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Highly sensitive detection of circulating tumour cells based on an ASV/CV dual-signal electrochemical strategy†

 Gang Wang,‡ Dan Han‡ and Qingyu Zhang *

Circulating tumour cells (CTCs), as a tumour marker, may provide more information in early diagnosis and accurate therapy of cancer patients. Electrochemical detection of CTCs has exhibited exceptional advantages. However, single-signal electrochemical detection usually has a high probability of false positives coming from interferents, operating personnel, and nonstandard analytical processes. Herein, a dual-signal strategy using anodic stripping voltammetry (ASV) and cyclic voltammetry (CV) for highly sensitive detection of CTCs was developed. When MCF-7 cells were present, aptamer DNA (DNA1)-magnetic beads (MBs) were captured by CTCs and detached from the biosensing electrodes. Following magnetic separation, polystyrene bead (PS)-CdS QDs labelled on MCF-7 cells were dissolved by HNO_3 and the intensity of the oxidation peak current of Cd^{2+} ions was proportional to the amount of MCF-7 cells in ASV ($y = 6.8929 \lg C_{\text{cells}} + 1.0357$ (C_{cells} , cells per mL); R^2 , 0.9947; LOD, 3 cells per mL). Meanwhile, the anodic peak currents of the remaining electrode in CV were also proportional to the amount of MCF-7 cells ($y = 3.7891 \lg C_{\text{cells}} + 52.3658$ (C_{cells} , cells per mL); R^2 , 0.9846; LOD, 3 cells per mL). An ASV/CV dual-signal biosensor for electrochemical detection of CTCs was achieved, which overcame the limitations of any single-signal mode and improved the detection reliability and precision.

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1. Introduction

Circulating tumour cells (CTCs), first described in 1869 by Prof. Ashworth, are shed from primary or metastatic tumours and spread into the peripheral blood leading to future metastasis.^{1,2} Numerous studies have demonstrated that CTCs can be regarded as a tumour marker in the monitoring of metastasis and postoperative recurrence.³ Meanwhile, as a “liquid biopsy”, CTC detection may provide additional information before fatal metastasis occurs *via* a convenient and non-invasive method. Therefore, it is essential to monitor CTCs in peripheral blood for early diagnosis, assessing the cancer status of patients, and prescribing personalized anticancer therapy. The analysis of CTCs is still challenging because of their low blood concentration (about one CTC per 10^9 hematopoietic cells). Currently, several approaches have been proposed for CTC detection, such as DNA nanomachine technology,⁴ microfluidic technology,^{5,6} magnetic microgel technology,⁷ and photoelectrochemical (PEC) technology.⁸ Nevertheless, the fundamental drawbacks of these approaches remain, including their limited sensitivity, complicated production technology, and standard sample

frequency. Therefore, the development of novel, useful, sensitive, and accurate methods for CTC detection is still absolutely essential.

Electrochemical (EC) methods offer outstanding benefits in detection because of their high sensitivity, quick response, simplicity, quick reaction, ease of use, and capacity for miniaturization.^{9,10} EC cytosensors can convert the analytical information between tumour cells and bio-probes to observable electrical readouts, including current, potential, or impedance.¹¹ Wang *et al.* reported a novel ultrasensitive EC biosensor for the detection of K562 cells by a signal amplification strategy based on multiple-layer CdS quantum dot (QD)-functionalized polystyrene microspheres (PSs) as a bio-probe and a graphene oxide (GO)-polyaniline (PANI) composite as a modified material of the capture electrode.¹² Xiang *et al.* described an EC cytosensor based on rolling circle amplification extension of electrode-immobilized primer/circular DNA complexes for efficiently capturing and sensitively detecting CTCs in whole blood. In their design, long DNA strands containing multiple repeated aptamer sequences were prepared through rolling circle amplification (RCA) reactions. These repeated sequences could contact cancer cells in a cooperative fashion, which significantly enhances the avidity to the target cells. By using rolling circle amplification extension to couple the multivalent binding probes with the catalytic enzymatic signal amplification, the target MCF-7 cells were efficiently captured and sensitively detected.¹³ Single-signal electrochemical detection often has

Department of Gastroenterology, Tianjin Medical University General Hospital, Anshan Road 154, Heping District, Tianjin, 300052, China. E-mail: zhangqy@tmu.edu.cn

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‡ These authors contributed equally.



a chance of false positives or false negatives coming from interferents, operational personnel, the instrument, and nonstandard analysis processes.^{14,15} Dual- or multi-signal detection has already been established to further improve the sensitivity and accuracy, including PEC and EC assays,¹⁶ colorimetric and PEC analysis,¹⁷ and colorimetric and fluorescence detection. The dual-signal technique not only possesses the properties of each mode, but also introduces reciprocal authentication in various modes, which significantly increases the reliability and precision of detection.^{18,19} For example, Kuang *et al.* created plasmonic circular dichroism- and luminescent-driven DNA-driven gold-upconversion nanoparticle (Au-UCNP) pyramids to identify miRNA in live cells.²⁰ For the purpose of detecting miRNA-210, Chen *et al.* created a magnetic-assisted EC-PEC dual-mode biosensing device.²¹ However, multi-mode detection is in the initial stages due to the high demand for biosensor structural design, sensing materials and choice of bio-probe, which serve to effectively respond to target changes of various signals.

Carbon nanotubes (CNTs), as a new class of advanced materials, have attracted tremendous interest due to their unique properties.²²⁻²⁴ CNTs can be divided into two categories, either single-walled CNTs (SWCNTs) or multi-walled CNTs (MWCNTs). Easy functionalization of CNTs makes them biocompatible for biological applications. Biomolecules,²⁵ nanoparticles (NPs)²⁶ and polystyrene nanofibers²⁶ attached to CNTs have been reported. In particular, CNT and metal-NP hybrids exhibit the properties of both CNTs and metal-NPs. Among them, Au-NPs, as the most stable biocompatible NPs, can be functionalized with CNTs through covalent or non-covalent linkages,^{27,28} and exhibit excellent electronic and mechanical properties. Moreover, quantum dots (QDs), with unique chemistry and physics, and confined size on the nanoscale (1–10 nm), have attracted attention from researchers for

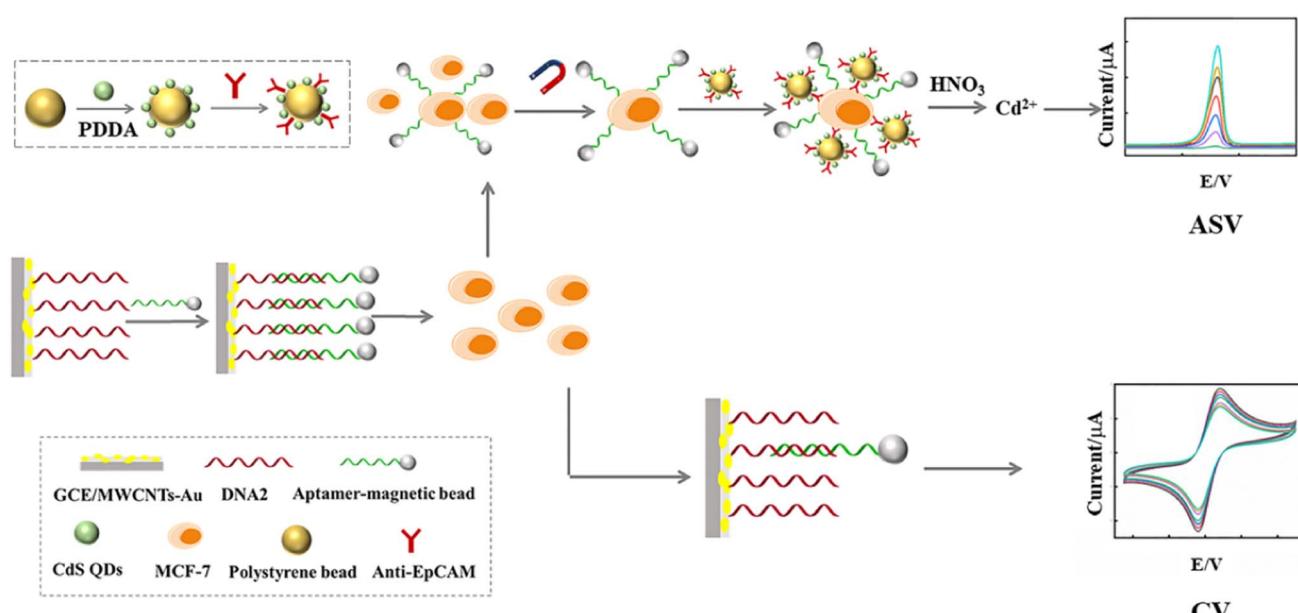
the construction of QD-based electrochemical sensors.²⁹⁻³¹ Depending on the excellent electrochemical and photophysical characteristics of QDs, the introduction of QDs can provide an increase in sensitivity and response speed in electrochemical sensors.^{31,32} Among them, cadmium sulfide (CdS) has found applications in various fields with a bulk band gap value of ~2.42 eV. CdS QDs have also been widely reported to prepare CdS QD/polystyrene (PS) composites for electrochemical detection due to PS's low specific weight, biocompatibility, flexibility, and high chemical resistance.^{33,34}

Herein, we have developed a dual-signal electrochemical strategy for the highly sensitive detection of CTCs based on anodic stripping voltammetry (ASV) and cyclic voltammetry (CV) as shown in Scheme 1. In our design, the capture electrode, GCE/multi-walled carbon nanotubes (MWCNTs)-Au/linked DNA (DNA2)-aptamer (DNA1)-magnetic beads (MB), was created using layer-by-layer assembly technology. When MCF-7 cells are present, DNA1-MB was captured and released from the capture electrode. Following magnetic separation, the MCF-7 cells were labelled with polystyrene beads (PS)-CdS quantum dots (QDs). After the CdS QDs were dissolved by HNO₃, Cd²⁺ ions were released. The intensity of the peak current of Cd²⁺ ions in ASV was proportional to the amount of MCF-7 cells. Meanwhile, the anodic peak current of the remaining electrode in CV was also proportional to the quantity of MCF-7 cells. An ASV/CV dual-signal electrochemical biosensor for the detection of CTCs was obtained, which would overcome the limitations of any single-signal detection and improve both the sensitivity and accuracy.

2. Experimental

2.1 Materials

The glassy carbon electrode (GCE), polyetherimide (PEI), multi-walled carbon nanotubes (MWCNTs), CdCl₂, MPA, Na₂S, poly



Scheme 1 Schematic representation of the ASV/CV dual-signal electrochemical detection of CTCs.



dimethyl diallyl ammonium chloride (PDDA), trisodium citrate solution (1% by weight) and HAuCl₄ were obtained from Aladdin Chemistry Co. Ltd (Shanghai, China). Bovine serum albumin (BSA), phosphate-buffered saline (PBS), potassium chloride, K₃[Fe(CN)₆], K₄[Fe(CN)₆]·3H₂O, trisodium citrate solution, polystyrene microspheres (PS), and anti-epithelial cell adhesion molecule antibody (anti-EpCAM) were obtained from Abcam Co., Ltd. Carboxyl-Fe₃O₄ magnetic microbeads (COOH-MB, 300–400 nm) were obtained from BaseLine ChromTech Research Centre (Tianjin, China). Glacial acetic acid, N-(3-(dimethylamino)-propyl)-N'-ethyl carbodiimide hydrochloride (EDC), N-hydroxy sulfosuccinimide (NHS), and the oligonucleotides used in this work were received from Sangon Biotechnology Co. Ltd (Shanghai, China).

MUC1 aptamer (NH₂-DNA1): 5'-NH₂-GGG AGA CAA GAA TAA ACG CTC AAG CAG TTG ATC CTT TGG ATA CCC TGG TTC GAC AGG AGG CTC ACA ACA GGC-3'.

Linker DNA2: 5'-SH-TTT TTT TTT TTT TTT TTT TTT TTT TTT GCC TGT TGT GAG-3'.

2.2 Preparation of the sensing electrode

2.2.1 Fabrication of the MWCNT-Au NP composite. First, carboxylated MWCNTs (COOH-MWCNTs) were synthesized as previously described.³⁵ 0.50 g MWCNTs were added to 40 mL of an acidic solution (H₂SO₄:HNO₃ = 3:1, volume ratio) and sonicated for 6 h. After being centrifuged and washed repeatedly with ultrapure water, COOH-MWCNTs were obtained. Then, 10 mg COOH-MWCNTs were mixed with PEI solution (0.1% wt) and sonicated for 30 min. After being filtered and washing with ultrapure water, the MWCNT-PEI composites were achieved and were dispersed in ultrapure water (about 1.5 mg mL⁻¹). Third, Au NPs were synthesized according to the previous methods.³⁶ Briefly, 1.6 mL aqueous trisodium citrate solution (1% by weight) was added into 100 mL boiling HAuCl₄ solution (0.25 mM) under vigorous stirring and reflux. The Au NPs were obtained after 30 min with a colour change from blue to wine red. Last, 10 mL Au NP aqueous solution (0.39 mg mL⁻¹) was added to 10 mL MWCNTs-PEI solution and stirred for 4 h to obtain MWCNTs-Au NPs.

2.2.2 Preparation of DNA1-MBs. 40 μL carboxyl-group-coated MBs (COOH-MBs) were transferred into a 1.5 mL Eppendorf tube and washed three times with 500 μL PBS buffer (10 mM phosphate, 0.1 M NaCl, pH 7.4). After magnetic separation, the MBs were dispersed in 500 μL PBS buffer (pH 7.4) and 100 μL capture NH₃-DNA1 (0.4 μM) was added. Under gentle shaking at 37 °C for 1 h, the DNA1-MBs were obtained by magnetic separation and washing with 500 μL of PBS buffer three times.

2.2.3 Preparation of the capture electrode. First, the GCE was polished with 0.3 and 0.05 μm Al₂O₃ powder and sonicated in alcohol and distilled water, respectively. Then, the GCE was dried by nitrogen gas at room temperature. In addition, 10 μL of MWCNT-Au composites (10 mg mL⁻¹) were added dropwise on the electrode to form the MWCNTs-Au film at room temperature. Subsequently, 5 μL (10 μM) linker DNA (DNA2) solution was dropped on the surface of the electrode and incubated for

12 h at room temperature. The DNA2 was modified on the electrode by Au-S bonds, which were formed from -SH in DNA2 and Au in the electrode of GCE/MWCNTs-Au. 1% BSA was used to block non-specific sites, and the electrode GCE/MWCNTs-Au/DNA2/BSA was immersed in PBS solution containing DNA1-MBs (10 μM), which came from -COOH in COOH-MB and -NH₃ in DNA1 through activation of EDC/NHS and heated to 95 °C for 5 min. Then, the mixture solution was cooled to room temperature and kept for 2 h. Finally, the capture electrode, GCE/MWCNTs-Au/DNA2-DNA1-MB, was achieved after being washed with 500 μL of PBS buffer three times.

2.3 Synthesis of the PS/CdS QDs/anti-EpCAM probe

First, CdS QDs were synthesized as Wang *et al.* reported *via* a one-pot route.¹² 0.4 mmol CdCl₂, 0.2 mmol MPA, and 0.2 mmol C₆H₅Na₃O₇ were dissolved into distilled water under vigorous stirring at a pH of 9. Then, 0.02 mmol Na₂S was added into the above solution under stirring overnight. MPA-capped CdS QDs were obtained after being washed with absolute ethanol and distilled water three times. Third, 50 μL COOH-PS (5%) was mixed with 1 mL PDDA (0.2 wt%) solution and sonicated for 20 min. After being washed three times by distilled water, 1 mL CdS QDs (3 × 10⁻⁶ M) were added and stirred for 12 h. And the above two procedures were repeated alternately three times. Then, the produced PS-CdS QD composites were activated by EDC and NHS at 37 °C and 10 μL anti-EpCAM (Ab, 1 μg mL⁻¹) solution was dropped into the active PS-CdS QDs under shaking for 6 h at 25 °C. After washing three times with distilled water, the PS/CdS QDs/anti-EpCAM probe was achieved.

2.4 ASV/CV dual-signal electrochemical detection of CTCs

The capture electrode, GCE/MWCNTs-Au/DNA2-DNA1-MB, was immersed in MCF-7 cell solutions with different concentrations and incubated for 1 h at 37 °C. First, after MCF-7 cells were captured, DNA1-MB separated from the capture electrode. After magnetic separation and being washed by PBS, 5 μL PS-CdS QDs-Ab probe solution was added into the captured MCF-7 cells by DNA1-MB. The Cd²⁺ solution dissolved by HNO₃ (200 μL, 0.1 M) was transferred into acetate buffer (HAc-NaAc buffer, 4.8 mL, 0.2 M, pH 5.2) containing 20 μg per mL Hg²⁺ for detection of metal ions by ASV. The electrode position was at -1.2 V for 30 min. A square-wave voltammetric waveform was recorded by stripping from -1.1 to 0.5 V (potential step, 4 mV; amplitude, 25 mV; frequency, 25 Hz). Based on the relationship between Cd²⁺ content and the MCF-7 cell concentration, MCF-7 cells were detected due to the sharp peak of Cd²⁺ in ASV at -0.72 V.^{37–39} On the other hand, after DNA1-MB had separated, the electrochemical characteristics of the capture electrode was investigated using cyclic voltammetry (CV) in 5 mM K₃Fe(CN)₆/K₄Fe(CN)₆ (0.1 M KCl). Based on the relationship between the anodic peak current intensity and the MCF-7 cell concentration, MCF-7 cells were detected. An ASV/CV dual-signal electrochemical biosensor for detection of CTCs was achieved.



3. Results and discussion

3.1 Characterization of MWCNTs, Au NPs and the MWCNT-Au NP composite

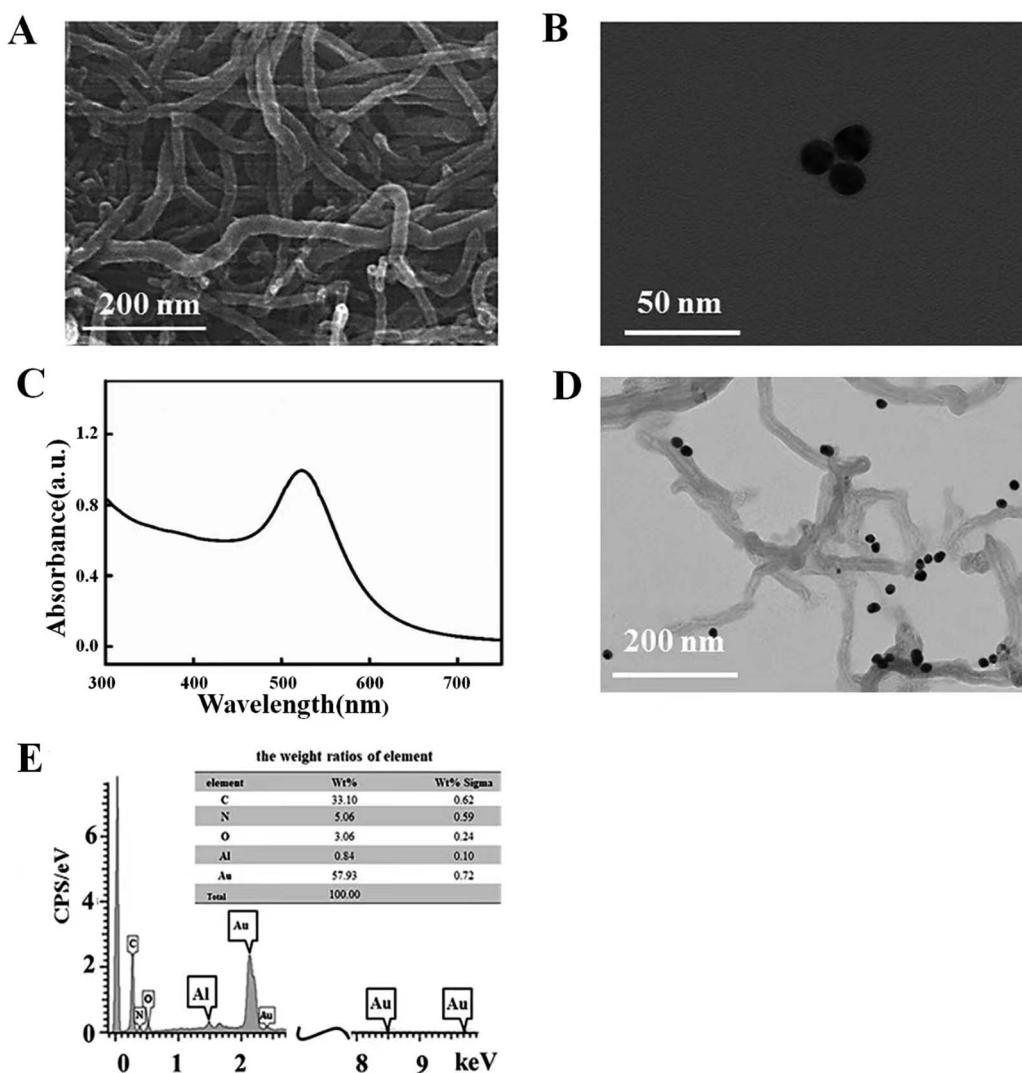
To reveal the morphology of the COOH-MWCNTs, they were examined using scanning electron microscopy (SEM). As illustrated in Fig. 1A, the MWCNTs exhibited tubular structures with a length of about twenty microns. According to Fig. 1B, the Au NPs showed a regular spherical form and a diameter of about 15 nm, whose absorption peak was at 520 nm as shown in Fig. 1C. The composite of MWCNTs-Au NPs was investigated by SEM as exhibited in Fig. 1D, and it was discovered that the Au NPs dispersed over the MWCNTs' exterior walls like dewdrops. As expected, the EDS pattern (Fig. 1E) displayed the distribution of C, N, O and Au, which provided direct evidence that the MWCNT-Au NP composite had been synthesized successfully.

3.2 Characterization of CdS QDs, PS and PS-CdS QDs

The TEM image of the synthesized MPA-QDs exhibited spherical shapes with an average size of 3.7 nm as shown in Fig. 2A. The UV-vis absorption and PL spectra of the CdS QDs were recorded in Fig. 2B. The absorption peak appeared at 410 nm and the fluorescence emission peak at 540 nm, indicating a Stokes shift of about 130 nm, which showed that there might be a lot of dangling bonds or flaws on the surface of the QDs, benefitting the attachment of PDDA. The size of the CdS QDs was predicted to be around 3.73 nm using empirical formula:⁴⁰

$$D = (-6.6521 \times 10^{-8})\lambda^3 + (1.9557 \times 10^{-4})\lambda^2 - (9.2352 \times 10^{-2})\lambda + 13.29 \quad (D, \text{the diameter of CdS QDs; } \lambda \text{ (nm), absorption peak position})$$

which was consistent with the above TEM results. Meanwhile, the morphology of PS and the PS-CdS QDs were investigated by TEM as shown in Fig. 2C and D. A uniform and smooth surface with a diameter of about 1 μm was visible on the PS. The TEM picture of the PS-CdS QDs, on the other hand,



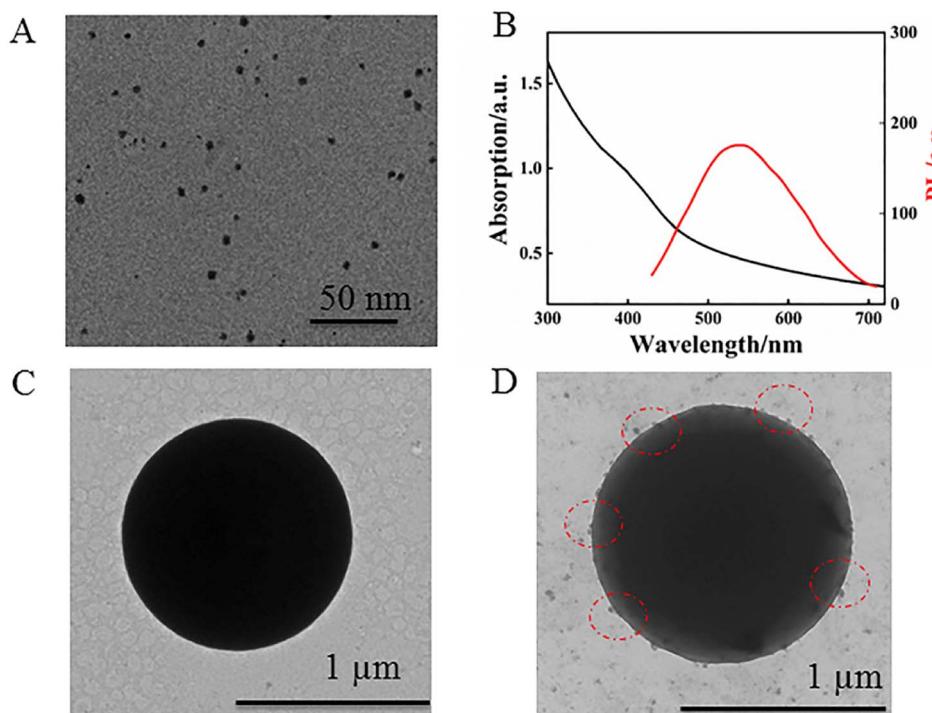


Fig. 2 (A) TEM image, and (B) UV-vis absorption and PL spectra of CdS QDs. (C) TEM image of PS and (D) PS–CdS QDs.

showed a few projecting sections dispersed over the PS surface, indicating that CdS QDs had formed on the surface of PS *via* PDDA by electrostatic attraction.

3.3 Characterization of the biosensor

Electrochemical impedance spectroscopy (EIS) and cyclic voltammetry (CV) were employed to investigate the assembly process of the biosensors. EIS and CV were performed in a support solution containing 5 mM $[\text{Fe}(\text{CN})_6]^{3-/-4-}$. In the EIS

spectrum, the semicircle at high frequencies represented the electron transfer resistance (R_{et}) and the linear counterpart at low frequencies meant the diffusion process, respectively.³⁹ Compared with the bare GCE as shown in Fig. 3A curve a and Fig. 3B curve a, due to its high electrical conductivity, the MWCNT–Au NP nanocomposite significantly promoted the interfacial electron transfer, exhibiting an ultra-small R_{et} in the ESI† (Fig. 3A curve b), while the peak current of CV increased (Fig. 3B curve b).

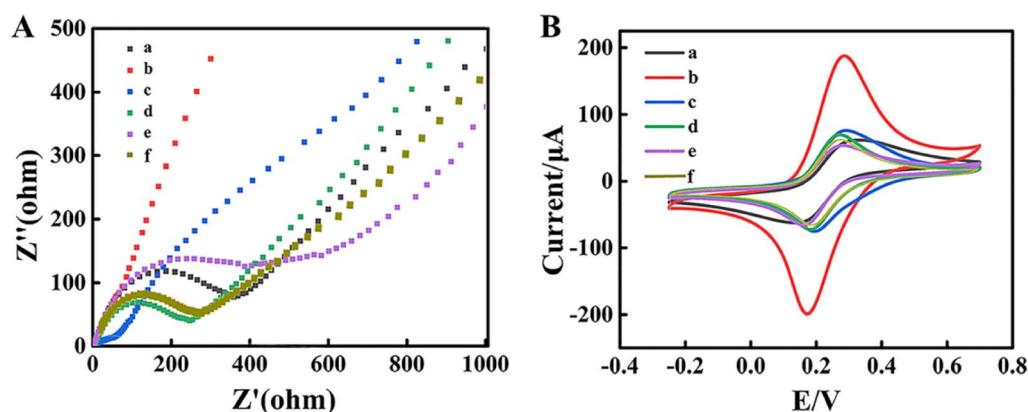


Fig. 3 (A) Electrochemical impedance spectroscopy of the (a) bare GCE, (b) GCE/MWCNTs–Au, (c) GCE/MWCNTs–Au/DNA2, (d) GCE/MWCNTs–Au/DNA2/BSA, (e) GCE/MWCNTs–Au/DNA2/BSA/DNA1-MB, and (f) remaining GCE/MWCNTs–Au/DNA2/BSA (after DNA1-MB was captured by MCF-7) in 0.1 M KCl solution containing 5 mM $\text{K}_3\text{Fe}(\text{CN})_6/\text{K}_4\text{Fe}(\text{CN})_6$ (impedance spectral frequency: 0.1–105 Hz, amplitude: 5 mV). (B) Cyclic voltammograms of the (a) bare GCE, (b) GCE/MWCNTs–Au, (c) GCE/MWCNTs–Au/DNA2, (d) GCE/MWCNTs–Au/DNA2/BSA, (e) GCE/MWCNTs–Au/DNA2/BSA/DNA1-MB, and (f) remaining GCE/MWCNTs–Au/DNA2/BSA (after DNA1-MB was captured by MCF-7) in 0.1 M KCl solution containing 5 mM $\text{K}_3\text{Fe}(\text{CN})_6/\text{K}_4\text{Fe}(\text{CN})_6$ (scan rate: 100 mV s⁻¹).



However, when the aptamer DNA1 coated on the GCE/MWCNTs-Au NPs, the R_{et} increased as shown in the EIS data (Fig. 3A curve c) and the peak current of CV enhanced (Fig. 3B curve c). Subsequently, after the immobilization of BSA on the GCE/MWCNTs-Au NPs/DNA2 electrode to block the residual nonspecific binding sites, the R_{et} increased (Fig. 3A curve d), which indicated that electron transfer was hindered by the presence of BSA. Meanwhile, the peak current of CV decreased. The impedance significantly increased when DNA1-MB hybridized with DNA2 on the surface of the GCE/MWCNTs-Au NPs/DNA2 (Fig. 3A curves e). As expected, the peak current of CV further decreased (Fig. 3B curves e). Finally, when MCF-7 cells are present, DNA1-MBs are captured and released from the capture electrode. The R_{et} of the remaining electrode decreased as shown in the EIS data (Fig. 3A curve e), and the peak current of CV increased.

3.4 Optimization of the experimental conditions

To obtain excellent performance of the EC biosensor for MCF-7 cell detection, some experimental parameters were discussed, including the molar ratio of Au NPs to MWCNTs, the quantity of DNA2 and DNA1, and the incubation time of DNA2 and DNA1-MB in CV, as shown in Fig. 4. In Fig. 4A, the anodic peak current in CV increased when the ratio of Au to MWCNTs decreased from 1 : 3 to 1 : 1, but there was almost no change after 1 : 1. As a result, the ideal ratio for electrode alteration was determined to be 1 : 1. In addition, in CV detection, the link DNA, DNA2, was another parameter that affected the EC signal. The peak current may be decreased by adding more DNA2, as shown in Fig. 4B.

However, additional DNA2, such as beyond 50 μ M, did not lead to a greater decrease in peak current, demonstrating a saturated linker of DNA2. Hence, the optimal concentration of DNA2 was 50 μ M. The amount of DNA1-MBs that hybridized with DNA2 on the electrode surface of the GCE/MWCNTs-Au NPs/DNA2 is shown in Fig. 4C. DNA1-MB caused a shift in the peak current. The peak current decreased with the increase of the ratio between DNA2 and DNA1. When the ratio of DNA2 to DNA1 was more than 1 : 1.5, there was essentially no change in the peak current, indicating full hybridization of DNA2 with DNA1. The effect of incubation time was also discussed as shown in Fig. 4D. The peak current decreased with the increase of the incubation time, and there was almost no obvious change after 60 min. Thus, 60 min was chosen as the incubation time for the construction of the electrochemical biosensor.

3.5 Dual-signal electrochemical detection of CTCs

The MCF-7 cells were captured by DNA1-MB, which was isolated from the GCE/MWCNTs-Au NPs/DNA2/DNA1-MB, and were labelled with PS-CdS QDs. Cd²⁺ ions from the collected PS-CdS QDs after magnetic separation were detected by ASV following being dissolved in HNO₃ solution. According to Fig. 5A, the number of MCF-7 cells had a direct correlation with the strength of the Cd²⁺ oxidation peak current at around -0.72 V. The peak current intensity was significantly correlated with the logarithm of the MCF-7 cell concentration in the range from 0 to 1.0×10^6 cells per mL. The linear regression equation was $y = 6.8929 \lg C_{cells} + 1.0357$ (C_{cells} , cells per mL) with a correlation coefficient of 0.9947 ($n = 6$), and the limit of detection (LOD)

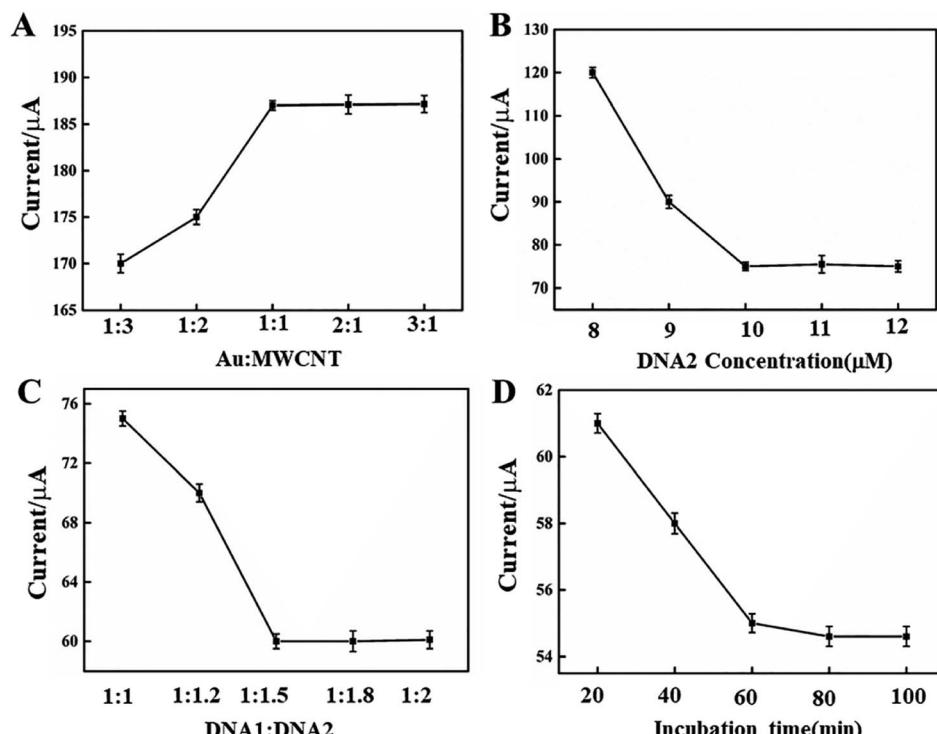


Fig. 4 The effect of (A) molar ratio of Au NPs : MWCNTs, (B) amount of DNA2, (C) amount of DNA1, and (D) incubation time of DNA2 and DNA1-MB in CV. Data are presented as mean \pm SD ($n = 5$ independent experiments).



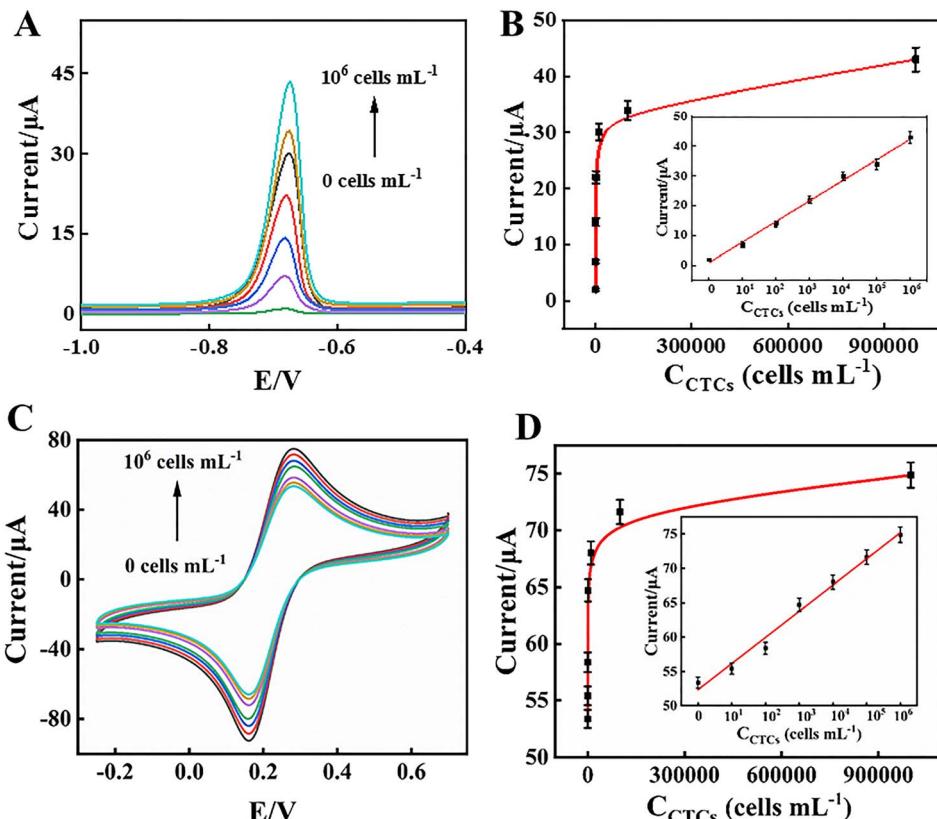


Fig. 5 (A) ASV and (C) CV responses of the electrochemical biosensor incubated with different concentrations of MCF-7 cells in 0.01 M phosphate buffer ($\text{pH} = 6.5$; 0, 10, 1×10^2 , 1×10^3 , 1×10^4 , 1×10^5 , 1×10^6 cells per mL); (B) and (D) calibration curve of the cytosensor for MCF-7 cells.

was 3 cells per mL ($\text{S/N} = 3$) as shown Fig. 5B. As a result, the electrochemical biosensor demonstrated the accurate tumour cell identification at low concentration. Furthermore, the anodic peak currents of the remaining electrode in CV after DNA1-MB had detached from it also presented a linear correlation (Fig. 5C) and the equation was $y = 3.7891 \lg C_{\text{cells}} + 52.3658$ (C_{cells} , cells per mL), with a coefficient of determination (R^2) of 0.9846 and a LOD of 3 cells per mL ($\text{S/N} = 3$), as shown in Fig. 5D. The ASV/CV dual-signal electrochemical biosensor for detection of CTCs indicated a high sensitivity compared with previous reports (Table 1).

The reproducibility of the biosensor was investigated from the response to 1.0×10^3 cells per mL with this dual-signal electrochemical strategy. A series of five electrodes were

prepared for CTC detection in the ESI† (Table 2). The relative standard deviation (RSD) of the measurements for the five electrodes was 2.70%, suggesting that the precision and reproducibility of the proposed biosensor was quite good. To determine the biosensor's selectivity, it was assessed in the presence of HeLa, L929, MCF-7, and a combination of all the

Table 1 Previous reports on electrochemical sensors for detecting CTCs

Electrochemical approaches	Electrochemical materials	Detection range (cells per mL)	Detection limit (cells per mL)	Ref.
Amperometry	Au NP/HRP-TMB	1×10^2 to 5×10^4	5	13
DPV	rGO/AuNPs/CuO nanzyme	50 to 7×10^3	27	41
SWV	Au/Fe ₃ O ₄ -modified graphene	5–500	3	42
LSV	AgNPs/nanofibers	10 to 5×10^5	6	43
EIS	Galactosylated gold-nanoisland	1×10^2 to 1×10^5	30	44
ASV/CV	MWCNTs-Au	0 to 1.0×10^6	3	This work



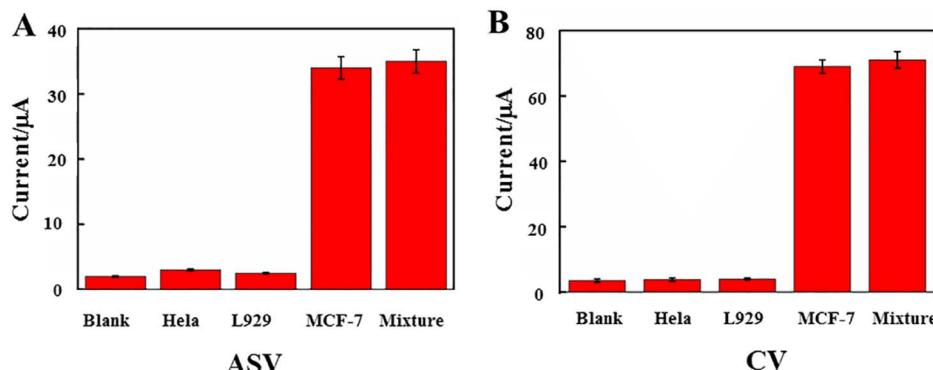


Fig. 6 The detection of different cells (HeLa, L929, MCF-7 and a mixture) in (A) ASV and (B) CV.

Table 3 Stability of the dual-signal electrochemical biosensor (CTC concentration, 10^3 cells per mL)

	Initial value	5 days		8 days		10 days		20 days	
		I_1	$I_1/I_0, \%$	I_2	$I_2/I_0, \%$	I_3	$I_3/I_0, \%$	I_4	$I_4/I_0, \%$
ASV (μA)	22.20	21.75	98.0	21.20	95.50	20.76	93.50	19.98	90.0
CV (μA)	64.80	62.53	96.5	61.75	95.30	59.75	92.20	57.64	89.0

cells (Fig. 6). Other interfering substances had almost no effect on the signal for the detection of MCF-7 cells. To test the stability of this dual-signal electrochemical biosensor, at five, eight, ten, and twenty days, the current response of the as-prepared biosensor used for the detection of CTCs (10^3 cells per mL) was measured, as shown in the ESI† (Table 3). The ASV/CV signal decreased respectively to about 98.0–90.0% and 96.5–89.0% of its initial value, suggesting that the stability of the proposed electrochemical biosensor was good.

4. Conclusions

In conclusion, we demonstrated an electrochemical method for the ultrasensitive detection of CTCs using an ASV/CV dual-signal. This sensor was dependent on the intensity of the Cd^{2+} ion oxidation peak current from the PS-CdS QDs tagged on the MCF-7 cells that were captured by DNA1-MB separated from the capture electrode, GCE/MWCNTs-Au NPs/DNA2/DNA1-MB, which was proportional to the number of MCF-7 cells in ASV, and showed a linear correlation in the range of 0 to 1.0×10^6 ($y = 6.8929 \lg C_{\text{cells}} + 1.0357$ ($R^2, 0.9947$; LOD, 3 cells per mL)). The anodic peak currents of the residual electrode in CV likewise showed a linear correlation after DNA1-MB had separated from the capture electrode ($y = 3.7891 \lg C_{\text{cells}} + 52.3658$ ($R^2, 0.9846$; LOD, 3 cells per mL)). An ASV/CV dual-signal detection strategy for CTCs was developed, which overcomes the limitations of any single-signal mode and improved the detection reliability and accuracy.

Conflicts of interest

The authors declare no competing interests.

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