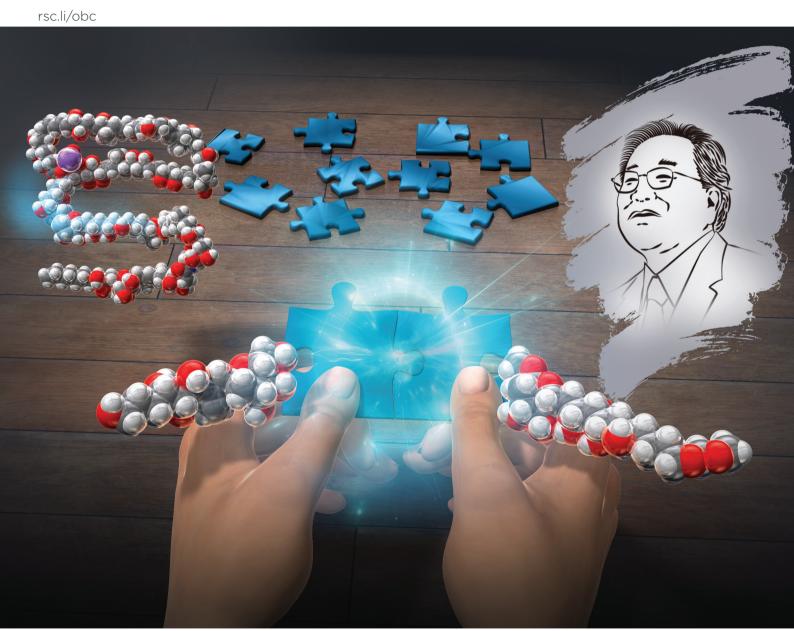
Volume 21 Number 44 28 November 2023 Pages 8785-8946

Organic & Biomolecular Chemistry



ISSN 1477-0520



PAPER

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View Article Online



Cite this: *Org. Biomol. Chem.*, 2023, **21**, 8837

Relative stereochemical determination of the C61–C83 fragment of symbiodinolide using a stereodivergent synthetic approach†‡

Hiroyoshi Takamura, **D** Kosuke Hattori, Takumi Ohashi, Taichi Otsu and Isao Kadota

Structural determination is required in the use of marine natural products to create novel drugs and drug leads in medicinal chemistry. Symbiodinolide, which is a polyol marine natural product with a molecular weight of 2860, increases the intracellular Ca²⁺ concentration and exhibits inhibitory activity against cyclooxygenase-1. Seventy percent of the structure of symbiodinolide has been stereochemically clarified. Herein, we report the elucidation of the relative configuration of the C61-C83 fragment, which is among the remaining thirty percent, using a stereodivergent synthetic strategy. We first assigned the relative configuration of the C61-C74 fragment. Two candidate diastereomers of the C61-C74 fragment were synthesized, and their NMR data were compared with those of the natural product, revealing the relative stereochemistry of this component. We then narrowed down the candidate compounds for the C69-C83 fragment from 16 possible diastereomers by analyzing the NMR data of the natural product, and we thus selected eight candidate diastereomers. Stereodivergent synthesis of the candidates for this fragment and comparison of the NMR data of the natural product and the eight synthetic products resulted in the relative stereostructural clarification of the C69-C83 fragment. These individually determined relative stereochemistries of the C61-C74 and C69-C83 fragments were connected via the common C69-C73 tetrahydropyran moiety of the fragments. Finally, the relative configuration of the C61-C83 fragment of symbiodinolide was determined. The stereodivergent synthetic approach used in this study can be extended to the stereochemical determination of other fragments of symbiodinolide.

Received 5th September 2023, Accepted 27th September 2023 DOI: 10.1039/d3ob01420q

rsc.li/obc

Introduction

Marine natural products (MNPs) are critical in drug discovery and development.¹ Structural determination of MNPs, including their planar structures and stereochemistries, is a fundamental research topic and a requirement in using MNPs to create novel drugs and drug leads in medicinal chemistry. Remarkable recent advancements in computational chemistry enhanced our ability to determine the configurations of structurally complex natural products.² Thus, over the decades, computational modeling of the ¹H and ¹³C nuclear magnetic resonance (NMR) chemical shifts of organic compounds resulted in marked increases in availability, accuracy, and

application. Nevertheless, if the target natural product contains conformationally flexible acyclic motifs and/or remote stereogenic centers, a synergistic combination of spectroscopic analysis and chemical synthesis is effective in unambiguous structural elucidation.³

Symbiodinolide (1, Fig. 1) is a secondary metabolite that was isolated from the cultured dinoflagellate Symbiodinium sp. by Uemura et al.4 Natural product 1 increased the intracellular free Ca²⁺ concentration at 7 nM against IMR-32 human neuroblastoma cells and exhibited cyclooxygenase-1 inhibitory activity (65% inhibition) at 2 µM. Furthermore, 1 ruptured the tissue surface of the flatworm Amphiscolops sp., which is a host organism containing the symbiotic dinoflagellate Symbiodinium sp., at 2.5 µM. In an isolation study, detailed 2D NMR analysis and chemical degradation of 1 revealed its planar structure and partial stereochemistry. The structure of 1 displays a molecular weight of 2860, 61 chiral centers, and an extended carbon skeleton that is highly functionalized with oxygen atoms. The large, complicated chemical structure of 1 hampers its configurational assignment. Therefore, our group investigated the chemical synthesis of each fragment of 1,

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[†]We dedicate this work to the memory of Prof. Daisuke Uemura, who sadly passed away on 13th April 2021.

[‡] Electronic supplementary information (ESI) available. See DOI: https://doi.org/10.1039/d3ob01420g

Fig. 1 Structure of 1.

including the stereoisomers, leading to the stereochemical clarification of the moieties corresponding to 70% of the structure of 1.5 Herein, we report the stereodivergent synthesis of the C61-C74 and C69-C83 fragments (3 and 4), resulting in the determination of the relative stereochemistry of the C61-C83 fragment 2 that is among the remaining 30%.

Results and discussion

Strategy for the stereochemical determination of fragment 2

The relative configuration of the tetrahydropyran domain at positions C69–C73 of 1 was verified via the $^3J_{\rm H,H}$ coupling constants and nuclear Overhauser effect (NOE) observations.4 The large coupling constants (${}^{3}J_{71,72}$ and ${}^{3}J_{72,73}$ = 8.3 Hz) indicate that H-71, H-72, and H-73 are oriented axially. The stereochemistry at the C69 position was elucidated using the NOEs of Ha-68/H-71 and Ha-68/H-73. To determine the stereostructure of fragment 2, we divided it into two fragments, i.e., fragments 3 and 4 (Scheme 1). The C69-C73 tetrahydropyran ring, with a known relative stereochemistry, is a common structure in frag-

Scheme 1 Strategy for the configurational determination of fragment

ments 3 and 4. Therefore, if the relative configurations of 3 and 4 are established individually, these two stereostructures may be connected by this common tetrahydropyran moiety to finally reveal the relative stereochemistry of fragment 2.

Fragment 3

The relative configurations of the three contiguous oxymethine stereogenic centers at the C64, C65, and C66 positions of 1 were determined via ¹H NMR data. ⁴ The coupling constants ${}^{3}J_{64,65}$ (5.5 Hz) and ${}^{3}J_{65,66}$ (7.4 Hz) indicate that the respective relative relationships of C64/C65 and C65/C66 are syn and anti. The relative stereochemistries at the C66 and C69 positions are not correlated, and thus, we propose two candidate diastereomers, 3a and 3b, for fragment 3, as shown in Fig. 2.

To stereochemically elucidate fragment 3 of 1, we planned to synthesize 3a and 3b and compare the NMR data of these synthetic products with those of natural product 1. In our retrosynthetic analysis, target molecules 3a and 3b may be synthesized via regio- and diastereoselective dihydroxylation at the C64 and C65 positions of dienes 5 and 6, respectively (Scheme 2). Alcohols 5 and 6, which are diastereomers at the C66 position, would be formed by coupling alkenyl iodide 7 and aldehyde 8. Aldehyde 8 would be synthesized by introducing a C4 side-chain into tri-O-acetyl-p-glucal (9) as a starting material.

We first examined the stereoselective synthesis of aldehyde 14, which is a branching synthetic intermediate of 3a and 3b (Scheme 3). The reaction of the known tetraacetate 10, which was prepared using 9 in one step, with 2-[(acetoxymethyl)allyl] trimethylsilane (11)8 proceeded in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) as a Lewis acid to yield allylated product 12 as a single diastereomer. 9 The stereo-

Fig. 2 Candidate diastereomers 3a and 3b for fragment 3

Scheme 2 Retrosynthetic analysis of candidate diastereomers 3a and

Scheme 3 Synthesis of alcohols 16 and 17.

chemistry at the C69 position was confirmed by the NOE correlations between H₂-68 and H-71, which indicate that the introduced side-chain moiety is positioned axially. Global deprotection of 12 via methanolysis and subsequent protection of the resulting tetraol using tert-butyldimethylsilyl chloride (TBSCl) afforded silyl ether 13 in 81% yield over two steps. The removal of the two primary tert-butyldimethylsilyl (TBS) groups of 13 with camphorsulfonic acid (CSA) in CH₂Cl₂/methanol (MeOH), followed by selective oxidation of the allylic alcohol component with MnO₂, produced aldehyde 14. The coupling reaction of an alkenyl lithium species, which was generated via lithium-iodine exchange between alkenyl iodide 15 10 and n-butyllithium (n-BuLi) in hexane, 11 with aldehyde 14 generated diols 16 and 17 in respective yields of 57% and 33%. 12 The C64/C65 geometries in 16 and 17 were confirmed as trans based on the coupling constants (${}^{3}J_{64,65} = 15.3$ Hz for **16** and 15.4 Hz for 17).

We then attempted to convert 16 to the first target molecule 3a (Scheme 4). Treatment of allylic alcohol 16 with OsO₄/ N-methylmorpholine N-oxide (NMO) in acetone/H₂O¹³ produced the desired tetraol 18, along with its structural isomer, wherein the exo-olefin moiety was dihydroxylated. 14 This obtained mixture could not be easily separated, and thus, the primary hydroxy group of 18 was protected with TBSCl/imidazole to furnish tetrakis-TBS ether 19, which was purified using silica gel column chromatography. The use of AD-mix-α or AD-mix-β¹⁵ in dihydroxylation did not improve the regioselectivity or chemical yield. The four TBS protective groups of 19 were removed using HCl in MeOH to quantitatively obtain the first target molecule, 3a. The second target molecule, 3b, was synthesized using 17 via a transformation similar to that employed in synthesizing 3a using 16 (Scheme 5). Thus, dihydroxylation of the disubstituted alkene moiety of 17 was conducted using OsO₄/NMO¹³ to afford the desired tetraol 20 ¹⁴ in 24% yield, along with recovered 17 in 33% yield. When the reaction time was extended, dihydroxylation of the exo-olefin

Scheme 4 Synthesis of candidate compound 3a.

Scheme 5 Synthesis of candidate compound 3b.

component of 20 was observed. Finally, tris-TBS ether 20 was deprotected using HCl in MeOH to furnish the second target molecule, 3b. Although we obtained candidate compounds 3a and 3b, the regioselectivities in the dihydroxylation of 16 and 17 were uncontrolled, and the chemical yields of 18 and 20 were low. Therefore, alternative synthetic routes to 3a and 3b were investigated.

In an alternative synthetic route, we temporarily and regioselectively protected the *exo*-olefin moieties of dienes **16** and **17** prior to dihydroxylation. The second synthesis of **3a** is shown in Scheme 6. After protecting diol **16** using TBSCl, regioselective hydroboration of the obtained diene was examined by changing the borane reagent. The use of BH₃·SMe₂ (SMe₂ = dimethylsulfide) or thexylborane was ineffective in the regioselective hydroboration of the *exo*-olefin moiety. When we treated the obtained diene with 9-borabicyclo[3.3.1]nonane (9-BBN) and performed an oxidative work-up, the desired alcohol **21** was obtained in 76% yield over two steps. Tosylation of **21**, iodination of the tosylate with NaI, and treatment of the iodoalkane with thiophenol (PhSH)/1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) afforded sulfide **22** in 68% yield

Scheme 6 Improved synthesis of candidate compound 3a.

over three steps. Diastereoselective dihydroxylation of alkene 22 with a stoichiometric amount of OsO₄ in pyridine^{13,16} afforded the desired diol 23 and its C64,C65-epimer (dr = 6.2:1). The alkene component of 24 was regenerated via the following sequence: (1) *meta*-chloroperoxybenzoic (mCPBA) oxidation of sulfide 23 to the corresponding sulfoxide and (2) elimination of the sulfoxide with NaHCO3 in xylene at reflux. Diol 24 and its C64,C65-epimer were separated, and 24 was obtained in 56% yield in three steps from 22. Finally, pentakis-TBS ether 24 was deprotected to afford 3a. The respective total number of steps and overall yield of 3a from 16 in the second synthesis, as shown in Scheme 6, were 9 and 29%. Although the number of steps was increased compared to that in the first synthesis (3 steps from 16, as shown in Scheme 4), the overall yield was improved compared to that in the first synthesis (23% from 16, as shown in Scheme 4). We then investigated the second synthesis of 3b using a conversion similar to that shown in Scheme 6 (Scheme 7). TBS protection of diol 17 and subsequent regioselective hydroboration with 9-BBN afforded alcohol 25 in 75% yield over two steps. Tosylation, nucleophilic iodination, and thioetherification afforded sulfide 26. Alkene 26 reacted with OsO4 in pyridine^{13,16} to produce diol 27 (dr = 4.0:1), which was then converted to alkene 28 via oxidation with mCPBA and elimination of the resulting sulfoxide. Global removal of the TBS protecting groups from 28 provided 3b. As in the case of 3a, although the transformation from 17 to 3b in the second synthesis (9 steps, Scheme 7) was lengthy, the overall yield of the

Scheme 7 Improved synthesis of candidate compound 3b.

second synthesis (29%) was higher than that of the first synthesis (24%, Scheme 5).

After obtaining candidate compounds 3a and 3b, we analyzed their 2D NMR spectra. The differences in the ¹H NMR chemical shifts representing the C66, C68, and C69 positions of 1 and the synthetic products 3a and 3b are graphically illustrated in Fig. 3.¹⁷ The chemical shift deviations of 3b were smaller than those of 3a, as the sums of the magnitudes of the differences in the ¹H NMR chemical shifts representing the C66, C68, and C69 positions were 0.23 (3a) and 0.15 (3b). Therefore, the relative configuration of fragment 3 of 1 is that of 3b.

Fragment 4

We then focused on the stereochemical clarification of fragment 4. This fragment contains four unidentified chiral

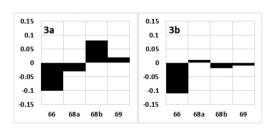


Fig. 3 Deviations in the ¹H NMR chemical shifts of the synthetic products **3a** and **3b** ($\Delta \delta = \delta_1 - \delta_3$ in ppm) relative to those of **1**. The x and y axes show the carbon number and $\Delta\delta$, respectively.

centers at positions C75, C76, C77, and C79, and thus, 16 diastereomers are possible. Prior to commencing the synthetic approach for structural determination, we analyzed the NMR data of natural product 1 to identify candidate compounds. The NMR chemical shifts of positions C75–C77 of 1, which are reported in an isolation study, 4 are shown in Table 1. A detailed comparison of these chemical shifts with data reported in the Kishi universal NMR database for 1,2,3-triols 18 clarified the relative relationships of C75–C76/C76–C77 as *syn/syn* or *anti/anti*. Therefore, eight hexaols, 4a–4h, as shown in Fig. 4, were used as candidate compounds for fragment 4.

The stereodivergent synthetic strategy⁶ for candidate compounds **4a-4h** is outlined in Scheme 8. The reaction of aldehyde **29** and alkyne **30** would give two propargylic alcohols, **31** and **32**, which are stereoisomeric at the C75 position (Scheme 8a). This coupling reaction is the first step in stereodivergence. Respective *trans*- and *cis*-alkenes **33** and **34** could be synthesized *via* the stereoselective reduction of alkyne **31**, which is the second stereodivergence step (Scheme 8b). The dihydroxylation of *trans*-alkene **33** would afford hexaol **4a**, and

Table 1 NMR chemical shifts of 1 in CD₃OD^a

Position	¹ H NMR	¹³ C NMR
75	3.89	69.6
76	3.18	78 . 5
77	3.95	70.5

^a Data reported in ref. 4. Chemical shifts are reported in ppm with reference to the internal residual solvent (¹H NMR: 3.30 ppm, ¹³C NMR: 49.0 ppm).

Fig. 4 Candidate diastereomers 4a-4h for fragment 4

hexaol **4b**, which is the C79-epimer of **4a**, could be synthesized *via* stereoinversion at the C79 position, which is the third stereodivergence step, after the dihydroxylation of **33**. Similarly, hexaols **4c** and **4d**, which are the respective C76-stereoisomers of **4a** and **4b**, could be produced using *cis*-alkene **34**. Furthermore, the other four candidates, **4e-4h**, could be prepared using propargylic alcohol **32**, with the respective *trans*- and *cis*-alkenes **35** and **36** as the key synthetic intermediates, *via* a similar transformation. We then commenced the stereodivergent synthesis of candidate compounds **4a-4h**.

First, we examined the synthesis of alkyne 41, which corresponds to the C76-C83 fragment (Scheme 9). The synthesis commenced with (S)-2-bromobutane-1,4-diol (37), which was prepared in two steps using 1-aspartic acid. 19 Compound 37 was treated with K₂CO₃ in CH₂Cl₂ to yield a known epoxy alcohol, 19 which underwent oxidation with 2,2,6,6-tetramethyl-1-piperidinyloxyl (TEMPO)/PhI(OAc)₂ ²⁰ and subsequent Wittig olefination with Ph₃P=CHCO₂Me in one-pot to afford unsaturated ester 38. The alkene moiety of 38 was chemoselectively hydrogenated with a Pd/C-ethylenediamine complex catalyst in tetrahydrofuran (THF)²¹ to provide the saturated ester, with the epoxide structure retained. The resulting epoxide was reacted with trimethylsilylacetylene/n-BuLi/BF₃·OEt₂ ²² (OEt₂ = diethyl ether) to produce alcohol 39. The hydroxy group of 39 was protected as the benzyloxymethyl (BOM) ether, and the ester group was reduced using LiAlH4 to yield alcohol 40. Protection of 40 with methoxymethyl chloride/N,N-diisopropylethylamine/tetra-n-butylammonium iodide (MOMCl/i-Pr2NEt/ TBAI), followed by removal of the trimethylsilyl (TMS) moiety with K₂CO₃ in MeOH, afforded alkyne 41 in 28% yield in seven steps from 37.

With coupling precursor **41** in hand, we then synthesized aldehyde **44**, which is the coupling partner of **41** (Scheme 10). Deacetylation of the known methyl glucoside **42**, 23 TBS protection of the resulting triol, and selective deprotection of the primary TBS ether provided alcohol **43** in 76% yield over three steps. One-carbon elongation from **43** was conducted *via* a combination of tosylation and cyanation to yield a nitrile, which was reduced with diisobutylaluminum hydride (DIBAL-H) to afford aldehyde **44**. Deprotonation of alkyne **41** using *n*-BuLi and subsequent addition of the resulting anion to aldehyde **44** afforded the desired alcohols **45** and **46** in respective yields of 28% and 57%. 24

We then converted **45** to **4a** (Scheme 11). Propargylic alcohol **45** was reduced using Red-Al²⁵ to yield *trans*-allylic alcohol **47**. The *trans*-geometry of **47** was determined based on the observed coupling constant (${}^3J_{76,77} = 15.4$ Hz). After protection of **47** with *tert*-butyldimethylsilyl trifluoromethanesulfonate (TBSOTf)/2,6-lutidine, *trans*-alkene **48**, with an allylic siloxy group, was dihydroxylated with OsO₄ in pyridine^{13,16} to afford the desired diol **49** in 28% yield and its C76,C77-epimer in 55% yield (dr = 1:2.0). These products could be separated *via* silica gel column chromatography. ^{14,26} The hydrogenation of BOM ether **49** with Pd(OH)₂/C in MeOH generated triol **50**. The three TBS groups of **50** were removed using tetra-*n*-buty-

Scheme 8 Stereodivergent synthetic strategy of candidate compounds 4a-4h.

Scheme 9 Synthesis of alkyne 41.

lammonium fluoride (TBAF) to produce candidate compound 4a. We then focused on the synthesis of candidate compound 4b, which is the C79-stereoisomer of 4a (Scheme 12). Diol 49, which is the dihydroxylation product in the synthesis of 4a, was acetylated to afford the diacetate, and the obtained BOM ether was deprotected to yield the corresponding C79-alcohol. The stereochemistry at the C79 position was inverted under Mitsunobu conditions²⁷ using *p*-nitrobenzoic acid/diethyl azodicarboxylate (DEAD)/triphenylphosphine (Ph₃P)²⁸ to produce p-nitrobenzoate 51. The simultaneous removal of the acetyl

Scheme 10 Synthesis of alcohols 45 and 46.

and benzoyl groups of triester 51 via methanolysis afforded the triol, which was reacted with TBAF to afford candidate compound 4b in 51% yield in five steps. The ¹H and ¹³C NMR data of 4b differed from those of 4a, and thus, the C79-stereo-

Scheme 11 Synthesis of candidate compound 4a.

Scheme 12 Synthesis of candidate compound 4b.

chemistry was inverted in the Mitsunobu reaction, leading to the formation of 51.

We then investigated the synthesis of candidate compound **4c**, which is the C76-epimer of **4a** (Scheme 13). The hydrogenation of alkyne **45** with a Lindlar catalyst produced *cis*-alkene **52**, with the geometry confirmed by the coupling constant ${}^3J_{76,77}=11.4$ Hz. The osmylation of *cis*-allylic alcohol **52** with OsO₄ in pyridine 13,16 generated the desired triol **53** in **53**% yield in two steps, 14 with the C76,C77-epimer of **53** obtained in 44% yield (dr = 1.2:1). The two diastereomers were separated using silica gel column chromatography. The deprotection of BOM ether **53** *via* hydrogenation provided tetraol **54**, which was desilylated using TBAF to afford candidate compound **4c**.

Scheme 13 Synthesis of candidate compound 4c.

Candidate compound **4d**, which is the C79-epimer of **4c**, was synthesized *via* a conversion similar to that used in synthesizing **4b** (Scheme **14**). Thus, acetyl protection of triol **53**, which is a synthetic intermediate of **4c**, removal of the BOM moiety, and Mitsunobu esterification^{27,28} with configurational inversion provided the desired *p*-nitrobenzoate **55**. The methanolysis of tetraester **55** afforded tetraol **56** in 59% yield in four steps from **53**. The stereochemical inversion from **53** to **55** *via* Mitsunobu esterification was confirmed by the difference in the NMR data of **56** and **54**. Finally, bis-TBS ether **56** was deprotected to produce candidate compound **4d**.

Scheme 14 Synthesis of candidate compound 4d.

Having synthesized candidate compounds 4a-4d by branching from the common synthetic intermediate 45 with the 75S absolute configuration, we then attempted to synthesize the other four candidate compounds, 4e-4h, using propargylic alcohol 46, with 75R stereochemistry, as the common intermediate. The transformation of 46 into candidate compound 4e is shown in Scheme 15. Propargylic alcohol 46 was treated with Red-Al²⁵ to produce trans-allylic alcohol 57 in 75% yield, and the trans-structure of alkene 57 was elucidated using the coupling constant ${}^{3}J_{76,77}$ = 15.6 Hz. Allylic alcohol 57 was subjected to the directed dihydroxylation in accordance with the protocol reported by Donohoe et al.²⁹ Thus, the reaction of 57 with OsO₄/tetramethylethylenediamine (TMEDA) in CH₂Cl₂ afforded the desired triol 58 in 47% yield14 and its C76,C77stereoisomer in 39% yield (dr = 1.2:1). The two triols were separated using silica gel column chromatography. The removal of the BOM protecting group via hydrogenation and the two TBS groups using TBAF provided candidate compound 4e. Triol 58 was then converted to candidate compound 4f via stereoinversion at the C79 position (Scheme 16). The acetylation of 58, removal of the BOM group, and Mitsunobu esterification 27,28 produced inverted p-nitrobenzoate 60, which underwent methanolysis to afford tetraol 61 in 59% yield over four steps. The configurational inversion under Mitsunobu conditions was verified by comparing the NMR data of 61 and its C79-epimer 59. The deprotection of bis-TBS ether 61 afforded candidate compound 4f in 66% yield.

We then attempted to convert 46 to candidate compound 4g, which is the C76-epimer of 4e (Scheme 17). Alkyne 46 was hydrogenated using a Lindlar catalyst to yield cis-alkene 62, with the geometry elucidated using the coupling constant $^{3}J_{76,77}$ = 10.8 Hz. Treatment of *cis*-allylic alcohol **62** with OsO₄

Scheme 15 Synthesis of candidate compound 4e

Scheme 16 Synthesis of candidate compound 4f

Scheme 17 Synthesis of candidate compound 4g

in pyridine^{13,16} afforded the desired triol 63 and its C76,C77stereoisomer in respective yields of 34% and 26% in two steps (dr = 1.3:1). Synthetic 63 and its diastereomer were separated via silica gel column chromatography. The removal of the BOM group of 63, followed by desilylation, yielded candidate compound 4g. Finally, we investigated the synthesis of candidate compound 4h, which is the C79-stereoisomer of 4g (Scheme 18). Triol 63 was converted to p-nitrobenzoate 65 via the following sequence: (1) acetylation, (2) deprotection of the BOM ether, and (3) Mitsunobu esterification. 27,28 Methanolysis of tetraester 65 and subsequent deprotection of bis-TBS ether 66 produced candidate compound 4h. The difference in the

Scheme 18 Synthesis of candidate compound 4h.

NMR data of **66** and **64** confirmed the stereoinversion in the Mitsunobu reaction during the synthesis of **65** using **63**.

With candidate compounds 4a-4h in hand, we compared the NMR data of these eight synthetic products with those of

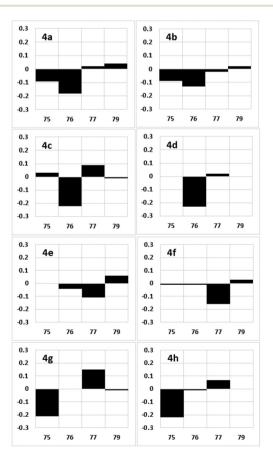


Fig. 5 Deviations in the ¹H NMR chemical shifts of the synthetic products **4a–4h** ($\Delta \delta = \delta_1 - \delta_4$ in ppm) relative to those of **1**. The x and y axes show the carbon number and $\Delta \delta_r$ respectively.

natural **1**. Deviations in the ¹H NMR chemical shifts of natural **1** and the synthesized samples **4a–4h** at the C75, C76, C77, and C79 positions were calculated, as shown in Fig. 5.¹⁷ The ¹H NMR characteristics of **4e** and **4f** were more similar to those of the natural product than those of the other candidate compounds. The total magnitudes of the differences in the ¹H NMR chemical shifts at the C75, C76, C77, and C79 positions are shown in Table 2, and those of **4e** and **4f** were 0.21. We then compared the ¹³C NMR chemical shifts of the eight synthetic candidates with those of the natural product. As shown

Table 2 Sums of the magnitudes of the differences in the NMR chemical shifts (ppm) of natural 1 and the synthetic products 4a-4h at the C75, C76, C77, and C79 positions

Compound	¹ H NMR	$^{13}\mathrm{C}\ \mathrm{NMR}$
4a	0.33	8.0
4b	0.26	5.4
4c	0.35	7.5
4d	0.25	4.9
4e	0.21	2.1
4e 4f	0.21	3.1
4g 4h	0.37	5.2
4h	0.30	3.2

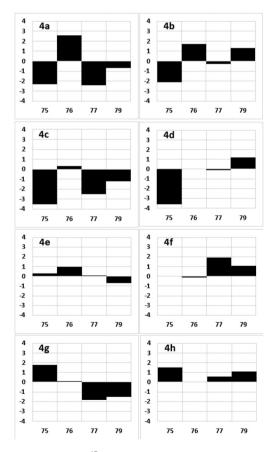


Fig. 6 Deviations in the ¹³C NMR chemical shifts of the synthetic products **4a**–**4h** ($\Delta\delta = \delta_1 - \delta_4$ in ppm) relative to those of **1**. The *x* and *y* axes show the carbon number and $\Delta\delta$, respectively.

Scheme 19 Relative configuration of fragment 2.

in Fig. 6, candidate compound 4e exhibited the most similar characteristics of candidate compounds 4a-4h to those of the natural product.¹⁷ The sum of the magnitudes of the deviations in the ¹³C NMR chemical shifts of 4e was 2.1, as shown in Table 2, which was the smallest of those calculated for 4a-4h. Therefore, the relative stereochemistry of fragment 4 is shown in 4e.

Relative stereochemistry of fragment 2

We determined the relative configurations of fragments 3 and 4 in 1 as those described by 3b and 4e, respectively, by synthesizing all candidate compounds for each fragment and comparing their NMR chemical shifts with those of the natural product (Scheme 19). The C69-C73 tetrahydropyran moiety is a common structure in both fragments. Therefore, by connecting the relative stereostructures 3b and 4e via the common C69-C73 component, we elucidated the relative stereochemistry of fragment 2 in 1 as that of 2a.

Conclusions

Fragment 2 is among the 30% of the structure of 1 that has not been previously clarified. To determine the stereochemistry of fragment 2, we divided it into fragments 3 and 4 and elucidated the relative configurations of these two fragments using a stereodivergent synthetic approach. For fragment 3, the first synthesis of candidate compounds 3a and 3b was performed by coupling aldehyde 14 and alkenyl iodide 15, with the dihydroxylation of dienes 16 and 17, wherein the regioselectivities were uncontrolled. To overcome this problem, in the second synthesis of 3a and 3b, the exo-olefin components of dienes 16 and 17 were temporally and regioselectively protected prior to dihydroxylation. The overall yields of 3a and 3b in the second synthesis were improved compared to those in the first synthesis. A comparison of the NMR chemical shifts of these two synthetic products and the natural product revealed that natural product 1 displayed the relative stereochemistry of 3b in fragment 3. For fragment 4, we first analyzed the NMR data of positions C75-C77 of the natural product and selected candidate compounds 4a-4h from 16 possible diastereomers. The stereodivergent, unified

synthesis of target molecules 4a-4h was performed via the coupling of alkyne 41 and aldehyde 44, stereoselective reduction of alkynes 45 and 46, and Mitsunobu inversion as the stereodivergence steps. The NMR chemical shifts of the synthetic products 4a-4h were compared with those of natural product 1, which clarified that the relative configuration in this domain of 1 was that of 4e. The tetrahydropyran component at positions C69-C73 is a common structure in fragments 3 and 4. Therefore, by overlapping the common tetrahydropyran moieties in 3b and 4e, the relative stereochemistry of fragment 2 of 1 is shown as 2a. The stereodivergent synthetic strategy used in this study can be employed for the configurational determination of other fragments of 1. Further study for the complete structural clarification of 1 is currently underway in our group.

Author contributions

H. T. conceived and directed the study. K. H., T. Ohashi, and T. Otsu conducted the syntheses and collected the data. All authors analyzed the collected data. H. T. prepared the manuscript.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We are grateful to Mr Soichiro Inoue (Okayama University, Okayama, Japan) for experimental support. We appreciate the assistance of the Division of Instrumental Analysis at Okayama University with NMR spectroscopy and high-resolution mass spectrometry. This study was supported by a Japan Society for the Promotion of Science KAKENHI grant (grant number: JP21H01938).

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