## **RSC Advances**



## **PAPER**

View Article Online
View Journal | View Issue



Cite this: RSC Adv., 2023, 13, 13511

# Stereoselective synthesis of (E)- $\alpha$ , $\beta$ -unsaturated esters: triethylamine-catalyzed allylic rearrangement of enol phosphates†

Yulong Zhang, Huichuang Guo, Qian Wu, Xiaojing Bi, \* Enxue Shi \* and Junhua Xiao \* \*

 $\alpha,\beta$ -Unsaturated esters are key structural motifs widely distributed in various biologically active molecules, and their Z/E-stereoselective synthesis has always been considered highly attractive in organic synthesis. Herein, we present a >99% (E)-stereoselective one-pot synthetic approach towards  $\beta$ -phosphoroxylated  $\alpha,\beta$ -unsaturated esters via a mild trimethylamine-catalyzed 1,3-hydrogen migration of the corresponding unconjugated intermediates derived from the solvent-free Perkow reaction between low-cost 4-chloroacetoacetates and phosphites. Versatile  $\beta,\beta$ -disubstituted (E)- $\alpha,\beta$ -unsaturated esters were thus afforded with full (E)-stereoretentivity by cleavage of the phosphoenol linkage via Negishi crosscoupling. Moreover, a stereoretentive (E)-rich mixture of a  $\alpha,\beta$ -unsaturated ester derived from 2-chloroacetoacetate was obtained and both isomers were easily afforded in one operation.

Received 12th April 2023 Accepted 27th April 2023

DOI: 10.1039/d3ra02430j

rsc.li/rsc-advances

 $\alpha$ ,β-Unsaturated carbonyl motifs, such as the relevant esters, amides, and aldehydes, are widely distributed in biologically active molecules as key structural components (Fig. 1).<sup>1-4</sup> Generally, the (Z) and (E)-isomers of those molecules possess very different living activities.<sup>5</sup> Moreover, ubiquitous  $\alpha$ ,β-unsaturated esters are also widely employed as useful intermediates for enantioselective hydrogenation,<sup>6</sup> allylic substitution,<sup>7</sup> conjugate addition,<sup>8</sup> and especially for the stereoselective generation of acyclic substituted alkenes in either (Z) or (E)-isomeric forms.<sup>9</sup>

Whilst numerous methods have been developed towards  $\alpha, \beta$ -unsaturated esters, <sup>10-13</sup> configuration-retentive transition-metal catalyzed (TMC) cross-coupling of alkenyl (pseudo)halides is universally recognized as one of the most practical methodologies. <sup>14</sup> Among the known non-classical pseudohalides, <sup>15</sup> diethylphosphoroxyl (DEP) functionality has been proved as a good leaving group in many organic reactions and the corresponding enol phosphates (EPs), possessing high stability and accessibility, were found to participate in various organic transformations. <sup>16</sup> Particularly, EPs have been utilized in many types of TMC coupling reactions including Suzuki–Miyaura, Stille, Negishi, and Heck reactions by cleavage of the enollinkage affording highly substituted alkenes. <sup>17</sup> However, the EPs-involved (Z) and (E)-stereocomplementary synthetic method towards  $\alpha,\beta$ -unsaturated esters with sufficient substrate

generality is still quite limited at present. The latest impressive approach was reported by Tanabe group, which employed Nmethylimidazole (NMI)-promoted phosphorylation of βketoesters to obtain (Z) and (E)- $\alpha$ , $\beta$ -unsaturated esters, but which suffers from pre-activation of the unstable diphenyl phosphorochloridate (DPPCl) and usage of strong metallic tertbutoxide bases.18 Based on our recent progress in regioselective solvent-free synthesis of EPs, 19 we envisioned that phosphoroxylated (Z) and/or (E)- $\alpha$ ,  $\beta$ -unsaturated esters may act as the universal synthon of α,β-unsaturated esters and should be facilely obtained from the commercially available and low-cost chloroacetoacetates and phosphites via a simple metal-free Perkow reaction. Herein, we wish to present a stereoselective one-pot synthetic approach towards β-phosphoroxylated (E)- $\alpha$ , $\beta$ unsaturated esters, which are subsequently converted into the corresponding disubstituted α,β-unsaturated esters by Negishi cross-coupling (Scheme 1).

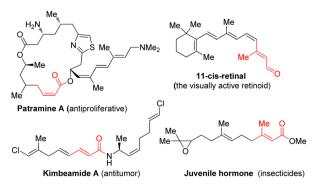


Fig. 1 Selected bioactive  $\alpha, \beta$ -unsaturated carbonyl motifs.

State Key Laboratory of NBC Protection for Civilian, Beijing 102205, P. R. China. E-mail: junhua@pku.edu.cn; exshi@sina.com; xiaojingbimail@yeah.net

† Electronic supplementary information (ESI) available: Full experimental details and analytical data. CCDC 2250165. For ESI and crystallographic data in CIF or other electronic format see DOI: https://doi.org/10.1039/d3ra02430j

#### Tanabe' work

Scheme 1 E-Stereoselective synthesis of  $\alpha$ ,  $\beta$ -unsaturated esters from enol phosphates.

Scheme 2 Perkow reaction of phosphite with chloroacetoacetate.

Since both 2-chloroacetoacetates and 4-chloroacetoacetates are capable of undergoing Perkow reaction with phosphites, we then took them together for comparison. Solvent-free Perkow reaction conditions were initially selected in view of high regioselectivity.<sup>19</sup> As shown in Scheme 2, reaction between  $(EtO)_3P$  and 2-chloroacetoacetate **2a** gave a mixture of (E) and (Z)-isomers of β-phosphoroxylated  $\alpha$ ,β-unsaturated ester **4a** in ratio of 2.6:1, whereas reaction between  $(EtO)_3P$  and 4-chloroacetoacetate **3a** gave the β-phosphoroxylated allylic ester **5a** as the only product. In other words, only moderate E/Z-stereoselectivity can be achieved if using 2-chloroacetoacetate, while no conjugated EP product can be obtained if using 4-chloroacetoacetate. However, according to Seeman's report that bases, such as NaH, are supposed to be able to promote 1,3-hydrogen relocation of allyl compounds, we then suspect that the unconjugated EP product **5a** may be able to be transformed into the conjugated one in a stereoselective way.<sup>20</sup>

Inspired by the above idea, we then turned to examine the possibility of the base-promoted 1,3-hydrogen rearrangement of **5a**. As shown in Table 1, among the eight kinds of bases examined, including inorganic *t*-BuOK, CH<sub>3</sub>ONa, NaOH, NaH,

Table 1 Optimization of base-promoted 1,3-hydrogen rearrangement of unconjugated  $\beta$ -phosphoroxylated allylic ester 5a  $^a$ 

Entry	Base	Load ( $x$ eq.)	Solvent	$T$ ( $^{\mathrm{o}}$ C)	Time (h)	$Yield^{b}$ (%)	$E/Z (4a)^c$
1	t-BuOK	1.2	THF	rt	24	0	_
2	CH <sub>3</sub> ONa	1.2	THF	rt	24	0	_
3	NaOH	1.2	THF	rt	24	0	_
4	NaH	1.2	THF	rt	24	0	_
5	$K_2CO_3$	1.2	THF	rt	24	0	_
6	$\mathrm{Et_{3}N}$	1.2	THF	rt	24	90	>99:1
7	Pyridine	1.2	THF	rt	24	0	_
8	(i-Pr) <sub>2</sub> NEt	1.2	THF	rt	24	20	>99:1
9	$Et_3N$	1.2	$CH_3CN$	rt	4	92	>99:1
10	$\mathrm{Et_{3}N}$	1.2	DCM	rt	20	90	>99:1
11	$\mathrm{Et_{3}N}$	1.2	$CH_3OH$	rt	22	83	>99:1
12	$\mathrm{Et_{3}N}$	1.2	DMF	rt	24	75	>99:1
13	$\mathrm{Et_{3}N}$	0.5	$CH_3CN$	rt	7	92	>99:1
14	$Et_3N$	0.1	CH <sub>3</sub> CN	rt	12	92	>99:1
15	$\mathrm{Et_{3}N}$	0.05	$CH_3CN$	rt	20	93	>99:1
16	$\mathrm{Et_{3}N}$	0.1	$CH_3CN$	0	24	95	>99:1
17	$Et_3N$	0.1	$CH_3CN$	80	4	90	>99:1

<sup>&</sup>lt;sup>a</sup> Reaction conditions: 5a (1.0 equiv.), base (x equiv.), solvent (3 ml). <sup>b</sup> Isolated yields. <sup>c</sup> Determined by NMR.

K<sub>2</sub>CO<sub>3</sub>, and organic Et<sub>3</sub>N, Pyridine (i-Pr)<sub>2</sub>NEt, only Et<sub>3</sub>N and (i-Pr)<sub>2</sub>NEt exhibited the supposed promoting abilities, affording the desired product E-4a, but encouragely both in >99% (E)stereoselectivity. Though only 20% yield was obtained by 1.2 equivalent (i-Pr)2NEt after 24 h reaction in THF at room temperature (Table 1, entry 8), while up to 90% yield was acquired by using Et<sub>3</sub>N (Table 1, entry 6). The following screening of solvents demonstrated that acetonitrile seemed to the best choice that the reaction could be accomplished in only 4 h and gave a higher yield of 92% (Table 1, entry 10). Further investigation about the loadage of Et<sub>3</sub>N showed that only 0.1 equivalent Et<sub>3</sub>N was sufficient to promote the rearrangement effectively, affording the comparative yield though with a few longer time of 12 h (Table 1, entry 14). Less loadage of Et<sub>3</sub>N and lower temperature both led to much longer reaction times (Table 1, entry 15&16). Though the reaction time could be shortened to 4 h at a higher temperature of 80 °C (Table 1, entry 17), we finally preferred the more benign room temperature for

Considering the convenience of experimental operation, we then turned into the possibility of one-pot manipulation. It was found that product  $\emph{E-4a}$  was afforded in 92% yield if using the crude intermediate 5a directly for the subsequent rearrangement reaction. Therefore, a mild  $\emph{E-}$ -stereoselective one-pot synthetic approach of  $\beta$ -phosphoroxylated  $\alpha,\beta$ -unsaturated esters was thus established: 3 (1.0 eq.) and P(III)-reagents (1.0 eq.) react 1 h at 40 °C neatly, then added triethylamine (0.1 eq.) and acetonitrile (3 mL), and further react about 12 h at room temperature.

the following preparations.

Having identified the optimal reaction conditions, we next set out to examine the scope of this new mild one-pot enol phosphorylation procedure (Table 2). As for the different *O*-alkyl

**Table 2** Scope of β-phosphoroxylated (E)- $\alpha$ , β-unsaturated esters a,b

4-chloroacetoacetate substrates, all the common P(III)-reagents possessing P–O, P–C, and/or P–N bonds gave the corresponding EPs in high yields. During the preparation of compounds **4e** and **4f**, the rearrangement reactions were found much accelerated probably due to the higher reactivities of phosphonite and phosphinite compared to phosphites. To demonstrate the practical utility, the reaction towards product **4a** was performed at the 50 mmol scale and 92% yield was obtained. The stereoscopic (*E*)-configuration of solid product **4f** was further confirmed by single crystal X-ray analysis.

With the *E*-stereospecific  $\beta$ -phosphoroxylated  $\alpha$ , $\beta$ -unsaturated esters in hand, we then investigated their stereoretentive Negishi cross-coupling to prepare the corresponding *E*-stereodefined disubstituted  $\alpha$ , $\beta$ -unsaturated esters. Among the typical catalysts screened including Pd(PPh<sub>3</sub>)<sub>4</sub>, Ni(acac)<sub>2</sub> and Pd(dppb) Cl<sub>2</sub>, the latter demonstrated the best performance in this Negishi reaction with only 0.02 equivalent loading by refluxing in acetonitrile. Various aromatic ArZnCl nucleophiles containing electron-donating and/or electron-withdrawing substituents at *ortho*, *meta*, and/or *para* positions were all tolerated well, affording the desired products in good to excellent yields (80–

**Table 3** Scope of (*E*)- $\alpha$ ,  $\beta$ -unsaturated esters *via* a stereoretentive Negishi cross-coupling reaction of **4a**  $^{a,b}$ 

 $<sup>^</sup>a$  Reaction conditions: 1 (1.0 mmol), 3 (1.0 mmol), Et<sub>3</sub>N (0.1 mmol), CH<sub>3</sub>CN (3.0 mL).  $^b$  Isolated yields.

 $<sup>^</sup>a$  Reaction conditions: 4a (1.0 mmol), ArMgBr (1.5 mmol), ZnCl $_2$  (1.5 mmol), Pd(dppb)Cl $_2$  (0.02 mmol), CH $_3$ CN (5.0 mL), reflux about 3 h.  $^b$  Isolated yields.

96%) without generating any stereochemical integrity (Table 3, 6a–6m). Disubstituted, condensed and hetero aromatic organometallic substrates also gave 85–92% yields of the products (Table 3, 6n–6r). However, it's regrettable that alkyl organozinc reagents was found unreactive under such conditions.

Furthermore, under the above optimal Negishi cross-coupling reaction conditions, both (Z) and (E) isomers of  $\alpha$ , $\beta$ -unsaturated esters **6a** could be easily achieved, just by one operation, directly from the (Z) and (E) mixture of **4a** in 22% and 70% yields respectively (Scheme 3).

According to the Cram's mechanistic interpretation for the allylic rearrangements, an intra-molecular pathway of the Et<sub>3</sub>N-promoted stereoselective 1,3-hydrogen rearrangement of the EPs **5a** was proposed because that the degree of the observed intramolecularity depended strongly on the base and solvent used. As shown in Scheme 4, triethylamine firstly removes the proton from the  $\alpha$ -carbon position of ester **5a**, resulting in a coplanar anionic allylic system by three carbon atoms. The hydrogen atom of the H–Et<sub>3</sub>N ammonium then bonds to both terminal carbon atoms to form the intermediate **Int**, collapse of which would then give the thermodynamically favourable conjugated  $\alpha$ ,  $\beta$ -unsaturated ester product *E*-**4a**.

In summary, a mild and environmental trimethylamine-catalyzed E-stereoselective 1,3-hydrogen allylic rearrangement of enol phosphates was firstly developed to afford versatile  $\beta$ -phosphoroxylated (E)- $\alpha$ , $\beta$ -unsaturated esters which can be then efficiently converted into the corresponding  $\beta$ , $\beta$ -disubstituted (E)- $\alpha$ , $\beta$ -unsaturated esters in high yields by a 100% stereoretentive Negishi cross-coupling reaction. Moreover, both (Z) and (E)- $\alpha$ , $\beta$ -unsaturated esters were able to be achieved in one manipulation when just employing 2-chloroacetoacetate instead of 4-chloroacetoacetate for the solvent and metal-free Perkow reaction.

It is interesting to note that more structure-diverse  $\alpha,\beta$ -unsaturated esters should be easily obtained by derivation reactions at the allylic position of  $\alpha,\beta$ -unsaturated esters and/or by utilizing 2-substituted 4-chloroacetoacetates as the starting materials.

Scheme 3 Preparation of (Z) and (E) isomers of 6a in one operation.

**Scheme 4** Proposed (*E*)-stereospecific allylic rearrangement mechanism.

## Conflicts of interest

There are no conflicts to declare.

### Notes and references

- (a) C. X. Zhuo and A. Furstner, J. Am. Chem. Soc., 2018, 140, 10514; (b) D. Romo, N. S. Choi, S. Li, I. Buchler, Z. Shi and J. O. Liu, J. Am. Chem. Soc., 2004, 126, 10582; (c) S. D. Marco, A. Cammas, J. Pelletier and I. E. Gallouzi, Nat. Commun., 2012, 3, 8963; (d) S. K. Naineni, J. Liang, B. Nagar and J. Pelletier, Cell Chem. Biol., 2021, 28, 825.
- (a) J. K. Nunnery, N. Engene, T. Byrum, T. F. Murray and W. H. Gerwick, *J. Org. Chem.*, 2012, 77, 4198; (b)
   S. Sekharan and K. Morokuma, *J. Am. Chem. Soc.*, 2011, 133, 19052; (c) E. A. Zhukovsky, P. R. Robinson and D. D. Oprian, *Science*, 1990, 251, 558.
- 3 M. Kiser and K. Golczak, Chem. Rev., 2014, 114, 194.
- 4 (a) P. Guo, Y. Zhang, L. Zhang and Q. Xia, J. Biol. Chem., 2021, 297, 101234; (b) Y. Ando, K. Matsumoto, K. Misaki, G. Mano, T. Shinada and S. G. Goto, Gen. Comp. Endocrinol., 2020, 289, 113394; (c) M. Nouzova, C. Rivera-Pérez and F. G. Noriega, Curr. Opin. Insect. Sci., 2018, 29, 49; (d) K. Li, Q. Jia and S. Li, Insect Sci., 2019, 26, 600.
- 5 (a) D. A. Evans, P. J. Coleman, L. C. Dias and A. N. Tyler, Angew. Chem., Int. Ed. Engl., 1997, 36, 2744; (b) D. A. Evans, D. M. Fitch, T. E. Smith and V. Cee, J. Am. Chem. Soc., 2000, 122, 10033; (c) D. A. Evans, P. H. Carter, E. M. Carreira, A. B. Charette, J. A. Prunet and M. Lautens, J. Am. Chem. Soc., 1999, 121, 7540; (d) I. Fleming, A. Barbero and D. Walter, Chem. Rev., 1997, 97, 2063.
- 6 (a) E. Negishi, Q. Hu, Z. Huang, M. Qian and G. Wang, Aldrichimica Acta, 2005, 38, 71; (b) J. Li, A. S. Grillo and M. D. Burke, Acc. Chem. Res., 2015, 48, 2297; (c) V. Hornillos, M. Giannerini, C. Vila, M. F. Mastral and B. L. Feringa, Chem. Sci., 2015, 6, 1394.
- 7 (a) K. Murakami and H. Yorimitsu, Beilstein J. Org. Chem.,
  2013, 9, 278; (b) M. G. Suero, E. D. Bayle, B. S. L. Collins and
  M. J. Gaunt, J. Am. Chem. Soc., 2013, 135, 5332; (c) F. Xue,
  J. Zhao, T. S. A. Hor and T. Hayashi, J. Am. Chem. Soc., 2015,
  137, 3189; (d) S. Wang and C. Xi, Org. Lett., 2018, 20, 4131.
- 8 (a) F. Guibe, *Tetrahedron*, 1998, **54**, 2967; (b) F. Guibe, *Tetrahedron*, 1997, **53**, 13509; (c) S. Escoubet, S. Gastaldi and M. Bertrand, *Eur. J. Org. Chem.*, 2005, 3855.
- 9 (a) F. W. Sum and L. Weiler, *Can. J. Chem.*, 1979, 57, 1431; (b) M. Ide and M. Nakata, *Synlett*, 2001, 1511.
- 10 (a) A. B. Flynn and W. W. Ogilvie, Chem. Rev., 2007, 107, 4698; (b) P. Polak, H. Vanova, D. Dvorak and T. Tobrman, Tetrahedron Lett., 2016, 57, 3684; (c) B. E. Maryanoff and A. B. Reitz, Chem. Rev., 1989, 89, 863; (d) Y. Ashida and Y. Tanabe, Chem. Rec., 2020, 20, 1.
- 11 (a) C. Gürtler and S. L. Buckwald, *Chem. Eur. J.*, 1999, 5, 3107; (b) M. Shindo, Y. Sato, T. Yoshikawa, R. Koretsune and K. Shishido, *J. Org. Chem.*, 2004, **69**, 3912.
- 12 (a) A. B. Lemay, K. S. Vulic and W. W. Oglivie, *J. Org. Chem.*,
   2006, 71, 3615; (b) J. S. Mercier, A. B. Flynn and
   W. W. Ogilvie, *Tetrahedron*, 2008, 64, 5472.

Paper

(a) Y. Yamamoto, N. Kirai and Y. Harada, *Chem. Commun.*,
 2008, 2010; (b) N. Kirai and Y. Yamamoto, *Org. Synth.*,
 2010, 87, 53; (c) Z. He, S. Kirchberg, R. Froehlich and
 A. Studer, *Angew. Chem., Int. Ed.*, 2012, 124, 3759.

- 14 (a) J. M. Baxter, D. Steinhuebel, M. Palucki and I. W. Davies, *Org. Lett.*, 2005, 7, 215; (b) A. Klapars, K. R. Campos, C. Chen and R. P. Volante, *Org. Lett.*, 2005, 7, 1185; (c) H. Nakatsuji, K. Ueno, T. Misaki and Y. Tanabe, *Org. Lett.*, 2008, 10, 2131; (d) H. Nakatsuji, H. Nishikado, K. Ueno and Y. Tanabe, *Org. Lett.*, 2009, 11, 4258.
- 15 Y. Ashida, K. Nakata, D. Yoshitake, Y. Sato, Y. Miyazaki and Y. Tanabe, *Asian J. Org. Chem.*, 2020, **9**, 604.
- 16 (a) A. L. Hansen, J. P. Ebran, M. Ahlquist, P. O. Norrby and T. Skrydstrup, Angew. Chem., Int. Ed., 2006, 45, 3349; (b) T. Hayashi, T. Fujiwa, Y. Okamoto, Y. Katsuro and M. Kumada, Synthesis, 1981, 1001; (c) K. C. Nicolaou, G. Q. Shi, G. P. Grtner and Z. Yang, J. Am. Chem. Soc., 1997, 119, 5467; (d) A. S. E. Karlstrçm, K. Itami and J. E. Bckvall, J. Org. Chem., 1999, 64, 1745; (e) J. Wu and Z. Yang, J. Org. Chem., 2001, 66, 7875; (f) J. P. Ebran, A. L. Hansen, T. M. Gøgsig and T. Skrydstrup, J. Am. Chem.

- Soc., 2007, **129**, 6931; (g) A. L. Hansen, J.-P. Ebran, T. M. Gøgsig and T. Skrydstrup, J. Org. Chem., 2007, 72, 6464; (h) W. You, Y. Li and M. K. Brown, Org. Lett., 2013, **15**, 1610.
- 17 (a) T. M. Gøgsig, A. T. Lindhardt and T. Skrydstrup, Org. Lett.,
  2009, 11, 4886; (b) J. Jiang, R. DeVita, G. Doss, M. Goulet and
  M. Wyvratt, J. Am. Chem. Soc., 1999, 121, 593; (c)
  A. L. Hansen, J. P. Ebran, T. M. Gøgsig and T. Skrydstrup,
  Chem. Commun., 2006, 39, 4137; (d) J. W. Coe, Org. Lett.,
  2000, 2, 4205.
- 18 H. Nakatsuji, Y. Ashida, H. Hori, Y. Sato, M. Taira and Y. Tanabe, Org. Biomol. Chem., 2015, 13, 8205.
- 19 (a) Y. Cao, Z. Gao, J. Li, X. Bi, L. Yuan, C. Pei, Y. Guo and E. Shi, RSC Adv., 2020, 10, 29493; (b) H. Guo, Y. Zhang, Z. Li, P. Zhao, N. Li and E. Shi, RSC Adv., 2022, 12, 14844.
- 20 S. G. Alcock, J. E. Baldwin, R. Bohlmann, L. M. Harwood and J. Seeman, J. Org. Chem., 1985, 50, 3526.
- (a) M. Hassam, A. Taher, G. E. Arnott, I. R. Green and W. A. L. van Otterlo, *Chem. Rev.*, 2015, 115, 5462; (b)
   D. J. Cram and R. T. Uyeda, *J. Am. Chem. Soc.*, 1964, 86, 5466.