     | 0                   | <b>3g</b> |
| 5     | <b>2h</b> | 47                     | 27                  | <b>3h</b> |
| 6     | <b>2i</b> | <b>90</b>              | <b>34</b>           | <b>3i</b> |

<sup>a</sup> Isolated yield after flash column chromatography through silica gel.  
<sup>b</sup> ee for **3d–3i** determined by chiral HPLC analysis or GC-MS analysis of diastereomeric derivatives.

catalysed reaction did not have any benefit on enantioselectivity, with reaction efficiency greatly reduced at –78 °C (Table 2, entry 12). THF proved to be the optimal reaction solvent, as employment of CH<sub>2</sub>Cl<sub>2</sub> and toluene led to slightly lower yield and ee, and in DMF the reaction was found not to favour allylated ester at all, highlighting the sensitivity of the ketene allylation to reaction conditions. Interestingly, Watanabe's group had found DMF to be a good solvent for the Pd(PPh<sub>3</sub>)<sub>4</sub>-catalysed allylation of diphenylketene with allyl phenyl ether.<sup>8</sup> When we investigated allyl phenyl ether as substrate, under the conditions of Table 2 entry 8, a complex mixture resulted.

With the optimal catalytic system in hand we then proceeded to explore the substrate scope of the reaction, examining variations of the ketene structure (R<sup>1</sup>, R<sup>2</sup> = aryl or alkyl), and of the allyl ether substitution pattern (Table 3). Most reactions proceeded, after purification, to afford the highly non-polar allyl ester products in good to excellent yields (61–93% for 13 examples) and purity, with some examples contaminated by non-polar byproducts/impurities, such as branched regioisomer (**3'**, Scheme 2).

Significantly, it was noted that good to excellent regioselectivity (rs) was displayed in all cases (rs 8:1–>32:1), favouring formation of the linear, less substituted product rather than the branched, more highly substituted product, e.g. rs 24:1 for **3j**; rs >32:1 for **3m**; rs 10:1 for **3t**; rs 32:1 for **3x**. The regioselectivity was interpreted in terms of preferential nucleophilic addition of ester enolate to the less sterically hindered end of the Pd(II)-π-allyl intermediate in **6** (Scheme 2). This outcome is in agreement with previously observed trends in most regioselectivity studies of Pd(0)-catalysed allylic alkylations of enolate species.<sup>1</sup>



**Table 2** Exploration of chiral ligands and reaction conditions for Pd(0)-catalysed ketene allylation reaction

| Entry | Chiral ligand  | Solvent                         | Temp.     | Yield <sup>a</sup> [%] | ee <sup>b</sup> [%] |
|-------|--|---------------------------------|-----------|------------------------|---------------------|
| 1     | ( <i>R</i> )-BINAP   | THF                             | rt        | 98                     | 4                   |
| 2     | ( <i>S,S</i> )-DACH naphthyl-Trost                           | THF                             | rt        | 93                     | 5                   |
| 3     | ( <i>R,R</i> )-ANDEN phenyl-Trost                            | THF                             | -25 °C    | 73                     | 5                   |
| 4     | (1 <i>R</i> ,1' <i>R</i> ,2 <i>S</i> ,2' <i>S</i> )-DUANPHOS | THF                             | rt        | 80                     | 3                   |
| 5     | ( <i>R,R</i> )-DIPAMP  | THF                             | rt        | 71                     | 8                   |
| 6     | ( <i>R</i> )-Cl-MeO-BIPHEP                                   | THF                             | rt        | 96                     | 7                   |
| 7     | ( <i>R</i> )-PHANEPHOS                                       | THF                             | rt        | 88                     | 7                   |
| 8     | <b>(<i>R</i>)-BINAPHANE</b>                                  | <b>THF</b>                      | <b>rt</b> | <b>90</b>              | <b>34</b>           |
| 9     | ( <i>S</i> )- <i>f</i> -BINAPHANE                            | THF                             | rt        | 79                     | -27                 |
| 10    | ( <i>S</i> )-BINAPINE  | THF                             | rt        | 86                     | -34                 |
| 11    | ( <i>R</i> )-BINAPHANE                                       | THF                             | -25 °C    | 75                     | 37                  |
| 12    | ( <i>R</i> )-BINAPHANE                                       | THF                             | -78 °C    | 13                     | -35                 |
| 13    | ( <i>R</i> )-BINAPHANE                                       | CH <sub>2</sub> Cl <sub>2</sub> | rt        | 72                     | 32                  |
| 14    | ( <i>R</i> )-BINAPHANE                                       | Toluene                         | rt        | 70                     | 32                  |
| 15    | ( <i>R</i> )-BINAPHANE                                       | DMF                             | rt        | <1                     | nd                  |

<sup>a</sup> Isolated yield after flash column chromatography through silica gel or iatrobeads. <sup>b</sup> ee for **3i** determined by chiral HPLC analysis (OD-H) or by derivatisation with (*S*)- $\alpha$ -methylbenzylamine and GC-MS analysis (ratio of diastereomers to determine ee).

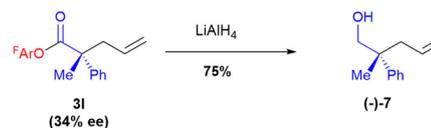
Gratifyingly, we generally found that with aryl-substituted allyl ethers ( $R^4 = \text{aryl}$ ) an increase in enantioselectivity to a good level (67–80% ee) was observed. A range of ketenes (seven in all) such as methylphenylketene, ethylphenylketene, *i*-butylphenylketene, methyl-*i*-propylketene, diphenylketene and alkylarylketenes bearing electron donating groups on the aryl ring performed well from a reactivity standpoint, albeit with quite varying effects on enantioselectivity (3–80% ee).<sup>13,14</sup> An *ortho*-donating group, in particular, was found to be detrimental to enantioselectivity, with an ee of 14% (*versus* 70% ee for no *ortho*-substituent, Table 3 entry 13 *vs.* entry 5) being obtained, and this was ascribed to the increase in steric bulk associated with the *ortho*-substituent leading to a mixture of enolate isomers from reversible addition of  $\text{ArO}^-$  to the ketene. The greatest influence on enantioselectivity was noted for the presence of an aryl substituent ( $R^4 = \text{Ph}$ ) on the  $\beta$ -allylic carbon of the allyl aryl ether (Table 3 entry 2 *vs.* entry 1 or entry 4 *vs.* entry 5), with an increase in ee of >30% compared to the unsubstituted case ( $R^4 = \text{H}$ ). The presence of a *para*-electron withdrawing group (*e.g.*  $\text{NO}_2$ ) on the  $R^4$  aryl group led to a further 5–10% increase in enantioselectivity (*e.g.* Table 3, entry 6 *vs.* entry 5). On the other hand, an *ortho*-electron withdrawing group on the  $R^4$  aryl substituent effectively shut down the reaction (Table 3, entry 7).

The absolute stereochemistry of **3i** was determined to be (*R*) by comparison of specific rotation value for the derived alcohol **7** with the literature value reported by Kanai and co-workers.<sup>5d</sup>

**Table 3** Substrate scope for Pd(BINAPHANE)-catalysed ketene allylation reaction

| Entry           | R <sup>1</sup> | R <sup>2</sup>                    | R <sup>3</sup>     | R <sup>4</sup>                                  | Yield <sup>a</sup> [%] | ee <sup>b</sup> [%] | 3 <sup>c</sup> |
|-----------------|----------------|-----------------------------------|--------------------|---|------------------------|---------------------|----------------|
| 1               | Et             | Ph                                | H                  | H   | 90 <sup>d</sup>        | 34                  | <b>3i</b>      |
| 2               | Et             | Ph                                | H                  | Ph  | 82                     | 70                  | <b>3j</b>      |
| 3               | Et             | Ph                                | H                  | 4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> | 89                     | 75                  | <b>3k</b>      |
| 4 <sup>d</sup>  | Me             | Ph                                | H                  | H   | 65                     | 34                  | <b>3l</b>      |
| 5               | Me             | Ph                                | H                  | Ph  | 43                     | 70                  | <b>3m</b>      |
| 6               | Me             | Ph                                | H                  | 4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> | 85                     | 80                  | <b>3n</b>      |
| 7               | Me             | Ph                                | H                  | 2-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> | <10                    | nd                  | <b>3o</b>      |
| 8               | Me             | Ph                                | H                  | CO <sub>2</sub> Me                              | 75                     | 79                  | <b>3p</b>      |
| 9 <sup>d</sup>  | Me             | Ph                                | CO <sub>2</sub> Et | H   | 68                     | 44                  | <b>3q</b>      |
| 10 <sup>e</sup> | Me             | Ph                                | Me                 | H   | 72                     | —                   | <b>3r</b>      |
| 11 <sup>d</sup> | <i>i</i> -Bu   | Ph                                | H                  | H   | 54                     | 41                  | <b>3s</b>      |
| 12              | <i>i</i> -Bu   | Ph                                | H                  | Ph  | 92                     | 68                  | <b>3t</b>      |
| 13              | Me             | 2-MeC <sub>6</sub> H <sub>4</sub> | H                  | Ph  | 77                     | 14                  | <b>3u</b>      |
| 14              | Me             | 4-MeO-Naphthyl                    | H                  | Ph  | 77                     | 69                  | <b>3v</b>      |
| 15 <sup>d</sup> | Me             | <i>i</i> -Pr                      | H                  | 4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub> | 93                     | 3                   | <b>3w</b>      |
| 16 <sup>e</sup> | Me             | <i>i</i> -Pr                      | H                  | CO <sub>2</sub> Me                              | 39                     | —                   | <b>3x</b>      |
| 17 <sup>e</sup> | Ph             | Ph                                | H                  | H   | 61                     | —                   | <b>3y</b>      |

<sup>a</sup> Isolated yield after flash column chromatography through silica gel or neutral silica (iatrobeads). <sup>b</sup> ee for **3** determined by chiral HPLC analysis (OD-H, AD-H and AD) or by derivatisation with (*S*)- $\alpha$ -methylbenzylamine and GC-MS analysis (ratio of diastereomers to determine ee). <sup>c</sup> Major regioisomer = linear isomer for all examples ( $r_s \geq 8:1$ ). <sup>d</sup> When  $\text{Pd}(\text{PPh}_3)_4$  (5 mol%) was employed: 51% yield for ( $\pm$ )-**3i**; 75% for ( $\pm$ )-**3l**; 71% for ( $\pm$ )-**3q**; 87% for ( $\pm$ )-**3s**; 77% for ( $\pm$ )-**3w**. <sup>e</sup>  $\text{Pd}(\text{PPh}_3)_4$  (5 mol%) used as catalyst.

**Scheme 3** Synthetic utility of ester products: elaboration to chiral alcohol.

The absolute configuration for all other examples were assigned to be (*R*) by analogy (Scheme 3).

In conclusion, we have developed a Pd(BINAPHANE)-catalysed enantioselective synthesis of allyl aryl esters, bearing an  $\alpha$ -quaternary centre, from allyl aryl ethers and disubstituted ketenes in good to excellent yields (up to 93%) and with moderate to good enantioselectivity (up to 80% ee). Future studies will seek to improve enantioselectivity through cooperative chiral Lewis base-Pd(L<sup>\*</sup>) catalysis and to explore the scope of the reaction with respect to *in situ* generated ketenes.<sup>5a,13e</sup>

Support has been provided by the RSC Research Enablement Grant (E22-5472593338) to N.J.K. Open access funding provided by IREL.

## Conflicts of interest

There are no conflicts to declare.



## Notes and references

- General Pd-catalysed allylic alkylation references: (a) O. Pamies, J. Margalef, S. Canellas, J. James, E. Judge, P. J. Guiry, C. Moberg, J.-E. Backvall, A. Pfaltz, M. A. Pericas and M. Dieguez, *Chem. Rev.*, 2021, **121**, 4373; (b) A. Y. Hong and B. M. Stoltz, *Eur. J. Org. Chem.*, 2013, 2745; (c) J. D. Weaver, A. Recio, III, A. J. Grenning and J. A. Tunge, *Chem. Rev.*, 2011, **111**, 1846; (d) B. M. Trost and M. L. Crawley, *Chem. Rev.*, 2003, **103**, 2921; (e) B. M. Trost and D. L. Van Vranken, *Chem. Rev.*, 1996, **96**, 395.
- S. Krautwald, D. Sarlah, M. A. Schafroth and E. M. Carreira, *Science*, 2013, **340**, 1065.
- F. A. Cruz and V. M. Dong, *J. Am. Chem. Soc.*, 2017, **139**, 1029.
- (a) X. Zhao, D. Liu, H. Guo, Y. Liu and W. Zhang, *J. Am. Chem. Soc.*, 2011, **133**, 19354; (b) G. Jiang and B. List, *Angew. Chem., Int. Ed.*, 2011, **50**, 9471.
- Catalytic enantioselective allylation of esters and acids: (a) K. J. Schwarz, J. L. Amos, J. C. Klein, D. T. Do and T. N. Snaddon, *J. Am. Chem. Soc.*, 2016, **138**, 5214; (b) H.-C. Lin, G. J. Knox, C. M. Pearson, C. Yang, V. Carta and T. N. Snaddon, *Angew. Chem., Int. Ed.*, 2022, **61**, e202201753; (c) R. Visse, M.-A. Mollemann and M. Braun, *Eur. J. Org. Chem.*, 2019, 4604; (d) T. Fujita, T. Yamamoto, Y. Morita, H. Chen, Y. Shimizu and M. Kanai, *J. Am. Chem. Soc.*, 2018, **140**, 5899; (e) M. Braun, P. Meletis and R. Visse, *Adv. Synth. Catal.*, 2011, **353**, 3380.
- X. Jiang, J. J. Beiger and J. F. Hartwig, *J. Am. Chem. Soc.*, 2017, **139**, 87.
- J. Liu, Z. Han, X. Wang, F. Meng, Z. Wang and K. Ding, *Angew. Chem., Int. Ed.*, 2017, **56**, 5050.
- (a) T.-a Mitsudo, M. Kadokura and Y. Watanabe, *J. Chem. Soc., Chem. Commun.*, 1986, 1539; (b) T.-a Mitsudo, M. Kadokura and Y. Watanabe, *J. Org. Chem.*, 1987, **52**, 1695.
- (a) M. Mondal, M. Panda, V. McKee and N. J. Kerrigan, *J. Org. Chem.*, 2019, **84**, 11983; (b) M. Mondal, M. Panda, N. W. Davis, V. McKee and N. J. Kerrigan, *Chem. Commun.*, 2019, **55**, 13558; (c) M. Mondal, S. Mitra, D. J. Twardy, M. Panda, K. A. Wheeler and N. J. Kerrigan, *Chem. – Eur. J.*, 2022, **28**, e202104391; (d) M. Mondal, S. Chen, N. Othman, K. A. Wheeler and N. J. Kerrigan, *J. Org. Chem.*, 2015, **80**, 5789; (e) S. Chen, A. A. Ibrahim, M. Mondal, A. J. Magee, A. J. Cruz, K. A. Wheeler and N. J. Kerrigan, *Org. Lett.*, 2015, **17**, 3248.
- (a) A. D. Allen and T. T. Tidwell, *Arkivoc*, part (i), **415**, 2016; (b) S. Chen, E. C. Salo and N. J. Kerrigan, in *Science of Synthesis Reference Library*, Asymmetric Organocatalysis, Lewis Base and Acid Catalysts, ed. B. List, Thieme, Stuttgart, 2012, vol. 1, ch. 1.1.10, p. 455; (c) D. H. Paull, A. Weatherwax and T. Lectka, *Tetrahedron*, 2009, **65**, 6771; (d) T. T. Tidwell, *Ketenes*, Wiley, New York, 2nd edn, 2006; (e) S. Takeuchi, N. Miyoshi and Y. Ohgo, *Chem. Lett.*, 1992, 551; (f) M. M. Li, Y. Wei, J. Liu, H. W. Chen, L. Q. Lu and W. J. Xiao, *J. Am. Chem. Soc.*, 2017, **139**(41), 14707.
- (a) H. Wack, A. E. Taggi, A. M. Hafez, W. J. Drury and T. Lectka, *J. Am. Chem. Soc.*, 2001, **123**, 1531; (b) J. Erb, D. H. Paull, L. Belding, T. Dudding and T. Lectka, *J. Am. Chem. Soc.*, 2011, **133**, 7536.
- (a) A. A. Ibrahim, P.-H. Wei, G. D. Harzmann, D. Nalla, M. Mondal, K. A. Wheeler and N. J. Kerrigan, *Tetrahedron*, 2021, **78**, 131838; (b) S. Chen, M. Mondal, A. A. Ibrahim, K. A. Wheeler and N. J. Kerrigan, *J. Org. Chem.*, 2014, **79**, 4920.
- Preparation of ketenes: (a) B. L. Hodous and G. C. Fu, *J. Am. Chem. Soc.*, 2002, **124**, 10006; (b) S. L. Wiskur and G. C. Fu, *J. Am. Chem. Soc.*, 2005, **127**, 6176; (c) A. D. Allen, L. M. Baigrie, L. Gong and T. T. Tidwell, *Can. J. Chem.*, 1991, **69**, 138; (d) J. E. Wilson and G. C. Fu, *Angew. Chem., Int. Ed.*, 2004, **43**, 6358; (e) S. Chen, A. A. Ibrahim, N. J. Peraino, D. Nalla, M. Mondal, M. Van Raaphorst and N. J. Kerrigan, *J. Org. Chem.*, 2016, **81**, 7824; (f) J. J. Douglas, G. Churchill, A. M. Z. Slawin, D. J. Fox and A. D. Smith, *Chem. – Eur. J.*, 2015, **21**, 16354; (g) M. Panda, M. Mondal, S. Chen, A. A. Ibrahim, D. J. Twardy and N. J. Kerrigan, *Eur. J. Org. Chem.*, 2020, 5752.
- Ketenes bearing electron-withdrawing groups (e.g. 4-CF<sub>3</sub>) on the aryl substituent were found to provide good reactivity (up to 64% yield), but enantioselectivity was not determined due to a lack of resolution on a variety of chiral HPLC columns.

