


 Cite this: *RSC Adv.*, 2023, **13**, 23870

Phthalic anhydride (PA): a valuable substrate in organic transformations

 Kobra Nikoofar * and Mansoorehsadat Sadathosainy

This review has been centralized on applications of phthalic anhydride (PA) as a valuable and significant heterocyclic substrate in two- and multicomponent organic reactions. The article has been subdivided into the following parts: (i) PA introduction by focusing on its characterization, synthesizing procedure, and multiple-aspect applications. In addition, the previous review articles based on PA have also been indicated; (ii) the applications of PA as a substrate have been subdivided into parts with a glance on the reaction components numbers; (iii) the applications of PA in esterification reactions; and (iv) some examples of PA in multistep synthesis. The review covers the corresponding literature up to the end of 2022. According to the abovementioned classifications, PA is a potent substrate to design a wide range of heterocyclic compounds that possess various kinds of properties and applications in chemistry, industry, and pharmaceuticals.

Received 21st May 2023

Accepted 25th July 2023

DOI: 10.1039/d3ra03378c

rsc.li/rsc-advances

1. Introduction

2-Benzofuran-1,3-dione (isobenzofuran-1,3-dione, 1,3-isobenzofuranidone, 1,3-dioxophthalan, phthalandione, 1,3-phthalandione, 1,2-benzenedicarboxylic acid anhydride, phthalic acid anhydride, 1,2-benzenedicarboxylic anhydride, 1,3-dihydro-1,3-dioxoisobenzofuran) with the common/popular name of phthalic anhydride (PA) is a versatile organic compound with the formula $C_6H_4(CO)_2O$. It is a white (or somehow colorless) powder (or flakes and sometimes needles) with a melting point of 131 °C, molar mass of 148.1 g mol⁻¹, and some acidic odor. Actually, this fused bicycle with the LD₅₀ = 1.530 mg kg⁻¹ (oral rat) is the anhydride form of phthalic acid. It is soluble in ethanol, acetone, water, and benzene but sparingly in ether and chloroform.¹ Contact with skin and eye may cause an allergic skin reaction and serious eye damage. Basketter and Kimber published a review article in 2016, in which the absence of both skin- and respiratory-sensitizing capacity of PA was confirmed based on interpreting predictive toxicology tests for skin sensitization.²

This white solid was first synthesized by Auguste Laurent in 1836. The first procedure involved liquid-phase mercury-catalyzed oxidation of naphthalene.³ In the modern process, vanadium pentoxide was used as the catalyst in a gas-phase reaction with naphthalene using molecular oxygen.⁴ Rindone group in 2010 prepared PA *via* the ozonation of naphthalene.⁵

An alternative synthetic procedure is also based on the oxidation of *o*-xylene by vanadium-based catalysts such as V₂O₅/

TiO₂ and anatase-supported vanadium oxide.^{6–8} In 2018, Co-Mn/H₃PW₁₂O₄₀@TiO₂ was also utilized as the catalyst for the selective vapor-phase oxidation of *o*-xylene to PA in the presence of green oxidant.⁹ Boger and Menegola also used extruded monoliths with high thermal conductivity as improved operated and economic catalysts to prepare PA from the oxidation of *o*-xylene.¹⁰ This technology was also studied in detail by Tronconi group in 2012.¹¹ The air oxidation of *o*-xylene over mesoporous V-Mo-MCM-41 molecular sieves was also applied for PA synthesis by Selvaraj and Lee in 2005.¹² PA can be also produced from phthalic acid.¹³ Nikolov's research group published a review article on the synthesis of PA from *o*-xylene in detail.¹⁴ Dias and Portela in 1997 issued a review article that discussed about various catalysts to gain PA from two main routes for its synthesis, which concerns the oxidation of *o*-xylene and naphthalene.¹⁵

Recently some green and novel protocols have been presented to obtain PA such as renewable production from biomass-derived furan and maleic anhydride.¹⁶ Deng group in 2021 also reported green production of PA from biobased furan and maleic anhydride by an acid resin catalyst.¹⁷ PA obtained renewably from 5-hydroxymethylfurfural (HMF) using MoO₃/Cu(NO₃)₂ as a catalyst.¹⁸ Sha group in 2016 designed a conceptual process to prepare PA from corn stover, an agricultural residue. Their techno-economic assessment contends energy integration alternatives as well as water consumption and life cycle greenhouse emissions.¹⁹ Ierapetritou and coworkers in 2015 accounted a novel route for PA production from hemi-cellulose solutions, in which they focused on synthetic process, technoeconomic analysis, and life cycle assessment (LCA) of the protocol.²⁰

Department of Organic Chemistry, Faculty of Chemistry, Alzahra University, P.O. Box 1993891176, Tehran, Iran. E-mail: k.nikoofar@alzahra.ac.ir; kobranikoofar@yahoo.com; Fax: +98 2188041344; Tel: +98 2188041344



Kinney and Pincus in 1951 reported PA production from catalytic air oxidation of some substrates such as certain higher aromatics and coal tar fractions.²¹ The PA synthesis by the oxidation of tar oils was also represented by Shelmerdine's group in 1953.²²

Prichard in 1956 obtained PA from the reaction of bromobenzene and carbon monoxide in the presence of sodium carbonate and nickel carbonyl.²³

Zazhigalov and Kizium in 2017 reported PA production by Diels–Alder reaction during *n*-pentane oxidation on vanadium phosphorus oxide (VPO) catalyst.²⁴ Kim and Yang in 2000 studied the synthesis of maleic and phthalic anhydrides from the mixture of cyclopentene and 1-pentene *via* their selective oxidations under different reaction parameters.²⁵

In 1919, Gibbs published the title “phthalic anhydride. I-introduction”, which discussed the different synthetic procedures of PA.²⁶ Gibbs in 1920 also interpreted the title “history of the preparation and properties of pure phthalic anhydride”.²⁷

The chemical engineering design of the PA preparation reactor is considered based on the computer model that simulates the operation of the reactor over a range of conditions that lead to both satisfactory mass transfer and heat transfer requirements.²⁸ Wainwright and Foster in 1979, reviewed the “catalysts, kinetics, and reactor design in phthalic anhydride synthesis”.²⁹

PA has been utilized in different fields of science and technology. In polymerization technologies, some examples such as preparation of ternary polymeric system including polysulfide (PS), diglycidylether of bisphenol A resin, and PA have been reported. PA plays a role in the curing reaction of this ternary polymer.³⁰ The copolymerization of PA with epoxides catalyzed by amine-bis(phenolate)chromium(III) complexes also performed in 2021.³¹ Amin's group in 2011 prepared two hyperbranched polyesteramides (HB1 and HB2) by the bulk reaction between PA and diisopropanolamine (DiPA) or diethanolamine (DEA), respectively. The effects of various solutions of HB1 and HB2 on the properties (such as measurements of water of consistency, setting times, bulk density, apparent porosity, and compressive strength for the cement pastes) of ordinary Portland cement (OPC) and Portland limestone cement (PLC) were studied.³² Duchateau's group constructed a partially renewable polyester *via* the catalytic ring-opening copolymerization of limonene oxide and PA.³³

Pooley and coworkers in 2005 copolymerized PA as an electrophilic monomer with aziridine or 2-methylaziridine as nucleophilic monomers in the absence of initiator under various experimental conditions.³⁴

PA was utilized as a bridge for the grafting of chitosan biopolymer onto wool fabric to prepare antibacterial agents.³⁵ It also played a role in the chemical modification of chitosan to obtain bacteria inhibitors.³⁶ In 2020, a novel biodegradable diblock/triblock poly(ester-bicarbonate)s was prepared from cyclohexene oxide (CHO), propylene oxide (PO), PA, and CO₂ in a one-pot/one-step protocol.³⁷ The surface modification of silk fiber using PA to graft the polysaccharide chitosan has been reported in 2009, which lead to the dyeing ability of the grafted silk. The grafted samples possess antibacterial potential.³⁸

PA is also a critical substrate to prepare a series of new optically active and thermally stable polyamides (PAs).³⁹ It plays the role of a chain end group in the fluorescent PMMA polymers. These polymers can detect the intermacromolecular reaction in reactive polymer blends at a low concentration using fluorescence-gel permeation chromatography.⁴⁰

It has also utilized as a key substrate for the multistep preparation of novel optically active polyamides derived from 5-(3-methyl-2-phthalimidylpentanoylamino)isophthalic acid.⁴¹ PA was also utilized for the preparation of chiral polyesters through enantioselective terpolymerization with racemic and meso-epoxides and also sequence-controlled block copolymers.⁴²

PA was also applied as a modifier in poly(butylene succinate-co-butylene adipate) (PBSA)/cellulose nanocrystal (CNC) composites. The modified PBSA/CNC composites demonstrated elevated mechanical properties, fast crystallization, and improved hydrophobicity.⁴³ PA has been added in the preparation procedure of poly(butylenes succinate)/cellulose nanocrystals (PBS/CNC) composite *via* melt blending, which yielded higher crystallinity and smaller crystals.⁴⁴

To obtain phthalylated cellulosic compounds with higher DS (degree of substitution), the chemical modification of sugarcane bagasse cellulose has been reported with PA in the 1-butyl-3-methyl imidazolium chloride solvent and 4-dimethylamino-pyridine (DAMP) catalyst.⁴⁵

PA is also significant in different fields of biological and pharmaceutical applications. It is a key substrate in the synthesis of symmetrical novel organoselenocyanates and diselenides dye stuffs, which was evaluated for the antitumor properties.⁴⁶ In 2012, *N*-benzoyl 3-nitro-phthalimide, which possesses anxiolytic activity in mice model, was prepared from PA as the substrate.⁴⁷

Bis[aqua-1,8-(1,2-dicarboxamido benzene)3,6-diazaoctane copper(II)/nickel(II)] tetrachloride, as two binuclear complex, was synthesized by a two-component one-pot metal template condensation between PA and 1,8-diamino-3,6-diazaoctane. The complexes are able to bind to calf thymus (CT)-DNA under physiological pH.⁴⁸ Lepoittevin *et al.* in 2021 investigated the reactivity of PA as a chemical respiratory sensitizer toward reconstructed human epidermis (RHE).⁴⁹

Spanedd and Bourel-Bonnet published a review article about the potential of cyclic anhydrides such as PA in bioconjugation to functionalize the biomolecules and carriers. The pH-dependent stability and reactivity, as well as the physical properties, can be tuned by the structure of the cyclic anhydride used. Thus, their application in smart delivery systems has become very important.⁵⁰

Alheety in 2021 prepared new complexes of PA (as ligand) with some cations (such as Co(II), Ni(II), Cu(II), Mn(II), and Zn(II)) in 1:1 molar ratio. The microbicide activity studies of the synthesized complexes against four types of bacteria (*E. coli*, *S. epidermidis*, *K. pneumoniae*, and *S. aureus*) was also reported. The complexes also demonstrated stability on laser beams for 10–30 s.⁵¹

Marzouk's group in 2016 designed and manufactured some new phthalazinones containing benzyl moiety *via* a multistep reaction started from PA. Some of the heterocycles



demonstrated antitumor activity.⁵² Yamaguchi and coworkers in 1998 constructed some 2-[2-(*l*-imidazolyl)alkyl]-1(2*H*)-phthalazinones as novel antiasthmatic agents with dual activities of thromboxane A2 synthetase inhibition and bronchodilation.⁵³ Costa Silva group in 2021 modified Chicha gum with PA to obtain a new biologically active material that demonstrated excellent inhibitory effect against *P. aeruginosa* and *K. pneumoniae* species (rating 100% inhibition) and could also inhibit *Escherichia coli* growth.⁵⁴ Bold *et al.* in 2000 synthesized new anilinophthalazines as potent and orally well absorbed inhibitors of the VEGF receptor tyrosine kinases useful as antagonists of tumor-driven angiogenesis. The key substrate for the preparation procedure starts from PA.⁵⁵

PA also has a special role in sensors. In 2016, phthaloylchitosan (PHCS) was synthesized simply and cost-effectively using chitosan and PA by microwave irradiation, which was applied to determine tyrosine with high sensitivity and good selectivity through carbon nanotube film-coated glassy carbon electrode.⁵⁶ Zhang *et al.* in 2022 reported a novel phthalic anhydride-based room-temperature phosphorescence (RTP) emitter with the lifetime longer than one second.⁵⁷

A novel fluorescent probe IMPD, based on imidazo[1,2-*a*]pyridine that contained PA moiety in its structure, was designed and synthesized by Huang's research group in 2021. The probe could detect hydrazine *via* its maleimide as the recognition group. In addition, the probe IMPD could dye the HepG2 cell with blue color in the presence of hydrazine.⁵⁸

Another notable application of PA is in the dye industry. It utilized to prepare the nanosized copper phthalocyanin blue (CuPc) pigments.⁵⁹ The phenol precipitation and dye bleaching capabilities of phthalic anhydride-modified horseradish peroxidase C (PA-HRP) were also investigated in 2000.⁶⁰

Polymeric surfactants have been synthesized by the reaction of maleic anhydride (MA), polyethylene glycol (PEG), and PA, which exhibit excellent surface-active properties (including surface tension, low-foaming, solubilization, and dispersant properties) in disperse dye systems.⁶¹

PA also plays a role in separation and waste removal. In 2016, the Gurgel group prepared chemically-modified sugarcane bagasse, named as carboxylate-functionalized sugarcane bagasse (SPA), *via* the reaction of PA with sugarcane bagasse. The SPA adsorbent was used to remove Co²⁺, Cu²⁺, and Ni²⁺ from aqueous solution in mono- and multicomponent systems in the batch mode.⁶²

Another aspect of the PA usage is in the field of catalyst and protecting group. It was utilized as a part of the co-catalytic system (in combination with Zn(OTf)₂) to promote Beckmann rearrangement.⁶³ The efficient and clean oxidation of sulfides to sulfones (not the probable sulfoxide) with urea-hydrogen peroxide in the presence of PA in ethyl acetate was also performed in 2018.⁶⁴ PA also worked as a remarkable protective group in the synthesis of a dipeptide (β -alanine-*L*-histidine) *via* two procedures, which are solution phase peptide synthesis (SPS) and solid phase peptide synthesis (SPPS).⁶⁵

Liu *et al.* in 2017 utilized PA as a low-temperature activator in the H₂O₂ bleaching system for cotton fabric. The performance of the H₂O₂/PA bleaching system was investigated by measuring

the CIE whiteness index (WI) of the bleached cotton fabric, H₂O₂ decomposition rate, and bursting strength, respectively.⁶⁶

Maldas and Kokta in 1990 considered the performance of PA as a coupling agent in wood fiber-filled polystyrene composites. Its presence evaluated the mechanical properties of the composite materials.⁶⁷

Duan in 2019 interpreted PA-promoted ring-opening cationic polymerization of cyclohexene oxide catalyzed by dinuclear chromium complex supported by piperazine-bridged [ONSO] ligand to obtain atactic poly(cyclohexeneoxide) polymer. The formation of carbocation species by interaction between PA and bimetallic chromium complexes is the real initiator center.⁶⁸

The hydrogenation of PA is another important reaction that has been evaluated extensively. Liquid phase selective hydrogenation of PA to phthalide (an important industrial intermediate for pharmaceuticals, fine chemicals, and organic synthesis) was reported in 2015 in the presence of Au/FeO_x-TiO₂ catalyst.^{69,70} Liquid phase hydrogenation of PA to phthalide over Au/TiO₂ catalysts (with different gold loadings) was reported in 2009.⁷¹ Different catalytic systems were reported for PA hydrogenation to phthalide, such as acid-tolerant intermetallic cobalt-nickel silicides as noble metal like catalysts,⁷² CoSi₂/CNTs,⁷³ hydrophobic activated carbon supported Ni-based acid-resistant catalyst,⁷⁴ and Al₂O₃-supported NiCu alloy.⁷⁵

PA is applicable in some other fields as well. Al-Sawaad and Alwaaly research group in 2021 reported bithiourea-phthalatonickel(II) complex (PTUNI) *via* the reaction of NiCl₂·6H₂O with thiourea (2 mol) and PA (1 mol). The complex was evaluated as a corrosion inhibitor for carbon steel alloy (C1010) against a corrosive medium of 0.1 M hydrochloric acid at 298 K.⁷⁶ Fouda group in 2013 also reported anhydride derivatives (such as PA) as corrosion inhibitors for carbon steel in hydrochloric acid solutions.⁷⁷ In 2022, glycerol/PA novel nanocomposite was introduced for microwave absorbing applications.⁷⁸

Velayutham and coworkers in 2009 demonstrated the synthesis and characterization of polyurethane (PUR) coatings derived from polyols synthesized with glycerol, PA, and oleic acid. The utilized polyols were designated as Alk28, Alk40, and Alk65, in which 28%, 40%, and 65% of oleic acid was present, respectively. The coatings obtained from polyol Alk28, with the lowest percentage of oleic acid content, exhibited the best overall physicochemical properties, followed by Alk40. PUR from polyol with the highest percentage of oleic acid content, Pualk65 coatings, were softer and their anticorrosive properties were less satisfactory.⁷⁹ Son in 1975 studied the role of PA in cure retardation of rubbers.⁸⁰

Based on the importance of PA in diverse fields of science and technology, many review articles have been written. Dubey and coworkers in 1996 published a review article about the importance of phthalic anhydride as a petrochemical agent.⁸¹ In 2002, a review article entitled "Some new applications of phthalic anhydride in organic synthesis" was published by the Iordache research group.⁸² In 2021, Elgharrawy reported a mini-article entitled "A review on phthalic anhydride industry and uses".⁸³ In 2016, Basketter and Kimber discussed about PA as



a chemical allergen in detail. They claimed that it displays a differentiated behavior, whereas most respiratory sensitizers are known also to give rise to delayed skin reactions; evidence for PA suggests that it only causes immediate type allergy.⁸⁴ Siddiqui and Javed research group study the computational, spectroscopic, Hirshfeld surface, electronic state, and molecular docking properties (with 21 different protein receptors) of PA.⁸⁵ Hong *et al.* in 2017 investigated the antiinflammatory effect of titrated extract of *Centella asiatica* in PA.⁸⁶ In 1977, a patient with occupational asthma caused by PA was reported.⁸⁷ Yanagimoto in 1956 discussed about the reaction between urea and PA under pressure.⁸⁸ The Martin group in 2015 published a book-chapter entitled “Anhydride-based multi-component reactions”, which contains some reactions of PA.⁸⁹

2. Applications of PA in two-component reactions

Katritzky and Yates in 1976 prepared 6*H*-benzimidazo[1,2-*b*][2,4]benzodiazepine-7,12-dione (**3**) *via* the catalyst-free reaction of PA (**1**) and 2-aminobenzimidazole (**2**) (Scheme 1).⁹⁰

Gitis group in 2000 reported the reaction of PA with 2-methylimidazole that formed amide.⁹¹ The authors also examined the reaction of maleic anhydride that leads to the formation of molecular complexes.

Hajipour *et al.* in 2000 interpreted the phthloyation microwave-assisted reaction of PA (**1**) with amino acids (**4**) under solvent-free conditions to obtain phthalimide derivatives without racemization (**5**) (Scheme 2).⁹² The substrates were mixed with silica gel and irradiated with microwave (900 W) for an appropriate time.

Billman and Harting in 1948 obtained the phthalyl derivatives of different kinds of amino acids (**5**) *via* the reaction of PA (**1**) with (**4**) at 180–185 °C within 15 min with 30–92% yield.⁹³ Zeng research group in 2004 also reported the *N*-phthaloylation of amino acids (Gly, Ala, Phe, Val, 10 mmol) with PA or phthalic acid (11 mmol) with amino acids at 130–135 °C under pressure (about 40 mmHg) for 15–30 min with 79.4–90.5% yield.⁹⁴ Kidd and King in 1948 also reported the preparation of phthalyl-L-glutamic acid.⁹⁵ Leite *et al.* in 2014 demonstrated the microwave-assisted synthesis of some phthaloyl amino acids from the reaction of PA (**1**) and amino acids (**4**, Gly, Ala, Val, Glu, Phe, Ile, and Trp) in a 1 : 1 molar ratio in TEA or 4-DMAP (0.5

mL) and DMF (three drops) within 2 min by 31.9–96%. The products presented antioral inflammatory activity comparable to thalidomide. Most of the compounds effectively suppressed nitric oxide production in murine cells stimulated with lipopolysaccharide.⁹⁶ Ökten and coworkers in 2022 reported the facile, expeditious, and cost-effective preparation of *N*-phthaloyl (*S*)-amino acids and evaluated their *in silico* activities against *Staphylococcus aureus*.⁹⁷

In 1958, the reaction of PA with different amino acids by refluxing in nonpolar solvents (such as benzene and toluene) in the presence of triethylamine was performed. By separating the water formed in the reaction, the phthalimide derivatives were prepared in good yields and without racemization. Phthaloylation without racemization may also be carried out in *N,N*-dimethylformamide medium.⁹⁸

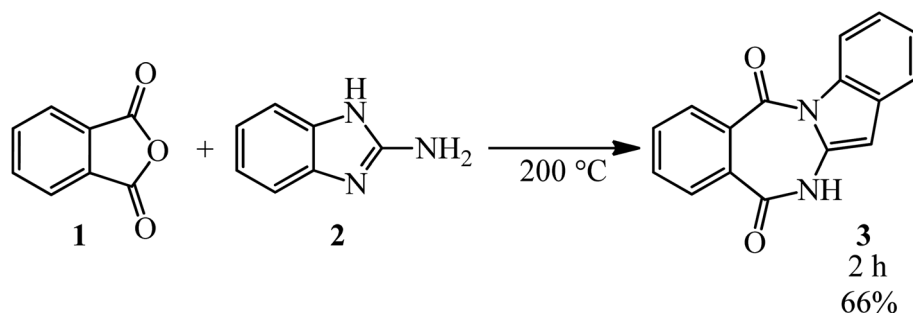
Homsı and Kasideh in 2015 obtained *N*-phthalimide amino acids from PA and amino acids (Gly, Ala, Phe, Val, Leu, and Asp) in refluxing glacial AcOH for 2 h with 66.8–95.8% yield. The synthesized compounds, which were purified through recrystallization from ethanol, were screened for their antimicrobial activity against four microorganisms, namely, *Streptococcus epidermidis*, *Escherichia coli*, *Mycobacterium tuberculosis*, and *Candida albicans*.⁹⁹

Safari *et al.* in 2009 achieved quinophthalone pigments (**7**) through the reaction of PA (**1**) and 2-methylquinolines (**6**) in the presence of BF₃/Et₂O as the catalyst under solvent-free and reflux conditions (Scheme 3).¹⁰⁰

Loghmani-Khouzani in 2004 also gained some quinophthalones (**7**) *via* the microwave-assisted (700 W) reaction of PA (**1**) and 2-methylquinolines (**6**) in a 1 : 1 molar ratio in the presence of silica gel (silica gel 60, 230–240 mesh Merck, 300 mg) within 2 min by 85–97%.¹⁰¹

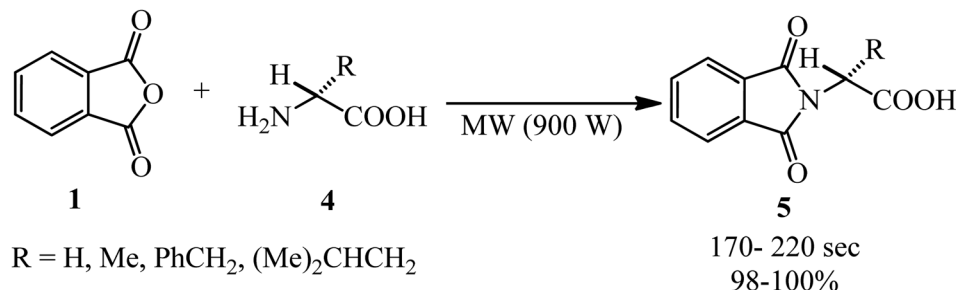
Phillips and Goss in 1926 obtained methyl-isopropyl-quinoline yellow (**11**) through the reaction of PA with methylisopropyl-quinaldine (**10**). The product (**10**) was also achieved from 2-amino-*p*-cymene (**8**) and paraldehyde (**9**) (Scheme 4).¹⁰²

Safari *et al.* in 2012 performed simple regioselective one-pot solvent-free reaction of 2-methylpyridines (**12**) and PA (**1**) in the presence of BF₃·nano SiO₂ (as a solid supported catalyst) at conventional heating and also under microwave irradiation to prepare the corresponding pyrophthalones (**13**) (Scheme 5).¹⁰³

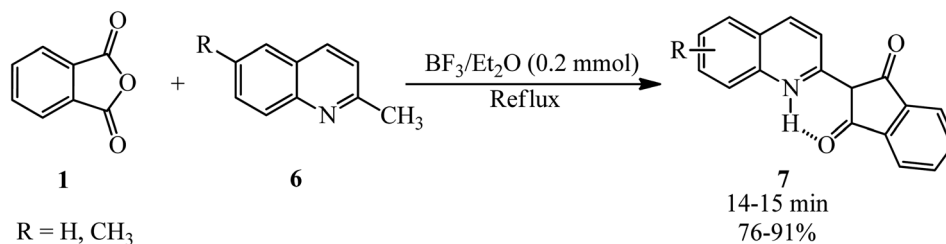


Scheme 1 Synthesis of 6*H*-benzimidazo[1,2-*b*][2,4]benzodiazepine-7,12-dione.

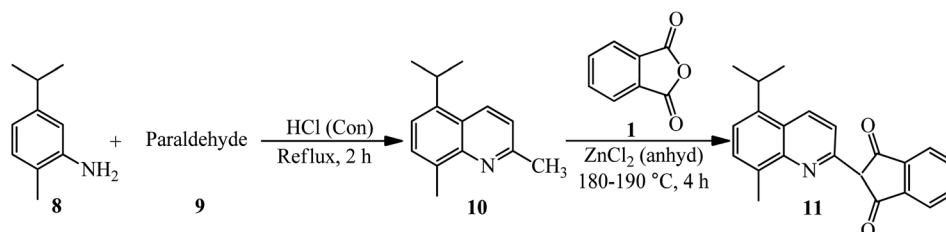




Scheme 2 Synthesis of phthalimide derivatives.



Scheme 3 Synthesis of quinophthalones.

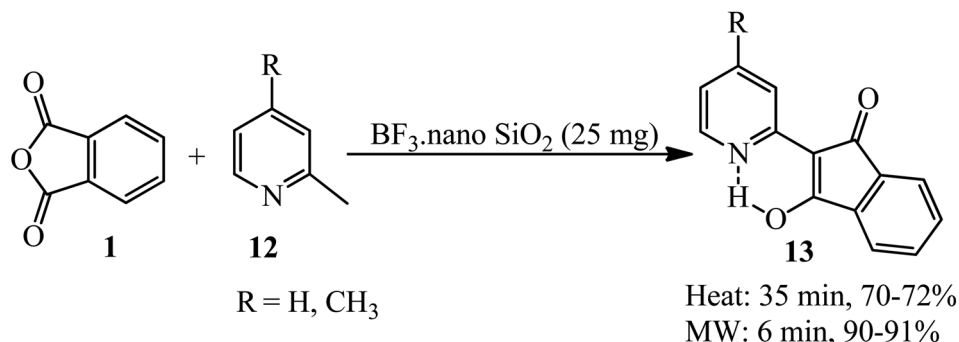


Scheme 4 Synthesis of methyl-isopropyl-quinoline yellow.

Cook and Martin in 1954 acquired orange crystals of 2-(6-methyl-2-pyridyl)-1,3-indanedione (**15**) *via* the reaction of PA (**1**) and 2,6-dimethylpyridine (**14**) in equimolar ratio under reflux conditions within 1.5 h. Their attempts to perform the condensation of both methyl groups through using 2 equimolar of PA were not successful, and only the mono-condensation product (**15**) as orange crystals was isolated (Scheme 6).¹⁰⁴ The

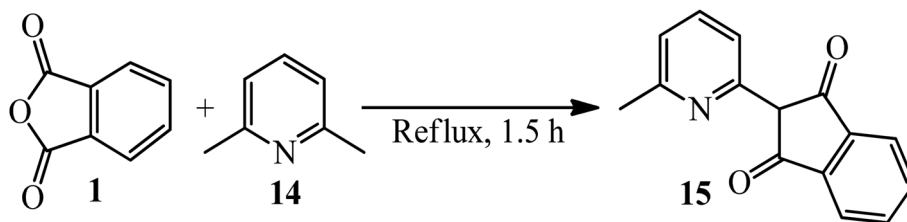
authors discussed about the chelating tendency of the obtained product with some bivalent-metal ions at 30 °C in a 75% (v/v) dioxane–water solution.

Bayat *et al.* in 2010 reacted PA (**1**) and dialkyl acetylenedicarboxylates (**17**) to afford the novel spirocyclic compounds (**18**, **19**) at room temperature *via* the promoting effect of triphenylphosphine (**16**) (Scheme 7).¹⁰⁵ Although Ph_3P has been utilized

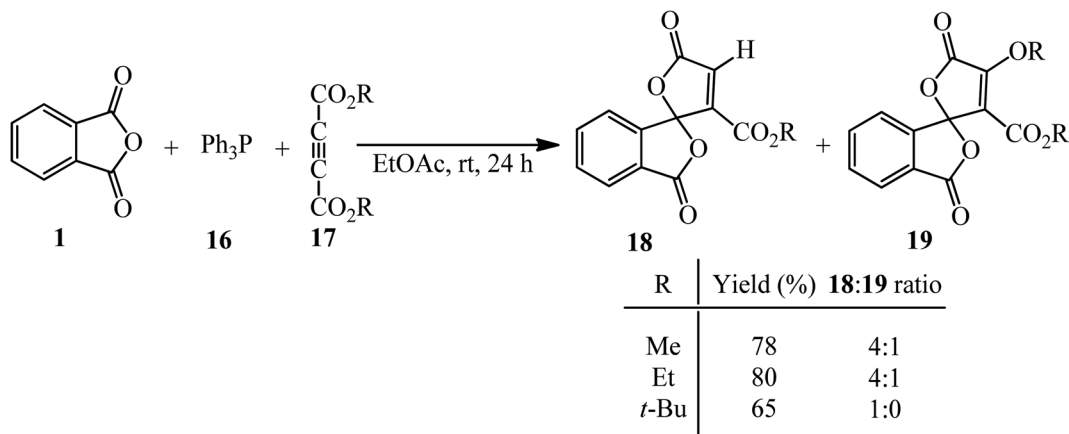


Scheme 5 Synthesis of pyrophthalones.





Scheme 6 Synthesis of (6-methyl-2-pyridyl)-1,3-indanedione.



Scheme 7 Synthesis of spirocyclic compounds based on PA.

in 1 mmol amount, similar to the other two substrates, but according to the proposed mechanism (Scheme 8) and also the products structure, its role is just that of a promoting agent. According to the mechanism, the first step is the formation of 1,3-dipolar intermediate (A), which attacked PA (1) to obtain zwitterionic intermediate (B), which cyclized to the spiro intermediate (D). Water attack to the positively charged phosphorus ion of (D) formed (F), followed by a proton transfer and loss of Ph_3PO , which led to product (18). In pathway A, intermediate (D) was attacked by the alkoxy anion, and the subsequent loss of Ph_3P gave compound (19).

Sharma's research group in 2012 synthesized some Schiff bases possessing imide moieties through multistep reactions initiated from the reaction of PA (1) and 4-aminobenzaldehyde (19) in dichloromethane to obtain 4-(1,3-dioxoisindolin-2-yl) benzaldehyde (20). In the second step (20), reacted with anilines (21) in the presence of glacial acetic acid to obtain the Schiff bases (22), which demonstrated analgesic and anti-inflammatory activities (Scheme 9).¹⁰⁶

Kumar *et al.* in 2014 described the reaction of PA (1) with amines (21) to obtain monoacid monoamides (A), which get the corresponding cyclic imide derivatives (23) in the presence of SOCl_2 . The PA also reacted with KOH to obtain monomethyl ester (B), which reacted with amines in the presence of 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimid hydrochlorid (DEC·HCl) and *N*-hydroxy benzotriazole (HOBt) to get the cyclic imide derivatives (24) (Scheme 10).¹⁰⁷

Habibi and Mohammadkhani Pordanjani in 2017 developed an efficient and easy catalytic protocol for the preparation of

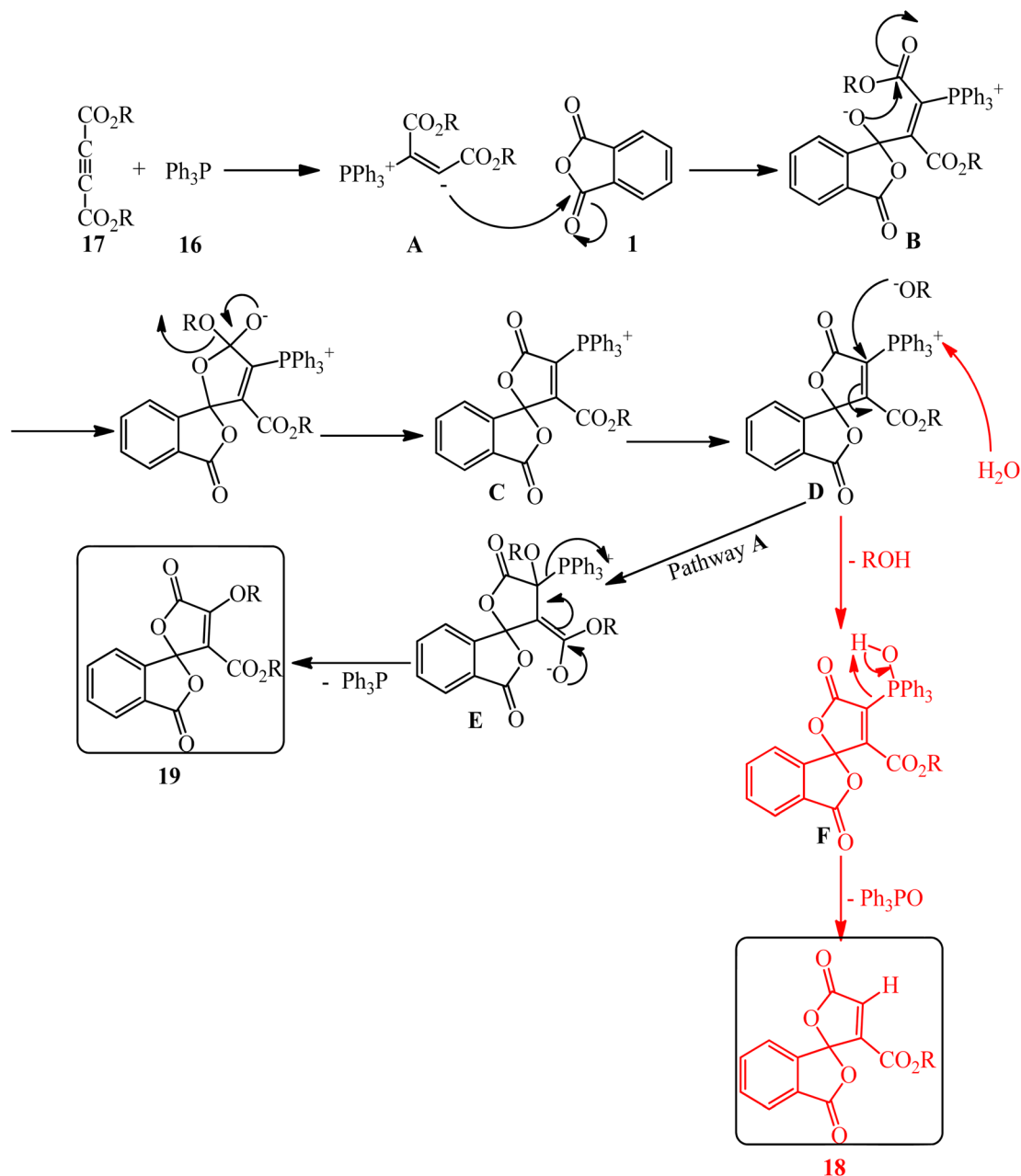
various isoindoline-1,3-diones (23) *via* the reaction of PA (1) with aromatic, aliphatic, and benzylic amines (21), using a catalytic amount of phthalimide *N*-sulfonic acid in ethanol at 80 °C (Scheme 11).¹⁰⁸

Chia group in 2019 prepared the isoindoline-1,3-diones *via* the reaction of anilines and PA in the presence of "water extract of onion peel ash" (WEOPA) (2 mL) as a green catalytic-solvent system at 80 °C within 8 min by 64–92%.¹⁰⁹ The WEOPA was recycled and reused for 5 runs without significant activity loss.

Aliabadi's group in 2014 obtained a series of phthalimides (23) through the reaction of PA (1) and anilines (21) in toluene solvent (reflux, 24 h) in the presence of Et_3N with 21–80% yield. The antiepileptic activity of the products was investigated using two experimental models, namely, maximal electroshock (MES) and pentylenetetrazole (PTZ), and the obtained results were compared with diazepam as the reference drug. The neurotoxicity of the compounds was also evaluated using the rotarod model. The presence of *para*-methoxy substituent in the product showed anticonvulsant activity in the MES (maximal electroshock) model. None of the tested compounds demonstrated acceptable protection in subcutaneous PTZ (pentylenetetrazole) model.¹¹⁰

Patel's group in 2022 achieved some new classes of isoindoline-1,3-diones (23) *via* the reaction of PA (1) and the primary amino-containing compounds (21) in refluxing glacial acetic acid as the solvent-catalyst within 3 h with 62–78% yield. The quantum chemistry-based investigations of the products as antimycobacterial agents (toward the *H37Rv* strain by a dual read-out assay method) was also performed. Computational





Scheme 8 Proposed mechanism of PA-based spirocyclic compounds synthesis.

studies such as density functional theory (DFT) study, molecular docking, and dynamic simulation studies illustrated the reactivity and stability of the synthesized compounds as *InhA* inhibitors.¹¹¹

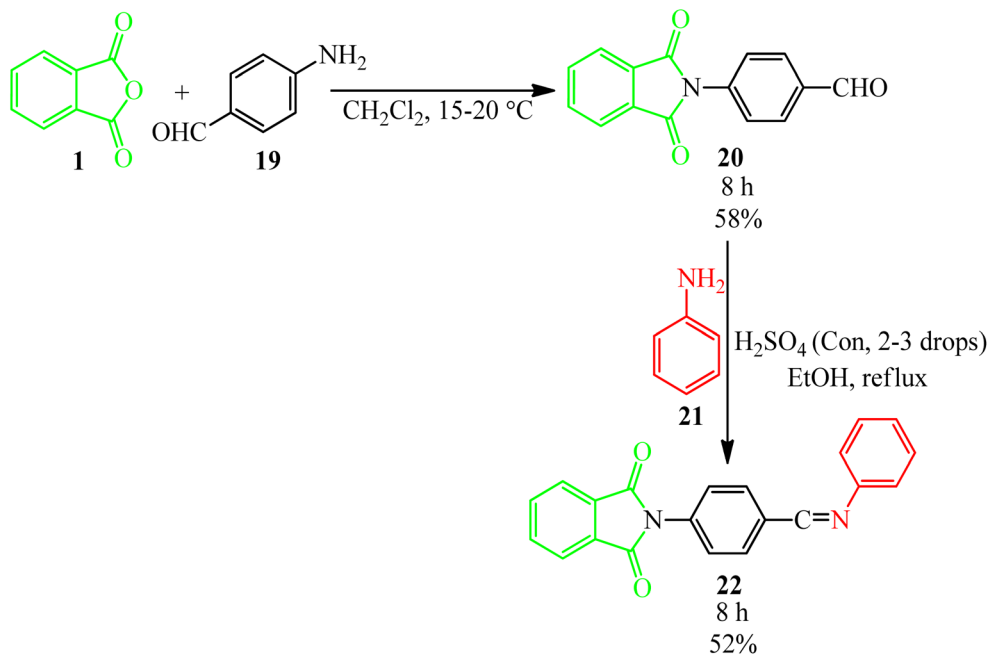
Hamdi *et al.* investigated some phthalimides *via* the reaction of PA (1) and amines (21) (aromatic, aliphatic, and benzylic) in 1 : 1.1 molar ratio in the presence of *p*-TSA (50 mg) in refluxing toluene within 3 h with 29–88% yield. The synthesized compounds were screened for their antimicrobial activities against Gram-positive bacterial strains (*Micrococcus luteus*, *Listeria monocytogenes*, *Staphylococcus aureus*, and *Bacillus cereus*), a Gram-negative bacterial strain (*Salmonella typhimurium*), and a fungus (*Candida albicans*). The cytotoxicity studies of the

phthalimides were conducted in two human cancer cell lines, namely, *MDA-MB-231* and *MCF-7*.¹¹²

Al-Mousawi in 2010 achieved 2-phenylisoindole-1,3-dione (23) *via* the reaction of PA (1) and aniline (21) under solvent-free conditions for 30 min in a focused microwave oven at 160 °C with 96% yield.¹¹³

Singh Bisht and Rajat Bisht in 2021 reported the reaction of PA with some amines (such as urea, glycine, aniline, and sulphanic acid) to yield various phthalimide derivatives (23) using domestic microwave with 70.7–80.21% yield. All synthesized derivatives were subjected to DDPH scavenging activity, which showed good to high antioxidant potential (69.56%) in the presence of ascorbic acid as the standard.¹¹⁴

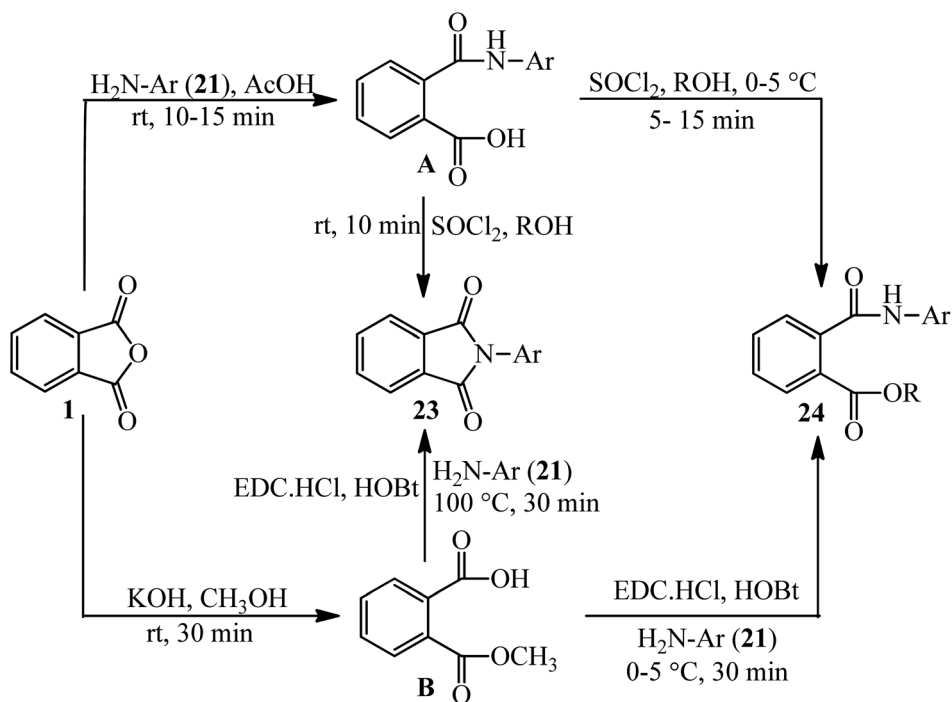




Scheme 9 Synthesis of 4-(1,3-dioxisoindolin-2-yl)benzaldehyde.

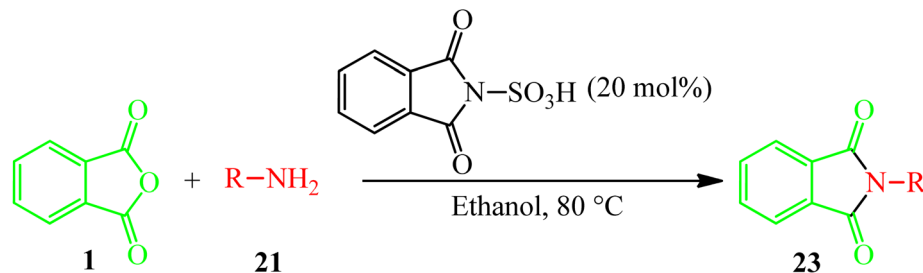
Wicks and Chen in 1979 mixed equimolar amounts of PA with two kinds of alcohols. First, the reaction of PA (**1**) and 2-amino-2-methyl-1-propanol (AMP) (**24**) yielded amide 1-AMP, imide 2-AMP, and/or ester 3-AMP. Amide 1-AMP undergoes an acyl shift to ester 3-AMP, apparently representing the first

example of an *N*- to *O*-acyl shift in the absence of strong acid. Second, the reaction of PA (**1**) and 2-aminoethanol (AE) (**25**) yielded only amide 1-AE or imide 2-AE. No acyl shift of amide 1-AE to ester 3-AE was detected in this case (Scheme 12).¹¹⁵

R = CH₃, C₂H₅Ar = C₆H₅, 4-CH₃C₆H₄, 2-CH₃C₆H₄, 4-ClC₆H₄, 4-OCH₃C₆H₄, 4-BrC₆H₄

Scheme 10 Cyclization of amidic acid esters.

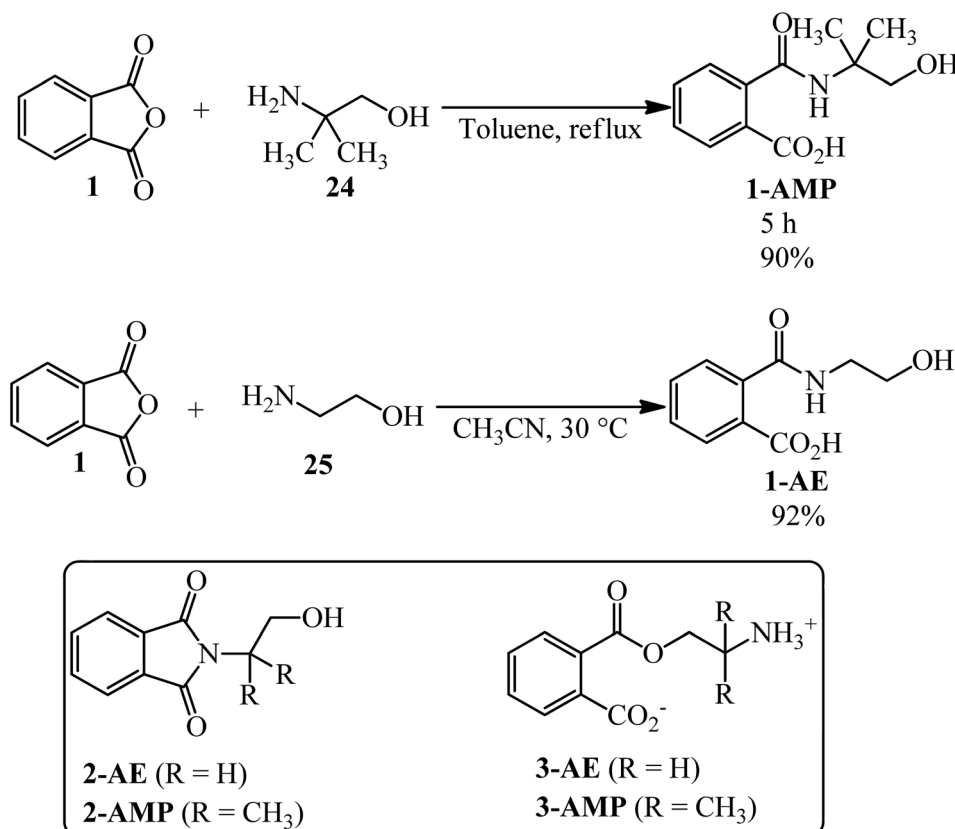




R = C₆H₅, C₆H₅CH₂, 4-MeC₆H₄, 2-MeC₆H₄, 4-MeOC₆H₄, 4-ClOC₆H₄, 4-IOC₆H₄, 2,4-Me₂OC₆H₃, 2,6-Me₂OC₆H₃, (CH₂)₃OH, 2-Pyridyl, 1-Naphthyl

2-4 h
86-93%

Scheme 11 Synthesis of isoindoline-1,3-diones.



Scheme 12 Reaction of PA with 2-amino-2-methyl-1-propanol (AMP) and 2-aminoethanol (AE).

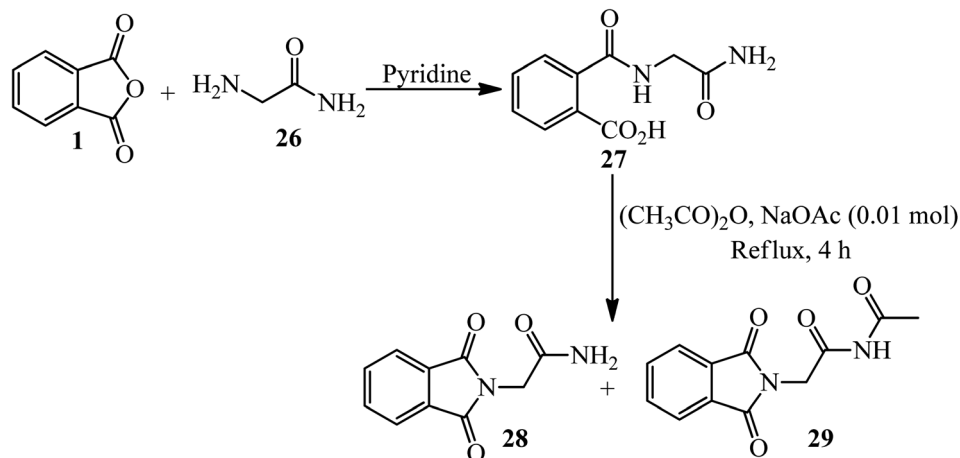
Hassanzadeh's group in 2017 supplied cyclic imides. In the first step, PA (**1**) reacted with glycine (**26**) in freshly distilled and dried pyridine under reflux conditions to yield the corresponding amic acid (**27**) within 5 h. The amic acid underwent ring closure with acetic anhydride and anhydrous sodium acetate to form imides (**28**, **29**), which were isolated *via* column chromatography (Scheme 13).¹¹⁶

In another procedure, the reaction of PA and 2-amino-benzylamine (**30**) in pyridine within 5 h under reflux conditions gave the corresponding cyclic imide (**31**) (Scheme 14).¹¹⁶ The imides were screened for their antimicrobial activities against three types of bacteria and one type of fungi.

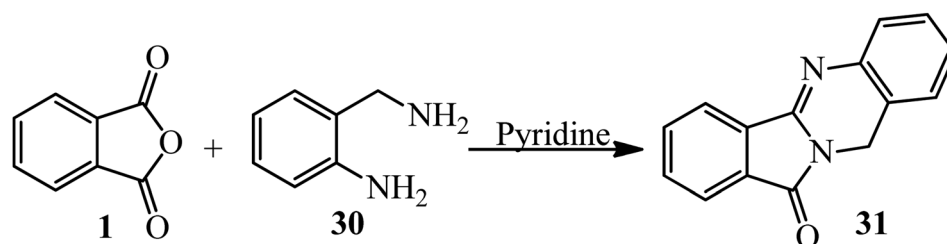
Phthalimide derived from benzylamine exhibited remarkable antimicrobial activity against *E. coli*.¹⁰⁹

Islami *et al.* in 2015 obtained 4-(2*H*-isoindol-2-yl)butanoic acid (**33**) by the reaction of PA (**1**) with 4-aminobutanoic acid (**32**), which was followed by the reaction with Mukaiyama's reagent (2-chloro-1-methylpyridinium iodide, **A**) in the presence of triethylamine in dichloromethane. In the second step, novel diaryl-3-[2-(2*H*-isoindol-2-yl)ethyl]azetidino-2-ones (**35**) was synthesized *via* the reaction of 4-(2*H*-isoindol-2-yl)butanoic acid (**33**) with aromatic imines (**34**). The reaction proceeded through the *in situ* generation of a novel ketene containing an isoindole ring (**C**) and the subsequent electrocyclic reaction of

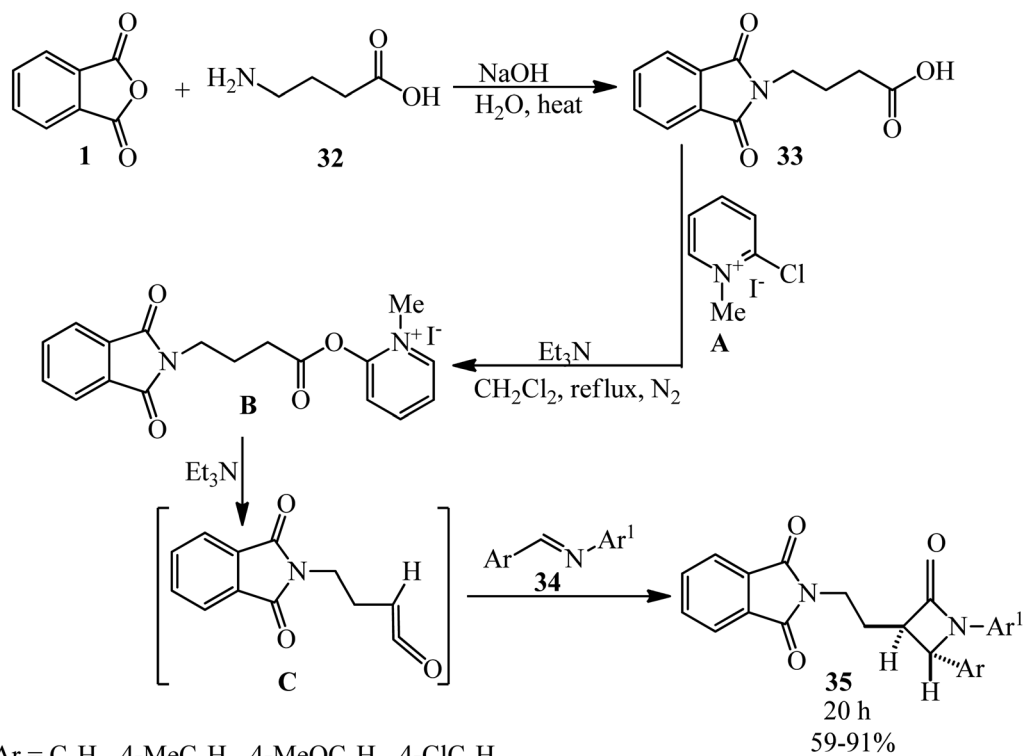


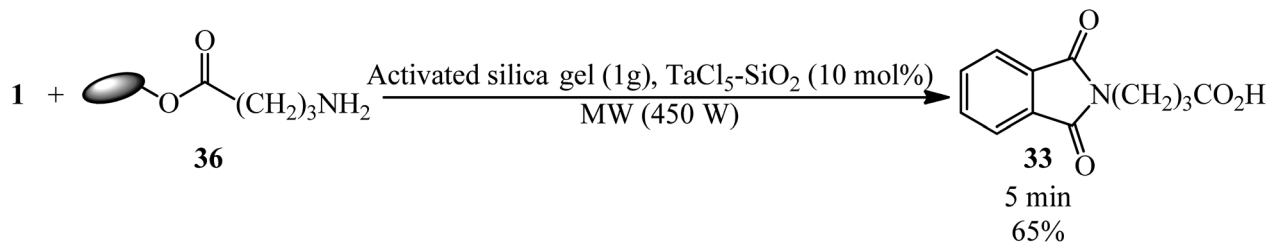


Scheme 13 Synthesis of imides.

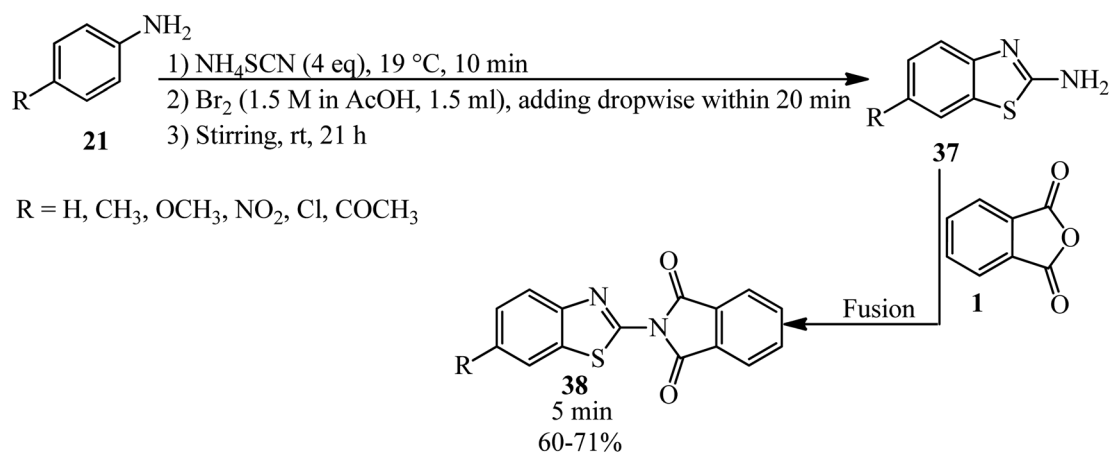


Scheme 14 Synthesis of cyclic imides.

Scheme 15 Stereoselective synthesis of 1,4-diaryl-3-[2-(2*H*-isoindol-2-yl)ethyl]azetidine-2-ones.



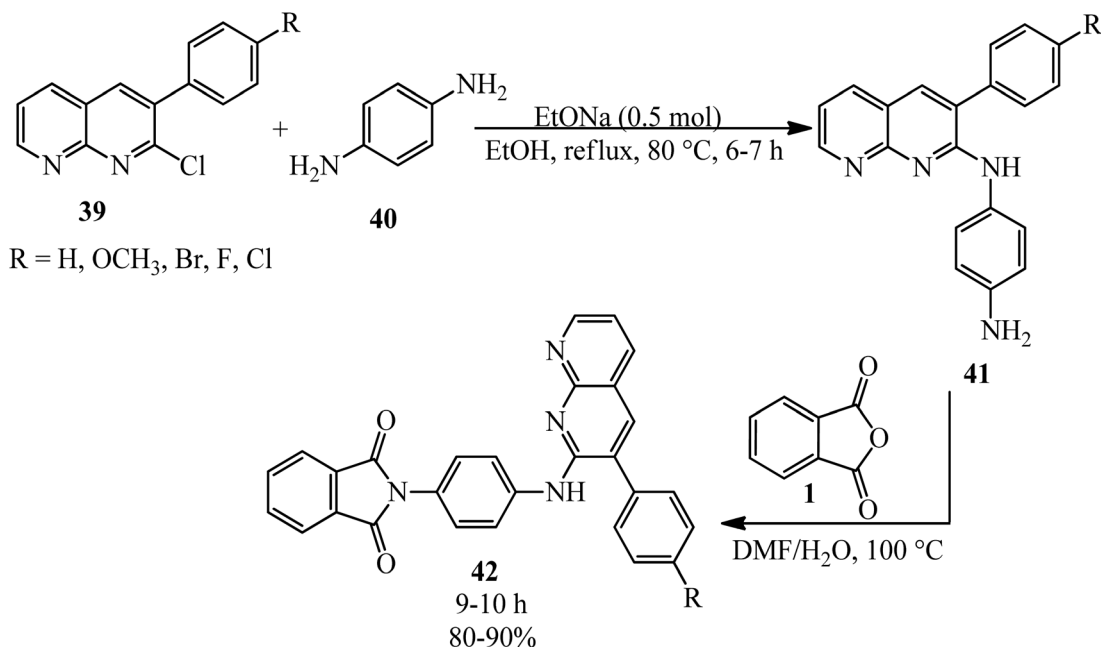
Scheme 16 Synthesis of 4-(1,3-2,3-dihydro-1H-2-isoidoly)butanoic acid.



Scheme 17 Synthesis of 2-(6-substituted benzo[d]thiazol-2-yl)isoidoline-1,3-diones.

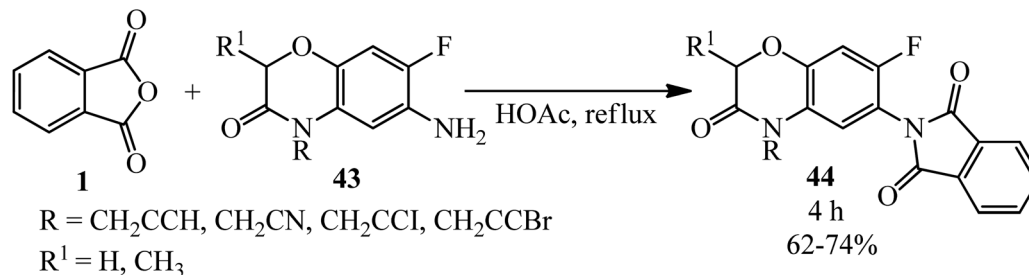
a zwitterionic intermediate. The reaction was found to be highly stereoselective, and the *trans*-isomers of (**E**) were obtained as the only products (Scheme 15).¹¹⁷

Chandrasekhar and coworkers in 2009 identified (**33**) as an *N*-alkyl imide of PA from a microwave-assisted (450 W) procedure of polymer bound γ -amino butyric acid (**36**) (1 mmol) and PA (1.25 mmol) in the presence of activated silica gel and TaCl_5 -

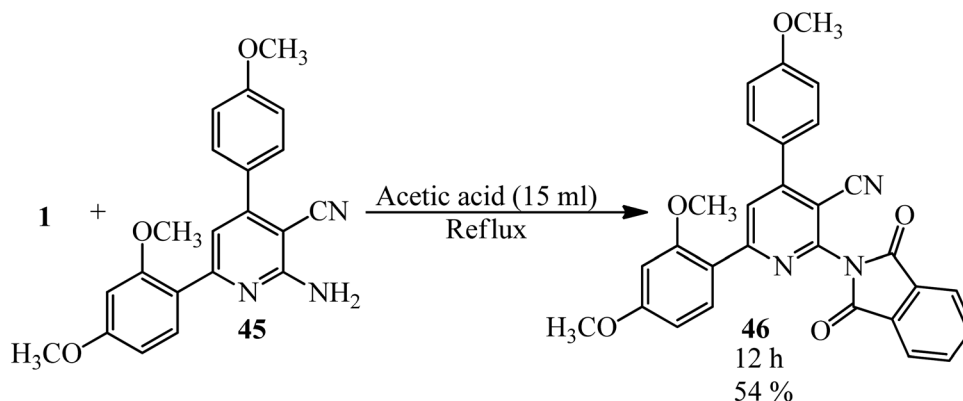


Scheme 18 2-(4-((3-Aryl-1,8-naphthyridin-2-yl)amino)phenyl)isoidoline-1,3-diones synthesis.





Scheme 19 Synthesis of 2-(7-fluoro-3-oxo-3,4-dihydro-2H-benzo[b][1,4]oxazin-6-yl)isoindoline-1,3-diones.



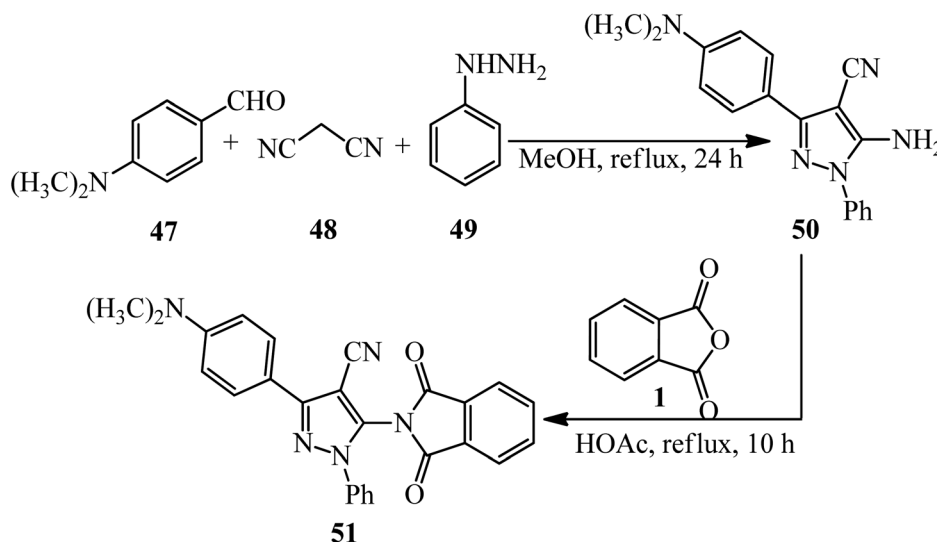
Scheme 20 Synthesis of 6-(2,4-dimethoxyphenyl)-2-(1,3-dioxoisoindolin-2-yl)-4-(4-methoxyphenyl)nicotinonitrile.

SiO₂ (Scheme 16).¹¹⁸ The organic product was isolated from the resin-silica mixture using TFA.

Pawar's group in 2014 synthesized 2-(6-substituted benzo[d]thiazol-2-yl)isoindoline-1,3-diones (**38**) from the heat-assisted reaction of PA and 6-substituted 2-amino benzothiazoles (**37**) within 5 min (Scheme 17).¹¹⁹ The 2-amino benzothiazoles (**37**) were prepared as using a previously reported method.¹²⁰ The antiinflammatory activity of the synthesized compounds was

determined using carrageenan-induced rat paw edema method (a mechanistic model for *in vivo* COX-2 inhibition). Compounds containing the Cl and COCH₃ substituents showed antiinflammatory activity comparable to the reference drug diclofenac at 100 mg kg⁻¹ doses.

2-(4-((3-Aryl-1,8-naphthyridin-2-yl)amino)phenyl)isoindoline-1,3-diones (**42**) was obtained by the treatment of substituted 3-aryl-1,8-naphthyridines (**41**) with PA in refluxing *N,N*-



Scheme 21 Synthesis of 3-(4-(dimethylamino)phenyl)-5-(1,3-dioxoisoindolin-2-yl)-1-phenyl-1H-pyrazole-4-carbonitrile.



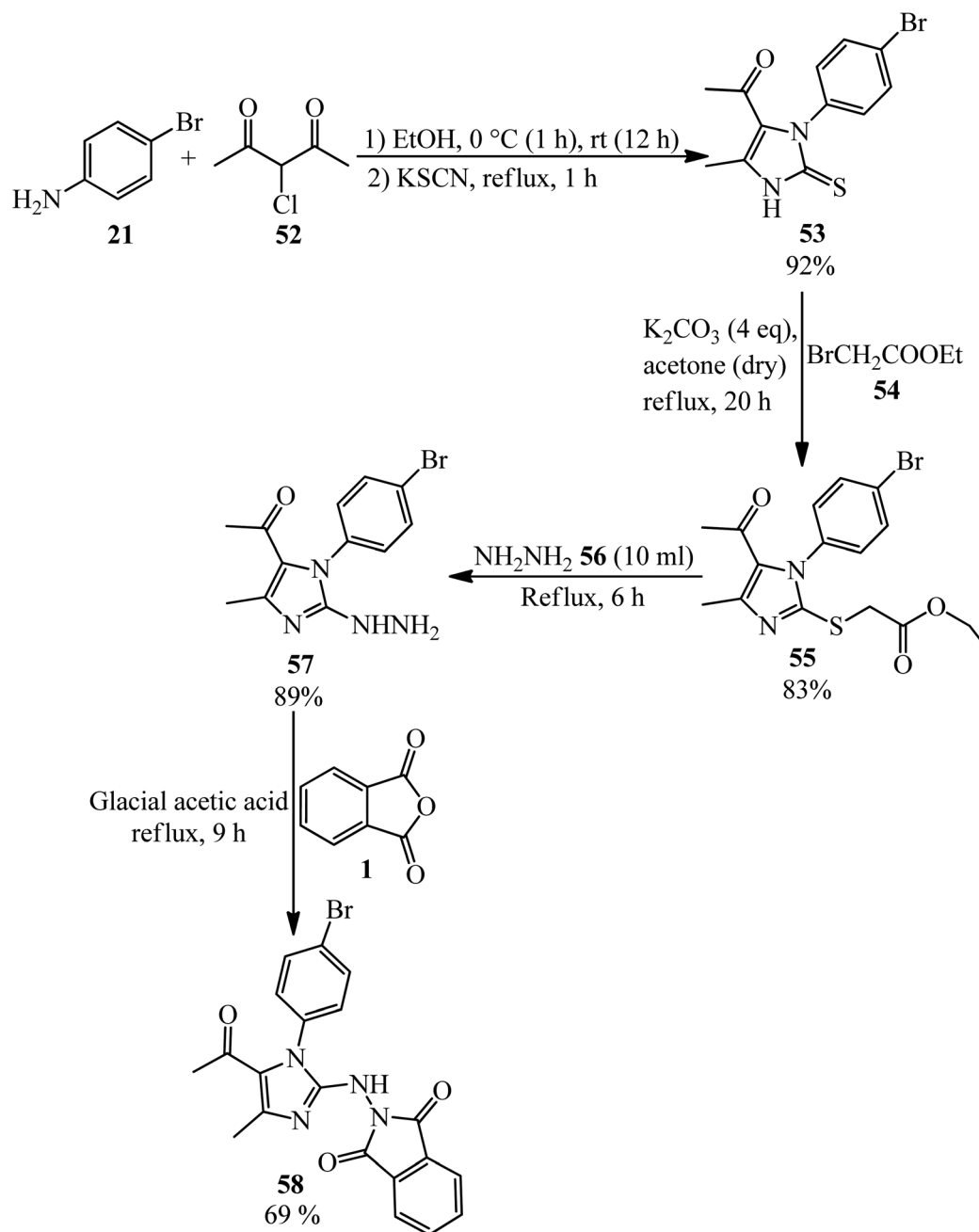
dimethylformamide containing 5% (v/v) water by Sakram group in 2018 (Scheme 18).¹²¹ Some of the products demonstrated moderate to good antimicrobial activity compared with the streptomycin reference.

The target compounds 2-(7-fluoro-3-oxo-3,4-dihydro-2H-benzo[*b*][1,4]oxazin-6-yl)isoindoline-1,3-diones (**44**) were constructed by Huang's group in 2005 *via* the reaction of PA (**1**) with 6-amino-7-fluoro-4-substituted-2H-3(4H)-benzo[*b*][1,4]oxazinones (**43**) in refluxing acetic acid within 4 h (Scheme 19).¹²² The products demonstrated protoporphyrinogen oxidase (protoph) inhibitory properties. The preliminary bioassay data displayed that some of them possessed promising herbicidal activities

comparable to that of the lead compound *B2055* (flumioxazin and its iodo analogue).

Anwer *et al.* in 2019 manufactured nicotinitriles attached to 1,3-dioxisoindoline ring (**46**) *via* the reaction of PA (**1**) and 2-amino-6-(2,4-dimethoxyphenyl)-4-(4-methoxyphenyl) nicotinitrile (**45**) in the presence of acetic acid that acted as the catalyst and solvent (Scheme 20).¹²³

Shaban *et al.* in 2020 also synthesized a new derivative of (**46**) named 2-(1,3-dioxisoindolin-2-yl)-4-(4-methoxyphenyl)-6-phenylnicotinitrile *via* the reaction of PA and 2-amino-4-(4-methoxyphenyl)-6-phenylnicotinitrile in acetic acid under



Scheme 22 Synthesis of 2-((5-acetyl-1-(4-bromophenyl)-4-methyl-1H-midazole-2-yl) amino)isoindoline-1,3-dione.



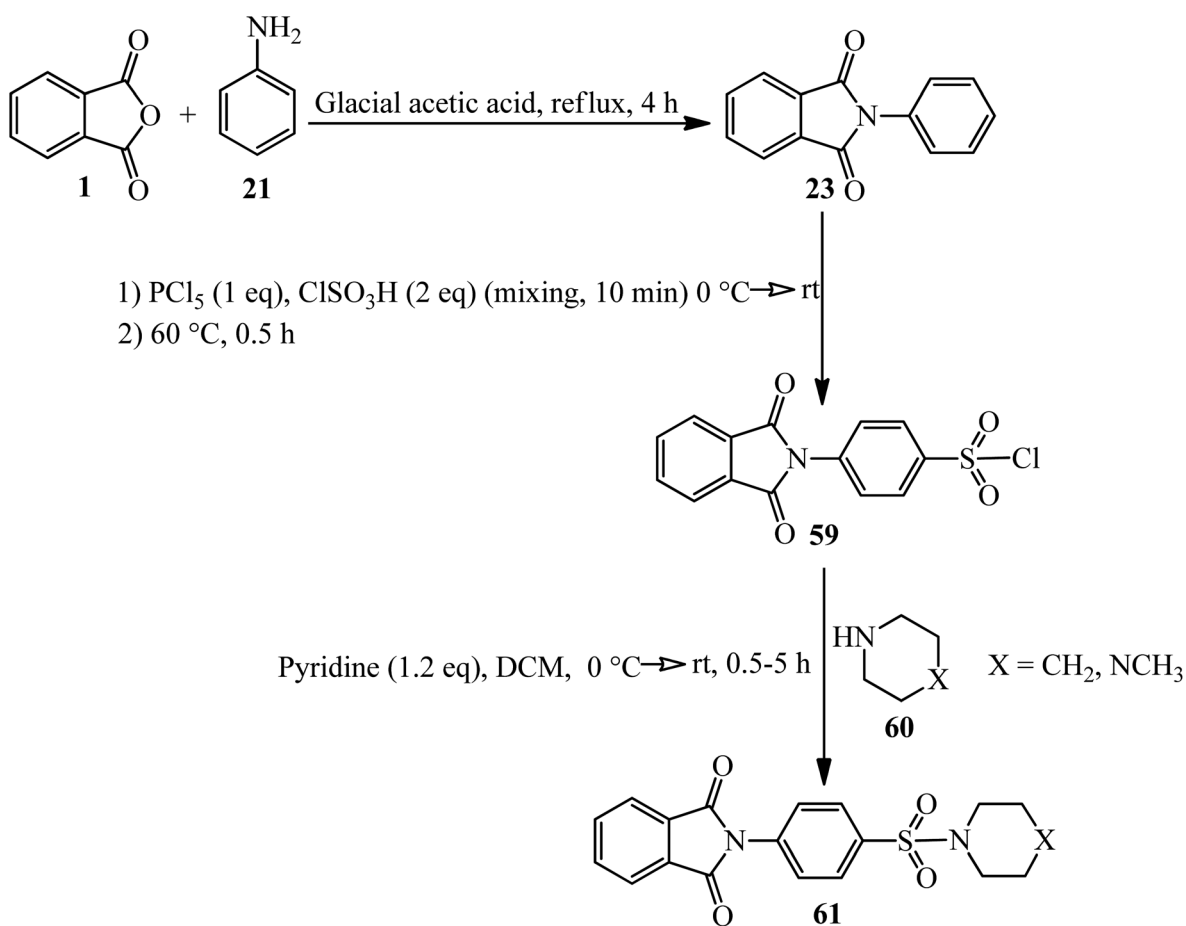
reflux conditions (22 h, 88%) and also microwave irradiation (3 min, 90.4%).¹²⁴

Anwer and Sayed in 2020 reported the reaction of PA (**1**) and 5-amino-3-(4-(dimethylamino)phenyl)-1-phenyl-1*H*-pyrazole-4-carbonitrile (**50**) to obtain 3-(4-(dimethylamino)phenyl)-5-(1,3-dioxoisindolin-2-yl)-1-phenyl-1*H*-pyrazole-4-carbonitrile (**51**). The adduct demonstrated high activities against Gram-negative and Gram-positive bacteria and high cytotoxic activity against *MCF7* and *HCT-116* (Scheme 21).¹²⁵ The pyrazole-containing substrate (**50**) was achieved *via* the one-pot coupling reaction of *N,N*-dimethylamino benzaldehyde (**47**), malononitrile (**48**), and phenyl hydrazine (**49**).

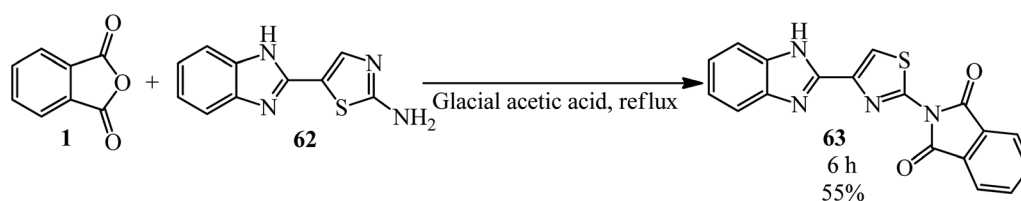
El-Shahat and Hasanin group in 2021 fabricated novel imidazole isoindoline-1,3-dione (**58**) from the reaction of PA

with 1-(1-(4-bromophenyl)-2-hydrazinyl-4-methyl-1*H*-imidazol-5-yl)ethenone (**57**) (Scheme 22).¹²⁶ According to Scheme 21, 1-(3-(4-bromophenyl)-5-methyl-2-thioxo-2,3-dihydro-1*H*-imidazol-4-yl)ethan-1-one (**53**) was obtained through the dropwise addition of 3-chloro-2,4-pentanedione (**52**) to a solution of 4-bromoaniline (**21**) in ethanol, followed by the addition of equimolar amount of KSCN. Subsequent alkylation of (**53**) with ethyl bromoacetate (**54**) led to the corresponding ethyl ester (**55**), which fused with hydrazine hydrate (**56**) to yield (**57**).

The Jayaprakash group in 2015 constructed some novel 4-(1,3-dioxo-2,3-dihydro-1*H*-isoindol-2-yl)benzene-1-sulphonamides (**61**). As shown in Scheme 22, in the first step, the 2-phenyl-1,3-isoindolinedione (**23**) was obtained by the reaction of PA (4 mmol) and aniline (3.4 mmol) in glacial acetic under nitrogen

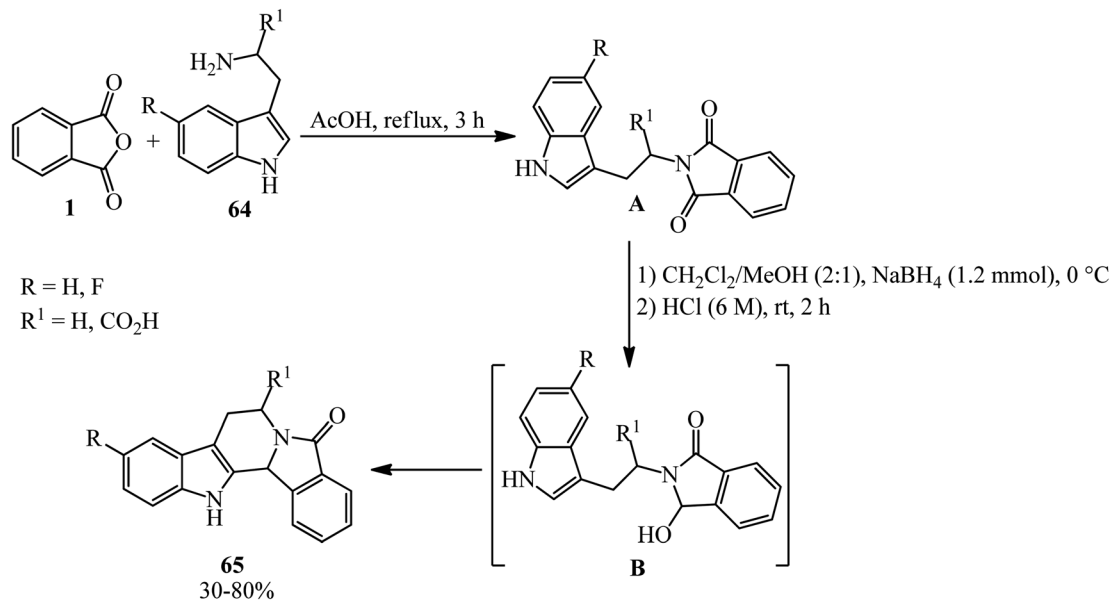


Scheme 23 Synthesis of 4-(1,3-dioxo-2,3-dihydro-1*H*-isoindol-2-yl)benzene-1-sulphonamides.



Scheme 24 2-(4-(1*H*-Benzo[*d*]imidazol-2-yl)thiazol-2-yl)isoindoline-1,3-dione synthesis.

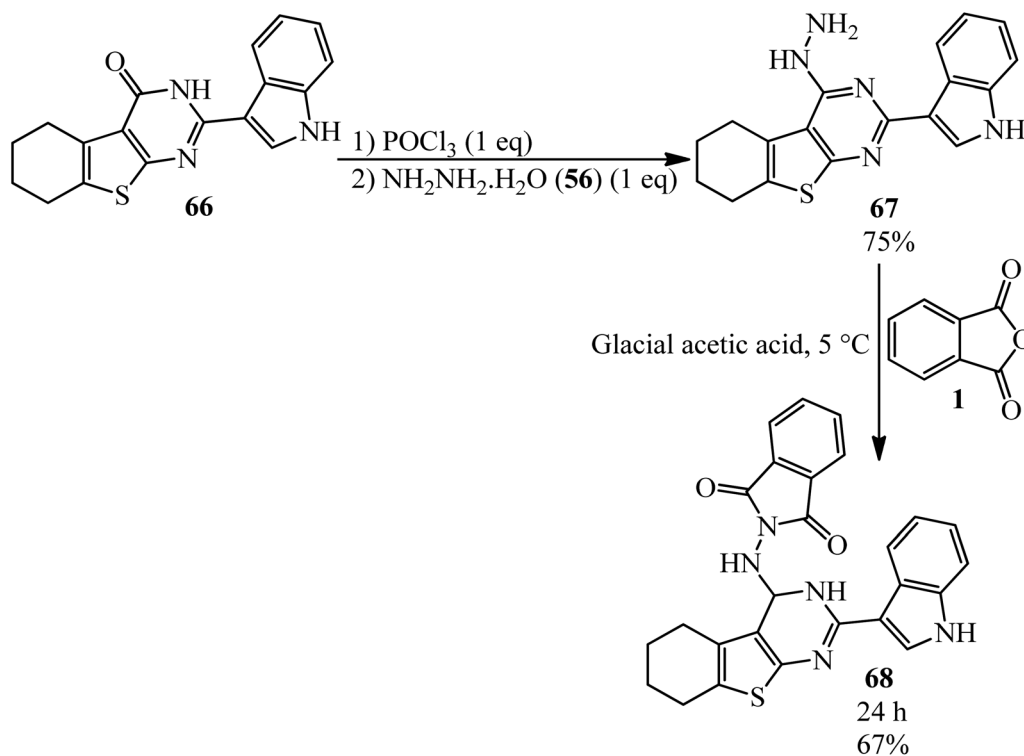


Scheme 25 Synthesis of tetrahydro- β -carbolines.

atmosphere at 120 °C within 2 h. In the next step, the product reacted with a mixture of chlorosulfonic acid (2 eq.) and phosphorus pentachloride (1 eq.) at 60 °C for 30 min. These two steps were performed based on previously-reported procedures.¹²⁷ The final products were obtained *via* the reaction of 4-(1,3-dioxoisindolin-2-yl)benzene-1-sulfonyl chloride (59) and piperidine/1-methylpiperazine (60) in the presence of pyridine

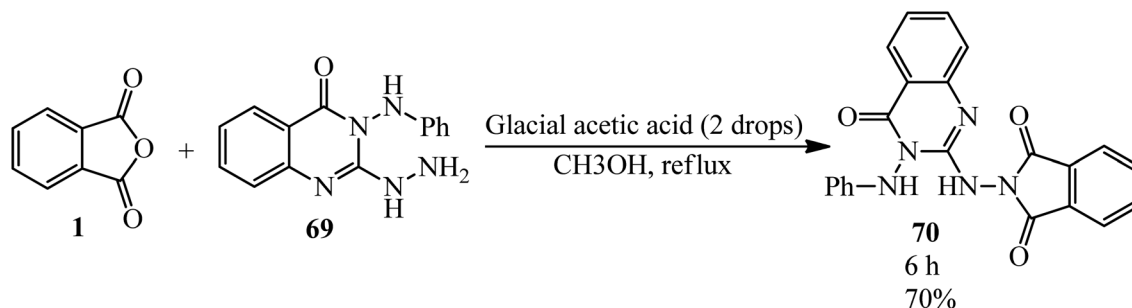
with 68–95% yield (Scheme 23).¹²⁸ The authors carried out molecular modelling studies on the adducts as Dengue virus 2 (*DENV2*) protease inhibitors.

Nofal group in 2014 reacted the starting compound [4-(1*H*-benzo[*d*]imidazol-2-yl)thiazol-2-amine] (62) with PA in refluxing glacial acetic acid for 6 h to obtain 2-(4-(1*H*-benzo[*d*]imidazol-2-

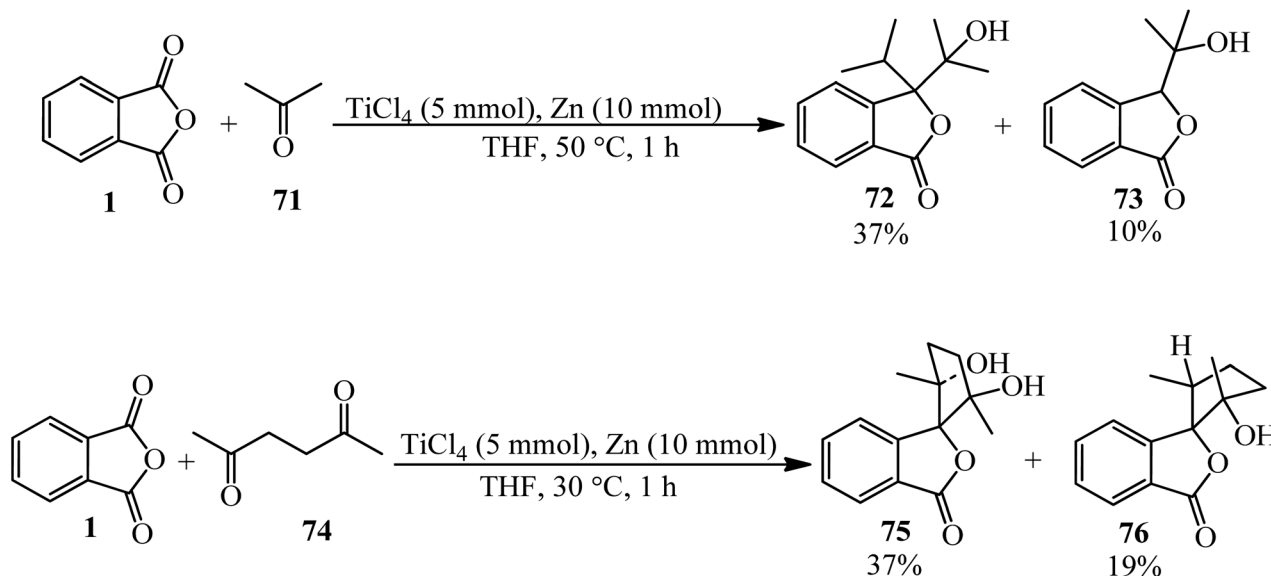


Scheme 26 A novel isoindoline-1,3-dione synthetic procedure.





Scheme 27 2-(4-Oxo-3-phenylamino-3,4-dihydroquinazolin-2-ylamino)-isoindole-1,3-dione synthesis.



Scheme 28 Reductive coupling of PA and ketones.

yl)thiazol-2-yl)isoindoline-1,3-dione (**63**) with 55% yield (Scheme 24).¹²⁹

Tetrahydro- β -carbolines (**65**), with strictosamide skeleton, were obtained by Liu in 2014 *via* intermolecular condensation, selective reduction, and intramolecular cyclization starting from PA (**1**) and tryptamines (**64**) in a 1 : 1.2 molar ratio (Scheme 25).¹³⁰ The reaction proceeds *via* the intermediate (**B**), which has not been isolated.

Abd El-All's research group in 2016 constructed 2-((2-(1*H*-indol-3-yl)-3,4,5,6,7,8-hexahydrobenzo[4,5]thieno[2,3-*d*]pyrimidin-4-yl)amino)isoindoline-1,3-dione (**68**) *via* the reaction of pyrimidinohydrazine (**67**) and PA. The target compound was tested for its activity against Influenza A Neuraminidase virus (*H3N2*), which was very potent (Scheme 26).¹³¹

In 2003, Saleh's group constructed 2-(4-oxo-3-phenylamino-3,4-dihydroquinazolin-2-ylamino)-isoindole-1,3-dione (**70**) by the reaction of PA with 2-hydrazinyl-3-(phenylamino)quinazolin-4(3*H*)-one (**69**) in refluxing methanol (Scheme 27).¹³²

Kise *et al.* in 2020 presented the reductive coupling of PA (**1**) with acetone (**71**) in a 1 : 10 molar ratio by Zn-TiCl₄ (in 2 : 1 molar ratio in THF) which, gave two-to-one (**72**) as the major and one-to-one (**73**) as the minor coupled products. The coupled

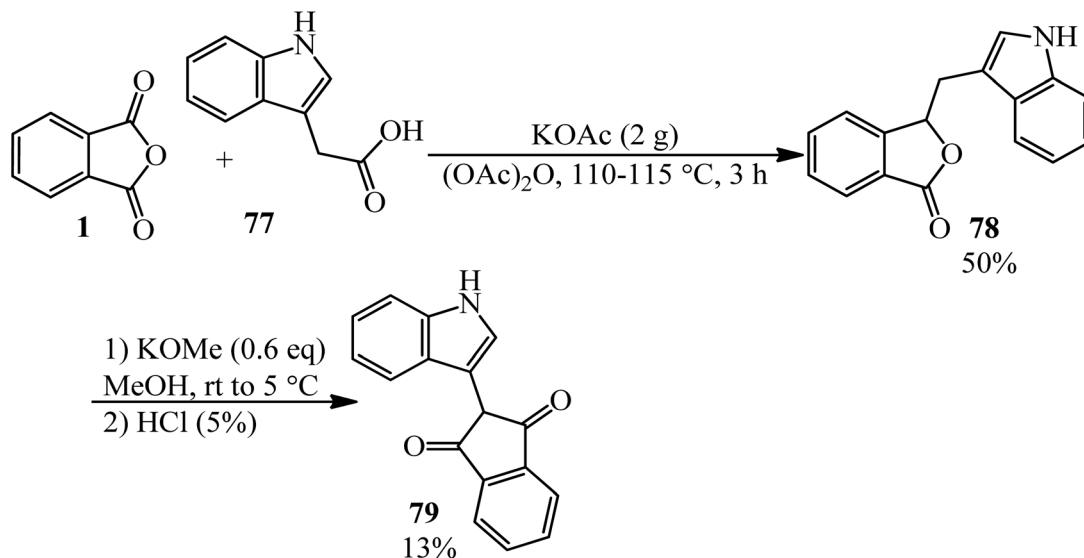
products were transformed to 3,3-diisopropyl-, 3-isopropylidene-, and 3-isopropylphthalides. In addition, the reductive coupling of PA (**1**) with acetonylacetone (**74**) in 1 : 5 molar ratio by Zn-TiCl₄ in THF gave 3-spirocyclopentanylpthalides as the selective *exo-cis* isomers (**75**, **76**) (Scheme 28).¹³³ Zn-TiCl₄ was utilized as the source of low-valent titanium.

Lácová in 1969 reacted indole-3-acetic acid (**77**) with PA, in a 0.56 : 1 molar ratio, under Perkin synthesis conditions (in the presence of potassium acetate and acetic anhydride) to form 3-(3-indolylmethine)phthalide (**78**), as yellow crystals, which underwent rearrangement in methanol in the presence of potassium methoxide to yield 2-(3-indolyl)indane-1,3-dione (**79**) (Scheme 29).¹³⁴

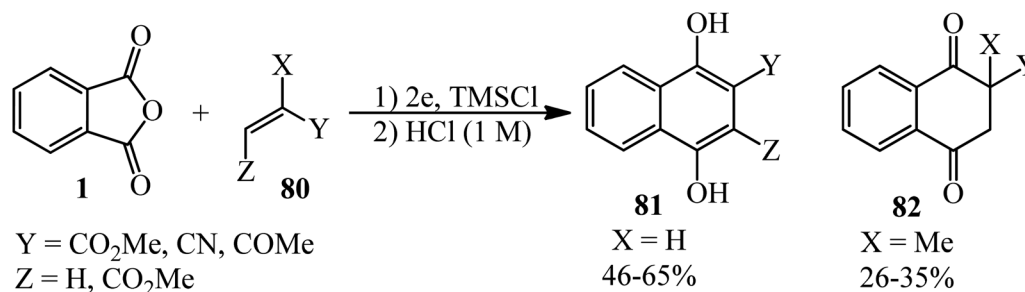
Kise and coworkers in 2020 demonstrated the electro-reduction coupling of PA (**1**) with α,β -unsaturated carbonyl compounds (**80**) in a 1 : 3 molar ratio in the presence of TMSCl (5 mmol) and subsequent treatment with HCl (1 M), which gave 1,4-dihydroxynaphthalenes (**81**) and 2-methyl 2,3-dihydro-1,4-diones (**82**) (Scheme 30).¹³⁵

An intermolecular nickel-catalyzed decarbonylative [4 + 2] cycloaddition between PA (**1**) with 1,3-dienes (**83**) in a 0.5 : 3





Scheme 29 Perkin synthesis of 2-(3-indolyl)indane-1,3-dione.



Scheme 30 Synthesis of dihydroxynaphthalenes and 2,3-dihydronaphthalene-1,4-diones.

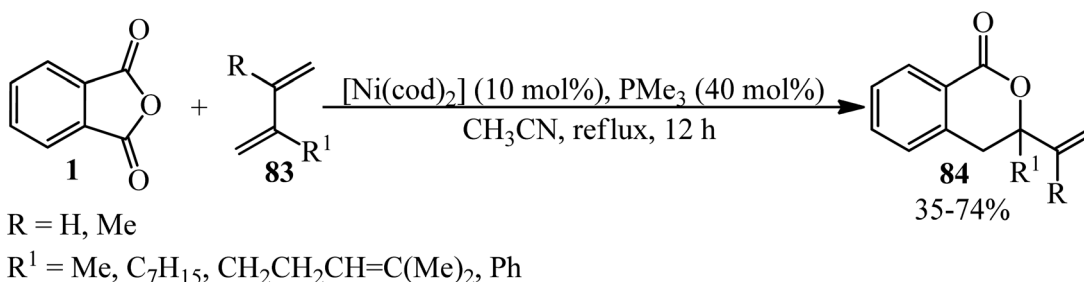
molar ratio in the presence of $[\text{Ni}(\text{cod})_2]$ as a Ni(0) precursor to afford substituted 3-vinyldihydroisocoumarins (**84**) was studied by the Kurahashi group in 2011 (Scheme 31).¹³⁶

According to the proposed mechanism, the oxidative addition of (**1**) to Ni(0)-bearing electron-rich phosphine ligands gave the nickelacycle (**A**) (Scheme 32). 7,8 subsequent decarbonylation provides oxanickelacycle (**B**). The insertion of (**83**) through the more electron-rich C=C bond to C-Ni bond leads to the more stable acyclic³-allylnickel intermediate (**D**), which undergoes nucleophilic addition of oxygen to π -allylnickel at

the more substituted carbon to give (**84**) and regenerates the starting Ni(0) complex.^{136,137}

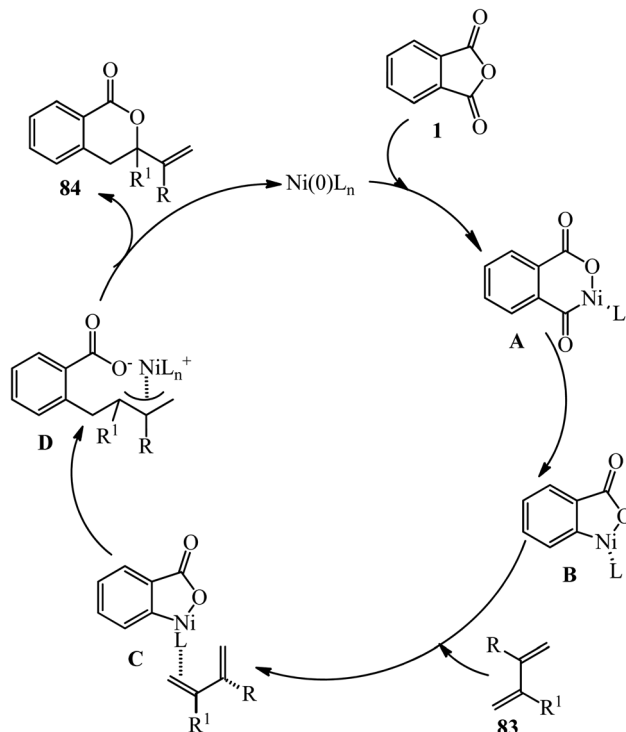
In 2022, the palladium-catalyzed decarbonylative/decarboxylative [4 + 2] annulation of PA (**1**) with cyclic diaryliodonium salt (**85**) to synthesize triphenylene (**86**) developed by Li and coworkers (Scheme 33).¹³⁸

Satchel and Stacey in 1971 investigated some products [2-(4,8-dihydroxy-1-naphthoyl)benzoic acid and its methylated form] (**88**, **89**) from a Friedel-Crafts reaction of PA (**1**) with 1,5-dihydroxynaphthalene (**87**). The reaction of 1,5-



Scheme 31 Substituted 3-vinyldihydroisocoumarins preparation.





Scheme 32 Plausible mechanism for the synthesis of substituted 3-vinylidihydroisocoumarins.

dimethoxynaphthalene (**90**) with PA and in the presence of excess aluminum chloride gave *o*-(8-hydroxy-4-methoxy-1-naphthoyl)benzoic acid (**91**) and *o*-(4,8-dihydroxy-1-naphthoyl) benzoic acid (**92**), while with an excess of phthalic anhydride, the main products are dimethoxy-acid and 4,8-bis-(2-carboxybenzoyl)-1,5-dimethoxynaphthalene (Scheme 34).¹³⁹

Naeimi and Shokrollah Brojerdi in 2014 gained anthraquinones (**94**) *via* the reaction of PA (**1**) and wide variety of substituted aromatic compounds such as (phenols, naphthols, alkyl benzenes, halo benzenes, and biphenyl) (**93**) using silica sulfuric acid (SSA) as a green, recyclable, and heterogeneous catalyst (Scheme 35).¹⁴⁰ The catalyst could be reused seven times without significant activity loss.

Nie and Liu research group in 2022 provided 2-ethyl-anthraquinone (2-EAQ) (**94**) using PA (**1**) and ethylbenzene (**93**) as a feedstock by the combination of acylation and dehydration

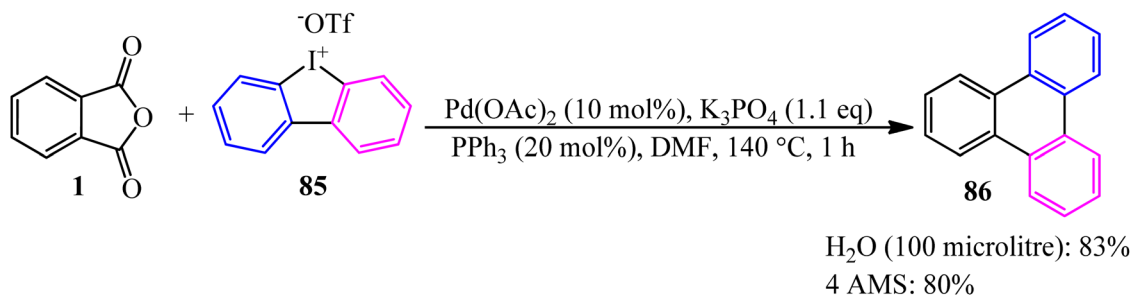
over a Sc-modified H β catalyst. Sc modification was used to create new strong Lewis acid and increased acid amount of H β .¹⁴¹

The catalytic performance of amide-AlCl₃ (DMA-2AlCl₃) ionic liquid analogs in synthesizing *o*-benzoylbenzoic acid (BBA) (**96**) from PA and benzene (**95**) was demonstrated. The catalyst was obtained by mixing AlCl₃ and DMA in 2 : 1 molar ratio at 100 °C for 3 h. Then, PA and benzene, in a 1 : 10 molar ratio were mixed at 40 °C within 5 h to obtain the product by 98.2% (Scheme 36).¹⁴² The authors proposed that Al₂Cl₇⁻, gained from DMA-2AlCl₃, attacks the anhydride and formed (**96**) *via* an electrophilic substitution reaction.

Naphtho[1,2-*b*]thiophene (**97**) reacted with PA (**1**) to prepare ketocarboxylic acid (**98**). Naphtho[2,1-*b*]thiophene (**99**) also reacted with PA to obtain 2-naphtho[2,1-*b*]thienyl *o*-carboxyphenyl ketone (**100**) with 70% yield. The Friedel-Crafts reaction between PA and thiophene (**101**) yielded ketoacid (**102**) (Scheme 37).¹⁴³

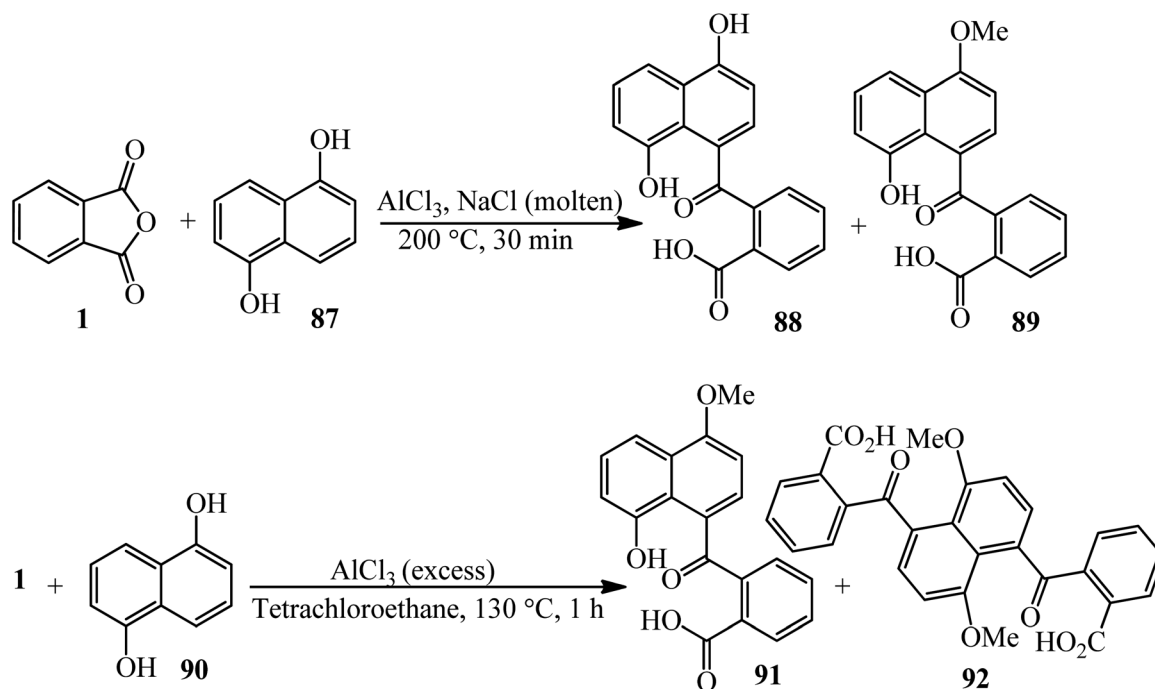
Naeimi and Namdari in 2009 described the direct preparation of anthraquinones (**94**) *via* the reaction of PA (**1**) and various benzene derivatives (**93**) in the presence of anhydrous AlCl₃ (0.11 mmol)/methanesulfonic acid (0.01 mmol) (LAMA) at 95–100 °C within 15–65 min with 6–93% yield. Benzenes that contain electron withdrawing substituents (such as nitrobenzene and 1,3-dichlorobenzene) had the lowest reactivity in this reaction (8% and 6% yields, respectively). The compound 1,3-dinitrobenzene did not get the corresponding product even performing the reaction overnight.¹⁴⁴

Shafiq group in 2021 reported newly synthesized anthraquinone-based pyrimidine derivatives. In the first step, the authors prepared the anthraquinone derivatives (**94**) *via* the modified procedure of Madje *et al.*¹⁴⁵ in which the PA and benzene derivatives in a 1:1.1 molar ratio were mixed in aqueous media at room temperature in the presence of alum (KAl(SO₄)₂·12H₂O), which was followed by the addition of concentrated HCl, washing with EtOAc, drying, and recrystallization from methanol. In the next step, the pyrimidine derivatives (**105**) (obtained by a procedure of Shafiq *et al.*¹⁴⁶) were obtained *via* the three-component reaction of benzaldehydes (**19**), urea (**103**), and ethyl acetoacetate (**104**). In the third step, different derivatives of the precursors (**94**) and (**105**) were mixed in the presence of catalytic amount of copper chloride and cupric oxide in methanol for 20–30 minutes at ambient temperature to obtain the anthraquinone-based pyrimidines

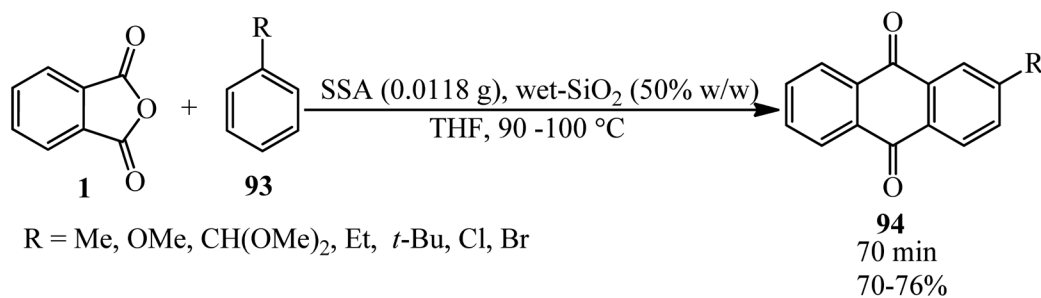


Scheme 33 Synthesis of triphenylene.

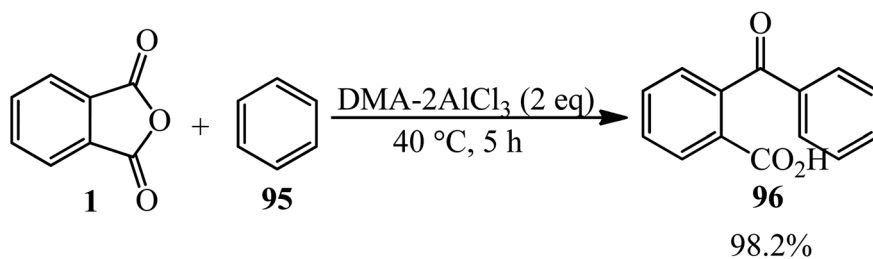




Scheme 34 The Friedel–Crafts reaction of 1,5-dihydroxynaphthalene and 1,5-dimethoxynaphthalene with PA.



Scheme 35 Synthesis of anthraquinones.



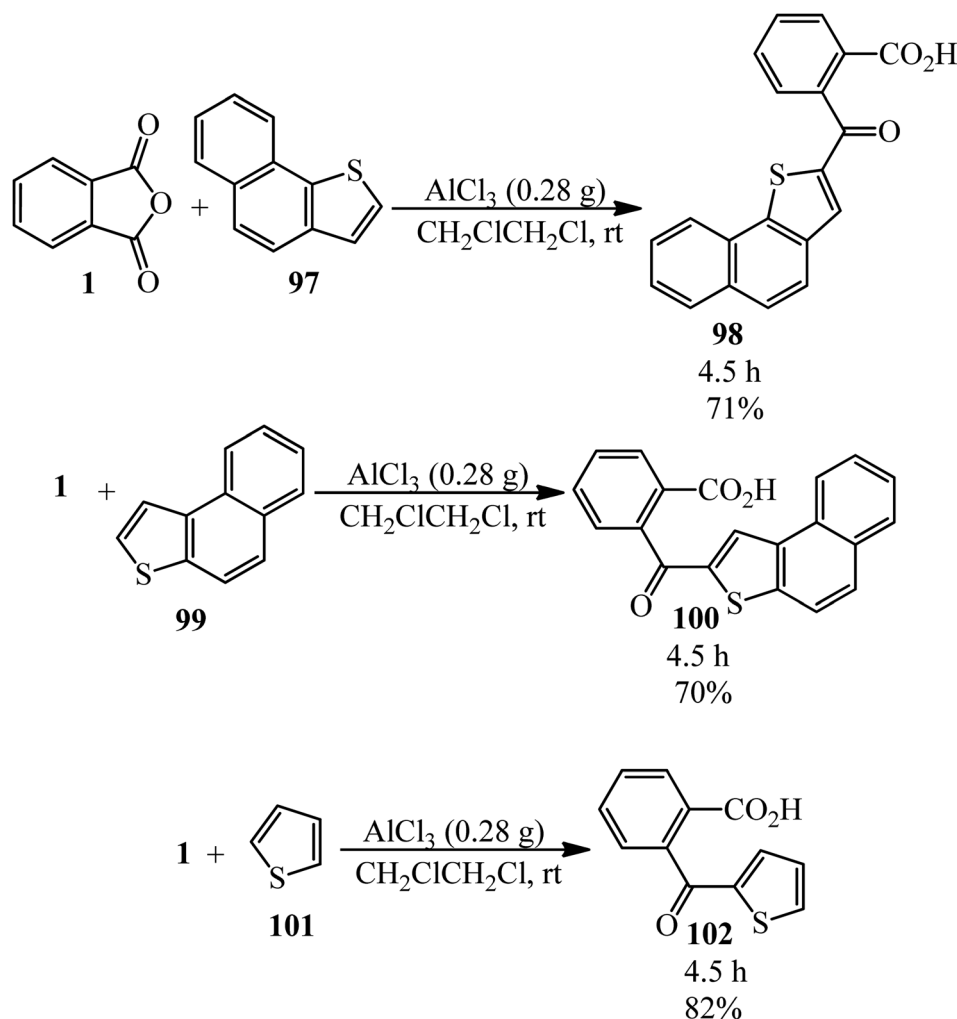
Scheme 36 Synthesis of *o*-benzoylbenzoic acid (BBA).

adducts (**106–109**) (Scheme 38).¹⁴⁷ The antioxidant, antidiabetic, molecular docking, and QSAR studies of the products were also examined.

Phillips in 1927 reported the synthesis of alizarin through a multistep reaction. First, the condensation of PA, *o*-dichlorobenzene, and anhydrous aluminum chloride into 3',4'-dichloro-2-benzoylbenzoic acid, and second, the conversion of this acid

by means of sulfuric acid into 2,3-dichloro-anthraquinone. Upon fusion with alkali, the dichloro-anthraquinone (**94**) converted into alizarin.¹⁴⁸ The authors confirmed the claim of Sprent and Dodd¹⁴⁹ that 2-chloro-anthraquinone is obtained by the condensation of PA and *o*-dichlorobenzene could not be confirmed.



Scheme 37 Reactions of naphtho[1,2-*b*]thiophene, naphtho[2,1-*b*]thiophene and thiophene with PA.

Kumar *et al.* in 2014 reported novel 2-amino-6-(1,4-dioxo-3,4-dihydrophthalazin)-2(1*H*)-yl-4-phenyl-4*H*-pyran-3,5-dicarbonitriles (**113**) through a two-step reaction. First, 3-(1,4-dioxo-3,4-dihydrophthalazin-(1*H*)-yl)-3-oxopropanenitrile (**111**) was obtained from reaction of PA (**1**) with ethyl cyanohydrazide (**110**) in the presence of acetic acid. The adduct reacted with benzaldehydes (**19**) and active methylene compounds such as malononitrile (**48**) and ethyl cyanoacetate (**112**) using *L*-proline as the catalyst in EtOH at ambient temperature to afford (**113**) (Scheme 39).¹⁵⁰

Abou-Elmagd group in 2012 achieved *N*-(1,3-dioxoisindolin-2-yl)-4-oxo-3,4-dihydrophthalazine-1-carboxamide (**115**) via the reaction of PA and 4-oxo-3,4-dihydrophthalazine-1-carbohydrazide (**114**) in the presence of glacial acetic acid in refluxing dioxane (Scheme 40).¹⁵¹

Simijonovic in 2020 described green aqua-mediated synthesis of benzamide-dioxoisindoline derivatives (**117**) via the reaction of PA and benzoyl hydrazides (**116**) at 80 °C under ultrasonic irradiation. All compounds were subjected to experimental determination of their antioxidative potential. The

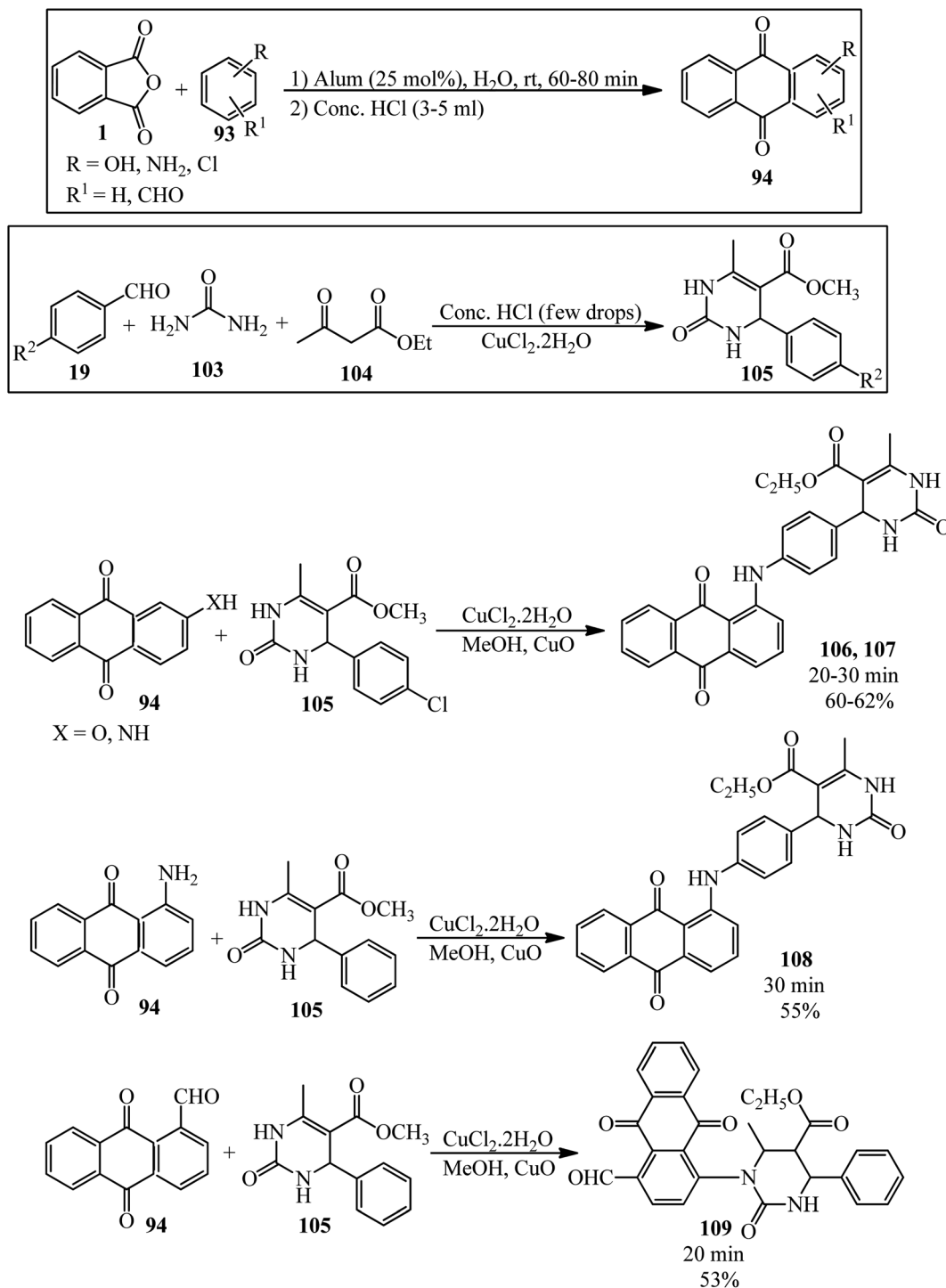
DPPH test revealed that newly synthesized phenolic compounds are the best antioxidants (Scheme 41).¹⁵²

Aly's group in 2022 interpreted that the coumarin hydrazide (**118**) reacted with PA (**1**) to furnish 2-(2-oxo-2*H*-chromene-3-carbonyl)-2,3-dihydrophthalazine-1,4-dione (**119**) by 76% yield (Scheme 42).¹⁵³

In 2019, 2-(6-iodo-4-oxo-2-undecylquinazolin-3(4*H*)-yl)aceto-hydrazide (**120**) reacted with PA (**1**) to produce 2-undecyl-4(3*H*)-quinazolinone (**121**), which was obtained by either conventional method or by the microwave-assisted technique. The adduct was tested *in vitro* against a panel of three human tumor cell lines, namely, Hepatocellular Carcinoma (liver) *HepG2*, colon cancer *HCT-116*, and mammary gland breast *MCF-7*, which demonstrated satisfactory activity (Scheme 43).¹⁵⁴

Zare's group in 2019 performed the reaction PA (**1**) and phenyl hydrazine (**49**) in (A): microdroplets (the reaction condition: N₂ atmosphere, DMF was demonstrated to be the best solvent for this microdroplet reaction, with a yield of product (**122**) up to 98% when the distance between the spray tip to the glass vial extended to 7 cm with reaction time of about 350 μs), and (B): bulk-phase without catalyst and without





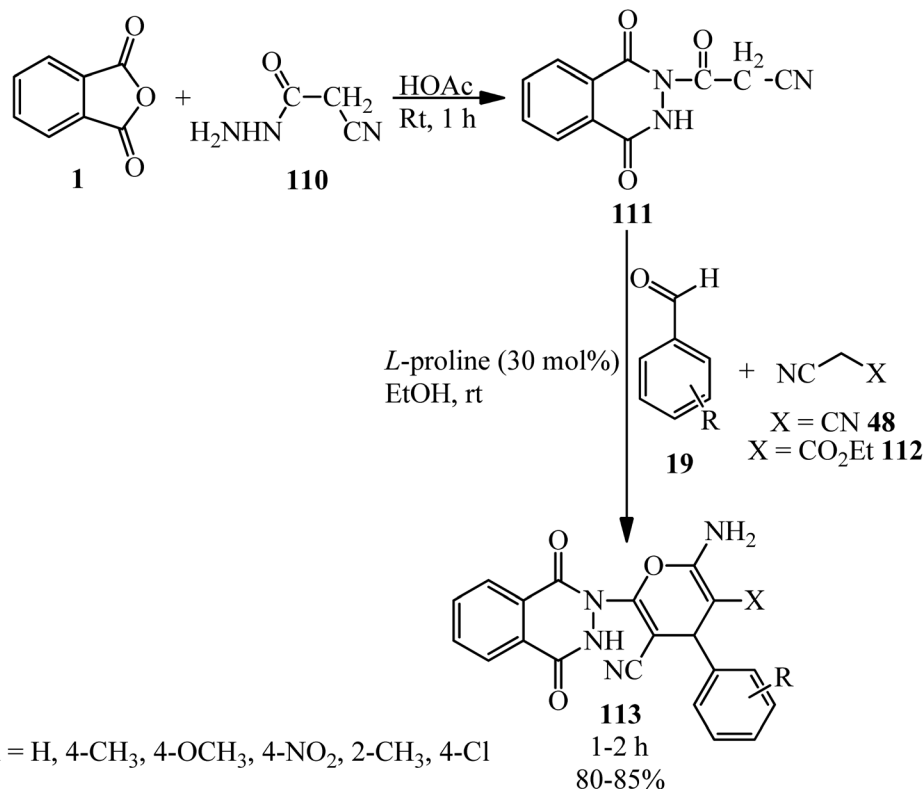
Scheme 38 Synthesis of anthraquinone-based pyrimidines.

refluxing (the concentration of the both substrates is 10 mM). The microdroplet reaction led to significant formation of (122) with a small amount of product (123), whereas the bulk-phase reaction of PA and (49) in acetic acid catalyst under high-temperature refluxing led to product (122), but without catalyst at room temperature only (123) was formed. Moreover, this reaction in microdroplets showed excellent selectivity, yielding only the important six-membered heterocyclic product (122).

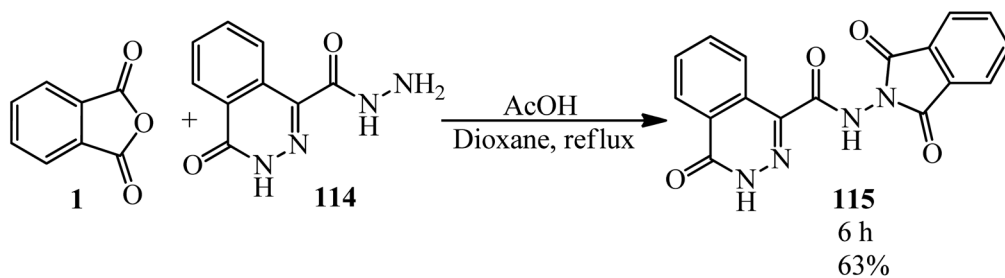
This behavior indicated an extremely different reaction pathway in the confined microdroplets. The authors claimed that the reaction rate and selectivity enhancement was attributed to a surface reaction in the confined microdroplets having low pH values, which is enhanced by the positive charging of the microdroplets (Scheme 44).¹⁵⁵

The Abbas group in 2021 achieved new Schiff bases and also their 1,3-oxazepines, derived from PA *via* a multistep reaction.

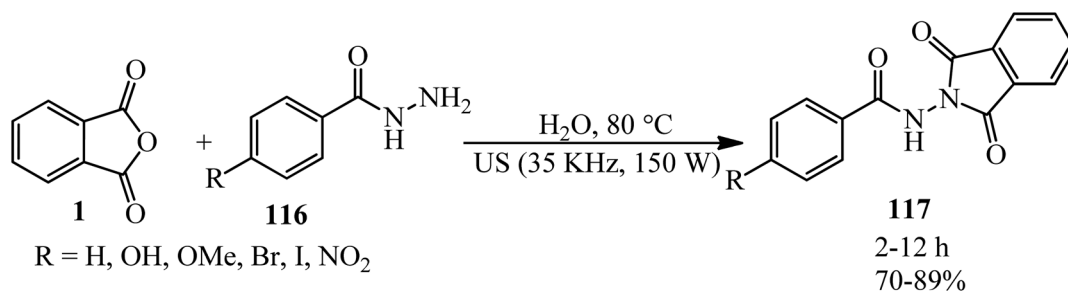




Scheme 39 Synthesis of 2-amino-6-(1,4-dioxo-3,4-dihydrophthalazin)-2(1H)-yl-4-phenyl-4H-pyran-3,5-dicarbonitriles.

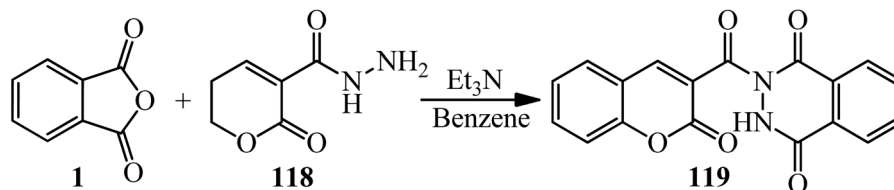
Scheme 40 *N*-(1,3-Dioxoisindolin-2-yl)-4-oxo-3,4-dihydrophthalazine-1-carboxamide preparation.

In the first step, 2-[*N*-(4-methyl phenyl)]phthalamic acid (**124**) was prepared through the reaction of PA (**1**) and 4-methyl aniline (**21**) in a 1 : 1 molar ratio. In the next step, the methylation of (**124**) gave 2-[*N*-(4-methyl phenyl)]phthalamide acetate (**125**), which reacted with hydrazine (**56**) to get 2-[*N*-(4-methyl phenyl)]phthalamic acid hydrazide (**126**). The Schiff bases (**127**) was

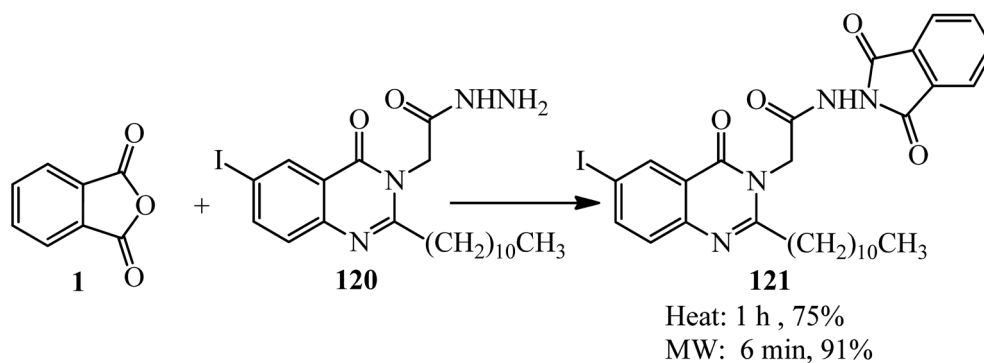


Scheme 41 Synthesis of benzamide-dioxoisindolines.

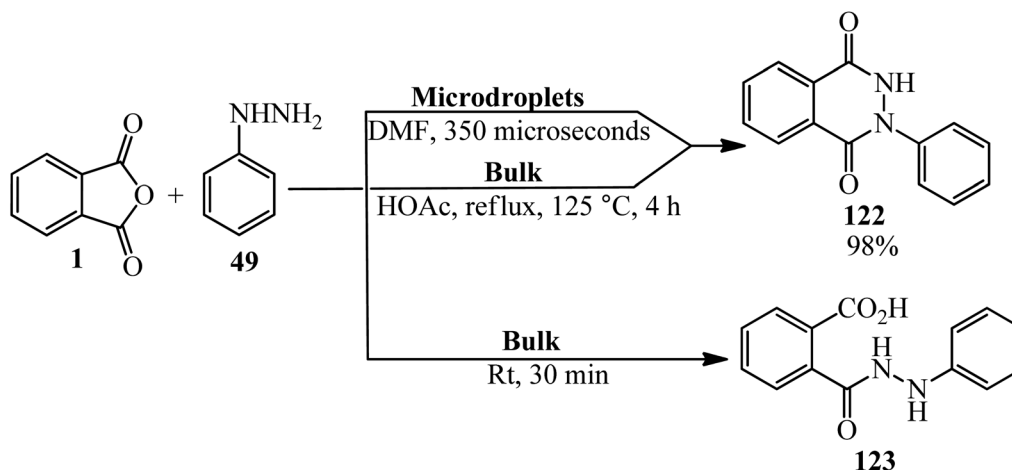




Scheme 42 Synthesis of 2-(2-oxo-2H-chromene-3-carbonyl)-2,3-dihydrophthalazine-1,4-dione.



Scheme 43 Production of 2-undecyl-4(3H)-quinazolinone.



Scheme 44 Investigating the reaction of PA and phenyl hydrazine.

obtained in refluxing ethanol in the presence of glacial acetic acid as catalyst. Finally, the 1,3-oxazepine derivatives of the Schiff bases (**128**) gained from the consequent reaction of (**127**) with PA in refluxing benzene (Scheme 45).¹⁵⁶

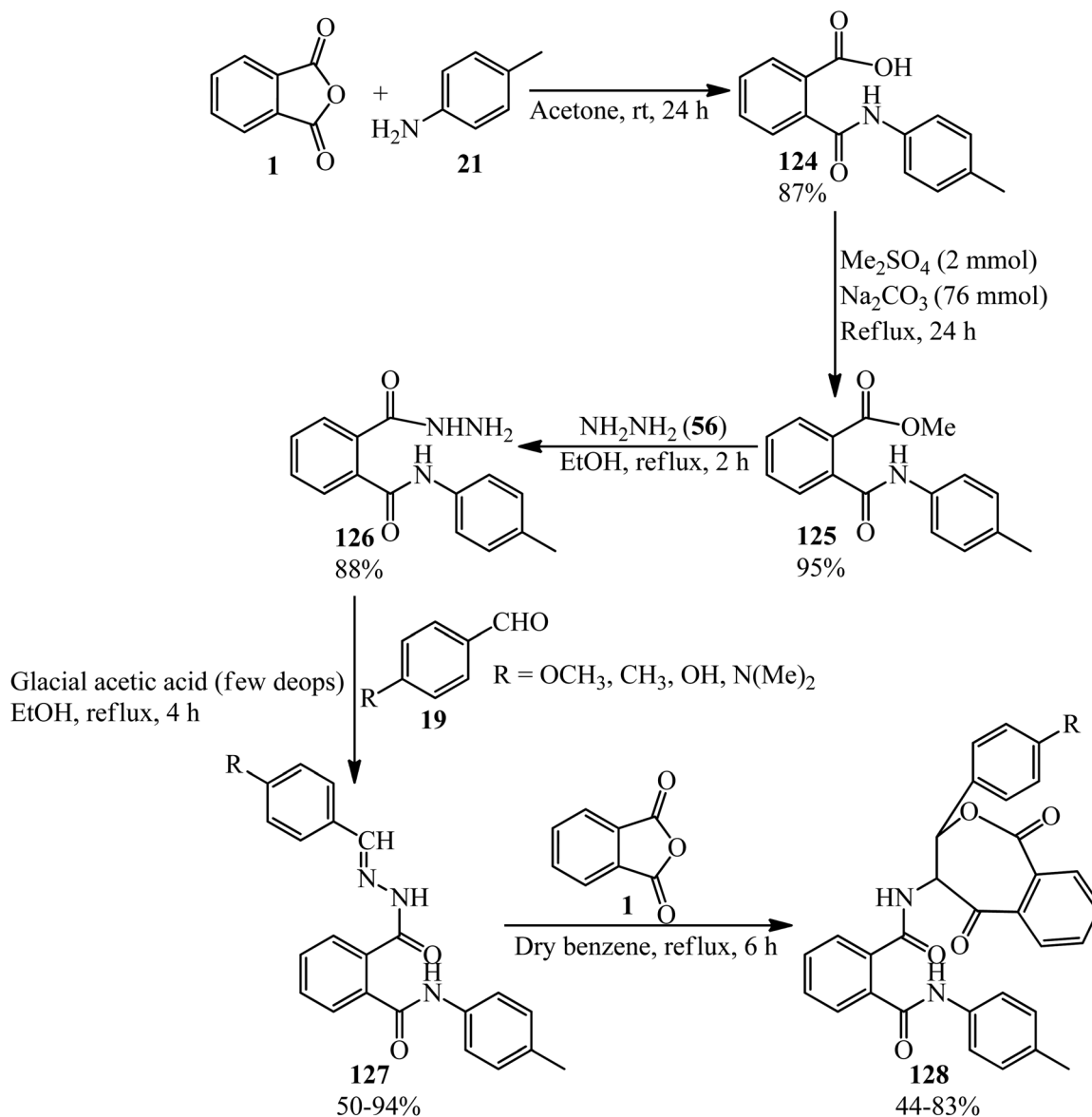
Hanoon in 2011 reported new 6-(2,3-dimethyl-5-oxo-4-phenyl-2,5-dihydro-1H-pyrazol-1-yl)-7-aryl-6,7-dihydrooxepine-2,5-diones (**131**) *via* the reaction of PA (**1**) and Schiff bases (**130**) in dry benzene. The Schiff bases was obtained *via* the condensation of various aromatic aldehydes (**19**) with 4-aminophenazone (**129**) in the presence of glacial acetic acid as the catalyst (Scheme 46).¹⁵⁷

In 2014, Serevičius group constructed a series of nonsymmetric 9,10-diphenylanthracenes (DPA) through a multistep reaction, which started from the acylation reaction of PA (**1**) and

substituted benzenes (**93**) (including pentyl and phenyl benzenes) to obtain substituted benzoylbenzoic acid (**96**) that underwent an intramolecular cyclization in heated acidic media (such as poly phosphoric acid or oleum 5%), which led to the formation of 2-substituted anthraquinones (**94**), which reacted with arylmagnesium bromide (**132**), resulting in 9,10-diaryl-9,10-dihydroxydihydroanthracene (**133**), which was reduced to form 2,9,10-trisubstituted anthracenes (**134**) (Scheme 47).¹⁵⁸ The authors investigations affirmed that DPA compounds are deep-blue emitters with enhanced charge transport properties.

The synthesis of 4-(4'-hydroxyaryl)(2H)phthalazin-1-ones (**137**) was done by Cheng's group in 2007. The procedure consists of a two-step reaction, starting from the Friedel-Crafts acylation reaction of six phenols (**135**) with PA (**1**) to obtain α -





Scheme 45 Preparation of Schiff bases and their 1,3-oxazepines derived from PA.

keto acids (136), followed by cyclization with excess amount of hydrazine hydrate (56) in good to excellent yields with high regioselectivity to get products (137), which are methyl-substituted phthalazinone-containing bisphenol-like monomers; the adducts were utilized as monomers to obtain a number of novel heterocyclic poly(arylene ether ketone)s through the reaction with an activated difluoro monomer based on a novel N–C coupling reaction. The obtained polymers demonstrated polymers with high T_g 's, good solubility, and excellent thermal stability (Scheme 48).¹⁵⁹

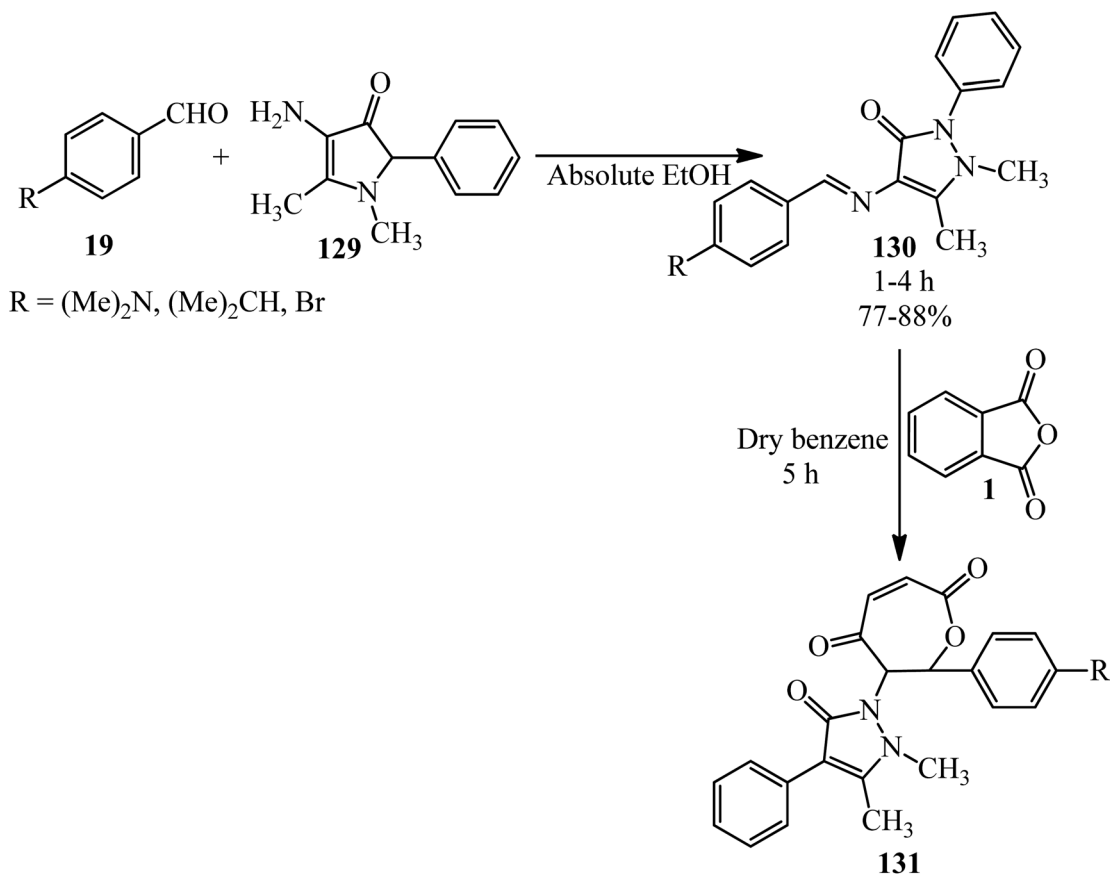
N-Sugar-substituted phthalimides (139) constructed in 2004 by Li's group *via* the reaction of PA (1) sugar azide (138) in a 15 : 1 molar ratio in the presence of tetrabutyl ammonium iodide (0.1 mmol, as catalyst) under essentially neutral conditions [which was performed by NaI (3 mmol) and Me_3SiCl (1.5 mmol)]

in dry acetonitrile at ambient temperature or 60 °C within 1–3 h with 78–92% yield (Scheme 49).¹⁶⁰

Kamal in 1994 also reported *N*-substituted phthalimides (139) from the reaction of the corresponding azides (138) and PA (1) in a 1 : 1 molar ratio, employing chlorotrimethylsilane and sodium iodide (*in situ* generation of iodotrimethylsilane) in acetonitrile at ambient temperature within 15 min with 95% yield.¹⁶¹

Nguyen and coworkers in 2012 performed the bisaddition of pyridinyl lithium (140) to PA, which yielded the monoaddition product 3-hydroxyisobenzofuranone (143) and 3,3-bis(6-methylpyridin-2-yl)isobenzofuran-1(3*H*)-one (144) as the byproduct. According to the mechanism, presumably, bisaddition arises from the ring opening of the initially formed alkoxyisobenzofuranone (141), leading to diaryl ketone (142), which





Scheme 46 Synthesis of new 6-(2,3-dimethyl-5-oxo-4-phenyl-2,5-dihydro-1*H*-pyrazol-1-yl)-7-aryl-6,7-dihydrooxepine-2,5-diones.

reacted with the organometallic reagent to give (**144**) (Scheme 50).¹⁶²

Acosta in 1995 synthesized 2-thenoylbenzoic acid (**146**) via the reaction of the PA (**1**) and 2-thienyllithium (**145**) by 80% yield. The product (**146**) cyclized in the presence of purified Mont K-10 (under microwave irradiation) to obtain thieno[2,3-*b*]-1,4-naphthoquinone (**147**) (Scheme 51).¹⁶³

Parham and Piccirilli in 1976 claimed that when equimolar amounts of PA (**1**) and phenyllithium (**148**) were added at -78°C , the yield of phthalide (**149**) was 78% based on phenyllithium. When the same ratios were maintained but the order of addition reversed, the yield of isolated (**149**) was 9%, while the yield of *o*-benzoylbenzoic acid (**96**) was 35%. Furthermore, the yield of *o*-benzoylbenzoic acid was further increased (55%) when excess (2 eq.) of PA was employed. In subsequent experiments, the aryllithium reagent was added rapidly to PA (2 eq.) in tetrahydrofuran at -100°C , in which good yields of substituted benzoylbenzoic acids (**96**) were obtained (Scheme 52).¹⁶⁴

Tanaka's group in 1994 identified the reaction of PA (**1**) and furan-3-yllithium (**150**) to obtain 2-(3-furanoyl)benzoic acid (**151**) though the inverse addition method. On the other hand, the lactonization of (**B**) gave 3,3-di-(3-furyl)-1,3-dihydroisobenzofuran-1-one (**152**). The reaction of (**151**) with LDA gave naphtho[2,3-*b*]furan-4,9-dione (**153**) as yellow needles (Scheme 53).¹⁶⁵ The authors attempted to obtain (**153**) via the

chlorination of (**151**) with thionyl chloride or PCl_5 , which was not successful.

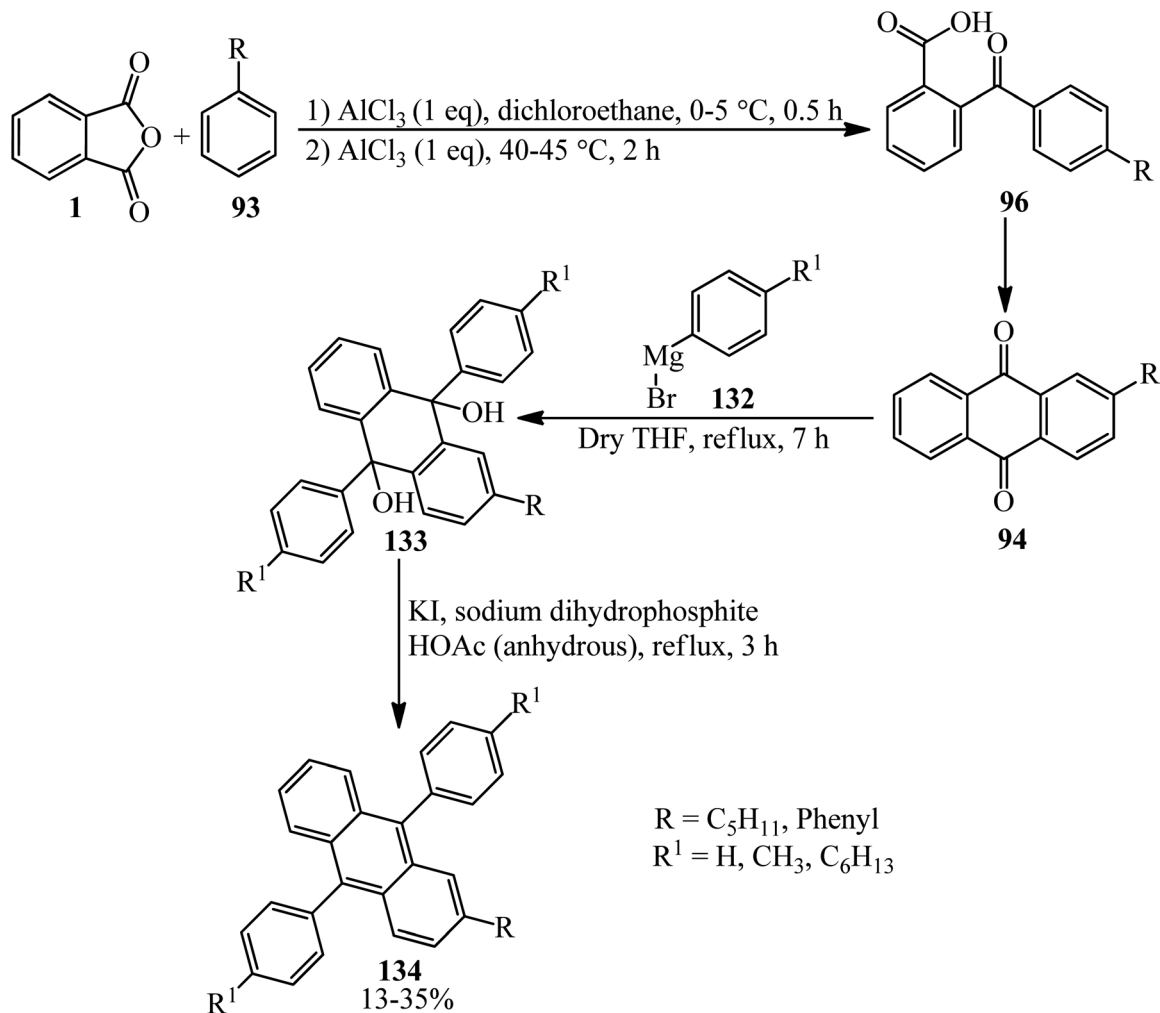
Martin and Seoane in 1992 claimed that 2-thenoylbenzoic acid (**146**), obtained from PA (**1**) and 2-thenoylmagnesium iodide (**154**), yielded thieno[2,3-*b*]-1,4-naphthoquinone (**147**) (Scheme 54).¹⁶⁶

Fieser and Hershbe in 1937 reported the reaction of PA and α -tetralylmagnesium bromide (**155**) yielded the 2-(α -tetraloyl)-benzoic acid (**156**) as colorless prisms, which was reduced to 2-((5,6,7,8-tetrahydronaphthalen-1-yl)methyl)benzoic acid (**157**) by high pressure hydrogenation (Scheme 55).¹⁶⁷

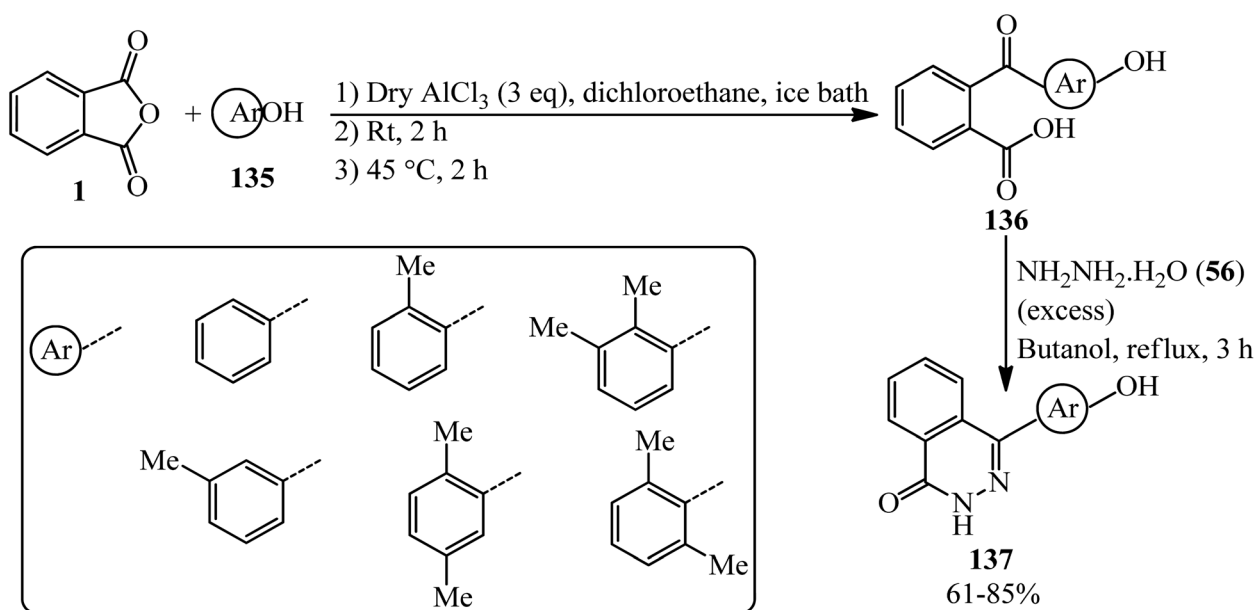
Baba in 1968 mixed PA (**1**) with triethylaluminum (**158**), which surprisingly got the reduced isobenzofuran-1(3*H*)-one (**159**) instead of the common ketoacid corresponding to the Grignard reaction (Scheme 56).¹⁶⁸

Benneville in 1941 demonstrated the reaction of PA (**1**) and organocadmium compounds (**160**) (including dialkyl- and diaryl-cadmium) to obtain keto acids (**96**). The method consists of more satisfactory yields than the ones achieved from the reaction of PA with the Grignard reagent and is more generally applicable than the Friedel-Crafts synthesis. At first, the organocadmium was obtained from mixing the Grignard reagent and anhydrous cadmium chloride by the method of Gilman and Nelson.¹⁶⁹ Then, a solution of PA in dry ether was added to organocadmium within fifteen to thirty minutes in an ice-bath, followed by refluxing (Scheme 57).¹⁷⁰



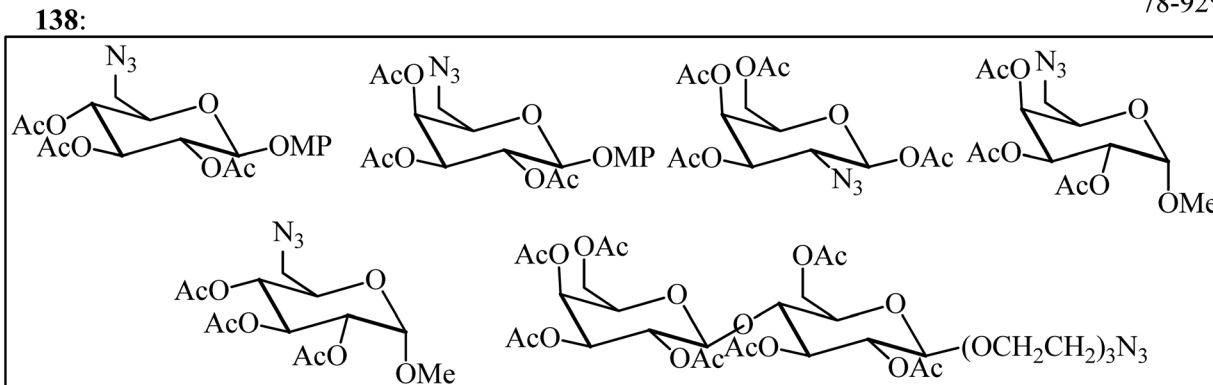
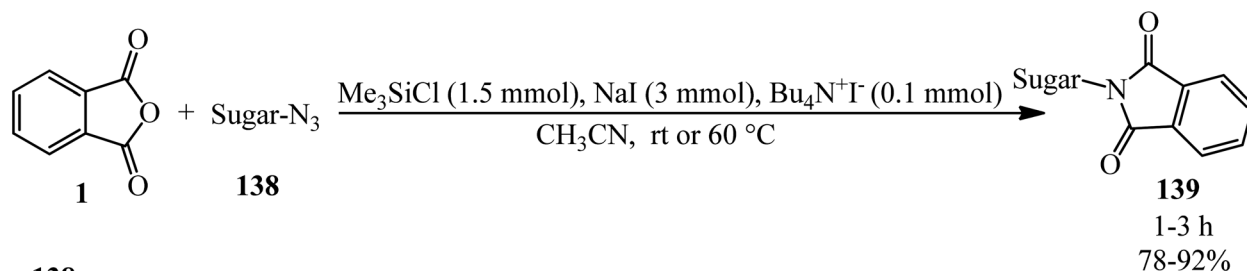


Scheme 47 Synthesis of nonsymmetric 9,10-diphenylanthracenes (DPA).

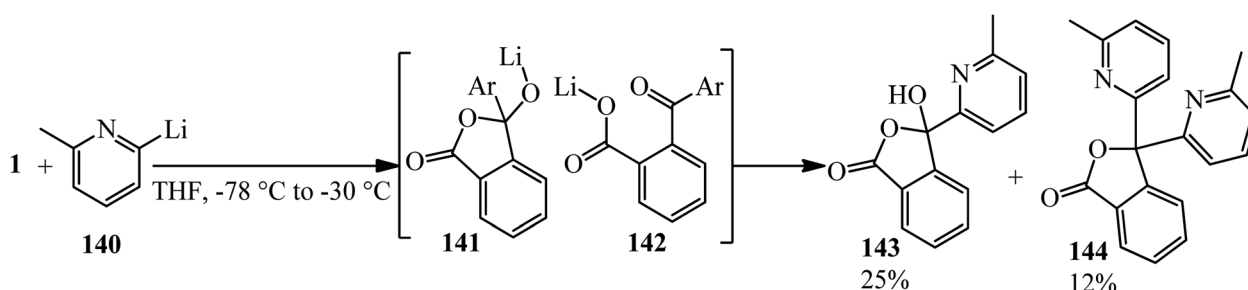


Scheme 48 Synthesis of 4-(4'-hydroxyaryl)(2H)phthalazin-1-ones.

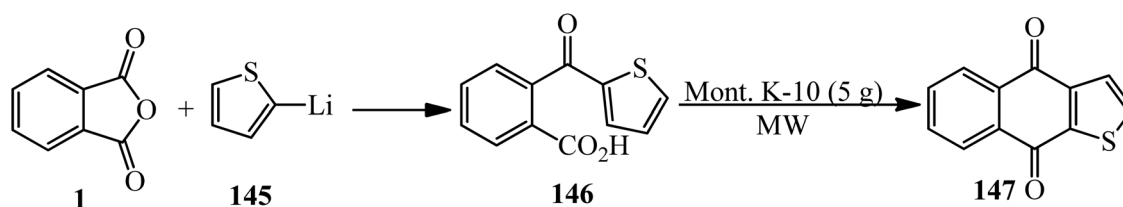




Scheme 49 N-Sugar-substituted phthalimides preparation.



Scheme 50 Addition of pyridinyl lithium to PA.

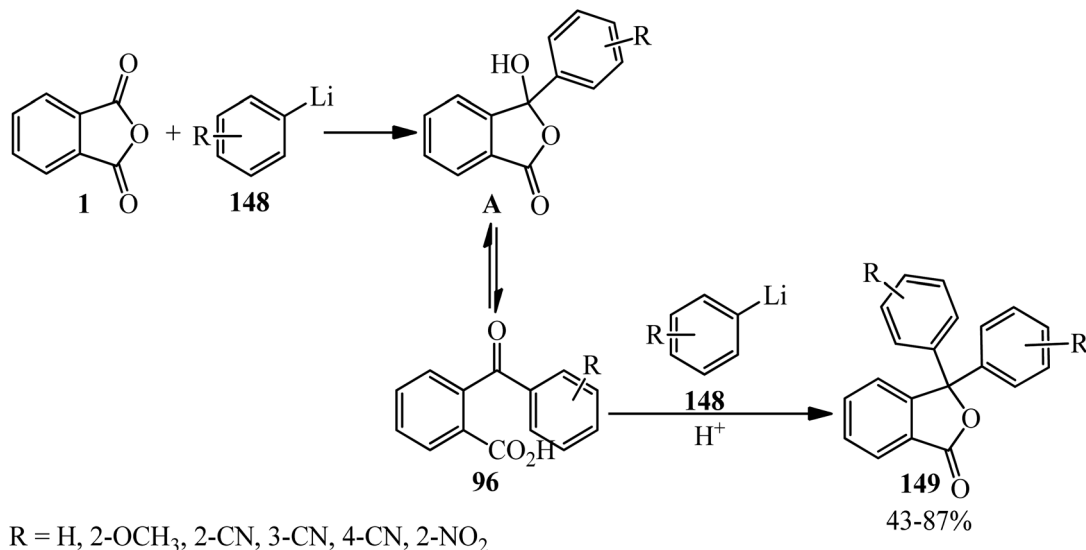
Scheme 51 Synthesis of thieno[2,3-*b*]-1,4-naphthoquinone.

In addition, McMullen in 1922 studied the Friedel–Crafts reaction of PA with benzene in the presence of AlCl_3 in detail.¹⁷¹

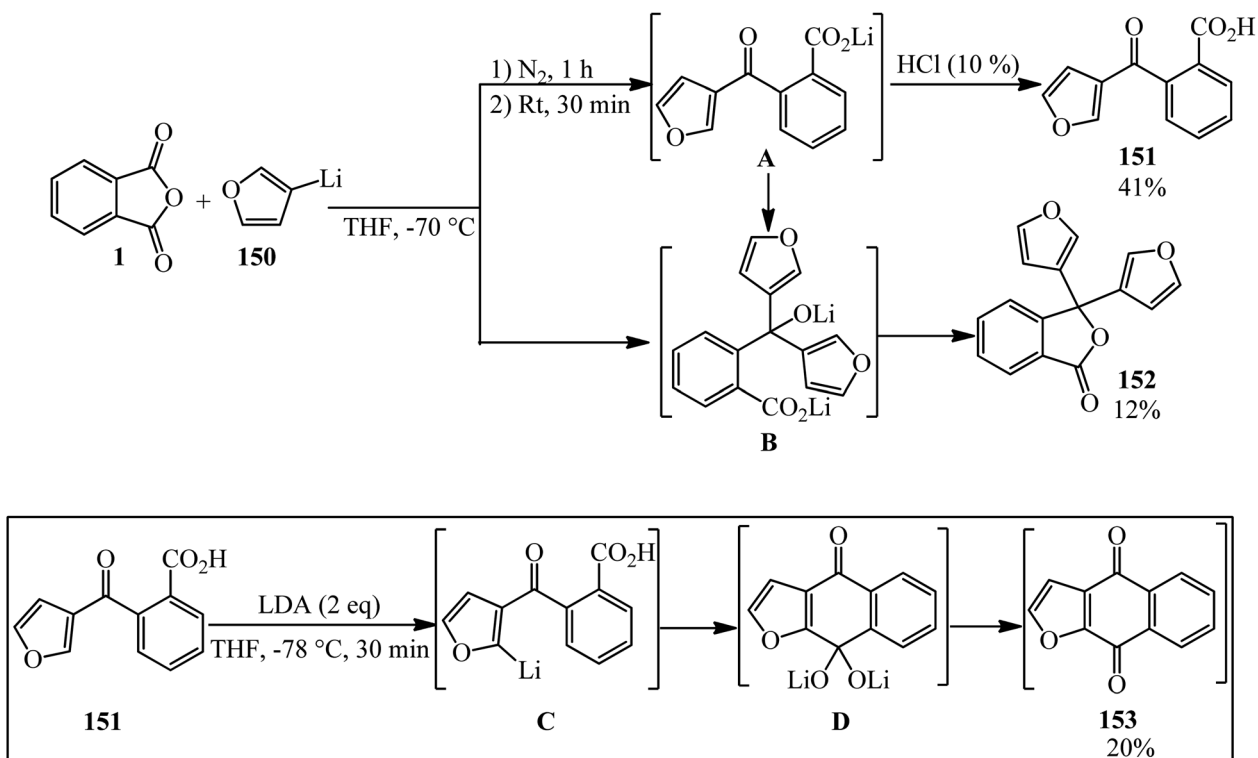
Rahman and Nahar in 1992 studied the reaction of arylcoppermagnesium reagents (prepared from ArMgX and CuI) with one equivalent of PA. The phenylcoppermagnesium reagent ($\text{Ar} = \text{C}_6\text{H}_5$) gave 2-benzoylbenzoic acid (**96**) and 3,3-diphenylphthalide (**149**) in 40% and 42% yield, respectively. The yield of 2-benzoylbenzoic acid (**96**) increased to 93% in the presence of dimethyl sulphide. Under these conditions, no phthalide was formed. On the other hand, lithium diphenylcuprate reacted

with PA in ether–hexane to give 2-benzoylbenzoic acid (**96**) and 3,3-diphenylphthalide (**149**) in 92% and 7% yields, respectively. The reaction of phenylcopper reagent (prepared from PhMgBr and CuI) and PA under similar conditions proceed slowly to get 2-benzoylbenzoic acid (**96**) in 15% yield. The reaction of phenylmagnesium bromide in the presence or absence of catalytic amounts of copper(i) iodide led to unsatisfactory results. The use of two equivalents of phenyllithium with one equivalent of PA, on the other hand, afforded 3,3-diphenylphthalide (**149**) in 77% yield. The authors believed that the species (**A**) is formed as the





Scheme 52 Investigation the reaction of PA and phenyllithium.

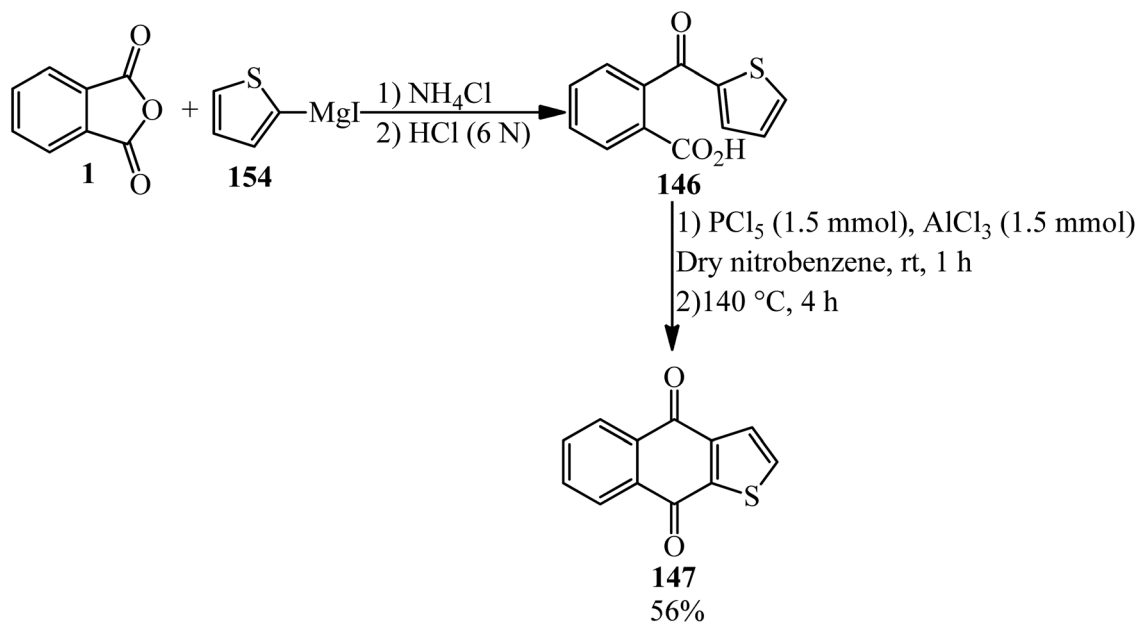


Scheme 53 Identification of the PA and furan-3-ylolithium reaction.

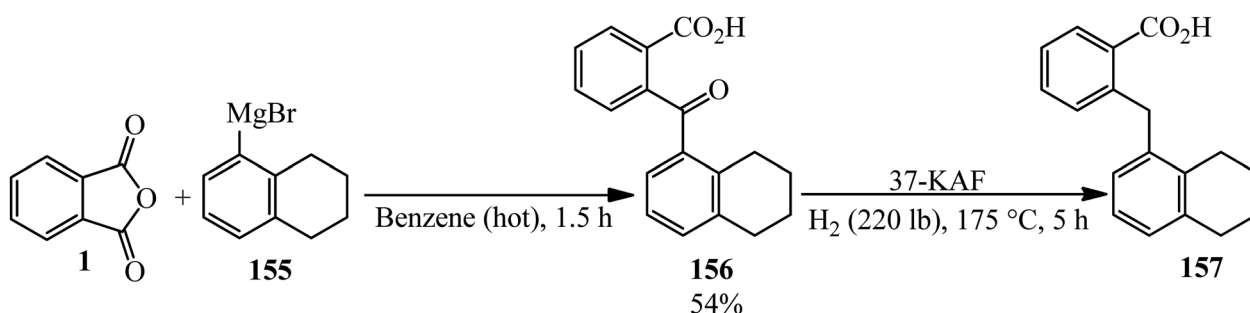
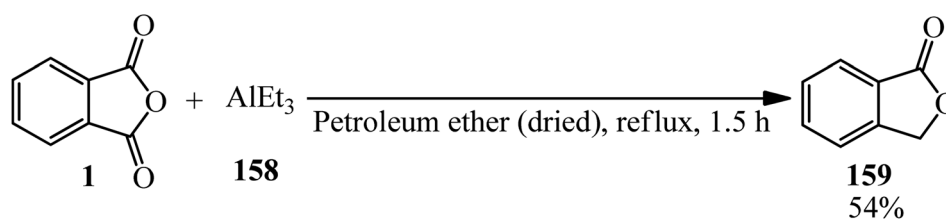
first step. If it is stable, it would be present in the system long enough to undergo an intramolecular ring opening (pathway A) to form 2-arylbzoate (C), which hydrolyzed to generate 2-arylbzoic acid (96). On the other hand, if (A) is very unstable, it may decompose intramolecularly by the transfer of an organic group and concomitant ring opening (pathway B) to produce (B), which converted into the lactone form, which is 3,3-diphenylphthalide (149) either during hydrolysis or by the elimination

of metal oxide prior to hydrolysis. Compared with diethyl ether, THF and dimethyl sulphide are expected to have greater stabilizing effects on species (A), presumably by forming more stable complexes; thus, in their presence, the reaction of (A) directed to pathway A. Since dimethyl sulphide is the most efficient of the ligands examined, it is not surprising that the yields of 2-benzoylbenzoic acid are the highest in its presence (Scheme 58).¹⁷²





Scheme 54 Reaction of PA and 2-thenoylmagnesium iodide.

Scheme 55 Reaction of PA and α -tetraallylmagnesium bromide.

Scheme 56 Reaction of PA with triethylaluminum.

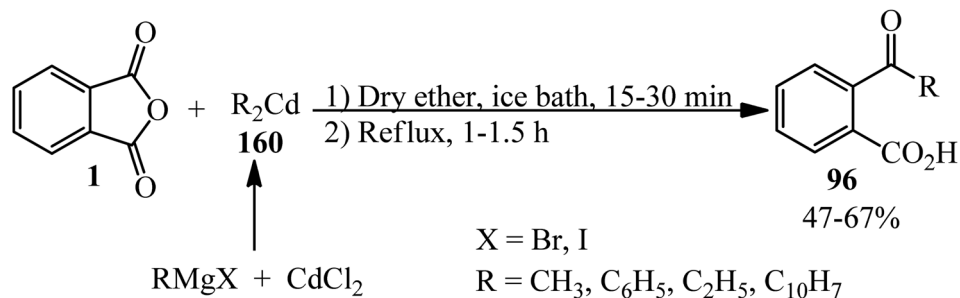
Wang in 2004 accounted an efficient method for the conversion of aldoximes (**161**) to nitriles (**162**) *via* the reaction of PA and aldoximes, in 1.01 : 1 molar ratio, which gained phthalic acid (**163**) as byproduct (Scheme 59).¹⁷³

According to the proposed mechanism, using PA (**1**) led to formation of the acylated aldoxime intermediate (**A**), which efficiently underwent a feasible [3,3]-sigmatropic rearrangement as the reaction pathway *via* this proposed six-membered transition state (**B**), which gave nitriles (**162**) without utilizing an external base (Scheme 60).¹⁷³

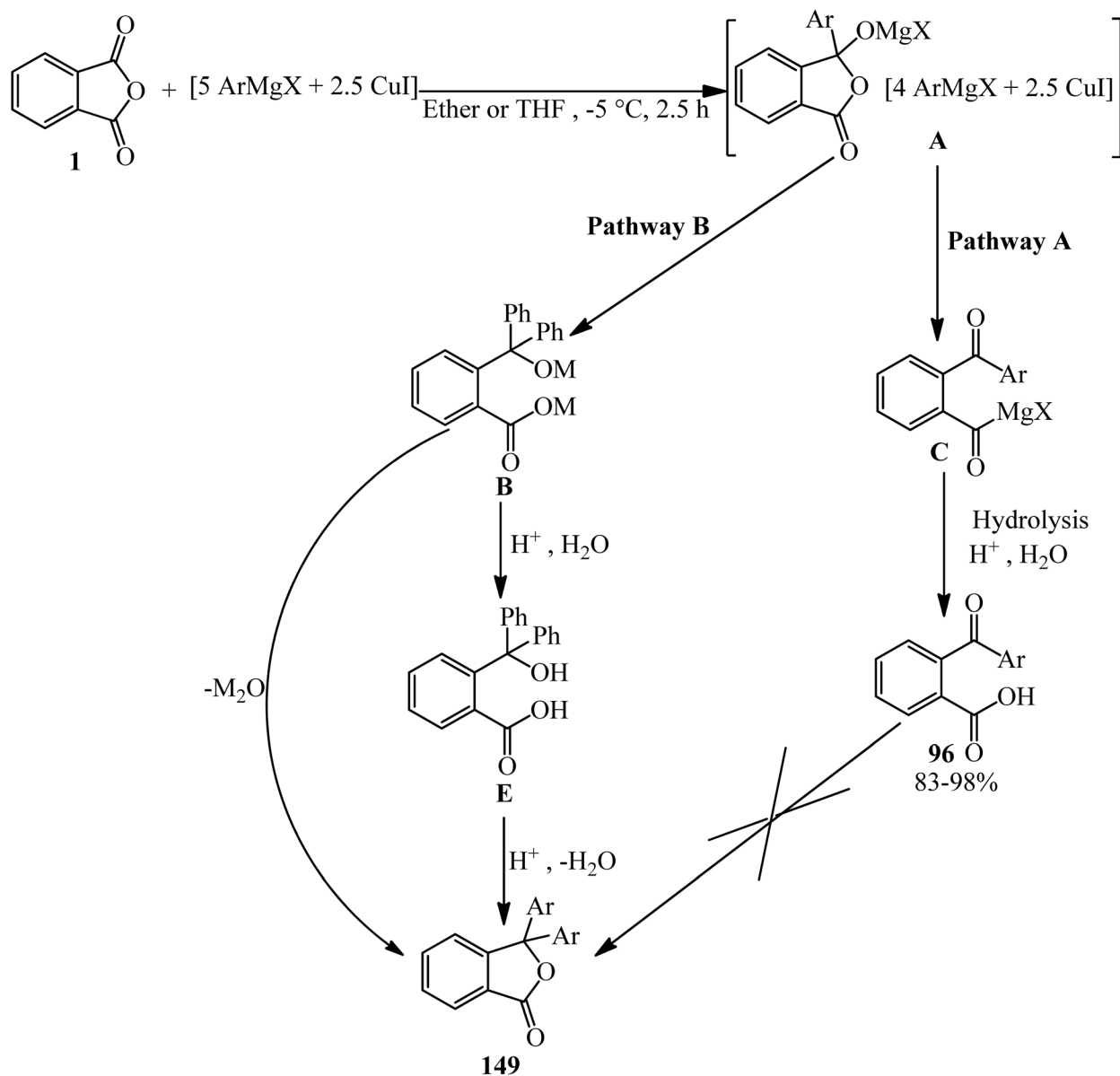
Heravi group in 2005 transformed amides (**164**) to carboxylic acids (**165**) upon reaction with PA under microwave irradiation in the absence of solvent (Scheme 61).¹⁷⁴ They utilized different amides (such as acetamide, acrylamide, and lactamide) successfully.

Kadhum group in 2022 supplied 2-(1,1-dimethyl-1*H*-benzo[*e*]indol-2(3*H*)-ylidene)-3-((4-(1,3-dioxoisindolin-2-yl)phenyl)imino)propanal (**167**) *via* the reaction of PA (**1**) and Schiff base ((*2E,3E*)-2-(1,1-dimethyl-1*H*-benzo[*e*]indol-2(3*H*)-ylidene)-3-(phenylimino)propanal) (**166**) (Scheme 62).¹⁷⁵





Scheme 57 Reaction of PA and organocadmium compounds.

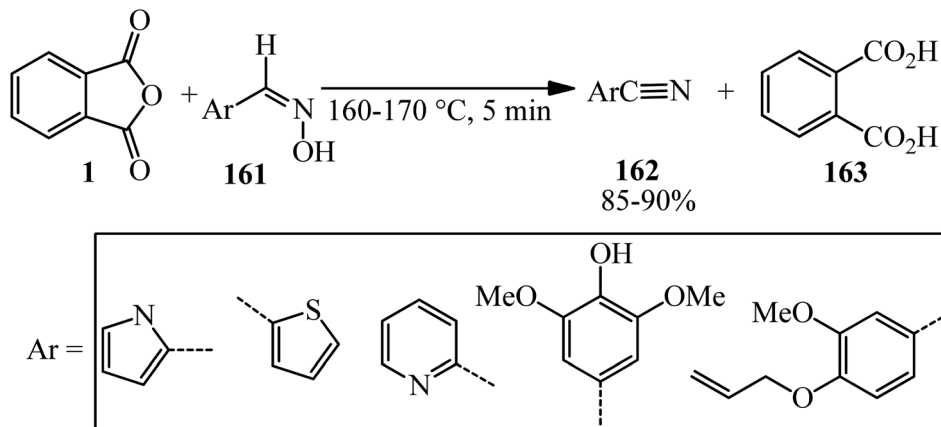


Scheme 58 Investigating the reaction of arylcoppermagnesium reagents with PA.

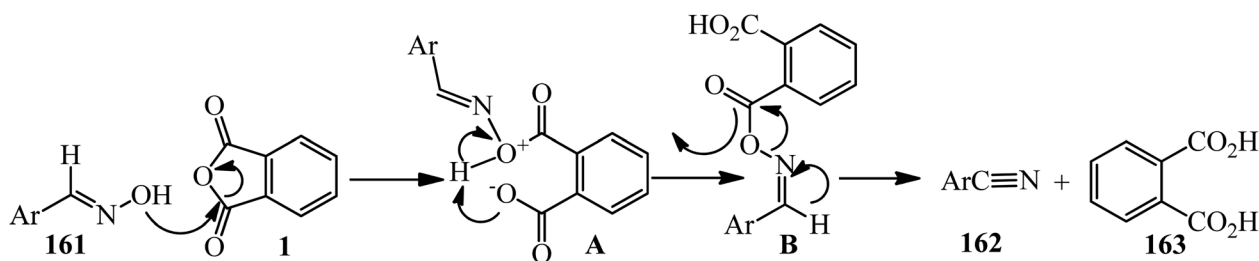
Matsubara group in 2011 dedicated the decarbonylative cycloadditions of PA (**1**) with allenes (**168**) in a 1 : 1.5 molar ratio using $\text{Ni}(\text{cod})_2$ as the Ni(0) precursor to give δ -lactones (**169**) in

a single step. The reaction represented an unprecedented insertion reaction of a carbon-carbon double bond into a carbon-oxygen bond (Scheme 63).¹⁷⁶ The asymmetric variant

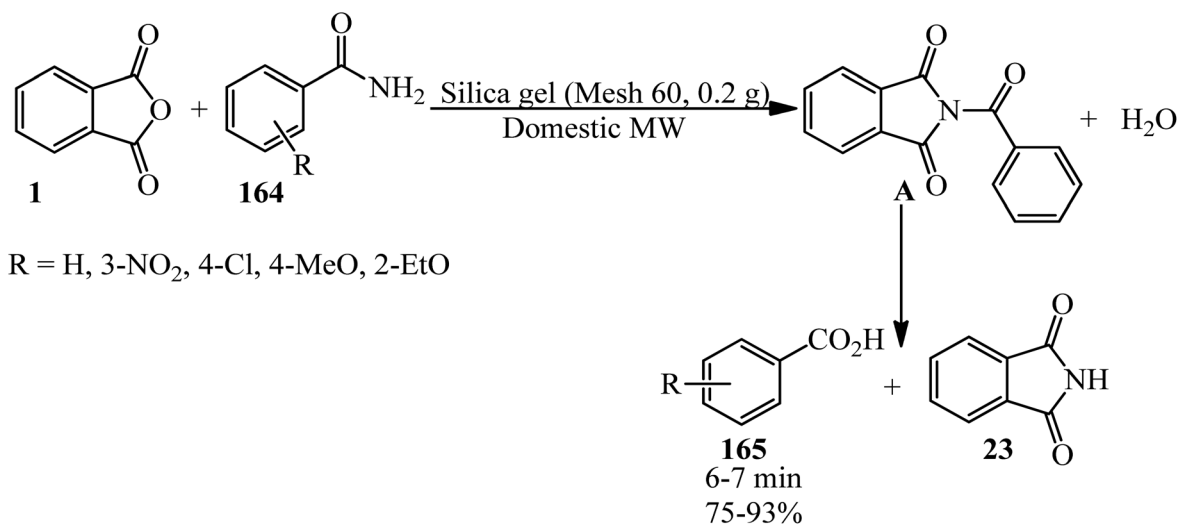




Scheme 59 Reaction of PA and aldoximes.



Scheme 60 Proposed mechanism for the reaction of PA and aldoximes.



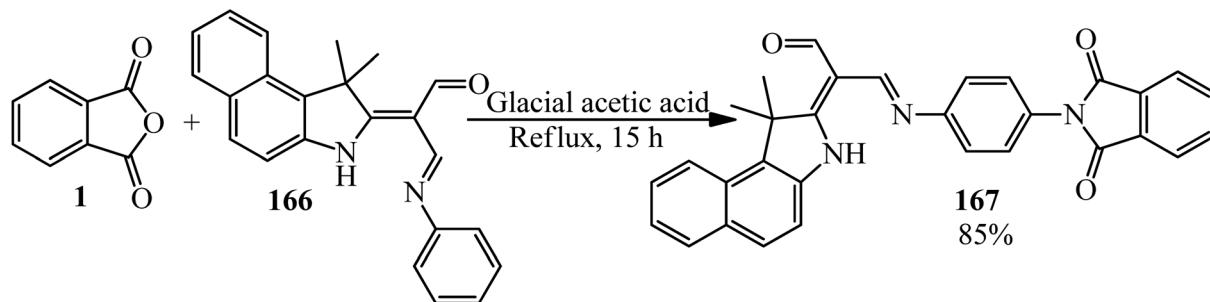
Scheme 61 Transformation of amides to carboxylic acids.

of the cycloaddition was also achieved using chiral phosphine ligands to provide δ -lactones enantioselectively (Scheme 64).¹⁷⁶

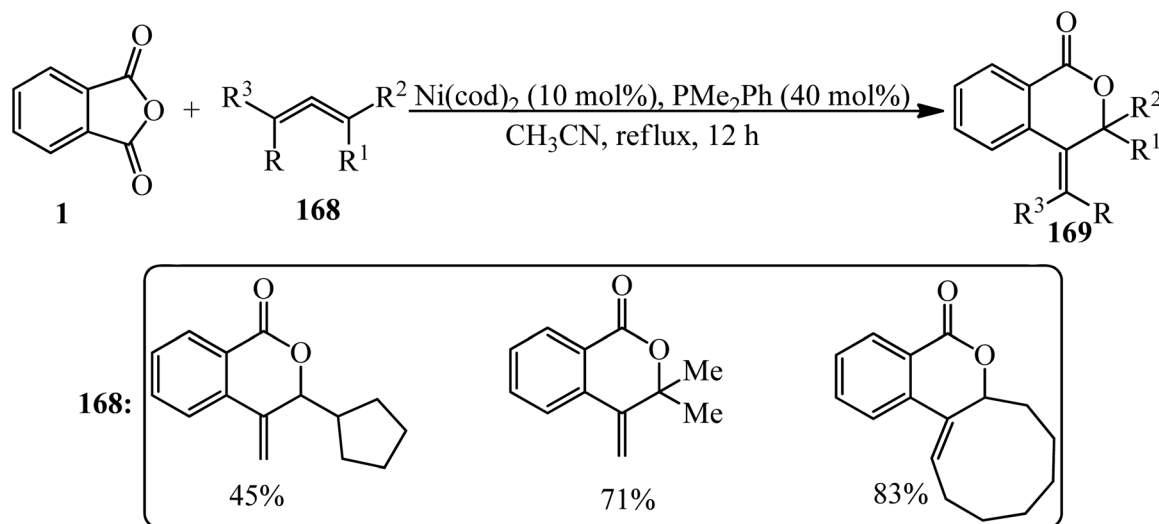
Láčová in 1986 considered the reaction of 2-benzothiazolylthioethanoic acid (**170**) (0.1 mol) with PA (0.2 mol) under conditions of Gabriel modification of Perkin synthesis. They claimed that in addition to the anticipated 3-(2-benzothiazolylthiomethylene)phthalide (**171**), four more compounds were identified as (*Z,Z*)-3,3'-thio-bis(methylenephthalide) (**172**), (*E,Z*)-

3,3'-thio-bis(methylenephthalide) (**173**), 1-(2-benzothiazolylthio)-1-inden-3-one (**174**), and dibenzothiazolyl disulfide (**175**). The starting compounds did not react in the presence of acetic anhydride since 2-benzothiazolylthioethanoic acid (**170**) was preferentially acetylated to yield 4-methylthiazol[2,3-*b*]benzothiazolium 5-carboxylate by an intramolecular condensation. Only (*Z*)-3-(2-benzothiazolylthiomethylene)phthalide (**171**) corresponded to a normal course of aldol synthesis (Scheme 65).¹⁷⁷

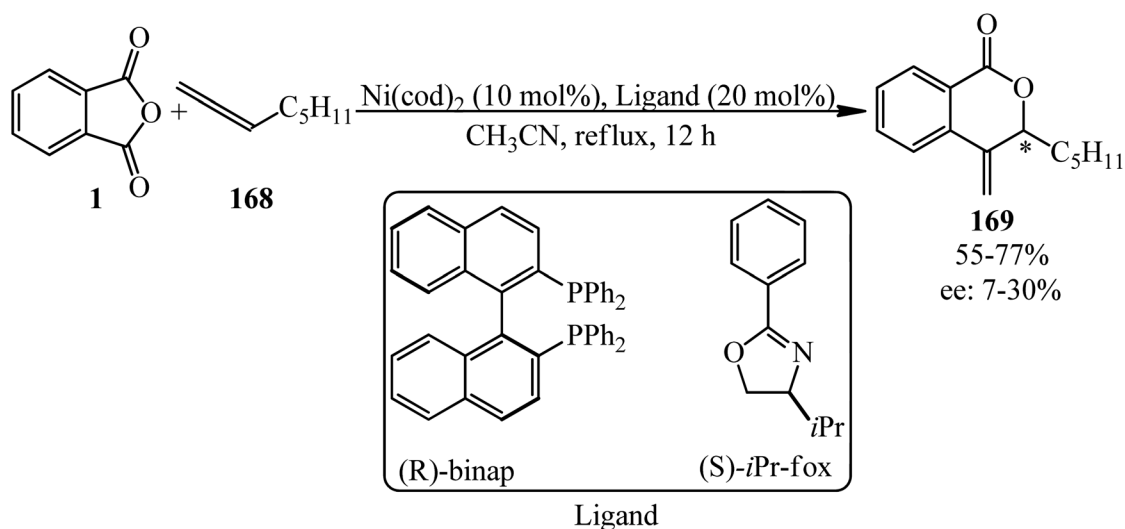




Scheme 62 Synthesis of 2-((1,1-dimethyl-1H-benzo[e]indol-2(3H)-ylidene)-3-((4-(1,3-dioxisoindolin-2-yl)phenyl)imino)propanal.



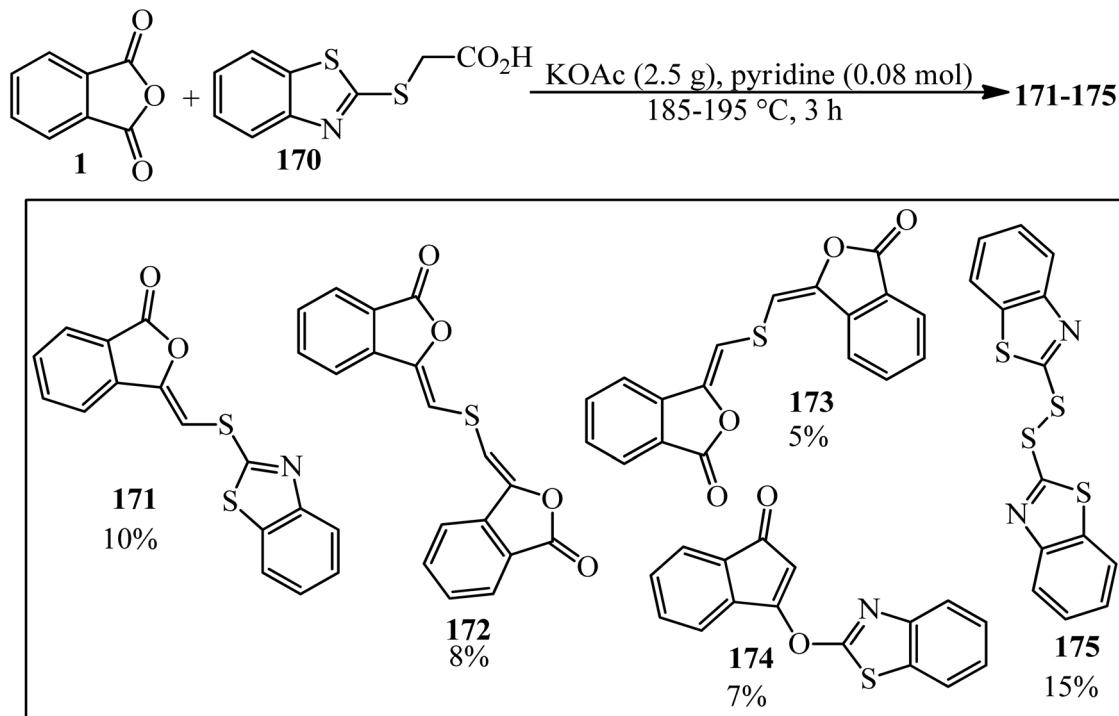
Scheme 63 Decarbonylative cycloadditions of PA with allenes.

Scheme 64 Enantioselective synthesis of δ -lactones using chiral phosphine ligands.

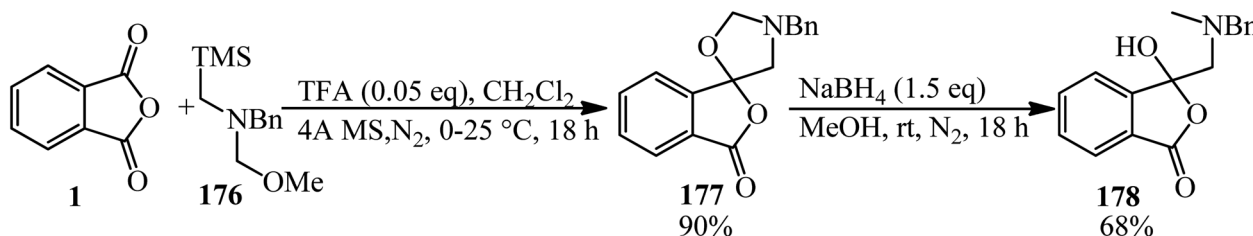
Meyer and Ryan group in 2015 reported that PA (1 eq) underwent a 1,3-dipolar cycloaddition reaction with *N*-benzylazomethine ylide (that formed *in situ* from *N*-(methoxymethyl)-

N-(trimethylsilylmethyl)benzylamine (176) (1.1 eq.) and a catalytic amount of trifluoroacetic acid) to produce unstable spiro(isobenzofuran-1,5'-oxazolidin)-3-ones (177), which was



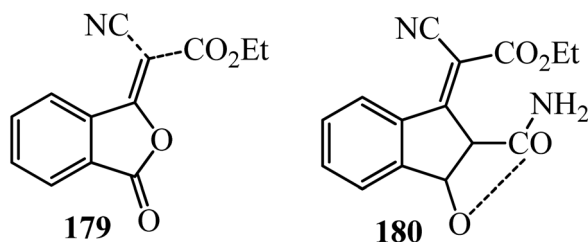


Scheme 65 Reaction of 2-benzothiazolylthioethanoic acid with PA.

Scheme 66 1,3-Dipolar cycloaddition of PA with *N*-benzylazomethine ylide.

reduced with sodium borohydride to afford 1(3*H*)-isobenzofuranones (**178**) (Scheme 66).¹⁷⁸

Renfrew and Bostock in 1977 performed the Knoevenagel condensation of PA (**1**) with ethyl cyanoacetate (**112**) in the presence of sodium as the catalyst in benzene for 4 h to furnish ethyl cyano(phthalidyldene)acetate (**179**) in 95% yield. Utilizing triethylamine as a base in toluene at 90 °C for 24 h, a yellow color was produced immediately, and upon refluxing,



Scheme 67 Ethyl cyano(phthalidyldene)acetate and (2)-ethyl 2-carbamoyl-8-cyano-3-hydroxybenzofulvene-8-carboxylate.

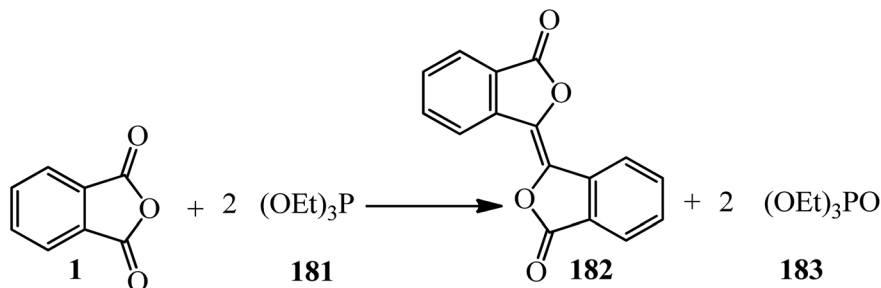
an intense orange color developed. On cooling, an orange oil separated, which on acidification gave a white solid in 64% yield, identified as (2)-ethyl 2-carbamoyl-8-cyano-3-hydroxybenzofulvene-8-carboxylate (**180**) (Scheme 67).¹⁷⁹

The Ramirez group in 1961 found that triethyl phosphite (**181**) effected the conversion of PA (**1**) in a 2 : 1 molar ratio into biphthalyl (**182**) in satisfactory yield (70%). The reaction was carried out in an excess of the phosphite as the solvent and the biphthalyl separated from the solution in nearly pure state. It was observed that most of the excess triethyl phosphite was isomerized to diethyl ethylphosphonate (**183**) during the reaction (Scheme 68).¹⁸⁰

3. Applications of PA in three-component and pseudo three-component reactions

In 2006, Habibi and Marvi obtained *N*-phthalimidophthalimide (**184**) from the reaction of PA (**1**) and hydrazine hydrate (**56**) in





Scheme 68 Synthesis of biphtalyl.

2 : 1 molar ratio in the presence of montmorillonite KSF and montmorillonite K-10 clays as natural heterogeneous catalysts with the help of microwave irradiation (600–780 W) under solvent-free conditions (Scheme 69).¹⁸¹ The pure product achieved through recrystallization by glacial acetic acid.

Habibi *et al.* in 2007 described a solvent-free and microwave-assisted procedure for the synthesis of phthalazino[2,3-*b*]phthalazine-5,7,12,14-tetraones (**186**) *via* the pseudo three-component reaction of PA (**1**) and (thio)semicarbazides (**185**) using montmorillonite K-10 clay. The catalyst demonstrated significant activity after recovering in another cycle (Scheme 70).¹⁸²

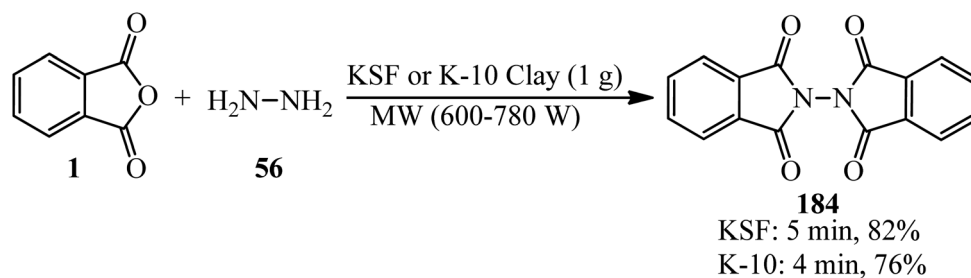
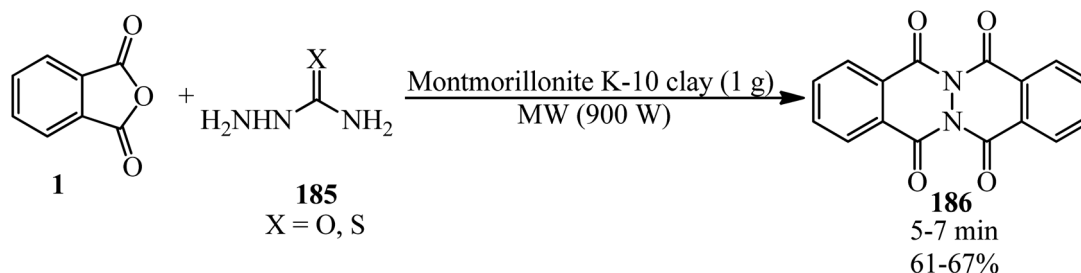
Maccioni's group in 2003 also gained the commands (**186**) *via* the catalytic role of acetic acid in refluxing isopropyl alcohol media within 1 h with 61–75% yield.¹⁸³ The antimicrobial properties of the products was also examined by the authors that was not satisfactory.

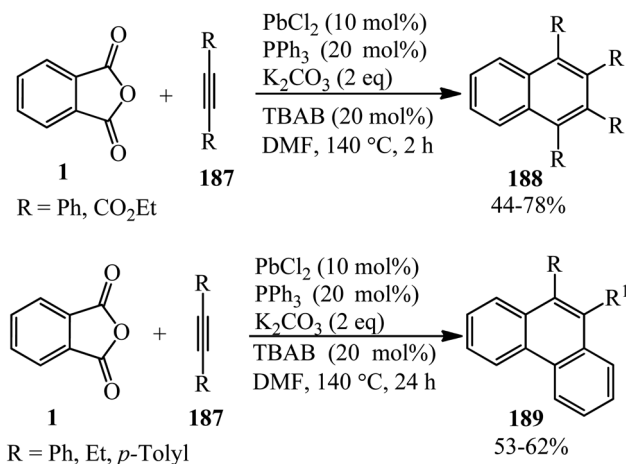
Jafarpour and coworkers in 2013 developed a new method for the decarboxylative and decarbonylative addition of cyclic anhydrides to alkynes. They performed the palladium-catalytic

benzannulation of PA (**1**) with alkynes (**187**) in a 1 : 3 and 1 : 2 molar ratio to obtain the corresponding polyfunctionalized sterically condensed naphthalenes (**188**) and phenanthrenes (**189**), respectively. The sequential liberation of CO₂ and CO occurred *via* the oxidative decomposition of anhydride (Scheme 71).¹⁸⁴

Fardpour *et al.* in 2019 constructed substituted vinylated phthalides (**191**) through a ruthenium-catalyzed cross-dehydrogenative coupling reaction of PA (**1**) with acrylates (**190**) in 1 : 2 molar ratio in the presence of Cu(OAc)₂ · H₂O as the oxidant in *N*-methyl-2-pyrrolidone (NMP) solvent (Scheme 72).¹⁸⁵ The reaction proceeded *via* C–H bond activation through a successive double vinylation accompanied by decarboxylation and annulation reaction.

Silva's research group in 2020 described the solvent-free synthesis of rhodamine dyes (**193**) *via* the reaction of PA and *m*-aminophenols (**192**) using Nb₂O₅ as the catalyst. The solvatochromic study of rhodamines was also performed (Scheme 73).¹⁸⁶ Rhodamine dyes possessed various applications due to

Scheme 69 Synthesis of *N*-phthalimidophthalimide.Scheme 70 Synthesis of phthalazino[2,3-*b*]phthalazine-5,7,12,14-tetraone.



Scheme 71 Synthesis of polyfunctionalized naphthalenes and phenanthrenes.

their properties, such as high molar absorptivity, high fluorescence quantum yield, photostability, and absorption and emission wavelengths in the visible region, which make them good candidates in electronic devices (such as lasers and OLEDs).

Eshghi *et al.* in 2015 explained the condensation of PA (**1**) with substituted phenols (**135**) in the presence of cobalt hydrogen sulfate under melt conditions that gave 3*H*-spiro [isobenzofuran-1,9'-xanthen]-3-one (**194**). In the case of phenol,

the product was 3,3-bis(4-hydroxyphenyl)isobenzofuran-1(3*H*)-one (**195**) (Scheme 74).¹⁸⁷

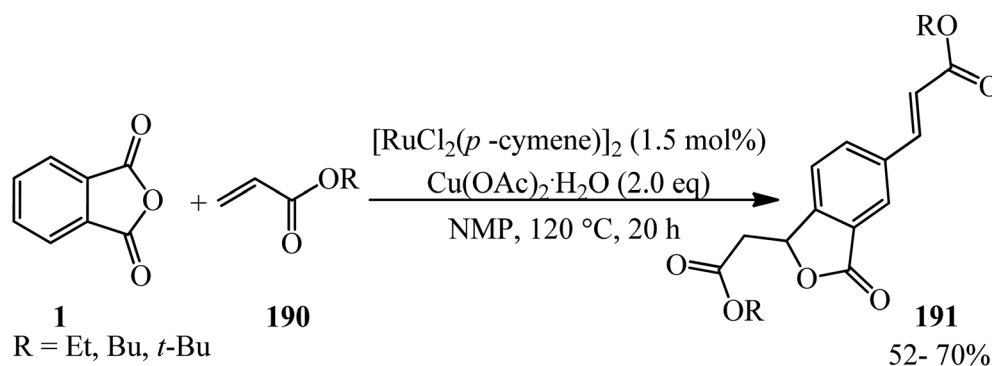
Shaabani's research group in 2002 reported the addition of PA (**1**), dialkyl acetylenedicarboxylates (**17**) and alkyl isocyanides (**196**), leading to highly functionalized γ -spiroiminolactones (**197**). The authors purposed preparing highly functionalized ketenimines (**198**) (pathway A), but the unusual γ -spiroiminolactones (**197**) were obtained in high yields (pathway B) (Scheme 75).¹⁸⁸

Shaabani *et al.* in 2009 also applied catalyst-free three-component method for the synthesis of a benzo-fused spiroacton (**197**) from the reaction of PA (**1**), dimethyl acetylenedicarboxylate (**17**), and cyclohexyl isocyanide (**196**) in dichloromethane at ambient temperature within 2 h with 82% yield. In fact, the zwitterion formed from isocyanide and dialkyl acetylenedicarboxylate reacted with PA to form the benzo-fused spiroacton (**197**).¹⁸⁹

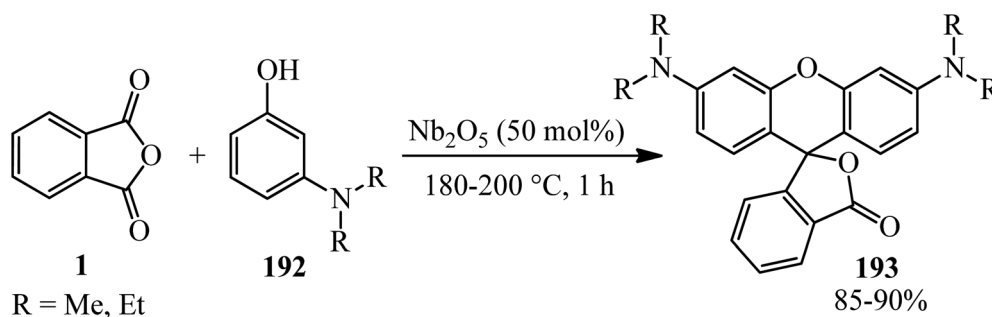
Mahmoodi's research group in 2010 demonstrated the regioselective synthesis of phthalazinones (**199**) from PA (**1**), phenyl hydrazine (**49**), and arenes (**93**) in the presence of efficient recyclable heterogeneous catalyst, montmorillonite-KSF, in high yields (Scheme 76).¹⁹⁰

The same research group in 2012 also reported ultrasound-assisted (45 kHz) preparation of phthalazinones (**199**) in the presence of the recyclable catalyst [bmim]Br/AlCl₃ (2 eq.) at 60 °C within 4–5 h by 65–75% yield.¹⁹¹

Thirupaiah and Vedula in 2013 developed a facile and efficient one-pot, three-component protocol for the synthesis of

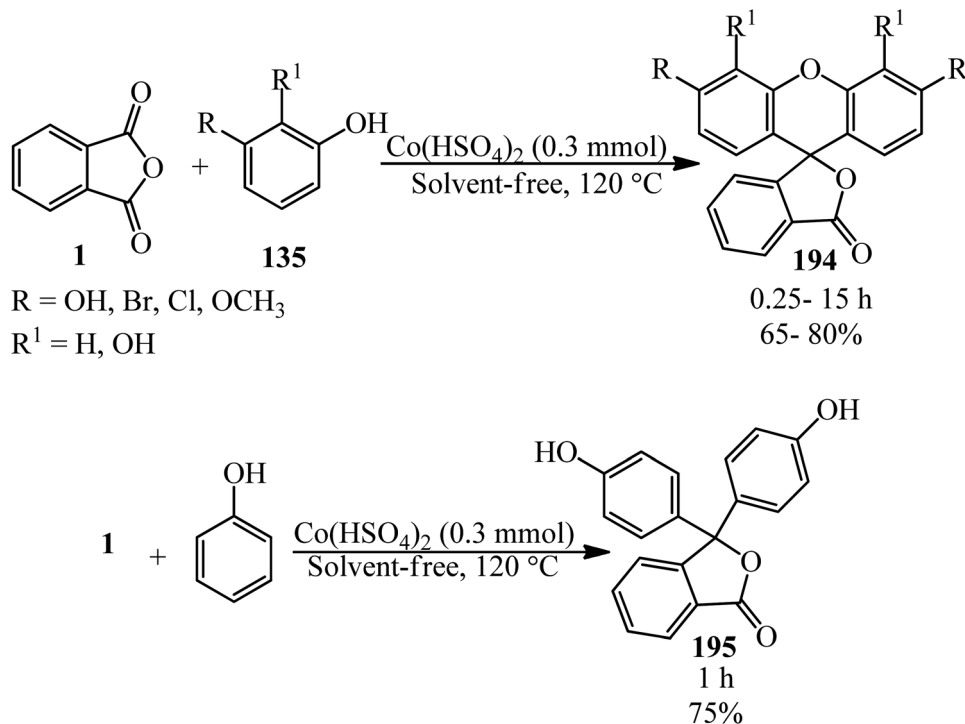


Scheme 72 Synthesis of vinylylated phthalides.



Scheme 73 Synthesis of rhodamines.





Scheme 74 Synthesis of 3*H*-spiro[isobenzofuran-1,9'-xanthen]-3-ones and 3-bis(4-hydroxyphenyl)isobenzofuran-1(3*H*)-one.

novel 2,3-dihydro-2-(6-(4-hydroxy-6-methyl-2-oxo-2*H*-pyran-3-yl)-7*H*-[1,2,4]-triazolo[3,4-*b*][1,3,4]thiadiazin-3-yl)phthalazine-1,4-dione (**202**) from the reaction of PA (**1**), 3-(2-bromoacetyl)-4-hydroxy-6-methyl-2*H*-pyran-2-one (**200**), and 4-amino-5-hydrazino-4*H*-[1,2,4]triazole-3-thiol (**201**) in acetic acid medium (Scheme 77).¹⁹² The product was purified by simple recrystallization from ethanol.

Chunduru and Vedula in 2013 reported the synthesis of aryl(hetaryl)-substituted thiazolylphthalazine-1,4-diones (**206**, **207**) via the reaction of PA (**1**), thiosemicarbazide (**203**), and phenacyl bromides (**204**)/3-(2-bromoacetyl)coumarins (**205**), respectively (Scheme 78).^{193,194}

Sujatha and Vedula in 2019 presented a novel one-pot multicomponent method for the synthesis of (*E*)-2-((benzylideneamino)-5-mercapto-4*H*-1,2,4-triazol-3-yl)-2,3-dihydrophthalazine-1,4-diones (**211**) via the reaction of PA (**1**), aromatic aldehyde (**19**), and 4-amino-5-hydrazino-4*H*-1,2,4-triazole-3-thiol (**201**) (Scheme 79).¹⁹⁵

Vedula *et al.* in 2020 developed the synthesis of a series of 2-(6-phenyl-7*H*-[1,2,4]triazolo[3,4-*b*][1,3,4]thiadiazin-3-yl)-2,3-dihydrophthalazine-1,4-diones (**212**) via a one-pot multicomponent reaction of PA (**1**), 4-amino-5-hydrazineyl-4*H*-1,2,4-triazole-3-thiol (**201**), and substituted 2-bromo-1-phenylethanones (**204**) in the presence of acetic acid under reflux conditions (Scheme 80).¹⁹⁶

Zare Fekri and Farjood Feshalami in 2020 investigated the multicomponent reaction of PA (**1**) and 2-aminobenzimidazole (**2**), and arenes (**93**) to synthesize *N*-(1*H*-benzo[*d*]imidazol-2-yl)-2-benzoylbenzamides (**213**), and also, the multicomponent synthesis of pyridazinones (**214**) via the reaction of PA (**1**),

phenyl hydrazine (**49**), and arenes (**93**) using *L*-proline-functionalized silicapropyl-modified nanomagnetic catalyst (Fe₃O₄@SiO₂-propyl@*L*-proline) (Scheme 81).¹⁹⁷

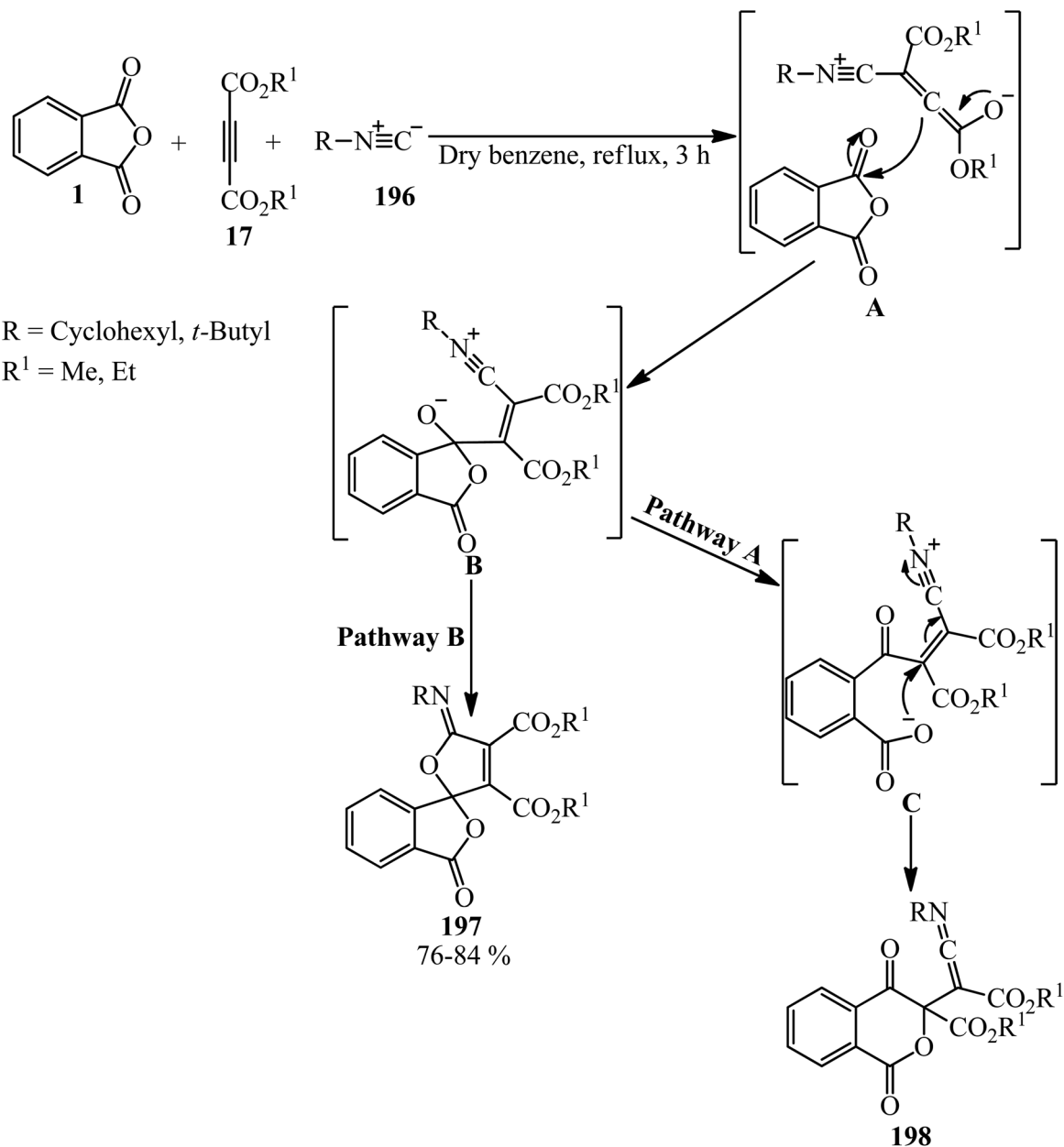
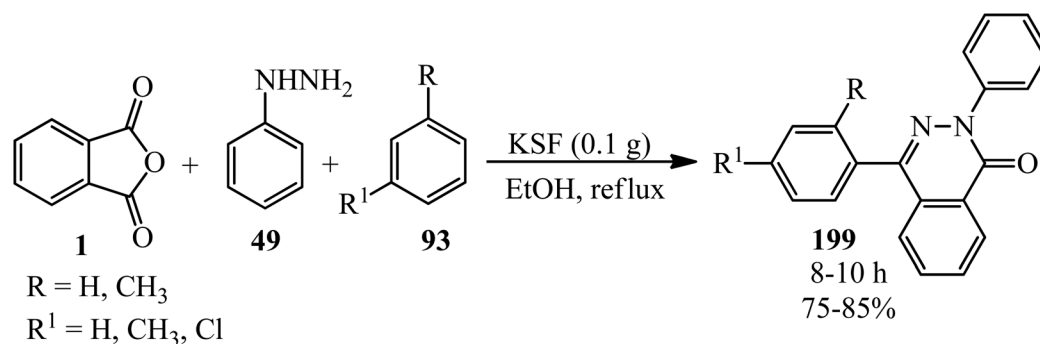
In 2021, they also obtained the amides (**213**) via the water-mediated the-component condensation of PA (**1**), 2-amino-benzimidazole (**2**), and (**93**) in the presence of nickel-ferrite silica-propyl supported glucosamine crystalline nanoparticles (NiFe₂O₄@SiO₂-propyl@glucosamine) (0.05 g) as an efficient, reusable, and heterogeneous catalyst within 1 h with 92–96% yield.¹⁹⁸

Bele and Darabantu in 2003 demonstrated the rapid synthesis of eighteen new 1,4-disubstituted phthalazines bearing an aryl or benzyl substituent at C-4 and a variety of aryloxy groups at C-1. The route A afforded the phthalazines (**215**) possessing a direct Ar–Ar linkage. Route B provided the *l*-chloro-4-benzylphthalazine (**219**). The neat condensation between (**215**, **219**) and selected phenols (**135**) resulted in the decomposition of the reaction mixtures. Indeed, three series of new phthalazine derivatives were obtained in refluxing xylene (**220**, **221**). All compounds were isolated simply by direct crystallization (Scheme 82).¹⁹⁹

Benjamin and Hijji in 2017 developed a novel green one-pot synthetic technique for the generation of thalidomide (**223**) via the reaction of PA (**1**), glutamic acid (**4**), and ammonium chloride (**222**) in a 1 : 1 : 1.1 molar ratio in the presence of catalytic amounts of 4-*N,N*-dimethylaminopyridine (DMAP) via microwave irradiation (Scheme 83).²⁰⁰

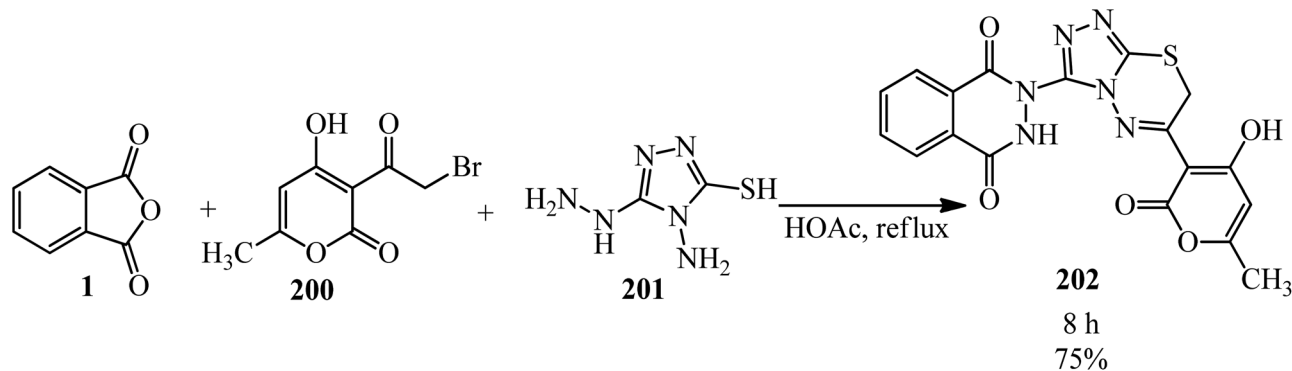
Garcia and Vilarrasa in 1986 prepared *N*-substituted phthalimides (**23**) via the reaction of PA (**1**), triphenylphosphine (**16**), and azides (**224**) in the presence of tetrabutyl ammonium



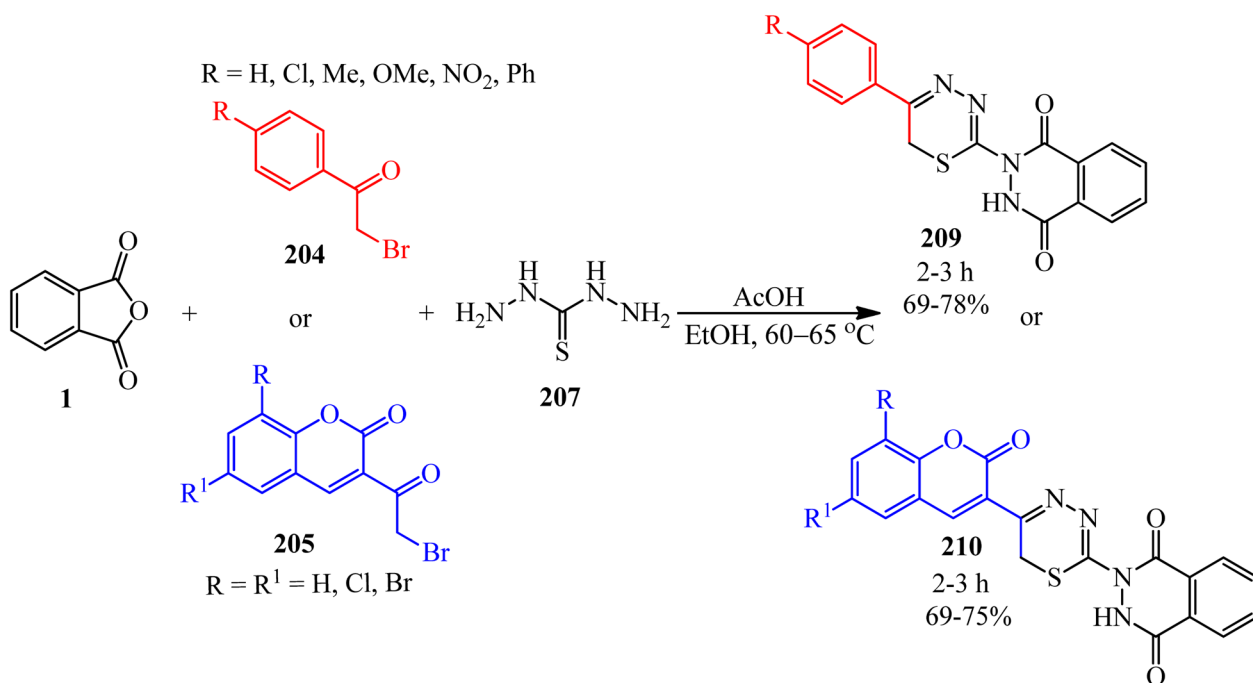
Scheme 75 Synthesis of functionalized γ -spiroiminolactones.

Scheme 76 Synthesis of phthalazinones.





Scheme 77 Synthesis of 2,3-dihydro-2-(6-(4-hydroxy-6-methyl-2-oxo-2H-pyran-3-yl)-7H-[1,2,4]-triazolo[3,4-b][1,3,4]thiadiazin-3-yl)phthalazine-1,4-dione.



Scheme 78 Synthesis of aryl(hetaryl)-substituted thiazolylphthalazine-1,4-diones.

cyanide as the catalyst and benzene or toluene as solvent (Scheme 84).²⁰¹

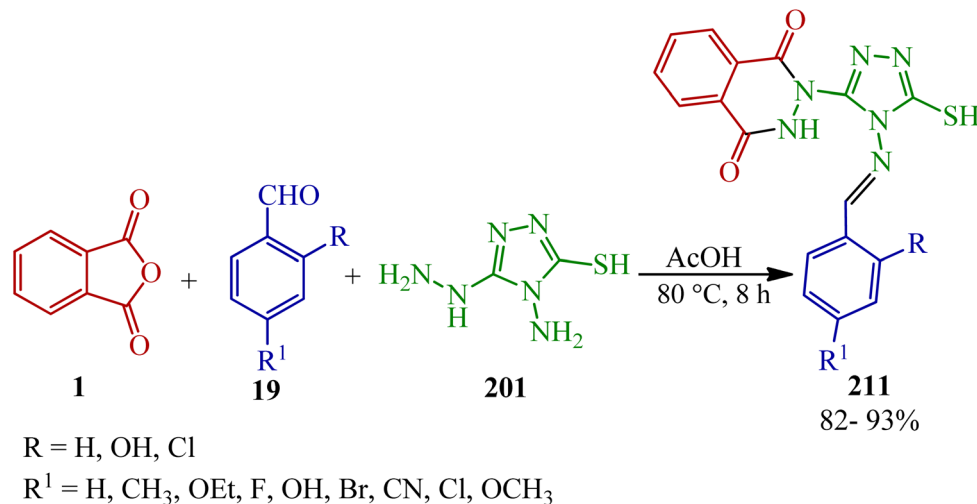
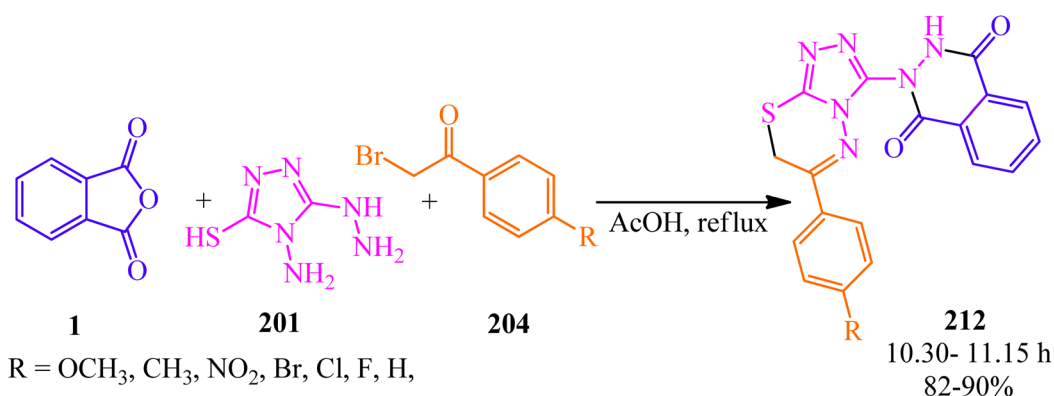
Deniau and coworkers in 2005 extended the asymmetric synthesis of diarylphosphine oxide-substituted isoindolinones (**229**) by a three-step reaction starting from PA (**1**) and (*S*)-1-amino-2-alkyloxymethylpyrrolidine (**225**) to prepare phthalhydrazides (**226**), which was reduced to 2-((*S*)-2-(alkoxymethyl)pyrrolidin-1-yl)-3-hydroxyisoindolin-1-one (**227**). In the second step, the final products (**229**) were obtained from the reaction of (**227**) with diarylphosphine oxides (**228**) (Scheme 85).^{202,203} The “de” of the products is more than 96% after recrystallization from hexane/toluene.

A green synthetic protocol was developed by Kalpana's group in 2021 for the synthesis of imidazo[4,5-*b*]pyrazine-conjugated benzamides (**231**) *via* the one-pot three-component reaction of PA (**1**), substituted anilines (**21**), and pyrazine-2,3-diamine (**230**)

in the presence of phosphoric acid as the catalyst in heated water (Scheme 86).²⁰⁴ The products were evaluated for their anticancer activity against liver and ovarian cancer cell lines (HepG2 and HeLa), which demonstrated moderate to good activities. In addition, molecular modeling investigations affirmed the crucial binding interactions of the target protein and the synthesized ligands. In addition, the permeability and bioavailability properties were predicted along with molecular descriptors such as shape index, molecular complexity, and molecular flexibility.

Rasheed *et al.* in 2021 accomplished the reaction of PA (**1**) and 4,4'-(2,8-dimethyl-2,3-dihydro-1*H*-benzo[*b*][1,5]diazepine-2,4-diyl)dianiline (**232**) in 2 : 1 molar ratio in the refluxing acetic acid solvent under microwave irradiation to obtain 2,2'-((2,8-dimethyl-2,3-dihydro-1*H*-benzo[*b*][1,5]diazepine-2,4-diyl)



Scheme 79 Synthesis of (*E*)-2-(benzylideneamino)-5-mercapto-4*H*-1,2,4-triazol-3-yl)-2,3-dihydrophthalazine-1,4-diones.Scheme 80 Synthesis of 2-(6-phenyl-7*H*-[1,2,4]triazolo[3,4-*b*][1,3,4]thiadiazin-3-yl)-2,3-dihydrophthalazine-1,4-diones.

bis(4,1-phenylene))bis(isoindoline-1,3-dione) (**233**) as a dark brown solid (Scheme 87).²⁰⁵

Eissa in 2014 obtained six *N,N*-substituted phthaldicarboximides (**235**), which was synthesized from the reaction of PA (**1**) with diamines (**234**). The synthesized compounds were screened for their antibacterial activity against four microorganisms, namely, *Staphylococcus aureus*, *Bacillus subtilis*, *Escherichia coli*, and *Klebsiella pneumonia*, and they were found to exhibit good to moderate antibacterial activity (Scheme 88).²⁰⁶

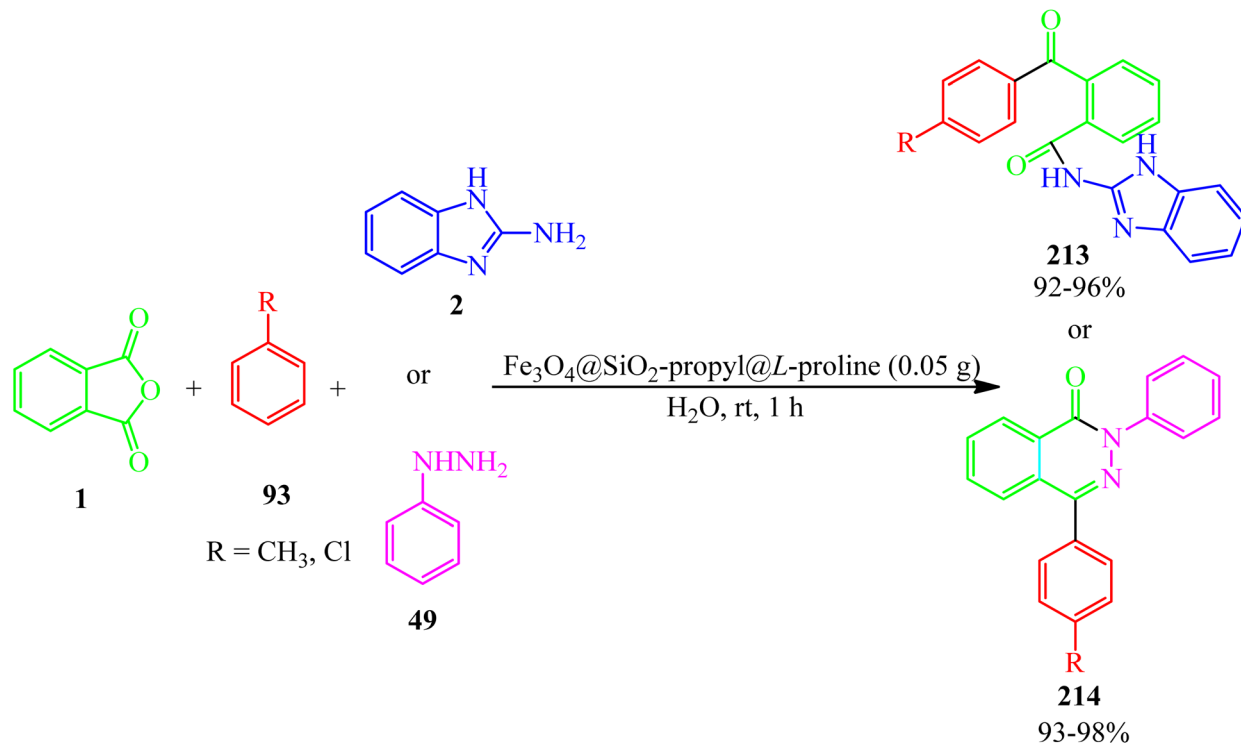
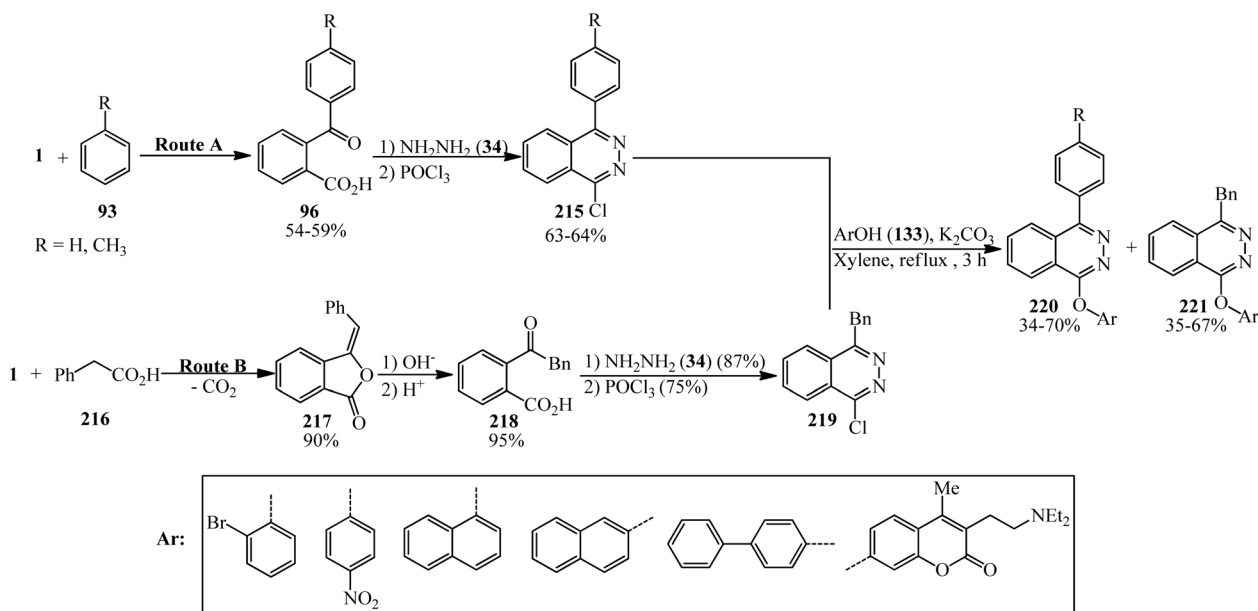
Chen's group in 2020 described an efficient cobalt-catalyzed intermolecular decarbonylative three-component carboamidation of alkynes (**187**) through the oxidative addition of cobalt into the N-C(O) bond of phthalimide and the subsequent decarbonylation. High regioselectivities achieved for unsymmetrical alkynes (including aryl-alkyl or aryl-aryl) to deliver polysubstituted isoquinolones (**237-240**) (Scheme 89).²⁰⁷

Zheng's group in 2019 described the efficient [2 + 2 + 2] benzannulation of PA (**1**) (or phthalic acid) with alkynes (**187**) in 1:1.33 molar ratio to prepare multisubstituted 1-naphthoic acids (**241**) via Ru-catalyzed C-H activation. The reaction

proceeded well using atmospheric oxygen as the sole oxidant with high atom/step economies (Scheme 90).²⁰⁸

Singh's group in 2015 demonstrated the transimidization reaction between PA (**1**), 2-amino-3-picoline (**242**), and 3-aminopropyl triethoxysilane (**243**) in a 1:1:1:1.125 molar ratio, which transited from 2-(3-methylpyridin-2-yl)isoindoline-1,3-dione (**23**) through a two-step reaction to synthesize *N*-(triethoxysilylpropyl)phthalimide (**244**). New silatranes (**246**, **247**) containing phthalimide as the exocyclic group were prepared by the transesterification reactions of (**244**) with triethanolamine/tris(isopropanolamine) (**245**), respectively. *N*-(silatranylpropyl)phthalimides (**246**, **247**) were evaluated for the preliminary antimicrobial activity using broth microdilution method, which shows that silatranes-possessing urea group exhibited good antimicrobial activity. The investigation of the UV-vis spectra of the silatranes proved fruitful for analyzing the hydrogen bonding of trifluoroacetic acid with phthalimide heterocycle possessing silatranes. Hydrogen bonding has shown major influence upon $n \rightarrow \pi^*$ transition, resulting in blueshift phenomenon (Scheme 91).²⁰⁹



Scheme 81 Synthesis of *N*-(1*H*-benzo[*d*]imidazole-2-yl)-2-yl)-2-benzoylbenzamides and pyridazinones.

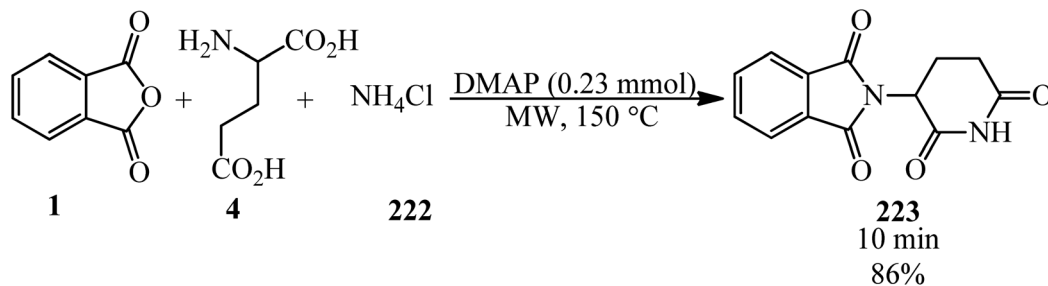
Scheme 82 Multistep synthetic procedure of new phthalazines.

Malik's group in 2022 accessed new imidazole-based *N*-phenylbenzamides (**249**) from one-pot three-component reaction PA, substituted anilines (**23**), and 2,3-diaminomaleonitrile (**248**) in the presence of HCl in refluxing ethanol. The cytotoxic evaluation revealed that some derivatives (with *para*-fluorine and *para*-methoxy substituent) were the most active compounds, which exhibited good activity against the tested

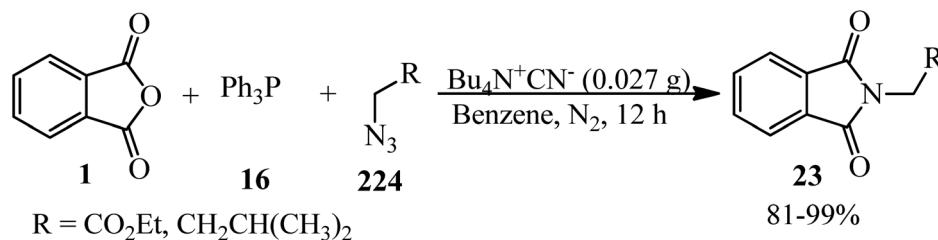
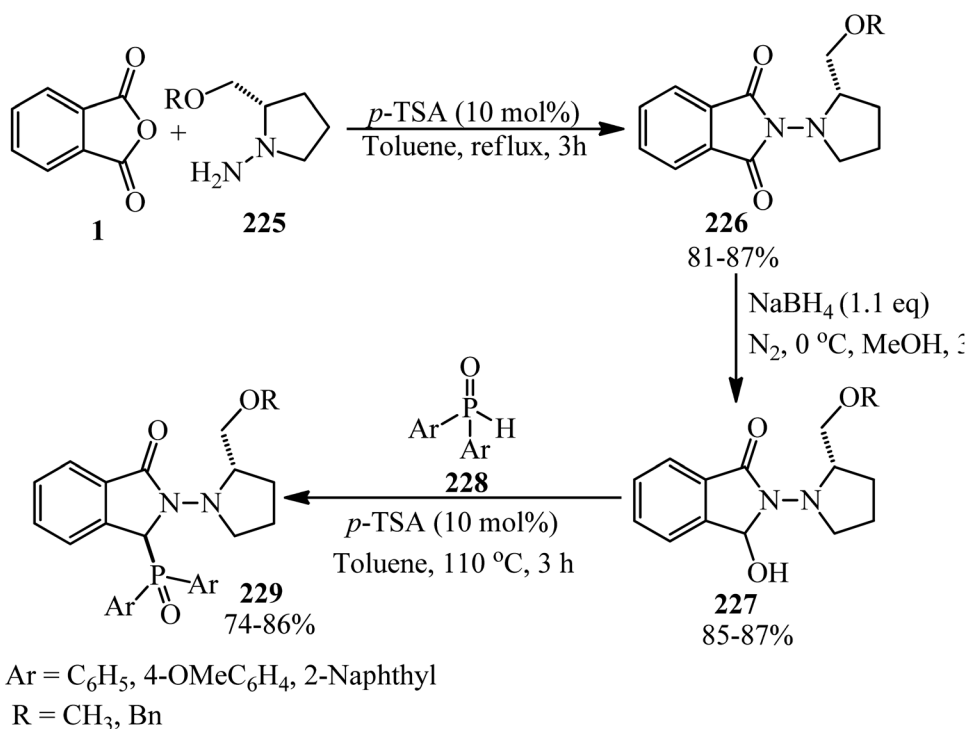
cancer cell lines (with single-digit IC₅₀ values). Computational studies and molecular dynamic simulations were also investigated (Scheme 92).²¹⁰

Xinwei's group in 2022 developed a rhodium(III)-catalyzed cascade reaction of phthalic anhydrides with cyclic 2-diazo-1,3-diketones (**250**) and methanol to obtain the esterified cyclohexenone-fused isocoumarins (**251**). The formation of C-C





Scheme 83 Synthesis of thalidomide.

Scheme 84 Synthesis of *N*-substituted phthalimides.

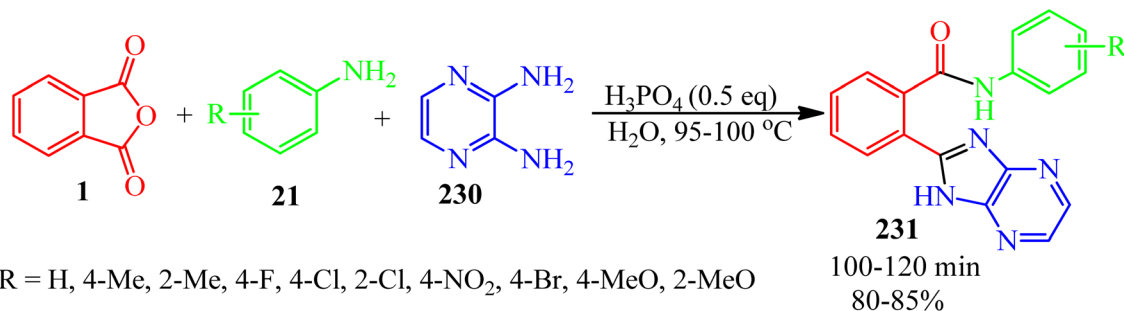
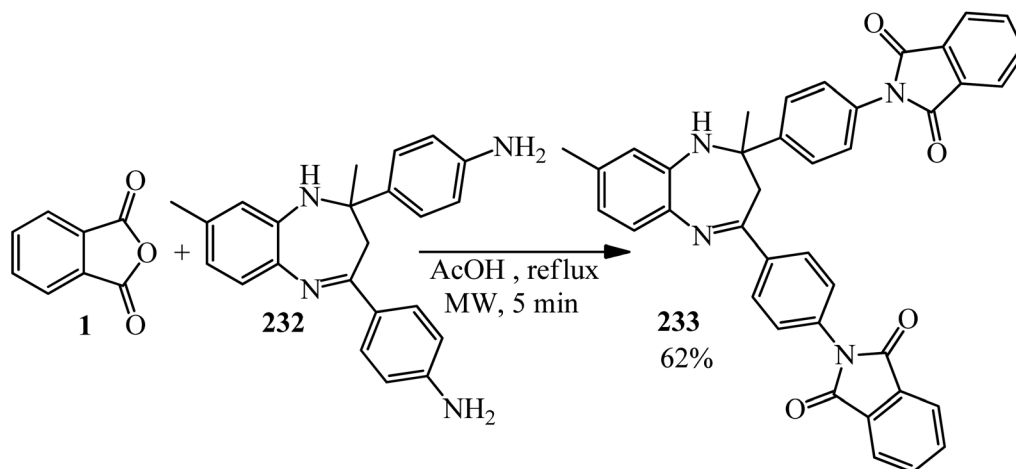
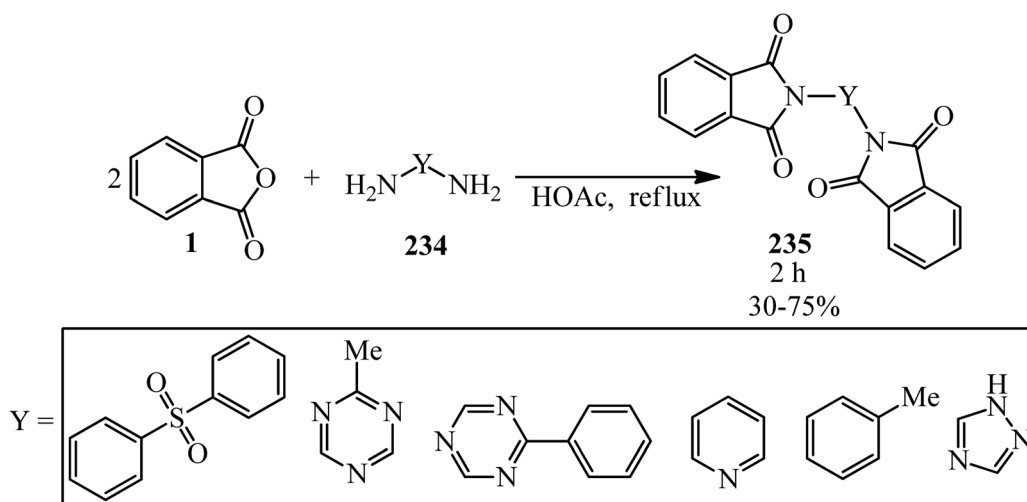
Scheme 85 Synthesis of diarylphosphine oxide-substituted isoindolinones.

and two C–O bonds through C–H activation, transannulative coupling, and subsequent annulation process occurred through this strategy. Furthermore, there is no need for an additive or base. Surprisingly, the reaction proceeded well under air atmosphere with broad substrate and good functional group tolerance (Scheme 93).²¹¹

4. Applications of PA in four-component and pseudo four-component reactions

Shaterian and Mohammadnia in 2012 prepared 1*H*-pyrazolo [1,2-*b*]phthalazine-5,10-diones (252) under ambient and

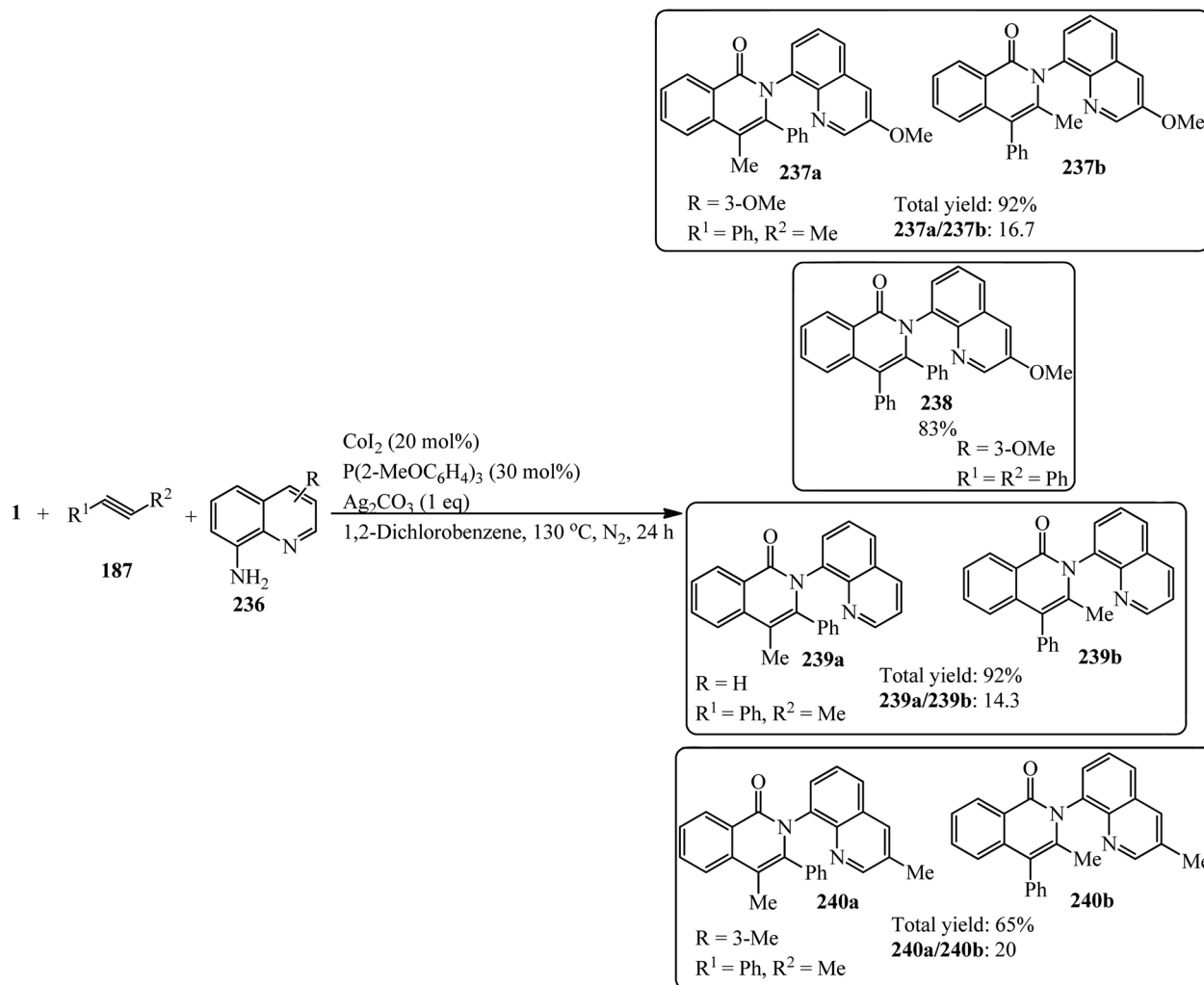


Scheme 86 Synthesis of synthesis of imidazo[4,5-*b*]pyrazine-conjugated benzamides.Scheme 87 Synthesis of 2,2'-((2,8-dimethyl-2,3-dihydro-1H-benzo[*b*][1,5]diazepine-2,4-diyl)bis(4,1-phenylene))bis(isoindoline-1,3-dione).Scheme 88 Preparation of *N,N*-substituted phthaldicarboximides.

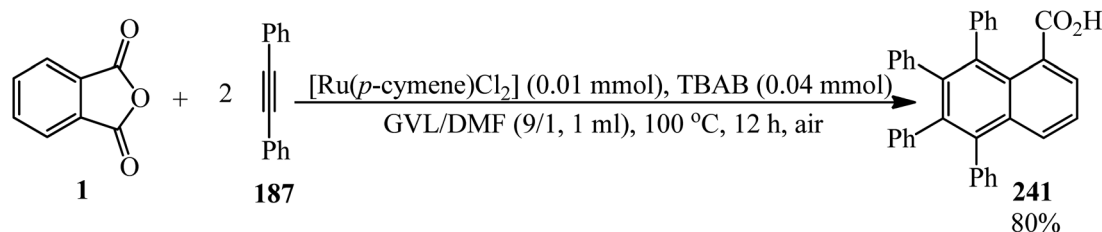
solvent-free conditions in the presence of mild basic ionic liquids, which are 1,8-diazabicyclo[5.4.0]undec-7-en-8-ium acetate (DBU[CH₃COO]), pyrrolidinium formate ([Pyrr][HCOO]), and pyrrolidinium acetate ([Pyrr][CH₃COO]) *via* the domino reaction of PA (**1**), aromatic aldehydes (**19**),

malononitrile (**48**)/ethyl cyanoacetate (**112**), and hydrazine monohydrate (**56**) (Scheme 94).²¹² The PA and (**56**) were mixed at 100 °C for 10 min to obtain solid phthalhydrazide intermediate (**A**). The subsequent addition of (**48**)/(**112**) and aromatic aldehydes (**19**) in addition with





Scheme 89 Regioselective synthesis of polysubstituted isoquinolones.



Scheme 90 Benzannulation of PA with alkyne.

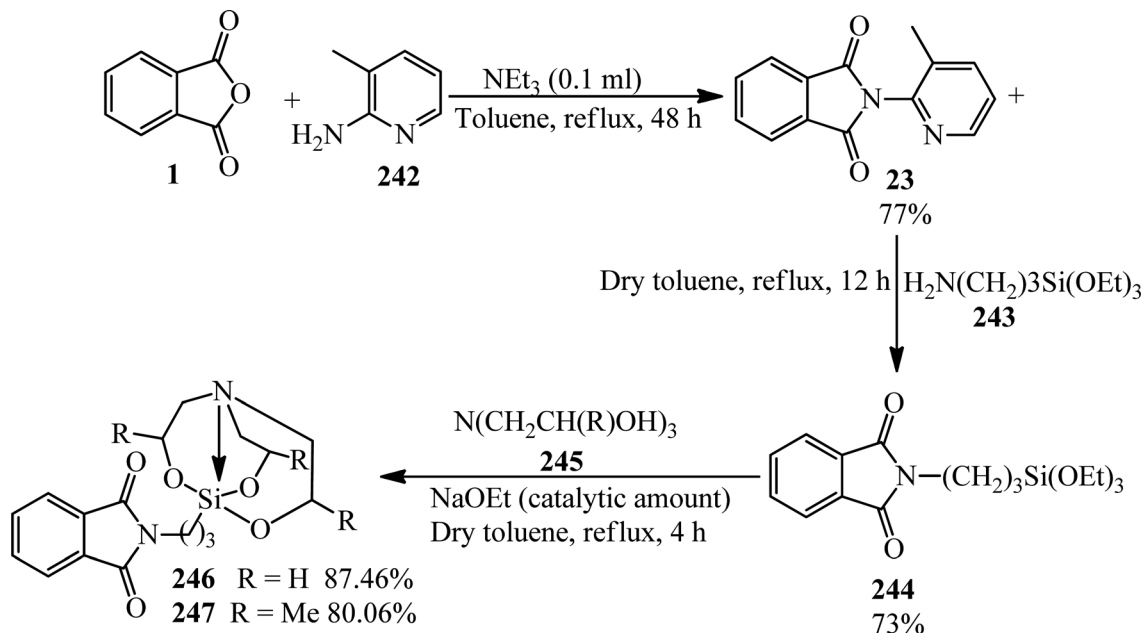
2 appropriate ionic liquid at ambient temperature yielded the products (252).

In 2014, the same group also obtained 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones (252) *via* the four-component domino one-pot condensation reaction of PA (1), aromatic aldehydes (19), malononitrile(48)/ethyl cyanoacetate (112), and hydrazine monohydrate (56) in the presence of magnetic Fe₃O₄ nanoparticles coated with (3-aminopropyl)-triethoxysilane (APTES-MNP, 10 mol%) as the catalyst under solvent-free conditions within 4–15 min with 94–86% yield.²¹³

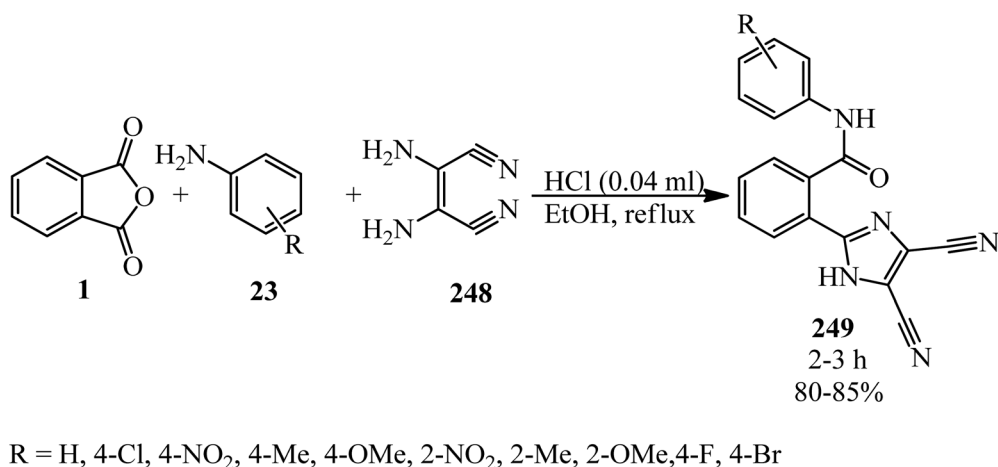
Many other research groups studied the domino four-component reaction of PA (1), aromatic aldehydes (19), malononitrile(48)/ethyl cyanoacetate (112), and hydrazine monohydrate (56) to prepare the corresponding 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones (252) in the presence of various catalytic systems successfully. For example, Ghomi *et al.* in 2014 utilized CuI nanoparticles (10 mol%) within 25–30 min by 82–93% to accelerate this transformation.²¹⁴

Some other researches utilized PA (1), aromatic aldehydes (19), malononitrile (48), and hydrazine monohydrate (56) to





Scheme 91 Transimidization reaction of PA, 2-amino-3-picoline, and 3-aminopropyl triethoxysilane.



R = H, 4-Cl, 4-NO₂, 4-Me, 4-OMe, 2-NO₂, 2-Me, 2-OMe, 4-F, 4-Br

Scheme 92 Synthesis of imidazole-based *N*-phenylbenzamides.

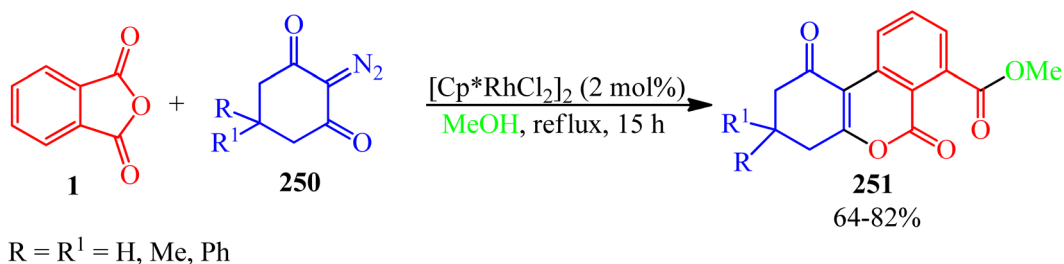
prepare 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones (**252**). They are as follows: (A) Ghorbani-Vaghei *et al.* in 2016 exploited piperidinium benzene-1,3-disulfonate nanomagnetic ionic liquid (NMIL, 20 mg) as a novel and reusable catalyst at 110 °C under solvent-free conditions within 40–80 min with 75–94% yield. The magnetized nanostructure was recycled and reused seven times without any loss of catalytic activity. In addition, they examined the four-component reaction in the presence of aliphatic aldehydes such as 3-phenylpropanal and fused aromatic candidates such as 2-naphthadehyde successfully.²¹⁵ (B) Lashkari's group in 2018 reported zinc acetate dihydrate (Zn(OAc)·2H₂O, 15 mol%) to accelerate the synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione derivatives (**252**) under solvent-free conditions at 70 °C with 93–83% yield within 2.5–4.5 h.²¹⁶ (C) Mohamadpour in 2020 discussed an ecosafe green

synthetic route to obtain (**252**) *via* the domino Knoevenagel–Michael cyclocondensation of PA (**1**), hydrazine monohydrate (**56**), aromatic aldehydes (**19**), and malononitrile (**48**) in the presence of carboxymethyl cellulose (CMC, 25 mol%) at 80 °C with a yield of 77–94% in 65–95 min period. Then, the catalyst was recovered after washing with ethyl acetate, filtering, and air drying, then reused for consecutive five runs with good yields and with insignificant CMC loss.²¹⁷ (D) Maghsoodlou and coworkers in 2016 achieved (**252**) *via* the domino reaction of PA (**1**), hydrazine monohydrate (**56**), aromatic benzaldehyde (**19**), and malononitrile (**48**) (1 : 1 : 1 : 1 molar ratio) with copper(II) acetate monohydrate (20 mol%) at 80 °C within 3–5 min with 73–89%.²¹⁸ (E) Amini's group in 2021 prepared (**251**) through a one-step reaction of PA (**1**), aromatic aldehydes (**19**), malononitrile (**48**), and hydrazine monohydrate (**56**) (in a 1 : 1 : 1 : 1.2

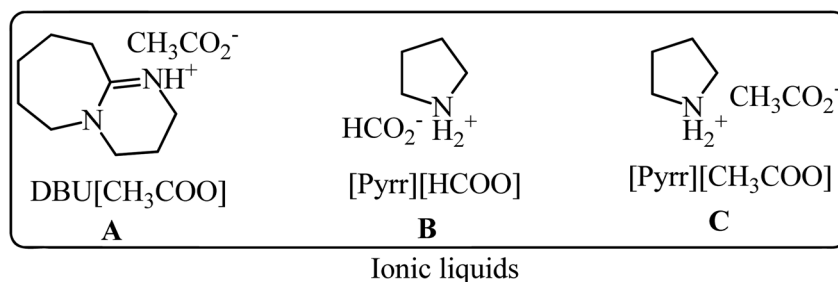
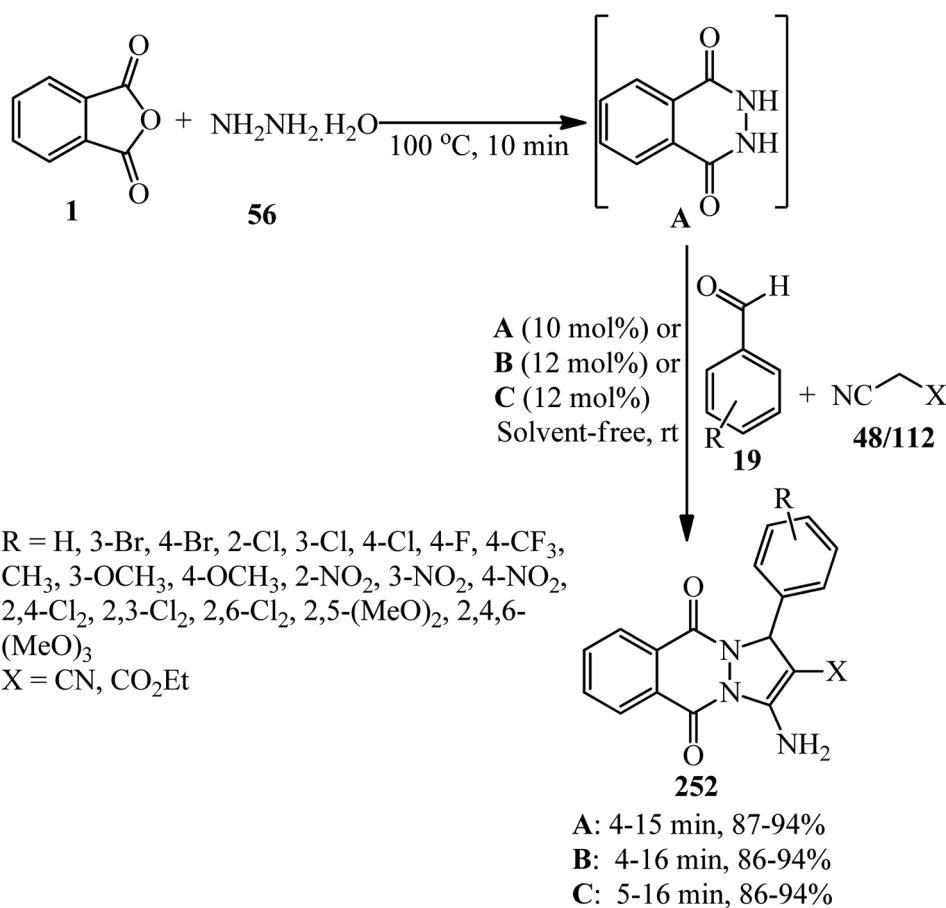


molar ratio) in the presence of $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ (10 mol%) in refluxing EtOH for 2–4 h within 81–96%. They also examined the procedure with ethanal (as an aliphatic candidate) to

prepare the corresponding products successfully. The authors executed the α -glucosidase inhibitory activity of the synthesized compounds using a source of the α -glucosidase enzyme



Scheme 93 Preparation of cyclohexenone-fused isocoumarins.



Scheme 94 Synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones.



(EC3.2.1.20, *Saccharomyces cerevisiae* at 20 U mg⁻¹ concentration). The results revealed that some of the products are efficient inhibitors of the α -glucosidase enzyme compared to the acarbose standard.²¹⁹ (F) Lingampalle's group in 2019 constructed (252) through the domino reaction, in which, firstly, equal molar ratio of PA (1) and monohydrate hydrazine (56) were mixed at 80 °C for 15 min. Then, aromatic aldehydes (19) and malononitrile (48) were added and heated at 100 °C under microwave irradiation (300 W) for 10–16 min in the presence of boric acid (10 mol%) catalyst, which led to the corresponding adducts (252) with 86–94% yield.²²⁰ (G) 3-Amino-1-aryl-5,10-dioxo-1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitriles (252) were also obtained by Abdesheikhi and Karimi-Jaberi in 2015 through the domino reaction of PA, hydrazine hydrate, benzaldehydes, and malononitrile in the presence of K₂CO₃ (0.1 g) in refluxing ethanol within 50–80 min by 86–96%.²²¹ (H) Kalhor's group in 2022 obtained novel multifunctional nanocatalyst (Mn/4-MePy-IL@ZY), namely, 4-methylpyridinium chloride ionic liquid grafted on Mn@zeolite-Y, and scanned its catalytic performance (10 mol%) in the four-component synthesis of pyrazolo[1,2-*b*]phthalazines (252) under mild reaction conditions (H₂O/EtOH, 1 : 1, 80 °C) within 8–12 min with 88–98% yields.²²²

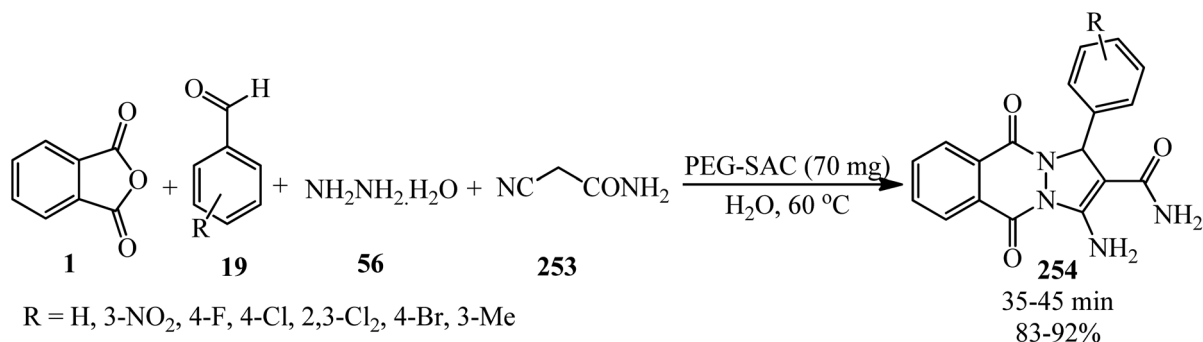
Jonnalagadda *et al.* in 2020 utilized eggshell powder (ESP, 20 mg) as a biodegradable and recyclable catalyst for the synthesis of pyrazolo-phthalazine derivatives (252) in aqueous media within 28–45 min by 93–98% at 60 °C. They performed the reaction in the presence of the substrates PA, hydrazine hydrate,

aldehydes, and activated methylene groups in a 1 : 1 : 1 : 1 molar ratio in a one-pot and one-step manner. They examined the method scope with different activated methylene compounds (malononitrile, ethyl cyanoacetate, and methyl cyanoacetate) and also several benzaldehydes and indole-3-carbaldehyde successfully.²²³

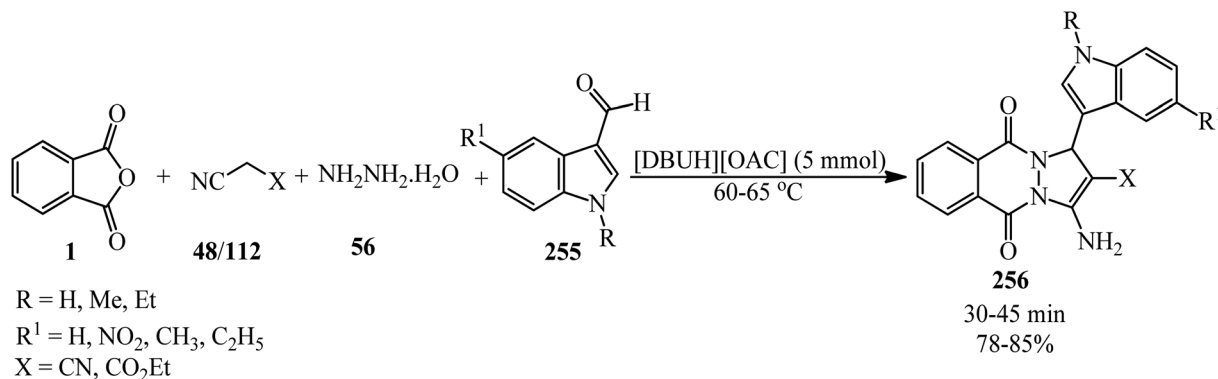
Roy's group in 2016 synthesized (252) through the domino reaction of PA (1), aromatic aldehyde (19), ethyl cyanoacetate (112), and hydrazine hydrate (56) in a 1 : 1 : 1 : 1.2 molar ratio in the presence of L-proline (10 mol%) and LiCl (5 mol%) in ethanol/water (1 : 1) media at 80 °C within 10–12 h by 87–90%. LiCl was used to elevate the yield of the reaction by elevating the electrophilicity of the aldehyde functionality.²²⁴

The Das group in 2014 prepared a novel biodegradable SO₃H-bearing carbonaceous solid catalyst (PEG-SAC, 70 mg) to achieve 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione carboxamides (254) through a domino aqueous media reaction of PA (1), hydrazine hydrate (56), (hetero)aromatic aldehydes (19), and malononitrile (48)/2-cyanoacetamide (253) at 60 °C. They also utilized aliphatic aldehydes such as isobutyraldehyde successfully. To improve the applicability of the Brønsted acid PEG-SAC catalyst, the authors also examined the preparation of multifunctionalized 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione carboxamides in a regio-controlled manner (no 1*H*-pyrazolo[1,2-*b*]phthalazine-2-carbonitriles observed) (Scheme 95).²²⁵

Raghavendra Siddaiah performed the domino synthesis of 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo



Scheme 95 Preparation of multifunctionalized 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dione carboxamides.



Scheme 96 Synthesis of 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazines.



[1,2-*b*]phthalazine derivatives (**256**) *via* the reaction of PA (**1**) with hydrazine hydrate (**56**), malononitrile (**48**)/ethyl cyanoacetate (**112**), and indole-3-carboxaldehydes (**255**) in the presence of the ionic liquid ([DBUH][OAc]) at 60–65 °C (Scheme 96).²²⁶

Reddy's group in 2014 also obtained 3-amino-1-(1*H*-indol-2-yl)-5,10-dioxo-5,10-dihydro-1*H*-pyrazolo[1,2-*b*]phthalazines (**256**) in a four-component domino reaction between hydrazine hydrate, *N*-substituted-indole-3-carboxaldehydes, malononitrile/ethyl cyanoacetate, and dialkylphthalates (methyl and ethyl) in the presence of InCl₃ (30 mol%) in refluxing ethanol within 1 h with 70–85% yield.²²⁷ Also, they utilized dialkylphthalates instead of PA; interestingly, the products (**256**) were obtained.

Shaabani's research group in 2012 developed the synthesis of structurally diverse 1*H*-pyrazolo[1,2-*b*]phthalazine-1,2-dicarboxylates (**257**) *via* a four-component reaction of PA (**1**), hydrazine hydrate (**56**), dialkyl acetylenedicarboxylates (**17**), and isocyanides (**196**) in ethanol/acetone (1 : 1) at room temperature in good to moderate yields (Scheme 97).²²⁸

Kumar's research group in 2013 presented one-pot four-component protocol for the synthesis of structurally diverse spirooxindoles, spiroannulated with chromenopyrazolophthalazines/pyranopyrazolophthalazines/indazolophthalazine (**262**, **263**, **264**) from the reaction of PA (**1**), hydrazine hydrate (**56**), isatins (**258**), and cyclic ketones [4-hydroxy-2*H*-chromen-2-one (**259**), 4-hydroxy-6-methylpyran-2-one (**260**), dimedone (**261**)] in aqueous alcoholic medium (H₂O/C₂H₅OH, 5 : 1) using catalytic amounts of sulphamic acid (SA) (Scheme 98).²²⁹ The reaction mechanism is considered to involve acid-catalyzed Knoevenagel condensation, Michael-type

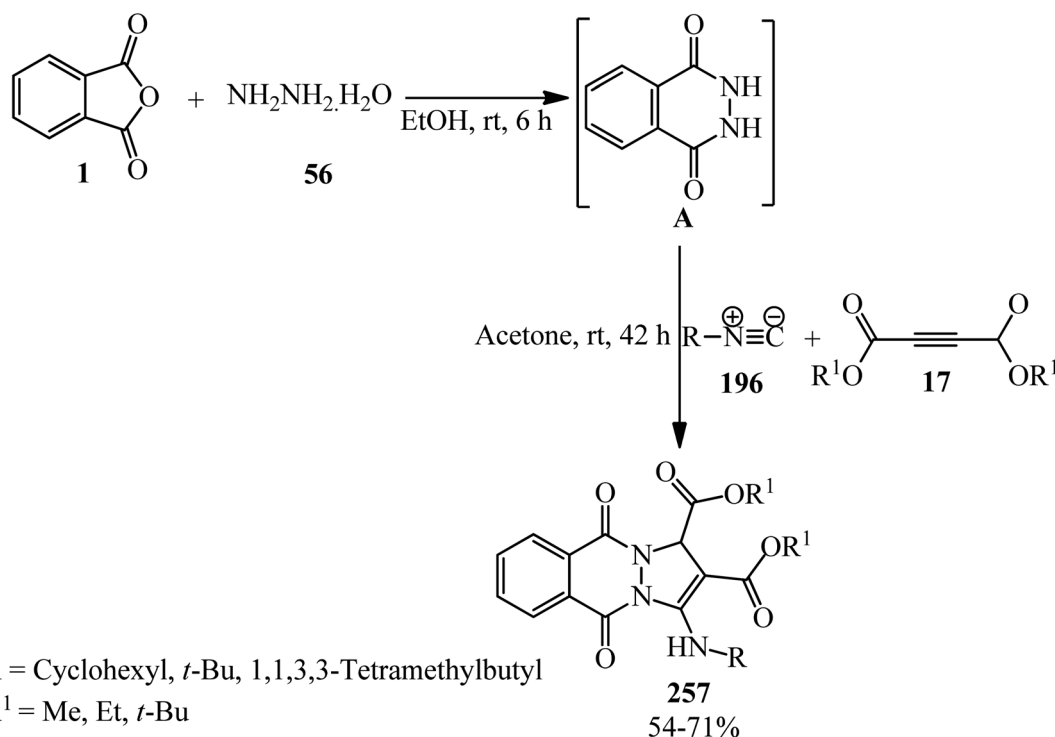
addition reaction, and consequent intramolecular dehydrative cyclization to obtain the pure products *via* recrystallization from ethanol.

Kumar's group in 2014 constructed indenopyrazolophthalazines (**266**) and pyrazolopyrimidophthalazines (**267**, **268**) *via* the reaction of PA (**1**), hydrazine hydrate (**56**), isatins (**258**), and 1,3-indandione (**265**)/barbituric acids (**266**) in the presence of deep eutectic solvent (DES, choline chloride : urea, 1 : 2) as a catalyst and reaction medium at 80 °C (Scheme 99).²³⁰ They also prepared chromenopyrazolophthalazines (**262**) and indazolophthalazine (**264**) in the same reaction conditions within 20–32 min with 88–90% and 86–93% yields, respectively.

Biabangard and Shaterian in 2015 obtained pyrazolopyrimidophthalazines (**267**) *via* the domino four-component reaction of PA (**1**), aromatic aldehydes (**19**), hydrazine hydrate (**56**), and barbituric acid (**266**), in a 1 : 1 : 1.2 : 1 molar ratio, using vitamin B1 supported on alumina (VB₁-Al₂O₃, 5 mol%) as a heterogeneous catalyst under solvent-free conditions at 70 °C within 10–14 min with 88–92% yield.²³¹ The recovered catalyst was reutilized for at least four runs without any loss of its activity.

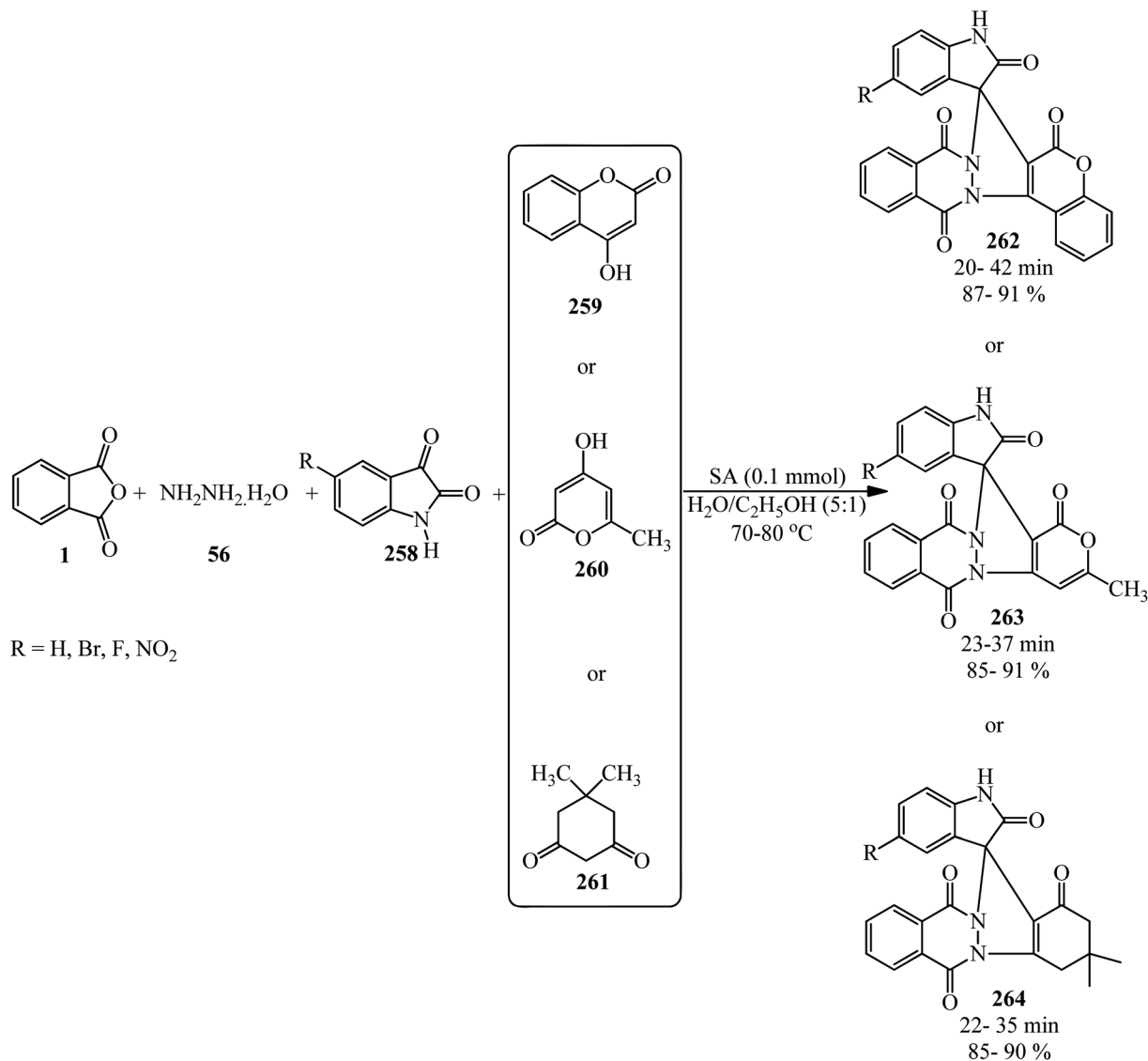
A novel catalyst obtained *via* the stabilization of methylene dipyridine nanoparticles on Fe₃O₄ (Fe₃O₄/SiO₂/propyltriethoxysilane/methylene dipyridine nanoparticles) was prepared by Sadeghzadeh and Nasserri in 2013, which was utilized to prepare pyrazolophthalazinyl spirooxindoles (**269**) *via* a four-component solvent-free reaction at room temperature (Scheme 100).²³²

Maleki and Sedigh Ashrafi in 2014 reported the synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-diones (**271**) and 1*H*-



Scheme 97 Synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-1,2-dicarboxylates.





Scheme 98 Synthesis of spirooxindoles spiroannulated with chromenopyrazolophthalazines/pyranopyrazolophthalazines/indazolophthalazine.

indazolo[1,2-*b*] phthalazine-1,6,11-triones (272) via the multi-component and one-pot reactions of PA (1), various aldehydes (19), hydrazinium hydroxide (56), and acyclic or cyclic 1,3-diketones (261, 270) using wet 2,4,6-trichlorotriazine (TCT) as a catalyst under solvent-free conditions (Scheme 101).²³³ Cyanuric acid was produced as the byproduct, which was removed by washing with water.

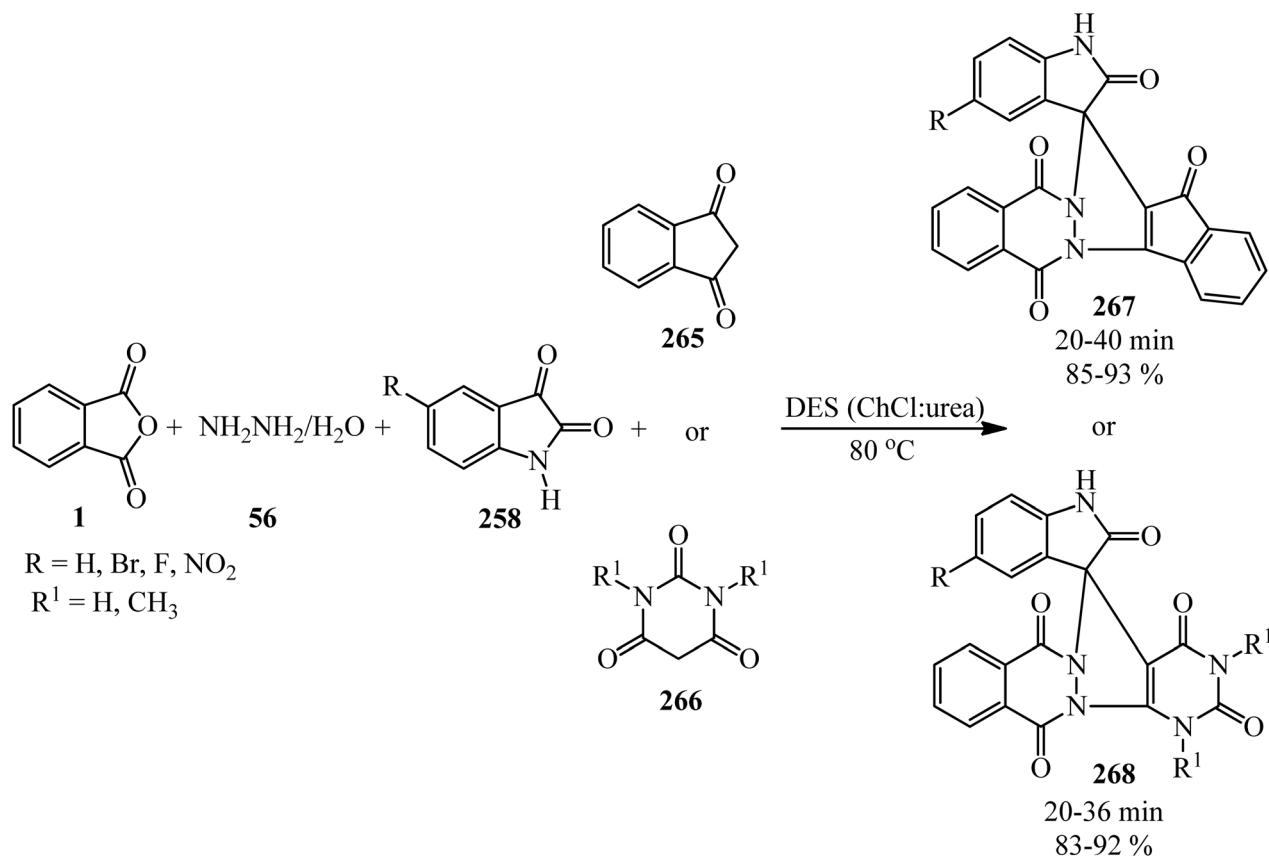
Esmailpour and Zahmatkesh *et al.* in 2017 investigated the synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-diones (271) 2*H*-indazolo[2,1-*b*]phthalazine-triones (272) using Fe₃O₄@SiO₂-imid-PMAⁿ nanoparticles (immobilization of H₃PMO₁₂O₄₀ nanoparticles (PMAⁿ) on an imidazole-functionalized Fe₃O₄@SiO₂) as an ecofriendly magnetic catalyst from the four-component condensation, solvent-free reaction of PA (1), aldehydes (19), hydrazinium hydroxide (56), 1,3-diketones (dime-done (261), 1,3-pentandion), and in 1:1:1.2:1 molar ratio under conventional thermal conditions (80 °C) or ultrasound-assisted conditions at room temperature.²³⁴ Also, to enlarge

the efficacy of the procedure, a vast range of aldehydes including aromatic, heteroaromatic, cyclic/acyclic aliphatic, and sterically-hindered candidates were used to perform the reaction successfully. They also examined the catalyst efficacy to prepare some new kinds of pyrazolo[1,2-*b*]phthalazine-5,10-diones (271) through the reaction of PA (1), aldehydes (19), hydrazine hydrate (56), and 1,3-diphenylpropanedione (273) (Scheme 102).

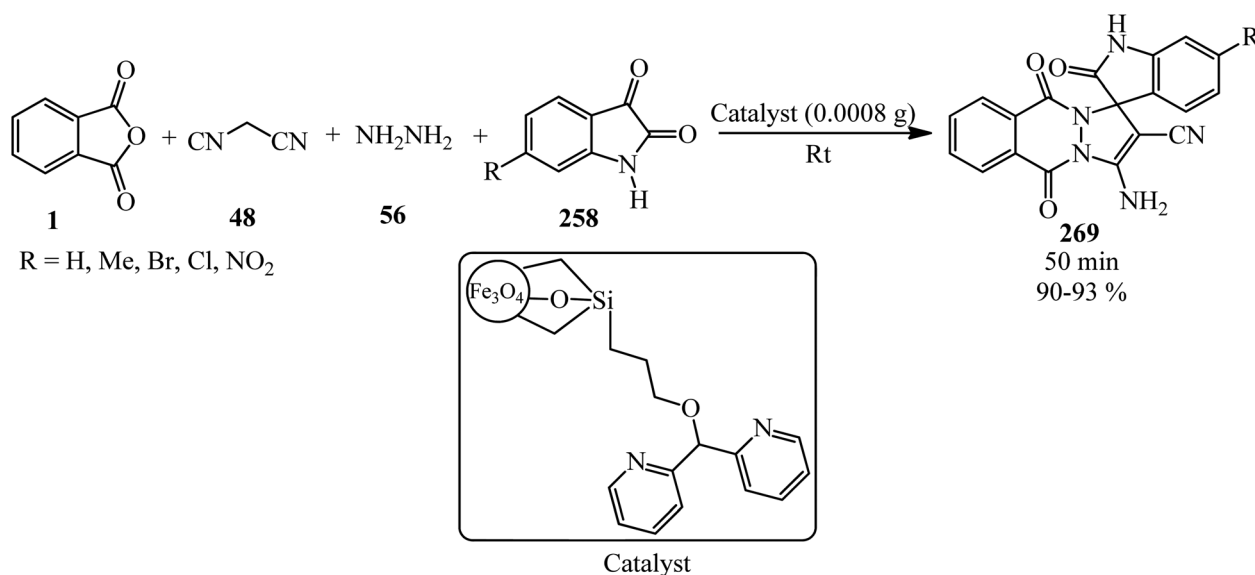
Safaei-Ghomi *et al.* in 2016 described the synthesis of some 1*H*-pyrazolo[1,2-*b*]phthalazine-diones (271) and 2*H*-indazolo[2,1-*b*]phthalazine-triones (272) via the domino reaction of PA (1), aromatic aldehydes (19), hydrazine monohydrate (56), and malononitrile (48)/ethyl cyanoacetate (112)/dime-done (261), respectively. The reaction proceeded in the presence of CuFe₂O₄ and ZrP₂O₇ nanocatalysts under solvent-free conditions (Scheme 103).²³⁵

Besides the abovementioned examples, many other research groups prepared different kinds of 2*H*-indazolo[2,1-*b*]





Scheme 99 Construction of indenopyrazolophthalazines and pyrazolopyrimidophthalazines.

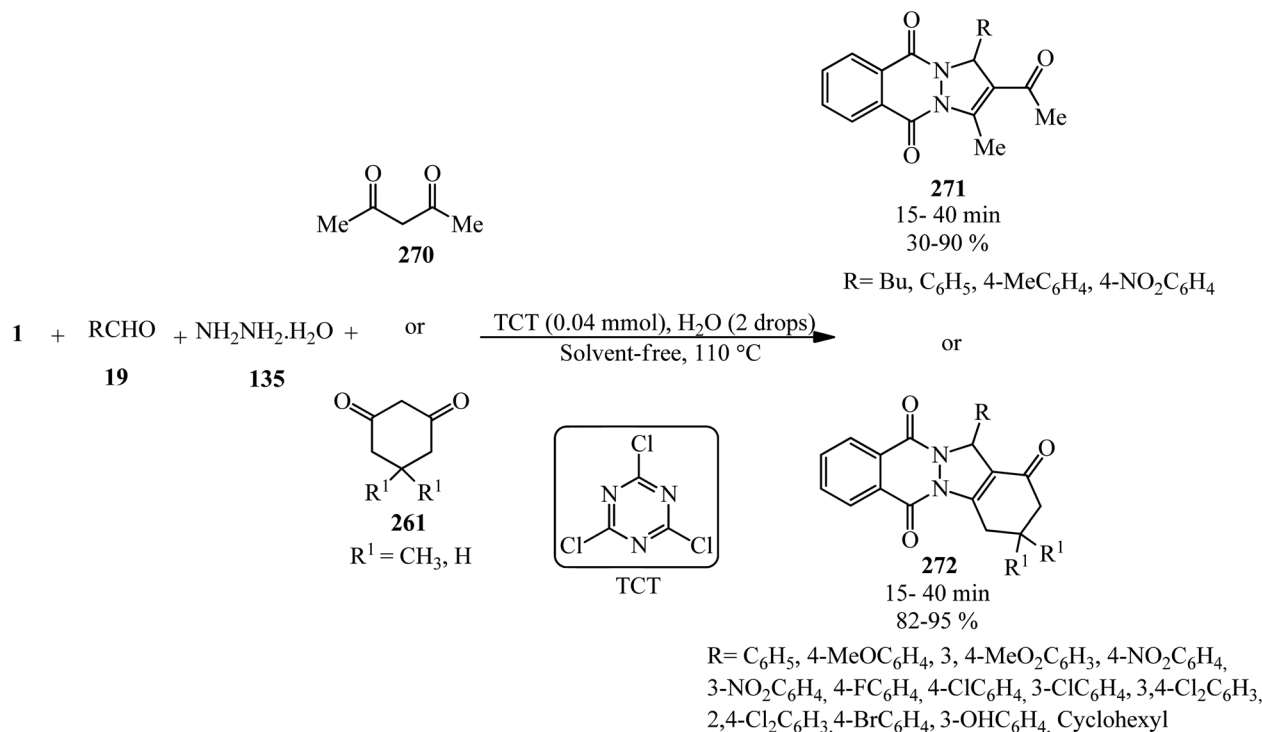


Scheme 100 Preparation of pyrazolophthalazinyli spirooxindoles.

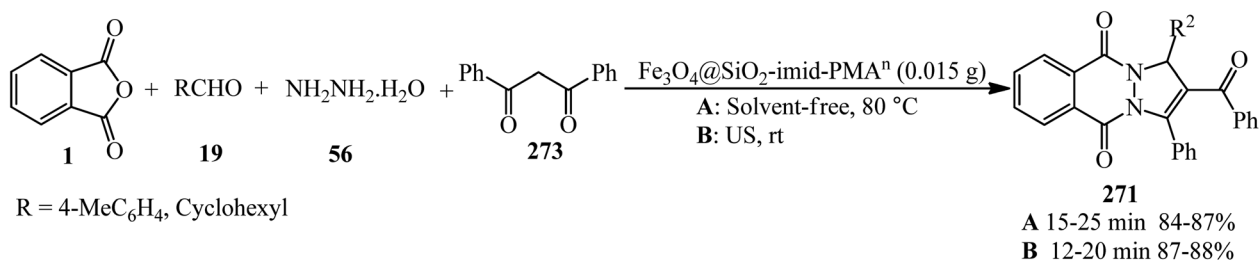
phthalazine-triones (272) via a four-component reaction of PA (1), different aldehydes (19), hydrazine hydrate (56), and dimesitylindane (261). (A) The Javidi and Esmaeilpour group in 2016 synthesized dendrimer-encapsulated phosphotungstic acid nanoparticles immobilized on nanosilica with surface amino

groups (dendrimer-PWAⁿ) and utilized it as a reusable catalyst to accelerate the green synthesis of (272) via a one-pot, one-step, and four-component condensation reaction of PA (1), aldehydes (19), hydrazinium hydroxide (56), and dimesitylindane (261) in a 1.2 : 1 : 1 : 1 molar ratio at both under solvent-free conditions at 80 °





Scheme 101 Synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-diones and 1H-indazolo[1,2-b] phthalazine-1,6,11-triones.

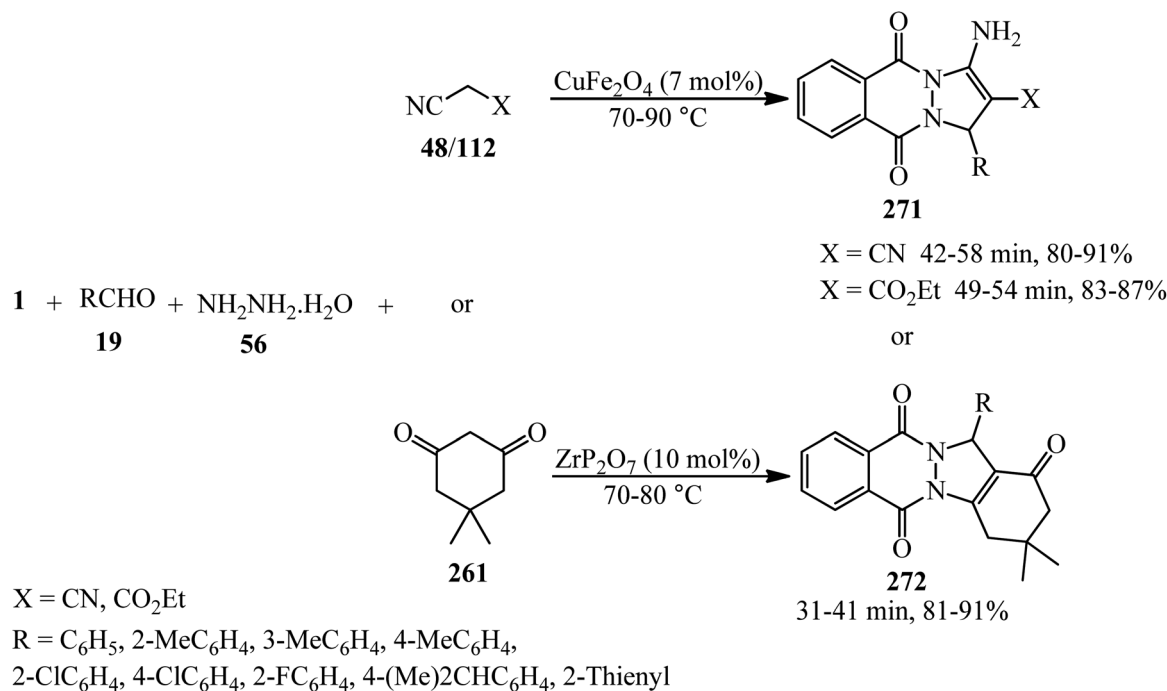


Scheme 102 Pyrazolo[1,2-b]phthalazine-5,10-diones obtained from PA, aldehydes, hydrazine hydrate, and 1,3-diphenylpropanedione.

C and ultrasound irradiation at room temperature.²³⁶ The protocol was successful for different classes of aldehydes such as aromatic, cyclic/linear aliphatic, heteroaromatics, and sterically-hindered candidates. An example of their procedure has been demonstrated in Scheme 104 (entry 1). In addition, the catalyst could be successfully recycled and reused at least for six runs without a significant loss in activity. (B) Shaterian and Rigi in 2014 prepared (272) using cellulose-SO₃H (4 mol%) as a solid-acidic reusable catalyst *via* the domino four-component condensation reaction under thermal (80 °C) solvent-free conditions (Scheme 104, entry 2).²³⁷ The protocol was unsuccessful in utilizing aliphatic aldehydes such as heptanal and octanal. (C) Mosaddegh and Hassankhani in 2011 described an efficient protocol for the one-pot, one-step synthesis of 2H-indazolo[2,1-b]phthalazine-1,6,11(13H)-triones (272) under thermal (125 °C) solvent-free conditions using PA (1), aromatic aldehydes (19), hydrazinium hydroxide (56), and dimesone (261) in a 1:1:1.2:1 molar ratio in the presence of

Ce(SO₄)₂·4H₂O (2.5 mol%) within 5–10 min (Scheme 104, entry 3).²³⁸ (D) Shaterian and Rigi in 2011 obtained (272) *via* the domino four-component condensation reaction from (1), aromatic aldehydes (19), (56), and dimesone (261) in a 1:1:1.2:1 molar ratio using 0.8 mol% starch sulfate as a bio-supported nonhygroscopic solid-acid catalyst under thermal (80 °C) solvent-free conditions within 4–15 min period with 72–92% yield (Scheme 104, entry 4).²³⁹ The recovered catalyst was reused four times without any loss of its activities. (E) The compounds (272) were achieved by Hasaninejed *et al.* in 2012 through the four-component one-step condensation reaction of PA (1), aromatic aldehydes (19), hydrazinium hydroxide (56), and 1,3-cyclohexanedione/dimesone in a 1:1:1.12:1 molar ratio at 80 °C using sulfuric acid-modified PEG-6000 (PEG-OSO₃H, 8 mol%) as a green, recyclable, and biodegradable polymeric catalyst, with the yield of 80–93% in 10–20 min (Scheme 104, entry 5).²⁴⁰ The authors calculated the environmental impact factor (E-factor)²⁴¹ to evaluate the minimizing





Scheme 103 Synthesis of 1*H*-pyrazolo[1,2-*b*]phthalazine-diones and 2*H*-indazolo[2,1-*b*]phthalazine-triones.

environmental impacts of their method. (F) Shaterian and Aghakhanizadeh in 2012 obtained (272) from the domino four-component solvent-free reaction of PA (1), arylaldehydes (19), hydrazine monohydrate (56), and dimedone/1,3-cyclohexanedione in a 1.2 : 1 : 1.4 : 1 molar ratio using reusable ionic liquids [including A: 2-pyrrolidonium hydrogen sulfate ([Hnhp][HSO₄]); B: (4-sulfobutyl)tris(4-sulfophenyl)phosphonium hydrogen sulfate; and C: triphenyl(propyl-3-sulphonyl)phosphonium toluenesulfonate] as acidic reusable catalysts (5 mol%). The catalysts recovered within 5 runs without significant activity loss. The time and yield of utilizing these 3 ILs at 80 °C are A: 8–11 min/70–90%; B: 4–8 min/83–92%; and C: 6–15 min/80–93%, respectively (Scheme 104, entry 6).²⁴² (G) Shekouhy and Hasaninejad in 2012 obtained (272) via the one-pot and one-step four-component reaction of PA (1), (hetero)aromatic aldehydes (19), hydrazine monohydrate (56), and dimedone (261) in a 1 : 1 : 1.2 : 1 molar ratio with a yield of 89–95% in 3–15 min using 1-butyl-3-methyl imidazolium bromide ([Bmim]Br, 0.5 g) as a neutral IL under ultrasonic irradiation and catalyst-free conditions at room temperature (Scheme 104, entry 7).²⁴³ (H) Veisi and Aminimanesh *et al.* in 2014 synthesized (272) in a one-pot and one-step reaction of PA (1), aliphatic/aromatic aldehydes (19), hydrazine monohydrate (56), and dimedone (261) in a 1 : 1 : 1.1 : 1 molar ratio in the presence of mildly basic ionic liquid *N,N,N,N*-tetramethylguanidinium acetate ([TMG][Ac], 10 mol%) in high yields (55–98%) at 80 °C within 10–30 min (Scheme 104, entry 8).²⁴⁴ (I) Habibi and Shamsian in 2015 utilized new reusable acid–base bifunctional ionic liquid, 1,4-dimethyl(4-sulfobutyl)piperazinium hydrogen sulfate ([DMSBP][HSO₄], 3 mol%), for the synthesis of (272) via the one-step four-component reaction of reaction of

PA (1), aromatic aldehydes (19), hydrazine monohydrate (56), and dimedone (261) in a 1 : 1 : 1.2 : 1 molar ratio with 85–94% yield in 3–15 min (Scheme 104, entry 9).²⁴⁵ Examining the reaction with octanal as an aliphatic aldehyde, it did not form the corresponding product. The catalyst was recovered from the aqueous medium, dried under vacuum, and reused in five successive runs without a substantial loss of activity. (J) The compounds (272) were also prepared by Ebrahimipour *et al.* in 2015 via the one-pot and one-step, and four-component reaction in the presence of three octahedral complexes containing 2-pyrazinecarboxylate (pzca), including [Ni(pzca)₂(H₂O)₂] (25 mol%), [Co(pzca)₂(H₂O)₂] (20 mol%), and [Cu(pzca)₂(H₂O)₂] (20 mol%) in acetic acid solvent at 50 °C within 10–40 min with 80–94% yields (Scheme 104, entry 10).²⁴⁶ (K) Bhosale *et al.* in 2017 developed the synthesis of (272) in the presence cesium chloride (15 mol%) in refluxing ethanol at 60 °C via the equimolar amounts of the substrates in a one-step and one-pot manner, in 3–4 h duration with a yield of 70–82% (Scheme 104, entry 11).²⁴⁷ (L) Gill's research group in 2017 demonstrated a synthetic route for 2*H*-indazolo[2,1-*b*]phthalazinetrione derivatives (272) through a one-pot, one-step four-component protocol in the presence of β-cyclodextrin as a supramolecular, biodegradable, and reusable catalyst in 80 °C aqueous media within 94–79% yield during 35–25 min (Scheme 104, entry 12).²⁴⁸ (M) Ebrahimipour and coworkers in 2017 reported the synthesis of (272) with a yield of 84–94% in 15–25 min in the presence of imidazole 2-acetamido-*N'*-(3-methoxy-2-oxidobenzylidene)benzohydrazonate-nickel(II) [Ni(L)(imi)], which is a tridentate Schiff base ligand (Scheme 104, entry 13).²⁴⁹ The Ni(II) complex showed promising antimicrobial activities against some



Entry	Conditions	R	Time (min)/yield (%)	Reference
1	Dendrimer-PWA ⁿ (0.14 mol%), solvent-free, 80 °C	<i>n</i> -Hexyl	30 min, 77%	[236]
	Dendrimer-PWA ⁿ (0.21 mol%), US, rt		25 min, 73%	
2	Cellulose-SO ₃ H (4 mol%), solvent-free, 80 °C	3-OMe-4-OHC ₆ H ₃	10 min, 78%	[237]
3	Ce(SO ₄) ₂ ·4H ₂ O (2.5 mol%), solvent-free, 125 °C	C ₆ H ₅	6 min, 78%	[238]
4	Starch sulfate (0.8 mol%), solvent-free, 80 °C	C ₆ H ₅	7 min, 89%	[239]
5	PEG-OSO ₃ H (8 mol%), solvent-free, 80 °C	2-Naphthyl	10 min, 90%	[240]
6	A or B or C (5 mol%), solvent-free, 80 °C	C ₆ H ₅	A: 7 min, 88%	[242]
			B: 6 min, 93%	
			C: 8 min, 90%	
7	[Bmim]Br (0.5 g), rt	C ₆ H ₅	10 min, 93%	[243]
8	[TMG][Ac] (10 mol%), 80 °C	C ₆ H ₅	15 min, 92%	[244]
9	[DMSBP][HSO ₄] (3 mol%), rt	C ₆ H ₅	7 min, 92%	[245]
10	[M(pzca) ₂ (H ₂ O) ₂], AcOH, 50 °C	C ₆ H ₅	A: 20 min, 93%	[246]
			B: 15 min, 94%	
			C: 30 min, 92%	
11	CsCl (15 mol%), EtOH, 60 °C	C ₆ H ₅	3 h, 70%	[247]
12	β-CD (15 mol%), H ₂ O, 80 °C	C ₆ H ₅	25 min, 90%	[248]
13	[Ni(L)(imi)] (25 mol%), AcOH, 50 °C	C ₆ H ₅	20 min, 94%	[249]
14	γ-Al ₂ O ₃ /BF ₃ /Fe ₃ O ₄ (8 mg), solvent-free, 80 °C	4-NO ₂ C ₆ H ₄	12 min, 95%	[250]
15	Fe ₃ O ₄ /CS-Alg/Ag NPs (0.3 mol%), solvent-free, 80 °C	C ₆ H ₅	30 min, 92%	[251]
16	Cell-DABCO-Cu (0.01 g), solvent-free, 80 °C	C ₆ H ₅	8 min, 89%	[252]
17	[TEAH][OAc] (0.2 mmol), solvent-free, 80 °C	C ₆ H ₅	13 min, 91%	[253]

Scheme 104 Investigation of the scope of the protocols to obtain 2*H*-indazolo[2,1-*b*]phthalazine-triones.

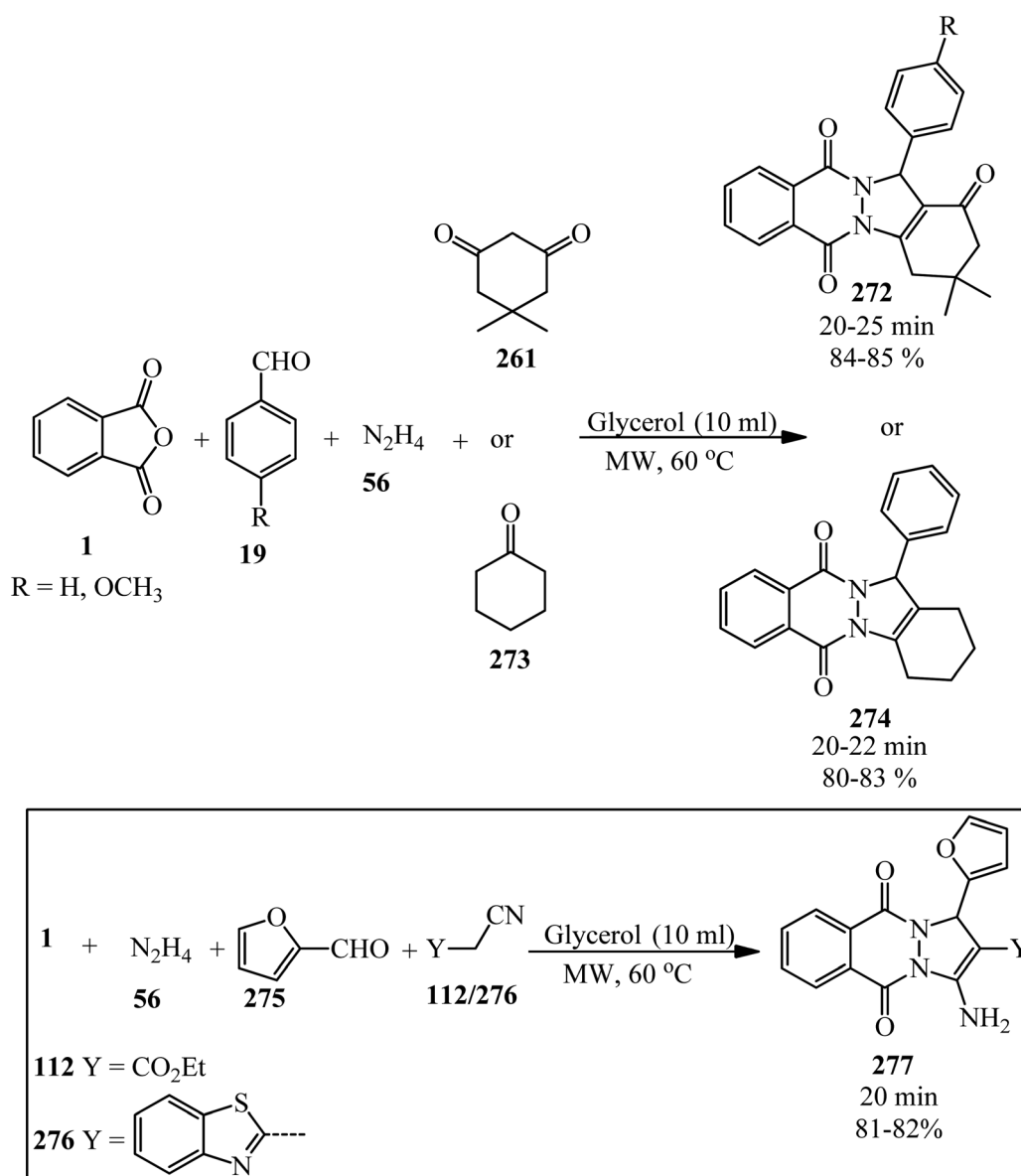
Gram-negative and Gram-positive bacteria such as *E. coli*, *S. aureus*, *P. aeruginosa*, and *B. cereus*. The catalyst was recovered by centrifugation and recrystallization from a mixture of

ethanol and water and reused at least three times with satisfactory results. (N) Bamoniri *et al.* in 2020 described a method for the synthesis of 2*H*-Indazolo[2,1-*b*]phthalazinetrione



derivatives (**272**) using nano $\gamma\text{-Al}_2\text{O}_3/\text{BF}_n/\text{Fe}_3\text{O}_4$ (8 mg) *via* the one-step reaction of equimolar of four substrates under solvent-free conditions, which yielded the corresponding adducts with 90–97% yield in 10–17 min (Scheme 104, entry 14).²⁵⁰ The catalyst was recycled by washing with CH_2Cl_2 , drying at 50 °C under vacuum for 1 h, and reused within 5 runs successfully. (O) Liu and Li *et al.* in 2020 prepared a novel modified core-shell magnetic nanocomposite by anchoring Ag NPs on magnetite core that was coated with chitosan-alginate dual biopolysaccharide ($\text{Fe}_3\text{O}_4/\text{CS}\text{-Alg}/\text{Ag}$ NPs). The novel nanocatalyst was utilized for the synthesis of (**272**) *via* one-step and one-pot reaction of the corresponding substrates at 80 °C within 0.2–1.5 h with 50–95% yield. The catalyst demonstrated human lung protective effects against α -Guttiferin. These events revealed that the catalyst suppressed Guttiferin-induced cell death in

a dose-dependent manner in lung MRC-5, CCD-19Lu, WI-38, and BEAS-2B cell lines. The catalyst was recovered easily using an external magnet and recycled for 10 successive times with minimal reduction in activity (Scheme 104, entry 15).²⁵¹ (P) In 2022, Naeimi and Zahedifar reported an immobilized copper(II) complex on microcellulose (cell-DABCO-Cu) as a novel catalyst to promote the preparation of (**272**) at 80 °C under solvent-free conditions in a one-step and one-pot manner (Scheme 104, entry 16).²⁵² In addition, the CuO nanoparticles with an average size of 40 nm were obtained by the direct calcination of Cell-DABCO-Cu. (Q) Mahmoodi *et al.* in 2020 obtained (**272**) *via* the one-pot, one-step, four-component, and solvent-free condensation of PA (**1**), aromatic aldehydes (**19**), hydrazine hydrate (**56**), and dimedone (**261**), which was accelerated with triethanolammonium acetate ([TEAH][OAc]) at 80 °C within 10–



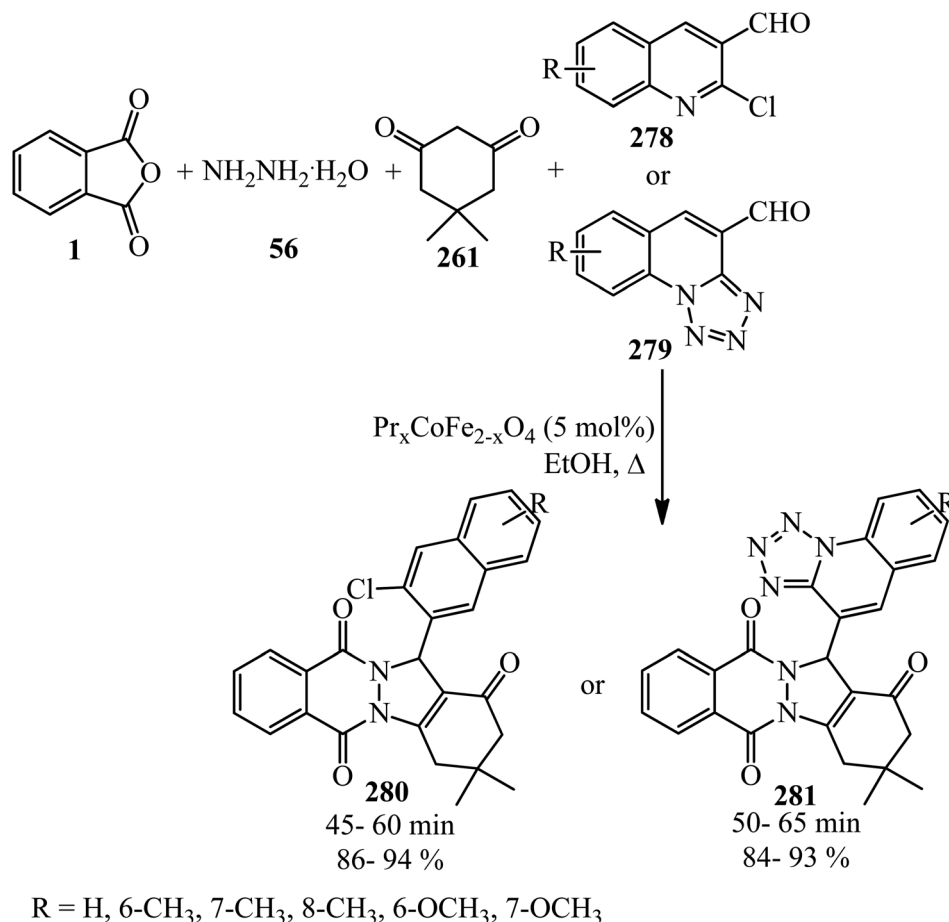
Scheme 105 Synthesis of 2H-indazolo[2,1-b]phthalazine-triones, 3,4-dihydro-1H-indazolo[1,2-b]phthalazine-6,11(2H,13H)-diones, and 3-amino-2-(benzo[d]thiazol-2-yl)-1-(furan-2-yl)-1H-pyrazolo[1,2-b]phthalazine-5,10-diones.



35 min with 89–95% yield (Scheme 104, entry 17).²⁵³ The authors examined the efficacy of the protocol through utilizing succinic anhydride instead of PA. They also prepared phthalazine-diones (271) by the reaction of PA (1), aromatic aldehydes (19), hydrazine hydrate (56), and alkyl cyanoacetates (112) (methyl and ethyl) successfully in the same reaction conditions within 22–37 min with 80–95%. (R) Tigote's group in 2017 used ZnFe_2O_4 nanoparticles (1.5 mol%) as a catalyst for this transformation *via* the one-pot and two-step reaction of aromatic aldehyde (19) and the dimedone (261), which was added into the mixture of PA (1) and hydrazine hydrate (56) in a 1 : 1 : 1 : 1.2 molar ratio at room temperature within 30–90 min with 76–89% yield.²⁵⁴ (S) Mahmoodi's group in 2020 used tetrabutyl phosphonium sulfate ($[\text{TBP}]_2\text{SO}_4$, 5 mol%) as a novel room-temperature ionic liquid (RTIL) to prepare (272) *via* the domino reaction of the mentioned four substrates at room temperature within 10–20 min with 89–92% yield. The authors examined the scope of the protocol utilizing succinic anhydride instead of PA. They also achieved phthalazine-diones through the reaction of phthalazine-diones (271) and the reaction of PA (1), aromatic aldehydes (19), hydrazine hydrate (56), and alkyl cyanoacetates (112) (methyl and ethyl) successfully in the same reaction conditions within 15–45 min with 65–92% yield.²⁵⁵

Some research groups utilized 1,3-cyclohexanedione (273) (in addition with dimedone) to prepare (272) and 3,4-dihydro-1*H*-indazolo[1,2-*b*]phthalazine-6,11(2*H*,13*H*)-diones (274), which consist of (A). Sadek *et al.* in 2019 developed a highly efficient, catalyst-free, one-pot, multicomponent synthesis of various indazolophthalazines (272, 274) in glycerol as a cheap, biodegradable, and commercially available promoting solvent and catalyst under controlled microwave heating (Scheme 105).²⁵⁶ The reaction mixture in glycerol was heated under reflux in a Milestone microwave lab station at 60 °C. They also synthesized (277) *via* the same reaction conditions.

(B) Pal's group in 2016 prepared the compounds (272, 274) *via* the one-pot four-component reaction of PA (1), variety of aldehydes (19) (including aliphatic and (hetero)aromatic aldehydes), hydrazine hydrate (56), and active methylene compounds (dimedone (261)/1,3-cyclohexanedione (273)) in the presence of magnetic Fe_3O_4 -glutathione core-shell (nano-FGT, 10 mg) at 80 °C under solvent-free conditions within 20–30 min with 87–97%. The magnetic catalyst could be separated easily by an external magnet and reused in five more consecutive runs without much decrease in the catalytic activities. The authors also examined the efficacy of the catalyst in the synthesis of two kinds of phthalazine-diones (271)



Scheme 106 Synthesis of phthalazine-quinolines.



via the reaction of PA (**1**), aldehydes (**19**) (which are 4-chloro-3-nitrobenzaldehyde and pentanal), aminonitriles/ethyl cyanoacetate (**112**), and hydrazine hydrate (**56**) in a 1 : 1 : 1 : 1.2 molar ratio in the same reaction conditions within 20–25 min with 92–97%.²⁵⁷

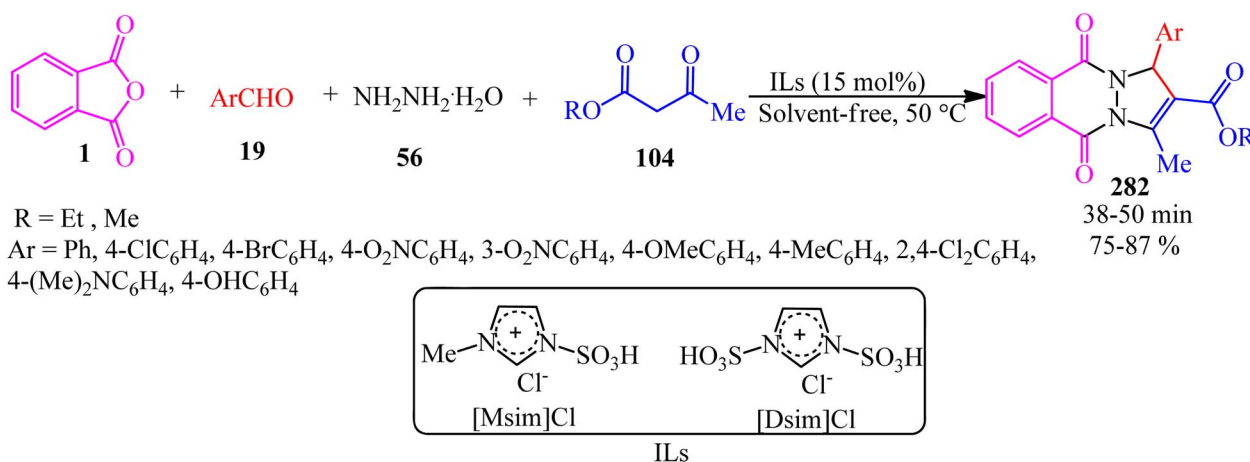
Zaheer *et al.* in 2016 exploited $\text{Pr}_x\text{CoFe}_{2-x}\text{O}_4$ ($x = 0.1$) nanoparticles to catalyze the efficient one-pot four-component reaction of PA (**1**), hydrazine hydrate (**56**), dimedone (**261**), and various quinoline aldehydes (**278**, **279**) to achieve phthalazine quinoline derivatives (**280**, **281**) (Scheme 106).²⁵⁸ The synthesized adducts were evaluated for antibiofilm activity against *P. aeruginosa* and *C. albicans*.

Zahedifar's research group in 2018 constructed a series of novel pyrazolo[1,2-*b*]phthalazine-2-carboxylates (**282**) via the one-pot four-component reaction of PA (**1**), aromatic aldehydes (**19**), hydrazine hydrate (**56**), and alkyl acetoacetates (**104**) in the presence of acidic ionic liquids such as 3-methyl-1-sulfo-1*H*-imidazol-3-ium chloride ([Msim]Cl), 1,3-disulfo-1*H*-imidazol-3-ium chloride ([Dsim]Cl), and triethyl(sulfo)ammonium chloride ([Et₃NSO₃H]Cl) as the catalyst (Scheme 107).²⁵⁹

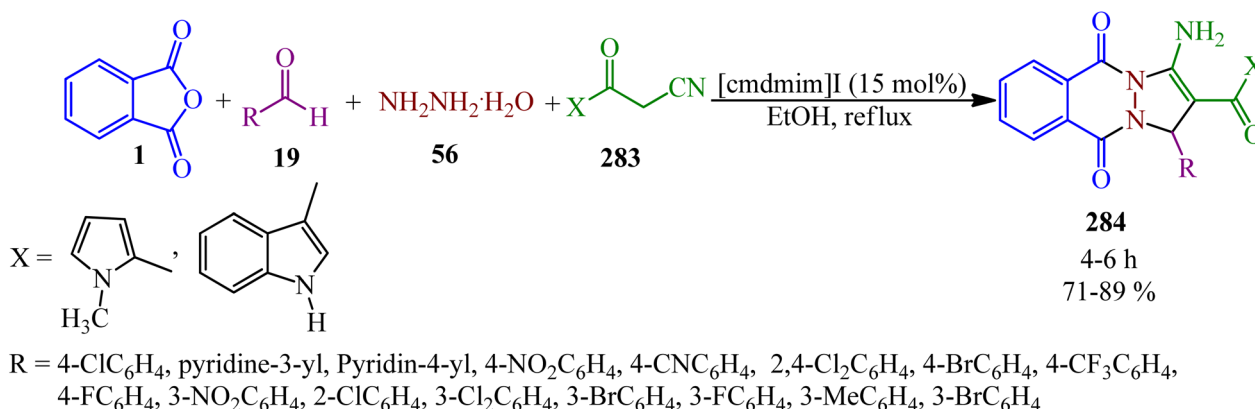
Jahanshahi and Mamaghani in 2019 described the synthesis of a wide range of novel 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-

dioness (**284**) in the presence of acetic acid-functionalized imidazolium salt (1-carboxymethyl-2,3-dimethylimidazolium iodide, [cmdmim]I), as a newly synthesized Brønsted acid catalyst. The reaction occurred via tandem Knoevenagel cyclocondensation of PA (**1**), aromatic aldehydes (**19**), hydrazine hydrate (**56**), and 3-(1-methyl-1*H*pyrrol-2-yl)-3-oxopropanenitrile (or 3-(1*H*-indol-3-yl)-3-oxopropanenitrile) (**283**) in the presence of [cmdmim]I in ethanol under reflux conditions (Scheme 108).²⁶⁰ The reusability of the catalyst was examined in 5 consecutive runs without any appreciable decrease in the activity.

In 2022, Vedula and coworkers provided alkyl/aralkyl/phenacyl thiotriazolyl isoindoline-1,3-diones (**287**) via the reaction of PA (**1**), hydrazine hydrate (**56**), dipotassium cyanodithioimidocarbonate salt (**285**), and alkyl/aralkyl/phenacyl bromides (**286**) in a 1 : 1.5 : 1 : 1 molar ratio using acetic acid and sodium acetate (0.1 mmol) via a one-pot four-component reaction at 80 °C (Scheme 109). The *in vitro* anticancer activity of the products revealed that some of them demonstrated cytotoxic assay against HeLa cancer cell lines. The compounds also subjected to their docking analysis and DFT calculations.²⁶¹

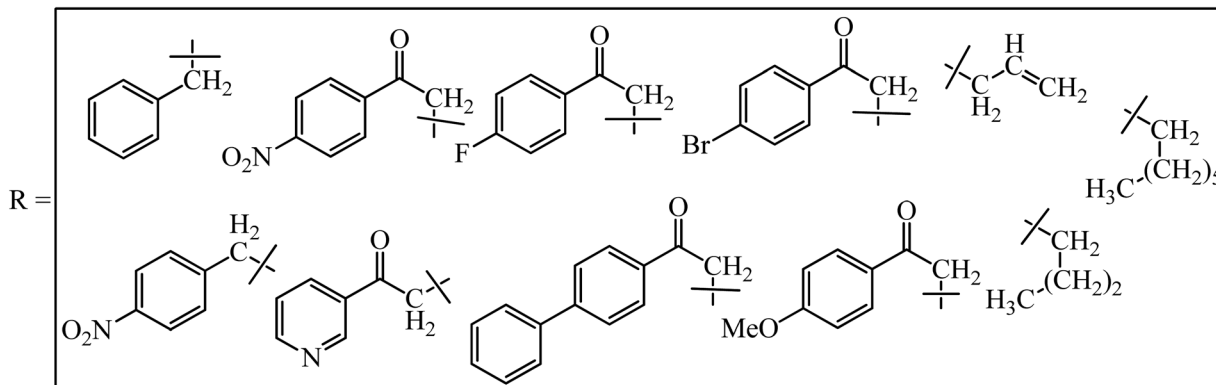
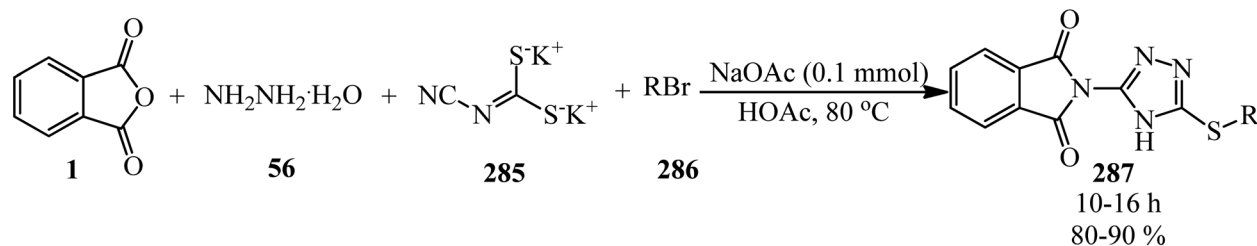


Scheme 107 Synthesis of pyrazolo[1,2-*b*]phthalazine-2-carboxylates.



Scheme 108 Synthesis of novel 1*H*-pyrazolo[1,2-*b*]phthalazine-5,10-dioness by [cmdmim]I.





Scheme 109 Synthesis of alkyl/aralkyl/phenacyl thiotriazolyl isoindoline-1,3-diones.

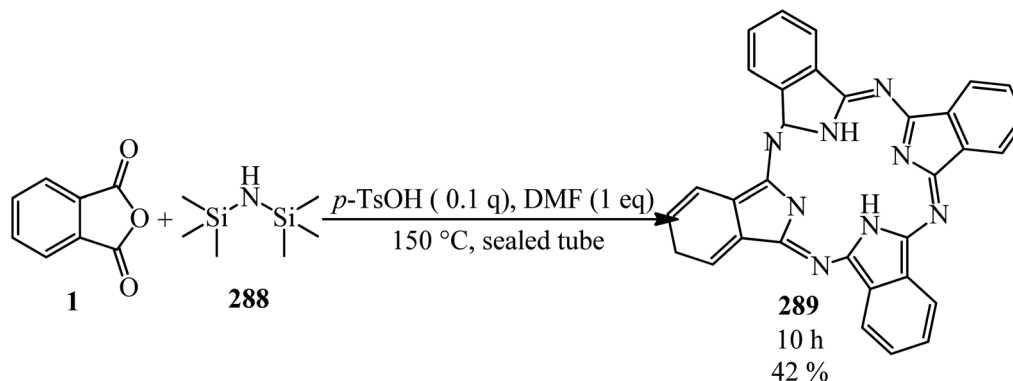
5. Applications of PA in five-component and higher-component reactions

Toru's group in 2003 reported the preparation of metal-free phthalocyanines from PA (1) with hexamethyldisilazane (HMDS) (288) in a 1 : 5 molar ratio (Scheme 110).²⁶² The transformation is actually a pseudo eight-component reaction.

Park *et al.* in 2004 developed a green procedure for the synthesis of metal phthalocyanines (291) (MPC: M = Cu, Mn, Al, Co, and Zn) through the condensation reaction of PA (1) (0.283 mol) and urea (103) (0.816 mol) with various metal chlorides (290) (7 g) in the presence of ammonium molybdate (5×10^{-4} mol) as a catalyst under microwave-assisted solvent-free conditions (Scheme 111).²⁶³ The procedure actually is a pseudo nine-component reaction.

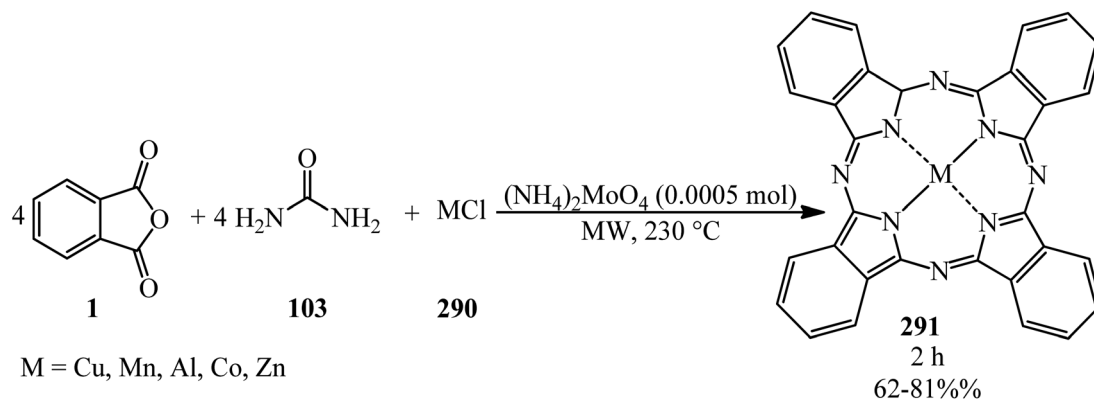
6. Applications of PA in esterification

Ahmad *et al.* in 2010 reported immobilized *Candida Antarctica* lipase, Novozym 435, as a biocatalyst to catalyze the esterification reaction of PA (1) and betulinic acid (292) to obtain 3-*O*-phthalyl betulinic acid (293) in *n*-hexane/chloroform. The effect of different parameters in the reaction process was predicted by the “response surface methodology” (RSM) technique. The comparison of the results of this model and experimental values revealed a good correspondence. The effect of enzyme amount was found to be less, while the reaction temperature, reaction time, and molar ratio strongly affected the ester yield. According to the “central composite rotatable design” (CCRD) optimization, in the presence of a betulinic acid to phthalic anhydride in a 1 : 1.11 molar ratio, the maximal yield of the ester (64.7%) was obtained using 145.6 mg enzyme at 53.9 °C in 20.3 h. This

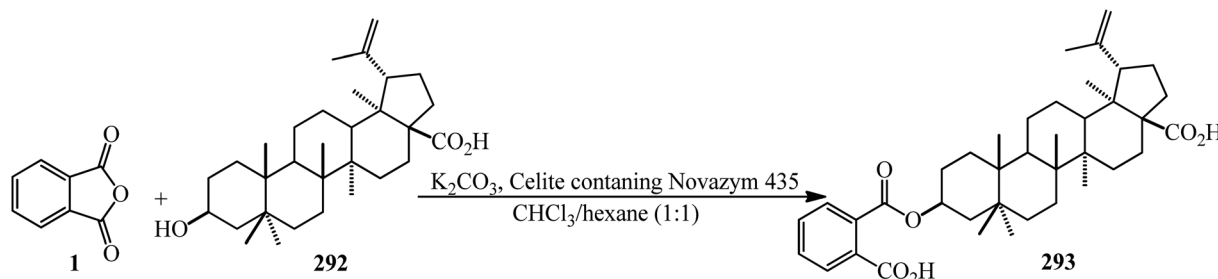


Scheme 110 Synthesis of phthalocyanines.





Scheme 111 Synthesis of metal phthalocyanines.



Scheme 112 Synthesis of 3-O-phthalyl betulinic acid.

predicted that the optimum conditions are in close correlation with the experimental results (Scheme 112).²⁶⁴

Ahmad's research group in 2010 also considered the reaction parameters of lipase-catalyzed esterification of betulinic acid (292) using PA (1) in organic solvent media.²⁶⁵ The lipase from *Candida antarctica* immobilized on an acrylic resin (Novozym 435) was employed for esterification. The influence of different reaction parameters, such as effect of single and mixed solvents, substrate molar ratio, reaction time, temperature, amount of enzyme, effect of inorganic bases, and effect of substrate support were investigated and optimized. The optimum conditions to obtain 3-O-phthalyl-betulinic acid (293) (61.8%) are: substrate molar ratio (betulinic acid : phthalic anhydride, 1 : 1), within 24 h at 55 °C, enzyme (176 mg), and Celite (170 mg) in 1 : 1 mixture of chloroform and *n*-hexane as a solvent in the presence of K₂CO₃ (as an inorganic base).

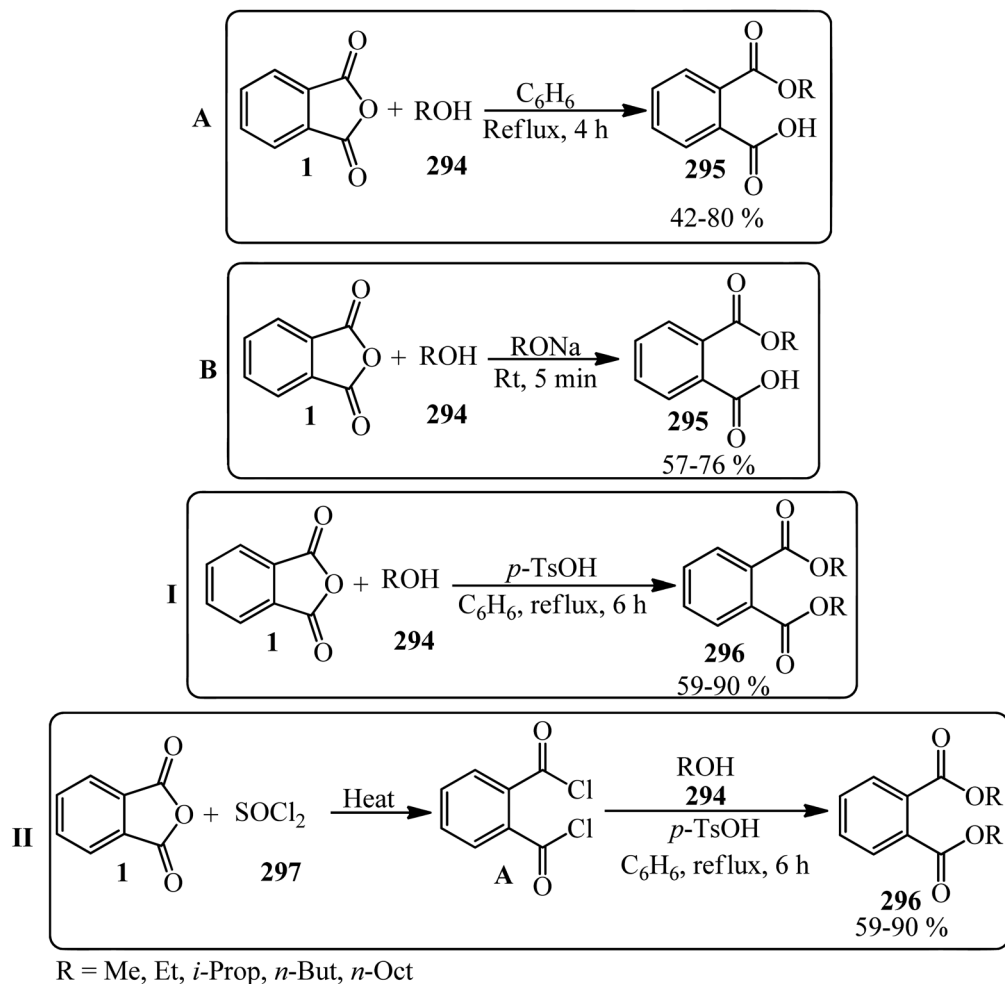
Dubey's research group in 1997 converted PA (1) to its monoesters through the reaction with simple alcohols (294) under a variety of conditions (Scheme 113).²⁶⁶ The monoesters (295) could be prepared *via* two procedures: (A) reaction of PA and alcohols in a 1 : 2.2 molar ratio, in refluxing benzene, (B) reaction of PA with alkyl alcohols in 1 : 3.8 molar ratio in the presence of sodium alkoxide (0.02 g) at room temperature with 5 min. The diesters (296) were also obtained *via* various procedures: (I) reaction of PA and alcohols in 1 : 3.2 molar ratio in benzene solvent in the presence of *p*-TSA (5 mg) under reflux conditions in a dean-stark apparatus within 6 h; (II) the reaction of PA with thionyl chloride (297) for 1 h under reflux conditions,

which was followed by separating thionyl chloride, and consequent addition of pyridine and an appropriate alcohol and refluxing for 3 h.

Fareghi-Alamdari and coworkers in 2017 manufactured two functionalized diacidic ionic liquids (FDAILs) including hydroxyl functionalized diacidic IL [3,3'-(2,2-bis(hydroxymethyl)propane-1,3-diyl)bis(1-methyl-1*H*-imidazole-3-ium) hydrogen sulfate, HFDAIL] and sulfonated diacidic IL [3,3'-(2,2-bis((sulfoxy)methyl)propane-1,3-diyl)bis(1-methyl-1*H*-imidazole-3-ium)bromide, SFDAIL] by a simple method in high yields, which served for the esterification reaction of anhydrides (PA, succinic anhydride, and maleic anhydride) with some alcohols in a 1 : 5 molar ratio to give corresponding dialkyl esters (296) as plasticizers under solvent-free conditions.²⁶⁷ HFDAIL showed higher catalytic performance in comparison with other reported catalysts, suggesting its high acidity and hydrophilic property. Phthalates, which are a class of neutral plasticizers. Are widely used in nonmilitary industries. Recycling experiments suggested that ILs could be reused seven times without a remarkable loss in their catalytic activity. According to the proposed mechanism, HFDAIL has two hydroxyl groups, which can absorb the produced water by hydrogen bonding. Thus, the esterification reaction equilibrium will shift to the right and the product yield will increase.

Fareghi-Alamdari's group also introduced two highly acidic, imidazolium-based, functionalized dicationic ionic liquids (FDCILs), which used (in 0.2 eq.) as efficient and green catalysts in the synthesis of phthalate plasticizers through the





Scheme 113 Synthesis of monoesters and diesters.

esterification of PA with alcohols (such as ethanol, *n*-propanol and *n*-butanol) at 110 °C. Among these two FDCILs, (FDCIL 1: [(dimethyl-4-sulfobutyl-ammonium)-1,2-ethane-1*H*-imidazolium sulfonic acid]hydrogen sulfate and FDCIL 2: 3,3'-(1,2-ethanediy)bis[1-(4-sulfobutyl)-1*H*-imidazolium sulfonic acid] hydrogen sulfate), the first one performed better. The catalytic activity of FDCIL is related to the density of acidic groups on it (higher acidity) and the length of the carbon chain (low lipophilic character) in the cationic part. The influences of the reaction temperature, catalyst dosage, and molar ratio of PA to alcohol on the esterification reaction were investigated. The reusability of the catalyst in these reactions was also studied. The yields were estimated by GC analysis.²⁶⁸ Phthalate plasticizers are the main plasticizers used as softening agents in various industrial applications. These compounds are mainly used as plasticizers for cellulosic resins and some vinyl ester resins, PVC, and nitrocellulose lacquers.²⁶⁹

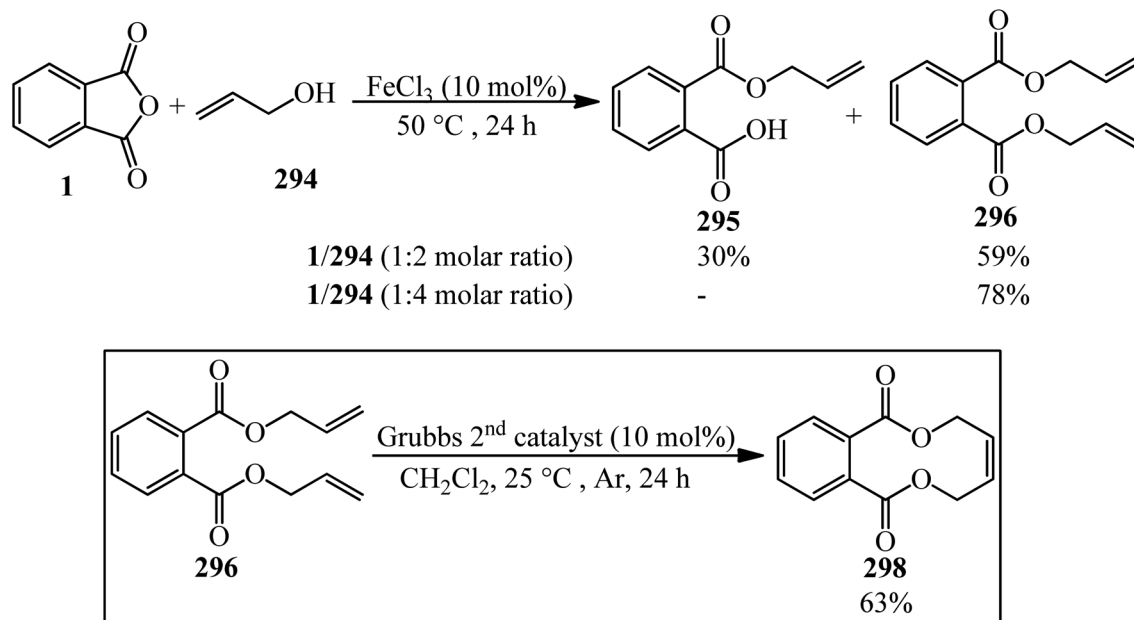
Fareghi-Alamdari in 2018 also prepared supported diacidic ionic liquid on magnetic silica nanoparticles (SDAIL@magnetic nano SiO₂) and investigated its catalytic activity (10 mol%) for the selective diesterification of alcohols (2-methoxy ethanol, allyl alcohol, 2-ethoxyhexanol, butanol, propanol, ethanol, and

methanol) with PA in a 5:1 molar ratio to afford the corresponding dialkyl plasticizers (296) under solvent-free conditions (65–180 °C) within 1–10 h. Under the optimized conditions, the conversion of PA was 100%, and the diester plasticizers were obtained with excellent yields (80–100%). The SDAIL@magnetic nano-SiO₂ catalyst showed good reusability and could be easily separated from the reaction mixture using an external magnet, washed with dichloromethane, and reused for the next runs for up to 8 runs without significant activity loss.²⁷⁰

Shahedi and Mansoori in 2018 studied the esterification reaction of PA (1) with various aliphatic and cycloaliphatic alcohols (such as propanol and butanol) in the presence of a Fe₃O₄@SiO₂-SO₃H (5 mol%) nanocatalyst. The two mentioned alcohols gave the corresponding diesters (296) in duration of 8–12 h with a yield of 71–97%.²⁷¹

Kulawska's group in 2011 considered the kinetics of the syntheses of higher aliphatic alcohols (C₇, C₉, C₁₁) phthalates in an isothermal, semi-batch reactor. In the first stage of the process, the formation of monoester (295) was very fast and irreversible. In the second stage, the esterification of monoester toward diester (296) in the presence sulfuric acid catalyst





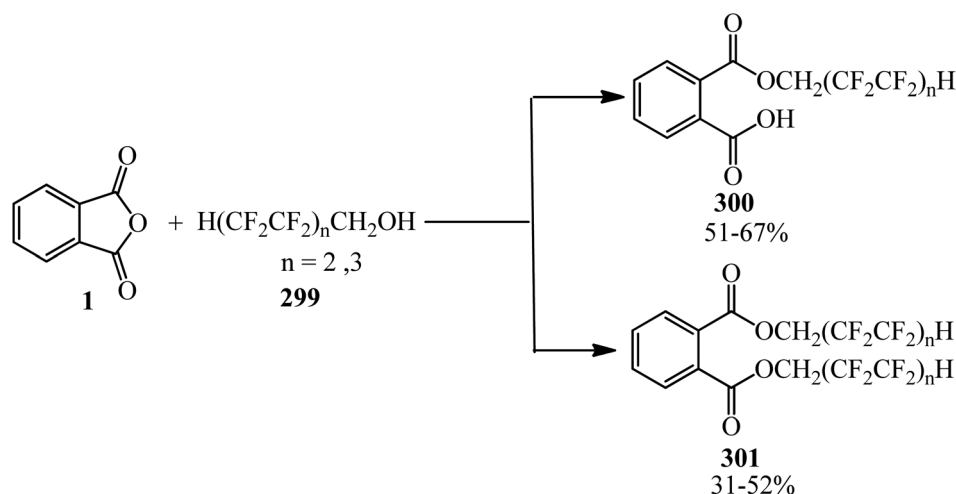
Scheme 114 Synthesis of allylic mono- and diesters of PA and macrolide.

progressed slowly. These reactions appeared to be first order with respect to the monoester and did not depend on the concentration of the alcohol.²⁷²

Mohammadpour Amini in 2003 evaluated the esterification of PA with 2-ethylhexanol and 1-butanol and ester decomposition of dioctyl phthalate (DOP) in the presence of different kinds of heteropolyacids such as (I) Keggin $\text{H}_3\text{PW}_{12}\text{O}_{40}$, $\text{H}_4\text{SiW}_{12}\text{O}_{40}$, and $\text{H}_4\text{SiMo}_{12}\text{O}_{40}$; (II) Wells–Dawson $\text{H}_6\text{P}_2\text{W}_{18}\text{O}_{62}$, $\text{H}_6\text{P}_2\text{W}_{17}\text{MoO}_{62}$; (III) Preyssler $\text{H}_{14}[\text{NaP}_5\text{W}_{29}\text{MoO}_{110}]$, $\text{H}_{14}[\text{NaP}_5\text{W}_{30}\text{O}_{110}]$. The heteropolyacids with Preyssler and Wells–Dawson structures (due to the higher number of acidic protons) and their molybdenum substituted derivatives (attributed to the reduction of Mo(vi) to Mo(v) and enhanced acidity) demonstrated higher activity in esterification and ester

decomposition reactions than Keggin type heteropolyacids. A complete conversion of PA to dioctyl phthalate and dibutyl phthalate are achieved in 2 h in the presence of molybdenum-substituted Preyssler heteropolyacid. In the decomposition of dioctyl phthalate in the presence of Preyssler heteropolyacid, 2-ethylhexene was formed in quantitative yield.²⁷³

The kinetics of esterification of PA with 2-ethylhexanol was studied by Bhutada and Pangarkar in 1986 using three kinds of catalysts such as tetrabutyl titanate (TBT), tetrabutyl zirconate (TBZ), and *p*-TSA. In each case, the kinetic parameters such as rate constant (order with respect to various reactants and catalyst), activation energy, and collision frequency were determined. The titanates and zirconates were found to be better than *p*-TSA in rate and rate/product quality, respectively.



Scheme 115 Synthesis of polyfluoroalkyl mono- and diesters.



The reaction shows general first order in the monoester and alcohol. Bimolecular acyl oxygen fission satisfactorily explains the kinetics.²⁷⁴

Gharibe in 2020 utilized the novel $\text{ZnAl}_2\text{O}_4/\text{SiO}_2$ (5 mol%) for the esterification of PA with 2-ethylhexanol to obtain dioctyl phthalate (DOP) in 99.2% yield in 45 min.²⁷⁵

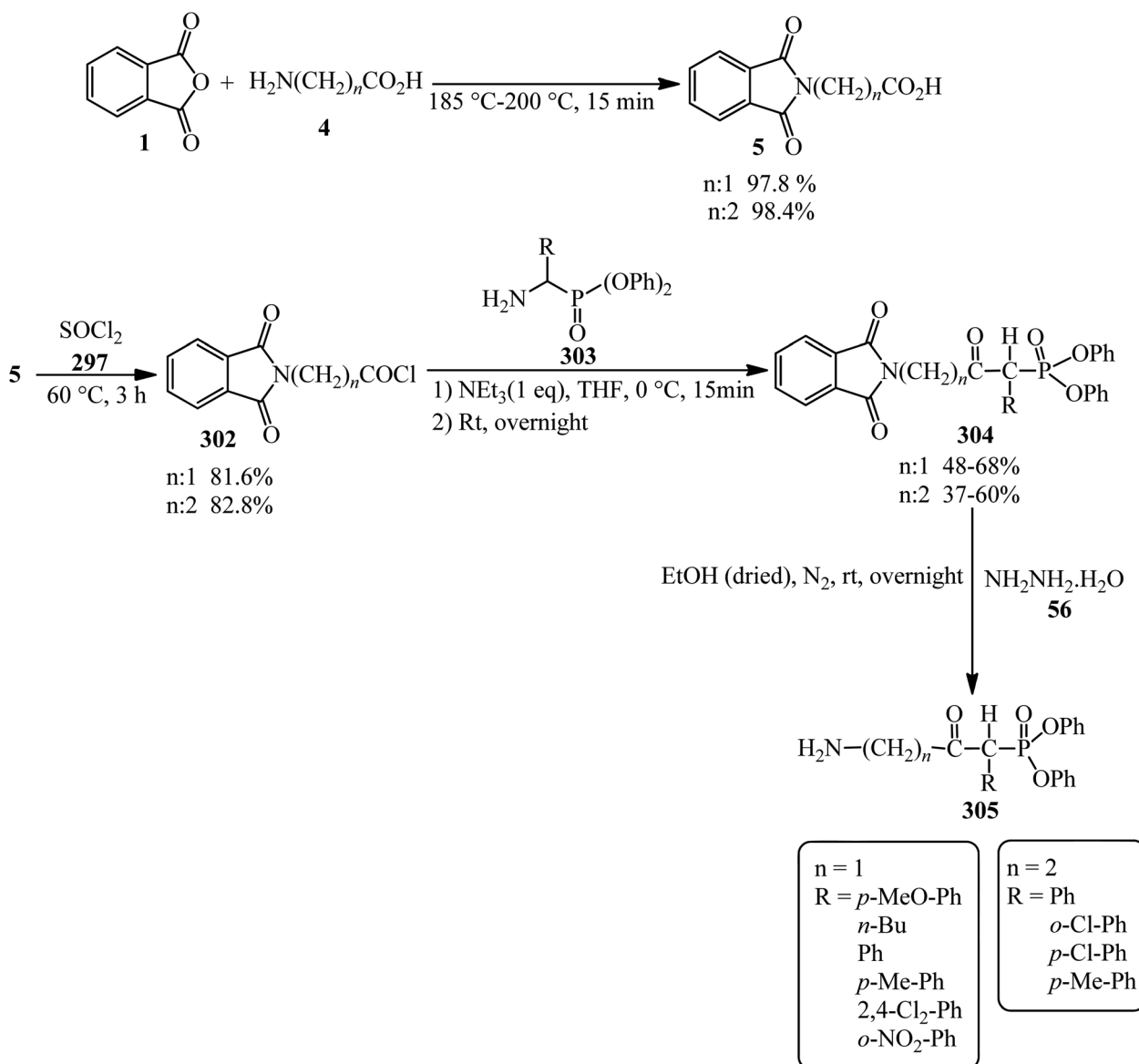
Brennecke's group in 1994 surveyed the pressure effect on the bimolecular rate constants for the esterification of PA with methanol in supercritical carbon dioxide (SC CO_2) at 40 °C and 50 °C. They observed a 25-fold decrease in the bimolecular rate constant based on bulk concentrations when increasing the pressure from 97.5 to 166.5 bar.²⁷⁶

Khodadadi Moghaddam and Gholami in 2006 inspected the esterification of PA with 2-octanol in the presence of sulfated titania ($\text{SO}_4^{2-}/\text{TiO}_2$ prepared by immersing titania powder in 1 N solution of sulfuric acid). The conversion

reached an equilibrium composition in 20 min for dioctyl phthalate without water removal from the reactor. The linear dependence of conversion to the catalyst amount exhibited that there are no mass transfer limitations in the reaction conditions.²⁷⁷

Yadav's group in 1992 represented the efficacy of several solid superacidic catalysts [such as phosphate, borated, and sulfated zirconia, sulphated iron oxide, phosphotungstic acid ($\text{P}_2\text{O}_5 \cdot 24\text{WO}_3 \cdot n\text{H}_2\text{O}$), and dodecatungstophosphoric acid ($\text{H}_3\text{-PO}_4 \cdot 12\text{WO}_3 \cdot n\text{H}_2\text{O}$)] in the preparation of the industrially important plasticizer dioctyl phthalate (DOP) from 2-ethylhexanol and PA.²⁷⁸

Bajracharya *et al.* in 2021 prepared phthalate monoesters (295) and diesters (296) *via* the reaction of PA (1) and alcohols (allyl, isoamyl, *n*-butyl, and benzyl alcohols) (294) through a one-pot two-step procedure in the presence of FeCl_3 as

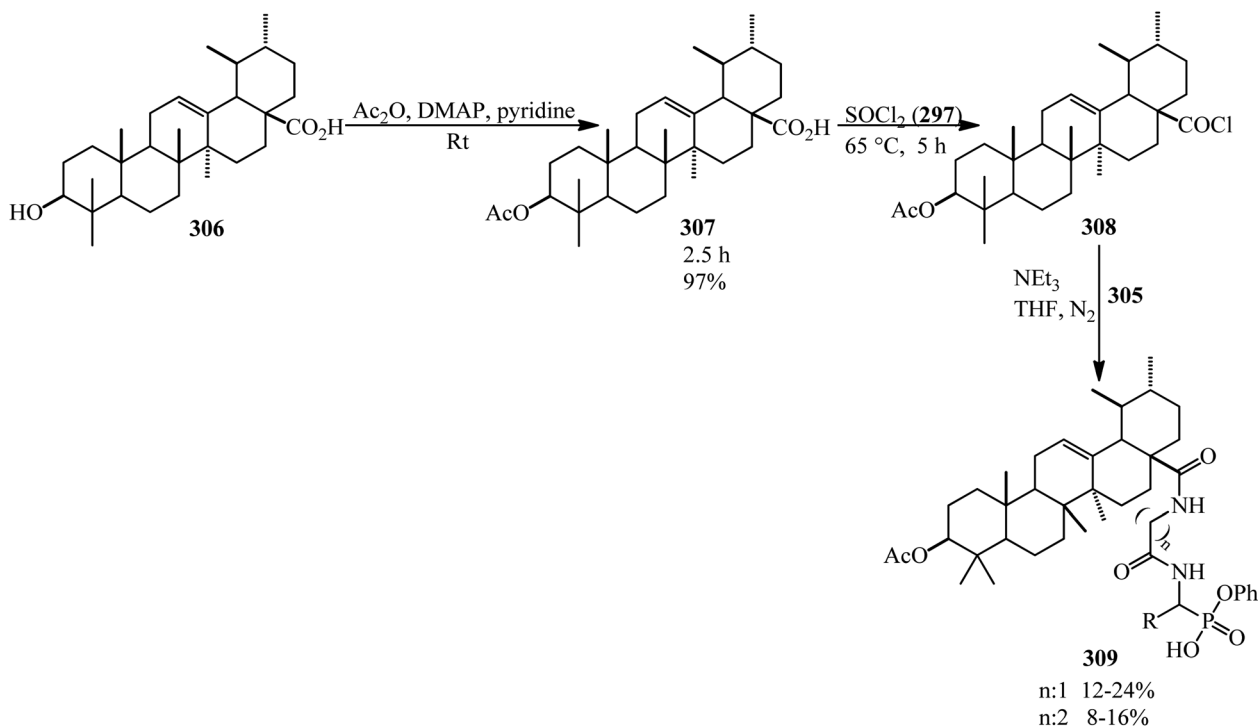


Scheme 116 Phosphonodipeptides and homolog preparation procedure.

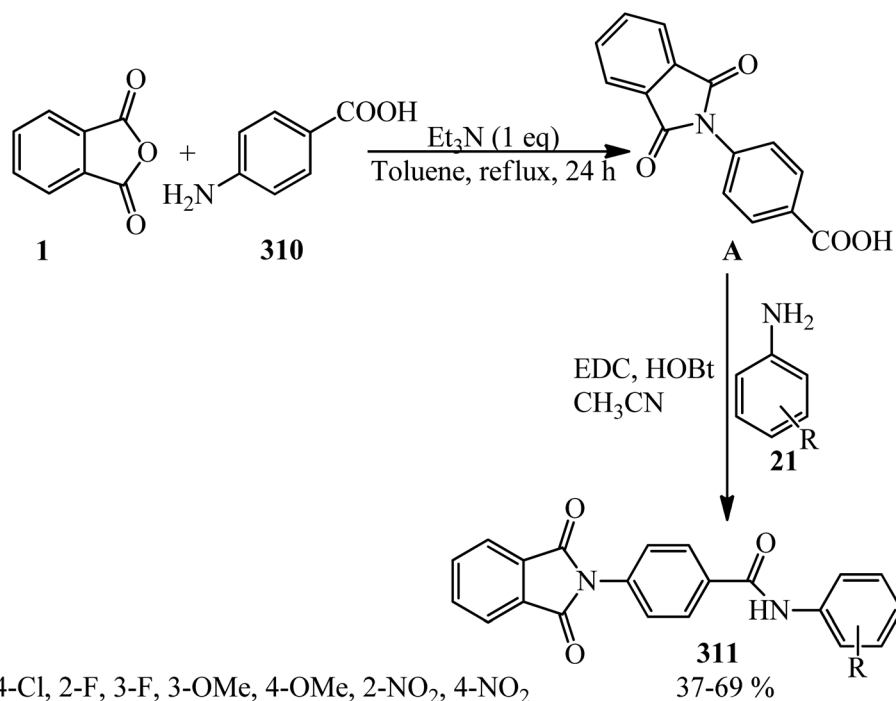


a catalyst. Utilizing the reactants in a 1 : 2 molar ratio at 50 °C achieved both the phthalate esters (**295**, **296**). Addition of excess amounts of alcohol at 50 °C or 100 °C yielded the selective preparation of diesters (**296**). The authors claimed that the synthesis of the adducts (**295**) was performed through

a facile addition displacement pathway, which was continued by Lewis acid catalysis providing phthalate diesters (**296**). The ring-closing metathesis of diallyl phthalate using the Grubb's 2nd generation catalyst led to macrolide (**298**) (Scheme 114).²⁷⁹



Scheme 117 Novel phosphonodipeptide conjugates of ursolic acid and their homologs.



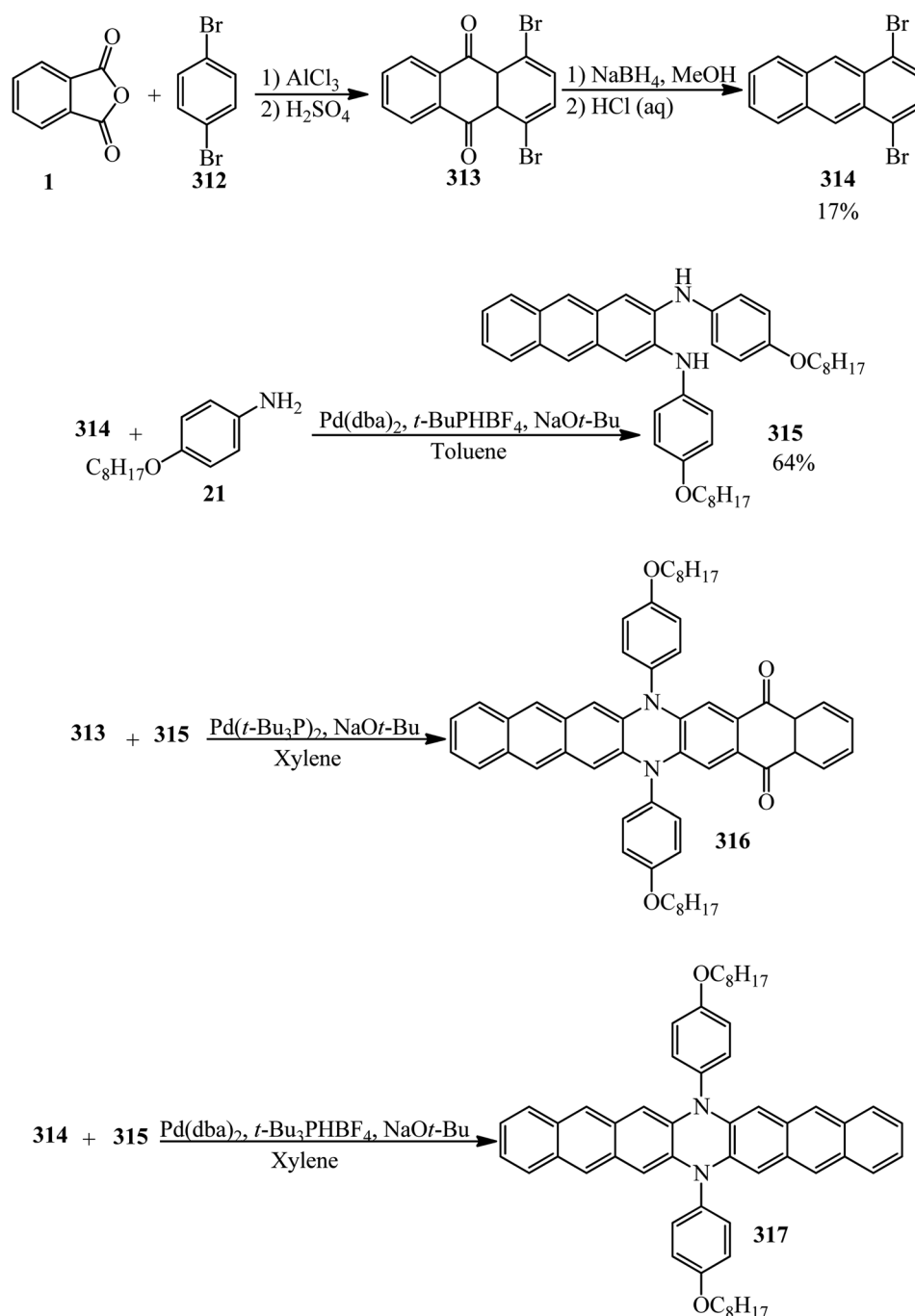
Scheme 118 Synthesis of 4-(1,3-dioxisoindolin-2-yl)-*N*-phenylbenzamides.



Polyfluoroalkyl mono- and diesters of PA were obtained by Rakhimova and Kudashev in 2011 through the reaction of PA (**1**) with polyfluorinated alcohols (**299**). The monesters (**300**) were achieved *via* the ultrasonic-assisted (40 kHz) reaction of equimolar amounts of PA and polyfluorinated alcohol in cyclohexanone for 2 h at 70 °C, which was followed by heating at 130 °C for 2 h to complete the transition of phthalic anhydride. The diesters (**301**) were also prepared as the same procedure in the presence of PA : alcohols in a 1 : 2 molar ratio (Scheme 115).²⁸⁰

Karimi Alavijeh and Amini in 2019 demonstrated that mesoporous MIL-101(Cr)-SO₃H and [Cr₃O(BDC-SO₃H)_{2.4}(BDC-SO₃-NH₃Bu)_{0.6}]_n (20% BuNH₂) acted as potent and robust catalysts in the esterification of PA (**1**) with various alcohols in a 1 : 5 molar ratio under solvent-free conditions.²⁸¹ After the completion of the reaction, the catalyst was separated by centrifugation, washing with dichloromethane, drying at 50 °C, and reusing in the next reaction for 6 runs without activity loss.

Fieser in 1931 discussed about the condensation of β-naphthol with phthalic anhydride in detail.²⁸²



Scheme 119 Synthetic procedure of 7,16-bis(4-(octyloxy)phenyl)-7,16-dihydrodinaphtho[2,3-*b*:2',3'-*i*]phenazine and 7,16-bis(4-(octyloxy)phenyl)-16,18a-dihydrodinaphtho[2,3-*b*:2',3'-*i*]phenazine-5,18(4aH,7H)-dione.



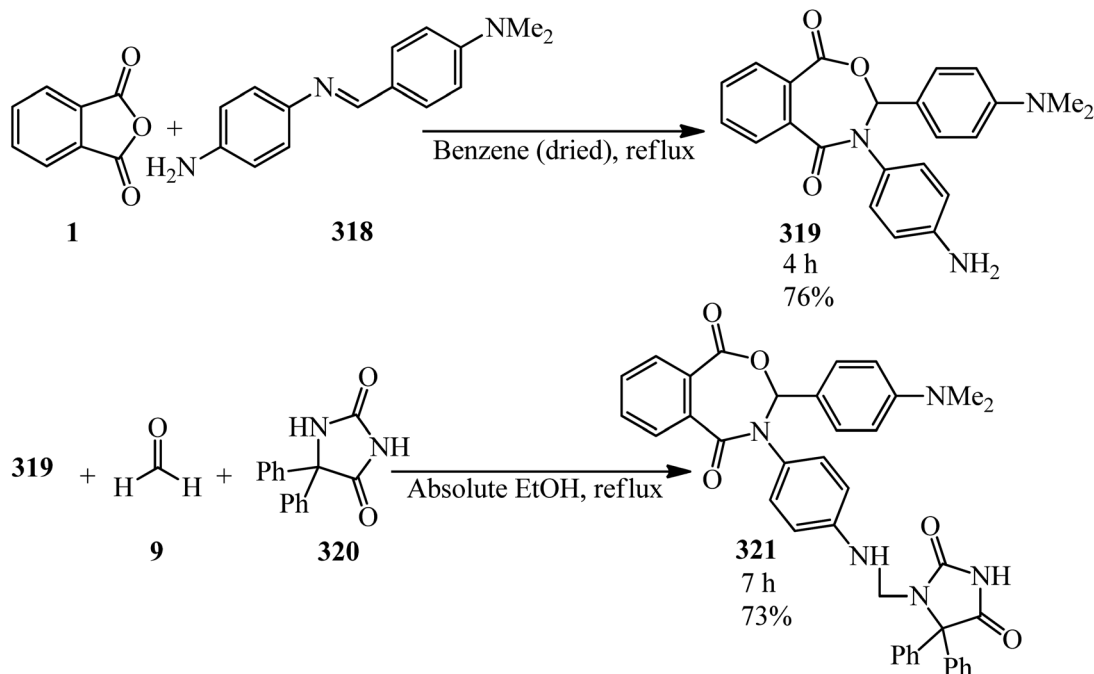
7. Some examples of PA utilization in multistep applicable products

PA could function as a key substrate in multicomponent sequential reactions, which led to applicable products in various fields such as drugs, natural products, and industry. Some examples have been discussed below.

Cavè *et al.* in 2008 prepared novel phosphonodipeptide conjugates of ursolic acid (3β -hydroxy-urs-12-en-28-oic acid) and their homologs *via* a series of reactions. First, the preparation of a series of α -phosphonodipeptides and homologs has been reported (Scheme 117). According to Scheme 116, the 1,3-dihydro-1,3-dioxo-2*H*-isoindole-2-acetic (or propanoic) acids (**5**) were obtained by mixing the grounded forms of PA (**1**) and amino acids (**4**) (glycine and β -alanine), which was converted to acyl chloride by SOCl_2 (**297**) to phthalimidoacetyl chloride (**302**). The subsequent reaction of (**302**) with α -aminophosphonates (**303**) yielded *N*-blocked phosphonodipeptides and homologs (**304**). The phosphonodipeptides and homologs (**305**) were obtained from hydrazinolysis with hydrazine hydrate (**56**). In the second step (Scheme 117), the introduction of phosphonodipeptides and their homologs (**305**) to naturally bioactive ursolic acid (UA) (**306**) at C-28 moiety afforded new classes of phosphonodipeptide conjugates of ursolic acid and their homologs (**309**), which was performed by (i) its acylation to 3β -acetoxy-urs-12-en-28-oic acid (**307**); (ii) chlorination to 3β -acetoxy-urs-12-en-28-oyl chloride (**308**), and (iii) final reaction with (**305**) to afford the (**309**) adducts.²⁸³ The phosphonodipeptide conjugates of UA and their homologs (**309**) are potent for anti-HT29 (human colon cancer cell line) and anti-HIV properties.

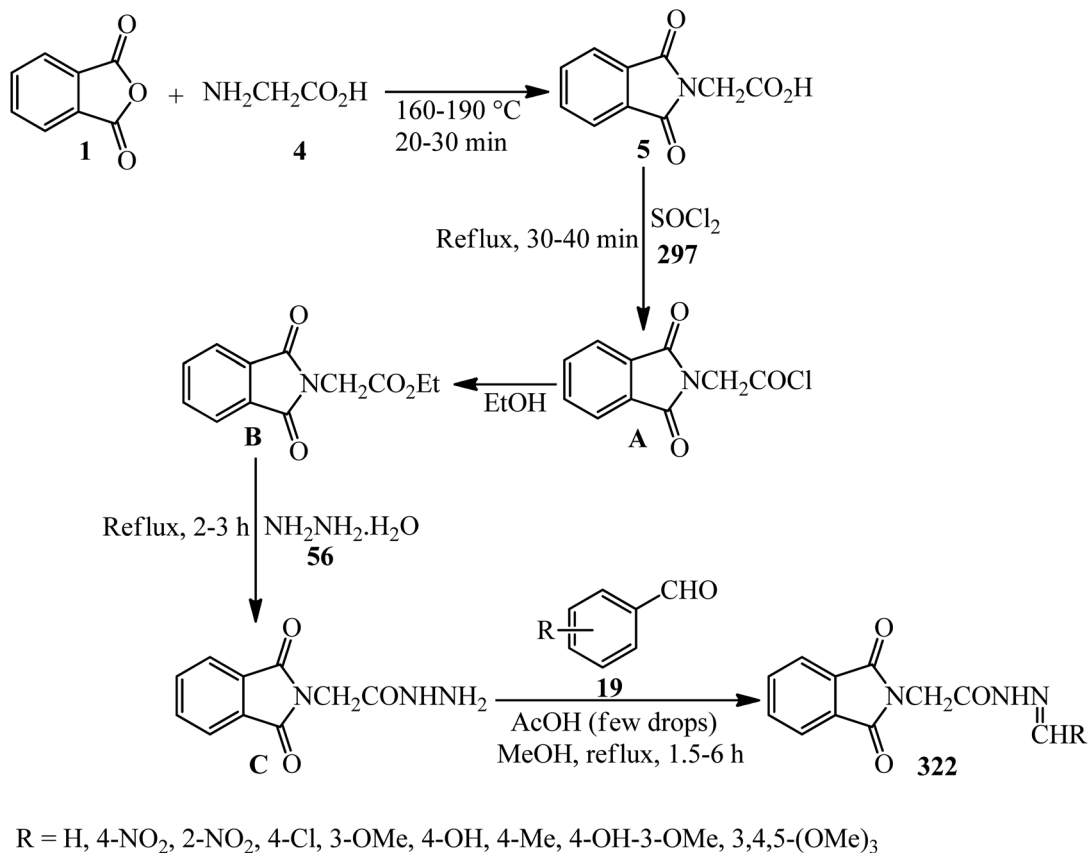
Aliabadi and coworkers in 2016 synthesized a new series of phthalimides named 4-(1,3-dioxoisindolin-2-yl)-*N*-phenylbenzamides (**311**) *via* the reaction of PA (**1**) with 4-aminobenzoic acid (**310**). The first step consisted of Gabriel-like reaction in the presence of triethylamine (as a proton acceptor) under reflux conditions to obtain 4-(1,3-dioxoisindolin-2-yl)benzoic acid (**A**), which was mixed with equimolar quantities of *N*-ethyl-*N*-dimethyl aminopropyl carbodiimide (EDC, as a carbodiimide coupling agent) and hydroxyl benzotriazole (HOBT, additive agent) in acetonitrile solvent and was stirred for 30 min at room temperature. Then, an equimolar quantity of appropriate anilines (**21**) was added to the reaction medium. On continuous stirring for 24 h, an amidation reaction through a carbodiimide coupling reaction yielded the adducts (**311**). The potential of the products as anti-Alzheimer agents was examined. The anti-acetylcholinesterase activity of the synthesized derivatives was assessed by Ellman's test. The product, with ortho nitro moiety, in this series exhibited the highest inhibitory potency ($\text{IC}_{50} = 1.1 \pm 0.25 \mu\text{M}$) compared to donepezil ($\text{IC}_{50} = 0.41 \pm 0.12 \mu\text{M}$) as the reference drug (Scheme 118).²⁸⁴

Iwanaga in 2019 prepared dinaphtho[2,3-*b*:2',3'-*i*]dihydrophenazine derivatives *via* a multistep reaction starting from PA. First, Friedel-Crafts reaction of PA (**1**) with 1,4-dibromobenzene (**312**), followed by dehydration, afforded 2,3-dibromoanthraquinone (**313**), which reduced stepwise to give 2,3-dibromoanthracene (**314**) with 17% overall yield. Pseudo three-component conventional Buchwald-Hartwig coupling of (**314**) with 4-octyloxyaniline (**21**)²⁸⁵ gave compound (**315**) as a yellow solid. The reaction of (**315**) with (**313**) and (**314**) gave 7,16-bis(4-(octyloxy)phenyl)-16,18*a*-dihydrodinaphtho[2,3-*b*:2',3'-*i*]phenazine-5,18(4*aH*,7*H*)-dione (**316**) and 7,16-bis(4-(octyloxy)phenyl)-



Scheme 120 3-(4-(Dimethylamino)phenyl)-4-(4-(((2,4-dioxo-5,5-diphenylimidazolidin-1-yl)methyl)amino)phenyl)-3,4-dihydrobenzo[e][1,3]xazepane-1,5-dione preparation.





Scheme 121 Synthesis of 2-(1,3-dioxo-1,3-dihydro-2H-isindol-2-yl)-N'-(substituted benzylidene)acetohydrazides.

7,16-dihydrodinaphtho[2,3-*b*:2',3'-*i*]phenazine (317), respectively. The X-ray crystallographic analysis of (317) indicated a planar structure. The photophysical properties were influenced by nitrogen atoms, resulting in extended π -conjugation for (317) and intramolecular donor-acceptor interactions for (316). The oxidation potentials of (316, 317) were similar due to the independence of the 2,3-diaminoanthracene unit (Scheme 119).²⁸⁶

4-(4-Aminophenyl)-3-(4-(dimethylamino)phenyl)-3,4-dihydrobenzo[*e*][1,3]oxazepine-1,5-dione (319) was constructed through the reaction of PA (1) and Schiff base (318). The Mannich reaction occurred between (319), paraformaldehyde (9), and phenytoin (320) in equimolar ratio to obtain 3-(4-(dimethylamino)phenyl)-4-(4-(((2,4-dioxo-5,5-diphenylimidazolidin-1-yl)methyl)amino)phenyl)-3,4-dihydrobenzo[*e*][1,3]oxazepine-1,5-dione (321). The authors investigated the preparation of some other phenytoin derivatives containing diazepines and quinazoline rings through the reaction of (318) with malic anhydride and phthalic imide. The anticancer and cytotoxicity properties of the products was tested through the effects of treating the *L20B* cell line (Scheme 120).²⁸⁷

Bala's group in 2014 introduced phthalic anhydride-based benzylidene-hydrazides (322) as novel, potential anti-inflammatory agents. The synthetic route started from PA (1) and glycine (4) to give phthaloylglycine (5), which was subjected to chlorination by thionyl chloride to form (A). Subsequent

esterification, followed by reaction with hydrazine hydrate (56), and the final reaction of intermediate (C) with benzaldehydes (19) yielded 2-(1,3-dioxo-1,3-dihydro-2H-isindol-2-yl)-N'-(substituted benzylidene)acetohydrazides (322). The research consists of computational studies. The products were screened for *in vivo* anti-inflammatory and analgesic activities by carrageenan-induced Rat Paw Oedema and tail immersion methods, respectively, using diclofenac sodium as a standard drug (Scheme 121).²⁸⁸

8. Conclusion

In this literature survey, the authors focused on applications of phthalic anhydride (PA), a cyclic fused heterocyclic anhydride, as a versatile substrate in organic reactions. Actually, PA is an efficient substrate and/or intermediate in several organic transformations due to its bifunctional cyclic anhydride moiety, which could be opened *via* the attack of various nucleophilic groups. In addition to the introduction part, which identified the compound and its overall utilities in various respects, the review has been subdivided into some parts, centralizing on the organic two- and multicomponent reactions containing PA as the substrate. The importance and applicability of PA in esterification are also discussed with examples. There is also a final part that discusses the preparation of heterocycles starting from PA *via* multistep protocols, which possess various applications



in pharmacology, treatment, natural products, industry, and some other fields of operative aspects of science and technology. In all the abovementioned parts, PA plays a crucial role as a substrate that cannot be changed with other heterocycles, which could be due to the intrinsic activity of the anhydride functional group. The authors hope that this review would be helpful and effectual for future research by chemists on PA and its analogs.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors are thankful to Alzahra University for their support.

References

- 1 S. C. Gad, Phthalic Anhydride, in *Encyclopedia of Toxicology*, Elsevier, 2014, pp. 934–936.
- 2 D. A. Basketter and I. Kimber, Phthalic anhydride: illustrating a conundrum in chemical allergy, *J. Immunotoxicol.*, 2016, **13**, 767–769, DOI: [10.1080/1547691X.2016.1177149](https://doi.org/10.1080/1547691X.2016.1177149).
- 3 A. Laurent, Sur l'acide naphthalique et ses combinaisons, *Ann. Chim. Phys.*, 1836, **61**, 113–125.
- 4 P. M. Lorz, F. K. Towae, W. Enke, R. Jäckh, N. Bhargava and W. Hillesheim, Phthalic acid and derivatives, *Ullmann's Encyclopedia of Industrial Chemistry*, Weinheim, Wiley-VCH, 2007, DOI: [10.1002/14356007.a20181.pub2](https://doi.org/10.1002/14356007.a20181.pub2).
- 5 B. Rindone, F. Saliu and R. Suarez-Bertoa, The synthesis of phthalic anhydride via ozonation of naphthalene, *Ozone: Sci. Eng.*, 2010, **32**, 161–165, DOI: [10.1080/01919511003788001](https://doi.org/10.1080/01919511003788001).
- 6 C. R. Dias, M. F. Portela and G. C. Bond, Oxidation of o-xylene to phthalic anhydride over V₂O₅/TiO₂ catalysts: part 4. mathematical modelling study and analysis of the reaction network, *J. Catal.*, 1996, **164**, 276–287, DOI: [10.1006/jcat.1996.0384](https://doi.org/10.1006/jcat.1996.0384).
- 7 S. Luciani, N. Ballarini, F. Cavani, C. Cortelli, F. Cruzolin, A. Frattini, R. Leanza and B. Panzacchi, Anatase-supported vanadium oxide catalysts for o-xylene oxidation: from consolidated certainties to unexpected effects, *Catal. Today*, 2009, **142**, 132–137, DOI: [10.1016/j.cattod.2008.08.044](https://doi.org/10.1016/j.cattod.2008.08.044).
- 8 J. N. Papageorgiou and G. F. Froment, Phthalic anhydride synthesis. reactor optimization aspects, *Chem. Eng. Sci.*, 1996, **51**, 2091–2098, DOI: [10.1016/0009-2509\(96\)00066-8](https://doi.org/10.1016/0009-2509(96)00066-8).
- 9 K. Masoomi, M. Ghiaci and A. M. Botelho do Rego, Selective vapor-phase oxidation of o-xylene to phthalic anhydride over Co-Mn/H₃PW₁₂O₄₀@TiO₂ using molecular oxygen as a green oxidant, *Appl. Organomet. Chem.*, 2018, **32**, e4461, DOI: [10.1002/aoc.4461](https://doi.org/10.1002/aoc.4461).
- 10 T. Boger and M. Menegola, Monolithic catalysts with high thermal conductivity for improved operation and economics in the production of phthalic anhydride, *Ind. Eng. Chem. Res.*, 2005, **44**, 30–40, DOI: [10.1021/ie040088f](https://doi.org/10.1021/ie040088f).
- 11 G. Groppi, E. Tronconi, C. Cortelli and R. Leanza, Conductive monolithic catalysts: development and industrial pilot tests for the oxidation of o-xylene to phthalic anhydride, *Ind. Eng. Chem. Res.*, 2012, **51**, 7590–7596, DOI: [10.1021/ie2021653](https://doi.org/10.1021/ie2021653).
- 12 M. Selvaraj and T. G. Lee, A novel route to produce phthalic anhydride by oxidation of o-xylene with air over mesoporous V-Mo-MCM-41 molecular sieves, *Microporous Mesoporous Mater.*, 2005, **85**, 39–51, DOI: [10.1016/j.micromeso.2005.05.046](https://doi.org/10.1016/j.micromeso.2005.05.046).
- 13 F. List and H. Alfs, Production of phthalic anhydride from phthalic acid, *US Pat. US3720692A*, United States, 1968.
- 14 V. Nikolov, D. Klissurski and A. Anastasov, Phthalic anhydride from o-xylene catalysis: science and engineering, *Catal. Rev.: Sci. Eng.*, 1991, **33**, 319–374, DOI: [10.1080/01614949108020303](https://doi.org/10.1080/01614949108020303).
- 15 C. R. Dias and M. F. Portela, Synthesis of phthalic anhydride: catalysts, kinetics, and reaction modeling, *Catal. Rev.: Sci. Eng.*, 1997, **39**, 169–207, DOI: [10.1080/01614949709353776](https://doi.org/10.1080/01614949709353776).
- 16 E. Mahmoud, D. A. Watson and F. Lobo, Renewable production of phthalic anhydride from biomass-derived furan and maleic anhydride, *Green Chem.*, 2014, **16**, 167–175, DOI: [10.1039/C3GC41655K](https://doi.org/10.1039/C3GC41655K).
- 17 X. Shao, L. Su, J. Zhang, Z. Tian, N. Zhang, Y. Wang, H. Wang, X. Cui, X. Hou and T. Deng, Green Production of phthalic anhydride from biobased furan and maleic anhydride by an acid resin catalyst, *ACS Sustainable Chem. Eng.*, 2021, **9**, 14385–14394, DOI: [10.1021/acssuschemeng.1c03974](https://doi.org/10.1021/acssuschemeng.1c03974).
- 18 (a) W. Jia, W. Li, X. Zhao, Y. Feng, M. Zuo, Y. Sun, X. Tang, X. Zeng and L. Lin, Insights into the catalytic mechanism of 5-hydroxymethylfurfural to phthalic anhydride with MoO₃/Cu(NO₃)₂ in one-pot, *Catal. Sci. Technol.*, 2021, **11**, 5656–5662, DOI: [10.1039/D1CY00940K](https://doi.org/10.1039/D1CY00940K); (b) W. Jia, Y. Sun, M. Zuo, Y. Feng, X. Tang, X. Zeng and L. Lin, One-pot synthesis of renewable phthalic anhydride from 5-hydroxymethylfurfural by using MoO₃/Cu(NO₃)₂ as catalyst, *Chemsuschem*, 2020, **13**, 640–646, DOI: [10.1002/cssc.201902590](https://doi.org/10.1002/cssc.201902590).
- 19 S. Giarola, Ch. Romain, Ch. K. Williams, J. P. Hallett and N. Shah, Techno-economic assessment of the production of phthalic anhydride from corn stover, *Chem. Eng. Res. Des.*, 2016, **107**, 181–194, DOI: [10.1016/j.cherd.2015.10.034](https://doi.org/10.1016/j.cherd.2015.10.034).
- 20 Zh. Lin, M. Ierapetritou and V. Nikolakis, Phthalic Anhydride production from hemicellulose solutions: techno-economic analysis and life cycle assessment, *AIChE J.*, 2015, **61**, 3708–3718.
- 21 C. R. Kinney and I. Pincus, Catalytic air oxidation of higher aromatics to phthalic anhydride, *Ind. Eng. Chem.*, 1951, **43**, 2880–2884, DOI: [10.1021/ie50504a062](https://doi.org/10.1021/ie50504a062).
- 22 J. Shelmerdine, F. Popper and D. McNeil, The production of phthalic anhydride by the oxidation of tar oils, *J. Appl. Chem.*, 1953, **3**, 513–521, DOI: [10.1002/jctb.5010031106](https://doi.org/10.1002/jctb.5010031106).



- 23 W. W. Prichard, Reactions of carbon monoxide at high temperature. I. a new synthesis of phthalic anhydrides, *J. Org. Chem.*, 1956, **78**, 6137–6139, DOI: [10.1021/ja01604a050](https://doi.org/10.1021/ja01604a050).
- 24 V. A. Zazhigalov and E. V. Kizium, Naphthoquinoneation of phthalic anhydride by Diels-Alder reaction during n-pentane oxidation on VPO catalysts and control of the process selectivity, *Theor. Exp. Chem.*, 2017, **53**, 194–198, DOI: [10.1007/s11237-017-9515-9](https://doi.org/10.1007/s11237-017-9515-9).
- 25 Y. H. Kim and H. S. Yang, Synthesis of maleic and phthalic anhydrides from the mixture of cyclopentene and 1-pentene, *Korean J. Chem. Eng.*, 2000, **17**, 357–364, DOI: [10.1007/BF02699053](https://doi.org/10.1007/BF02699053).
- 26 H. D. Gibbs, Phthalic anhydride. I-introduction, *Ind. Eng. Chem.*, 1919, **11**, 1031–1032, DOI: [10.1021/ie50119a010](https://doi.org/10.1021/ie50119a010).
- 27 H. D. Gibbs, History of the preparation and properties of pure phthalic anhydride, *Ind. Eng. Chem.*, 1920, **12**(10), 1017–1018, DOI: [10.1021/ie50130a031](https://doi.org/10.1021/ie50130a031).
- 28 M. S. Ray, Case Study-Phthalic anhydride reactor design, *Chemical engineering design project, a case study approach*, CRC press; London, 2nd edn, 1989, ch. 10.
- 29 M. S. Wainwright and N. R. Foster, Catalysts, kinetics and reactor design in phthalic anhydride synthesis, *Catal. Rev.: Sci. Eng.*, 1979, **19**, 211–292, DOI: [10.1080/03602457908068056](https://doi.org/10.1080/03602457908068056).
- 30 A. Soroush, H. Rezaie Haghghat and S. H. Sajadina, Thermal and mechanical properties of polysulfide/epoxy copolymer system: The effect of anhydride content, *Polym. Adv. Technol.*, 2014, **25**, 184–190, DOI: [10.1002/pat.3221](https://doi.org/10.1002/pat.3221).
- 31 W. Bukowski, A. Bukowska, A. Sobota, M. Pytel and K. Bester, Copolymerization of phthalic anhydride with epoxides catalyzed by amine-bis(phenolate) chromium(III) complexes, *Polymers*, 2021, **13**, 1785, DOI: [10.3390/polym13111785](https://doi.org/10.3390/polym13111785).
- 32 A. Amin, H. H. M. Darweesh, S. M. M. Morsi and M. M. H. Ayoub, Effect of phthalic anhydride-based hyperbranched polyesteramide on cement characteristics, *J. Appl. Polym. Sci.*, 2011, **120**, 3054–3064, DOI: [10.1002/app.33484](https://doi.org/10.1002/app.33484).
- 33 E. Hosseini Nejad, A. Paoniasari, C. G. W. van Melis, C. E. Koning and R. Duchateau, Catalytic ring-opening copolymerization of limonene oxide and phthalic anhydride: toward partially renewable polyesters, *Macromolecules*, 2013, **46**, 631–637, DOI: [10.1021/ma301904y](https://doi.org/10.1021/ma301904y).
- 34 S. A. Pooley, G. S. Canessa and B. L. Rivas, Spontaneous copolymerization of phthalic anhydride with aziridine or 2-methylaziridine, *Eur. Polym. J.*, 1995, **31**, 547–553, DOI: [10.1016/0014-3057\(94\)00204-5](https://doi.org/10.1016/0014-3057(94)00204-5).
- 35 M. Ranjbar-Mohammadi, M. Arami, H. Bahrami, F. Mazaheri and N. M. Mahmoodi, Grafting of chitosan as a biopolymer onto wool fabric using anhydride bridge and its antibacterial property, *Colloids Surf., B*, 2010, **76**, 397–403, DOI: [10.1016/j.colsurfb.2009.11.014](https://doi.org/10.1016/j.colsurfb.2009.11.014).
- 36 E. M. A. Braz, S. C. C. C. Silva, C. A. R. S. Brito, L. M. Brito, H. M. Barreto, F. A. A. Carvalho, L. S. S. Junior, A. O. Lobo, J. A. Osajima, K. S. Sousa and E. C. Silva-Filho, Spectroscopic, thermal characterizations and bacteria inhibition of chemically modified chitosan with phthalic anhydride, *Mater. Chem. Phys.*, 2020, **240**, 122053, DOI: [10.1016/j.matchemphys.2019.122053](https://doi.org/10.1016/j.matchemphys.2019.122053).
- 37 S. Ye, W. Wang, J. Liang, S. Wang, M. Xiao and Y. Meng, Metal-free approach for a one-pot construction of biodegradable block copolymers from epoxides, phthalic anhydride, and CO₂, *ACS Sustainable Chem. Eng.*, 2020, **8**, 17860–17867, DOI: [10.1021/acssuschemeng.0c07283](https://doi.org/10.1021/acssuschemeng.0c07283).
- 38 S. Davarpanah, N. M. Mahmoodi, M. Arami, H. Bahrami and F. Mazaheri, Environmentally friendly surface modification of silk fiber: chitosan grafting and dyeing, *Appl. Surf. Sci.*, 2009, **255**, 4171–4176, DOI: [10.1016/j.apsusc.2008.11.001](https://doi.org/10.1016/j.apsusc.2008.11.001).
- 39 S. Mallakpour and S. Sepehri, Preparation of new optically active polyamides containing a L-Phenylalanine, phthalimide side-chain via the diisocyanate route by microwave energy: comparison with conventional heating, *Des. Monomers Polym.*, 2008, **11**, 535–546, DOI: [10.1163/156855508X363825](https://doi.org/10.1163/156855508X363825).
- 40 B. Moona, T. R. Hoyea and Ch. W. Macosko, Synthesis and application of fluorescently labeled phthalic anhydride (PA) functionalized polymers by ATRP, *Polymer*, 2002, **43**, 5501–5509, DOI: [10.1016/S0032-3861\(02\)00406-8](https://doi.org/10.1016/S0032-3861(02)00406-8).
- 41 S. Mallakpour and M. Taghavi, A facile, microwave-assisted synthesis of novel optically active polyamides derived from 5-(3-methyl-2-phthalimidylpentanoylamino)isophthalic acid and different diisocyanates, *Eur. Polym. J.*, 2008, **44**, 87–97, DOI: [10.1016/j.eurpolymj.2007.10.027](https://doi.org/10.1016/j.eurpolymj.2007.10.027).
- 42 (a) J. Li, Y. Liu, W.-M. Ren and X.-B. Lu, Enantioselective terpolymerization of racemic and meso-epoxides with anhydrides for preparation of chiral polyesters, *Proc. Natl. Acad. Sci. U. S. A.*, 2020, **117**, 15429–15436, DOI: [10.1073/pnas.2005519117](https://doi.org/10.1073/pnas.2005519117); (b) H.-Y. Ji, D.-P. Song, B. Wang, L. Pan and Y.-S. Li, Organic Lewis pairs for selective copolymerization of epoxides with anhydrides to access sequencecontrolled block copolymers, *Green Chem.*, 2019, **21**, 6123–6132, DOI: [10.1039/c9gc02429h](https://doi.org/10.1039/c9gc02429h).
- 43 X. Zhang and Y. Zhang, Poly(butylene succinate-co-butylene adipate)/cellulose nanocrystal composites modified with phthalic anhydride, *Carbohydr. Polym.*, 2015, **134**, 52–59, DOI: [10.1016/j.carbpol.2015.07.078](https://doi.org/10.1016/j.carbpol.2015.07.078).
- 44 X. Zhang and X. Wang, Polybutylene succinate/cellulose nanocrystals: role of phthalic anhydride in squeeze oriented bionanocomposites, *Carbohydr. Polym.*, 2018, **196**, 254–261, DOI: [10.1016/j.carbpol.2018.04.124](https://doi.org/10.1016/j.carbpol.2018.04.124).
- 45 W. Li, L. Wu, C. Di, C. Liu and R. Sun, DMAP-catalyzed phthalylation of cellulose with phthalic anhydride in [bmim]Cl, *Bioresour.*, 2011, **6**, 2375–2385.
- 46 S. Shaaban, M. A. Arafat, E. Gaffer and W. S. Hamama, Synthesis and anti-tumor evaluation of novel organoselenocyanates and symmetrical diselenides dyestuffs, *Der Pharma Chem.*, 2014, **6**, 186–193.
- 47 F. Hassanzadeh, M. Rabbani, G. A. Khodarahmi and M. Moosavi, Synthesis and evaluation of the anxiolytic activity of some phthalimide derivatives in mice model of anxiety, *Iran. J. Pharm. Res.*, 2012, **11**, 109–115.



- 48 S. Mathur and S. Tabassum, Template synthesis of novel carboxamide dinuclear copper (II) complex: spectral characterization and reactivity towards calf-thymus DNA, *BioMetals*, 2008, **21**, 299–310, DOI: [10.1007/s10534-007-9119-2](https://doi.org/10.1007/s10534-007-9119-2).
- 49 M.-Th. Khong, V. Berl, L. Kuhn, Ph. Hammann and J.-P. Lepoittevin, Chemical modifications induced by phthalic anhydride, a respiratory sensitizer, in reconstructed human epidermis: a combined HRMAS NMR and LC-MS/MS proteomic approach, *Chem. Res. Toxicol.*, 2021, **34**, 2087–2099, DOI: [10.1021/acs.chemrestox.1c00172](https://doi.org/10.1021/acs.chemrestox.1c00172).
- 50 M. V. Spanedd and L. Bourel-Bonnet, Cyclic anhydrides as powerful tools for bioconjugation and smart delivery, *Bioconjugate Chem.*, 2021, **32**, 482–496, DOI: [10.1021/acs.bioconjchem.1c00023](https://doi.org/10.1021/acs.bioconjchem.1c00023).
- 51 B. D. Salih, A. H. Dalaf, M. A. Alheety, W. M. Rashed and I. Q. Abdullah, Biological activity and laser efficacy of new Co(II), Ni(II), Cu(II), Mn(II) and Zn(II) complexes with phthalic anhydride, *Mater. Today*, 2021, **43**, 869–874, DOI: [10.1016/j.matpr.2020.07.083](https://doi.org/10.1016/j.matpr.2020.07.083).
- 52 M. I. Marzouk, S. A. Shaker, A. A. Abdel Hafiz and Kh. Z. El-Baghdady, Design and synthesis of new phthalazinone derivatives containing benzyl moiety with anticipated antitumor activity, *Biol. Pharm. Bull.*, 2016, **39**, 239–251, DOI: [10.1248/bpb.b15-00656](https://doi.org/10.1248/bpb.b15-00656).
- 53 M. Yamaguchi, K. Kamei, T. Koga, M. Akima, T. Kuroki and N. Ohi, Novel antiasthmatic agents with dual activities of thromboxane A₂ synthetase inhibition and bronchodilation. 1. 2-[2-(1-Imidazolyl)alkyl]-1(2EI)-phthalazinones, *J. Med. Chem.*, 1993, **36**, 4052–4060, DOI: [10.1021/jm00077a009](https://doi.org/10.1021/jm00077a009).
- 54 S. C. C. Costa Silva, E. M. A. Braz, C. A. R. S. Brito, M. M. M. Alves, F. A. A. Carvalho, H. M. Barreto, A. L. Oliveira, D. A. Silva and E. C. Silva-Filho, Phthalic anhydride esterified chicha gum: characterization and antibacterial activity, *Carbohydr. Polym.*, 2021, **251**, 117077, DOI: [10.1016/j.carbpol.2020.117077](https://doi.org/10.1016/j.carbpol.2020.117077).
- 55 G. Bold, K.-H. Altmann, J. Frei, M. Lang, P. W. Manley, P. Traxler, B. Wietfeld, J. Brü ggen, E. Buchdunger, R. Cozens, S. Ferrari, P. Furet, F. Hofmann, G. Martiny-Baron, J. Mestan, J. Rösel, M. Sills, D. Stover, F. Acemoglu, E. Boss, R. Emmenegger, L. Lässer, E. Masso, R. Roth, Ch. Schlachter, W. Vetterli, D. Wyss and J. M. Wood, New anilinophthalazines as potent and orally well absorbed inhibitors of the VEGF receptor tyrosine kinases useful as antagonists of tumor-driven angiogenesis, *J. Med. Chem.*, 2000, **43**, 2310–2323, DOI: [10.1021/jm9909443](https://doi.org/10.1021/jm9909443).
- 56 F. Chekin and S. Bagheri, Tyrosine sensing on phthalic anhydride functionalized chitosan and carbon nanotube film coated glassy carbon electrode, *Russ. J. Electrochem.*, 2016, **52**, 174–180, DOI: [10.1134/S1023193515120034](https://doi.org/10.1134/S1023193515120034).
- 57 Z. Yin, Q. Sun, L. Chen, M. Du, Y.-Y. Huang, Q. Peng, G. Zhang and D. Zhang, New phthalic anhydride-based room-temperature phosphorescence emitter with lifetime longer than one second, *Adv. Opt. Mater.*, 2023, **11**, 2202224, DOI: [10.1002/adom.202202224](https://doi.org/10.1002/adom.202202224).
- 58 D. Li, Q. Wang, N. Rao, Y. Zhang, Y. Le, L. Liu, L. Li, L. Huang and L. Yan, Development of Imidazo[1,2-*a*]pyridine-based probe for detection of hydrazine and its applications in imaging of HepG2 cell, *J. Mol. Struct.*, 2021, **1239**, 130521, DOI: [10.1016/j.molstruc.2021.130521](https://doi.org/10.1016/j.molstruc.2021.130521).
- 59 M. Edrissi, B. Nasernejad and B. Sayedi, Novel method for the preparation of copper phthalocyanin blue nanoparticles in an electrochemical cell irradiated by microwave, *Int. J. Eng., Trans. B*, 2007, **20**, 257–262.
- 60 A. M. O'Brien and C. Ó Fágáin, Dye bleaching and phenol precipitation by phthalic anhydride-modified horseradish peroxidase, *J. Chem. Technol. Biotechnol.*, 2000, **75**, 363–368, DOI: [10.1002/\(SICI\)1097-4660\(200005\)](https://doi.org/10.1002/(SICI)1097-4660(200005)75:3<363::AID-JCTB1097-4660(200005)>3.0.CO;2-1).
- 61 K. M. Chen and C. C. Tsai, Synthesis and surface activity of maleic anhydride-polyethylene glycol-phthalic anhydride polymeric surfactants, *J. Am. Oil Chem. Soc.*, 1988, **65**, 1346–1349, DOI: [10.1007/BF02542419](https://doi.org/10.1007/BF02542419).
- 62 S. N. C. Ramos, A. L. P. Xavier, F. S. Teodoro, L. F. Gil and L. V. A. Gurgel, Removal of cobalt(II), copper(II), and nickel(II) ions from aqueous solutions using phthalate-functionalized sugarcane bagasse: Monoand multicomponent adsorption in batch mode, *Ind. Crops Prod.*, 2016, **79**, 116–130, DOI: [10.1016/j.indcrop.2015.10.035](https://doi.org/10.1016/j.indcrop.2015.10.035).
- 63 Z.-F. Xu, T. Zhang and W. Hong, O-phthalic anhydride/Zn(OTf)₂ co-catalyzed Beckmann rearrangement under mild conditions, *Tetrahedron*, 2019, **75**, 3113–3117, DOI: [10.1016/j.tet.2019.04.056](https://doi.org/10.1016/j.tet.2019.04.056).
- 64 M. Lutz, M. Wenzler and I. Likhovorik, An efficient oxidation of sulfides to sulfones with urea-hydrogen peroxide in the presence of phthalic anhydride in ethyl acetate, *Synthesis*, 2018, **50**, 2231–2234, DOI: [10.1055/s-0037-1609446](https://doi.org/10.1055/s-0037-1609446).
- 65 M. Khosravi, R. Rahimi, M. Rabbani and E. Safavi, Microwave-assisted solid-phase(SPPS) and solution-phase (SPS) synthesis of biological dipeptide (β-alanine-L histidine), *Quat. J. Appl. Chem. Res.*, 2017, **11**, 8–18.
- 66 K. Liu, X. Zhang and K. Yan, Development of o-phthalic anhydride as a low-temperature activator in H₂O₂ bleaching system for cotton fabric, *Cellulose*, 2018, **25**, 859–867, DOI: [10.1007/s10570-017-1578-1](https://doi.org/10.1007/s10570-017-1578-1).
- 67 D. Maldas and B. V. Kokta, Influence of phthalic anhydride as a coupling agent on the mechanical behavior of wood fiber-polystyrene composites, *J. Appl. Polym. Sci.*, 1990, **41**, 185–194, DOI: [10.1002/app.1990.070410116](https://doi.org/10.1002/app.1990.070410116).
- 68 M. Wang, J. Huang, Y. Xu, B. Han and Zh. Duan, Phthalic anhydride promoted ring-opening polymerization of cyclohexene oxide catalyzed by dinuclear chromium complex supported by piperazine-bridged [ONSO] ligand, *Polym. Int.*, 2019, **68**, 1704–1709, DOI: [10.1002/pi.5874](https://doi.org/10.1002/pi.5874).
- 69 Y. Liu, Z. Wei, T. Xing, M. Lu and X. Li, Performance of Au/FeO_x-TiO₂ catalyst for liquid phase selective hydrogenation of phthalic anhydride to phthalide, *J. Ind. Eng. Chem.*, 2015, **3**, 321–327, DOI: [10.1016/j.jiec.2014.08.036](https://doi.org/10.1016/j.jiec.2014.08.036).



- 70 Y. X. Liu, T. F. Xing, Y. M. Luo, X. N. Li and W. Yan, Au/FeO_x-TiO₂ as an efficient catalyst for the selective hydrogenation of phthalic anhydride to phthalide, *Chin. Chem. Lett.*, 2010, **21**, 1322–1325, DOI: [10.1016/j.ccl.2010.06.042](https://doi.org/10.1016/j.ccl.2010.06.042).
- 71 Y. Liu, T. Xing, Z. Wei, X. Li and W. Yan, Liquid phase selective hydrogenation of phthalic anhydride to phthalide over titania supported gold catalysts, *Catal. Commun.*, 2009, **10**, 2023–2026, DOI: [10.1016/j.catcom.2009.07.021](https://doi.org/10.1016/j.catcom.2009.07.021).
- 72 L. Zhang, X. Chen, Y. Chen, Z. Peng and Ch. Liang, Acid-tolerant Intermetallic Cobalt-nickel silicides as noble metal like catalysts for selective hydrogenation of phthalic anhydride to phthalide, *Catal. Sci. Technol.*, 2019, **9**, 1108–1116, DOI: [10.1039/C8CY02258E](https://doi.org/10.1039/C8CY02258E).
- 73 L. Zhang, X. Chen, S. Jin, J. Guan, Ch. T. Williams, Zh. Peng and Ch. Liang, Rapid microwaves synthesis of CoSi₆/CNTs as novel catalytic materials for hydrogenation of phthalic anhydride, *J. Solid State Chem.*, 2014, **217**, 105–112, DOI: [10.1016/j.jssc.2014.05.021](https://doi.org/10.1016/j.jssc.2014.05.021).
- 74 Y. Liu, Y. Gu, Y. Hou, Y. Yang, S. Deng and Z. Wei, Hydrophobic activated carbon supported Ni-based acid-resistant catalyst for selective hydrogenation of phthalic anhydride to phthalide, *J. Chem. Eng.*, 2015, **275**, 271–280, DOI: [10.1016/j.ccej.2015.04.051](https://doi.org/10.1016/j.ccej.2015.04.051).
- 75 T. Xiao, J. Xie, J. Cheng, X. Dai, S. Lu, R. Zuo, Z. Li and Zh. Yang, Al₂O₃ supported NiCu alloy as a stable catalyst for selective hydrogenation of phthalic anhydride to phthalide, *Appl. Catal., A*, 2023, **660**, 119189, DOI: [10.1016/j.apcata.2023.119189](https://doi.org/10.1016/j.apcata.2023.119189).
- 76 I. M. Al-Jubanawi, H. Z. Al-Sawaad and A. A. Alwaaly, Synthesis characterization and corrosion inhibition of thiourea and phthalic anhydride complex with Ni(II) for carbon steel alloy C1010 0.1 M hydrochloric acid, *Surf. Eng. Appl.*, 2021, **57**, 595–606, DOI: [10.3103/S1068375521050057](https://doi.org/10.3103/S1068375521050057).
- 77 A. S. Fouda, A. M. El-desoky and D. M. Ead, Anhydride derivatives as corrosion inhibitors for carbon steel in hydrochloric acid solutions, *Int. J. Electrochem. Sci.*, 2013, **8**, 8823–8847.
- 78 B. A. R. Zaki, F. L. Rashid and M. N. Al-Baiati, Glycerol/phthalic anhydride novel nano composite for microwave absorbing applications, *Rev. Compos. Mater. Av.*, 2022, **32**, 133–139.
- 79 T. S. Velayutham, W. H. Abd Majid, A. B. Ahmad, G. Y. Kang and S. N. Gan, Synthesis and characterization of polyurethane coatings derived from polyols synthesized with glycerol, phthalic anhydride and oleic acid, *Prog. Org. Coat.*, 2009, **66**, 367371, DOI: [10.1016/j.porgcoat.2009.08.013](https://doi.org/10.1016/j.porgcoat.2009.08.013).
- 80 P. N. Son, The chemistry of phthalic anhydride in cure retardation of rubbers, *Rubber Chem. Technol.*, 1976, **49**, 118–125, DOI: [10.5254/1.3534939](https://doi.org/10.5254/1.3534939).
- 81 P. K. Dubey, S. M. G. Mohiuddin and R. Kumar, Phthalic anhydride - a valuable petrochemical, *Asian J. Chem.*, 1996, **8**, 341–345.
- 82 F. Iordache, S. A. Iordache, I. Costea, C. Simion, A. Simion, D. Mihaescu and F. Badea, Some new applications of phthalic anhydride in organic synthesis, *Rev. Roum. Chim.*, 2002, **46**, 835–845, DOI: [10.1002/CHIN.200246249](https://doi.org/10.1002/CHIN.200246249).
- 83 A. S. Elgharbawy, A Review on phthalic anhydride industry and uses, *Trends Pet. Eng.*, 2021, **1**, 1–2.
- 84 D. A. Basketter and I. Kimber, Phthalic anhydride: illustrating a conundrum in chemical allergy, *J. Immunotoxicol.*, 2016, **13**, 767–769, DOI: [10.1080/1547691X.2016.1177149](https://doi.org/10.1080/1547691X.2016.1177149).
- 85 A. Fatima, Gh. Khanum, A. Sharma, K. Garima, S. Savita, I. Verma, N. Siddiqui and S. Javed, Computational, spectroscopic, Hirschfeld surface, electronic state and molecular docking studies on phthalic anhydride, *J. Mol. Struct.*, 2022, **1249**, 131571, DOI: [10.1016/j.molstruc.2021.131571](https://doi.org/10.1016/j.molstruc.2021.131571).
- 86 J. H. Park, J. Y. Choi, D. J. Son, E. K. Park, M. J. Song, M. Hellström and J. T. Hong, Anti-inflammatory effect of titrated extract of centella asiatica in phthalic anhydride-induced allergic dermatitis animal model, *Int. J. Mol. Sci.*, 2017, **18**, 738, DOI: [10.3390/ijms18040738](https://doi.org/10.3390/ijms18040738).
- 87 E. H. Chester, H. J. Schwartz, C. B. Payne JR and S. Greenstein, Phthalic anhydride asthma, *Clin. Exp. Allergy*, 1977, **7**, 15–20, DOI: [10.1111/j.1365-2222.1977.tb01419.x](https://doi.org/10.1111/j.1365-2222.1977.tb01419.x).
- 88 T. Yanagimoto, The reaction between urea and phthalic anhydride under pressure, *Int. Rev. Phys. Chem.*, 1956, **25**, 71–77.
- 89 K. S. Martin, J. T. Shaw and A. Younai, Anhydride-based multi-component reactions, *Multicomponent reactions in organic synthesis* ed. J. Zhu, Q. Wang and M. Wang, John Wiley & Sons, 2014, ch. 13, pp. 379–400.
- 90 A. R. Katritzky and F. Yates, Phthaloylation of amino-azoles and amino-azines, *J. Chem. Soc., Perkin Trans. 1*, 1976, 309–312, DOI: [10.1039/P19760000309](https://doi.org/10.1039/P19760000309).
- 91 Y. M. Evtushenko, B. E. Zaitsev, V. M. Ivanov and K. M. Gitis, The reaction of maleic and phthalic anhydrides with 2-methylimidazole, *Chem. Heterocycl. Compd.*, 2000, **36**, 1054–1057, DOI: [10.1023/A:1002781815067](https://doi.org/10.1023/A:1002781815067).
- 92 A. R. Hajipour, S. E. Mallakpour and G. Imanzadeh, Microwave-assisted rapid synthesis of phthalimide derivatives, *Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem.*, 2001, **40**, 250–251.
- 93 J. H. Billman and W. F. Harting, Amino acids. V.1 phthalyl derivatives, *J. Am. Chem. Soc.*, 1948, **70**, 1473–1474, DOI: [10.1021/ja01184a051](https://doi.org/10.1021/ja01184a051).
- 94 Q. Zeng, Z. Liu, B. Li and F. Wang, Mild and effective N-phthaloylation of amino acids, *Amino Acids*, 2004, **27**, 183–186, DOI: [10.1007/s00726-004-0109-1](https://doi.org/10.1007/s00726-004-0109-1).
- 95 D. A. Kidd and F. E. King, Preparation of phthalyl-L-glutamic acid, *Nature*, 1984, **162**, 776, DOI: [10.1038/162776b0](https://doi.org/10.1038/162776b0).
- 96 A. C. L. Leite, F. F. Barbosa, M. V. de Oliveira Cardoso, D. R. M. Moreira, L. C. D. Coêlho, E. B. da Silva, G. B. de Oliveira Filho, V. M. O. de Souza, V. R. A. Pereira, L. C. Reis, P. M. P. Ferreira, C. Pessoa, A. Gonc, A. Wanderley, F. V. B. Mota and T. G. da Silva, Phthaloyl amino acids as anti-inflammatory and



- immunomodulatory prototypes, *Med. Chem. Res.*, 2014, **23**, 1701–1708, DOI: [10.1007/s00044-013-0730-1](https://doi.org/10.1007/s00044-013-0730-1).
- 97 A. Saddiqa, I. Shahzadi, M. Usman, O. Çakmak and S. Ökten, Facile, expeditious and cost-effective preparation of N-phthaloyl (S)-amino acids and their in silico activities against Staphylococcus Aureus, *Org. Prep. Proced. Int.*, 2022, **54**, 583–593, DOI: [10.1080/00304948.2022.2113724](https://doi.org/10.1080/00304948.2022.2113724).
- 98 A. K. Bose, F. Greer and Ch. C. Price, A procedure for phthaloylation under mild conditions, *J. Org. Chem.*, 1958, **23**, 1335–1338, DOI: [10.1021/jo01103a025](https://doi.org/10.1021/jo01103a025).
- 99 A. Homsı and A. Kasıdeh, Synthesis of some N-phthalimide amino acids derivatives and evaluation their biological activity, *Int. J. ChemTech Res.*, 2015, **8**, 1817–1825.
- 100 S. D. Khalili, S. H. Banıtabai Koupaie and J. Safari, Lewis acid catalyzed synthesis of quinophthalone pigments under solvent-free conditions, *Sci. Iran.*, 2009, **16**, 1–6.
- 101 H. Loghmani-Khouzani, M. M. Sadeghi and J. Safari, Silica gel catalyzed synthesis of quinophthalone pigments under solvent-free conditions using microwave irradiation, *Molecules*, 2002, **7**, 135–139, DOI: [10.3390/70200135](https://doi.org/10.3390/70200135).
- 102 M. Phillips and M. J. Goss, The preparation and properties of methylisopropyl-quinoline yellow, *J. Am. Chem. Soc.*, 1926, **48**, 823–826, DOI: [10.1021/ja01414a051](https://doi.org/10.1021/ja01414a051).
- 103 J. Safari, S. H. Banıtaba and S. D. Khalili, BF₃-nano SiO₂ as a catalytic system for one-pot green synthesis of pyrophthalone derivatives under microwave conditions, *Arabian J. Chem.*, 2012, **5**, 419–424, DOI: [10.1016/j.arabjc.2010.09.031](https://doi.org/10.1016/j.arabjc.2010.09.031).
- 104 J. R. Cook and D. F. Martin, Chelating tendencies of some substituted 2-(pyridyl)-1,3-indanediones with bivalentmetal ions, *J. Inorg. Nucl. Chem.*, 1964, **26**, 571–577, DOI: [10.1016/0022-1902\(64\)80290-6](https://doi.org/10.1016/0022-1902(64)80290-6).
- 105 M. Bayat, H. Imanieh and M. H. Farjam, Triphenylphosphine-promoted synthesis of spiroketals from phthalic anhydride with dialkyl acetylenedicarboxylates, *Synth. Commun.*, 2010, **40**, 2475–2482, DOI: [10.1080/00397910903267913](https://doi.org/10.1080/00397910903267913).
- 106 S. Sharma, A. D. K. Jain, A. Aggarwal and N. S. Gill, Synthesis, characterization and pharmacological evaluation of novel Schiff bases of imide moiety, *J. Med. Sci.*, 2012, **12**, 61–69, DOI: [10.3923/jms.2012.61.69](https://doi.org/10.3923/jms.2012.61.69).
- 107 P. P. Kumar, Y. D. Reddy, Y. B. Kumari, B. R. Devi and P. K. Dubey, n expeditious synthesis of imides from phthalic, maleic and succinic anhydrides and chemoselective C=C reduction of maleic amide esters, *Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem.*, 2014, **53**, 392–398.
- 108 D. Habibi and H. M. Pordanjani, Phthalimide-N-sulfonic acid, an efficient catalyst for the synthesis of various isoindoline-1,3-dione derivatives, *Chem. Pap.*, 2017, **71**, 2293–2299, DOI: [10.1007/s11696-017-0223-7](https://doi.org/10.1007/s11696-017-0223-7).
- 109 P. W. Chia, P. S. Chee, M. Z. H. Aziz, S. F. M. Radzi and S.-Y. Kan, Water extract of onion peel ash: an efficient green catalytic system for the synthesis of isoindoline-1,3-dione derivatives, *Malaysian J. Anal. Sci.*, 2019, **23**, 23–30, DOI: [10.17576/mjas-2019-2301-03](https://doi.org/10.17576/mjas-2019-2301-03).
- 110 A. Aliabadi, B. Gholamine and T. Karimi, Synthesis and antiseizure evaluation of isoindoline-1,3-dione derivatives in mice, *Med. Chem. Res.*, 2014, **23**, 2736–2743, DOI: [10.1007/s00044-013-0870-3](https://doi.org/10.1007/s00044-013-0870-3).
- 111 I. Ahmad, R. H. Pawara, R. T. Girase, A. Y. Pathan, V. R. Jagatap, N. Desai, Y. O. Ayipo, S. J. Surana and H. Patel, Synthesis, molecular modeling study, and quantum-chemical-based investigations of isoindoline-1,3-diones as antimycobacterial agents, *ACS Omega*, 2022, **7**, 21820–21844, DOI: [10.1021/acsomega.2c01981](https://doi.org/10.1021/acsomega.2c01981).
- 112 H. Jelali, L. Mansour, E. Deniau, M. Sauthier and N. Hamdi, An efficient synthesis of phthalimides and their biological activities, *Polycyclic Aromat. Compd.*, 2022, **4**, 1806–1813, DOI: [10.1080/10406638.2020.1809468](https://doi.org/10.1080/10406638.2020.1809468).
- 113 S. M. Al-Mousawi, M. A. El-Asasery and M. H. Elnagdi, Enaminones in heterocyclic synthesis: a novel route to tetrahydropyrimidines, dihydropyridines, triacylbenzenes and naphthofurans under microwave irradiation, *Molecules*, 2010, **15**, 58–67, DOI: [10.3390/molecules15010058](https://doi.org/10.3390/molecules15010058).
- 114 A. S. Bisht and R. Bisht, Microwave assted synthesis of phthalimide amino derivatives with their antioxidant potential, *Curr. Trends Biotechnol. Pharm.*, 2021, **3**, 23–27, DOI: [10.18231/j.ctppc.2021.007](https://doi.org/10.18231/j.ctppc.2021.007).
- 115 Z. W. Wicks Jr and G.-F. Chen, Reactions of phthalic anhydride with 2-amino alcohols, *J. Org. Chem.*, 1979, **44**, 1244–1247, DOI: [10.1021/jo01322a011](https://doi.org/10.1021/jo01322a011).
- 116 E. Jafari, N. Taghi jarah-Najafabadi, A. Jahanian-Najafabadi, S. Poorirani, F. Hassanzadeh and S. Sadeghian-Rizi, Synthesis and evaluation of antimicrobial activity of cyclic imides derived from phthalic and succinic anhydrides, *Res. Pharm. Sci.*, 2017, **12**, 526–534, DOI: [10.4103/1735-5362.217433](https://doi.org/10.4103/1735-5362.217433).
- 117 B. Hosseinkhani, M. R. Islami and S. Hosseinkhani, Highly Stereoselective synthesis of isoindole derivatives containing an azetidinone ring, *Synlett*, 2015, **26**, 2277–2279, DOI: [10.1055/s-0035-1560066](https://doi.org/10.1055/s-0035-1560066).
- 118 S. Chandrasekhar, M. B. Padmaja and A. Raza, Solid phase-solid state synthesis of N-alkyl imides from anhydrides, *Synlett*, 1999, 1597–1599, DOI: [10.1055/s-1999-2894](https://doi.org/10.1055/s-1999-2894).
- 119 S. Pawar, S. Sawant, A. Nerkar and A. Bhosale, In silico design, synthesis and pharmacological screening of novel 2-(6-substituted benzo[d]thiazol-2-yl)isoindoline-1,3-diones as potential COX-2inhibitors for anti-inflammatory activity, *Int. J. Pharm. Pharm. Sci.*, 2014, **6**, 353–357.
- 120 P. Jimonet, F. Audiau, M. Barreau, J.-Ch. Blanchard, A. Boireau, Y. Bour, M.-A. Coléno, A. Doble, G. Doerflinger, C. D. Huu, M.-H. Donat, J. M. Duchesne, P. Ganil, C. Guérémy, E. Honoré, B. Just, R. Kerphirique, S. Gontier, P. Hubert, P. M. Laduron, J. Le Blevec, M. Meunier, J.-M. Miquet, C. Nemecek, M. Pasquet, O. Piot, J. Pratt, J. Rataud, M. Reibaud, J.-M. Stutzmann and S. Mignani, Riluzole series. synthesis and in Vivo “antigliutamate” activity of 6-substituted-2-benzothiazolamines and 3-substituted-2-imino-benzothiazolines, *J. Med. Chem.*, 1999, **42**(15), 2828–2843, DOI: [10.1021/jm980202u](https://doi.org/10.1021/jm980202u).



- 121 D. Ravi, B. Sakram, K. Ashok, S. Rambabu, B. Sonyanaik, A. Kurumanna and P. Madhu, Novel synthesis of substituted N-(3-aryl-1,8-naphthyridin-2-yl) and N,N'-bis(3-aryl-1,8-naphthyridin-2-yl)benzene-1,4-diamines and 2-(4-((3-aryl-1,8-naphthyridin-2-yl)amino)phenyl)isoindoline-1,3-diones and their biological evaluation, *Synth. Commun.*, 2018, **48**, 50–58, DOI: [10.1080/00397911.2017.1386790](https://doi.org/10.1080/00397911.2017.1386790).
- 122 M.-Z. Huang, K.-L. Huang, Y.-G. Ren, M.-X. Lei, L. Huang, Z.-K. Hou, A.-P. Liu and X.-M. Ou, Synthesis and herbicidal activity of 2-(7-fluoro-3-oxo-3,4-dihydro-2H-benzo[b][1,4]oxazin-6-yl)isoindoline-1,3-diones, *J. Agric. Food Chem.*, 2005, **53**, 7908–7914, DOI: [10.1021/jf051494s](https://doi.org/10.1021/jf051494s).
- 123 K. E. Anwer, G. H. Sayed, H. H. Hassan and M. E. Azab, Conventional and microwave synthesis of some new pyridine derivatives and evaluation their antimicrobial and cytotoxic activities, *Egypt. J. Chem.*, 2019, **62**, 707–726, DOI: [10.21608/ejchem.2018.5115.1452](https://doi.org/10.21608/ejchem.2018.5115.1452).
- 124 S. Y. Mansour, G. H. Sayed, M. I. Marzouk and S. S. Shaban, Synthesis and anticancer assessment of some new 2-amino-3-cyanopyridine derivatives, *Synth. Commun.*, 2021, **51**, 1160–1170, DOI: [10.1080/00397911.2020.1870698](https://doi.org/10.1080/00397911.2020.1870698).
- 125 K. E. Anwer and G. H. Sayed, Conventional and microwave reactions of 1,3-diaryl-5,4-enaminonitrile-pyrazole derivative with expected antimicrobial and anticancer activities, *J. Heterocycl. Chem.*, 2020, 1–15, DOI: [10.1002/jhet.3946](https://doi.org/10.1002/jhet.3946).
- 126 M. El-Shahat, W. I. El-Sofany, A. A. Soliman and M. Hasanin, Newly synthesized imidazolotriazole, imidazolotriazine, and imidazole-pyrazole hybrid derivatives as promising antimicrobial agents, *J. Mol. Struct.*, 2021, **1250**, 131727, DOI: [10.1016/j.molstruc.2021.131727](https://doi.org/10.1016/j.molstruc.2021.131727).
- 127 L. M. Lima, P. Castro, A. L. Machado, C. A. M. Fraga, C. Lugnier, V. L. G. de Moraes and E. J. Barreiro, Synthesis and anti-inflammatory activity of phthalimide derivatives, designed as new thalidomide analogues, *Bioorg. Med. Chem.*, 2002, **10**, 3067–3073, DOI: [10.1016/S0968-0896\(02\)00152-9](https://doi.org/10.1016/S0968-0896(02)00152-9).
- 128 A. K. Timiri, S. Subasri, M. Keshewani, V. Vishwanathan, B. N. Sinha, D. Velmurugan and V. Jayaprakash, Synthesis and molecular modelling studies of novel sulphonamide derivatives as dengue virus 2 protease inhibitors, *Bioorg. Chem.*, 2015, **62**, 74–82, DOI: [10.1016/j.bioorg.2015.07.005](https://doi.org/10.1016/j.bioorg.2015.07.005).
- 129 Z. M. Nofal, E. A. Soliman, S. S. Abd El-Karim, M. I. El-Zahar, A. M. Srour, S. Sethumadhavan and T. J. Maherc, Synthesis of some new benzimidazole-thiazole derivatives as anticancer agents, *J. Heterocycl. Chem.*, 2014, **51**, 1797–1806, DOI: [10.1002/jhet.1886](https://doi.org/10.1002/jhet.1886).
- 130 Y.-N. Sun, C.-L. Wang, N. Zhang, Zh. Wang, Zh.-L. Liu and J.-L. Liu, Synthesis of tetrahydro- β -carboline from phthalic anhydrides and tryptamine, *Chin. Chem. Lett.*, 2014, **25**, 1503–1506, DOI: [10.1016/j.ccllet.2014.06.012](https://doi.org/10.1016/j.ccllet.2014.06.012).
- 131 A. S. Abd El-All, S. M. S. Atta, H. M. F. Roaiah, E. M. Awad and M. M. Abdalla, New potent SARS-CoV 3C-like protease inhibitors derived from thieno[2,3-d]pyrimidine derivatives, *Arch. Pharm. Chem. Life Sci.*, 2016, **349**, 202–210, DOI: [10.1002/ardp.201500407](https://doi.org/10.1002/ardp.201500407).
- 132 M. A. Saleh, Y. A. Hafez, F. E. Abdel-hay and W. I. Gad, Synthesis and Reactions of 3-amino-2-methyl-3H-[1,2,4] triazolo[5,1-b]quinazolin-9-one and 2-hydrazino-3-phenylamino-3H-quinazolin-4-one, *J. Heterocycl. Chem.*, 2003, **40**, 973–978, DOI: [10.1002/jhet.5570400604](https://doi.org/10.1002/jhet.5570400604).
- 133 N. Kise, S. Yamamoto and T. Sakurai, Reductive coupling of phthalic anhydrides with aliphatic ketones by low-valent titanium: unusual two-to-one coupling for preparation of 3,3-disubstituted phthalides, *Tetrahedron*, 2020, **76**, 130820, DOI: [10.1016/j.tet.2019.130820](https://doi.org/10.1016/j.tet.2019.130820).
- 134 M. Lácová, Condensation of 3-indoleacetic acid with phthalic anhydride by the Gabriel and Oglialoro-modification of the Perkin synthesis, *Chem. Pap.*, 1969, **23**, 450–453.
- 135 N. Kise, S. Yamamoto and T. Sakurai, Electroreductive coupling of phthalic anhydrides with α,β -unsaturated carbonyl compounds: synthesis of 1,4-dihydroxynaphthalenes, *J. Org. Chem.*, 2020, **85**, 13973–13982, DOI: [10.1021/acs.joc.0c02000](https://doi.org/10.1021/acs.joc.0c02000).
- 136 K. Fujiwara, T. Kurahashi and S. Matsubara, Nickel-catalyzed decarbonylative cycloaddition of phthalic anhydrides with 1,3-dienes, *Chem. Lett.*, 2011, **40**, 322–323, DOI: [10.1246/cl.2011.322](https://doi.org/10.1246/cl.2011.322).
- 137 (a) E. M. O'Brien, E. A. Bercot and T. Rovis, Decarbonylative cross-coupling of cyclic anhydrides: introducing stereochemistry at an sp^3 carbon in the cross-coupling even, *J. Am. Chem. Soc.*, 2003, **125**, 10498–10499, DOI: [10.1021/ja036290b](https://doi.org/10.1021/ja036290b); (b) J. B. Johnson, E. A. Bercot, J. M. Rowley, G. W. Coates and T. Rovis, Ligand-dependent catalytic cycle and role of styrene in nickel-catalyzed anhydride cross-coupling: evidence for turnover-limiting reductive elimination, *J. Am. Chem. Soc.*, 2007, **129**, 2718–2725, DOI: [10.1021/ja067845g](https://doi.org/10.1021/ja067845g); (c) B. M. Trost and F. Chen, Transition metal mediated eliminations in anhydrides and thioanhydrides, *Tetrahedron Lett.*, 1971, **12**, 2603–2607, DOI: [10.1016/S0040-4039\(01\)96930-8](https://doi.org/10.1016/S0040-4039(01)96930-8); (d) K. Sano, T. Yamamoto and A. Yamamoto, Kinetic and thermodynamic control of asymmetric induction by chiral diphosphine in ring contraction of (chiral diphosphine) $NiCH_2CH_2CH_2COO$ to (chiral diphosphine) $NiCH(CH_3)CH_2COO$, *Chem. Lett.*, 1984, **13**, 941–944, DOI: [10.1246/cl.1984.941](https://doi.org/10.1246/cl.1984.941); (e) K. Sano, T. Yamamoto and A. Yamamoto, preparation of Ni- or Pt-containing cyclic esters by oxidative addition of cyclic carboxylic anhydrides and their properties, *Bull. Chem. Soc. Jpn.*, 1984, **57**, 2741–2747; (f) T. Yamamoto, K. Sano and A. Yamamoto, Effect of ligand on ring contraction of six-membered nickel-containing cyclic esters, $LnNiCH_2CH_2CH_2COO$, to their five-membered-ring isomers, $LnNiCH(CH_3)CH_2COO$. kinetic and thermodynamic control of asymmetric induction by chiral diphosphines in the ring contraction, *J. Am. Chem. Soc.*, 1987, **109**, 1092–1100, DOI: [10.1021/ja00238a017](https://doi.org/10.1021/ja00238a017); (g) R. Fischer, D. Walther, R. Kempe, J. Sieler and B. Schönecker, Metallacyclische acyl-carboxylate des nickels: Reaktivität der acylgruppe und synthesespotential bei kreuzkopplungsreaktionen mit alkyhalogeniden, *J. Organomet. Chem.*, 1993, **447**, 131–



- 136, DOI: [10.1016/0022-328X\(93\)80282-G](https://doi.org/10.1016/0022-328X(93)80282-G); (h) K. Fujiwara, T. Kurahashi and S. Matsubara, Decarbonylative cycloaddition of phthalimides with 1,3-dienes, *Org. Lett.*, 2010, **12**, 4548–4551, DOI: [10.1021/ol101842y](https://doi.org/10.1021/ol101842y).
- 138 L. Huang, R. Xie, Ch. Wen, Y. Yang, Y. Wang, S. Ren, B. Huang and S. Li, Decarbonylative/decarboxylative [4+2] annulation of phthalic anhydrides and cyclic iodonium towards triphenylenes, *Org. Biomol. Chem.*, 2022, **20**, 3913–3916, DOI: [10.1039/D2OB00500J](https://doi.org/10.1039/D2OB00500J).
- 139 M. P. Satchel and B. E. Stacey, Reactions of 1,5-disubstituted naphthalenes with phthalic anhydride, *J. Chem. Soc. C*, 1971, 469–474, DOI: [10.1039/J39710000469](https://doi.org/10.1039/J39710000469).
- 140 H. Naeimi and S. S. Brojerdi, Facile and efficient one-pot synthesis of anthraquinones from benzene derivatives catalyzed by silica sulfuric acid, *Polycyclic Aromat. Compd.*, 2014, **34**, 504–517, DOI: [10.1080/10406638.2014.910238](https://doi.org/10.1080/10406638.2014.910238).
- 141 C. Ji, H. Wang, S. Sun, Q. Li, J.-J. Zou, S. Yu, H. Shi, G. Nie and S. Liu, One-pot synthesis of 2-ethylanthraquinone from phthalic anhydride and ethylbenzene over a Sc-modified H β catalyst, *Chem. Eng. Sci.*, 2022, **251**, 117480, DOI: [10.1016/j.ces.2022.117480](https://doi.org/10.1016/j.ces.2022.117480).
- 142 Y. Wang, G. Wang, Y. Yan, X. Zhao and Y. Wang, Acylation reaction of benzene with phthalic anhydride catalyzed by amide-AlCl₃ ionic liquid analogs, *Acta Pet. Sin.*, 2022, **3**, 1001–8719, DOI: [10.3969/j.issn.1001-8719.2022.02.008](https://doi.org/10.3969/j.issn.1001-8719.2022.02.008).
- 143 M. L. Tedjamulia, Y. Tominaga, R. N. Castle and M. L. Lee, The synthesis of dinaphthothiophenes, *J. Heterocycl. Chem.*, 1983, **20**, 1143–1148, DOI: [10.1002/jhet.5570200502](https://doi.org/10.1002/jhet.5570200502).
- 144 H. Naeimi and R. Namdari, Rapid, efficient and one pot synthesis of anthraquinone derivatives catalyzed by Lewis acid/methanesulfonic acid under heterogeneous conditions, *Dyes Pigm.*, 2009, **81**, 259–263, DOI: [10.1016/j.dyepig.2008.10.019](https://doi.org/10.1016/j.dyepig.2008.10.019).
- 145 B. R. Madje, K. F. Shelke, S. B. Sapkal, G. K. Kakade and M. S. Shingare, An efficient one-pot synthesis of anthraquinone derivatives catalyzed by alum in aqueous media, *Green Chem. Lett. Rev.*, 2010, **3**, 269–273, DOI: [10.1080/17518251003776877](https://doi.org/10.1080/17518251003776877).
- 146 N. Shafiq, S. Ashraf, S. Parveen and B. Ali, A one-pot rapid synthesis, docking study and biological evaluation of some tetrahydropyrimidine-5-carboxylates, *Indian J. Heterocycl. Chem.*, 2019, **29**, 249–253.
- 147 G. Zarren, N. Shafiq, U. Arshad, N. Rafiq, S. Parveen and Z. Ahmad, Copper-catalyzed one-pot relay synthesis of anthraquinone based pyrimidine derivative as a probe for antioxidant and antidiabetic activity, *J. Mol. Struct.*, 2021, **1227**, 129668, DOI: [10.1016/j.molstruc.2020.129668](https://doi.org/10.1016/j.molstruc.2020.129668).
- 148 M. Phillips, The preparation of alizarin from phthalic anhydride and ortho-dichlorobenzene, *J. Am. Chem. Soc.*, 1927, **49**, 473–478, DOI: [10.1021/ja01401a021](https://doi.org/10.1021/ja01401a021).
- 149 C. Sprent and W. Dodd, *Brit. Pat.*, 1923, vol. 204, p. 528.
- 150 P. P. Kumar, Y. D. Reddy, C. V. R. Reddy, B. R. Devi and P. K. Dubey, One pot, three-component synthesis of novel 3,4-dihydrophthalazin-2(1H)-yl-4-phenyl-4H-pyrans, *Tetrahedron Lett.*, 2014, **55**, 2177–2182, DOI: [10.1016/j.tetlet.2014.02.020](https://doi.org/10.1016/j.tetlet.2014.02.020).
- 151 M. R. Mahmoud, W. S. I. Abou-Elmagd, H. A. Derbala and M. H. Hekal, Synthesis and spectral characterisation of some phthalazinone derivatives, *J. Chem. Res.*, 2012, 75–82.
- 152 V. M. Milovanovi, Z. D. Petrovi, S. Novakovi, G. A. Bogdanovi, V. P. Petrovi and D. Simijonovi, Green synthesis of benzamide-dioxoisindoline derivatives and assessment of their radical scavenging activity-experimental and theoretical approach, *Tetrahedron*, 2020, **76**, 131456, DOI: [10.1016/j.tet.2020.131456](https://doi.org/10.1016/j.tet.2020.131456).
- 153 A. A. Aly, M. A. Abo-Riya, A. A. El-Shiekh and A. I. Elkala, Synthesis and biological evaluation of new series of phthalazinone, pyridazinone, oxadiazole, pyrazolone scaffolds based on coumarin-3-carbohydrazide, *Benha J. Appl. Sci.*, 2022, **7**, 21–29.
- 154 M. H. Hekal, F. S. M. A. El-Azm and S. R. Atta-Allah, Ecofriendly and highly efficient microwaveinduced synthesis of novel quinazolinone-undecyl hybrids with in vitro antitumor activity, *Synth. Commun.*, 2019, **49**, 2630–2641, DOI: [10.1080/00397911.2019.1637001](https://doi.org/10.1080/00397911.2019.1637001).
- 155 D. Gao, F. Jin, X. Yan and R. N. Zare, Selective synthesis in microdroplets of 2-phenyl-2,3-dihydrophthalazine-1,4-dione from phenyl hydrazine with phthalic anhydride or phthalic acid, *Chem. - Eur. J.*, 2019, **25**, 1466–1471, DOI: [10.1002/chem.201805585](https://doi.org/10.1002/chem.201805585).
- 156 S. F. Abbas, J. H. Tomma and E. T. Ali, Synthesis and characterization of new Schiff bases and their 1,3-oxazepines derived from phthalic anhydride, *Sys. Rev. Pharm.*, 2021, **12**, 260–265.
- 157 H. D. Hanoon, Synthesis and characterization of new seven-membered heterocyclic compounds from reaction of new Schiff-bases with maleic and phthalic anhydrides, *J. Chem.*, 2011, **41**, 77–89.
- 158 T. Serevičius, R. Komskis, P. Adome' nas, O. Adomeniene, V. Jankauskas, A. Gruodis, K. Kazlauskas and S. Juršėnas, Non-symmetric 9,10-diphenylanthracene-based deep-blue emitters with enhanced charge transport properties, *Phys. Chem. Chem. Phys.*, 2014, **16**, 7089–7101, DOI: [10.1039/c4cp00236a](https://doi.org/10.1039/c4cp00236a).
- 159 L. Cheng, L. Ying, J. Feng, C. H. Y. Wang, J. L. Li and Z. Xu, Novel heterocyclic poly(arylene ether ketone)s: synthesis and polymerization of 4-(40-hydroxyaryl)(2H)phthalazin-1-ones with methyl groups, *J. Polym. Sci., Part A: Polym. Chem.*, 2007, **45**, 1525–1535, DOI: [10.1002/pola.21880](https://doi.org/10.1002/pola.21880).
- 160 S.-N. Zhang, Zh.-J. Li and M.-S. Cai, Synthesis of N-sugar-substituted phthalimides and their derivatives from sugar azides and phthalic anhydride, *Carbohydr. Res.*, 2004, **339**, 1419–1420, DOI: [10.1016/j.carres.2004.03.032](https://doi.org/10.1016/j.carres.2004.03.032).
- 161 A. Kamal, E. Laxman, N. Laxman and N. V. Rao, Efficient one-pot synthesis of n-substituted phthalimides~aphthalimides from azides and anhydrides by iodotrimethylsilane, *Tetrahedron Lett.*, 1998, **39**, 8733–8734, DOI: [10.1016/S0040-4039\(98\)01937-6](https://doi.org/10.1016/S0040-4039(98)01937-6).
- 162 H. N. Nguyen, V. J. Cee, H. L. Deak, B. Du, K. P. Faber, H. Gunaydin, B. L. Hodous, S. L. Hollis, P. H. Krolkowski, Ph. R. Olivieri, V. F. Patel, K. Romero, L. B. Schenkel and S. D. Geuns-Meyer, Synthesis of 4-substituted chlorophthalazines,



- dihydrobenzoazepinediones, 2-pyrazolylbenzoic acid, and 2-pyrazolylbenzohydrazide *via* 3-substituted-3-hydroxyisoindolin-1-ones, *J. Org. Chem.*, 2012, 77, 3887–3906, DOI: [10.1021/jo3000628](https://doi.org/10.1021/jo3000628).
- 163 A. Acosta, P. de la Cruz, P. De Miguel, E. Diez-Barra, A. de la Hoz, F. Langa, A. Loupy, M. Majdoub, N. Martin, C. Sanchez and C. Seoane, Microwave assisted synthesis of heterocyclic fused quinones in dry media, *Tetrahedron Lett.*, 1995, 36, 2165–2168, DOI: [10.1016/0040-4039\(95\)00202-N](https://doi.org/10.1016/0040-4039(95)00202-N).
- 164 W. E. Parham and R. M. Piccirilli, Preparation of arylbenzoic acid. reaction of aryllithium reagents with phthalic anhydride, *J. Org. Chem.*, 1976, 41, 1268–1269, DOI: [10.1021/jo00869a046](https://doi.org/10.1021/jo00869a046).
- 165 J. Koyanagi, K. Yamamoto, K. Nakayama and A. Tanaka, A short-step synthesis of naphtho[2,3-b]furan-4,9-dione, *J. Heterocycl. Chem.*, 1994, 31, 1303–1304, DOI: [10.1002/jhet.5570310534](https://doi.org/10.1002/jhet.5570310534).
- 166 P. de la Cruz, N. Martin, F. Miguel and C. Seoane, Novel pi-extended thiophene-fused electron acceptors for organic metals, *J. Org. Chem.*, 1992, 57, 6192–6198, DOI: [10.1021/jo00049a027](https://doi.org/10.1021/jo00049a027).
- 167 L. Fieser and E. B. Hershberg, 1',2',3',4'-Tetrahydro-10-isopropyl-1,2-benzanthracene, *J. Am. Chem. Soc.*, 1937, 59(11), 2331–2335, DOI: [10.1021/ja01290a069](https://doi.org/10.1021/ja01290a069).
- 168 Y. Baba, The reaction of triethylaluminum with phthalic anhydride, *Bull. Chem. Soc. Jpn.*, 1968, 41, 1020–1021, DOI: [10.1246/bcsj.41.1020](https://doi.org/10.1246/bcsj.41.1020).
- 169 H. Gilman and J. F. Nelson, Relative reactivities of organometallic compounds. IX. cadmium, *Recl. Trav. Chim.*, 1936, 55, 518–530, DOI: [10.1002/recl.19360550607](https://doi.org/10.1002/recl.19360550607).
- 170 P. L. de Benneville, The synthesis of keto acids and ketones by the reaction of acid anhydrides with cadmium alkyls, *J. Org. Chem.*, 1941, 6, 462–466, DOI: [10.1021/jo01203a012](https://doi.org/10.1021/jo01203a012).
- 171 T. C. McMullen, The Friedel and Crafts reaction with phthalic anhydride, *J. Am. Chem. Soc.*, 1922, 44(9), 2055–2060, DOI: [10.1021/ja01430a027](https://doi.org/10.1021/ja01430a027).
- 172 M. T. Rahman and S. K. Nahar, The reaction of arylcoppermagnesium and other organometallic reagents with phthalic anhydride, *J. Organomet. Chem.*, 1992, 425, 201–208, DOI: [10.1016/0022-328X\(92\)80036-W](https://doi.org/10.1016/0022-328X(92)80036-W).
- 173 E.-Ch. Wang, K.-S. Huang, H.-M. Chen, Ch.-Ch. Wu and G.-J. Lin, An Efficient method for the preparation of nitriles *via* the dehydration of aldoximes with phthalic anhydride, *J. Chin. Chem. Soc.*, 2004, 51, 619–627, DOI: [10.1002/jccs.200400093](https://doi.org/10.1002/jccs.200400093).
- 174 M. M. Heravi, D. Zargarani, R. Hekmat Shoar and S. Khaleghi, Hydrolysis of amides to carboxylic acids using phthalic anhydride under microwave irradiation and solvent-free conditions, *J. Chem. Res.*, 2005, 119–120.
- 175 S. H. Kadhum, W. B. Ali and A. A. Jarullah, Synthesis and characterization of some compounds derived from 2-(1,1-dimethyl-1,3-dihydro-2H-benzo[e]indol-2-ylidene) malonaldehyde, *HIV Nurs.*, 2022, 22, 3625–3631.
- 176 Y. Ochi, T. Kurahashi and S. Matsubara, Decarbonylative cycloaddition of phthalic anhydrides with allenes, *Org. Lett.*, 2011, 13, 1374–1377, DOI: [10.1021/ol200044y](https://doi.org/10.1021/ol200044y).
- 177 M. Lácová, Reaction of 2-benzothiazolylthioethanoic acid with phthalic anhydride under conditions of Gabriel modification of Perkin synthesis, *Chem. Pap.*, 1986, 40, 95–102.
- 178 H. Santos, A. Distiller, A. M. D'Souza, Q. Arnoux, J. M. White, A. G. Meyer and J. H. Ryan, 1,3-Dipolar cycloadditions with an azomethine ylide, *Org. Chem. Front.*, 2015, 2, 705–712, DOI: [10.1039/C5QO00062A](https://doi.org/10.1039/C5QO00062A).
- 179 A. H. Renfrew and S. B. Bostock, Reaction of phthalic anhydride with ethyl cyanoacetate: a route to the 2H-indeno[2,1-c]pyridine-1,9-dione, 2H-indeno[2,1-c]pyridine-3,9-dione, 2H-indeno[2,1-c]pyridazine-3,9-dione, and indeno[2,1-c]pyran-1,9-dione systems, *J. Chem. Soc., Perkin Trans. 1*, 1977, 84–90, DOI: [10.1039/P19770000084](https://doi.org/10.1039/P19770000084).
- 180 F. Ramirez, H. Yamanaka and O. H. Basedow, New reactions of phosphite esters: the conversion of phthalic anhydride into biphtalyl by trialkyl phosphites and into phthalide-3-phosphonates by dialkyl phosphites, *J. Am. Chem. Soc.*, 1961, 83, 173–178, DOI: [10.1021/ja01462a034](https://doi.org/10.1021/ja01462a034).
- 181 D. Habibi and O. Marvi, Montmorillonite KSF and montmorillonite K-10 clays as efficient catalysts for the solventless synthesis of bismaleimides and bisphthalimides using microwave irradiation, *Arkivoc*, 2006, 8–15, DOI: [10.3998/ark.5550190.0007.d02](https://doi.org/10.3998/ark.5550190.0007.d02).
- 182 D. Habibi, N. Mahmoudi and O. Marvi, Green Procedure for the synthesis of phthalazino[2,3-b]phthalazine-5,7,12,14-tetraones, *Synth. Commun.*, 2007, 37, 3165–3171, DOI: [10.1080/00397910701545247](https://doi.org/10.1080/00397910701545247).
- 183 M. C. Cardia, S. Distinto, E. Maccioni, L. Bonsignore and A. DeLogu, Synthesis and biological activity evaluation of differently substituted 1,4-dioxo-3,4-dihydrophthalazine-2(1H)-carboxamides and -carbothioamides, *J. Heterocycl. Chem.*, 2003, 40, 1011–1015, DOI: [10.1002/jhet.5570400609](https://doi.org/10.1002/jhet.5570400609).
- 184 F. Jafarpour, H. Hazrati and S. Nouraldinmoussa, Three-bond breaking of cyclic anhydrides: easy access to polyfunctionalized naphthalenes and phenanthrenes, *Org. Lett.*, 2013, 15, 3816–3819, DOI: [10.1021/ol401318v](https://doi.org/10.1021/ol401318v).
- 185 M. Fardpour, A. Darvish, E. Kianmehr and A. N. Kharat, Ru-catalyzed synthesis of substituted phthalides through C–H bond activation and functionalization, *Tetrahedron Lett.*, 2019, 60, 699–702, DOI: [10.1016/j.tetlet.2018.12.005](https://doi.org/10.1016/j.tetlet.2018.12.005).
- 186 B. H. S. Torquato da Silva, L. O. Riehl, G. Carvalho dos Santos, J. C. Roldao and L. C. Silva-Filho, Solvent-free synthesis using Nb₂O₅ and a theoretical-experimental study of solvent effect in new rhodamine dyes, *ChemistrySelect*, 2020, 5, 1455–1463, DOI: [10.1002/slct.201904486](https://doi.org/10.1002/slct.201904486).
- 187 H. Eshghi, M. Eftekhari, M. Rahimizadeh, M. Hassanpour and M. Bakavoli, Efficient Solvent free synthesis of 12-aryl-8,9,10,12-tetrahydrobenzo[a]-xanthen-11-ones and 3H-spiro[isobenzofuran-1,9'-xanthen]-3-one derivatives using cobalt hydrogensulfate as a green, heterogeneous and reusable catalyst, *J. Org. Chem. Res.*, 2016, 2, 20–27, DOI: [10.22036/ORG.CHEM.2016.13465](https://doi.org/10.22036/ORG.CHEM.2016.13465).
- 188 A. Shaabani, M. B. Teimouri and H. R. Bijanzadeh, The reaction of Alkyl Isocyanides and Dialkylacetylene Dicarboxylates with Phthalic Anhydride Derivatives: A



- Novel Synthesis of γ -spiroiminolactones, *J. Chem. Res.*, 2002, **2002**, 381–383, DOI: [10.3184/030823402103172383](https://doi.org/10.3184/030823402103172383).
- 189 A. Shaabani, A. H. Rezayan, S. Ghasemi and A. Sarvary, A mild and efficient method for the synthesis of 2,5-dihydro-5-imino-2-methylfuran-3,4-dicarboxylates *via* an isocyanide-based multicomponent reaction, *Tetrahedron Lett.*, 2009, **50**, 1456–1458, DOI: [10.1016/j.tetlet.2009.01.069](https://doi.org/10.1016/j.tetlet.2009.01.069).
- 190 L. Zare, N. O. Mahmoodi, A. Yahyazadeh, M. Mamaghani and K. Tabatabaieian, An efficient chemo- and regioselective three-component synthesis of pyridazinones and phthalazinones using activated KSF, *Chin. Chem. Lett.*, 2010, **21**, 538–541, DOI: [10.1016/j.ccllet.2009.11.032](https://doi.org/10.1016/j.ccllet.2009.11.032).
- 191 L. Zare, N. O. Mahmoodi, A. Yahyazadeh and M. Nikpassand, Ultrasound-promoted regio and chemoselective synthesis of pyridazinones and phthalazinones catalyzed by ionic liquid [bmim]Br/AlCl₃, *Ultrason. Sonochem.*, 2012, **19**, 740–744, DOI: [10.1016/j.ultsonch.2011.11.008](https://doi.org/10.1016/j.ultsonch.2011.11.008).
- 192 B. Thirupaiah and R. R. Vedula, Novel one-pot multicomponent synthesis of substituted 2,3-dihydro-2-(6-(4-hydroxy-6-methyl-2-oxo-2H-pyran-3-yl)-7H-[1,2,4]triazolo[3,4-b][1,3,4]thiadiazin-3-yl)phthalazine-1,4-diones and substituted 3-[3-(N'-benzylidene-hydrazino)-7H-[1,2,4]triazolo[3,4-b][1,3,4]thiadiazin-6-yl]-4-hydroxy-6-methylpyran-2-ones, *Synth. Commun.*, 2014, **44**, 513–519, DOI: [10.1080/00397911.2013.817584](https://doi.org/10.1080/00397911.2013.817584).
- 193 V. S. R. Chunduru and R. R. Vedula, One-pot synthesis of aryl(hetaryl)-thiazolylphthalazine-1,4-diones *via* a multicomponent approach, *Chem. Heterocycl. Compd.*, 2013, **49**, 586–590, DOI: [10.1007/s10593-013-1285-2](https://doi.org/10.1007/s10593-013-1285-2).
- 194 V. S. R. Chunduru and V. R. Rao, Synthesis of aryl and heteryl 1,3,4-thiadiazinyl-phthalazine-1,4-dione derivatives *via* a multicomponent approach, *Synth. Commun.*, 2013, **43**, 923–929, DOI: [10.1080/00397911.2011.604147](https://doi.org/10.1080/00397911.2011.604147).
- 195 K. Sujatha and R. R. Vedula, Novel one-pot multicomponent synthesis of (E)-2-(benzylideneamino)-5-mercapto-4H-1,2,4-triazolo-3-yl-2,3-dihydrophthalazine-1,4-dione derivatives, *Synth. Commun.*, 2019, **50**, 129–135, DOI: [10.1080/00397911.2019.1689571](https://doi.org/10.1080/00397911.2019.1689571).
- 196 P. C. Jilloju, M. Srikanth, S. V. Kumar and R. R. Vedula, One-pot, multi-component synthesis of substituted 2-(6-phenyl-7H-[1,2,4]triazolo[3,4-b][1,3,4]thiadiazin-3-yl)-2,3-dihydrophthalazine-1,4-diones, *Polycyclic Aromat. Compd.*, 2021, **41**, 2043–2054, DOI: [10.1080/10406638.2019.1709212](https://doi.org/10.1080/10406638.2019.1709212).
- 197 L. Z. Fekri and M. F. Feshalami, Green, practical and scalable multicomponent reaction for the synthesis of amides and pyridazinones from arenes using L-Proline functionalized silicapropyl modified nanomagnetic as a heterogeneous bronsted acid catalyst, *Polycyclic Aromat. Compd.*, 2022, **42**, 690–710, DOI: [10.1080/10406638.2020.1749091](https://doi.org/10.1080/10406638.2020.1749091).
- 198 L. Z. Fekri and M. F. Feshalami, A new multicomponent reaction for the synthesis of amides, *Org. Prep. Proced. Int.*, 2021, **53**, 291–300, DOI: [10.1080/00304948.2021.1880217](https://doi.org/10.1080/00304948.2021.1880217).
- 199 C. Bele and M. Darabantu, New 1,4-disubstituted phthalazines: synthesis, structure and herbicidal evaluation, *Heterocycl. Commun.*, 2003, **9**, 641–646, DOI: [10.1515/HC.2003.9.6.641](https://doi.org/10.1515/HC.2003.9.6.641).
- 200 E. Benjamin and Y. M. Hijji, A Novel green synthesis of thalidomide and analogs, *J. Chem.*, 2017, **2017**, 6436185, DOI: [10.1155/2017/6436185](https://doi.org/10.1155/2017/6436185).
- 201 J. Garcia, J. Vilarrasa, X. Bordas and A. Banaszek, New synthetic “tricks” one-pot preparation of N-substituted phthalimides from azides and phthalic anhydride, *Tetrahedron Lett.*, 1986, **27**, 639–640, DOI: [10.1016/S0040-4039\(00\)84061-7](https://doi.org/10.1016/S0040-4039(00)84061-7).
- 202 E. Deniau, D. Enders, A. Couture and P. Grandclaudeon, Asymmetric synthesis of 3-hetero-substituted 2,3-dihydro-1H-isoindol-1-ones, *Tetrahedron: Asymmetry*, 2005, **16**, 875–881, DOI: [10.1016/j.tetasy.2005.01.012](https://doi.org/10.1016/j.tetasy.2005.01.012).
- 203 E. Deniau, D. Enders, A. Couture and P. Grandclaudeon, A new synthetic route to highly enantioenriched 3-substituted-2,3-dihydro-1H-isoindol-1-ones, *Tetrahedron: Asymmetry*, 2003, **14**, 2253–2258, DOI: [10.1016/S0957-4166\(03\)00409-9](https://doi.org/10.1016/S0957-4166(03)00409-9).
- 204 K. Kalpana, R. Syed, J. Saranya, M. Rafi and B. Ravi Kiran, Synthesis and theoretical study of novel imidazo[4,5-b]pyrazine-conjugated benzamides as potential anticancer agents, *Russ. J. Org. Chem.*, 2021, **57**, 1487–1494, DOI: [10.1134/S1070428021090153](https://doi.org/10.1134/S1070428021090153).
- 205 M. K. Rasheed, D. A. Al-rifaie and D. I. Madab, Synthesis and evaluation of antibacterial and antifungal activity of new 1,5-Benzodiazepine derivatives contain cyclic imides and Mannich bases, *Int. J. Pharm. Res.*, 2021, **13**, 2175, DOI: [10.31838/ijpr/2021.13.01.348](https://doi.org/10.31838/ijpr/2021.13.01.348).
- 206 H. H. Eissa, Synthesis of bisphthaldicarboximide *via* the reaction of phthalic anhydridewith diamines and evaluation of its biological activity, *Int. J. ChemTech Res.*, 2014, **6**, 95–101.
- 207 X.-T. Min, D.-W. Ji, H. Zheng, B.-Zh. Chen, Y.-Ch. Hu, B. Wan and Q.-A. Chen, Cobalt-catalyzed regioselective carboamidation of alkynes with imides enabled by cleavage of C–N and C–C bonds, *Org. Lett.*, 2020, **22**, 3386–3391, DOI: [10.1021/acs.joc.9b00926](https://doi.org/10.1021/acs.joc.9b00926).
- 208 H. Chen, L. Ouyang, J. Liu, W.-J. Shi, G. Chen and L. Zheng, Synthesis of multisubstituted 1-naphthoic acids *via* Ru-catalyzed C–H activation and double-alkyne annulation under air, *J. Org. Chem.*, 2019, **84**, 12755–12763, DOI: [10.1021/acs.joc.9b00926](https://doi.org/10.1021/acs.joc.9b00926).
- 209 G. Singh, A. Saroa, S. Girdhar, S. Rani, S. Sahoo and D. Choquesillo-Lazarte, Synthesis, characterization, electronic absorption and antimicrobial studies of N-(silatranylpropyl)phthalimide derived from phthalic anhydride, *Inorg. Chim. Acta*, 2015, **427**, 232–239, DOI: [10.1016/j.ica.2015.01.011](https://doi.org/10.1016/j.ica.2015.01.011).
- 210 M. S. Malik, R. I. Alsantali, Q. M. S. Jamal, Z. S. Seddigi, M. Morad, M. A. Alsharif, E. M. Hussein, R. S. Jassas, M. M. Al-Rooqi, Z. Abduljaleel, A. O. Babalghith, H. M. Altass, Z. Moussa and S. A. Ahmed, New imidazole-based N-phenylbenzamide derivatives as potential



- anticancer agents: key computational insights, *Front. Chem.*, 2022, **9**, 808556, DOI: [10.3389/fchem.2021.808556](https://doi.org/10.3389/fchem.2021.808556).
- 211 S. Mingzhou, Zh. Lanlan, Ch. Miaomiao, H. Wangcheng, H. Xinwei and L. Hongjian, Synthesis of esterified/fused isocoumarins via Rh-catalyzed C-H activation/transannulative coupling/annulation of phthalic anhydrides with cyclic 2-diazo-1,3-diketones and methanol, *Chin. J. Org. Chem.*, 2022, **42**, 3816–3823, DOI: [10.6023/cjoc202205039](https://doi.org/10.6023/cjoc202205039).
- 212 H. R. Shaterian and M. Mohammadnia, Mild basic ionic liquids catalyzed new four-component synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-diones, *J. Mol. Liq.*, 2012, **173**, 55–61, DOI: [10.1016/j.molliq.2012.06.007](https://doi.org/10.1016/j.molliq.2012.06.007).
- 213 H. R. Shaterian and M. Mohammadnia, Mild preparation of 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives with magnetic Fe₃O₄ nanoparticles coated by (3-aminopropyl)-triethoxysilane as catalyst under ambient and solvent-free conditions, *Res. Chem. Intermed.*, 2014, **40**, 371–383, DOI: [10.1007/S11164-012-0969-Z/TABLES/3](https://doi.org/10.1007/S11164-012-0969-Z/TABLES/3).
- 214 J. Safaei-Ghomi, H. Shahbazi-Alavi, A. Ziarati, R. Teymuri and M. R. Saberi, A highly flexible green synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives with CuI nanoparticles as catalyst under solvent-free conditions, *Chin. Chem. Lett.*, 2014, **25**, 401–405, DOI: [10.1016/j.cclet.2013.11.046](https://doi.org/10.1016/j.cclet.2013.11.046).
- 215 R. Ghorbani-Vaghei, J. Mahmoodi and Y. Maghbooli, Preparation and characterization of nanomagnetic piperidinium benzene-1,3-disulfonate ionic liquid as a novel, green and heterogeneous catalyst and its use in the synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-diones and 1H-pyrazolo[1,2-a]pyridazine-5,8-diones under solvent-free conditions, *Appl. Organomet. Chem.*, 2017, **31**, e3717, DOI: [10.1002/aoc.3717](https://doi.org/10.1002/aoc.3717).
- 216 F. Mohamadpour, M. Lashkari, R. Heydari and N. Hazeri, Four-component clean process for the eco-friendly synthesis of 1H-pyrazolo [1,2-b]phthalazine-5,10-dione derivatives using Zn(OAc)₂·2H₂O as an efficient catalyst under solvent-free conditions, *Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem.*, 2018, **57**, 843–851.
- 217 F. Mohamadpour, Carboxymethyl cellulose (CMC) as a recyclable green catalyst promoted eco-friendly protocol for the solvent-free synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives, *Polycyclic Aromat. Compd.*, 2022, **42**, 1091–1102, DOI: [10.1080/10406638.2020.1768412](https://doi.org/10.1080/10406638.2020.1768412).
- 218 F. Mohamadpour, M. T. Maghsoodlou, R. Heydari and M. Lashkari, Copper(II) acetate monohydrate: an efficient and ecofriendly catalyst for the one-pot multi-component synthesis of biologically active spiroopyrans and 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives under solvent-free conditions, *Res. Chem. Intermed.*, 2016, **42**, 7841–7853, DOI: [10.1007/s11164-016-2565-0](https://doi.org/10.1007/s11164-016-2565-0).
- 219 M. Hosseinpour Tehrani, S. A. Mirshokraie, M. Khoobi and M. Amini, Synthesis of pyrazolo[1,2-b]phthalazine-5,10-dione derivatives: a new class of α -glucosidase inhibitors, *Pharm. Biomed. Res.*, 2021, **7**, 121–132, DOI: [10.18502/pbr.v7i2.7365](https://doi.org/10.18502/pbr.v7i2.7365).
- 220 S. R. Kolsepatil, A. V. Sapkal, S. Chandole and D. L. Lingampalle, One pot synthesis of pyrazolo phthalazine dione derivatives under microwave irradiation, *Rasayan J. Chem.*, 2019, **12**, 415–420.
- 221 M. Abdesheikhi and Z. Karimi-Jaberi, Four-component synthesis of 3-amino-1-aryl-5,10-dioxo-1H-pyrazolo[1,2-b]phthalazine-2-carbonitrile derivatives promoted by potassium carbonate, *J. Chem. Res.*, 2015, **39**, 482–483, DOI: [10.3184/174751915X14382558688543](https://doi.org/10.3184/174751915X14382558688543).
- 222 M. Kalhor, Z. Orouji and M. Khalaj, 4-Methylpyridinium chloride ionic liquid grafted on Mn@zeolite-Y: design, fabrication and performance as a novel multi-functional nanocatalyst in the four-component synthesis of pyrazolophthalazine-diones, *Microporous Mesoporous Mater.*, 2022, **329**, 111498, DOI: [10.1016/j.micromeso.2021.111498](https://doi.org/10.1016/j.micromeso.2021.111498).
- 223 N. Kerru, L. Gummidi, S. V. H. S. Bhaskaruni, S. N. Maddila and S. B. Jonnalagadda, One-pot green synthesis of novel 5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine derivatives with eco-friendly biodegradable eggshell powder as efficacious catalyst, *Res. Chem. Intermed.*, 2020, **46**, 3067–3083, DOI: [10.1007/s11164-020-04135-6](https://doi.org/10.1007/s11164-020-04135-6).
- 224 H. N. Roy, M. Rana, A. Z. Al Munsur, K.-I. Lee and A. K. Sarker, Efficient and convenient synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives mediated by L-proline, *Synth. Commun.*, 2016, **46**, 1370–1376, DOI: [10.1080/00397911.2016.1192650](https://doi.org/10.1080/00397911.2016.1192650).
- 225 K. Pradhan, S. Paul and A. R. Das, Synthesis of a diversified combinatorial library of 1H-pyrazolo[1,2-b]phthalazine-5,10-dione derivatives applying sustainable carbon-based solid acid catalyst involving a domino four-component reaction, *Monatsh. Chem.*, 2014, **145**, 1343–1352, DOI: [10.1007/s00706-014-1195-8](https://doi.org/10.1007/s00706-014-1195-8).
- 226 G. J. Raghavendra and V. Siddaiah, One-pot four component synthesis of 3-amino-1-(1H-indol-2-yl)-5,10-dioxo-5,10-dihydro-1H-pyrazolo[1,2-b]phthalazine derivatives mediated by [DBUH][OAc], *Russ. J. Gen. Chem.*, 2018, **88**, 1892–1898, DOI: [10.1134/S1070363218090219](https://doi.org/10.1134/S1070363218090219).
- 227 Y. D. Reddy, B. S. Narayana, Ch. V. R. Reddy and P. K. Dubey, Four-component domino reaction for the synthesis of 1-(1H-indol-2-yl)-1H-pyrazolo[1,2-b]phthalazine-5,10-diones, *Synth. Commun.*, 2014, **44**, 3037–3046, DOI: [10.1080/00397911.2014.928326](https://doi.org/10.1080/00397911.2014.928326).
- 228 S. Keshipour, S. Shojaei and A. Shaabani, A novel one-pot isocyanide-based four-component reaction: synthesis of highly functionalized 1H-pyrazolo[1,2-b]phthalazine-1,2-dicarboxylates and 1H-pyrazolo[1,2-a]pyridazine-1,2-dicarboxylates, *Tetrahedron*, 2012, **68**, 6141–6145, DOI: [10.1016/j.tet.2012.05.078](https://doi.org/10.1016/j.tet.2012.05.078).
- 229 S. Khandelwal, A. Rajawat and M. Kumar, An efficient and environmentally benign four-component domino protocol for synthesis of spirooxindoles spiroannulated with fused heterosystems of privileged heterocycles, *Curr. Bioact. Compd.*, 2013, **9**, 203–210, DOI: [10.2174/15734072113099990014](https://doi.org/10.2174/15734072113099990014).
- 230 A. Rajawat, S. Khandelwal and M. Kumar, Deep eutectic solvent promoted efficient and environmentally benign



- four-component domino protocol for synthesis of spirooxindoles, *RSC Adv.*, 2014, **4**, 5105, DOI: [10.1039/c3ra44600j](https://doi.org/10.1039/c3ra44600j).
- 231 A. Biabangard and H. Reza Shaterian, Vitamin B1 supported on alumina as an efficient heterogeneous catalyst for synthesis of tetrahydro-2,6-dioxypyrimidin-4-yl-2,3-dihydrophthalazine-1,4-dione derivatives, *J. Iran. Chem. Soc.*, 2015, **12**, 1529–1534, DOI: [10.1007/s13738-015-0623-y](https://doi.org/10.1007/s13738-015-0623-y).
- 232 M. Sadeghzadeh and M. A. Nasserri, Methylene dipyridine nanoparticles stabilized on Fe₃O₄ as catalysts for efficient, green, and one-pot synthesis of pyrazolophthalazinyl spirooxindoles, *Catal. Today*, 2013, **217**, 80–85, DOI: [10.1016/j.cattod.2013.07.018](https://doi.org/10.1016/j.cattod.2013.07.018).
- 233 B. Maleki and S. Sedigh Ashrafi, In Situ generated 'HCl': A highly efficient catalyst for one-pot synthesis of 1H-indazolo[1,2-b]phthalazine-1,6,11-triones and 1H-pyrazolo[1,2-b]phthalazine-5,10-diones under solvent-free conditions, *J. Mex. Chem. Soc.*, 2014, **58**, 159–167, DOI: [10.29356/jmcs.v58i2.172](https://doi.org/10.29356/jmcs.v58i2.172).
- 234 M. Esmaeilpour, J. Javidi, F. N. Dodeji and S. Zahmatkesh, Solvent-free, sonochemical, one-pot, four-component synthesis of 2H-indazolo[2,1-b]phthalazine-triones and 1H-pyrazolo[1,2-b]phthalazine-diones catalyzed by Fe₃O₄@SiO₂-imid-PMAN magnetic nanoparticles, *Res. Chem. Intermed.*, 2021, **47**, 2629–2652, DOI: [10.1007/s11164-016-2462-6](https://doi.org/10.1007/s11164-016-2462-6).
- 235 J. Safaei-Ghomi, A. Hatami, H. Shahbazi-Alavi and A. Ziarati, CuFe₂O₄ and ZrP₂O₇ nanoparticles as highly efficient catalysts for the one-pot synthesis of phthalazine derivatives under solvent-free conditions, *Sci. Iran.*, 2016, **23**, 27052716, DOI: [10.24200/sci.2016.3979](https://doi.org/10.24200/sci.2016.3979).
- 236 M. Esmaeilpour, J. Javidi and F. Dehghani, Preparation, characterization and catalytic activity of dendrimer-encapsulated phosphotungstic acid nanoparticles immobilized on nanosilica for the synthesis of 2H-indazolo[2,1-b]phthalazine-triones under solvent-free or sonochemical conditions, *J. Iran. Chem. Soc.*, 2016, **13**, 695–714, DOI: [10.1007/s13738-015-0782-x](https://doi.org/10.1007/s13738-015-0782-x).
- 237 H. R. Shaterian and F. Rigi, One-pot, four-component synthesis of 2H-indazolo[2,1-b]phthalazine-triones catalyzed by cellulose-SO₃H as a reusable heterogeneous and efficient catalyst, *Res. Chem. Intermed.*, 2014, **40**, 1989–1995, DOI: [10.1007/s11164-013-1096-1](https://doi.org/10.1007/s11164-013-1096-1).
- 238 E. Mosaddegh and A. Hassankhani, A rapid, one-pot, four-component route to 2H-indazolo[2,1-b]phthalazine-triones, *Tetrahedron Lett.*, 2011, **52**, 488–490, DOI: [10.1016/j.tetlet.2010.08.099](https://doi.org/10.1016/j.tetlet.2010.08.099).
- 239 H. R. Shaterian and F. Rigi, Starch sulfate as an efficient and biodegradable polymer catalyst for one-pot, four-component reaction of 2H-indazolo[2,1-b]phthalazine-triones, *Starch*, 2011, **63**, 340–346, DOI: [10.1002/star.201000144](https://doi.org/10.1002/star.201000144).
- 240 A. Hasaninejad, M. R. Kazerooni and A. Zare, Solvent-free, one-pot, four-component synthesis of 2H-indazolo[2,1-b]phthalazine-triones using sulfuric acid-modified PEG-6000 as a green recyclable and biodegradable polymeric catalyst, *Catal. Today*, 2012, **196**, 148–155, DOI: [10.1016/J.CATTOD.2012.05.026](https://doi.org/10.1016/J.CATTOD.2012.05.026).
- 241 R. A. Sheldon, Catalysis: the key to waste minimization, *J. Chem. Technol. Biotechnol.*, 1997, **68**, 381–388, DOI: [10.1002/\(SICI\)1097-4660\(199704\)](https://doi.org/10.1002/(SICI)1097-4660(199704)68:3<381::AID-JCTB10974660(199704)>3.0.CO;2-1).
- 242 H. R. Shaterian and M. Aghakhanizadeh, Four-component synthesis of 2H-indazolo[2,1-b]phthalazine-1,6,11(13H)-trione derivatives, *J. Org. Chem. Res.*, 2012, **15**, 1060–1064, DOI: [10.1016/j.crci.2012.09.005](https://doi.org/10.1016/j.crci.2012.09.005).
- 243 M. Shekouhy and A. Hasaninejad, Ultrasound-promoted catalyst-free one-pot four component synthesis of 2H-indazolo[2,1-b]phthalazine-triones in neutral ionic liquid 1-butyl-3-methylimidazolium bromide, *Ultrason. Sonochem.*, 2012, **19**, 307–313, DOI: [10.1016/j.ultsonch.2011.07.011](https://doi.org/10.1016/j.ultsonch.2011.07.011).
- 244 H. Veisi, A. Aminimanesh, N. Khankhani and R. Ghorbani-Vaghei, Protic ionic liquid [TMG][Ac] as an efficient, homogeneous and recyclable catalyst for one-pot four-component synthesis of 2H-indazolo[2,1-b]phthalazine-triones and dihydro-1H-pyrano[2,3-c]pyrazol-6-ones, *RSC Adv.*, 2014, **4**, 25057–25062, DOI: [10.1039/C4RA03514C](https://doi.org/10.1039/C4RA03514C).
- 245 D. Habibi and A. Shamsian, An efficient and recyclable bifunctional acid-base ionic liquid for synthesis of 1H-indazolo[1,2-b]phthalazinetriones, *Res. Chem. Intermed.*, 2015, **41**, 6245–6255, DOI: [10.1007/s11164-014-1736-0](https://doi.org/10.1007/s11164-014-1736-0).
- 246 B. Afzalian, J. T. Mague, M. Mohamadi, S. Yousef Ebrahimpour, B. Pouramiri and E. T. Kermani, Ni(II), Co(II), and Cu(II) complexes incorporating 2-pyrazinecarboxylic acid: Synthesis, characterization, electrochemical evaluation, and catalytic activity for the synthesis of 2H-indazolo[2,1-b]phthalazine-triones, *Chin. J. Catal.*, 2015, **36**, 1101–1108, DOI: [10.1016/S1872-2067\(14\)60318-1](https://doi.org/10.1016/S1872-2067(14)60318-1).
- 247 S. S. Choudhare, G. S. Khansole and V. N. Bhosale, Synthesis of phenyl 1H-indazolo[1,2-b]phthalazine trione derivatives with their biological evaluation, *Indo Am. J. Pharm. Sci.*, 2017, **2017**, 4272–4279, DOI: [10.5281/zenodo.1095543](https://doi.org/10.5281/zenodo.1095543).
- 248 A. V. Chate, P. K. Bhadke, M. A. Khande, J. N. Sangshetti and C. H. Gill, β-Cyclodextrin as a supramolecular catalyst for the synthesis of 2H-indazolo[2,1-b]phthalazine-trione derivatives in water and their antimicrobial activities, *Chin. Chem. Lett.*, 2017, **28**, 1577–1582, DOI: [10.1016/j.ccllet.2017.03.007](https://doi.org/10.1016/j.ccllet.2017.03.007).
- 249 F. Khajoei Nejad, M. Khosravan, S. Y. Ebrahimpour and F. Bisceglie, A mixed-ligand quinazoline-based Ni(II) Schiff base complex: Synthesis, characterization, crystal structure, antimicrobial investigation and catalytic activity for the synthesis of 2H-indazolo[2,1-b]phthalazine-triones, *Appl. Organomet. Chem.*, 2018, **32**, e3907, DOI: [10.1002/aoc.3907](https://doi.org/10.1002/aoc.3907).
- 250 M. Mahmoodi Fard Chegeni, A. Bamoniri and B. B. F. Mirjalili, A versatile protocol for synthesis of 2H-indazolo[2,1-b]phthalazine triones using γ-Al₂O₃/BFn/Fe₃O₄ as an efficient magnetic nano-catalyst, *Polycyclic Aromat. Compd.*, 2022, **42**, 391–399, DOI: [10.1080/10406638.2020.1735457](https://doi.org/10.1080/10406638.2020.1735457).



- 251 Y. Han, Y. Gao, X. Cao, M. M. Zangeneh, S. Liu and J. Li, Ag NPs on chitosan-alginate coated magnetite for synthesis of indazolo[2,1-b]phthalazines and human lung protective effects against α -Guttiferin, *Int. J. Biol. Macromol.*, 2020, **164**, 2974–2986, DOI: [10.1016/j.ijbiomac.2020.08.183](https://doi.org/10.1016/j.ijbiomac.2020.08.183).
- 252 A. Naeimi and M. Zahedifar, Immobilized copper complex on biodegradable cellulose: catalytic application for the green synthesis of 2H-indazolo[1,2-b]phthalazine trione derivatives and synthesis of CuO nanoparticles, *Iran. J. Chem. Chem. Eng.*, 2022, **40**, 49–57, DOI: [10.30492/IJCCE.2020.121610.3971](https://doi.org/10.30492/IJCCE.2020.121610.3971).
- 253 N. Amirmahani, N. O. Mahmoodi, M. Malakootian and A. Pardakhty, Introducing new and effective catalysts for the synthesis of pyridazino[1,2-a]indazole, indazolo[2,1-b]phthalazine and pyrazolo[1,2-b]phthalazine derivatives, *MethodsX*, 2020, **7**, 100823, DOI: [10.1016/j.mex.2020.100823](https://doi.org/10.1016/j.mex.2020.100823).
- 254 R. M. Tigote, K. P. Haval and S. K. Kazi, ZnFe₂O₄ nanoparticles: an efficient and reusable catalyst for 2H-indazolo[2,1-b]phthalazine-triones synthesis under solvent free condition, *J. Med. Chem. Drug Discov.*, 2017, **2**, 654–665.
- 255 N. Amirmahani, N. O. Mahmoodi, M. Malakootian and A. Pardakhty, [TBP]₂SO₄ ionic liquid catalyst for 4MCR of pyridazinoindazole, indazolophthalazine and pyrazolophthalazine derivatives, *Mol. Diversity*, 2022, **26**, 15–25, DOI: [10.1007/s11030-020-10153-8](https://doi.org/10.1007/s11030-020-10153-8).
- 256 O. S. Zaky, M. A. Selim, M. M. Ebied and K. U. Sadek, Glycerol: A promising benign solvent for catalyst-free one-pot multi-component synthesis of 1H-pyrazolo[1,2-b]phthalazine-5,10-diones and 2H-indazolo[2,1-b]phthalazine-triones under controlled microwaves irradiation, *J. Heterocycl. Chem.*, 2019, **56**, 2796–2803, DOI: [10.1002/JHET.3659](https://doi.org/10.1002/JHET.3659).
- 257 B. Dam, M. Saha, R. Jamatia and A. K. Pal, Nano-ferrite supported glutathione as a reusable nanoorganocatalyst for the synthesis of phthalazine-trione and dione derivatives under solvent free condition, *RSC Adv.*, 2016, **6**, 54768–54776, DOI: [10.1039/C6RA06376D](https://doi.org/10.1039/C6RA06376D).
- 258 Z. Zaheer, F. A. K. Khan, J. N. Sangshetti, R. H. Patil and K. S. Lohar, Novel amalgamation of phthalazine-quinolines as biofilm inhibitors: one-pot synthesis, biological evaluation and in silico ADME prediction with favorable metabolic fate, *Bioorg. Med. Chem. Lett.*, 2016, **26**, 1696–1703, DOI: [10.1016/j.bmcl.2016.02.057](https://doi.org/10.1016/j.bmcl.2016.02.057).
- 259 B. Pouramiri, R. G. Far and M. Zahedifar, Acidic ionic liquids: highly efficient catalysts for one-pot four-component synthesis of pyrazolo[1,2-b]phthalazines under solvent-free conditions, *Chem. Heterocycl. Compd.*, 2018, **54**, 1056–1060, DOI: [10.1007/s10593-018-2391-y](https://doi.org/10.1007/s10593-018-2391-y).
- 260 P. Jahanshahi and M. Mamaghani, Chemodivergent, multicomponent-tandem facile synthesis of novel 1H-pyrazolo[1,2-b]phthalazine-5,10-dione using acetic acid functionalized imidazolium salt {[cmdmim]⁺]} as a recyclable catalyst, *New J. Chem.*, 2019, **43**, 8266–8278, DOI: [10.1039/C9NJ00993K](https://doi.org/10.1039/C9NJ00993K).
- 261 P. Venkatesham, P. Shyam, Pooja, R. Chedupaka and R. R. Vedula, Facile one-pot multi-component synthesis, characterization, molecular docking studies, biological evaluation of 1,2,4-triazolo isoindoline-1,3-diones and Their DFT calculations, *Polycyclic Aromat. Compd.*, 2023, **43**(3), 2283–2301, DOI: [10.1080/10406638.2022.2042333](https://doi.org/10.1080/10406638.2022.2042333).
- 262 H. Uchida, H. Yoshiyama, P. Y. Reddy, S. Nakamura and T. Toru, Novel synthesis of metal-free phthalocyanines from phthalimides and phthalic anhydrides with hexamethyldisilazan, *Synlett*, 2003, 2083–2085, DOI: [10.1055/s-2003-41484](https://doi.org/10.1055/s-2003-41484).
- 263 K. S. Jung, J. H. Kwon, S. M. Shon, J. P. Ko, J. S. Shin and S. S. Park, Microwave synthesis of metal phthalocyanines under solvent-free conditions, *J. Mater. Sci.*, 2004, **39**, 723–726, DOI: [10.1023/B:JMASC.0000011541.91490.88](https://doi.org/10.1023/B:JMASC.0000011541.91490.88).
- 264 F. B. H. Ahmad, M. G. Moghaddam, M. Basri and M. B. Abdul Rahman, Enzymatic synthesis of betulinic acid ester as an anticancer agent: optimization study, *Biocatal. Biotransform.*, 2010, **28**, 192–200, DOI: [10.3109/10242421003753795](https://doi.org/10.3109/10242421003753795).
- 265 M. Ghaffari M, F. B. H. Ahmad, M. Basri and M. B. Abdul Rahm, Lipase-catalyzed esterification of betulinic acid using phthalic anhydride in organic solvent media: study of reaction parameters, *J. Appl. Sci.*, 2010, 337–342, DOI: [10.3923/jas.2010.337.342](https://doi.org/10.3923/jas.2010.337.342).
- 266 P. K. Dubey, S. M. Ghouse Mohiuddin and D. Ramesh, Reactions of phthalic anhydride with alcohols, *Asian J. Chem.*, 1997, **9**, 379–387.
- 267 R. Fareghi-Alamdari, M. Nadiri Niri and H. Hazarkhani, Synthesis and characterization of a new hydroxyl functionalized diacidic ionic liquid as catalyst for the preparation of diester plasticizers, *J. Mol. Liq.*, 2017, **227**, 153–160, DOI: [10.1016/j.molliq.2016.11.121](https://doi.org/10.1016/j.molliq.2016.11.121).
- 268 N. Zekri, R. Fareghi-Alamdari and Z. Khodarahmi, Functionalized dicationic ionic liquids: Green and efficient alternatives for catalysts in phthalate plasticizers preparation, *J. Chem. Sci.*, 2016, **128**, 1277–1284, DOI: [10.1007/s12039-016-1127-8](https://doi.org/10.1007/s12039-016-1127-8).
- 269 G. Wypych, *Handbook of Plasticizers*, ChemTec Publishing, New York, 2004, ch. 2, pp 36–43.
- 270 R. Fareghi-Alamdari, M. Nadiri Niri, H. Hazarkhani and N. Zekri, Diacidic ionic liquid supported on magnetic-silica nanocomposite: a novel, stable, and reusable catalyst for selective diester production, *J. Iran. Chem. Soc.*, 2018, **15**, 2615–2629, DOI: [10.1007/S13738-018-1450-8](https://doi.org/10.1007/S13738-018-1450-8).
- 271 Z. Shahedi and Y. Mansoori, Fe₃O₄@SiO₂-SO₃H nanoparticles: An efficient magnetically retrievable catalyst for esterification reactions, *J. Part. Sci. Technol.*, 2018, **4**, 67–79, DOI: [10.22104/jpst.2018.2811.1117](https://doi.org/10.22104/jpst.2018.2811.1117).
- 272 M. Kulawska, H. Moroz and A. Kasprzyk, Kinetic investigations on the esterification of phthalic anhydride with N-heptyl, N-nonyl or N-undecyl alcohol over sulfuric acid catalyst, *React. Kinet., Mech. Catal.*, 2011, **104**, 9–15, DOI: [10.1007/s11144-011-0337-9](https://doi.org/10.1007/s11144-011-0337-9).
- 273 M. Arabi, M. Mohammadpour Amini, M. Abedini, A. Nemati and M. Alizadeh, Esterification of phthalic



- anhydride with 1-butanol and 2-ethylhexanol catalyzed by heteropolyacids, *J. Mol. Catal. A: Chem.*, 2003, **200**, 105–110.
- 274 S. R. Bhutada and V. G. Pangarkar, Esterification of phthalic anhydride with 2-ethylhexanol, *J. Chem. Technol. Biotechnol.*, 1986, **36**, 61–66, DOI: [10.1002/jctb.280360204](https://doi.org/10.1002/jctb.280360204).
- 275 S. Gharibe, ZnAl₂O₄/SiO₂ as an efficient nanocatalyst for esterification of phthalic anhydride by 2-ethylhexanol, *Iran. J. Sci. Technol., Trans. A: Sci.*, 2020, **44**, 1349–1355, DOI: [10.1007/s40995-020-00899-z](https://doi.org/10.1007/s40995-020-00899-z).
- 276 J. B. Ellington, K. M. Park and J. F. Brennecke, Effect of local composition enhancements on the esterification of phthalic anhydride with methanol in supercritical carbon dioxide, *Ind. Eng. Chem. Res.*, 1994, **33**(4), 965–974, DOI: [10.1021/ie00028a026](https://doi.org/10.1021/ie00028a026).
- 277 M. Khodadadi Moghaddam and M. R. Gholami, Preparation of Titania based solid strong acid and study of its catalytic activity in esterification of phthalic anhydride and sebacic acid, *Mater. Lett.*, 2006, **60**, 715–719, DOI: [10.1016/j.matlet.2005.09.069](https://doi.org/10.1016/j.matlet.2005.09.069).
- 278 T. S. Thorat, V. M. Yadav and G. D. Yadav, Esterification of phthalic anhydride with 2-ethylhexanol by solid superacidic catalysts, *Appl. Catal.*, 1992, **90**, 73–96, DOI: [10.1016/0926-860X\(92\)85050-L](https://doi.org/10.1016/0926-860X(92)85050-L).
- 279 G. B. Bajracharya, R. Koju, S. Ojha, S. Nayak, S. Subedi and H. Sasai, Plasticizers: synthesis of phthalate esters via FeCl₃-catalyzed nucleophilic addition of alcohols to phthalic anhydride, *Results Chem.*, 2021, **3**, 100190, DOI: [10.1016/j.rechem.2021.100190](https://doi.org/10.1016/j.rechem.2021.100190).
- 280 N. A. Rakhimova and S. V. Kudashev, Features of reactions of phthalic anhydride with polyfluorinated alcohols, *Russ. J. Gen. Chem.*, 2011, **81**, 1529–1534.
- 281 M. Karimi Alavijeh and M. M. Amini, Synthesis and characterization of butylamine-functionalized Cr(III)-MOF-SO₃H: synergistic effect of the hydrophobic moiety on Cr(III)-MOF-SO₃H in esterification reactions, *Polyhedron*, 2019, **173**, 114142, DOI: [10.1016/j.poly.2019.114142](https://doi.org/10.1016/j.poly.2019.114142).
- 282 L. F. Fieser, The condensation of β-naphthol with phthalic anhydride, *J. Am. Chem. Soc.*, 1931, **53**, 3546–3560, DOI: [10.1021/ja01360a051](https://doi.org/10.1021/ja01360a051).
- 283 S.-L. Deng, I. Baglin, M. Nour and Ch. Cavé, Synthesis of phosphonodipeptide conjugates of ursolic acid and their homologs, *Heteroat. Chem.*, 2008, **19**, 55–65, DOI: [10.1002/hc.20396](https://doi.org/10.1002/hc.20396).
- 284 A. Mohammadi-Farani, S. Soltani Darbandi and A. Aliabadi, Synthesis and acetylcholinesterase inhibitory evaluation of 4-(1,3-dioxoisindolin-2-yl)-N-phenyl benzamide derivatives as potential anti-alzheimer agents, I, *Iran. J. Pharm. Sci.*, 2016, **15**, 313–320, DOI: [10.22037/ijpr.2016.1868](https://doi.org/10.22037/ijpr.2016.1868).
- 285 D.-Y. Kim, S.-A. Lee, H. Kim, S. M. Kim, N. Kim and K.-U. Jeong, An azobenzene-based photochromic liquid crystalline amphiphile for the remote-controllable light shutter, *Chem. Commun.*, 2015, **51**, 11080–11083, DOI: [10.1039/C5CC02834E](https://doi.org/10.1039/C5CC02834E).
- 286 T. Iwanaga, N. Asano, H. Yamada and S. Toyota, Synthesis and photophysical properties of dinaphtho[2,3-b:2',3'-i] dihydrophenazine derivatives, *Tetrahedron Lett.*, 2019, **60**, 1113–1116, DOI: [10.1016/j.tetlet.2019.03.035](https://doi.org/10.1016/j.tetlet.2019.03.035).
- 287 M. A. Mahmoud, A. Y. Abed and S. M. D. Al-Nuzal, Synthesis and anticancer evaluation of new heterocyclic compounds in cell line L20B in vitro, *J. Pharm. Sci. Res.*, 2019, **11**, 901–904.
- 288 A. Kajal, S. Bala, S. Kamboj and V. Saini, Synthesis, characterization, and computational studies on phthalic anhydride-based benzylidene-hydrazide derivatives as novel, potential anti-inflammatory agents, *Med. Chem. Res.*, 2014, **23**, 2676–2689, DOI: [10.1007/s00044-013-0848-1](https://doi.org/10.1007/s00044-013-0848-1).

