


 Cite this: *RSC Adv.*, 2023, **13**, 20342

Photon avalanche assisted upconversion via customizing the green emission†

 Manisha Prasad and Vineet Kumar Rai *

The developed SnWO₄ phosphors incorporated with Ho³⁺, Yb³⁺ and Mn⁴⁺ ions have been explored under 980 nm laser irradiation. The molar concentration of dopants has been optimized to 0.5 Ho³⁺, 3.0 Yb³⁺ and 5.0 Mn⁴⁺ in SnWO₄ phosphors. The upconversion (UC) emission from the codoped SnWO₄ phosphors has been substantially amplified up to 13 times and described based on the energy transfer and charge compensation. On incorporating the Mn⁴⁺ ions in the Ho³⁺/Yb³⁺ codoped system the sharp green luminescence shifted to reddish broadband emission due to the photon avalanche mechanism. The processes accountable for the concentration quenching have been described based on critical distance. The interaction responsible for the concentration quenching in Yb³⁺ sensitized Ho³⁺ and Ho³⁺/Mn⁴⁺:SnWO₄ phosphors is considered to be dipole-quadrupole and exchange interaction type, respectively. The activation energy 0.19 eV has been determined, and the phenomenon of thermal quenching is discussed using a configuration coordinate diagram.

 Received 2nd May 2023
 Accepted 5th June 2023

DOI: 10.1039/d3ra02894a

rsc.li/rsc-advances

1. Introduction

Manganese-based upconverting phosphors have recently been popular because of their extensive use in various promising applications, including solid-state lighting, displays, upconverters, temperature monitoring, plant growth and biological applications.^{1–3} Manganese (Mn) can be utilized as an activator in downshifting (DS) and upconversion (UC) phenomena.^{4–6} According to Blasse, the emission from Mn⁴⁺ lies in the deep red region, resulting from the ²E → ⁴A₂ transition, which indicates that the Mn⁴⁺ incorporated phosphor can be a good candidate for red emission.⁷ Mn is accepted as a single luminescence center and can be paired with other atoms such as Bi, Cr, and rare earths (RE) such as Ce, Pr, Dy, Sm, Yb, Ho, Nd, Er and Tm.^{8–18} Researchers are interested in red-emitting Mn²⁺/Mn⁴⁺ ions in aggregation with Eu³⁺ ions.^{19,20} Codoping of manganese with lanthanide initiates a mutual energy transfer process, which enhances the photoluminescence properties. The production of dimers can sometimes improve the emission intensity.¹⁸ The optical characteristics of Mn can be easily influenced by the oxidation state (*i.e.*, 2+, 4+) and the interaction of the crystal field. Besides the transition metals, fixed emissions can be obtained by doping the RE ions. On doping the lanthanides, the crystal field symmetry of the dopants breaks and due to the mixing of the 4fⁿ and 4fⁿ⁻¹5d electronic

configurations having opposite parity, the electric dipole transitions become allowed.^{21–25}

Among the various metal tungstates (MWO₄), SnWO₄ exists in two phases: α-SnWO₄ is favoured at low synthesis temperature below 670 °C, whereas β-SnWO₄ is commonly obtained beyond this temperature.²⁶ The SnWO₄ has a variety of applications, such as electrochemical performance, humidity sensing, photodynamic tumour therapy, photocatalytic degradation, water splitting, resistive switching memory, *etc.*^{27–30} The SnWO₄ is a diamagnetic semiconductor with a large bandgap and low phonon energy that can be employed for upconverting applications.^{26,27,31}

In the present work, Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors have been prepared and investigated through several characterization techniques such as X-ray diffraction (XRD), field emission scanning electron microscopy (FESEM), Fourier transform infrared spectroscopy (FTIR), Raman spectroscopy, X-ray photon spectroscopy (XPS), UV-vis-NIR spectroscopy and frequency upconversion study. In the coactivated system, the green emission of Ho³⁺ ions has been tuned with manganese ions. The detailed study of concentration quenching, the thermal stability and thermal quenching have been performed based on the UC measurements.

2. Experimentation

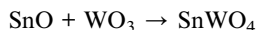
2.1 Synthesis of phosphors

A variety of SnWO₄ phosphors at varying concentration of Ho³⁺, Yb³⁺ and Mn⁴⁺ ions have been produced using the conventional solid-state synthesis technique. All stoichiometric raw materials SnO, WO₄, Ho₂O₃, Yb₂O₃ and MnO₂ were taken in high purity according to the following chemical expressions:

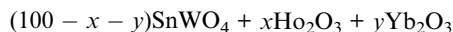
Laser and Spectroscopy Laboratory, Department of Physics, Indian Institute of Technology (Indian School of Mines), Dhanbad, 826004, Jharkhand, India. E-mail: vineetkrrai@yahoo.co.in; vineetkrrai@iitism.ac.in; Tel: +91-326-223 5404/5282

† Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3ra02894a>

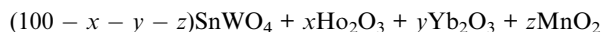




where, $x = 0.1, 0.3, 0.5$ and 0.7 mol%



where, $x = 0.5; y = 1.0, 1.5, 3.0$ and 4.0 mol%



where, $x = 0.5; y = 3.0; z = 1.0, 5.0, 10.0, 15.0$ and 20.0 mol%

The accurately weighing precursors were mixed in the mortar pestle. After grinding for 90 minutes using volatile liquid acetone, the mixture was maintained at 1100°C in a muffle furnace for three hours. The phosphors were retrieved and used for additional characterizations after naturally cooling to ambient temperature.

2.2 Phosphors characterization

To determine the crystal formation and lattice parameters, the optimized $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors were analyzed by X-ray diffractometer in $10^\circ \leq 2\theta \leq 70^\circ$ range. FESEM consisting of an airlock compartment has been used for surface morphological analysis and chemical components analyzed with EDAX. The vibrational bands have been detected with the FTIR and

Raman spectroscopy. XPS confirms the valence states and binding energies of the elements in phosphors. Diffuse reflectance spectra (DRS) have been monitored in the UV-vis-NIR region. The developed phosphors have been irradiated with a continuous wave laser source to achieve frequency UC. The monochromator used for this purpose consists of a triple grating and a photomultiplier tube. The luminescence lifetimes measurements have been carried out by using HORIBA 'Lifetime Spectrometer' and micro second xenon pulse lamp excitation. Thermal stability experiments have been carried out using a multimeter, thermocouple, and a small heater. Calculations of CIE coordinates have been performed using the GoCIE software.

3. Results and discussion

3.1 Structural characterization

3.1.1. XRD analysis. Fig. 1(a) displays the crystalline nature of the codoped SnWO_4 phosphors. The diffraction patterns of phosphors are consistent with the JCPDS file 01-070-1497, of the $\beta\text{-SnWO}_4$ system.²⁶ $\beta\text{-SnWO}_4$ has a cubic structure with space group number 198 and space group $P\bar{1}23$. The unit cell consists of SnO_6 octahedra and $[\text{WO}_4]^{2-}$ tetrahedra with lattice constant $a = b = c = 7.2989 \text{ \AA}$, $\alpha = \beta = \gamma = 90^\circ$, volume = 388.84 \AA^3 . According to the Debye-Scherrer equation, the crystallize size is given by,³²

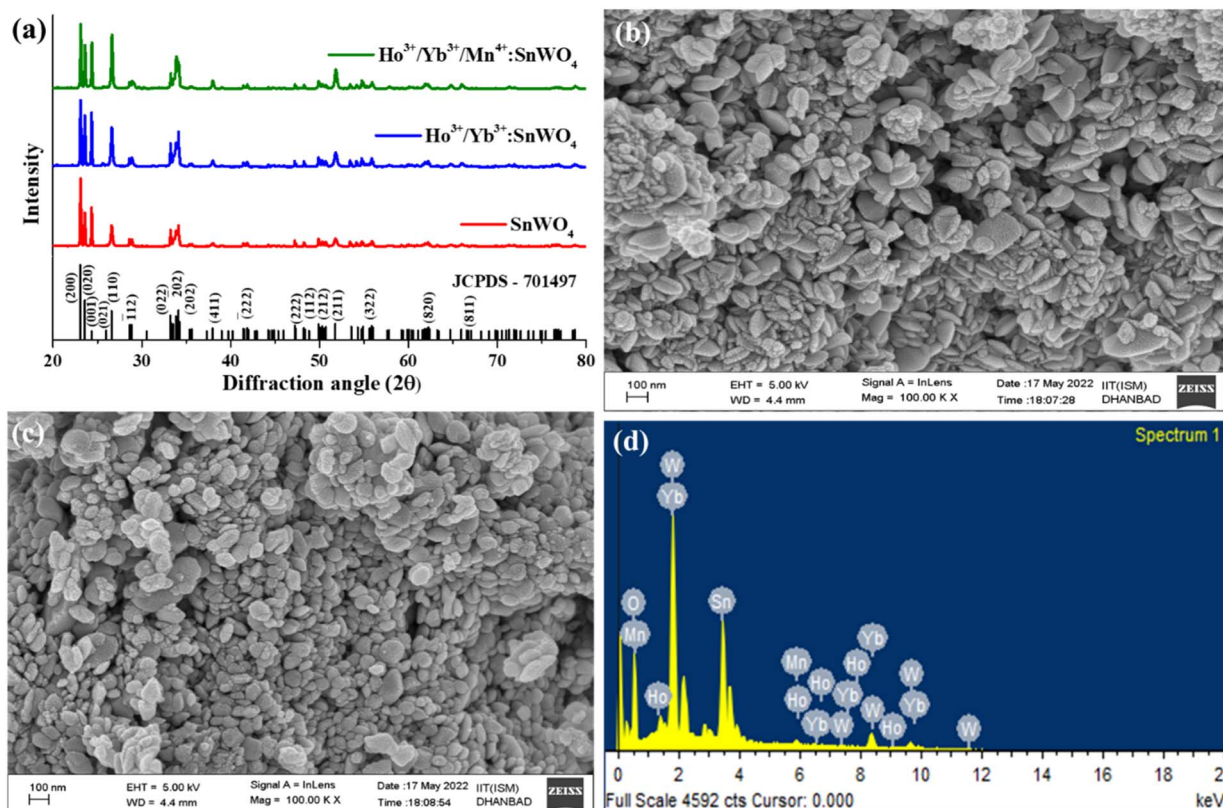


Fig. 1 (a) XRD spectra of SnWO_4 host, optimized $\text{Ho}^{3+}/\text{Yb}^{3+}$ and $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors. The FESEM images of (b) $\text{Ho}^{3+}/\text{Yb}^{3+}$ (c) $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors. (d) EDAX of $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors.



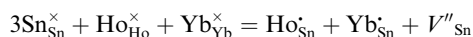
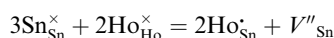
$$D = \frac{0.89\lambda}{\beta \cos \theta} \quad (1)$$

where ' λ ' implies the wavelength of X-ray, ' β ' designates the full width at half maxima and ' θ ' denotes Bragg's diffraction angle. The estimated crystal size for SnWO₄, Ho³⁺/Yb³⁺:SnWO₄ and Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors are determined to be 66.95 nm, 62.54 nm and 60.39 nm respectively. Further, the positive slope value obtained from the Williamson–Hall equation demonstrates the tensile strain created in the phosphors due to dopants ions.³³ The decrease in strain followed by the decrease in slope values (0.017, 0.016, 0.009) with codoping may be the reason for the decrease in crystallite size.^{32,34}

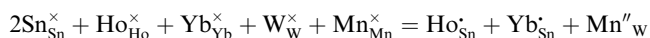
The absence of any extra peak in the developed phosphors, indicates that there is no impurity phase. The intense diffraction peaks present in the phosphors correspond to (200), (020), (001), (110), ($\bar{1}12$), (022), (202), (202), (411), (222), (222), (112), (212), (211), (322), (820) and (811) *hkl* planes.³⁵ Again, using relation between the radius of the host cation ' $R_m(\text{CN})$ ' and doped ion ' $R_d(\text{CN})$ ', the acceptable percent of doping (D_r) can be confirmed as,^{32,34}

$$D_r = 100 \times \frac{R_m(\text{CN}) - R_d(\text{CN})}{R_m(\text{CN})} \quad (2)$$

where the values of ionic radii for Sn²⁺, W⁶⁺, Ho³⁺, Yb³⁺ and Mn⁴⁺ are 0.69 Å, 0.60 Å, 0.90 Å, 0.86 Å, 0.53 Å in VI coordination, respectively.³⁶ The D_r value for Sn²⁺ and Ho³⁺/Yb³⁺ is 30% and 24%, respectively. Also, for W⁶⁺ and Mn⁴⁺ ions, the D_r value is 11%. The calculated D_r values in the present case is less than 30%. This implies that the dopants successfully occupy the cationic sites of the host. The substitution of Sn²⁺ by the RE ions creates Sn vacancy in the prepared Ho³⁺:SnWO₄ and Ho³⁺/Yb³⁺:SnWO₄ phosphors, due to which the SnWO₄ system shows charge imbalance that can be demonstrated by using the following Kroger Vink notation given,^{6,34,37}



The vacancy of oxygen ($\text{W}_{\text{W}}^{\times} + \text{Mn}_{\text{Mn}}^{\times} = \text{Mn}_{\text{W}}'' + V_{\text{O}}^{\cdot}$) in the SnWO₄ sites is produced when Mn⁴⁺ is integrated into SnWO₄ phosphors and W⁶⁺ is replaced by Mn⁴⁺. Thus, it can be applied to the method described below to compensate the Sn vacancy V_{Sn} :



3.1.2. FE-SEM and EDAX. The FESEM images {Fig. 1(b) and (c)} show the morphological properties of prepared phosphors. The image taken on 100 nm scale shows that the particles are agglomerated on the surface. Additionally, particle size appears to be decreasing from Ho³⁺/Yb³⁺ to Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors. The presence of all elements *viz.*, tin, tungsten, oxygen, holmium, ytterbium and manganese are confirmed by an elemental analysis using EDAX.

3.1.3. FTIR and Raman analysis. The vibrational features of the SnWO₄ host have been accomplished using the FTIR spectrum over the 400–4000 cm⁻¹ range {Fig. 2(a)}. The phonon energy corresponds to the maxima at 604 cm⁻¹ and 668 cm⁻¹. The asymmetric stretching bonds of the WO₆ group are responsible for these vibrations. The peaks at 768 cm⁻¹ and 850 cm⁻¹ resemble the stretching vibration of W–W and W–O–W bonds. The stretching and bending mode of the hydroxyl (–OH) group appears at 1634 cm⁻¹ and 3435 cm⁻¹.^{35,38}

Further, the developed phosphors have been characterized with Raman spectra in 100–1000 cm⁻¹ range {Fig. 2(b)}. The detected Raman peaks indicate that the cubic β -phase SnWO₄ phosphors have been successfully formed. According to the group theory, β -SnWO₄ consists of 36 vibration peaks, 18 of them are Raman, and rest of them are infrared active. In Mulliken notation, it is given by 6A, 6E, 6E, 18T.²⁷ Several Raman shifts are detected at 135 cm⁻¹, 275 cm⁻¹, 327 cm⁻¹, 718 cm⁻¹ and 807 cm⁻¹.²⁸ The stretching vibration of W–O bonds in the WO₆ octahedra appears at 718 cm⁻¹ and 807 cm⁻¹.³⁵ The peak intensity decreases on doping Ho³⁺/Yb³⁺/Mn⁴⁺, which shows the successful incorporation of dopants. When the dopants are added, the bond distance between two atoms decreases, and there is a reduction in the phonon excitation probability.³⁹

3.1.4. XPS analysis. The chemical analysis and binding energy of the elements present in the codoped phosphors have been confirmed with XPS spectroscopy. The spectrum (binding energy vs. count per second) has been recorded from 0–1200 eV {Fig. 2(c)}. The XPS spectrum shows that all dopants and host elements (Ho, Yb, Mn, Sn, W and O) are present in the phosphors. The peak observed at ~284 eV due to C1s is used for binding energy calibration. The survey scan shows the characteristic peaks of Sn (4d) at 25 eV. The spin–orbit doublets are detected at 484 eV (3d_{5/2}) and 495 eV (3d_{3/2}).³⁵ Additionally, the 3p_{3/2} and 3p_{1/2} states are responsible for the peaks at 715 eV and 756 eV. The XPS peaks for tungsten are found at 35 eV (4f_{7/2}), 247 eV (4d_{5/2}), 259 eV (4d_{3/2}), 426 eV (4p_{3/2}) and 495 eV (4p_{1/2}).^{6,40} The XPS peaks of oxygen are observed at 21 eV (O2s), 530 eV (O 1s) and 974 eV (OKLL).⁶ As the concentration of dopants is very low in the developed SnWO₄ phosphors, the intensity of Ho³⁺/Yb³⁺/Mn⁴⁺ is comparatively small. The 4d_{3/2} and 4d_{5/2} orbitals are responsible for the binding energy maxima, which are positioned at 159 eV and 184 eV respectively and at 7 eV due to the Yb (4f_{5/2}).⁵ The +4-oxidation state of manganese has been identified with peaks at 642 eV and 654 eV. It corresponds to the 2p_{3/2} and 2p_{1/2} orbitals due to spin–orbit splitting.⁴¹

3.2 Optical characterization

3.2.1. Diffuse reflectance spectra and bandgap analysis. The SnWO₄, codoped Ho³⁺/Yb³⁺ and Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors have been characterized with UV-vis spectroscopy to find the absorption bands in the 250–1250 nm wavelength range {Fig. 2(d)}. The spectra were calibrated with the standard BaSO₄ powder. There is no absorption peak for the SnWO₄ host. The doped/codoped SnWO₄ phosphors consist of absorption peaks at 538 nm and 640 nm that correspond to the ⁵I₈ → ⁵F₄, ⁵S₂ and ⁵I₈ → ⁵F₅ transitions. The 970 nm peak shows the ²F_{7/2}



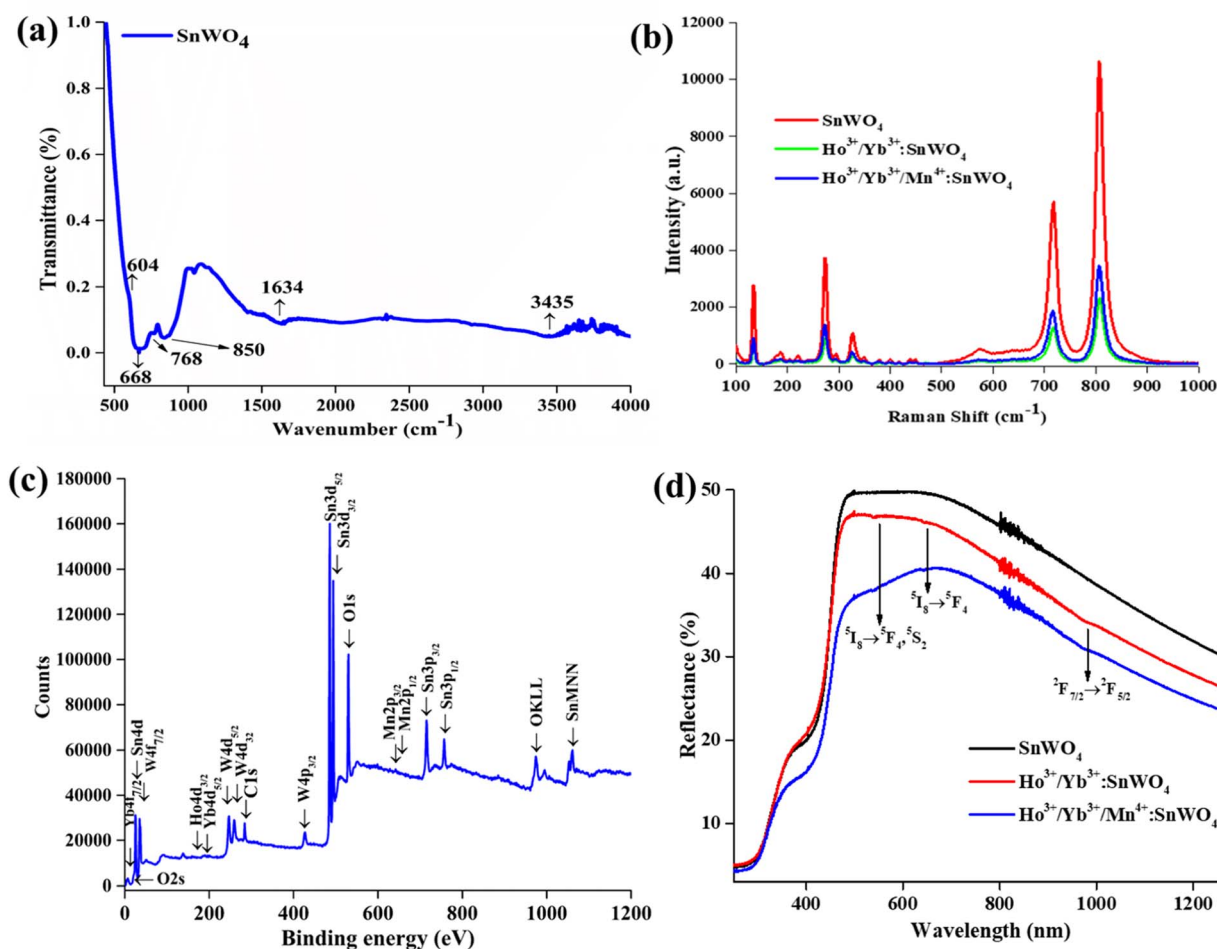


Fig. 2 (a) FTIR spectrum of SnWO₄ host, (b) Raman spectra of SnWO₄, Ho³⁺/Yb³⁺ and Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors, (c) XPS of Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors (d) UV-vis spectra of SnWO₄, Ho³⁺/Yb³⁺ and Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors.

→ ²F_{5/2} absorption transition from Yb³⁺. Applying the following relationship between the Kubelka Monk function and Tauc, the bandgap has been calculated from the DRS spectra,²⁵

$$[F(R_{\infty})/hv] = C(hv - E_g)^n \quad (3)$$

where, $F(R_{\infty})$ stands for Kubelka–Munk function which satisfies the following relation,

$$F(R_{\infty}) = (1 - R)^2/2R = K/S \quad (4)$$

where ‘ R ’ stands for diffuse reflectance of the spectrum, ‘ K ’ stands for absorption coefficient and ‘ S ’ stands for scattering coefficient. $F(R_{\infty})$ depicts the reflection of the infinitely thick sample against a reference at each wavelength, ‘ C ’ is a constant, ‘ hv ’ stands for photon energy, ‘ E_g ’ for the bandgap and $n = \frac{1}{2}$ for direct bandgap. The obtained values of bandgap are 3.60 eV, 3.54 eV and 3.66 eV, corresponding to SnWO₄, Ho³⁺/Yb³⁺:SnWO₄ and Ho³⁺/Yb³⁺/Mn⁴⁺:SnWO₄ phosphors, respectively. The results show that with the doping of larger atoms in the Sn²⁺ sites, the bandgap decreases, and with the smaller manganese doping, the bandgap increases.^{26,34}

3.2.2. Photoluminescence study and effect of Mn⁴⁺ codoping. Fig. 3(a) displays the UC emission of produced phosphors in the spectral range of 350–900 nm after excitation at 980 nm radiation. The emission from doped/codoped SnWO₄ phosphors has been enhanced by altering the concentration of Ho³⁺ (0.1, 0.3, 0.5 and 0.7 mol%)/Yb³⁺ (1.0, 1.5, 3.0 and 7.0 mol%)/Mn⁴⁺ (1.0, 5.0, 10.0, 15.0 and 20.0 mol%). The luminescence peaks found at ~541 nm and ~645 nm, arising from Ho³⁺ ions, are responsible due to the ⁵F₄, ⁵S₂ → ⁵I₈ and ⁵F₅ → ⁵I₈ transitions, respectively. Codoping of Yb³⁺ ions enhanced the obtained UC emission intensity. Due to the Yb³⁺ ion’s sensitizing effect, the green emission is intensified by 9 times that of Ho³⁺:SnWO₄ phosphors.

The UC emission intensity (I) and pump power (p) is correlated by the expression $I \propto p^n$, where n is the number of pump photons needed to populate the emitting levels. The calculated ‘ n ’ value for the ⁵F₄, ⁵S₂ → ⁵I₈ transition is 1.3. The decreased slope value in the Ho³⁺/Yb³⁺ doped SnWO₄ system is due to upconversion saturation at high power density, which is a consequence of the involvement of processes like energy transfer and non-radiative relaxation.^{42–44} The energy level diagram has been used to explain excitation and emission



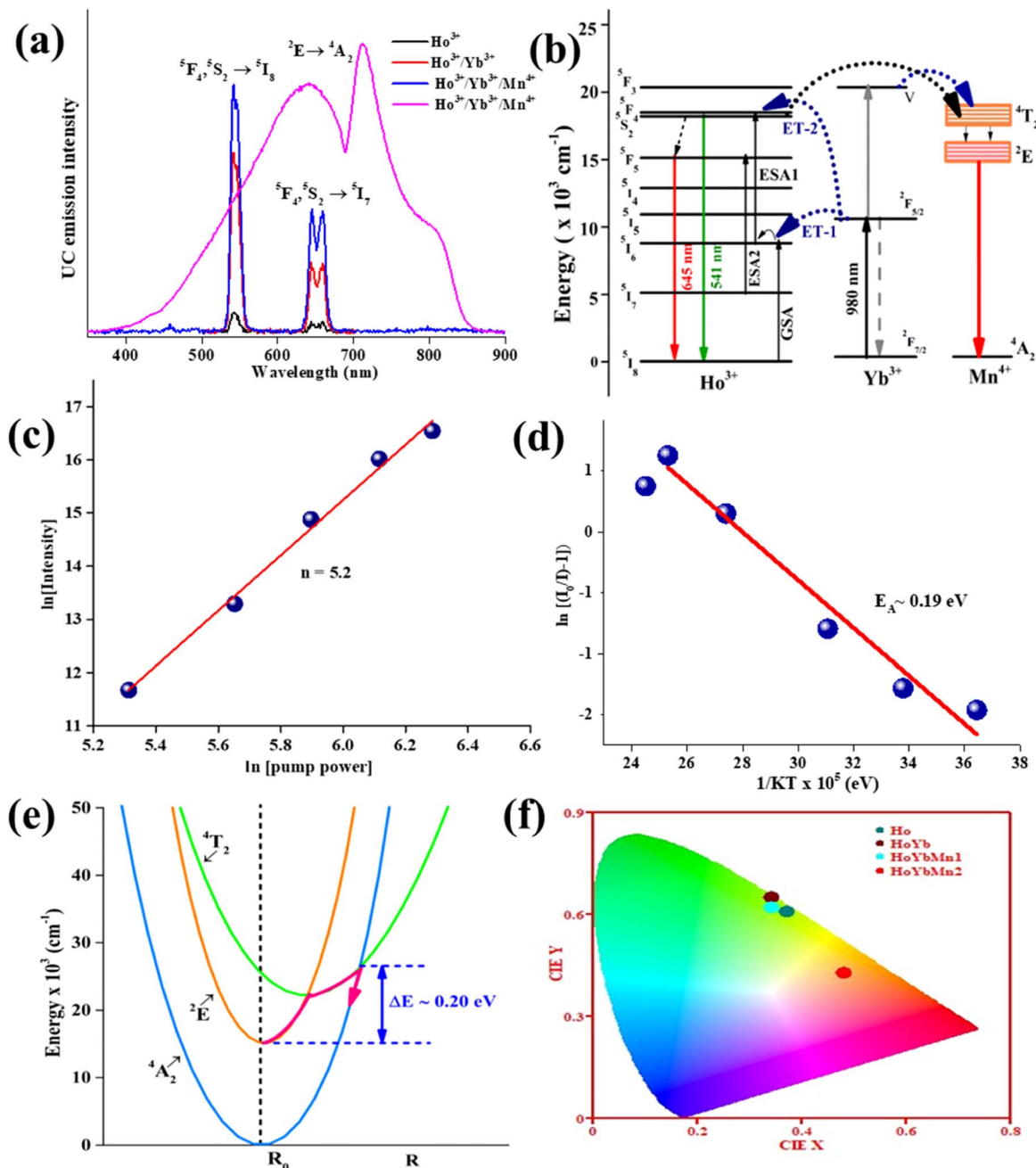


Fig. 3 (a) UC emission spectra of Ho^{3+} , $\text{Ho}^{3+}/\text{Yb}^{3+}$, $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors upon 980 nm excitation, (b) schematic energy level diagram of $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors, (c) slope value 'n' for broadband in $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors, (d) activation energy of $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors, (e) configuration coordinate diagrams showing thermal quenching in $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors (f) CIE coordinates for Ho^{3+} , $\text{Ho}^{3+}/\text{Yb}^{3+}$, $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors.

mechanisms of the Ho^{3+} , Yb^{3+} and Mn^{4+} dopants in SnWO_4 {Fig. 3(b)}. The luminescence from $\text{Ho}^{3+}/\text{Yb}^{3+}$ is the result of ESA, GSA and ET processes.^{5,45,46} The electrons are stimulated to the $^5\text{I}_6$ level by taking the 980 nm excitation energy through GSA, and the $^5\text{F}_4$, $^5\text{S}_2$ levels of Ho^{3+} ions are then populated by absorbing the second photon (ESA1). Again, the $^5\text{F}_4$ and $^5\text{S}_2$ levels are populated by the energy transfer from Yb^{3+} ions through ET-1 and ET-2 processes. The $^5\text{F}_4$, $^5\text{S}_2 \rightarrow ^5\text{I}_8$ transition results in the green emission at 541 nm. A part of the population at $^5\text{F}_4$, $^5\text{S}_2$ levels relaxes non-radiatively (involving 5 phonons) to

populate the next lower $^5\text{F}_5$ level. Also, the ESA2 from the $^5\text{I}_7$ level populates the $^5\text{F}_5$ level from where the $^5\text{F}_5 \rightarrow ^5\text{I}_8$ transition corresponding to the red emission at 645 nm occurs.

Further, on introducing the Mn^{4+} ions in the $\text{Ho}^{3+}/\text{Yb}^{3+}:\text{SnWO}_4$ phosphors, some fascinating results have been obtained. Here, Mn^{4+} acts as an intensity enhancer at low pump photon density and a spectra modifier on tuning the pump power density. In the phosphors, after codoping Mn^{4+} , the UC emission intensity of the green band is enhanced by 1.4 times and 13 times compared to the $\text{Ho}^{3+}/\text{Yb}^{3+}$ and Ho^{3+} , respectively. On increasing the pump



power ($<23.6 \text{ W cm}^{-2}$) the UC emission intensity enhances, and at higher pump power ($\geq 23.6 \text{ W cm}^{-2}$) there is a change in the sharp UC peak to broadband ranging from 400–900 nm due to the ${}^2\text{E} \rightarrow {}^4\text{A}_2$ transition. At 45.1 W cm^{-2} visible broadband occurs, and the intensity goes on increasing on a further increase of pump power density. The broadband emitting $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors is found with slope value 5.2 {Fig. 3(c)}, demonstrating the role of the photon avalanche (PA), which is supported by the cross-relaxation process.^{47–51} The single $\text{Mn}^{4+}:\text{SnWO}_4$ is not excited with 980 nm excitation because there is no equivalent energy level. Hence, the energy transfer from either Ho^{3+} or Yb^{3+} or both is the only possible way to populate the energy levels of Mn^{4+} . The energy transfer from $\text{Ho}^{3+}/\text{Yb}^{3+}$ ions to ground state ${}^4\text{A}_2$ of Mn^{4+} becomes prominently active when the pump photon number increases sufficiently. The two photons trigger other two-photons and further to four and sixteen photons and so on, simultaneously for a photon avalanche process, which populates the excited ${}^4\text{T}_2$ level of Mn^{4+} . Mn^{4+} ions are populated in the ${}^2\text{E}$ level through the non-radiative relaxation from the ${}^4\text{T}_2$ level. Owing to the ligand field dependency of Mn^{4+} ions, the broadband spectra corresponding to the ${}^2\text{E} \rightarrow {}^4\text{A}_2$ transition has two overlapped peaks at 640 and 710 nm, which is located in the red and deep red region. The peak at 710 nm arises in the broad spectra because of the simultaneous generation and annihilation of phonons caused by the electronic states of 3d electrons and lattice vibrations.^{52–56}

To further reveal the spectroscopic properties, the luminescence lifetimes measurements of $\text{Ho}^{3+}/\text{Yb}^{3+}$ and $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors have been performed upon 361 nm excitation. The ESI Fig. S(a) and S(b)† show the rise and decay time of green (530 nm) and red (645 nm) emissions corresponding to the ${}^5\text{F}_4, {}^5\text{S}_2 \rightarrow {}^5\text{I}_8$ and ${}^5\text{F}_5 \rightarrow {}^5\text{I}_8$ transitions, respectively, in $\text{Ho}^{3+}/\text{Yb}^{3+}$ and $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors. The decay curves are fitted to the exponential rise and decay function given by $[I = I_0[\exp(-) - \exp(-t/\tau_d)]]$ where ' I ' and ' I_0 ' are emission intensity at time ' t ' and ' 0 ', ' τ_r ', ' τ_d ' are the rise and decay times.^{57–59} The rise time is found to be 0.92 μs and 1.41 μs for green emission and 0.73 μs and 1.12 μs for red emission for $\text{Ho}^{3+}/\text{Yb}^{3+}$ and $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors. The decay time for $\text{Ho}^{3+}/\text{Yb}^{3+}$ and $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors are found to be 5.83 μs and 5.79 μs for green emission and 5.75 μs and 5.81 μs for red emission, respectively. There is a slight change in decay time has been observed. The decay time is the lifetime of the excited levels of the rare earth ions involved in the emission process. However, there is an increase in rise time for both 530 nm and 645 nm emissions has been observed, which shows an increment in the efficient energy transfer process among the dopant ions.

3.2.3. Concentration quenching and energy transfer interaction. On increasing the percentage of activator ions beyond 0.5 mole percent of Ho^{3+} and 5.0 mole percent of Mn^{4+} the decrease in luminescence intensity is recorded. Because the distance among dopant ions becomes smaller, and the sensitizer ions are insufficient to transfer the energy to such a smaller distance. In the $\text{Ho}^{3+}/\text{Yb}^{3+}:\text{SnWO}_4$ phosphors, the mean distance among Ho^{3+} ions have a significant influence on the energy transfer between them. As a result, the critical distance has been calculated using the Blesse equation,³⁴

$$R_c = 2 \left[\frac{3V}{4\pi X_c N} \right]^{1/3} \quad (5)$$

where ' V ' stands for unit cell volume, ' X_c ' for critical concentration, and ' N ' for the quantity of activator ions occupying each lattice site. In present case, $V = 388.84 \text{ \AA}^3$, X_c is 0.5 mol%, $N = 4$. After calculation, the R_c value is found to be 7.18 \AA . If R_c is more than 5, it means that rare earth interactions are not the result of exchange interactions.^{34,46} Hence, the multipolar interaction precedes the mechanism, which may be established by the relationship given by Dexter and Van Uitert,⁶⁰

$$\frac{I}{C} = K [1 + \beta C^{Q/3}]^{-1} \quad (6)$$

where ' I ' stands for the emission intensity corresponding to each concentration and ' C ' stands for the dopant ion concentration. ' K ' and ' β ' are constants, while ' Q ' denotes the nature of the interaction between the quenching center and luminescence center. The exchange interaction, electric dipole–dipole, electric dipole–quadrupole, and electric quadrupole–quadrupole interactions are represented by the distinct values of ' Q ' = 3, 6, 8, and 10. Since the value of $Q/3$ has been determined to be 2.60, Q is roughly equivalent to 8. This implied that the concentration quenching of Ho^{3+} ions in the SnWO_4 phosphors is caused by electric dipole–quadrupole interaction.^{53,61} However, by considering the concentration of Mn^{4+} ions, the energy transfer process from Ho^{3+} to Mn^{4+} can be identified. The found value of R_c is 3.2 \AA , which is less than 5 \AA . This shows that Ho^{3+} transfers energy to Mn^{4+} via short range exchange interaction in the $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors.⁶² The efficient energy migration is decided by the critical distance. It becomes important as the dopant's concentration increases. Beyond the critical concentration, further increase in the dopant's concentration brings the dopants too close (*i.e.*, below critical distance) and the attractive potential rises. The obtained value of critical distance confirms that at high concentration (5.0 mol% Mn^{4+}), *i.e.*, when there is a small distance between Mn^{4+} ions, the excitation energy effectively moves among Mn^{4+} ions till it hits a killing site and then decays non-radiatively. Thus, the concentration quenching decreases the UC emission intensity.

3.2.4. Thermal stability, configuration coordinates and CIE diagram. The temperature dependent UC spectra of $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors have been recorded over 303–473 K temperature range to determine the thermal stability. Temperature is a measurable parameter that controls the dynamics of the UC process. Thus, it becomes necessary to check any optical behavior change, such as spectral position, lifetime, CIE, CCT, *etc.*, before the device fabrication. As temperature increases, the intensity of UC emission decreases to 46% at 423 K compared to the initial intensity.^{5,62–65} Thermal quenching at higher temperatures is because of the non-radiative multiphonon relaxation caused by lattice vibrations. The relevant relation between phonon density (n), energy gap ($\hbar\omega$), Boltzmann constant (K) and the absolute temperature (T) can be given by equation $\langle n \rangle = 1 / \left[1 - \left(\frac{\hbar\omega}{KT} \right) \right]$.^{5,53} The activation energy can be determined by using the following Arrhenius equation,^{58,66}



$$I = \frac{I_0}{1 + C \exp\left(-\frac{E_A}{kT}\right)} \quad (7)$$

The calculated value of activation energy E_A for the broad emitting $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors is 0.19 eV {Fig. 3(d)}. Greater the activation energy of phosphors, more is the thermal stability. The activation energy and thermal quenching due to increased temperature can be illustrated by using the configuration coordinate {Fig. 3(e)}. The figure shows the $^4\text{A}_2$, ^2E and $^4\text{T}_2$ energy level parabola in terms of potential curve equilibrium position or internuclear distance (R). According to the configuration coordinate, the parabolas of $^4\text{A}_2$ and ^2E energy states do not cross each other. Thus, the quenching of UC intensity is not possible through $^4\text{A}_2$ and ^2E crossover. With an increase in temperature, thermal quenching is caused by the wide splitting of the $^4\text{T}_2$ and $^4\text{T}_1$ spin-quartet states of Mn^{4+} . As a result, there is horizontal displacement of the $^4\text{T}_2$ parabola. The ^2E , $^2\text{T}_1$, $^2\text{T}_2$ and $^4\text{A}_2$ levels are originated from the same t_2^3 electronic orbital and the $^4\text{T}_1$ and $^4\text{T}_2$ levels are formed from another t_2^2e orbital. Therefore, the ground state $^4\text{A}_2$ and excited state $^4\text{T}_2$ (or $^4\text{T}_1$) parabolas exhibit a significant lateral displacement. A greater displacement indicates a more intense electron–phonon interaction.⁶⁷ The lowest sublevel originating from the $^4\text{T}_2$ state is shifted nearer to the emitting ^2E state; due to that, the nonradiative transitions from ^2E to $^4\text{T}_2$ are promoted, which probably returns to the $^4\text{A}_2$ ground state.⁶⁸ Thus, thermal quenching occurs in the phosphors owing to thermally generated crossover from the $^4\text{T}_2$ excited state to the $^4\text{A}_2$ ground state.

To visualize the color emission, the CIE coordinates of all developed phosphors have been determined {Fig. 3(f)}. The found CIE coordinates are (0.37, 0.61), (0.34, 0.65), (0.34, 0.62) and (0.48, 0.43) for Ho^{3+} , $\text{Ho}^{3+}/\text{Yb}^{3+}$, $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}$ and broad emitting $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}$ codoped SnWO_4 phosphors, respectively. The observation shows that upon Yb^{3+} ions codoping in $\text{Ho}^{3+}:\text{SnWO}_4$ phosphors, the CIE coordinates shifted towards a more greenish region, *i.e.*, the incorporation of Yb^{3+} enhances the purity. However, due to the codoping of Mn^{4+} , the CIE coordinates shift towards reddish region. Also, the CCT values have been calculated for all the phosphors, and the calculated values are 4937 K, 5403 K, 5395 K and 2579 K, respectively. The observation shows that from single Ho^{3+} doped to $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}$ codoped phosphors, the CCT value shifted from cool light to warm light region.⁶⁹ Thus, the present phosphors can be used in green and broad visible upconverters.

4. Conclusion

The green and broad emitting $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors synthesized effectively by the solid-state reaction approach have been structurally and optically studied. Charge compensation boosted the green emission of Ho^{3+} in codoped SnWO_4 phosphors by 13 times after incorporating Mn^{4+} ions. The broad UC emission is induced by the photon avalanche energy transfer process. Critical distance has been used to demonstrate the concentration quenching owing to exchange

interaction in the $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors. The temperature-dependent luminescence quenching of the broad emitting $\text{Ho}^{3+}/\text{Yb}^{3+}/\text{Mn}^{4+}:\text{SnWO}_4$ phosphors has been described by using a configuration coordinate diagram. From these above experimental investigations, it can be inferred that the generated phosphors may be utilized for broadband visible optical devices.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

Ms Manisha Prasad is thankful to IIT(ISM), Dhanbad for providing the research fellowship.

References

- H. D. Nguyen and R. S. Liu, Narrow-band red-emitting Mn^{4+} -doped hexafluoride phosphors: synthesis, optoelectronic properties, and applications in white light-emitting diodes, *J. Mater. Chem. C*, 2016, **4**, 10759–10775.
- M. Prasad and V. K. Rai, Frequency Upconversion in UCNPs Containing Transition Metal Ions, *Upconverting Nanoparticles: From Fundamentals to Applications*, 2022, pp. 141–169.
- R. Cao, F. Zhang, C. Cao, X. Yu, A. Liang, S. Guo and H. Xue, Synthesis and luminescence properties of $\text{CaAl}_2\text{O}_4:\text{Mn}^{4+}$ phosphor, *Opt. Mater.*, 2014, **38**, 53–56.
- X. Xu, J. Ren, N. Huang, H. Zeng, G. Chen, D. Kong and H. Tao, Broadly tunable emission from Mn-doped zinc gallogermanate phosphors through composition modification, *Opt. Mater. Express*, 2014, **4**, 2433–2440.
- M. Prasad and V. K. Rai, Influence of transition metal ions: Broad band upconversion emission in thermally stable phosphors, *J. Alloys Compd.*, 2020, **837**, 155289.
- M. Prasad and V. K. Rai, Simultaneous effects of synthesis temperature and dopants on MgWO_4 UC phosphors, *Methods Appl. Fluoresc.*, 2022, **10**, 034004.
- L. Dong, L. Zhang, Y. Jia, Y. Xu, S. Yin and H. You, $\text{ZnGa}_{2-y}\text{Al}_y\text{O}_4:\text{Mn}^{2+}, \text{Mn}^{4+}$ Thermo-chromic Phosphors: Valence State Control and Optical Temperature Sensing, *Inorg. Chem.*, 2020, **59**, 15969–15976.
- L. Dong, L. Zhang, Y. Jia, B. Shao, W. Lü, S. Zhao and H. You, Site occupation and luminescence of novel orange-red $\text{Ca}_3\text{M}_2\text{Ge}_3\text{O}_{12}:\text{Mn}^{2+}, \text{Mn}^{4+}$ ($\text{M} = \text{Al}, \text{Ga}$) phosphors, *ACS Sustainable Chem. Eng.*, 2020, **8**, 3357–3366.
- H. Zhou, N. Guo, M. Zhu, J. Li, Y. Miao and B. Shao, Photoluminescence and ratiometric optical thermometry in $\text{Mn}^{4+}/\text{Eu}^{3+}$ dual-doped phosphor via site-favorable occupation, *J. Lumin.*, 2020, **224**, 117311.
- L. Li, Y. Pan, Z. Chen, S. Huang and M. Wu, Tunable luminescence and energy transfer properties of Bi^{3+} and Mn^{4+} co-doped $\text{Ca}_{14}\text{Al}_{10}\text{Zn}_6\text{O}_{35}$ phosphors for agricultural applications, *RSC Adv.*, 2017, **7**, 14868–14875.



- 11 H. Zhang, Y. Chen, X. Zhu, K. Liu, H. Zhou, X. Sun, Y. Feng, *et al.* The deep red $\text{Ca}_2\text{YZr}_2\text{Al}_3\text{O}_{12}$: Mn^{4+} phosphor and enhanced emission by Bi^{3+} doping, *J. Lumin.*, 2021, **236**, 118131.
- 12 Q. Sai, C. Xia, H. Rao, X. Xu, G. Zhou and P. Xu, Mn, Cr-codoped MgAl_2O_4 phosphors for white LEDs, *J. Lumin.*, 2011, **131**, 2359–2364.
- 13 X. Zhang and M. Gong, Single-phased white-light-emitting NaCaBO_3 : Ce^{3+} , Tb^{3+} , Mn^{2+} phosphors for LED applications, *Dalton Trans.*, 2014, **43**, 2465–2472.
- 14 W. Liang and Y. Wang, Energy transfer between Pr^{3+} and Mn^{2+} in $\text{K}_2\text{YZr}(\text{PO}_4)_3$: Pr, Mn phosphor, *Mater. Chem. Phys.*, 2011, **127**, 170–173.
- 15 X. Qu, L. Cao, W. Liu, G. Su and P. Wang, Luminescence properties of CdSiO_3 : Mn^{2+} , RE^{3+} (RE = Sm, Dy, Eu) phosphors, *J. Alloys Compd.*, 2009, **487**, 387–390.
- 16 R. Valiente, O. S. Wenger and H. U. Güdel, Near-infrared-to-visible photon upconversion process induced by exchange interactions in Yb^{3+} -doped RbMnCl_3 , *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2001, **63**, 165102.
- 17 Z. T. Chen, E. H. Song, M. Wu, S. Ding, S. Ye and Q. Y. Zhang, Bidirectional energy transfer induced single-band red upconversion emission of Ho^{3+} in KZnF_3 : Mn^{2+} , Yb^{3+} , Ho^{3+} nanocrystals, *J. Alloys Compd.*, 2016, **667**, 134–140.
- 18 Y. Lv, Y. Jin, T. Sun, J. Su, C. Wang, G. Ju, Y. Hu, *et al.* Visible to NIR down-shifting and NIR to visible upconversion luminescence in $\text{Ca}_{14}\text{Zn}_6\text{Ga}_{10}\text{O}_{35}$: Mn^{4+} , Ln^{3+} (Ln = Nd, Yb, Er), *Dyes Pigm.*, 2019, **161**, 137–146.
- 19 J. Liao, Q. Wang, H. R. Wen, H. Yuan, S. J. Liu, J. Fu and B. Qiu, First observation of mutual energy transfer of Mn^{4+} – Er^{3+} via different excitation in $\text{Gd}_2\text{ZnTiO}_6$: Mn^{4+} / Er^{3+} phosphors, *J. Mater. Chem. C*, 2017, **5**, 9098–9105.
- 20 E. H. Song, S. Ding, M. Wu, S. Ye, Z. T. Chen, Y. Y. Ma and Q. Y. Zhang, Tunable white upconversion luminescence from Yb^{3+} - Tm^{3+} - Mn^{2+} tri-doped perovskite nanocrystals, *Opt. Mater. Express*, 2014, **4**, 1186–1196.
- 21 X. Qin, X. Liu, W. Huang, M. Bettinelli and X. Liu, Lanthanide-activated phosphors based on 4f–5d optical transitions: theoretical and experimental aspects, *Chem. Rev.*, 2017, **117**, 4488–4527.
- 22 K. Ogasawara, S. Watanabe, Y. Sakai, H. Toyoshima, T. Ishii, M. G. Brik and I. Tanaka, Calculations of complete $4f^n$ and $4f^{n-1} 5d^1$ energy level schemes of free trivalent rare-earth ions, *Jpn. J. Appl. Phys.*, 2004, **43**, L611.
- 23 I. Gupta, S. Singh, S. Bhagwan and D. Singh, Rare earth (RE) doped phosphors and their emerging applications: A review, *Ceram. Int.*, 2021, **47**, 19282–19303.
- 24 G. Liu, Advances in the theoretical understanding of photon upconversion in rare-earth activated nanophosphors, *Chem. Soc. Rev.*, 2015, **44**, 1635–1652.
- 25 M. Prasad and V. K. Rai, Thermally stable upconverting $\text{Na}_3\text{Zr}_2(\text{SiO}_4)_2\text{PO}_4$: Er^{3+} / Yb^{3+} phosphors for displays and optical thermometry, *J. Alloys Compd.*, 2022, **911**, 164968.
- 26 W. Jeitschko and A. W. Sleight, Synthesis, properties and crystal structure of β - SnWO_4 , *Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem.*, 1972, **28**, 3174–3178.
- 27 J. L. Solis, J. Frantti, V. Lantto, L. Häggström and M. Wikner, Characterization of phase structures in semiconducting SnWO_4 powders by Mössbauer and Raman spectroscopies, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1998, **57**, 13491.
- 28 F. A. Alharthi, M. A. Alsaiani, M. S. Jalalah, M. Shashank, A. A. Alghamdi, J. S. Algethami and N. Ganganagappa, Combustion synthesis of β - SnWO_4 -rGO: Anode material for Li-ion battery and photocatalytic dye degradation, *Ceram. Int.*, 2021, **47**, 10291–10300.
- 29 C. Seidl, J. Ungelenk, E. Zittel, T. Bergfeldt, J. P. Sleeman, U. Schepers and C. Feldmann, Tin tungstate nanoparticles: a photosensitizer for photodynamic tumor therapy, *ACS Nano*, 2016, **10**, 3149–3157.
- 30 P. Han, B. Sun, S. Cheng, F. Yu, B. Jiao and Q. Wu, An optoelectronic resistive switching memory behavior of Ag/α - SnWO_4 /FTO device, *J. Alloys Compd.*, 2016, **681**, 516–521.
- 31 M. Zhang, Z. Cui, R. Song, B. Lv, Z. Tang, X. Meng, W. Bu, *et al.* SnWO_4 -based nanohybrids with full energy transfer for largely enhanced photodynamic therapy and radiotherapy, *Biomaterials*, 2018, **155**, 135–144.
- 32 L. Mukhopadhyay, V. K. Rai, R. Bokolia and K. Sreenivas, 980 nm excited Er^{3+} / Yb^{3+} / Li^+ / Ba^{2+} : NaZnPO_4 upconverting phosphors in optical thermometry, *J. Lumin.*, 2017, **187**, 368–377.
- 33 P. Ghosh, J. Oliva, E. D. L. Rosa, K. K. Haldar, D. Solis and A. Patra, Enhancement of upconversion emission of LaPO_4 : $\text{Er}@\text{Yb}$ core–shell nanoparticles/nanorods, *J. Phys. Chem. C*, 2008, **112**, 9650–9658.
- 34 S. Pattnaik and V. K. Rai, Impact of charge compensation on optical and thermometric behaviour of titanate phosphors, *Mater. Res. Bull.*, 2020, **125**, 110761.
- 35 S. R. Ede and S. Kundu, Microwave synthesis of SnWO_4 nanoassemblies on DNA scaffold: a novel material for high performance supercapacitor and as catalyst for butanol oxidation, *ACS Sustainable Chem. Eng.*, 2015, **3**, 2321–2336.
- 36 R. D. Shannon, Revised effective ionic radii and systematic studies of interatomic distances in halides and chalcogenides, *Acta Crystallogr., Sect. A: Cryst. Phys., Diffr., Theor. Gen. Crystallogr.*, 1976, **32**, 751–767.
- 37 S. Wang, Y. J. Han, L. Shi, M. X. Jia, Y. L. Tong, N. N. Bao, S. L. Niu, *et al.* Charge compensation assisted enhanced photoluminescence derived from Al^{3+} -codoped NaLaMgWO_6 : Mn^{4+} phosphors for plant growth lighting applications, *J. Lumin.*, 2020, **226**, 117438.
- 38 H. N. Sumedha, M. A. Alsaiani, M. S. Jalalah, M. Shashank, F. A. Alharthi, N. Ahmad, N. Ganganagappa, *et al.* Rapid Microwave Synthesis of β - SnWO_4 Nanoparticles: An Efficient Anode Material for Lithium Ion Batteries, *Crystals*, 2021, **11**, 334.
- 39 T. Badapanda, V. Senthil, S. Panigrahi and S. Anwar, Diffuse phase transition behavior of dysprosium doped barium titanate ceramic, *J. Electroceram.*, 2013, **31**, 55–60.
- 40 S. Wang, H. Gao, C. Chen, Q. Li, C. Li, Y. Wei and L. Fang, Effect of phase transition on optical and photoluminescence properties of nano- MgWO_4 phosphor prepared by a gamma-ray irradiation assisted polyacrylamide gel method, *J. Mater. Sci.: Mater. Electron.*, 2019, **30**, 15744–15753.



- 41 F. Grote, R. S. Kühnel, A. Balducci and Y. Lei, Template assisted fabrication of free-standing MnO₂ nanotube and nanowire arrays and their application in supercapacitors, *Appl. Phys. Lett.*, 2014, **104**, 053904.
- 42 A. Nadort, J. Zhao and E. M. Goldys, Lanthanide upconversion luminescence at the nanoscale: fundamentals and optical properties, *Nanoscale*, 2016, **8**, 13099–13130.
- 43 M. Pollnau, D. R. Gamelin, S. R. Lüthi, H. U. Güdel and M. P. Hehlen, Power dependence of upconversion luminescence in lanthanide and transition-metal-ion systems, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2000, **61**, 3337.
- 44 M. K. Mahata, T. Koppe, T. Mondal, C. Brüsewitz, K. Kumar, V. K. Rai, U. Vetter, *et al.* Incorporation of Zn²⁺ ions into BaTiO₃: Er³⁺/Yb³⁺ nanophosphor: an effective way to enhance upconversion, defect luminescence and temperature sensing, *Phys. Chem. Chem. Phys.*, 2015, **17**, 20741–20753.
- 45 M. Mondal and V. K. Rai, Ho³⁺-Yb³⁺: YMoO₄ core@ shell nanoparticles for enhanced visible upconversion and security applications, *J. Alloys Compd.*, 2018, **750**, 304–311.
- 46 Z. T. Chen, E. H. Song, M. Wu, S. Ding, S. Ye and Q. Y. Zhang, Bidirectional energy transfer induced single-band red upconversion emission of Ho³⁺ in KZnF₃: Mn²⁺, Yb³⁺, Ho³⁺ nanocrystals, *J. Alloys Compd.*, 2016, **667**, 134–140.
- 47 G. Bilir, G. Ozen, M. Bettinelli, F. Piccinelli, M. Cesaria and B. Di Bartolo, Broadband Visible Light Emission From Nominally Undoped and Cr³⁺ Doped Garnet Nanopowders, *IEEE Photonics J.*, 2014, **6**, 1–11.
- 48 Y. Zhu, W. Xu, C. Li, H. Zhang, B. Dong, L. Xu, H. Song, *et al.* Broad white light and infrared emission bands in YVO₄: Yb³⁺, Ln³⁺ (Ln³⁺ = Er³⁺, Tm³⁺, or Ho³⁺), *Appl. Phys. Express*, 2012, **5**, 092701.
- 49 L. Marciniak, W. Strek, D. Hreniak and Y. Guyot, Temperature of broadband anti-Stokes white emission in LiYbP₄O₁₂: Er nanocrystals, *Appl. Phys. Lett.*, 2014, **105**, 173113.
- 50 X. Chen, W. Xu, Y. Zhu, P. Zhou, S. Cui, L. Tao, H. Song, *et al.* Nd₂O₃/Au nanocomposites: upconversion broadband emission and enhancement under near-infrared light excitation, *J. Mater. Chem. C*, 2014, **2**, 5857–5863.
- 51 W. Xu, X. Min, X. Chen, Y. Zhu, P. Zhou, S. Cui, H. Song, *et al.* Ag-SiO₂-Er₂O₃ nanocomposites: Highly effective upconversion luminescence at high power excitation and high temperature, *Sci. Rep.*, 2014, **4**, 5087.
- 52 D. Chen, Y. Zhou and J. Zhong, A review on Mn⁴⁺ activators in solids for warm white light-emitting diodes, *RSC Adv.*, 2016, **6**(89), 86285–86296.
- 53 *Upconverting Nanoparticles: From Fundamentals to Applications*, ed. V. K. Rai, John Wiley & Sons, 2022.
- 54 K. Sankarasubramanian, B. Devakumar, G. Annadurai, L. Sun, Y. J. Zeng and X. Huang, Novel SrLaAlO₄: Mn⁴⁺ deep-red emitting phosphors with excellent responsiveness to phytochrome PFR for plant cultivation LEDs: synthesis, photoluminescence properties, and thermal stability, *RSC Adv.*, 2018, **8**, 30223–30229.
- 55 T. Hu, H. Lin, Y. Cheng, Q. Huang, J. Xu, Y. Gao, Y. Wang, *et al.* A highly-distorted octahedron with a C_{2v} group symmetry inducing an ultra-intense zero phonon line in Mn⁴⁺-activated oxyfluoride Na₂WO₂F₄, *J. Mater. Chem. C*, 2017, **5**, 10524–10532.
- 56 Y. Lv, Y. Jin, T. Sun, J. Su, C. Wang, G. Ju, Y. Hu, *et al.* Visible to NIR down-shifting and NIR to visible upconversion luminescence in Ca₁₄Zn₆Ga₁₀O₃₅: Mn⁴⁺, Ln³⁺ (Ln = Nd, Yb, Er), *Dyes Pigm.*, 2019, **161**, 137–146.
- 57 J. Grigorjevaite and A. Katelnikovas, Up-Converting K₂Gd(PO₄)(WO₄): 20% Yb³⁺, Ho³⁺ Phosphors for Temperature Sensing, *Materials*, 2023, **16**, 917.
- 58 M. Prasad and V. K. Rai, Coactivated cyan emitting phosphors in optical thermometry using thermally and non-thermally coupled levels, *Mater. Res. Bull.*, 2022, **160**, 112116.
- 59 B. Bondzior, D. Stefańska, T. H. Q. Vü, N. Miniajluk-Gawel and P. J. Dereń, Red luminescence with controlled rise time in La₂MgTiO₆: Eu³⁺, *J. Alloys Compd.*, 2021, **852**, 157074.
- 60 M. Prasad, M. Mondal, L. Mukhopadhyay, S. Pattnaik and V. K. Rai, Photoluminescence investigation in tungstate-based materials, *Mater. Today: Proc.*, 2021, **46**, 6388–6391.
- 61 Z. Jia and M. Xia, Blue-green tunable color of Ce³⁺/Tb³⁺ coactivated NaBa₃La₃Si₆O₂₀ phosphor via energy transfer, *Sci. Rep.*, 2016, **6**, 1–9.
- 62 W. H. Du, N. Zhang, N. Z. Zhuo, L. Y. Xie, T. Jiang, Y. H. Zhu, H. B. Wang, *et al.* Photoluminescence properties and energy transfer of apatite-type Sr₃LaNa(PO₄)₃F: Ce³⁺, Tb³⁺ phosphors, *Mater. Res. Express*, 2019, **6**, 126209.
- 63 K. Sankarasubramanian, B. Devakumar, G. Annadurai, L. Sun, Y. J. Zeng and X. Huang, Novel SrLaAlO₄: Mn⁴⁺ deep-red emitting phosphors with excellent responsiveness to phytochrome PFR for plant cultivation LEDs: synthesis, photoluminescence properties, and thermal stability, *RSC Adv.*, 2018, **8**, 30223–30229.
- 64 L. Mukhopadhyay and V. K. Rai, Thermally stable red emitting xenotime phosphate nanophosphors for displays, *Mater. Res. Bull.*, 2020, **121**, 110628.
- 65 H. Chen, H. Lin, Q. Huang, F. Huang, J. Xu, B. Wang, Y. Wang, *et al.* A novel double-perovskite Gd₂ZnTiO₆:Mn⁴⁺ red phosphor for UV-based w-LEDs: structure and luminescence properties, *J. Mater. Chem. C*, 2016, **4**, 2374–2381.
- 66 S. Pattnaik and V. K. Rai, Tailoring of upconversion luminescence of Al³⁺ engineered titanate phosphor for non-invasive thermometry, *Methods Appl. Fluoresc.*, 2022, **10**, 034002.
- 67 B. Henderson and G. F. Imbusch, *Optical spectroscopy of inorganic solids*, Oxford University Press, 2006, vol. 44.
- 68 T. Senden, R. J. van Dijk-Moes and A. Meijerink, Quenching of the red Mn⁴⁺ luminescence in Mn⁴⁺-doped fluoride LED phosphors, *Light: Sci. Appl.*, 2018, **7**, 1–13.
- 69 J. Dutta and V. K. Rai, Zirconia based Ho³⁺-Yb³⁺ codoped upconverting nanophosphors for green light emitting devices applications, *Methods Appl. Fluoresc.*, 2018, **6**, 025003.

