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Pd(II)-Catalyzed enantioselective C–H olefination toward the synthesis of *P*-stereogenic phosphinamides[†]

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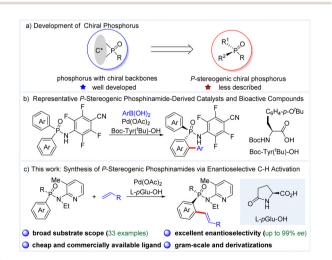
P-Stereogenic phosphorus compounds are important structural elements in chiral ligands or organocatalysts. Herein, we report a Pd(II)-catalyzed enantioselective C-H olefination toward the synthesis of *P*-stereogenic phosphinamides using cheap commercially available \bot -*p*Glu-OH as a chiral ligand. A broad range of *P*-stereogenic phosphinamides were gained in good yields with high enantioselectivities (33 examples, up to 77% yield, 99% ee) *via* desymmetrization and kinetic resolution.

Chiral phosphorus compounds play an important role in both laboratory and industrial synthesis due to their extensive applications in asymmetric synthesis as chiral ligands or organocatalysts.¹ However, most of them have chirality located in their backbones, such as a chiral axis, planar-chirality, or a stereogenic carbon center (Scheme 1a). Although better chiral induction may be gained during the transformation because the stereogenic phosphorus atom is coordinated in closer proximity to the catalytic center,² the application of *P*-stereogenic phosphorus compounds is less described due to the lack of general and efficient methods for their enantioselective preparation.³ Classical synthetic methods usually rely on resolution processes or the use of chiral auxiliaries.⁴ In recent years, more and more approaches including desymmetrization reactions⁵ and catalytic asymmetric methods have been developed.⁶

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In the past decade, transition metal-catalyzed asymmetric C-H functionalization has been developed rapidly and proved to be a powerful strategy to access chiral building blocks from simple starting materials.7 In particular, Pd(II)/mono-N-protected amino acid (MPAA)-catalyzed enantioselective C-H activation developed by Yu and coworkers has received extensive attention for the diverse reactivities of palladacycles and the use of cheap and commercially available ligands.⁸ In 2015, Han and coworkers successfully applied this catalytic system for the synthesis of P-stereogenic phosphinamides using Boc-Tyr(^tBu)-OH as a chiral ligand (Scheme 1b).9 Phosphinamide bearing a 2,3,5,6tetrafluoro-4-cyanophenylamino (Ar^F) directing group was important for this transformation. And it also should be noted that the arylation products could be used as chiral organocatalysts in the desymmetric enantioselective reduction of cyclic 1,3-diketones.¹⁰ Since then, efficient synthesis of P-stereogenic phosphorus compounds via asymmetric C-H functionalization has been developed enabled by different catalytic systems.11 However, until



Scheme 1 Asymmetric synthesis of *P*-stereogenic phosphinamides *via* enantioselective C–H activation.

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recently, another Pd(π)/MPAA-catalyzed enantioselective C–H activation towards the synthesis of chiral phosphorus compounds was reported by our group using cheap and commercially available ι -pyroglutamic acid (ι -*p*Glu-OH) as a chiral ligand.¹² A newly developed *N*-ethyl-*N*-(3-methylpyridin-2-yl)amino directing group was crucial for the reactivity. Herein, we report the synthesis of *P*-stereogenic phosphinamides *via* Pd(π)-catalyzed enantioselective C–H olefination (Scheme 1c). A broad range of *P*-stereogenic phosphinamides, up to 77% yield and 99% ee) *via* desymmetrization and kinetic resolution. The *P*-stereogenic phosphinamide products could be easily transformed to *P*-chiral phosphine oxides.

We initiated our investigation by optimizing the Pd(II)catalyzed C-H olefination of phosphoramide 1a with butyl acrylate 2a. To our delight, the product 3a was obtained in 54% yield with 93% ee in the presence of $Pd(OAc)_2$ (10 mol%), L-pGlu-OH (20 mol%), Ag₂CO₃ (0.05 mmol), and benzoquinone (0.1 mmol) in tBuOH for 12 h under air (Table 1). Next, we screened the solvents such as PhCl, dioxane and t-AmylOH, and t-AmylOH gave the desired product in a higher yield and enantioselectivity. Other silver salts were also investigated, and no better result was observed (entries 5-7). Increasing the temperature to 70 °C or decreasing the temperature to 50 °C resulted in a lower yield (entries 8 and 9). Control experiments showed that L-pGLu-OH and silver salt were crucial to this reaction (entries 10 and 11). In fact, the phosphinamide bearing a 2,3,5,6-tetrafluoro-4-cyanophenylamino directing group (Ar^F), which was used in Han's work, could not afford the olefination product (entry 12).

With the optimal reaction conditions in hand, we then investigated the scope of this enantioselective transformation (Table 2). Diaryl-phosphinamides with electron-donating groups

Table 1 Optimization of the reaction conditions ^a					
	P ^P N _E t 1a	+ CO2 ⁿ E	L- [Ag BC	Ac) ₂ (10 mol%) p/Glu-OH] (2.5 equiv) Q (1.0 equiv) Veent, T, 12 h	Me N N Et CO ₂ "Bu 3a
Entry	Solvent	[Ag]	$T(^{\circ}C)$	Yield of $3a^{b}$ (%)	ee of 3a ^c (%)
1	t-BuOH	Ag_2CO_3	60	54	93
2	PhCl	Ag_2CO_3	60	20	81
3	Dioxane	Ag_2CO_3	60	63	89
4	t-AmylOH	Ag_2CO_3	60	66	95
5	t-AmylOH	Ag_2SO_4	60	32	92
6	t-AmylOH	AgOAc	60	41	92
7	t-AmylOH	Ag_3PO_4	60	47	95
8	t-AmylOH	Ag_2CO_3	50	43	94
9	t-AmylOH	Ag_2CO_3	60	57	94
10	t-AmylOH	No [Ag]	60	10	53
11^d	t-AmylOH	Ag_2CO_3	10	4	0
12^e	t-AmylOH	Ag_2CO_3	70	nr	_

^{*a*} Reaction conditions: **1a** (0.1 mmol), **2a** (0.2 mmol), $Pd(OAc)_2$ (0.01 mmol), [Ag] (0.05 mmol), L-pGlu-OH (0.02 mmol), BQ (0.1 mmol), solvent (1.0 mL), air for 12 h. ^{*b*} Isolated yield. ^{*c*} The ee was determined by chiral HPLC. ^{*d*} Without L-pGlu-OH. ^{*e*} 2,3,5,6-Tetrafluoro-4-cyanophenylamino (Ar^F) directing group was used.

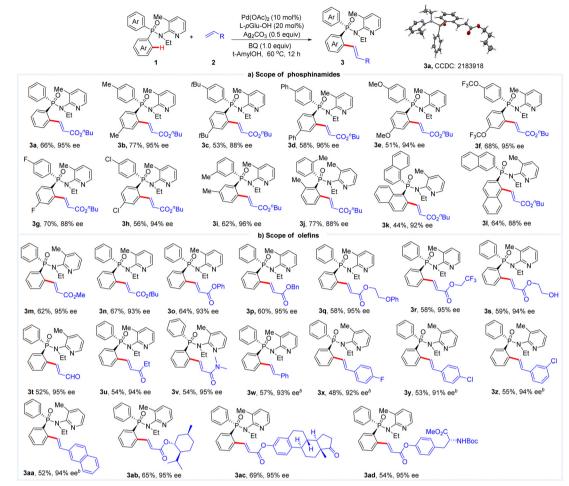
(Me, tBu, Ph and OMe) or electron-withdrawing groups (OCF₃, F, and Cl) on the para-position gave the corresponding olefination products in moderate to good yields with high enantioselectivities (3b-3h, 50-77% yield, 88-96% ee). meta-Me diarylphosphinamide 1i gave the corresponding product 3i in 62% yield with 96% ee. Diaryl-phosphinamides with steric properties were also tolerated well, and ortho-Me, 1-naphthyl phosphinamide and 2-naphthyl substituted phosphinamide afforded the respective olefination products in good yields with good to high enantioselectivities (3j-3l, 88-92% ee). The scope of the coupling partner, olefins, was also evaluated (Table 2b). A broad range of acrylates bearing Me, tBu, Ph, Bn and hydroxyethyl worked well, giving the desired olefination products in good yields with high ee values (3m-3q, 93-95% ee). In addition, acrylates with useful functional groups, trifluoromethyl and hydroxyl, were still compatible and gave 3r in 60% yield with 95% ee and 3s in 59% yield with 94% ee. Other electronically biased olefins, such as acrylaldehyde, α , β unsaturated ketone, and acrylamide proceeded well to give the corresponding products in good results (3t-3v, 54-58% yield, 94-95% ee). Various styrenes also reacted smoothly using Ag₂SO₄ as an oxidant, affording the desired products in moderate yields with high enantioselectivities (3w-3aa, 48-57% yield, 91-94% ee). What's more, chiral phosphinamide bearing core structures of natural products, including 1-menthol, tyrosine and estrone, were also obtained in good yields with high enantioselectivities (3ab-3ad, 54–69% yield, 95% ee). The absolute configuration of 3a was ascribed by X-ray diffraction analysis and extrapolated to the other products.13

To our delight, kinetic resolution was also compatible under the same conditions (Table 3). Racemic phosphinamides *rac*-4 with Me, Et and iPr substituted afforded the olefination products (R)-5 and unreacted (S)-4 in good to excellent enantioselectivities.

We then conducted gram-scale preparation and derivatizations to demonstrate the synthetic utility of this protocol. The olefination of **1a** on a 3.0 mmol scale with **2a** afforded **3a** in 66% yield (920.8 mg) with 90% ee (see the ESI† for details). The directing group of olefination product **3w** could be removed in TfOH, giving product **6w** in 72% yield with 92% ee (Scheme 2). The reaction of **6w** with Grignard reagents afforded the desired chiral phosphine oxides **7a** to **7d** in good yields with minimum enantiomeric loss.

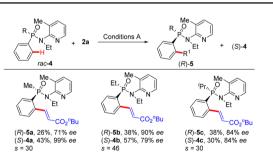
Several experiments were conducted to gain mechanistic insights. The H/D exchange experiment was carried out firstly. The recovered **1a** was incorporated with no deuterium when **1a** reacted in the absence of **2a**. The kinetic isotope effect (KIE) experiments were then conducted and a KIE value of $k_{\rm H}/k_{\rm D}$ = 2.38 was obtained. These results indicate that the C–H cleavage step is irreversible and likely the rate-determining step (see the ESI† for details). We then revealed the chiral induction model through DFT calculations. The C–H bond activation transition states of the two phenyl groups (colored by blue and black) of phosphonamide are located (Fig. S2a, ESI†). In TS2-*R*, there is a significant steric repulsion between the methyl group on the directing group and the phenyl group of phosphamide, so the favorable conformer for generating the *R* enantiomer is TS1-*R*. In the transition states that generate the *S* enantiomer, due to

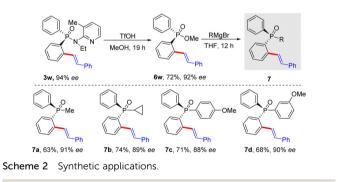
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^a Reaction conditions: 1 (0.1 mmol), 2 (0.2 mmol), Pd(OAc)₂ (0.01 mmol), L-pGlu-OH (0.02 mmol), Ag₂CO₃ (0.05 mmol), BQ (0.1 mmol), t-AmylOH (1.0 mL) for 12 h at 60 °C in a sealed reaction tube, isolated yield. ^b Ag₂SO₄ (0.05 mmol) was used.

Table 3 Kinetic resolution^a





а Reaction conditions: rac-4 (0.2 mmol), 2a (0.4 mmol), Pd(OAc)₂ (0.02 mmol), L-pGlu-OH (0.04 mmol), Ag₂CO₃ (0.1 mmol), BQ (0.2 mmol), t-AmylOH (1.0 mL) at 60 °C for 12 h under air in a sealed reaction tube, isolated yield. Selectivity factors: $s = \ln[(1 - C)(1 - ee_4)]/\ln[(1 - C)(1 + ee_4)]/\ln[(1 - C$ (ee_4)], $C = (ee_4)/(ee_4 + ee_5)$.

the favorable π - π stacking in TS1-S, its energy is 1.6 kcal mol⁻¹ lower than that of TS2-S. Therefore, the enantioselectivity is determined by TS1-*R* and TS1-*S*. The energy difference of 1.9 kcal mol^{-1} is consistent with the high ee value in the experiment. We also

calculated the stereoselective outcome for the kinetic resolution result when one of the phenyl is replaced with methyl. The energy span that determines the enantioselectivity is reduced to 0.9 kcal mol^{-1} (TS1-*R*-(Me) vs. TS1-*S*-(Me)), which is consistent with the experimental results (71% ee).

In summary, we have developed the efficient synthesis of P-stereogenic phosphinamides via Pd(II)-catalyzed enantioselective olefination. A broad range of P-stereogenic phosphinamides were

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gained in moderate to good yields with high enantioselectivities *via* desymmetrization and kinetic resolution (33 examples, up to 77% yield, 99% ee). Gram-scale preparation was also compatible with good yield and high ee value. And the olefination *P*-stereogenic phosphinamide products could be transformed to potentially useful *P*-chiral phosphine oxides. The practical applications of these new *P*-stereogenic phosphinamides are currently being explored.

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Conflicts of interest

There are no conflicts to declare.

Notes and references

- (a) W. Tang and X. Zhang, Chem. Rev., 2003, 103, 3029; (b) S. J. Connon, Angew. Chem., Int. Ed., 2006, 45, 3909; (c) H. Fernández-Pérez, P. Etayo, A. Panossian and A. Vidal-Ferran, Chem. Rev., 2011, 111, 2119; (d) P. W. N. M. van Leeuwen, P. C. J. Kamer, C. Claver, O. Pàmies and M. Diéguez, Chem. Rev., 2011, 111, 2077; (e) D. Parmar, E. Sugiono, S. Raja and M. Rueping, Chem. Rev., 2014, 114, 9047.
- 2 (a) A. Grabulosa, *P-Stereogenic Ligands in Enantioselective Catalysis*, RSC, Cambridge, 2011; (b) W. Tang and X. Zhang, *Chem. Rev.*, 2003, **103**, 3029.
- 3 For select reviews, see: (a) J. S. Harvey and V. Gouverneur, *Chem. Commun.*, 2010, **46**, 7477; (b) I. Wauters, W. Debrouwer and C. V. Stevens, *Beilstein J. Org. Chem.*, 2014, **10**, 1064.
- 4 (a) K. M. Pietrusiewicz and M. Zablocka, Chem. Rev., 1994, 94, 1375;
 (b) A. Grabulosa, J. Granell and G. Muller, Coord. Chem. Rev., 2007, 251, 25;
 (c) O. I. Kolodiazhnyi, Tetrahedron: Asymmetry, 2012, 23, 1. For select examples, see: (d) Z. S. Han, N. Goyal, M. A. Herbage, J. D. Sieber, B. Qu, Y. Xu, Z. Li, J. T. Reeves, J.-N. Desrosiers, S. Ma, N. Grinberg, H. Lee, H. P. R. Mangunuru, Y. Zhang, D. Krishnamurthy, B. Z. Lu, J. J. Song, G. Wang and C. H. Senanayake, J. Am. Chem. Soc., 2013, 135, 2474; (e) S. Rast, B. Mohar and M. Stephan, Org. Lett., 2014, 16, 2688; (f) E. Bergin, C. T. O'Connor, S. B. Robinson, E. M. McGarrigle, C. P. O'Mahony and D. G. Gilheany, J. Am. Chem. Soc., 2007, 129, 9566; (g) Z. S. Han, H. Wu, Y. Xu, Y. Zhang, B. Qu, Z. Li, D. R. Caldwell, K. R. Fandrick, L. Zhang, F. Roschangar, J. J. Song and C. H. Senanayake, Org. Lett., 2017, 19, 1796.
- ⁵ For selected examples, see: (a) Z. Huang, X. Huang, B. Li, C. Mou, S. Yang, B.-A. Song and Y. R. Chi, *J. Am. Chem. Soc.*, 2016, **138**, 7524;
 (b) B. Pérez-Saavedra, N. Vázquez-Galiñanes, C. Saá and M. Fañanás-Mastral, *ACS Catal.*, 2017, 7, 6104; (c) Y. Zhang, F. Zhang, L. Chen, J. Xu, X. Liu and X. Feng, *ACS Catal.*, 2019, **9**, 4834; (d) B. M. Trost, S. M. Spohr, A. B. Rolka and C. A. Kalnmals, *J. Am. Chem. Soc.*, 2019, **141**, 14098; (e) Y. Huang, Y. Li, P.-H. Leung and T. Hayashi, *J. Am. Chem. Soc.*, 2014, **136**, 4865; (f) Y. Toda, M. Pink and J. N. Johnston, *J. Am. Chem. Soc.*, 2014, **136**, 14734.
- 6 For selected examples, see: (a) V. S. Chan, I. C. Stewart, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2006, 128, 2786;
 (b) C. Scriban and D. S. Glueck, J. Am. Chem. Soc., 2006, 128, 2788;

(c) C. Scriban, D. S. Glueck, J. A. Golen and A. L. Rheingold, Organometallics, 2007, 26, 1788; (d) B. J. Anderson, M. A. Guino-o, D. S. Glueck, J. A. Golen, A. G. DiPasquale, L. M. Liable-Sands and A. L. Rheingold, Org. Lett., 2008, 10, 4425; (e) V. S. Chan, M. Chiu, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2009, 131, 6021; (f) X.-T. Liu, Y.-O. Zhang, X.-Y. Han, S.-P. Sun and O.-W. Zhang, J. Am. Chem. Soc., 2019, 141, 16584; (g) J. R. Moncarz, N. F. Laritcheva and D. S. Glueck, J. Am. Chem. Soc., 2002, 124, 13356; (h) C. Korff and G. Helmchen, Chem. Commun., 2004, 530; (i) S. Pican and A.-Gaumont, Chem. Commun., 2005, 2393; (j) T. J. Brunker, B. J. Anderson, N. F. Blank, D. S. Glueck and A. L. Rheingold, Org. Lett., 2007, 9, 1109; S. Zhang, J.-Z. Xiao, Y.-B. Li, C.-Y. Shi and L. Yin, J. Am. Chem. Soc., 2021, 143, 9912; (k) N. F. Blank, J. R. Moncarz, T. J. Brunker, C. Scriban, B. J. Anderson, O. Amir, D. S. Glueck, L. N. Zakharov, J. A. Golen, C. D. Incarvito and A. L. Rheingold, J. Am. Chem. Soc., 2007, 129, 6847; (l) V. S. Chan, R. G. Bergman and F. D. Toste, J. Am. Chem. Soc., 2007, 129, 15122; Q. Dai, W. Li, Z. Li and J. Zhang, J. Am. Chem. Soc., 2019, 141, 20556; (m) R. Beaud, R. J. Phipps and M. J. Gaunt, J. Am. Chem. Soc., 2016, 138, 13183; (n) Q. Dai, L. Li and J. Zhang, Angew. Chem., Int. Ed., 2021, 60, 27247; (o) X.-T. Liu, X.-Y. Han, Y. Wu, Y.-Y. Sun, L. Gao, Z. Huang and Q.-W. Zhang, J. Am. Chem. Soc., 2021, 143(30), 11309; (p) W.-H. Wang, Y. Wu, H.-T. Wang, P.-J. Qi, W.-N. Lan and Q.-W. Zhang, Nat. Synth., 2022, 1, 738.

- 7 (a) J. Wencel-Delord and F. Colobert, Chem. Eur. J., 2013, 19, 14010–14017; (b) C. Zheng and S.-L. You, RSC Adv., 2014, 4, 6173; (c) C. G. Newton, S.-G. Wang, C. C. Oliveira and N. Cramer, Chem. Rev., 2017, 117, 8908; (d) T. G. Saint-Denis, R.-Y. Zhu, G. Chen, Q.-F. Wu and J.-Q. Yu, Science, 2018, 359, eaao4798; (e) J. Loup, U. Dhawa, F. Pesciaioli, J. Wencel-Delord and L. Ackermann, Angew. Chem., Int. Ed., 2019, 58, 12803; (f) G. Liao, T. Zhang, Z.-K. Lin and B.-F. Shi, Angew. Chem., Int. Ed., 2020, 59, 19773; (g) T. Yoshino, S. Satake and S. Matsunaga, Chem. Eur. J., 2020, 26, 7346; (h) T. K. Achar, S. Maiti, S. Jana and D. Maiti, ACS Catal., 2020, 10, 13748; (i) O. Vyhivskyi, A. Kudashev, T. Miyakoshi and O. Baudoin, Chem. Eur. J., 2021, 27, 1231; (j) Q. Zhang and B.-F. Shi, Acc. Chem. Res., 2021, 54, 2750; (k) Y.-Q. Han and B.-F. Shi, Acta Chim. Sin., 2023, 81, 1522.
- 8 (a) Q. Shao, K. Wu, Z. Zhuang, S. Qian and J.-Q. Yu, Acc. Chem. Res., 2020, 53, 833; (b) K. M. Engle and J.-Q. Yu, J. Org. Chem., 2013, 78, 8927; (c) R. Giri, B.-F. Shi, K. M. Engle, N. Maugel and J.-Q. Yu, Chem. Soc. Rev., 2009, 38, 3242; (d) B.-F. Shi, N. Y.-H. Maugel, Y.-H. Zhang and J.-Q. Yu, Angew. Chem., Int. Ed., 2008, 47, 4882.
- 9 Z.-J. Du, J. Guan, G.-J. Wu, P. Xu, L.-X. Gao and F.-S. Han, J. Am. Chem. Soc., 2015, 137, 632.
- (a) X.-L. Qin, A. Li and F.-S. Han, J. Am. Chem. Soc., 2021, 143, 2994;
 (b) G.-J. Wu, D.-X. Tan and F.-S. Han, Acc. Chem. Res., 2021, 54, 4354.
- 11 (a) J. Diesel and N. Cramer, ACS Catal., 2019, 9, 9164; (b) P.-F. Qian, J.-Y. Li, T. Zhou and B.-F. Shi, Synthesis, 2022, 4784. For selected examples see: (c) Z.-Q. Lin, W.-Z. Wang, S.-B. Yan and W.-L. Duan, Angew. Chem., Int. Ed., 2015, 54, 6265; (d) L. Liu, A. A. Zhang, Y. Wang, F. Zhang, Z. Zuo, W. X. Zhao, C. L. Feng and W. Ma, Org. Lett., 2015, 17, 2046; (e) G. Xu, M. Li, S. Wang and W. Tang, Org. Chem. Front., 2015, 2, 1342; (f) Y. Sun and N. Cramer, Angew. Chem., Int. Ed., 2017, 56, 364; (g) Y. Sun and N. Cramer, Chem. Sci., 2018, 9, 2981; (h) P. Hu, L. Kong, F. Wang, X. Zhu and X. Li, Angew. Chem., Int. Ed., 2021, 60, 20424; (i) C.-W. Zhang, X.-Q. Hu, Y.-H. Dai, P. Yin, C.-Y. Wang and W.-L. Duan, ACS Catal., 2022, 12, 193; (j) S.-Y. Song, Y. Li, Z. Ke and S. Xu, ACS Catal., 2021, 11, 13445; (k) G. R. Genov, J. L. Douthwaite, A. S. K. Lahdenperä, D. C. Gibson and R. J. Phipps, Science, 2020, 367, 1246; (l) Q.-J. Yao, J.-H. Chen, H. Song, F.-R. Huang and B.-F. Shi, Angew. Chem., Int. Ed., 2022, e202202892; (m) J.-H. Chen, M.-Y. Teng, F.-R. Huang, H. Song, Z.-K. Wang, H.-L. Zhuang, Y.-J. Wu, X. Wu, Q.-J. Yao and B.-F. Shi, Angew. Chem., Int. Ed., 2022, e202210106.
- 12 T. Zhou, L.-J. Fan, Z.-J. Chen, M.-X. Jiang, P.-F. Qian, X. Hu, K. Zhang and B.-F. Shi, *Org. Lett.*, 2023, **25**, 5724.
- 13 CCDC 2183918 (**3a**) contain the supplementary crystallographic data for this paper⁺.