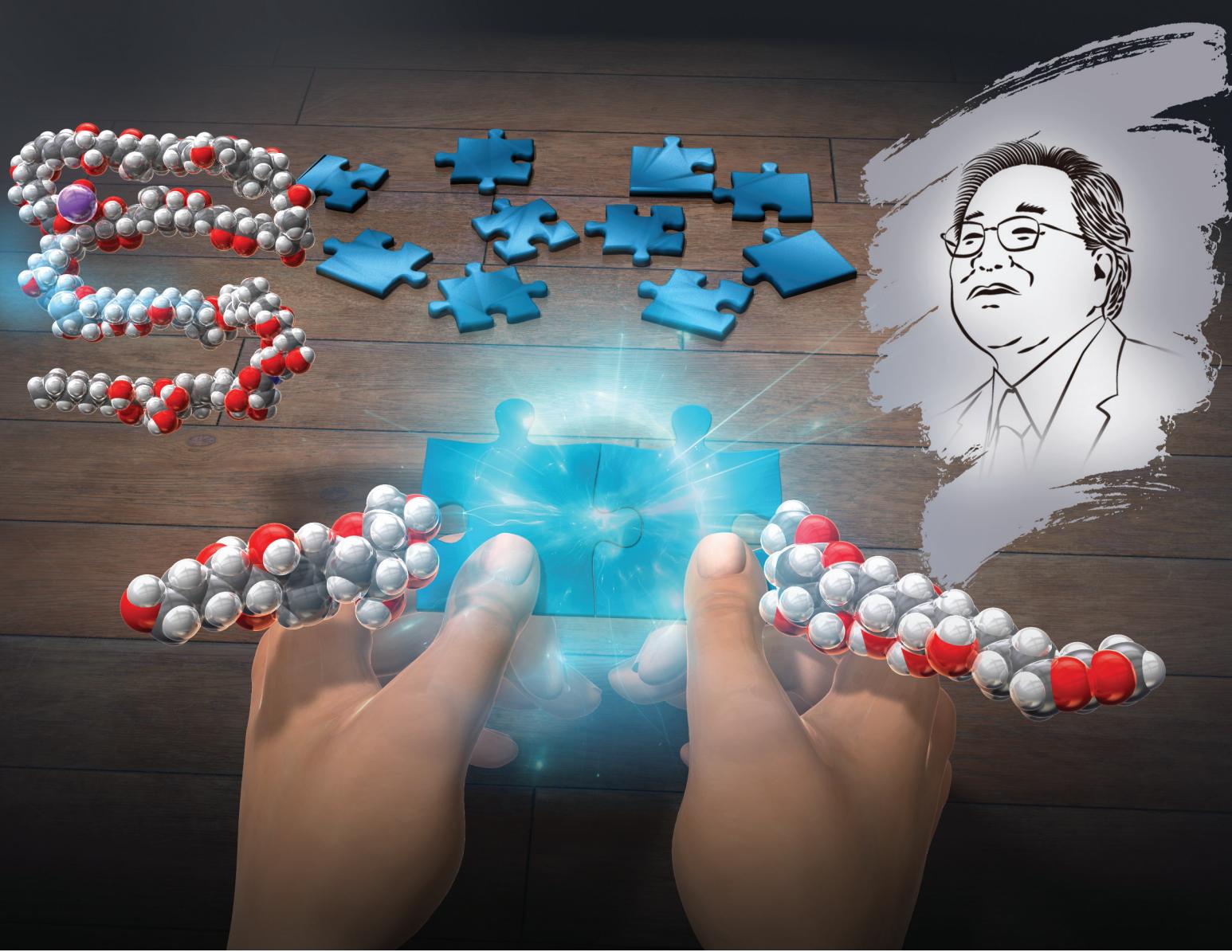


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Relative stereochemical determination of the C61–C83 fragment of symbiodinolide using a stereodivergent synthetic approach†‡

Hiroyoshi Takamura, * Kosuke Hattori, Takumi Ohashi, Taichi Otsu and Isao Kadota

Structural determination is required in the use of marine natural products to create novel drugs and drug leads in medicinal chemistry. Symbiodinolide, which is a polyol marine natural product with a molecular weight of 2860, increases the intracellular Ca^{2+} concentration and exhibits inhibitory activity against cyclooxygenase-1. Seventy percent of the structure of symbiodinolide has been stereochemically clarified. Herein, we report the elucidation of the relative configuration of the C61–C83 fragment, which is among the remaining thirty percent, using a stereodivergent synthetic strategy. We first assigned the relative configuration of the C61–C74 fragment. Two candidate diastereomers of the C61–C74 fragment were synthesized, and their NMR data were compared with those of the natural product, revealing the relative stereochemistry of this component. We then narrowed down the candidate compounds for the C69–C83 fragment from 16 possible diastereomers by analyzing the NMR data of the natural product, and we thus selected eight candidate diastereomers. Stereodivergent synthesis of the candidates for this fragment and comparison of the NMR data of the natural product and the eight synthetic products resulted in the relative stereostructural clarification of the C69–C83 fragment. These individually determined relative stereochemistries of the C61–C74 and C69–C83 fragments were connected via the common C69–C73 tetrahydropyran moiety of the fragments. Finally, the relative configuration of the C61–C83 fragment of symbiodinolide was determined. The stereodivergent synthetic approach used in this study can be extended to the stereochemical determination of other fragments of symbiodinolide.

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Introduction

Marine natural products (MNPs) are critical in drug discovery and development.¹ Structural determination of MNPs, including their planar structures and stereochemistries, is a fundamental research topic and a requirement in using MNPs to create novel drugs and drug leads in medicinal chemistry. Remarkable recent advancements in computational chemistry enhanced our ability to determine the configurations of structurally complex natural products.² Thus, over the decades, computational modeling of the ^1H and ^{13}C nuclear magnetic resonance (NMR) chemical shifts of organic compounds resulted in marked increases in availability, accuracy, and

application. Nevertheless, if the target natural product contains conformationally flexible acyclic motifs and/or remote stereogenic centers, a synergistic combination of spectroscopic analysis and chemical synthesis is effective in unambiguous structural elucidation.³

Symbiodinolide (**1**, Fig. 1) is a secondary metabolite that was isolated from the cultured dinoflagellate *Symbiodinium* sp. by Uemura *et al.*⁴ Natural product **1** increased the intracellular free Ca^{2+} concentration at 7 nM against IMR-32 human neuroblastoma cells and exhibited cyclooxygenase-1 inhibitory activity (65% inhibition) at 2 μM . Furthermore, **1** ruptured the tissue surface of the flatworm *Amphiscolops* sp., which is a host organism containing the symbiotic dinoflagellate *Symbiodinium* sp., at 2.5 μM . In an isolation study, detailed 2D NMR analysis and chemical degradation of **1** revealed its planar structure and partial stereochemistry. The structure of **1** displays a molecular weight of 2860, 61 chiral centers, and an extended carbon skeleton that is highly functionalized with oxygen atoms. The large, complicated chemical structure of **1** hampers its configurational assignment. Therefore, our group investigated the chemical synthesis of each fragment of **1**,

Department of Chemistry, Graduate School of Natural Science and Technology, Okayama University, 3-1-1 Tsushima-naka, Kita-ku, Okayama 700-8530, Japan.
E-mail: takamura@cc.okayama-u.ac.jp

†We dedicate this work to the memory of Prof. Daisuke Uemura, who sadly passed away on 13th April 2021.

‡ Electronic supplementary information (ESI) available. See DOI: <https://doi.org/10.1039/d3ob01420g>



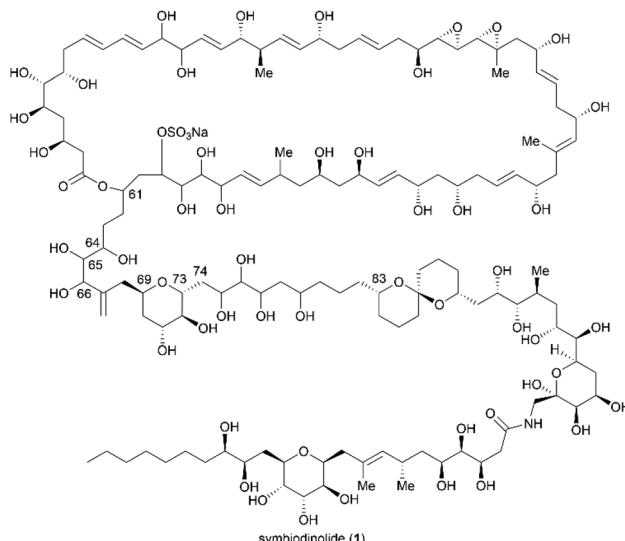


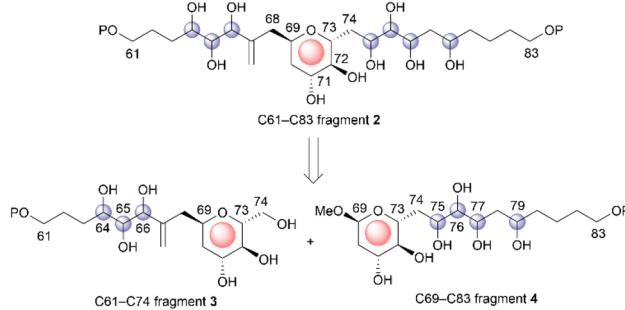
Fig. 1 Structure of 1.

including the stereoisomers, leading to the stereochemical clarification of the moieties corresponding to 70% of the structure of 1.⁵ Herein, we report the stereodivergent synthesis⁶ of the C61–C74 and C69–C83 fragments (3 and 4), resulting in the determination of the relative stereochemistry of the C61–C83 fragment 2 that is among the remaining 30%.

Results and discussion

Strategy for the stereochemical determination of fragment 2

The relative configuration of the tetrahydropyran domain at positions C69–C73 of 1 was verified *via* the $^3J_{H,H}$ coupling constants and nuclear Overhauser effect (NOE) observations.⁴ The large coupling constants ($^3J_{71,72}$ and $^3J_{72,73} = 8.3$ Hz) indicate that H-71, H-72, and H-73 are oriented axially. The stereochemistry at the C69 position was elucidated using the NOEs of Ha-68/H-71 and Ha-68/H-73. To determine the stereostructure of fragment 2, we divided it into two fragments, *i.e.*, fragments 3 and 4 (Scheme 1). The C69–C73 tetrahydropyran ring, with a known relative stereochemistry, is a common structure in frag-



Scheme 1 Strategy for the configurational determination of fragment 2.

ments 3 and 4. Therefore, if the relative configurations of 3 and 4 are established individually, these two stereostructures may be connected by this common tetrahydropyran moiety to finally reveal the relative stereochemistry of fragment 2.

Fragment 3

The relative configurations of the three contiguous oxymethine stereogenic centers at the C64, C65, and C66 positions of 1 were determined *via* 1H NMR data.⁴ The coupling constants $^3J_{64,65}$ (5.5 Hz) and $^3J_{65,66}$ (7.4 Hz) indicate that the respective relative relationships of C64/C65 and C65/C66 are *syn* and *anti*. The relative stereochemistries at the C66 and C69 positions are not correlated, and thus, we propose two candidate diastereomers, 3a and 3b, for fragment 3, as shown in Fig. 2.

To stereochemically elucidate fragment 3 of 1, we planned to synthesize 3a and 3b and compare the NMR data of these synthetic products with those of natural product 1. In our retrosynthetic analysis, target molecules 3a and 3b may be synthesized *via* regio- and diastereoselective dihydroxylation at the C64 and C65 positions of dienes 5 and 6, respectively (Scheme 2). Alcohols 5 and 6, which are diastereomers at the C66 position, would be formed by coupling alkenyl iodide 7 and aldehyde 8. Aldehyde 8 would be synthesized by introducing a C4 side-chain into tri-O-acetyl-D-glucal (9) as a starting material.

We first examined the stereoselective synthesis of aldehyde 14, which is a branching synthetic intermediate of 3a and 3b (Scheme 3). The reaction of the known tetraacetate 10, which was prepared using 9 in one step,⁷ with 2-[(acetoxyethyl)allyl] trimethylsilane (11)⁸ proceeded in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) as a Lewis acid to yield allylated product 12 as a single diastereomer.⁹ The stereo-

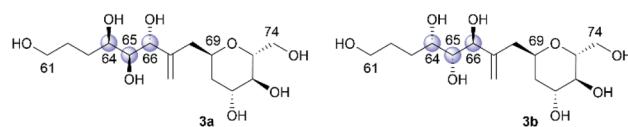
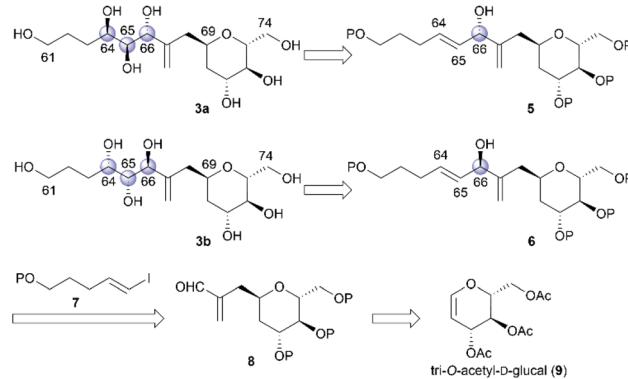
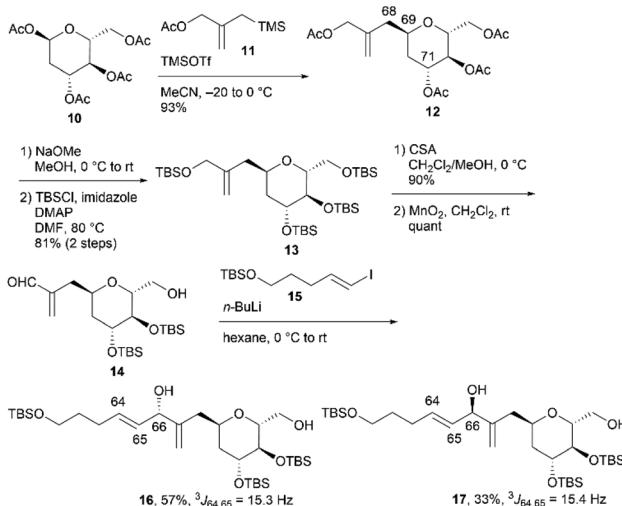


Fig. 2 Candidate diastereomers 3a and 3b for fragment 3.



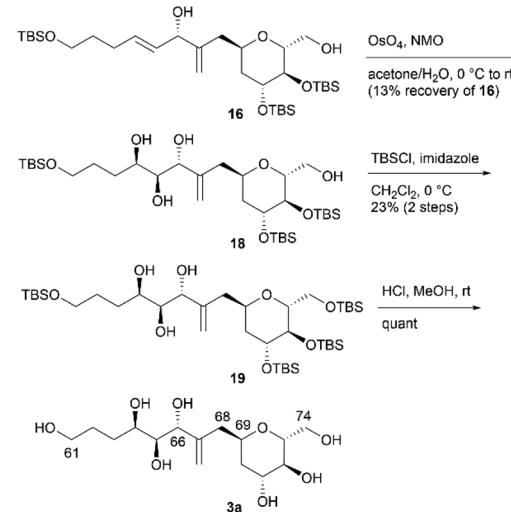
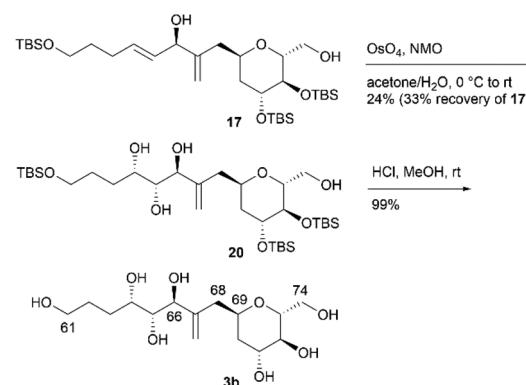
Scheme 2 Retrosynthetic analysis of candidate diastereomers 3a and 3b.



Scheme 3 Synthesis of alcohols 16 and 17.

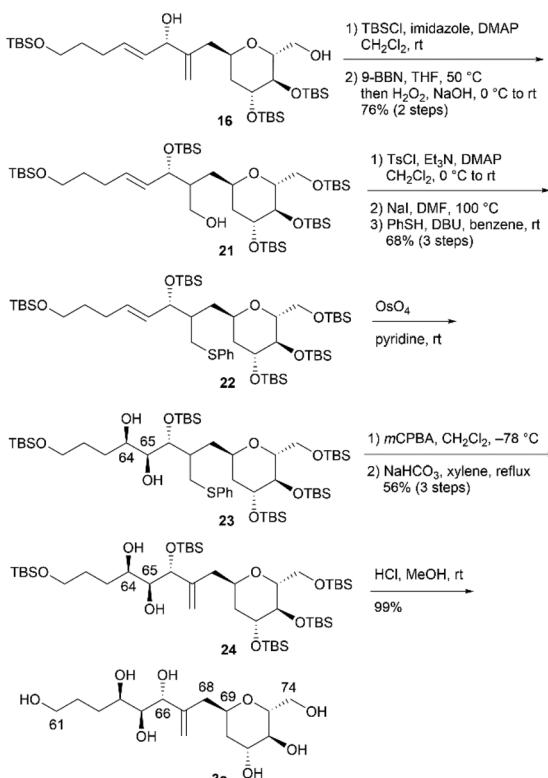
chemistry at the C69 position was confirmed by the NOE correlations between H₂-68 and H-71, which indicate that the introduced side-chain moiety is positioned axially. Global deprotection of **12** *via* methanolysis and subsequent protection of the resulting tetraol using *tert*-butyldimethylsilyl chloride (TBSCl) afforded silyl ether **13** in 81% yield over two steps. The removal of the two primary *tert*-butyldimethylsilyl (TBS) groups of **13** with camphorsulfonic acid (CSA) in CH₂Cl₂/methanol (MeOH), followed by selective oxidation of the allylic alcohol component with MnO₂, produced aldehyde **14**. The coupling reaction of an alkenyl lithium species, which was generated *via* lithium-iodine exchange between alkenyl iodide **15**¹⁰ and *n*-butyllithium (*n*-BuLi) in hexane,¹¹ with aldehyde **14** generated diols **16** and **17** in respective yields of 57% and 33%.¹² The C64/C65 geometries in **16** and **17** were confirmed as *trans* based on the coupling constants (³J_{64,65} = 15.3 Hz for **16** and 15.4 Hz for **17**).

We then attempted to convert **16** to the first target molecule **3a** (Scheme 4). Treatment of allylic alcohol **16** with OsO₄/N-methylmorpholine *N*-oxide (NMO) in acetone/H₂O¹³ produced the desired tetraol **18**, along with its structural isomer, wherein the *exo*-olefin moiety was dihydroxylated.¹⁴ This obtained mixture could not be easily separated, and thus, the primary hydroxy group of **18** was protected with TBSCl/imidazole to furnish tetrakis-TBS ether **19**, which was purified using silica gel column chromatography. The use of AD-mix- α or AD-mix- β ¹⁵ in dihydroxylation did not improve the regioselectivity or chemical yield. The four TBS protective groups of **19** were removed using HCl in MeOH to quantitatively obtain the first target molecule, **3a**. The second target molecule, **3b**, was synthesized using **17** *via* a transformation similar to that employed in synthesizing **3a** using **16** (Scheme 5). Thus, dihydroxylation of the disubstituted alkene moiety of **17** was conducted using OsO₄/NMO¹³ to afford the desired tetraol **20**¹⁴ in 24% yield, along with recovered **17** in 33% yield. When the reaction time was extended, dihydroxylation of the *exo*-olefin

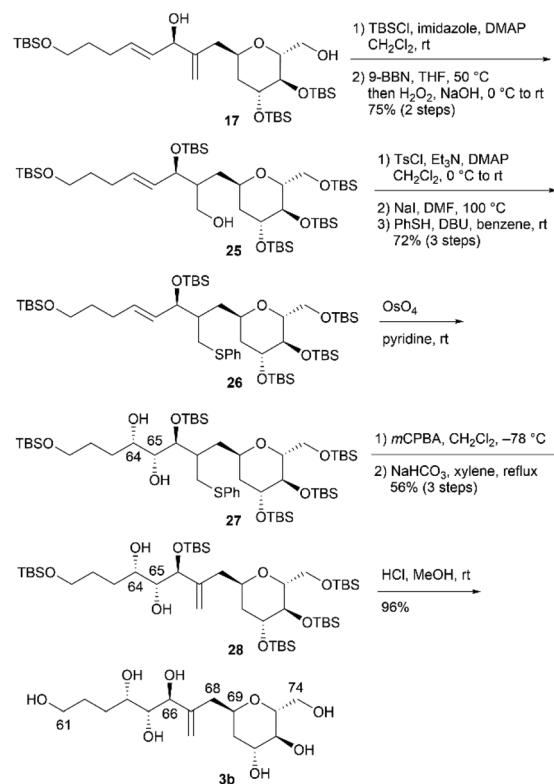
Scheme 4 Synthesis of candidate compound **3a**.Scheme 5 Synthesis of candidate compound **3b**.

component of **20** was observed. Finally, tris-TBS ether **20** was deprotected using HCl in MeOH to furnish the second target molecule, **3b**. Although we obtained candidate compounds **3a** and **3b**, the regioselectivities in the dihydroxylation of **16** and **17** were uncontrolled, and the chemical yields of **18** and **20** were low. Therefore, alternative synthetic routes to **3a** and **3b** were investigated.

In an alternative synthetic route, we temporarily and regioselectively protected the *exo*-olefin moieties of dienes **16** and **17** prior to dihydroxylation. The second synthesis of **3a** is shown in Scheme 6. After protecting diol **16** using TBSCl, regioselective hydroboration of the obtained diene was examined by changing the borane reagent. The use of BH₃-SMe₂ (SMe₂ = dimethylsulfide) or the xylborane was ineffective in the regioselective hydroboration of the *exo*-olefin moiety. When we treated the obtained diene with 9-borabicyclo[3.3.1]nonane (9-BBN) and performed an oxidative work-up, the desired alcohol **21** was obtained in 76% yield over two steps. Tosylation of **21**, iodination of the tosylate with NaI, and treatment of the iodoalkane with thiophenol (PhSH)/1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) afforded sulfide **22** in 68% yield



Scheme 6 Improved synthesis of candidate compound 3a.



Scheme 7 Improved synthesis of candidate compound 3b.

over three steps. Diastereoselective dihydroxylation of alkene 22 with a stoichiometric amount of OsO_4 in pyridine^{13,16} afforded the desired diol 23 and its C64,C65-epimer (dr = 6.2 : 1). The alkene component of 24 was regenerated *via* the following sequence: (1) *meta*-chloroperoxybenzoic acid (*m*CPBA) oxidation of sulfide 23 to the corresponding sulfoxide and (2) elimination of the sulfoxide with NaHCO_3 in xylene at reflux. Diol 24 and its C64,C65-epimer were separated, and 24 was obtained in 56% yield in three steps from 22. Finally, pentakis-TBS ether 24 was deprotected to afford 3a. The respective total number of steps and overall yield of 3a from 16 in the second synthesis, as shown in Scheme 6, were 9 and 29%. Although the number of steps was increased compared to that in the first synthesis (3 steps from 16, as shown in Scheme 4), the overall yield was improved compared to that in the first synthesis (23% from 16, as shown in Scheme 4). We then investigated the second synthesis of 3b using a conversion similar to that shown in Scheme 6 (Scheme 7). TBS protection of diol 17 and subsequent regioselective hydroboration with 9-BBN afforded alcohol 25 in 75% yield over two steps. Tosylation, nucleophilic iodination, and thioetherification afforded sulfide 26. Alkene 26 reacted with OsO_4 in pyridine^{13,16} to produce diol 27 (dr = 4.0 : 1), which was then converted to alkene 28 *via* oxidation with *m*CPBA and elimination of the resulting sulfoxide. Global removal of the TBS protecting groups from 28 provided 3b. As in the case of 3a, although the transformation from 17 to 3b in the second synthesis (9 steps, Scheme 7) was lengthy, the overall yield of the

second synthesis (29%) was higher than that of the first synthesis (24%, Scheme 5).

After obtaining candidate compounds 3a and 3b, we analyzed their 2D NMR spectra. The differences in the ^1H NMR chemical shifts representing the C66, C68, and C69 positions of 1 and the synthetic products 3a and 3b are graphically illustrated in Fig. 3.¹⁷ The chemical shift deviations of 3b were smaller than those of 3a, as the sums of the magnitudes of the differences in the ^1H NMR chemical shifts representing the C66, C68, and C69 positions were 0.23 (3a) and 0.15 (3b). Therefore, the relative configuration of fragment 3 of 1 is that of 3b.

Fragment 4

We then focused on the stereochemical clarification of fragment 4. This fragment contains four unidentified chiral

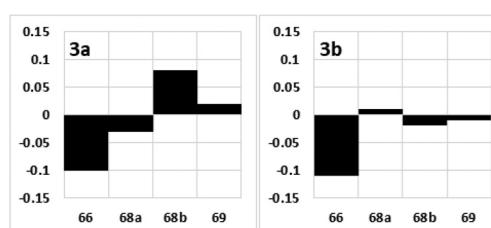


Fig. 3 Deviations in the ^1H NMR chemical shifts of the synthetic products 3a and 3b ($\Delta\delta = \delta_1 - \delta_3$ in ppm) relative to those of 1. The x and y axes show the carbon number and $\Delta\delta$, respectively.



centers at positions C75, C76, C77, and C79, and thus, 16 diastereomers are possible. Prior to commencing the synthetic approach for structural determination, we analyzed the NMR data of natural product **1** to identify candidate compounds. The NMR chemical shifts of positions C75–C77 of **1**, which are reported in an isolation study,⁴ are shown in Table 1. A detailed comparison of these chemical shifts with data reported in the Kishi universal NMR database for 1,2,3-triols¹⁸ clarified the relative relationships of C75–C76/C76–C77 as *syn/syn* or *anti/anti*. Therefore, eight hexaols, **4a**–**4h**, as shown in Fig. 4, were used as candidate compounds for fragment 4.

The stereodivergent synthetic strategy⁶ for candidate compounds **4a**–**4h** is outlined in Scheme 8. The reaction of aldehyde **29** and alkyne **30** would give two propargylic alcohols, **31** and **32**, which are stereoisomeric at the C75 position (Scheme 8a). This coupling reaction is the first step in stereodivergence. Respective *trans*- and *cis*-alkenes **33** and **34** could be synthesized *via* the stereoselective reduction of alkyne **31**, which is the second stereodivergence step (Scheme 8b). The dihydroxylation of *trans*-alkene **33** would afford hexaol **4a**, and

Table 1 NMR chemical shifts of **1** in CD_3OD ^a

Position	^1H NMR	^{13}C NMR
75	3.89	69.6
76	3.18	78.5
77	3.95	70.5

^a Data reported in ref. 4. Chemical shifts are reported in ppm with reference to the internal residual solvent (^1H NMR: 3.30 ppm, ^{13}C NMR: 49.0 ppm).

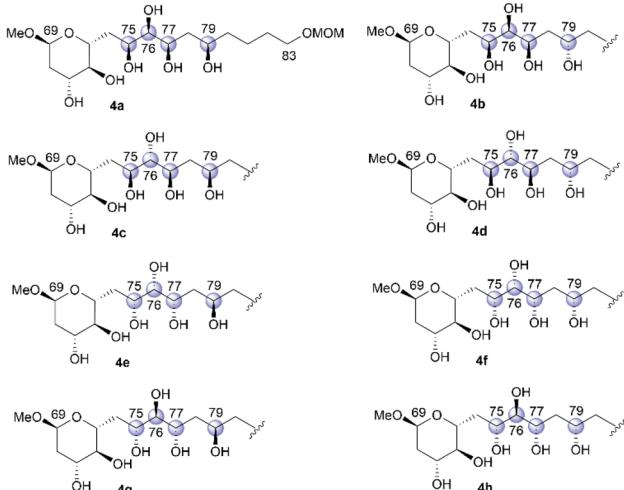


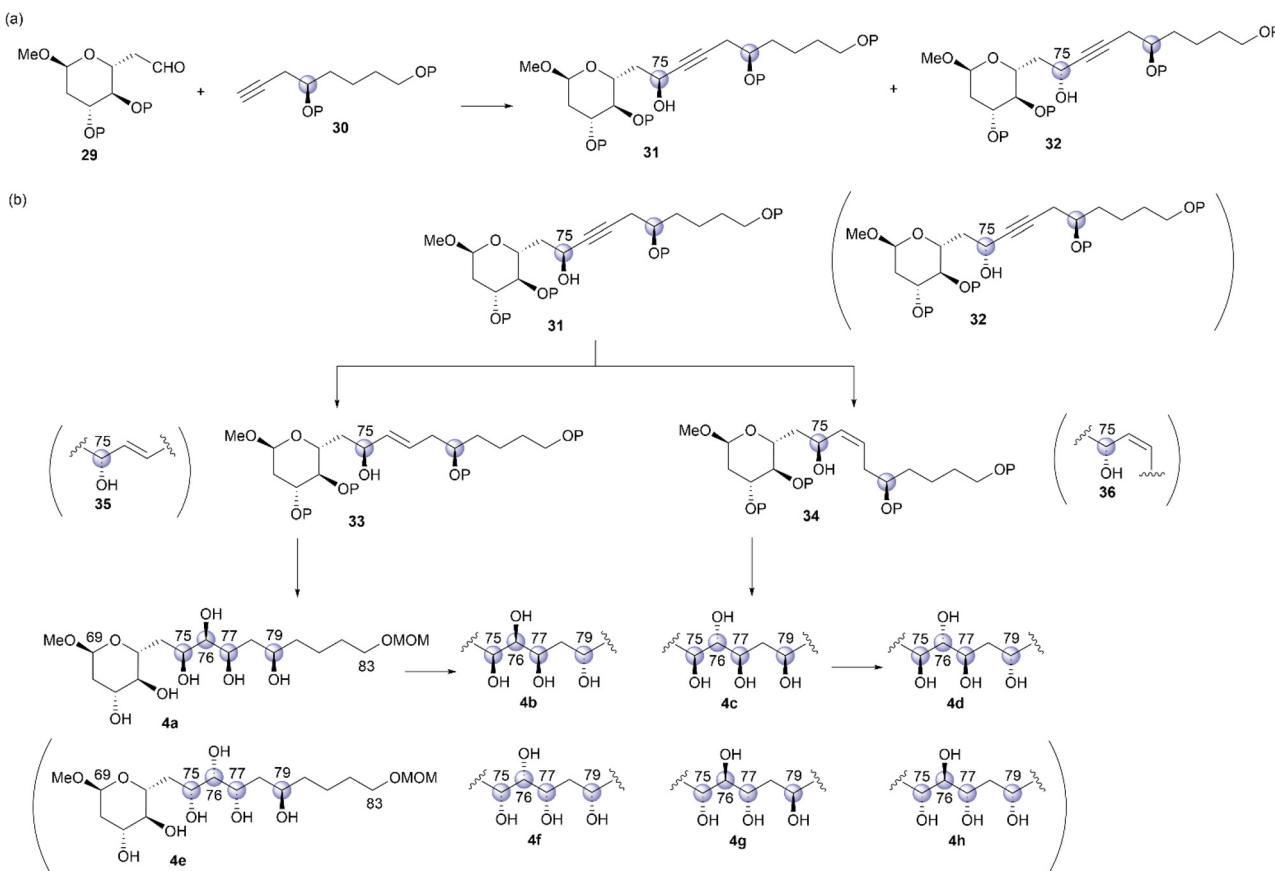
Fig. 4 Candidate diastereomers **4a**–**4h** for fragment 4.

hexaol **4b**, which is the C79-epimer of **4a**, could be synthesized *via* stereoinversion at the C79 position, which is the third stereodivergence step, after the dihydroxylation of **33**. Similarly, hexaols **4c** and **4d**, which are the respective C76-stereoisomers of **4a** and **4b**, could be produced using *cis*-alkene **34**. Furthermore, the other four candidates, **4e**–**4h**, could be prepared using propargylic alcohol **32**, with the respective *trans*- and *cis*-alkenes **35** and **36** as the key synthetic intermediates, *via* a similar transformation. We then commenced the stereodivergent synthesis of candidate compounds **4a**–**4h**.

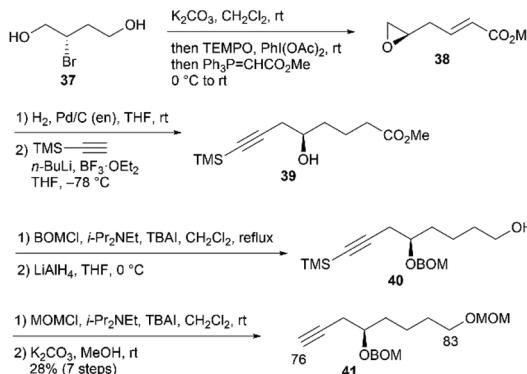
First, we examined the synthesis of alkyne **41**, which corresponds to the C76–C83 fragment (Scheme 9). The synthesis commenced with (*S*)-2-bromobutane-1,4-diol (**37**), which was prepared in two steps using L-aspartic acid.¹⁹ Compound **37** was treated with K_2CO_3 in CH_2Cl_2 to yield a known epoxy alcohol,¹⁹ which underwent oxidation with 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO)/ $\text{PhI}(\text{OAc})_2$ ²⁰ and subsequent Wittig olefination with $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Me}$ in one-pot to afford unsaturated ester **38**. The alkene moiety of **38** was chemoselectively hydrogenated with a Pd/C -ethylenediamine complex catalyst in tetrahydrofuran (THF)²¹ to provide the saturated ester, with the epoxide structure retained. The resulting epoxide was reacted with trimethylsilylacetylene/*n*-BuLi/ $\text{BF}_3\text{-OEt}_2$ ²² (OEt_2 = diethyl ether) to produce alcohol **39**. The hydroxy group of **39** was protected as the benzyloxymethyl (BOM) ether, and the ester group was reduced using LiAlH_4 to yield alcohol **40**. Protection of **40** with methoxymethyl chloride/*N,N*-diisopropylethylamine/tetra-*n*-butylammonium iodide (MOMCl/*i*-Pr₂NEt/TBAI), followed by removal of the trimethylsilyl (TMS) moiety with K_2CO_3 in MeOH, afforded alkyne **41** in 28% yield in seven steps from **37**.

With coupling precursor **41** in hand, we then synthesized aldehyde **44**, which is the coupling partner of **41** (Scheme 10). Deacetylation of the known methyl glucoside **42**,²³ TBS protection of the resulting triol, and selective deprotection of the primary TBS ether provided alcohol **43** in 76% yield over three steps. One-carbon elongation from **43** was conducted *via* a combination of tosylation and cyanation to yield a nitrile, which was reduced with diisobutylaluminum hydride (DIBAL-H) to afford aldehyde **44**. Deprotonation of alkyne **41** using *n*-BuLi and subsequent addition of the resulting anion to aldehyde **44** afforded the desired alcohols **45** and **46** in respective yields of 28% and 57%.²⁴

We then converted **45** to **4a** (Scheme 11). Propargylic alcohol **45** was reduced using Red-Al²⁵ to yield *trans*-allylic alcohol **47**. The *trans*-geometry of **47** was determined based on the observed coupling constant ($^3J_{76,77} = 15.4$ Hz). After protection of **47** with *tert*-butyldimethylsilyl trifluoromethanesulfonate (TBSOTf)/2,6-lutidine, *trans*-alkene **48**, with an allylic siloxy group, was dihydroxylated with OsO_4 in pyridine^{13,16} to afford the desired diol **49** in 28% yield and its C76,C77-epimer in 55% yield (dr = 1 : 2.0). These products could be separated *via* silica gel column chromatography.^{14,26} The hydrogenation of BOM ether **49** with $\text{Pd}(\text{OH})_2/\text{C}$ in MeOH generated triol **50**. The three TBS groups of **50** were removed using tetra-*n*-butyl-

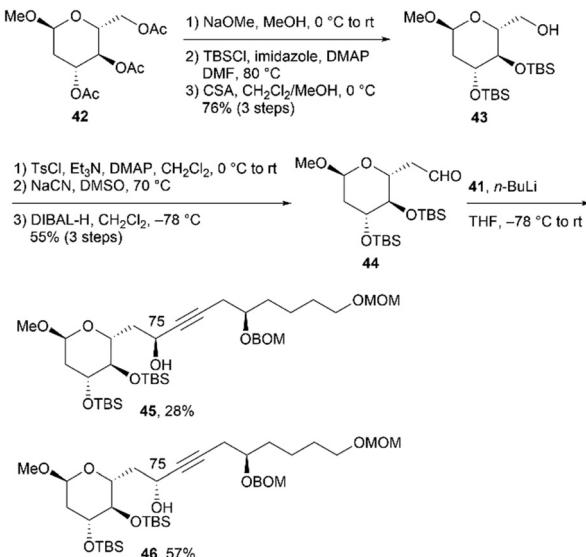


Scheme 8 Stereodivergent synthetic strategy of candidate compounds 4a–4h.



Scheme 9 Synthesis of alkyne 41.

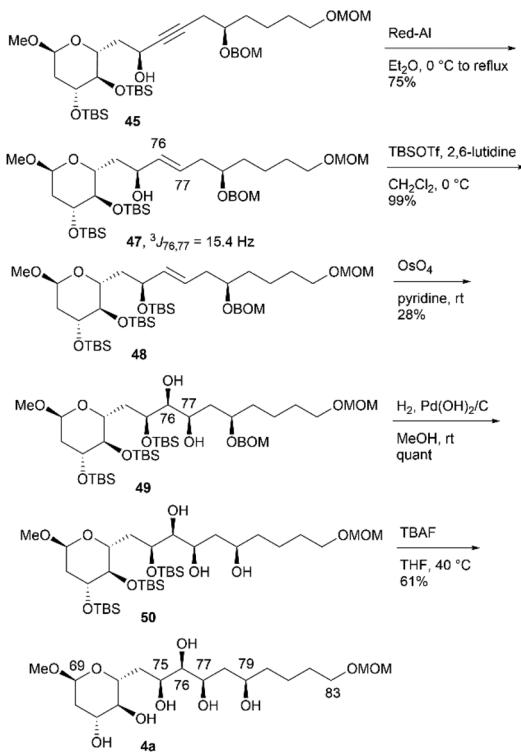
lithium aluminium hydride (LiAlH4) to produce candidate compound **4a**. We then focused on the synthesis of candidate compound **4b**, which is the C79-stereoisomer of **4a** (Scheme 12). Diol **49**, which is the dihydroxylation product in the synthesis of **4a**, was acetylated to afford the diacetate, and the obtained BOM ether was deprotected to yield the corresponding C79-alcohol. The stereochemistry at the C79 position was inverted under Mitsunobu conditions²⁷ using *p*-nitrobenzoic acid/diethyl azodicarboxylate (DEAD)/triphenylphosphine (*Ph*₃P)²⁸ to produce *p*-nitrobenzoate **51**. The simultaneous removal of the acetyl



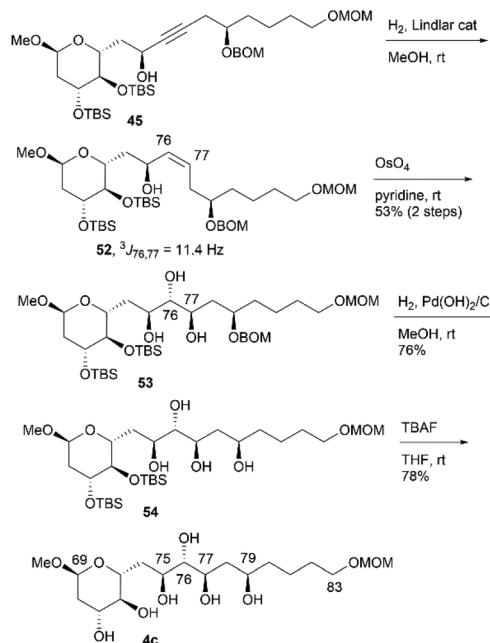
Scheme 10 Synthesis of alcohols 45 and 46.

and benzoyl groups of triester **51** via methanolysis afforded the triol, which was reacted with TBAF to afford candidate compound **4b** in 51% yield in five steps. The ¹H and ¹³C NMR data of **4b** differed from those of **4a**, and thus, the C79-stereo-

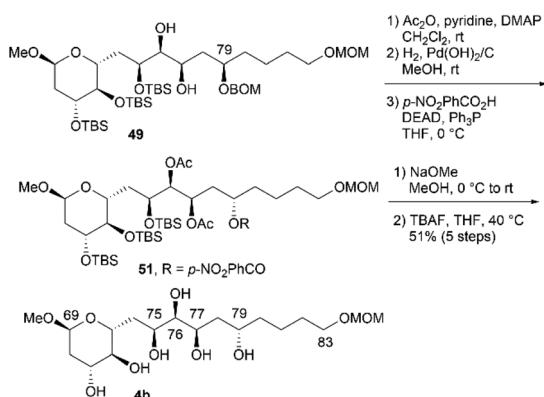




Scheme 11 Synthesis of candidate compound 4a.



Scheme 13 Synthesis of candidate compound 4c.

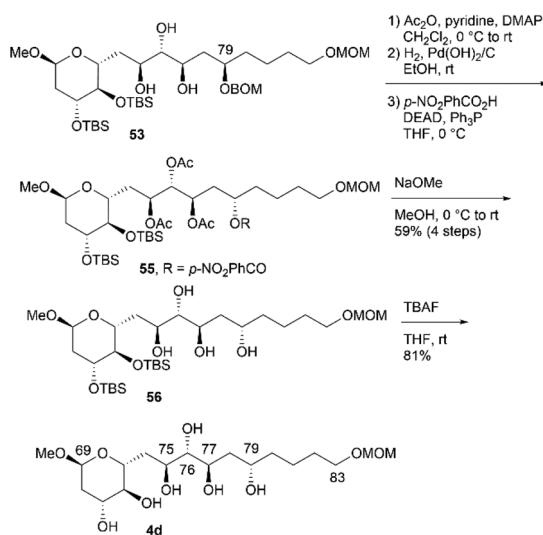


Scheme 12 Synthesis of candidate compound 4b.

chemistry was inverted in the Mitsunobu reaction, leading to the formation of 51.

We then investigated the synthesis of candidate compound 4c, which is the C76-epimer of 4a (Scheme 13). The hydrogenation of alkyne 45 with a Lindlar catalyst produced *cis*-alkene 52, with the geometry confirmed by the coupling constant $^3J_{76,77} = 11.4$ Hz. The osmylation of *cis*-allylic alcohol 52 with OsO₄ in pyridine^{13,16} generated the desired triol 53 in 53% yield in two steps,¹⁴ with the C76,C77-epimer of 53 obtained in 44% yield (dr = 1.2 : 1). The two diastereomers were separated using silica gel column chromatography. The deprotection of BOM ether 53 via hydrogenation provided tetraol 54, which was desilylated using TBAF to afford candidate compound 4c.

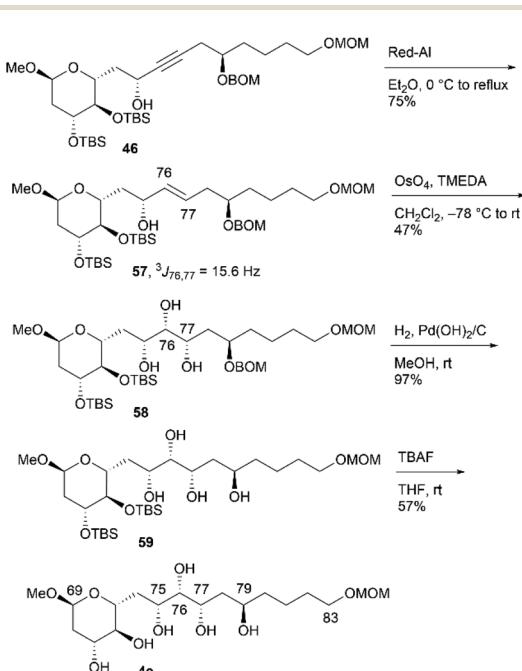
Candidate compound 4d, which is the C79-epimer of 4c, was synthesized *via* a conversion similar to that used in synthesizing 4b (Scheme 14). Thus, acetyl protection of triol 53, removal of the BOM moiety, and Mitsunobu esterification^{27,28} with configurational inversion provided the desired *p*-nitrobenzoate 55. The methanolysis of tetraester 55 afforded tetraol 56 in 59% yield in four steps from 53. The stereochemical inversion from 53 to 55 *via* Mitsunobu esterification was confirmed by the difference in the NMR data of 56 and 54. Finally, bis-TBS ether 56 was deprotected to produce candidate compound 4d.



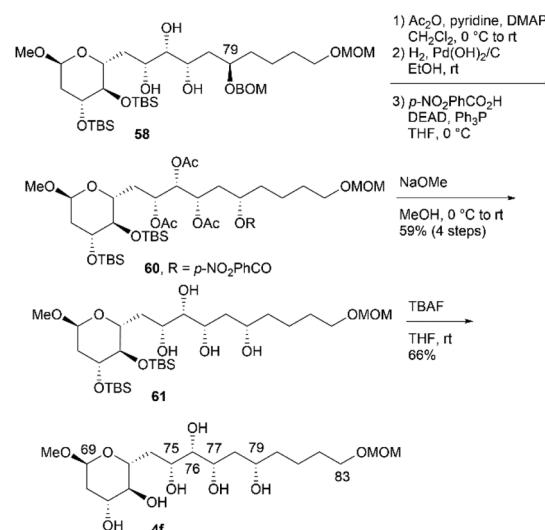
Scheme 14 Synthesis of candidate compound 4d.

Having synthesized candidate compounds **4a–4d** by branching from the common synthetic intermediate **45** with the *75S* absolute configuration, we then attempted to synthesize the other four candidate compounds, **4e–4h**, using propargylic alcohol **46**, with *75R* stereochemistry, as the common intermediate. The transformation of **46** into candidate compound **4e** is shown in Scheme 15. Propargylic alcohol **46** was treated with Red-Al²⁵ to produce *trans*-allylic alcohol **57** in 75% yield, and the *trans*-structure of alkene **57** was elucidated using the coupling constant $^3J_{76,77} = 15.6$ Hz. Allylic alcohol **57** was subjected to the directed dihydroxylation in accordance with the protocol reported by Donohoe *et al.*²⁹ Thus, the reaction of **57** with OsO₄/tetramethylethylenediamine (TMEDA) in CH₂Cl₂ afforded the desired triol **58** in 47% yield¹⁴ and its C76,C77-stereoisomer in 39% yield (dr = 1.2 : 1). The two triols were separated using silica gel column chromatography. The removal of the BOM protecting group *via* hydrogenation and the two TBS groups using TBAF provided candidate compound **4e**. Triol **58** was then converted to candidate compound **4f** *via* stereoinversion at the C79 position (Scheme 16). The acetylation of **58**, removal of the BOM group, and Mitsunobu esterification^{27,28} produced inverted *p*-nitrobenzoate **60**, which underwent methanolysis to afford tetraol **61** in 59% yield over four steps. The configurational inversion under Mitsunobu conditions was verified by comparing the NMR data of **61** and its C79-epimer **59**. The deprotection of bis-TBS ether **61** afforded candidate compound **4f** in 66% yield.

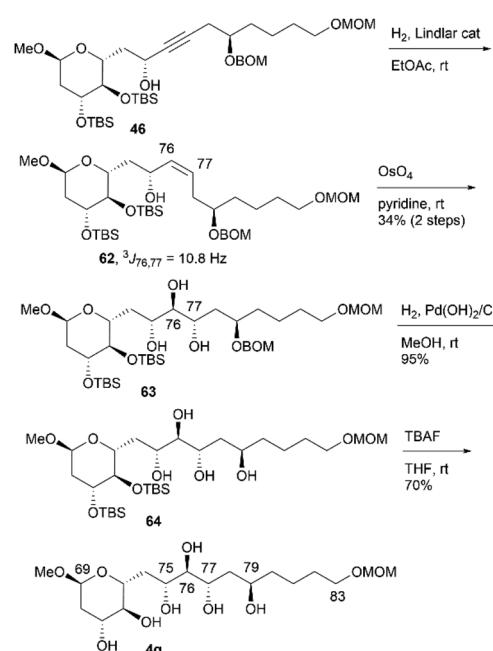
We then attempted to convert **46** to candidate compound **4g**, which is the C76-epimer of **4e** (Scheme 17). Alkyne **46** was hydrogenated using a Lindlar catalyst to yield *cis*-alkene **62**, with the geometry elucidated using the coupling constant $^3J_{76,77} = 10.8$ Hz. Treatment of *cis*-allylic alcohol **62** with OsO₄



Scheme 15 Synthesis of candidate compound **4e**.



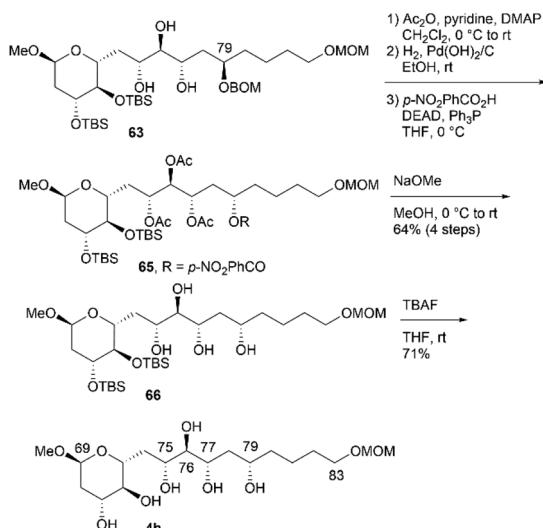
Scheme 16 Synthesis of candidate compound **4f**.



Scheme 17 Synthesis of candidate compound **4g**.

in pyridine^{13,16} afforded the desired triol **63** and its C76,C77-stereoisomer in respective yields of 34% and 26% in two steps (dr = 1.3 : 1).¹⁴ Synthetic **63** and its diastereomer were separated *via* silica gel column chromatography. The removal of the BOM group of **63**, followed by desilylation, yielded candidate compound **4g**. Finally, we investigated the synthesis of candidate compound **4h**, which is the C79-stereoisomer of **4g** (Scheme 18). Triol **63** was converted to *p*-nitrobenzoate **65** *via* the following sequence: (1) acetylation, (2) deprotection of the BOM ether, and (3) Mitsunobu esterification.^{27,28} Methanolysis of tetraester **65** and subsequent deprotection of bis-TBS ether **66** produced candidate compound **4h**. The difference in the





Scheme 18 Synthesis of candidate compound 4h.

NMR data of **66** and **64** confirmed the stereoinversion in the Mitsunobu reaction during the synthesis of **65** using **63**.

With candidate compounds **4a–4h** in hand, we compared the NMR data of these eight synthetic products with those of

natural **1**. Deviations in the ^1H NMR chemical shifts of natural **1** and the synthesized samples **4a–4h** at the C75, C76, C77, and C79 positions were calculated, as shown in Fig. 5.¹⁷ The ^1H NMR characteristics of **4e** and **4f** were more similar to those of the natural product than those of the other candidate compounds. The total magnitudes of the differences in the ^1H NMR chemical shifts at the C75, C76, C77, and C79 positions are shown in Table 2, and those of **4e** and **4f** were 0.21. We then compared the ^{13}C NMR chemical shifts of the eight synthetic candidates with those of the natural product. As shown

Table 2 Sums of the magnitudes of the differences in the NMR chemical shifts (ppm) of natural **1** and the synthetic products **4a–4h** at the C75, C76, C77, and C79 positions

Compound	^1H NMR	^{13}C NMR
4a	0.33	8.0
4b	0.26	5.4
4c	0.35	7.5
4d	0.25	4.9
4e	0.21	2.1
4f	0.21	3.1
4g	0.37	5.2
4h	0.30	3.2

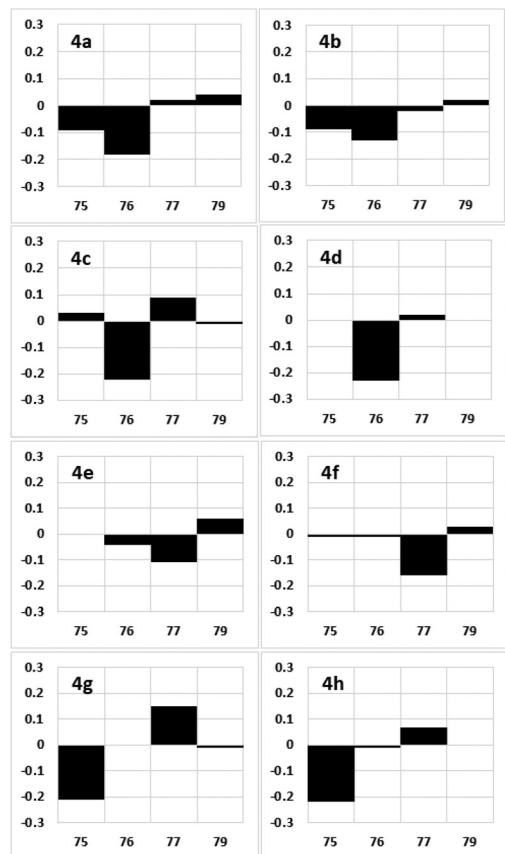


Fig. 5 Deviations in the ^1H NMR chemical shifts of the synthetic products **4a–4h** ($\Delta\delta = \delta_1 - \delta_4$ in ppm) relative to those of **1**. The x and y axes show the carbon number and $\Delta\delta$, respectively.

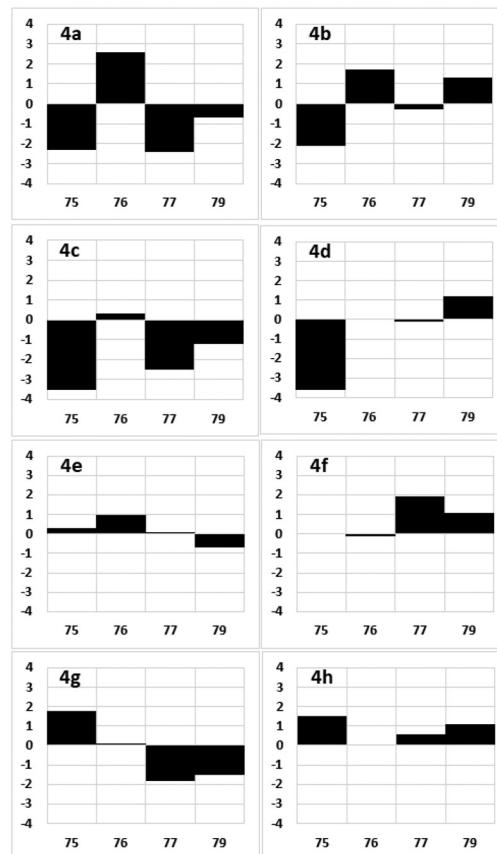
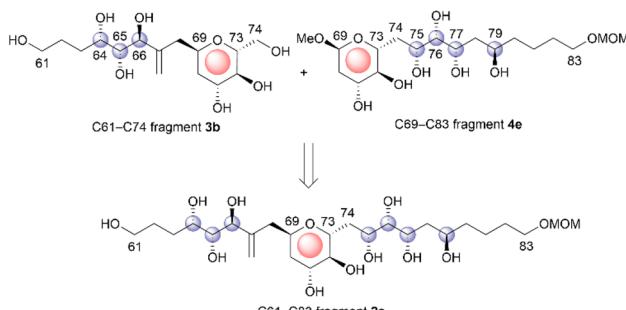


Fig. 6 Deviations in the ^{13}C NMR chemical shifts of the synthetic products **4a–4h** ($\Delta\delta = \delta_1 - \delta_4$ in ppm) relative to those of **1**. The x and y axes show the carbon number and $\Delta\delta$, respectively.





Scheme 19 Relative configuration of fragment 2.

in Fig. 6, candidate compound **4e** exhibited the most similar characteristics of candidate compounds **4a–4h** to those of the natural product.¹⁷ The sum of the magnitudes of the deviations in the ¹³C NMR chemical shifts of **4e** was 2.1, as shown in Table 2, which was the smallest of those calculated for **4a–4h**. Therefore, the relative stereochemistry of fragment **4** is shown in **4e**.

Relative stereochemistry of fragment 2

We determined the relative configurations of fragments **3** and **4** in **1** as those described by **3b** and **4e**, respectively, by synthesizing all candidate compounds for each fragment and comparing their NMR chemical shifts with those of the natural product (Scheme 19). The C69–C73 tetrahydropyran moiety is a common structure in both fragments. Therefore, by connecting the relative stereostructures **3b** and **4e** via the common C69–C73 component, we elucidated the relative stereochemistry of fragment **2** in **1** as that of **2a**.

Conclusions

Fragment **2** is among the 30% of the structure of **1** that has not been previously clarified. To determine the stereochemistry of fragment **2**, we divided it into fragments **3** and **4** and elucidated the relative configurations of these two fragments using a stereodivergent synthetic approach. For fragment **3**, the first synthesis of candidate compounds **3a** and **3b** was performed by coupling aldehyde **14** and alkenyl iodide **15**, with the dihydroxylation of dienes **16** and **17**, wherein the regioselectivities were uncontrolled. To overcome this problem, in the second synthesis of **3a** and **3b**, the *exo*-olefin components of dienes **16** and **17** were temporally and regioselectively protected prior to dihydroxylation. The overall yields of **3a** and **3b** in the second synthesis were improved compared to those in the first synthesis. A comparison of the NMR chemical shifts of these two synthetic products and the natural product revealed that natural product **1** displayed the relative stereochemistry of **3b** in fragment **3**. For fragment **4**, we first analyzed the NMR data of positions C75–C77 of the natural product and selected candidate compounds **4a–4h** from 16 possible diastereomers. The stereodivergent, unified

synthesis of target molecules **4a–4h** was performed *via* the coupling of alkyne **41** and aldehyde **44**, stereoselective reduction of alkynes **45** and **46**, and Mitsunobu inversion as the stereodivergence steps. The NMR chemical shifts of the synthetic products **4a–4h** were compared with those of natural product **1**, which clarified that the relative configuration in this domain of **1** was that of **4e**. The tetrahydropyran component at positions C69–C73 is a common structure in fragments **3** and **4**. Therefore, by overlapping the common tetrahydropyran moieties in **3b** and **4e**, the relative stereochemistry of fragment **2** of **1** is shown as **2a**. The stereodivergent synthetic strategy used in this study can be employed for the configurational determination of other fragments of **1**. Further study for the complete structural clarification of **1** is currently underway in our group.

Author contributions

H. T. conceived and directed the study. K. H., T. Ohashi, and T. Otsu conducted the syntheses and collected the data. All authors analyzed the collected data. H. T. prepared the manuscript.

Conflicts of interest

There are no conflicts to declare.

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