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# Stereoselective synthesis of (E)- $\alpha$ , $\beta$ -unsaturated esters: triethylamine-catalyzed allylic rearrangement of enol phosphates†

Yulong Zhang, Huichuang Guo, Qian Wu, Xiaojing Bi, \* Enxue Shi \* and Junhua Xiao \* \*

 $\alpha,\beta$ -Unsaturated esters are key structural motifs widely distributed in various biologically active molecules, and their Z/E-stereoselective synthesis has always been considered highly attractive in organic synthesis. Herein, we present a >99% (E)-stereoselective one-pot synthetic approach towards  $\beta$ -phosphoroxylated  $\alpha,\beta$ -unsaturated esters via a mild trimethylamine-catalyzed 1,3-hydrogen migration of the corresponding unconjugated intermediates derived from the solvent-free Perkow reaction between low-cost 4-chloroacetoacetates and phosphites. Versatile  $\beta,\beta$ -disubstituted (E)- $\alpha,\beta$ -unsaturated esters were thus afforded with full (E)-stereoretentivity by cleavage of the phosphoenol linkage via Negishi crosscoupling. Moreover, a stereoretentive (E)-rich mixture of a  $\alpha,\beta$ -unsaturated ester derived from 2-chloroacetoacetate was obtained and both isomers were easily afforded in one operation.

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 $\alpha$ ,β-Unsaturated carbonyl motifs, such as the relevant esters, amides, and aldehydes, are widely distributed in biologically active molecules as key structural components (Fig. 1).<sup>1-4</sup> Generally, the (Z) and (E)-isomers of those molecules possess very different living activities.<sup>5</sup> Moreover, ubiquitous  $\alpha$ ,β-unsaturated esters are also widely employed as useful intermediates for enantioselective hydrogenation,<sup>6</sup> allylic substitution,<sup>7</sup> conjugate addition,<sup>8</sup> and especially for the stereoselective generation of acyclic substituted alkenes in either (Z) or (E)-isomeric forms.<sup>9</sup>

Whilst numerous methods have been developed towards  $\alpha, \beta$ -unsaturated esters, <sup>10-13</sup> configuration-retentive transition-metal catalyzed (TMC) cross-coupling of alkenyl (pseudo)halides is universally recognized as one of the most practical methodologies. <sup>14</sup> Among the known non-classical pseudohalides, <sup>15</sup> diethylphosphoroxyl (DEP) functionality has been proved as a good leaving group in many organic reactions and the corresponding enol phosphates (EPs), possessing high stability and accessibility, were found to participate in various organic transformations. <sup>16</sup> Particularly, EPs have been utilized in many types of TMC coupling reactions including Suzuki–Miyaura, Stille, Negishi, and Heck reactions by cleavage of the enollinkage affording highly substituted alkenes. <sup>17</sup> However, the EPs-involved (Z) and (E)-stereocomplementary synthetic method towards  $\alpha,\beta$ -unsaturated esters with sufficient substrate

generality is still quite limited at present. The latest impressive approach was reported by Tanabe group, which employed Nmethylimidazole (NMI)-promoted phosphorylation of βketoesters to obtain (Z) and (E)- $\alpha$ , $\beta$ -unsaturated esters, but which suffers from pre-activation of the unstable diphenyl phosphorochloridate (DPPCl) and usage of strong metallic tertbutoxide bases.18 Based on our recent progress in regioselective solvent-free synthesis of EPs, 19 we envisioned that phosphoroxylated (Z) and/or (E)- $\alpha$ , β-unsaturated esters may act as the universal synthon of α,β-unsaturated esters and should be facilely obtained from the commercially available and low-cost chloroacetoacetates and phosphites via a simple metal-free Perkow reaction. Herein, we wish to present a stereoselective one-pot synthetic approach towards β-phosphoroxylated (E)- $\alpha$ ,βunsaturated esters, which are subsequently converted into the corresponding disubstituted α,β-unsaturated esters by Negishi cross-coupling (Scheme 1).

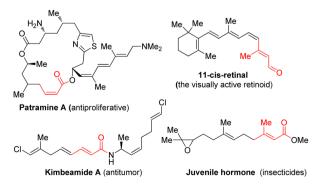


Fig. 1 Selected bioactive  $\alpha, \beta$ -unsaturated carbonyl motifs.

State Key Laboratory of NBC Protection for Civilian, Beijing 102205, P. R. China. E-mail: junhua@pku.edu.cn; exshi@sina.com; xiaojingbimail@yeah.net

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#### Tanabe' work

Scheme 1 E-Stereoselective synthesis of  $\alpha$ , $\beta$ -unsaturated esters from enol phosphates.

Scheme 2 Perkow reaction of phosphite with chloroacetoacetate.

Since both 2-chloroacetoacetates and 4-chloroacetoacetates are capable of undergoing Perkow reaction with phosphites, we then took them together for comparison. Solvent-free Perkow reaction conditions were initially selected in view of high regioselectivity.<sup>19</sup> As shown in Scheme 2, reaction between  $(EtO)_3P$  and 2-chloroacetoacetate **2a** gave a mixture of (E) and (Z)-isomers of β-phosphoroxylated  $\alpha$ , β-unsaturated ester **4a** in ratio of 2.6:1, whereas reaction between  $(EtO)_3P$  and 4-chloroacetoacetate **3a** gave the β-phosphoroxylated allylic ester **5a** as the only product. In other words, only moderate E/Z-stereoselectivity can be achieved if using 2-chloroacetoacetate, while no conjugated EP product can be obtained if using 4-chloroacetoacetate. However, according to Seeman's report that bases, such as NaH, are supposed to be able to promote 1,3-hydrogen relocation of allyl compounds, we then suspect that the unconjugated EP product **5a** may be able to be transformed into the conjugated one in a stereoselective way.<sup>20</sup>

Inspired by the above idea, we then turned to examine the possibility of the base-promoted 1,3-hydrogen rearrangement of **5a**. As shown in Table 1, among the eight kinds of bases examined, including inorganic *t*-BuOK, CH<sub>3</sub>ONa, NaOH, NaH,

Table 1 Optimization of base-promoted 1,3-hydrogen rearrangement of unconjugated  $\beta$ -phosphoroxylated allylic ester 5a  $^a$ 

Entry	Base	Load ( $x$ eq.)	Solvent	$T$ ( $^{\mathrm{o}}$ C)	Time (h)	$Yield^{b}$ (%)	$E/Z (4a)^c$
1	t-BuOK	1.2	THF	rt	24	0	_
2	CH <sub>3</sub> ONa	1.2	THF	rt	24	0	_
3	NaOH	1.2	THF	rt	24	0	_
4	NaH	1.2	THF	rt	24	0	_
5	$K_2CO_3$	1.2	THF	rt	24	0	_
6	$\mathrm{Et_{3}N}$	1.2	THF	rt	24	90	>99:1
7	Pyridine	1.2	THF	rt	24	0	_
8	(i-Pr) <sub>2</sub> NEt	1.2	THF	rt	24	20	>99:1
9	$Et_3N$	1.2	$CH_3CN$	rt	4	92	>99:1
10	$\mathrm{Et_{3}N}$	1.2	DCM	rt	20	90	>99:1
11	$\mathrm{Et_{3}N}$	1.2	$CH_3OH$	rt	22	83	>99:1
12	$\mathrm{Et_{3}N}$	1.2	DMF	rt	24	75	>99:1
13	$\mathrm{Et_{3}N}$	0.5	$CH_3CN$	rt	7	92	>99:1
14	$Et_3N$	0.1	CH <sub>3</sub> CN	rt	12	92	>99:1
15	$\mathrm{Et_{3}N}$	0.05	$CH_3CN$	rt	20	93	>99:1
16	$\mathrm{Et_{3}N}$	0.1	$CH_3CN$	0	24	95	>99:1
17	$Et_3N$	0.1	$CH_3CN$	80	4	90	>99:1

<sup>&</sup>lt;sup>a</sup> Reaction conditions: 5a (1.0 equiv.), base (x equiv.), solvent (3 ml). <sup>b</sup> Isolated yields. <sup>c</sup> Determined by NMR.

K<sub>2</sub>CO<sub>3</sub>, and organic Et<sub>3</sub>N, Pyridine (i-Pr)<sub>2</sub>NEt, only Et<sub>3</sub>N and (i-Pr)<sub>2</sub>NEt exhibited the supposed promoting abilities, affording the desired product E-4a, but encouragely both in >99% (E)stereoselectivity. Though only 20% yield was obtained by 1.2 equivalent (i-Pr)2NEt after 24 h reaction in THF at room temperature (Table 1, entry 8), while up to 90% yield was acquired by using Et<sub>3</sub>N (Table 1, entry 6). The following screening of solvents demonstrated that acetonitrile seemed to the best choice that the reaction could be accomplished in only 4 h and gave a higher yield of 92% (Table 1, entry 10). Further investigation about the loadage of Et<sub>3</sub>N showed that only 0.1 equivalent Et<sub>3</sub>N was sufficient to promote the rearrangement effectively, affording the comparative yield though with a few longer time of 12 h (Table 1, entry 14). Less loadage of Et<sub>3</sub>N and lower temperature both led to much longer reaction times (Table 1, entry 15&16). Though the reaction time could be shortened to 4 h at a higher temperature of 80 °C (Table 1, entry 17), we finally preferred the more benign room temperature for the following preparations.

Considering the convenience of experimental operation, we then turned into the possibility of one-pot manipulation. It was found that product  $\emph{E-4a}$  was afforded in 92% yield if using the crude intermediate 5a directly for the subsequent rearrangement reaction. Therefore, a mild  $\emph{E-}$ stereoselective one-pot synthetic approach of  $\beta$ -phosphoroxylated  $\alpha,\beta$ -unsaturated esters was thus established: 3 (1.0 eq.) and P(m)-reagents (1.0 eq.) react 1 h at 40 °C neatly, then added triethylamine (0.1 eq.) and acetonitrile (3 mL), and further react about 12 h at room temperature.

Having identified the optimal reaction conditions, we next set out to examine the scope of this new mild one-pot enol phosphorylation procedure (Table 2). As for the different *O*-alkyl

**Table 2** Scope of β-phosphoroxylated (*E*)- $\alpha$ ,β-unsaturated esters<sup>a,b</sup>

4-chloroacetoacetate substrates, all the common P(III)-reagents possessing P–O, P–C, and/or P–N bonds gave the corresponding EPs in high yields. During the preparation of compounds **4e** and **4f**, the rearrangement reactions were found much accelerated probably due to the higher reactivities of phosphonite and phosphinite compared to phosphites. To demonstrate the practical utility, the reaction towards product **4a** was performed at the 50 mmol scale and 92% yield was obtained. The stereoscopic (*E*)-configuration of solid product **4f** was further confirmed by single crystal X-ray analysis.

With the *E*-stereospecific  $\beta$ -phosphoroxylated  $\alpha,\beta$ -unsaturated esters in hand, we then investigated their stereoretentive Negishi cross-coupling to prepare the corresponding *E*-stereodefined disubstituted  $\alpha,\beta$ -unsaturated esters. Among the typical catalysts screened including Pd(PPh<sub>3</sub>)<sub>4</sub>, Ni(acac)<sub>2</sub> and Pd(dppb) Cl<sub>2</sub>, the latter demonstrated the best performance in this Negishi reaction with only 0.02 equivalent loading by refluxing in acetonitrile. Various aromatic ArZnCl nucleophiles containing electron-donating and/or electron-withdrawing substituents at *ortho*, *meta*, and/or *para* positions were all tolerated well, affording the desired products in good to excellent yields (80–

Table 3 Scope of (E)- $\alpha$ , β-unsaturated esters *via* a stereoretentive Negishi cross-coupling reaction of **4a**  $^{a,b}$ 

 $<sup>^</sup>a$  Reaction conditions: 1 (1.0 mmol), 3 (1.0 mmol), Et<sub>3</sub>N (0.1 mmol), CH<sub>3</sub>CN (3.0 mL).  $^b$  Isolated yields.

 $<sup>^</sup>a$  Reaction conditions: 4a (1.0 mmol), ArMgBr (1.5 mmol), ZnCl $_2$  (1.5 mmol), Pd(dppb)Cl $_2$  (0.02 mmol), CH $_3$ CN (5.0 mL), reflux about 3 h.  $^b$  Isolated yields.

96%) without generating any stereochemical integrity (Table 3, 6a-6m). Disubstituted, condensed and hetero aromatic organometallic substrates also gave 85–92% yields of the products (Table 3, 6n-6r). However, it's regrettable that alkyl organozinc reagents was found unreactive under such conditions.

Furthermore, under the above optimal Negishi cross-coupling reaction conditions, both (Z) and (E) isomers of  $\alpha$ , $\beta$ -unsaturated esters **6a** could be easily achieved, just by one operation, directly from the (Z) and (E) mixture of **4a** in 22% and 70% yields respectively (Scheme 3).

According to the Cram's mechanistic interpretation for the allylic rearrangements, an intra-molecular pathway of the  $Et_3N$ -promoted stereoselective 1,3-hydrogen rearrangement of the EPs 5a was proposed because that the degree of the observed intramolecularity depended strongly on the base and solvent used. As shown in Scheme 4, triethylamine firstly removes the proton from the  $\alpha$ -carbon position of ester 5a, resulting in a coplanar anionic allylic system by three carbon atoms. The hydrogen atom of the H–Et $_3N$  ammonium then bonds to both terminal carbon atoms to form the intermediate Int, collapse of which would then give the thermodynamically favourable conjugated  $\alpha$ ,  $\beta$ -unsaturated ester product E-4a.

In summary, a mild and environmental trimethylamine-catalyzed E-stereoselective 1,3-hydrogen allylic rearrangement of enol phosphates was firstly developed to afford versatile  $\beta$ -phosphoroxylated (E)- $\alpha$ , $\beta$ -unsaturated esters which can be then efficiently converted into the corresponding  $\beta$ , $\beta$ -disubstituted (E)- $\alpha$ , $\beta$ -unsaturated esters in high yields by a 100% stereoretentive Negishi cross-coupling reaction. Moreover, both (Z) and (E)- $\alpha$ , $\beta$ -unsaturated esters were able to be achieved in one manipulation when just employing 2-chloroacetoacetate instead of 4-chloroacetoacetate for the solvent and metal-free Perkow reaction.

It is interesting to note that more structure-diverse  $\alpha,\beta$ -unsaturated esters should be easily obtained by derivation reactions at the allylic position of  $\alpha,\beta$ -unsaturated esters and/or by utilizing 2-substituted 4-chloroacetoacetates as the starting materials.

Scheme 3 Preparation of (Z) and (E) isomers of 6a in one operation.

**Scheme 4** Proposed (*E*)-stereospecific allylic rearrangement mechanism.

### Conflicts of interest

There are no conflicts to declare.

#### Notes and references

- (a) C. X. Zhuo and A. Furstner, J. Am. Chem. Soc., 2018, 140, 10514; (b) D. Romo, N. S. Choi, S. Li, I. Buchler, Z. Shi and J. O. Liu, J. Am. Chem. Soc., 2004, 126, 10582; (c) S. D. Marco, A. Cammas, J. Pelletier and I. E. Gallouzi, Nat. Commun., 2012, 3, 8963; (d) S. K. Naineni, J. Liang, B. Nagar and J. Pelletier, Cell Chem. Biol., 2021, 28, 825.
- (a) J. K. Nunnery, N. Engene, T. Byrum, T. F. Murray and W. H. Gerwick, *J. Org. Chem.*, 2012, 77, 4198; (b)
   S. Sekharan and K. Morokuma, *J. Am. Chem. Soc.*, 2011, 133, 19052; (c) E. A. Zhukovsky, P. R. Robinson and D. D. Oprian, *Science*, 1990, 251, 558.
- 3 M. Kiser and K. Golczak, Chem. Rev., 2014, 114, 194.
- 4 (a) P. Guo, Y. Zhang, L. Zhang and Q. Xia, J. Biol. Chem., 2021, 297, 101234; (b) Y. Ando, K. Matsumoto, K. Misaki, G. Mano, T. Shinada and S. G. Goto, Gen. Comp. Endocrinol., 2020, 289, 113394; (c) M. Nouzova, C. Rivera-Pérez and F. G. Noriega, Curr. Opin. Insect. Sci., 2018, 29, 49; (d) K. Li, Q. Jia and S. Li, Insect Sci., 2019, 26, 600.
- 5 (a) D. A. Evans, P. J. Coleman, L. C. Dias and A. N. Tyler, Angew. Chem., Int. Ed. Engl., 1997, 36, 2744; (b) D. A. Evans, D. M. Fitch, T. E. Smith and V. Cee, J. Am. Chem. Soc., 2000, 122, 10033; (c) D. A. Evans, P. H. Carter, E. M. Carreira, A. B. Charette, J. A. Prunet and M. Lautens, J. Am. Chem. Soc., 1999, 121, 7540; (d) I. Fleming, A. Barbero and D. Walter, Chem. Rev., 1997, 97, 2063.
- 6 (a) E. Negishi, Q. Hu, Z. Huang, M. Qian and G. Wang, Aldrichimica Acta, 2005, 38, 71; (b) J. Li, A. S. Grillo and M. D. Burke, Acc. Chem. Res., 2015, 48, 2297; (c) V. Hornillos, M. Giannerini, C. Vila, M. F. Mastral and B. L. Feringa, Chem. Sci., 2015, 6, 1394.
- 7 (a) K. Murakami and H. Yorimitsu, Beilstein J. Org. Chem.,
  2013, 9, 278; (b) M. G. Suero, E. D. Bayle, B. S. L. Collins and
  M. J. Gaunt, J. Am. Chem. Soc., 2013, 135, 5332; (c) F. Xue,
  J. Zhao, T. S. A. Hor and T. Hayashi, J. Am. Chem. Soc., 2015,
  137, 3189; (d) S. Wang and C. Xi, Org. Lett., 2018, 20, 4131.
- 8 (*a*) F. Guibe, *Tetrahedron*, 1998, **54**, 2967; (*b*) F. Guibe, *Tetrahedron*, 1997, **53**, 13509; (*c*) S. Escoubet, S. Gastaldi and M. Bertrand, *Eur. J. Org. Chem.*, 2005, 3855.
- 9 (a) F. W. Sum and L. Weiler, Can. J. Chem., 1979, 57, 1431; (b)
   M. Ide and M. Nakata, Synlett, 2001, 1511.
- 10 (a) A. B. Flynn and W. W. Ogilvie, Chem. Rev., 2007, 107, 4698; (b) P. Polak, H. Vanova, D. Dvorak and T. Tobrman, Tetrahedron Lett., 2016, 57, 3684; (c) B. E. Maryanoff and A. B. Reitz, Chem. Rev., 1989, 89, 863; (d) Y. Ashida and Y. Tanabe, Chem. Rec., 2020, 20, 1.
- 11 (a) C. Gürtler and S. L. Buckwald, *Chem. Eur. J.*, 1999, 5, 3107; (b) M. Shindo, Y. Sato, T. Yoshikawa, R. Koretsune and K. Shishido, *J. Org. Chem.*, 2004, **69**, 3912.
- 12 (a) A. B. Lemay, K. S. Vulic and W. W. Oglivie, J. Org. Chem., 2006, 71, 3615; (b) J. S. Mercier, A. B. Flynn and W. W. Ogilvie, Tetrahedron, 2008, 64, 5472.

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(a) Y. Yamamoto, N. Kirai and Y. Harada, *Chem. Commun.*,
 2008, 2010; (b) N. Kirai and Y. Yamamoto, *Org. Synth.*,
 2010, 87, 53; (c) Z. He, S. Kirchberg, R. Froehlich and
 A. Studer, *Angew. Chem., Int. Ed.*, 2012, 124, 3759.

- 14 (a) J. M. Baxter, D. Steinhuebel, M. Palucki and I. W. Davies, *Org. Lett.*, 2005, 7, 215; (b) A. Klapars, K. R. Campos, C. Chen and R. P. Volante, *Org. Lett.*, 2005, 7, 1185; (c) H. Nakatsuji, K. Ueno, T. Misaki and Y. Tanabe, *Org. Lett.*, 2008, 10, 2131; (d) H. Nakatsuji, H. Nishikado, K. Ueno and Y. Tanabe, *Org. Lett.*, 2009, 11, 4258.
- 15 Y. Ashida, K. Nakata, D. Yoshitake, Y. Sato, Y. Miyazaki and Y. Tanabe, *Asian J. Org. Chem.*, 2020, **9**, 604.
- 16 (a) A. L. Hansen, J. P. Ebran, M. Ahlquist, P. O. Norrby and T. Skrydstrup, Angew. Chem., Int. Ed., 2006, 45, 3349; (b) T. Hayashi, T. Fujiwa, Y. Okamoto, Y. Katsuro and M. Kumada, Synthesis, 1981, 1001; (c) K. C. Nicolaou, G. Q. Shi, G. P. Grtner and Z. Yang, J. Am. Chem. Soc., 1997, 119, 5467; (d) A. S. E. Karlstrçm, K. Itami and J. E. Bckvall, J. Org. Chem., 1999, 64, 1745; (e) J. Wu and Z. Yang, J. Org. Chem., 2001, 66, 7875; (f) J. P. Ebran, A. L. Hansen, T. M. Gøgsig and T. Skrydstrup, J. Am. Chem.

- Soc., 2007, **129**, 6931; (g) A. L. Hansen, J.-P. Ebran, T. M. Gøgsig and T. Skrydstrup, J. Org. Chem., 2007, 72, 6464; (h) W. You, Y. Li and M. K. Brown, Org. Lett., 2013, **15**, 1610.
- 17 (a) T. M. Gøgsig, A. T. Lindhardt and T. Skrydstrup, *Org. Lett.*, 2009, 11, 4886; (b) J. Jiang, R. DeVita, G. Doss, M. Goulet and M. Wyvratt, *J. Am. Chem. Soc.*, 1999, 121, 593; (c) A. L. Hansen, J. P. Ebran, T. M. Gøgsig and T. Skrydstrup, *Chem. Commun.*, 2006, 39, 4137; (d) J. W. Coe, *Org. Lett.*, 2000, 2, 4205.
- 18 H. Nakatsuji, Y. Ashida, H. Hori, Y. Sato, M. Taira and Y. Tanabe, Org. Biomol. Chem., 2015, 13, 8205.
- 19 (a) Y. Cao, Z. Gao, J. Li, X. Bi, L. Yuan, C. Pei, Y. Guo and E. Shi, RSC Adv., 2020, 10, 29493; (b) H. Guo, Y. Zhang, Z. Li, P. Zhao, N. Li and E. Shi, RSC Adv., 2022, 12, 14844.
- 20 S. G. Alcock, J. E. Baldwin, R. Bohlmann, L. M. Harwood and J. Seeman, J. Org. Chem., 1985, 50, 3526.
- 21 (a) M. Hassam, A. Taher, G. E. Arnott, I. R. Green and W. A. L. van Otterlo, *Chem. Rev.*, 2015, 115, 5462; (b)
  D. J. Cram and R. T. Uyeda, *J. Am. Chem. Soc.*, 1964, 86, 5466.