RSC Advances



PAPER

View Article Online
View Journal | View Issue



Cite this: RSC Adv., 2023, 13, 22335

Fluorinated hydrogel nanoparticles with regulable fluorine contents and T_2 relaxation times as 19 F MRI contrast agents†

Ziwei Duan,^a Changjiang Liu,^a Junjie Tang,^a Ruling Zhang,^a Danfeng Peng,^b Ruitao Lu,^b Zong Cao*^a and Dalin Wu^b*^a

Medical imaging contrast agents that are able to provide detailed biological information have attracted increasing attention. Among the new emerging imaging contrast agents, 19 F magnetic resonance imaging contrast agents (19 F MRI CAs) are extremely promising for their weak background disturbing signal from the body. However, to prepare 19 F MRI CAs with a long T_2 relaxation time and excellent biocompatibility in a simple and highly effective strategy is still a challenge. Herein, we report a new type of 19 F MRI hydrogel nanocontrast agents (19 F MRI HNCAs) synthesized by a surfactant-free emulsion polymerization with commercial fluorinated monomers. The T_2 relaxation time of 19 F MRI HNCA-1 was found to be 25–40 ms, guaranteeing its good imaging ability *in vitro*. In addition, according to an investigation into the relationship between the fluorine content and 19 F MRI signal intensity, the 19 F MRI signal intensity was not only determined by the fluorine content in 19 F MRI HNCAs but also by the hydration microenvironment around the fluorine atoms. Moreover, 19 F MRI HNCAs demonstrated excellent biocompatibility and imaging capability inside cells. The primary exploration demonstrated that 19 F MRI HNCAs as a new type of 19 F MRI contrast agent hold potential for imaging lesion sites and tracking cells *in vivo* by 19 F MRI technology.

Received 29th April 2023 Accepted 10th July 2023

DOI: 10.1039/d3ra02827e

rsc.li/rsc-advances

Introduction

Magnetic resonance imaging (MRI) has gained increasing attention in modern medical diagnosis, because it can allow collecting high-quality information of soft tissues without the use of harmful radioactive nuclides. 1-3 Traditionally, MRI technology depends on the local differences in proton spin densities and the relaxation rates of water protons in vivo. However, the above-mentioned two parameters do not have significant differences between detecting sites and healthy tissues in the body, resulting in a failure to obtain sufficient contrast or provide precise information. As a result, contrast agents (CAs) are normally introduced in MRI examination to enhance the image contrast by regulating the relaxation properties of the neighboring water protons.^{4,5} At present, the normally introduced ¹H MRI CAs in clinical use are paramagnetic and superparamagnetic meta ion-based species, such as gadolinium chelates and iron oxide nanoparticles. 6,7 Although a significant enhancement of MRI performance by applying

To further improve the safety of MRI, researchers have started to develop other types of MRI CAs based on heteronuclear atoms, such as fluorine.11,12 Different from 1H MRI CAs that influence the relaxation properties of nearby water protons, the fluorine atoms in 19F MRI CAs are able to be visualized directly by MRI equipment.13 In addition, 19F has a 100% natural abundance and a gyromagnetic ratio close to hydrogen (83% compared with hydrogen), giving 19F MRI excellent imaging resolution.14 Another advantage of 19F MRI is the images have a higher imaging contrast due to the absence of 19F atoms in the detecting domains.15 In addition, the signal intensity in ¹⁹F MRI is proportional to the fluorine content, enabling the quantitative application of ¹⁹F MRI. ¹⁶ ¹⁹F MRI is able to provide more insightful information of soft tissues in vivo over ¹H MRI since the fluorine atoms in vivo are mainly embedded in the solid matrices of teeth and bones in vivo. 17 In principle, any species containing fluorine atoms, including small molecules, 18,19 macromolecules, 20 and nano-objectives, 21 has potential as ¹⁹F MRI CAs.

paramagnetic and superparamagnetic meta ion-based compounds has been achieved, the ¹H MRI CAs still possess some limitations, especially their application security, ⁸ which should be seriously considered. For example, gadolinium-based CAs are able to cause nephrogenic systemic fibrosis in patients with impaired kidney function according to a recent symmetrically study. ^{9,10}

[&]quot;School of Biomedical Engineering, Shenzhen Campus of Sun Yat-Sen University, Shenzhen, 518107, China. E-mail: wudlin6@mail.sysu.edu.cn; caozhong@mail.sysu.edu.cn

bShenzhen International Institute for Biomedical Research, Shenzhen, 518109, China † Electronic supplementary information (ESI) available. See DOI: https://doi.org/10.1039/d3ra02827e

Perfluorocarbons (PFCs), because of their high fluorine content, have been widely investigated as ¹⁹F MRI CAs.²² In practice, PFCs are not the best molecules as ¹⁹F MRI CAs, because of their serious accumulation in the liver and spleen.^{23,24} In addition, the low boiling point of PFCs prohibits this type of ¹⁹F MRI CAs from achieving long-term storage in vitro and circulation in vivo, even though the PFC emulsions are stabilized by robust surfactants. 25,26 The fluorine content in 19F MRI CAs is not the only parameter determining the imaging result of ¹⁹F MRI CAs. For high-efficiency ¹⁹F MRI CAs, fluorine atoms should hold very excellent local mobility in 19F MRI CAs under hydration condition, which can ensure a T_2 relaxation time of ¹⁹F MRI CAs above 10 ms to guarantee sufficient signal collecting intensity during measurement.27 It is well known that the strong hydrophobicity of fluorine atoms causes them to aggregate in 19F MRI CAs, further lose their local mobility, and shorten the T_2 relaxation time under hydration conditions.²⁸ In order to solve this critical problem, various strategies have been explored by researchers, such as copolymerizing fluorine-based monomers with various hydrophilic monomers to create a hydration environment around fluorine atoms,29,30 synthesizing hydrophilic monomers appending fluorine atoms,31 and developing smart "On-Off" signal amplifying 19F MRI CAs echoing physiological triggers.32,33 Changkui Fu et al. recently reported water-soluble fluoropolymers with sulfoxide side-china groups as low-fouling ¹⁹F MRI CAs ($T_2 = 373-431$ ms). Their results demonstrated that introducing water-soluble groups around fluorine atoms in ¹⁹F MRI CAs is an impactful strategy to prolong the T_2 relaxation time and enhance the imaging efficacy in vitro and in vivo.34 More directly, Jinhao Gao et al. applied water-soluble fluorine-containing ionic liquids ($T_2 = 4.4$ s) as fluorine markers in ¹⁹F MRI. They encapsulated a fluorinecontaining ionic liquid inside porous silica nanoparticles and the ionic liquids were sealed by pH-responsive polymers on the surface of the porous silica nanoparticles. Their hybrid system acted as a pH-responsive "Off-On" 19 F MRI CA. 35 Apart from the mobility of fluorine atoms in the ¹⁹F MRI CAs, the magnetically equivalency of the fluorine nuclei is another determinative factor in the ¹⁹F MRI CAs performance, because un-equivalent

In order to achieve high performance in ¹⁹F MRI, the influence of the topological structures (linear, block, hyperbranched, star-like) and nanostructures (micelles, vesicles, and worm-like micelles) of ¹⁹F MRI CAs on the imaging performance have also been explored in depth,^{37–43} because the mobility of the fluorine atoms in ¹⁹F MRI CAs is determined by the topological structures, while the uptake and accumulation behavior of the ¹⁹F MRI CAs inside disease lesions are influenced by their nanostructures.44,45 For example, Kristofer J. Thurecht et al. explored the influence of the topology structure on the T_2 relaxation times and imaging performance of 19F MRI CAs by synthesizing a series of hyperbranched polymeric scaffolds. The longest T_2 relaxation time of 71 ms could be obtained by incorporating hydrophilic oligo(ethyl glycol) ($M_{\rm w}=400$) in their hyperbranched 19F MRI CAs.29 In the last two decades, polymerization-induced self-assembly (PISA) as a heterogeneous polymerization technology has been developed as an

fluorine nuclei can result in blurry MR images.36

extraordinary powerful method to fabricate different nano-structures of amphiphiles, and accordingly, PISA has been also received increasing attention in 19 F MRI CAs preparation. Wei Zhao *et al.* first prepared poly(oligo(ethylene glycol) methyl ether acrylate-co-2,2,2-trifluoroethyl acrylate) (poly(OEGA-co-TFEA)) as a macro-RAFT chain transferring agent (macro-CTA). After that, AIBN initiated the PISA of styrene and 3-vinyl benzaldehyde with macro-CTA in isopropanol to form 19 F MRI CAs with micelles, worm-like micelles, and polymersomes nano-structures. The results revealed that the 19 F MRI CAs with a worm-like nanostructure had the best cell-uptake behavior compared with the others. The T_2 relaxation times of all the 19 F MRI CAs were from 176 ms to 179 ms in D_2O . This work indicated the great potential of utilizing PISA to prepare 19 F MRI CAs and to adjust their morphologies. 46

The imaging performance of the newly synthesized ¹⁹F MRI CAs have been improved considerably in the last two decades. However, most of the high-performance ¹⁹F MRI CAs developed to date require many synthetic steps and tedious purification procedures in their preparation process, which prohibits most of the synthesized 19F MRI CAs from practical application in clinical use. Thus, exploring facile strategies to fabricate highperformance 19F MRI CAs with acceptable preparation procedures still requires attention and effort. Some research has revealed that to achieve high-resolution in vivo ¹⁹F MRI, a very high concentration of ¹⁹F MRI CAs (above 50 mM) is normally administrated,47-49 which indicates that prolonging the circulation time of the ¹⁹F MRI CAs in vessels *in vivo* and promoting their accumulation inside the imaging lesion in vivo are pivotal as well. Thus, fabricating high-performance 19F MRI CAs with a long T_2 relaxation time and long circulation time in vivo and with excellent biocompatibility by a simple, efficient, and economical strategy demands increasing attention from researchers.

Herein, we present a new type of 19F MRI hydrogel nanoparticle contrast agents (19F MRI HNCAs) synthesized by a surfactant-free emulsion polymerization of the commercial fluorinated monomers (Fig. 1). We hypothesize that the hydration state in hydrogel nanoparticles is able to render the fluorine atoms with enough mobility, allowing the T_2 relaxation times of the ¹⁹F MRI HNCAs to be over 10 ms. Our explorations demonstrated a T_2 relaxation time of the ¹⁹F MRI HNCAs of 25– 40 ms at concentrations ranging from 5-40 mg mL⁻¹ under a magnetic field strength of 9.4 T. They displayed excellent ¹⁹F NMR intensity in $D_2O/H_2O(1/9, v/v)$ and were able to image well in vitro. In addition, the ¹⁹F MRI HNCAs demonstrated excellent biocompatibility and great colloid stability in aqueous solution (PBS buffer) for more than one year at room temperature. Most essentially, we found that the 19F NMR intensity and imaging efficiency mainly depended on the mobility of fluorine atoms rather than the fluorine content inside the ¹⁹F MRI HNCAs. Increasing the fluorine content in the ¹⁹F MRI HNCA blindly without considering the mobility of the fluorine atoms resulted in an attenuation of the ¹⁹F NMR signal intensity and imaging capability. The present research results have great significance in the design and preparation of other advanced ¹⁹F MRI CAs.

TBA TFMA

Polymeric Nanoparticles

19F MRI HNCAs

19F NMR/MRI

Fig. 1 $\,$ Schematic illustration of the synthesis process and constitution of the 19 F MRI HNCAs

Experimental section

Materials

Paper

Methacryloyl chloride (MAC) (95%), 2,2,2-trifluoroethyl methacrylate (98%), tertiary-butyl acrylate (TBA) (99%), 2,2,2-trifluoroethyl methacrylate (TFMA) (98%), sodium vinylbenzenesulfonate (NaVBS) (90%), potassium persulfate (KPS) (99.9%), trifluoroacetic acid (TFAA) (99%), and n-hydroxysuccinimide (NHS) (95%) were purchased from Shanghai Acmec Biochemical. Bis-(2-hydroxyethyl) disulfide (BHDSH) (>90%) was a product of Shanghai Aladdin Biochemical Technology. Triethylamine (99.5%), methanol (≥99.9%), 3,6-diamino-9-[2-(methoxycarbonyl)phenyl]-xanthylium chloride (Rhodamine-123) (98%), and 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC·HCL) (≥97%) were products of Anhui Senrise Technology. Deuterium oxide (D2O) (99.9%) and chloroformd (CDCl₃) (99.8%) were purchased from Shanghai Macklin Biochemical Technology. Dichloromethane (DCM) ($\geq 98\%$), formic acid (FA) (≥88%), ethyl acetate (EA) (99.5%), and petroleum ether (PE) (≥99.9%) were purchased from Guangzhou Brand. Dimethyl sulfoxide (DMSO) (99.9%) was obtained from Beijing Innochem Technology. Tertiary-butyl acrylate and 2,2,2-trifluoroethyl methacrylate were passed through Al₂O₃ to remove the stabilizer before usage. Phosphotungstate negative staining solution (PNSS) (2%, pH 7.0) was purchased from Leagene Biotechnology. Dulbecco's modified eagle medium (DMEM) with 4.5 g L⁻¹ p-glucose, 1-glutamine, 110 mg mL⁻¹ sodium, fetal bovine serum (FBS), and antibiotic-antimycotic (AA) were purchased from Thermo Fisher Scientific. Phosphate-buffered saline (PBS) was purchased from Beijing Solarbio Science & Technology. 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) was purchased from Shanghai Aladdin Biochemical Technology. DAPI staining solution (DAPI) (98%) (2 μg mL⁻¹) was a product of Wuhan Servicebio Technology. Paraformaldehyde (PFA) was purchased from biosharp.

Synthesis of disulfide dimethacrylate (2,2'-DED)

2,2'-DED was synthesized according to the literature.⁵⁰ BHDSH (1.5425 g, 10 mmol) and triethylamine (4.1699 mL, 30 mmol)

were dissolved in DCM (30 mL) in a two-necked flask equipped with argon (Ar). After the mixture was cooled down by immersion in an ice bath, MAC (2.6135 g, 25 mmol) was added dropwise. The reaction continued under an Ar atmosphere at room temperature for 24 h. The reaction was extracted by DCM and washed with water and brine, dried, and concentrated under vacuum. The crude product was purified by silica gel column chromatography (EA: PE = 1:15) to give a lightly yellow colored liquid 1.8920 g (72% yield). ¹H NMR (CDCl₃, 400 MHz, ppm): 6.13 (2H, s, 2H), 5.59 (2H, q, 2H), 4.41 (4H, t, 4H), 2.98 (4H, t, 4H), 1.94 (6H, s, 6H). The NMR spectra was provided in the ESI (Fig. S6†).

19F NMR/MRI

Synthesis of ¹⁹F MRI HNCAs-1

The polymeric nanoparticles (PNs) were synthesized by a surfactant-free emulsion polymerization. NaVBS (40 mg) and KPS (20 mg) were dissolved in a 40 mL mixture of $\rm H_2O$ and methanol (90/10, v/v). TBA, TFMA, and 2,2'-DED were then added into above solution. Then, oxygen was removed by bubbling argon through the solution for 20 min. The polymerization was carried out under an argon atmosphere at 70 °C for 18 h. The produced PNs were washed three times with ethanol and finally with $\rm H_2O$. ¹⁹F MRI HNCAs-1 were finally prepared by hydrolyzing the *tert*-butyl groups of the PNs in a mixture of TFAA and FA (80/20, v/v). The product was finally dialyzed against $\rm ddH_2O$ for 48 h with changing the $\rm ddH_2O$ three times.

Synthesis of rhodamine-123 modified ¹⁹F MRI HNCAs-1 (¹⁹F MRI HNCAs-1 Rh-123)

 $^{19} F$ MRI HNCAs-1 (100 mg) were first dispersed in anhydrous DMSO (2 mL) in a two-necked flask under an argon atmosphere. Then, EDC·HCL (1 mg, 0.005216 mmol) and NHS (600 µg, 0.005214 mmol) were added into the above DMSO solution. The flask was immersed in an ice bath. After 30 min, rhodamine-123 (200 µg, 0.0005252 mmol) was added. The mixture was allowed to react at room temperature in the dark for 36 h. The rhodamine-123 modified $^{19} F$ MRI HNCAs-1 were washed three times with ethanol and finally with DMSO. The obtained $^{19} F$

MRI HNCAs-1 Rh-123 were dispersed in DMSO and stored in the refrigerator at -4 °C.

Transmission electron microscopy (TEM) analysis

TEM measurements were carried out using an HT7800 SEM instrument (Hitachi High-Tech Scientific Solutions, Japan). The operating pressure of electricity was set to 120 kV. The sample was stained by PNSS as a contrast agent before the measurements.

Dynamic light scattering (DLS) and zeta potential measurements

DLS and zeta potential measurements were carried out using a Zetasizer Nano-ZS90 instrument (Malvern Instruments, Malvern, U.K.) equipped with a 4.0 mV He–Ne laser operating at 633 nm and a detection angle of 173°. The measuring temperature was set as 25 °C. The number-weighted hydrodynamic diameter was obtained from analysis of the autocorrelation functions using the method of cumulants. Three measurements were made for each sample with 60 s equilibrium time before each measurement. The concentration of the objectives was 5 mg mL $^{-1}$ in PBS.

¹⁹F Nuclear magnetic resonance (¹⁹F NMR)

 19 F NMR experiments were carried out on a Bruker AVANCE III HD Ascend 400 MHz spectrometer using a 12 μ s pulse width, relaxation delay of 1 s, acquisition time of 0.72 s, and 32 scans at 25 °C. All the chemical shifts are given herein in ppm.

Spin-lattice relaxation times (T_1)

The T_1 times were measured on a Bruker AVANCE III HD Ascend 400 MHz spectrometer using the standard inversion-recovery pulse sequence at 25 °C. The samples were dissolved in a mixture of D₂O/PBS (10/90, v/v) with a certain concentration. For each measurement, the relaxation delay was 1 s and the number of scans was 8.

Spin-spin relaxation times (T_2)

The T_2 times of the ¹⁹F HNCAs were measured using the Carr-Purcell–Meiboom–Gill (CPMG) pulse sequence at 25 °C. The ¹⁹F MRI HNCAs were dispersed in a mixture of D_2O/PBS (10/90, v/v) with a certain concentration. The relaxation delay was 2 s, and the number of scans was 32. For each measurement, the echo times were set from 10 to 1200 s, and 14 points were collected. The decay in amplitude of the spin echo could be described by a single-exponential function, which allowed calculating the T_2 relaxation times.

¹⁹F MRI imaging

Images of phantoms containing the 19 F MRI HNCAs solutions were acquired on a Bruker BioSpec 94/20 USR MRI at 9.4 T. The 19 F MRI HNCAs solutions (concentrations from 0 mg mL $^{-1}$ to 40 mg mL $^{-1}$ in PBS) were loaded in 2 mL Schering bottles. 1 H MRI images were acquired using the RARE sequence (field of view = 4 cm, slice thickness = 20 mm, $T_{\rm E} = 8.5$ ms, $T_{\rm R} = 1000$

ms, number of averages = 32, FOV = $40 \times 40 \text{ mm}^2$, matrix = 32 \times 32). ¹⁹F MRI images were acquired using the FLASH sequence (field of view = 4 cm, slice thickness = 20 mm, $T_{\rm E}$ = 1.3 ms, $T_{\rm R}$ = 800 ms, number of averages = 32, FOV = 40 \times 40 mm², matrix = 32 \times 32).

Cytotoxicity studies

To quantify a potential impact on cell viability, the MTT assay was applied. Cells were cultured with ^{19}F MRI HNCAs (concentrations from 0 mg mL $^{-1}$ to 10 mg mL $^{-1}$) in a 96-well plate in a final volume of 200 μL for 24 h. The MTT solution (20 μL) was added into each well and incubated for 4 h. The liquid in each well of the plate was suction removed, and washed with PBS once. The DMSO (150 μL) was added into each well of the plate, and subsequently incubated on a shaker for 5 min. Finally, the absorbance of the cells was measured using a multifunctional microplate detector (BioTek Instrument, U.S.) at 490 nm.

Confocal microscopy analysis

Imaging was conducted to identify if the 19 F MRI HNCAs-1 were internalized by the cells. Cells were seeded at a density of 100 000 cells per dish onto a confocal dish. Next, 5 mg mL $^{-1}$ rhodamine-123 modified 19 F MRI HNCAs were added and incubated for 4 h. The cells were fixed for 15 min in 4% PFA, before being rinsed for 3 \times 5 min in PBS. Subsequently, DAPI was added to the fixed cells and incubated for 10 min, before being rinsed for 3 \times 5 min in PBS. Images were acquired using an Olympus FV3000 laser confocal microscope and were analyzed using XV Viewer software.

Biosafety

The biosafety of the 19F MRI HNCAs-1 was assessed through hematoxylin and eosin (H&E) staining and biochemical index analysis at 5 days after the intravenous injection of the 19 F MRI HNCAs-1 (100 μL, 15 mg mL⁻¹) and PBS (blank) and intratumoral injection of the ¹⁹F MRI HNCAs-1 (50 μL, 15 mg mL⁻¹) and PBS (blank) in to BALB/c mice. The mice were subsequently killed and the major organs, including heart, liver, spleen, lung, and kidney, were collected, and fixed in 4% PFA solution and embedded in paraffin. Embedded tissue specimens were sectioned and photographed by a microscope. Also, 500 µL blood samples from the mice were collected in 1.5 mL centrifuge tubes, followed by centrifugation at 3000 rpm for 10 min at 4 °C, and the upper serum was collected for biochemical analysis. The serum biochemistry tests included aspartate transaminase (AST), albumin (ALB), total protein (TP), urea (UREA), creatinine (CRE), and cholesterol (CHO). All the experiments concerning the mice samples were approved by the Institutional Animal Care and Use Committee, Sun Yat-sen University (Approval No.: SYSU-IACUC-2020-B1108). We also state that informed consent was obtained for any experimentation with human subjects.

¹⁹F MRI signal detection inside 4T1 cells

After incubating ¹⁹F MRI HNCAs-1 (concentration: 10 mg mL⁻¹ in DMEM) with 4T1 cells at 37 °C for 24 h, the 4T1 cells (\sim 6 ×

106) were trypsinized and collected into a 1.5 mL centrifuge tube. Subsequently, the 4T1 cells were centrifuged and washed three times by pure PBS solution. The washed cells were resuspended in 300 µL clean DMEM and then subjected to five freeze-thawing cycles. Finally, the cells were centrifuged and the supernatants were collected for ¹⁹F NMR measurement. The cultured DMEM pure solution (the co-culture medium) and the last PBS solution that was used to wash the 4T1 cells were characterized by 19F NMR.

Results and discussion

¹⁹F MRI HNCAs synthesis and characterization

The detailed synthesis procedure and constitution of the ¹⁹F MRI HNCAs are presented in Fig. 1 and the Experimental section. The surfactant-free emulsion co-polymerization of tertiary-butyl acrylate (TBA) and trifluoroethyl methacrylate (TFMA) was applied to synthesize the ¹⁹F MRI HNCAs in this work, because of its simplicity, scale-up potential, and potential for diameter manipulation. TFMA with magnetically equivalent fluorine atoms was chosen as the fluorine marker. By turning the feed ratio of TBA and TFMA during the polymerization, the fluorine content inside the 19F MRI HNCAs and the 19F NMR intensity can be easily regulated in principle. During the polymerization, 1% (mole ratio to total monomers) crosslinking agent ((2,2'-dithiodiethanol) diacrylate (2,2'-DED)) (Fig. S6†) was

added to maintain the morphology of the 19F MRI HNCAs in aqueous solution. In order to verify the validity of this synthesis protocol, 19F MRI HNCAs-1 with a monomer mole ratio of 9:1 (TBA: TFMA = 9:1) were first synthesized. Around 30 g polymeric nanoparticles (PNs) (the precursor of the ¹⁹F MRI HNCAs-1) could be prepared by this approach every time.

Fig. 2 presents the dynamic light scattering (DLS) results, showing the PNs-1 with a diameter of 96 \pm 26 nm (Fig. 2A, black line). The zeta potential value of PNs-1 was -37.7 ± 0.3 mV (Fig. 2B, black column) indicating the sufficient negative charge on the surface and their good dispersity in aqueous solution. The TEM image demonstrated the PNs-1 had a spherical structure (Fig. 2C). In order to synthesize the ¹⁹F MRI HNCAs-1, the tert-butyl groups inside the PNs-1 were deprotected in the presence of trifluoroacetic acid (TFAA) and formic acid (FA). According to the DLS results (Fig. 2A, red line), the 19F MRI HNCAs-1 had a diameter of 230 \pm 73 nm. The increase in the diameter was caused by the swelling of the hydrophilic poly(acrylic acid)s in the ¹⁹F MRI HNCAs-1. Inside ¹⁹F MRI HNCAs-1, the hydrophilic poly(acrylic acid)s were able to enhance the hydration effect for the nearby fluorine atoms. The zeta potential value of the 19 F MRI HNCAs-1 was -22.4 ± 0.4 mV, as shown in Fig. 2B (red column), demonstrating that the negative charges (SO₃²⁻) were not located on surface of the ¹⁹F MRI HNCAs-1, which could be caused by the increased flexibility of the polymer chains and the migration of the negative charges

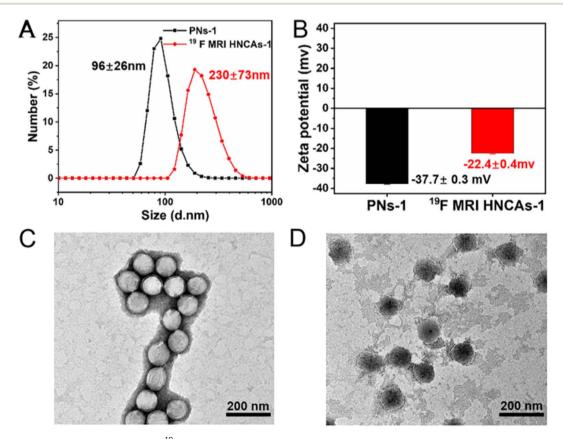


Fig. 2 Structure characterization of PNs-1 and ¹⁹F MRI HNCAs-1. (A) Hydrodynamic diameter values measured by DLS, (B) zeta potential values in H₂O. (C) TEM image of the PNs-1 and (D) TEM image of the ¹⁹F MRI HNCAs-1.

(SO₃²⁻) into the core of the ¹⁹F MRI HNCAs-1. The TEM image in Fig. 2D presented the round morphology of the ¹⁹F MRI HNCAs-1, clearly demonstrating that using 1% crosslinking agent (2,2′-DED) was enough to maintain the integrity of the ¹⁹F MRI HNCAs-1 during the hydrolysis reaction. This is important as nanosized objectives are apt to aggregate in aqueous solution to reduce their total surface energy, especially in the presence of ions. The ¹⁹F MRI HNCAs-1 could maintain their stable hydrodynamic diameter values and PDI for at least 30 days in PBS buffer at room temperature, as shown in Fig. S1,† demonstrating that the ¹⁹F MRI HNCAs-1 could be stored for a long time.

Magnetic resonance properties of the ¹⁹F MRI HNCAs

As discussed in the introduction, the mobility of fluorine atoms is restricted by the hydrophobic interaction among fluorine atoms and tert-butyl groups and dipole-dipole interactions among different fluorine atoms in PNs. We hypothesized that the mobility of fluorine atoms in PNs-1 could be optimized by deprotecting the hydrophobic tert-butyl groups, and accordingly, the ¹⁹F NMR intensity of the ¹⁹F MRI HNCAs-1 could be promoted, because the left hydrophilic carboxylic groups could enhance the hydration condition around the fluorine atoms. As shown in Fig. 3A, the 19F NMR intensity of the PNs-1 at a chemical shift of -73 ppm in D₂O/H₂O (1/9, v/v) was negligible, while the 19 F NMR intensity (chemical shift of -73 ppm) increased remarkably after removing the tert-butyl groups for ¹⁹F MRI HNCAs-1 in D₂O/H₂O (1/9, v/v) at the same concentration of nanoobjects (10 mg mL⁻¹). This encouraging result verified our above hypothesis directly. Compared with the ¹H MRI technology, one of the greatest advantages of ¹⁹F MRI is that the signal intensity is proportional to the fluorine content in ¹⁹F MRI CAs. As shown in Fig. 3B and S2,† the signal-to-noise ratio (SNR) was linearly dependent on the concentration of ¹⁹F MRI HNCAs-1 in aqueous solution $(2.5-40 \text{ mg mL}^{-1})$, indicating the 19F MRI HNCAs-1 are suitable for quantitative MRI measurements. Because the carboxylic acid groups can undergo protonation and deprotonation when the pH values of the

solution changes, this process may change the hydration environment of fluorine atoms and influence the imaging property of ¹⁹F MRI HNCAs-1. Accordingly, ¹⁹F NMR measurements were also carried out at varying pH values from 3.0 to 11.0. As shown in Fig. S3,† the ¹⁹F signal of ¹⁹F MRI HNCAs-1 maintained a consistent intensity at this pH range, demonstrating the imaging performance of ¹⁹F MRI HNCAs-1 was not influenced by the pH of the surroundings.

Beside the mobility of fluorine atoms, the fluorine content in ¹⁹F MRI CAs also influence the ¹⁹F NMR intensity and the ¹⁹F MRI imaging behavior.44 Thus, by adjusting the feed mole ratio of monomer TBA and TFMA in the polymerization process, another four 19F MRI HNCAs samples with different fluorine contents were synthesized by applying the same procedure and conditions as well. Compared with the ¹⁹F MRI HNCAs-1, the fluorine contents of the newly synthesized 19F MRI HNCAs were enhanced with increasing the feed ratio of TFMA in the polymerization. The detailed monomer ratios and the newly synthesized 19F MRI HNCAs characteristics are presented in Table S1.† The diameters of the newly synthesized ¹⁹F MRI HNCAs were between 180 nm and 250 nm. As shown in Fig. 4A, before deprotecting the tert-butyl groups, the ¹⁹F NMR intensities of the newly synthesized 19F MRI HNCAs did not appear at a chemical shift of -72.9 ppm, while the 19 F NMR intensities appeared again at a chemical shift of -72.9 ppm after the tertbutyl groups were removed, which was consistent with the result of the ¹⁹F MRI HNCAs-1. In principle, ¹⁹F MRI CAs with a higher fluorine content should display a higher ¹⁹F NMR intensity. However, for our 19F MRI HNCAs, 19F MRI HNCAs-1, which had the lowest fluorine content, displayed the strongest ¹⁹F NMR intensity at the same fluorine concentration of 0.7 mg mL⁻¹ and aqueous medium volume of 0.5 mL during the ¹⁹F NMR intensity measurements (Fig. 4A). This could be attributed to the fluorine atoms easily suffering from self-aggregation due to their strong dipole-dipole interactions and super hydrophobicity in aqueous solution. Compared with the other four newly prepared ¹⁹F MRI HNCAs, the low fluorine content inside ¹⁹F MRI HNCAs-1 resulted in the least self-aggregation and the

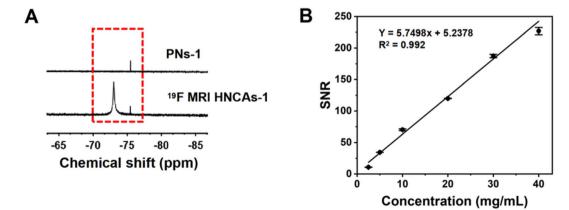


Fig. 3 19 F NMR characterization of 19 F MRI HNCAs-1 in D₂O/H₂O (1/9, v/v). (A) 19 F NMR intensity of the PNs-1 and 19 F MRI HNCAs-1 with trifluoroacetate sodium as an internal reference (nano-objectives concentration: 10 mg mL $^{-1}$) and (B) SNR values of the 19 F MRI HNCAs-1 at different concentrations (2.5–40 mg mL $^{-1}$).

Paper

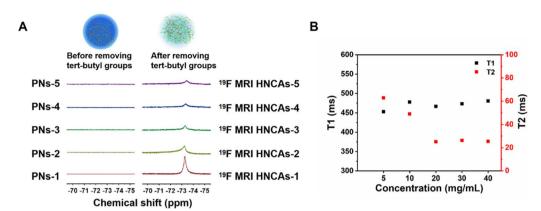


Fig. 4 (A) 19 F NMR intensity of the synthesized PNs and 19 F MRI HNCAs with a fluorine concentration of 0.7 mg mL $^{-1}$ in D₂O/H₂O (1/9, v/v) and (B) T_1 relaxation times and T_2 relaxation times of the ¹⁹F MRI HNCAs-1 at concentrations of 5–40 mg mL⁻¹ in D₂O/H₂O (1/9, v/v).

greatest mobility of fluorine atoms, which finally caused it to display the strongest ¹⁹F NMR intensity.

For our research in this article, because ¹⁹F MRI HNCAs-1 had the best ¹⁹F NMR performance in aqueous solution compared with the others, the 19F MRI HNCAs-1 were chosen for further study. It is well known that calcium ions existing in the cell nutrition medium and bodily fluids in vivo can complex with carboxylic acid to form supramolecular crosslinking points,51 which can slow down the chain segments movement of poly(acrylate acid)s. Because abundant carboxylic acids are located inside the ¹⁹F MRI HNCAs-1, we speculated that the calcium ions probably are able to change the fluorine atoms nearby microstructure and influence the 19F NMR intensity of the ¹⁹F MRI HNCAs-1. As shown in Fig. S4,† the ¹⁹F NMR intensity of the 19F MRI HNCAs-1 in aqueous solution containing calcium ions with a concentration of 6 mg mL⁻¹ was almost the same with that in a calcium ions-free aqueous solution, which demonstrated that the formation of supramolecular crosslinking points between carboxylic acids and calcium ions surrounding fluorine atoms could not prohibit the mobilities of the fluorine atoms and decrease the ¹⁹F NMR intensity of the ¹⁹F MRI HNCAs-1.

In aqueous solution, a longer T_2 relaxation time of ¹⁹F MRI CAs is essential for generating high-resolution ¹⁹F MRI images in vitro and in vivo. Normally, an ideal ¹⁹F MRI CA should have a T_2 relaxation time longer than 10 ms for achieving excellent performance in in vivo imaging.²⁷ Therefore, the T_1 and T_2 relaxation times of the 19F MRI HNCAs-1 were measured using the inversion-recovery and Carr-Purcell-Meiboom-Gill (CPMG) pulse sequences, respectively. The T_2 relaxation time of ¹⁹F MRI HNCAs-1 at concentrations from 5 mg mL⁻¹ to 10 mg mL⁻¹ were all over 40 ms without drastic change, indicating that the fluorine atoms maintained their constant mobility in this concentration range, as shown in Fig. 4B. The T_2 relaxation time longer than 10 ms for 19F MRI HNCAs-1 was attributed to the good mobility of fluorine atoms inside the hydration microenvironment caused by the nearby hydrophilic carboxyl groups. However, the T_2 relaxation time decreased to around 25 ms when the concentration of the 19F MRI HNCAs-1 increased to

20 mg mL $^{-1}$, while the T_2 relaxation times of around 25 ms were stably maintained in the concentration range of 20-40 mg mL^{-1} . Obviously, ¹⁹F MRI HNCAs-1 had T_2 relaxation times longer than 10 ms in the concentration range of 5-40 mg mL⁻¹, demonstrating the 19 F MRI HNCAs-1 have great potential for use in imaging. The T_1 relaxation times of ¹⁹F MRI HNCAs-1 in the concentration range of 5-40 mg mL⁻¹ were maintained as 450-500 ms without any obvious concentration dependency.

¹⁹F MRI imaging behavior of the ¹⁹F MRI HNCAs-1 in vitro

As discussed above, the ¹⁹F MRI HNCAs-1 displayed an excellent 19 F NMR intensity and long T_2 relaxation time in aqueous solution as shown in Fig. 4, thus the 19F MRI HNCAs-1 should be able to perform well in ¹⁹F MRI *in vitro* in principle. As shown in Fig. 5, hot-spot images of the ¹⁹F MRI HNCAs-1 aqueous solution at concentrations of 5-40 mg mL⁻¹ were successfully obtained. Moreover, the brightness of the hot-spot images was linearly dependent on the concentration of the ¹⁹F MRI HNCAs-1. In comparison, the aqueous solution of the PNs-1 (before deprotecting the tert groups) did not show an ¹⁹F MRI intensity in the same concentration range and measuring condition. The in vitro hot-spot imaging results were agreement with the ¹⁹F NMR outcome in Fig. 3A, showing that removing the hydrophobic tert-butyl groups was able to endow the ¹⁹F MRI HNCAs-1 with a strong fluorine atoms mobility and ¹⁹F NMR intensity.

Cell-uptake behavior and biocompatibility of the 19F MRI **HNCAs-1**

The successful imaging behaviors of the ¹⁹F MRI HNCAs-1 aqueous solution in vitro encouraged us to explore their application potential at the cellular level. The cell-uptake behavior of the ¹⁹F MRI HNCAs-1 was first investigated by flow cytometry in both red and green channels. In order to trace the 19F MRI HNCAs-1 by flow cytometry, the ¹⁹F MRI HNCAs-1 were first modified by rhodamine-123 with EDC·HCl and NHS. As shown in Fig. S5,† the fluorescence emission at 530 nm in the fluorescence spectrum demonstrated the successful modification of rhodamine-123 on the ¹⁹F MRI HNCAs-1. As shown in Fig. 6A, at 1 h post incubation of the 19F MRI HNCAs-1 with 4T1 cells

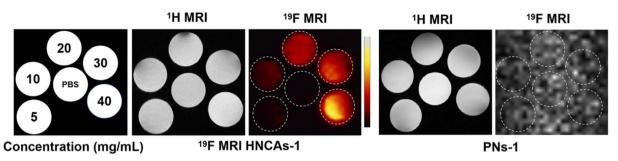


Fig. 5 ¹H/¹⁹F MRI performances of the ¹⁹F MRI HNCAs-1 and PNs-1 aqueous solution at concentrations of 5–40 mg mL⁻¹ in vitro.

(breast cancer cell), the fluorescence histograms demonstrated that the ¹⁹F MRI HNCAs-1 had been endocytosed by 4T1 already. The fast uptake behavior of the ¹⁹F MRI HNCAs-1 by 4T1 cells could probably be attributed to the extreme hydrophilicity of the carboxylic acid groups located on the surface of the ¹⁹F MRI HNCAs-1. The uptake amount of the ¹⁹F MRI HNCAs-1 by 4T1 cells reached a maximum at 4 h post incubation. After that, the amount of ¹⁹F MRI HNCAs-1 inside the 4T1 cells started to decrease due to the metabolism process inside the 4T1 cells. The mean fluorescence intensities of the 4T1 cell exploration outcomes were in agreement with the above fluorescence histograms results, as shown in Fig. 6B. In addition, in order to confirm the cell-uptake behavior, CLSM was used to visualize the presence of rhodamine-123-modified ¹⁹F MRI HNCAs-1

inside the 4T1 cells. As shown in Fig. 6C, the Z-slices through the cellular equator presented obvious blue and green fluorescence signals, largely within the cytoplasm of the 4T1 cancer cells, demonstrating the successful cellular uptake at 4 h post incubation of the $^{19}{\rm F}$ MRI HNCAs-1 with 4T1 cells. The successful uptake behavior demonstrated that the $^{19}{\rm F}$ MRI HNCAs-1 hold potential in imaging species and metabolic processes inside cells.

As practicable ¹⁹F MRI CAs, the biocompatibility and toxicity of the ¹⁹F MRI HNCAs-1 for cells and mice are determining factors. The cytotoxicity of the ¹⁹F MRI HNCAs-1 was first investigated against 4T1 cells and HUVEC (human umbilical vein endothelial cells) by 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assays. As shown in

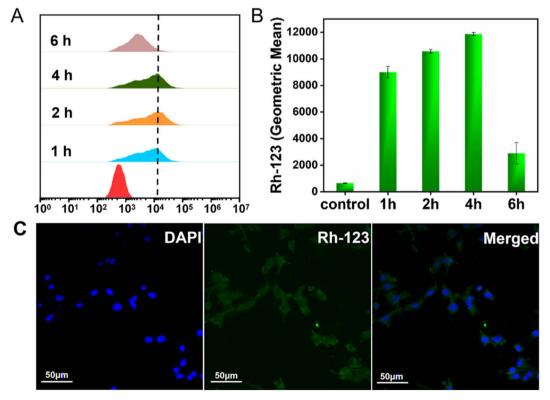


Fig. 6 19 F MRI HNCAs-1 cell-uptake-behavior characterization. (A). Fluorescence histograms of the 4T1 cells after incubation with 19 F MRI HNCAs-1 at incubation times of 1–6 h, (B) mean fluorescence intensities of the 4T1 cells after incubation with 19 F MRI HNCAs-1 at incubation times of 1–6 h, (C) CLSM images of the uptake behavior of 19 F MRI HNCAs-1 inside 4T1 cells.

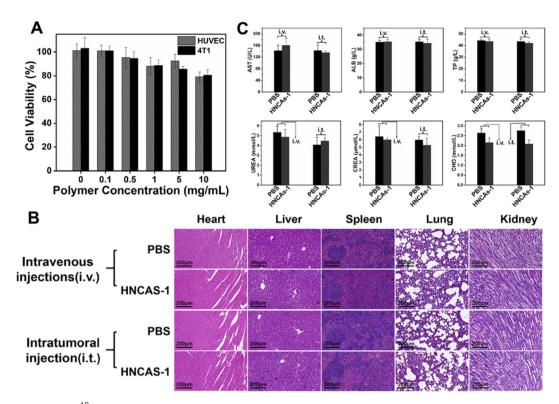


Fig. 7 Biosafety results of the ¹⁹F MRI HNCAs-1. (A) Cells viabilities against 4T1 and HUVEC cells, (B) H&E staining images of the main organs of mice at 5 days post administrating the ¹⁹F MRI HNCAs-1, (C) physiology indices values of the mice at 5 days post administrating the ¹⁹F MRI HNCAs-1.

Fig. 7A, the 19 F MRI HNCAs-1 at concentrations of 0.1 to 10 mg mL $^{-1}$ showed cell viability ratios above 80% after 24 h incubation, demonstrating the 19 F MRI HNCAs-1 had excellent

biocompatibility for both cancer and mammalian cells. In order to systematically access the biocompatibility and safety in mice, 100 μ L 19 F MRI HNCAs-1 with a concentration of 15 mg mL $^{-1}$

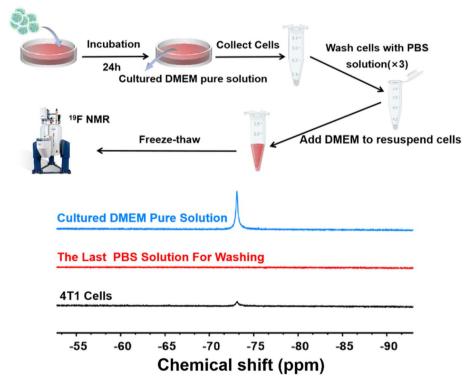


Fig. 8 ¹⁹F NMR intensity of ¹⁹F MRI HNCAs-1 within 4T1 cells.

was administrated into mice *via* lateral tail vein injection and intratumoral injection. At 5 days post-injection of the ¹⁹F MRI HNCAs-1, the important organs (heart, liver, spleen, lung, and kidney) of the mice were harvested and subjected to histological examination. The H&E staining images of the main organs indicated that they maintained their typical physiological structures without any tissue damage, as shown in Fig. 7B. In addition, the main physiology indices of blood, including albumin (ALB), aspartate transaminase (AST), creatinine (CRE), total protein (TP), cholesterol (CHOL), and urea (UREA), were at similar levels to the PBS control group, both maintaining a healthy level, as shown in Fig. 7C. All the above exploration results demonstrated that the ¹⁹F MRI HNCAs-1 possess satisfactory biosafety.

¹⁹F MRI capability of the ¹⁹F MRI HNCAs-1 inside 4T1 cells

The ¹⁹F MRI HNCAs-1 displayed excellent ¹⁹F NMR intensity, ¹⁹F MRI performance in vitro, highly efficient cell-uptake behavior, and excellent biosafety against cells and mice. As a result, we hypothesized that the ¹⁹F MRI HNCAs-1 would be able to image inside cells. To test this hypothesis, the19 F MRI HNCAs-1 were co-cultured with 4T1 cells for 24 h. The detailed purification processes are shown in Fig. 8 and described in the Experimental section. As shown in Fig. 8, the cultured DMEM pure solution showed ¹⁹F NMR intensity (blue line) demonstrating that not all the ¹⁹F MRI HNCAs-1 could be phagocytosed by 4T1 cells during the co-culture. In order to confirm that the ¹⁹F MRI HNCAs-1 could image inside 4T1 cells after being phagocytosed, the cocultured 4T1 cells were carefully washed with PBS three times to wash away any free ¹⁹F MRI HNCAs-1 outside the 4T1 cells. The 19 F NMR signal at a chemical shift of -73 ppm of the PBS washing solution (the third-time of washing solution) was not detectable (red line), demonstrating the free 19 F MRI HNCAs-1 outside the 4T1 cells had been fully washed away. After the freeze-thawing processes, the remaining 4T1 cells were transferred and characterized by ¹⁹F NMR in DMEM. The appearance of the ¹⁹F NMR signal at a chemical shift of -73 ppm (black line) indicated that the ¹⁹F MRI HNCAs-1 inside 4T1 cells still maintained their 19F MRI capability. The above results demonstrated again that the 19F MRI HNCAs-1 could be endocytosed by 4T1 cells. More importantly, the ¹⁹F NMR intensity of the endocytosed ¹⁹F HNCAs-1 inside 4T1 was still retained.

Conclusion

We developed a simple and high-performance surfactant-free emulsion polymerization strategy to synthesize ¹⁹F MRI CAs utilizing commercial fluorinated monomers. The research results demonstrated that donating a good hydration microenvironment surrounding the fluorine atoms could promote the mobility of the fluorine atoms and enhance the ¹⁹F MR imaging performance of the ¹⁹F MRI CAs. However, an intensive ¹⁹F NMR intensity could not be achieved by just increasing the amount of fluorinated monomer during the synthesis process or the fluorine content in the ¹⁹F MRI HNCAs. The ¹⁹F MRI HNCAs-1 displayed long-term storage stability, robust

functional stability, great biosafety *in vivo*, and excellent ¹⁹F MR imaging capability *in vitro*. For that reason, we believe that the ¹⁹F MRI HNCAs-1 could serve as a powerful ¹⁹F MRI contrast agent for not only the precise imaging of lesion sites (such as tumor, inflammation) but also could be used for cell tracking *in vivo* non-invasively. Additionally, the strategy used to prepare the ¹⁹F MRI CAs in this article has a certain referential significance for the preparation of new types of ¹⁹F MRI CAs.

Conflicts of interest

The authors declare they have no competing financial interests.

Acknowledgements

Dalin Wu acknowledges the financial support of the National Natural Science Foundation of China (22075327) and the Natural Science Foundation of Guangdong Province (2022A1515011392). In addition, Zheng Hu was acknowledged for her constructive suggestion on running the cell experiments.

References

- 1 P. Debbage and W. Jaschke, *Histochem. Cell Biol.*, 2008, **130**, 845–875.
- 2 E. T. Ahrens and J. W. M. Bulte, *Nat. Rev. Immunol.*, 2013, **13**, 755–763.
- 3 J. Wahsner, E. M. Gale, A. Rodríguez-Rodríguez and P. Caravan, *Chem. Rev.*, 2019, **119**, 957–1057.
- 4 D. Hao, T. Ai, F. Goerner, X. Hu, V. M. Runge and M. Tweedle, *J. Magn. Reson. Imaging*, 2012, **36**, 1060–1071.
- 5 J. L. Major and T. J. Meade, *Acc. Chem. Res.*, 2009, **42**, 893-903.
- 6 P. Caravan, J. J. Ellison, T. J. McMurry and R. B. Lauffer, Chem. Rev., 1999, 99, 2293–2352.
- 7 H. B. Na, I. C. Song and T. Hyeon, Adv. Mater., 2009, 21, 2133–2148.
- 8 M.-F. Bellin, Eur. J. Radiol., 2006, 60, 314-323.
- 9 J. T. Heverhagen, G. A. Krombach and E. Gizewski, *Rofo*, 2014, **186**, 661–669.
- 10 P. S. N. Rowe, L. V. Zelenchuk, J. S. Laurence, P. Lee, W. M. Brooks and E. T. McCarthy, *Am. J. Physiol.*, 2015, 309, F764–F769.
- 11 J. C. Knight, P. G. Edwards and S. J. Paisey, *RSC Adv.*, 2011, 1, 1415–1425.
- 12 E. T. Ahrens, B. M. Helfer, C. F. O'Hanlon and C. Schirda, Magn. Reson. Med., 2014, 72, 1696–1701.
- 13 A. Mali, E. L. Kaijzel, H. J. Lamb and L. J. Cruz, J. Controlled Release, 2021, 338, 870–889.
- 14 E. T. Ahrens and J. Zhong, NMR Biomed., 2013, 26, 860–871.
- 15 E. T. Ahrens, R. Flores, H. Xu and P. A. Morel, *Nat. Biotechnol.*, 2005, 23, 983–987.
- 16 M. Srinivas, P. A. Morel, L. A. Ernst, D. H. Laidlaw and E. T. Ahrens, Magn. Reson. Med., 2007, 58, 725–734.
- 17 J.-X. Yu, R. R. Hallac, S. Chiguru and R. P. Mason, *Prog. Nucl. Magn. Reson. Spectrosc.*, 2013, **70**, 25–49.

- 18 G. D. Kenny, K. P. Shaw, S. Sivachelvam, A. J. P. White, R. M. Botnar and R. T. M. de Rosales, J. Fluorine Chem., 2016, 184, 58-64.
- 19 J. Zhang, Y. Yuan, Y. Li, H. Yang, H. Zhang, S. Chen, X. Zhou, Z. Yang and Z.-X. Jiang, J. Org. Chem., 2020, 85, 6778-6787.
- 20 Z.-X. Jiang and Y. B. Yu, J. Org. Chem., 2010, 75, 2044-2049.
- 21 M. Ogawa, S. Nitahara, H. Aoki, S. Ito, M. Narazaki and T. Matsuda, Macromol. Chem. Phys., 2010, 211, 1602-1609.
- 22 J. Ruiz-Cabello, B. P. Barnett, P. A. Bottomley and J. W. M. Bulte, NMR Biomed., 2011, 24, 114-129.
- 23 I. Tirotta, V. Dichiarante, C. Pigliacelli, G. Cavallo, G. Terraneo, F. B. Bombelli, P. Metrangolo and G. Resnati, Chem. Rev., 2015, 115, 1106-1129.
- 24 R. Holman, O. Lorton, P. C. Guillemin, S. Desgranges, C. Contino-Pépin and R. Salomir, Front. Chem., 2022, 9, 810029.
- 25 I. N. Kuznetsova, Pharmaceutical Chemistry Journal, 2003, 37, 415-420.
- 26 Y. Huang, A. M. Vezeridis, J. Wang, Z. Wang, M. Thompson, R. F. Mattrey and N. C. Gianneschi, J. Am. Chem. Soc., 2017, 139, 15-18.
- 27 D. Jirak, A. Galisova, K. Kolouchova, D. Babuka and M. Hruby, Magn. Reson. Mater. Phys., Biol. Med., 2019, 32,
- 28 Y. Wang, X. Tan, A. Usman, Y. Zhang, M. Sawczyk, P. Král, C. Zhang and A. K. Whittaker, ACS Macro Lett., 2022, 11, 1195-1201.
- 29 K. J. Thurecht, I. Blakey, H. Peng, O. Squires, S. Hsu, C. Alexander and A. K. Whittaker, J. Am. Chem. Soc., 2010, 132, 5336-5337.
- 30 B. E. Rolfe, I. Blakey, O. Squires, H. Peng, N. R. B. Boase, C. Alexander, P. G. Parsons, G. M. Boyle, A. K. Whittaker and K. J. Thurecht, J. Am. Chem. Soc., 2014, 136, 2413-2419.
- 31 Z. Guo, M. Gao, M. Song, Y. Li, D. Zhang, D. Xu, L. You, L. Wang, R. Zhuang, X. Su, T. Liu, J. Du and X. Zhang, Adv. Mater., 2016, 28, 5898-5906.
- 32 Y. Yuan, S. Ge, H. Sun, X. Dong, H. Zhao, L. An, J. Zhang, J. Wang, B. Hu and G. Liang, ACS Nano, 2015, 9, 5117-5124.
- 33 H. Lin, X. Tang, A. Li and J. Gao, Adv. Mater., 2021, 33, 2005657.

- 34 C. Fu, B. Demir, S. Alcantara, V. Kumar, F. Han, H. G. Kelly, X. Tan, Y. Yu, W. Xu, J. Zhao, C. Zhang, H. Peng, C. Boyer, T. M. Woodruff, S. J. Kent, D. J. Searles and A. K. Whittaker, Angew. Chem., Int. Ed., 2020, 59, 4729-4735.
- 35 X. Zhu, X. Tang, H. Lin, S. Shi, H. Xiong, Q. Zhou, A. Li, Q. Wang, X. Chen and J. Gao, Chem, 2020, 6, 1134-1148.
- 36 K. L. Peterson, K. Srivastava and V. C. Pierre, Front. Chem., 2018, 6, 160-181.
- 37 C. Zhang, Y. Zhou, Q. Liu, S. Li, S. Perrier and Y. Zhao, Macromolecules, 2011, 44, 2034-2049.
- 38 K. Wang, H. Peng, K. J. Thurecht, S. Puttick and A. K. Whittaker, Polym. Chem., 2014, 5, 1760-1771.
- 39 K. Wang, H. Peng, K. J. Thurecht, S. Puttick and A. K. Whittaker, Biomacromolecules, 2015, 16, 2827-2839.
- 40 S. Li, J. Han and C. Gao, Polym. Chem., 2013, 4, 1774-1787.
- 41 O. Munkhbat, M. Canakci, S. Zheng, W. Hu, B. Osborne, A. A. Bogdanov and S. Thayumanavan, Biomacromolecules, 2019, 20, 790-800.
- 42 N. J. Warren and S. P. Armes, J. Am. Chem. Soc., 2014, 136, 10174-10185.
- 43 X. Tang, X. Gong, A. Li, H. Lin, C. Peng, X. Zhang, X. Chen and J. Gao, Nano Lett., 2020, 20, 363-371.
- 44 Y. Mo, C. Huang, C. Liu, Z. Duan, J. Liu and D. Wu, Macromol. Rapid Commun., 2023, 2200744.
- 45 D. Janasik and T. Krawczyk, Chem. Eur. J., 2022, 28, e202102556.
- 46 W. Zhao, H. T. Ta, C. Zhang and A. K. Whittaker, Biomacromolecules, 2017, 18, 1145-1156.
- 47 C. Zhang, S. S. Moonshi, Y. Han, S. Puttick, H. Peng, B. J. A. Magoling, J. C. Reid, S. Bernardi, D. J. Searles, P. Král and A. K. Whittaker, Macromolecules, 2017, 50, 5953-5963.
- 48 C. Zhang, S. S. Moonshi, W. Wang, H. T. Ta, Y. Han, F. Y. Han, H. Peng, P. Král, B. E. Rolfe, J. J. Gooding, K. Gaus and A. K. Whittaker, ACS Nano, 2018, 12, 9162-9176.
- 49 X. Tang, A. Li, C. Zuo, X. Liu, X. Luo, L. Chen, L. Li, H. Lin and J. Gao, ACS Nano, 2023, 17, 5014-5024.
- 50 C. Miao, F. Li, Y. Zuo, R. Wang and Y. Xiong, RSC Adv., 2016, 6, 3013-3019.
- 51 M. B. Gindele, K. K. Malaszuk, C. Peter and D. Gebauer, Langmuir, 2022, 38, 14409-14421.