RSC Advances



PAPER

View Article Online
View Journal | View Issue



Cite this: RSC Adv., 2023, 13, 18812

Received 11th May 2023 Accepted 14th June 2023

DOI: 10.1039/d3ra03136e

rsc.li/rsc-advances

Influence of the physical properties of the layered oxyselenides Bi₂YO₄Cu₂Se₂ through Ni doping

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We have synthesized a series of Ni-doped layered oxyselenides $Bi_2YO_4Cu_{2-x}Ni_xSe_2$ ($0 \le x \le 0.4$). The crystal structure and physical properties were studied through X-ray diffraction, and electric and thermo transport measurements. We also performed DFT calculations to study the electric structure of the designed $Bi_2YO_4Ni_2Se_2$, which is similar to that of KNi_2Se_2 .

Introduction

In recent decades, compounds with layered crystal structures have attracted people's increasing attention due to their interesting physical properties, such as superconductivity, ¹⁻³ charge density wave, ⁴⁻⁹ topological insulation ¹⁰ *etc.* In particular, the discovery of iron-based superconductors in 2008 aroused wide concern all over the world. ² Up to now, the mechanism of high-temperature superconductivity remains a mystery, which requires the joint efforts of theoretical and experimental researchers. According to previous research experience, high-temperature superconductors tend to adopt layered crystal structures. And it has been an important way to search for new superconductors through exploring new layered compounds.

Among the known iron-based superconductors, they all have the Fe_2Pn_2/Fe_2Ch_2 (Pn = P, As; Ch = S, Se, Te) conducting layer, which is believed playing an important role in the superconductivity. In fact, all the iron-based superconductors have analogues in CuCh based compounds. For example, the 1111type LaOFeP/As and LaOCuS/Se adopt the same crystal structure. The 122-type KFe₂Se₂ and KCu₂Se₂, the 32 522-type Sr₃Sc₂O₅Fe₂As₂ and Sr₃Sc₂O₅Cu₂S₂ etc. All have the same situations.13-15 Many iron-based superconductors were discovered by substituting Fe₂Pn₂ layers for Cu₂Ch₂ layers in the known CuCh-based compounds. In 2002, a new layered CuSebased compound Bi₂YO₄Cu₂Se₂ was reported and we studied the physical and electric properties lately. 16-18 The Cu₂Se₂ layer is defined as the conducting layer and the Bi₂YO₄ layer is defined as the block layer. As every block layer provide one electron to conducting layer, which causes a mixed state of Cu²⁺/Cu⁺, it exhibits p-type metallic behaviour which is rare in

Results and discussion

The crystal structure of Bi₂YO₄Cu₂Se₂ is shown in Fig. 1(a), which is alternatively stacked by Bi₂YO₄ layer and Cu₂Se₂ layer. Fig. 1(b) shows the crystal structure of KNi₂Se₂, which has the ThCr₂Si₂-type structure.²⁰ Comparing these two structures, it is not difficult to find that their crystal structure is the same except the block layer. Fig. 2 shows the X-ray diffraction patterns of Bi₂YO₄Cu₂-_xNi_xSe₂. All the diffraction peaks of the doped samples match the Bragg peaks of Bi₂YO₄Cu₂Se₂, which indicates high quality samples. The fitting lattice parameters of Bi₂YO₄Cu₂-_xNi_xSe₂ are shown in Fig. 2(b). The lattice parameter *a* and *c* for Bi₂YO₄Cu₂Se₂ are 3.862 and 24.31 Å respectively, which is corresponding to previous results.^{16,17} The lattice parameter a is decreased with increasing Ni content, while the lattice parameter *c* is the opposite. It indicates that the in-plane

CuCh-based compounds. Similarly, p-type metallic ground state is also found in KCu₂Se₂, in which the K layer also provides one electron to Cu₂Se₂ layer per unit cell. In 2012, the analogue KNi₂Se₂ was reported superconducting below 0.8 K, which has aroused people's extensive research interest. 19 If the Cu₂Se₂ layer in Bi₂YO₄Cu₂Se₂ was replaced by Fe₂Pn₂ or Ni₂Pn₂ layers, we think new superconductors may be achieved. We try our best to synthesize Bi₂YO₄Ni₂Se₂ through solid state reaction. Unfortunately, all the attempts ended in failure. On the other hand, it is also valuable to study the influence of the physical proper-ties of the layered oxyselenide Bi₂YO₄Cu₂Se₂ through Ni doping at Cu site. In this paper, we successfully doped 20% Ni at Cu site in Bi₂YO₄Cu₂Se₂. The temperature coefficient of resistivity can be effectively adjusted through Ni doping. Furthermore, we studied the thermoelectric properties of the Ni-doped samples. Finally, we study the electric structure of the designed Bi₂YO₄Ni₂Se₂ through the first principle calculation, which exhibit similar electronic structure with KNi2Se2. This will tempt people to explore the synthesis of Bi₂YO₄Ni₂Se₂ through other methods.

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Fig. 1 The crystal structure of Bi₂YO₄(Cu,Ni)₂Se₂ (a) and KNi₂Se₂ (b)

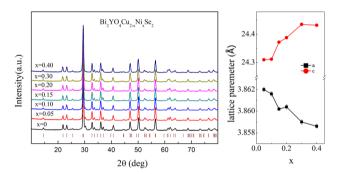


Fig. 2 The X-ray diffraction patterns and the fitted lattice parameters of $Bi_2YO_4Cu_{2-x}Ni_xSe_2$ measured at 300 K.

interaction is strengthened and the interlayer interactions are weakened. The reported lattice parameters a and c of KCu₂Se₂ are 4.008 and 13.643 Å respectively. Compared with KCu₂Se₂, the thicker block layer in Bi₂YO₄Cu₂Se₂ makes the parameter a smaller. In other words, the in-plane interaction of Bi₂YO₄-Cu₂Se₂ is stronger than that of KCu₂Se₂. For KNi₂Se₂, the fitted lattice parameter a and c according to experiment result are 3.89295 and 13.3158 Å respectively and the calculated lattice parameter a and c according to DFT calculation are 3.9680 and 13.0479 Å respectively. The lattice parameter a of KNi₂Se₂ is larger than that of Bi₂YO₄Cu₂Se₂. According to the above results, the lattice parameter a of Bi₂YO₄Ni₂Se₂, if it exists, should be larger than that of KNi₂Se₂. This may be the reason we cannot doped much more Ni in Bi₂YO₄Cu₂Se₂.

Fig. 3 shows the temperature dependence of the electric resistivity of Bi₂YO₄Cu_{2-x}Ni_xSe₂. For Bi₂YO₄Cu₂Se₂, the resistivity increases with increasing temperature from 2 to 300 K, which indicates a metallic transport behaviour and is similar with previous results.¹⁷ The value of resistivity is increased with increasing Ni content. Compared with KNi₂Se₂, the resistivity is relatively higher. Unfortunately, no superconducting behaviour was found until 2 K for all the Ni doped Bi₂YO₄Cu₂Se₂.

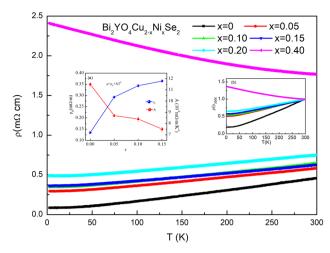


Fig. 3 The temperature dependence of the resistivity of Bi_2YO_4 - $Cu_{2-x}Ni_xSe_2$ from 2 to 300 K. Inset (a) is the parameter ρ_0 and A fitted through the Fermi liquid behavior equation with different Ni con-tent. Inset (b) is the reduced resistivity of $Bi_2YO_4Cu_{2-x}Ni_xSe_2$.

According to previous study,17 the metallic behaviour originates from the mixed valences of Cu ion in Cu₂Se₂ layer. Every Bi₂YO₄ unit provides one electron to Cu₂Se₂ unit, so every two Cu atoms only pro-vide 3 electrons to keep balance. While when Ni is doped at Cu site, every Ni atom can provide two electrons, namely electronic doping. For undoped Bi₂YO₄Cu₂Se₂, the carriers are mainly p-type. Regardless of other factors, Ni doping at Cu site decreases the carrier con-centration and the resistivity would increase with increasing Ni content. Our experimental results are consistent with the viewpoint. As shown in Fig. 3, the reduced resistivity increases with increasing Ni content. Furthermore, the Bi₂YO₄Cu_{1.6}Ni_{0.4}Se₂ exhibits semiconducting/insulating transport behaviour from 2 to 300 K. The resistivity (x = 0–0.2) obey the Fermi liquid behaviour from 2 K to 50 K. We fitted the resistivity curves through the Fermi liquid behaviour equation $\rho = \rho_0 + AT^2$. Where ρ_0 represents the residual resistivity. The results are shown in the inset (a) of Fig. 3. As can be seen, the residual resistivity in-creases with x. As known, the residual resistivity ρ_0 is in direct proportion to $m^*\hbar/n\tau_0$, where m^* , n and \hbar/τ_0 represent the carrier effective mass, carrier concentration and the scattering rate related to the disordered potential. For Bi₂YO₄Cu_{1.6}Ni_{0.4}Se₂, it exhibits semiconducting transport behaviour that the resistivity decreases with arising temperature. Further study is needed to clarify the ground state of Ni doped Bi₂YO₄Ni₂Se₂. According to previous study, the ground state of KNi₂Se₂ is metallic.²⁰ Our following DFT calculation result also indicates the metallic ground state of Bi2YO4Ni2Se2. It should not turn it to semiconductor through Ni doping. The semiconducting-like phenomenon should not be intrinsic, which may be caused by defect or grain boundary introduced with Ni doping.

Fig. 4 shows the thermoelectric properties of $\mathrm{Bi_2YO_4Cu_2}_{-x^2}\mathrm{Ni}_x\mathrm{Se_2}$ (x=0,0.2,0.3). Fig. 4(b) shows the Seebeck coefficient curves of $\mathrm{Bi_2YO_4Cu_2}_{-x}\mathrm{Ni}_x\mathrm{Se_2}$. The Seebeck coefficient also exhibits metallic behaviour from 100 to 300 K, increasing with temperature linearly according to the equation

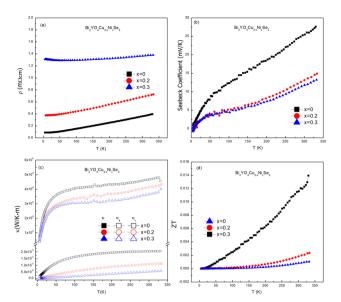


Fig. 4 The thermoelectric properties of Bi₂YO₄Cu_{2-x}Ni_xSe₂

$$S = \frac{8\pi^2 k_{\rm B}^2 Tm^*}{3eh^2} \left(\frac{\pi}{3n}\right)^{2/3}$$

where $k_{\rm B}$ is the Boltzmann constant, m^* is the effective mass of carrier, and n is the carrier concentration, which applies for metals or degenerate semiconductors under assumption of parabolic band and acoustic phonon scattering approximation. The Seebeck coefficients are positive for all the samples ranging from 5 to 330 K, indicating p-type carriers dominated in these materials, which is consistent with previous results. The Seebeck coefficient decreases with increasing Ni content. Fig. 3(c) shows the total thermal conductivity curves. The total thermal conductivity decreases with increasing Ni content. The total thermal conductivity can be divided into the electron component (κ_e) and the lattice component (κ_l): $\kappa = \kappa_e + \kappa_l$. The electron component is proportional to the electrical conductivity according to Wide-Franz law, $\kappa_e = L\sigma T = LT/\rho$. The decreasing resistivity indicates the electron component of the total thermal conductivity increase. Then the lattice thermal conductivity decreases with increasing Ni content, which originates from the increasing defect scattering induced by Ni. The dimensionless figure of merit is defined as $ZT = S^2/\rho \kappa T$, which is the measure of the efficiency of a thermoelectric material. We calculated the ZT of the synthesized samples, which is shown in Fig. 4(d). Although the induce of Ni de-crease the resistivity of the sample, the ZT decrease with increasing Ni content because of the great degree reduction of the Seebeck coefficient. The introduction of Ni is unfavourable to the improvement of thermoelectric properties in Bi₂YO₄Cu₂Se₂.

The plane-wave projector augmented method as implement in the Vienna *ab initio* simulation package (VASP) was used to calculate the density of states (DOS) and energy band. The Perdew-Bruke-Ernzerhof (PBE) form of generalized gradient approximation (GGA) was chosen as the exchange-correlation potential.²³ Due to the ex-change-correlation effects of the strongly localized Cu and Ni 3d electrons, we adopted the DFT +

U method. We tested different values of the screened Coulomb parameter U. The added Coulomb interactions do not change the ground state. An energy cut-off of 520 eV was used in all cases. The convergence criterion for energy was set to be 10⁻⁶ eV per unit cell and the forces on all relaxed atoms were less than 0.01 eV A^{-1} . According to previous study, nonmagnetic states is adopted to calculate the electronic band structure. Fig. 5 shows the band structure of Bi₂YO₄Cu₂Se₂ and Bi₂YO₄Ni₂Se₂. The calculated band structure of Bi₂YO₄Cu₂Se₂ is consistent to previous result. The Fermi energy located at the valence band, indicating the p-type metallic ground state, which is consistent with the experiment results. Furthermore, as the replace of Cu by Ni, the Fermi energy move deeper in valence band, namely similar p type metallic ground state. The conduction band is attached to the valence band. The semiconducting behaviour in Ni-doped Bi₂YO₄Cu₂Se₂ observed in our experiment may be non-intrinsic. For Bi₂YO₄Cu₂Se₂, the bands exhibit complicated character near the Fermi surface. Along the $Z-\Gamma$ line, the energy dispersion is very weak. The energy dispersion is strong in the layer-parallel directions, which originates from the strong hybridizations between Cu and Se. Similar quasi-twodimensional properties are found in Bi₂YO₄Ni₂Se₂, which is consistent with KNi2Se2.20 There are also two bands crossing the Fermi energy, indicting the multi-band character. Compared the band structure of Bi₂YO₄Ni₂Se₂ and KNi₂Se₂, the structure of the valence band is very similar. Moreover, the thick Bi₂YO₄ layer make the 2D property more obvious. The ThCr₂Si₂-type super-conductor KNi₂Se₂ was discovered with a superconducting transition temperature of 0.8 K. We believe the Bi₂YO₄Cu₂Se₂ may be a superconducting material if it can be successful synthesized.

Experimental

 ${\rm Bi_2YO_4Cu_{2-x}Ni_xSe_2}$ sample was prepared by reacting a stoichiometric mixture of ${\rm Bi_2O_3}$, ${\rm Y_2O_3}$, Y, Cu, Ni and Se. The raw materials weighed by stoichiometric ratio were mixed. After thoroughly grounding in an agate pestle and mortar, the mixtures were pressed into pellets under 12 MPa. The pellets

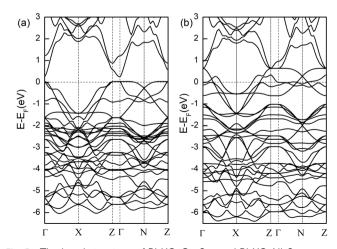


Fig. 5 The band structure of $Bi_2YO_4Cu_2Se_2$ and $Bi_2YO_4Ni_2Se_2$.

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were then sealed under vacuum (<10⁻⁴ Pa) in the silica tubes which had been baked in drybox for 1-2 h at 150 °C. The ampoules were heated to 830 °C and maintained at this temperature for 24 h. Finally, the furnace is shut down and cools to room temperature naturally. The obtained samples were reground, pelletized, and heated for another 24 hours at 830 °C followed by furnace cooling. The X-ray powder diffraction patterns were recorded at room temperature on a PANalytical diffractometer (X'Pert PRO MRD) with Cu Ka radiation (40 kV, 40 mA). Structural refinement of powder samples was carried out by using Rietica software. The electrical resistivity and thermoelectric properties were measured using a Quantum Design PPMS (Physical Properties Measurement System) from 300 to 2 K.

Conclusions

In summary, we have synthesized a series of layered oxyselenides Bi₂YO₄Cu_{2-x}Ni_xSe₂ through solid state reaction. There crystalline structures and physical properties were thoroughly studied. Along with the increase of Ni content, the resistivity increase. It exhibits semiconducting transport behaviour in Bi₂YO₄Cu_{1.6}Ni_{0.4}Se₂, which may be caused by the diffraction of impurity or defects. The crystal structure and band structure of Bi₂YO₄Ni₂Se₂ are like those of KNi₂Se₂. We suspect that Bi₂YO₄Ni₂Se₂ should also exhibit superconductivity if it can be successfully synthesized.

Author contributions

Conceptualization, L. X. and J. W.; methodology, L. X.; validation, Z. L., W. K. and C. W.; formal analysis, Y. Z.; investigation, S. T.; resources, S. T.; data curation, L. X.; writing—original draft preparation, S. T.; writing—review and editing, L. X.; all authors have read and agreed to the published version of the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This work was supported by the Research Start-up Funds of Shandong University of Science and Technology (No. 415058).

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