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Ternary pentagonal BXN (X = C, Si, Ge, and Sn) sheets with high piezoelectricity†

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The discovery of new and stable two-dimensional pentagonal materials with piezoelectric properties is essential for technological advancement. Inspired by recently reported piezoelectric materials *penta*-BCN and *penta*-BSiN, we proposed *penta*-BGeN and *penta*-BSnN as new members of the *penta*-family based on first-principles calculations. Comprehensive analyses indicated that both *penta*-BGeN and *penta*-BSnN are thermodynamically, dynamically, mechanically, and thermally stable. In terms of mechanical stability, the elastic constant decreased as lower elements in group 4A of the periodic table were used. Therefore, *penta*-BGeN and *penta*-BSnN are softer than *penta*-BCN and *penta*-BSiN. In terms of piezoelectric properties, piezoelectric stress and strain tensors increase following the same pattern. In group 4A, *penta*-BSnN had the highest intrinsic piezoelectricity, especially the e_{22} piezoelectric stress. Typically, the piezoelectric strain d_{ij} coefficient increases with material softness; *penta*-BSnN possessed the highest d_{ij} . Thus, due to its inherent piezoelectricity, *penta*-BSnN has tremendous potential as a nanoscale piezoelectric material.

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1 Introduction

Since the discovery of hexagonal graphene, the theoretical exploration and production of two-dimensional (2D) materials have advanced rapidly, even beyond the conventional honeycomb-like structure. A new pentagonal structure, *penta*-graphene, that resembles the Cairo pentagonal tiling proposed by Zhang *et al.*,¹ has been extensively investigated through first-principles calculations, although it cannot yet be experimentally synthesized. Its wide band gap of 3.25 eV and high strength could lead to potential applications in nanoelectronics and nanomechanical devices. Other prominent examples are SiC₂, CN₂, BN₂, B₂C, PdS₂, AlN₂, PtN₂, PdN₂, and NiS₂.^{2–9}

In addition to binary compounds, ternary compounds offer a higher degree of freedom for tuning their electronic and structural properties. The most prominent examples are *penta*-BCN and hydrogenated *penta*-BCN,^{10,11} which possesses a strong piezoelectric response and is thermodynamically, mechanically, and thermally stable according to density functional theory

(DFT). Two dimensional piezoelectric materials have the ability to convert electrical energy into mechanical energy and can find utility in a variety of applications, including sensors, flexible electronics, energy harvesting, and energy storage.^{12,13} Subsequently, *penta*-PdPSe and *penta*-PdPS were experimentally synthesized.¹⁴ However, *penta*-PdPSe is non-piezoelectric materials. However, in recent research, the incorporating sulfur into its surface can induce piezoelectricity from non-piezoelectricity in *penta*-PdPSe.¹⁵

Here, we deployed first-principles calculations to search for a novel member of the ternary pentagonal sheet family with piezoelectric properties. Inspired by the piezoelectricity of *penta*-BCN, we proposed *penta*-BGeN and *penta*-BSnN as two new members of the *penta*-2D sheet family. We found that both are thermodynamically, dynamically, mechanically, and thermally stable. *Penta*-BSnN manifests the high piezoelectricity because of its noncentrosymmetric and semiconducting characteristics. Therefore, *penta*-BSnN is a promising candidate for future nanoscale piezoelectric materials, especially for electro-mechanical devices.

2 Computational method

First-principles calculations using the framework of DFT are employed with the generalized gradient approximation (GGA) by Perdew–Burke–Ernzerhof (PBE)¹⁶ with the plane-wave basis projector augmented wave (PAW)¹⁷ method, as defined in the Vienna *Ab initio* Simulation Package (VASP).^{18,19} A kinetic energy

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cutoff of 500 eV is used for the plane-wave expansion. The structures are optimized until the force on each atom is less than $0.01 \text{ eV } \text{\AA}^{-1}$, and the energy convergence is set to 10^{-5} eV . The Monkhorst-Pack²⁰ scheme with $9 \times 9 \times 1$ is employed for the first Brillouin zone integration. A vacuum layer of 20 \AA is added along the direction perpendicular to the layer to prevent interactions between a crystal layer and its neighboring image. To verify the dynamic stability, the PHONOPY package²¹ with an energy cutoff of 500 eV and a single k -point at Γ are used to calculate the phonon spectra over a $5 \times 5 \times 1$ supercell. To confirm the thermodynamic stability, *ab initio* molecular dynamics (AIMD) simulations were deployed under the constant volume and temperature (NVT) ensemble, where the temperature (T) was controlled using the Nose-Hoover thermostat.^{22–25} The Heyd-Scuseria-Ernzerhof (HSE06)^{26,27} hybrid functional is deployed to calculate the electronic band structure because the standard approximation severely underestimates the band gaps.

3 Results and discussion

We first constructed *penta*-BXN ($X = \text{C, Si, Ge, and Sn}$) based on the reported atomic configuration of BCN in ref. 10 by replacing the C atoms with Si, Ge, or Sn atoms, as shown in Fig. 1. The pentagonal primitive unit cell, represented by the square in Fig. 1, consists of 6 atoms, including 2 B atoms, 2 X ($X = \text{C, Si, Ge, or Sn}$) atoms, and 2 N atoms. The calculated lattice constants (a and b) and buckling height of *penta*-BXN ($X = \text{C, Si, Ge, and Sn}$) are listed in Table 1. Our calculated lattice constants for *penta*-BCN and *penta*-BSiN matched previous studies.^{10,28} The lattice constants a and b increased as we go down a group in the periodic table because the atomic radii increase with elements further down a group. The buckling height h also increases the further down a group an element is.

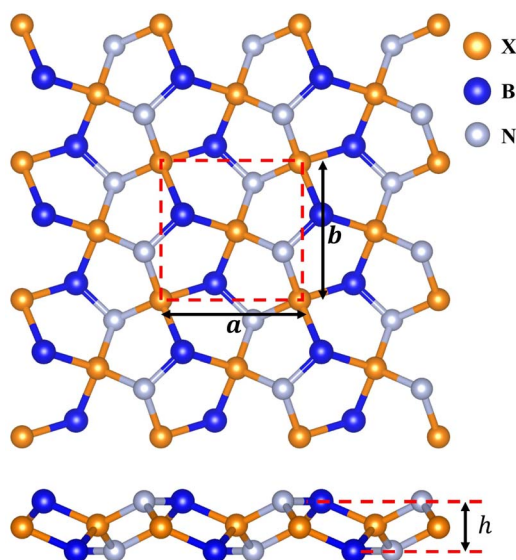


Fig. 1 3×3 supercell of pentagonal BXN ($X = \text{C, Si, Ge and Sn}$) geometrical structures.

Table 1 The calculated lattice constants a and b in \AA , bucking height in \AA , and the band gaps E_g^{PBE} and E_g^{HSE} in eV

	a	b	h	E_g^{PBE}	E_g^{HSE}
<i>penta</i> -BCN	3.670	3.631	1.314	1.70	2.91
<i>penta</i> -BSiN	4.438	4.401	1.602	1.48	2.33
<i>penta</i> -BGeN	4.607	4.590	1.640	1.59	2.42
<i>penta</i> -BSnN	5.045	5.059	1.700	1.05	1.73

In binding stability, the cohesive energy per atom was calculated, given by the following equation;^{29,30}

$$E_{\text{coh}} = \frac{E_{\text{tot}}(\text{BXN}) - \sum_i n_i E_i}{n_i}, \quad (1)$$

where $E_{\text{tot}}(\text{BXN})$ is the total energy of the *penta*-BXN ($X = \text{C, Si, Ge and Sn}$); n_i is the number of Ni, P, and S atoms in the unit cell; and E_i represents the total energies of an isolated single B, X ($X = \text{C, Si, Ge or Sn}$), and N atoms. Our calculated cohesive energies of the *penta*-BXN ($X = \text{C, Si, Ge and Sn}$) are -7.69 , -6.83 , -6.24 and $-5.84 \text{ eV per atom}$ for *penta*-BCN, *penta*-BSiN, *penta*-BGeN and *penta*-BSnN, respectively. The negative values indicate that all *penta*-BXN are more energetically favorable than the isolated atoms. We can see that *penta*-BCN has the lowest cohesive energy per atom. It means that *penta*-BCN has a strongest binding interaction among *penta*-BXN. In comparison, the previous report cohesive energy of *penta*-graphene is $-6.84 \text{ eV per atom}$.³¹ For the binary *penta*-carbides XC_2 , the cohesive energy of *penta*-SiC₂, *penta*-GeC₂, and *penta*-SnC₂ are -5.29 , -4.69 , and $-3.83 \text{ eV per atom}$, respectively.³²

The dynamic stability of *penta*-BCN and *penta*-BSiN has previously been verified by calculating their phonon dispersion.^{10,28} Herein, we verify the dynamic stability of *penta*-BGeN and *penta*-BSnN by constructing phonon band structure. The dynamic stability manifests in phonon spectra containing positive frequencies throughout the Brillouin zone. In contrast, if the spectra display imaginary frequency modes (negative values), the structures contain non-restorative forces against a displacement of atoms, indicating dynamic instability. Fig. 2(a) and (b) show the calculated phonon dispersions of *penta*-BGeN and *penta*-BSnN, respectively. Thus, as shown in

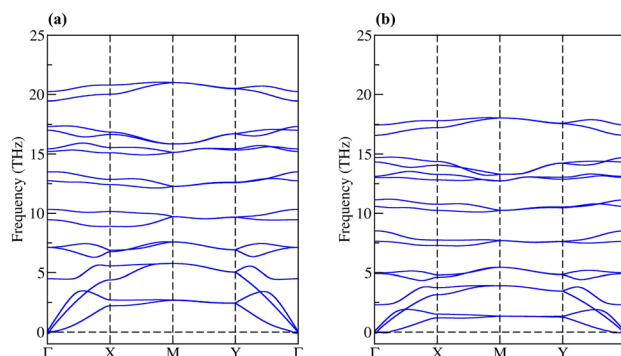


Fig. 2 Calculated phonon band structures of (a) *penta*-BGeN and (b) *penta*-BSnN.



Fig. 2(a) and (b), there are no imaginary modes in the Brillouin zone. These spectra confirm that *penta*-BXN (X = C, Si, Ge, and Sn) are dynamically stable. We note that the imperceptible negative frequencies at Γ , which are only visible when the spectra are magnified, are a well-known computational error that can be effectively eliminated using the local-density approximations (LDA) functional. These negative frequencies of the phonon dispersion have also been observed in *penta*-PdSe₂.³³

Mechanical stability is an important property for strain engineering. Such stability can be verified from the linear elastic constants, C_{11} , C_{22} , C_{12} , and C_{66} , which are elements in the stiffness tensor. Here, the elastic constants are obtained using the finite difference method. The calculated elastic constants of *penta*-BXN (X = C, Si, Ge, and Sn) are listed in Table 2. Our calculated elastic constants were comparable to those previously calculated for *penta*-BCN and *penta*-BSiN,^{10,28} which are listed in Table 2. Typically, structures are considered mechanically stable when the linear elastic constants satisfy the conditions $C_{11}C_{22} - C_{12}^2 > 0$ and $C_{66} > 0$. Notably, these conditions were met for all pentagonal BXN, confirming the mechanical stability of *penta*-BXN (X = C, Si, Ge, and Sn). In comparison, the elastic constant C_{ij} decreased with increasing atomic radii (*i.e.*, descending a group in the periodic table).

Next, we evaluated the 2D Young's modulus in the [100] and [010] directions (*i.e.*, along the plane), which is obtained by $E_a = (C_{11}^2 - C_{12}^2)/C_{11}$ and $E_b = (C_{22}^2 - C_{12}^2)/C_{22}$. Poisson's ratios in the [100] and [010] directions are $\nu_a = C_{12}/C_{11}$ and $\nu_b = C_{12}/C_{22}$, respectively. Young's modulus and Poisson's ratio for all pentagonal BXNs are listed in Table 2. The values of *penta*-BCN and *penta*-BSiN are close to those previously reported, $E_a = 223.45 \text{ Nm}^{-1}$ and $E_b = 189.03 \text{ Nm}^{-1}$ for *penta*-BCN, and $E_a = 113 \text{ Nm}^{-1}$ and $E_b = 109 \text{ Nm}^{-1}$ for *penta*-BSiN. The Young's modulus decreased when we went down a group in the periodic table, while Poisson's ratio increased. These indicate that the pentagonal materials softened as larger elements in group 4A were chosen, and the materials were likely to deform in directions perpendicular to [100] and [010]. Similarly, the elastic constants and Young's modulus of *penta*-XC₂ (X = C, Si, Ge, and Sn) have been reported.³² The calculated Young's modulus values for *penta*-XC₂ (X = C, Si, Ge, and Sn) are 266.22, 138.54,

121.65, and 81.23 Nm^{-1} , respectively. It can be observed that in *penta*-XC₂ (X = C, Si, Ge, and Sn), the Young's modulus decreases as we move down the group in the periodic table. This behavior is similar to that of *penta*-BXN, where larger elements tend to soften the material. The near-zero Poisson's ratios of *penta*-BCN indicate that *penta*-BCN maintains its dimensions under compression and extension along the in-plane directions.

We used eqn (1) and (2) to calculate Young's modulus (E) and Poisson's ratio (ν) along an arbitrary angle θ ,^{34–36} which is the angle relative to the x direction, in order to further investigate how the mechanical properties of *penta*-BXN depend on its crystal orientation.

$$E(\theta) = \frac{C_{11}C_{12} - C_{12}^2}{C_{11}s^4 + C_{22}c^4 + \left(\frac{C_{11}C_{12} - C_{12}^2}{C_{66}} - 2C_{12}\right)c^2s^2}, \quad (2)$$

$$\nu(\theta) = -\frac{\left(C_{11} + C_{12} - \frac{C_{11}C_{12} - C_{12}^2}{C_{66}}\right)c^2s^2 - C_{12}(c^4 + s^4)}{C_{11}s^4 + C_{22}c^4 + \left(\frac{C_{11}C_{12} - C_{12}^2}{C_{66}} - 2C_{12}\right)c^2s^2}, \quad (3)$$

where $c = \cos \theta$ and $s = \sin \theta$. The calculated results for *penta*-BGeN and *penta*-BSnN are plotted in Fig. 3, while the calculated results for *penta*-BCN and *penta*-BSiN have been provided previously.^{10,28} The maxima of the in-plane Young's modulus in *penta*-BGeN and *penta*-BSnN are located around the [110] direction (40 for *penta*-BGeN and 45 for *penta*-BSnN). Meanwhile, the minima in-plane Young's modulus of both are along [010] (90). In *penta*-BGeN, the minimum and maximum Poisson's ratios are found along [110] (43) and [010] (90), respectively. In *penta*-BSnN, the minimum and maximum Poisson's ratios are located at the same angle as *penta*-BGeN but have higher positive values. Because the in-plane Young's modulus and Poisson's ratios depend on crystal orientation, *penta*-BXN (X = C, Si, Ge, and Sn) exhibit anisotropic mechanical properties that can be explained by their lattice constants, where $a \neq b$.

Using mapping of indices, the relationship among the elastic (C_{jk}), piezoelectric stress (e_{ik}), and strain tensors (d_{ij}) is

$$e_{ik} = d_{ij}C_{jk} \quad (4)$$

Specifically, for a pentagonal system, the relationship between the three variables becomes

Table 2 The elastic constants C_{jk} (Nm^{-1}), Poisson's ratio in [100] (ν_a) and [010] (ν_b) directions and Young's modulus (Nm^{-1}) in [100] (E_a) and [010] (E_b) directions

	Previous work		This work			
	BCN ¹⁰	BSiN ²⁸	BCN	BSiN	BGeN	BSnN
C_{11}	223.56	114.46	222.31	116.97	100.59	66.82
C_{12}	4.90	12.76	4.03	13.06	10.69	17.85
C_{22}	187.39	111.21	187.39	112.60	98.45	66.64
C_{66}	105.38	48.97	105.38	49.36	47.09	32.57
ν_a	0.022	0.11	0.018	0.112	0.106	0.267
ν_b	0.026	0.11	0.022	0.116	0.109	0.268
E_a	223.45	113	222.24	115.51	99.46	62.05
E_b	189.03	109	187.30	111.08	97.30	61.86

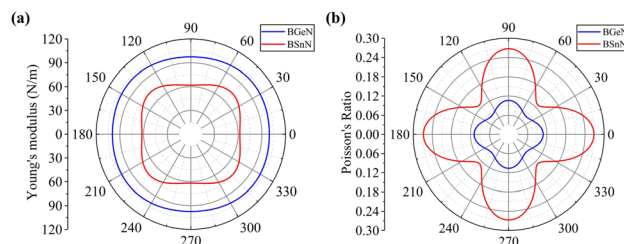


Fig. 3 Variation of (a) in-plane Young's modulus, and (b) Poisson's ratio for *penta*-BGeN and *penta*-BSnN. Note that solid blue and red lines represent *penta*-BGeN and *penta*-BSnN, respectively.



$$\begin{pmatrix} 0 & 0 & e_{16} \\ e_{21} & e_{22} & 0 \\ 0 & 0 & e_{36} \end{pmatrix} = \begin{pmatrix} 0 & 0 & d_{16} \\ d_{21} & d_{22} & 0 \\ 0 & 0 & d_{36} \end{pmatrix} \begin{pmatrix} C_{11} & C_{12} & 0 \\ C_{21} & C_{22} & 0 \\ 0 & 0 & C_{33} \end{pmatrix}. \quad (5)$$

Piezoelectric tensors e_{ik} are calculated using density functional perturbation theory (DFPT) as implemented in the VASP package. The elastic stiffness tensors are calculated by using the finite difference method mentioned before. Therefore, the elements of the strain tensors, d_{16} , d_{21} , d_{22} and d_{36} , are derived:

$$d_{16} = \frac{e_{16}}{C_{66}}, \quad (6)$$

$$d_{21} = \frac{C_{21}e_{22} - C_{22}e_{21}}{C_{21}C_{12} - C_{11}C_{22}}, \quad (7)$$

$$d_{22} = \frac{C_{12}e_{21} - C_{11}e_{22}}{C_{21}C_{12} - C_{11}C_{22}}, \quad (8)$$

$$d_{36} = \frac{e_{36}}{C_{66}} \quad (9)$$

According to eqn (5)–(8), the piezoelectric strain tensors d_{ij} , which are listed in Table 3, are derived from C_{jk} , as listed in Table 2, and e_{ik} , as listed in Table 3. Our calculated piezoelectric strain tensors, d_{ij} , and piezoelectric stress tensors, e_{ik} , were comparable to those of previous work in *penta*-BCN and *penta*-BSiN,^{10,28} which are also listed in Table 3. Note that the piezoelectric stress tensors e_{21} and e_{22} refer to the induced polarization along the *x*- and *y*-directions, respectively, when the strain is applied along the *y*-direction. Meanwhile, e_{16} and e_{36} represent the induced polarization along the *xy*-direction when the strain is applied along the *x*- and *y*-directions, respectively. The negative values of e_{ik} and d_{ij} refer to the opposite direction to the strain displacement. From the calculation results, the piezoelectric stress e_{ik} and strain d_{ij} tensors tended to increase as the substituted atoms went down the group in the periodic table, which is illustrated in ESI Fig. S1.† It can be inferred that the softer materials possess larger d_{ij} coefficients. The piezoelectric strain tensor d_{ij} of *penta*-BSnN is largest because *penta*-BSnN possessed the smallest elastic constants. Typically, the

piezoelectric strain d_{ij} coefficient increases with material softness.³⁷ The piezoelectric stress tensor e_{22} of *penta*-BSnN is about 3.3 times higher than that of *penta*-BCN. In addition, all of the piezoelectric coefficients (e_{21} , e_{16}) of *penta*-BXN are higher than those of two dimensional *h*-BN ($e_{11} = 1.38 \times 10^{-10} \text{ C m}^{-1}$).³⁸

The thermal stabilities of *penta*-BGeN and *penta*-BSnN were examined *via ab initio* molecular dynamic simulations. Supercells with a $3 \times 3 \times 1$ duplication of the unit cell are used. The molecular dynamics simulations are performed for 8 ps with a time step of 1 fs at 1200 K. Fig. 4(a) shows the fluctuation of the potential energy of *penta*-BGeN throughout the simulation at 1200 K. *Penta*-BGeN is thermally stable at 1200 K. However, *penta*-BSnN is unable to stand at 1200 K, as shown in ESI Fig. S2.† Therefore, we lower the simulation temperature to 600 K. Finally, *penta*-BSnN possesses thermal stability around 600 K, as illustrated in Fig. 4(b).

The electronic properties of *penta*-BGeN and *penta*-BSnN were first explained by the calculated Heyd-Scuseria-Ernzerhof (HSE06)²⁷ band structure, as shown in Fig. 5 When the PBE functional¹⁶ was used, the electronic band gap showed indirect band gaps of 1.59 and 1.05 eV for *penta*-BGeN and *penta*-BSnN, as listed in Table 1. The standard DFT underestimates the band gap.^{39–41} Therefore, the HSE06 functional,^{26,27} which mixes 75% HF exchange and 25% PBE exchange, was also employed to address this issue. With the use of this functional, a larger HSE band gap of 2.42 eV was obtained for *penta*-BGeN while a larger HSE band gap of 1.73 eV was achieved for *penta*-BSnN. Both had an indirect band gap character. As shown in ESI Fig. S1,† the valence band maximum (VBM) of *penta*-BGeN and *penta*-BSnN are found at Γ point when HSE06 was used, while the conduction band minimum (CBM) was established along the $\Gamma - X$ path for *penta*-BGeN, and the along $\Gamma - Y$ path for *penta*-BSnN.

Next, we will discuss the effect of band edge position on photocatalytic water splitting. Recent density functional theory (DFT) studies have shown that *penta*-materials are among the possible candidates for photocatalytic water splitting.^{28,32} In water splitting, the band edges position of VBM must cover the redox/oxidation level potentials of water. The standard redox potentials with pH of oxygen evolution reaction (OER) is $E_{\text{O}_2/\text{H}_2\text{O}} = -5.67 + \text{pH} \times 0.059 \text{ eV}$, while standard redox potentials with pH of hydrogen evolution reaction (HER) can be express as $E_{\text{H}^+/\text{H}_2} = -4.44 + \text{pH} \times 0.059 \text{ eV}$.^{42,43} In the present study, we present the calculated positions of the VBM and CBM with respect to the vacuum level, as shown in ESI Fig. S3.† Our results reveal that

Table 3 Piezoelectric stress tensors e_{ik} in unit of $\times 10^{-10} \text{ C m}^{-1}$ and piezoelectric strain tensors d_{ij} of *penta*-BXN (X = C, Si, Ge and Sn) in the unit of pm V^{-1}

	Previous work		This work			
	BCN ¹⁰	BSiN ²⁸	BCN	BSiN	BGeN	BSnN
e_{21}	1.93	3.78	1.93	2.35	2.14	1.93
e_{22}	−1.24	−2.40	−1.24	−2.27	−2.68	−4.11
e_{16}	1.80	0.84	1.80	1.50	1.86	2.22
e_{36}	−0.08	−0.26	−0.081	−0.068	−0.113	−0.121
d_{21}	0.878	3.59	0.882	2.27	2.45	4.88
d_{22}	−0.678	−2.57	−0.680	−2.28	−2.98	−7.47
d_{16}	1.72	1.71	1.70	3.03	3.96	6.81
d_{36}	−0.076	−0.53	−0.076	−0.139	−0.240	−0.372

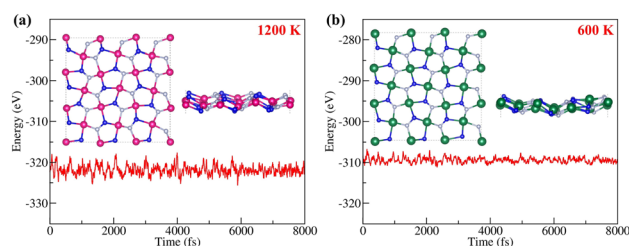


Fig. 4 Total energy fluctuation with time during the AIMD simulation at (a) 1200 K of *penta*-BGeN and (b) 600 K of *penta*-BSnN.



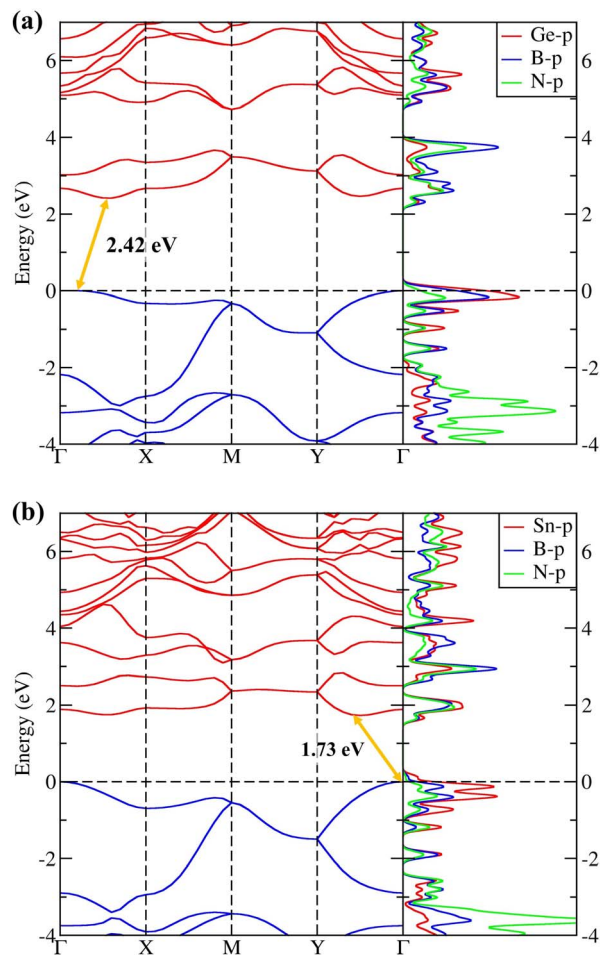


Fig. 5 The electronic band structure and partial Density of States (pDOS) of (a) *penta*-BGeN and (b) *penta*-BSnN.

the VBM shifts rapidly with increasing pH, while the CBM remains relatively constant. Although, at a pH of 7 (as indicated by the red dashed line), the CBM band edges of all the BXN (X = C, Si, Ge, and Sn) are above the hydrogen evolution potential (H^+/H_2), only the *penta*-BCN material has the VBM band edges below the oxygen evolution potential (O_2/H_2O). These findings suggest that *penta*-BCN is a promising material for water-splitting processes, as it can catalyze both the hydrogen and oxygen evolution reactions at pH 7. Additionally, a decrease in pH (as indicated by the black dashed line) leads to a downward shift in the oxidation and reduction potentials of water, which favors the HER but is detrimental to the OER. Therefore, *penta*-BCN is the only BXN material suitable for OER under highly acidic conditions (pH = 0) in water-splitting processes.

The band gaps of 2D materials can potentially be modulated by strain. Thus, we also investigated how the band gaps of *penta*-BGeN and *penta*-BSnN change with additional biaxial strain from equilibrium (*i.e.*, zero strain) to the critical strain point. In order to avoid the interaction of periodic conditions, the $2 \times 2 \times 1$ supercell has been adopted to release the special constraints. The values of two critical strains, (ϵ_{C1} and ϵ_{C2}), can be inferred from Fig. 6. The first is the point where the

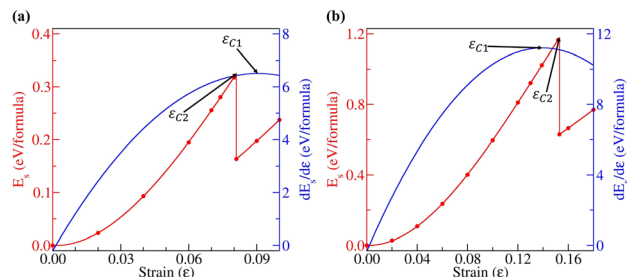


Fig. 6 Strain-stress relations for (a) *penta*-BGeN and (b) *penta*-BSnN with biaxial strains. The vertical dashed lines indicate the critical structure instability.

derivative curve reaches its maximum turning point. This point is widely known as the bond-breaking point. The second is the point at which the plastic range begins to manifest irreversible deformations (reconstruction process). Above ϵ_{C2} , the strain energy always increases, and the system maintains its reconstructive structure. In *penta*-BGeN, ϵ_{C1} occurred at nearly $\epsilon = 0.14$ while it occurred at nearly $\epsilon = 0.08$ in *penta*-BSnN. This means that *penta*-BGeN could withstand biaxial strain before mechanical failure at these critical points. However, in *penta*-BGeN, structural instability occurred at ϵ_{C2} , at which point the structure starts to collapse.

4 Conclusions

By using DFT-based calculations, we theoretically explored *penta*-BGeN and *penta*-BSnN as the new members in the *penta*-2D family. Both *penta*-BGeN and *penta*-BSnN are dynamically, mechanically, and thermally stable, as comprehensively verified by the phonon dispersion, elastic constants, and molecular dynamics simulations, respectively. *Penta*-BGeN and *penta*-BSnN are semiconductors with indirect band gaps of 2.42 and 1.73 eV, respectively. *Penta*-BGeN and *penta*-BSnN are soft materials because their elastic constants decrease as we go down the group in the periodic table. The elements of the piezoelectric stress and strain tensors increase as we move down a group in the periodic table. In the same family, *penta*-BSnN yields the highest intrinsic piezoelectricity, especially the e_{22} piezoelectric stress. Typically, the piezoelectric strain, d_{ij} , coefficient increases with material softness, yielding the greatest d_{ij} for *penta*-BSnN. Thus, *penta*-BSnN has promising prospects as a nanoscale piezoelectric material due to its inherent piezoelectricity.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 S. Zhang, J. Zhou, Q. Wang, X. Chen, Y. Kawazoe and P. Jena, *Proc. Natl. Acad. Sci. U. S. A.*, 2015, **112**, 2372–2377.
- 2 G. R. Berdiyrov and M. E.-A. Madjet, *RSC Adv.*, 2016, **6**, 50867–50873.
- 3 S. Zhang, J. Zhou, Q. Wang and P. Jena, *J. Phys. Chem. C*, 2016, **120**, 3993–3998.
- 4 J. Li, X. Fan, Y. Wei and G. Chen, *Sci. Rep.*, 2016, **6**, 31840.
- 5 F. Li, K. Tu, H. Zhang and Z. Chen, *Phys. Chem. Chem. Phys.*, 2015, **17**, 24151–24156.
- 6 Z. Cheng, X. Zhang, H. Zhang, J. Gao, H. Liu, X. Yu, X. Dai, G. Liu and G. Chen, *Phys. Chem. Chem. Phys.*, 2021, **23**, 6278–6285.
- 7 Y.-S. Lan, X.-R. Chen, C.-E. Hu, Y. Cheng and Q.-F. Chen, *J. Mater. Chem. A*, 2019, **7**, 11134–11142.
- 8 J. Li, X. Fan, Y. Wei, H. Liu, S. Li, P. Zhao and G. Chen, *Sci. Rep.*, 2016, **6**, 1–10.
- 9 C.-T. Wang and S. Du, *Phys. Chem. Chem. Phys.*, 2020, **22**, 7483–7488.
- 10 K. Zhao, Y. Guo, Y. Shen, Q. Wang, Y. Kawazoe and P. Jena, *J. Phys. Chem. Lett.*, 2020, **11**, 3501–3506.
- 11 T. Thanasarnsurapong, K. Dabsamut, T. Maluangnont, J. T-Thienprasert, S. Jungthawan and A. Boonchun, *J. Appl. Phys.*, 2021, **129**, 095101.
- 12 S. D. Mahapatra, P. C. Mohapatra, A. I. Aria, G. Christie, Y. K. Mishra, S. Hofmann and V. K. Thakur, *Adv. Sci.*, 2021, **8**, 2100864.
- 13 M. B. Ghasemian, T. Daeneke, Z. Shahrababaki, J. Yang and K. Kalantar-Zadeh, *Nanoscale*, 2020, **12**, 2875–2901.
- 14 P. Li, J. Zhang, C. Zhu, W. Shen, C. Hu, W. Fu, L. Yan, L. Zhou, L. Zheng, H. Lei, *et al.*, *Adv. Mater.*, 2021, **33**, 2102541.
- 15 K. Dabsamut, I. Chatratin, T. Thanasarnsurapong, S. Jungthawan and A. Boonchun, *J. Alloys Compd.*, 2023, 169640.
- 16 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, **77**, 3865.
- 17 P. E. Blöchl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953.
- 18 G. Kresse and J. Furthmüller, *Comput. Mater. Sci.*, 1996, **6**, 15–50.
- 19 G. Kresse and J. Furthmüller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 11169.
- 20 H. J. Monkhorst and J. D. Pack, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1976, **13**, 5188.
- 21 A. Togo and I. Tanaka, *Scr. Mater.*, 2015, **108**, 1–5.
- 22 S. Nosé, *J. Chem. Phys.*, 1984, **81**, 511–519.
- 23 N. Shuichi, *Prog. Theor. Phys. Suppl.*, 1991, **103**, 1–46.
- 24 W. G. Hoover, *Phys. Rev. A*, 1985, **31**, 1695.
- 25 D. Frenkel and B. Smit, *Understanding molecular simulation: from algorithms to applications*, Elsevier, 2001, vol. 1.
- 26 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2003, **118**, 8207–8215.
- 27 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2006, **124**, 219906.
- 28 M. J. Varjovi, M. Kilic and E. Durgun, *Phys. Rev. Mater.*, 2022, **6**, 034004.
- 29 K. Dabsamut, T. Thanasarnsurapong, I. Chatratin, T. Maluangnont, S. Jungthawan and A. Boonchun, *J. Phys. Chem. C*, 2022, **126**, 19455–19461.
- 30 K. Dabsamut, I. Chatratin, T. Thanasarnsurapong, T. Maluangnont and A. Boonchun, *Phys. Chem. Chem. Phys.*, 2023, **25**, 3815–3819.
- 31 N. Sathishkumar, S.-Y. Wu and H.-T. Chen, *Chem. Eng. J.*, 2020, **391**, 123577.
- 32 M. E. Kilic and K.-R. Lee, *Phys. Rev. Mater.*, 2021, **5**, 065404.
- 33 A. D. Oyedele, S. Yang, L. Liang, A. A. Puzetzy, K. Wang, J. Zhang, P. Yu, P. R. Pudasaini, A. W. Ghosh, Z. Liu, *et al.*, *J. Am. Chem. Soc.*, 2017, **139**, 14090–14097.
- 34 E. Cadelano, P. L. Palla, S. Giordano and L. Colombo, *Phys. Rev. B*, 2010, **82**, 235414.
- 35 H. Wang, X. Li, P. Li and J. Yang, *Nanoscale*, 2017, **9**, 850–855.
- 36 Y. Zhao, X. Li, J. Liu, C. Zhang and Q. Wang, *J. Phys. Chem. Lett.*, 2018, **9**, 1815–1820.
- 37 R. Hinchet, U. Khan, C. Falconi and S.-W. Kim, *Mater. Today*, 2018, **21**, 611–630.
- 38 K.-A. N. Duerloo, M. T. Ong and E. J. Reed, *J. Phys. Chem. Lett.*, 2012, **3**, 2871–2876.
- 39 K. Dabsamut, T. Maluangnont, P. Reunchan, J. T-Thienprasert, S. Jungthawan and A. Boonchun, *Appl. Phys. Lett.*, 2022, **120**, 203101.
- 40 H. Xiao, J. Tahir-Kheli and W. A. Goddard III, *J. Phys. Chem. Lett.*, 2011, **2**, 212–217.
- 41 M. K. Chan and G. Ceder, *Phys. Rev. Lett.*, 2010, **105**, 196403.
- 42 G. W. VanLoon and S. J. Duffy, *Environmental chemistry: a global perspective*, Oxford university press, 2017.
- 43 C. Dues, W. G. Schmidt and S. Sanna, *ACS Omega*, 2019, **4**, 3850–3859.

