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Strong anisotropic second-order nonlinear optical responses in 0D lead-free chiral perovskite single-crystalline microwire arrays†

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Low-dimensional hybrid organic–inorganic chiral perovskites with intrinsic non-centrosymmetric structures have shown promising potential for second-order nonlinear optical applications due to their structural diversity and large lattice distortions. Simultaneously, “structure-level” low-dimensional single crystal perovskites have attracted wide attention to meet the increasing demand of on-chip integrated nonlinear optical devices. Herein, we synthesized two new chiral hybrid 0D perovskites $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ based on the non-toxic element bismuth by introducing chiral *R/S*-MBA as the organic component. The highly ordered assembly microwire arrays were further prepared using a capillary-bridge confined technique. These high-crystallinity chiral perovskite microwire arrays preserve their structures with high distortion levels and achieve high-efficiency frequency conversion across a wide range of bands and different polarization states with a high polarization ratio of up to 0.992, which is superior to those of the corresponding polycrystalline films that cause undesirable light scattering. Such non-toxic Bi-based low-dimensional chiral perovskite single-crystalline microwire arrays with excellent NLO performances might provide a strategy for NLO-integrated applications in advancing photonic devices.

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1. Introduction

Nonlinear optics (NLO) is the branch of science concerned with the nonlinear regime of the interaction between intense lasers and matter, which is of great significance in the development of laser technology and spectroscopy.^{1,2} As a typical second-order NLO response, second-harmonic generation (SHG) refers to a nonlinear process of converting two photons with the same frequency into one with twice the frequency based on the asymmetric structure, boosting a series of

advanced applications ranging from laser manufacturing and engineering, biological microscopic imaging, to telecommunication and military technology.^{3–7} At present, most commercial SHG materials are inorganic materials due to their relatively high second-order NLO susceptibility, high mechanical strength, and good physicochemical stability.^{8–10} However, the inherent shortcomings of inorganic materials, such as the incompatibility of flexible substrates and processing difficulties, significantly hinder their potential application in flexible, miniaturized, and integrated NLO devices. Thus, it is imperative to develop novel NLO materials to promote the advancement of NLO fields.

A hybrid halide perovskite with abundant chemical structures and optical-electrical properties has emerged as a promising material for optoelectronic applications, such as light-emitting diodes, photodetectors, field-effect transistors, solar cells, etc.^{11–19} Recently, the introduction of chiral cations has bolstered multiple exotic optical, electrical, and spintronic properties of perovskites. Particularly, the breaking of lattice inversion symmetry induced by chiral cations opens a new avenue for chiral halide perovskites in the field of NLO.^{20–24} Compared to their three-dimensional counterparts, low-dimensional perovskite materials ranging from two, one, to zero dimensions (0D) exhibit a more pronounced NLO responses due to the light-matter interaction, various excitonic

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effects, structural diversity, and large lattice distortions.^{25–32} Meanwhile, the large bandgap of these low-dimensional perovskites leads to their transparency over a broader spectral range, providing a wide wavelength window for SHG applications.^{26,27} However, most low-dimensional chiral perovskites contain toxic Pb^{2+} , seriously impeding their future commercialization and industrialization for NLO devices. The adjacent VA group cation Bi^{3+} , which possesses a similar ionic radius and electronegativity but less toxicity compared to the toxic Pb^{2+} , has been recently developed as an isoelectronic alternative to Pb^{2+} .^{33–36} The introduction of Bi^{3+} can also promote the formation of low-dimensional perovskite structures to further improve the stability.^{35,36} Thus, the non-toxic Bi -based low-dimensional perovskites will become the new and important explored sector of perovskite application accelerating the development of integrated NLO devices.

Besides the “material-level” low dimension, the “structure-level” low dimension, emphasizing the morphology, especially the array patterning, is also an essential step toward integrated nonlinear optical devices.^{31,32,37–43} Long-range ordered perovskite single-crystalline arrays with unique optical anisotropy and cleavage surface necessary for nonlinear optics can compensate for the defects, such as scattering of light in films and inherently dispersive refractive indices of bulk crystals, attracting more attention for compact integrated devices to meet the increasing demand of integrated photonic circuits and on-chip nanophotonics.^{43–45} However, the controllable manufacturing and patterning of high-quality NLO perovskite single-crystal arrays have become one of the key bottlenecks in the next generation of nanophotonic devices.

In this work, we report the synthesis of two new 0D chiral non-toxic Bi -based perovskites, $(R/S\text{-MBA})_4Bi_2Cl_{10}$, featuring an intrinsic non-centrosymmetric polar crystal structure, and the observation of efficient second-order NLO responses from their high crystallinity and pure crystallographic orientation microwire arrays assembled by capillary-bridge confined techniques. The systematic study of power-, wavelength-, and polarization-dependent SHG responses demonstrates the excellent second-order NLO performance of those chiral perovskite microwire arrays with a wide transparent region. In contrast to the polycrystalline film counterparts, $(R/S\text{-MBA})_4Bi_2Cl_{10}$ single-crystalline microwire arrays exhibit excellent NLO performances, such as a stronger SHG response and a higher polarization ratio of up to 0.992. This work suggests the potential applications of low-dimensional lead-free chiral perovskites in NLO applications and will vigorously accelerate the development of integrated optical devices.

2. Experimental section

2.1. Synthesis of perovskite crystals

Bi_2O_3 (466 mg, 1 mmol) and *R*-(+)- or *S*-(−)- α -methylbenzylamine (485 mg, 4 mmol) were added into aqueous HCl solution (20 mL) with magnetic stirring. The solution was heated to 150 °C for 5 h until it is clear and then

was slowly cooled to room temperature. Colorless perovskite crystals were obtained after 2 h, which were filtered, washed with chlorobenzene, and dried in a vacuum oven for purification.

2.2. Fabrication of microwire arrays and films

The synthesized crystals were dissolved in anhydrous dimethylsulfoxide (DMSO) or *N,N*-dimethylformamide (DMF) to prepare perovskite precursor solution with different concentrations. The perovskite precursor solution (10 μ L, 30 mg mL^{-1} , DMSO) was first added to a silicon micropillar template with asymmetric wettability. Then a cleaned Si or quartz glass substrate was pressed on it to form a sandwiched assembly system, which was moved to a vacuum oven at 75 °C for 24 h until the solution evaporated completely. On peeling off the micropillar template, the perovskite single-crystalline microwire arrays on the target substrate can be obtained. To prepare polycrystalline films, the perovskite precursor solution (50 μ L, 200 mg mL^{-1} , DMF) was firstly spin-coated onto the substrate (3000 rpm, 45 s), and then transferred to a hot plate at 70 °C for 10 min.

2.3. Characterization

Single-crystal X-ray diffraction (SCXRD) analyses were performed on a Bruker SMART APEX II CCD diffractometer with $Mo\text{ K}\alpha$ ($\lambda = 0.71073\text{ \AA}$) radiation at 170 K. The crystal structure was solved by direct methods using SHELXs and refined on F^2 by full-matrix least-squares techniques using the program SHELXTL within the Olex2 crystallographic package. The XRD patterns were measured on an X-ray diffractometer (Bruker, D8 focus, Germany) with monochromatized $Cu\text{ K}\alpha$ radiation ($\lambda = 1.5406\text{ \AA}$). Rocking curves were recorded using a high-resolution X-ray diffractometer (Rigaku Corporation, SmartLab, Japan). The ultraviolet-visible-near infrared ray (UV-vis-NIR) transmittance spectrum was collected on a UV-vis-NIR spectrophotometer (Agilent, Cary 7000, USA) and a Fourier transform infrared spectrometer (Varian, Excalibur 3100, USA). The circular dichroism (CD) spectra were recorded using a circular dichroism spectrometer (JASCO, J-815, Japan). The morphology and elemental distribution were characterized using a scanning electron microscope (Hitachi, S-4800, Japan). The chemical state of elements was investigated using an X-ray photoelectron spectroscope (Thermo Fisher, ESCALAB 250Xi, UK). Thickness measurements were performed on an atomic force microscope (Bruker, ICON2-SYS, Germany). Transmission electron microscopy (TEM) and selected area electron diffraction (SAED) characterization was carried out using a transmission electron microscope (JEOL, 2100, Japan). The grazing-incidence wide-angle X-ray scattering (GIWAXS) patterns were measured using a WZX-SAXS/WAXS system (Xenocs, XEUS, France). The static CA was investigated using a contact angle system (Dataphysics, OCA20, Germany).

2.4. Calculation details

The Vienna *ab initio* simulation package (VASP) was employed to perform the density functional theory (DFT) calculations

based on the frozen-core all-electron projector augmented wave (PAW) method. The generalized gradient approximation (GGA) of the Perdew–Burke–Ernzerhof (PBE) formulation was selected to describe the exchange–correlation energy. For the plane wave, a 400 eV kinetic energy cutoff with a Gaussian smearing of 0.1 eV in width was adopted. For structural optimization, the Monkhorst–Pack k -point grid with a resolution of 0.2 Å⁻¹ was selected to sample the reciprocal space. Considering the van der Waals (vdW) interactions, the DFT-D3 correction with Becke–Johnson damping was adopted during calculations. For electronic properties, the convergence criterion was set to 1.0×10^{-6} eV in energy and 0.005 eV Å⁻¹ in force.

2.5. Nonlinear optical measurements

Powder second-order NLO properties were examined using the Kurtz–Perry method under 2090 nm Q-switched laser irradiation. The SHG intensities were semi-quantitatively determined by using AgGaS₂ with the same sieved particle size as the reference. The SHG measurements of microwires and films were performed using a confocal microscope with a Ti:sapphire femtosecond pump (Spectra physics Mai Tai, <150 fs, 80 MHz, 740–900 nm). A 50× objective lens (NA = 0.80) was used to focus the laser onto a spot with a diameter of about 4 μm. SHG mapping was performed using an MStarter 100 confocal microscope (Metatest Optoelectronic). The linear

polarization direction of the incident pump was tuned with a half-wave plate.

3. Results and discussion

3.1. Crystal structure analysis

To obtain the symmetry-breaking perovskite structure required for SHG, colorless plate crystals of 0D chiral perovskites (*R*/*S*-MBA)₄Bi₂Cl₁₀ were synthesized by a temperature-lowering crystallization method (see the Experimental section and Fig. S1†). Briefly, chiral *R*-(+)- or *S*-(−)-α-methylbenzylamine (*R*- or *S*-MBA) and Bi₂O₃ with a 4:1 stoichiometry were dissolved in aqueous HCl solution, and then the crystal precipitated with the decrease of temperature. To reveal the chiral crystal structure, SCXRD was performed, and the crystallographic data are listed in Tables S1–3.† The powder XRD patterns of (*R*/*S*-MBA)₄Bi₂Cl₁₀ were consistent with the simulated ones and proved the high-phase purity, as presented in Fig. 1a. Both (*R*/*S*-MBA)₄Bi₂Cl₁₀ crystallize in the Sohncke space group of monoclinic *P*2₁ with identical unit cell parameters, suggesting non-centrosymmetric and polar crystal structures. Scanning electron microscopy (SEM) images of the exfoliated crystal show smooth surfaces, confirming the high crystallinity of the as-prepared perovskite crystals (Fig. S1†). Fig. 1b, c and Fig. S2† present the crystal structures of (*R*/*S*-MBA)₄Bi₂Cl₁₀.

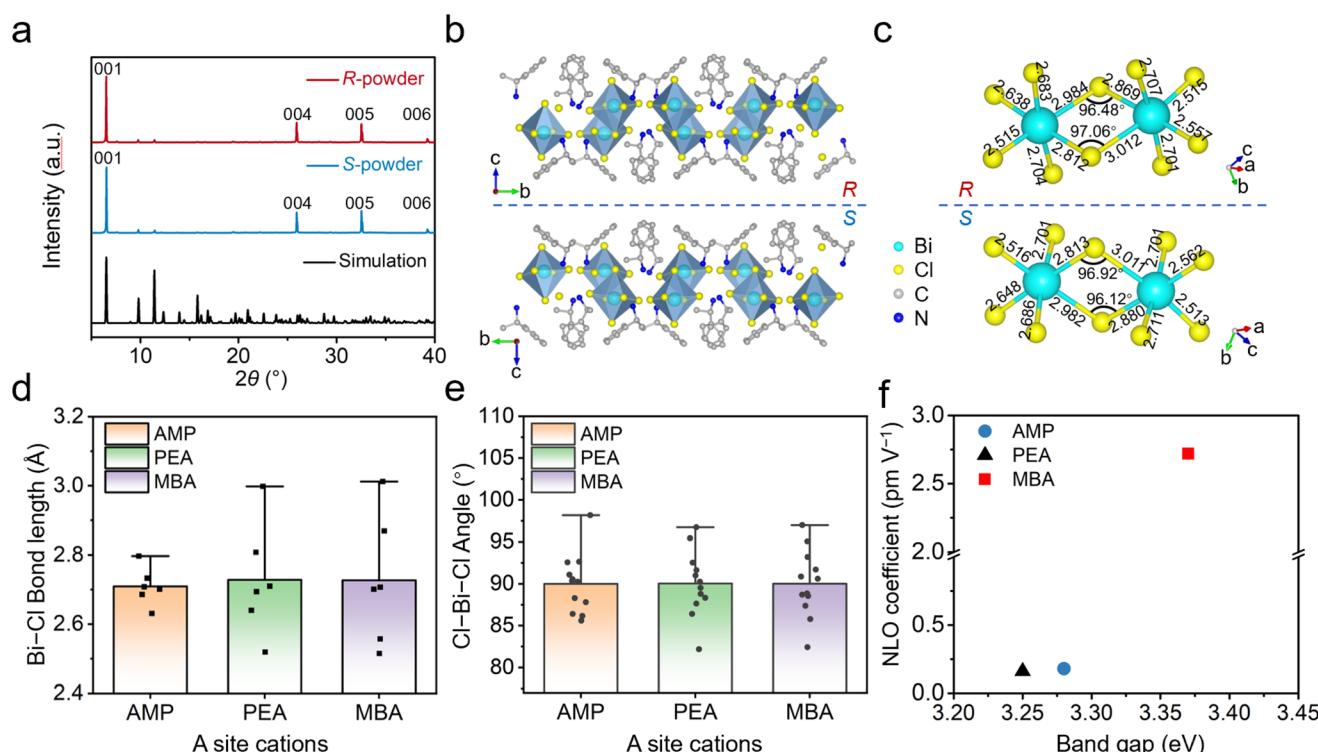


Fig. 1 (a) XRD patterns of (*R*/*S*-MBA)₄Bi₂Cl₁₀ powder. (b) Crystal structures of the enantiomers of (*R*/*S*-MBA)₄Bi₂Cl₁₀ along the *a*-axis. Hydrogen atoms of chiral ammonium are omitted for clarity. (c) The $[Bi_2Cl_{10}]^{4-}$ building block composed of two edge-connected octahedra in (*R*/*S*-MBA)₄Bi₂Cl₁₀. Statistics of (d) Bi–Cl bond lengths, (e) Cl–Bi–Cl angles, and (f) band gaps and NLO coefficients of perovskites with different A-site cations.

Obviously, these two enantiomers are a mirror image of each other (Fig. 1b). With the introduction of large MBA^+ cations, these two 0D perovskites are composed of isolated $[\text{Bi}_2\text{Cl}_{10}]^{4-}$ edge-sharing dimers, maintaining the electrical neutrality of the system. The $-\text{NH}^{3+}$ groups of the chiral MBA^+ cations around the $[\text{Bi}_2\text{Cl}_{10}]^{4-}$ building blocks interact with Cl^- ligands by the $\text{N}-\text{H}\cdots\text{Cl}-\text{Bi}$ hydrogen bond to stabilize the structure (Fig. S3†), resulting in chirality transfer and the asymmetric tilt and distortion of BiCl_6 octahedra. The inhomogeneous $\text{Bi}-\text{Cl}$ bond lengths and $\text{Bi}-\text{Cl}-\text{Bi}$ bond angles observed in those distorted octahedra are 2.515–3.012 Å and 96.48–97.06° for $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$, and 2.513–3.011 Å and 96.12–96.92° for $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$, respectively (Fig. 1e and f). It has been reported that the A-site cations have a significant effect on the structure, properties, and device performance.⁴⁶ The subtle distortions from one structure to the next could be behind the observed differences in optical properties. To characterize the effects of A-site cations on the deformation of octahedra, we compared the metal–halide ($\text{Bi}-\text{Cl}$) bond lengths and halide–metal–halide ($\text{Cl}-\text{Bi}-\text{Cl}$) bond angles in the octahedra of 0D perovskite structures with different A-site cations, 2-picolyamine (AMP),³⁵ *R*-1-(4-fluoro)phenylethylamine (PEA),⁴⁷ and *R*-MBA (MBA) in our system, as presented in Fig. 1d, e and Fig. S4.† Obviously, the average values of $\text{Bi}-$

Cl bond lengths and $\text{Cl}-\text{Bi}-\text{Cl}$ bond angles tend to be uniform in three octahedra, while the value distribution of MBA octahedra is more dispersed and more deviated from that of the perfect octahedron. Axial $\text{Cl}-\text{Bi}-\text{Cl}$ angles follow the same distribution rule (Fig. S5†). All these indicate that the MBA -octahedra have the largest distortion level, which is more conducive to the enhancement of nonlinear optical effects. As expected, the perovskite with the MBA spacer in our work exhibits a higher band gap and NLO coefficient discussed in detail below (Fig. 1f). Additionally, we further calculated the degree of octahedral distortion which can be quantified using the following formulas:

$$\lambda_{\text{otc}} = \frac{1}{6} \sum_{n=1}^6 [(l_n - l_0)/l_0]^2 \quad (1)$$

$$\sigma^2 = \frac{1}{11} \sum_{n=1}^{12} (\theta_n - 90^\circ)^2 \quad (2)$$

where l_n and l_0 are the individual and average bond lengths of the $\text{Bi}-\text{Cl}$ bond and θ_n is the $\text{Cl}-\text{Bi}-\text{Cl}$ bond angle. The calculated distortion level of two $[\text{BiCl}_6]^{3-}$ octahedra are $\lambda_{\text{otc}} = 2.9 \times 10^{-3}$, 4.0×10^{-3} , $\sigma^2 = 11.4$, 16.0 for $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and $\lambda_{\text{otc}} = 2.8 \times 10^{-3}$, 4.0×10^{-3} , $\sigma^2 = 11.2$, 15.6 for $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$,

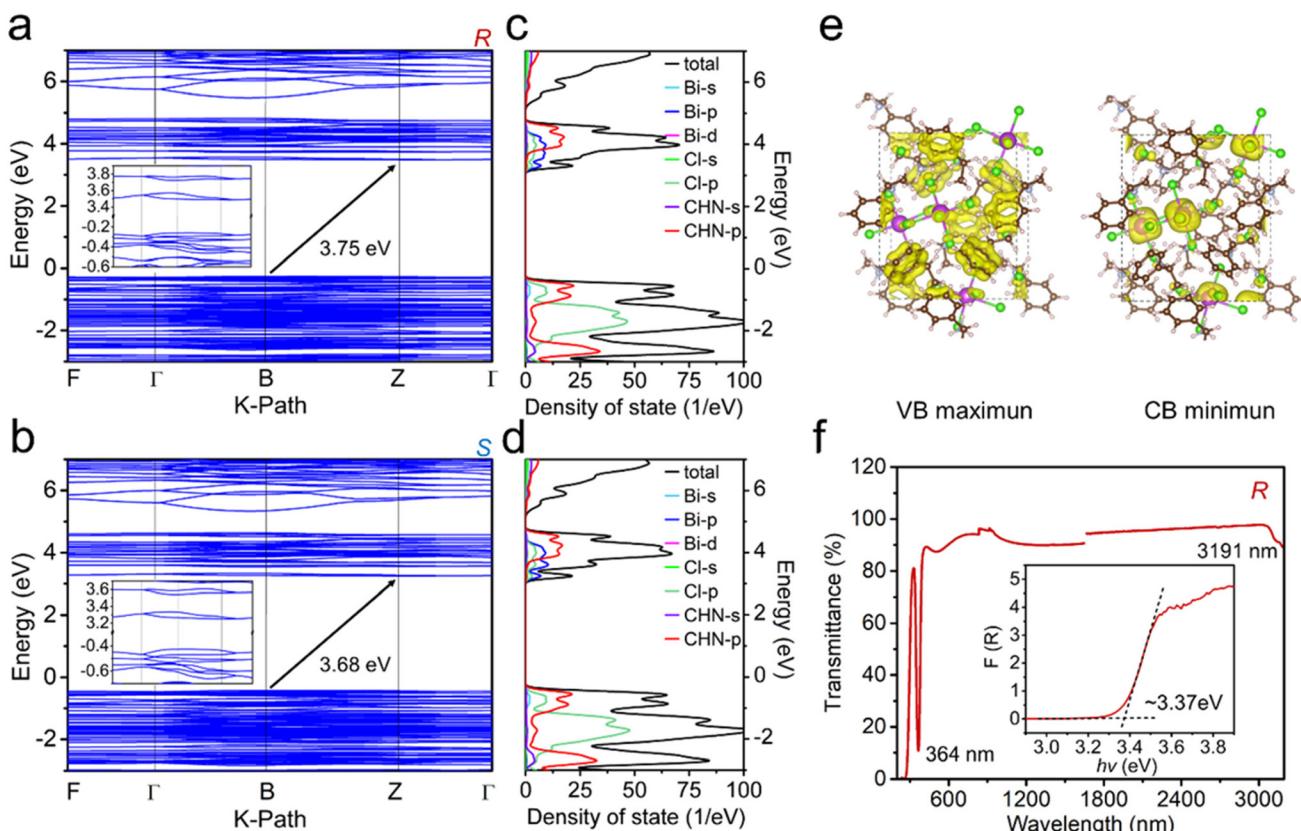


Fig. 2 Calculated electronic band structures of (a) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (b) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$. Calculated partial density of states of (c) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (d) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$. (e) Isosurface plots of the wave functions of the VB maximum and CB minimum of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$. (f) UV-NIR transmittance spectrum and optical bandgap calculated based on the Tauc plot (inset) of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$.

which are larger than those of the most reported perovskites.⁴⁸ This loss of structural inversion symmetry and the high octahedral distortion level could effectively facilitate material polarization, thus contributing to the large SHG coefficient.

To further reveal the electronic properties of the $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ chiral perovskite, the band structures and partial densities of states (PDOS) have been calculated based on DFT. As shown in Fig. 2a and b, the valence band (VB) maximum is located at B points of the Brillouin zone, while the conduction band (CB) minimum is located at the Z points, indicating an indirect bandgap characteristic of both $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$. The calculated bandgap energies of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ are 3.75 eV and 3.68 eV, respectively, which are slightly higher than the experimentally measured value of 3.37 eV for both $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ based on the Tauc plot (Fig. 2f and Fig. S6†), which is probably due to the exclusion of the spin-orbit coupling effect in the self-consistent field (SCF) calculations.⁴⁹ The PDOS around the Fermi levels in Fig. 2c and d reveals that the VB maximum is primarily composed of the CHN-p orbital, while the CB minimum is mainly contributed by the Bi-p and Cl-s orbitals. It means that both anionic $[\text{Bi}_2\text{Cl}_{10}]^{4-}$ groups and R/S-MBA organic cations have a significant influence on the band gap energy, as well as the optical properties of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$. Moreover, the isosurface plot of the wave functions of the VB maximum also

shows that the charge density is mainly distributed on R/S-MBA organic cations, while the CB minimum is mainly distributed on $[\text{Bi}_2\text{Cl}_{10}]^{4-}$ groups (Fig. 2e and Fig. S7†). The UV-vis-NIR transmittance spectra of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ show a very wide window with a high transparency range from 364 nm to 3191 nm, indicating that it could be effectively pumped and applied from the visible light to near-infrared region (Fig. 2f and Fig. S6†).

3.2. Fabrication and characterization of chiral perovskite single-crystalline microwire arrays

To achieve reliable processing technology to meet the large-scale integration of NLO devices, we employed a capillary-bridge confined technique to manipulate the nucleation and growth of perovskite crystals, patterning highly ordered assembly microwire arrays with high crystallinity (see the Experimental section and Fig. S8†). Briefly, by introducing the asymmetric-wettability micropillar templates with hydrophilic tops and hydrophobic sidewalls (Fig. S9†), the perovskite precursors will be confined between the hydrophilic pillar tops and the substrate to form a sandwiched assembly system; thus the nucleation site can be precisely controlled. Simultaneously, the asymmetric modification steers a directional dewetting process, determining the directional growth of the assembly of microwire arrays. As the solvent evaporated

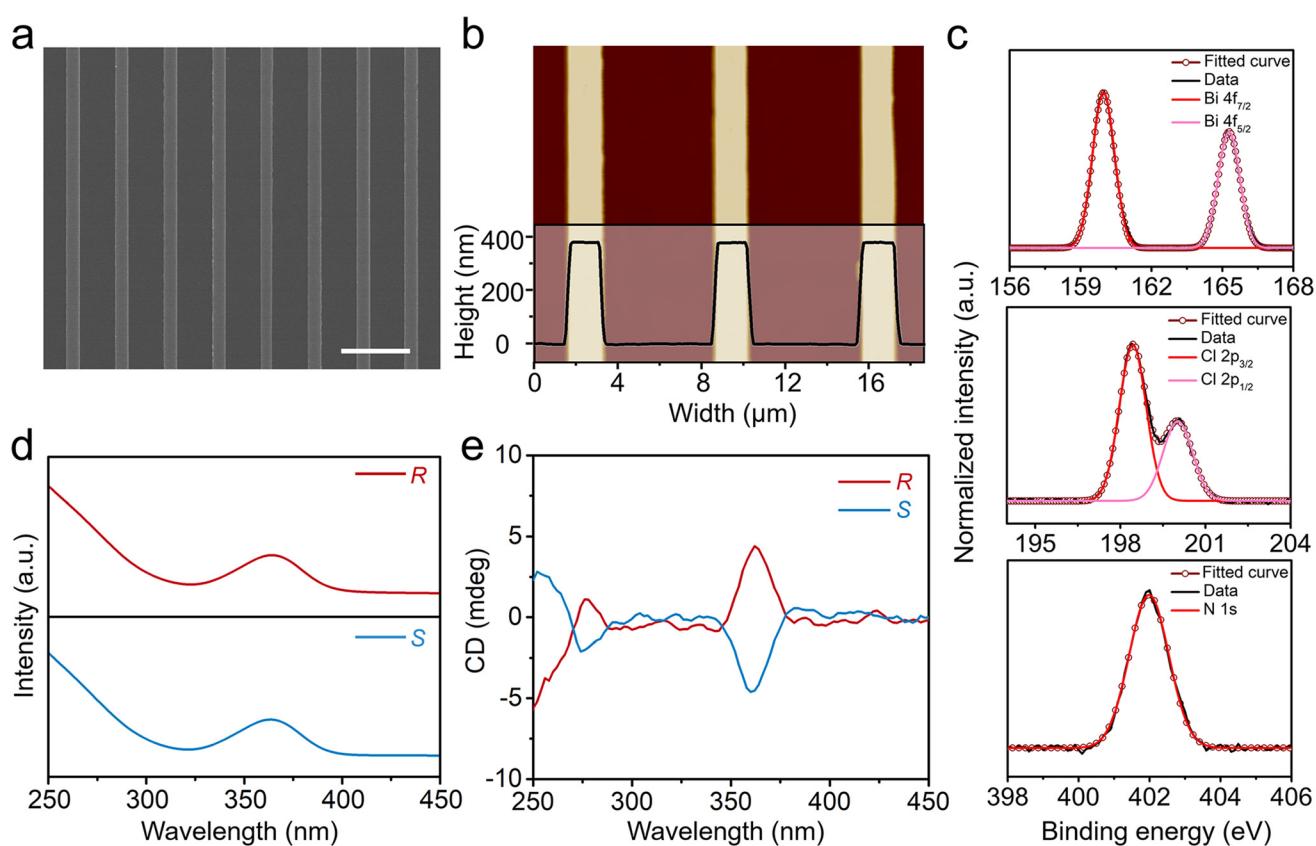


Fig. 3 (a) SEM image and (b) AFM image of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. Scale bar in (a): 10 μm . (c) Absorption and (d) CD spectra of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. (e) XPS spectra of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays showing Bi 4f, Cl 2p, and N 1s regions.

completely, the lead-free chiral perovskite microwire arrays of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ with high crystallinity and a pure crystallographic orientation were fabricated.

To study the morphology and surface roughness of the as-fabricated chiral perovskite NLO microwire arrays, SEM and atomic force microscopy (AFM) were performed. As shown in Fig. 3a and Fig. S10,† the SEM images present high-quality perovskite microwire arrays with a precise location, a strictly isometric arrangement, and homogeneous size over a large area. AFM images further illustrate the smooth surface and uniform size of microwires with a height of about 370 nm, a width of about 1.9 μm , and an adjacent distance of about 5.1 μm (Fig. 3b and Fig. S11†). The elemental components and distribution of the chiral perovskite microwires were analyzed by X-ray photoelectron spectroscopy (XPS) and energy dispersive spectroscopy (EDS). The XPS results exhibit two characteristic peaks in the Bi 4f region with binding energies of 160.0 eV for $4f_{7/2}$ and 165.3 eV for $4f_{5/2}$, two characteristic peaks in the Cl 2p region with binding energies of 198.4 eV for $2p_{3/2}$ and 200.0 eV for $2p_{1/2}$, and one characteristic peak in the N 1s region with a binding energy of 402.0 eV, revealing the stable chemical state of those elements within microwires (Fig. 3c and Fig. S12†). EDS elemental mappings in Fig. S13† further confirm the presence of Bi, Cl, C, and N elements, which are uniformly distributed in those microwires. The EDS result shows that the atomic ratio of Bi/Cl is 1/4.97, which is consistent with the stoichiometry of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ (Bi/Cl, 1/5). We also studied the optical and chiroptical properties of the chiral

perovskite microwire arrays. As shown in Fig. 3d, the UV-vis absorption of both $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays shows an absorption peak at 364 nm. The CD spectrum of the $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays in Fig. 3e exhibits obvious CD signals with bisignated features different from those of the corresponding chiral *R/S*-MBA precursors,⁵⁰ which confirms the intrinsic chirality and non-centrosymmetric polar crystal structures of the $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays.

As the NLO response depends on the anisotropy structure and crystallinity of the perovskite crystals,⁵¹ we further carried out XRD, GIWAXS, TEM, and SAED measurements to clarify the crystallinity and crystallographic orientation of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. The XRD patterns of the microwire arrays are presented in Fig. 4a, b and Fig. S14†; they show periodic sharp diffraction peaks that can be assigned to the $(00l)$ planes, reflecting that the (001) plane is parallel to the substrate. As shown in the insets, the rocking curves of (001) peaks show a sharp arc-like feature with the full width at half maximum (FWHM) as low as 0.2°, indicating the high crystallinity of the microwire arrays. The GIWAXS patterns in Fig. 4c and d show sharp discrete diffraction spots, further confirming the high-crystallinity (001) -oriented $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires along the out-of-plane direction. TEM images illustrate that the microwires are well-shaped with sharp edges and smooth surfaces, and the corresponding SAED patterns exhibit sharp diffraction spots, verifying the single crystallinity and the preferential growth along the (010) direction of those chiral perovskite microwires (Fig. 4e and f). The predicted $(R/S$

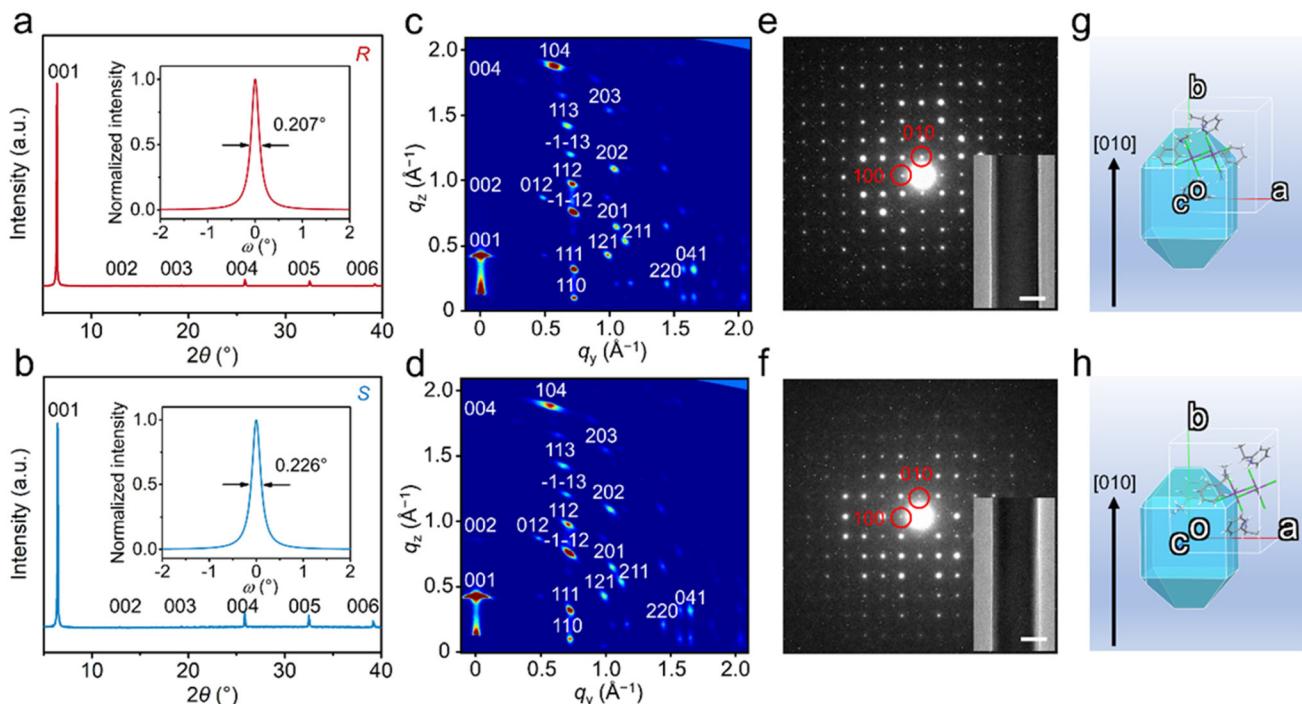


Fig. 4 XRD patterns and rocking curves corresponding to the (001) plane (inset) of (a) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (b) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. GIWAXS images of (c) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (d) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays, verifying the (001) out-of-plane direction. TEM images and SAED patterns of (e) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (f) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires suggest the smooth surface of microwires with a growth direction of $[010]$. Scale bar: 1 μm . Simulated BFDH morphology of (g) $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and (h) $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ crystals.

$(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ crystal morphology based on Bravais–Friedel–Donnay–Harker (BFDH) calculations is presented in Fig. 4g and h, revealing the dominant growth direction along the [010] zone axis, which is in line with the SAED results.

3.3. The second-order nonlinear optical properties

To assess the nonlinear optical properties of the chiral perovskites featuring a non-centrosymmetric structure, we preliminarily evaluated the SHG signals of polycrystalline powder samples using the Kurtz–Perry method with a pump wavelength of 2090 nm, and the sieved AgGaS_2 crystals serve as references. As shown in Fig. S15,† $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ polycrystalline samples exhibit remarkable SHG responses, which are estimated to be $0.04 \times \text{AgGaS}_2$ with the particle size ranging from 150 to 200 μm . Since the intensity of the SHG response is proportional to the square of the second-order NLO coefficient, it can be estimated that the effective NLO coefficient of the title compound is approximately 2.72 pm V^{-1} under a 2090 nm laser. This value is larger than those of most of the similar organic–inorganic hybrid perovskites reported recently, demonstrating its potential applications in the field of infrared nonlinear optics (Table S4†).^{27,28,31} Then we studied the second-order NLO properties of the as-fabricated chiral perov-

kite single-crystalline microwire arrays in detail utilizing a home-built femtosecond laser setup (Fig. S16†). Fig. 5a illustrates a diagram of the SHG process in those perovskite microwires. As shown in Fig. 5b, under laser irradiation with a wavelength of 800 nm, a clear set of response signals with a central wavelength of half the pump wavelength can be observed, corresponding to the expected second-order nonlinear process. The power-dependent SHG signals in Fig. 5b can be well fitted linearly on a logarithmic scale with slopes of 1.95 for $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and 1.96 for $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ (Fig. 5c), displaying a quadratic relationship between the SHG intensity and the incident laser power, which confirms that the strong visible signals are generated from the second-order NLO process. Fig. 5d depicts the SHG intensity mapping image of an individual $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire, which clearly demonstrates the high quality and homogeneity of the as-fabricated single-crystalline microwire arrays. To evaluate the stability of the as-prepared microwire arrays, SHG spectra of the $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire were collected every minute with laser on. No obvious degradation was observed after continuous radiation, indicating the high stability of the perovskite microwire arrays (Fig. S17†). To verify the high-efficiency SHG based on the microwire arrays, we also performed power-dependent SHG measurements on their polycrystalline film counterpart. As

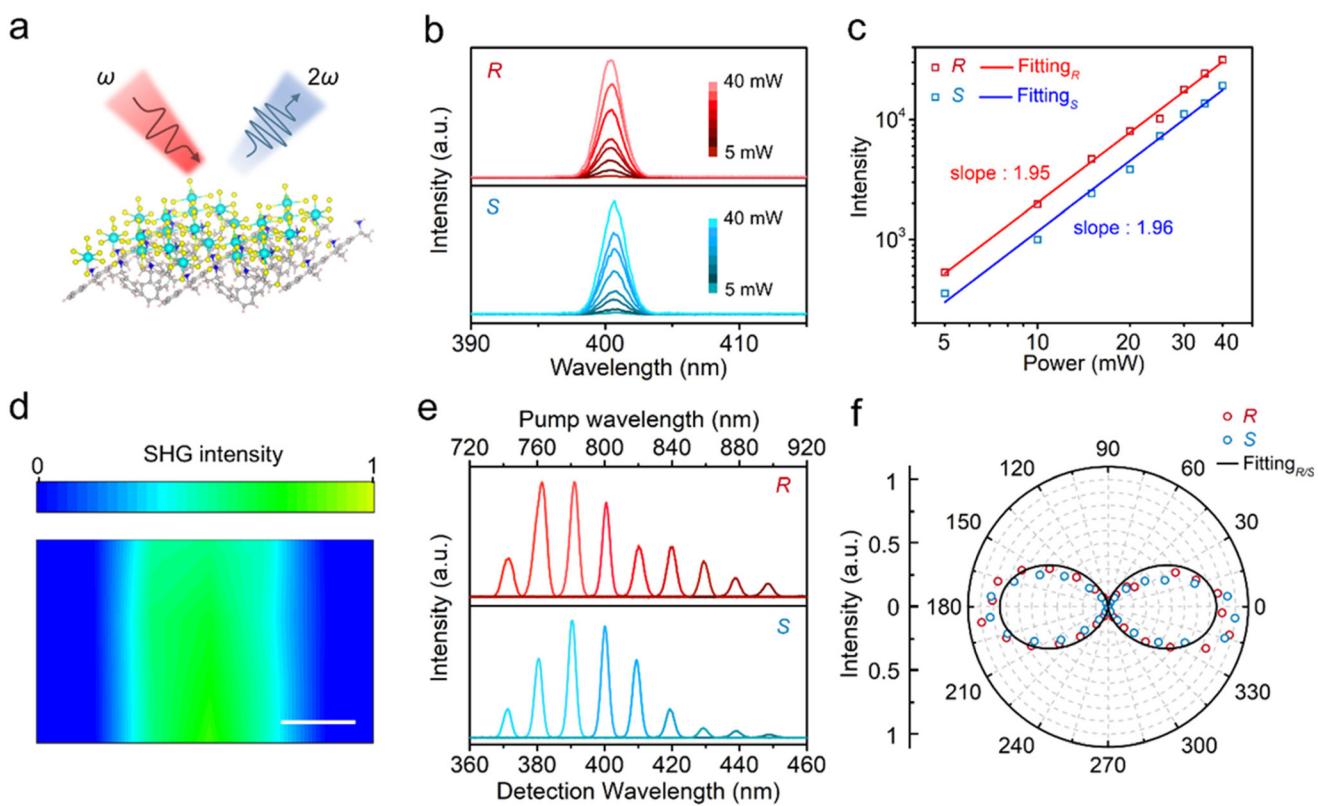


Fig. 5 (a) Schematic diagram of the SHG process. (b) Power-dependent SHG intensity of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays with a pump wavelength of 800 nm. (c) The logarithmic plot of SHG intensity as a function of the incident power corresponding to (b). (d) SHG mapping of a typical $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire. Scale bar: 1 μm . (e) Wavelength-dependent SHG intensity of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. (f) Normalized polarization-dependent SHG intensity of $(R/S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwire arrays. The fitted curves of $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ and $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ coincide and have been indicated with a black line.

expected, the $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ film shows a weaker SHG response compared to the single-crystalline microwire under the same excitation conditions (Fig. S18a–c†). The formula for SHG conversion efficiency $\eta_{2\omega}$ is shown below:

$$\eta_{2\omega} = P_{2\omega}/P_{\omega}^2 \quad (3)$$

where $P_{2\omega}$ is the SHG intensity and P_{ω} is the laser power; the stronger SHG intensity implies a higher SHG conversion efficiency of the microwire arrays, which can be attributed to the suppressed grain boundaries and lower propagation loss (Fig. S19†).

The wavelength-dependent SHG intensity of $(R\text{-S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires measured in the tuning pump wavelength range from 740 to 900 nm with a step of 20 nm is presented in Fig. 5e, showing the strong SHG signals at the second harmonic wavelength from 370 to 450 nm. Both $(R\text{-S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires show no difference in the SHG signal changes at different second-harmonic wavelengths and exhibit the highest SHG intensity at 390 nm, which is mainly due to the strong self-absorption and phase mismatch.^{25,31} Since the inherent structural asymmetry usually implies an anisotropic SHG response, we further investigate the linear polarization-resolved SHG intensity of $(R\text{/S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires. The polarization direction of the incident laser can be changed with the azimuthal angle from 0° to 360° by tuning the half-wave plate, where 0° corresponds to that the polarization direction of the incident laser is parallel to the long axis microwire, the crystallographic *b*-axis. Fig. 5f presents the polarization-dependent SHG intensity of $(R\text{/S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires, which was fitted with a $\cos^4\theta$ function, showing distinct anisotropy characteristics. The SHG signals exhibit a typical twofold pattern, with maximum intensity observed in the parallel direction (0°) and minimum intensity observed in the vertical direction (90°). The polarization ratio is calculated using the following formula:

$$\rho = (I_{\max} - I_{\min})/(I_{\max} + I_{\min}) \quad (4)$$

The polarization ratios ρ are calculated to be as high as 0.987 for $(R\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires and 0.992 for $(S\text{-MBA})_4\text{Bi}_2\text{Cl}_{10}$ microwires, which are higher than those of most reported perovskites.^{27,28,31} In addition, compared with their polycrystalline film counterpart with a polarization ratio of 0.272, microwires also display a notably higher SHG anisotropy, demonstrating the high crystallinity, pure crystallographic orientation, and excellent NLO performances of microwires fabricated by capillary-bridge confined techniques (Fig. S18d†).

4. Conclusions

In summary, we synthesized lead-free chiral 0D perovskites $(R\text{-S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ with high distortion levels and successfully constructed high-quality single-crystalline microwire arrays by employing the capillary-bridge confined techniques. The as-prepared chiral $(R\text{/S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ single-crystalline microwire

arrays feature an intrinsic non-centrosymmetric polar crystal structure and exhibit a wide transparent region, excellent SHG conversion efficiency and high polarization ratio of up to 0.992. The simultaneous realization of large nonlinear anisotropy and high-efficiency SHG responses with a broadband transparency window in $(R\text{/S-MBA})_4\text{Bi}_2\text{Cl}_{10}$ single-crystalline microwire arrays would pave the way toward the engineering and development of advanced technological applications in nanophononics and optoelectronics.

Conflicts of interest

There are no conflicts to declare.

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