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Stereoselective synthesis of (*E*)- α,β -unsaturated esters: triethylamine-catalyzed allylic rearrangement of enol phosphates†

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α,β -Unsaturated esters are key structural motifs widely distributed in various biologically active molecules, and their *Z/E*-stereoselective synthesis has always been considered highly attractive in organic synthesis. Herein, we present a >99% (*E*)-stereoselective one-pot synthetic approach towards β -phosphoroxylated α,β -unsaturated esters via a mild trimethylamine-catalyzed 1,3-hydrogen migration of the corresponding unconjugated intermediates derived from the solvent-free Perkow reaction between low-cost 4-chloroacetoacetates and phosphites. Versatile β,β -disubstituted (*E*)- α,β -unsaturated esters were thus afforded with full (*E*)-stereoretentivity by cleavage of the phosphoenol linkage via Negishi cross-coupling. Moreover, a stereoretentive (*E*)-rich mixture of a α,β -unsaturated ester derived from 2-chloroacetoacetate was obtained and both isomers were easily afforded in one operation.

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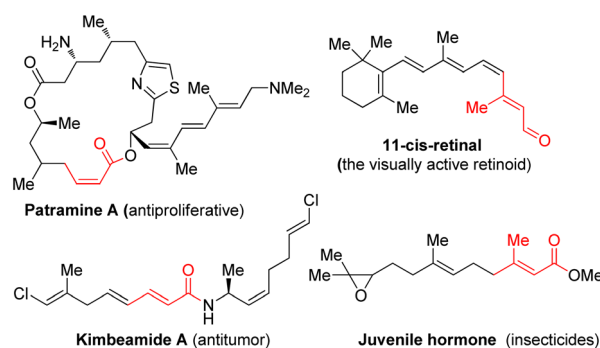
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α,β -Unsaturated carbonyl motifs, such as the relevant esters, amides, and aldehydes, are widely distributed in biologically active molecules as key structural components (Fig. 1).^{1–4} Generally, the (*Z*) and (*E*)-isomers of those molecules possess very different living activities.⁵ Moreover, ubiquitous α,β -unsaturated esters are also widely employed as useful intermediates for enantioselective hydrogenation,⁶ allylic substitution,⁷ conjugate addition,⁸ and especially for the stereoselective generation of acyclic substituted alkenes in either (*Z*) or (*E*)-isomeric forms.⁹

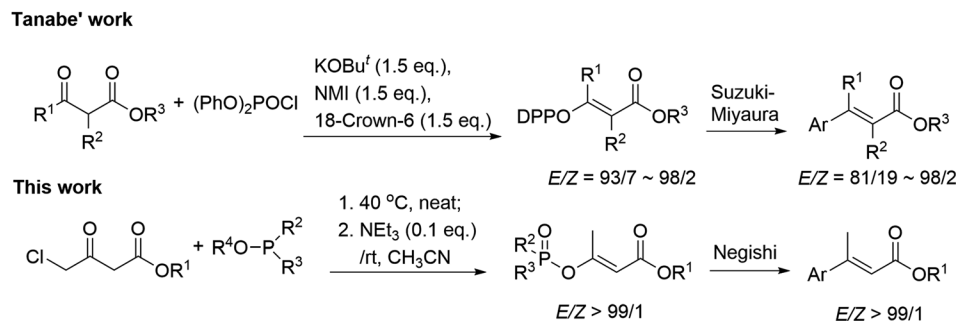
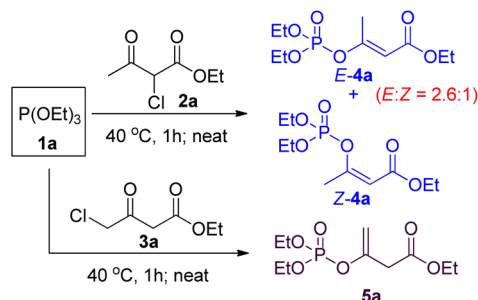
Whilst numerous methods have been developed towards α,β -unsaturated esters,^{10–13} configuration-retentive transition-metal catalyzed (TMC) cross-coupling of alkenyl (pseudo)halides is universally recognized as one of the most practical methodologies.¹⁴ Among the known non-classical pseudohalides,¹⁵ diethylphosphoroxyl (DEP) functionality has been proved as a good leaving group in many organic reactions and the corresponding enol phosphates (EPs), possessing high stability and accessibility, were found to participate in various organic transformations.¹⁶ Particularly, EPs have been utilized in many types of TMC coupling reactions including Suzuki–Miyaura, Stille, Negishi, and Heck reactions by cleavage of the enol-linkage affording highly substituted alkenes.¹⁷ However, the EPs-involved (*Z*) and (*E*)-stereocomplementary synthetic method towards α,β -unsaturated esters with sufficient substrate

generality is still quite limited at present. The latest impressive approach was reported by Tanabe group, which employed *N*-methylimidazole (NMI)-promoted phosphorylation of β -ketoesters to obtain (*Z*) and (*E*)- α,β -unsaturated esters, but which suffers from pre-activation of the unstable diphenyl phosphorochloridate (DPPCl) and usage of strong metallic *tert*-butoxide bases.¹⁸ Based on our recent progress in regioselective solvent-free synthesis of EPs,¹⁹ we envisioned that phosphoroxylated (*Z*) and/or (*E*)- α,β -unsaturated esters may act as the universal synthon of α,β -unsaturated esters and should be facilely obtained from the commercially available and low-cost chloroacetoacetates and phosphites via a simple metal-free Perkow reaction. Herein, we wish to present a stereoselective one-pot synthetic approach towards β -phosphoroxylated (*E*)- α,β -unsaturated esters, which are subsequently converted into the corresponding disubstituted α,β -unsaturated esters by Negishi cross-coupling (Scheme 1).


 Fig. 1 Selected bioactive α,β -unsaturated carbonyl motifs.

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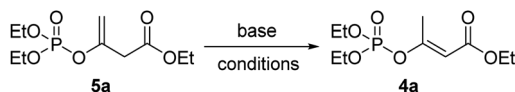

Scheme 1 *E*-Stereoselective synthesis of α,β -unsaturated esters from enol phosphates.

Scheme 2 Perkow reaction of phosphite with chloroacetoacetate.

Since both 2-chloroacetoacetates and 4-chloroacetoacetates are capable of undergoing Perkow reaction with phosphites, we then took them together for comparison. Solvent-free Perkow reaction conditions were initially selected in view of high

regioselectivity.¹⁹ As shown in Scheme 2, reaction between $(\text{EtO})_3\text{P}$ and 2-chloroacetoacetate **2a** gave a mixture of (*E*) and (*Z*)-isomers of β -phosphoroxylated α,β -unsaturated ester **4a** in ratio of 2.6 : 1, whereas reaction between $(\text{EtO})_3\text{P}$ and 4-chloroacetoacetate **3a** gave the β -phosphoroxylated allylic ester **5a** as the only product. In other words, only moderate *E/Z*-stereoselectivity can be achieved if using 2-chloroacetoacetate, while no conjugated EP product can be obtained if using 4-chloroacetoacetate. However, according to Seeman's report that bases, such as NaH, are supposed to be able to promote 1,3-hydrogen relocation of allyl compounds, we then suspect that the unconjugated EP product **5a** may be able to be transformed into the conjugated one in a stereoselective way.²⁰

Inspired by the above idea, we then turned to examine the possibility of the base-promoted 1,3-hydrogen rearrangement of **5a**. As shown in Table 1, among the eight kinds of bases examined, including inorganic *t*-BuOK, CH_3ONa , NaOH, NaH,

Table 1 Optimization of base-promoted 1,3-hydrogen rearrangement of unconjugated β -phosphoroxylated allylic ester **5a**^a

Entry	Base	Load (x eq.)	Solvent	<i>T</i> (°C)	Time (h)	Yield ^b (%)	<i>E/Z</i> (4a) ^c
1	<i>t</i> -BuOK	1.2	THF	rt	24	0	—
2	CH_3ONa	1.2	THF	rt	24	0	—
3	NaOH	1.2	THF	rt	24	0	—
4	NaH	1.2	THF	rt	24	0	—
5	K_2CO_3	1.2	THF	rt	24	0	—
6	Et_3N	1.2	THF	rt	24	90	>99 : 1
7	Pyridine	1.2	THF	rt	24	0	—
8	$(i\text{-Pr})_2\text{NEt}$	1.2	THF	rt	24	20	>99 : 1
9	Et_3N	1.2	CH_3CN	rt	4	92	>99 : 1
10	Et_3N	1.2	DCM	rt	20	90	>99 : 1
11	Et_3N	1.2	CH_3OH	rt	22	83	>99 : 1
12	Et_3N	1.2	DMF	rt	24	75	>99 : 1
13	Et_3N	0.5	CH_3CN	rt	7	92	>99 : 1
14	Et_3N	0.1	CH_3CN	rt	12	92	>99 : 1
15	Et_3N	0.05	CH_3CN	rt	20	93	>99 : 1
16	Et_3N	0.1	CH_3CN	0	24	95	>99 : 1
17	Et_3N	0.1	CH_3CN	80	4	90	>99 : 1

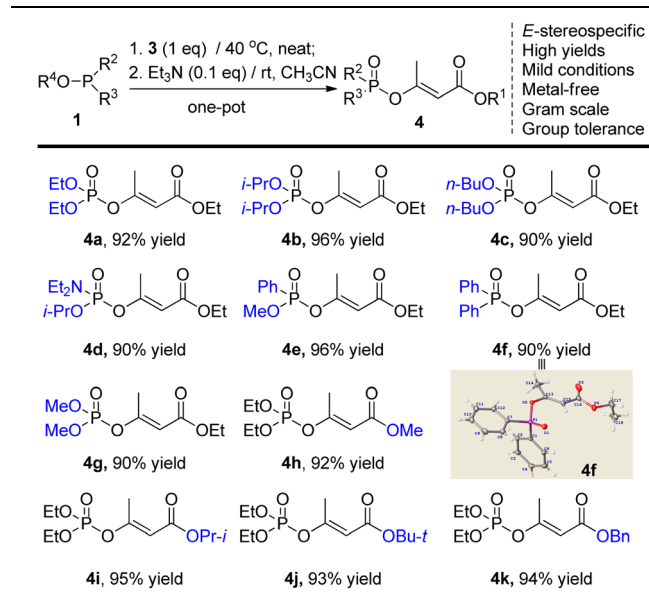
^a Reaction conditions: **5a** (1.0 equiv.), base (x equiv.), solvent (3 ml). ^b Isolated yields. ^c Determined by NMR.



K_2CO_3 , and organic Et_3N , Pyridine (i -Pr) $_2$ NEt, only Et_3N and (i -Pr) $_2$ NEt exhibited the supposed promoting abilities, affording the desired product **E-4a**, but encouragingly both in >99% (*E*)-stereoselectivity. Though only 20% yield was obtained by 1.2 equivalent (i -Pr) $_2$ NEt after 24 h reaction in THF at room temperature (Table 1, entry 8), while up to 90% yield was acquired by using Et_3N (Table 1, entry 6). The following screening of solvents demonstrated that acetonitrile seemed to be the best choice that the reaction could be accomplished in only 4 h and gave a higher yield of 92% (Table 1, entry 10). Further investigation about the dosage of Et_3N showed that only 0.1 equivalent Et_3N was sufficient to promote the rearrangement effectively, affording the comparative yield though with a few longer time of 12 h (Table 1, entry 14). Less dosage of Et_3N and lower temperature both led to much longer reaction times (Table 1, entry 15&16). Though the reaction time could be shortened to 4 h at a higher temperature of 80 °C (Table 1, entry 17), we finally preferred the more benign room temperature for the following preparations.

Considering the convenience of experimental operation, we then turned into the possibility of one-pot manipulation. It was found that product **E-4a** was afforded in 92% yield if using the crude intermediate **5a** directly for the subsequent rearrangement reaction. Therefore, a mild *E*-stereoselective one-pot synthetic approach of β -phosphoroxylated α,β -unsaturated esters was thus established: **3** (1.0 eq.) and P(III)-reagents (1.0 eq.) react 1 h at 40 °C neatly, then added triethylamine (0.1 eq.) and acetonitrile (3 mL), and further react about 12 h at room temperature.

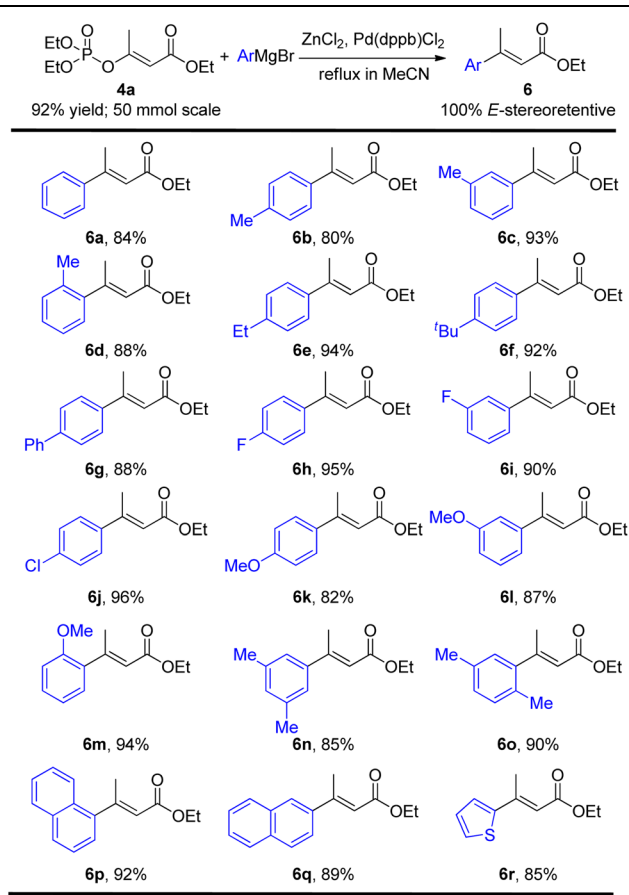
Having identified the optimal reaction conditions, we next set out to examine the scope of this new mild one-pot enol phosphorylation procedure (Table 2). As for the different *O*-alkyl

Table 2 Scope of β -phosphoroxylated (*E*)- α,β -unsaturated esters^{a,b}

^a Reaction conditions: **1** (1.0 mmol), **3** (1.0 mmol), Et_3N (0.1 mmol), CH_3CN (3.0 mL). ^b Isolated yields.

4-chloroacetoacetate substrates, all the common P(III)-reagents possessing P–O, P–C, and/or P–N bonds gave the corresponding EPs in high yields. During the preparation of compounds **4e** and **4f**, the rearrangement reactions were found much accelerated probably due to the higher reactivities of phosphonite and phosphinite compared to phosphites. To demonstrate the practical utility, the reaction towards product **4a** was performed at the 50 mmol scale and 92% yield was obtained. The stereoscopic (*E*)-configuration of solid product **4f** was further confirmed by single crystal X-ray analysis.

With the *E*-stereospecific β -phosphoroxylated α,β -unsaturated esters in hand, we then investigated their stereoretentive Negishi cross-coupling to prepare the corresponding *E*-stereo-defined disubstituted α,β -unsaturated esters. Among the typical catalysts screened including $Pd(PPh_3)_4$, $Ni(acac)_2$ and $Pd(dppb)Cl_2$, the latter demonstrated the best performance in this Negishi reaction with only 0.02 equivalent loading by refluxing in acetonitrile. Various aromatic $ArZnCl$ nucleophiles containing electron-donating and/or electron-withdrawing substituents at *ortho*, *meta*, and/or *para* positions were all tolerated well, affording the desired products in good to excellent yields (80–

Table 3 Scope of (*E*)- α,β -unsaturated esters via a stereoretentive Negishi cross-coupling reaction of **4a**^{a,b}

^a Reaction conditions: **4a** (1.0 mmol), $ArMgBr$ (1.5 mmol), $ZnCl_2$ (1.5 mmol), $Pd(dppb)Cl_2$ (0.02 mmol), CH_3CN (5.0 mL), reflux about 3 h. ^b Isolated yields.

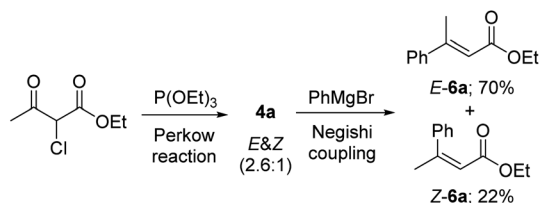
96%) without generating any stereochemical integrity (Table 3, **6a–6m**). Disubstituted, condensed and hetero aromatic organometallic substrates also gave 85–92% yields of the products (Table 3, **6n–6r**). However, it's regrettable that alkyl organozinc reagents was found unreactive under such conditions.

Furthermore, under the above optimal Negishi cross-coupling reaction conditions, both (*Z*) and (*E*) isomers of α,β -unsaturated esters **6a** could be easily achieved, just by one operation, directly from the (*Z*) and (*E*) mixture of **4a** in 22% and 70% yields respectively (Scheme 3).

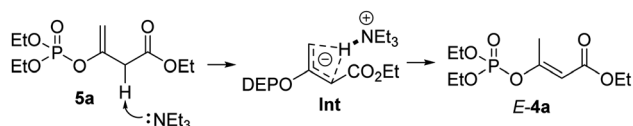
According to the Cram's mechanistic interpretation for the allylic rearrangements, an intra-molecular pathway of the Et_3N -promoted stereoselective 1,3-hydrogen rearrangement of the EPs **5a** was proposed because that the degree of the observed intramolecularity depended strongly on the base and solvent used.²¹ As shown in Scheme 4, triethylamine firstly removes the proton from the α -carbon position of ester **5a**, resulting in a coplanar anionic allylic system by three carbon atoms. The hydrogen atom of the $\text{H-Et}_3\text{N}$ ammonium then bonds to both terminal carbon atoms to form the intermediate **Int**, collapse of which would then give the thermodynamically favourable conjugated α,β -unsaturated ester product *E*-**4a**.

In summary, a mild and environmental trimethylamine-catalyzed *E*-stereoselective 1,3-hydrogen allylic rearrangement of enol phosphates was firstly developed to afford versatile β -phosphoroxylated (*E*)- α,β -unsaturated esters which can be then efficiently converted into the corresponding β,β -disubstituted (*E*)- α,β -unsaturated esters in high yields by a 100% stereoretentive Negishi cross-coupling reaction. Moreover, both (*Z*) and (*E*)- α,β -unsaturated esters were able to be achieved in one manipulation when just employing 2-chloroacetoacetate instead of 4-chloroacetoacetate for the solvent and metal-free Perkow reaction.

It is interesting to note that more structure-diverse α,β -unsaturated esters should be easily obtained by derivation reactions at the allylic position of α,β -unsaturated esters and/or by utilizing 2-substituted 4-chloroacetoacetates as the starting materials.



Scheme 3 Preparation of (*Z*) and (*E*) isomers of **6a** in one operation.



Scheme 4 Proposed (*E*)-stereospecific allylic rearrangement mechanism.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- (a) C. X. Zhuo and A. Furstner, *J. Am. Chem. Soc.*, 2018, **140**, 10514; (b) D. Romo, N. S. Choi, S. Li, I. Buchler, Z. Shi and J. O. Liu, *J. Am. Chem. Soc.*, 2004, **126**, 10582; (c) S. D. Marco, A. Cammas, J. Pelletier and I. E. Gallouzi, *Nat. Commun.*, 2012, **3**, 8963; (d) S. K. Naineni, J. Liang, B. Nagar and J. Pelletier, *Cell Chem. Biol.*, 2021, **28**, 825.
- (a) J. K. Nunnery, N. Engene, T. Byrum, T. F. Murray and W. H. Gerwick, *J. Org. Chem.*, 2012, **77**, 4198; (b) S. Sekharan and K. Morokuma, *J. Am. Chem. Soc.*, 2011, **133**, 19052; (c) E. A. Zhukovsky, P. R. Robinson and D. D. Oprian, *Science*, 1990, **251**, 558.
- M. Kiser and K. Golczak, *Chem. Rev.*, 2014, **114**, 194.
- (a) P. Guo, Y. Zhang, L. Zhang and Q. Xia, *J. Biol. Chem.*, 2021, **297**, 101234; (b) Y. Ando, K. Matsumoto, K. Misaki, G. Mano, T. Shinada and S. G. Goto, *Gen. Comp. Endocrinol.*, 2020, **289**, 113394; (c) M. Nouzova, C. Rivera-Pérez and F. G. Noriega, *Curr. Opin. Insect. Sci.*, 2018, **29**, 49; (d) K. Li, Q. Jia and S. Li, *Insect Sci.*, 2019, **26**, 600.
- (a) D. A. Evans, P. J. Coleman, L. C. Dias and A. N. Tyler, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 2744; (b) D. A. Evans, D. M. Fitch, T. E. Smith and V. Cee, *J. Am. Chem. Soc.*, 2000, **122**, 10033; (c) D. A. Evans, P. H. Carter, E. M. Carreira, A. B. Charette, J. A. Prunet and M. Lautens, *J. Am. Chem. Soc.*, 1999, **121**, 7540; (d) I. Fleming, A. Barbero and D. Walter, *Chem. Rev.*, 1997, **97**, 2063.
- (a) E. Negishi, Q. Hu, Z. Huang, M. Qian and G. Wang, *Aldrichimica Acta*, 2005, **38**, 71; (b) J. Li, A. S. Grillo and M. D. Burke, *Acc. Chem. Res.*, 2015, **48**, 2297; (c) V. Hornillos, M. Giannerini, C. Vila, M. F. Mastral and B. L. Feringa, *Chem. Sci.*, 2015, **6**, 1394.
- (a) K. Murakami and H. Yorimitsu, *Beilstein J. Org. Chem.*, 2013, **9**, 278; (b) M. G. Suero, E. D. Bayle, B. S. L. Collins and M. J. Gaunt, *J. Am. Chem. Soc.*, 2013, **135**, 5332; (c) F. Xue, J. Zhao, T. S. A. Hor and T. Hayashi, *J. Am. Chem. Soc.*, 2015, **137**, 3189; (d) S. Wang and C. Xi, *Org. Lett.*, 2018, **20**, 4131.
- (a) F. Guibe, *Tetrahedron*, 1998, **54**, 2967; (b) F. Guibe, *Tetrahedron*, 1997, **53**, 13509; (c) S. Escoubet, S. Gastaldi and M. Bertrand, *Eur. J. Org. Chem.*, 2005, 3855.
- (a) F. W. Sum and L. Weiler, *Can. J. Chem.*, 1979, **57**, 1431; (b) M. Ide and M. Nakata, *Synlett*, 2001, 1511.
- (a) A. B. Flynn and W. W. Ogilvie, *Chem. Rev.*, 2007, **107**, 4698; (b) P. Polak, H. Vanova, D. Dvorak and T. Tobrman, *Tetrahedron Lett.*, 2016, **57**, 3684; (c) B. E. Maryanoff and A. B. Reitz, *Chem. Rev.*, 1989, **89**, 863; (d) Y. Ashida and Y. Tanabe, *Chem. Rec.*, 2020, **20**, 1.
- (a) C. Gürtler and S. L. Buckwald, *Chem. - Eur. J.*, 1999, **5**, 3107; (b) M. Shindo, Y. Sato, T. Yoshikawa, R. Koretsune and K. Shishido, *J. Org. Chem.*, 2004, **69**, 3912.
- (a) A. B. Lemay, K. S. Vulic and W. W. Ogilvie, *J. Org. Chem.*, 2006, **71**, 3615; (b) J. S. Mercier, A. B. Flynn and W. W. Ogilvie, *Tetrahedron*, 2008, **64**, 5472.



- 13 (a) Y. Yamamoto, N. Kirai and Y. Harada, *Chem. Commun.*, 2008, 2010; (b) N. Kirai and Y. Yamamoto, *Org. Synth.*, 2010, **87**, 53; (c) Z. He, S. Kirchberg, R. Froehlich and A. Studer, *Angew. Chem., Int. Ed.*, 2012, **124**, 3759.
- 14 (a) J. M. Baxter, D. Steinhuebel, M. Palucki and I. W. Davies, *Org. Lett.*, 2005, **7**, 215; (b) A. Klapars, K. R. Campos, C. Chen and R. P. Volante, *Org. Lett.*, 2005, **7**, 1185; (c) H. Nakatsuji, K. Ueno, T. Misaki and Y. Tanabe, *Org. Lett.*, 2008, **10**, 2131; (d) H. Nakatsuji, H. Nishikado, K. Ueno and Y. Tanabe, *Org. Lett.*, 2009, **11**, 4258.
- 15 Y. Ashida, K. Nakata, D. Yoshitake, Y. Sato, Y. Miyazaki and Y. Tanabe, *Asian J. Org. Chem.*, 2020, **9**, 604.
- 16 (a) A. L. Hansen, J. P. Ebran, M. Ahlquist, P. O. Norrby and T. Skrydstrup, *Angew. Chem., Int. Ed.*, 2006, **45**, 3349; (b) T. Hayashi, T. Fujiwa, Y. Okamoto, Y. Katsuro and M. Kumada, *Synthesis*, 1981, 1001; (c) K. C. Nicolaou, G. Q. Shi, G. P. Grtner and Z. Yang, *J. Am. Chem. Soc.*, 1997, **119**, 5467; (d) A. S. E. Karlström, K. Itami and J. E. Bckvall, *J. Org. Chem.*, 1999, **64**, 1745; (e) J. Wu and Z. Yang, *J. Org. Chem.*, 2001, **66**, 7875; (f) J. P. Ebran, A. L. Hansen, T. M. Gøgsig and T. Skrydstrup, *J. Am. Chem. Soc.*, 2007, **129**, 6931; (g) A. L. Hansen, J.-P. Ebran, T. M. Gøgsig and T. Skrydstrup, *J. Org. Chem.*, 2007, **72**, 6464; (h) W. You, Y. Li and M. K. Brown, *Org. Lett.*, 2013, **15**, 1610.
- 17 (a) T. M. Gøgsig, A. T. Lindhardt and T. Skrydstrup, *Org. Lett.*, 2009, **11**, 4886; (b) J. Jiang, R. DeVita, G. Doss, M. Goulet and M. Wyvratt, *J. Am. Chem. Soc.*, 1999, **121**, 593; (c) A. L. Hansen, J. P. Ebran, T. M. Gøgsig and T. Skrydstrup, *Chem. Commun.*, 2006, **39**, 4137; (d) J. W. Coe, *Org. Lett.*, 2000, **2**, 4205.
- 18 H. Nakatsuji, Y. Ashida, H. Hori, Y. Sato, M. Taira and Y. Tanabe, *Org. Biomol. Chem.*, 2015, **13**, 8205.
- 19 (a) Y. Cao, Z. Gao, J. Li, X. Bi, L. Yuan, C. Pei, Y. Guo and E. Shi, *RSC Adv.*, 2020, **10**, 29493; (b) H. Guo, Y. Zhang, Z. Li, P. Zhao, N. Li and E. Shi, *RSC Adv.*, 2022, **12**, 14844.
- 20 S. G. Alcock, J. E. Baldwin, R. Bohlmann, L. M. Harwood and J. Seeman, *J. Org. Chem.*, 1985, **50**, 3526.
- 21 (a) M. Hassam, A. Taher, G. E. Arnott, I. R. Green and W. A. L. van Otterlo, *Chem. Rev.*, 2015, **115**, 5462; (b) D. J. Cram and R. T. Uyeda, *J. Am. Chem. Soc.*, 1964, **86**, 5466.

