Chemical Science



EDGE ARTICLE

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2023, 14, 11532

dall publication charges for this article have been paid for by the Royal Society of Chemistry

Received 8th July 2023 Accepted 30th September 2023

DOI: 10.1039/d3sc03493c

rsc.li/chemical-science

Synergistically remodulating H⁺/Ca²⁺ gradients to induce mitochondrial depolarization for enhanced synergistic cancer therapy†

Xiaoni Wang,^a Xiyang Ge,^a Xiaowen Guan,^a Jin Ouyang ^b and Na Na ^a

The remodulation of H^+/Ca^{2+} gradients in the mitochondria matrix could be effective to induce mitochondria depolarization for the enhancement of cancer therapy. However, it is still challenged by H^+ homeostasis, insufficient Ca^{2+} , uncoordinated regulations, and inefficient loading/delivery strategies. Herein, a supramolecular DNA nanocomplex (Ca@DNA-MF) was prepared to synergistically remodulate H^+/Ca^{2+} gradients for mitochondrial depolarization. Upon targeted functionalization and TME-triggered delivery, multiple reagents were released in cancer cells for synergistic three-channel mitochondrial depolarization: the gene reagent of siMCT4 blocked the LA metabolism to induce mitochondrial acidification by downregulating monocarboxylate transporter 4 (MCT4); released Ca^{2+} disrupted Ca^{2+} homeostasis to facilitate Ca^{2+} -based mitochondrial depolarization; specifically, TME-activated glutathione (GSH) depletion facilitated efficient generation of hydroxyl radicals (*OH), further enhancing the mitochondrial depolarization. The remodulation not only triggered apoptosis but also led to ferroptosis to generate abundant ROS for efficient LPO-based apoptosis, providing a synergistic strategy for enhanced synergistic cancer therapy.

Introduction

Rational regulation of organelles, to administer cellular behaviour and fate, has attracted great attention for the intervention and treatment of diseases. ¹⁻³ As powerhouses and energy providers, mitochondria have been reported to play an important role in the occurrence of many major diseases and their regulation can effectively induce cancer cell death. Generally, the activity of mitochondria depends on the proton (H⁺) gradient of the inner mitochondrial membrane. To form the proton gradient, ATP is normally generated and provides energy to maintain cell metabolism and survival. ⁴⁻⁶ Meanwhile, Ca²⁺ homeostasis in mitochondria also contributes to the regulation of H⁺ gradients for maintaining the normal function of mitochondria. ⁷⁻⁹ Accordingly, modulating the H⁺ gradient and Ca²⁺ homeostasis would lead to mitochondrial damage and cell apoptosis *via* opening of mitochondrial permeability

C for the subsequent lysis of caspase-3 protein. ^{10,11} Therefore, it can be imaged that the synergistic regulation of H⁺/Ca²⁺ influx to the mitochondrial matrix will lead to efficient mitochondrial depolarization for enhanced cancer therapy.

Nevertheless, the mitochondrial depolarization could be

transition pores. This would facilitate the release of cytochrome

Nevertheless, the mitochondrial depolarization could be a complex process, which was normally hindered in the heterogeneous and dynamic tumor microenvironment (TME). 12-14 Firstly, TME homeostasis of tumor cells dramatically hinders the mitochondrial depolarization. For example, pH homeostasis could be maintained by the effusion of intracellular by-products of glycolysis metabolite and lactic acid (LA) metabolism in the TME.15,16 Besides, cytoplasm and mitochondria can serve as important Ca²⁺ storage hubs to maintain intracellular Ca2+ homeostasis, which greatly limited the mitochondrial depolarization.^{17,18} Furthermore, the toxic reactive oxygen species (ROS), a pivotal agent for mitochondrial membrane permeability, is easily depleted by abundant endogenous antioxidant glutathione (GSH), leading to poor therapeutic effectiveness. 19-22 Therefore, multichannel synergistic regulations to balance TME homeostasis for sensitized mitochondrial depolarization are required.

Meanwhile, the effective loading and delivery of therapeutic agents as well as TME-responsive drug release are limited, ^{23–25} which encounter undesirable leakage of "off-target" toxicity and adverse side effects. ^{26–28} Therefore, the meticulous loading of multiple reagents into nanocarriers is essential for designing an ideal collaborative system. Fortunately, upon programmable

^aKey Laboratory of Radiopharmaceuticals, Ministry of Education, College of Chemistry, Beijing Normal University, Beijing 100875, China. E-mail: nana@bnu.edu.cn

^bDepartment of Chemistry, College of Arts and Sciences, Beijing Normal University at Zhuhai, Zhuhai City, Guangdong Province 519087, China

[†] Electronic supplementary information (ESI) available: Experimental section; general information and additional data for material characterization, including the data of fluorescence spectra and TEM. Detailed DNA sequence, 4T1 cell imaging, colocalization experiments, flow cytometric analyses, *ex vivo* imaging of organs and tumor tissues, and H&E staining for major tissue sections. See DOI: https://doi.org/10.1039/d3sc03493c

Edge Article Chemical Science

assembly, deoxyribonucleic acid (DNA)-based materials exhibited stimulus-responsiveness (such as pH, enzymes, and metal ions), controlled network structures and convenient modifications, which showed great potential for cancer therapy.²⁹⁻³⁴ Therefore, a multifunctional DNA nanoplatform is expected to induce cell apoptosis upon efficient delivery of multiple reagents for safe and synergistic mitochondrial depolarization.

Herein, a supramolecular DNA nanocomplex (Ca@DNA-MF) was fabricated upon the dynamic assembly of multiple mitochondrial depolarization reagents into a DNA nanonetwork. Via the hybridization of designed DNA primers, the tiny "seeds" of Fe/Mn-polyphenol coordination polymers were packed by siRNA (siMCT4)-integrated DNA nanonetworks. The obtained spherical nanocomplex was further biomineralized with Ca²⁺ in the presence of hyaluronic acid (HA) to obtain Ca@DNA-MF for targeting tumor cells. Upon targeting functionalization with HA, Ca@DNA-MF accumulated at tumor sites. This was followed by efficient Ca²⁺ release to break Ca²⁺ homeostasis, leading to Ca²⁺-based mitochondrial depolarization. Simultaneously, ATP-triggered siMCT4 downregulated MCT4 expression on the tumor cell membrane to restrain lactate acid (LA) efflux and induce TME acidification. This can further accelerate the overloading of Ca2+ in the mitochondrial matrix. Specifically, the drastic cytoplasmic acidification also enhanced the Fenton reaction to generate adequate ROS. This further increased the mitochondrial membrane permeability and dissipated the proton gradient. This was achieved by the synergistic consume of excessive endogenous glutathione (GSH) by redox systems of Fe³⁺/Fe²⁺ and Mn²⁺/Mn^{x+}. Consequently, via metabolic reprogramming and cancer cell disturbance, synergistic remodulation of H+/Ca2+ gradients was obtained for efficient mitochondrial depolarization. Upon triggering apoptosis and ferroptosis to enhance ROS generation, the remodulation was efficient for enhanced synergistic cancer therapy with low toxicity.

Results and discussion

Design of the Ca@DNA-MF nanocomplex

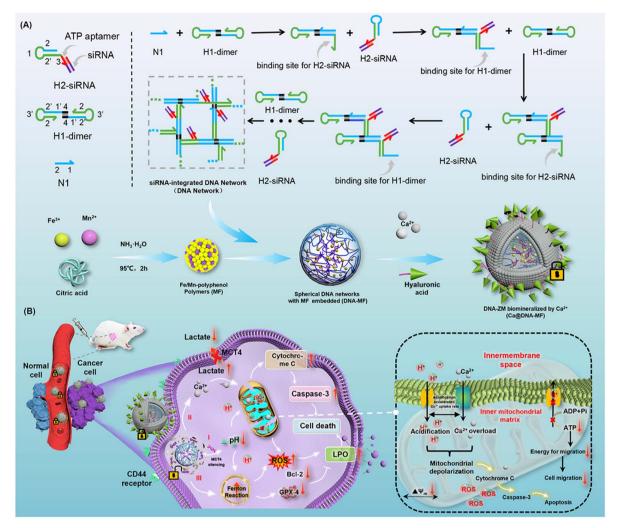
The present nanocomplex of Ca@DNA-MF was prepared upon multiple procedures of dynamic self-assembly. As shown in Scheme 1A, the primers of N1, H1-dimer and ATP aptamerintegrated H2-siRNA hairpins were designed. In the presence of N1, H1-dimer hybridized with H2 hairpins to form a DNA network under the hybridization chain reaction (C-HCR).35,36 Both the ATP aptamer and a specific MCT4 inhibitor of siMCT4 were integrated into H2 hairpins. This would facilitate ATPdriven dissociation for gene-mediated metabolic reprogramming, which depended on LA levels in high glycolytic cancer cells. Simultaneously, the "seeds" of Fe/Mn-polyphenol polymers (MF) were prepared by thermal decomposition of Fe³⁺ and Mn²⁺ in NH₃·H₂O, which then accumulated into "seeds" in the presence of citric acid. Thereafter, the "seeds" were added into the aforementioned DNA nanocomplex to obtain spherical MFembedded DNA networks (DNA-MF) upon electrostatic interactions. Finally, the spherical nanocomplex was biomineralized by Ca²⁺ in the presence of HA. Therefore, Ca@DNA-MF was

obtained for the controllable release of therapy reagents, avoiding reagent leaking in the *vivo* fluid circulation and eliminating side effects toward normal tissues.

In the vivo fluid circulation (Scheme 1B), Ca@DNA-MF sitespecifically targeted tumor sites, because the surface HA specifically recognized highly expressed CD44 membrane proteins on the surface of cancer cells.37-39 Then, upon endocytosis, the Ca-biomineralized cover was on-demand "unlocked" in the slightly acidic TME. This initiated the subsequent synergistic multichannel mitochondrial depolarization. Firstly (channel I), the siMCT4 in the DNA networks was released upon the interaction between the integrated ATPaptamer and the cytoplasmic ATP in cancer cells. The ATPdriven siMCT4 could downregulate the MCT4 expression on the tumor cell membrane to initiate intracellular retention of LA. This resulted in drastic acidification of cytoplasm and further induced mitochondria matrix acidification for mitochondrial depolarization. Simultaneously, the acidification of the mitochondria matrix would synergistically promote the Ca²⁺ overloading upon the Ca2+ release from the cover of the DNA nanocomplex, being another channel for mitochondrial depolarization (channel II). Thirdly, along with the decomposition of Ca@DNA-MF, the seeds of Fe/Mn-polyphenol polymers were exposed to release Fe3+, which immediately in situ reduced to Fe²⁺ by endogenous GSH. Subsequently, promoted by Mn²⁺, Fe²⁺ participated in the efficient Fenton-like catalytic reduction of endogenous H2O2 to cytotoxic 'OH.40-43 This would further facilitate the mitochondrial depolarization upon the increase of mitochondrial membrane permeability for penetrating more H⁺/Ca²⁺ into the mitochondrion (channel III). Additionally, the acidosis in channel I would also enhance the Fenton-like 'OH generation to facilitate the potential of apoptosis with LA metabolic modulation. Consequently, upon the synergistic "network" of multichannel mitochondrial depolarization, an increase of ROS can be eventually obtained. This would exhibit increased expression of both cytochrome C and caspase-3 (apoptotic-pathway-related proteins), along with the decrease of Bcl-2 and GPX-4 protein.

Characterization of the Ca@DNA-MF nanocomplex

To confirm the successful preparation of the Ca@DNA-MF nanocomplex, a series of characterization methods have been carried out. As demonstrated by transmission electron microscopy (TEM) characterization, the seeds of MF nanoparticles exhibited a tiny spherical morphology at a monodisperse size of 6 nm (Fig. 1A-MF). Meanwhile after being embedded into 100 nm of the DNA networks (Fig. S1†), larger spherical DNA-MF particles of approximately 120 nm diameter were obtained (Fig. 1A DNA-MF). Subsequently, after biomineralization with Ca²⁺ and HA, the size of the Ca@DNA-MF nanocomplex reached about 140 nm. Demonstrated by high-angle annular dark field-scanning TEM (HAADF-STEM) and elemental mapping images, the uniform element distribution of both DNA-MF (Fig. S2†) and Ca@DNA-MF (Fig. 1B) was revealed, verifying the successful assembly-based preparation. Furthermore, the stepped increase of particle size during the



Scheme 1 Synthesis and working procedures of Ca@DNA-MF. (A) The schematic illustration for the synthesis. (B) The working procedures for the synergistic multichannel depolarization of the mitochondrial. (I) The channel of MCT4 silencing for acidosis. (II) The channel of Ca²⁺ overload. (III) The channel of Fenton-like oxidative stress.

preparation was further confirmed by dynamic laser scattering (DLS) analysis (Fig. 1C). Besides, zeta potential characterization revealed the successful integration of MF (+18.64 mV) with DNA networks (-21.25 mV), which formed DNA-MF (-6.54 mV). This was followed by successful biomineralization by ${\rm Ca}^{2^+}$, exhibiting the increased zeta potential of Ca@DNA-MF (-4.76 mV) (Fig. 1D).

The stepped-synthesis of the DNA network-based Ca@DNA–MF was also verified by polyacrylamide gel electrophoresis (PAGE). As shown in Fig. 1E, the C-HCR primers of N1 (lane 2), H1-dimer (lane 3), H2 (lane 4), and the siRNA mimic of a ssDNA (lane 5) exhibited relatively faster migration due to the different molecular weights. When the ssDNA was hybridized to H2 upon the complementation of the 5' sequence of ssDNA to the 3' end of H2, the obtained product of H2-ssDNA migrated much slower (lane 6). This indicated that the siRNA could be efficiently tethered to H2 at the 3' end with the ATP aptamer overhung, which would facilitate the ATP-initiate release of siRNA. Meanwhile the C-HCR product of the DNA network (lane 7) and

the final Ca@DNA-MF (lane 8) were trapped without any separation, verifying the successful synthesis of DNA networks *via* C-HCR. In addition, Ca@DNA-MF was quite stable in complicated biological environments, demonstrated by the fact that no DNA fragment was recorded after incubating with Ca@DNA-MF in 10% fetal bovine serum (FBS)-containing culture medium (Fig. S3†). The DLS results also demonstrated the good stability of Ca@DNA-MF under different physiological conditions (DI water, PBS, and RPMI 1640), which facilitated biological applications (Fig. S4†).

Furthermore, the related chemical composition and element valence state of the stepped synthesis products were verified by X-ray photoelectron spectroscopy (XPS). The co-existence of Ca, Fe and Mn was well recorded in the XPS spectra of MF, DNA–MF and the Ca@DNA–MF nanocomplex (Fig. S5† and 1F). In the XPS high-resolution spectra of Ca@DNA–MF, the Ca²⁺ peaks (Ca 2p_{3/2} at 346.5 and Ca 2p_{1/2} at 346.5 eV) (Ca 2p spectrum, Fig. 1G(i)), peaks of Fe²⁺ (709.7 and 724.08 eV) and Fe³⁺ (711.7 and 726.12 eV) (Fe 2p spectrum, Fig. 1G(ii)), and Mn²⁺ peaks

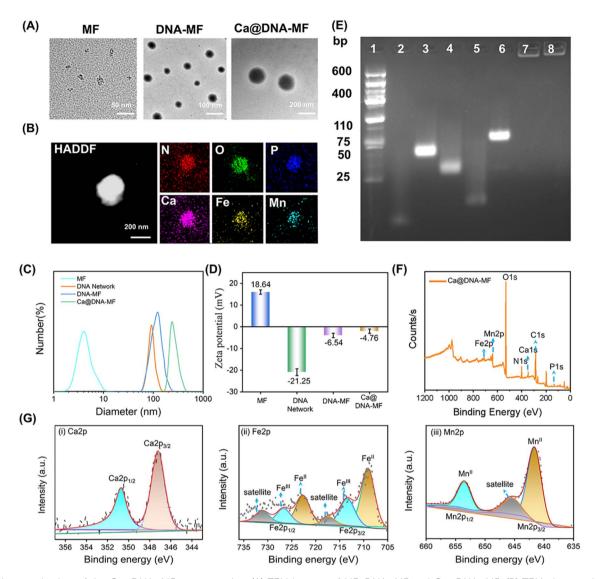


Fig. 1 Characterization of the Ca@DNA-MF nanocomplex. (A) TEM images of MF, DNA-MF and Ca@DNA-MF. (B) TEM elemental mapping images of N, O, P, Ca, Fe and Mn. (C) DLS analysis. (D) Zeta potentials. (E) Polyacrylamide gel electrophoresis images of stepped products in C-HCR-based preparation of Ca@DNA-MF. (1) Ladder, (2) N1, (3) H1-dimer, (4) H2, (5) ssDNA, (6) H2-ssDNA, (7) DNA network, and (8) Ca@DNA-MF. (F) XPS spectrum of Ca@DNA-MF. (G) XPS high-resolution spectrum of Ca 2p (i), Fe 2p (ii) and Mn 2p (iii).

(Mn $2p_{3/2}$ at 641.14 and Mn $2p_{1/2}$ at 653.07 eV) (Mn 2p spectrum, Fig. 1G(iii)) were well recorded. It should be noted that the important Fenton-like regent of Fe species exhibits the coexistence of Fe²⁺ and Fe³⁺, demonstrated by the Fe 2p_{1/2} and Fe 2p_{3/2} peaks. Consequently, the successful synthesis of the Ca@DNA-MF nanocomplex was confirmed.

The acidity-activated ROS generation and GSH consumption by Ca@DNA-MF

As designed, the mitochondrial depolarization could be facilitated by siMCT4-initiated acidosis (channel I in Scheme 1) and Ca²⁺ (channel II). Meanwhile, the acidity would further activate GSH depletion and hydroxyl radical ('OH) generation by a Fenton-like reaction (channel III). This was a synergistic process, in which the acidification of channel I also facilitated

the other two channels. To verify the synergistic roles of acidification for the other two mitochondrial dysfunction channels, the characterization of ion release, 'OH generation and GSH consumption was carried out. Herein, the ion release (channel II) was evaluated by coupled plasma optical emission spectrometry (ICP-OES) tests. The generation of 'OH and the depletion of GSH in channel III were evaluated by monitoring the decreased absorption of 3,3',5,5'-tetramethylbenzidine (TMB) and 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB), respectively. The redox recycling between Fe³⁺/Fe²⁺ and Mn²⁺/Mn^{x+} facilitated the GSH depletion and 'OH generation (Fig. 2A).

As demonstrated by ICP-OES analysis (Fig. 2B), the release of Ca, Fe and Mn was inefficient under physiological conditions at pH 7.4. This indicated the good stability and low systemic toxicity of Ca@DNA-MF, preventing the leakage of reagents into

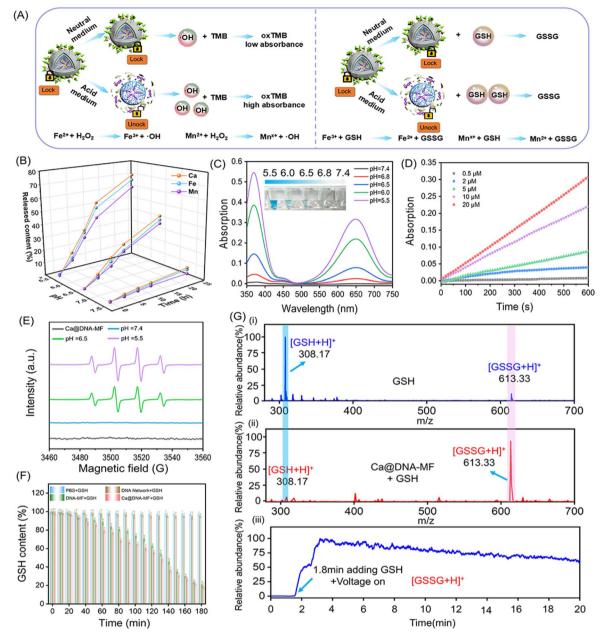


Fig. 2 Evaluation of the acidosis-based synergistic multichannel mitochondrial dysfunctions by Ca@DNA-MF. (A) Schematic illustration for the catalytic 'OH generation, which facilitated the oxidation of TMB (colorless) to oxTMB (blue) or oxidation of GSH to GSSG. (B) Profiles of the release of Ca^{2+} , Fe^{3+} and Mn^{2+} during the incubation of Ca@DNA-MF in HEPES under different pH conditions. (C) TMB oxidation by 100 μ M H_2O_2 in the presence of Ca@DNA-MF under different conditions. (D) Time-dependent absorbance of oxTMB by different amounts of H_2O_2 in HEPES buffer (pH = 6.5) with Ca@DNA-MF presented. (E) ESR spectra under different pH conditions using 5,5-dimethyl-1-pyrroline N-oxidation (DMPO) as the 'OH scavenger. (F) Absorption of GSH after the depletion by different nanomaterials. (G) GSH residual concentration in PBS (i), Ca@DNA-MF (ii) and Ca@DNA-MF at different times (HEPES pH 6.5) (iii).

blood circulations. While a relatively faster ion release was observed under acidic conditions (pH 5.5 and pH 6.5), and the highest release was exhibited at pH 5.5. Therefore, the TME (acidic condition)-initiated collapse of Ca@DNA–MF facilitated the subsequent mitochondrial dysfunction by Ca²⁺ (channel II) and by Fenton-like therapy (channel III). In addition, TME responsiveness would minimize the premature leakage-induced systematic toxicity and concurrently facilitate precise spatiotemporal therapy at tumor sites.

Besides, the Fenton-like therapy also exhibited aciddependent enhancements. For the Fenton-like therapy, the delivered Fe³⁺ was firstly *in situ* reduced to Fe²⁺ to effectively consume GSH. This was confirmed by the increased absorption of the Fe²⁺ probe (1,10-phenanthroline) at 510 nm. As a result, the obtaining of Fe²⁺ was enhanced under acidic conditions, indicated by the higher absorption at pH 5.5 than at pH 6.5 and 7.4 (ESI Fig. S6†). This was consistent with the pH-responsive morphological variations in TEM and DLS characterization (Fig. S7†). Therefore, acidosis can facilitate the release of the mitochondrial dysfunction reagents, which would initiate the subsequent multichannel mitochondrial depolarization.

Inspired by the reduction of Fe³⁺ to Fe²⁺, the toxic hydroxyl radical ('OH) was generated via a Fenton-like reaction to decompose H₂O₂. This was evaluated by 3,3',5,5'-tetramethylbenzidine (TMB) oxidation. As demonstrated (Fig. 2C and S8†), the absorption of oxTMB (generated by 'OH oxidation) increased with the decrease of the pH value, exhibiting increased blue colors (the inset image). Meanwhile poor catalytic activity was observed at pH 7.4, which further confirmed the activation of catalytic activity for generating 'OH under acidic conditions. This was in accordance with the decreased H2O2 signals (with titanium sulfate as the probe, Fig. S9†) upon H₂O₂ consumption. Besides, the steady-state kinetics of the catalytic generation of 'OH by peroxidase-like coversion were also examined. The kinetic curves of the catalytic conversion of different concentrations of H₂O₂ to 'OH were recorded in a slightly acidic environment (Fig. 2D and S10†), which resulted in the maximum initial velocity $(V_{\rm max})$ and a Michaelis-Menten constant $(K_{\rm m})$ of 4.37 \times 10^{-8} M s⁻¹ and 15.06 μ M, respectively. This 'OH generation was also confirmed by the increased FL signals of terephthalic acid (TA) upon the oxidation of TA by 'OH (Fig. S11†). In addition, the acidic-dependent generation of 'OH was also confirmed by electron spin resonance (ESR) spectroscopic analysis, which exhibited increased characteristic 'OH signals (1:2:2:1) with the decrease of pH values (Fig. 2E). Besides, no obvious O2. signal was observed in the Ca@DNA-MF system and the groups in the presence of air or H₂O₂ at pH 5.5, 6.5 and 7.4 (Fig. S12†). Therefore, the good catalytic capability of Ca@DNA-MF was confirmed, which can generate efficient 'OH for the subsequent mitochondrial dysfunction in channel III.

Furthermore, the depletion of GSH in channel III was evaluated by comparison of 'OH and GSH under different conditions. Moreover, the depletion of intracellular GSH to GSSG is beneficial for improving the efficiency of Fenton-like reactions. As demonstrated, with adding GSH into the H₂O₂-Ca@DNA-MF system, the amount of 'OH increased to a certain extent (0-9 mM), indicated by the TMB degradation test (Fig. S13A†). This is in accordance with the depletion of GSH (Fig. 2F) for converting Fe³⁺ into Fe²⁺, which facilitated the 'OH generation by the Fenton-like process (Fig. S13C†). Meanwhile with too much GSH added (higher than 15 nM), 'OH obviously decreased due to the consumption of 'OH by excessive GSH (Fig. S13B†). Simultaneously, the depletion of GSH was also confirmed by the decreased absorption of 5,5'-dithiobis-(2nitrobenzoic acid) (DTNB) during the incubation (Fig. S14†). Furthermore, the GSH consumption was monitored by ambient mass spectrometry, which directly extracted and ionized samples without any pre-treatment (Section 4.4 in the ESI†). As demonstrated by both off-line (Fig. 2G(i) and (ii)) and on-line examinations (Fig. 2G(iii)), with Ca@DNA-MF treatment, the characteristic GSH ion of $[GSH + H]^+$ (at m/z 308) dramatically decreased along with the significant increase of [GSSG + H]⁺. Therefore, the depletion of endogenous GSH

(lower than 10 mM of GSH expression in the TME) could facilitate the generation of 'OH for synergetic Fenton-like therapy (channel III).⁴⁴

The ATP-triggered intracellular delivery of mitochondrial dysfunction reagents

To explore the on-demand rapid ATP-triggered release and gene knockdown in cells, siRNA expression was evaluated for examining the release behavior. For visualized monitoring, DNA networks were labelled with red FL dye of TAMRA, whose FL signals were quenched by BHQ2 linked to the siRNAs upon fluorescence resonance energy transfer (FRET). As illustrated in Fig. 3A, in the presence of ATP, the siRNA was released along with the turn-on red signals due to the breaking of the FRET system. This process was initiated by ATP in the TME, which was demonstrated by the highest FL signals in the group of Ca@DNA-MF (functionalized with the ATP aptamer), relative to the negligible fluorescence signal in the nATP-Ca@DNA-MF group (without the ATP aptamer) (Fig. 3B, S15A and B†). In addition, this release rate was a time-dependent process, which was illustrated by PAGE analysis of H2-siRNA products after incubation for different times. As shown in Fig. 3C, the amount of H2-siRNA decreased during the incubation and satisfactory release can be achieved within 0.5 h. Meanwhile, no significant dissociation was recorded for the ssDNA complex with a scramble sequence of the ATP aptamer (donated as nATP-H2) even after incubation for 3 h (Fig. S15C†). Therefore, the ATPtriggered siRNA reagent release has been successfully designed, which would facilitate intracellular delivery and therapy applications.

Thereafter, the delivery of the mitochondrial dysfunction reagents was examined by confocal laser scanning microscopy (CLSM) imaging of 4T1 cell lines. As a result, a significantly increased red signal of TAMRA (indicating the release of siR-NAs) was recorded during the incubation of the 4T1 cells with ATP aptamer-functionalized Ca@DNA-MF, even exhibiting obvious red signals at 4 h (Fig. 3D(i)). Meanwhile without the ATP aptamer (nATP-Ca@DNA-MF), a quite weak signal of TAMRA was observed even after 24 h of incubation (Fig. 3D(ii)). The important role of the ATP aptamer in the siRNA delivery was further demonstrated by flow cytometry analysis. As a result (Fig. 3E), a dramatically higher uptake of the reagents than that without the ATP aptamer was exhibited, implying effective cellular uptake for therapeutic applications. Time-dependent cell internalization via endocytosis was also demonstrated according to subcellular localization experiments. Indicated by Pearson's correlations (Fig. S16A†), the initial overlapped signals of Ca@DNA-MF and lysosomal were separated with time prolonging, suggesting efficient endosomal escape of Ca@DNA-MF. Meanwhile without the Ca cover, the DNA network group was stably trapped in lysosomes (Fig. S16B†). This was generated from the rapid decomposition of the Ca cover in acidic lysosomes, which facilitated the production of ROS to improve lysosomal escape for the efficient delivery of multiple reagents.45

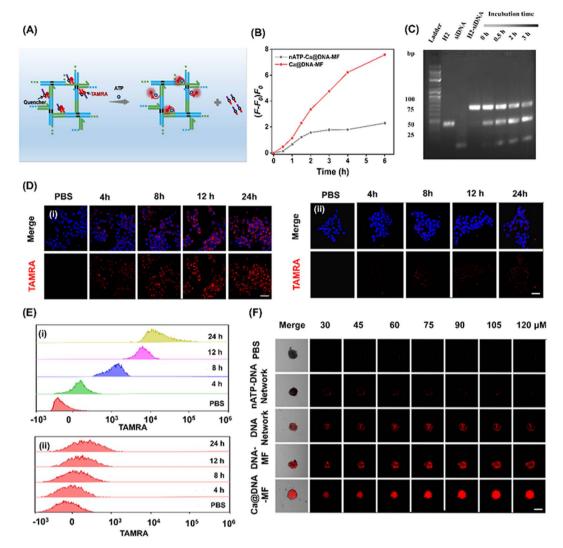


Fig. 3 Examination of the ATP-triggered delivery of mitochondrial dysfunction reagents. (A) Schematic illustration of ATP-triggered release of siRNAs, which emitted "turn-on" FRET-based red FL signals of TAMRA that was labelled on siRNAs. (B) The increase of FL signals *versus* time in the presence of Ca@DNA-MF with or without the ATP-aptamer linked, respectively. (C) PAGE image for products of ATP-triggered collapse of H2-ssDNA (2 μM), after incubation at 37 °C for different times. Intercellular ATP was 5–10 mM. (D) Cell imaging of 4T1 cells after being incubated with Ca@DNA-MF with (i) and without (ii) the ATP-aptamer and the nuclear dye Hoechst (blue). Red signal of TAMRA indicated the location of siRNA. Scale bars: 50 μm. (E) Mean FL of flow cytometry data with (i) and without the ATP aptamer (ii). (F) CLSM Z-stack scanning of 3D multicellular spheroids of 4T1 cells treatment with different nanomaterials. Scale bars: 500 μm.

Besides, the modification with HA was also important for tumor-targeted uptake, verified by the 3 times weaker cellular uptake of DNA-MF (without HA modified) than that of Ca@DNA-MF (Fig. S17 and S18†). The tumor-targeted intracellular delivery was further confirmed by the imaging comparison of different cell lines, including cancer cells (MCF-7, A549 and HeLa) and normal cells (HEK-293T, with low CD44 antigen expression). As a result, 20 times stronger cellular uptake was exhibited for the cancer cells than the normal cells, attributed to low expression of the CD44 antigen that selectively bound to HA on Ca@DNA-MF (Fig. S19A and B†). The role of HA in the selective targeting of tumor cells was also confirmed by flow cytometry analysis (Fig. S19C†). The study on the cellular uptake pathway demonstrated that the entrance of Ca@DNA-MF into 4T1 cells was inhibited by specific endocytic inhibitors

and at low temperature. This indicated that the cellular uptake of Ca@DNA-MF was an energy-dependent endocytosis pathway (Fig. S20†). Additionally, the tumor infiltration properties were also evaluated by the imaging of 4T1 cell-based multicellular spheroids (MTSs) after being treated with different TAMRA-labeled nanonetworks. As demonstrated in Fig. 3F, no obvious red signal was observed in each depth group without the ATP-aptamer (nATP-DNA network). Alternatively, the strongest FL signals of tumor spheroids were recorded after being incubated with Ca@DNA-MF, indicating the best infiltration and targeted properties with both the ATP aptamer and HA functionalized on Ca@DNA-MF. In addition, the acidic conditions facilitated the reagent delivery, demonstrated by the much obvious red signals (Fig. S21A and B†) and higher cell uptakes (Fig. S21C†) at lower pH values. Therefore, the present Ca@DNA-MF exhibited the

Edge Article

efficient tumor-targeted and safe delivery of mitochondrial dysfunction reagents with minimized off-target toxicity.

The performance of the mitochondrial dysfunction by remodulating of H⁺/Ca²⁺ gradients

The influence of Ca@DNA-MF on mitochondrial pH, morphological membrane potential (MMP), Ca²⁺ expression and ATP production in 4T1 cells was further evaluated. As illustrated in Fig. 4A, the level of free Ca²⁺ increased observably, owing to the disturbance of intramitochondrial Ca2+ homeostasis upon Ca@DNA-MF decomposition. The released siMCT4 initiated drastic intracellular acidification and facilitated the H⁺ influx/ acidification in the mitochondrial matrix. This was further synergistically promoted by the overloaded Ca²⁺, which resulted

in the dramatically increased penetration of H⁺/Ca²⁺ for depolarization in the internal mitochondrial matrix. Meanwhile, the mitochondrial depolarization would decrease the ATP expression, which induced low cell migration with low migration energy. More significantly, the efficient mitochondrial depolarization finally induced apoptosis along with the increase of apoptotic-pathway-related proteins such as cytochrome C and caspase-3.

To evaluate the performance of siMCT4-initiated intracellular acidification, the intracellular pH values were determined by cell imaging with BCECF-AM as the pH probe. With 4T1 cells as models (Fig. 4B), the intracellular pH was about 7.3 after being treated with PBS. Meanwhile the intracellular pH decreased and the lowest value was recorded after being treated

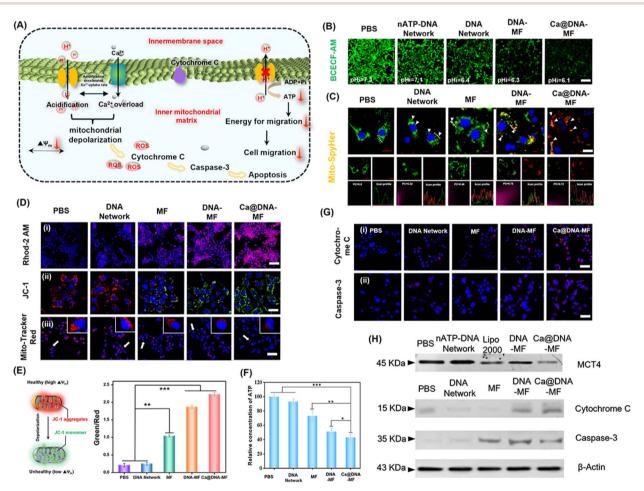


Fig. 4 Examination of the mitochondrial dysfunction of 4T1 cells by different nano-networks. (A) Schematic illustration of mitochondria depolarization caused by remodulate H^+/Ca^{2+} gradients. (B) Imaging of extracellular (PBS) and intracellular (cytoplasm) pH levels with DCECF-DA as the pH probe, which exhibited increased green signals at higher pH values. Scale bars: 100 µm. (C) Imaging of mitochondrial pH values in 4T1 cells transfected with pH-sensitive Mito-SypHer. The increased ratio of red (480 nm)/green (430 nm) indicated higher pH values. Scale bars: 10 μ m. (D) Mitochondrial imaging for Ca²⁺ evaluation (showing purple signals with Rhod-2 AM as the probe). (i) Scale bars: 100 μ m. Mitochondrial membrane potential measurement (with JC-1 as the probe, exhibiting the conversion of signals from red to green with mitochondrial dysfunction). (ii) Scale bars: 10 µm. The mitochondrial distribution with Mitochondria-Tracker Red as the probe to exhibit red signals (iii). Scale bars: 50 μ m. (E) Corresponding quantitative analysis of (D). ***p < 0.001 and n.s.: no significance. Data presented as mean \pm SD (n = 3). (F) Relative concentration of intracellular ATP. Bars represent mean \pm SD (n=3 independent samples) (*P < 0.05; ***P < 0.001, calculated using a two-tailed unpaired t test). (G) Expression of cytochrome C (i) and total caspase-3 (ii) in 4T1 cells after different treatments for 24 h. Scale bars: 50 μm. (H) Western blotting assay of MCT4 inhibition and the mitochondria apoptosis indicated by the contents of cytochrome C and caspase-3. Control: \(\beta\)-actin.

with Ca@DNA-MF (pH 6.1). This confirmed the siMCT4initiated cytosolic acidity via blocking of LA effusion (channel I). Furthermore, the highest intracellular LA level of Ca@DNA-MF further confirmed the successful MCT4 blockage (Fig. S22†). Besides, the pH values in the supernatant of Ca@DNA-MF further validated similar results to the above investigations (Fig. S23†). Therefore, the siMCT4-induced acidosis could be regulated by the LA metabolic content to remodel the tumor intracellular environment. To further examine the influx of H⁺ into the mitochondrial matrix, the mitochondrial pH values (pH-mito) were monitored with a pH-sensitive probe of mitochondrial FL protein Mito-SypHer. As a result (Fig. 4C), the significant H+ influx into mitochondria was recorded after being treated with Ca@DNA-MF, which was indicated by the obvious fluorescence emission-shift from green to red (white arrows). This would therefore facilitate the decrease of H gradient of the inner/outer mitochondria, for achieving mitochondrial depolarization. The mitochondrial dysfunction was further confirmed by the round-shaped 4T1 cells, accompanied by obvious matrix swelling and mitochondrial fission after being treated with Ca@DNA-MF (Fig. S24†).

In addition, the expression of Ca²⁺ in both cytoplasm and mitochondria was evaluated to support the Ca2+-based mitochondrial depolarization in channel II. From the cell imaging of cytosolic Ca²⁺, the strongest FL signal of Ca²⁺ (indicated by the green probe of Fluo-4 AM) was recorded in the group of Ca@DNA-MF (Fig. S25†), further confirming the pH-activated intracellular release of Ca²⁺. Besides, the intracellular acidification in channel I would accumulate H⁺ in the mitochondrial intermembrane. This would trigger the opening of the mitochondrial permeability transition pore, accelerating a significant influx of H⁺ and Ca²⁺ and resulting in mitochondrial dysfunction. 46 Thereafter, the Ca²⁺ expression in mitochondria was evaluated by the mitochondrial Ca²⁺ probe of Rhod-2 AM (Fig. 4D(i)), which indicated the highest Ca2+ signal in the Ca@DNA-MF group. This Ca²⁺ expression was more significant than that in the DNA-MF group without Ca2+ coating, which further confirmed the good synergistic mitochondrial dysfunction of Ca²⁺ in channel II.

To evaluate the performance of the mitochondrial depolarization, a series of the characterization techniques have been employed. Firstly, the membrane potential $(\Delta \psi)$ was evaluated by using a JC-1 probe to indicate the proton gradient across the mitochondrial membrane. The increased ratio of the green/red signal indicated the decrease of $\Delta \psi$ upon the conversion of JC-1 aggregates (red) into monomers (green) via the mitochondrial depolarization (Fig. 4E). As a result (Fig. 4D(ii)), the Ca@DNA-MF group displayed the lowest $\Delta \psi$, resulting in an \sim 6.7-fold green/red ratio relative to the untreated one (Fig. 4E). This further confirmed the satisfactory mitochondrial depolarization by Ca@DNA-MF. This was also in accordance with the smallest number of mitochondria with severe mitochondrial damage by Ca@DNA-MF, which was demonstrated by the intracellular distribution of mitochondria staining with Mito-Tracker Deep Red (Fig. 4D(iii)). In addition, the most significant mitochondrial permeability transition pore (mPTP) activation upon mitochondrial depolarization with Ca@DNA-MF

confirmed by the calcein-AM loading/CoCl₂ quenching strategy (Fig. S26†).⁴⁷ Consequently, the significant mitochondrial depolarization by Ca@DNA-MF has been verified.

Next, intracellular adenosine triphosphate (ATP) was measured as a direct embodiment of LA metabolic reprogramming. As expected, the lowest ATP level was recorded in the Ca@DNA-MF group (Fig. 4F), which was attributed to the oxidative phosphorylation inside mitochondria. Given that ATP was defined as the cell "power plant" of mitochondria, the ATPdependent cell migration was inhibited upon mitochondrial dysfunction and was demonstrated by the wound healing assay (Fig. S27†). In addition, the corresponding biomarkers of the mitochondria dysfunction (including the inhibition of MCT-4, the release of cytochrome C and the activation of the caspase family such as caspase-3) were also evaluated, upon mitochondria-related apoptotic cascades. As illustrated by imaging of 4T1 cells (Fig. 4G), the significantly increased signals of cytochrome C and caspase-3 were released from the mitochondria into the cytosol after being treated with Ca@DNA-MF. This was in accordance with the western blotting assay, which exhibited significantly decreased expression of MCT-4, and upregulated expression of cytochrome C and caspase-3 (Fig. 4H and S28†). Therefore, the significant mitochondrial dysfunction via the remodulating of H⁺/Ca²⁺ gradients by Ca@DNA-MF has been confirmed, which would facilitate the subsequent efficient apoptosis of cancer cells.

Generation of ROS for LPO-based cell apoptosis

Upon the synergistic multichannel mitochondrial dysfunction, the mitochondrial was damaged to facilitate the generation of ROS for LPO-based apoptosis. As illustrated in Fig. 5A, upon specific targeting of HA (on Ca@DNA-MF) to the CD44 receptor on the membrane of cancer cells, Ca@DNA-MF was endocytosed into cancer cells. This facilitated the release of multiple regents for mitochondrial dysfunctions in three channels, which induced the significant enhanced generation of ROS (i.e., H⁺/Ca²⁺ transport modulation and GSH depletion). Consequently, the high expression of ROS initiated the decrease of reductive protein of Bcl-2, which prevents apoptotic mitochondrial signaling by inhibiting the oligomerization and activation of proapoptotic factors. In addition, the increased ROS would initiate the apoptosis upon oxidation of R-OH to R-OOH, with the product of MDA as the significant biomarker. Besides, the oxidized GSH (GSSG) in channel III would also result in the decreased expression of reductive GPX-4. Considering that the reductive GPX-4 exhibited the capability of lipid reduction, the apoptosis would be further facilitated due to the inhibition of lipid reduction with the low expression of GPX-4.

Firstly, to examine the generation of ROS and level of LPO for the cell apoptosis, 4T1 cells were incubated with different nanomaterials for comparison. As demonstrated (Fig. 5B and C), the most significant ROS signals were recorded for 4T1 cells treated with Ca@DNA-MF, indicated by the ROS probe of dichlorofluorescein diacetate (DCFH-DA). This confirmed the accumulation of ROS by Ca@DNA-MF, which would elicit the mitochondrial depolarization and redox dyshomeostasis for the

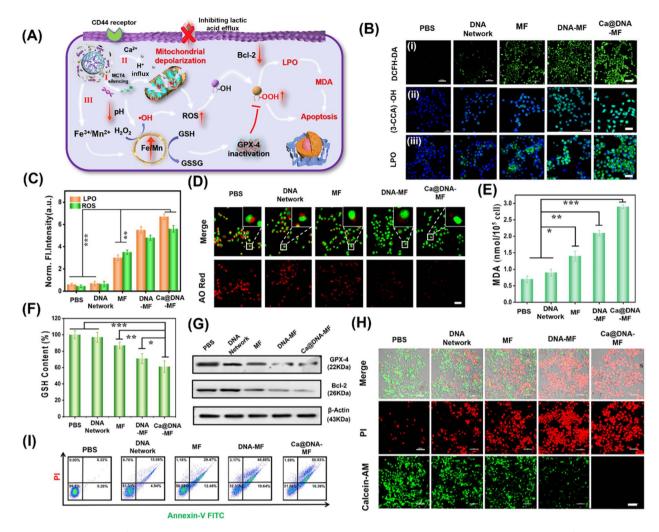


Fig. 5 Evaluation of the LPO-based mitochondrial dysfunction after treating 4T1 cancer cells with different groups of PBS, DNA networks, MF, DNA-MF and Ca@DNA-MF. (A) Schematic illustration of ROS generation for the LPO-based mitochondrial dysfunction. (B) CLSM images. (i) Intracellular ROS levels with DCFH-DA as the indicator. Scale bars: 100 μm. (ii) 'OH levels. Scale bars: 50 μm. (iii) LPO levels. Scale bars: 10 μm. The 4T1 cells were incubated with different nanonetworks for 24 h. Scale bars: 100 μm. (C) The fluorescence intensity of LPO and *OH in (B). (D) The imaging of the endosomal membrane content after 24 h of the treatments, based on red signals from AO staining. Scale bars: 10 μm. (E) Content of the LPO product of MDA after treating with different nanomaterials. The statistical analysis was performed in contrast to a control group (*p < 0.05, **p < 0.01, ***p < 0.001, and t-test). (F) GSH content. (G) Western blotting assay of intracellular GPX-4 and Bcl-2 proteins, indicating the therapy performance. (H) CLSM images of 4T1 cells upon different treatments and stained with calcein-AM/PI; scale bars: 100 µm; (I) flow cytometric apoptosis analysis of 4T1 cells stained with Annexin V-FITC/PI.

therapy. Besides, the highest fluorescence levels of 'OH and LPO were observed in the Ca@DNA-MF group and no obvious signal was recorded in the control group (Fig. 5B). This indicated the tremendous upregulation of LPO by enhancing the ROS generation for efficient cell apoptosis, which was also consistent with the flow cytometry analysis of the LPO level (Fig. 5B(iii) and S29†). This was also confirmed by the obvious endosomal membrane rupture, which was indicated by dramatically decreased red signals of acridine orange (AO) (Fig. 5D).

Furthermore, some biomarkers or important therapyrelated molecules were evaluated. Being a typical end product of LPO, the intracellular malondialdehyde (MDA) was highest when cells were treated with Ca@DNA-MF, in accordance with LPO imaging (Fig. 5E). Given that GSH depletion can inactive GPX-4 for cell apoptosis, GSH consumption activity was determined after various treatments. As expected, the most efficient depletion of GSH by Ca@DNA-MF was also confirmed (Fig. 5F), which was generated from the conversion of Fe³⁺ into Fe²⁺ for efficient Fenton-like therapy. The down-regulation expression of GPX-4 was determined by western blotting assay (Fig. 5G-GPX-4), which could be attributed to the oxidation of GSH to GSSG. In addition, the expression of Bcl-2 (the key anti-apoptosis protein) also decreased, which was generated from the modulation of metabolic phenotypes in the tumor cells (Fig. 5G-Bcl-2 and S30†). Therefore, efficient mitochondrial dysfunction-based apoptosis was obtained by Ca@DNA-MF, due to the

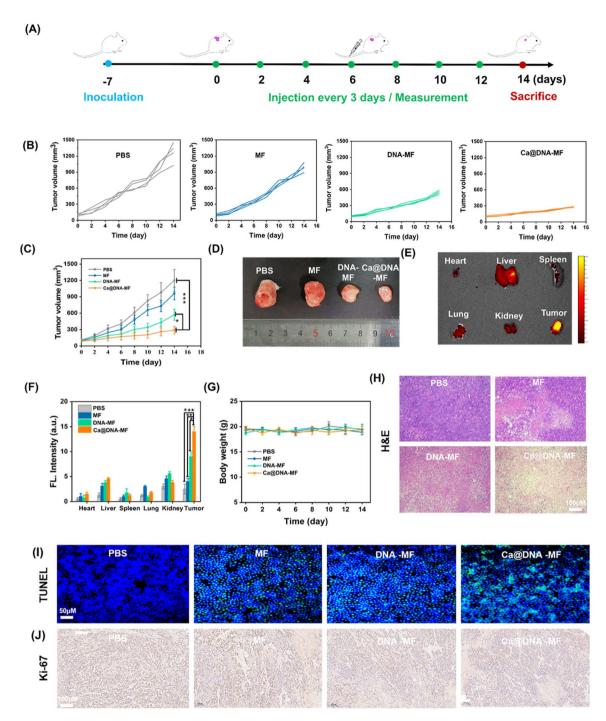


Fig. 6 In vivo explorations on tumor targeting and antitumor activities. (A) Scheme of the establishment of the 4T1 tumor xenograft model and the treatment process. (B) Individual tumor growth curves and (C) relative growth curves of 4T1 tumors in different treatment groups (*p < 0.05, **p < 0.01, ***p < 0.001, and t-test). (D) Photographs of excised 4T1 tumors on the 14th day after different treatments. (E) FL images of tissue and (F) quantitative results of the average FL signals of the major organs (*p < 0.05, **p < 0.01, ***p < 0.001, and t-test). (G) Body-weight of 4T1 tumor-bearing female mice after different treatments. (H) Histochemical analyses (H&E) stained, (I) TUNEL-stained and (J) Ki67-stained of tumor tissues after different treatments for 14 days.

provoking of excessive ROS by Ca@DNA-MF *via* multistage synergistic processes (channel II to channel III).

To investigate the *in vitro* therapeutic effects of Ca@DNA–MF, cell viability was evaluated using standard MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-*H*-tetrazolium bromide)

assay. As a result, Ca@DNA–MF exhibited dose-dependent cytotoxicity on 4T1 cells at a CD50 value of 128.4 μg mL⁻¹ (ESI Fig. S31A†). In addition, negligible influence on the cell viability was observed for the HEK-293T cell (85.6%) compared to the 4T1 cancer cell (19.8%), which owe to the high express H_2O_2 and

Edge Article Chemical Science

GSH levels in cancer cells. Furthermore, enhanced cytotoxicity was recorded at lower pH values (ESI Fig. S31C†), which was in accordance with the designed acidosis-based synergistic multichannel mitochondrial depolarization. The higher cytotoxicity under acidic conditions could be generated from the overloading of Ca²⁺ in the mitochondrial matrix upon the enhanced release of Ca²⁺ at lower pH values. More significantly, more ROS species would be generated due to the enhanced Fenton-like reactivity under acidic conditions, which consequently resulted in enhanced cytotoxicity. Furthermore, the performance of cellular apoptosis by Ca@DNA-MF was evaluated by calcein-AM and PI cell staining. As shown in Fig. 5H, the most obvious dead cells were obtained by Ca@DNA-MF after 24 h of incubation. Moreover, Annexin-V-FITC/PI-based flow cytometry analysis was also carried out (Fig. 5I). As a result, the apoptosis rates for the groups of PBS, DNA networks, MF, DNA-MF, and Ca@DNA-MF were 0.22%, 13.06%, 29.47%, 44.88%, and 50.53%, respectively. Therefore, the multichannel mitochondrial dysfunction behavior of Ca@DNA-MF for inducing admirable cancer cell apoptosis was confirmed.

In vivo examinations

Finally, the in vivo therapeutic performance of Ca@DNA-MF was assessed with 4T1 tumor-bearing mice as the models. The PBS, MF, DNA-MF, and Ca@DNA-MF were intravenously injected every three days by the tail vein, and the tumor volumes and body weights were monitored every 3 days for 14 days (Fig. 6A). As indicated by the tumor growth curves (Fig. 6B and C), the best tumor inhibition efficacy was obtained after being treated with the present Ca@DNA-MF, resulting the significant tumor suppression of 70.4% on the 14th day. This was in accordance with the smallest tumor for the group treated with Ca@DNA-MF (Fig. 6D). The distinctively higher inhibition efficiency of the Ca@DNA-MF group than that of the DNA-MF group indicated the significant permeability and targeted effect of the nano-reagents covered by Ca@HA. Besides, the accumulation and retention of Ca@DNA-MF at tumor sites were confirmed by the most significant FL signals (from the TAMRAlabeled Ca@DNA-MF) at the tumor, demonstrated by both FL images (Fig. 6E and S32†) and the quantitative data (Fig. 6F). Therefore, the present Ca@DNA-MF exhibited efficient therapeutic efficacy upon the efficient targeting of tumors.

Moreover, the biosafety of Ca@DNA-MF was evaluated by both hemolysis assay and *in vivo* experiments. As a result, no significant hemolysis was observed during incubating mice blood with even 200 μg mL⁻¹ of Ca@DNA-MF (Fig. S33†). In addition, on obvious change of the bodyweight of the mice was observed during the *in vivo* treatment (Fig. 6G), also indicating the good biosafety of the formulations. Thereafter, the hematoxylin and eosin (H&E) staining of the major organs and tumor tissues was employed. As a result, severe fibrosis with an incomplete cellular morphology and many disintegrated nuclei was observed in the Ca@DNA-MF group, and no evident tissue necrosis was observed in the other groups (Fig. 6H). In addition, no obvious tissue necrosis was also recorded in other major organs after being treated with Ca@DNA-MF (Fig. S34†).

Furthermore, the most significant cell apoptosis and the tumor cell proliferation inhibition by Ca@DNA–MF were confirmed *via* evaluating apoptosis-related proteins by terminal deoxynucleotidyl transferase-mediated dUTP nick-end labelling (TUNEL) (Fig. 6I) and Ki67 staining (Fig. 6J). This was generated from the deactivation of Ca@DNA–MF in the normal tissues, without the activation species of high ATP expression or lower pH values. This can avoid the premature leakage of reagents ahead of arriving at the tumor, potentially reducing the off-target toxicity in normal tissues and achieving tumor-specific therapy.

Conclusion

A supramolecular DNA nanocomplex of Ca@DNA-MF was successfully prepared for the remodulation of H⁺/Ca²⁺ gradients to induce mitochondrial depolarization. The remodulation not only triggered apoptosis but also led to ferroptosis through the enhancement of ROS, which obtained the synergistical enhancement of cancer therapy. Upon targeted and TMEtriggered intracellular delivery, multiple reagents including Ca²⁺, the MCT4 gene silencing inhibitor of siMCT4 and Fentonlike reagents were released into cancer cells. Subsequently, MCT4 gene silencing induced acidosis to enhance the conversion of endogenous H₂O₂ into highly toxic 'OH. Simultaneously, GSH was oxidized to GSSG to facilitate ferroptosis upon the irreversible GPX-4 inactivation and lipid peroxidation process. Therefore, based on the three channels of the therapy, multiple and synergistic mitochondrial depolarization was obtained via the remodulating of H⁺/Ca²⁺ gradients. This dramatically initiated the significant generation of excessive ROS species by regulating the H⁺/Ca²⁺ gradient influx in the matrix, TMEactivated GSH depletion and abundant 'OH generation. Therefore, the intrinsic redox homeostasis was disturbed for LPO-based apoptosis. By both cell imaging and in vivo experiments, synergistic cancer therapy was achieved with low toxicity, providing an efficient strategy for enhanced cancer therapy.

Ethical statement

All animal experiment protocols were reviewed and approved by the Animal Care and Use Committee of Institute of Beijing Normal University and complied with all relevant ethical regulations.

Data availability

All relevant data is presented in the manuscript and ESI.† Raw data is available upon request by email to the corresponding author.

Author contributions

X. Wang and N. Na conceived and designed the project. X. Wang performed the experiments. X.Ge, X. Guan supported figure

preparations and J. Ouyang supported the characterizations. N. Na directed the whole research.

Conflicts of interest

Chemical Science

There are no conflicts to declare.

Acknowledgements

We gratefully acknowledge the National Natural Science Foundation of China (NNSFC, 22274012 and 21974010).

Notes and references

- M. Fan, J. Zhang, C. Tsai, B. Orlando, M. Rodriguez, Y. Xu,
 M. Liao, M. F. Tsai and L. Feng, *Nature*, 2020, 582, 129–133.
- 2 K. Lee, S. Park, K. Lee, S. Kim, H. Kim, Y. Meroz, L. Mahadevan, K. Jung, T. K. Ahn, K. K. Parker and K. Shin, *Nat. Biotechnol.*, 2018, **36**, 530–535.
- 3 W. Li, S. Yin, Y. Shen, H. Li, L. Yuan and X. Zhang, *J. Am. Chem. Soc.*, 2023, **145**, 3736–3747.
- 4 S. Cassim, M. Vučetić, M. Ždralević and J. Pouyssegur, *Cancers*, 2020, **12**, M1119.
- 5 L. Cui, A. M. Gouw, E. L. LaGory, S. Guo, A. N. ttarwala, Y. Tang, J. Qi, Y. Chen, Z. Gao, K. M. Casey, A. A. Bazhin, M. Chen, L. Hu, J. Xie, M. Fang, C. Zhang, Q. Zhu, Z. Wang, A. J. Giaccia, S. S. Gambhir, W. Zhu, D. W. Felsher, M. D. Pegram, E. A. Goun, A. Le and J. Rao, *Nat. Biotechnol.*, 2021, 39, 357–367.
- 6 E. Gouaux and R. MacKinnon, Science, 2005, 310, 1461-1465.
- 7 S. Matsuyama, J. Llopis, Q. L. Deveraux, R. Y. Tsien and J. C. Reed, *Nat. Cell Biol.*, 2000, 2, 318–325.
- 8 P. Zheng, B. Ding, G. Zhu, C. Li and J. Lin, *Angew. Chem., Int. Ed.*, 2022, **61**, e202204904.
- 9 P. Zheng, B. Ding, R. Shi, Z. Jiang, W. Xu, G. Li, J. Ding and X. Chen, Adv. Mater., 2021, 33, 2007426.
- 10 C. Giorgi, S. Marchi and P. Pinton, *Nat. Rev. Mol. Cell Biol.*, 2018, **19**, 713–730.
- 11 Y. Mari, C. Katnik and J. Cuevas, Cell Calcium, 2010, 48, 70-82.
- 12 C. Xue, M. Li, C. Liu, Y. Li, Y. Fei, Y. Hu, K. Cai, Y. Zhao and Z. Luo, *Angew. Chem., Int. Ed.*, 2021, **60**, 8938–8947.
- 13 J. Cheng, Y. Zhu, X. Xing, J. Xiao, H. Chen, H. Zhang, D. Wang, Y. Zhang, G. Zhang, Z. Wu and Y. Liu, *Theranostics*, 2021, 11, 5418–5429.
- 14 C. Zhang, L. Chen, Q. Bai, L. Wang, S. Li, N. Sui, D. Yang and Z. Zhu, ACS Appl. Mater. Interfaces, 2022, 14, 27720–27732.
- C. Yang, M. Wang, M. Chang, M. Yuan, W. Zhang, J. Tan,
 B. Ding, P. Ma and J. Lin, J. Am. Chem. Soc., 2023, 145, 7205–7217.
- 16 S. Puri and K. Juvale, Eur. J. Med. Chem., 2020, 199, 112393.
- 17 L. Xu, G. Tong, Q. Song, C. Zhu, H. Zhang, J. Shi and Z. Zhang, ACS Nano, 2018, 12, 6806–6818.
- 18 P. Zheng, B. Ding, R. Shi, Z. Jiang, W. Xu, G. Li, J. Ding and X. Chen, *Adv. Mater.*, 2021, **33**, e2007426.
- 19 X. Meng, D. Li, L. Chen, H. He, Q. Wang, C. Hong, J. He, X. Gao, Y. Yang and B. Jiang, ACS Nano, 2021, 15, 5735–5751.

- 20 D. Wang, H. Wu, C. Wang, L. Gu, H. Chen, D. Jana, L. Feng, J. Liu, X. Wang and P. Xu, *Angew. Chem.*, *Int. Ed.*, 2021, 133, 3038–3044.
- 21 S. Guo, Z. Li, J. Feng, W. Xiong, J. Yang, X. Lu, S. Yang, Y. Xu, A. Wu and Z. Shen, *Nano Today*, 2022, 47, 101663.
- 22 W. Wang, F. Fu, Z. Huang, W. Wang, M. Chen, X. Yue, J. Fu, X. Feng, Y. Huang, C. Wu and X. Pan, ACS Nano, 2022, 16, 8370–8387.
- 23 Y. Chen, Y. Liu, C. Guo, C. Yin, C. Xie and Q. Fan, *Adv. Funct. Mater.*, 2023, 33, 2209927.
- 24 G. Guan, C. Zhang, H. Liu, Y. Wang, Z. Dong, C. Lu, B. Nan, R. Yue, X. Yin, X. Zhang and G. Song, *Angew. Chem., Int. Ed.*, 2022, **61**, e202117229.
- 25 C. Zhang, W. Bu, D. Ni, S. Zhang, Q. Li, Z. Yao, J. Zhang, H. Yao, Z. Wang and J. Shi, *Angew. Chem., Int. Ed.*, 2016, 55, 2101–2106.
- 26 B. Yu, B. Choi, W. Li and D. Kim, *Nat. Commun.*, 2020, 11, 3637.
- 27 B. Yang, Q. Liu, X. Yao, D. Zhang, Z. Dai, P. Cui, G. Zhang, X. Zheng and D. Yu, ACS Appl. Mater. Interfaces, 2019, 11, 38395–38404.
- 28 C. Yao, H. D. Qi, X. Jia, Y. Xu, Z. Tong, Z. Gu and D. Yang, *Angew. Chem., Int. Ed.*, 2022, **61**, e202113619.
- 29 Y. Dong, C. Yao, Y. Zhu, L. Yang, D. Luo and D. Yang, *Chem. Rev.*, 2020, **120**, 9420–9481.
- 30 X. Guo, F. Li, C. Liu, Y. Zhu, N. Xiao, Z. Gu, D. Luo, J. Jiang and D. Yang, *Angew. Chem., Int. Ed.*, 2020, **59**, 20651–20658.
- 31 X. Guan, F. Meng, H. Tan, X. Wang, J. Li, J. Wei, J. Ouyang and N. Na, *Chem. Sci.*, 2022, **13**, 8657–8666.
- 32 P. Winterwerber, S. Harvey, D. Y. W. Ng and T. Weil, *Angew. Chem.*, *Int. Ed.*, 2020, **59**, 6144–6149.
- 33 C. Whitfield, M. Zhang, P. Winterwerber, Y. Wu, D. Y. W. Ng and T. Weil, *Chem. Rev.*, 2021, **121**, 11030–11084.
- 34 H. Pei, N. Lu, Y. Wen, S. Song, Y. Liu, H. Yan and C. Fan, *Adv. Mater.*, 2010, **22**, 4754–4758.
- 35 J. Wang, J. Chao, H. Liu, S. Su, L. Wang, W. Huang, I. Willner and C. Fan, *Angew. Chem., Int. Ed.*, 2017, **56**, 2171–2175.
- 36 X. Shen, W. Xua, J. Ouyang and N. Na, *Chin. Chem. Lett.*, 2022, 33, 4505–4516.
- 37 G. Nabil, R. Alzhrani, H. O. Alsaab, M. Atef, S. Sau, A. K. Iyer and H. E. Banna, *Cancers*, 2021, **13**, 898–925.
- 38 W. S. Sun, J. X. Yang, M. F. Hou, S. W. Xie, L. Q. Xiong, B. Li and C. Zhang, *Adv. Funct. Mater.*, 2021, **31**, 2009116–2009132.
- 39 R. Wang, H. Yang, A. R. Khan, X. Yang, J. Xu, J. Ji and G. Zhai, J. Colloid Interface Sci., 2021, 598, 213–228.
- 40 Z. Sun, T. Wang, J. Wang, J. Xu, T. Shen, T. Zhang, B. Zhang, S. Gao, C. Zhao, M. Yang, F. Sheng, J. Yu and Y. Hou, *J. Am. Chem. Soc.*, 2023, 145, 11019–11032.
- 41 Y. Zhu, P. Gong, J. Wang, J. Cheng, W. Wang, H. Cai, R. Ao, H. Huang, M. Yu, L. Lin and X. Chen, *Angew. Chem., Int. Ed.*, 2023, **62**, e202218407.
- 42 J. Wang, Z. Sun, S. Wang, C. Zhao, J. Xu, S. Gao, M. Yang, F. Sheng, S. Gao and Y. Hou, J. Am. Chem. Soc., 2022, 144, 19884–19895.
- 43 C. Cao, X. Wang, N. Yang, X. Song and X. Dong, *Chem. Sci.*, 2022, **13**, 863–889.

44 Q. Sun, S. Zhang, X. Wei, T. Yang, J. Wang and M. Chen, *Anal. Chim. Acta*, 2022, **1221**, 340100.

Edge Article

- 45 X. Wang, X. Shen, J. Li, X. Ge, J. Ouyang and N. Na, *Anal. Chem.*, 2022, **94**, 16803–16812.
- 46 F. Shen, S. Dai, N. Wong, S. Deng, A. Wong and D. Yang, *J. Am. Chem. Soc.*, 2020, **142**, 10769–10779.
- 47 P. He, F. Liu, M. Li, M. Ren, X. Wang, Y. Deng, X. Wu, Y. Li, S. Yang and J. Song, *Adv. Healthcare Mater.*, 2023, 12, 2203106.