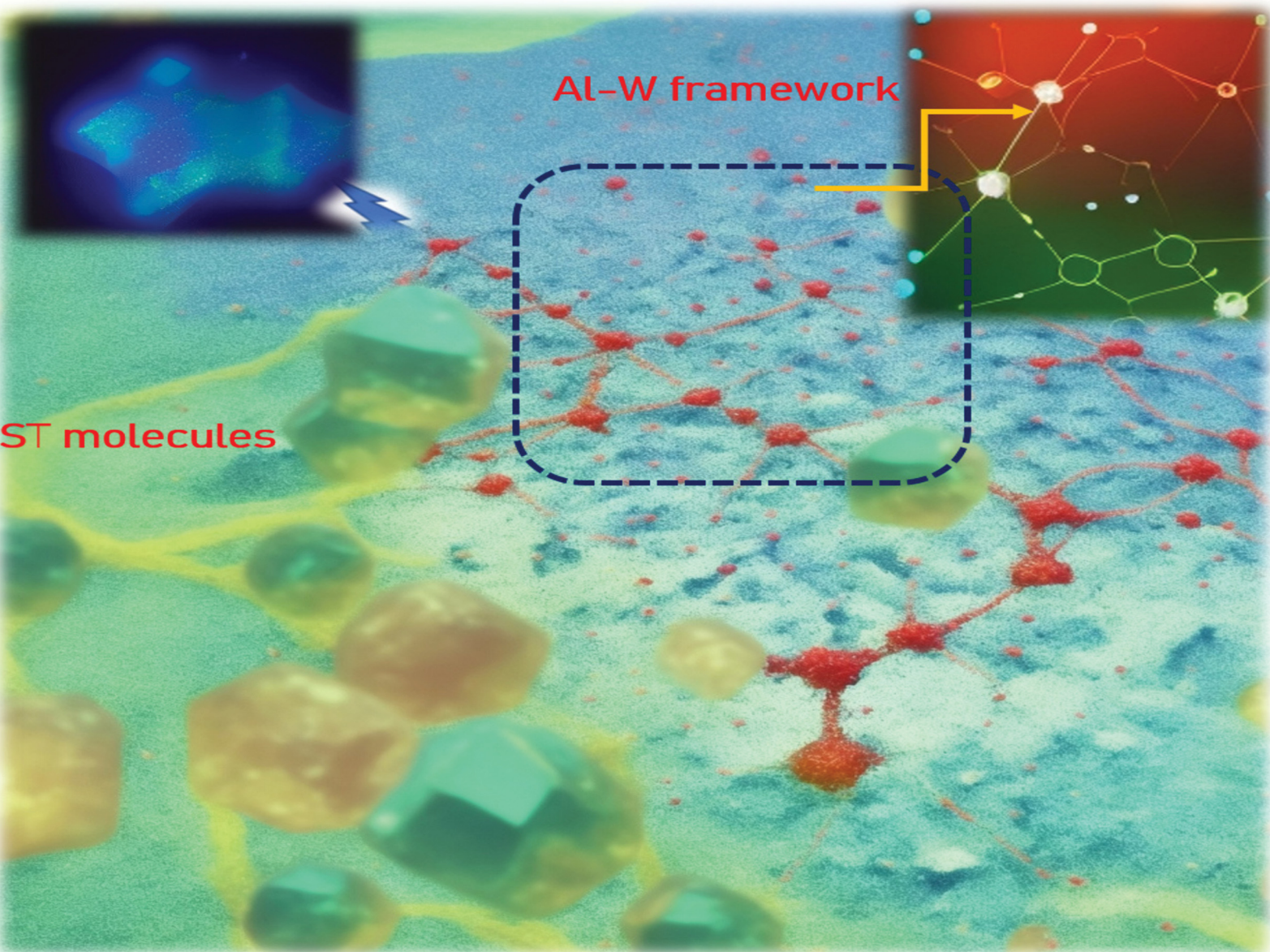


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**PAPER**

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“Synergistic effect” based novel and ultrasensitive approach  
for the detection of serotonin using DEM-modulated  
bimetallic nanosheets

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# “Synergistic effect” based novel and ultrasensitive approach for the detection of serotonin using DEM-modulated bimetallic nanosheets†

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Neurotransmitters have been of immense scientific interest due to their importance as human-health biomarkers. Several reports suggest necessary improvisations in the sensing capabilities of these neurotransmitters. Herein, the authors report a novel synthesis methodology for bimetallic aluminum–tungsten (Al–W) nanosheets, with the hybrid nanostructure showing high specificity toward serotonin neurotransmitters. The inspiration to design hybrid metallic nanosheets depends on the inherited optical properties of the parent precursors. The interstate conversion (ISC) between Al–W nanosheets promoted photoluminescent behavior with serotonin. The PL study shows that serotonin drastically enhanced  $\lambda_{em}$  at 335 nm. The importance of emission below the visible spectrum is to modulate any possible aggregation-induced emissions, which earlier troubled analytical chemists. The understanding of the selective detection of serotonin from a group of similar neurotransmitters is discussed with nanomolar quantification. The quantified detection limit using Al–W nanosheets is 0.05 nM with high linearity ( $R^2 = 0.9906$ ). Furthermore, real-world quantification studies have been performed on human urine and serum samples with  $R^2$  of 0.9938 and 0.9801, respectively.

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## Introduction

Our physiology is altered by hormones and neurotransmitters, which also control our behavior, hormone levels, sleep, and body temperature. The disruption of the circadian clock could have a major negative impact on our health and mental state.<sup>1,2</sup> Serotonin (5-hydroxytryptamine, ST) is a neurotransmitter that is produced in the raphe nuclei of the brainstem and transported to various parts of the brain, playing an important role as a biomarker in several diseases such as depression.<sup>3,4</sup> Alzheimer's<sup>5</sup> is regarded as the most distinct biomarker for carcinoid tumors,<sup>6</sup> atherosclerotic cardiovascular diseases (ASCVD), and chronic heart failure (CHF).<sup>7</sup> Abnormal levels of ST also affect sleep patterns, aggression, and appetite control. Numerous mental and neural illnesses, diabetes, vascular

problems, and post-traumatic stress disorder,<sup>8,9</sup> are triggered by excessive serotonin levels. Thus, the accurate detection of ST is urgently required to research of all diseases related to ST levels. Serotonin is related to tryptophan and melatonin; it acts as a precursor to melatonin, which is a hormone that controls insulin signaling, circadian rhythm, and energy balance.<sup>10</sup> Hamon & Glowinski explained the chemistry of tryptophan in the year 1974, which is subsequently converted into serotonin.<sup>11</sup> For these reasons, we have tried to detect tryptophan and melatonin also, but satisfactory results were not obtained.

To date, serotonin levels were determined using HPLC,<sup>12</sup> immunoassays,<sup>5,13</sup> chemiluminescence,<sup>14,45</sup> machine learning<sup>15</sup> radioimmunoassays,<sup>16</sup> fluorescence detection,<sup>17–21</sup> microdialysis, and fast-scan cyclic voltammetry<sup>22,23,46</sup> methods. The risk, diagnostics, pharmacological response, and therapeutic aim of 5-HT levels in cardiac disorders were measured using the HPLC method by Shimokawa's group in 2017.<sup>24</sup> In recent years, electrochemical sensing has also been explored to detect serotonin with different types of nanomaterials,<sup>6,25</sup> however, the main disadvantage of using electrochemical sensors<sup>23</sup> is lack of reusability and reproductively. Although some of the above techniques have shown great selectivity and sensitivity for ST detection, they need complex preparations, are time consuming, and destroy the analytes. At the same time, some require dyes such as labels and molecular frameworks for fluorescent detection. Most of the techniques need very expensive instruments. Therefore, fluorescence sensors are gaining significant attention in biosensing as

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they are simple and economical sensors with high sensitivity and selectivity.

Utilizing organic precursors for preparing nanomaterials can help in the molecular mixing of metals, prevent the loss of metal salts, and improve the homogeneity of solutions. In addition, this technique can enhance the morphology and texture of nanostructures. To create a symmetrical nanomaterial with high purity and reactivity, metals need more active sites, which are provided by organic precursors,<sup>26</sup> this opening new opportunities for the synthesis of bimetallic nanosheets. Diethylmalonate is a nontoxic and convenient natural product that is used in many organic syntheses as it provides mono and disubstitution and leads to a symmetrical and stable structure. It also acts as an intermediate in the synthesis of vitamins, drugs, and agrochemicals.<sup>27,28</sup> Thus, considering the aforementioned factors, diethylmalonate (DEM) was used as an organic precursor to prepare the 2D nanosheets.

Therefore, a label-free, cost-effective, and fast operational method is needed for ST detection. Herein, we propose a novel strategy for serotonin detection by the generation of fluorescence-induced “synergistic effect” with the reaction of nonfluorescent nanosheets. We prepared the Al–W bimetallic nanosheets for the detection of ST using the “turn-on fluorescence” method, wherein stronger fluorescent property in serotonin was obtained with simple chemical modifications by the reaction with our novel and cost-effective bimetallic nanosheets. The ultrasonication method was used to synthesize the Al–W nanosheets. We derivatized serotonin with our Al–W nanosheets, which led to drastically enhanced optical properties as compared to earlier reports.<sup>47</sup> The synergistic effect can be seen selectively with serotonin, where serotonin reacted with Al–W nanosheets in basic medium (0.1 N NaOH and certain amount of PBS) and produced blue fluorescence at  $\lambda_{em}$  335 nm on  $\lambda_{ex}$  300 nm excitation, and an enhancement in fluorescence intensity was observed with increasing ST concentration. The detection limit for the proposed serotonin detection methodology was calculated to be 0.05 nM. We also determined serotonin in other biological fluids such as human

urine and blood serum. The selectivity experiment was performed and it was found that serotonin has high selectivity over other amines and neurotransmitters.

## Materials and methods

### Reagents

Tungsten chloride ( $WCl_4$ ) and aluminum chloride ( $AlCl_3$ ) were purchased from Alfa Aesar, USA. Diethylmalonate (DEM) (99%) was purchased from Lancaster Eastgate, Morecambe, England. All monoamines including serotonin, tryptophan, *N*-epinephrine, cysteamine, melatonin, lomefloxacin, and dopamine were purchased from Sigma Aldrich, USA.

### Instrumentation

The optical properties were observed using an EVOLUTION 201 (Thermo SCIENTIFIC, USA) UV-Vis absorption spectrophotometer, while the fluorescence spectra were recorded using a Hitachi Fluorescence Spectrophotometer F-2700 (Hitachi, Japan). The structural characteristics of the nanomaterials were studied using a JEOL (Japan) JEM2100 transmission electron microscope (TEM) and high-angle annular dark-field scanning transmission electron microscopy (HAADF-STEM) imaging. The particle size was determined using an ELSZ-2000 (Otsuka Electronic, Japan) dynamic light scattering (DLS) instrument, and the crystallinity of the material was examined by X-ray diffraction (XRD, Bruker D8 Advance, Philips, Netherlands). Raman measurements were performed by depositing the as-prepared self-assembled samples on a glass substrate using the setup  $\lambda = 633$  nm laser (JOBIN-YVON T64000, USA), while XPS was performed using an Auger electron spectrometer (JEOL, Japan).

### Synthesis of bimetallic nanosheets

20 mg  $WCl_4$  and 20 mg  $AlCl_3$  (1 : 3 moles of W : Al) were added into 10 mL DEM and placed for probe sonication for 15 min.

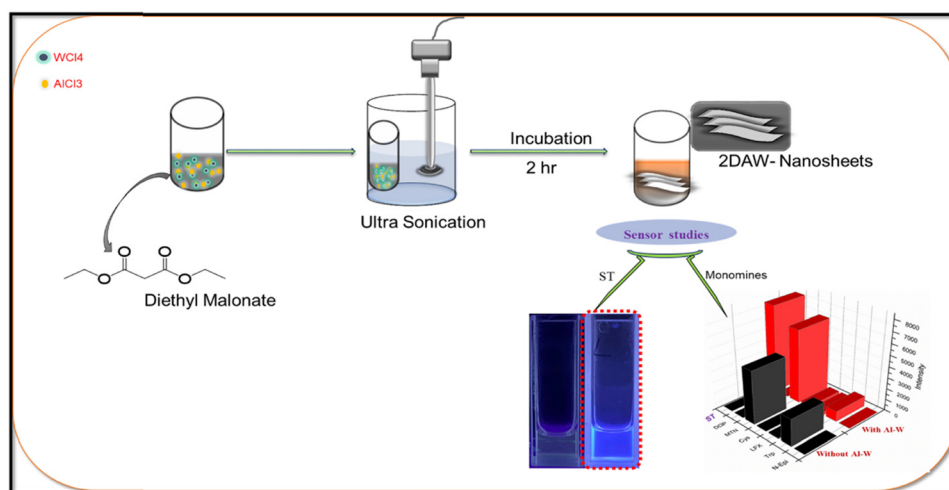


Fig. 1 Schematic presentation to show the reaction procedures for the preparation of Al–W nanosheets and sensor studies for the detection of serotonin with the selectivity 3D graph.

The mixture was kept for incubation for further 2 h, and bimetallic nanosheets were then obtained, as shown in Fig. 1. Further, this solution was characterized and used for sensing applications.

### Detection of serotonin

Firstly, the initial 10 mM stock solution of serotonin was prepared in 0.1 N NaOH. Afterward, increasing concentration (10 nM–1 mM) of ST from the stock solution was added with 30  $\mu$ L Al–W Ns, and the solution was diluted with PBS (pH 7.4) to a final volume of 1 mL. Then, the fluorescence emission spectra of the abovementioned solution was recorded at 300 nm excitation wavelength. All the experiments were done three times in parallel to ensure the recorded emission values.

### Selectivity

To observe the selectivity of the current biosensor toward serotonin, we performed an array of experiment with other biological monoamines neurotransmitters (chemical structures are shown in Fig. S1, ESI<sup>†</sup>) such as tryptophan, *N*-epinephrine, cysteamine, melatonin, lomefloxacin, and dopamine with the same concentration in PBS solution as that of serotonin. The fluorescence emission readings were recorded at 300 nm excitation wavelength.

### Real sample preparation

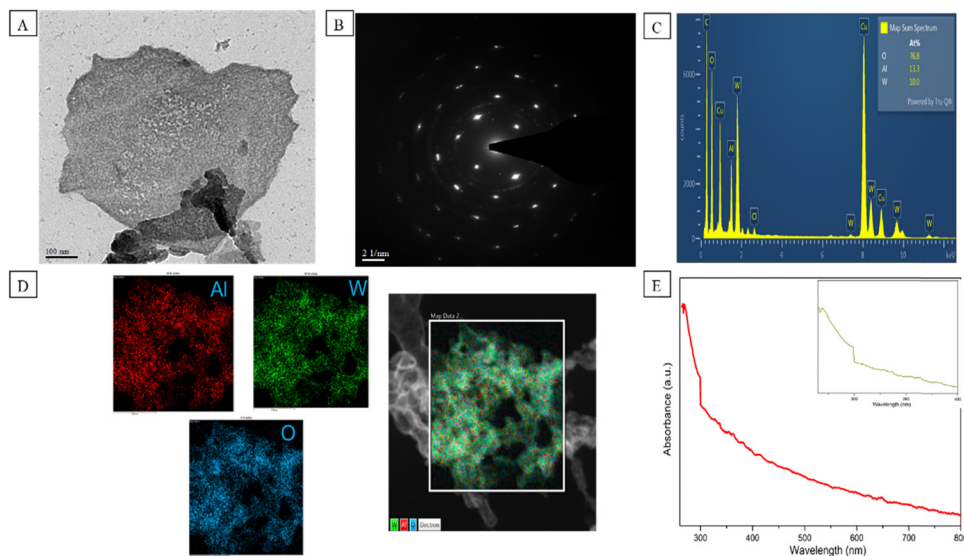
Again, a stock solution of serotonin was prepared in fresh filtered urine, and then four different concentrations (500 nM, 1  $\mu$ M, 5  $\mu$ M, 10  $\mu$ M) of ST from the above stock solution (in urine) were added into 30  $\mu$ L Al–W nanosheets. The solution was diluted with PBS to a final volume of 1 mL. The emission readings were recorded at 300 nm excitation. The same

experiment was also performed with blood serum sample instead of urine.

## Results and discussion

### Characterization of the bimetallic nanosheets

Many structural techniques were used for the characterization of Al–W nanosheets. First, we studied the TEM and HAADF-STEM images to confirm the morphology and size of Al–W nanosheets, as shown in Fig. 2A. Fig. 2A displays the sheet-like structure on a 100 nm scale. The crystalline nature of the nanosheets is evident in Fig. 2B, which confirms the nonhexagonal lattice with the planes (011) and (021) obtained from the SAED pattern. The elements of the nanosheets were confirmed by EDAX (Fig. 2C) with the Al:W:O ratio of 1.33:1:7.68. The HAADF-STEM image in Fig. 2D shows the elemental analysis of Al–W nanosheets, where red color is for Al metal, green color for W, and blue color for O. Furthermore, the UV-absorbance peaks are shown for Al–W nanosheets in Fig. 2E, and the spectra shows an intense peak at about 270 nm (inset), indicating the presence of Al<sup>29,30</sup> and a small hump at 300 nm, showing the presence of W in bimetallic nanosheets.<sup>31</sup> Some controlled experiments were also performed for pure Al and W nanosheets, with W and Al with different 1:1 of moles in DEM. The UV absorbance analyses for W and Al separately synthesized in DEM are shown in Fig. S2A and S2B (ESI<sup>†</sup>) before and after the addition of ST. The fluorescence investigations between W (DEM) and Al (DEM) with and without ST are shown in Fig. S2D and S2E (ESI<sup>†</sup>). We found that there were no discernible changes in color or intensity even after the addition of ST at a very high concentration (10 mM). From all the control experiments, we found out that a perfect mole ratio of W and Al was required to synthesize the Al–W



**Fig. 2** Characterization of the Al–W nanosheets (A) surface morphology showing layered nanostructure, (B) SAED pattern showing perfect crystalline phases, (C) elemental ration shown in EDS pattern, (D) high-angle annular dark-field scanning transmission electron microscopy image of Al–W nanosheets, showing the pattern for elements W (green color), Al (red color), O (blue color), and single element mapping of W (green color), O (blue color), and Al (red color), (E) UV spectra of Al–W nanosheets (zoom-in image in the inset shows a hump at 270 nm).

nanosheets, and the average size of Al-W nanosheets was measured to be 200 nm by DLS, which is shown in Fig. S3 (ESI<sup>†</sup>).

Next, the  $\mu$ -Raman spectra of Al-W nanosheets synthesized in DEM were recorded, as shown in Fig. 3A. Four major peaks were observed for Al-W nanosheets. The peaks at 1570 (G band) and 1339  $\text{cm}^{-1}$  are attributed to Al<sup>32,33</sup> peaks at 803 and 685  $\text{cm}^{-1}$ , showing the Al-W ratio in the nanostructure; we also conducted Raman spectroscopy for the different Al:W (1:1) ratio. However, as can be seen in Fig. S4 (ESI<sup>†</sup>), there was no such sign of the presence of W and Al in the nanostructure. In Fig. 3B and Fig. 3C, we carried out X-ray photoelectron spectroscopy (XPS) of Al-W nanosheets. As shown in Fig. 3B and C, the X-ray photoelectron spectroscopy (XPS) study reveals the bonding of oxygen with tungsten and aluminum. The XPS results of W can be fit into three peaks; the two dominating peaks centered at 34.069 and 36.19 eV corresponding to typical binding energies of W 4f with +6 valence state (W 4f<sub>7/2</sub> and W 4f<sub>5/2</sub>, respectively), a small peak at 40.0 eV, confirm the presence of the W 4f in the 5+ state. The presence of W in the +5 state shows the existence of oxygen vacancies, which can be filled by Al.<sup>34–37</sup> In Fig. 3C, the XPS peaks of Al 2p are presented. Two shoulders are observed in this spectrum at 72.7 and 73.4 eV binding energies. The major peak at 73.4 eV refers to Al<sub>2</sub>O<sub>3</sub>; beside that, a small broadening peak at 72.7 eV is related to Al-metal bonds.<sup>38–40</sup> Two peaks for O 1s are shown in Fig. 3D, which clearly featured the presence of asymmetric O 1s in Al-W nanosheets. The most intensive and broad peak at 530.8 eV is related to oxygen present in the lattice (O<sup>2-</sup>) of W, and the small hump at 528.8 eV also corresponds to metal-oxide, which may refer to Al-O bonds.<sup>36,41,42</sup> The survey scan is shown in Fig. S5 (ESI<sup>†</sup>).

## Sensor performance

The distinctive characteristics of the Al-W nanosheets after the interaction with serotonin and the interaction of Al-W

nanosheets with serotonin was evaluated by UV-Vis absorbance spectroscopy, as shown in Fig. 4A. Fig. 4A shows three broad bands at 209 nm and 230 nm and a peak at 272 nm. First, the absorption peak increase from 200 nm to 250 nm is due to serotonin<sup>43,44</sup> and the peak is observed at 272 nm for Al-W nanosheets with a small shift. The presence of ST causes the “turn-on” blue fluorescence emission effect at 335 nm after excitation at 300 nm within a few minutes (Fig. 4B). To check the stability of the sensor with ST, we carried out time-dependent fluorescence studies. As shown in Fig. 4B, the luminescent intensity gradually increases until the 5th day but from the 6th day, it started decreasing, which also can be observed in images under a UV lamp, shown in Fig. 4 inset. To demonstrate the significance of Al-W nanosheets for the generation of fluorescence characteristics in the system, a control experiment for serotonin was also conducted simply using DEM solvent (Fig. S6, ESI<sup>†</sup>). To investigate the sensitivity of the sensor, we carried out the fluorescence titration experiment by adding different concentrations of serotonin into Al-W nanosheets. We took a broad range of serotonin concentration varying from 10 nanomolar to 1 millimolar. As shown in Fig. 4C, with the increasing addition of ST concentration (10 nM to 1 mM), the fluorescence intensity of solution (Al-W + ST) at 335 nm increased on 300 nm excitation with a linear correlation value ( $R^2$ ) of 0.9906. There is also a strong linear relationship between ST concentration (10 nM<sup>-1</sup> mM) and intensity ratio (Fig. 4D). The limit of detection for ST was calculated to be 0.05 nM. We compared the LOD with some previous literature and found out that the LOD of ST in this work is improved up to 10–20 times compared to other reported sensors (Table S1, ESI<sup>†</sup>). The interaction between Al-W and ST is experimentally proved by the zeta potential results, as shown in Fig. S7 (ESI<sup>†</sup>). The presence of extremely electron-rich metals causes the zeta potential of Al-W in the buffer solution (PBS) to be  $-49.72$  mV, which induces a general negative charge

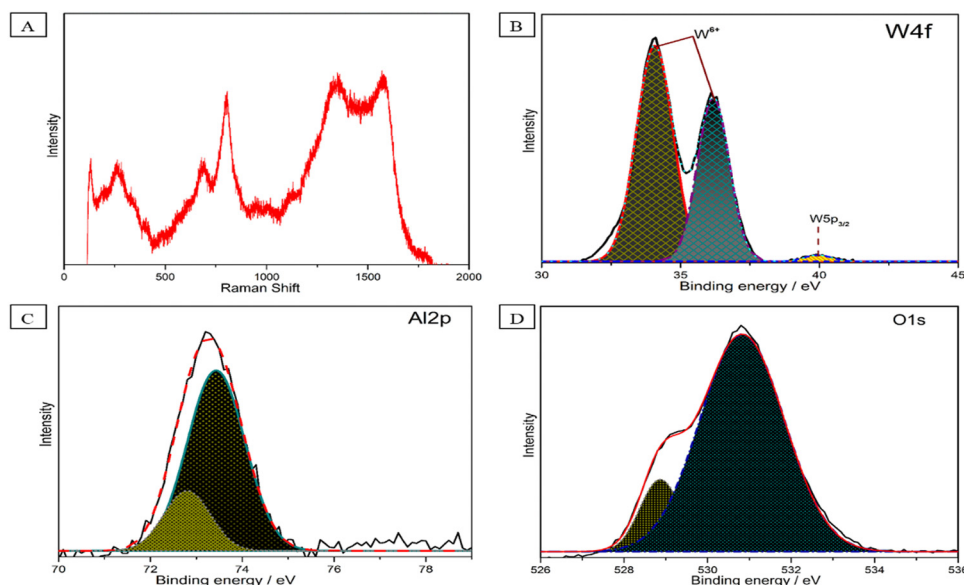


Fig. 3 (A) Raman spectra of Al-W nanosheets. XPS spectra fitting of (B) W 4f, (C) Al 2p, and (D) O 1s for Al-W nanosheets.

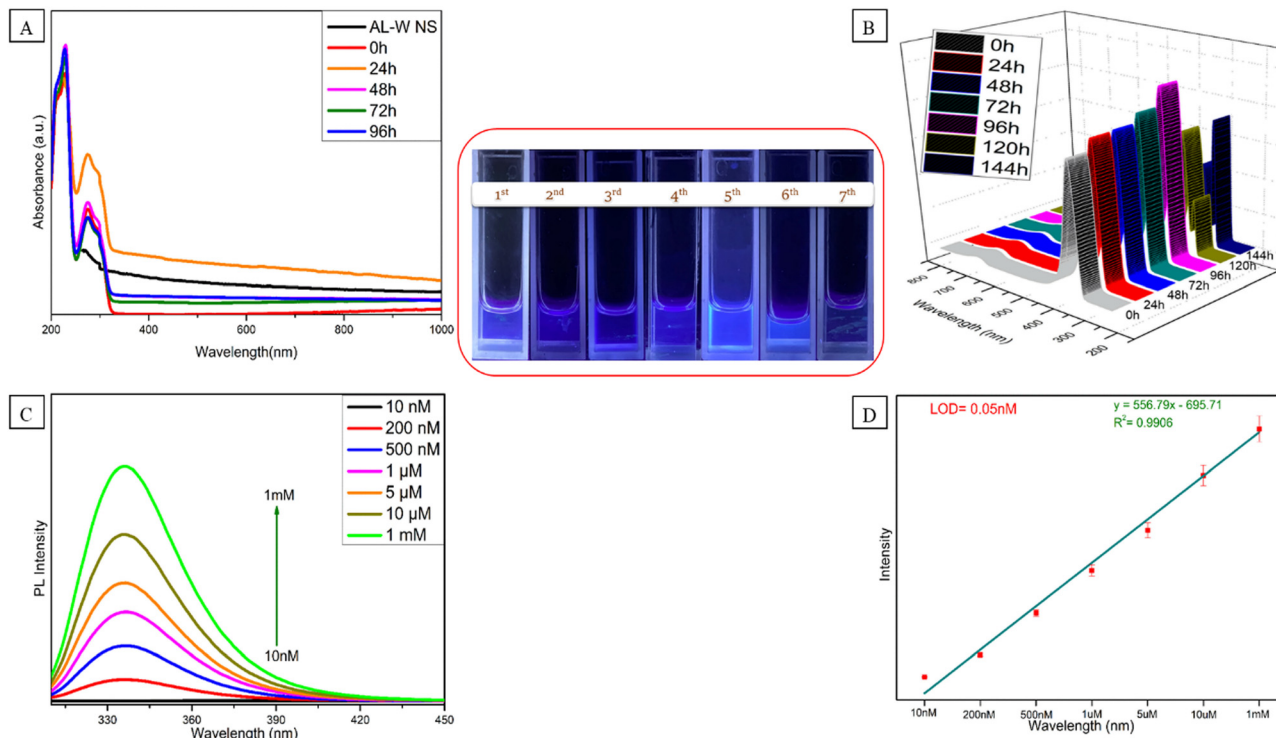


Fig. 4 Serotonin assay. (A) Variation in UV spectra shown to study the interaction between the Al–W nanosheets and serotonin, (B) recorded PL emission for 7 consecutive days (UV light driven images shown in inset), (C) emission spectra with the increasing concentration of ST, (D) linear fitting relationship between the PL intensity and ST concentration.

on the material. However, the measured zeta potential for the Al–W + ST solution is  $-9.31$  mV.

### Real sample analysis

As serotonin is a very important neurotransmitter in human body, after realizing the significance of ST levels in human body and how the alteration of ST level can affect the diagnosis of many critical carcinoid diseases, we further assessed the feasibility of the sensor in human urine and blood serum. Keeping this in mind, we prepared four concentrations (500 nM, 1, 5 and 10  $\mu$ M) of ST in human urine and blood serum. The emission spectra were recorded in the same way as the above method (Fig. 4C). As shown in Fig. 5A and C, increasing fluorescence response was evaluated by an increase in the concentration from 500 nM to 10  $\mu$ M in human urine and blood serum samples. In a wide range of serotonin concentrations (500 nM–10  $\mu$ M), the emission intensity rises linearly with increasing serotonin concentration. The linear regression equation between the concentration of serotonin and intensity for the urine and blood serum samples is also expressed with  $R^2 = 0.9938$  and  $R^2 = 0.9801$  in Fig. 5B and C, respectively. The recovery results for urine and blood serum samples were also calculated, as shown in Fig. S9 (ESI<sup>†</sup>). To prove the sensor ability, the developed sensor was compared with the reported sensor based on different methods with their LOD, range, and results, as shown in Table S2 (ESI<sup>†</sup>). From the previous reported work, it can be concluded that we have developed an ultrasensitive sensor for the detection of serotonin in biological

fluids as well and which may be due to the synergistic effect that occurs between Al–W nanosheets and serotonin.

To ensure the selectivity of the fluorescence sensor for ST in biological systems, the fluorescence response behavior of the sensor toward several neurotransmitters and biological drugs (norepinephrine, tryptophan, lomefloxacin, cysteamine, melatonin, dopamine, and serotonin) was tested under the experimental conditions. Fluorescence results were taken without and with Al–W nanosheets (grey bar-without Al–W and purple bar-with Al–W), as shown in Fig. 6; only ST showed the obvious results. The reaction with ST instantly showed the turn-on fluorescence effect with a great increase in fluorescence intensity.

However, there were no significant changes with other neurotransmitters and PL emission spectra, and the images related to each monoamine taken under UV lamp are shown in Fig. S8 (ESI<sup>†</sup>). Melatonin caused a slight increase in fluorescence after reaction with the analyte (Al–W nanosheets). Thus, the developed method showed satisfactory results only for ST and confirmed the selectivity for ST. From the above results, the Al–W nanosheets showed the greatest affinity and ultrasensitivity toward serotonin in comparison with the literature (Table S1 and S2, ESI<sup>†</sup>).

After concluding all the results, the possible mechanism for luminescence can be attributed to the synergistic effect. By looking at the UV spectra and zeta potential results before and after the addition of ST in Al–W nanosheets, it is possible that because of the presence of highly rich electrons metals such as W and Al in a certain ratio mentioned above, the spin state of W can lead to the delocalization of the spin states of Al, which may

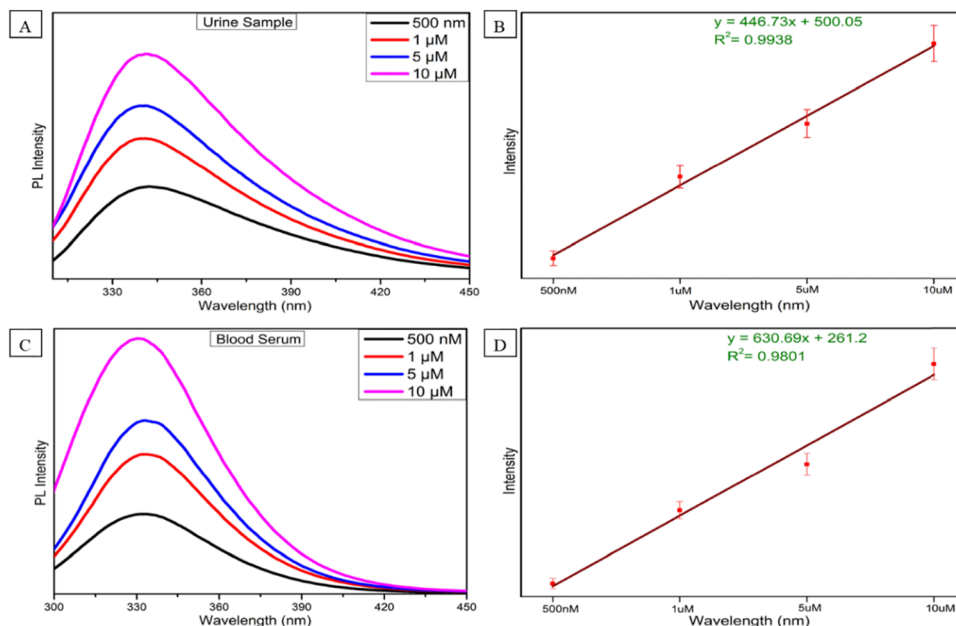


Fig. 5 Real world sample studies in urine and blood samples. (A) PL Emission responses of ST different concentration in human urine, (B) linear relationship corresponding to ST in urine sample, (C) PL emission responses of different concentrations of ST in blood serum, (D) linear relationship corresponding to ST in the blood serum sample.

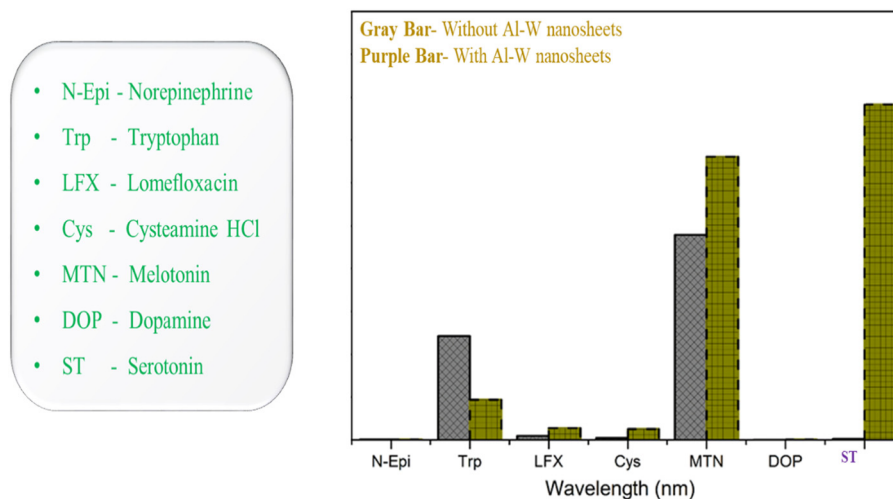


Fig. 6 Fluorescence response of the Al-W nanosheets toward ST and a variety of interfering substances (gray bar-without nanosheets, purple bar-with nanosheets).

increase the Al electron density and improve the adsorption behavior of the Al-W nanosheets. Thus, these electrons can be transferred from the Al-W nanosheets to the LUMO of ST, and serotonin's intrinsic fluorescence, which comes from its indole moiety, can be used to analyze it. Finally, the turn on fluorescence effect occurs with high emission intensity (Fig. 7).

## Conclusion and future road

As highlighted in this article, the clinical significance of serotonin is very crucial and important for the diagnosis and

treatment of the expanding range of diseases and disorders. For this purpose, a simple turn on fluorescence method was developed by introducing the novel Al-W nanosheets with serotonin. Time-saving ultrasonication method was used for the first time to synthesize Al-W nanosheets. We also introduced Al and W metals for the first time together. Owing to the synergistic effect of Al-W nanosheets and ST, extraordinary performance was observed with a detection limit of 0.05 nM in a wide linear range from 10 nM to 1 mM. The developed sensor was also tested in biological fluids such as human urine and blood serum for the quantitative analysis of ST, and the calculated results were excellent

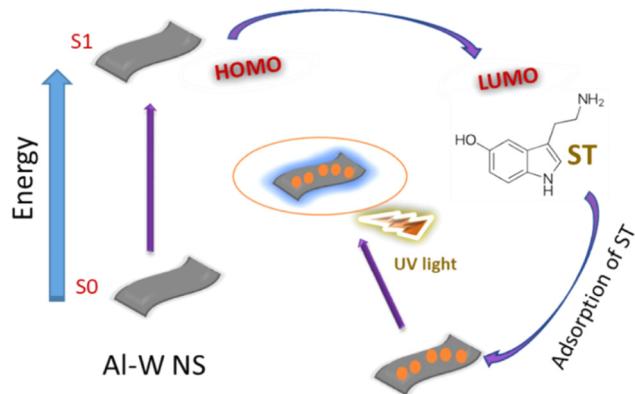


Fig. 7 The synergistic effect generated the luminescence mechanism for the sensing of ST.

with a recovery of more than 100% in urine and blood serum samples. Therefore, from the above results and observation, we successfully developed a new ultrasensitive sensor for the detection of ST, which includes simple sample preparation and cost-effective methods. In the future, the integration of health with time can be resolved with such excellent ultrasensitive biosensors, which can offer new therapeutic opportunities by delivering a thorough understanding of receptor functions.

## Author contributions

Deepak Dabur: proposed and designed the methodology and carried out all the experiments, data management, and wrote the manuscript. Nallin Sharma: review and revised the manuscript. Hui Fen-Wu: principal investigator and getting funding support, supervising all experiments, checking all data on group meeting weekly and expert guidance to the model and work plan, and revision/proof on the paper.

## Conflicts of interest

There are no conflicts to declare.

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