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A microbiological system for screening the interference of XNA monomers with DNA and RNA metabolism†

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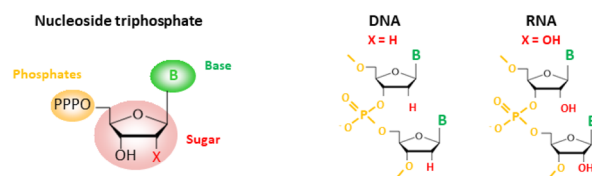
We explored the toxicity and mutagenicity of a wide range of xenobiotic nucleoside triphosphates to an *Escherichia coli* strain equipped with a nucleoside triphosphate transporter. This bacterial test provides a tool to evaluate and guide the synthesis of nucleotides for applications such as the propagation of non-natural genetic information or the selection of potential drugs.

Genetic information is stored and conveyed by only two types of nucleic acids DNA and RNA (Scheme 1), in all living species. Over the past few decades, different systems have revealed that genetic information can also be carried and propagated by non-natural nucleic acids.^{1,2} These chemically modified molecules, called XNA (for Xenobiotic Nucleic Acids),^{3,4} exhibit modified nucleobases, sugars or backbones or a combination of these modifications and even novel bases or base pairs. XNAs have mainly been considered to improve the physicochemical or pharmacological properties of nucleic acids. These characteristics make them potentially useful for numerous biotechnological, therapeutic and diagnostic applications,⁵ but also for the understanding of molecular mechanisms and the origins of life.⁶

In the natural world, bacteriophage genomes exhibit the widest chemical diversity of non-canonical nucleobases.⁷ We have recently elucidated the biosynthetic pathway of a non-canonical nucleotide: aminoadenine (Z) which completely replaces adenine of siphoviruses genomes⁸ and characterized a specific DNA polymerase responsible for its incorporation.⁹ These natural variations mostly prevent cleavage by prokaryotic host restriction endonucleases. The expression of genetic information by non-natural nucleic acids to increase the spectrum of coding possibilities is a major objective of biotechnology and synthetic biology. Another important outcome of this approach is also to limit genetic pollution by minimizing the possible interactions of these non-canonical molecules with

the rest of the living world.¹⁰ In this sense, considerable efforts have been made in recent years to produce non-natural base pairs to expand the genetic alphabet and produce new functionalities.¹¹⁻¹³ In a previous study, we constructed an *E. coli* strain equipped with a dNTP transporter from an algal plastid that allows the transport of nucleoside triphosphates into the bacterium.¹⁴ Using this *E. coli* strain, we here tested the *in vivo* toxicity and the mutagenic effect of our chemically modified nucleoside triphosphates containing modifications in the sugar, nucleobase or backbone and also in different combinations (Fig. 1 and 2). The ability of modified nucleosides to interact with the nucleic acids of the cell is an important prerequisite to be tested for their use to efficiently carry genetic information *in vivo*.

To test the toxicity of chemically modified nucleoside triphosphates *in vivo*, a dilution of an overnight pre-culture of the *E. coli* strain equipped with a dNTPs transporter (XE858) was spread on a rich medium plate in the presence of a 100 mM solution of nucleoside triphosphates placed in the central well of the plate. Toxicity was measured by the size of the inhibition halo observed after the overnight growth of the strain on plate (Fig. 3A). A positive control of toxicity was performed with the dideoxyCTP (ddCTP), a replication chain terminator. As a negative control, the strain devoid of dNTP transporter was used under the same conditions (Fig. 3B). To test the mutagenic effect of these XNA triphosphates transported into bacterial



Scheme 1 Natural nucleic acids.

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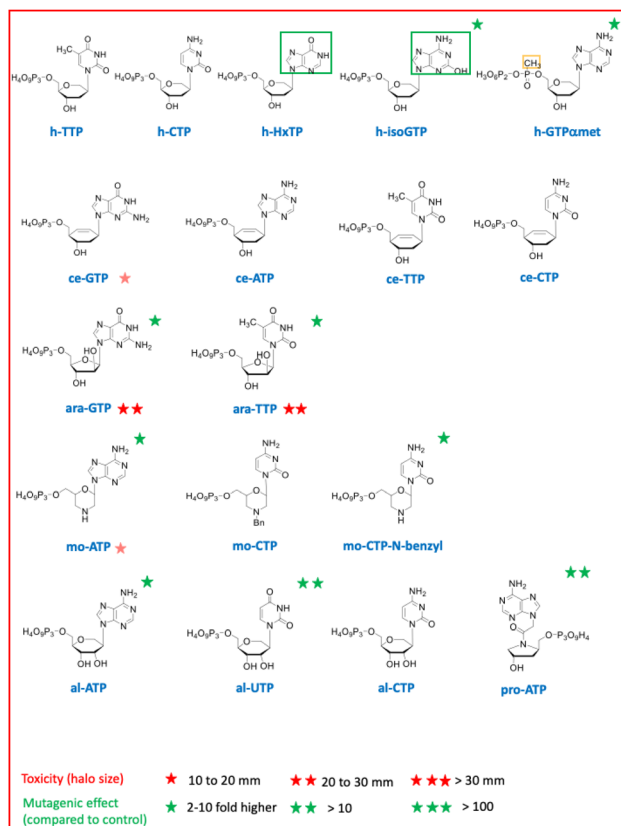


Fig. 1 List of chemically synthesized nucleoside triphosphates with non-natural sugar combined with a non-natural nucleobase framed in green or a modification of the phosphate moiety framed in yellow. (h: hexitol, ce: cyclohexene, ara: arabinose, mo: morpholino, al: alitol, pro: prolinol).

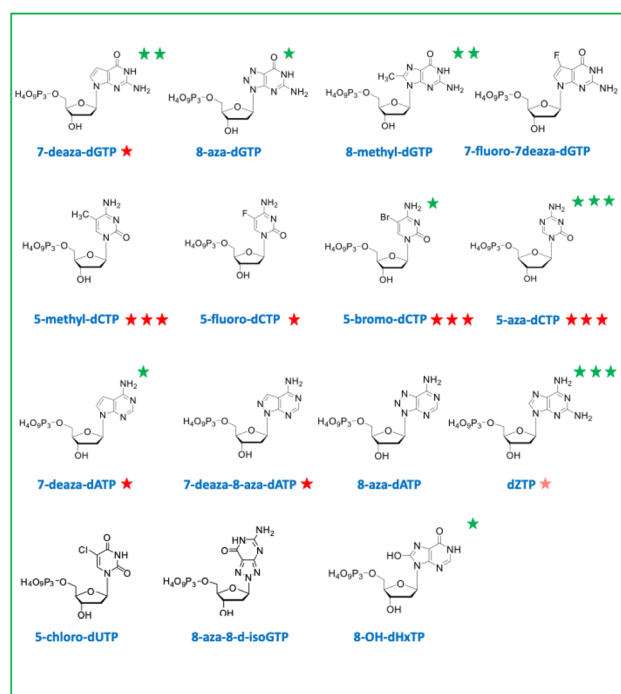


Fig. 2 List of chemically synthesized deoxyribonucleotide triphosphates with non-canonical nucleobases.

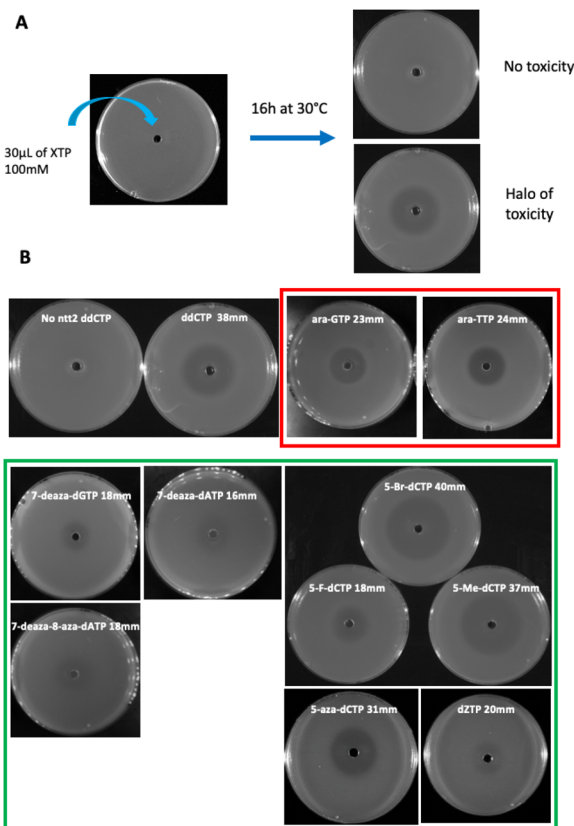
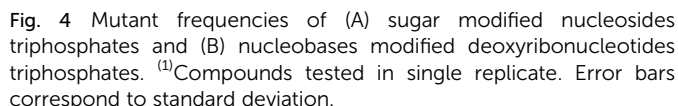


Fig. 3 Toxicity on plates (A) toxicity test description (B) pictures of the toxic compounds revealed in our selection strain and their relative halo diameters.

cells, the XE858 strain was incubated overnight with different concentrations of nucleoside triphosphates and spread on LB rifampicin plates. Rifampicin resistance, most frequently results from a mutation in the *rpoB* gene encoding for the beta subunit of bacterial RNA polymerase. This selection provides a direct assay of the mutagenicity of nucleoside triphosphate by interfering with DNA replication. 5-Aza-dCTP (decitabine triphosphate) known to generate G- > C transversions when incorporated into DNA was used as positive control of mutagenesis. It was previously hypothesized that decitabine could undergo a ring-opening step, followed by hydrolysis to yield a structure capable of pairing with cytosine. Thus, decitabine triphosphate incorporated as dCTP can pair with a cytosine inducing G to C mutations.¹⁵ Mutation frequencies were determined by the ratio of rifampicin-resistant clones to the number of viable clones counted on LB plates at each concentration (Fig. 4).

Chemically synthesized nucleotides with different sugar moieties (hexitol, cyclohexene or arabinose) (Fig. 1) were initially chosen for their chemical and enzymatic stability. As XNA polymerases able to copy DNA in XNA and back to DNA were successfully developed, these XNA were established to be capable of evolution.^{2,3} HNA were also selected for the development of efficient aptamers and catalysts. Using an *E. coli* selection, we have also previously shown that genetic



Hexitol triphosphate and cyclohexene triphosphate did not induce a mutagenic effect *in vivo* when coupled to a canonical base (Fig. 4A). However, a slight mutagenic effect was observed when hexitol triphosphates were coupled to a non-canonical and ambiguous base with the h-isoGTP or to a phosphate modification with the h-GTPzmet. This result suggests that these sugar

For a thymine alternative, an almost complete substitution of this base (98.4%) with chlorouracil was achieved in *E. coli* genomic DNA by metabolic engineering and evolution.²⁰ As expected, the corresponding deoxyribochlorouracil triphosphate showed no toxicity or mutagenic effect confirming its potential for synthetic genome construction and other applications such as aptamers.²¹ Since 7-deaza-dATP, 8-aza-dATP and 7deaza-8aza-dATP can be polymerized with comparable efficiency by cellular DNA polymerases, using a polyT template *in vitro*,²² the 8-aza-dA would be the best candidate, regarding its characteristics as a triphosphate *in vivo* (no toxicity or mutagenic effect), to pair with chlorouracil. The mutagenic effect of 8-OH-dHxTP, 12 to 71 fold higher than the spontaneous mutation rate and closely related to the concentration used, was expected because this nucleobase can base pair with an A or a C after a 180° rotation around the glycosidic bond (Fig. 2 and 4B).

Amino adenine deoxyribonucleotide triphosphate, which totally replaces adenine in genomes of some siphoviruses exhibited an unexpectedly high mutagenic effect and low toxicity suggesting weak interaction with RNA and efficient incorporation into DNA. The high mutagenic effect probably results from the spontaneous deamination of amino adenine to guanine after incorporation of the dZTP into DNA.²³ However, it remains to be determined whether an *E. coli* enzyme would catalyze this *in situ* deamination reaction on DNA.

Conclusions

In this *in vivo* system, the fact that at least one compound of each type of modification is toxic or mutagenic strongly indicates that the synthetic compounds cross the bacterial membrane and reach replication and transcription machineries. In general, we observed that the most mutagenic nucleoside triphosphates are little toxic and the most toxic are little mutagenic. A prominent exception to this pattern is 5-aza-dCTP, which is known to be highly mutagenic and which we found to be highly toxic. Altogether, our bacterial screening system provides a simple and effective way for assessing both the toxicity and mutagenicity of synthetic nucleotides. In the future, noncanonical nucleoside triphosphates inducing either of these two effects should be avoided as building blocks of XNA carriers of genetic information *in vivo*. The interactions of certain nucleoside triphosphates with cellular components, would certainly require further investigation to understand the underlying mechanisms. The combination of organic synthesis of non-canonical DNA and RNA monomers and the construction of reporter strains should allow the screening of numerous drug candidates and new types of nucleic acids propagated *in vivo*.

Author contributions

VP, PM and PH conceived the project. VP and PM designed experiments. AB, MA, and CH performed experiments. VP wrote the manuscript. All authors reviewed the manuscript.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 P. Herdewijn and P. Marlière, *Chem. Biodiversity*, 2009, **6**, 791–808.

- 2 V. B. Pinheiro, A. I. Taylor, C. Cozens, M. Abramov, M. Renders, S. Zhang, J. C. Chaput, J. Wengel, S.-Y. Peak-Chew, S. H. McLaughlin, P. Herdewijn and P. Holliger, *Science*, 2012, **336**, 341–344.
- 3 V. B. Pinheiro and P. Holliger, *Curr. Opin. Chem. Biol.*, 2012, **16**, 245–252.
- 4 J. C. Chaput and P. Herdewijn, *Angew. Chem., Int. Ed.*, 2019, **58**, 11570–11572.
- 5 K. Duffy, S. Arangundy-Franklin and P. Holliger, *BMC Biol.*, 2020, **18**, 112.
- 6 A. I. Taylor, V. B. Pinheiro, M. J. Smola, A. S. Morgunov, S. Peak-Chew, C. Cozens, K. M. Weeks, P. Herdewijn and P. Holliger, *Nature*, 2015, **518**, 427–430.
- 7 P. Weigele and E. A. Raleigh, *Chem. Rev.*, 2016, **116**, 12655–12687.
- 8 D. Sleiman, P. S. Garcia, M. Lagune, J. Loc'h, A. Haouz, N. Taib, P. Röthlisberger, S. Gribaldo, P. Marlière and P. A. Kaminski, *Science*, 2021, **372**, 516–520.
- 9 V. Pezo, F. Jaziri, P.-Y. Bourguignon, D. Louis, D. Jacobs-Sera, J. Rozenski, S. Pochet, P. Herdewijn, G. F. Hatfull, P.-A. Kaminski and P. Marlière, *Science*, 2021, **372**, 520–524.
- 10 M. Schmidt, *Bioessays*, 2010, **32**, 322–331.
- 11 S. A. Benner, N. B. Karalkar, S. Hoshika, R. Laos, R. W. Shaw, M. Matsuura, D. Fajardo and P. Moussatche, *Cold Spring Harbor Perspect. Biol.*, 2016, **8**, a023770.
- 12 K. Hamashima, M. Kimoto and I. Hirao, *Curr. Opin. Chem. Biol.*, 2018, **46**, 108–114.
- 13 V. T. Dien, S. E. Morris, R. J. Karadeema and F. E. Romesberg, *Curr. Opin. Chem. Biol.*, 2018, **46**, 196–202.
- 14 V. Pezo, C. Hassan, D. Louis, B. Sargueil, P. Herdewijn and P. Marlière, *ACS Synth. Biol.*, 2018, **7**, 1565–1572.
- 15 L. Jackson-Grusby, P. W. Laird, S. N. Magge, B. J. Moeller and R. Jaenisch, *Proc. Natl. Acad. Sci. U.S.A.*, 1997, **94**, 4681–4685.
- 16 V. Pezo, F. W. Liu, M. Abramov, M. Froeyen, P. Herdewijn and P. Marlière, *Angew. Chem., Int. Ed.*, 2013, **52**, 8139–8143.
- 17 G. Ceulemans, A. Van Aerschot, B. Wroblewski, J. Rozenski, C. Hendrix and P. Herdewijn, *Chem.-Eur. J.*, 1997, **3**, 1997–2010.
- 18 A. W. Feldman, E. C. Fischer, M. P. Ledbetter, J.-Y. Liao, J. C. Chaput and F. E. Romesberg, *J. Am. Chem. Soc.*, 2018, **140**, 1447–1454.
- 19 E. Eremeeva, M. Abramov, L. Margamuljana and P. Herdewijn, *Chem.-Eur. J.*, 2017, **23**, 9560–9576.
- 20 P. Marlière, J. Patrouix, V. Döring, P. Herdewijn, S. Tricot, S. Cruveiller, M. Bouzon and R. Mutzel, *Angew. Chem. Int. Ed. Engl.*, 2011, **50**, 7109–7114.
- 21 C. Gasse, M. Zaarour, S. Noppen, M. Abramov, P. Marlière, S. Liekens, B. De Strooper and P. Herdewijn, *Chembiochem*, 2018, **19**, 754–763.
- 22 E. Eremeeva, M. Abramov, P. Marlière and P. Herdewijn, *Org. Biomol. Chem.*, 2017, **15**, 168–176.
- 23 M. Levy and S. L. Miller, *Proc. Natl. Acad. Sci.*, 1998, **95**, 7933–7938.